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Third International Cloud Condensation Nuclei Workshop

Reno Workshop Objectives, Accomplishments, Instrument Descriptions and Review Papers

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The Desert Research Institute University of Nevada System Reno, Nevada

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DEDICATION

It is with great sorrow that we recall the untimely loss of our good friend and colleague, Professor Roger Lincoln Steele.

A man of great talent and undeniable style, Professor Steele contributed extensively to Workshops such as this one, from his scientific involvement in the First International Workshop at Lannemezan, France in 1967, to his responsibilities as Chairman of the Organizing/ Steering Committee during the Second Workshop held in Ft. Collins, Colorado in 1970.

Professor Steele will remain in our minds as an individual of uncommon strength and unique character, and it is to his memory that we dedicate these Proceedings.



ROGER LINCOLN STEELE February 12, 1924 – November 10, 1980

CCN WORKSHOP PARTICIPANTS

Standing, left to right:

A. Gagin, H. Nuzitsa, U. Katz, J. Dea, W. Hoppel, J. Hudson, J. Jiusto, H. Gerber, R. Borys, R. Serpolay, E. Hindman, V. Keller, M. Politovitch, T. Ohtake, W. Mach, R. Hucek, R. McKenzie, W. Kocmond, F. Rogers, T. Wojciechowski, M. Kitchen, J. Fitzgerald, G. Ayers, G. Lala

Kneeling, left to right:

D. Alofs, W. Megaw, R. Leaitch, T. Mee, J. Anderson, D. Rogers, R. Ruskin, M. Trueblood



The Third International Cloud Condensation Nuclei Workshop was held at the Desert Research Institute, Reno, Nevada, October 6-17, 1980. The goals of the Workshop were to intercompare CCN measurement technology and to perform a limited number of experiments of fundamental scientific interest. A total of 39 scientists representing 20 institutions were in attendance. Twenty-five instruments were tested, including size characterization devices and two Aitken counters. The test acrosols were supplied to the instruments by an on-line generation system, thereby eliminating the need for storage bags. Some of the main conclusions reached during the two-week Workshop were as follows:

(1) Test aerosols of pure soluble salts, both monodisperse and polydisperse, can be provided with stability in output concentration to about $\pm 3\%$ per hour;

(2) Of nine static diffusion chambers (SDC), the five best units (averaged) agreed to within 20% of the NRL mobility analyzer and to within 10% at 1% supersaturation;

(3) Four of the five continuous flow diffusion (CFD) chambers agreed with each other to within about 15% at 0.7% supersaturation and about 20% at 0.3% supersaturation;

(4) The best CFD's and SDC's agreed to within about 15%;

(5) Two of four isothermal haze chambers agreed with each other to within about 40%;

(6) Analysis of the results showed that most instruments' estimation of the CCN spectral slope, k, and the known dry aerosol size distribution slope, β , confirmed the theoretical relationship k= 2/3 β .

THIRD INTERNATIONAL CLOUD CONDENSATION NUCLEI WORKSHOP

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In 1967 the First International Workshop on Condensation and Ice Nuclei was convened under the guidance of Prof. Henri Dessens at his Centre de Recherches Atmospheriques in Lannemezan, France. The Workshop provided a unique opportunity for scientists from all parts of the world to exchange ideas on measurement techniques and to compare instruments for observing ice and condensation nuclei.

A number of important lessons were learned from the first Workshop; principal among them was the need for improved methods of particle generation and characterization, and also the requirement to provide a continuous and stable source of nuclei to the instruments. Solutions to these problems were provided by the combined efforts of several scientists, and at the Second International Workshop, held at Colorado State University in Ft. Collins, Colorado, in August 1970, substantial progress was made in generating, storing and delivering aerosols in the desired size ranges to the more than 25 CCN and Ice Nucleus measuring devices.

In retrospect, the Ft. Collins Workshop was an enormous success, providing researchers an opportunity to learn about the strengths and limitations of the then current instruments used for measuring atmospheric nuclei. At the conclusion of the Workshop, it was determined that, since the needs of CCN and IN counters are so vastly different in terms of aersol concentrations, separate workshops would be preferable in the future for comparing these two major types of instrumentation. Indeed, this has been the case and, in 1975, an independent Ice Nucleus Workshop was held at the University of Wyoming in Laramie, Wyoming. The objective of that Workshop was to measure and compare methods of ice nucleus measurement; additional useful experiences were gained in methods of generating and delivering ice-forming nuclei to the instrument. Ten years were to pass before the present Workshop was convened at the Desert Research Institute (DRI) in Reno, Nevada. In the intervening period, progress has been made in instrument design technology, and several new measurement concepts have emerged. Notable among them have been the various types of continuous flow diffusion (CFD) chambers, the isothermal "haze" chambers which are able to operate at extremely low supersaturations, the highly efficient CFD spectrometers and the automated static diffusion chambers. Detailed discussions of each of these types of CCN instruments are provided within these proceedings.

The advances in CCN counter technology have been more than matched by the growth and improvements in aerosol size characterization technology. As just one example, electrical aerosol analyzers are now an off-the-shelf item in use in laboratories throughout the world, whereas at the time of the Ft. Collins Workshop, only a few groups possessed prototype devices.

In view of the many recent innovations in CCN counters, as well as the vast improvements that have been made in particle characterization technology, it was the unanimous decision of the Ad Hoc Commission of the Nucleation Committee, International Commission on Cloud Physics (ICCP), that a high quality instrument comparison workshop be conducted. With the support of the National Science Foundation and the National Aeronautics and Space Administration, such a Workshop was conducted at DRI, Reno, Nevada, October 6-17, 1980, with W. Kocmond and J. Jiusto serving as sponsor Principal Investigators. It is the purpose of these proceedings to document the results and activities of the two-week Reno Workshop.

SECTION II. OBJECTIVES

In the broadest sense, the Third International CCN Workshop had two principal objectives:

- To intercompare CCN measurement technology over a wide range of instrument operating conditions, and
- (2) to perform one or perhaps two fundamental cloud physics experiments of interest to the scientific community.

Because of the strong desire to check each instrument's detection limits and response characteristics under varying operating conditions, most experiments were directed toward this goal. Deliberations by a Workshop Steering Committee (Gagin, Jiusto, Kassner, Kocmond, Megaw, Ruskin and Radke) in conjunction with DRI representatives produced several recommendations with regard to the scientific details of the experiments that were performed. The principal recommendations were:

- (1) that the composition of the test aerosols should include NaCl, AgI and $(NH_4)_2SO_4$;
- (2) aerosol concentrations should be in the range of a few hundred to a few thousand per $\rm cm^3$ active at 0.1 to 1.0% supersaturation;
- (3) ambient air monitoring should be a part of each day's runs;
- (4) on-line continuous generation of aerosols with high stability (+2%) would be preferred over use of a storage bag to supply aerosols to the instruments;

- (5) aerosol samples should be forced through a duct, rather than drawn by suction in order to avoid possible contamination from negative line pressures; and
- (6) both polydisperse and monodisperse aerosols should be generated for the intended instrument checks.

In addition to these recommendations, the consensus held that a few steep sloped aerosol distributions should be produced for tests of instrument responsiveness and also one or two experiments of very high (> 2000 cc⁻¹ active at 1.0%) and very low (< 200 cc⁻¹ active at 1% supersaturation) aerosol concentrations should be performed to test each instrument's limits of detection.

With such a broad range of instrument calibration checks, there was less opportunity to conduct specific experiments of fundamental scientific interest. A number of suggestions were considered which included e.g. tests of Kohler theory for a range of particle sizes; comparisons of experiment and theory regarding condensation on particles of limited wettability; checks of temperature dependence of CCN activity in terms of S_c ; and attempts to identify conditions under which particle multiplication may occur.

Partial answers to several of these questions were obtained from the results of experiments performed during the course of the Workshop. Details regarding the findings can be found in the individual summary papers.

Welcoming Address

by

H.W. Georgii Institut fur Meteorologie und Geophysik Frankfurt, Germany

Ladies and Gentlemen:

It is a privilege for me to welcome you to the Third International Workshop sponsored by the Committee on Nucleation of the International Commission on Cloud Physics and supported by the National Science Foundation and NASA. We are here as the guests of the Desert Research Institute of the University of Nevada, and my special thanks and gratitude go to the host, Professor Warren Kocmond and his associates who provided these beautiful facilities for us. To all of you, I want to express our gratitude. This workshop is mainly devoted to the measurement of cloud condensation nuclei. As a matter of fact, the Second International Workshop, held during August 1970 at Fort Collins had to deal, among other goals, with the following subjects:

(1) Survey of the state-of-the-art in the field of measurement of CCN, (2) compare the operating characteristics of various types of CCN counters, and (3) humidity activation characteristics in various types of CCN counters. While the main effort of the 1970 Workshop, and also of the 1975 Workshop held at the University of Wyoming in Laramie, was devoted to the measurement of ice nuclei, six CCN counters were also tested at Ft. Collins. In the meantime, during the last decade, the importance of CCN for cloud and precipitation physics became more evident. We therefore found it appropriate to devote this Workshop predominantly to the measurement of cloud condensation nuclei.

We became aware in the meantime that only aerosol particles activated at a supersaturation of about 0.5% are of interest to the cloud physicist. We have improved our knowledge on the distribution of CCN in the troposphere and its relation to the atmospheric aerosol in general. This is important, since concentration and composition of cloud nuclei influence directly the average size of cloud drops, the number concentration of cloud drops, the colloidal stability of clouds and the optical density of clouds. The importance of sulfate containing particles as potential cloud nuclei was emphasized during the last ten years by many investigators, and it was confirmed that over oceans the sea-salt particles are only a small fraction of the total maritime cloud condensation nuclei. It could be shown that also over the oceans a large fraction of cloud nuclei is composed of ammonium sulfate or sulfuric acid.

Not long ago, it was assumed that Aitken nuclei are of little importance as cloud nuclei. According to more recent observations, we have to assume that a certain fraction of Aitken nuclei is activated as cloud nuclei. However, the results are still somewhat controversial. A large concen-

tration of Aitken particles does not always lead to a large number of CCN. While measurements in the plumes of large cities in the United States showed an increase of CCN downwind of pollution sources, this was not observed in Israel. In general, it can be said that the concentrations of cloud nuclei over the continent range from 100 to 1000/cc while over the oceans they range from some tens to a few hundred/cc. These values are in good agreement with the drop concentrations in continental or maritime clouds. From this point of view, it appears that the cloud nuclei counters used in these investigations detect the right fraction of the atmospheric aerosol activated in the process of cloud formation. One problem, still unsolved, is the possible long-term trend of the cloud nuclei concentration on a global scale. One major source of cloud nuclei is probably the emission of reactive gases and the subsequent formation of secondary nuclei by gas to particle conversion. A long-term increase of the cloud nuclei population will certainly influence the efficiency of the rain-forming process.

During the 1970 Workshop, five CCN counters of the thermal diffusion principle showed satisfactory results and good agreement with natural aerosols. The agreement among the instruments was less satisfactory for artificial aerosols. In the meantime, more sophisticated instruments have been developed and we are looking forward with interest to the experimental phase of this Workshop. Cloud nuclei have become a more and more important fraction of the atmospheric aerosols. I therefore believe that this Workshop is very timely and provides the necessary international platform to study and to discuss the progress which had been made during the last ten years and to give the necessary directives to researchers in this field.

Welcoming and Keynote Address

by

Sean A. Twomey* University of Arizona Tucson, Arizona

A. Historical Perspectives

Early workers (e.g., P. Squires) were aware that there were many more condensation nuclei than cloud droplets; the realization came that it was important to make measurements at low supersaturations rather than with Aitken counters. Discussions in the early 1950's turned to the problem of detecting and counting the small cloud droplets that would be produced at small, cloud-like supersaturations.

This summary of Dr. Twomey's Welcoming and Keynote remarks has been composed based upon notes provided by the author.

The creation of small supersaturations under isothermal conditions (rather than by adiabatic expansion) seemed highly desirable, leading to the conception of the chemical diffusion chamber. At that time (middle 1950's), Wieland in Switzerland had constructed a thermal gradient diffusion chamber and used it to nucleate and grow droplets at small supersaturations. Detection and counting in Wieland's chamber was accomplished by examination of a sticky layer on the chamber floor, into which droplets fell and were hopefully preserved. Twomey introduced photographic detection, and soon the thermal gradient diffusion chamber, with large diameter-to-height ratio, cooled at the base (for convective stability and to avoid transient supersaturations), became the standard device. Photography became the standard detection method, although problems such as nominally "fast" film being sometimes slower than nominally "slower," but less grainy films, remained to be sorted out. Enough devices were in operation that an early comparison workshop could be held at NRL's Chesapeake Bay Annex in 1965.

Whatever the detection method, it will have a threshold - a minimum detectable size of droplet, and it is evidently vital to determine or estimate that, since otherwise one may be counting inactivated (haze) droplets, or not counting all activated droplets. At very low supersaturations an inactivated droplet may be several microns radius and a clear distinction between "haze" and "cloud" (in terms of size) no longer exists. Under such conditions, a simple counting procedure is hardly sufficient and some method of sizing is required, essentially calling for a different technology.

At about the same time, NRL's airborne CCN studies began, and soon investigators were able to compare CCN counts to cloud droplet concentrations. The comparison results were, in general, satisfactory.

B. <u>Points Concerning the Sizes and Composition of</u> <u>CCN</u>

An aspect of typical cumulative CCN distributions fortunate for cloud physics is that there is found a convenient and simple proportionality between critical supersaturation and numbers of activated cloud droplets:

 $N = cs^k$,

where k is typically less than unity and often around 0.5 or so. The parameter k could equally well have been much larger, or the simple power law could have been replaced by some less tractable functional relationship. At very low critical supersaturations, k does tend to increase to greater than unity, creating a more difficult situation where the number of droplets activated is critically dependent upon small variations in the ambient supersaturation, and hence on the detailed time evolution of temperature before and during the condensation processes.

The sizes of natural CCN have been shown by several experiments to be close to the minimum size that Köhler's theory (of the critical supersaturation of pure soluble electrolytes) would allow. Again, it is fortunate that most CCN of interest seem to be soluble compounds, not simply insoluble, wettable particles of various contact angles. It has long seemed that the most likely candidates for these soluble compounds were sodium chloride, ammonium sulfate, and sulfuric acid. Ammonium sulfate continues to be regarded as a predominant constituent.

C. Gaps in the Understanding of CCN

The sources of CCN are not completely understood, although gas-to-particle processes must be a major contributor. The speaker described observations of CCN production occurring over several-hour periods in Arizona, as well as diurnal cycles in the CCN count at the Robertson site in Australia. Related observations by J. Hudson and J. Jiusto were mentioned. CCN are apparently rather transient in nature, with a lifetime of no more than a couple of days. Indeed they cannot exhibit a diurnal cycle if their life expectancy is much longer.

The region of the size spectrum between Aitken nuclei and CCN is another unknown; the ratio of Aitken particle concentrations to CCN concentrations is highly variable, and it seems unlikely that any simple extrapolation or interpolation to connect the two will be sufficient. And to completely understand the progenitors of CCN, it is important to obtain information on the size range from Aitken nuclei on down to molecular clusters, although this region may not be <u>directly</u> relevant to CCN.

The opposite end of the spectrum - from CCN up to "giant" nuclei - is sometimes of influence in cloud droplet growth calculations. It is certainly the critical range in slow condensation processes such as many fogs may be.

A critical instrumental shortcoming is the minimum detectable size of nuclei; some aerosol workers quote 0.01 μm as the minimum sensitive size for particular expansion counters, too large to be helpful in resolving the problems just mentioned.

Welcoming Address

by

Vincent J. Schaefer State University of New York Albany, NY

I am very pleased to have the opportunity to participate in this Workshop on instruments for measuring cloud condensation nuclei. I am particularly intrigued to see the number of young scientists now involved in this interesting and important aspect of cloud physics.

About 25 years ago, Ted Rich of the General Electric Company and I (then Director of Research of the Munitalp Foundation) gathered most of the persons interested in atmospheric nuclei and held a three-day conference on this fascinating subject. Only about 20 scientists and engineers could be found in the United States. During this conference, we identified most of the problems that are still with us, but our techniques then were somewhat primitive when compared with the sophisticated electronic equipment now being used in the laboratory, as well as in the field and on aircraft. I was intrigued with the comments of Dr. Georgii, which in his absence were read by Dr. Dieter Stein. I am indeed sorry to learn that due to illness Dr. Georgii will not be attending this Workshop. I had looked forward to seeing him and hope he is well on the road to recovery. Several statements in his remarks were particularly interesting to me, since I have spent a great deal of time in many parts of the world in an attempt to establish the aerosol concentration patterns in a wide variety of environments. My findings agree completely with the patterns mentioned by Dr. Georgii.

Using the portable and highly reliable Gardner counter with which most of you are familiar, I have found consistent patterns in the concentration of both condensation and cloud condensation nuclei in very clean as well as in very polluted air.

It is well recognized that the Gardner counter provides a very good indication of the concentration of Aitken nuclei when used to provide supersaturations in excess of 300 percent. I have found that this instrument can also be used to provide consistent, as well as semi-quantitative measurements of the number of cloud condensation nuclei in an air sample. If instead of using 20 to 27 inches of mercury vacuum only one scale division is used, the Gardner provides a rather good measurement of the concentration of particles active as water condensing nuclei (CCN) at 1 percent supersaturation.

There are several reasons why this method of using this instrument is frowned upon! Ted Rich who, as you know, invented the so-called Gardner counter was the first to tell me that such a procedure was highly irregular and meaningless. He cited intricate relationships in the life cycle of the cloud droplets forming on nuclei at various supersaturations which he had measured during the development of his instrument.

Despite this disapproval, I have made many thousands of measurements using this instrument in its low vacuum mode as well as high and median settings. I have found that the results provide an extremely consistent measurement of the concentration of cloud condensation nuclei in a particular type of environment. Thus in particle-free air I read zero particles. In very clean air, such as the stratosphere, mid-ocean, the Fiji Islands, mountain summits, caverns, deep forests, and similar places far from man's influence, my measurements of cloud condensation nuclei (CCN) range from 0 to 150 CCN cm⁻³ when the Aitken particle concentration ranges from 200 to 1000 particles per cubic centimeter.

At the other extreme, in cities, the plumes of heavy industries, large airports, superhighways, vehicular tunnels, and the like, the values I find range from 800 to 3000 cloud condensation nuclei and 50,000 to 300,000 Aitken nuclei pe. cubic centimeter.

I have classified the natural and anthropogenic aerosol concentration measured mostly in the Northern Hemisphere and have summarized these findings in three volumes which have recently been published by our Research Center and which I'll be glad to send to participants of this Workshop while the supply lasts. I hasten to add that what I have just said does not in any way suggest that the Workshop which is starting with this meeting is not extremely important. If progress is to be made in our cloud physics studies, it is extremely important that all measuring instruments used to establish scientific facts such as the concentration of atmospheric particles should agree with each other. Only then will we be able to communicate with each other in a meaningful way.

As most of you who have worked in the atmosphere know, the concentration of airborne particles varies over a considerable range, no matter where it is measured. It is necessary when establishing an aerosol "climatology" at a given geographic location to determine the range in concentration over a period of time, its diurnal pattern and the effects that appear with wind direction and atmospheric stability. It is also necessary to measure seasonal variations.

On the west coast of Ireland, for example, with a southerly or southeasterly flow the particle concentration pattern shows the characteristic of continental air (1000 CCN cm⁻³/40,000 AN cm⁻³) while with a northerly or northwesterly flow it is Polar Maritime (100 CCN cm⁻³/800 AN cm⁻³). Even in a region noted for high pollution levels such as the Los Angeles Basin, a strong persistent flow of oceanic air will drive the Basin pollution through the mountain passes and even over the crest of the Sierra to affect the air quality of Las Vegas, Phoenix, Flagstaff and more distant places. When this occurs, the particulate levels which could have been 3000 CCN cm⁻³/300,000 AN cm⁻³ can drop to 500 CCN cm⁻³/10,000 AN cm⁻³. When this happens, the visibility is greatly improved and it is possible to see what a delightful place the Basin would be if its population was reduced by an order of magnitude!

I hope when this Workshop is ended and we collectively gain a higher appreciation of the performance of our instruments that extensive field measurements can be mounted so as to establish confidence in the meaning of our data as they relate to atmospheric visibility, the genesis and nature of storms, the stability of clouds and the formation of precipitation, as well as the basic causes of such specific occurrences as acid rain.

I look forward to participating in this Workshop and look forward to meeting you all and seeing your equipment in satisfactory operation.

Have fun!

. In developing a plan for the Third International CCN Workshop, initial efforts were carried out by an Ad Hoc Commission of the Nucleation Committee of the International Committee on Cloud Physics (ICCP); this Ad Hoc Commission, appointed by Professor Georgii, Chairman of the Nucleation Committee, included J. Kassner, J. Megaw, L. Radke, K. Whitby, and J. Jiusto, Chairman. The Ad Hoc Commission returned the recommendation that the Desert Research Institute's offer to host the meeting should be accepted and that a Steering Committee composed of A. Gagin, J. Jiusto, J. Kassner. W. Kocmond, J. Megaw, L. Radke, and R. Ruskin should proceed with the planning of the specific logistical and scientific details of the meeting.

The Steering Committee functioned until the end of the Workshop on October 17, 1980, as the principal scientific decision-making body. A group completely local to DRI, the "Local Arrangements Committee," handled logistical matters such as participant travel arrangements and lodging, the shipping of instruments to and from Reno, and the creation of suitable laboratory space. Individual members of the DRI staff took responsibility for technical matters such as design of an aerosol generation system which would meet the goals set by the Steering Committee, determinations of instrument power supply and heat rejection requirements, and tentative placement of instruments along the aerosol supply duct.

In the meantime, correspondence between the offices of the Principal Investigators and interested individuals, together with announcements in the Bulletin of the American Meteorological Society and the *Journal de Recherches Atmospheriques*, resulted in a tentative list of over 40 participants, later to be somewhat reduced to the actual attendees as shown in Table 1.

As the effort proceeded into the summer months of 1980, it was decided that some members of the Steering Committee could meet during the International Cloud Physics Conference, Clermont-Ferrand, France (July 1980). Three Steering Committee members who attended this meeting held discussions with a contingent of three DRI representatives; out of the talks came a number of important recommendations that were incorporated into the Workshop Objectives (see Section II').

At DRI, U. Katz took primary responsibility for the aerosol generation and distribution system (Figure 1); a more thorough description of the system can be found in Section V. Aerosol sizing, also described in Section V, was "officially" provided by the Naval Research Laboratory at the request of the Steering Committee.

The final physical and logistical arrangements were handled by the Local Arrangements Committee, mainly through Mrs. Jo Janowski and the staff of the DRI machine shop. Under their supervision, the large components of the aerosol delivery system were installed, provision was made for both 50 Hz



Figure 1. Schematic of Aerosol Generation System

and 60 Hz electricity totalling over 80 kw delivered to outlets spaced along the entire length of the aerosol duct, and 12 new laboratory benches were built. Travel and lodging arrangements were finalized for 22 participants from the USA and 7 from overseas. A total of 17 instruments of various types had to be transported to DRI; in the case of instruments from overseas, customs inspections and clearances were involved.

With most participants and instruments in Reno by October 6, the Workshop opened with welcoming remarks by Prof. W.C. Kocmond, Prof. H.-W. Georgii (represented by Mr. D. Stein) and Dr. V. Schaefer. A keynote address was given by Dr. S. Twomey. Two weeks of intense activity then began.

The Steering Committee met daily to plan the test aerosols to be used; generally their deliberations were conducted at the beginning of each day, while instrument operators readied their devices and monitored outside ambient aerosol. Table 2 summarizes the 30 experiments actually performed. The specifications for each test aerosol varied day-to-day, in response partly to earlier suggestions originating in the Steering Committee or coming from individual participants:

(a) High CCN concentrations (relevant to the atmosphere in areas of high pollution, volcanic plumes, etc.) to establish the practical upper limit of applicability of various instruments;

 (b) Low CCN concentrations (e.g., less than 100 cm⁻³ at 1%), to check instrument performance in clean maritime or polar environments;
 (c) A bimodal size distribution (two mono-

(c) A bimodal size distribution (two monodisperse, soluble aerosols of different nominal sizes), to check the ability of instruments to resolve a correspondingly bimodal CCN activity spectrum.

With "feedback" from participants, these suggestions were incorporated into the rather tight experiment schedule (suggestion "a" is reflected in experiments 9 and 24, "b" in 4 and 23, and "c" in 10). Additional suggestions during the Workshop led to experiment 21, a test of the ability of instruments to give a consistent and accurate reading over a prolonged period of time, CCN concentration being held constant; experiment 25, a test of the "zero" of instruments sampling particle-free air, and experiment 29, a test of the possible CCN activity of a hydrophobic aerosol, paraffin wax.

Daily activities included frequent discussions between members of the Steering Committee and the larger body of participants. Steering Committee members assisted with the presentation of collected data after each experiment; in addition, participants presented informal talks on their equipment on a daily basis.

At the end of the Workshop, the Steering Committee completed its function by requesting all participants to supply a description of their instruments together with comments on performance during these experiments; those reports are assembled in Section V. In addition, various individuals were asked to submit reports on special interest topics, such as reviews of the CCN counters by generic type; those reports are to be found in Section VI.

The final action of the assembled group was to provide verbal and written response to the Steering Committee's request for criticisms and suggestions for improvements. The following points resulted:

(a) Although previously debated, the question of whether or not to standardize data-taking such that all CCN counters would measure CCN concentrations at a few, given "set-point" supersaturations should be reviewed again. The practice at this Workshop was to avoid specific, standard settings (because of different instrument requirements) and to provide as complete a spectrum as possible. "Set-points" would simplify data comparisons but, of course, not all instruments are designed to operate at the same supersaturations.

(b) It was intended that the results of each experiment would be quickly entered onto computer file. In practice, this effort, which ideally could have provided data listings and computer graphics in short turn-around time, was hampered by difficulties in design and planning as well as equipment problems. Future Workshops would benefit from rapid turn-around of the data, but better control of the experimenter input is also needed.

(c) Power and aerosol supply were in reasonably good shape; ambient heat rejection (air conditioning) and physical space had less margin of comfort but were satisfactory. Noise level was high at times; one participant suggested that a vacuum manifold (replacing numerous individual pumps) would alleviate this problem.

In summary, most participants felt that the Workshop proceeded quite smoothly and that the objectives set forth by the Ad Hoc Commission of the Nucleation Committee were successfully achieved. In making this determination, special mention must be made of the close cooperation between participants, Steering Committee and DRI hosts. Several social events were held during the 12 day Workshop, including an open house hosted by Dr. Cliff Murino, President of DRI. The strong sense of friendship and good times that develop in a working environment such as this will always be remembered.

TABLE 1. CCN WORKSHOP PARTICIPANTS

Name/Affiliation	Instrument	Duct No.	Name/Affiliation	Instrument	Duct No.
Dr. Jeffrey B. Anderson Space Science Lab NASA-MSFC Alabama 35812			Dr. James G. Hudson Desert Research Institute University of Nevada System P.O. Box 60220 Reno. NV 89506	CFDCC 3SS CFDCC IHC	14,16,18
Dr. Greg Ayers CSIRO Box 134, Epping, NSW 2121 Sydney, Australia	STGDCC	13	Dr. James Jiusto Atmos. Sci. Res. Center State Univ. of NY at Albany 1400 Washington Avenue	STGDCC	10
Dr. Darryl Alofs Graduate Center for Cloud Physics Res. Univ. of Missouri-Rolla 109 Norwood Hall Rolla, MO 65401	CFDCC	21	Albany, NY 12222 Dr. J. Kassner Graduate Center for Cloud Physics Res. 109 Norwood Hall Univ. of Missouri-Rolla	CFDCC	21
Mr. Randolph D. Borys Colorado State University Dept. of Atmospheric Sciences Ft. Collins, CO 80523	STGDCC	25	Rolla, MO 65401 Dr. Ulrich Katz Desert Research Institute University of Nevada System	Aero. Gen.	
Mr. Jack Dea Desert Research Institute University of Nevada System P.O. Box 60220 Reno, NV 89506	Aero. Gen.		P.O. Box 60220 Reno, NV 89506 Dr. Vernon Keller Space Sciences Lab NASA-MSFC		
Dr. S. Domonkos University of Washington Dept. of Atmos. Sciences Seattle, WA 98195 Dr. L.R. Faton	4SS CFDCC	20	Alabama 35812 Mr. Gary Keyser Desert Research Institute University of Nevada System P. 0 Roy 60220	DRI-NASA CFDCC	15
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Naval Research Laboratory Washington, DC 20375 Dr. Abe Gagin Cloud Physics Laboratory Department of Meteorology	STDGCC	9	England Prof. Warren C. Kocmond Desert Research Institute University of Nevada System P.O. Box 60220	DRI-NASA CFDCC	15
Hebrew University Jerusalem, Israel Dr. Herman E. Gerber Naval Research Laboratory Washington, DC 20375			Reno, NV 89506 Dr. G.G Lala Atmos. Sci. Res. Center State Univ. of NY at Albany 1400 Washington Avenue	STGDCC	10
Dr. Edward Hindman II Colorado State University Dept. of Atmospheric Sciences Ft. Collins, CO 80523	IHC	26	Albany, NY 12222 Mr. R. Leaitch Dept. of Physics York University 4700 Keele Street	Diffusion Tube	7
Dr. W.A. Hoppel, Code 4320 Atmospheric Physics Branch Naval Research Laboratory Washington, DC 20375	Aero. Sizing	12	Downsview, Toronto Canada M3J1P3 Dr. Ray McKenzie Chemistry Building	Pollak	28
Mr. Richard Hucek Florida State University Dept. of Meteorology Tallahassee, FL 32306	CFDCC Impactor	22, 23	National Bureau of Standards Washington, DC 20234	121	

Name/Affiliation	Instrument	Duct No.	Name/Affiliation	Instrument	Duct No
Dr. William H. Mach Florida State University Department of Meteorology Tallahassee, FL 32306	CFDCC Impactor	22,23	Dr. David Rogers University of Wyoming Dept. of Atmos. Science Box 3038, Univ. Station Laramie. WY 82071	STGDCC Aerosol Sizing	1,2
Mr. Thomas'R. Mee Mee Industries, Inc. 1629 S. Del Mar Avenue San Gabriel, CA 91776	STGDCC	8	Dr. R. Ruskin Naval Research Laboratory Washington, DC 20375	STGDCC	17
Dr. W.J. Megaw Department of Physics York University	Diffusion Tube	7	Dr. V.J. Schaefer State University of New York Albany, NY 12222		
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Dr. T. Ohtake	Impactor/	27	Bourder, CO 80307		
Geophysical Institute University of Alaska Fairbanks, AK 99701	Photomicrography	y	Dr. D. Stein Institut fur Met. und Geophy. D6000 Frankfurt a. Main-l Feldbergstr. 47, Germany		
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Ms. Marsha Politovitch University of Wyoming Dept. of Atmos. Science Box 3038, Univ. Station	STGDCC Aerosol Sizing	1,2	Rolla, MO 65401 Dr. Sean Twomey Inst. of Atmospheric Science		
Laramie, WY 820/1			Tucson, AZ 95721		
Dr. Lawrence F. Radke Dept. of Atmos. Sciences University of Washington Seattle, WA 98195	4SS CFDCC	20	Mr. C.H. Wilson NASA Langley Research Center MS 404B Hampton VA 23665		
Dr. C. Fred Rogers Desert Research Institute University of Nevada System P.O. Box 60220 Reno, NV 89506	DRI-NASA CFDCC n	15	Dr. T. Wojciechowski Naval Research Laboratory Atmospheric Physics Branch Washington, DC 20375	STGDCC	17

No'.	Date		Aerosol A=Ambient; M=Monodisperse; P=Polydisperse
0	Tues 7 Oct	AM	M - (NH ₄) ₂ SO ₄
۱	Tues	PM	P - NaCl - oscillating concentration
2	Tues	PM	P - NaCl - higher concentration
3	Wed 8 Oct	Ам	A - quite fluctuating
4	Wed	AM	M - NaCl - low concentration
5	Wed	PM	M - NaCl - medium concentration
6	Wed	PM	Α
7	Thurs 9 Oct	AM	A - aborted - duct blockage
8	Thurs	АМ	M - NaCl - slight drift down
9	Thurs	AM	M - NaCl - higher concentration
10	Thurs	РМ	Bimodal - NaCl - "flat k"
11	Thurs	РМ	Α
12	Fri 10 Oct	АМ	Α
13	Fri	AM	$P - (NH_4)_2 SO_4$
14	Fri	PM	$P - (NH_4)_2 SO_4$
15	Fri	PM	M - (NH ₄) ₂ SO ₄
16	Fri	РМ	A
17	Mon 13 Oct	АМ	A
18	Mon	АМ	M - (NH ₄) ₂ SO ₄
19	Mon	АМ	M - (NH ₄) ₂ SO ₄
20	Mon	PM	M - (NH ₄) ₂ SO ₄
21	Mon	РМ	$M - (NH_4)_2 SO_4 - time variations$
22	Tues 14 Oct	AM	$P - (NH_4)_2 SO_4 - medium concentration$
23	Tues	PM	$P - (NH_4)_2 SO_4 - 1ow concentration$
24	Tues	PM	$P - (NH_4)_2 SO_4 - high concentration$
25	Wed 5 Oct	АМ	Filtered air - noise check
26	Wed	АМ	Α
27	Wed	PM	P - AgI, "insoluble"
28	Wed	РМ	M - AgI, "insoluble"
29	Wed	PM	P - paraffin, hydrophobic

Table 2. List of Experiments

SECTION V. INSTRUMENT DESCRIPTIONS

CONTRIBUTION TO CCN WORKSHOP REPORT FROM UNIVERSITY OF WYOMING GROUP

David C. Rogers and Marcia K. Politovich University of Wyoming Laramie, Wyoming



1. APPARATUS

The University of Wyoming's CCN counter is a static, horizontal, parallel plate thermal gradient diffusion chamber of rather conventional design. Its intended use is primarily for field measurements, hence the small physical size and straight-forward simplicity of operation. The plate separation is 0.9 cm, and the inside chamber dimensions are 8.5 cm \times 10.0 cm; aspect ratio is 9.4:1. Activated CCN grow to visible size droplets which are photographed with a 35 mm camera (Nikon F, Micro-Nikkor-P lens f3.5,55 mm plus M2 extension tube, Tri-X film developed at ASA 1600). Illumination is provided by a Helium-Neon laser (0.6328 μm wavelength, multimode 5 mw) oriented at an angle of 23° from the forward direction of the camera's optical axis. The multimode character of the laser provides a flat-top intensity profile which serves to reduce uncertainties about the size of the illu-minated volume. The angle of 23° in the forward direction was chosen to take advantage of the first broad peak in the Mie scattering function for water droplets which are in the size range 3 to 7 μm diameter. The circular laser beam is 0.18 cm in diameter and is centered in the chamber to illuminate the middle 20% of the vertical distance. A fixed width on the film is used to define the horizontal dimension of the sample volume; this volume is .034 cm³, in the form of an elongated cylinder.

The temperature difference between the top and bottom plates is measured by precision thermistors which are flush-mounted between the aluminum plates and the surface wicking material (blotting paper). This temperature measurement is displayed to $\pm 0.1^{\circ}$ C and is also used by an electronic circuit to control the temperature difference. The measurement is compared with a value selected by the operator, and the difference is used to control the current to thermoelectric modules which cool the bottom plate. The range of temperature differences normally used extends from 3 to 7°C; this results in a range of supersaturations of approximately 0.3 to 2%.

Sample air passes through a temperature preconditioning chamber just before entering the CCN chamber. The preconditioning chamber is maintained slightly warmer than the top plate of the CCN chamber. In this manner, transient supersaturations are minimized. Air samples are brought into the chamber under suction and, after thorough flushing, the chamber outlet is closed. A time delay of several seconds between this closing time and the photography allows droplets to grow large enough to be photographed but not so large that they fall out of the illuminated volume; this delay is controlled and decreases with larger supersaturations.

Earlier calibration experiments using monodisperse latex particles determined the photographic minimum detectable particle size is less than 0.7 μm diameter. Aerosol losses in the entrance region have been measured as negligible by comparing size distributions of various aerosols before and after passing through the chamber. Finally, earlier comparisons with theory and other CCN chambers have been performed for monodisperse and polydisperse salt aerosols as well as natural aerosols. Comparisons between the University of Wyoming's (UW) chamber and that of the Desert Research Institute (DRI) were made during March 1978. These comparisons are briefly summarized in Tables 1 and 2, and Figure 1.

TABLE 1. CFD-SDL COMPARISONS

Super-	Concentration	(no. cm ⁻³)	CFD/SDL
saturation(%)	SDL	CFD	
.28	95	168	1.77
.36	207	288	1.39
.50	164	219	1.34
.65	243	352	1.43
1.02	1310	3596	2.75
.35	320	241	0.75
1.00	327	497	1.52
1.00	371	613	1.65
	368	544	1.48
0.2	485	750	1.55
1.0	485	810	1.67
	Super- saturation(%) .28 .36 .50 .65 1.02 .35 1.00 1.00 1.00 0.2 1.0	Super- saturation(%) Concentration SDL .28 95 .36 207 .50 164 .65 243 1.02 1310 .35 320 1.00 327 1.00 3/1 .368 0.2 485	Super- saturation(%) Concentration (no. cm ⁻³) SDL CFD .28 95 168 .36 207 288 .50 164 219 .65 243 352 1.02 1310 3596 .35 320 241 1.00 327 497 1.00 3/1 613 .368 544 0.2 485 750 1.0 485 810

CFD/SDL avg. 1.57 ± .47

Omitting 2 extremes, avg. 1.53 \pm .14

TABLE 2. CRITICAL SUP	ERSATURATION
-----------------------	--------------

Aerosol	Critical	Super	saturation(%)
	SDL	CFD	Theory
NaCl .03 µm diameter	.36	.32	.35
NaCl .05 µm diameter	.75	.68	.75
AgI-NH,I complex	<0.2	<.25	.07
(0.2 µm dia.)			

SDL	=	Static Diffusion Liquid CCN Chamber (UW)
CFD	=	Continuous Flow Diffusion CCN Chamber (DRI)



Figure 1. CCN spectra measured in Laramie, March 1978. SDL = Static Diffusion Liquid CCN Chamber (UW); CFD = Continuous Flow Diffusion CCN Chamber (DRI).

2. WORKSHOP PARTICIPATION

Our main interests in participating in the International CCN Workshop were to compare the University of Wyoming's CCN chamber with the others present and to discuss with and learn from the participants various aspects of CCN measurement techniques, their current thinking, and advances in the ten years since the last workshop.

Unfortunately, problems with the electronics in our device arose during the Workshop and prevented us from operating the chamber above about 0.3% supersaturation, so our comparison experiments were attenuated.

3. DATA USAGE

The emphasis of our work with the CCN counter has been in the application of CCN data obtained from it, rather than development of the instrument itself. We have kept the design simple and the size small to enable us to transport it to remote sites for in situ sampling. We include here several examples to illustrate the manner in which the data are used.

Figure 2 shows a C-K plot of measurements obtained at and near our cloud observation facility at Elk Mountain in southeastern Wyoming. These measurements were taken to characterize winter CCN populations in southeastern Wyoming, particularly near Elk Mountain. The data suggest a trend of increasing C-values later in the spring while retaining similar K-values. There was significant snow cover in and upwind of the sampling location during January and February. In later months, snow cover was sparse if present at all.



Figure 2. C-K plot of the CCN spectra measured during the winter of 1978-79 near Elk Mountain. Sample months and locations are indicated.



Figure 3. Droplet measurements obtained in an orographic cloud which formed over Elk Mountain on 16 January 1979. Data sources are indicated (ASSP is the Axially Scattering Spectrometer Probe, Particle Measuring Systems, Boulder, CO, and CGS is the University of Wyoming's soot-coated impactor slide droplet sampler). CCN predictions represent application of Twomey's (1959) relationship between CCN spectrum parameters and droplet concentrations to measurements taken upwind of Elk Mountain. (O) represents an updraft of 50 cm/s, (I) represents the range of updrafts 25-100 cm/s. A 20-minute time lag was added to CCN data to account for transport time from the CCN sampling site to the droplet measuring site.

An important use of our CCN data has been in comparisons of droplet concentrations derived from the upwind CCN spectra (using Twomey's, 1959, equation¹ and Young's, 1979, cloud model²) with in situ measurements from the observatory and instrumented aircraft in the Elk Mountain cap clouds. Figure 3 shows the results of such a comparison on 16 January 1979. The CCN predictions track the actual measurements from the observatory well. Droplet concentrations measured by the aircraft (N10UW) were higher than those from the observatory (EMO) on this day, which we attribute to variations in the vertical structure of aerosol concentrations in the boundary layer, which were measured by the aircraft, rather than substantial increases in updraft speeds at that level.

These figures are from a paper by M.K. Politovich which is being prepared for publication.

¹Twomey, S., 1959: The nuclei of natural cloud formation, Part II: The supersaturation in natural clouds and the variations in cloud droplet concentration. *Geophysics Pura. Appl.*, 43, 243-249.

²Young, K.C., 1974: A numerical simulation of wintertime, orographic precipitation. Part I: Description of model microphysics and numerical techniques. J. Atmos. Sci., 31, 1735-1748.

DESCRIPTION AND DISCUSSION OF THE NRL TGDCC

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The NRL thermal gradient diffusion cloud chamber (TGDCC) is identical to that described in the Proceedings of the Second International Workshop on Condensation and Ice Nuclei (Grant, 1971). The chamber consists of two plates 7.5 cm in diameter separated by 1.25 cm and covered with saturated filter paper. The cylindrical wall separating the plates is glass. The top plate is at room temperature and the bottom plate is cooled with a thermoelectric cooler. The temperature difference is measured with several sets of thermocouples. Α high intensity mercury arc lamp with appropriate lens and collimating slits which are imaged in the cloud chamber defines an illuminated volume 0.15 cm thick and 0.3 cm high by 1.5 cm long. This illuminated volume is viewed at 90° scattering angle by an 8 mm camera and a video camera system.

The video system permits immediate playback and stopframe counting of the cloud droplet concentration in the chamber. The 12-inch monitor screen is marked off to identify the viewing dimensions for ease in counting the droplet images.

At the Workshop, samples for a given experiment were admitted to the TGDCC directly from the sampling duct which was slightly pressurized. The count was recorded both on video tape and on a number of frames of an 8 mm movie film. The 8 mm Bolex camera is operated manually while viewing the chamber through the camera. The first frame is taken one to two seconds after closing the valve and about 10 to 15 frames are taken manually until it is clear that the maximum particle count has passed. The photographic results were not available at the Workshop. The video data were read immediately after each experiment. The method for obtaining the count supplied at the Workshop is as follows: two (or three) successive samples were recorded at each supersaturation. The recordings from these samples were played back and visually examined to obtain the succession of frames for each sample where the maximum count occurred. Several of these frames were counted and the maximum count obtained. The maximum counts from two (or three) samples were averaged to obtain the data submitted at the Workshop.

The photographic recordings have not yet been analyzed but it has always been our experience in the past with atmospheric aerosols that the two methods track very well but that the photographic count is always about 10 to 15% higher. We attribute this to the smaller detectable size obtained with the photographic method. The minimum detectable size for our photographic system is estimated to be about a half micron whereas the minimum detectable size for the video system is estimated to be just under one micron.

There are definite limits on the supersaturation range for which the results of the TGDCC are valid. The range of validity is usually given as 0.2 to 1%. However, even within this range there can be significant errors depending upon the nuclei spectrum (size distribution) being measured. The accuracy limitations of TGDCC's in general and our chamber in particular were investigated by Hoppel and Wojciechowski (1976).

The smaller particles which have critical supersaturations about the same as the maximum supersaturation in the TGDCC are not nucleated until the chamber has reached its steady state value and then grow more slowly than the larger particles which are nucleated before the chamber has reached equi-The object is to find a period of time librium. when the less active particles have grown to minimum detectable size and the larger (more active) particles have not yet started to fall out. A time which satisfies both criteria may not exist for all size distributions. If such a period of time does exist, then it should evidence itself by a plateau in the curve of number detected versus time. Α well-defined plateau was not found to exist in the data on natural continental aerosols presented by Hoppel and Wojciechowski (1976). Alofs and Carstens (1976) did a numerical simulation of the TGDCC which predicted large errors depending upon the nuclei distribution and minimum detectable size.

Another source of uncertainty in the TGDCC is statistical in nature and has to do with the fact that the number of particles in a small volume will deviate from the true macroscopic mean with a standard deviation given by the square root of the mean. The typical number of particles in our viewing volume varies from about 20 for nuclei concentrations of 300 cm^{-3} to 200 for nuclei concentrations of 3000 cm^{-3} . At the lower end there is therefore a standard deviation (in a large number of measurements) due to real natural variations of over 20%. This uncertainty, of course, can be reduced by averaging more measurements from the same macroscopic sample of air.

For a monodisperse nuclei sample the plateau in number versus time should be much more pronounced than in the case of a steep size distribution as is usually the case for continental aerosols. At the Workshop, we had a unique opportunity to sample nearly monodisperse aerosols as well as polydisperse and natural aerosols. The Workshop data, therefore, offered a unique opportunity to look for a plateau in number as a function of time.

Figures 1 through 4 show the results of counting every tenth frame (every third of a second) on the videotape starting shortly after the motion in the chamber subsided. There has been no attempt to synchronize the starting points to the same time after the valve was closed. Therefore, there may be a maximum of a half second offset in the plotted times from one sample to the next. The supersaturation in all four figures was about 0.72%.

Figures 1 and 2 are for monodisperse NaCl and $(NH_4)_2SO_4$ with two and three runs, respectively. Both of these figures evidence more of a plateau type behavior than we have seen in those ambient samples which we have previously examined in detail. We assume that the variations from one frame to the next after the maximum is caused by the unequal rates at which particles fall into or out of the sensitive volume. Any difference in the level of the plateau from one sample to the next



Figure 1



Figure 2

would be due to natural fluctuations discussed above and should have a standard deviation equal to the square root of the number of particles counted.

Figures 3 and 4 are for polydisperse $(NH_4)_2SO_4$ and ambient air, respectively. Here the curves are peaked more strongly as would be expected with little evidence of a plateau. The arrows along the right side of the figures indicate the values obtained at the Workshop by the method of analysis indicated earlier.

For many years, NRL has used the standard method cited earlier for determining the CCN concentration from the video recording. This procedure of averaging the maximum count obtained on several successive recordings at the same supersaturation results in concentrations which are somewhat higher than concentrations calculated from an average across the plateau. We have persisted in analyzing our data in this manner for several reasons: (1)the time required to count enough frames to define a plateau (or lack of one) is prohibitive on a routine basis; (2) comparison of results using our photographic system and video system gives values for the video system which are 10 to 15% lower than those obtained with the photographic system, which has a smaller minimum detectable size; (3) most importantly, if there is no plateau then the maximum value should be closest to the correct value; and (4) our results over many years are internally consistent since we have not changed this procedure.



Figure 3



Figure 4

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 J. Appl. Meteon., 15, 107-112.

ACKNOWLEDGEMENTS

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THE CCN COUNTER OF THE I.O.P.G. OF PUY DE DOME: MAIN CHARACTERISTICS AND RESULTS OF MEASUREMENTS

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1. DESCRIPTION OF THE INSTRUMENT

The experimental device used by the I.O.P.G. of Puy de Dome for counting CCN during the Workshop is a cylindrical static thermal diffusion chamber with horizontal plates, 7.0 cm in diameter and 2.1 in thickness (volume V \approx 80 cc). The system includes a multimode laser beam (power > 5mW), illuminating a part of the median volume of the chamber in which the maximal supersaturation is developing. A video set (TV camera + monitor) displays a motion picture of the droplets generated by the CCN and observed in a volume V = 7.5 10⁻³cc. For obtaining a better signal-to-noise ratio (a better contrast), the TV tube was selected so that its current-wavelength characteristic was in good agreement with the (He, Ne) laser radiation. The TV image is recorded on a video tape recorder, thus allowing analysis shortly after the time of measurement.

Instead of counting by sight, a device called "Image Processing Unit", originally provided for counting in real time, is used. The upper plate is electrically heated and the lower plate is at ambient temperature. Water vapor is supplied by a wet mineral paper filter. The desired temperature difference ΔT (limited at 5°C) between the horizontal plates is displayed and then regulated by an automatic control system, which includes a series of eight differential thermocouples imbedded in the plates. This mounting has the advantage of not only amplifying the voltage but also of taking into account an average temperature on each plate.

The air sample under analysis is introduced in the chamber at a flow rate of 1.0 &/m, and a section of the air intake tube is heated, allowing air sample introduction into the chamber at a temperature of about 3°C above the lower plate temperature. The supersaturation range covered by this equipment is generally between 0.1 and 1.5% and the time required for obtaining a spectrum in this supersaturation range is approximately 30 to 40 minutes. But in conditions of ambient temperature above $25^{\circ}C$ (as occurred during the Workshop), it is not possible to exceed a supersaturation of 0.8% because of the ΔT limitation (see above). Because the device cannot use another greater value of observed sample volume, the accuracy of the measurement is poor at the low extremity of supersaturation tion range (about 50%).

2. SELECTIVE DATA ANALYSIS

2.1 Global Survey

With such a device, 20 supersaturation spectra (plus one point for Experiment No. 13) corresponding to our complete participation in the Workshop have been outlined and classified according to their shape and position in regard to the series of others established for each experiment.

The following main characteristics of our spectra, appearing in Table 1 with the same capital letter, have been distinguished like that: (A) similar to and well placed among the others in the series; (B) same type of concavity as for the majority in the series; (C) close to the upper (or lower) extremes of the series; (D) smaller CCN concentrations at low supersaturation and/or greater CCN concentrations at higher supersaturation; (E) CCN concentrations abnormally high along the most part of the spectra, with some of them being higher than those obtained with Pollak or TSI counters; (F) lack of plateau; and (G) type of concavity differing from that of the majority.

For each supersaturation spectrum denoted by the number of the experiment, one, two or three, but no more, of these main characteristics have been summarized in Table 1 by a cross in the corresponding square. On Line C, we distinguish respectively by an "s" or "i" index, the fact that the spectrum is close to the upper or lower envelope of the series. In comparing the measurement sheet and the TSI recorded graph corresponding to Experiment No. 16, it can be observed that the abnormal concavity of the spectrum might correspond to a measurement carried out during a peak of aerosol discharge (cross in bracket, Line G).

Through this analysis, Table 1 appears to be divided into two parts: (A, B, C) - grouping the characteristics which reveals a behaviour not far from a "supposed mean" behaviour for a CCN counter; and (D, E, F, G) - grouping the characteristics which reveals an anomaly of behaviour. It is obvious that the percentage of crosses shown in the

first group (A,B,C) is higher than that of the other group, i.e., 62% against 38%.

Character -	T	Experiment n°																			
ization	4	6	8	9	10	П	12	13	14	15	16	17	18	19	20	22	23	24	26	27	28
Α	X			Ø	6J	X						X	X	X	1	X					
В	Γ	X	$\overline{\mathcal{N}}$		X	X	Ø					Ø	X			X	X	X			X
С	1	Ø,					Ø,	K,	×.		Ķ		Γ					X		Xç	
D	T												X		X		X		X	X	X
E	1		X										Ι		X						
F	X									\mathbb{X}				X							
G	Γ	Γ		X					X	X	80								X		

Table 1

Another more realistic analysis consists of dividing the supersaturation spectra in three categories:

1. The crosses pertaining to a given spectrum are shown in the upper part of Table 1. This situation, corresponding to a good behaviour of the device, concerns 9 spectra, i.e., 43% of the whole.

2. The crosses pertaining to a given spectrum are shown in the lower part of Table 1. This situation, corresponding to a frankly bad behaviour of the device, concerns only 3 spectra, i.e., 14% of the whole spectra.

3. The crosses pertaining to a given spectrum are distributed on both parts of Table 1. This situation corresponds to an intermediate behaviour and concerns the remaining spectra, i.e., 43%.

Moreover, it is possible to point out that: (a) the number of spectra for which the crosses are found either on Line C or D is relatively high, i.e., 13 out of 21; and (b) on Line C, the number of crosses affected by the "s" index (i.e., 5) is higher than the number of crosses affected by the "i" index (i.e., 2). Both factors denote a trend in the device to overestimate the CCN concentration, especially in the range of high supersaturation.

2.2 Individual Comparison

Comparisons have been made between our results and spectra obtained using similar equipment (i.e., static diffusion chamber) on corresponding experiments. At times, the spectra were in close agreement, such as: NRL (n° 6-11-12-17); CSIRO (n° 9-10-14-17-28); SUNY (n° 9-14); Hebrew University (n° 9-14-17-18-24). However, the comparison also revealed differences in the spectra, with the concentration ratio reaching 2 to 3 with NRL (n° 9); CSIRO (n° 4-6-8-20); CSU (n° 4-23); SUNY (n° 6-8-9-10) and Hebrew University (n° 10-28). The discrepancy with the Wyoming equipment was still larger (n° 11-12-14-22-23). Surprisingly, for about 62% of the experiments, a number of our spectra were found in reasonable agreement with those resulting from measurements with the DRI continuous flow diffusion chamber (n° 9-12-14-17-18-19-22-27-28) or the Missouri-Rolla haze chamber (N° 6-10-11-12-13-18-19-22-27-28).

2.3 Stability of the Measurements

In Experiment No. 21, which tested the repeatability of the measurements, two runs were carried out continuously during approximately 16 minutes and 30 minutes at supersaturations 0.50% and 0.26%, respectively. At the same time, the aerosol to be analyzed was delivered at a stable concentration level of 860 cc⁻¹ (measured with the TSI equipment) or 1000 cc⁻¹ (measured with the CCN Pollak counter). After calculating the corresponding standard deviation, the average concentration values were 1210 \pm 105 at 0.50% supersaturation for the 1st run and 190 \pm 70 at 0.26% supersaturation for the 2nd run.

Two features are emerging from these results: (1) a sensible overestimate of the concentration; and (2) a standard deviation which reaches an acceptable percentage (9%) of the concentration values at mean or high supersaturation, while it is not acceptable (37%) at low supersaturation. This is mainly due to the fact that the examined volume cannot be adjusted to discrete supersaturation ranges.

3. CONCLUSIONS

In performing the data analysis of the measurements achieved during the Workshop with our device, a global approach was preferred rather than an individual analysis, in order to illustrate some main characteristics in the behaviour of the device with respect to a "mean behaviour" resulting from a general survey of all the equipments involved in each experiment. In this regard, our device seems to have a behaviour not unlike a "mean behaviour", although it tends generally to overestimate the CCN concentrations measured near the high supersaturations and sometimes underestimates the concentrations close to 0.1% or 0.2% of supersaturation. Despite the fact that it belongs to a type of static diffusion chamber, it shows, however, similar spectra to those obtained with other types of chambers (continuous flow diffusion chamber and haze chamber).

In the spring of 1980, at the site selected by WMO for a possible Precipitation Enhancement Project in the area of Valladolid, Spain, the CSIRO device and our static thermal diffusion chamber were placed side by side for the purpose of analyzing the same natural air sampling. Although at that time, the air intake of our chamber was not yet equipped with a heater, the main difference between the two devices was the way in which the AT between the plates was achieved. In fact, in the CSIRO device the lower horizontal plate is cooled. The comparison of the results displayed CCN concentrations from twice to three times higher with our device, so that our CCN measurements on the P.E.P. site in 1979 and 1980 were questioned.

Such discrepancies between these devices were again found in some Workshop experiments; however, it was possible to observe that, in a number of other experiments at the least equivalent, the corresponding spectra were rather close to each other (see Section 2.2). Neither the difference in the methods used to obtain ΔT nor the differences in the geometry of the chambers suffice to explain such variable results. Nevertheless, it is obvious that the comparisons made during meetings of a workshop type are able to greatly improve our knowledge of the behaviour of the device involved. As

Personal communication of Dr. Warren King from CSIRO.

for our equipment, it seems that the best way to try to reduce its trend to overestimate the CCN concentrations should consist in cooling the lower plate instead of heating the upper one and in searching for a better diameter-to-depth ratio in order to improve the stability inside the chamber.

4. ACKNOWLEDGEMENTS

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UMR DUAL MODE CCN COUNTER (MODES: CFD PLUS HAZE)

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1. INSTRUMENT DESCRIPTION

This instrument and its performance characteristics have recently been described in detail (Alofs, 1978; Alofs, *et al.*, 1979). Therefore only a brief description of it is given here.

The chamber consists of two vertical plates 100 cm long in the vertical direction and 13 cm wide, with a 0.8 cm spacing between the plates. The sample flows downward between the plates in a small diameter stream surrounded by filtered air. The sample flow into the chamber is usually 0.008 liters/min, determined by the pressure drop through a capillary tubing 0.25 mm diameter. This flow branches off of a larger delivery flow of 5 liters/ min, to avoid diffusional losses.

The droplets formed on the CCN are counted and sized using a Climet Model 201 optical particle counter (OPC) with an 8-channel pulse height analyzer (Climet Model 210). An advantage of the Climet over the Royco 225 is that the response curves increase monotonically with size (no dips to produce multivalue sizes for a given pulse height). The Cooke and Kerker (1978) calculations are used to correct for index of refraction. A sheath air inlet was constructed for the Climet OPC, similar in design to that used in the Royco 225. The rate of flow into the Climet OPC is 0.35 to 1.0 liters/ min.

The haze mode is used for the nuclei active at supersaturations (S) from 0.0133 to 0.173%. In this mode the two plates are kept at the same

temperature (25°C). The values of S are determined from the drop sizes, using the relation S = .08/d, with S in % and with the drop diameter d in μ m. For nuclei with .068 < S < .173, the total chamber flow (sample plus filtered air) is l liter/min, giving a residence time of 39 sec. For lower S, the chamber flow is 0.35 liter/min and the time is 110 sec. In either case, all of the flow is drawn into the OPC.

In the CFD mode the supersaturation is determined by the temperature difference between the two plates. This temperature difference is controlled by water baths, and is measured with a mercuryglass thermometer immersed sequentially in each water bath. Separate experiments show that the temperature difference at the plate surfaces (measured with thermocouples on the air side of the filter paper) equals the bath temperature difference to within 5%.

About 40 minutes is required to obtain a spectrum (5 values of S in the CFD mode, plus 7 values of S in the haze mode). About half of this time is used to adjust the temperatures and flows; the rest is used in actual counting of the nuclei. Generally the measurement begins in the CFD mode, with the hot plate at 25° C and the cold plate at 20° C. The cold plate temperature is raised in steps (the baths heat very quickly, 1000 watts of heating versus about 50 watts of cooling). The chamber flow rate is usually reduced as S is decreased. The time for flushing the chamber is quite short in the CFD mode, but amounts to about 5 minutes in the haze mode at 0.35 liters/min.

2. PERFORMANCE AT THE WORKSHOP

The analysis that follows was performed based on the computer printouts of the data, as supplied to us in February, 1981.

Consider first the monodisperse sodium chloride and ammonium sulfate experiments. Let S_{CU} denote the critical supersaturation determined from the size given by U. Katz. On a plot of CCN count versus S, these aerosols show a plateau. At a count of 50% of the plateau value, let the corresponding S be defined as the measured critical super-Define the parameter X by X =saturation, Sc. (S_C - S_{CU})/S_{CU}. Table 1 shows the average (\vec{X}) and standard deviation (σ) of X for eight instruments that we judged to be giving above average perfor-mance. There are two groups of experiments, with mance. $S_{CII} > 0.1$ shown on the top 6 rows, and $S_{CII} < 0.1$ shown on the bottom 4 rows. In each group the instruments are arranged according to $\sigma,$ lowest σ on top.

In the top group of Table 1, it can be seen that our instrument gave \overline{X} = +0.0296 and σ = 0.107

INCEDUMENT	EXDENTMENTS	SCU	x • (Sc ^{-S} cu ^{1/S} cu_				
INSTRUMENT	EXPERIMENTO	RANGE	AVERAGE,X	STD.DEV.,σ			
#18, DRI-CFD Hudson-Squires type	4,5,10,15, 18,20	>0.1	-0.0565	0.103			
#21, UMR CFD mode	4,5,10,15, 18,20	>0.1	+0.0296	0.107			
#17, NRL Twomey type	4,5,10,18	>0.1	-0.171	0.125			
#12, NRL elect. class.	4,5,10,15, 18,20	>0.1	+0.0169	0.156			
#15, DRI-CFD NASA	10,15,18, 20	>0.1	+0.0067	0.158			
#10, SUNY static, scattering	5,10,18	>0.1	-0.0019	0.315			

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#21, UMR haze mode	8,9,10,19	<0.1	+0.109	0.0453
#12, NRL elect. class.	8,9,10,19	<0.1	+0.0932	0.0897
#14, DRI haze	8,9,10,19	<0.1	+0.221	0.190
#11, NRL haze	8,9,10,19	<0.1	+0.184	0.321

TABLE 1. Critical Supersaturation (S_C) of Monodisperse Aerosols. S_{CU} = Value from U. Katz Electrical Classifier.

which is about the same performance as we have reported in the literature (Alofs, *et al.*, 1979). Except for instruments #17 ($\bar{X} = -0.171$) and #10 ($\sigma = 0.315$), the other instruments (#18, #12, #15) gave performance of similar quality, with $|\bar{X}| < 0.06$ and $\sigma < 0.16$.

Now consider the bottom 4 rows of Table 1; that is, experiments with $S_C < 0.1$. A very good performance is indicated by our instrument in the haze mode ($\bar{X} = +0.109$, $\sigma = 0.0453$). The NRL electrical classifier showed somewhat higher σ (0.089) and the haze chambers of DRI and NRL showed considerably higher σ . These instruments (#14 and #11) both gave X \approx 0.4 for experiments #8 and #9, and |X| < .05 for experiments #10 and #19. These instruments also both use Royco Model 225 optical counters, which have a multiple value response function for water drops in the range 1 to 2 µm diameter (Cooke and Kerker, 1975). Experiments #8 and #9 give haze drops of about 2 µm diameter, which is within the ambiguous size range of the Royco and probably explains the decrease in sizing accuracy.

Table 2 shows another type of comparison. The concentration ratio (R = ours \div other) was computed for S = 0.03, 0.1, 0.3, and 1.0. The average (\overline{R}) and standard deviation (σ) of R were then computed over the available set of experiments for each instrument.

In Table 2, consider first the Hudson-Squires type CFD of DRI. At S = 0.3, \overline{R} = 1.04 and at S = 1, \overline{R} = 1.18. Thus our counts were 4-18% higher than theirs. This is pretty good agreement in our opinion. However, σ/\overline{R} = 0.275 at S = 0.3, which is definitely higher than we expected in view of the 1% agreement that Hudson and Squires (1976) obtained with a pair of their CFD's. At S = 1, the value of σ/\overline{R} is considerably lower (0.132), and is the lowest value of σ/\overline{R} in Table 2.

In the haze mode, our instrument compares well with the NRL haze chamber (\overline{R} = .974 at S = .03, \overline{R} = 1.067 at S = .1) but gives considerably higher concentration than the DRI haze chamber (\overline{R} = 2.2) and lower concentration than the NRL electrical classifier. The standard deviations at S = .03 and S = 0.1 are higher than in Table 1, but are still not too bad considering the problems involved in sizing water drops with optical counters.

		Average Ratio, R, Univ. of Mo. (#21) + Other			Relative Std. Deviation c/R				
OTHER INSTRUMENT	SUPERSATURATION (\$) EXPERIMENTS	.03	.1	. 3	1.	.03	.1	. 3	1.
#18, DRI-CFD Hudson-Squires type	5,8,9,10,11,12,13, 14,15,16,18,19,20, 22,23,26,27,28			1.04	1.18			0.275	0.132
#17, NRL Twomcy type	1,2,3,4,5,8,9,10, 11,12,14,15,18,19, 20,22,23,26,27			0.943	0.937			0.258	0.215
#10, SUNY static,scattering	5,8,9,10,12,13,14, 15,18,19,20,22,23, 26,27,28			0.885	1.27			0.221	0,200
#12, NRL clect. class.	1,2,4,5,8,9,10,13, 14,15,18,19,20,22, 23,24	0.678	0.632	0.899	0.915	0.302	0.418	0.284	0.209
#11, NRL, haze	1,2,3,6,8,9,10,11, 12,14,15,19,20,22, 23,24,26	0.974	1.067			0.600	0.382		
#14, DRI haze	8,9,10,11,12,14,15, 16,17,19,20,22,23, 24,26	2.17	2.18			0.722	0.333		

TABLE 2. Ratio, University of Missouri ÷ Other Instrument

3. ACKNOWLEDGEMENTS

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A CLOUD CONDENSATION NUCLEUS SPECTROMETER DESIGNED FOR AIRBORNE MEASUREMENTS

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Abstract

A portable, vertical plate cloud condensation nucleus spectrometer, suitable for mounting aboard a small aircraft, is described. This instrument, which incorporates several unique design features, is automated and can provide rapid, simultaneous measurements of the concentrations of cloud condensation nuclei at four supersaturations.

1. INTRODUCTION

In 1970, at the Second International Workshop on Condensation and Ice Nuclei, all but one of the cloud condensation nucleus (CCN) counters were horizontal static-diffusion chambers operating in a manner little changed from those used in the 1950's [1,2]. Only one of the static chambers was fully automated [3]. All of the counters had significant limitations in time resolution and all required an operating time of several minutes or more to measure the supersaturation spectrum. Recently, variations on the continuous flow CCN counters developed in the 1960's [4,5,6] have produced a number of automated and semi-automated CCN counters which use single particle optical counters as detectors. These new devices offer the potential for both significantly improved accuracy and time resolution in CCN measurements, although most have provided only limited improvements in quickly measuring the CCN activity spectrum.

The CCN spectrometer to be described in this paper is a simple extension of the continuous flow diffusion chamber (CFDC) operated in the vertical mode, much like the instruments described in [7], [8] and [9]. The principal new design features of our instrument are its rapid time response and ability to rapidly measure the CCN activity spectrum. Also, the instrument is small enough to be mounted and operated in a research aircraft of modest size.

2. PHYSICAL LAYOUT AND AIRFLOW

The activity spectrum of CCN in an air sample is measured rapidly by simultaneously operating the spectrometer at four supersaturations. The CCN spectrometer consists of four essentially independent CFDC's in a parallel array (Fig. 1). Each CFDC is 61 cm long, 10 cm wide, and consists of two vertical parallel plates 1 cm apart, maintained at different temperatures. Each CFDC has a single particle optical counter located at the center of its base.

The instrument operates in the following sequential manner:

i) An air sample is drawn through the duct at the top of the CFDC's.

ii) A fraction of the airflow exits the sample duct through a symmetrical, streamlined slit at the bottom of the duct. The rest of the sample flow is discarded.

iii) As the air enters the top of one of the CFDC's, it is sheathed with two temperatureconditioned, particle-free "curtains" of air (called the "sheath flows").

iv) The three airstreams travel in laminar flow, with negligible intermixing, down each of the CFDC's where, after 8 cm of travel, they encounter a saturated felt on the cold plate. At a further distance downstream, the warm plate is also covered with wet felt. This results in the airstream becoming supersaturated and CCN are activated to form droplets. The total length of felt on the



Figure 1. Schematic showing the airflow and optics of the University of Washington's CCN Spectrometer.

warm plate is roughly inversely proportional to the supersaturation in the chamber.

v) As the airstreams leave each of the CFDC's they are split along the vertical, central plane of the sample air, with half of the total flow being removed on each side at the base of the CFDC. A small fraction of the airflow is removed as the stream is divided by a small tube which extends up into the airflow. The droplets in this fraction of the airflow are subsequently counted by optical counters (see \$4).

vi) The position of the airstream with respect to the inlet to the optical counters can be precisely adjusted by changing the flow rates in the sheath flows; this allows the airflow to be exactly centered on the inlet to the optical counters.

vii) The two halves of the total airflow are then filtered, dried and reinserted on the back sides of the warm and cold plates. These two airstreams travel up the plates and are in thermal equilibrium with the plates; when they arrive at the top of the CFDC they form the warm and cold sheath flows.

3. SUPERSATURATION CONTROL

A novel method is used for providing four distinct temperature differences between the warm and cold plates of the four CFDC's. As shown in Fig. 2, a large aluminum plate is cooled nearly uniformly by a small mechanical refrigerator. The warm plate, which is made of copper, is connected to the cold plate by a brass end plate and at the opposite end it is heated by the hot refrigerator gases. The copper plate is vertically segmented between each CFDC with a thin insulator. The four copper segments stabilize at four distinct, and nearly uniform, temperatures. From measurements of these temperatures, and the temperature of the cold plate, the supersaturation in each of the CFDC's is computed by a microprocessor. Note that the supersaturations in all four CFDC's are adjusted by a single control. The four supersaturations normally achieved are approximately 0.2%, 0.5%, 1.0% and 1.5%.

4. OPTICAL DETECTION AND DATA PROCESSING

The CCN activated in the CFDC's are detected as droplets in an optical box (Fig. 2). The optical box contains a three mW He-Ne laser and four identical photo-detectors placed beneath each of the sample outlet tubes from the four CFDC's. The streams of droplets pass through the laser beam and are viewed by the photo-detectors at a forward scattering angle of $45^{\circ} \pm 5^{\circ}$. The photometers readily detect all droplets greater than 0.5 µm in diameter. The available growth times in the CFDC's (5-20s) are more than adequate to grow the activated cCN into droplets that are much larger than any unactivated haze droplets. Thus, by appropriate sensitivity adjustments, activated droplets are not.



Figure 2. Schematic of thermal and electrical components of the CCN Spectrometer.

During the CCN Workshop the range of supersaturations achieved was limited by a malfunction of the refrigerator unit.

A microprocessor is programmed to process the data in several ways. It can display the supersaturation and CCN concentrations measured in each of the four CFDC's on command from the aircraft's central computer. Alternatively, the microprocessor can delay output and accumulate data until certain statistical counting criteria are met.

5. CONCLUSIONS

We have designed and built a compact and reasonably simple CCN spectrometer for airborne use. In its first full-scale testing, at the Third International Measurement Workshop on CCN, it compared well with both conventional CCN counters and large continuous flow CCN counters designed for use on the ground. This instrument is currently undergoing final laboratory testing and modification before flight testing.

6. ACKNOWLEDGEMENTS

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STATIC DIFFUSION CLOUD CHAMBER

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1. CHAMBER GEOMETRY

The thermal diffusion chamber is based upon an earlier design of Twomey and Davidson, the cylindrical glass walls of which have been replaced by an annular perspex (plexiglass) spacer 12.5 mm in height, which is fitted with a very thin stainless steel liner that contacts both top and bottom plates so as to linearize the edge temperature gradient. Gagin and Terliuc (1968) used a somewhat thicker liner for this purpose.

The optical arrangement is similar to that used by Lala and Jiusto (1977). In the present design heated windows have been installed at appropriate places to allow a waisting light beam from a projector - lamp/lens combination to pass through the chamber. A window is placed at 45° with respect to the beam to view the forward scattered light with a photocell and lens. A further window at 90° to the beam allows the number of drops in the illuminated cloud to be recorded photographically.

Summary of Chamber Geometry:

Inside diameter 75 mm Height 12.5 mm Scattered light at 45° recorded with a peak detector Cloud droplets recorded photographically at 90° Sample volume 10 mm x 3 mm x 3 mm

2. SAMPLING SYSTEM

Air is drawn continuously through the equipment at about 1 - $1.25 \text{ dm}^3/\text{min}$ by means of a simple aerator (aquarium) pump. The flow bypasses the chamber until a measurement is required, at which point air is admitted by turning a four-way selector valve to the "sample" position. A two-way selector valve in the air circuit enables the sample air to be passed through an absolute filter so that periodic zero checks can be made.

Sampled air enters the chamber from six peripheral holes in the underside of the top plate after passing twice around the edge in a gallery to attain the same temperature as the top plate.

The chamber and associated air lines are sealed from the ambient atmosphere so that, if required, the equipment may be operated at pressures below ambient. A prototype of this design has been successfully operated in the Division's pressurized aircraft.

2.1 Supersaturation Range

Readings are taken at five fixed points: 0.25%, 0.5%, 0.75%, 1.0% and 1.25%.

2.2 Detection System

Light source: Sylvania EJV Projector Lamp

150W - 21V operated at approximately 19V, illuminating at a distance of 45 mm a 2.4 mm square of ground glass screen. Light from this screen is passed through infrared absorbing glass and focused by means of a Bell and Howell projector lens (Fl.2; focal length 51 mm) to give a beam waisting down to a 3 mm square section over a 15 mm length at the center of the diffusion chamber. The central 10 mm is used for droplet detection.

Camera: Canon AE-1 fitted with a Vivitar

55 mm Macro F2.8 lens, power-winder and data back. Film used is Tri-x (400 A.S.A.) processed for 600 A.S.A. Photographs are taken at 1/4 sec. and F4, though the effective aperture is much smaller (higher F number) since the camera views the chamber through a small window set in a short tube. Photographs of a mm graticule placed at 45° to the incident light beam and camera window confirmed both the beam cross sectional area and that depth of field was sufficient to encompass the 3 mm width of the beam.

Photocell detector: An EG and G Electro-Optics Silicon Photovoltaic Detector type PV-215 coupled to a two-stage amplifier having an output gain of 100V per micro-amp of cell current. Amplifier output is fed to a peak detector having a digital panelmeter readout. A manual reset is used. The photocell views the scattered light from virtually the same sample volume as the camera, but at an angle of 45°, and through an 8 mm focal length lens. A mask in front of the lens serves to define the viewed sample volume.

3. TEMPERATURE CONTROL

Temperature of the top and bottom plates is measured using semi-conductor transducers (National Semiconductor type LM3911 or similar). The transducers are coupled to amplifiers to give outputs of 100mV per K and have been set to within 1 or 2 hundredths of a K at ice point and at 293.15K. Digital panelmeters read the top plate temperature to 0.1K and the top-bottom temperature difference to 0.01K, though the bottom plate temperature varies by \pm .04K due to "hunting" of the controller. Thermocouples have been built into both plates to provide an independent check on the performance of the temperature measurement and control circuitry.

Bottom plate cooling is by means of a Komatsu thermoelectric cooling element, type KSF-2012. Simple circuitry uses the temperature transducer outputs to establish and maintain automatically the required ΔT , regardless of top plate temperature variations.

4. CALIBRATION

Calibration amounts to relating the peak value of scattered light intensity to the number of droplets in the sample volume at the time at which the peak occurred. In the present case, a chart recorder was used to register, simultaneously, variation in scattered light intensity and the time at which photographs of the scattering volume were taken (usually a series of 5-7 frames at 1 per second). It was then a simple matter to determine which frame corresponded most closely in time to the peak in the scattered light intensity. In this way, calibration curves were constructed for each supersaturation in the form of plots relating "film count (cm⁻³)" to "peak reading".

The simple analysis of a thermal gradient chamber discussed by Lala and Jiusto (1977) suggests the peak value in scattered light intensity is dependent on supersaturation and droplet concentration only and, in particular, that the number of droplets present in the sample volume at the time the peak is reached is only two-thirds of the initial nucleus concentration, independent of supersaturation. Thus our final estimate of CCN concentration is made by multiplying the "film count" by a factor of 1.5.

Calibration curves used at the workshop were derived from room air samples by blending varying amounts of filtered and unfiltered air to obtain a range of peak readings.

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AN AIRBORNE ISOTHERMAL HAZE CHAMBER

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1. INTRODUCTION

Thermal gradient diffusion cloud chambers (TGDCC) are used to determine the concentrations of cloud condensation nuclei (CCN) with critical supersaturations (S_C) greater than or equal to about 0.2% (Hoppel and Wojciechowski, 1976). The concentrations of CCN active at S_C \lesssim 0.2% can be determined using an isothermal haze chamber (IHC) following the work of Alofs (1978), Hoppel (1979) and Hudson (1980). They have shown that CCN spectra over a supersaturation range $0.01 \leq S_C \leq 1.0\%$ can be obtained from simultaneous measurements with a TGDCC and an IHC. Their instruments are designed to operate either in the laboratory or in the field.

The IHC discussed in this paper and operated at the CCN Workshop was designed for use in a light aircraft. It is based on the IHC design of Hudson (1980). The objectives of our participation in the Workshop were to (1) compare the response of the airborne IHC to the theoretical response (based on accurate sizes, numbers and compositions of aerosol particles), (2) compare the response of the IHC to the response of the larger laboratory IHC's, and (3) compare the response of the IHC to the response of the CSU-Mee TGDCC (Borys, 1980) in the region of overlapping supersaturations. The Mee instrument resembles that constructed by Lala and Jiusto (1977).

2. DESCRIPTION OF THE CHAMBER

Hudson (1980) provides a schematic of the IHC and the operating principles. Here we will provide details of the airborne version of Hudson's chamber which we have constructed. The line drawings for the chamber are given in Figure 1. The air flow and water flow through the chamber are given in Figure 2.

The supersaturation range of the instrument is determined by the size range of the solution droplets (in equilibrium at 100.0% RH) which the ROYCO sensor can detect. The equilibrium droplet size is related to its critical supersaturation as follows:

$$r_{100} = 4.1 \times 10^{-6} / S_c$$
 (1)

where r is in centimeters and S_C is in percent, after Laktionov (1972), Hoppel and Fitzgerald (1977), Alofs (1978) and Hudson (1980). The ROYCO sensor latex-sphere calibration was revised for water droplets by Fitzgerald (1980, private communication) following Cooke and Kerker (1975); see Figure 3. The sensor was adjusted such that the critical supersaturation thresholds corresponding to the droplet sizes from (1) were 0.15%, 0.11%, 0.041%, 0.027% and 0.016%, respectively.

The ROYCO Model 225 optical particle counter which is described by Liu, *et al.* (1974) was used to size and count the solution drops produced in the IHC. The instrument was operated at a 60 s sampling interval. Consequently, it took 1 minute for a CCN spectra to be obtained.

The maximum flow F to permit droplet equilibrium in the chamber was estimated following a procedure outlined by Hudson (1980). The results of the variations in particle concentrations as a function of F were similar to the results reported by Hudson. We found the maximum value of F should be \sim 35 cm³s⁻¹. Values of F greater than this value would provide insufficient time for droplets to reach equilibrium size before passing through the ROYCO sensor.

The volume sampled for a CCN spectra corresponds to the flow rate f through the sensor multiplied by the sample period: $\sim 1 \text{ cm}^3 \text{s}^{-1} \times 60 \text{ s} = 60 \text{ cm}^3$. This result is valid because the main flow F is essentially particle free.

There is no cooling or heating requirements for the chamber; the chamber is isothermal.



Figure 1. Isothermal haze chamber and main components: Aluminum cylinder with inside walls covered with Whatman 41 cellulose high-volume filter paper; plexiglass top into which flows particle-free air, sample air and water by gravity feed from the 500 ml reservoir; the ROYCO Model 225 sensor which sizes and counts the solution droplets which form in the chamber; the sump for receiving the water which drains from the filter paper.



Figure 2. Airflow and water flow schematic.



Figure 3. Calibration of ROYCO sensor for latex spheres. The calibration for pure water drops was from Fitzgerald (1980, private communication).

The minimum detectable drop size is 0.62 μm diameter (see Figure 3). This is the minimum size setting before electronic noise interferes.

3. RESULTS

During Experiments 1 through 13, the IHC was operated in the same fashion (f = 1.3 cm³s⁻¹, F = 35 cm³s⁻¹). It was clear from these early experiments that the concentrations of CCN detected by the airborne IHC were significantly below the concentrations detected by the laboratory IHC's (see the Table). For example, in the Table, the airborne IHC detected an average of 94 times fewer CCN active at S_C \leq 0.05% than did the laboratory IHC's.

One reason for the CCN concentration differences between the laboratory and airborne IHC's was thought to be a malfunctioning CSU ROYCO 225 sensor. The sensor was checked with the York University ROYCO 225 sensor and no significant difference was found between the dry particle size distributions measured by the two instruments. Further, during Experiments 14 and 15, the spare NRL ROYCO 225 sensor was connected to the CSU IHC and the CCN concentrations continued to be too low. Consequently, the CSU ROYCO 225 sensor appeared to be operating normally.

Another reason for the CCN concentration differences was thought to be a subsaturated chamber. Consequently, a humidifier (courtesy of Dr. Fred Rogers) was placed in the main-flow air line downstream of the filters during Experiment 17. The humidifier remained in the line from Experiment 18 to the end of the Workshop. The results in the Table show that the humidification reduced the differences in concentrations between the lab IHC's and the airborne IHC. For example, the factor of 94 difference mentioned earlier reduced to a factor of 7.5 following humidification. The Table presents a comparison of the results from theory and from the airborne IHC. Following humidification, the airborne IHC measured 9.0 times fewer CCN active at 0.05% and 7.9 times fewer CCN active at 0.14% than predicted to occur by theory.

Figure 4 illustrates the results obtained from the simultaneously operating CSU IHC and CSU TGDCC. Also plotted are the results obtained using measured particle sizes and the theoretical relationship between the dry $(NH_4)_2$ SO4 particle size and its critical supersaturation from Fitzgera¹d (1973):

$$r_{\rm d} = 1.53 \times 10^{-6} {\rm s_c}^{-2/3}$$
 (2)

where $r_{\rm d}$ is in centimeters and $S_{\rm C}$ is in percent. It can be seen from Figure 4 that the results from the two instruments tracked the theoretical response but with concentrations significantly less than the theoretical response. Further, the supersaturation ranges of the two instruments did not overlap because the lowest value of the TGDCC range was 0.2% as established by Borys (1980) and the smallest droplet detected by the ROYCO was 0.62 μm diameter which is equivalent to a $S_{\rm C}$ of 0.13%. Nevertheless, it can be seen the slopes of the curves in this region approximate the theoretical slope.

	Humidity?	Experiment	Theory*	NRL	UMR	DRI	CSU	Theory/CSU	Lab IHC's**/CSU
$N(S_{C} \le 0.05\%) \text{ cm}^{-3}$	No Humidity	8 9 10 11	300 600 66 -	120 270 30 90	100 250 17 26	80 200 8 -	1.5 5.0 0.09 1.0	200 120 733 - 351	67 48 203 58 94
	Humidity	19 20 22 23 24	110 0.4 5 1 10	- 2 1 0.5 5	65 2 1 1 7	25 2 1 0.5 4	4.0 0.1 1.4 0.15 4	$ \begin{array}{c c} 28 \\ 4 \\ 3.6 \\ 6.7 \\ 2.5 \end{array} 9.0 $	11 20 0.7 4.4 1.3
$N(S_{C} \le 0.14\%) cm^{-3}$	No Humidity	8 9 10 11	420 800 270 -	290 600 240 350	300 850 210 460	240 350 100 -	120 300 30 70	$\begin{array}{c} 3.5\\2.7\\9.0\\-\end{array}$ 4.1	2.3 2.0 6.1 5.8 4.1
	No Humidity	19 20 22 23 24	1000 140 150 28 400	- 190 100 17 170	1000 140 100 13 230	350 58 30 5 70	250 30 17 2 50	4.0 4.7 8.8 14 8.0 7.9	2.7 4.3 4.5 5.1 3.1 3.1

TABLE Selected CCN Concentration Measurements for S $_{\rm C}$ \leq 0.05 and \leq 0.14%

The dry-particle sizes and numbers were known and their S could be calculated using theoretical relationships developed by Fitzgerald (1973).

** The Lab IHC value is the average of the NRL, UMR and DRI measurements.


Figure 4. Results from the CSU JHC and the CSU TGDCC for Experiment No. 24 (left) and Experiment No. 22 (right). The particles in both cases were polydisperse $(NH_4)_2SO_4$.

4. DISCUSSION

How subsaturated must the airborne IHC have been to account for the differences between the theoretical results and the measurements? From Figure 4 it can be seen that if the S_C values were shifted from 0.1% to about 0.05% then the theoretical results and measurements would be identical. The r100 corresponding to these two S_C values are 0.41 μ m and 0.82 μ m, respectively, using (1). Consequently, we may have been measuring the concentrations of droplets with radii $\geq 0.41 \ \mu$ m at an unknown RH. The unknown RH can be estimated from a graphical representation (Figure 5) of the relationship between S_C , requilibrium and RH ($\leq 100.0\%$) provided by Hoppel (1980, private communication). As can be seen from Figure 5 $\geq 0.41 \ \mu$ m at 100.0% RH become droplets $\geq 0.41 \ \mu$ m at 20.0% RH, this subsaturation would explain all of the differences between theory and measurements.

How long a growth time is required for the largest solution drop measured $(r_{100} = 2.5 \ \mu\text{m})$ to closely approximate its equilibrium size? Hoppel (1980, private communication) has calculated the growth times to reach the critical radius (r_c) ; r_c is greater than r_{100} . The times are shown in Figure 6. Interpolating between his worst case values (condensation coef. = 0.036) the growth time to reach 2.5 μ m is about 280 s. The growth time available in the chamber is about 307 s (35 cm³s⁻¹/126 cm² x 83 cm). Consequently, sufficient time

exists for the largest solution drops to reach equilibrium. Insufficient growth times cannot explain the differences between theory and the measurements.

Could the water droplet calibration for the ROYCO sensor account for the differences between the theoretical results and the measurements? To do this, the droplets with diameters indicated to be 0.62, 0.82 and 2.0 μm , according to the calibration in Figure 3, would have had to indicate 1.13, 1.64 and 3.56 μm , respectively. These latter values are a dramatic departure from the calibration. The calibration works adequately with the NRL ROYCO 225 which is connected to their IHC. Consequently, it is believed the sensor calibration cannot account for all of the differences between theory and measurements.

The only particle losses in the system occur across the limiting-orifice (Figure 2). The losses were measured to be at most a factor of 1.3. This loss cannot account for much of the factors of 3.9 and 7.5 reported in the Table.

What is the possibility that droplets entering the ROYCO sensor at equilibrium size (RH = 100%) shrink in size by the time they pass through the optics? It can be seen from Figure 1 that the sensor inlet protrudes into the IHC. Therefore, the time for a particle to travel from the inlet to the optics is approximately 1.1 ms (7.9 x 10^{-3} x 5.1 cm/37 cm³s⁻¹). Since the sheath air entering the sensor was slightly warmer and dryer than air entering the sensor from the IHC, conditions



Figure 5. Relationship between n_{dry} , $n_{equilibrium}$, RH ($\leq 100.0\%$) and S_c for NaCl particles (close upproximation to (NH4)_2SO4 particles). From Hoppel (1980, private communication).

for droplet evaporation were present. Alofs (1978) calculated effects of droplet shrinkage on $S_{\rm C}$ for different droplet residence times. Applying our l.l ms time to his worst case situation (95% RH in the optics) it was found that the effect would be confined to $S_{\rm C} \geq 0.1\%$ and at most would change the inferred $S_{\rm C}$ of 0.16% to about 0.15%. Consequently, droplets will not shrink significantly due to the extremely short residence time.



Figure 6. Growth times for dry-particles to reach their critical sizes (r_c) . From Hoppel (1980, private communication).

Finally, errors in the theoretically derived CCN response should be negligible. Gerber, et al. (1977) demonstrated experimentally the soundness of the theoretical relationship between r_d and S_c given by (2). The same particle sizing and counting equipment used by Gerber, et al., was used by Hoppel during the CCN Workshop to provide the measurements for deriving the CCN concentrations.

5. CONCLUSIONS

The following conclusions can be made based on the results obtained from the CCN Workshop. The CCN concentrations measured with the airborne IHC were lower than theoretically predicted by factors ranging between 7.9 and 9.0. The CCN concentrations measured with the airborne IHC were lower than the concentrations measured with the larger laboratory IHC's by factors ranging between 3.9 and 7.5. The bounds of the supersaturation ranges of the airborne IHC ($S_C \leq 0.16\%$) and the CSU-Mee TGDCC

 $(S_C \geq 0.2\%)$ do not overlap. Nevertheless, the slopes of the interpolated data between the bounds agree favorably with the theoretical slopes. Slight subsaturations in the IHC plus uncertainties with the sensor calibration are the most probable causes for the discrepancies between the measured and predicted CCN concentrations.

6. ACKNOWLEDGEMENTS

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A DESCRIPTION OF THE UK METEOROLOGICAL OFFICE CCN COUNTER

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The Meteorological "Office CCN Counter is a modified Mee Industries Model 130 CCN Counter which was purchased in 1977.

1. THE CHAMBER

The chamber is a thermal gradient diffusion type with the upper plate at room temperature and the lower plate thermoelectrically cooled. The light from a 25W projector bulb is focused in a beam of approximate dimensions 1x3 mm in the central part of the chamber. The light scattered by drops is viewed in the forward direction (40°) by a microscope and a sensitive photodetector.

Experiments have been performed to relate the peak output from the photocell with the maximum number of drops in the beam recorded photographically. The microscope is used to check the calibration at low concentrations of CCN.

2. ANCILLARIES

Modifications to the original Mee design are as follows:

(a) In order to reduce the time necessary to obtain an activation spectrum, the thermal capacity of the cooling system has been reduced by mounting the thermoelectric cooler adjacent to the bottom plate. In order to reduce the temperature difference between the plates (ΔT) and hence the supersaturation, the current is reversed through the cooler, instead of a separate heater being employed.

(b) The temperature of the top and bottom plates is monitored using precision surface mounting thermistors (YSI 400) bonded directly to the back of the sintered bronze plate. Horizontal temperature gradients across the lower plate are minimized by placing a brass disk, into which water drainage channels have been machined, directly below the plate and on top of the Peltier cooler.

(c) The new control circuits enable a sampling sequence in which an activation spectrum consisting of CCN measurements at a predetermined number of supersaturations and over any desired range to be made. ΔT is stepped up and back down through this range automatically. There is also a circuit which monitors ΔT and prevents new air samples being taken until the desired level of temperature stability in the chamber has been reached. At present, this is set to $d(\Delta T)/dt \leq 0.01^{\circ}Cs^{-1}$. A typical portion of this sequence is shown in Figure 2 below.

(d) The chief use of the instrument is onboard the Meteorological Research Flight C-130 aircraft. To make the instrument suitable for this use, it has been repackaged and is powered by a 115V, 400 Hz aircraft supply. For use in groundbased field projects, data logging is performed by an HP9830 calculator.



Figure 1. A view of the chamber; the control electronics, power supplies, etc. are in a separate box, not shown).

Mee Industries Inc., Rosemead, CA 91770



Figure 2. A typical sampling sequence. The "display" is a digital display on the front panel and the "CCN concentration" is the peak photocell output after amplification. A four-point activation spectrum in the range of supersaturation 0-1.5% requires approximately 10 minutes to complete.

3. PERFORMANCE

From experience in the operation of the modified Mee counter at the CCN Workshop and during field projects, it has become clear that there are a number of deficiencies in the design which have yet to be corrected.

 (a) The low intensity light source results in a poor signal-to-noise ratio in the photodetector output.

(b) Calibration is difficult because photography of the drops growing in the chamber is hindered by the low illumination.

(c) The high fall speed of the drops viewed through the microscope is evidence of flow inside the chamber air sample. This raises doubts about whether a quasi-thermodynamic equilibrium is achieved.

(d) The sintered bronze plates are difficult to clean and may be subject to contaminants.

(e) Condensation occasionally occurs on the chamber windows, reducing the amount of scattered light focused onto the photodetector.

4. CONCLUSION

The modifications made to the Mee CCN counter have improved the temperature control and stability and also reduced the time taken to produce an activation spectrum. However, measurements of CCN concentration using this instrument have shown that its performance is still affected by design deficiencies.

5. ACKNOWLEDGEMENT

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CLOUD CONDENSATION NUCLEUS COUNTER BY IMPACTOR SAMPLING TECHNIQUE

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Unlike typical CCN counters, this device counts the numbers of water droplets condensed on aerosol particles sampled on a microcover glass at various different relative humidities. The relative humidities ranged from 75% to a calculated value of 110%. A schematic of the apparatus is shown in Figure 1. The individual CCN can be identified in an optical micrograph and scanning electron micrograph and may be inspected for their chemical composition later.

Sampling is made by means of an impactor and a vacuum pump of about 3 liters or less air volume onto an area of 3.35 mm by 0.25 mm of a microcover glass (0.2 mm thick). Then the cover glass is transferred to a small humidifying chamber where temperature is controlled by a thermoelectric cooler at the bottom of the chamber. The chamber has a closed glass top, and the circular side wall of the chamber has a blotting paper soaked with a saturated aqueous solution of sodium chloride, giving an equilibrium relative humidity of 75% in the temperature range between 20° and 70° C.

As the cover glass is cooled down by applying a direct current to the thermoelectric cooler, it is chilled to a lower temperature producing a relative humidity of 100%. Additional current produces relative humidities beyond water saturation. The exact saturation point is identified by the observation of dew on the cover glass which was previously coated with aluminum, using a vacuum evaporation to form a mirror. Simultaneously, the temperature difference between the cover glass and the air is observed by means of a thermojunction and a microvolt meter from the starting point of cooling, giving the "0" reading. Humidity values are determined from interpolation and extrapolation of the microvolt reading at the point dew forms. A usual microvolt reading at the dew point is about 150 microvolts for a 25% humidity difference. In this case, every 6 microvolt change indicates a 1% relative humidity change. Consequently, a 12 microvolt reading higher than that necessary to obtain the dew point gives a nominal 2% supersaturation. Due to various cooling water temperatures, the reading at the dew point has been found to change slightly.

The water drops condensed are photographed for counting by a Polaroid micrograph camera or a normal microscope camera. Since the sizes of the water drops on a cover glass are larger than 5 microns, the resolution of an optical microscope is adequate.

The greatest advantage of this technique is to allow the inspection of sizes and chemical composition of the individual nuclei by an electron microscope combined with an X-ray energy spectrometer with a specimen cooling device to prevent possible heating on volatile nuclei. Such analysis is in progress for the aerosols sampled over the Arctic Ocean. The second advantage is that the aerosols sampled can be stored for a long time and their condensation ability can be examined later.



Figure 1

However, as it has probably been seen, this technique may not give an accurate absolute concentration of cloud nuclei. The poor collection efficiency on the sampler may limit the collection of aerosol particles smaller than 0.1 microns diameter, although this may be improved by use of a better sampling technique (possibly by a microorifice impactor). If the vacuum or air speed at the nozzle is reduced, small particles will be missing from the cover glass. Thermal precipitator will result in the particles being evaporated. Cooling thermal precipitator using liquid nitrogen will result in frosting on the substrate.

Another problem is the humidity value on a substrate where the aerosols were sampled. Water vapor molecules tend to diffuse to the particles and some condensed water drops will restrict many adjacent aerosols from condensing to water drops as pointed out by Lala and Jiusto (1972) for the case of ice nucleation. Effective relative humidity on substrate may therefore never reach water saturation. For CCN counts at room temperatures, humidity reaches saturation values confirming by observations of visible dew on a very clean mirror. However, the actual degree of supersaturation may not exceed by more than 0.1%, even though the calculated value of supersaturation may be 2%. The real supersaturation degrees for calculated humidity values remain unknown. In order to break the microscale boundary layer over hygroscopic particles, circulation of humidity-controlled air should be adequate. Preliminary experiments have confirmed this procedure to be successful.

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DIFFUSION TUBE

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1. EQUIPMENT DESCRIPTION

The diffusion tube is designed to operate below about 0.25% of water supersaturation. It is simply a long tube lined on the inside with a damp chamois cloth, and heated isothermally to a few degrees centigrade above the incoming air.

The diffusion coefficient for water vapour is slightly larger than that for heat, making it possible to supersaturate the airflow. This is the same principle by which transient supersaturations may occur in parallel plate cloud chambers. Our elementary analysis considers only the diffusion of vapour and heat from the walls into the moving air.

The droplet sampling tube which inserts into the main diffusion tube draws the central 4 or 5% of the total volume from the tube. It is only necessary then to compute the supersaturation for the air stream in the centre of the tube. A typical supersaturation profile is shown in Figure 1 (solid line). It is dependent upon the relative humidity and temperature of the incoming air and the temperature of the tube walls. The abscissa represents the tube length divided by the total flow rate. The cutoff is chosen at 2.0 to permit reasonably isokinetic droplet sampling and because the relative humidity is reduced to nearly 100%.



Figure 1. Relative humidity profile, average supersaturation = 0.11%. Z = tube length = 87 cm.



Figure 2. Numerical growth study for profile shown in Fig. 1. Dry distribution is nearly monodispense.

TABLE 1

Channel Used Dowr to S.S.(%	(NH ₄) ₂ SO ₄ D100 Value Cr.S.S.(%)	NaCl D100 Value Cr.S.S.(%)	Channel Threshold Dia. (µm)	Royco Channel No.
0.08	0.088	0.066	1.25	1
0.07	0.073	0.055	1.5	2
0.05	0.056	0.038	2.0	3
0.035	0.037	0.025	2.5	4
0.02	0.024	0.016	3.0	5

In order to define a single supersaturation from this transient, we average the water in excess of 100% relative humidity over the period from the point of reaching 100% to Z/flow equal to 2.0. This we define as the operating supersaturation. The dashed line in Figure 1 indicates this average for this curve. The validity of this approach has been investigated numerically. A dry salt distribution was grown along both paths in Figure 1. The resulting distributions are given in Figure 2. Studies have indicated agreement between droplet distributions is good below 0.1% supersaturation and adequate up to 0.2% at least.

The most serious difficulty occurs with the interpretation of the CCN. Since the available growth time (about 10-12 seconds) is short, for small S.S.'s, the droplets are small (1-4 μ m dia.) and confusion may result with inactivated haze droplets, whose DIOO values are greater than our threshold. During the Workshop, only one channel on our Royco O.P.C. was used to count the droplets. The threshold level was set at 1.5 µm dia.; as a result some haze particles were counted at supersaturations below about 0.06%. This was noted in Experiment 22 and particularly during our second run in Experiment 26. We have tried to overcome this by evaluating the CCN according to Table 1. The CCN are determined from the counts in Channel 1 down to an operating supersaturation of about 0.08%. Channel 2 is then used to 0.07%, etc.

Experiments with monodisperse aerosols generated with our own classifier have suggested the following:

1. Very good resolution is possible at low supersaturations, down to at least 0.05%.

2. Our estimation of the operating supersaturation is high by about 30% of our value.

3. Results for similar conditions are very reproducible.

The combination of 1 and 3 are encouraging, but work must be continued to assess the proper supersaturations. The high values are thought to be attributable to free convective activity and a slight temperature gradient between the wall temperature sensor and the actual wet surface. Both are being investigated. We expect that at higher supersaturations the free convection will be a greater problem because of larger temperature gradients involved. This is in agreement with a general decrease in accuracy with increasing supersaturation noted during our classifier experiments. The problem may be somewhat alleviated by beginning with higher initial humidity levels.

The computed operating supersaturations for different wall temperatures and humidity conditions, with an incoming air temperature of 20.0° C, are plotted in Figure 3. It is quite obvious the supersaturation is dependent equally upon the initial relative humidity as well as temperature differences between the wall and air. The dependence, on both parameters, lessens towards the lower supersaturations thus indicating better accuracy, in terms of temperature and humidity measurements, for the small values.



Figure 3. Operating supersaturations as a function of the temperature difference between the wall and ambient, for different incoming relative humidities.

2. OPERATION

The total flow rate is 2.55 L/m; there are two input flows. Humidified filtered air is mixed 3 or 4 to 1 with the air to be measured. A Royco 225 samples about 250 cc/m from the centre of the air stream (0.2 of the radius). The sampling tube extends 12 cm into the tube bottom (see Figure 4), with the remaining flow (about 2.3 L/m) being drawn out by two tubes at the very bottom. The residence time in the sampling tube is no more than 0.1 seconds.

The relative humidity of the incoming air is kept between 65 and 80%. The water bath is heated to about 8 or 9° C above the ambient. Measurements are taken in between adding cool water to lower the wall temperature. Initially it was hoped to vary the incoming humidity while keeping the temperature gradients constant but we encountered some difficulty with the response time of the humidity sensor to large changes.

The temperatures are recorded with YSI precision thermistors $(\pm 0.1^{\circ}C)$. There are six located in the tube walls. The measured temperature appears stable and quite isothermal. The humidity is measured with dry and wet bulb thermistors. The wet bulb thermistor is encased in a cotton wick and thoroughly wetted.



Figure 4. Schematic, in section, of diffusion tube. Aluminum tube is about 100 cm in length, plastic entrance tube is 30 cm long, and sampling tube is 16.5 cm.

AN AUTOMATIC LIGHT SCATTERING CCN COUNTER

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1. PRINCIPLE OF OPERATION

The ASRC cloud condensation nucleus counter is a static thermal diffusion chamber which has been modified to include an optical system for the determination of droplet concentration by the measurement of scattered light. The principle of operation is the same as that described by Lala and Jiusto (1977). The determination of concentration is made by measurement of the peak scattered light signal from the cloud of growing droplets which is a function of both the droplet concentration and chamber supersaturation. Because the formation of the peak is related to the rate of growth of the droplets and sedimentation, both of which are determined by supersaturation, the system calibration can be uniquely determined by comparison with an absolute counter such as a static diffusion chamber This apwith a photographic recording system. proach to the measurement of droplet concentration in the cloud chamber has made possible the design of a compact system with low power requirements which can be operated automatically under elec-tronic control to provide real time measurement of cloud nucleus concentration.

CCN SYSTEM COMPONENTS

2.1 Diffusion Chamber

The cloud chamber is a cylindrical volume bounded by temperature controlled wet plates at the top and bottom of a cylindrical plastic wall. The chamber diameter is 7.6 cm with a plate separation of 1.0 cm with a volume of 45.6 cm^3 .

The temperature of the top and bottom surface of the chamber is measured by means of integrated circuit transistor thermometers (Analog Devices, Inc. AD590) which have a high sensitivity with excel-lent linearity and long term stability. The sensors are embedded in the center of the aluminum plates approximately 0.1 cm below the surface. The difference in the plate temperature, as sensed by the transistor thermometer, is used as the feedback signal for a closed loop temperature regulator which maintains the temperature difference at a fixed reference level by controlling the current to a thermoelectric cooler attached to the lower plate. This system has been designed to have a long term stability of $\pm 0.02^{\circ}$ C, while having the capability of rapidly changing the plate temperature by heating or cooling (8 sec to stabilize to within ±0.05°C for a 1°C change in plate temperature).

The moisture supply for both the top and bottom surface is provided by water saturated blotter paper. Complete saturation of the upper surface is insured by a connection to an external reservoir through holes in the plate with capillary forces providing the flow of water required. A water supply is not provided for the lower surface because once wet, the normal operation of the chamber maintains saturation of the blotter by the vertical diffusion of water vapor from the upper surface.

2.2 Illumination System

The original design of the system described by Lala and Jiusto (1977) used polychromatic illumination from a tungsten halogen lamp. At that time, it was thought that broadband illumination was necessary to insure the smoothing of the Mie scattering peaks required to produce a scattered light signal which was monotonically increasing with droplet Subsequent calculations of the scattering of size. monochromatic light has shown that the angular integration over the scattering volume and collection aperture provides sufficient smoothing of the scattering peaks. The present design, taking advantage of this, uses monochromatic illumination from an infrared-emitting diode (General Electric F5D1) with a nominal emission wavelength of 880 mm. In order to provide sufficient illumination, the diode is operated in the pulsed mode at 6.25 kHz. The diverging illumination from the diode is collimated by a simple lens to form a cylindrical beam with a diameter of 0.4 cm. The principal advantages of using the infrared diode are that its emission wavelength matches the peak sensitivity of the scattered light detector, the power requirements are small in comparison to the tungsten lamp, and the long lifetime of the diode.

2.3 Scattered Light Detection System

The scattered light detection system is identical to that used in the previous design. A lens located at a 45 degree scattering angle forms an image of the cloud on a slit which serves to eliminate stray light to define the length of the scattering volume (1.5 cm). Directly behind the slit is a hybrid photodetector-amplifier (Bell and Howell 529) which converts the scattered light to a voltage signal. The low level light signal is amplified and converted to a D.C. voltage signal by means of a synchronous detector. The overall system gain produces a sensitivity of 500 μ v/droplet at a supersaturation of 0.5%.

2.4 Measurement and Control System

All data handling and timing requirements of the system are handled by a single board microcomputer (Rockwell International AIM65). Communication between the computer system and the temperature control and light scattering system is performed by analog to digital and digital to analog converters. By means of this interface, the computer is able to control all of the system functions, as well as the processing and display of the data. All user control of the system is handled through computer software, making it possible to alter the system operating parameters without having to configure new control circuits for special applications. The use of computer control allows for the optimization of the overall system performance, as well as complete flexibility in the application of the system for CCN measurements.

SYSTEM OPERATION

A measurement cycle can be initiated either by the computer or by an input from the operator. After receiving the start command, the computer determines the chamber top plate temperature and calculates the necessary temperature difference required to produce the desired supersaturation. Βv means of the digital to analog converter, the com-puter establishes the temperature reference and waits for the actual temperature difference to settle to within ±0.05°C for two seconds before proceeding. After temperature stabilization, the computer opens the sample valves and purges the chamber by means of a small pump. During the last second of the six second sample cycle, an average value of the background signal is determined. The sample cycle is terminated by stopping the pump and closing the valve after a one-half second delay. During the period which follows, the computer con-tinuously samples the scattered light signal and waits for the occurrence of a peak defined by the signal falling below 80% of the highest previous reading. At this time, the peak value minus the background signal is converted to a concentration and displayed along with the chamber operating condition. The complete cycle is completed in 20 to 45 seconds, depending on the supersaturation used.

SYSTEM CALIBRATION

The calibration of the system is obtained by direct comparison with a second thermal gradient diffusion chamber equipped for photographic recording. The chamber and temperature control system for the photographic unit are identical to the automatic system. The illumination for the photographic system is provided by a 100 watt mercury



Fig. 1

arc lamp and a large aperture lens system which forms a rectangular beam with an 0.2 cm by 0.4 cm cross section. Droplets in the illuminated volume are photographed at right angles to the beam with a modified oscilloscope camera.

The calibration procedure consists of taking a 200 liter sample of ambient aerosol which is used as a source for both systems during simultaneous measurements. Concentrations are adjusted by diluting the sample with clean air and repeating the measurement after stabilization. Figure 1 shows a plot of the scattered light signal against the concentration determined by the photographic system at a supersaturation of 1%. This procedure is repeated at five supersaturations between 0.25% and 1% which is used to determine the system calibration.

5. ACKNOWLEDGEMENTS

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DESCRIPTION OF THE NRL ISOTHERMAL HAZE CHAMBER

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The background and rationale for using the isothermal haze chamber (IHC) concept to extend the range of CCN measurements to lower supersaturations than is possible with the thermal gradient diffusion chamber is given in the overview section. In the IHC the critical supersaturation is inferred from the measurement of the size of particles which have grown to their equilibrium size at exactly 100% RH. Here the Naval Research Laboratory (NRL) isothermal haze chamber will be described, and the reason for its unique design will be explained.

There are three major design constraints which had to be met:

l. Calculations of the equilibrium size as a function of relative humidity show that the humidity in the IHC must be within about 0.01% of saturation if the inferred critical supersaturation is not to be significantly in error.

2. The smaller the critical supersaturation of the particle, the larger is its equilibrium size. The larger its equilibrium size, the longer it takes to reach that size. Over 100 sec of growth time is needed for a particle with a critical supersaturation of about 0.015% to reach 95% of its equilibrium size at saturation. Much longer times are necessary for particles with smaller critical supersaturation. The second design criteria was that the chamber must provide over 100 sec of residence time at high humidity. It is this limitation which places the lower limit of about 0.015% on the critical supersaturation of our IHC.

3. The commercial optical counter (Royco) which was available has a sample flow rate of 47 cc per sec. Our decision to stick with the manufacturer's flow rate determined the third major design constraint.

The NRL IHC has cylindrical symmetry as is shown schematically in Figure 1. The design is a two-tube design. The sample first enters a small tube lined with ceramic saturated with water. The pressure drop across the small inlet orifice is less than a quarter of an inch of water and this pressure drop is used to monitor the sample flow rate. The sample flow rate which is about 3 cc per sec was recalibrated at the altitude of the Workshop prior to its start. Filtered air enters through four ports on the outer circumference of a humidifier which consists of passages between concentric cylinders lined with saturated, sponge-like material (see Figure 1). The filtered air, after leaving the humidifier, flows downward through the annular ceramic-lined passage between two tubes and forms a sheath flow around the sample flow beyond the point where the sample emerges from the small tube. At the bottom of the IHC the sample, together with sheath flow (47 cc per sec), enters the optical particle counter where the humidified particles are sized into five size channels between about 0.25 and $3 \mu m$ radius.

The distance required for air passing through a wet-walled tube to reach saturation depends only on the volume flow rate and length of tube and not on the radius of the tube. The reason for using the small tube for initial humidification of the sample is to bring the sample to high humidity in the shortest distance possible. This can best be accomplished by keeping the sample flow, which is very small, isolated from the larger sheath flow until both are fully humidified. The small tube in our IHC is 45 cm long, has a diameter of 1.27 cm and a volume flow rate of 3 cc per sec. Solutions to the diffusion equation show that a sample entering at 50% RH is within 0.004% of saturation by the time it has passed 10 cm down the tube. The residence time in the small tube is about 20 sec.

When the sample joins the sheath air, there is an increase in the cross-sectional area of about 36 and an increase in flow of about 15, resulting in a net increase in residence time per unit length of 2.3. The total residence time in the large tube is about 90 sec, giving a total residence time for growth at high relative humidity of about 110 sec.



Figure 1. NRL Isothermal Haze Chamber

The value of the filtered sheath flow is multiple: (a) the filtered sheath flow helps prevent gravitational fallout by confining the sample to the central region of the chamber; (b) sheath air is necessary for dilution. The large number of small particles which swell to optically detectable sizes at high humidities will swamp the optical detection system and cause coincidence counting problems if the sample is not diluted with sheath air; and (c) the total air flow required by the optical particle counter is supplied with the minimum overall length by introducing the humidified sheath air after the sample has already reached high humidity.

During a measurement period, the optical counter was on a one-minute counting cycle controlled by its internal timer. However, resetting, starting and reading the BCD output was accomplished with the same HP 9825 computer used to control the NRL mobility analyzer described elsewhere. During each size distribution taken with the mobility analyzer, 12 individual one-minute readings of the NRL IHC were taken and printed out and, at the end of the size distribution, the average of the 12 readings for each size channel was printed out. For most experiments, it was the average of the 12 one-minute samples that was supplied to the Workshop.

The factory calibration of the optical counter was done with latex spheres which have a much different index of refraction than does the water droplets. A size calibration for water droplets was estimated from the factory calibration and the work of Cook and Kerker (1975).

It is our opinion that the NRL IHC functioned well throughout the Workshop with only a minor problem created by the failure of one of the metering pumps which wet the ceramic. This failure required periodic, partial disassembly and manual wetting on a schedule less frequent than would have been done if the pump had been functioning.

The absolute accuracy of the measurement is difficult to assess. The biggest source of possible error in our system is thought to be the stability and accuracy of the calibration of the optical particle size spectrometer.

The largest size channel corresponds to a critical supersaturation of 0.014%. The growth time required for particles of this size to reach their equilibrium size exceeds the 110 second residence time in our IHC. Since the supersaturation spectrum is always very steep in this region, the contribution of these larger particles which have not yet attained their equilibrium size to smaller size channels is negligible. However, failure of these particles to reach their equilibrium size could result in a significant lowering of the count in the size channel corresponding to the smallest critical supersaturation.

Acknowledgements

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MODIFIED MEE INDUSTRIES STATIC THERMAL GRADIENT DIFFUSION CLOUD CHAMBER

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1. EQUIPMENT DESCRIPTION

The purpose of attending the CCN Workshop was to calibrate and determine the usable operating range of the cloud chamber. The instrument consisted of a Mee Model 130 Cloud Condensation Nucleus Counter (Serial #3). The original optical bench, light source, and detection components were removed and replaced with a simple laser illumination-photographic counting system. The purpose of the modification was to permit discrete droplet counting rather than the original method of scattered light detection and to extend the minimum concentration detection limit to approximately $10/cm^3$ for use in remote areas where aerosol concentrations were expected to be low. The system consisted of a 5 mw HeNe laser with an operating wavelength of 0.65 microns. The laser beam was incident to the camera field of view at a 40° angle, the same angle used in the original instrument configuration utilizing a photo diode as a scattered light detector. In both arrangements, the detector, in this case the camera, sensed the forward scattered light at the given angle from the center of the parallel plate thermal gradient diffusion chamber. The film used was Kodak 2475 recording film developed at 3000 ASA. Photographs were made at 1/2 second exposure through a microscope attachment to the chamber producing a magnification of $3.47 \times \text{on the}$ film negative. The observed volume was 0.011 cm³. Temperature and *AT* were controlled by thermoelectric cooling of the bottom plate with the upper plate at ambient temperature.

The operation procedure consists of flushing the chamber for 10 seconds at a sample flow rate of 4 liters per minute. An observation is then made through the eyepiece to determine the time required for maximum droplet number concentration to occur. This time is then used for subsequent measurements at the same selected chamber conditions. One to 30 exposures are made on each frame to obtain 50-100 droplet images per frame, depending upon the CCN This produces uncertainties of 10concentration. 15% in the CCN concentration determination. At the upper limit of 30 exposures used, concentrations of 10 cm^{-3} can be determined with uncertainties of 50%. The time to determine a four-point CCN spectrum of low concentrations (<500 cm⁻³ at 0.5% SS) is 90 minutes. At high concentrations the time reduces to about 30 minutes. Since the droplet counting is from images of droplets on film, the minimum detectable concentration and uncertainty can be selected by the operator with the subsequent loss or gain of time resolution. The instrument in its present form can only be operated manually.

2. RESULTS OF THE WORKSHOP

The experiments performed at the Workshop were undertaken to confirm the calculated sample volume, determine the usable range of supersaturation, and minimum detectable size.

Problems with chamber leaks were discovered for the experiments 0-10. Hence the data from these experiments cannot be used for comparison. Comparison of absolute CCN concentrations with other state-of-the-art continuous flow diffusion chambers at supersaturations near 1% indicated that the volume used produced CCN concentrations well within the range of CCN concentrations determined at the Workshop. This agreement is interpreted to mean the sample volume was correct. Direct measurements of the beam geometry done in the laboratory indi-cated a factor of 4-5 error. This error is apparently due to the larger apparent visible beam diameter versus the actual usable beam diameter given the droplet illumination, chamber optical geometry, microscope optics and film characteristics. It is therefore suggested that the only way in which the sample volume should be determined is by experimentation with known droplet sizes and numbers via known aerosol generation and sampling.

By comparison of the results from experiments 19 and 15, it was determined that the minimum usable supersaturation for this chamber was between 0.09% and 0.22%, respectively. By using the dry aerosol data from experiments 15, 18, 19 and 20, and the number of droplets detected at indicated supersaturation of 0.1%, the minimum supersaturation the instrument was able to achieve was 0.11%, 0.17%, 0.09%, and 0.19%, respectively. Therefore, 0.20% was determined as the lowest usable supersaturation.

A minimum detectable size of about one micron radius was determined, based upon the smallest observed image on the film, the known magnification of the image by the microscope attachment, and the artificial enlarging of the image by diffraction of the relatively long wavelength monochromatic light through the small diameter lenses of the microscope. This agrees roughly with a droplet size of 0.7 microns radius at which the maximum in light scattering efficiency occurs for the wavelength light used.

The chamber response to known aerosol at different supersaturations compared favorably to theory, both in the predicted slope for polydisperse aerosol and the determination of the critical supersaturation for monodisperse aerosol. However, there appears to be some tendency to undercount at mid-range supersaturations giving an anomalous concave upward appearance to the curve for some experiments, but not all. The cause of this effect is unknown. Further testing will be required to look more closely at this effect.

TABLE 1. INSTRUMENT PARAMETERS

Detection System:

Laser light source	5 milliwatt
	0.65 micron wavelength
Photographic counting	2475 recording film
	ASA 3000
Minimum detection size	∿l µm radius

Sampling:

Flow rate	4 lpm for 10 sec
Sample volume	0.011 cm ³
Time for spectra	30-90 min
(4 pts)	50 50 mm

Chamber Specifications:

Aspect ratio	4.6:1
Temperature control	Thermoelectric cooling
	of lower plate
Usable supersatura-	0.2 - 2.0%
tion range	

SIMULTANEOUS OPERATION OF THREE CCN COUNTERS AND AN ISOTHERMAL HAZE CHAMBER AT THE 1980 INTERNATIONAL CCN WORKSHOP

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1. INTRODUCTION

The Desert Research Institute (DRI) operated four devices for the detection and characterization of cloud condensation nuclei (CCN) and fog condensation nuclei (FCN) during the International CCN Workshop, Reno, Nevada, October 6-17, 1980. In chronological order of development, the CCN devices are the conventional continuous-flow diffusion chamber (CFD), the rapid-cycle CCN spectrometer, and the instantaneous CCN spectrometer. The FCN detection device is an isothermal haze chamber (IHC). These four instruments will be discussed in sequence, and general comments on their performance at the Workshop will be given.

2. CONVENTIONAL CONTINUOUS-FLOW DIFFUSION CHAMBER

2.1 Instrument Description

One of the DRI Continuous Flow Diffusion (CFD) Chambers (Hudson and Squires, 1973, 1976) was used in this Workshop. Since this instrument is largely unchanged since 1976, only the more important features will be described here, along with the values of some of the operating parameters used during the Workshop.

The CFD was operated in the vertical mode; that is, the parallel plates were vertical although the sample traveled in a horizontal direction. The plate separation was 1.3 cm while the width of the plates (this is actually a vertical distance when the chamber is operated in the vertical mode) is 30 cm. The sample enters the working volume of the chamber through a slit in a transverse tube which is at one end of the chamber midway between the plates. It is confined to a 1 mm thick zone about the central plane by the two particle-free air flows which go around each plate. The sample slit is about one-third of the chamber width so that in the other dimension, the sample comes no closer than 10 cm from the "side" walls of the chamber. This allows a minimum aspect ratio of 8. The length of the plates is 40 cm which is the total distance the sample travels through the instrument while the length of the wet paper on the cold plate is 38 cm which allows a 2 cm dry space for the sample after its introduction. The wet paper on the warm plate does not begin for another 10 cm as it is 28 cm long.

2.2 Operation

The main flow of particle-free air enters the chamber at two ports on the backsides of the diffusion plates. These flows then pass over the plates and then turn 180° at their ends where these two flows merge with each other and with the sample flow. By this time, the air should have attained a

reasonably complete temperature equilibrium with each plate. The total flow of air then moves between the plates to the end of the chamber where all of the air is pulled through an optical particle counter (OPC). It is actually the pump in the OPC which produces the air flow through the chamber and the slight underpressure ($\sim 8"$ of H_2O) within the chamber. A valve on the flowmeter which measures the main flow, F, can be adjusted to change F which, during the Workshop, varied from 20 to 58 cm³/sec. Thus the velocity in the central plane of the chamber was v = 3/2 F/A, where A = 39 cm² is the cross-sectional area. Thus, v ranged from 0.77 to 2.23 cm³/sec so that the time which the sample spent in the supersaturated volume of the chamber varied between approximately 12 and 36 seconds. The lower flow rates (longer times) were generally used for the lower supersaturations where longer growth times are necessary while faster flows or shorter times were used for the higher supersaturations where fallout could be a problem.

The sample flow rate is controlled by a glass capillary tube which restricts the flow to specific values depending on the pressure difference across the tube. The rate of flow through a given capillary was calibrated for the pressure differences to be used while it is in operation. For most of the Workshop experiments, the sample flow rate was about 0.6 $\rm cm^3 sec^{-1}$ although it varied from 0.1 $\rm cm^3 sec^{-1}$ to 1 $\rm cm^3 sec^{-1}$ during the course of the Workshop.

Supersaturations in this chamber during the Workshop ranged from 0.02% to 1.5%. Although supersaturations below 0.1% are normally beyond the useful range of the instrument, certain monodisperse aerosols may yield useful results at such low supersaturations. The time required to complete a spectrum depends on how many supersaturations are used. Although a smaller number of steps generally requires less time, if more steps are used, the time required to change plate temperatures is reduced since the steps are smaller. Typically, it takes at least five minutes to change plate temperatures and at least three minutes to acquire useful data at a given supersaturation.

The drops exit the chamber along with the total air flow. All of this passes through the active volume of the Royco 225 optical particle counter (OPC). In this instrument, the drops (or particles) are counted individually as they scatter light when they pass through an incandescent light beam. The scattered light is gathered by a PM tube which produces a signal which is roughly proportional to the amount of light scattered which is somewhat proportional to the particle size. Since the light beam is larger than the cross-sectional area of the particle stream, all particles which enter



Figure 1. Physical Schematic of the Rapid Cycle CCN Spectrometer.

Dimensions:

Legend

- 1. Carrier flow entrance
- 2. Entrance manifold
- 3. Cold plate wicking surface
- 4. Diffuser screen
- 5. Cold thermal plate
- 6. Sample injection tube
- 11. Sheath flow entrance
 12. OPC exhaust
 L = 44.45 cm
 L1 = 10 cm

7.

8.

9.

10.

Plate spacing = 1.6 cm

Width = 29 cm

the OPC scatter light. The chances of more than one particle being within the beam simultaneously are insignificant unless the OPC count is more than 100,000/min so the drops are usually counted and sized individually. When count rates were higher than this, lower sample flows were used.

The plate temperatures are controlled by circulating fluid from constant temperature baths. Plate temperatures are monitored by stainless steel thermisters which are imbedded in the plates just below the wet filter paper on its surface.

This instrument participated in and yielded valid data for all of the Workshop experiments.

3. RAPID CYCLE CCN SPECTROMETER

This instrument was built as a prototype for the NASA low gravity Atmospheric Cloud Physics Laboratory (ACPL). Rapid changes in supersaturation are accomplished by injecting a surge of fluid into the temperature-controlled plates. This displacing fluid is at a temperature different from the original plate temperatures. Thus the change in temperature of the plates is accomplished by proper mixing of an appropriate amount of fluid from a reservoir at an extreme temperature. A hot and a cold reservoir are on hand for this purpose. Microprocessor control is used to inject the fluid into the plates so that a smooth cycle of supersaturations can be obtained. Thus four or five supersaturations can be examined within a period of 5 to 10 minutes, depending upon the desired accuracy and the particle concentration. This same period of time is required just to make one change in supersaturation with the conventional CFD's.

Warm thermal plate

Exhaust manifold

Warm plate wicking surface

Optical particle counter

Figure 1 is a schematic diagram of the rapid cycle CCN spectrometer. It is quite similar even in dimensions to the earlier versions of the CFD's. A notable exception to this is the elimination of the flow around the back sides of each plate. In this version, the carrier flow enters the chamber at the opposite end of the chamber from the optical bench (Royco 225) unlike the previous CFD's, where the carrier flow entered at the optical bench end of the chamber behind both plates only to flow toward the other end of the chamber around the plates and back between the plates. The laminar flow so accomplished is obtained in the spectrometer by the use of a diffuser screen. A stainless steel wicking surface is used instead of filter paper as the wet surface on the inside of the plates.

A hypothetical experiment is shown in Figure 2. Each supersaturation can be maintained as long as desired; each subsequent supersaturation can be established and stable within about 30 seconds.

Each thermal plate is accurately controlled in temperature such that the temperature difference between the two plates is stable and known to



Figure 2. Temperature control of the thermal plates of the rapid cycle CCN spectrometer is programmed to follow the sequence shown below.

Event Sequence:

- Stabilization: ∿15-20 min is required to allow the CFD to become equilibrated at the initial temperature settings.
- 2. "LWAIT" Period: The experiment is initiated by a manual command at the start of "LWAIT". This period is long enough to allow sufficient time for counting particles in the OPC.
- 3. "TSTEP": Counting stops at the end of the "LWAJT" period. TSTEP is a temp. parameter

within about ±0.01°C for each supersaturation. Figure 3 is a schematic of the hydraulic circuit for the warm thermal plate. A centrifugal pump circulates water through the thermal plate and a housing containing four electrically resistive immersion heaters in a coolant flow of about 60 cm³/sec The coolant is also circulated over a small glass bead thermistor just downstream of the pump; this method of temperature measurement provides an accurate estimate of the average temperature because of the thorough mixing present at the pump exit. The thermal plate consists of a covered channeled metal surface with large distribution manifolds at both ends. The large manifolds assure that flow through each of the 18 water channels is uniform. Placing the exit and entrance ports on the same side of the thermal plate assures kinematic mixing of the water in the primary circuit. The time required for a parcel of water to flow completely around the primary circuit is one to two seconds.

Temperature control is provided by a secondary hydraulic circuit consisting of a thermoelectric module (TEM) powered "cold source" heat exchanger and a small gear pump. Since this is a closed hydraulic system, operation of the gear pump (trickle pump) displaces cold water into the primary circuit and returns warmer water to the cold source heat exchanger. By correctly metering the proper water flow, any equilibrium temperature of the thermal plate above the cold source temperature can be maintained. Very fine temperature control can be achieved by metering slightly more cold water than is required (slightly overcooling the primary circuit) and adding electrical energy with the use of required for reaching the next supersaturation level in the experiment protocol. The computer control servos around the value Tw defined by this variable. Time required for this step is $\sim 2-5$ sec.

4. "PAUSE": Allows time for system to stabilize at new temp. settings. This has been a period of ~30 sec in preliminary e periment. The end of "PAUSE" initiates another "LWAJT" period at a new supersaturation. The instrument continues to sequence until ∆T reaches a predetermined minimum.

the immersion heater. Both immersion heater output and flow of water through the trickle pump are under servo control of the control computer.

A second pump is included in the secondary hydraulic circuit in order to provide rapid changes of thermal plate temperature during the "TSTEP" period shown in Figure 2. The surge pump is operated full on for a period ranging from about 1 to 7 seconds, depending upon the magnitude of the de-sired temperature decrease in the warm plate. A relatively large parcel of cold water is displaced into the primary circuit during this period, causing a rapid drop in temperature. A period of about 10-30 seconds is required for temperature equilibrium. The surge pump is not used again until another change in plate temperature is desired. The hydraulic circuit for the cold plate is similar to that described above except that the surge pump for the cold plate is attached to a separate TEM-powered heat exchanger (Hot Source), controlling the temperature of the water above the mean temperature of the experiment. Operation of the surge pump for the cold circuit raises the temperature of the cold plate, such that the temperature difference between the plates is reduced in steps as the experiment progresses.

The ΔT measurement is accomplished with the use of a ten-element thermopile mounted on the back surface of each thermal plate. The thermocouple junctions are chromel-constantan and are potted in good thermal contact with an aluminum-base with thermally conductive epoxy. The thermopile is assembled in the form of a yoke; the thin thermo-



Figure 3. Schematic representation of hydraulic circuit of the rapid cycle CCN spectrometer.

Legend

- 1. Thermal plate
- 2. Thermistor
- 3. Centrifugal pump
- 4. Servo heater
- 5. 6. Trickle pump
- Surge pump
- 7. Cold source heat exchanger
- 8. Thermoelectric module
- 9. Water jacket
- 10. Waste heat radiator
- 11. Fan

couple leads are sandwiched between two layers of grounded copper foil tape for support and electrical shielding. The entire bundle is mounted on a thin (0.04 cm x 2.5 cm) strip of fiberglass that forms a yoke around the two thermal plates. The base of the yoke mounts on the thermopile amplifier, located at the bottom edge of the CFD chamber, reducing the lead length of the thermocouples to a minimum. The resolution of measurement is 0.0024°C.

Carrier flow and sample flow rates are similar to those used in the earlier model CFD's. The high degree of precision and accuracy achieved with the earlier models is preserved in the spectrometer. In fact, the increased temperature uniformity and measurement sensitivity allows for increased accuracy. Moreover, this accuracy holds true during the supersaturation cycles of the spectrometer. Thus, the entire spectrum can be monitored with precision and accuracy limited primarily by statistics.

Figure 4 displays an example of an F-plateau in the rapid cycle CCN spectrometer. This shows that the nucleus concentration is independent of the time the sample is exposed to the supersatu-Thus all particles are allowed to activate ration. but not fall out. This is more fully explained in Hudson and Squires (1976). Figure 4 also displays a comparison in particle concentration between the rapid cycle spectrometer and the earlier model CFD's. Agreement was within 3% which is about the same as the experimental uncertainty.



Figure 4. Relative concentration of CCN detected by the rapid cycle spectrometer vs. carrier (main) flow through the chamber. Here the count is normalized to the deduced concentration in a conventional CFD operating side-by-side. The supersaturation in both instruments was fixed at 0.80%.

The rapid cycle CCN spectrometer operated very well during the International Workshop; during many of the experiments, it was actually operating under computer control with no operator present. There were three problems, two of which were instrumentrelated, and a third which may be classified as operator error. First, the chamber had been newly cleaned and reassembled at the time of the Workshop, and was operating with a background count of order 10-20 cm⁻³ (at 1% supersaturation). This count is about an order of magnitude higher than the normal background for this device; after the Workshop it was found to be entirely due to a slightly loose fitting on the carrier flow inlet. Second, the losses of CCN in the sample inlet system were higher than designed for, due in large part to the presence of charged aerosols in many of the experiments. This problem is discussed in greater detail by Hudson and Alofs in a companion paper discussing CFD design and performance. Fin-ally, due both to occasional, necessary operator absences and operator unfamiliarity with CFD operation at low supersaturations, the main, carrier air flow was not always adjusted properly as the micro-processor stepped the chamber through its range of supersaturations. The result generally manifested itself in a tendency to undercount.

INSTANTANEOUS SPECTROMETER 4.

4.1 Equipment Description

This instrument was built along the same lines as the DRI continuous flow diffusion (CFD) chamber (Hudson and Squires, 1976). The most important feature of this instrument is that it uses the sizes of the drops detected by the optical counter to deduce the critical supersaturations of the nu-Since several size thresholds can be used, clei. this allows the possibility of simultanously determining the number, N, vs. critical supersaturation S_c , for several S_c 's. This is difficult in a conventional CFD where the drops usually achieve a nearly monodisperse size distribution regardless of the range of S_{C} 's in the sample aerosol (Hudson, 1976).

4.2 Theory of Operation

The instantaneous spectrometer, however, contains three supersaturation steps which disperse the drop spectrum over a wider size range. This range is further widened since the sample is exposed to these supersaturations in ascending order. The device, which is shown in Figure 5, is a series of three CFD's inside one chamber. It contains a sequence of three pairs of temperature controlled plates so that a sample aerosol can be exposed to three separate supersaturations (S_1 , S_2 , S_3). This means that in the first zone only the largest nuclei become activated drops. That is, only those nuclei with S_C 's below S_1 grow into droplets while the remaining nuclei remain as unactivated haze drops. After being exposed to this constant supersaturation, these drops approach a monodisperse distribution.



Figure 5. Schematic of the instantaneous CCN spectrometer.

Legend: $L_1 = 48 \text{ cm}; L_2 = 8.4 \text{ cm}; L_3 = 10.4 \text{ cm};$ $<math>\overline{L_4} = 38.4 \text{ cm}$ (1st supersaturation zone - S_1 , \overline{I}_3 , \overline{I}_4); $L_5 = 12 \text{ cm}$ (2nd supersaturation zone - S_2 , \overline{I}_2 , \overline{I}_5); $L_6 = 8 \text{ cm}$ (3rd supersaturation zone - S_3 , \overline{I}_1 , \overline{I}_6).

<u>Supersaturation</u>: S1<S2<S3 <u>Plate Temp</u>: T1<T2<T3<T4<T5<T6

(1) Carrier flow entrance; (2) Diffuser screen; (3) Sample injection tube; (4) Cold plate wicking surface; (5) Warm plate wicking surface; (6) First warm section, T_4 ; (7) First cold section, T_3 ; (8) Second warm section, T_5 ; (9) Second cold section, T_2 ; (10) Third warm section, T_6 ; (11) Third cold section, T_1 ; (12) Temperature bath at T_4 ; (13) Temperature bath at T_3 ; (14) Temperature bath at T_5 ; (15) Temperature bath at T_2 ; (16) Temperature bath at T_6 ; (17) Temperature bath at T_1 ; (18) Exhaust to OPC.

In the next zone nuclei with $S_1 < S_C < S_2$ become activated and grow into cloud droplets with similar sizes. In the meantime the drops which were already activated in the first zone grow even larger in the second zone. In fact, their growth rate is speeded up due to the higher driving supersaturation in zone two. Thus, the nuclei with $S_C < S_1$ grow even larger and somewhat more monodisperse and at the end of the second zone a bimodal drop distribution should result. Finally, the third zone activates the smallest nuclei (largest S_C) with $S_2 < S_C < S_3$ and a trimodal distribution should result.

The most significant result is not the trimodal distribution but the fact that the drop size spectrum has a wider spread than it has in a CFD. In the spectrometer the drop concentration is less sensitive to drop size and it is easier to discriminate nucleus $S_{\rm C}\,$'s based on drop sizes. Therefore, a small change in the drop size thresholds results in a smaller change in apparent concentration than would be the case with the monodisperse distribution in a CFD. Thus it is much more feasible to relate drop sizes to S_{C} and to establish size thresholds which correspond to certain $S_{C}{\,}^{\prime}s$. If there were no other factors than S_c affecting drop size, then a trimodal drop distribution with clear separations between modes would always result. In that case, size discrimination could be made between the modes and a definite N vs. ${\rm S}_{\rm C}$ spectrum could be made which would correspond to the three supersaturations used in the chamber. In such a case, a cumulative distribution would have three plateaus where the number concentration would be constant over a range of sizes. In the CFD there is one drop size plateau which ensures that all nuclei are activated but that none fall to the floor so that a direct determination of N vs. $\rm S_C$ can be made with S_C being the applied supersaturation in the chamber. There are some situations when the instantaneous spectrometer has three drop size plateaus which then allow direct determinations of N vs. S_C for the three S_C 's. However, in most situations the modes are not completely separated (Fig. 6) and instead of plateaus in the cumulative distribution, we find decreases in the slope of N vs. r (Fig. 7). Although this is a much better situation than in the CFD, the lack of a plateau limits the accuracy of direct measurements of N vs. Sr. Accuracy can be increased by setting the voltage thresholds so that the number concentration in the spectrometer matches that in a CFD monitoring the same sample at a specific supersaturation.

INSTANTANEOUS SPECTROMETER NUMBER VS SIZE



0.93% 0.60% 0.23%

Figure 6. Relative number of drops vs. relative sizes (voltages) for the instantaneous spectrometer. Note that this is a differential and not a cumulative plot.

Although this can also be done with two CFD's (Hudson, 1976) (where one of the CFD's takes the role of the spectrometer and the other one is used to calibrate the first CFD), the process works much



Figure 7. Relative cumulative number of drops vs. size in the instantaneous spectrometer. Applied supersaturations were 1%, 0.4% and 0.15%.

better with the instantaneous spectrometer where there is nearly a constant concentration over some parts of the size range. This considerably reduces the requirement for stability of the various operating parameters. With the instantaneous spectrometer it has been possible to keep the operating parameters constant enough that voltage settings can be used for many days or weeks with continued good accuracy.

4.3 Description

Figure 5 shows most of the dimensions of the instantaneous spectrometer. The plates are separated by 1.6 cm while the plate width is 29 cm. This chamber was also operated with the plates vertical and sample moving horizontally. The main flow through the chamber was 50 cm³ sec⁻¹ throughout the Workshop. This resulted in a particle velocity of v = 1.62 cm sec⁻¹ so that the sample spent about 31 sec in the chamber; 18.6 sec at S₁, 7.4 sec at S₂, and 5 sec at S₃.

As with the DRI CFD, a Royco 225 optical particle counter is used as the detecting device for the instantaneous spectrometer. In addition a 512 channel analyzer (MCA) (Northern Scientific, Inc.) is also interfaced to the Royco to increase particle size resolution so that greater detail in the concentration vs. size spectrum can be displayed.

The plate temperatures were roughly the same for the entire Workshop so that the supersaturations were nearly constant at $S_1 = 0.30\%$, $S_2 = 0.55\%$ and $S_3 = 0.90\%$. The droplet size thresholds were

set by matching the number concentrations in the instantaneous spectrometer with the concentrations measured with the CFD set at the three different plate temperatures in the instantaneous device. The largest drops corresponded with the lowest supersaturations, etc. All of the drops which could be detected down to the smallest sizes ($\sim 0.2 \mu m$ radius water drops) corresponded to the number of CCN active at the highest supersaturation in the spectrometer.

Channel 2 was set for about 1.42 μm radius water drops while Channel 3 was set for 1.75 μm radius water drops. A slight number vs. drop size plateau was observed here and the concentration of CCN in the CFD at 0.55% supersaturation (which was S2 in the spectrometer) was found to be always less than the number of drops in Channel 2 but more than that found in Channel 3 of the instantaneous spectrometer. This meant that nuclei with Sc of 0.55% produced drops within the size range of 1.42 μm and 1.75 μm radius in that particular configuration of the instantaneous spectrometer. Thus, the average of Channels 2 and 3 were used to deduce the number of CCN active at 0.55% in the spectrometer. Channel 4 was set at 2.77 μm radius water drops. In a similar fashion, these corresponded to 0.30% Sc. It was found necessary to make a small adjustment in the size thresholds only once during the Workshop.

The sample flow rate was usually the same as the CFD, 0.60 $\rm cm^3 sec^{-1}$, although it was at times as low as 0.1 $\rm cm^3 sec^{-1}$. The plates were also controlled by the same regulator baths and the same types of thermisters were embedded in the plates. This chamber differed from the CFD in three other respects: (1) There were no flows of particle-free air around the backside of the plates. Instead, a diffuser screen was used to eliminate any turbulence; (2) A metal mesh screen was used instead of filter paper for the moist plate surfaces; and (3) Instead of dripping water onto the plates as in the CFD's, water was fed to the metal screens by capillary action from a reservoir of distilled water.

4.4 Operation

Several tests can be performed to check the performance of the instantaneous spectrometer. When the upstream lowest supersaturation, S1, is increased, the larger sized droplet peak increases and becomes larger as it should. When the higher downstream supersaturation, S_3 , is increased, the magnitude of the smaller sized peak is increased and there is an increase in its size. Under these conditions, the larger sized peak is only shifted to a slightly larger size. These observations are all in keeping with the operating principles. Thus, sizes which allow separations between the peaks can be chosen. Moreover, the Royco voltage thresholds can be set so that certain size channels can be used to monitor the concentration at specific supersaturations. The size channels can be adjusted so that an individual drop size plateau can be obtained for each supersaturation (see Hudson and Squires, 1976). This assures that all drops which should have been activated at a certain supersaturation were activated and counted. Changes in the downstream supersaturation, S₃, do not affect the detected concentration active for instance at S₁ or S₂.

The spectrum of three supersaturations was available simultaneously as soon as the OPC counted and printed out the numbers. Agreement with the DRI CFD was very good and consistent throughout the Workshop as shown by the results.

5. ISOTHERMAL HAZE CHAMBER

The measurement of CCN can be extended to lower values of supersaturation by using the isothermal haze chamber (IHC) first described by Laktionov (1972). The basic operating principle of the IHC relates to the fact that the equilibrium size of a haze droplet, r_{100} , at 100% RH (supersaturation = 0) is uniquely related to the critical supersaturation S_c of the nucleus. Following Laktionov (1972) and Alofs (1978), at T = 20°C the relationship is

$$r_{100} = \frac{4.1 \times 10^{-6}}{S_{c}}$$
(1)

where r is in centimeters and S_c is in percent. According to Hoppel and Fitzgerald (1977), this relationship is unique if the particle is at least 1% soluble. Eq. (1) can be applied if drops are grown to their equilibrium sizes in a saturated environment and then counted as a function of size.

The Desert Research Institute IHC is a device which subjects sample aerosol to 100% relative humidity for 100-200 s. In most cases, this is enough time to allow the drops to attain their equilibrium sizes at which time they are counted and sized by an optical particle counter (OPC Royco 225). The size is then related to the critical supersaturation $S_{\rm C}$ so that an N vs. $S_{\rm C}$ curve can be drawn.

This instrument was built along the same lines as the continuous flow diffusion (CFD) chamber (Hudson and Squires, 1976), in that the sample occupies only a small volume of the cloud chamber which is made up mostly of particle-free air. Figure 8 is a schematic diagram of the DRI IHC. The geometry, of course, is different from the CFD in that the IHC is a right circular cylinder, while the CFD is a rectangular parallelepiped. In the IHC, the sample travels vertically downward (the sample always travels horizontally in the DRI CFD) so that fallout of the large drops still carries them into the optical counter instead of onto the walls where they would be lost.

As with the CFD the total flow F of air can be changed without altering the sample flow f. This allows the capability of changing the time which the sample spends in the saturated volume of the chamber. Therefore, like the CFD, it is possible to give the sample nuclei various growth times before counting and sizing by the OPC which is attached at the outflow end of the cylinder which is the bottom of the chamber.

Using this device, it is possible to experimentally determine whether the drops have had sufficient time to reach their equilibrium sizes. If a plateau exists in the relationship between F and the number of drops larger than specific threshold sizes (Fig. 9), then it is reasonable to assume that these drops have reached their equilibrium sizes r_{100} . According to calculations made by Laktionov (1972), 100 s are required to grow nuclei with $S_c = 0.02\%$ to within 95% of their r_{100} which



Figure 8. Schematic diagram of the DRI isothermal haze chamber (IHC) and associated apparatus.

is 2 μm . The available time in the DRI IHC usually approaches 200 s. The existence of an F plateau ensures that drops have enough time to grow to their equilibrium size and that they do not evaporate in the OPC.

Various flows of air through the IHC imply various velocities through the optical counter. In theory, this should not affect the size calibration of the instrument which should only depend on pulse height and not pulse width. Although Alofs (1978) has found that this dependence is not always true at extremely low flows, the flows used in the DRI IHC were never that low. In fact, the size calibration was found to be the same within experimental error for the various flows which were used.

The response of the Royco OPC depends on the index of refraction of the particles. The instrument is calibrated with latex spheres which have an index of refraction of 1.59. Calculations for various OPC's including a Royco have been made by Cooke and Kerker (1975). Although this is probably the best treatment of the subject, it has many shortcomings in this application. The limits of the scattering angles are slightly different for the Royco 225's used in this study and the values for the Royco 245 referred to by Cooke and Kerker. The largest discrepancy is the angle β (the upper limit of the scattering cone) which Cooke and Kerker refer to as 25° for the Model 245; the manufacturer quotes a value of 28° for the Model 225 while the author measured 27.4° on one of the 225 instruments. This discrepancy may or may not be enough to change the results of the Mie scattering calculations. Moreover, Cooke and Kerker did not perform a calculation for index of refraction 1.59 which is the index of refraction of the latex spheres used



Figure 9. Number of drops exceeding various size thresholds as a function of the total flow in the JHC. The numbers were normalized to the concentration in a simultaneously operating CFD at 0.15% supersaturation. This figure illustrates a plateau in number concentration for various growth times for various critical supersaturations.

to calibrate the device. The best one can do is to interpolate between the 1.54 and 1.70 curves and compare this with the response curve for index 1.33. A correction curve can then be developed so that the instrument calibration performed with latex spheres can be applied to water drops. This index of refraction correction was then applied to the IHC.

This device operated fairly well during the Workshop, but there are indications that the smallest droplets, those grown on FCN of highest critical supersaturation, may have suffered evaporation in the Royco OPC. The consequence of evaporation of this sort is that the IHC tends to undercount in the region of operation overlapping CFD chamber operation, or about 0.1% to 0.2% supersaturation.

6. ACKNOWLEDGEMENTS

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Development of the rapid cycle CCN spectrometer was supported by the NASA Marshall Space flight Center, under Contract NAS8-31470. Principal Investigators for this contract were Patrick Squires and Warren Kocmond. The contributions of Sam Keck, Peter Wagner, and Darrel Reid to this project are gratefully acknowledged. 7. REFERENCES

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1. INTRODUCTION

It is common practice to generate laboratory aerosols of soluble materials with pneumatic atomizers. In a typical device, a solution of the substance to be aerosolized is injected into a jet of air and the liquid is broken up into very small droplets. After forced evaporation, a dry aerosol of the solute is produced. In most atomizers, a solid surface is placed into the droplet-laden jet such that the larger droplets are impacted and a narrower size distribution is produced.

In preparation for the Workshop, a number of commercially available devices were tested that operated on the above principle. Devices included the DeVilbiss No. 644^{*}, the Nano-Mist^{**} and other inhalation nebulizers, and the TSI^{***} Model 3076 Constant Output Atomizer (COA), specifically intended for laboratory use. The schematic example (Figure 1) shows a vertical cross-section of the TSI-COA which amounts to a slightly modified version of the Collison Atomizer (May, 1973). Its distinctive feature is the ducted portion of the jet within which the liquid is injected, in contrast to the open jet mostly used in various other inhalation nebulizers (as shown schematically in Figure 3).

Despite differences in design, all the atomizers tested suffered from short- and/or long-term fluctuations of their output particle number concentrations. Some of the devices displayed a constant unsteady behavior while others, such as the TSI-COA, occasionally performed well but then flipped into a very unstable mode for prolonged periods. For many types of laboratory investigations, especially where several instruments are involved as in the case of the CCN Workshop, it is essential that temporal stability of test aerosols be maintained within a few percent. The aim of the present study was to elucidate the mechanisms responsible for atomizer instabilities and to find methods for alleviating these problems.

2. STUDY OF OUTPUT INSTABILITIES

In most atomizers, it is not possible to observe the critical nozzle area during operation nor can the nozzle geometry be modified to effect changes in the performance. In order to overcome these difficulties a variable geometry atomizer (VGA) with a transparent housing was designed. Using this device, the relative position and the size of the air jet, liquid feed tube and impaction surface could be varied. Atomizer performance was



Figure 1. Cross-section of TSI constant output atomizer (from TSI Manual).

assessed by analyzing its output in an electrostatic classifier (EC, TSI Model 3071) and an electrical aerosol detector (EAD, TSI Model 3068) the signal of which was recorded on a stripchart.

Initially, it was thought that build-up of electric charges on non-conductive components of the aerosol flow system were responsible for the sometimes very abrupt changes in aerosol output. However, this interpretation had to be ruled out on the basis of experiments with systems that were made totally conductive.

A series of tests in conjunction with other aerosol measuring devices and with various atomizers confirmed that the observed output fluctuations were not artifacts of the aerosol instruments but, indeed, originated in the atomizer system. The same tests also indicated that the output variations occurred over the entire particle size range, though not necessarily to exactly the same extent. In the case of the TSI-COA, the average output in the unstable operating mode contains a considerably higher proportion of larger particles than in the steady mode as Figure 2 attests.

DeVilbiss Company, Somerset, PA.

Eastfield Corporation, Noroton, CN.

TSI Incorporated, St. Paul, MN.



Figure 2. Size distributions of NaCl aerosol generated by TSI-COA at 30 psi with a 0.01% solution in stable and unstable operating modes.

Visual observation of the VGA in operation uncovered several phenomena causing unsteady aerosol outputs. By monitoring the impact area of the jet on a transparent plate, it was found that the size and position of that area was fluctuating in correlation with the particle output fluctuations. Measurements, on the other hand, showed a strong dependence of the output from the mutual position of jet and liquid feed tube. Closer scrutiny (by stereo microscope) indicated:

(a) that in case of wide (~1 mm diameter) liquid feed tubes, the jet and associated turbulence caused the liquid surface to oscillate and, as a consequence, to release liquid at irregular intervals into different portions of the jet, thus generating output irregularities as observed, e.g., in the DeVilbiss nebulizer. This is schematically depicted in Figure 3a, while a narrow tube as in Figure 3b alleviates the problem;



Figure 3. Schematic of open-jet atomizer, (a) with wide liquid feed tube causing irregular output; and (b) with narrow liquid feed providing improved output stability.

(b) that salt deposits at the rim of liquid feed tubes were forming after a period of operation - sooner when working with concentrated solutions, later for dilute solutions. Both the liquid discharge into the jet as well as the jet itself are influenced by the presence of the salt accumulation resulting in erratic output behavior. Removal of the minute salt obstacle immediately restored a steady output; and

(c) that gradually accumulated solution from spray droplets being whirled around between nozzle and impaction surface or solution directly from the liquid feed tube is periodically entrained, or drips into, the jet resulting in temporary output maxima. Concentrated solutions aggravate the problem due to their tendency for foaming, as frequently observed in the TSI-COA where the narrow vertical cavity promotes this undesirable effect.

3. DESIGN AND PERFORMANCE OF THE IMPROVED CON-STANT OUTPUT ATOMIZER (ICOA)

With temporal stability being the main criterion, a new atomizer design was considered essential in order to eliminate or minimize the above mentioned causes of fluctuations found in other units. Thus in a succession of test models, a geometrical arrangement of air jet, liquid feed tube and impaction surface evolved that showed none of the previously described problems.

The top of Figure 4 depicts a cross-section and face-view of the air nozzle and liquid feed combination that evolved. By allowing the liquid to travel in a short, open channel from the tube end to the air nozzle (aided by capillarity and Venturi effect), the formation of salt deposits as well as entrainment of spilled-over liquid was eliminated. A sufficiently large housing (typically \sim 1 & volume) and the impaction surface positioned





Figure 4. Top: Schematic of ICOA nozzle. Bottom: Schematic of ICOA operational assembly.

several (\sim 4) centimeters from the nozzle provided enough space for the spray to dissipate without forming deposits that would drip into the jet. The exit cone of the 0.41 mm nozzle allowed the liquid to enter the jet from all sides and thus further prevented accumulation of liquid near the nozzle.

By operating the unit with a liquid metering pump (such as FMI-RHOCKC*) as shown in Figure 4, bottom, at a rate of 0.05 to 2.0 ml min-1, size spectra as shown in Figure 5 could be obtained at less than $\pm 1\%$ deviation from the average particle number concentration for periods of one hour or more. An illustration of this can be seen in Figure 6 which shows concentration data for the



Figure 5. Aerosol size distributions obtained with the DRJ-JCOA at the three indicated air pressures with 0.025% NaCl solution. Upper scale refers to original droplet diameters as inferred from particle sizes and solution concentration.



Fluid Metering, Inc., Oyster Bay, NY.

beginning and end of a six hour run. Occasional minor long term drifts of the output (typically under 2%) have tentatively been traced to the liquid metering pump. The atomizer has been operated successfully at air pressures between 10 and 100 psi; below 10 psi, the output is slightly less stable.

4. ACKNOWLEDGEMENTS

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SECTION VI. SUMMARY - REVIEW PAPERS

CCN COMPARISONS OF STATIC DIFFUSION CHAMBERS

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ABSTRACT

Nine CCN counters of the static diffusion (SDC) type were compared with one another and with continuous flow diffusion (CFD) chambers. The nine SDCs showed a considerable amount of variation, largely attributable to newness and/or lack of prior calibration of some units. The five more consistent instruments agreed quite Well, to within at least 20% of the NRL mobility anlayzer and to within 10% at 1% supersaturation. There was satisfactory agreement between the more reliable SDC and CFD chambers.

1. INTRODUCTION

At the first International Workshop on Cloud Nuclei in Lannemezan, France (1967), it was not necessary to distinguish between types of instruments for measuring CCN concentrations. Only a few were in existence and present, and all were of the Twomey-type (1963), based on a method of Wieland (1956). In these thermal gradient diffusion chambers, nucleated droplets were detected individually on photographic film. Even at the second Workshop in Fort Collins, Colorado (1970), five of the six units present again were of the thermal diffusion (static) chamber type.

As is evident from the instrument descriptions, a whole host of CCN counters were represented at this Workshop in Reno. They may be classified by type as static diffusion chambers (SDC), continuous flow diffusion chambers (CFD), and low supersaturation "haze" chambers. In this preliminary analysis, we will concentrate on the performance of the static diffusion chambers, although some brief reference to the other instrument types may be appropriate.

At the 1970 Fort Collins Workshop, the five thermal diffusion chambers (excluding one commercial unit) agreed quite well, to within $\pm 30\%$ of the mean (Ruskin and Kocmond, 1971). Chamber operation and expansions were essentially at three supersaturations S: 0.3, 0.75 and 1.0%. By contrast we operated at Reno over a broader supersaturation range, included more S points, and did not prescribe standard S values. In retrospect, the latter was unfortunate and made the data analyses from computer printout data quite laborious and time consuming. The inclusion of many S levels did enhance the data base and allow for, among other things, consideration of CCN supersaturation spectra shapes (e.g., Jiusto and Lala article). Some of the basic questions that such an analysis can address are as follows:

a. How well did the SDC units intercompare?

b. Where discrepancies occurred, were there logical explanations?

c. Was there general agreement between SDCs and the more sophisticated CFD units?

d. Has the science of CCN concentration measurements advanced during the past decade?

Clearly in light of the wealth of data produced and time limitations involved, we will only touch on some of these complex questions. Hopefully, more definitive answers will emerge after further analyses of the data and future experiments stimulated by this International Workshop.

2. CCN INSTRUMENTS

The static diffusion chambers operated at the Workshop are listed in Table 1. Of the nine instruments, only one - the NRL counter - was essentially the same as that used at Fort Collins. This instrument has been used extensively over the years in field programs and in comparisons with other counters. Thus it is considered a useful reference for intercomparisons. The Hebrew University counter was present at all three workshops, though at Reno it employed a different detection method.

As is evident, there has been a trend in recent years away from individual droplet counting in static diffusion chambers to measuring the total scattered light from a sample volume of droplets and transformation to CCN concentrations. Some of the advantages and disadvantages of each approach are discussed in Section 3.4.

Five of the nine SDCs employed a scattered light principle. A related approach apparently was first used by Radke and Hobbs (1969) but with "side" scatter detection over a broad angle, necessary recording of the Mie peak, and a sizable chamber geometry. The commercial unit of Mee, Inc. introduced near-forward light scattering which allows for considerable counter size reduction and portability. Lala and Jiusto (1977) established the theoretical relationship for forward scatter chambers linking supersaturation (drop growth), light scatter intensity and drop concentration; full automation and data reduction via microprocessor control were achieved.

TABLE 1

CCN Counters - SDC Type

Instrument No.	Figure Symbol	Organization	Operator	Droplet (Nuclei) Detection	Light Source
2	0	U. of Wyoming	Rogers/Politovitch	Photographic	He-Ne Laser (5 mW)
5	В	British Met. Office	Kitchen	Light Scatter	25 W Tungsten
8	Е	Mee, Inc.	Mee	Light Scatter	25 W Tungsten
9	н	Hebrew U.	Gagin/Nuzitsa	Light Scatter	100 W Halogen
10	S	SUNY	Lala/Jiusto	Light Scatter	I.R. Emit. Diode (25 mW)
13	I	CSIRO	Ayers	Light Scatter	150 W Tungsten
17	N	NRI.	Wojciechowski	Video Camera	200 W Mercury Arc
24	F	France	Serpolay	Video Camera	He-Ne Laser (5 mW)
25	С	CSU	Borys	Photographic	He-Ne Laser (5 mW)

In essence the CCN instruments of SUNY, Hebrew University and CSIRO are reasonably similar in principle and all incorporate a back-up standard photographic method for calibration. The British Met Office and CSU units are modified Mee, Inc. commercial counters. The instrument from France employs a laser light beam and a video camera detector, while the Wyoming counter utilizes laser illumination and records droplet images with a 35 mm camera.

Several of the SDCs had undergone recent redesign or modification, partly in preparation for the Workshop. Others were newly assembled and admittedly brought to the Workshop to test their performance and obtain first-time cross calibrations. This was counter to the original plan of comparing "proven" instruments, but the more open approach undoubtedly added to the overall learning process. Under such circumstances, one might expect a fair spread in the data obtained, and indeed it occurred.

The NRL mobility analyzer (electrical classifier) provided accurate aerosol size information from which supersaturation spectra could be computed with aerosols of known chemical composition (see Hoppel article). These derived spectra also served as a useful reference for instrument comparison.

3. RESULTS

3.1 Overall SDC Performance

One method of evaluating the data was to examine the CCN concentration measurements made with all SDC instruments on all experiments at designated supersaturations. The supersaturations considered were 0.2%, 0.5% and 1.0%. As noted previously, not all counters operated at these specific S values, so a good deal of judicious extrapolation was necessary. The 0.2% supersaturation is at the lower bound of reliability for static diffusion chambers because of problems associated with haze discrimination, non-uniform drop sizes (light

scatter method), and temperature control. As one approaches 0.5 to 1.0%, higher reliability should be expected.

Table 2 lists by experiment the coefficient of variation, and the average CCN concentration \overline{N} of all SDCs (5 to 9) participating in a given experiment compared with the NRL SDC, the NRL mobility analyzer (MA), and the UMR continuous flow chamber. The coefficient of variation V is a dimensionless function of the standard deviation and the arithmetic mean, i.e.,

$$V = \sigma/\overline{N}$$
 or $V(\%) = 100 \sigma/\overline{N}$.

V is a relative measure of the dispersion of the data and, in a very loose sense for these data, can be thought of as the percentage to each side of the mean that will encompass most of the CCN concentration values.

Table 2 reveals a number of noteworthy items:

1. The three classes of experiments - involving polydisperse salts, monodisperse salts, and ambient air - were not too dissimilar in terms of the presented variables. Thus for simplicity one can concentrate on the overall averages of the bottom row.

2. σ/\overline{N} values of about 0.5 at 0.2% and 0.5% S were comparable and not overly satisfying. At 1% S the degree of dispersion had reduced to 0.32.

3. The concentration ratios (\overline{N} divided by the count of each of the three indicated instruments) show that, on average, the SDCs were under-estimating CCN concentrations. This was graphically evident at the Workshop where certain instruments were obviously registering low counts. Inadequate light intensity or an inefficient optics system are usually the sources of such difficulties. Possible calibration error, depending on the standard used, can also introduce discrepancies.

TABLE 2

Average Performance of All SDC Instruments (Coefficient of Variation and Concentration Ratios vs. S)

		0.2% S			0.5% S				1.0% S				
a.	Polydisperse Salts	<u>σ/N</u>	N/NRL	<u>n/ma</u>	N/UMR	_σ/N	N/NRL	<u>n/ma</u>	N/UMR	<u> </u>	N/NRL	N/MA	N/UMR
	Exps. 1, 2, 10, 13, 14, 22, 23, 24*	0.51	0.80	0.84	1.01	0.50	0.70	0.58	0.72	0.33	0.86	0.73	0.79
Ъ.	Monodisperse Salts												
	Exps. 4, 5, 8, 9, 15, 18, 19, 20*	0.46	0.75	0 .89	0.87	0.54	0.70	0.80	1.02	0.31	0.87	0.91	0.95
c.	Ambient Air												
	Exps. 3, 6, 11, 12 16, 17, 26	0.54	0.67	-	0.91	0.57	0.76	-	0.71	0.33	0.85	-	0.84
	Overall Average	0.50	0.74	0 .8 7	0.93	0.54	0.72	0.69	0.82	0.32	0.86	0.82	0.86

*0.25% S rather than 0.20% S

TABLE 3

Average Performance of Five SDC Instruments: NRL, CSIRO, SUNY, Hebrew University, and France (Coefficient of Variation and Concentration Ratios vs. S)

		···	0.2	% S			0.5	% S			1.0	% S	
a.	Polydisperse Salts	<u> </u>	<u>N/NRL</u>	N/MA	N/UMR	<u>σ/N</u>	N/NRL	\overline{N}/MA	N/UMR	<u>σ/Ν</u>	N/NRL	<u>n/ma</u>	N/UMR
	Exps. 10, 13, 14, 22, 23, 24*	0.42	0.79	0.95	1.09	0.29	0.77	0.68	0.84	0.19	0.85	0.83	0.89
ь.	Monodisperse Salts												
	Exps. 5, 8, 9, 15, 18, 19, 20*	0.36	0.84	0.98	0.97	0.30	0.84	0.90	1.08	0.31	0.93	0.99	0.97
c.	Ambient Air Exps. 6, 11, 12,	0.48	0.74	-	1.03	0.35	0.91	-	0.89	0.22	0.94	-	0.99
	Overall Average	0.42	0.79	0.97	1.03	0.31	0.84	0.79	0.94	0.24	0.91	0.91	0.95

*0.25% S rather than 0.2% S

Because of problems with some instruments, a degree of smoothing was in order. Table 3 presents identical information for just five of the SDC counters whose performance was considered more reliable at the Workshop; these were the units from NRL, CSIRO, SUNY, Hebrew University and France. Even these units on certain experiments registered values that were out of the mainstream. For inclusion of an experiment in this analysis, at least four of the five instruments had to be participating.

The coefficient of variation (Table 3) improved considerably, decreasing to 0.42, 0.31 and 0.24 as supersaturation increased from 0.2 to 0.5 to 1.0%. σ/\overline{N} was somewhat better (smaller) for the laboratory aerosol tests than for ambient air, due largely to greater ambient aerosol fluctuations. Such temporal fluctuations (even with lab aerosols at times) can introduce misleading scatter in the data because the CCN counters possessed different sample processing times and also could not readily be synchronized in time to common S levels.

The overall concentration ratios of Table 3 present a similar but more consistent picture than Table 2 data. At 0.2% the SDC's appeared to underestimate (0.79) with respect to the NRL counter but were within 3% of both the NRL mobility analyzer (0.97) and UMR CFD (1.03). At 0.5% the SDC's apparently underestimated concentrations by about 6 to 20%. At the higher 1% supersaturation, the concentration ratios were all quite respectable, reaching 0.91 to 0.95 values. In general, the SDC's were in somewhat better agreement with the UMR counter than with the other two references. Note that similar agreement would have been obtained with the DRI CFD (instrument #15) which tracked very closely with the UMR counter.

3.2 Individual Experiments and Comparisons

Figures 1-5 depict some representative SDC experiments (not necessarily the best nor the worst cases, but ones that were reasonably well controlled with many participants). The CCN concentrations vs. S data were taken from the printout sheets provided each participant after the Workshop; then averaged where appropriate if several runs were made; and plotted by computer in the fashion shown. The solid lines represent CCN-S spectra computed by NRL from particle size data obtained with the NRL mobility analyzer. Refer to Table 1 for instrument symbol designation.

In the monodisperse aerosol experiments shown in Figures 1 and 2, the critical supersaturations were ${\le}0.2\%$ so that CCN concentrations should be uniform over the indicated S range. Indeed, most data points were within 20% of the plateau. In three other monodisperse aerosol experiments, S_C fell within the SDC operating range of 0.2-1% S. A glance at the Workshop data indicates that most instruments were capable of detecting S_C to within ±0.1% S.

It should be noted that in Figure 1 the British Met Office instrument data were not plotted and in Figure 2 the CSU data were ignored. Both instruments were experiencing difficulties and undercounting. While the Mee, Inc. instrument is represented in Fig 1 (E symbols), it was periodically malfunctioning during the limited days it was



Figure 1. Static Diffusion Chamber Values and Mobility Analyzer Curve (see Table 1 for instrument symbol legend).



Figure 2. Static Diffusion Chamber Values and Mobility Analyzer Curve.



Figure 4. Static Diffusion Chamber Values and Mobility Analyzer Curve.



Figure 6. CFD Instrument Values and Mobility Analyzer Curve (see instrument symbols in Sec.3.3 of text).

at the Workshop. It was a new factory unit and reportedly rushed to the Workshop without the customary time for checkout.

Figures 3 and 4 illustrate polydisperse aerosol experiments. Most points in these two experiments lie below the mobility analyzer curve, with considerably more data spread in Experiment 24 (Figure 4). In these and several other cases, the NRL, CSIRO and SUNY instruments tracked reasonably close to one another, with the Hebrew University unit also close at the higher supersaturations. Overall the consistent NRL chamber data matched most closely the NRL mobility analyzer data.

An ambient air case is shown in Figure 5 with considerable data spread. Both the "B" and "C" point instrument are presumably in an early stage of development. The remaining data points cluster reasonably well.

3.3 SDC and CFD Chambers

Figures 6-8 represent CFD chamber data for three of the same experiments just described. Instrument symbols are as follows:



Figure 7. CFD Instrument Values and Mobility Analyzer Curve.



Figure 8. CFD Instrument Values in Ambient Air.

Symbol	Instrument No.	Group
n	15	DRI (NASA unit)
d	16	DRI (Spectrometer)
D	18	DRI
М	21	UMR
W	20	Univ. of Washington
Y	7	York University (different concept/see Leaitch and Megaw)

One cannot help but be impressed by the relatively small degree of data spread, particularly in the ambient air case (Figure 8). The fact that only five CFD instruments are involved, that three of these were developed at one institution, and that all have benefitted from extensive engineering and intercomparisons certainly are constructive factors. The results perhaps contain an instructive message for those wishing to develop CCN units.

Comparing Figures 6 and 7 with Figures 1 and 4, it is seen that most CFD points were also just below the NRL classifier curve. By carefully examining these figures (and the ambient case), it is evident that the CFD data and more reliable cluster of higher SDC data points overlapped quite well. In short, we sugest that a well-designed SDC and a well-designed CFD will both perform well. That is evident from the concentration ratios of Table 3 (five SDC's vs. UMR CFD) and from Alofs' determination (described elsewhere in this report) that the UMR CFD and NRL SDC systems agreed on average to 6% at 0.3 and 1.0% S.

3.4 SDC Types

As mentioned previously, SDC instruments now take two principal forms: individual droplet imaging and counting via TV monitors and regular cameras or total light scattering from an ensemble of droplets. The results from this Workshop are not definitive in terms of whether either class is inherently superior. At least one or two of each type appeared to give above average performance while others did not.

An objection to the light scattering method is that at the lower supersaturations, drop sizes are less uniform and less apt to yield accurate concentration data. Similarly with individual droplet detection methods at low S, it is very difficult to discriminate haze from activated droplets. Also with low nucleus concentrations, some subjective judgement can enter as to what frame or portion of a sample volume to count. With either type system, very small sample volumes (light beams) can produce erratic results in low concentration aerosols.

Allowing that Twomey-type chambers with an experienced operator may possess higher inherent accuracy, there is much to be said for the light scattering technique. The latter approach lends itself to more miniaturization, portability, and automatic operation and recording of data. This facilitates field studies of CCN concentrations (and spectral slopes k) on fine time and spatial scales that otherwise would be formidable. A flexible compromise configuration, in evidence at the Workshop, is an SDC that incorporates both the optical and light scattering concepts with the camera operation used at least for periodic calibration.

4. CONCLUDING REMARKS

In summary, the International CCN Workshop provided a wealth of data and information on different classes of instruments and their performance. In view of the massive quantity of data available, much of which could not be thoroughly covered, this report contribution should be considered preliminary.

In general, the more reliable SDCs performed acceptably most of the time. Some are clearly in need of modification and further development. Even those with above-average performance were occasionally out of the mainstream and improvements were undoubtedly suggested to their instrument developers.

The combined average CCN concentrations from five SDC's agreed to within at least 20% with the NRL mobility analyzer, and to within 10% at 1% supersaturation. Agreement between the NRL SDC and the mobility analyzer was even closer. The coefficient of variation (σ/\overline{N}) of the five instruments was 0.42 at 0.2% S, 0.31 at 0.5% S and 0.24 at 1.0% S.

The spread of SDC data was considerable and in excess of that at the prior Fort Collins Workshop. We largely attribute this to the greater number of instruments present, many of which were relatively new and untested previously. By contrast, four of five CFD units present showed closer agreement with one another, in part for just the opposite reasons cited above as well as for indispensible long-term engineering. The above-average SDC's agreed most satisfactorily (within about 10-15%) with the UMR and DRI CFD's over their common S range.

We began by posing the general question as to whether the science of CCN concentration measurements has advanced over the past decade. We believe it has! Certainly the capability of measur-ing at supersaturations lower by about an order of magnitude with haze chambers and certain CFD's is an all-important achievement; now virtually all cloud and fog condensation nuclei can be discrimi-nated. CFD instruments have ushered in a high degree of engineering sophistication and accuracy for many applications. The Twomey-type counter, as represented by the NRL SDC, continues to maintain an enviable level of performance. It will undoubt-edly remain a useful standard for SDC units. The introduction of light scattering techniques for SDC performance, while not necessarily adding to accuracy, provides for much simpler operation and data processing. The net result may be CCN information with a time and spatial resolution heretofore unavailable.

5. ACKNOWLEDGEMENTS

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CCN-SUPERSATURATION SPECTRA SLOPES (k)

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ABSTRACT

Theoretically the slope k of a CCN-supersaturation spectrum (N = cS^{K}) should equal two-thirds of the slope of the total (soluble) aerosol sizedistribution. Workshop results tended to verify this relation. As has been noted before, the k values are markedly different depending on whether one is measuring ambient CCN concentrations at supersaturations S above or below $\circ 0.1-0.2\%$. The larger k values for S $\leq 0.1\%$ is consistent with the greater decrease in large particle concentration with increasing size. Over the S range of 0.02% to 2%, two power fits (and k values) may sometimes suffice for a reasonable approximation of the CCN distribution. At other times, and with laboratory generated aerosols, such an approach is inadequate and requires refinement.

1. INTRODUCTION

Since the inception of cloud condensation nucleus measurements, it has been observed experimentally that nucleus concentration N versus supersaturation S can be expressed reasonably well by the power function

$$N = cS^{K}$$
 (Twomey, 1959). (1)

With thermal gradient diffusion chambers operating above $\sim 0.2\%$, slope values k typically range from about 0.4 to 1.0. Early indications were that lower values are associated with maritime aerosols and the higher values with continental aerosols (Twomey, 1959; Kocmond, 1965; Jiusto, 1967). Twomey and Wojciechowski (1969) in an extensive aircraft measurement program over various parts of the world reported average k values aloft of 0.5 and 0.7 (0.2 to several % S) for continental and maritime air, respectively.

From Köhler theory for droplet growth, it is evident that the critical supersaturation and size of soluble nuclei are related by S $_{\rm C}$ \approx $r^{-3/2}$. Similarly a Junge power-function relation exists between particle size and cumulative particle concentration: N_{Cum} \propto $r^{-\beta}$. Combining these relations with expression (1) above lead to:

$$k = 2/3 \beta$$
 (2)

where β is the slope of the total soluble aerosol size distribution. Junge and McLaren (1971) demonstrated that the shape or slope k of a CCN supersaturation spectrum is more dependent on aerosol size distribution than on aerosol composition, provided the particles are at least 10% soluble. Added confirmation was provided by an analysis of data from the 1970 Fort Collins Nucleus Workshop (Fitz-gerald, 1973).

Because Junge total aerosol size distributions often have slopes $\beta \gtrsim 3$ for particle radii >0.1 µm,

CCN k slopes of 2 are sometimes mentioned. This misconception is occasionally encountered in the literature. At chamber supersaturations of 0.2-1%, nuclei much smaller than 0.1 μm radius are activated as well. Because of their much greater abundance, the resultant k values should be weighted to the flatter portion of the Junge curve and hence yield values less than 2 and generally less than 1.

Conversely, at low supersaturations <0.1-0.2%, where larger condensation nuclei are activated, one should expect higher k values. Such evidence was first reported via direct measurements by Laktionov (1973). Subsequently, the development of "haze" chambers operating at low supersaturations confirmed the trend (e.g., Hoppel, 1978; Hudson, 1980).

Clearly, the topic of CCN spectra shape over a broad range of supersaturation deserves further resolution. This International CCN Workshop, with its variety of chambers, offered an opportunity to explore such questions as:

a. Is the relation $k = 2/3 \beta$ generally valid?

b. Over what S range can one safely apply the empirical function $N = cS^{K}$ with a single k value?

c. For ambient air, will two k values suffice for the respective supersaturation ranges less than or greater than about 0.2%?

2. APPROACH

Detailed particle size distributions were obtained at the Workshop by NRL with their highly advanced mobility analyzer. (See section by Hoppel and 23 size spectra so produced). From the NRL cumulative size distributions, one can readily calculate slopes (β_i) between two size increments and also average β over any desired size range. Figure 1 illustrates the equivalent k (2/3 β) values versus size for the polydisperse NaCl aerosol generated during Experiment 2. The critical supersaturation S_C scale corresponding to NaCl dry particle size is also shown. For the two salts employed at the Workshop, the approximate function relations between critical supersaturation and dry particle size are:

$$S_c = 1.31 \times 10^{-11} r_d^{-3/2}$$
 (NaC1)
 $S_c = 2.01 \times 10^{-11} r_d^{-3/2}$ (NH₄)₂S0₄ (3)

For experiments with ambient air, it was tentatively assumed that aerosol composition could be approximated by ammonium sulfate particles.

CCN spectra values of k were calculated for selected instruments and experiments, based on final Workshop data printouts provided each participant. Slopes k were determined for rather narrow supersaturation increments (three successive S values) as dictated by each instrument's operating levels. Also k was caluclated for broad S ranges over which each instrument normally functions, i.e., ~ 0.2 -1.5% for static diffusion chambers (SDC) and continuous flow diffusion chambers (CFD) and ~ 0.01 -0.2% for haze chambers.

Listed in Table 1 are the CCN instruments considered in this preliminary analysis.

TABLE 1

CCN Instrument Numbers, Types, and Organizations

Instru- ment No.	Group	Operator	Chamber Type		
10	SUNY	Lala	SDC		
11	NRL	Норре1	Haze		
13	CSIRO	Ayers	SDC		
14	DRI	Hudson	Haze		
15	DRI	Rogers	CFD		
17	NRL	Wojciechowski	SDC		
21	UMR	Alofs	Dual CFD and Haze		

3. RESULTS

3.1 CCN Slopes Above and Below 0.2% S

Table 2 presents the k values for broad S ranges above and below ${\sim}0.2\%$. Theoretical values computed from NRL particle size data (k = 2/3 β) are indicated followed by actual measured values (N= cS^{K}) for given CCN chambers.

Overall, the comparisons between CCN instruments and between calculated and measured k values are quite respectable. This is particularly true when one considers that the instruments were not operated over identical S ranges or time intervals. The former diluting effect can be appreciated from Figure 1, where it is evident that a modest shift in the S range covered would lead to differing k averages, particularly at the large particle (low S) end of the spectrum. Also, time variations in ambient aerosol concentrations were considerable and would influence the comparative results from different instruments requiring anywhere from 10 minutes to 30 minutes, respectively, to complete a The NRL mobility aerosol (size) CCN spectrum. analyzer data represent the smoothed average of typically 2 to 4 size spectra (${\sim}25$ min each) over the duration of an experiment.

While the number of cases is limited, one may note that for ambient aerosols, the slope values for S > 0.2% were typically less than 1, and for S < 0.2% were much steeper (k \gtrsim 2 to 3). This is consistent with prior field measurements and with Junge type aerosol distributions. Laboratory-generated aerosols also showed a steeper slope k at low S, reflecting an analogous greater decrease in large particle concentration with increasing size.

If k values for ambient aerosol monotonically increased with size over, say, the 0.1-1.0% S range as shown for a lab aerosol (Figure 1 and others to follow), then a single k value would hardly suffice.

3.2 CCN Slopes over Narrow Size (S) Increments

Finer increments of the size and supersaturation spectra were examined in terms of their asso-



Figure 1. Slope k derived from mobility analyzer data as a function of dry particle radius (critical supersaturation computed). In this and subsequent figures, each plotted point typically represents the average over the interval bounded by immediately adjacent points.

ciated k values. The results are shown in Figures 2-6 for laboratory and ambient aerosols. Again the solid line represents k values constructed from the NRL MA particle size distributions. Superimposed are measured k values as obtained with the four indicated CCN chambers. Some observations are as follows:

With the possible exception of Experiment 17 (Figure 6), the agreement between theoretical and measured k values is considered good for S > 0.2% for both ambient and laboratory aerosols. Agreement is also reasonably good for S < 0.2% for laboratory aerosols (Figures 2 and 3), but not for ambient air (Figures 4 and 5). In ambient air, the typically higher measured k values for S < 0.2% may be due to sample time and concentration variations or to inherent accuracy limitations of sparse data. However, it may be that particles larger than ~ 0.1 µm with suitable fractions of soluble material to serve as effective CCN have steeper aerosol distribution. Evidence is accumulating (e.g., Meszaros, 1968; Winkler, 1970) to the effect that large particles > 0.1 µ have considerably lower solubility ratios than smaller particles.

In the 0.2-1.0% S range, the ambient aerosols tested do seem to possess a reasonably flat k plateau. This generally was not the case in the .02 to 0.2% S range.

Experiment 17 (Figure 6) was a case in which the ambient aerosol concentration fluctuated greatly with time. A good deal of the scatter in the



Figure 2. Comparison of k values from CCN spectra (four instruments) and from mobility analyzer derived slopes (solid line) -- laboratory aerosol.



Figure 3. Comparison of k values from CCN spectra (four instruments) and from mobility analyzer denived slopes (solid line) -- laboratory aerosol.

	The million multiple and type										
				SDC		CFL)		F	Haze	٦
Ex No.	periment Aerosol	NRL MA+	10	13	17	15	21	NRL MA†	11	14	21
2	NaC1	0.69		0.50	0.49		0.79	2.48	1.78	2.57	1.63
14	(NH ₄) ₂ SO ₄	0.89	0.63	0.71	0.67	0.67	1.0	2.47	2.81	3.31	2.66
23	(NH ₄) ₂ SO ₄	0.96	0.66	1.49	1.07	0.93	1.33	2.62	4.80	4.61	4.16
23	(NH4)2504	1.21	0.90	1.01	1.01	0.93	1.60	3.91	4.82	5.22	3.85
6	Ambient	0.83	0.75	0.74	0.85	0.91		2.28	3.08	3.76	1.83
11	Ambient	0.62		0.37	0.39	0.44	0.51	2.02	2.60	2.29	3.80
17	Ambient	0.93	1.08	0.83	0.63	1.27	2.32	1.95	2.98	2.14	2.71

TABLE 2 CCN Supersaturation Slopes k^{*} Instrument Number and Type

*For SDC and CFD Chambers - S range from \sim 0.2 to 1.5% For Haze Chambers - S range from \sim .01 to 0.2%

+NRL Mobility Analyzer (MA), sometimes referred to in other Workshop data as the NRL Electrical Classifier (EC).
data can undoubtedly be attributed to this factor. Under such circumstances, CCN instruments with short time constants are distinctly advantageous. Otherwise a storage vessel for "grab samples" should be employed. If the NRL mobility analyzer curve is truly representative (and not also influenced by temporal aerosol variations), it is evident that a single k value in the 0.1-1% S range would be questionable.

CONCLUSIONS

Much more is known about CCN concentration variations than about the shape (k) of the CCN supersaturation spectrum. While experiments at the Workshop were not specifically designed to focus on k values, some relevant information was obtained as a byproduct. Some preliminary insights gained were as follows:

a. From Köhler theory and ambient aerosol distributions, it is predicted that $k = 2\beta/3$, where β is the slope of the (soluble) total aerosol size distribution. Workshop results tended to verify this relation.

b. Over the typical S range (0.2 to 1%) of many CCN counters (SDC's and CFD's), the CCN spectrum slope in the ambient air cases was generally uniform; such is necessary for applicability of a single-valued power function of the form N = cS^k . A uniform k does not hold for the laboratory generated aerosols, and is not always an accurate approximation in ambient air.



Figure 4. Comparison of k values from CCN spectra (four instruments) and from mobility analyzer derived slopes (solid line) -- ambient aerosol.



Figure 5. Comparison of k values from CCN spectra (four instruments) and from mobility analyzer derived slopes (solid line) -- ambient aerosol.



Figure 6. Comparison of k values from CCN spectra (four instruments) and from mobility analyzer derived slopes (solid line) -- ambient aerosol.

d. At supersaturations from ~0.01-0.2%, k values were much larger and seemingly less constant over the S range of interest. Future measurements will help determine whether a single power fit is adequate in this range.

The measured k values for ambient air in this low S range were typically larger than those calculated from an assumed soluble aerosol size distribution. This may merely reflect insufficient data and lack of suitable experimental control. However, another hypothesis to examine is whether these larger CCN particles, with suitable effective solubility ratios, decrease in number more rapidly with size than the total aerosol distibution.

f. CCN concentrations can change quite rapidly with time in ambient air. Unless fast-response CCN counters are used or large point-sample containers employed, erroneous k values and power functions can result.

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MEASUREMENT OF THE AEROSOL SIZE DISTRIBUTION WITH NRL'S MOBILITY ANALYZER

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ABSTRACT

The size distribution of the aerosol sample generated at the International CCN Workshop was measured with the NRL mobility analyzer/size spectrometer in the size range between 0.0057 and 0.57 μ m radius. A description of the instrumentation and data analysis is given, together with the measured size distributions calculated for each of 23 experiments carried out at the Workshop.



The size distribution of aersol particles between 0.0057 μ m and 0.57 μ m radius was measured with NRL's mobility analyzer. The basic instrument is shown in Figure 1. The aerosol sample is first brought to charge equilibrium by passing the sample through a region of bipolar ionization. The sample air enters the mobility analyzer through a slit in the outer cylindrical wall where it is confined to a thin laminar layer along the wall by filtered sheath air. A small amount of filtered air is extracted through a slit in the inner electrode. When a voltage is applied to the inner electrode, those particles which are charged and lie in a narrow mobility range will be withdrawn with the extracted air. By measuring the particle concentration in the extracted air as a function of voltage, the mobility distribution of the charged fraction can be obtained. From the mobility distribution of the charged fraction (of one polarity) and the equilibrium (Boltzmann) charge distribution, the size distribution can be calculated. The theory of the analyzer and the analytical procedure for obtaining the size distribution has been given by Hoppel (1978).

The concentration of particles in the extracted air is much lower than in the sample air. The first data obtained with the mobility analyzer was taken by measuring the concentration in the extracted air with a Pollak counter. Although the longtube Pollak counter is probably the most sensitive of the light scattering CN counters, its use imposed rather severe limitations on the sensitivity of the system. Useful size distributions could be obtained in atmospheres which had high concentrations of particles but when the total count dropped below about 1000 particles per $\rm cm^3$ the system could not be used. Even at higher concentrations the sensitivity was not adequate to detect particles in all size channels over the specified range.

In an attempt to increase the sensitivity of the system, a single particle condensation nucleus counter was developed at NRL, which now allows much greater sensitivity and has made it possible to automate the system. While this new continuous particle counter is currently used as part of the mobility analyzer system, it is novel and will possibly have other applications in regard to CCN While this new continuous and CN. and CN. Figure 2 shows the design of the segmented thermal gradient CN counter for single particle counting. The sample air enters along the axis of the chamber and is surrounded by filtered, humidified sheath air. The walls are water saturated with alternating segments maintained at alternating temperatures. The basic principle is similar to that employed in the TGDCC where a supersaturation exists between horizontally oriented hot and cold plates with saturated walls. The TGDCC is limited to measuring particles activated in a narrow range of supersaturations between about 0.2% and 1.0% due to well-documented reasons related to fall-out during the time required to establish the equilibrium supersaturation in the TGDCC. The advantage of the





Figure 1. Diagram of NRL Mobility Analyzer.

segmented chamber is that particles falling vertically are not lost and the segmented geometry suppresses large scale convective motions which would occur between vertically oriented hot and cold walls. The criteria in designing the segmented chamber is that the air flow rate must be such that the residence time in a single segment is short compared to the time required for moisture or heat to diffuse in from the walls to the center; whereas the residence time of the sample in the entire chamber must be longer than the diffusion time. When this criterion is met, the air along the axis will reach a temperature and vapor pressure which is approximately midway between the temperature and vapor pressure of each segment. Along the axis the asymtotic supersaturation will be the same as exists in the center of a TGDCC working at the same temperature difference. Figure 3 shows the development of the supersaturation within the chamber. Details of the design and analysis of the chamber are given by Hoppel, Twomey, and Wojciechowski (1979, 1980). The particles are nucleated and (1979, 1980). grown to optically detectable sizes in the segmented chamber and then transmitted directly into an optical counter for single particle counting. When the segmented chamber is used with the mobility analyzer, a small portion of the output sample from the mobility analyzer is passed from the analyzer to the segmented chamber for counting. At the Workshop, the chamber was operated at a temperature difference of about 20° C which results in a super-saturation of about 20 percent.

The system (mobility analyzer and segmented chamber) has been operated using either 21 or 11 size channels to cover the specified range. After considerable experience with atmospheric measurements, it was decided to standardize to 11 channels when the system was automated. The reason for using fewer channels is that it takes about 18 minutes to cover 11 channels and about twice that long for 21 channels. In the atmosphere the increased resolution offered by 21 channels is less important than making the measurement as quickly as possible because of temporal changes in concentrations. On those occasions at the CCN Workshop when very stable nearly monodisperse particles were generated, it would have been helpful to have had the added resolution of 21 channels.



Figure 2. Design of NRL segmented CN counter for single particle counting.



Figure 3. Supersaturation within segmented CN counter.

The system is automated with the aid of Hewlett Packard 9825 desktop computer with a real time clock. The voltage is changed every 84 seconds by a 16 bit word written to a 16 bit parallel port which controls the output of the high voltage power sup-Sixty seconds are allowed for flushing and plv. stabilization at which time the single particle counter (Royco) is reset and two 12 second counts are recorded, averaged, and stored. The Royco is controlled through a binary port and read with a BCD interface. At the end of the cycle (12 voltage steps including zero volt background count) the computer calculates the mobility distribution and size distribution by an iterative procedure (given by Hoppel, 1978) which corrects for the effect of multiply charged particles according to the Boltz-A computer controlled mann charge distribution. plotter then produces real time plots of the differential and cumulative size distribution. These size distributions were available immediately after Usually several (two to four) each experiment. individual size distributions were taken during each experiment.

After the Workshop, the raw data from the individual runs were examined and averaged. An average size distribution was then calculated for each experiment for which data was available. It is believed that the averaged size distribution is preferred to the individual distribution in all experiments for the following reasons: (1) temporal variations in the concentration may occur between voltage steps and this will distort the size distribution. The more data there is to be averaged, the less will be the distortion due to temporal variations; (2) even if the temporal variations are small, variations due to counting statistics of the samples are reduced by averaging; (3) there are occasions when the high supersaturation in the segmented chamber results in the formation of a droplet at the mouth of the Royco. When this droplet is drawn through the Royco a large spurious count is encountered. This spurious count is easily recognized by examining the count in the different Royco channels recorded as raw data. The spurious reading caused by the droplet results in a large disparity between channels which is easily detected. When the system is working correctly, the count is nearly the same in all Royco channels. In the averaged data, the spurious points are removed before averaging.

The size distributions calculated from the averaged data for each experiment are shown in the 23 figures given at the end of this contribution.

Experiments 27 and 28 were for silver iodide aerosol. The accuracy of the size distribution for these experiments cannot be trusted but the distributions are included for sake of completeness. Sivler iodide is quite insoluble and therefore many of the particles may not have been nucleated at the 20% supersaturation in the segmented chamber. Some compensation was made by increasing the supersaturation during these runs by an unrecorded amount. The raw data from the optical counter did not behave normally in that the number of particles was not the same in all channels, i.e., the particles did not appear to grow in the manner expected for nucleated droplets.

It should be pointed out that the largest channel is uncorrected for multiple charging. In order to correct a channel for the presence of larger particles with multiple charges, some knowledge of the concentration of larger particles must be available. It is therefore impossible to make this correction in the largest channel on the basis of analyzer data. However, if the size distribution is decreasing very rapidly at the largest size (as is nearly always the case), this correction will be small. Since the largest channel is uncorrected it must be viewed with suspicion.

Also, when the cumulative count drops below a few particles per cc, the count cannot be trusted because this corresponds to a raw count which is close to background levels. (This is not an inherent lower limit since this lower limit can be easily decreased further by increasing the dilution ratio in the segmented chamber. The dilution ratio is presently about 1 to 15.)

The accuracy with which the boundaries of the size channels are determined is more easily evaluated than the accuracy with which the total number of particles in that channel can be evaluated. The sizing accuracy is related to the accuracy with which the mobilities are determined. The midpoint mobility of an interval is given by

$$k_1 = \frac{\Phi_1}{4\pi CV}$$
 (esu) = $\frac{\epsilon \Phi_1}{CV}$ (mks)

Where ϕ is the volume flow rate of the sheath air, C the electrical capacity, and V the voltage. The accuracy of C and V are not in question. The airflows were remeasured by the "bubble" method at the altitude of the Workshop and are calibrated to within 3%. The accuracy to which the airflows are set and maintained throughout a given run is about A 5% error in flow results in a 5% error in 5%. mobility. A 5% error in mobility translates into approximately a 5% error in radius at the larger sizes and a 10% error in radius at the smaller sizes. The Stokes-Cunningham-Millikin relationship is used to calculate the radius from mobility. If this relationship introduces no further errors, then the maximum error in the size boundary is certainly less than 15%. It should be pointed out that the radius in the size measurement is the equivalent drag radius which may be slightly different than the "mass" radius required in the Kohler theory.

The accuracy with which the number concentration of particles in a given size channel is known depends on several factors: (1) the detection system, (2) the shape of the size distribution and, (3) the validity of the Boltzmann charge distribution. The nucleated droplets are detected by the Royco particle counter which, in this application, is used as a single particle counter and not as a sizing device. It is generally acknowledged that the absolute counting accuracy of optical counters is much better than their sizing ability. The errors due to counting are believed to be small compared to factors (2) and (3). The method of extracting the mobility distribution assumes that the distribution is nearly linear across any given channel. This assumption is not justified in the case when the distribution is strongly peaked in a single channel as was the case for many of the monodisperse experiments at the Workshop. Since the transmission function is the greatest at the center of the interval and drops to zero at each extreme, it would appear that if the peak in the monodisperse size distribution occurs at the center of one of the (preset) channels, then the total number of particles would be overestimated, whereas if the peak occurs near the boundary of the channel, the total number would be underestimated. Below a radius of about 0.02 μm the validity of the

Boltzmann distribution is in doubt. There are experimental results which indicate that the Boltzmann law hold down to 0.01 μm (Lui and Pui, 1974 and Servaas and Krider, 1977), but on the other hand, there is theoretical justification to indicate that the real distribution departs from Boltzmann below about 0.03 μm and that the ratio of charged to uncharged particles is larger than predicted by the Boltzmann law.

As a result of these three factors, it is difficult to quantify the accuracy of the total numbers of particles measured with the mobility analyzer. However, we believe that the second order mobility analyzer is the most accurate method of measuring the size distribution presently available.

Acknowledgements

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Experiment #1 - Average of 3 Runs











Experiment #18 - Average of 3 Runs

Experiment #20 - Average of 4 Runs



Experiment #22 - Average of 3 Runs (2,3 and 4) [taken between 1100 and 1220; early run (#1) was taken when the count was higher and remains as distributed at Workshop].







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Experiment #28 - Average of 2 Runs

PERFORMANCE OF THE CONTINUOUS FLOW DIFFUSION CHAMBERS

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ABSTRACT

A brief comparative description is made of the five continuous flow chambers which participated in the Workshop. Overall, comparisons for the various types of experiments - monodisperse, polydisperse and ambient aerosol - showed agreement among these chambers to within 15% in most cases. A careful analysis of the results indicated that a proper accounting of certain parameters would bring about much closer agreement among four of these instruments.

1. COMPARATIVE DESCRIPTION OF INSTRUMENTS

There were five continuous flow diffusion chambers in the Workshop. Three of these were from the Desert Research Institute (DRI), one was from the University of Missouri-Rolla (UMR), and the other was from the University of Washington. One of the DRI chambers was the conventional continuous flow diffusion (CFD) chamber, designated C. The second DRI chamber was the rapid cycle spectrometer built for NASA, designated N. The third DRI chamber was the instantaneous spectrometer, designated I. The UMR instrument acted in a dual capacity since it also was used as an isothermal haze chamber. This article will be restricted to analysis of its operation in the CFD mode. The designations M and W will, respectively, denote the UMR and University of Washington instruments.

All of these chambers moved samples of aerosol through parallel plate diffusion chambers after which the drops which formed were detected by optical particle counters. The parallel plates in all five chambers were vertically oriented although the plates to be horizontal to suppress convection. However, the widespread acceptance of vertical plates seems to indicate that this convection is not a serious problem. The direction of the sample flow was not the same in all chambers since the sample moves horizontally in the three DRI chambers (C, N and I) and vertically downward in the M and W chambers. In all five chambers, the sample was confined to a plane midway between the plates by particle-free air. The sample was spread out into a lamina in the plane midway between the plates for all chambers except UMR where the sample was an axial stream. The sample was thus confined in these chambers to keep it in the zone of maximum supersaturation which is approximately midway between the plates. In the DRI and UMR chambers, all of the sample which entered the chambers was detected by the optical particle counters (OPC's), whereas the Washington chamber detected only a small fraction of the sample which was carried through a central tube to an optical box. In all of the chambers, the time of exposure to each

supersaturation was adjusted for maximum performance by using suitable lengths of the wet zones and by adjusting the total flows. The times used in each of the chambers seemed to be consistent when similar supersaturations were considered. The wet surfaces were filter paper for the C and M chambers, metal for the N and I chambers, and felt for the W CFD.

Plate temperatures in the C and I chambers were controlled by water baths whereas the W chamber had refrigeration coils in direct contact with the cold plate. The warm plate of the W chamber was heated by the exhaust gases from the refrigerator along one edge of the warm plate. The N chamber also used circulating fluid but the regulation of that fluid was done with computercontrolled thermoelectric modules.

The M plate temperatures were monitored by checking the reservoir temperatures with mercury thermometers. The C and I CFD's used thermistors imbedded in the plates whereas the N CFD used a thermopile to measure ΔT . The W chamber also used thermisters to monitor the plate temperatures.

The M, C and N chambers obtained spectra by changing plate temperatures. The former two were changed manually while the N CFD changed temperatures according to a prearranged computer program. The I and W chambers did not normally change plate temperatures. The I chamber simultaneously maintained three supersaturations by using three chambers in series and using the droplet distributions from one OPC to deduce the concentrations for three supersaturations. The W instrument had four chambers in parallel and four detectors to monitor the concentrations at four supersaturations.

The DRI chambers (C, I and N) all used Royco 225 optical particle counters while the M chamber used a Climet 201 OPC. The W chamber used four noncommercial OPC's using a single laser for illumination and four photodiodes for detectors. This system does not give as much size discrimination as the commercial OPC's, but accurate sizing is not usually necessary at the higher supersaturations where the W instrument was normally operated.

2. PARTICIPATION AND EXPERIMENT TYPES

Of the 29 experiments, five did not yield data suitable for comparison purposes. Experiments 3, 7 and 17 were ambient aerosols which were not constant enough in time to permit useful comparisons. Experiment 25 was a noise test which all of the participating CFD's passed. Experiment 29 was a truly insoluble aerosol (paraffin) which did not show activated droplets in any of the continuous chambers. Of the 24 useful experiments, the M instrument participated in all except Experiment 21 where its haze chamber mode was used exclusively. The C chamber was involved in all 24 experiments. The DRI spectrometers (N and I) participated in 20 of the 24 experiments as they missed the first 4 (No. 1,2,4 and 5). The W instrument participated in 10 experiments.

The most important division of the experiments is between the monodisperse aerosols on the one hand and the polydisperse and ambient aerosol experiments on the other hand. These two distinct types of aerosols require separate analysis. One conventional way of describing the CCN spectra of aerosols is by noting the concentration, C, at one reference supersaturation (e.g. 1%) and the slope, K, of the distribution. Such a characterization is not applicable to some of the Workshop aerosols, because the slope of the distribution sometimes varied greatly over the range of supersaturations.

3. MONODISPERSE AEROSOL EXPERIMENTS

A monodisperse aerosol is an extreme CCN spectrum because there is a very steep slope over a narrow range of S and a flat distribution over other parts of the spectrum. Hence, this type of spectrum would not be characterized very well by C and K. Instead, a monodisperse aerosol should be characterized by the total number of drops active on the plateau and by the supersaturation (S_{C}) in the very steep region.

Experiments with a monodisperse aerosol allow a determination of the counting ability of the instruments because, as long as the chamber can attain a supersaturation greater than S_C , the concentration should be constant. Thus this concentration is insensitive to supersaturation levels in the chambers. In fact there should be agreement between the CFD's and Aitken particle counters for monodisperse aerosols. On the other hand, S_C , the second parameter which characterizes the monodisperse aerosols should be a very sensitive test of the supersaturations in the chambers.

3.1 Total Concentration

There were 10 monodisperse experiments, Nos. 4,5,8,9,15,18,19,20,21, and 28. An examination of the experimental plots shows that the M, C, I and N chambers always exhibit a constant concentration for supersaturations above the critical supersaturation, S_C of the aerosol when the chambers operated at high enough supersaturations. The W chamber

observed a flat distribution for only one (No. 9) of the four monodisperse experiments in which it participated.

Table 1 shows the total CCN concentrations for each instrument; that is, the concentrations at and above the flat portion of the distribution beyond which no more nuclei are detected. Also listed is the percentage count compared to the M chamber. At the bottom is the average and standard deviation of the comparison with M. The data in Table 1 is averaged over the entire duration of the experiment. Although the aerosol generator was usually quite constant, this data is subject to error due to the fact that the instruments may have been operating at the high supersaturations during different time periods over which the concentration may have changed.

Table 2 was devised in an attempt to cover the above shortcoming. It was derived by choosing time periods during the monodisperse experiments when there was simultaneous data from at least two chambers. Simultaneous Pollak and TSI Aitken (CN) nucleus data is also presented. These should be in agreement with the CCN counters since these are monodisperse aerosols. Also listed in the table is the percentage difference of each CCN counter from the Pollak counter (or TSI when the Pollak data was not available). The last row shows the average and standard deviation of the comparison with the CN counters. These tables are quite consistent showing that Table 2 is fairly representative of the discrepancies between the chambers. This is in spite of the fact that the data in Table 2 is incomplete and rather unequally distributed. Simultaneous times could not be found for some instruments for some experiments while more than one case could be found for some experiments and instruments. Hence Table 2 is more precise, while the data from Table 1 is more representative.

Table 1 shows that the three DRI instruments were in excellent agreement with each other but that they were about 15% lower than the UMR chamber. Since the DRI instruments are so similar, this would seem to represent a systematic error between the UMR and DRI instruments, especially since the standard deviations for the DRI instruments are about equal. As pointed out above, this discrepancy in concentration for monodisperse aerosol reflects actual differences in counting rather than a problem with supersaturation. Either the sample volume or the counting efficiencies are different or the same sample is not being seen by each chamber.

Exp. No.	м	C Ncm	-3		N	I		W	· · · · · · · · · · · · · · · · · · ·
4	260	220	85%			· · · ·			
5	1300	950	73%						
8	310	295	95%	260	84%	295	95%	400	129%
9	∿610	490	79%	540	89%	535	88%	440	72%
15	1630	1290	7 9%	1290	79%	1290	79%		,.
18	1080	900	83%	920	85%	900	83%	1080	100%
19	1030	990	96%	990	96%	990	96%		
20	1030	900	87%	910	88%	970	89%	1020	99%
21	no data	880				880			
28	430	370	80%	330	77%	380	88%		
Averag	es	85%	+ 7%	85%	<u>+</u> 6%	88%	+ 6%	100% +	23%

TABLE 1. MONODISPERSE TOTAL COUNT

Exp.	No. Time	Polla	ak TSI	М		(C	1	N	I	
4	1010-10; 1141-11;	26 262 <u>+</u> 56 282 <u>+</u>	2 275-280 6 300	269	103%	230	82%				
5	1428-144	l0 [∿1304	1] 2800	1405	108%						
	1537-15	53 124	l 2750			968	78%				
8	1005-103	37 273	300-270	308 <u>+</u> 22	113%	284 <u>+</u> 17	104%	304 	111%	297	109%
9	1207-122	21 572	2 570-550	649	113%			642	112%	∿616	108%
	1245-124	17	510	544	*						
	1247-130)2	510-470			504				530	
15	1510-152	9 1683	3 3200-3300	1810	108%	1509	90%	1564	93%	1563	93%
18	1014-102	21 112	2 2400	1168	105%			965	87%		
	1044-10	3 1190	2500			971	82%			968	81%
19	1145-120	06 1052	910-890	1125	106%	1009	95%	1047	99%	1020	96%
20	1400-142	25 1068	900-910	1137	106%			1040	97%	990	93%
	1427-142	29 "	910	1117	105%	961	90%	966	90%	976	91%
	1513-15	6 [108]	900			900	84%			926	86%
28	1532-154	8 39	2 210	438	112%	400	102%		100%	408	104%
	Average	5		108 <u>+</u>	3.6	89 <u>+</u>	9.2	98 <u>+</u>	9	96 🕂	9

TABLE 2. MONODISPERSE TOTAL COUNT, SIMULTANEOUS CASES

* = Haze Mode

 ψ = Abbreviated Experiment

Figures 1-4 were plotted to see if the discrepancies were dependent on particle size. Data from Table 2 was used because of its superior precision. The data from the three DRI chambers (Figs. 2-4) is quite similar and shows a definite size dependence whereas the UMR chamber (Fig. 1) shows only weak size dependence. Loss of particles due to diffus-ion is a size dependent process which could be responsible for some of the discrepancies between these continuous flow chambers and the TSI. Diffusion loss also depends on the sample flow rate and the distance which the sample has to travel. Since these quantities were nearly identical for the DRI chambers, and the DRI chambers yielded similar data, loss by diffusion seems to be a good possibility. Furthermore, diffusional losses for the UMR chamber are expected to be smaller, because the smaller sample flow branched off of a larger delivery flow. Thus the UMR data reflects this fact (Fig. 1).

Figures 5 and 6 show the theoretical diffusion loss rate for the UMR and DRI chambers, respectively. The agreement between the experimental and theoretical curves indicates that diffusion losses can account for the size dependent relationship between the CFD's and the TSI and thus the differences between the UMR and DRI chambers. However, the data seems to show a linear relationship whereas the theory shows a parabolic relationship. Another possible cause of size dependent particle loss is by electrostatic effects. Since the monodisperse particles were produced from an Electrostatic Classifier, they are all charged particles. It has been found that just touching the plastic tubing or even the conducting tubing through which the sample aerosol passed caused reductions in the apparent concentration. This is especially true for the smaller particles which have higher mobilities.

These results indicate that, if theoretical diffusional losses were accounted for, the agreement between the UMR and DRI continuous flow chambers would be much better than Table 1 would indicate. With such a correction, it would appear that the UMR and DRI chambers agree to about 5 percent on the plateau concentration for monodisperse aerosols.

3.2 Critical Supersaturation, S_c

The other parameter used for the monodisperse aerosols is S_c , the critical supersaturation, corresponding to the peak in the differential aerosol size distribution. The measured value of S_c was defined as the supersaturation which yielded one half of the concentration found at the high supersaturations where the concentrations were constant (and presumably all particles were being counted). These one-half values for each chamber and for each monodisperse experiment are listed in Table 3. Also listed are the theoretical values, based on the size from the TSI electrostatic classifier used to generate the aerosols. As to the theoretical



Figure 1. Ratio of concentrations in UMR chamber to those in the TSI CN counter for monodisperse particles of various sizes. Data taken from Table 2.



Figure 2. Same as Fig. 1, for DRJ conventional CFD.



Figure 3. Same as Fig. 1, for DRJ NASA chamber.



Figure 4. Same as Fig. 1, for DRI instantaneous chamber.



Figure 5. Calculated diffusion losses for the UMR chamber for various particle sizes.



Figure 6. Same as Fig. 5, for DRI chambers. All three DRI chambers would have the same diffusion loss characteristics.

Exp. No.	Size/Com- position	Theoreti- cal S _c	M		C		N	1]	[W	
4 5	0.036 NaCl 0.036 Nacl	0.50 0.50	0.52 0.54	104% 108%	0.52 0.52	104% 104%						
8	0.18 NaC1	0.045	0.052	116%	0.084	187%						
9	0.18 NaC1	0.045	0.05	111%	0.065	144%						
15	0.092 Amon	0.18	0.21	86%	0.17	81%	0.2	95%				
18	0.04 Amon	0.64	0.62	85%	0.61	84%	0.54	74%	0.70	96%	0.33	45%
19	0.14 Amon	0.091	0.11	100%	0.12	109%	0.13	118%				
20	0.08 Amon	0.22	0.20	69%	0.18	69%	0.22	85%			0.30	115%
28	AgI		0.33		0.40		0.38		0.40			
Aver [Exc	ages luding Exp.	8,9]	97% 92%	+ 16% + 15%	110% 92%	+ 38% + 16%	93% 93%	+ 18% + 18%				

TABLE 3. MONODISPERSE S CUTOFF

relation between dry size and $S_{\rm C}$, we note that various sources disagree somewhat. For example, for ammonium sulfate at 20°C, Hanel (1976) lists $S_{\rm C}$ values of 1.91% and 0.0532% at dry diameters of 0.02 μm and 0.2 μm respectively. Jiusto and Lala (their equation 3, this proceedings volume) give approximate relation which, for the corresponding conditions give higher values of $S_{\rm C}$ (2.01% and 0.0635%,respectively). For sodium chloride, the two sources agree much more closely. In Table 3, we have used the values from Hanel.

It can be seen in Table 3 that the M, C and N chambers showed excellent agreement with theory and each other. The I chamber was really not participating in this aspect of the experiment since it was always operating at three fixed supersaturations throughout the Workshop. Experiments 8 and 9 were actually below the range of the DRI chambers and within the haze mode of the UMR chamber. Exclusion of these two experiments reveals a more realistic test of the diffusion chambers.

The agreement of the UMR and DRI chambers on $S_{\rm C}$, indicates that these chambers were sensing the same supersaturations. The agreement with theory shows that the deduced supersaturations were probably accurate.

POLYDISPERSE AND AMBIENT EXPERIMENTS

These CCN distributions were more conventional so they could usually be characterized by more typical parameters such as C and K from N = CSK, where C is the concentration at 1% supersaturation and K is the slope of the log-log distribution. However, the calculation of K rests on the assumption of a constant slope and even when this condition is fulfilled, the calcuation of K is one step removed from the data. For these reasons, we decided that a better method of analysis of this data would consist of direct comparisons of all of the chambers at the two most extreme supersaturations which can be used. This would test C and K in a more direct manner without the necessity of a straight line distribution. The two supersaturations chosen were 0.70% and 0.30%.

Tables 4 and 5 show the results of this analysis. There is not a standard instrument to relate the measurements to but, since there are so many similarities between the three DRI instruments, we have again chosen to relate all measurements to the UMR chamber. The fact that the DRI chambers show better agreement with UMR at 0.30% is consistent with the results of the monodisperse experiments. Thus the smallest CCN seemed to be undercounted in the DRI chambers, suggesting that once again diffusion losses or electrostatic charging is a problem. The magnitude of the discrepancy is also quite similar to that found in the monodisperse experiments.

Figure 7 shows that there was probably some undercounting by the DRI instruments at very high concentrations. This data is rather inconclusive except for the concentration of $10,000 \text{ cm}^{-3}$ (Exp.24). Such a graph tests the possibility of effects such as vapor depletion in the cloud chamber or coincidence losses in the OPC. It appears that one of these is a problem in the DRI chambers at very high concentrations.



Figure 7. Ratio of concentration in DRI conventional chamber to the UMR chamber vs. concentration in the DRI chamber. Data taken from Table 4.

TABLE 4. POLYDISPERSE	AND	AMBIENT,	0.7%
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Exp. No.	Composition	м		С	1	N		I		W
1	NaC1	960	700	73%					410	43%
2	NaC1	2000	1300	65%					820	41%
6	Ambient	1200	1125	94%	1300	108%	1175	98%		
10	Bimodal/NaC	1 800	950	119%	800	100%	700	88%	1150	144%
11	Ambient	1250			1100	88%	1030	82%	1040	83%
12	Ambient	2100	2000	95%	2000	95%	2000	95%	1000	48%
13	Ammon. Sul.	2250	2350	104%	2150	96%	2000	89%	1200	53%
14	Ammon. Sul.	3150	2150	68%	2300	73%	2100	67%		
16	Ambient	840	880	105%	800	9 5%	820	98%		
22	Ammon. Sul.	1200	720	60%	720	60%	700	58%		
23	Ammon.Sul.	250	225	90%	235	94%	225	90%		
24	Ammon. Sul.	10800	6 500	60%	5000	46%	5200	48%		
26	Ambient	1100	1140	104%	1150	105%	1140	104%		
26	Run 2/Amb.	1250			1280	102%	1150	92%		
27	Ag I	775	615	79%	400	52%	650	84%		
27	Run 2	510			320	63%	500	98%		
	Averages		86% +	20%	84%	+ 21%	85%	+ 16%	69%	+ 40%

TABLE 5. POLYDISPERSE AND AMBIENT, 0.3%

Exp. No.	Composition	М		С	I	N		I
1	NaC 1	500	435	87%			•• • • • • •	
2	NaC 1	1010	890					
6	Ambient	940	800	85%	700	74%	940	100%
10	Bimodal/NaCl	315	260	83%	280	89%	300	95%
11	Ambient	960	820	85%	700	73%	700	73%
12	Ambient	1000	1200	120%	1200	170%	1200	170%
13	Ammon. Sul.	1300	1010	78%	1010	78%	990	76%
14	Ammon. Sul.	1500	1100	73%	1130	75%	1000	67%
16	Ambient	600	670	112%	540	90%	420	70%
22	Ammon. Sul.	525	450	86%	310	59%	415	79%
23	Ammon. Sul	110	115	105%	100	91%	120	109%
24	Ammon. Sul.	3050	2800	92%	1300	43%	2500	87%
26	Ambient	600	600	100%	460	77%	660	110%
26	Run 2/Amb.	620			620	100%	620	100%
27	AqI	42	21	50%	65	155%	70	167%
27	Run 2	13			25	192%	42	3239
28	Ag I		0.7		101			
	Averages		8 9 % ·	+ 18%	94%	+ 39%	112%	+ 66%

5. CONCLUSIONS

The five continuous flow diffusion chambers showed good agreement with each other in spite of differences such as three types of optical particle counters and three different wicking materials. With the various aerosol distributions and types of particles, the chambers nearly always agreed to within 15% in number concentrations and supersaturation determinations. Error analysis showed that diffusion losses could account for most of these differences.

6. REFERENCES

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REVIEW OF ISOTHERMAL HAZE CHAMBER PERFORMANCE

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ABSTRACT

Isothermal haze chambers (IHC) were present at the 1980 International CCN Workshop (Reno, Nevada); the theory of this method of characterizing cloud condensation nuclei (CCN) over the critical supersaturation range of about 0.01% to 0.2% is reviewed, and guidelines for the design and operation of IHC's are given. IHC data from the International Workshop are presented and critically analyzed. Two of the four IHC's agreed to about 40% over the entire range of S_c 's. A third chamber showed similar agreement with the first two over the lower part of the S_c range but only a factor of two agreement at higher S_c 's. Some reasons for the discrepancies are given.

1. INTRODUCTION

Since the Second International Workshop on Condensation and Ice Nuclei in 1970, the measurement of cloud condensation nuclei (CCN) has been extended to the range of 0.015%-0.15% supersaturation through the use of the isothermal haze chamber (IHC), a new type of CCN counter first described by Laktionov (1972). The principle of operation of the IHC derives from the almost unique relationship which exists between the critical supersaturation, S_c , of a particle and its equilibrium size, r_{100} , at 100% RH. In the IHC, nuclei are grown to their equilibrium sizes in an environment of 100% RH and are then counted as a function of size. From these data and the relationship between S_c and r_{100} , one obtains the CCN supersaturation spectrum. Alofs and Podzimek (1974) are credited with bringing Laktionov's work to the attention of the Western cloud physics community.

Four groups operated IHC's at the International CCN Workshop, Reno, NV, 6-17 October 1980: the Desert Research Institute (DRI), the University of Missouri at Rolla (UMR), the Naval Research Laboratory (NRL), and Colorado State University (CSU). All of these instruments are described in companion papers. In addition, descriptions of the DRI and UMR instruments have been published by Hudson (1980) and Alofs (1978), respectively.

The purpose of this review is threefold: (1) to describe the $S_c - r_{1,0,0}$ relationship; (2) to discuss some of the critical design aspects of the IHC; and (3) to present some results of the IHC intercomparisons conducted at the Workshop.

From the relationship between $S_{\rm C}$ and dry particle radius ($r_{\rm O}$) and that between $r_{1,p,0}$ and $r_{\rm O}$ one obtains the following relationship between $S_{\rm C}$ and $r_{1,p,0}$.

$$S_{c}(\%) = 38.5 \left(\frac{2\sigma_{c}}{\rho_{w}R_{v}T}\right) \left(\frac{\sigma_{c}\hat{1}_{1}\circ\sigma}{\sigma_{1}\circ\sigma\hat{1}_{c}}\right)^{2} r_{1}\circ\sigma^{-1}$$
(1)

In this equation $\sigma_{\rm C}$ and i_c are the surface tension and van't Hoff factor of the solution droplet when it has attained its critical radius; σ'_{10} , and i₁, are the surface tension and van't Hoff factor of the droplet in equilibrium at 100% RH; $\rho_{\rm W}$ is the density of water; T is the temperature; and R_V is the gas constant of water vapor.

The relationship between S_c and r_1 depends on particle composition only to the extent that the value of σ_c (σ_c i σ_1 , σ_1 , σ_c)² depends on composition. If we assume that $i_{100} = i_c$ and that $\sigma_{100} = \sigma_c = \sigma_w$, σ_w being the surface tension of pure water, then Eq. (1) reduces to

$$S_{c}(\%) = 38.5 \left(\frac{2\sigma_{w}}{\rho_{w}R_{v}T}\right) r_{100}^{-1},$$
 (2)

which is the relationship used by Laktionov (1972). At T = 20° C, and with r expressed in microns, Eq. (2) may be written

$$S_{c}(\%) = \frac{0.041}{r_{100}}$$
 (3)

Taking into account the temperature dependence of σ_W , we find that S_C varies approximately as $T^{-3/2}$ for a fixed value of r_{100} .

Corrandini and Tonna (1979) analyzed the S_C vs. $r_{1.00}$ relationship for the main electrolytes composing continental and marine aerosol particles. They

ing continental and marine aerosol particles. They found that the uncertainty in the actual electrolytic composition of CCN results in a maximum deviation from Eq. (2) of approximately 6%.

Hoppel and Fitzgerald (1977) examined the effect of insoluble material in aerosol particles on the relationship between S_c and r, . Their calculations showed that insoluble material will not cause a significant departure from Eq. (2) as long as soluble material accounts for at least 5% of the dry particle mass.

The largest departure from the Laktionov relationship [Eq. (2)] is likely to be caused by the presence of surface-active organic materials in the particles which will lower the droplet surface tension below the value for pure water. In the presence of organic material Eq. (2) overestimates S_c by (σ_w/σ_c^{-}) , where σ_c^{-} is the actual surface tension of the droplet at its critical radius. The surface tension of the droplets formed upon atmospheric aerosol particles is not well known. Hanel (1976) made a few measurements of the surface tension of dilute aqueous solutions of atmospheric aerosol samples collected in central Europe and found values 10% to 30% below the value of pure water.

3. CRITICAL DESIGN ASPECTS OF THE IHC

In addition to the errors in IHC data arising from the lack of specific knowledge about the chemical composition of the aerosol particles, the accuracy of CCN spectra obtained with the IHC also depends on the accuracy with which the equilibrium particle size distribution at 100% RH can be determined. In order to achieve an accurate measurement of this size distribution, the humidifying section of the IHC must be truly isothermal, the relative humidity of the air sample must be brought acceptably close to 100%, the aerosol particles must be allowed enough growth time to reach r and the droplets must be accurately counted and sized. We shall now discuss each of these requirements in turn.

3.1 Isothermal Operation

The diffusivities of heat and water vapor in air are not equal (at standard conditions, the latter exceeds the former by about 15%). Therefore, if an air sample containing CCN flows through a tube having wet walls, the temperature of which increases in the direction of flow, a small supersaturation will be experienced by the CCN. Conversely, if the temperature of the tube decreases along the direction of flow, undersaturation may result. Under either of these conditions, the size of the CCN can depart significantly from r . It is important, therefore, that the humidifying section of the IHC be accurately isothermal and that its temperature be known to sufficient accuracy.

It is, of course, also important to take into account the inherent dependence of r upon temperature. From Eq. (2) we find that knowledge of T to an accuracy of 1°C easily suffices for 1% accuracy in S_c .

3.2 Humidification of the Air Sample

It is not possible to humidify the air entering the IHC to exactly 100% RH since this would require an infinitely long chamber. In practice, then, the sample is brought close enough to saturation that the error in S_c resulting from the departure of the actual equilibrium particle size from r_{10} is acceptably small. Figure 1 shows the relative error in S_c as a function of the deviation in relative humidity from saturation. The quantity 1-S plotted along the abscissa is one minus the saturation ratio. The error has been computed for different values of dry particle radius (r_0) having the indicated values of S_c . For the purpose of these calculations it was assumed that particles are composed of pure ammonium sulfate. We see from Fig. 1, for instance, that in order to operate an IHC as low as $S_c = 0.015\%$ with an error (in S_c) of no more than 10%, the sample must be humidified to 99.99% RH. If we require an error of less than 5%then we must humidify the sample to 99.996% RH. It is also seen that, should the maximum humidity achieved be only 99.95%, then errors in S_c of as much as 40% can be expected. If the CCN Spectrum can be described by the relationship N = CS_c^k , then the error ΔN in particle concentrations due to an error ΔS_c in critical supersaturation is given by $\Delta N/N = -k\Delta S_c/S_c$.

We now turn to the question of how long the humidifying section must be to attain the desired relative humidity. Since most IHC's are of cylindrical geometry, we shall consider the case of air flow through a tube. For flow through a wetted tube, the increase in relative humidity depends only on the length of the tube and the volume flow rate and not on the radius (R) of the tube. The reason for this is that both the time constant for diffusion of water vapor from the walls and the residence time of the sample in the tube are proportional to R^2 . Figure 2 shows the relative humidity at the center of a wetted tube as a function of distance from the inlet and volume flow rate, ϕ , for the case of an air sample having an initial relative humidity of 50%. Figure 2 is based on analytical solutions to the diffusion equations for laminar flow through tubes (Goldstein, 1965). These solutions assume Poiseuille flow and neglect the axial diffusion term. For initial relative humidi-ties of 30% and 70%, the tube length required to raise the relative humidity of the sample above 99.9% is about 5% greater and 10% less, respectively, than the length needed for air entering at 50% RH.

The volume flow rate through the humidifying section is usually determined by the flow requirements of the optical particle counter (OPC) used to count and size the droplets. The Royco 225 optical counter, for instance, uses a flow of 47 $\rm cm^3s^{-1}$ (or 2.8 liters/min). For this flow rate, a tube length of 140 cm is required to humidify the sample from 50% RH to 99.99% RH.

3.3 <u>Residence Time Needed to Achieve Droplet</u> Equilibrium

When an aerosol particle is exposed to an environment of increasing relative humidity its actual size will lag behind its equilibrium size. The smaller the critical supersaturation of a particle, the larger is its equilibrium size, r_{100} , and the longer it takes to grow to that size. Therefore, the humidifying section of the IHC must provide enough residence time at ~100% RH so that the CCN can grow acceptably close to r_{100} .

Laktionov (1972) presented, without documentation, calculated values of the time required for droplets to grow to within 5% of r in an environment of 100% RH. These values have been used in determining the length of, or time required in, the humidifying section of the IHC. Recently, Robinson and Scott (1980) computed growth times of water solution droplets in an IHC, using a new droplet growth equation derived from first-principles kinetic theory. This equation was shown to agree with a conventional growth equation to within 5% for droplets smaller than 1 μ m and to within 0.1% for large radii. Table 1 is taken directly from their paper and shows the time needed for nuclei of varying S_c to grow from their dry size to 95% and 99% of r in a saturated environment. The computations of Robinson and Scott assumed that nuclei are composed entirely of sodium chloride and that the condensation coefficient of water is 0.036. Table 1 also lists Laktionov's values of the time needed



0.00

Figure 2. Relative humidity of the air at the center of a wet-walled tube as a function of distance down the tube and volume flow rate, ϕ . The initial relative humidity of the air is 50%.

Figure 1. Relative error in S, in percent, as a function of the deviation in relative humidity in the IHC from saturation. The error is shown for six values of dry particle radius, r_o , having the indicated values of S_c .

s _c (%)	0.103	0.063	0.051	0.041	0.033	0.026	0.021	0.016
r _{۱۰۰} (µm)	0.40	0.65	0.80	1.0	1.25	1.6	2.0	2.5
t(S) to .95 r ₁₀₀	1.6	4.6	7.4	12.5	21.3	38.9	68.1	121
t(S) to .99 r ₁₀₀	3.3	9.8	16.7	26.6	45.5	84.1	148	262
t(S) [Laktionov] to .95 r	4	10	15	25	40	65	100	180

TABLE 1. TIME REQUIRED FOR DROPLET GROWTH FROM DRY SIZE TO 95 AND 99 PERCENT OF r AT S = 1.0, T = 20°C

to grow to 95% of $r_{1,0}$. It is seen that Laktionov's growth times are significantly greater than those computed by Robinson and Scott.

Since Laktionov did not give the details of his calculations, we do not know the reason for the discrepancy in growth times. If we accept the values of Robinson and Scott, then we see that an IHC must provide over 120 s of growth time at a relative humidity of ~100% if it is to give accurate results at supersaturations as low as 0.015%.

A conservative estimate of the total length needed for the humidifying section may be obtained by adding together the length needed to humidify the air to an acceptable closeness to 100% RH (see previous section) and the additional length necessary to provide 120 s of residence time at ~100% RH.

3.4 Droplet Sizing and Counting

The accuracy of CCN spectra obtained with the IHC also depends on the accuracy with which the droplets emerging from the isothermal (humidifying) chamber are counted and sized. In order to obtain an accurate measurement of the droplet size distribution, the droplets must enter the OPC without serious modification of their numbers or sizes, and the response curve of the OPC must be well known, both in terms of absolute accuracy and resolution.

(a) Entry of Droplets into the OPC

The IHC droplet size distribution can be modified as a result of impaction of droplets on the walls of the inlet tube of the OPC and by evaporation shrinkage of the droplets before they enter the light beam. Alofs (1978) made a thorough investigation of both of these problems as part of an analysis of the performance of his dual-range cloud nucleus counter. In Alofs' device, these two problems may be amplified, compared to other IHC's, because the droplets must pass through a tube 1.3 mm diameter by 12 cm long to reach the OPC.

Alofs (loc. cit.) checked the problem of impaction by varying the flow into his OPC (a Royco at that time) while keeping the main flow through the humidifying chamber constant. The tests were made for 2.5 μ m and 5.0 μ m radius droplets. For each of these sizes, it was found that a plateau of count vs. Royco flow existed (where presumably impaction is not a problem) and that at higher flow rates the count dropped off due to impaction. Losses due to impaction set in at a flow rate of ~1.0 liters/min in the case of the 2.5 μm droplets and at a flow of ~0.5 liters/min for the 5.0 μm droplets. Alofs attributed the loss in counts to the fact that the aerosol stream was not exactly lined up with the 1.3 mm inlet tube. It seems reasonable to expect that in an IHC in which the droplets do not pass through a narrow tube to reach the OPC, higher flow rates can be used before impaction becomes a problem.

The inlet tube of the OPC will normally be warmer than the isothermal chamber temperature due to electrically-generated internal heat in the OPC. This will result in some evaporation of the droplets before they are sized. Alofs (loc. cit.) calculated the change in droplet size due to evaporation in the OPC inlet tube of his device. One may conclude from these calculations that, for nuclei with 0.01% $\leq S_C \leq$ 0.1%, evaporation will be relatively insignificant in an IHC which uses the full manufacturer's flow rate through the OPC (2.8 liters/min in the case of the Royco 225), even if the inlet tube is as much as 0.5°C warmer than the isothermal chamber. However, if the droplet residence time in the OPC inlet is significantly increased due to the use of a reduced flow rate, then it is necessary to avoid evaporation. Alofs showed that the problem of evaporation (of nuclei with S_C \leq 0.1%) could be effectively eliminated by thermostatting the OPC so that the inlet tube is not more than 0.1°C warmer than the humidifying section.

The problems of impaction and evaporation are both very difficult to avoid completely by a priori design provisions. It seems best to allow enough adjustability in the IHC design, that these phenomena can be avoided by experimental adjustments.

(b) Sizing Accuracy of the OPC

In contrast with OPC use with continuous-flow, parallel-plate counters, the OPC on an IHC must not only count, but must also size accurately in order to give the concentration of nuclei as a function of S_C . This means that the response curve (i.e., signal strength as a function of particle radius) of the particular OPC used must be accurately known for the case of water spheres. OPC's are calibrated by the manufacturer with polystyrene latex spheres which have an index of refraction, m, of 1.59. Therefore, use of the OPC with the IHC to size water droplets (m = 1.33) means that the manufacturer's calibration curve must be corrected for index of refraction.

Cooke and Kerker (1975) used a Mie scattering computer program to predict the theoretical response curves of several widely used commercial OPC's, including Royco and Climet devices, for a range of values of refractive index. These curves, for all but the Climet and Royco 220 devices, show an initial monotonic increase in signal with increasing particle size, followed by a region of multivalued response. For particle sizes larger than about 1 μ m radius, the response of OPC's again becomes single valued. The Royco 225, a popular device used by Hudson (1980) and others, is not treated by Cooke and Kerker, but its response should be close to that of the Royco 245, which is treated.

The calculated response curves of Cooke and Kerker are presently the best available basis for extrapolating the manufacturer's polystyrene latex calibration to work with water droplets. In using these curves to correct for index of refraction, a problem arises in that the response curves are calculated for refractive indices 1.33, 1.45, 1.54, and 1.70, but not for index 1.59, that of polystyrene. The operators of IHC's, then, have been forced to interpolate between the curves presented by Cooke and Kerker, a procedure which is made difficult by the oscillatory, mathematically complicated nature of these curves. As pointed out by Hodkinson and Greenfield (1965), the initial monotonically increasing portions of the response curves represent the large-radius end of the Rayleigh scattering region, the upper limit of which occurs at a particle radius of about 0.05 μ m. As long as the monotonic character of the response curves persists above the Rayleigh scattering region, the response can be expected to vary as $(m^{2}-1)^{\frac{2}{2}}$ $(m^{2}+2)^{-\frac{2}{2}}$, and the interpolation between curves to provide response values for other indices of refraction may be carried out on that basis. However, the oscillatory (multivalued) region of many OPC response curves sets in at values of radius well below 0.5 μ m, marking the end of any resemblance to the behavior in the Rayleigh region. Outside the Rayleigh region, the complicated nature of the scattering functions (wherein m appears both as the coefficient and argument of Bessel and Legendre functions) precludes easy interpolation. One then has the choice of either repeating the work of Cooke and Kerker for m = 1.59 or simply doing a visual interpolation based on the Cooke and Kerker curves. There may, of course, be considerable error involved in determining the response curve by the latter method.

OPC size resolution on the order of 0.1 μm is required to be able to distinguish between nuclei differing by as much as 0.02% in S_C. Whitby and Willeke (1979) have estimated that most OPC's in good adjustment will artificially broaden a monodisperse spectrum to the extent that the resulting geometric standard deviation is of order 1.1. This would imply, for example, that slightly over 68% of the count of an ideal 1.0 micron aerosol would be interpreted by the OPC as belonging to aerosol sizes between 0.91 and 1.1 μm .

The actual calibration of any OPC should be checked by the user but, in general, sizing errors are a serious concern, especially if the OPC has a multivalued response function in the radius range of interest.

4. HAZE CHAMBER DATA FROM THE INTERNATIONAL WORKSHOP

4.1 Preliminary Comments

While IHC's were operational during all 29 experiments of the International Workshop, useable data resulted from only a subset of the total number of experiments when the aerosol was large enough to register and when other problems such as fluctuations were not present. In addition, the four instruments (#11, Naval Research Laboratory; #14, Desert Research Institute; #21, University of Missouri; #26, Colorado State University) were not always simultaneously operational. Therefore, the data to be discussed is a subset of the total data file, which will generally cover those experiments where the aerosol was suitable for IHC and all four IHC's were operational.

A few remarks about the instruments themselves are worthwhile. IHC #21 is unlike the others in that it uses a Climet OPC for detection and sizing of the haze droplets. IHC #11 is actually a second-generation device, built by the Naval Research Laboratory after considerable experience had been gained with an earlier model. In particular, a very efficient sheath air pre-humidifier is found on this version. IHC #14 and #26 are related in terms of design; #26 was constructed following, to some extent, the mechanical details of #14 but was only newly-completed at the time of this Workshop. Some difficulties in the operation of #26 were experienced, but allowance should be made for the fact that it was a newly-constructed device. Finally, one should generally note that IHC's are required to respond to concentrations that span many (eg, 3-5) orders of magnitude, a considerable dynamic range which is taxing under the best of circumstances.

In the process of analyzing the data, it was discovered that somewhat different procedures were used to deal with the index of refraction problem cited in Section 3.4(b) of this report. Although DRI and NRL both used Royco OPC's it was determined that subtle differences in the procedure used to correct for the index of refraction resulted in some significant differences in the data. Specifically, this meant that different Sc's were deduced from voltage threshold responses which would have shown identical sizes for the polystyrene latex spheres used for size calibrations. After this was discovered, the DRI data were redone using the method similar to that used by NRL. Since the CSU data were treated identically to NRL data, this modification then allowed all Royco-equipped IHC's to be compared on an equivalent basis. It is difficult to determine which procedure is best, although the NRL procedure does yield higher concentrations, at a given S_c , in better agreement with the UMR chamber and with the mobility analyzer.

The operators of the UMR chamber accounted for the index of refraction in an entirely different While the procedures used for the Roycomanner. equipped chambers were based solely on theory, the UMR IHC relied most heavily on an empirical technique. Size thresholds were determined by passing known sized NaCl nuclei through the UMR IHC; the nuclei would form drops which were assumed to be the equilibrium size of solution droplets at 100% This procedure was followed for one or two R.H. sizes and the rest of the curve was completed by using the functional shape of the Cooke and Kerker results. This circumvents the index of refraction problems which have been described in Section 3.4(b) and in the last paragraph for the Roycos.

4.2 Best and Typical Agreement Cases

In Figure 3, the IHC results of Experiment 20 (monodisperse ammonium sulfate aerosol) are shown; the data provide the best agreement between IHC's found during the Workshop. The curve labeled #14' is the original DRI data; #14 has been redone with the index of refraction correction treated similarly to the data of #11 and #26. This yielded much better agreement with #11 and #21. Note that the theoretical CCN spectra derived from the NRL mobility analyzer data (#12) are also plotted; these were obtained by using the theoretical (Kohler) relationship between S_c and dry particle size for ammonium sulfate aerosols to transform the NRL size distribution data into a CCN spectrum. (IHC #26 did not report data for this experiment.)

Figure 4 shows a count-versus- S_c plot of Experiment 9, monodisperse sodium chloride, and represents the more "typical" case from the Workshop. Once again, it is apparent that #14' shows much closer agreement with #11, #12, and #21. The difference between #14 and #14' shows the sensitivity to the method used to correct for the index of refraction. In contrast to Figure 3, it was more frequently the case that the count magnitudes were in the order shown in Figure 4 at the higher S_c 's: IHC #11 and #21 highest, #14' next, then #14, and #26 lowest. At the lowest supersaturations, however, curves 14', 11, and 21 were in no consistent



Figure 3. Concentrations registered by isothermal haze chambers and NRL mobility analyzer as a function of supersaturation setting, for Experiment 20. Aerosol was monodisperse ammonium sulfate. Mobility analyzer data are labelled by instrument Number 12. Haze chamber data are similarly labelled; see text for instrument numbers.

order and agreement between the three IHC's was much better.

Not shown in Figure 4 are the counts of the continuous-flow CCN counters. In Figure 3 this data shows a "plateau" in count; the mean $S_{\rm C}$ of this nominally monodisperse aerosol was 0.045%, with the smallest CCN in the distribution exhibiting ${\rm S}_{\rm C}$ as high as 0.06% to 0.1% supersaturation. For this experiment the counts from those devices generally fell in the range 500-600, or at about the same magnitude as indicated at 0.15% by IHC #11, #21, and #14. IHC's #21 and #14 (#14' also) detected a plateau in number vs. S_c which allowed them to estimate the S_c of the aerosol. These were all overestimates of the theoretical value 0.045%; #21 saw 0.055%, #14' estimated 0.073%, #14 0.086%. Although IHC #11 did not detect a plateau this may have just been due to the OPC channel threshold selections. If 0.15% were assumed to be a plateau reading, #11 would estimate 0.073%. Experiment #8 was the only other experiment which used an $S_{\rm C}$ in the THC name 0.045% as Experiment 9. Although the concentration was lower in Experiment 8, the results were nearly identical to Experiment 9. IHC #26 often exhibited a kink in the region around and immediately below 0.1%; this kink may correspond to the double-valued region of the particular Royco in use on IHC #26.

The relative order (by magnitude of count at a given S_c) of the four instruments was preserved

throughout most of the experiments. Considering that three of the devices utilized a Royco OPC, the double-valued response region of which falls in the approximate S_C interval of 0.04% to 0.08% (r_{100} of 0.5x10⁻⁴cm to 1.0x10⁻⁴cm), it may be inferred that at least a given Royco OPC seems to consistently place signals of the same magnitude in the same electronic channel, even though the signals may fall within the double-valued region. Otherwise, interpretation of data from the IHC's using the Royco (IHC #11, #14, #26) would be next to impossible in the 0.04% to 0.08% S_C range. (Even granting this consistency, it is not too surprising to see discrepancies in this troublesome region of the spectrum.)

4.3 <u>Relative IHC Results as a Function of</u> Experiment

Figures 5 and 6 simply display the relative counts of each IHC as a function of experiment number, where the relative count is shown as a percentage of the NRL mobility analyzer count at the same S ($S_c = 0.09\%$ for Fig. 5 and 0.15\% for Fig. 6). The curves on Figs. 5 and 6 show discontinuities where data are inappropriate or missing. These S_c values were chosen somewhat arbitrarily; our intention is simply to display relative performance at representative points along the S_c spectrum.

In Figure 5 once again #14' is plotted. As in the last two figures, this represents DRI data which has been corrected for the index of refraction in a manner similar to that used by the other



Figure 4. Same as Figure 3, but for Experiment 9. Aerosol was monoidisperse sodium chloride.





Figure 5. Isothermal haze chamber data compared to NRL mobility analyzer data, plotted as a function of experiment number, for a supersaturation setting of 0.09%.

two Royco equipped IHC's. No. 14' is not plotted in Figure 6 because at 0.15% the ratio of #14' to #14 is only about 1.15 except in Experiments #22, 23 and 24 where it is about 1.5, causing only a small change in the relative positions of the data from the various instruments.

In examining these figures, it is apparent that the relative order of the instruments is generally preserved although #11 and #21 seem to alternate as the highest reading IHC. A similar figure for $S_c = 0.04\%$ was constructed, but had very large gaps in data, and considerably more scatter than seen in Figs. 5 and 6. Nevertheless, the order was like that of Figure 5, except that #14' alternated with #11 and #21 for highest readings. These data cover ambient aerosols (Experiments 6 and 11), mono-disperse sodium chloride (8 and 9), polydisperse sodium chloride (1,2,10), monodisperse ammonium sulfate (15, 19,20) and polydisperse ammonium sulfate (13, 14,22,23,24). The maxima and minima of the curves of the four instruments sometimes seem to occur in unison, but are not clearly related to any Rather, it appears particular type of aerosol. that a significant variable might be the response of the NRL mobility analyzer, more than the type of aerosol under study. From experiment to experiment, small changes were occasionally made in the sample flow rate (time of exposure of the sample to saturation) of instrument #14. Although these should have been within the region where counts have plateau values as a function of carrier flow, this may not have always been the case for all of the OPC droplet size thresholds used.

Figure 6. Same as Figure 5, but for a supersaturation setting of 0.15%.

Figure 6 takes on additional significance if we regard 0.15% as representative of the supersaturation range where haze chamber data overlaps that of the static and continuous-flow CCN counters. Inspection of the Workshop data shows that the NRL mobility analyzer usually read somewhat higher than the approximate average of the static and continuous counters, by a factor between one and two. Therefore, if those CCN counters were also shown on Figure 6, their results would fall in about the same range as given by the curve for IHC #21, a result that is not too surprising because this IHC always gave data which was continuous with the higher supersaturation data given by the dual-mode operation of chamber #21. Thus, both IHC #21 and IHC #11 tended to agree well with other chambers around 0.15% supersaturation. The other two IHC's were less often in agreement in this region, although often #14 seemed to give data which joined well with the data of a continuous CCN counter from the same institution, #15.

Although Figures 5 and 6 indicate that there is a considerable counting discrepancy between the various IHC's, it should be kept in mind that when the slope of the CCN spectrum is very steep (as was often the case in the range of S_C of the IHC's), a small absolute error in S_C can result in a large error in particle count. Thus small discrepancies in the sizes assigned to given water droplets by various OPC's could lead to large differences in count; ignoring for the moment all other sources of the discrepancies, the primary calibrations (even of OPC's from one manufacturer) of OPC's differ from instrument to instrument. In regard to the present data, clearly the most desirable approach would have been to compare the OPC's (at least the three Roycos) on a non-water aerosol such as polystyrene latex before or during the Workshop.

4.4 Conclusions

The general tradition in comparisons of CCN counters has been to regard the instruments which register the highest counts as the most correct. That tradition should never be accepted without scrutiny, for instruments may overcount as well as undercount, although generally there are more reasons for the latter. In the case of these com-parisons, there are additional measurements which drive one to the conclusion that the higher-counting instruments are indeed to be favored; those measurements are the continuous-flow and static CCN counts, in the case of monodisperse aerosols where a plateau in the count-versus- S_c is to be found, and the results of the NRL mobility analyzer. It would appear, then, that instruments #11 and #21 offer some advantages over the other devices.

The differences between #14 and #14' point out the problems involved in correcting for the index of refraction. Although these differences at first appeared to be very minor, they resulted in signifi-cant differences in the region of supersaturation between 0.04% and 0.08%, corresponding to the double valued region of the Royco. The steepness of the aerosol distribution in this region results in a very high sensitivity to this variability. Fortunately all of the IHC's usually avoided using voltage thresholds in this region of the spectrum. Hence the differences involved in correcting for index of refraction were usually not as serious as they could have been. Although the differences which resulted from variations in the procedure are significant they are not overwhelming. Nevertheless, it would have been much better to use a uniform method of correcting for index of refraction at the time of the Workshop as corrections made later are not certain to be identical. In the future, it would be desirable to look into the whole question more thoroughly.

IHC #21 is different in that it uses a Climet OPC. However, it is difficult to compare the treatment applied by its operator to the index of refraction problem, which unlike the Royco-equipped chambers, does not strictly rely on theoretical considerations. Hence it is difficult to determine if the Climet OPC offers advantages in sizing water droplets based upon calibration with polystyrene latex spheres of a different index of refraction. Nonetheless, the UMR IHC does show good number concentration agreement with the NRL mobility analyzer as well as better agreement with the theoretical S_{c} 's in the monodisperse experiments (8 and 9). This is understandable since the UMR chamber was essentially calibrated with a mobility classi-fier, which may be a better method for applied CCN counting. Nevertheless even though the Royco equipped IHC's have considerable calibration problems they do at least use an independent method of calibration. There are more occasions when IHC #11 counts higher than the NRL mobility analyzer, than is the case with IHC #21, but such occasions are rare enough that no conclusion can be drawn.

All IHC's in this study were basically stable, rugged devices which at least showed an output data

signal which was proportional to the input aerosol concentration, despite the taxing dynamic range and requirement that OPC accuracy and resolution be pushed to the limit. These devices clearly have a promising future in certain measurements of hygroscopic aerosols, such as the counting of fog condensation nuclei, or even in new applications such as studies of aerosol retention in the human lungs.

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AEROSOL GENERATION AND DISTRIBUTION SYSTEM FOR THE THIRD INTERNATIONAL CLOUD CONDENSATION NUCLEI WORKSHOP

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ABSTRACT

In order to obtain identical samples participating CCN instruments and aerosol characterizing equipment were located along and connected to a 8.2 cm diameter aluminum tube through which the test aerosols were pumped directly from the source at a rate of 200 to 600 & min⁻¹ and at very slight overpressure.

Of the total of 29 experiments, 18 were carried out with artificial NaCl or $(NH_4)_2SO_4$ aerosols. These were generated from salt solutions by pneumatic atomizers of special design (by DRI) to ensure high constancy of the aerosol output concentration. In three experiments with insoluble CCN (AgI, paraffin wax) the aerosols were generated thermally. In some of the tests, an electrostatic classifier was used for narrowing the particle size distributions.

1. INTRODUCTION

Comparisons or cross-calibrations of small particle instrumentation are often performed, for convenience's sake, with the aid of most easily obtainable aerosols, such as the ones present in room or atmospheric air, the particles of which are usually not well characterized and are mostly of a rather complex nature. However, since different aerosol instruments generally differ in their response spectra, these differences can only be assessed and ultimately understood if aerosols of distinct characteristics are utilized. Thus, in order to achieve the objectives of the Workshop, it was imperative to provide test aerosols of high monodispersity and chemical purity, and to take the utmost care in conveying the sample to the instruments.

2. AEROSOL DISTRIBUTION SYSTEM

Based on the favorable experience in previous Workshops (WS), the primary requirement of supplying all instruments simultaneously with as nearly identical samples as possible was met by locating the instruments along a sample duct (or manifold) in which a flow rate was maintained that was high compared to the sample flow extracted from the duct by the instruments.

A further requirement was to provide a steady flow of aerosol in which variations of particle concentration with time were less than a few percent ($\leq \pm 3\%$, according to recommendations from the Steering Committee). In order to achieve this criterion, the choice was to either use a large storage bag (as had been the case at the Ft. Collins WS) or to operate a very constant aerosol source.

Several considerations led to the decision to deliver aerosols directly into the sampling duct instead of using a large bag. With regard to artificial aerosols, generation techniques at the DRI had improved over the years to the point where runs of one to two hours could be achieved with about 90% probability of keeping the particle con-centration within $\pm 5\%$ of a mean value whereby the short term fluctuations (which are generally more detrimental for instrument comparisons) usually were even smaller; superimposed on this was a slow drift in the particle concentration. Furthermore, examination of the Ft. Collins WS data (Grant, 1971) indicated that samples stored in the 54 m³ bag were decaying at a considerable rate; total NaCl particle concentration, -10% hr⁻¹; large sub-micron particles, \sim -50% hr⁻¹; and -20 to -30% hr⁻¹ for the CCN of atmospheric samples active at 1% supersaturation. These values did not point to an advantage of a bag system, not even for the case of ambient aerosol when compared with the low level of fluctuations in the prevailing situations of westerly winds usually experienced at the DRI during routine direct measurements of the atmospheric aero-Another disadvantage of a bag system is its sol. finite volume dictating the duration of an experi-ment (which may be too short for certain investigations; some participants had indicated the desirability of prolonging some experiments well beyond an hour).

Since the DRI aerosol laboratory was already equipped with a duct system to aspirate outside air through an inlet 9 m above the roof (i.e., ~ 25 m above ground level) for distribution to instruments, it was relatively easy to extend that system to the WS room. Figure 1 shows the floor plan of the 150 m² WS area and the layout of the sample duct. The numbers 1 to 28 refer to the sampling outlets spaced about 0.6 m apart along the 8.2 cm ID aluminum duct. The sampling outlets were of 5 mm stainless steel tubing having a 90° bend inside the duct, pointing into the air flow; this resulted in isokinetic sampling conditions for instrument sampling rates of 0.0037 times the flow rate in the duct (which was varied between 200 and 700 k min⁻¹ from one experiment to another). Also shown are the sampling locations of all the instruments (identified by type and participating organization). In order to minimize the risk of losing particles

Unfortunately, unusually calm weather persisting during the period of the WS, combined with local industrial sources, was causing considerable fluctuations in aerosol concentration.



through charge accumulation on surfaces, all instruments were connected to the sample duct by electrically conductive latex rubber tubing which had been tested previously and found to have no adverse effects on aerosol sampling.

Although the distance between the first and last sample outlet was 18 meters, the sample time delay between first and last instrument was only 8 to 25 seconds depending on the sample flow velocity of 0.6 to 2 m s⁻¹. Since significant changes in aerosol characteristics (mainly concentration) generally were associated with considerably longer time constants, simultaneous sampling by all instruments can be assumed for all but a few circumstances. By measuring the total particle concentration at sample outlets 3 and 28 with the same instrument (alternating between the two outlets) it was determined that no significant decay of the aerosol along the duct was taking place.

In order to prevent contaminants from entering the sampling duct, the latter was kept at slightly higher than room pressure, i.e., 0.05 to 0.5 mb, depending on sample flow rate. When using artificial aerosols, the compressed air generation system provided the needed overpressure while, in the case of ambient air sampling, an axial fan (Rotron Model TN3A2) built into the sampling duct upstream of the WS area produced the pressure difference. It was determined experimentally that the fan caused a reduction in total particle concentration in the ambient aerosol by about 5% but, more importantly, that its motor did not generate any particles.

AEROSOL GENERATION SYSTEM

For reasons mentioned previously, more emphasis was placed on work with artificial, well defined CCN which were used in 21 of the experiments while the ambient aerosol served in only eight comparison tests.

In order to cover the basic types of soluble, insoluble and hydrophobic aerosols, NaCl and $(NH_{4})_{2}SO_{4}$ were selected for the first category due to their relevance to the real atmospheric aerosol. The choice for the water insoluble CCN material fell on pure AgI, mainly due to the relative ease with which this highly insoluble substance can be dispersed; it has to be pointed out that these AgI particles most likely do not resemble the ones generated for cloud seeding purposes which usually possess a water soluble component. Originally, the candidate material for hydrophobic CCN was Teflon (briefly used at the Ft. Collins WS), but the lack of a satisfactory dispersal method led to the use of paraffin wax instead.

Parameters to be varied other than the chemical make-up were the total particle number concentration (a few hundred to several thousand per cm^3), and the particle size distribution ("monodisperse", polydisperse and mode at several different sizes).

The experimental set-up for producing these aerosols is shown schematically in Figure 2. The aerosol generator, in this case a pneumatic atomizer for the production of water soluble CCN from salt solutions, is depicted at the top right of the figure. While many types of atomizers, especially inhalation therapy nebulizers, could, in principle, have been used for the present purpose, the requirements of constant output and prolonged operation demanded the design of a specialized and more versatile device as described in detail by Dea and Katz (1981) in a subsequent article.

In order to minimize contamination while providing a wide range of pressures needed for control of output parameters, the atomizer was operated with high grade compressed air from commercial tanks. The salt solutions were prepared from J.T. Baker and Fisher Scientific reagent grade salts and HPLC water (J.T. Baker Chemical Co.). The concentration of the salt solutions was varied according to the particle size requirements of individual experiments. After emerging from the atomizer, the fine mist of solution droplets was diluted about



Figure 2. Schematic of Aerosol Generation System.

1:1 with dry filtered air, mainly to promote rapid evaporation of the droplets, but also to help reduce coagulation which is considerable at typical number densities of the order of 10^8cm^{-3} .

After complete removal of liquid water from the particles in the diffusion dryer (TSI Model 3062), the aerosol was brought to charge equilibrium by passing it through a charge neutralizer equipped with a 10 mCi ${\rm Kr}^{85}$ source. While this procedure helped reduce particle losses caused by charge effects, it was mainly preparatory to extracting a quasi-monodisperse fraction from the total atomizer output by means of an electrostatic classifier (EC, TSI Model 3071). Since theory and operation of this device are discussed extensively in a variety of papers (Knutson, 1975, 1976; Hoppel, 1978), only the following points most relevant to the WS application will be mentioned: The fact that the particles are classified by electrical mobility means that, in addition to particles of the desired size, a small but not negligible percentage of larger, multiply-charged particles were included in the nominally monodisperse aerosol leaving the EC. Also, the population of particles having essentially the selected size is not truly monodisperse but has a size distribution of finite width which depends on the flow conditions through the EC - a typical geometric standard deviation being about 1.2. Since laminar flow inside the EC is essential, the aerosol flow in and out of the EC is limited, in the present case, to about 100 cm³s⁻¹. Due to this condition, it was necessary to discard excess aerosol prior to entering the EC (the filter shown in that line on Figure 2 serves to protect the flow control valve and flow meter from salt deposits). Furthermore, this flow limitation in the EC, combined with a maximum particle number concentration imposed by coagulation rates, resulted in a maximum rate on the order of 10^7s^{-1} at which "monodisperse" particles could be introduced into the sampling duct.

In order to achieve the necessary flow in the sampling duct (cf. previous section), the aerosol was injected ax-ially into a turbulent stream of 200 to 600 g min⁻¹ clean dilution air. The 3 cm diameter dilution and mixing section shown in the center of Figure 2 incorporated an optional 200 & buffer volume to smoothen high frequency fluctuations in aerosol concentration. Due to the large volume required, the dilution air had to be drawn from the house compressed air system which, in turn, necessitated particularly careful purification with oversized absolute and charcoal filters, especially in view of the fact that the dilution air constituted 97 to 99% of the air in the final aerosol delivered to the sampling duct. For flows under 600 & min⁻¹, it was possible to keep the rate constant within two percent.

As the bottom part of Figure 2 indicates, the artificial aerosol was fed into the sampling duct 26 m from the first sampling outlet, just downstream of the axial fan. A 6 cm ball valve was used to seal off the ambient aerosol branch of the duct during artificial aerosol experiments.

The upper left-hand side of Figure 2 displays the location of an auxiliary aerosol generation system which was essentially a duplicate of the one described above. The auxiliary system was equipped with commercial inhalation type nebulizers and operated with filtered compressed air from the house supply. This system was used during some preliminary WS experiments and also during Experiment No. 10 where two "monodisperse" aerosols with different particle size were mixed.

In those experiments where no monodisperse aerosol was desired, the EC was by-passed as indicated by the dashed line in Figure 2. In contrast to the monodisperse case, the problem facing us here was to produce a sufficiently low particle concentration to be meaningful and acceptable for CCN instruments. Reduction of the air pressure on the atomizer to only a few psi drastically lowered the number output but, at the same time, reduced the output stability markedly. Similarly, when nearly all the aerosol originating from the atomizer was removed by aspiration into the excess aerosol line, the small differential flow into the sampling system underwent amplification of the small irregularities in the atomizer or excess flows. However, by applying both methods in moderation it was possible to provide a final particle concentration in the sampling duct as low as 400 cm^{-3} .

For the few experiments carried out with other than soluble CCN, the atomizer in Fig. 2 was replaced with the devices shown in Figs. 3 and 4.

AgI (purified, Fisher Scientific) was aerosolized thermally by bringing a small amount of the substance $(\sim 1 \text{ g})$ slightly above its melting point (550°C) by means of the arrangement shown in Fig. 3. The AgI was placed in a tantalum boat which was mounted on two copper electrodes inside a Pyrex The temperature of the Ta-boat was conflask. trolled by manually regulating the low voltage heating current. A stream of dry N2 (several \imath min^-1) directed at the hot AgI served to quench the vapors thus inducing recondensation of AgI particles. The resulting aerosol left the generator through the side tubulation and subsequently underwent the same handling as the soluble aerosols. In order to provide assurance that no parts of the generator other than the AgI were participating in aerosol forma-tion, the device was tested without AgI; not until reaching temperatures well above 600° C did particle generation from the hot Ta set in. While temporal stability of the polydisperse AgI aerosol was quite within the required tolerances, the monodisperse case suffered from deviations in concentration of up to 10% from the mean.



Figure 3. Agl Aerosol Generator

Preparation of the paraffin wax aerosol followed similar procedures. About 50 ml of paraffin were kept molten in a Pyrex flask at constant temperature by immersion in a bath of boiling water. A jet of dry N₂ from a capillary tube was blown vertically onto the liquid paraffin surface, again triggering particle formation through a quenching action. The polydisperse aerosol fluctuated in concentration by about 5%. Since the particle concentration was only in the vicinity of 400 cm⁻³ and most CCN instruments were able to detect only a small percentage of the hydrophobic particles, no experiment with monodisperse paraffin aerosol was carried out.

In order to facilitate interpretation of the CCN instruments' response to the test aerosols, a few WS participants measured size distributions



Figure 4. Paraffin Aerosol Generator

(Hoppel, 1981; Rogers, 1981) and total number concentrations (Rogers and McKenzie, 1981), while Mach and Hucek (1981) performed some chemical analyses; the reader is referred to the respective papers for additional details.

4. ACKNOWLEDGEMENTS

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ELEMENTAL COMPOSITION OF AEROSOLS IN FOURTEEN EXPERIMENTS OF THE CLOUD CONDENSATION NUCLEI WORKSHOP

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ABSTRACT

International Cloud Condensation Nuclei (CCN) Workshop aerosols were collected with two Ci Impactors and analyzed with proton induced X-ray emission (PIXE) for chemical composition and to detect if contamination was present. One of the impactors sampled the generated aerosols; the other impactor sampled the generated aerosols; the other impactor sampled droplets from a diffusion cloud chamber. The purpose of the experiments was to test the feasibility of a study of the transfer of chemical elements from the fine particle sizes to the coarse particle sizes, after CCN are activated and cloud droplets are formed. The data indicated that sulfur-containing aerosols did exhibit the expected transfer.

1. INTRODUCTION

The aerosols generated during fourteen experiments of the International Cloud Condensation Nuclei Workshop were collected with seven stage, single orifice, Battelle-type cascade impactors (Mitchell and Pilcher, 1959) and analyzed with proton induced X-ray emission (PIXE) (Johansson et al, 1975) for chemical composition. Seven experiments were associated with the generation of aerosols of ammonium sulfate, three with sodium chloride, two with silver iodide, and two with ambient aerosols. A sample of AgI powder used to generate the aero-sols was also analyzed with PIXE. Katz and Dea (1980) described the source of this AgI powder. The aerosols were analyzed for the mass in nano-grams of Na, Mg, Al, Si, P, S, Cl, K, Ca, Ti, Cr, V, Mn, Fe, Ni, Cu, Zn, Br, Pb, Ag. The aerosols generated during the workshop did not always contain these elements; however, the aerosols were analyzed to detect any contamination. Several experiments were combined into one sampling period with the impactors. The aerosols were sampled in the experiments: 8-9, 10, 13-14, 15, 18-19, 20, 22, 24, 26, 27, 28.

2. DESCRIPTION OF INSTRUMENT AND ANALYSIS

Two cascade impactors were used to collect the aerosols. One impactor was connected to the tube that supplied aerosols to each experimenter, and a second impactor was placed behind a continuous flow diffusion (CFD) chamber that is similar to the CFD described by Hudson and Squires (1973, 1976). Both impactors have six stages with a Mylar film for collecting the aerosols and one stage with a 0.4 μ m Nuclepore filter. The six stages separate the sampled aerosol into six size ranges with > 8, 8-4, 4-2, 2-1, 1-0.5, 0.5-0.25 μ m aerodynamic diameter. Aerosols with aerodynamic diameters less than 0.25 μ m were collected on the seventh stage with the Nuclepore filter. This Nuclepore filter limited the flow through the impactor to approximately 1 1/min. The first five Mylar films were coated with Vaseline and the sixth Mylar film was coated with paraffin. Each set of six coated Mylar films and one Nuclepore filter were matched with Mylar films and Nuclepore filter that were not exposed to the aerosol but were analyzed with PIXE to establish the background chemical composition of the filters.

The continuous flow diffusion chamber has two aluminum plates that are 50 cm long, 33 cm wide, and separated by 1.5 cm. The temperature of each plate is maintained with a circulating water bath, and the inner surface of each plate is covered with a black cotton cloth. Each cloth is wetted with water from the appropriate water bath. The chamber was normally operated with a temperature difference across the plates that produced a supersaturation of approximately 1%. The aerosol is sampled at the entrance of the chamber through a capillary tube that restricted the flow to about 0.25 l/min. A sheath flow of air maintained the aerosol in the middle of the chamber and provided air to supply the 1 l/min required for the cascade impactor at the exit port of the chamber.

In the analysis of the aerosols with PIXE, the aerosol sample is exposed to 5 MeV protons. The resulting X-ray spectrum is decomposed with a modified version of a computer program described by Kaufmann et al, 1977. Constants in this program are specified after the analysis of the spectra from a set of standards, which are samples of compounds or elements whose mass is known to within $\pm 5\%$. The analysis by PIXE can provide a mass measurement to within ± 5 nanograms.

3. DISCUSSION OF EXPERIMENTS

The purpose of these experiments with the two cascade impactors was to test the feasibility of a study of the transfer of chemical elements from the fine particle sizes, where d < 0.5 μ m, to the coarse particle sizes, where d > 4 μ m, after condensation of water vapor produces droplets. The chemical elements in those aerosol particles that are cloud condensation nuclei (CCN) will be transferred to the larger sizes; those elements in the aerosol particles that are not CCN, however, will remain at the smaller sizes. Several experiments with the two cascade impactors did indicate a possible transfer of sulfur. For example, a comparison of the sulfur collected on each stage of the impactor during experiments 13 and 14 is shown in Figure 1. These two experiments were sampled with the same filters to obtain more mass on each stage of the



Figure 1. Comparison of mass of sulfur on seven stages of two Cascade Impactors. Stage 1 collects aerosol particles with an aerodynamic diameter $d < 0.25 \ \mu m$, and stage 7 collects particles with $d > 8 \ \mu m$. The mass collected by the impactor behind the CFD is shown with the dashed line, and the mass supplied to the CFD is shown with the solid line. The mass collected by each impactor has been normalized with the total mass collected on the impactor.

impactors. Stage 1 is the Nuclepore filter that collects aerosol particles with $d < 0.25~\mu\text{m}$, and stage seven collects particles with $d > 8~\mu\text{m}$. There is a smaller amount of sulfur, where the amount is expressed as a percent of the total mass of sulfur collected on all stages, on the Nuclepore filter and a larger amount on the seventh stage of the impactor behind the CFD.

Two problems made these experiments with the cascade impactors difficult. There was usually insufficient mass of a chemical element on a filter to permit a good determination of the total mass of each element on the filter. And some sulfur, as well as other elements, was usually found on stage seven of the impactor connected to the tube that supplied the aerosol from the aerosol generator. The CFD will be redesigned to provide more aerosol for the impactor. Furthermore, a filter can be placed in front of the CFD to prevent aerosol with $d > 8 \ \mu m$ from entering the chamber.

The analysis of experiment 26, which was conducted with ambient aerosols that were in the room normally used to generate the aerosols, indicated 296 nanograms of bromine on the Nuclepore of the impactor connected to the tube. The Nuclepore on the CFD had 264 nanograms. The bromine was not found on the filters that were analyzed to provide an analysis of the chemical elements in the filters. The bromine, therefore, was on aerosols with $d < 0.25~\mu m$; and these aerosols were neither deliquescent nor sufficiently hygroscopic to be CCN at the imposed supersaturation of approximately 1%.

The sample of AgI powder used to generate the AgI aerosols was analyzed with PIXE to search for contamination with trace elements. No contamination was found in the analysis of this AgI powder.

4. ACKNOWLEDGEMENT

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COMPARISON BETWEEN TWO AITKEN COUNTERS AND WITH CLOUD CONDENSATION NUCLEI COUNTERS AT THE 1980 INTERNATIONAL CCN WORKSHOP

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ABSTRACT

Activities at the 1980 International CCN Workshop (Reno, Nevada) included using a Pollak counter and a TSI Model 3020 Condensation Nucleus Counter to monitor the test aerosol concentration. The per-formance of these two counters has been intercompared and, when monodisperse CCN were used as the test aerosol, these two counters have also been compared to one of the CCN counters which gave consistently good performance and is representative of the CCN counters involved in the Workshop. The Pollak and the TSI 3020 counters agreed to within about 10% at concentrations below 1000 cm⁻³, whereas above that concentration, substantial systematic differences were observed. When the test aerosol was monodisperse and the CCN counters were operated at supersaturations that should nucleate all the aerosol, the Pollak counter read about 6% to 20% lower than the CCN counter in the concentration range from 250 cm^{-3} to 1800 cm^{-3} . This discrepancy is similar to that reported by Emmanuel and Squires (1969). The TSI 3020 read about 20% lower than the CCN counter at concentrations less than 1000 cm $^{-3}$, and a factor of about two higher at concentrations above 1000 cm $^{-3}$. Post-workshop calibration of the Pollak counter indicates that its performance was essentially in agreement with expectations for the instrument. Post-workshop evaluation of the TSI 3020 performance suggests the cause of the poor performance above 1000 $\rm cm^{-3}$ and indicates the care one must use to be sure the instrument is operated in such a way as to fulfill all the theoretical assumptions its proper performance is based upon.

1. INTRODUCTION

The main purpose of the October 6-17, 1980 International Cloud Condensation Nuclei Workshop, held at Reno, Nevada, was the intercomparison and performance evaluation of instruments which utilize applied supersaturations of the order of 1%. Such supersaturations should activate most cloud condensation nuclei (CCN), i.e., those nuclei capable of initiating cloud droplet growth in atmosphericallyrealistic conditions. Two inherently different instruments - Aitken counters - were also operated during the Workshop; these devices detect particles which nucleate at supersaturations of the order of 200% (which includes the special class of CCN). The Aitken counter data were used to monitor the stability of the output of the Workshop's test aerosol generation facility. The main purpose of this paper, however, is to discuss the intercomparison of the two Aitken counters - a Pollak counter and a TSI Model 3020 Condensation Nucleus Counter - and to compare them with the CCN counters when the test aerosol was a monodisperse aerosol under experimental conditions where it should be activated and detected by all the various devices.

The Nolan-Pollak Aitken counter has, in its present configuration, been in use about 20 years (Pollak and Metnieks, 1960). Detailed descriptions of this device and of its immediate predecessors are readily available (e.g., Metnieks and Pollak, 1959; Pollak and Metnieks, 1960; Pollak and Metnieks, 1957). [We shall follow the custom of referring to this device in an abbreviated manner as the "Pollak" counter, although the name of P.J. Nolan was also associated with the critical, early years of development of the instrument; see for example Nolan and Pollak (1946) - a work establishing the technical relationship and comparison of the Pollak counter to the classical Aitken counter.]

Subsequent workers examined various aspects of the operation and calibration of the Pollak counter. Kassner, et al. (1968) responded to ear-lier work criticizing the adiabaticity of the device, but concluded the reported discrepancies were a function of inappropriate diagnostic techniques rather than the Pollak counter itself. Emmanuel and Squires (1969) used a laboratory version Aitken counter in which activated droplets were recorded photographically to calibrate a Pollak counter over the approximate concentration range from zero to 1000 cm⁻³. The resulting calibration curve gave concentrations about 30% higher than the curve of Pollak and Metnieks (1960). Liu, et al. (1975), responding to reports that the Pollak counter did not agree well with aerosol concentration measurements by the electrical mobility analyzer, reported careful comparison studies which indicated agreement to better than 10% between these two vastly different measurement techniques for concentrations less than 104cm-3. In general, then, the Pollak counter has found widespread acceptance and use, particularly at concentrations below about 104cm-3 where vapor depletion by growing droplets does not significantly depress the applied supersaturation. The Pollak counter used in this study is No. 7 of the series manufactured by R. Gussman.

The newer Aitken counter at the Workshop was a TSI Model 3020 Condensation Nucleus Counter*

BGI Incorporated, Waltham, MA 02154

^{**}TSI Incorporated, St. Paul, MN 55164

(Serial No. 12). This counter samples a continuous, uninterrupted flow of aerosol through a saturation chamber (using butyl alcohol as the working fluid) and a chilling section, 25°C colder than the saturator, where the supersaturation is achieved. The commercial instrument was developed directly from the laboratory design described by Bricard, et al. (1976). The principle of operation and performance were discussed at the Ninth International Nucleation Conference (Argawal, et al., 1977). To the best of our knowledge, this report is the first published critical evaluation of the TSI Model 3020.

2. EXPERIMENTAL ARRANGEMENT

The layout of the Workshop laboratory is shown in Figure 1, indicating the arrangement of the various participants along the sample supply duct. The two Aitken counters received their sample from the same sample port, which was the last port on the supply line. Other papers in this volume discuss the details and results of the other instruments, and the nature of the aerosol generation and supply system. Twenty-nine separate experiments were carried out during the course of the Workshop (one of which had to be terminated due to an accidental duct blockage). A chronological listing and description of the type of aerosol used in each experiment is indicated in Table 1.

Excluding Experiment "0", for which no Aitken data were taken, there were ten experiments involving artificial monodisperse aerosols, ten involving artificial polydisperse aerosols, and eight involving natural (ambient) polydisperse aerosols. This comparison is mainly concerned with the monodisperse aerosol experiments, in which the aerosols were obtained by passing the polydisperse output from an atomizer through an "electrostatic classifier" which selects a monodisperse fraction of aerosol based upon electrical mobility. Aerosols thus prepared are not strictly monodisperse; however,

examination of the results of these experiments in the form of cumulative CCN counts plotted as a function of individual CCN counter supersaturation setting usually shows a "plateau" or region where counts do not increase above a certain supersaturation, indicating that the size distribution is in fact very narrow. The results of these kinds of measurements are discussed in more depth in the papers describing the performance and intercomparison of the continuous-flow and static diffusion chambers. For our purposes, it is sufficient to note that when such a "plateau" exists, the nuclei concentration measured with an Aitken counter should be equal to the concentration measured on the "plateau" by a CCN counter, i.e., the existence of the "plateau" demonstrates that all the nuclei in the sample have been activated at the supersaturation setting of the CCN counter at which the "plateau" occurs and all nuclei will, of course, have been activated in the Aitken counters which operate at much higher supersaturations.

The operation of the Aitken counters was generally according to standard practice. The Pollak counter was operated according to the recommended procedure of Metnieks and Pollak (1959, op. cit.) with two exceptions: the light beam was slightly convergent (as recommended by Pollak and Metnieks (1960, op. cit.)), and 152 mm Hg over-pressure was used instead of the sea level value of 160 mm Hg in order to compensate for the effect of laboratory altitude above sea level upon the amount of liquid water released during the expansion (cf. Emmanuel and Squires, 1969, op. cit.).

The TSI 3020 CNC was operated according to manufacturer's instructions, with extra care given to maintain the sample flow rate through the instrument at the prescribed value of 5 cm³/sec. Because the instrument uses a mass flow meter to measure and control flow, it is necessary to adjust the instrument flow control in response to altitude



No.	Date	A=A	Aerosol mbient; M=Monodisperse; P=Polydisperse
0	Tues 7 Oct	AM	$M - (NH_4)_2 SO_4$
ı	Tues	PM	P - NaCl - oscillating concentration
2	Tues	PM	P - NaCl - higher concentration
3	Wed 8 Oct	AM	A - quite fluctuating
4	Wed	AM	M - NaCl - low concentration
5	Wed	PM	M - NaCl - medium concentration
6	Wed	PM	A
7	Thurs 9 Oct	AM	A - aborted - duct blockage
8	Thurs	AM	M - NaCl - slight drift down
9	Thurs	AM	M - NaCl - higher concentration
10	Thurs	PM	Bimodal - NaCl - "flat k"
11	Thurs	PM	A
12	Fri 10 Oct	AM	Α
13	Fri	AM	$P - (NH_4)_2 SO_4$
14	Fri	PM	$P - (NH_4)_2 SO_4$
15	Fri	PM	$M - (NH_4)_2 SO_4$
16	Fri	PM	Α
17	Mon 13 Oct	AM	A
18	Mon	AM	M - (NH ₄) ₂ SO ₄
19	Mon	AM	M - (NH ₄) ₂ 50 ₄
20	Mon	PM	M - (NH ₄) ₂ 50 ₄
21	Mon	PM	M - $(NH_4)_2SO_4$ - time variations
22	Tues 14 Oct	АМ	$P - (NH_4)_2SO_4 - medium concentration$
23	Tues	PM	$P - (NH_4)_2 SO_4 - 1ow concentration$
24	Tues	PM	$P - (NH_4)_2 SO_4 - high concentration$
25	Wed 5 Oct	АМ	Filtered air - noise check
26	Wed	АМ	Α
27	Wed	РМ	P - AgI, "insoluble"
28	Wed	PM	M - AgI, "insoluble"
29	Wed	PM	P - paraffin, hydrophobic

TABLE 1. LIST OF EXPERIMENTS

effects on the density of air if one wants to maintain the volumetric flow rate at 5 cm³/sec. When checked with a volumetric flowmeter (a bubble meter), the volumetric flow rate through the instrument was 4.92 cm³/sec. The TSI 3020 CNC continuous real-time output, displayed on a chart recorder, provided a valuable monitor of the Workshop aerosol generation and delivery system stability.

3. EXPERIMENTAL RESULTS

A. Intercomparison of the Pollak and TSI CNC

The results of simultaneous measurements of the concentration of both polydisperse and monodisperse aerosols by the two instruments are shown in The results obtained with the TSI CNC Figure 2. are dependent upon the operating mode of the instrument, i.e., whether it was in the single-count mode or the photometric mode. The agreement between the two instruments is quite good when the TSI was operating in the single-count mode (lower curve in Figure 2), with the Pollak reading consistently about 10% lower. When the TSI operated in the photometric mode (upper curve) a nonlinear response occurred in the concentration range from 1000 cm⁻³ to 2000 cm⁻³, and above that concentration the TSI gave values twice those of the Pollak. The discontinuity in the results of the TSI between the two operating modes and the nonlinear portion of the response curve will be discussed below in the section on TSI Performance Evaluation. We have post-Workshop evidence to suggest that the Pollak performed according to expectations and we conclude that the upper curve in Figure 2 indicates that the TSI performance suffered from a systematic calibration error in the photometric mode. This error did not seriously detract from the value of the instrument as a continuous monitor of the relative aerosol concentration during the course of the various experiments performed during the Workshop.



Figure 2. Comparison of the Pollak and TSI 3020 CN Counters.

We can also compare the Aitken counters with a representative CCN counter when all three instruments were sampling monodisperse aerosols. Monodisperse sodium chloride of nominal diameter 0.036 μm (Experiments 4 and 5) and 0.18 μm (Experiments 8 and 9) and monodisperse ammonium sulfate of nominal diameters 0.092, 0.04, 0.14 and 0.08 μm (Experiments 15, 18, 19, and 20, respectively) were the test aerosols. Monodisperse silver iodide of nominal diameter 0.12 μm was the test aerosol in Experiment 28. (The tenth monodisperse aerosol experiment was Experiment 21, a repeat of No. 20 for the purpose of evaluating the measurement reproducibility of the various instruments; it is not included here because the CCN counter selected for comparison to the Aitken counters was operated as an isothermal haze chamber during that experiment.)

B. Intercomparison of the Pollak and TSI with a CCN Counter

It has not been a trivial task to select a "representative" CCN counter for use in this comparison. Many devices, both static diffusion and continuous-flow diffusion types, appeared to operate well during the course of the Workshop. We have selected the continuous-flow diffusion (CFD) chamber from the University of Missouri at Rolla for the comparison, as it consistently provided concentration values that were higher in comparison to the average of all CCN counters, but not so high as to indicate a malfunction. Results from this chamber on monodisperse aerosols usually exhibited a well-defined "plateau" in the plot of cumulative CCN concentration as a function of supersaturation value. The higher-than-overall-average count values for this chamber may suggest it had fewer sampling and detection losses. In another paper in this volume, comparing this chamber (hereinafter referred to as the UMR CFD) with the other CFD chambers in the Workshop, it was identified as showing good comparative performance (Hudson and Alofs, 1981).

Figure 3 shows a comparison of the Pollak counter and the UMR CFD using count values taken from the "plateau" regions. Error bars on the data simply indicate the magnitude of the statistical counting error. Only the lowest value shows per-fect agreement (Exp. 24). The UMR CFD gives higher values by 20% at the lower concentration experi-ments to 6% at the higher concentration experiments. It is interesting to recall that Emmanuel and Squires (1969, op. cit.) reported that the Pollak using the 1960 calibration values, under-counted in comparison to their photographic device, by about 30% over the concentration range from about 200 $\rm cm^{-3}$ to 1000 $\rm cm^{-3}.$ In a post-Workshop comparison of the DRI Pollak with another Pollak counter maintained as a standard by Dr. Austin Hogan, the DRI Pollak read high by some 10% to 20% (to be discussed below). Taking this discrepancy into consideration, the UMR CFD would be in agreement with the photographic device results reported by Emmanuel and Squires. These data, then, support the suggestion that a good CFD chamber will typically indicate on the order of 10% to 30% higher values than a Pollak counter when sampling monodisperse aerosol under conditions where the CFD is operating in the "plateau" for aerosols of varied chemical composition and of moderate concentration. (The activity of silver iodide as CCN is still a matter of some debate; here we present the one available data point (Experiment 28) and note that it is unremarkable in position on the plot.))


COUNT REGISTERED BY UMR CFD CHAMBER (CCN cm-3)

Figure 3. Comparison of the Pollak Counter and the UMR CFD CCN Counter.

A comparison of the results of the TSI CNC and the UMR CFD is shown in Figure 4 where, again, the values for the UMR CFD are the cumulative concentration values determined from the "plateau". The UMR CFD results correlate very well with the TSI (with the possible exception of silver iodide), being consistently 20% higher over the range up to 1000 cm⁻³, where the TSI is operating in the single-count mode. As one would anticipate from the data presented in Figure 2, the TSI reads higher by a factor of about 2 when operating in the photometric mode.

4. POST-WORKSHOP CALIBRATION OF THE POLLAK COUNTER

The DRI Pollak counter (No. 7) was built and calibrated according to the published procedures of its inventors, but had not been compared against another Pollak counter in recent years before the Workshop. At the kind invitation of Drs. Jim Jiusto and Austin Hogan of the Atmospheric Sciences Research Center, State University of New York at Albany, a post-Workshop calibration was performed in February, 1981 in their laboratory. Pollak counter No. 7 was first compared to No. 12 "as is", and then was dismantled and the internal alignment of the light beam was checked. The "as is" calibration gave the result that the DRI Pollak No. 7 consistently read 10% to 20% higher than No. 12 over the concentration range for 400 $\rm cm^{-3}$ to 70,000 12 $\rm cm^{-3}.$ Upon dismantling and inspecting No. 7, it was found that although the light beam was of the proper diameter at the photocell end of the fog tube, it was off-axis by about 0.4 cm. Correction of this misalignment did not, however, change the systematic discrepancy that existed between the two instruments.

Several possible sources of such a systematic difference were explored, for example, by interchanging the photocells and the microammeters which provide the output signals and readings of the Pollak counters, and by interchanging operators. None of these changes removed the discrepancy. A potentially important parameter suggested in the paper of Pollak and Metnieks (1960, op.cit.) led us to investigate the rapidity with which the cleanair overpressure was established in the two instruments. Due to minor differences in the plumbing of these instruments, No. 12 achieved overpressure at a slower rate than No. 7. Standardizing both devices to the overpressure rate of No. 12 led to excellent agreement, with differences now less than would be anticipated due to statistical counting errors. It appears that the time to achieve overpressure in No. 7 (the Pollak used in the Workshop) was less than that recommended by Pollak and Metnieks (1960, op.cit.), whereas the time required for No. 12 was longer than the recommended time. We can only estimate that Pollak No. 7 was reading higher during the Workshop than an "ideal" Pollak would have, by an amount less than 10 to 20% of the number concentration.

5. TSI PERFORMANCE EVALUATION

A brief description of the operational modes and the associated underlying assumptions of the TSI CNC is necessary in order to evaluate its performance during the Workshop. The sample stream is first saturated with alcohol and then passed through a chilled condenser tube where supersaturation occurs; the nuclei are activated and the resulting droplets grow to 5 μm to 10 μm diameter. The sample of droplets then passes into a viewing/ detection volume. There are two modes of operation



Figure 4. Companison of the TSI 3020 CN Counter and the UMR CFD CCN Counter.

for detecting the droplets. At concentrations below 1000 cm⁻³, electrical pulses generated by light scattered from individual droplets are counted and this count rate is converted to particle concentration and displayed on the front panel and is also available as an analog output. At particle concentrations above 10^3 cm⁻³, the photodetector circuit measures the light scattered from all the droplets present in the viewing volume at any given time. The photodetector output is calibrated as a function of concentration, and the results of the calibration are incorporated into the electronics of the instrument so that it displays the correct concentration.

In the single count mode the instrument operates in an absolute counting mode and the accuracy of the concentration is determined only by the accuracy of the flow rate of the sample and correction for coincidence losses occurring due to two or more droplets passing through the viewing volume simultaneously. (Statistical errors of counting for low count values can be reduced to desired levels by increasing counting times.) Because the flow monitor and control system is based upon a mass flow meter, the volumetric flow rate through the instrument will be sensitive to altitude and must be corrected if the instrument is operated at altitudes other than that at which it was set up. One can either reset the flow to give 5 cc^{3}/sec^{3} or calculate the actual volumetric flow and apply an appropriate correction factor to the indicated concentration (i.e., multiply by [5 cm³sec⁻¹/(actu-al volumetric flow in cm³sec⁻¹)]). The instrument we used in the Workshop was adjusted to correct the flow to $4.92\ {\rm cm}^3 {\rm sec}^{-1}$.

In the photometric mode additional conditions must be met to assure that the instrument performs in calibration. Because the final droplet size determines the amount of light scattered per original activated nuclei, any parameter that affects droplet size must be controlled to maintain the conditions for which the photometric mode was calibrated. Changes in altitude in this case will affect not only the flow rate (and therefore the growth time in the condenser) but may also affect vapor diffusivity and thermal diffusivity. These effects have not yet been well studied for this instrument. A second precaution one must be aware of is the effect of dirty optics in the photodetec-In the single count mode, this will have no tor. affect because one is just counting pulses, not scattered intensity. However, in the photometric However, in the photometric mode all scattered light contributes to the count rate, whether it is scattered from dirty optics or from droplets in the viewing volume. The error will be most pronounced at low values in the photometric mode and will become increasingly less important as the concentration increases. We believe that this is the cause of the non-linear portion of the TSI response when compared to the Pollak in the log-log plot in Figure 2. Dirty optics will pro-duce a constant, small D.C. offset that will be integrated into the signal in the photometric mode signal. As the total signal increases, it becomes a negligible contribution to the signal and the response then appears linearly related to the Pol-lak response. (Unfortunately there has not been opportunity since the Workshop to inspect the optics as they were during the Workshop in order to confirm this explanation of that portion of the curve.) The linear portion of the photometric mode response is parallel to the single count mode re-

sponse when extrapolated to the photometric mode concentration range. This suggests that the discrepancy is due to an error in the calibration factor for the instrument under the operating conditions encountered at the Workshop. One instrumental parameter that is useful in helping determine if the calibration may be in error is the photodetector pulse height. For the instrument we used, the value when the CNC was calibrated was 0.30 Volts peak-to-A post-Workshop reading indicated it was peak. 0.55 to 0.6 Volts, indicating that the instrument would not be expected to be in proper calibration. It is tempting to note that the ratio of the actual pulse height to the correct pulse height is the same as the difference between the photometric mode results and the extrapolated single count mode values, i.e, a factor of about 2. This cannot. however, be applied in such a straightforward manner because calibration will also depend on other parameters. It is sufficient here to note that there is reasonable evidence to indicate that the photometric mode results are high because the calibration is not applicable under the conditions in which we used the instrument, but that the linear response with concentration indicates that the TSI CNC will still give reliable relative concentration values under these conditions. These results again demonstrate the importance of understanding the underlying principles of the detection/measurement process of the instrument.

6. CONCLUSIONS

The performance of a Pollak counter (No. 7) of nearly standard specifications was compared with that of a set of CCN counters as represented by the UMR CFD under conditions where both were sampling the same monodisperse aerosol. Nine experiments are represented in the comparison data which indicates that the Pollak underestimates the concentration by 10% to 30% over the concentration range from about 260 cm⁻³ to 1800 cm⁻³ in comparison to the UMR CFD. This discrepancy is of the same sign, but of slightly smaller magnitude than that reported earlier by Emmanuel and Squires (1969, op.cit.) for a comparison of the Pollak counter with an "absolute" Aitken counter which recorded the drop-let concentration photographically for subsequent counting.

A TSI 3020 CNC was also compared with the Pollak and the UMR CFD counters. It gave concentration values 20% lower than the UMR CFD in the concentration range from 260 cm⁻³ to 1000 cm⁻³ (i.e., when the TSI was in the single-count mode). It gave values 10% higher than the Pollak over the same concentration range. Above 1000 cm⁻¹, when the TSI operated in the photometric mode, it gave values about 2 times higher than the Pollak or the UMR CFD.

This work, then, has demonstrated very good agreement between three instruments of vastly different design and detection/measurement principles the Pollak counter, a continuous-flow CCN counter, and a continuous-flow CNC. The present results essentially confirm the earlier work of Emmanuel and Squires (1969, op.cit.) which indicated a small positive correction to the calibration of the Pollak at aerosol concentrations below 1000 cm⁻³. The apparent discrepancy between the Pollak and the UMR CFD is somewhat accentuated by the fact that the UMR CFD consistently read higher than the average of the other CCN counting devices in the Workshop. The comparison study by Hudson and Alofs (1981, op.cit.) indicates that the UMR CFD was operating properly and providing accurate data. The excellent agreement with the TSI, which is a self-calibrating or absolute detector when in the single-count mode, is particularly gratifying and provides additional confidence in the validity of the large store of Aitken concentration values that have been obtained through the use of the Pollak counter over the past 20 some years.

The TSI 3020 CNC exhibited very good performance in the single-count mode and, despite loss of absolute accuracy in the photometric mode (as discussed in the text), it still provided valuable real-time relative concentration measurements which were of great utility in assessing the stability of the test aerosol generation system.

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Certain commercial equipment, instruments or materials are identified in this paper in order to adequately specify the experimental procedure. Such identification does not imply recommendation or endorsement by the National Bureau of Standards nor does it imply that the materials or equipment are necessarily the best available for the purpose.

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Two instruments were used to size dry aerosols for the CCN experiments. The Naval Research Laboratory (NRL) instrument uses an electrical mobility analyzer (Hoppel, 1978) to separate the aerosol size fractions and a diffusion chamber (Hoppel, et al., 1979) to grow and count the particles. The University of Wyoming (WYO) Aerosol Monitoring System (Rogers and Vali, 1981) consists of three devices which are monitored and controlled by a minicomputer. The devices are: a Thermo-systems Electrical Aerosol Size Analyzer (Model 3030) which covers the size range 0.005 to 0.18 μ m radius, an Environment One (E1) Condensation Nucleus Monitor (Model Rich 100) for Aitken particles, and an optical particle counter (Climet Model 6064A optics and in-house electronics) to cover the size range 0.15 to 8 µm radius.



Figure 1. Aerosol size distributions of Experiment 8: average of two Naval Research Laboratory (NRL) measurements and of three University of Wyoming (WYO) measurements. Also shown are Aitken counter values (El = Environment One, Model Rich LOO, TSJ = Thermo-systems Model 3020, P = Pollak Counter and G =Gardner Counter1.



Figure 2. As for Figure 1, but aerosol from Experiment 27: average of four NRL measurements and of two WYO measurements.

Measurements from these two systems were usually found to be in agreement during the CCN Work-Two examples are presented here to compare shop. aerosol size distribution measurements of the two instruments: Experiment 8 (monodisperse NaCl) and Experiment 27 (polydisperse AgI). Differential (dN/dR) and cumulative plots are shown for both instruments for Experiment 8 in Figure 1 and for Experiment 27 in Figure 2; also shown are Aitken particle measurements for comparison. The monodisperse peak of Experiment 8 was well defined by both instruments, although information below 0.05 µm was not discernible for the WYO system; this is likely due to the low concentration, lower than the mobil-ity analyzer is expected to detect reliably. Both instruments indicated peak concentrations in the range 0.08 to 0.1 μ m radius and total particle concentrations of 400 to 500 cm⁻³. The total concentrations from the mobility analyzers are approximately twice the values obtained from the Aitken counters.

Measurements of the shape of the aerosol size distribution of Experiment 27 were in substantial agreement below 0.1 μ m, even to detecting a slight peak at 0.2 μ m radius. At larger sizes, the NRL instrument measured a higher concentration than the WYO optical counter, and at smaller sizes, the WYO mobility analyzer measured higher concentrations than NRL. Both instruments indicated that the greatest concentrations occurred at the small sizes, but the Aitken particle counters offer little help in resolving the discrepancy in integrated total concentration, since their values range over a factor of 3. Perhaps the chemical composition and shape of the aerosol was partly responsible, as thermally generated AgI is expected to be wettable but unsoluble and may have a linear rather than globular shape.

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APPENDIX A

TABULATED DIFFERENTIAL SIZE SPECTRA RESULTS FROM NRL MOBILITY ANALYZER

FOR FURTHER DETAILS, REFER TO WILLIAM A. HOPPEL PAPER, "MEASUREMENT OF THE AEROSOL SIZE DISTRIBUTION WITH NRL'S MOBILITY ANALYZER" CONTAINED IN SECTION VI.

NOTE THAT ABSENCE OF AN ENTRY IN DN/DR COLUMN INDICATES CONCENTRATION WAS VERY LOW OR ZERO.

Experiment #1 - Average of 3 Runs

<u>Channe1</u>	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	1.69x10 ⁵
2	1.83x10 ⁻⁵ cm	3.24x10 ⁵
3	1.15x10 ⁻⁵ cm	4.19x10 ⁶
4	7.53x10 ⁻⁶ cm	1.87x10 ⁷
5	5.03x10 ⁻⁶ cm	5.80x10 ⁷
6	3.46x10 ⁻⁶ cm	1.55x10 ⁸
7	2.37x10 ⁻⁶ cm	2.53x10 ⁸
8	1.52x10 ⁻⁶ cm	3.09x10 ⁸
9	1.15x10 ⁻⁶ cm	6.05x10 ⁸
10	8.10x10 ⁻⁷ cm	3.12x10 ⁸
11	5.75×10^{-7} cm	2.23×10^8

Experiment #2 - Average of 2 Runs

Cumulative Radius	DN/DR
3.09x10 ⁻⁵ cm	2.11x10 ⁴
1.83x10 ⁻⁵ cm	1.72x10 ⁶
1.15x10 ⁻⁵ cm	1.07x10 ⁷
7.53x10 ⁻⁶ cm	4.41x10 ⁷
5.03x10 ⁻⁶ cm	1.15x10 ⁸
3.46x10 ⁻⁶ cm	2.67x10 ⁸
2.37x10 ⁻⁶ cm	5.14x10 ⁸
1.52x10 ⁻⁶ cm	7.05x10 ⁸
1.15x10 ⁻⁶ cm	1.35x10 ⁹
8.10x10 ⁻⁷ cm	8.22x10 ⁸
5.75x10 ⁻⁷ cm	7.14x10 ⁸
	$\frac{\text{Cumulative Radius}}{3.09 \times 10^{-5} \text{ cm}}$ $1.83 \times 10^{-5} \text{ cm}$ $1.15 \times 10^{-5} \text{ cm}$ $7.53 \times 10^{-6} \text{ cm}$ $3.46 \times 10^{-6} \text{ cm}$ $2.37 \times 10^{-6} \text{ cm}$ $1.52 \times 10^{-6} \text{ cm}$ $1.52 \times 10^{-6} \text{ cm}$ $1.15 \times 10^{-6} \text{ cm}$ $8.10 \times 10^{-7} \text{ cm}$ $5.75 \times 10^{-7} \text{ cm}$

8.10x10⁻⁷cm

 5.75×10^{-7} cm

1

2

3

4

5

6

7

8

9

10

11

Experiment #3 - One Run Only Channe 1 Cumulative Radius DN/DR 6.40x10⁶ 3.09x10⁻⁵cm 3.33x10⁷ 1.83x10⁻⁵cm 1.15x10⁻⁵cm 9.17x10⁷ 7.53x10⁻⁶cm 1.40x10⁸ 5.03x10⁻⁶cm 4.44x10⁸ 3.46x10⁻⁶cm 7.06x10⁸ 2.37x10⁻⁶cm 8.23x10⁸ 2.81x10⁹ 1.52x10⁻⁶cm 8.97x10⁹ 1.15x10⁻⁶cm

Experiment #4 - Average of 4 Runs

Channe 1	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	
2	1.83x10 ⁻⁵ cm	
3	1.15×10 ⁻⁵ cm	
4	7.53x10 ⁻⁶ cm	
5	5.03x10 ⁻⁶ cm	
6	3.46×10 ⁻⁶ cm	2.70x10
7	2.37x10 ⁻⁶ cm	3.26x10
8	1.52x10 ⁻⁶ cm	3.57x10 ⁶
9	1.15×10 ⁻⁶ cm	3.68x10 ⁶
10	8.10x10 ⁻⁷ cm	3.54x10
11	5.75x10 ⁻⁷ cm	

Experiment #5 - Average of 3 Runs

Channel	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	
2	1.83x10 ⁻⁵ cm	
3	1.15×10 ⁻⁵ cm	
4	7.53x10 ⁻⁶ cm	
5	5.03x10 ⁻⁶ cm	
6	3.46x10 ⁻⁶ cm	3.54x10 ⁶
7	2.37x10 ⁻⁶ cm	1.51x10 ⁸
8	1.52x10 ⁻⁶ cm	1.55x10 ⁹
9	1.15x10 ⁻⁶ cm	1.60x10 ⁸
10	8.10×10 ⁻⁷ cm	6.88x10 ⁶
11	5.75x10 ⁻⁷ cm	

Experiment #6 - Average of 3 Runs

Channe 1	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	1.06x10 ⁵
2	1.83x10 ⁻⁵ cm	2.66x10 ⁶
3	1.15x10 ⁻⁵ cm	3.06x10 ⁷
4	7.53x10 ⁻⁶ cm	1.17x10 ⁸
5	5.03x10 ⁻⁶ cm	1.74x10 ⁸
6	3.46x10 ⁻⁶ cm	4.19x10 ⁸
7	2.37x10 ⁻⁶ cm	9.75x10 ⁸
8	1.52x10 ⁻⁶ cm	2.15x10 ⁹
9	1.15x10 ⁻⁶ cm	7.26x10 ⁹
10	8.10x10 ⁻⁷ cm	8.03x10 ⁹
11	5.75x10 ⁻⁷ cm	2.54x10 ⁹

3.41x10⁹

9.41x10⁸

Experiment #8 - Average of 2 Runs

Channe l	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	1.08x10 ⁴
2	1.83x10 ⁻⁵ cm	6.22x10 ⁵
3	1.15x10 ⁻⁵ cm	7.83x10 ⁶
4	7.53x10 ⁻⁶ cm	7.67x10
5	5.03x10 ⁻⁶ cm	6.13x10 ⁶
6	3.46x10 ⁻⁶ cm	1.77x10 ⁵
7	2.37x10 ⁻⁶ cm	
8	1.52x10 ⁻⁶ cm	8.27x10 ⁴
9	1.15x10 ⁻⁶ cm	2.54x10 ⁴
10	8.10x10 ⁻⁷ cm	1.76x10 ²
11	5.75x10 ⁻⁷ cm	

Experiment #9 - Average of 3 Runs

<u>Channel</u>	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	4.86x10 ⁴
2	1.83x10 ⁻⁵ cm	1.49x10 ⁵
3	1.15x10 ⁻⁵ cm	1.58x10/
4	7.53x10 ⁻⁶ cm	1.39x10 ⁸
5	5.03x10 ⁻⁶ cm	8.82x10 ⁶
6	3.46x10 ⁻⁶ cm	6.89x10 ^b
7	2.37x10 ⁻⁶ cm	-
8	1.52x10 ⁻⁶ cm	2.21x10 ⁵
9	1.15x10 ⁻⁶ cm	4.72×10^4
10	8.10x10 ⁻⁷ cm	3.19x10 ²
11	5.75x10 ⁻⁷ cm	

Experiment #10 - Average of 2 Runs

Channel	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	
2	1.83×10 ⁻⁵ cm	
3	1.15x10 ⁻⁵ cm	1.33x10 ⁶
4	7.53x10 ⁻⁶ cm	1.19x10 ⁷
5	5.03×10 ⁻⁶ cm	3.38x10 ⁷
6	3.46x10 ^{~6} cm	4.75x10 ⁷
7	2.37x10 ⁻⁶ cm	5.06x10 ⁷
8	1.52x10 ⁻⁶ cm	1.07x10 ⁹
9	1.15x10 ⁻⁶ cm	8.61x10 ⁸
10	8.10x10 ⁻⁷ cm	1.58x10 ⁷
11	5.75x10 ⁻⁷ cm	

Experiment #11 - Average of 3 Runs

Channel	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	1.90x10 ⁵
2	1.83×10 ⁻⁵ cm	1.77x10 ⁶
3	1.15x10 ⁻⁵ cm	2.81x10 ⁷
4	7.53x10 ⁻⁶ cm	8.86x10 ⁷
5	5.03x10 ⁻⁶ cm	1.03x10 ⁸
6	3.46x10 ⁻⁶ cm	2.89x10 ⁸
7	2.37x10 ⁻⁶ cm	2.35x10 ⁸
8	1.52×10 ⁻⁶ cm	1.59x10 ⁹
9	1.15x10 ⁻⁶ cm	5.69x10 ⁸
10	8.10x10 ⁻⁷ cm	1.36x10 ⁹
11	5.75x10 ⁻⁷ cm	3.95x10 ⁹

Experiment #13 - Average of 3 Runs

<u>Channel</u>	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	5.49x10 ⁴
2	1.83x10 ⁻⁵ cm	2.65x10 ⁵
3	1.15x10 ⁻⁵ cm	4.76x10 ⁶
4	7.53x10 ⁻⁶ cm	4.98x10 ⁷
5	5.03x10 ⁻⁶ cm	1.93x10 ⁸
6	3.46x10 ⁻⁶ cm	4.93x10 ⁸
7	2.37x10 ⁻⁶ cm	1.02x10 ⁹
8	1.52x10 ⁻⁶ cm	1.63x10 ⁹
9	1.15x10 ⁻⁶ cm	3.39x10 ⁹
10	8.10x10 ⁻⁷ cm	2.44x10 ⁹
11	5.75x10 ⁻⁷ cm	2.26x10 ⁹

Experiment #14 - Average of 3 Runs

<u>Channel</u>	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	4.22x10 ⁴
2	1.83x10 ⁻⁵ cm	6.92x10 ⁵
3	1.15x10 ⁻⁵ cm	1.07x10 [/]
4	7.53×10 ⁻⁶ cm	5.86x10 ⁷
5	5.03x10 ⁻⁶ cm	1.50x10 ⁸
6	3.46×10 ⁻⁶ cm	6.23x10 ⁸
7	2.37x10 ⁻⁶ cm	1.16x10 ⁹
8	1.52x10 ⁻⁶ cm	1.54×10^{9}
9	1.15x10 ⁻⁶ cm	3.04x10 ⁹
10	8.10x10 ⁻⁷ cm	1.98x10 ⁹
11	5.75x10 ⁻⁷ cm	2.13x10 ⁹

Experiment #15 - Average of 2 Runs

<u>Channel</u>	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	
2	1.83x10 ⁻⁵ cm	3.09×10^{4}
3	1.15x10 ⁻⁵ cm	9.70x10 ⁵
4	7.53x10 ⁻⁶ cm	3.26x10 ⁷
5	5.03x10 ⁻⁶ cm	1.84x10 ⁸
6	3.46x10 ⁻⁶ cm	5.43x10 ⁸
7	2.37x10 ⁻⁶ cm	
8	1.52x10 ⁻⁶ cm	
9	1.15x10 ⁻⁶ cm	1.93x10 ⁶
10	8.10x10 ⁻⁷ cm	
11	5.75x10 ⁻⁷ cm	5.20x10

Experiment #17 - Average of 2 Runs

<u>Channel</u>	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	1.48x10 ⁵
2	1.83x10 ⁻⁵ cm	7.51x10 ⁵
3	1.15x10 ⁻⁵ cm	6.71x10 ⁶
4	7.53x10 ⁻⁶ cm	3.53x10 ⁷
5	5.03x10 ⁻⁶ cm	1.77x10 ⁸
6	3.46x10 ⁻⁶ cm	4.49x10 ⁸
7	2.37x10 ⁻⁶ cm	9.97x10 ⁸
8	1.52x10 ⁻⁶ cm	1.43x10 ⁹
9	1.15x10 ⁻⁶ cm	4.67 x10 ⁹
10	8.10x10 ⁻⁷ cm	4.12x10 ⁹
11	5.75x10 ⁻⁷ cm	4.23x10 ⁹

Experiment #18 - Average of 3 Runs

<u>Channel</u>	Cumulative Radius	DN/DR	Channe1	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm		1	3.09x10 ⁻⁵ cm	
2	1.83x10 ⁻⁵ cm		2	1.83x10 ⁻⁵ cm	
3	1.15x10 ⁻⁵ cm		3	1.15x10 ⁻⁵ cm	1.92x10 ⁵
4	7.53x10 ⁻⁶ cm		4	7.53x10 ⁻⁶ cm	1.05x10 ⁷
5	5.03x10 ⁻⁶ cm	1.77x10 ⁵	5	5.03x10 ⁻⁶ cm	6.84x10 ⁷
6	3.46×10 ⁻⁶ cm	1.93×10 ⁷	6	3.46×10 ⁻⁶ cm	4.29x10 ⁸
7	2.37×10 ⁻⁶ cm	2.71x10 ⁸	7	2.37x10 ⁻⁶ cm	1.59x10 ⁷
8	1.52x10 ⁻⁶ cm	1.23x10 ⁹	8	1.52×10 ⁻⁶ cm	
9	1.15x10 ⁻⁶ cm	1.17x10 ⁹	9	1.15×10 ⁻⁶ cm	9.44x10 ⁵
10	8.10x10 ⁻⁷ cm	3.10x10 ⁸	10	8.10×10 ⁻⁷ cm	
11	5.75x10 ⁻⁷ cm	1.21x10 ⁸	11	5.75x10 ⁻⁷ cm	

Experiment #19 - Average of 4 Runs

Channe1	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	_
2	1.83x10 ⁻⁵ cm	4.95x10 ⁵
3	1.15x10 ⁻⁵ cm	1.29x10 ⁷
4	7.53x10 ⁻⁶ cm	1.03x10 ⁸
5	5.03x10 ⁻⁶ cm	3.04×10^8
6	3.46x10 ⁻⁶ cm	8.90x10 ⁷
7	2.37x10 ⁻⁶ cm	1.49x10 ⁷
8	1.52x10 ⁻⁶ cm	
9	1.15x10 ⁻⁶ cm	4.04x10 ⁴
10	8.10x10 ⁻⁷ cm	6.39x10 ²
11	5.75x10 ⁻⁷ cm	

Experiment #20 - Average of 4 Runs

Channe 1	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	
2	1.83x10 ⁻⁵ cm	
3	1.15x10 ⁻⁵ cm	9.62×10 ⁴
4	7.53x10 ⁻⁶ cm	1.03x10 ⁷
5	5.03x10 ⁻⁶ cm	7.06x10 ⁷
6	3.46x10 ⁻⁶ cm	4.96 ×10 ⁸
7	2.37x10 ⁻⁶ cm	3.14×10 ⁷
8	1.52x10 ⁻⁶ cm	3.00x10 ⁵
9	1.15x10 ⁻⁶ cm	
10	8.10x10 ⁻⁷ cm	
11	5.75x10 ⁻⁷ cm	

Experiment #21 - Average of 4 Runs

A	-	4
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Experiment #22 - Average of 3 Runs (2,3 and 4) [taken between 1100 and 1220; early run (#1) was taken when the count was higher and remains as distributed at Workshop].

<u>Channel</u>	Cumulative Radius	DN/DR
1	3.09x10 ⁻⁵ cm	2.11x10 ⁴
2	1.83x10 ⁻⁵ cm	4.59×10^{4}
3	1.15x10 ⁻⁵ cm	3.37x10 ⁵
4	7.53x10 ⁻⁶ cm	6.21x10 ⁶
5	5.03x10 ⁻⁶ cm	6.29x10 ⁷
6	3.46×10^{-6} cm	1.67x10 ⁸
7	2.37×10^{-6} cm	3.43x10 ⁸
8	1.52×10^{-6} cm	8.69x10 ⁸
9	1.15×10^{-6} cm	4.13x10 ⁸
10	$8.10 \times 10^{-7} \text{ cm}$	2.15x10 ⁸
11	5.75×10^{-7} cm	1.52x10 ⁸

Experiment #23 - Average of 4 Runs

Channe 1	Cumulative Radius	DN/DR
1	3.09×10 ⁻⁵ cm	2.64x10 ⁴
2	1.83x10 ⁻⁵ cm	
3	1.15x10 ⁻⁵ cm	8.20x10 ⁴
4	7.53x10 ⁻⁶ cm	1.25x10 ⁶
5	5.03x10 ⁻⁶ cm	1.64×10^{7}
6	3.46x10 ⁻⁶ cm	6.55x10/
7	2.37×10 ⁻⁶ cm	1.19x10 ⁸
8	1.52x10 ⁻⁶ cm	1.19x10 ⁸
9	1.15x10 ⁻⁶ cm	1.50×10^8
10	8.10x10 ⁻⁷ cm	5.93×10^{7}
11	5.75x10 ⁻⁷ cm	5.32x10 ⁷

Experiment #24 - Average of 4 Runs

Cumulative Radius	DN/DR
3.09x10 ⁻⁵ cm	
1.83x10 ⁻⁵ cm	3.09x10 ⁴
1.15x10 ⁻⁵ cm	7.78x10 ⁵
7.53x10 ⁻⁶ cm	1.59x10/
5.03x10 ⁻⁶ cm	2.78x10 ⁸
3.46x10 ⁻⁶ cm	1.72x10 ⁹
2.37x10 ⁻⁶ cm	4.04x10 ⁹
1.52x10 ⁻⁶ cm	3.60x10 ⁹
1.15x10 ⁻⁶ cm	5.28x10 ⁹
8.10x10 ⁻⁷ cm	2.52x10 ⁸
5.75x10 ⁻⁷ cm	1.15×10 ⁹
	$\frac{\text{Cumulative Radius}}{3.09 \times 10^{-5} \text{ cm}}$ $1.83 \times 10^{-5} \text{ cm}$ $1.15 \times 10^{-5} \text{ cm}$ $7.53 \times 10^{-6} \text{ cm}$ $3.46 \times 10^{-6} \text{ cm}$ $2.37 \times 10^{-6} \text{ cm}$ $1.52 \times 10^{-6} \text{ cm}$ $1.52 \times 10^{-6} \text{ cm}$ $1.15 \times 10^{-6} \text{ cm}$ $8.10 \times 10^{-7} \text{ cm}$ $5.75 \times 10^{-7} \text{ cm}$

Experiment #27 - Average of 4 Runs

channel cullurative Radius	N/UK
1 3.09x10 ⁻⁵ cm 2.1	1x10 ⁵
2 1.83x10 ⁻⁵ cm 1.1	1x10 ⁶
3 1.15x10 ⁻⁵ cm 1.3	2x10 ⁶
4 7.53x10 ⁻⁶ cm 8.7	6x10 ⁶
$5 5.03 \times 10^{-6} \text{cm}$ 1.2	4x10 ⁸
$6 \qquad 3.46 \times 10^{-6} \text{cm} \qquad 6.6$	0x10 ⁸
7 $2.37 \times 10^{-6} \text{cm}$ 1.2	0x10 ⁹
8 1.52×10 ⁻⁶ cm 1.4	1x10 ⁹
9 1.15x10 ⁻⁶ cm 3.1	6x10 ⁸
10 8 10×10^{-7} cm 2.7	'6x10 ⁹
$11 5.75 \times 10^{-7} \text{cm}$ 2.6	6x10 ⁹

Experiment #28 - Average of 2 Runs

Channe 1	Cumulative Radius	DN/DR
I	3.09x10 ⁻⁵ cm	
2	1.83x10 ⁻⁵ cm	
3	1.15×10 ⁻⁵ cm	_
4	7.53×10 ⁻⁶ cm	1.78x10 ⁷
5	5.03x10 ⁻⁶ cm	1.65×10^{8}
6	3.46×10 ⁻⁶ cm	2.23x10 ⁷
7	2.37x10 ⁻⁶ cm	
8	1.52×10^{-6} cm	3.72x10 ⁶
9	1.15×10^{-6} cm	8.17x10 ⁶
10	8.10×10^{-7} cm	
11	5.75x10 ⁻⁷ cm	

APPENDIX B

In this section, the complete 1980 International CCN Workshop data file is presented, in the form of both tables and graphs of CCN counted as a function of supersaturation setting, for all experiments. The reader is urged to consult the companion review papers (covering the static diffusion chambers, continuous-flow chambers, and isothermal haze chambers by generic type) for important commentaries and points necessary to interpret these data. For example, as discussed in "Review of Isothermal Haze Chamber Performance", the interpretation of haze chamber results is dependent upon understanding the method used to calibrate the chamber optical counters for the sizing of water droplets. Useful summaries of the data will also be found in the review papers; discussions of the methods of obtaining data from the various instruments will generally be found in the relevent instrument description papers in Section V.

The data file is presented exactly as the data were received during the Workshop, with the exception of three post-Workshop revisions, all of which were cases of straightforward errors in the reporting of data. These revisions occurred in the cases of data from instruments #13 (CSIRO, G. Ayers), #14 (DRI, J. Hudson), and #25 (CSU, R. Borys); below we summarize, in the words of each instrument operator, the change in data together with reasons.

Instrument #13 - Ayers

I learned upon my return to CSIRO that the newly-generated calibration curves relate "film count" to "peak reading", and that to get to an estimate of the "true" CCN concentration, the "film count" must be increased by 50%. Thus all the concentration values I fed into your data system should be multiplied by a factor of 1.5. This factor is explained at more length in the notes. I regret that I was unaware of it and hope that it does not cause too much inconvenience.

Instrument #14 - Hudson

In the process of writing the review paper on the IHC's, I looked back into my original Workshop notebook. I discovered that, from experiment #12 to the end of the Workshop, the threshold supersaturations for channel 3 were mislabeled in the data file. According to the original notebook, this should have been 0.11%. However, it was written into the Workshop data sheets as 0.085% which it had actually been set at for a couple of earlier experiments.

Instrument #25 - Borys

CCN concentrations were revised for the CSU static diffusion chambers for experiments 17-20. The changes were made because of errors in reading the original photographic film (i.e., counting the droplet images visually) by being out of sequence one frame for a 36 exposure roll of film. The results were spectra shifted to the left on plots done at DRI, thus artificially increasing the concentrations over all supersaturations. This was easily done since there were plateaus in the curves and the error wasn't noticed until the data were reviewed in detail after returning to CSU. The net effect is to simply correct a "human error". There was no physical instrumental correction made.

CCN WORKSHOP NUCLEI MEASUREMENTS DATA SHEET

Experiment No:	Date:	/ / (yr) (mo) (da				
Purpose:						
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Run No:'	Instrument No:	Duct No:	Observer:			

NUCLEI MEASUREMENTS

Sample	Time	Plate T	emp °C	Volume		Nuclei	_	Fst	2	
Start	End	Cold	Hot	Sampled (£)	S _c (%)	Count	N	ĸ	Remarks ⁻	
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•										5
										6
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 1 If more than one spectrum is run, please record the data on another sheet with a new run number.

²Only the first 25 characters of the remarks column will be recorded and displayed via computer processing.

CCN WORKSHOP DATA CODE

Inst. No.	Group	Graphics Code
2	Wyoming	0
5	British Met Office	В
7	York Univ.	Y
9	Hebrew Univ.	н
10	SUNY	S
11	NRL	Nh (Haze)
13	CSIRO	I
14	DR I	Dh (Haze)
15	DR I	Dn (Spec. NASA)
16	DRI	Di (Spec. Instant)
17	NRL	Ns (SDC)
18	DRI	Dc (CFD)
20	Washington	W
21	Missouri	м
22	Florida	
23	Florida	
24	France	F
25	CSU	Cs (SDC)
26	CSU	Ch (Haze)
27	Alaska	А
28	NBS	Gt
29	NBS	Gp



EXPERIMENT # 01

PURPOSE

INSTRUMENT COMPARISON WITH POLYDISPERSE ARTIFICIAL CCN

DESCRIPTION OF EXPERIMENT

DATE START END OT DCT.1980 1458 1615 NUCLEI TYPE

NACL. POLYDISPERSE

GENERATION METHOD

DRI ATOMIZER. 18 PSI

SIZE DISTRIBUTION SHAPING

NONE

REMARKS

OUTPUT FLUCTUATES +- 8% Sample wet bulb 55 (f), Dry Bulb 81 (f), Dil. Airflow 670 L/Min.

WEATHER SYNOPSIS

٩N

MEAN WEATHER CONDITIONS DURING EXPERIMENT

	 AVG WD (DEG)	NA
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	AVG TEMP (C)	AN
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80/10/07

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EXPERIMENT # 02

80/10/07

PURPOSE

INSTRUMENT COMPARISON WITH POLYDISPERSE ARTIFICIAL CCN

DESCRIPTION OF EXPERIMENT

TIME START END

07 DCT+1980 1640 1740 DATE

NUCLEI TYPE

NACL POLYDISPERSE

GENERATION METHOD

DRI ATOMIZER. 30 PSI

SIZE DISTRIBUTION SHAPING

NONE

REMARKS

STABLE OUTPUT ONLY AFTER 1700. BETTEK STABILITY THAN #1: DIL. AIRFLOW 570 L/MIN.

WEATHER SYNOPSIS

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MEAN WEATHER CONDITIONS DURING EXPERIMENT

	AVG WD (DEG)	<b>N</b>
6	AVG WS (M/S)	47
DE CONDITIONS	AVG PRESS (MB)	N A
00151	А С. RH (8) 	AN
	AVG TEMP (C)	A N
ţs	AVG PRESS (MB)	A N
CONDITION	А (6, RH (8)	A N
ROOI	AVG TEMP (C)	42

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EXPER	I ME NT	*: 02					NUCLET	MEASURE	MENTS				PROCES	SING DATE: 81/07/29
		OBSERVER				ΡL	ATE TEMP		VOLUME			200	FcT	
NO.	N0.	NAME	0 2	START		COLD	H01 (C)	DELTA	SAMPLED (L)	SAT (%)	COUNT	)	) 7	REMARKS
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RUN:	2													
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RUN:	2													
			~~~~	1728 1728 1728 1728	1733 1733 1733 1733 1733						257 27 23 23 23 23 23 23 23 24 24 24 24 24 24 24 24 24 24 24 24 24	20 20 20 20 20 20 20 20 20 20 20 20 20 2		ISOTHERMAL MALE CHAM
RUN:	1													
8888888 111111	<b>66666</b>	4444 2222 2222 2222 2222 2242 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 224 2 2 2 2 2 2 2 2 2 2 2 2 2	ุกกงกงก	1704 1706 1708 1712 1712	1705 1709 17109 1713	06470 2646 204450 204450 204450 204450 204450 204450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 2000 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 20450 2000 20450 2000 200	0000000 0000000 00000000 00000000 000000	000040 040044 040044		00 00 00 00 00 00 00 00 00 00 00 00 00	8440 1100440 110000 110000 110000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 1000000			C. VALUE LISTED ARE PRESET ON INSTR. AND ARE APPROXIMATE ONLY. DROPLET POPULATIONS

EXPER	INENT	#: 05					NUCLET	MEASURI	EMENTS				PROCESS	51NG DATE: 81/07/29
NOCH	DUCT NO.	OBSERVER NAME	NO.	SAMPLE START	TIME	COLD PL	ATE TEMP HOT (C)	DELTA	VOLUME SAMPLED (L)	SUPER SAT (%)	NUCLE I COUNT	S + Z	EST	REMARKS
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RUN: 18 18 18	1 16 18		ውውው	1700 171 1730	1706 1714	21•10 22•60	26.50 25.25 25.25	245 6950 0550	000 000 000 000 000	1.15 .97 .27		1691 1645 840		
NUNUN NUNUN N	00000	RADXE RADXE RADXE RADXE RADXE RADXE	22222	171155 177155 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 71755 717557 717557 717557 717557 717557 717557 717577 717577 717577 717577 717577 7175777 7175777 7175777 717577777777	1730 1730 1730 1730					.57 .857 1.145 1.34	750 810 11550 11550			LARGE SUPPLY VARIATN Early in Pun Migh SS channel Failed
	-			11111111 00102020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 100202020 1002020 1002020 1002020 1002020 1002020 1002020 1002020 1002020 1002020 1002020 1002020 1002020 1002020 1002020 1002020 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 10000 100000 10000 10000 1000000	00000000000000000000000000000000000000	0000000000 0000000000 000000000000000	000000000 000000000 000000000000000000	4MN000000		•46	01 00 00 00 00 00 00 00 00 00 00 00 00 0			
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EXPEF	MACH	•0•	26666	RUN:	~~~~~~





80/10/08

EXPEPIMENT # 03

PURPOSE

INSTRUMENT COMPARISON WITH AMBIENT AEROSOL

DESCRIPTION OF EXPERIMENT

TIME START END 08 0CT+1980 0805 0930 DATE

NUCLEI TYPE

AMBIENT

GENERATION METHOD

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SIZE DISTRIBUTION SHAPING

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REMARKS

HIGHLY FLUCTUATING

WEATHER SYNOPSIS

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MEAN WEATHER CONDITIONS DURING EXPERIMENT

		 AVG WD	AN
	10	AVG WS (M/S)	AN
	DE CONDITION	AVG PRESS (MB)	AN
	0UTSI	 AVG RH (\$)	N A
1		AVG TEMP (C)	AN
	ţs	AVG PRESS (MB)	4 7
	M CONDITION	AVG RH	20.0
	ROO	AVG TEMP (C)	21.0

81/07/29																		ALUES APE N INSTR. APPROXIMATE
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	SAMPLE			00000000000000000000000000000000000000		00000000000000000000000000000000000000		0831 0834 0837		0845 0845 0845 0845 0845		0915 0918 0921 0924		0935 0932 0932 0935		00000 00000 000000 00000		00000 4444 00000
	NO.				•	~~~~~~		លលាមា		លលាលា		លល់លំហ		տատա				ุกญญ
#: 03	OBSERVE NAME							GAGIN GAGIN Gagin Gagin		6461 <i>N</i> 66661 <i>N</i> 66666		0000 001 0000 0000 0000 0000 0000 0000		2222 20122 20122 2006 2006 2006 2006		11111 100000 10000000 111111		A A A A A A A A A A A A A A A A A A A
RIMENT	DUCT NO.		1	លាបារបាលបារាជា	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	ապատանան	1	ውውውው	N	ወወወወ	e	ውውውው	4	ውውውው	1		-	 
EXPEI	N N N N N N N N N N N N N N N N N N N		RUN:	ເບເບັນເປັນເປັນເປັນເປັນເປັນເປັນເປັນເປັນເປັນເປ	RUN:	տատարատատ	RUN:	ውውውው	RUN:	ውውውው	RUN:	ወወወው	RUN:	ውውውው	RUN:		RUN:	

,29																10		
ING DATE: 81/07/	REMARKS															VOL SAMP X 10**5		E=2.1. 0.5 UM
PROCESS	EST															4		
	S I Z					170 9.3		206533 20633 20633 20633		1414054 0110574005 020574005 0400574005 040054005 040054005 040054005 040054005						64274 64627 64627 64627 64662 7 64662 7 64662 7 64662 7 64662 7 64662 7 64662 7 64662 7 64662 7 64662 7 64662 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 6476 7 7 6476 7 7 7 7		57
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	SUPER SAT (%)	1.25				• 2] • 13 • 093		485 - 1985 1				• 74 • 49 • 336		114 0089 03589 22589		0.0 40.047 1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0		.16
MEASUREMENTS	VOLUME SAMPLED (L)					000 000 000										0000 0000 0000 0000 0000 0000 0000		1.5
	DELTA	5.42		N44NN44WW 488344N⊶44 488-10-18-0								4m0 900		000000				0.00
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EXPEDIMENT # 04

80/10/08

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PURPOSE

DETERMINE SC AGREEMENT THROUGH USE OF VERY LARGE K (SWARPLY MONODISPERSE AEROSOL).

DESCRIPTION OF EXPERIMENT

TIME START END 0945 1338 DATE 08 OCT+1980 NUCLEI TYPE

NACL+ MONODISPERSE

GENERATION METHOD

DRI ATOMIZER. NO PREDILUTION

SIZE DISTRIBUTION SHAPING

EC: 486V, NOM. 0.036 UM DIA. 3.3 L/MIN. SAMPLE FLOW

REMARKS

T 73 (F), TD 50 (F), DIL, FLOW 300 L/MIN. Note disturbance between 1035 and 1040

WEATHER SYNOPSIS

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MEAN WEATHER CONDITIONS DURING EXPERIMENT

ROOM CONDITIONS *****************

OUTSIDE CONDITIONS

AV6 KD (DEG) ٩N AVG WS ٩N AVG PRESS (MB) ٩N AVG RH (\$) N N AVG TEMP NA AVG PRESS (MB) 852 AVG RH 20.0 AVG TEMP 22.0

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SDC/CFD RESULTS

haze chamber results

EXPEDIMENT # 05

PURPOSE

AS # 4, BUT HIGHER AEROSOL CONCENTRATION

DESCRIPTION OF EXPERIMENT

DATE TIME END

NUCLEI TYPE

NACL. MONODISPERSE

GENERATION METHOD

DRI ATOMIZER. NO PREDILUTION. 70 PSI.

SIZE DISTRIBUTION SHAPING

EC. 486V (NOM. 0.036 UM DIA.) 6.7 L/MIN. SAMPLE FLOW. 250 L/MIN DIL. FLOW.

REMARKS

T 74(F), TD 50(F)

WEATHER SYNOPSIS

٩N

MEAN WEATHER CONDITIONS DURING EXPERIMENT

AVG UD (DEG) NA AVG WS (H/S) Na OUTSIDF CONDITIONS AVG PRESS (MB) AN AVG RH ٩N AVG TEMP (C) AVG PRESS (HB) ******************************** 852 ROOM CONDITIONS AVG RH (%) 21.0 AVG TEMP 23.0

EXPER	IMENT	#: 05 Docedare				ć	NUCLEI	MEASURE	EMENTS				PRACES	ING DATE:	81/07/29
NOCH	NUCT NO.		NO.	STARTE T	LIME END	COLD	ALE LEND HOT (C)	DELTA	VOLUME SAMPLED (L)	SUPER SAT (\$)	NUCLFI	8 2 I 2 2	EST K	REMARKS	
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RUN:	N														
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RUN:	e														
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PURPOSE

INSTRUMENT COMPARISON WITH AMBIENT AEROSOL

DESCRIPTION OF EXPERIMENT

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MEATHER SYNOPSIS

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MEAN WEATHER CONDITIONS DURING EXPERIMENT

AVG WD (DEG) 90 AVG WS (M/S) 0.0 OUTSIDE CONDITIONS AVG PRESS (MB) AN AV6 RH (%) ٩N AVG TEMP (C) NA AVG PRESS (MB) 852 ROOM CONDITIONS АVG RH (%) 21.0 AVG TEMP 25.0

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IMENT	#: 06 085Erver				PL	NUCLEI ATE TEMP	MEASUREI	MENTS				PROCESS	ING DATE: 81/07/29
	ZAKE	NO.	SAMPLE Start	TIME	COLD	FOH CC)	DELTA	VOLUME SAMPLED (L)	SUPER SAT (%)	COUNT	085 2 1 2	EST K	REMARKS
			16455 16455	1651	22.18 21.44 21.37	26.57 26.49 26.49	4 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6			1022.9	997.6 1088.5	6 9 9 8	N=1202 S 0.77
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EXPE	21MENT	*: 05					NUCLEI	MEASUR	EMENTS				PROCES	51NG DATE: 81/07/29
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555	1551	A Second Se Second Second Seco		1652	1653	20.40	25.00	09.60 9.60 9.60		84 4 25 5 25	1956			DIV RAW COUNT AY 12.4 TO GET CONC.
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1666 166	1666	HUDSON HUDSON	ውውው	1637 1637 1637	1641 1641 1641	21.40 21.70 22.25	26.10 25.40 24.85	04.70 9.70	0024	•35• •250		1 853 1 366 1 866		
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1665 166	999 	HUDSON	ውውው	1659 1659 1659	1702 1702 1702	21.40 21.70 22.25	26.10 25.40 24.85	3.70 2.60	•0018 •0018 •0018	-115		1951 1356 880		
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166	666 222	HUDSON HUDSON HUDSON	ውውው	1713 1713 1713	1716 1716 1716	21.40 21.70 22.25	26.10 25.40 24.85	3.70 2.60	.0018 .0018 .0018	55.		1851 1376 792		
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009 111	000 111	NOSONHUSON	ውውው	1730	1732 1732 1732	21.40 21.70 22.25	26.10 25.40 24.85	2.60 2.60	•0018 •0018 •0018	222		1754 1316 746		
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NUCLEI MEASUREMENTS

EXPER	IMENT	*: 05					NUCLEI	MEASURE	HENTS				PROCESS	ING DATE: 81/07/29
NO CH	DUCT NO.	OBSERVER NAME	• 0 N	SAMPLE START	TIME	COLD PL/	ATE TEMP HOT (C)	DELTA	VOLUME SAMPLED (1)	SUPER SAT (%)	NUCLFI	8 I X	EST	REMARKS
1989	00000	NOOSOON NOOSOON NOOSOON NICOOSOON	10000	1713 1720 1720	 1716 1723 1732	00000 00000 00000 00000 00000	255.40 255.40 255.45	2-10 2-10 1-60	00218	.175	812 788 733 233 233			
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2222	2122	ALOFS ALOFS ALOFS ALOFS		1713 1713 1713 1713	1714 1714 1714 1714	255.00 255.00 255.00	00000 5222 5225	0000 0000		068 068 114 173	11 3447 6735			
RUN:	v													
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RUN:	T													
26	26	HINDMAN	Ŷ	1628	1717	23.00	23.00	0.00	1.2	.16	49	64		E=l.5, D>≠ 0.5 UM

EXPERI	MENT	*: 05				NUCLEI	MEASUR	EMENTS				PROCES	SING DATE:	91/07/29
	1010	OBSERVER	2 104 2	71 MC	Ъ	ATE TEMP	_		0.3011.2		500			
		NAME NO.	START	END	COLD	TOH TOH	DELTA	SAMPLED		COUNT	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	- N¥	REMARKS	
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26 RUN:	26 1	HINDMAN 6	1628	1717	23.00	23.00	00-0	1.2	.016	.0049	6400.		E= 0.0020.	D>= 5.0UM
27	27	0HTAKE 19 0HTAKE 19 0HTAKE 19	1620 1620 1620	1620 1620				••• ••• •••		ທານທ ອີເດີຍ ອີເດີຍ ອີເດີຍ			102% 103% 115%	





EXPEPIMENT # 08

PURPOSE

HAZE CHAMBER TEST WITH LARGE PARTICLES

DESCRIPTION OF EXPERIMENT

END	1152
START	0955
DATE	09 OCT,1980

NUCLEI TYPE

NACL. MONODISPERSE. D=0.18 UM

GENERATION METHOD

DRI ATOMIZER. 40 PSI. 0.025 \$ SOLUTION.

SIZE DISTRIBUTION SHAPING

EC. 8550V. MONODISPERSE FLOW 6 L/MIN. Dilution flow 210 L/MIN.

REMARKS

SLIGHT DRIFT OF PARTICLE CONC. TO LOWER VALUES.

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS DURING EXPERIMENT

	 AVG WD (DEG)	٩N
S	AVG WS	AN
IDE CONDITION	AVG PRESS (MB)	NA
00151	AVG RH (%)	٩N
	AVG TEMP (C)	A N
4S	 AVG PRESS (MB)	854
M CONDITIO	 AVG RH	22.0
ROO	AVG TEMP (C)	26.0

SSING DATE: 81/07/29	REMARKS				NOTION ODEDATION	ONLY ABOVE 0.15%												
PROCE	ы К Т																	
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	NUCLE I COUNT					00000 100000 100000		90000000000000000000000000000000000000		4400 6010 0000		44 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8		4400 004 0000		0000 0000 0000 0000		44 00000000000000000000000000000000000
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EXPER	IMENT	#: 08					NUCLEI	MEASURI	EMENTS				PROCES	51NG DATE: 81/07/29
NOCH	DUCT NO.	OBSERVER NAME	NO.	SAMPLE START	TIME	COLD PL	ATE TEMP HOT (C)	DELTA	VOLUME SAMPLED (L)	SUPER SAT (%)	NUCLEI	а 8 † 3 8	EST	REMARKS
   4 4 4 4 4 4 	   4 4 4 4 4 4 	ZZZZZZZ ZZZZZZZ ZZZZZZZ ZZZZZZZ ZZZZZZ	1000000 1 1	044444 000000					0000000 000000000000000000000000000000		200 200 200 200 200 200 200 200 200 200	     		F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55 F==+488.55F==+588.55 F==+488.55F==+588.55 F==+488.55F==+588.55F==+588.55 F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55F==+588.55
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ING DATE:	REMARKS							VOL. SAP CHAMBER		EEE EEE EEE EEE EEE EEE EEE EEE EEE EE		109% R.H
PROCESS	EST											
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	NUCLFI	13.3	300 630 630	0000 0000 00-00-00 00-00-00 00-00-00		00000 4041 00000				117 71 73 73 73 73 75 75 75 75 75 75 75 75 75 75 75 75 75		1.426
	SUPER SAT	640 ·	• 69 • 87 1 • 23					0011 0042 0042 0042		• 15 • 15 • 05 • 05 • 016		
MENTS	VOLUME SAMPLED	(L) .0018						900000 900000 000000				1
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SDC/CFD RESULTS

HAZE CHAMBER RESULTS

EXPERIMENT # 09

PURPOSE

HAZE CHAMBER TESTS (LARGE PARTICLES)

DESCRIPTION OF EXPERIMENT

)ATE	TIME START	END
9 OCT.1960	1152	1319

NUCLEI TYPE

NACL. MONODISPERSE. D=0.18 UM

GENERATION METHOD

DRI ATOMUZER. 40 PSI. 0.5% SOLUTION.

SIZE DISTRIBUTION SHAPING

EC. B550V; MONODISPERSE FLOW 6 L/MIN. Dilution flow 210 L/MIN.

REMARKS

SLIGHT DRIFT IN CONC TO LOWER VALUES DUE TO CHANGE IN MONODISPERSE FLOW

WEATHER SYNOPSIS

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MEAN WEATHER CONDITIONS DURING EXPERIMENT

	AVG WD (DEG)	<b>N</b>
S	AVG WS	٩N
IDE CONDITION	AVG PRESS (MB)	A N
OUTSI	AVA RH	A M
	AVG TEMP (C)	4 14
4S	AVG PRESS (MB)	854
M CONDITIO	AVG RH (\$)	22.0
ROO	AVG TEMP (C)	26.0

59																		CHAM
81/07/2																		IAL HAZE (
ING DATE:	REMARKS																	ISOTHERM
PROCESS	E N N		00.0															
	S 40 S 12		392 392								806.5 800.9 742.7 698.6 672.7				852.6 751.2	695.8 698.2 647.6		636
	NUCLF1 COUNT				720 640 500		760 650 620 620		0000		20000000000000000000000000000000000000		812 75 75 75 75 75 75 75 75 75 75 75 75 75		822 8822 7382 7628 7628 7628 7628 7628 7628 7628 76	7210.1 7210.1 7110.4 6633.1 632.1		636
	SUPER SAT (%)		• 16 • 36		1.01 1.01 1.04 1.0				1.00				ИЩФЮФЬФФЩ ••••••		សមាម សមាម សមាម សមាម សមាម សមាម សមាម សមាម			.15
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NUCLET	ATE TEMP HOT (C)		25.30		26.90 26.90 27.00 27.10		26.80 26.90 26.70 27.10		24-00 24-00 27-20		001-100 01-1446 001-1446 001-1446 00100000000000000000000000000000000		0-1-1-1-000 1-1-1-1-000 0-0-0-0-00000000		0045 6045 756 756 757 757 757 757 757 757 757 75	000000 000000 000000		
	COLD PL		23.30		21.80 23.30 23.70 25.20		22.00 23.20 24.10 25.20		231.80 23.50 23.50 25.00		222 222 222 222 222 222 222 222 222 22		4610000141 40000011110 10000001		200 20 20 20 20 20 20 20 20 20 20 20 20	000000 449996 1000000 1000000 1000000000000000000		
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	SAMPLE		1229 1231		1206 1208 1210		1218 1220 1223 1223		12320		10000000000000000000000000000000000000		2009894684		1221 1221 1221	1221 1221 1221 1221 1221 1221		1251
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21 MF NT	DUCT NO.		NN	1	ውውውው	~	ወወወወ	Ē	ውውውው	-		N	000000000	Ð	0000		-	11
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SSING DATE: 81/07/29	REMARKS			POSS. ZERO RESET EAR		Тангорована тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала тала		F=25. D=0.38 F=255. D=0.67 F=255. D=0.67 F=25. D=2.47 F=25. D=2.47 P=25. D=2.47		BACKGROUND ABOUT 10 PER CC. TEMPS APPROX IN ABSOLUTE. OK IN DELT READING ROYCO CH. 2 LESS BACGROUND . NEAR ISOTHERMAL ON LAST POINT		
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	NUCLE1 COUNT	108894 10889 1411 1		₩₽₽₽₽₽₽₽₽₽₽₽₽₽₽ 8440£4458749₩₽79 ₩₽₽₽₽₽₽₽₽₽₽₽₽₽₽₽₽₽		N N N N N N N N N N N N N N		1 8 8 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		୶ୠୠଊୢଊଊୠୠ୶ ୢ୳ୡଡ଼୲୷ଡ଼ଢ଼ଢ଼୴୴ଊଡ଼ ୴ୠଢ଼୷୶୴ୠ୴ଡ଼ଊ		0000000 
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	DBSERVER											
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EXPER	NO.		RUN:		RUN:	*********	RUN:	****	RUN:	ມດາດປະການການ	RUN:	9999999 

EXPEF	<b>ZIMENT</b>	*: 09					NUCLET	MEASURE	MENTS				PROCESS	ING DATE: 81/07/2	6
MACH NO.	DUCT NO.	OBSERVER NAME	N0	SAMPLE START	TIME	COLD	ATE TEMP HOT (C)	DELTA	VOLUME SAMPLED (L)	SUPER SAT (%)	NUCLFI	S I 2 0	Est K	REMARKS	
1 1 + 0 + 0 + 0 + 1 + 1 + 1 + 1 + 1 + 1 + 1 + 1 + 1 + 1	INCORRE INCORRE	CT EXPER. NO.	6 6	1247	1250	21.70 22.30	25.50 24.80	3.80 2.50	.0018	-58 -25	552 551	8 9 8 8			5 3 5
RUN: *16 16	2 NC0RRE 16	CT EXPER. NO. HUDSON HUDSON	ው ውው	1259 1259 1259	1302 1302 1302	21.40 21.70 22.30	26.10 25.50 24.80	4.70 3.80 2.50	• 0024 • 0024 • 0024	50 0 50 0 10	508 507 507				
RUN:	н 17777 77777		~~~~~	1206 1209 1210 1211	1200 1210 1211 1211 1212					500-100 004-00 000-00	809 809 799.2 769.2 769.6	2098 2098 2099 2099 2099 2099 2099			
RUN: 1881: 1881: 1881:	1 100886 18 18 18	CHUDSON CHERSPER HUDSON HUDSON	ው ውውው	1203 1228 1228 1259	1208 1230 13550	24.00 23.75 21.65 21.65	25 05 255 05 255 20 2400	1.05 2.45 5.45 5.75	.003 .0016 .0018 .0028	• 0 • 3 • 0 • 3 • 1 • 1 • 1 • 1 • 1 • 1 • 1 • 1 • 1 • 1	6 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7				
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81/07/29				02=10 UM		NOMINAL
SING DATE:		REMARKS				99% R.H. 108% R.H.
PROCES						
	5	6 I Z				
		COUNT				42.318 70.531
	03010	SAT SAT SAT		027		
EMENTS	3441 104	SAMPLED	1	•••		.1225
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NUCLE	LATE TEM	H01	25.00	52.00 52.00		
	ā	COLD	25.00	200 222 222		
	TINC		1315			1205
	S IGMD	START	1213	1213		1205
	SERVER	NO.		000		19
60 :#	190	NAME		NAMONIH		OHTAKE OHTAKE
IMENT	T	.00	26	660 000	-	27
EXPEF	1044		26	000 000	RUN:	27





EXPEPIMENT # 10

PURPOSE

BIMODAL NACL TO GET K-PLATEAU

DESCRIPTION OF EXPERIMENT

DATE TIME END START END 09 DCT+1980 1417 1620

NUCLEI TYPE

NACL. 0.1% SOLUTION FOR SMALL. 0.5% FOR LARGE SIZE.

GENERATION METHOD

ATOMIZERS: DRI. 40 PSI. NANOMIST 19PSI.

SIZE DISTRIBUTION SHAPING

ECL, 0.10M PIA., EC2 0.03 UM PIA. FLOW 410 L/MIN.

## REMARKS

TOTAL N STEADY, BUT SMALL SIZE DRIFT UP, LARGE ONE DOWN.

WEATHER SYNOPSIS

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MEAN WEATHER CONDITIONS DURING EXPERIMENT

KOOM CONDITIONS

AVG VD (DEG) NA AVG WS (M/S) -----AVG PRESS (MB) ٩N AVA RH (5) ۹ Z AVG TEMP (C) ٩Z AVG PRESS (MB) 851 AVG RH (%) 22.0 AVG TEMP 26.0

OUTSIDE CONDITIONS

EXPE	LIMENT	*: 10					NUCLET	MEASUR	EMENTS				PROCES	51NG DATE: 81/07/29	
MACH NO.	DUCT NO.	OBSERVER NAME	NO.	SAMPLE START	TIME	COLD PL	ATE TEMP HOT (C)	DELTA	VOLUME SAMPLED (L)	SUPER SAT (%)	NUCLFI	S I Z	E A T	REMARKS	
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ING DATE: 81/07/29	REMARKS	ПП ПП ПП ПП ПП ПП ПП ПП ПП ПП	F=39	555555555 567555555 577755555 5775555 575555 57555 57555 5755 5755 5755 5755 5755 5755 5755 5755 5755 5755 5755 5755 5755 5755 5755 5755 5755 5755 5755 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 575 57		BACKROUND NOW 10 10 16 PER CC. Reading Royco CH. Tempes are approx. In Absolute, ok in Deltat Is CH 3 CMT								
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EXPERI	INENT .	*: 10					NUCLET	MEASUR	EMENTS				PROCESS	ING DATE:	81/07/29
NOCH NOCH	DUCT NO.	0BSERVER	10	TARTE	TIME END	COLD	LATE TEMP HOT (C)	DELTA	VOLUME SAMPLED (L)	SUPER SAT (%)	NUCLFI COUNT	ORS N	E ST	REMARKS	
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EXPERIMENT # 11

PURPOSE

INSTRUMENT COMPARISON WITH AMBIENT AIR.

DESCRIPTION OF EXPERIMENT

DATE TIME END 514RT END 09 DCT+1980 1636 1725

NUCLEI TYPE

AMBIENT

GENERATION METHOD

٩N

SIZE DISTRIBUTION SHAPING

**N** 

REMARKS

FLOW RATE 700 L/MIN.

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS DURING EXPERIMENT

AVG WD (DEG) 120 AVG WS 1.0 OUTSIDE CONDITIONS AVG PRESS (HB) 851 AVG RH ٩N AVG TEMP (C) NA AVG PRESS (MB) 851 ROOM CONDITIONS AVG RH (%) 21.0 AVG TEMP 29.0

EXPER	IMENT	#: 11					NUCLET	MEASUR	EMENTS				PROCESS	SING DATE:	81/07/29
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EXPE	RIMENT	#: 11					NUCLEI	MEASUR	EMENTS				PROCES	SING DATE: 81/07/29
NOCH NOCH	DUCT NO.	0BSERVER NAME	°02	SAMPLE	TIME	COLD PL	ATE TEMP HOT (C)	DELTA	VOLUME SAMPLED (L)	SUPER SAT (%)	NUCLEI	ORS S I Z	EST	REMARKS
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haze chamber results

SDC/CFD RESULTS

80/10/10

PURPOSE

INSTRUMENT COMPARISON WITH AMBIENT AEROSOL.

DESCRIPTION OF EXPERIMENT

TIME START END 0830 1002 DATE 10 OCT+1980

NUCLEI TYPE

AMBIENT

GENERATION WETHOD

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SIZE DISTRIBUTION SHAPING

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REMARKS

T 23.5 (C). TD 12 (C)

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS DURING EXPERIMENT

AVG WS (M/S) OUTSIDE CONDITIONS AVG PRESS (M9) AVG RH AVG TEMP (C) AVG PRESS (MB) ROOM CONDITIONS AVG RH (\$) AVG TEPP

AVG WD (DEG) 360

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		SUPER SAT (\$)	1.005		00000000000000000000000000000000000000		.068 .069 .114 .173		40500		01100000000000000000000000000000000000				
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I MEASURE	a	DELTA			NWN000000		0000 0000 0000		00000 00404 M44N-		0000000 N&CENSN NM4NP				
NUCLE	LATE TEM	то С)			00000000000000000000000000000000000000		0000 0000 0000 0000 0000		00000 0000 0000 0000 0000 0000 00000 0000		0000000 •••000000 ••••00000 ••••000000 ••••000000				
	•	COLD	8 9 9 9 9 9		00000000000000000000000000000000000000		0000 0000 0000 0000 0000		00000 50000 50000 50000		800000 800000 8000000 8000000 80000000 8000000				
			8 8 9		00000000000000000000000000000000000000		0000 00000 00000		0884 0915 0915 0915 0915		0000000 40-00000 40-0444 40-04444		00000000000000000000000000000000000000		6633
		START	000		00000000000000000000000000000000000000		4444 0000 0000		00000 40000 40000 0000		00000000000000000000000000000000000000		ᲝᲝᲝᲝᲝᲝᲝ ୫ ୫ 444444 ©©©©©©©©©		0833
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EXPER			000	RUN:		RUN:	2222	RUN:	44444	RUN:	ທທທທທທາ ດດາດດີດດີດ	RUN:	~~~~~~~~~ ^^^^^^	:NUA	27



SDC/CFD RESULTS

HAZE CHAMBER RESULTS

PURPOSE

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POLYDISPERSE FOR MODERATE K.

DESCRIPTION OF EXPERIMENT

TIME START END 10 0CT+1960 1017 1300 DATE

NUCLEI TYPE

(NH4) 2504

GENERATION METHOD

NANOWIST, 5.5 PSI, PREDILUTION 150 BR., 1.77 G/L SOLUTION.

SIZE DISTRIBUTION SHAPING

NONE

REMARKS

DILUTION FLOW RATE 590 L/MIN. UNSTABLE. SEE STRIPCHART FOR REFERENCE.

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS DURING EXPERIMENT

AVG ND (DEG) NA AVG WS ٩X OUTSIDE CONDITIONS AVG PRESS (MB) AN AVG RH (#) ٩ AVG TEMP (C) ٨N AVG PHESS (HB) 850 ROOM CONDITIONS AVG RH 21.0 AVG TEMP (C) 27.0

80/10/10

81/09/26	I												UCTUATIONS		. CHAMBER		
SING DATE:	REMARKS												AEROSOL FL		ISOTHERMAL		
PROCES	EST																
	8 I Z		823 867												78 188 19 18 19 19 19 19 19 19 19 19 19 19 19 19 19		
	NUCLEI				2340 850 1850 18650		2420 1090 240 240 240		0440 440 1		900 100 100 100 100 100 100 100 100 100		20032000 200320 200320 200320 200120 200120 200120 2000 20000 2000000		1990 1990 1990 1990 1990 1990 1990 1990		1058 1680
	SUPER SAT (\$)		.23		1. 1. 1. 196		98 102 16		500 500 500 500 500 500 500 500 500 500		ດທິດທິດ–ທິດຈາກດານ ທີ່ທີ່ດີດ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີດີ ທີ່ມີມີດີ ທີ່ມີມີດີ ທີ່ມີມີດີ ທີ່ມີມີດີ ທີ່ມີມີດີ ທີ່ມີມີດີ ທີ່ມີມີດີ ທີ່ມີມີດີ ທີ່ມີມີດີ ທີ່ມີມີດີ ທີ່ມີມີ ທີ່ມີມີ ທີ່ມີມີ ທີ່ມີມີ ທີ່ມີມີ ທີ່ມີມີ ທີ່ມີມີ ທີ່ມີມີ ທີ່ມີມີ ທີ່ມີ ທີ່ມີມີ ທີ່ມີ ມີ ມີ ມີ ມີ ມີ ມີ ມີ ມີ ມີ ມີ ມີ ມີ ມ		ин-400-400-и -		•000 •00 •00 •00 •00 •00		έν. •
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NUCLEJ	-ATE TEMF HOT (C)	5 5 5 5	26.00 26.20		2000 200 200 200 200 200 200 200 200 20		24.10 26.30		264.00 266.20 286.40 286.40 286.40		ຒຎຎຎຎຎຎຎ ຎຉຎຎຎຎຎຎຎຎ ຎຎຎຎຎຎຎຎຎ		ຨຨຬຏຠຬຬຠຬຨ ຉຨຉຉຉຉຉຉຨຑຉ ຎຎຎຎຎຎ				28.50 28.50
	COLD		23.90		22.70 24.60 25.50 26.10		25.60 26.30		233.20 255.10 26.20 26.20				+ 80018-80000 0000-10000 0000-0000 0000-0000 0000-0000 0000-0000-0000 0000-0000-0000 0000-0000-0000 0000-0000-0000 0000-0000-0000 0000-0000-0000 0000-0000-0000 0000-0000-0000 0000-0000-0000 0000-0000-0000 0000-0000-0000 0000-0000-0000 0000-0000-0000 0000-0000-0000 000000				26.01
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	SAMPLE START		1205 1208		1144 1152 1152		1159 1201 1203		1213 1216 1216		44000000000000000000000000000000000000		00000000000000000000000000000000000000		12238 2388 2388 2388 2388 2388 2388 2388		1145
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9/26			ECORDS	ECORDS			PER CC DUCE		
ING DATE: 81/0	REMARKS		PHOTOGRAPHIC R Taken	PHOTOGRAPHIC R Taken			BACKGROUND 10 AVERGING LAST COUNTS PER SS MAY NEED TO REI MAIN FLOW FOR LOWEST SS PTS LAST PNT CH 3		
PROCESS	EST								
	80 . N								
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	SAMPLE	1150	1152 1155 1156 1156	112443 2445 2445 2445 2445 2445 2445 2445	88888888888888888888888888888888888888	1251 1251 1251	1739955000 9779111000 1777111000 17771111111		
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81/09/26		0 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1					MEASUREMENT 150	T DOWN. CHAMBER.
SING DATE:		REMARKS					ONLY ONE AT S=000.	INSTRUMEN REPAIRING
PROCES	F 0 F							
	500							
		COUNT	1069 1169 1169 1169 1169	990 1910 29332 2938 2938 2938	1103 1410 2230	0001 0000 000 0	530	
		SUPER SAT (*)	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1 4061 406	.59 .89 .117	18 18 19 19 19 19 19 19 19 19 19 19	.15	
MENTS	1	VOLUME SAMPLED (L)	++mmm 00000 00000	0.00 0.00 0.00 0.00 0.00 0.00 0.00 0.0				
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	T IS OBSFRVE	NAME	22000 2000 2000 2000 2000 2000 2000 20	NOSQUH 2022QUH 2022QUH 2022QUH 2022QUH	00 00 00 00 00 00 00 00 00 00 00 00 00	118.000 118.0000 118.00000 118.00000 118.00000 118.00000 118.00000 118.00000 118.00000 118.0000000000	SERPOLAY SERPOLAY	HI NDMAN HI NDMAN
		DUCT NO.	96096		00000 70000 7	-	1 24 24	1 26 26
10.2	EXPE	MACH	22000	RUN: 18 18 18 18	RUN 2000 8000		RUN: 24	RUN: 26







PURPOSE

SAME AS #13.

DESCRIPTION OF EXPERIMENT

NUCLEI TYPE

(NH4)2504. POLYDISPSERSE.

SIZE DISTRIBUTION SHAPING

NONE

REMAKKS

STABILITY VARIES, SEE STRIPCHART FOR ADJUSTMENTS.

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS DURING EXPERIMENT

AVG ND (DEG) NA AVG WS (M/S) AN OUTSIDE CONDITIONS AVG PRESS (MB) ٩N AVG RH (#) ٩ -----AVG TEMP (C) ž AVG PRESS (MB) 850 ROOM CONDITIONS АVG RH (\$) 21.0 AVG TEMP 27.0

80/10/10

81/09/26																	
SING DATE:	REMARKS																
PROCES	EST K																
	S 80 0 1 2		903 950												773 773 79.9 1.6 1.6		
	NUCLEI				2480 1050 380		1950 2200 2200		2670 9550 300		11221 1221 1221 1221 1222 1222 1222 12		00000000000000000000000000000000000000		779 977 9.6 1.		758 758 7 06 7 06 7 06 7 06 7 06 7 06 7 06 7 06
	SUPER SAT (F)		.23		1 1 1 1 2 1 2 1 2 1 2 1 2 1		12597		1.08 		ທເກເກເດີນ ດັບບັບເມື່ອ ••••• •••••		, , , , , , , , , , , , , , , , , , ,		0000 10000 10000		••••• •0391 •••••
EMENTS	VOLUME SAMPLED (L)						~~~~		~~~~		*********		**********				
I MEASURI	DELTA		2.40		N		4.80 3.70 2.70		0000 0000 00000 00000		ທທພທພທຈາທທ ທູດວາເທີນພູວວ ຈູມວວນ-ຈູທູທີ່ມີສຸດ		00000000000000000000000000000000000000				
NUCLE	ATE TEM		21.60		26.00 26.000 26.000 26.000 26.000 26.0000000000		28.30 28.30 28.30 28.30 28.30 28.30 28.30 28.30 28.30 29.30 20.30 20.30 20.30 20.30 20.30 20.30 20.50		84400 8440 8440 8440 8440 8440 8440 844		๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛						
	COLD		19.20		00000 9999 9999 9999 9999 9999 9999 99		60000 500 500 500 500 500 500 500 500 50		0000 4000 4000 4000		00000000000000000000000000000000000000		00000000000000000000000000000000000000				
			1429 1431												11111 2000 2000 2000 2000 2000 2000 200		1111 4441 4416 41666 10666
	START START 		1428 1430		19461 19467 19509 19500 10000 10000000000		8000 1111 1111		4441 4441 1916 1918		11111111111 100000000000 440000000000 4400000000		00000000000000000000000000000000000000		1337 13337 13337 1337 1337 1337		1999999 195999 195999 195999 195999 195999 19599 10599 1000 1000
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EXPER	IMENT	#:]4				i	NUCLEI	MEASUR	EMENTS				PROCESS	SING DATE: 81/09/26
MACH	DUCT NO.	OBSERVER NAME	N0.	SAMPLE START	TIME	COLD PL	ATE TEMP HOT (C)	DELTA	VOLUME SAMPLED	SUPER SAT (\$)	NUCLEI COUNT	S I S	EST K	REMARKS
RUN:	I													
	<u>66666</u>	A A A A A A A A A A A A A A A A A A A	งงงงง	1352 1353 1356 1356 1358	1353 13553 13554 135554 1355566 135556 135556 135556 135556 1355566 135556	240 26 26 26 26 26 26 26 26 26 26 26 26 26	000000 00000 00000 00000 000000	50804 50804 50804 50804		. 55 . 155 1. 25 1. 25	9999900 999990 999990 999990 999990 999990 999990 999990 999990 999990 999990 99900 99000 90000 90000 90000 90000 90000 90000 90000 90000 90000 90000 90000 90000 90000 90000 90000 90000 9000000			
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RUN:	ŗ													
444444444 ***	4444444444 Марарарияна		ውውውውውውውውውው የትርጉ የቀቀም የቀቀም የቀቀም የቀም የቀም የቀም የቀም የቀም የቀም		ຏຏຏຏຏຏຏຏຏຏຏ ຎຎຎຎຎຎຎຎຎຎ ຎຎຎຎຎຎຎຎຎ				00000000000000000000000000000000000000		и 100004096 160904696 1010694696 1010694696			FFFFFFFFFFFF 444400000 4411 44440000000 444440000000
RUN:	1	DAFR. FRFD		7551	AEE I	18.36	25,06	6.70		1 . A6	3178			BACKGROUND CNT
1	າດຕາມຕາມດາຍ ເປັນເປັນເປັນເປັນເປັນເປັນເປັນເປັນເປັນເປັນ		๛๛๛๛๛๛๛๛๛ ๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛		000-000000	04900440740 04000440740 04000440 04000440 04000440	00000000000000000000000000000000000000	0,000,400,000 - CLOOO (1) (0 - CLOOO		947000040 947000000 9470000000	00000000000000000000000000000000000000			10 PER CC. DROP FR CC. DROP AT LOVER SS BAD AT LOVER SS LOVERED MAIN FLO FROM TO 1315. Last few min 15 Pct.
RUN:	-													
			თ დდდდდდდ	COC0000444 NNNNNNMM 4444444 TTTTTTTTT	444000000 NNNN0000 444444444 MUNN00000	211-35 211-35 211-35 211-35 211-35 211-35 211-35 211-35 211-35 21-	ທທທທທທທ ຈນຈະດາຈະບຈ ພພວະບວນເວັດ ເວັດເວັດເວັດ ເວັດເວັດເວັດ ເວັດເວັດ ເວັດເວັດ ເວັດເວັດ ເວັດເວັດ ເວັດເວັດ ເວັດເວັດ ເວັດເວັດ ເວັດເວັດ ເວັດ	4WN4WN4WN - 401-601-60 Nonnonnon	44444444 NUNUNUNUN COCOCOCOCO COCOCOCOCO COCOCOCOCO COCOCOCO COCOCOCOCO COCOCOCOCO COCOCOCOCO COCOCOCOCO COCOCOCOCO COCOCOCOCOCO COCOCOCOCO COCOCOCOCO COCOCOCOCOCOCO COCOCOCOCO COCOCOCOCOCOCOCO COCOCOCOCOCOCOCO COCOCOCOCOCOCOCOCO CO	ຬໞໞຎຬຎຎຬຎຎ ຎຎຎຎຎຎ	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2			
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PURPOSE

MONODISPERSE AEROSOL WITH SIZE USEFUL TO BOTH HAZE AND CCN INTRUMENTS. I.E., B APPRUX. = 0.092 UM.

DESCRIPTION OF EXPERIMENT

TIME START END DATE

NUCLEI TYPE

(NH4)2 S04

GENERATION METHOD

DRI ATOMIZER. 50 PSI. PUMP 2.

SIZE DISTRIBUTION SHAPING

EC. 2780 V.. MONODISPERSE FLOWS 6 L/MIN.

REMARKS

DILUTION FLOW 390 L/MIN.

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS DURING EXPERIMENT

AVG ND (DEG) NA AVG WS ٩N OUTSIDE CONDITIONS AVG PRESS (MR) AVG RH (%) ¥ N AVG TEMP (C) ٩N AVG PRESS (MB) -----850 ROOM CONCITIONS ******** AVG RH (%) 21.0 AVG TEMP (C) 27.0

80/10/10

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	NUCLEI COUNT	1755 15455 1673 1635	1410 1410 1410 1440 1440	212	115055 115055 115055 11505 11505 11505 11505 11505 11505 11505 11505 11505 11505 11505 11505 11505 11505 11505 11505 11505 11505 11505 11505 11505 11505 11505 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 1155 11		• • • • • • • • • • • • • • • • • • •		1111111 0000000000 0000000000000000000		1111111111 8000000000000000000000000000		
	SUPER SAT (#)	. 75 . 75 1.25	- - - - - - - - - - - - - - - - - - -	• 1 < 9	1 • • • • • • • • • • • • • • • • • • •						,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		
	VOLUME SAMPLED (L)												
	DELTA	50-40 19-09 19-09	004400 604000 404000	5 5 5 7 1 1	54000000 5400040 55500040				₽9048880999 ₽9899999 9989999999		400400400 66666666 666666666 6666666666		
	LATE TEMF Hot (C)	60000 10005 10005	000000 0000 0000 000000 000000	20.20	00000000 N00000 000000 0000000 MMMNNNM				00.3642000 0044400000 0044400000 00000000000		00000000000000000000000000000000000000		
	COLD	255 255 255 255 255 255 255 255 255 255	000000 000000 000000 000000 0000000000	26.05	, , , , , , , , , , , , , , , , , , ,				00000000000000000000000000000000000000		20000000000000000000000000000000000000		
	END	1514 1514		1530	1166124 166116 166116 166116 10986424 10986424 109864242		ຏຏຏຏຏຏຏຏຏ ຎຎຎຎຎຉຉຉຉ ຎຎຎຎຎຉຉຉຉຉ		4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4		00000000000000000000000000000000000000		
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	NO.	1 0000 1	๛๛๛๛	ŝ	๛๛๛๛๛๛		<b>ႧႧႧႧႧႧႧႧႧ</b> Ⴇ		๛๛๛๛๛๛๛๛๛๛ ๛๛๛๛๛๛๛๛๛๛๛ ๛๛๛๛๛๛๛๛๛๛๛๛๛๛		<b>ውውውውውውውውው</b>		
	OUSERVE												
#: 15	NAME	A Y E R S S S S S S S S S S S S S S S S S S	<b>4444</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>77777</b> <b>777777</b> <b>777777</b> <b>777777</b> <b>7777777</b> <b>7777777</b> <b>7777777777</b>	AYERS	<b>A A A A A A</b> <b>A A A A A A A</b> <b>A A A A A A A A A</b> <b>A A A A A A A A A A A A A A A A A A A </b>				88888888888888888888888888888888888888				
RIMENT	DUCT NO.		N 0000000	<u>າຕ</u> ຕ		-	*********	1	ດດດດາດດາດດາດ ເມັນເປັນເປັນເປັນເປັນເປັນເປັນເປັນເປັນເປັນເປ	-		ŝ	
EXPE	NO.		N NAMERO	I 3	<u> </u>	RUN:	*****	RUN:	 ນານນານດາດດາວດາວ	RUN:	2222222222	RUN:	

NUCLEI MEASUREMENTS

81/09/26							X 10++5	T DOWN.
SING DATE:	REMARKS						VOL SAMP	INSTRUMEN REPAIRING
PROCES	EST K							
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	NUCLEI	1927591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 977591 9775910000000000000000000000000000000000	1253 1589 1589 1589 1963 1860	400041 800041 810640 8010580	NF NØF FRØM 440 MORS - 000 50 MORS - 00000 MORS - 000000 MORS - 000000 MORS - 000000 M	150 400 1530 2180	966 966 2739 2739 2739 2739 2739 2739 266 2739 266 2739 266 2739 266 2739 266 2739 266 2739 266 2739 2739 2739 2739 2739 2739 2739 2739	
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MENTS	VOLUME SAMPLED (L)	NNN444		 		~~~~~ ~~~~~~	8110 810 810 810 810 810 810 810	
I MEASURE	P DELTA	4005005		N4WN4 000000 000000000000000000000000000	40000000000000000000000000000000000000	-NN44		
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	COLD	2211-200 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 221-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-2400 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-240 201-2400 201-2400 201-2400 201-2400 201-2400 201-2400 201-2400 201-2400 201-2400 201-2400 201-2400 201-2400 201-2400 201-2400 201-2400 201-2400 200-2400 200-2000 200-2000 200-2000 200-200000000		000000 000000 000000 000000 000000 00000	00000000000000000000000000000000000000	284.00 284.00 284.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 286.50 200 286.50 200 200 200 200 200 200 200 200 200 2	0000000 0000000 10000000 10000000 1000000	
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	NO.	1000000 1	~~~~~	ውውውውውው	00000000000000000000000000000000000000	ອນດ້ານຄາຍ ເບັນເບັນເບັນ	നനനനനന	60
#: 15	UBSERVER NAME		C C C C C C C C C C C C C C C C C C C		11777777777777777777777777777777777777	SERPOLAY SERPOLAY SERPOLAY SERPOLAY SERPOLAY	8-88-9 0084 0084 0084 0084 008 008 008 008 008	NAMANI H Hundhani H
RIMENT	DUCT NO.	000000				55555 20000	ດທິດດິດດິດ ວັງວິດດິດດິດ -	1 26 26
EXPE	MACH	000000	RUN: 17 17 17	NUN 881 888 888 888 888 888 888 888 888 88			ສ ວັດທານທານ ເຊິ່າ	RUN: 26 26

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SDC/CFD RESULTS

HAZE CHAMBER RESULTS

PURPOSE

INSTRUMENT COMPARISON WITH AMBIENT AEROSOL.

DESCRIPTION OF EXPERIMENT

NUCLEI TYPE

AMBIENT

GENERATION METHOD

٩N

SIZE DISTRIBUTION SHAPING

NA

REMARKS

NONE

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS DURING EXPERIMENT

AVG WD (DEG) 270 AVG WS 4.0 OUTSIDE CONDITIONS AVG PRESS (MB) -----AVG RH (\$) NA AVG TEMP (C) -----AVG PPESS (MB) ------850 ROOM CONDITIONS AVG RH (%) 20.0 AVG TEMP 29.0

80/10/10

EXPER	IMENT	#: 16					NUCLEI	MEASURE	EMENTS				PROCESS	SING DATE:	81/09/26
		OBSERVER				7	ATE TEMP								
		NAME	NO.	START		COLD	101 101	DELTA	VULUME SAMPLED (L)	SUPER SAT (#)	COUNT	S I Z	EST	REMARKS	
			ł												
RUN:	I														
NN	NN	POLITOVICH POLITOVICH	202	1656 1658	1657 1700	23.90 23.80	26.00	2.40		.17		456 456			
RUN:	-														
លលល់	លហហរប			1635 1637 1638	1635 1637 1637	22.20	0000 6666 4444 NNN	0000 1400 NM41			40 40 40 40 40 40 40 40 40 40 40 40 40 4				
RUN:	` ~		:				C • • • •			1.6	0				
មកម្មសាល់ស្នា	սուսուսուս		~~~~~~~	0100000 6444440 6000000 64444400	0004999 444444 444444	00000000000000000000000000000000000000	COCCOCO 6666660 4444440 NNNNNNN	4000000 4004040 4004040			00490000 0049000 00490000				
RUN:	r														
ውውውው	<b>0000</b>	GAGIN Gagin Gagin Gagin	លលល់ល	1636 1638 1648 1668		23.00 25.10 25.10 25.10	28610 28610 286510 28650 28650	0000 0000 0000	~~~~		1310 270 100				
:NUA	2														
ውውውው	ውውውው	GAGIN Gagin Gagin Gagin	ഗഗഗഗ	16659 16559 16572 16572		22.70 24.30 25.00 26.10	28.30 28.30 28.20 28.10	0000 4000 ••••	~~~~	1.14 .61	1550 2000 2000 2000				
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RUN:	-														
		AYERS Ayers Ayers Ayers	ุกกงกง	500004 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 64444 644444 644444 64444 64444 644444 644444 6444444	16441 16441 16441 16441	26.98 255.982 255.982 24.75	6660¢ 19444 66606 66606 66606	₩₩400 ₩₩400 ₩₩400		- 52 - 755 - 755	488 870 915 1240 1290				
RUN:	~														
<b>00000</b>		AYERS Ayers Ayers Ayers	ุกกกกก	90098 9009 9009 9008	166564 665544 5655447	664834 265526 266834 266834	000000 MMNNNN 00000 MMMMMM	00400 00400 00400		1.25 75 255	1140 1020 870 750				
RUN:															
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56			010.					1110	
SING DATE: 81/09/	REMARKS	11111 10000 10000 10000 10000	MAIN FLO ABT 130 CNTS, BACKGROUND AEROSOL CHANGING AS FCN, TIME 7 MAIN FLO 1287					IMPORTANT FLUCTUA Loy Level AT Abou 1647	3000 × 0703
PROCES	EST K								AL.1
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SDC/CFD RESULTS

HAZE CHAMBER RESULTS

PURPOSE

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INSTRUMENT COMPARISON WITH AMBIENT AEROSOL.

DESCRIPTION OF EXPERIMENT

DATE	START	END
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NIICI ET TVDE		

NUCLEI TYPE

AMBIENT

GENERATION METHOD

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SIZE DISTRIBUTION SHAPING

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REMARKS

20% R.H.

WEATHER SYNOPSIS

NONE

WEAN WEATHER CONDITIONS DURING EXPERIMENT

ROOM CONDITIONS

OUTSIDE CONDITIONS

AVG VD (DEG) VAR AVG WS <1.0 AVG PRESS (MB) AVG RH (%) 21.0 AVG TEMP (C) 26.0

80/10/10

109/26																	AEROSO		
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80/10/13

INSTRUMENT COMPARISON WITH SMALL MONODISPERSE CCN. PURPOSE

DESCRIPTION OF EXPERIMENT

TIME START END 0E11 0101 DATE 13 OCT+1980

NUCLEI TYPE

(NH4)2 S04, MONODISPEHSE, D=0.04 UM.

GENERATION WETHOD

DRI ATOMIZER, 70 PSI, 1.77 G/L SOLUTION.

SIZE DISTRIBUTION SHAPING

EC. 634V, MONODISPERSE, FLOW RATE = 5.8 L/MIN. DILUTION FLOW 570 L/MIN.

REMARKS

NONE

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS DURING EXPERIMENT

AVG WD (DEG) ...... AVG WS OUTSIDE CONDITIONS *********************************** AVG PRESS (MB) AVG RH AVG TEMP (C) AVG PRESS (MB) 848 ROOM CONDITIONS AVG RH (%) 21.0 AVG TEMP (C) 26.0

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81/09/26				NOMINAL
SING DATE!	REMARKS			97% R.H. 100% R.H. 109% R.H.
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EXPER	MACH NO.	500000 000000	RUN 20000 N	RUN: 27 27



SDC/CFD RESULTS

HAZE CHAMBER RESULTS

EXPERIMENT # 19

PURPOSE

INSTRUMENT COMPARISON WITH LARGE MONODISPERSE AEROSOL (FOR INCLUSION OF HAZE Instrument).

DESCRIPTION OF EXPERIMENT

DATE TIME END

NUCLEI TYPE

(NH4)2 S04, MONODISPOERSE, D=0.14 UM.

GENERATION METHOD

DRI ATOMIZER. 70 PSI. 1.77 G/L SOLUTION.

SIZE DISTRIBUTION SHAPING

EC. 5800V. FLOWRATE 6 L/MIN. DILUTION FLOW 310 L/MIN.

REMARKS

NONE

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS DURING EXPERIMENT

HOOM CONDITIONS

OUTSIDE CONDITIONS

	EMP AVG RH AVG PRESS AVG TEMP AVG RH AVG PRESS AVG WS AVG WC) (%) (%) (MB) (C) (%) (MB) (M/S) (DEG) 	0 21.0 848	CO	AV6 RH (%) 21.•	AVG PRESS (MB) 	AVG TEMP (C)	A V G R H (%)	AVG PRESS (MB)	AVG HS (M/S)	AVG WD AVG WD (DEG)
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80/10/13

EXPER	21 MENT	#1]9						NUCLET	I MEASURI	EMENTS				PROCESS	ING DATE:	81/09/26
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ESSING DATE1 81/09/26	REMARKS					WATER IN ROYCO		
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EXPER	М∧СН N0.	18	RIIN:			812 22 22 22 22 22 22 22 22 22 22 22 22 2	RUN 00000	RUN:



EXPERIMENT # 20

PURPUSE

INSTRUMENT CUMPARISON WITH AEROSOL SIZE SUITABLE FOR CCN AND HAZE CHAMBERS.

DESCRIPTION OF EXPERIMENT

TIME START END 1352 1545 DATE

13 OCT+1980

NUCLEI TYPE

(NH4) 2 S04, PONODISPERSE, D=0.08 UM.

GENERATION METHOD

DRI ATOMIZER. 70 PSI. 1.77 G/L SOLUTION.

SIZE DISTRIBUTION SHAPING

EC. 222V. FLOWRATE 4.4 L/MIN.. DILUTION FLOW 370 L/MIN.

REMARKS

NONE

WEATHER SYNOPSIS

NONE

	SNOL	S AVG WS AVG WD (M/S) (DEG)
	IDE CONDIT	AVG PRES (MB)
	OUTS	AVG RH (%)
[MENT		AVG TEMP (C)
S DURING EXPER	NS	A VG PKE SS A VG P
	M CONDITIO	А VG RH (%) - (%) 21. ¢
MEAN WEATHER	ноо	AVG TEMP (C) 29.0

80/10/13

PROCESSING DATE: 81/09/26	OBS FST	L REMARKS	498088888888888888888888888888888888888		562 705		19 19 19 19 19 19 19 19 19 19 19 19 19 1				940.7 1009.8 1007.4 945 963.3				6 8 8 9 6 9 6 0 6 0 6 0 6 0 6 0 6 0 6 0 6 0 6 0 6 0		
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	00000000000000000000000000000000000000
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81/09/26			5 6 7 7 1 5												
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haze chamber results

EXPERIMENT # 21

PURPOSE

TIME VAHIATION TEST TO DETERMINE REPRODUCABILITY OF INSTRUMENT.

TIME Start End SIZE DISTRIBUTION SHAPING 1545 1726 DESCRIPTION OF EXPERIMENT GENERATION METHOD WEATHER SYNOPSIS DATE SAME AS 20 13 OCT 1980 SAME AS 20 SAME AS 20 NUCLEI TYPE REMARKS NONE

NONE

AVG WD (DEG) AVG WS (M/S) ******************************* OUTSIDE CONDITIONS AVG PRESS (MH) AVG RH (%) AVG TEMP (C) MEAN WEATHER CONDITIONS DURING EXPERIMENT AVG PHESS (MB) 848 ROOM CONDITIONS AVG RH (%) 21.0 AVG TEMP (C) 28.0

80/10/13

81/09/26	 								. = 64.23 1UND /EL=100CM-3		
SING DATE:		REMARKS							STD. DEV. Count Arc Noise Lev		
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PROCESSING DATE: 81/09/26	JCLEI OBS EST REMARKS Junt - K Remarks - N	910.2 910.2 922.1 922.1 928.4 905.7 905.7 905.6 STD. DEV. = 26.5	902.7 905.6 Stör bev. ************************************	1197.9 38 POINTS (1 POINT PER MINUTE) STD. DEV. = 79.9	933.9 <u>40</u> POINTS (1 POINT
	SUPER NU SAT CO	1000000	ດຈີດດຈີດຈີດຈີດຈີດຈີດຈີດຈີດຈີດຈີດ ລົດດີຈີຈີ ດີດດີດຈີດ ຈີດດີດລວດດີດດີດ ແພບເຫຍແບບເບເບເບເບເບເບເບເບເບເບເບເບເບເບເບເບເບເບ	I	ų
MENTS	VOLUME SAMPLED (L)	******	•••••••••••••••••••••••••••••••••••••••		
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12	OUSERVER	معمعهم	~	SSS	Va
T #: 2	T N N			AY6 AY6	4 4 6
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SSING DATE: 01/09/26		REMARKS	CC. AVERAGING OVER I MINUTE SIZE GOOD DROP SIZE PLATEAUS ANTW FLOW ABOUT 1300 COUNTS ROOM TEMP 21.75 C.		AVERAGING OVER 1 MINUTE.		POOR DROP SIZE PLAT. MAIN FLOW HAY Bain Countign at 1310 Countign at Reading Royco Channel 2. May Be Apply 1. May Sive Backgroundes Corre Ction at This	•	AVERAGING OVER ONE MINUTE.		STAND. DEV 9 STAND. DEV 9 STAND. DEV 9 STAND. DEV 1 STAND. DEV 8 STAND. DEV 8 STAND. DEV 8		STAND DEV.210 Stand Dev.21				
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		SUPER SAT (%)	-114 -069 -068		.173 .089 .089		.173 .114 .089		.173 .089 .068		.173 .114 .089		.173 .114 .089		.173 .114 .089		173 114 089		.173 .114 089 068		.173 .114 .089 .068		113 114
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EXPER		NOCH NOCH	1222	RUN:	5555	RUN:	3335	RUN:	5555	RUN:	2222	RUNS	2222	RUN:	2122	RUN:	2122	RUN:	2122	RUN:	2125	RUN:	21

PROCESSING DATE: 81/09/26		ESI REMARKS	9 55 9 F F F F F F F F F F F F F F F F F																				
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		COUNT	26.6		45.388 5.085 5.095 5.095 5.095 5.095 5.095 5.005		185 46.4 54.9		197.4 51.2 29.6 7.56		504 504 504 504 504 504 504 504 504 504		187.1 26.7 25.7 5.6		0.05 0.05 0.05 0.05 0.05 0.05 0.05 0.05		186.5 51.9 25.8 6.7		189 48.4 56.8		187.5 52.55 28.45 77.3		C 701
		SAT R	689 699 699		173 089 068		.173 .114 .089		173 114 089		173 114 069 068		.173 .114 .089		173 114 089 068		173 114 068 068		173 114 089		.173 .114 089		
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		START	1629 1629		1631 1631 1631		4444 6000 1111		1637 1637 1637 1637		1641 1641 1641		4444		1650 1650 1650		16555 65555 165555 16555 1655555 165555 165555 165555 165555 1655555 1655555 1655555 1655555 1655555 1655555 16555555 1655555 16555555 16555555 1655555555		1658 1658 1658 1658		1701 1701 1701 1701		1701
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1/09/26											• • 5 • • 5		5 • •		
SING DATE: 8		REMARKS									STD. DEV.=24 VOL SAMP X1 INST. CALIR		STD. DEV.=3 VOL SAMP X1 INST CALIB.		+/- 4.5 +/- 1.0
PROCES	551														
	200	<u>g</u> ız	F F G T T								1856		317		15.6 3.3
		COUNT	6.3		198.3 53.6 27.4		4 189 7 26 7 4		1 247 5747 500 100 100 100 100 100 100 100 100 100		1856		317		15.6 3.3
	02013	SAT (#)	.068		113 089 068		173 173 089 089		173 1194 069 068		1.1		.26		• 055
EMENTS		SAMPLED (L)									5.3		10.6		3.4 1.9
MEASURE		DELTA	0.00		0000 0000		0000 0000		0000 0000		5.10		2.60		00 00 00
NUCLEI	ATE TEMP	ξĉ	25.00		00000 00000 00000 00000 00000		00000 00000 00000 00000 00000		00000 00000 00000 00000 00000		26.70		26.90		26.00 26.00
	2	COLD	25.00		255.00 25		00000 52200 52200 52500 52000 52000 52000 52000 5000000		00000 2000 5000 5000 5000 5000 5000 500		21.60		24.30		26.00
	1111		1705		1708 1708 1708 1708		1712		1715 1715 1715						1626 1751
	3 10114 3	START	1704		1707 1707 1707				171 171 4171 4171						1546 1627
	EHVER	Ņ									000		<u>നനന</u>		99
#: 21	OBSI	NAME	AL OF S		AL UFS AL UFS AL UFS AL UFS		AL OF S AL OF S AL OF S AL OF S		ALOFS ALOFS ALOFS ALOFS		HORYS Horys Horys		HORYS RORYS HORYS		HINDMAN HINDMAN
2 I ME N T			21	23	2222	54	7222	25	2222	-	ມີ ທີ່ມີ ທີ່ມີ	2	ນເວມ ດີເວັນ	1	26
EXPEI	10.00		21	RUN:	2222	RUN:	2222	RUN:	2222	RUNI	ດເດເດ ທັກກາ	RUNS	ເຊິ່ງ ເຊິ່ງ	RUN:	26 26



EXPERIMENT # 22

PURPOSE

INSTRUMENT COMPARISON WITH PULYDISPERSE AEROSOL. MEDIUM CONC.

DESCRIPTION OF EXPERIMENT

NUCLEI TYPE

(NH4)2 S04

GENERATION METHOD

DRI ATOMIZER, 30 PSI, 0.065 G/L SOLUTION, MIGM DILUTION.

SIZE DISTRIBUTION SHAPING

NONE

REMARKS

1020-1040 DHIFT UP 10%, 1040 DROP APPROX. 20%, 1052 UROP APPROX. 20%.

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS DURING EXPERIMENT

	AVG WD (DEG)	
	AVG WS	
DE CONDITIONS	AVG PRESS (MB)	
00151	AVG RH (%)	
	AVG TEMP (C)	
5	 AVG PRESS (MB)	849
CONDITION	 AVG RM (*)	22.0
ROOM	VG TEMP	25.0

80/10/14

ATE: 81/09/26	IRKS													TEMPERATURE CTUALLY Frence Between Tand Abbient. 0%						
SING D	REMI	i												TON TON TUBER						
PROCES	EST									00-1										
	580 	2		222 222 222 222 222 222 222 222 222 22		490 490 90 90 90 90 90 90 90 90 90 90 90 90 9								00000000000000000000000000000000000000						
	NUCLE I COUNT							-40 4074 2040		80000000 9040400 9040400 9040400		001-14 9919-10				1650 540 210		1250 350 140		
	SUPER			17 23 17		.17 .23 .31				୰ ຎ ຑຓຑຎ໙ ∙©.4~~4© ~ • • • • • •		90044000 40.004 10.004		46414690 46614690 46614690		1.16 .59		1.09		
MENTS	VOLUME													~~~~~~~~ ©©©©©©©©©©©© ~~~~~~~~~~~~~~~~		~~~~		~~~		
MEASURE	DELTA			~~~~ • • • • • • • • • • • • • • • • •		2.40		0000 0430 049		940-000 		447740 970000 970000 970000				2000 2000 2000		000 000 000		
NUCLEI	ATE TEMP HOT			21.10 21.30 21.60		22.60		8000 8000 8000 8000 8000 8000 8000 800		00000000000000000000000000000000000000		0000000 7000000 7000000 70000000 800000000		00000000000000000000000000000000000000		27.80 24.00 24.00		27.80 27.90 25.80		
	0100 14			19.00 18.90 18.80		20.50 20.50 19.90		21.60 20.20 18.80 17.50		11000000000000000000000000000000000000		2000 2000 2000 2000 2000 2000 2000 200				22.30 25.10 25.10		22.50 24.60 25.80		
	TIME END			0011		1129 1133 1133		1029 1031 1031 1032		4000000 4000000 4000000				69426656 1111111111111111111111111111111111						
	SAMPLE			1057 1059 1102		1128 1130 1132		1030 1031 1031 1031		0000000 0000000 4000000 4000000 40000000				11111110 11111110 122493 12249 12049 12000000000000000000000000000000000000		1032 1035 1035		1049 1055 1058		
	NON CONTRACTOR	:		2000		200		2000		~~~~~~~~~~~		~~~~~~		*******		ഹഗവ		ທທານ		
	BSERVE			DAVE		DAVE DAVE DAVE														
#: 22	NAME			ROGERS. ROGERS.		ROGERS. ROGERS.		KITCHEN KITCHEN KITCHEN KITCHEN		NNNNN KKITTCHE NNN TCCHER NNT TCCHER NT					GAGIN Gagin Gagin		GAGIN Gagin Gagin			
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EXPER	MACH		RUN:	NNN	RUN:	กกก	RUN:	ມທາກ	RUNS	ບເບັນບາຍເບັນ	RUN:	ເດເດຍເດຍເດັ	RUN:	~~~~~~	RUNI	300	RUN:	ንውው	RUN:	

81/09/26									באטסטר ו							
SING DATE:	REMARKS									AFTER 1052		AFTER 1052				
PROCES	EST															
	SHO N					390.7 558.7 844.2 948.2 1009				313.9 478 608 714.9 747.9				N 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		154
	NUCLEI	10		1200 360 120		40 00 00 00 00 00 00 00 00 00		1 1 00 000 000 000 000 000 000 000 000	940	00-001-00000 20-0010000 20-0000000 20-0000000000		00000000000000000000000000000000000000		22 28 29 29 29 29 29 29 29 29 29 29 29 20 20 20 20 20 20 20 20 20 20 20 20 20		154
	SUPER SAT (%)	.19		1.16 .53 .17		ຑຑຑຑຑຑຑຑຑຑຌຌຌ ຎຎຑຑ຺ຨ຺ຨໞ຺ຩ ຺຺຺຺຺຺຺		МЩФФФФФ , , ,	r•1	ທະຫາກອາຊາດ ທີ່ດີດີດີດີ່ • •≁∽ • • • • • • • • • • • • • • • • • • •		~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~		 		.15
MENTS	VOLUME SAMPLED (L)	1.2		~~~~		*********	r	********	•			**********				
MEASURE	DELTA	2.30		0000 00 00 00 00 00 00 00 00 00 00 00 0		00401		4000440000 000004400 000004440	00.00	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~		4M90109M04M 0.0-1::::-4000 0.0-1:::-4000 0.0-1:::-4000 0.0-1::				
NUCLEI	ATE TEMP HOT (C)	28.10		27.80 27.80 27.40 27.40		~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~			14.00	L-LULWIN NU44UUN 4444444 NUNUNUNN		-1-00000084 000000000 444440000000 00000000				
	0100 61	25.80		20000 400 200 200 200 200 200 200 200 20		0054 48100 0854 48100000000000000000000000000000000000		00000000000000000000000000000000000000		6-12-20 		~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~				
	TIME													000000 000000 100000000000000000000000		1125
	SAMPLE START	1112		1116 1118 1121 1126		49090000000000000000000000000000000000		111111111 000000000 000000000000000000	1401	901009849984984 11111111111 11111111111111111		40424000000000000000000000000000000000		0001100 000000 00000000000000000000000		1105
	R NON	19 1		លលាល					-	 		<u></u>		888888 NNNNN		28
*: 22	OBSERVE NAME	GAGIN		GAGIN Gagin Gagin Gagin		**************************************		222222222 	LALA	72777777777777777777777777777777777777		**************************************		F1726EHAL0 F1726EHAL0 F1726EHAL0 F1726EHAL0 F1726EHAL0		FITZGEHALD
IMENT	DUCT NO.	6	4	ውውውው	-		~	000000000	3 10	0000000000	4	0000000000	I		~	11
EXPER	MACH	10	RUN:	ውውውው	RUN:		RUNI		LUN:	000000000000000000000000000000000000000	RUN:		RUNS		RUN:	11

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PROCESSING DATE: 81/09/26	EST REMARKS								BACKGROUND NOW KNOWN AS FCN OF SS* TARTED AEASURENTS STARTED AT 100 AT 900 AT 164EST PLAT P00R AT 164EST SS P01NT, AVERAGING
	OBS -	212					ň		
	NUCLEI				1	4074794000 4074794000 6484097999000 00000000000000000000000000000	1350	001 40 00 000 40 00 000 00 0000000 00000000	1035 909 972 972 1523 572
	SUPER SAT	4404 6600 100		 		, , , , , , , , , , , , , , , , , , ,		~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	
MENTS	VOLUME SAMPLED								
I MEASURE	DELTA			ຨຬຎ <i>ຠໟຬຓຩຩຎ</i> ຨຨຨຨຓຓຎຎຏ ຨຬຎຨຌຬຎຬຎຎຨຨຨຨຨຬ ຎຎຨຨຏຎຨຨຌຨຨຌຬ		04000000000000000000000000000000000000	າມ ເມີນ ເມີນ		000400 000400 000400
NUCLEI	LATE TFMF HOT (C)			20202020000000000000000000000000000000		00000000000000000000000000000000000000	255		200000 544450 544400 554400 5000 5000 50
	COLD	8 8 8 8 8 8		00000000000000000000000000000000000000		00000000000000000000000000000000000000	500		20.54 19.11 19.11 20.84 20.85
	T I ME	11255		00000000000000000000000000000000000000			1155	£££££\$\$\$\$\$\$600000 NNNNN AAAAAMMMM TTTTTTTTTTNNNNN TTTTTTTTTTTTTT	461111 4961112 11223
	SAMPLE START	11000		40200000000000000000000000000000000000			1155	00000000000000000000000000000000000000	600004 11110004 111111111111111111111111
	vo.			๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛		๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛	10101	ტ დდდდდდდდდდდდდ	33399999 990000
#: 22	NAME ORSERVE	FITZ6EHALD FITZ6EHALD FITZ6EHALD FITZ6EHALD FITZ6EHALD					AYERS		202222 222222 000000 11111 1000000 200000 200000 200000
9 I MENT	DUCT NO.		-		2			************	- -
EXPE	MACH NO.		RUNI	~~~~~~	RUN:	~~~~~~~	133 133	************	8 NUNUNUN NUNUNUN

OCESSING DATE: 81/09/26	ST REMARKS	OVER 4 PNTS EA SS. AVG LAST 2 PNTS. AVG LAST 2 PNTS. Raised Main Flo.						AFTER CONC CHANGE 1052								
æ	S = N							631.5 828.8 932.4 11339.6 1228.4		384.8 828.8 9328.8 11399.6 1228.4						
	NUCLEI	259 259 1018	2003 2003 2003 2003	0		80-480-4804 4800-4600 4800-4600		631 828 828 828 828 828 828 858 858 858 858		384.8 828.8 932.6 1139.6 1228.4		000000 440000 000000		4011 4001 4001 4001 4001 4001 4001 4001		178
	SUPER SAT (#)	1.04	•10 •10 •10			຺຺຺຺຺຺຺຺຺຺ ຺຺຺຺຺຺຺຺຺຺຺຺຺ Գ֎ՠԳ֎ՠԳ֎ՠ		401-M- N04- 401-M-		457-0-		440000- 50-000- -				.173
MENTS	VOLUME SAMPLED (L)		0000 0000 0000			000000000 000000000 0000000000 00000000										
MEASURE	DELTA	00-00 4/2221	0000 - 000 - 000	00500 00500 00500		400400400 666666666 800000000						000000 22022- 200000				00
NUCLEI	ATE TEMP HOT (C)	0225 0225 0225 00000	0000 •••• •000 •••• •000	NNNNNN 100000 100000		ດວທດວກດວກ ຍາຍຄາດວາດດາ ທານທານທານ						0235200 460000 0000000 0000000		000000 000000 000000 000000 000000 00000		25.00
	COLD PL	50 50 50 50 50 50 50 50 50 50 50 50 50 5	221.40 221.40	00000 04400 04400 00000		20000000000000000000000000000000000000						000000 900000 900000 900000 900000		-00000 0000-00 0000-00 000000 000000000		25.00
	END END	1143 1143 1143	1057			00000000000000000000000000000000000000		111111 200000 444000		1136 1142 1142 1151		00000000000000000000000000000000000000		PU00000		1105
	START	11138	1055	111111 5339 1112111 11121 11111 11111 11111 11111 11111 11111 1111		200000444 20000000000000000000000000000		10044 10054 10050 10052 10052		11134 11134 11134 11134 11134 1134 1134		40004 1.410-N 1.111-1 1.1111-1 1.1111-1 1.1111-1 1.1111-1 1.1111-1 1.1111-1 1.1111-1 1		1026 1031 1035 1039 1042 1042 1042 1042 1042 1042 1042 1042		1104
	0N	10000	იიი	ንወወወወው		ტტტტტტტტტ		~~~~~~		<u>n</u> unua 		ንውውውውው		000000 000000 000000		26
#: 22	0HSERVER											2000 2000 2000 2000 2000 2000 2000 200		TRUEBL 000 TRUEBL 000 TRUEBL 000 TRUEBL 000 TRUEBL 000 TRUEBL 000 TRUEBL 000		TRUEBLOOD .
2 I ME N T	DUCT NO.			000000	~	000000000 	1		~		1		l	~~~~~	<u>م</u>	12
FXPEF	MACH NO.		RUN: 16	<u></u>	RUN:	~~~~ <u>~</u> ~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	RUN:	~~~~~	RUN:	~~~~~	:NUA	****	RUN:		:NUA	21

EI 81/09/26		S			I ONS 10NS		MP X 10**5		00000000000000000000000000000000000000	:	H. NOMINAL
SING DAT		REMARK	8 8 9 1 1 1		CONCEN VARIAT VARIAT		VOL SA		4000 -00000 		
PROCES	•	EST	8 8 8 8				• 92				
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		COUNT	1407 - 0 1407 - 0 1407 - 0 1407 - 0 1407 - 0 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		00000000000000000000000000000000000000		212 339 835		11 2.10 2.12 2.12 2.12		
		SUPER SAT (%)	14689000000000000000000000000000000000000		4040000044 0		0446 0446				
MENTS		VOLUME SAMPLED (L)			********		21.22		*****		ດ ທີ່ມີ ທີ່ມີ
MEASURE	•	DELTA	000000040M0N0 00000004040 00000004040		-00000000 00000000 0000000000000000000		0000 4000 4000		00000 00000 •••••		
NUCLEI	ATE TEMP	101 (C)	ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ດີ ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທີ່ທ		0000000000 M42000000 M420000000 M000000000		23.90 23.90 23.90 23.90 23.90		00000 00000 00000 00000 00000		
	Ч	COLD	00000-00000 00000-00000 00000-00000 00000-00000 00000-00000 00000-00000 00000-00000 00000-00000 0000-00000 0000-00000 0000-00000 0000-00000 0000-0000 0000-0000 0000-0000 0000-0000 0000-0000 0000-0000 0000-0000 0000-0000 0000-0000 0000 0000-0000 0000 00000-0000 0000 00000 00000 00000 00000 00000 0000		000000000 04004000 04004000 000000000		21.50 21.50 20.40 19.00		00000 00000 00000 00000 00000 00000		
					400000000 000000000 000000000000000000		1107 1153 1208 1225		1216 1216 1216 1216	:	
		START			00000000000000000000000000000000000000		1011 1202 1223		1150 1150 1150 1150		
		. ON	00000000000000000000000000000000000000		ທິດທີ່ທີ່ທີ່ທີ່ທີ່ມີ ເປັນທີ່ທີ່ທີ່ທີ່ທີ່ເຫັນທີ່ທີ່ເປັນທີ່ທີ່ ເປັນທີ່ເປັນທີ່ເປັນທີ່ເປັນທີ່ເປັນທີ່ເປັນທີ່ເປັນທີ່ເປັນທີ່ເປັນທີ່ເປັນທີ່ເ		ጣጣጣጣ		୶୶୶୶୶	5	500
#: 22	OBSERVE	NAME	TRUE BL 000 TRUE B		SERPOLAY SEERPOLAY SEERPOLAY SEERPOLAY SEERPOLAY SEERPOLAY SEERPOLAY SEERPOLAY		80478 80478 80478 80478		NNNN NNNN NNNNN NNNNN NNNNN NNNNN NNNNN NNNN		OHTAKE OHTAKE
RIMENT	10.10	N0.		I	44444444 MMMMMMMM	-	ມ ບັນນັ້ນ ເ	I	00200 0000		
EXPE	10			RUNI		RUNS	ວເວເດເດ ບັບບັບບັ	RUN:	63336 55555	SUN:	~~~



EXPERIMENT # 23

PURPOSE

INSTRUMENT COMPARISON WITH LOW CUNCENTRATION POLYDISPERSE ARTIFICAL AFROSOL.

DESCRIPTION OF EXPERIMENT

DATE	TIME START	ÉND	
14 OCT+1980	1323	1516	

NUCLEI TYPE

(NH4)2 SO4+ POLYDISPEHSE.

GENERATION METHOD

DRI ATOMIZER. 40 PSI. HIGH DILUTION. 0.065 G/L SOLUTION.

SIZE DISTRIBUTION SHAPING

NONE

REMARKS

NONE

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS LURING EXPERIMENT

AVG WD (DEG) AVG WS (M/S) OUTSIDE CONDITIONS AVG PRESS (MA) AVG RH (*) AVG TEMP (C) AVG PHESS (Mb) ROOM CONDITIONS AVG RH (\$) AVG TEMP 25.0

849

21.0

80/10/14

EXPER	IMENT	#1 23					NUCLET	MEASURE	MENTS				PROCES	SING DATE: A1/09/26
NO CH	00.	UBSERVER	log I	STARLE	T I ME	COLD	ATE TEMP +01 (C)	DELTA	VOLUME SAMPLED (L)	SUPER SAT (%)	NUCLEI COUNT	SIZ	EST	HEMARKS
RUN:	-													
ŝ	NN	ROGERS, DAVE Rogers, Dave	22	1336 1336	1337 1340	22.20	24.40	2.10		.23		59 103		
RUN:	N													
NN	ŝ	POLITOVICH POLITOVICH	202	1412 1415	1413 1416	22.50	24.60	2.10		17 53.		53 106		
RUN:	1													
ഹാ	n un	KITCHEN KITCHEN	25	1323	1323	18.40	22.10	3.70 4.70		40 90	38			
ഹസം	ທເກເກ		000 777	1000	1326			000 		0 0 NI • • 0 •	900 900 900			
ŝ	n.	KITCHEN	12	1328	1328	14.90	22.20	3.30		.45	22			
BUN:	-													
~~	~ ~	LEALTCH	41	1322	1323		7.60			•176		29.7		HOT TEMPERATURE
	~~~		444, 	7900 7900 7900	14-01 14-01 14-01							58 39 9 9 9 9 9 9 9		DIFFERENCE BETWEEN TUBE AND AMBIENT.
~~	~~	LEAITCH	44	1406 1410	1407 1411					•164 •16		ດ ດາດເ 4 ດາ		INPUT HUMIDITY IS 70%
~~~~	~~~~	LEALTCH LEALTCH LEALTCH	444	444 000 000	1467		202 9 9 9 9			124		2.4		
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æ	8	MEE	17	1400		25.20	27.50	2.30		2	220			
c c	ac ac :	T ک التانیا ک	~~	000		24.80	27.50	2.70		(11) ••0 •	900 91 91 91			
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	~													
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RUN:	~													
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ውው	•00	GAGIN	ហហ	1355		25.50	00 4 4 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	00F 00F			100			
RUN:	m													
30	D D	GAGIN GAGIN	ഗഗ	1425		24.70	24.50 24.50	5.10 3.80	~~	1.01	900 900			
o o	00	GAGIN Gagin	ന്ന	05.41 05.41		25.60	28.50 28.60	2.20	1.2		400			
RUN:	1													
01	0	LALA	e i	1325		22.60	25.12	2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2	••	.25	105.9	6 2VI		
00	-0	LALA	ה היי	2251		22.04	50°-02	100 100 100	••		155.7	c•101		

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. 31 V SM1	ING DAIL!	REMARKS	0 6 6 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8									
		EST F	2 8 9 0									
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ROCESSING DATE: 81/09/26	EST Remarks			USING HIGHER MAIN FLOW USING THIS MORNINGS RACKGROUND AS FCN OF SS, MULT TIMES 2 LOWERING MAIN FLO.									
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80/10/14

PURPOSE

TEST HIGH PARTICLE CONCENTRATION EFFECTS.

DESCRIPTION OF EXPERIMENT

TIME STAHT END DATE

NUCLET TYPE

(NH4)2 SO4, POLYDISPEHSE, HIGH CONC.

GENERATION METHOD

DRI ATOMIZER. 50 PSI. 0.065 G/L SOLUTION.

SIZE DISTRIBUTION SHAPING

NONE

REMARKS

NONE

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS DURING EXPERIMENT

HOOM CONDITIONS

OUTSIDE CONDITIONS

AVG WD (DEG) AVG WS (M/S) AVG PRESS (MB) AVG RH (¥) AVG TEMP (C) AVG PRESS (MB) 648 AVG RH (%) 21.0 AVG TEMP (C) 24.0

SING DATE: 81/09/26	REMARKS		HOT TEMPERATURE IS Actually difference Between ture And Ambient Input Humidity IS 72-80 %					
PROCES	EST K							
	SHO	864 848	20-00000000000000000000000000000000000				1870.9 2982.5 4166.2 5117 5828.5	
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ING DATE: A1/09/26	REMARKS													F=4]
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PROCESSING DATE: 81/09/26	EST REMARKS			· 12/.62 REPLACED CAPILARY HIGH SS PLLARY HIGH MAIN FLOW. LOWERED MAIN FLOW. LOWERED MAIN FLOW. TO WERED MAIN FLOW.				
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#1 24	NAME							HUDSON
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EXPER	IMENT	#: 24					NUCLET	MEASURE	MENTS				PROCESS	ING DATE: F	81/09/26
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## NUCLEI MEASUREMENTS

PROCESSING DATE: 01/09/26

1 81/09/26			H. NOHINAL
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		SAMPLED	.73
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		START	1601
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B-138

PURPOSE

CLEAN AIR FOR NOISE CHECK IN INSTRUMENT.

DESCRIPTION OF EXPERIMENT

TIME START END 0820 0921 DATE 15 OCT+1980

NUCLEI TYPE

٩N

GENERATION METHOD

٩N

SIZE DISTRIBUTION SHAPING

٩N

REMARKS

NONE

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS DURING EXPERIMENT

OUTSIDE CONDITIONS AVG PRESS (MR) AVG RH (*) AVG TEMP AVG PRESS ROOM CONDITIONS AVG RH (%) AVG TEMP

148

22.0

27.0

AVG WD (DEG)

AVG WS (M/S)

80/10/15

EXPER	INENT	#: 25					NUCLEI	MEASURE	EMENTS				PROCESS	ING DATE:	81/09/26
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(PEH)	[ ME N ]	#: 25					NUCLEI	MEASURI	EMENTS				PROCES	SING DATE: B1/09/26
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B-142

42

PURPOSE

INSTRUMENT COMPAHISON WITH AMBIENT AIR.

DESCRIPTION OF EXPERIMENT

TIME START END	0 0936 1130	الد
DATE	15 OCT+198	NUCLEI TYP

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GENERATION METHOD

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SIZE DISTRIBUTION SHAPING

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REMARKS

AIR THROUGH OPEN WINDOW IN AEROSOL LAB DUCT ELBOWS IN LAB OPENED.

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS UURING EXPERIMENT

	 AVG WD (DEG)	
S	 AVG WS	
IDE CONDITION	AVG PRESS (MR)	
00151	 AVG RH (%)	
	AVG TEMP (C)	
45	AVG PRESS (MB)	847
CONDITION	AVG RH (%)	22.0
ROOM	AVG TEMP (C)	28.0

80/10/15

EXPE	RIMENT	#: 26					NUCLE	I MEASUR	EMENTS				PROCES!	SING DATE: 81/09/26
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B-149

PURPOSE

INSTRUMENT COMPARISON WITH INSOLUBLE, WETTABLE AEROSOL.

DESCRIPTION OF EXPERIMENT

TIME START END 1151 1415 DATE 15 OCT.1980 NUCLEI TYPE

AGI

GENERATION METHOD

THERMAL, N2 CARRIER GAS, HIGH DILUTION.

SIZE DISTRIBUTION SHAPING

A N

REMARKS

SLOW UPWARD URIFT OF TOTAL NUMBER CONC.

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS UURING EXPERIMENT

ROOM CONDITIONS

OUTSIDE CONDITIONS

AVG TEMP (C)	AVG RH (%)	AVG PHESS (MB)	AVG TEMP	AVG RH (*)	AVG PRESS (MR)	AVG WS	AVG WD
25.0	22.0	844					

80/10/15

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SING DATE: 81/09/26	REMARKS				MAIN FLOW ABOUT 2300 USING THIS MORNINGS USING THIS MORNINGS SMALLER CAPILLARY GOOD CAPILLARY SIZE PLAT GOING RAD. SIZE PLAT GOING RAD.		SECOND RUN AT LOWER AVER FLOW, ABT, 1800 AVERDENG LAST 4 COUNTS PER SS. USING SMALER CAPILLARY.						
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ROCESSING DATE: 81/09/26

REMARKS

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SDC/CFD RESULTS



PURPOSE

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INSOLUBLE MONODISPERSE AEROSOL TO COMPARE PLATEAUX.

DESCRIPTION OF EXPERIMENT

TIME STAKT END	1535 1642		
DATE	15 OCT+1980	NUCLEI TYPE	AGI

GENERATION METHOD

THERMAL, N2 CARRIER, TA BOAT.

SIZE DISTRIBUTION SHAPING

EC, 4357V, 6 L/MIN. AEROSOL FLOW.

HEMARKS

QUASIPERIODIC FLUCTUATIONS +/- 10%, APPROX, 7 MIN, PERIOD.

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS UURING EXPERIMENT

	 AVG WD (DEG)
S	AVG WS
IDE CONDITION	AVG PRESS (MB)
OUTS	AVG RH (%)
	AVG TEMP
4S	AVG PRESS (MB)
M CONDITIO	AVG RH (%)
ROO	AVG TEMP (C)

847

22.0

25.0

80/10/15

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11:0000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 10:000 11:0000 10:000 10:000 10:000 <t< td=""></t<>
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5 1623 22.60 27.00 4.40 1.2 76 450 5 1628 23.80 27.00 4.40 1.2 .76 450 23.80 27.20 2.10 1.2 .46 40 2.1.20 2.1 20 2.17 20

SING DATE1 81/09/26	REMARKS						MAIN FLOW ABT. 1820. Aerosol fluctuations	SECOND RUN WITH Migher Main Flow. About 2280.
PROCES	EST							
	S I N							
	NUCLE I COUNT	1000 000000000000000000000000000000000	-000440000 6040604001 604040000	0000000 00440004 00440044	000000 40000-0 000000	00324 0128 0128 058	801400 40000 640000 640000	9999 9999 9999
	SUPER SAT	1 1 1 4 1 4 1 0 0 0 0 0 0 0 0 0 0 0 0 0		11 23 23 24 24 24 24 24 24 24 24 24 24 24 24 24	11. 25 25 25 25 25 25 25 25 25 25 25 25 25			1.86 1.86 1.15
4ENTS	VOLUME SAMPLED (L)					000000 000000 000000 000000 000000 00000		
MEASUREN	DELTA		MW444AAA Punguashood 400004-000 400004-0000 400004-000	00444400 40000000 200000000000000000000	04-400 04-400		0-044000	6.70 5.33 33
NUCLEI	ATE TEMP HOT (C)	00.24000-500 00.2400-500 00.2400-500 00.20000000	111111100 111111100 1111111100 11111111	0000000000 0000000 00000000 0000000000	2222 2222 2222 2222 2222 2222 2222 2222 2222		33943	40. 440 541 540 540 540
	CULD PL	06909999999999999999999999999999999999	4034999901 409999990 409099999 4090999999 4090999999 40909999999 409099999999	200000000 40400000 4040000000 40400000000	2000 2000 2000 2000 2000 2000 2000 200		00000000000000000000000000000000000000	18.36 18.74 19.11
	TIME			~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	111111 1000000 442000 8460460	1607 1607 1607 1607	1111111111 1000000000 1000000000000000	1629 1633 1633
	SAMPLE START	10000000000000000000000000000000000000	00000000000000000000000000000000000000	ຸ ທຽນນານທານາ ພຍຍຜຸ∢∢ ∢ຽ ເວັອອວຼວຼ່ມທຸ∳ຍ	111111 100000 1440000 1440000	ປີ ແລະ ແລະ ແລະ ແລະ ແລະ ແລະ ແລະ ແລະ ແລະ ແລະ	1111111111 121111111111 12111111111111	1628 1632 1632
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CESSING DATE: 81/09/26	T REMARKS	FLOW CAPILLARY.							VOL SAMP X 10++5	
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	08S								2346 2346 231	-
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	DUCT NO.	155	-	222222 522222 5				4444	1 252 252 252	1 26
EXPER	MACH NO.	155	N N N N N N N N N N N N N N N N N N N	RUN 1666666 1666666	A NU 1881 1888 1888 1888 1888 1888 1888 1				RUN: 2000 :	RUN: 26

81/09/26			0 0 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1				NOMINAL
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	1310114	COUNT		0	80.90 21.1. 0.1.0 0.1.0	.024	2.94 2.94
	01010	SAT (%)	140	.016		.016	
		SAMPLED		1.4	~~~~	1.2	ານເບ ອີອີ
		DELTA		0.00		00.0	
	CATE TEME	101 (0)	0000	20.00	0000 0000 500 500 500 500 500 500 500 5	26.00	
	ã	COLD	2000 200 200 200	26.00	266.00 266.000 266.0000000000	26.00	
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	3 10114 3	START	2020	1602	1618 1618 1618 1618	1618	1544
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INENT		N0.	266	26 2	2000 2000	1 56	12
EXPER			002 002	26 RUN:	0000	26 RUN:	27

NUCLEI MEASUREMENTS

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80/10/15

PURPOSE

COMPARE INSTRUMENTS WITH HYDROPHOBIC AEROSOL.

DESCRIPTION OF EXPERIMENT

DATE TIME END

NUCLEI TYPE

PARAFFIN (HOUSEHOLD TYPE).

GENERATION METHOD

NZ JET INTO LIQUED PAHAFFIN SURFACE.

SIZE DISTRIBUTION SHAPING

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REMARKS

VERY BROAD DISTRIBUTION.

WEATHER SYNOPSIS

NONE

MEAN WEATHER CONDITIONS UURING EXPERIMENT

HOOM CONDITIONS

OUTSIDE CONDITIONS

AVG TEMP AVG RH AVG PRESS AVG WS AVG WD (C) (%) (MA) (M/S) (DEG)
AVG PHESS (MB)
AVG RH (%)
AVG TEMP (C)

EXPER	IMENT	#: 29					NUCI EI	MEASURI	EMENTS				PROCES	SING DATE: 81/09/26	
MACH NO.	DUJCT NO.	OBSERVER NAME	. 0 	SAMPLE START	TIME	COLD PL	ATE TEMP HOT (C)	DELTA	VOLUME Sampled (L)	SUPER SAT (1)	NUCLE I COUNT	S I N	EST KT	REMARKS	
RUN				6 6 8 8 8				1 9 1 1	6 9 9 9 9 9 9 9		8	6 8 9 9	3 8 8 8		:
N :N	NN	POL I TUVI CH	200	1731	1732	23.50	25.70 25.80	2.50		.17		169 235			
RUN: 7777	-	LEEALTCCH LEEALTCCH LTCCH LTCCH	4444 	1656 1706 1726	1657 1707 1717 1727		9999 •••99 ••050 ••050		 9888 9888 9888 9889 9889 9889 988	000300 M444 M		0404 0404			
2000 10 10 10	0000 -	GAAGIN Gaagin Agin Agin	លលាលល	1725 1725 1729		21.80 23.20 25.10 25.10	27.20 27.90 27.20	NWW9 10000	~~~~	1.17 .53 .17	0000				
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RUN:	1 13	AYERS AYERS	NN	1712	1719									NO PARTICLES IN RANGE 0.25 TO 1.25%	ھ
8 N 21111111 N 200000000	 ມາມານທານທານ	RADOGERARSS CONTRACTOR CONTR	നനനനനനന സസസസന്തന സസസസസന്തന	06 117711 09 117711 09 177711 09 177711 09 17771 09 17771 09 17771 09 17771 09 17771 09 17771 09 17771 09 17771 09 17771 09 17771 09 17771 09 17771 09 17771 09 177711 00 1777110 00 1777110 00 1777110 00 1777110 10 17771100000000	0951177711 11777711 1177771 1177777 1177777 1177777 117777 117777 117777 117777 117777 117777 117777 117777 117777 117777 117777 117777 117777 117777 117777 117777 1177777 1177777 1177777 1177777 1177777 11777777	00000000000000000000000000000000000000	00000000000000000000000000000000000000	₽₽₽₽₽₽₽₽₽₽₽₽₽₽₽₽₽₽₽₽₽₽₽₽ ₽₽₽₽₽₽₽₽₽₽₽₽			 			MAIN FLOW ABT 2280. May be seeing only Large way particles	• •
RUN: 18 18	1 18 18	NOSQUH NOSQUH	ውውው	1701 1709 1754	1703 1711 1758	23.20 21.85 23.60	26.45 27.80 25.85		.0018 .0018 .0018		1.52 1.76				
н С пилил С пилил С пилил	44444 MMMMA T	SERPOLAY SERPOLAY SERPOLAY SERPOLAY SERPOLAY SERPOLAY	ດເດເດເດ ທີ່ທີ່ຫຼື	1705 17705 17720 17750	1708 1715 1725 1732 1739	00000 50000 5000 5000 5000 5000 5000 5	10000 1000000	00000 1000000	*****	446 1987 10 10 10 10 10 10 10 10 10 10 10 10 10	00000 444				
RUN: 25	1 25	BORYS	n	1714	1716	16.00	22.90	6.90	4.24	2.15	1132	1132		VOL SAMP X 10**5	

81/09/26			,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,			Q		INAL	
SING DATE:		REMARKS	1 1 1 1 1 1 1 1 1		0.01/1	+/-0.006		75% 118% NOM	
PROCES	101	j Y							
	300	çı z	410		502 44 7	• 038 0			
		COUNT	410		203	• 038 0		~~	
		SAT (#)	1.08		• 15	016			
EMENTS	3011 107	SAMPLED	10.6		***			6.7	
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16.	ABSTRACT								
	Accurate knowledge of o	cloud condensat	tion nuclei size o	listributions i	is para-				
	mount to improving our under	rstandiny of cl	loud microphysical	phenomena. 1	The prob-				
	lems associated with obtain	ing this knowle	edge are large and	l varied; there	efore, many				
l I	technique development effort	ts to resolve t	these problems are	e in progress a	around the				
	globe. The objective of the Third International Cloud Condensation Nuclei Workshop								
	was to bring a large number of the development efforts together to insure an exchange								
	of results and progress being made in the hope that the attainment of satisfactory								
	solutions could be accelerat	ted. At this s	stage it appears a	as if the works	shop was				
	very successful.								
	The complexity and mag	nitude of the w	vorkshop were such	that it could	be success-				
	fully accomplished only unde	er the combined	d sponsorship of t	the National A	eronautics				
	and Space Administration and the National Science Foundation. The workshop was								
	supported by contract NAS8-33820; this document constitutes the final report under								
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