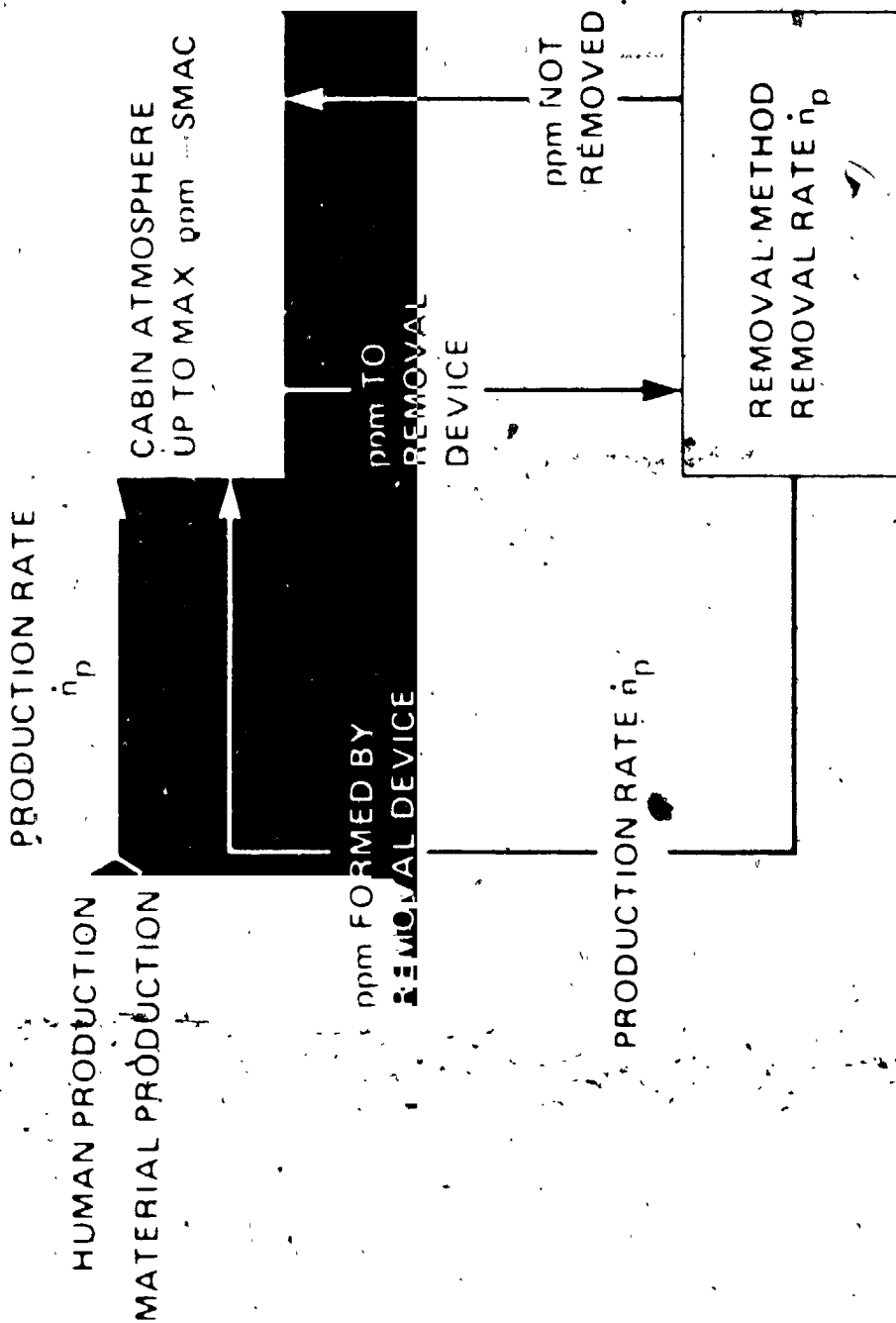


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INTRODUCTION

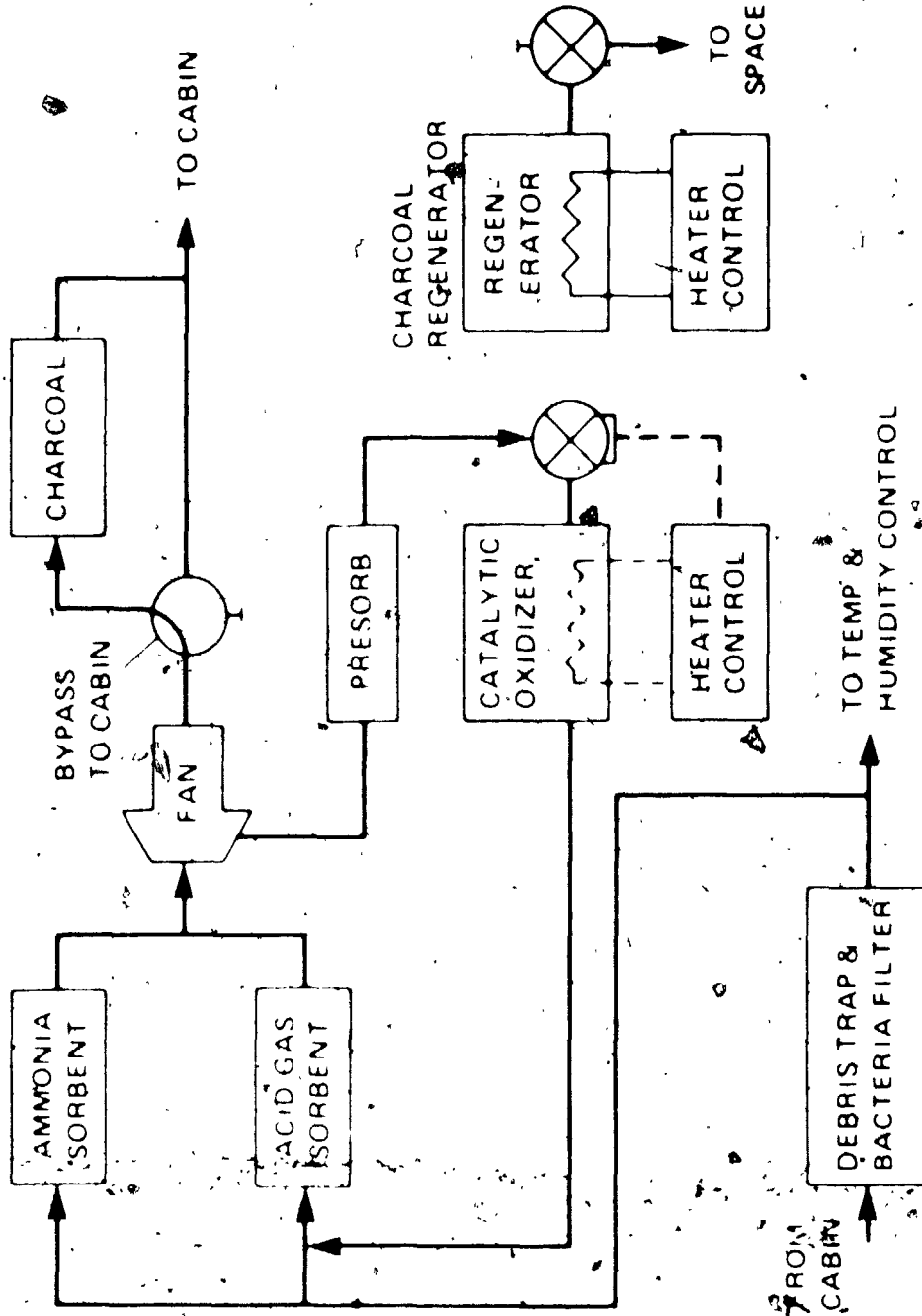
This talk summarizes some of the potential problems associated with acid gas sorbents, activated charcoal beds and the catalytic oxidizer proposed for spacecabin trace contaminant control.

CONTAMINANT FLOW CYCLE IN CLOSED ATMOSPHERIC SYSTEM



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TRACE CONTAMINANT CONTROL SYSTEM PROPOSED FOR A SPACE STATION



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REPRESENTATIVE SPACECABIN
CONTAMINANTS

Acetone	Formaldehyde
Acetaldehyde	Hydrogen
Acetylene	Hydrogen Chloride
Allyl Alcohol	Hydrogen Fluoride
Ammonia	Hexene-1
Amyl Alcohol	n-Hexane
Benzene	Hexamethylcyclotrisiloxane
n-Butane	Hydrogen Sulfide
Butene-1	Indole
cis-Butene-2	Isopropyl Alcohol
trans-Butene-2	Isobutyl Alcohol
n-Butyl Alcohol	Methylene Chloride
Butyraldehyde	Methyl Chloroform
Butyric Acid	Methyl Ethyl Ketone
Carbon Disulfide	Methyl Isopropyl Ketone
Carbon Monoxide	Methyl Alcohol
Chlorine	3-Methyl Pentane
Chloroacetone	Methane
Chlorobenzene	Monomethylhydrazine
Caprylic Acid	Methyl Mercaptan
Chloropropane	Nitric Oxide
Cyclohexane	Nitrogen Tetroxide
Cyclohexanol	Nitrous Oxide
Cyanamide	Propylene
1, 1-Dimethylcyclohexane	Isopentane
trans-1, 2-Dimethylcyclohexane	n-Pentane
2, 2-Dimethylbutane	Propane
1, 4-Dioxane	n-Propylacetate
Dimethylhydrazine	Propyl Mercaptan
Ethyl Alcohol	Phenol
Ethyl Acetate	Skatole
Ethylene Dichloride	Sulfur Dioxide
Ethylene	Toluene
Ethylene Glycol	Trichloroethylene
trans-1, 2-Dimethylcyclohexane	Tetrachloroethylene
Ethyl Sulfide	1, 1, 3-Trimethylcyclohexane
Ethyl Mercaptan	Tetrafluoroethylene
Freon 1	Freon 21
Freon 11	Valeric Acid
Freon 12	Vinyl Chloride
Freon 113	Vinylidene Chloride
Freon 114	m-Xylene
Freon 115	o-Xylene
Freon 117	p-Xylene

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BOILING POINTS OF
SPACECRAFT CONTAMINANTS

Component ⁽¹⁾	Normal Boiling Point, °C ⁽²⁾
Acetonitrile (C ₂ H ₃ N)	81.8
Benzene (C ₆ H ₆)	80.1
t-Butanol (C ₄ H ₁₀ O)	82.9
Cyclohexane (C ₆ H ₁₂)	80.7
1,2-Dichloroethane	83.7
o-Dichlorobenzene (C ₆ H ₄ Cl ₂)	179.0
Ethyl Acetate (C ₄ H ₈ O ₂)	-77.1
Freon 12	-30
Freon 113	48.2
Furan (Furfuran) (C ₄ H ₄ O)	32
Isopropanol (C ₃ H ₈ O)	82.5
Methyl Chloroform (1,1,1-Trichloroethane) (C ₂ H ₃ Cl ₃)	74.2
Methylethylketone (C ₄ H ₈ O)	79.6
Vinyl Chloride (C ₂ H ₃ Cl)	-13.8

(1) Green, B. D. and J. I. Strickland, 1974, Vol. 99, Third European Electro-Optics Conference (1974), 38-32

(2) CO₂ condenses (to solid) at -78.5°C at 1 atm

BASIC ADSORBENT BEDS AS ATMOSPHERIC CONTAMINANT REMOVAL DEVICES

- BASIC ADSORBENT BEDS HAVE BEEN SHOWN TO REMOVE --
CO₂, HCl, H₂S, Cl₂ AND SO₂ WITH SOME EFFECTIVENESS AND
TO BE INEFFECTIVE FOR NO₂, CH₃SH, AND CHF₃ (FREON 23)
- DATA ON ADSORPTION OF ALL CONTAMINANTS ON BASIC BEDS
IS INCOMPLETE WITH RESPECT TO EFFECT OF TEMPERATURE,
CO₂ CONC, HUMIDITY, CONTAMINANT CONCENTRATION, REACTION
RATES, AND BED CAPACITIES -- NO DATA IS AVAILABLE FOR
MANY CONTAMINANTS.

CONTAMINANT CONTROL BY ADSORPTION ON BASIC BEDS

LiOH, Li_2CO_3 , MnO_2 , Na_2CO_3 , CaCO_3

- PROBLEM OF DETERMINING WHAT SPECIES WILL BE ADSORBED AND WHAT CONDITIONS FAVOR ADSORPTION
- PROBLEM OF ADSORPTION RATE AND BED CAPACITY IN STOICHIOMETRIC GAS-SOLID REACTION

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IDENTIFIABLE RESEARCH AREA

INVESTIGATIONS SHOULD BE MADE OF

- THE RANGE OF CONTAMINANTS REMOVABLE BY A BASIC ADSORBENT BED.
- THE STABILITY OF REMOVAL EFFICIENCIES FOR HCl, HF, Cl₂ AND F₂ IN EXTENDED SERVICE.
- THE REASON FOR LOW AND MODERATE REMOVAL EFFICIENCY.
- TEMPERATURE AND HUMIDITY OPTIMIZATION AND THE COMPATIBILITY OF OPTIMIZED CONDITIONS WITH OTHER SUB SYSTEMS.

ATMOSPHERIC TRACE CONTAMINANT CONTROL BY CATALYTIC OXIDATION
PROBLEM OF MULTIPLE OXIDATION PRODUCTS

- INLET SPECIES
H₂O O₂ NH₃ RH RS RX RN N₂
- REACTOR AT 300-700 F
Pt Pd Al₂O₃ OR MnO₂ CuO AgO
- POSSIBLE RADICAL SPECIES IN BED
R H O X N S O O OH
- POSSIBLE EFFLUENT SPECIES
CO₂ H₂O R O R H HX NO N₂O NO₂ N₂O₄ N₂ SO₂ R'X X₂
- HIGH TEMPERATURE NEEDED FOR DIFFICULT TO OXIDIZE SPECIES
- SPECIES FRAGMENTS IN REACTOR CAN COMBINE TO FORM NEW COMPOUNDS
- H₂O + CO₂ ARE DESIRED PRODUCTS BUT CANNOT BE FORMED BY X N S COMPOUNDS
- NUMBER OF ACTIVE SITES FOR OXIDATION OF ANY PARTICULAR SPECIES LIMITED BY COMPETITIVE ADSORPTION OF OTHER SPECIES
- SUSTAINED CATALYST ACTIVITY DEPENDS ON MAINTENANCE OF PHYSICAL STRUCTURE AND CHEMICAL ACTIVITY AVOID SINTERING AND POISONING

CATALYTIC OXIDATION OF ATMOSPHERIC TRACE CONTAMINANTS

IDENTIFIABLE RESEARCH AREA

- EVALUATE THE DESIRABILITY AND EFFICIENCY OF CATALYTIC OXIDATION AS A TRACE CONTAMINANT CONTROL METHOD WHEN FEED GAS CONTAINS COMPOUND OF NITROGEN, SULFUR, AND THE HALOGENS
 - INVESTIGATE PRODUCT IDENTIFICATION AND MASS BALANCE WITH S, N, OR X COMPOUNDS IN FEED
 - INVESTIGATE TRANSIENT REDUCTIONS IN ACTIVITY OF CATALYST BED CAUSED BY S, N, OR X COMPOUNDS IN FEED
 - INVESTIGATE PERMANENT LOSS IN CATALYST ACTIVITY OR POISONING FROM S, N, OR X COMPOUNDS IN FEED

SPECIFIC PROBLEMS IN ATMOSPHERIC TRACE CONTAMINANT REMOVA
CONTAMINANT CONTROL BY ADSORPTION ON CHARCOAL BEDS

PROBLEM OF BLOCKING OF ADSORPTION OF LIGHTER
(MORE VOLATILE) SPECIES BY

- HEAVIER (LESS VOLATILE) CONTAMINANT SPECIES
- CONTAMINANT SPECIES REACTING OR CHEMISORBING
ON CHARCOAL
- WATER ADSORBED FROM HUMID INLET GAS STREAM

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ATMOSPHERIC TRACE CONTAMINANT CONTROL BY
ADSORPTION ON CHARCOAL BEDS

WORK TO DATE HAS SHOWN HUMIDITY OF INLET
GAS STREAM MAY:

- REDUCE ~~AD~~ ADSORPTION
- ENHANCE ADSORPTION
- PREFERENTIALLY BLOCK SOME SPECIES
- CHANGE OPTIMUM BED TEMPERATURE FOR
CONTAMINANT REMOVAL

IDENTIFIABLE RESEARCH AREA: HUMIDITY EFFECTS

INVESTIGATIONS ARE NEEDED TO:

- CONFIRM BLOCKAGE EFFECTS OF WATER VAPOR
- INVESTIGATE HUMIDITY EFFECTS IN ADSORPTION OF SO₂, DIMETHYL HYDRAZINE, MONO-METHYL HYDRAZINE, DIOXANE, CYANAMIDE, HCN, METHYL ETHYL KETONE, N₂O₄, ETHYLENE GLYCOL, ALLYL ALCOHOL AND OTHER POLAR CONTAMINANTS
- PROVIDE MORE RELIABLE GENERALIZED EXPRESSIONS FOR HUMIDITY EFFECTS TO AID DESIGN ANALYSIS FOR MISSION REQUIREMENTS

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NEED FOR FURTHER RESEARCH ON ATMOSPHERIC TRACE
CONTAMINANT CONTROL METHODS

- NO ONE METHOD CAN DO COMPLETE TASK
- IMPORTANT PROBLEMS REMAIN LARGELY UNSOLVED PREDICTION OF EFFECTS OF CONTAMINANT MIXTURES HUMIDITY VARIATION AND OFF DESIGN LOADS BED REGENERATION AND LONG TERM ACTIVITY
- MORE DEFINITIVE EXPERIMENTS ARE NEEDED TO PROVIDE THE DESIGN DATA NECESSARY FOR CONTAMINANT CONTROL SYSTEMS WITH
 - HIGHER RELIABILITY FACTOR
 - BETTER USE OF WEIGHT AND VOLUME AVAILABLE
 - LOWER POWER PENALTY
 - GREATER MARGIN FOR COMFORT AND HEALTH
- INTERFACE MUST BE MAINTAINED BETWEEN CONTROL METHODS RESEARCH AND MATERIALS AND PHYSIOLOGICAL RESEARCH AND DESIGN ENGINEERING