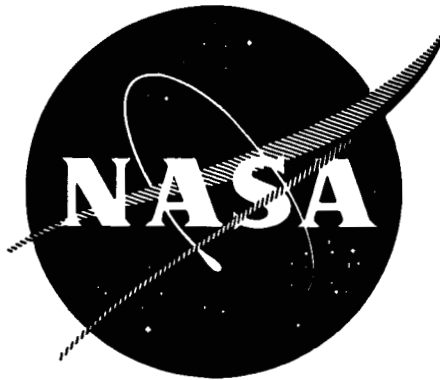




EVALUATION PROGRAM
for
SECONDARY SPACECRAFT CELLS

ACCEPTANCE TEST
OF
GULTON INDUSTRIES, INCORPORATED
12 & 20 AMPERE - HOUR ADHYDRODE CELLS

prepared for
GODDARD SPACE FLIGHT CENTER
CONTRACT W11,252B

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QUALITY EVALUATION LABORATORY
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QUALITY EVALUATION LABORATORY
UNITED STATES NAVAL AMMUNITION DEPOT
CRANE, INDIANA

EVALUATION PROGRAM
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QE/C 67-1

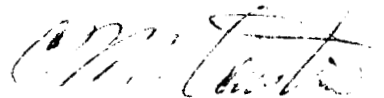
21 JANUARY 1967

PREPARED UNDER THE DIRECTION OF

E. C. Bruess

E. C. BRUESS
Manager, Electrochemical

APPROVED BY



C. M. AUSTIN
By direction

Enclosure (1)

REPORT BRIEF

GULTON 12 AND 20 AMPERE-HOUR ADHYDRODE NICKEL CADMIUM

SECONDARY SPACECRAFT CELLS

- Ref: (a) National Aeronautics and Space Administration Purchase Order Number W11,252B
(b) NASA ltr BRA/VBK/pad of 25 September 1961 w/BUWEPS first end FQ-1:WSK of 2 October 1961 to CO NAD Crane
(c) Preliminary Work Statement for Battery Evaluation Program of 25 August 1961

I. TEST ASSIGNMENT BRIEF.

A. In compliance with references (a) and (b), evaluation of Gulton Industries, Inc. 12 and 20 ampere-hour Adhydrode Secondary Spacecraft Cells was begun according to the program outline of reference (c).

B. The object of this evaluation program is to gather specific information concerning secondary spacecraft cells. Information concerning performance characteristics and limitations, including cycle life under various electrical and environmental conditions, will be of interest to power systems designers and users. Cell weaknesses, including causes of failure of present designs, will be of interest to suppliers as a guide to product improvement.

C. Twenty-two 12 ampere-hour and twenty 20 ampere-hour cells were purchased from Gulton Industries, Inc., Metuchen, New Jersey, by National Aeronautics and Space Administration (NASA). These cells are rated at 12 and 20 ampere-hours respectively by the manufacturer and include the adhydrode or absorption hydrogen type auxiliary electrode.

II. CONCLUSIONS.

A. From the results of this test, it can be concluded that:

1. The ceramic seals of these cells, manufactured by Gulton Industries, Inc. are satisfactory as evidenced by no leakers out of the 42 cells tested.

2. The capacities of the cells were in the acceptable range of 14.82 to 16.50 ampere-hours for the 12 ampere-hour cells and 23.30 to 28.70 ampere-hours for the 20 ampere-hour cells respectively.

III. RECOMMENDATIONS.

A. It is recommended that these Gulton Industries, Inc. 12 and 20 ampere-hour adhydrode cells be accepted on the basis of the acceptance test results.

RESULTS OF ACCEPTANCE TESTS

OF

12 AND 20 AMPERE-HOUR ADHYDRODE NICKEL CADMIUM

SECONDARY SPACECRAFT CELLS

MANUFACTURED BY

GULTON INDUSTRIES, INC.

I. INTRODUCTION.

A. On 18 July 1966, this activity began acceptance tests on twenty-two 12 ampere-hour and twenty 20 ampere-hour cells. These tests were completed on 17 August 1966.

II. TEST CONDITIONS.

A. All acceptance tests were performed at an ambient temperature between 23° C and 27° C at existing relative humidity and atmospheric pressure, and consisted of the following:

1. Phenolphthalein Leak Test.
2. Capacity Test.
3. Cell Short Test.
4. Immersion Seal Test.
5. Overcharge Test.
6. Internal Resistance Test of the Adhydrode.
7. Internal Resistance Test of the Cell.
8. Immersion Seal Test.

B. All charging and discharging was done at constant current (± 5 percent). Cells were charged in series but discharged individually.

III. CELL IDENTIFICATION AND DESCRIPTION.

A. Cells were identified by the manufacturer's serial numbers which were from 167 to 199, although not consecutively, for the 12.0 ampere-hour cells and from 121 to 140 consecutively for the 20.0 ampere-hour cells.

B. The 12.0 and 20.0 ampere-hour adhydrode cells are both rectangular in shape with average heights (base to top of positive terminal), lengths, widths and weights as follows:

Type	Height (Inches)	Length (Inches)	Width (Inches)	Weight (Grams)
12.0 A.H.	4.623	0.890	2.975	569.5
20.0 A.H.	7.120	0.888	2.986	925.6

The individual cell dimensions and weights are given in Tables I and II for the 12.0 and 20.0 ampere-hour cells respectively. Figure 1 is a photograph showing both the Gulton Industries, Inc. 12.0 and 20.0 ampere-hour adhydrode cells.

C. The cell containers or cans, and the cell covers are made of cold rolled steel. Both terminals on each battery type are insulated from the cell covers by ceramic seals and protrude through the cover as solder type terminals.

D. These cells, rated by the manufacturer at 12.0 and 20.0 ampere-hours respectively, were supplied in a discharged (with shorting wire) condition.

IV. TEST PROCEDURES AND RESULTS.

A. Phenolphthalein Leak Test.

1. The phenolphthalein leak test is a determination of the condition of the welds and ceramic seals on receipt of the cells. This test was performed with a phenolphthalein spray indicator solution of one-half of one percent concentration.

2. There were no signs of leakage on any of the 42 cells subjected to the leak test.

B. Capacity Test.

1. The capacity test is a determination of the cell capacity at the $c/2$ discharge rate, where c is the manufacturer's rated capacity, to a cutoff voltage of 1.00 volt per cell. The discharge was made after a 1-hour open circuit period following the 16-hour charge at the $c/10$ rate. A total of three capacity checks were made at this activity. The cells were discharged individually, but were recharged in series.

2. In order to gather data on the characteristics of the adhydrode, 51 ohms resistance was used between the adhydrode and the negative terminal for the first capacity check; 24 ohms was used for the second capacity check; and an open circuit of infinite resistance was used for the third capacity check.

3. Since complete capacity data, including adhydrode characteristics with the three resistance values, was not submitted by the manufacturer, it was not possible to compare the manufacturer's results with those of this activity. The 12 ampere-hour individual cell capacities ranged from 14.82 to 16.50 ampere-hours for an average of 15.78 ampere-hours. The 20 ampere-hour individual cell capacities ranged from 23.30 to 28.70 ampere-hours for an average of 26.20 ampere-hours. The cell capacities together with the adhydrode voltage characteristics for the 12 and 20 ampere-hour cells are tabulated in Tables III and IV respectively. Characteristic 2-hour rate discharge curves for the two types are shown in Figures 2 and 3.

C. Cell Short Test.

1. The cell short test is a means of detecting slight shorting conditions which may exist because of imperfections in the insulating materials, or damage to the element in handling or assembly.

2. Following completion of the third capacity discharge test, each individual cell was loaded with a resistor of value giving a c/1 to c/5 discharge rate and allowed to stand 16 hours with the resistor acting as a shorting device. At the end of 16 hours, the resistors were removed and the cells were placed on open circuit stand for 24 hours. Any cell whose voltage did not recover to 1.15 volts or higher was rejected.

3. The open circuit cell voltages, 24 hours after removal of the shorting resistors, ranged from 1.17 to 1.20 volts for an average of 1.19 volts on the 12 ampere-hour cells and from 1.20 to 1.21 volts for an average of 1.20 volts on the 20 ampere-hour cells.

4. There were no rejects of any of the cells subjected to the cell short test. The voltage values for the 42 accepted cells are shown in Tables III and IV.

D. Immersion Seal Test.

1. The immersion seal test is a means of detecting leakage of a seal or weld. The test was performed before and after the overcharge test sequence to determine the presence and cause of leaks.

2. The cells were placed under water in a bell jar container. A vacuum of 20 inches of mercury was held for 3 minutes. Cells discharging a steady stream of bubbles were considered rejects.

3. There were no rejects in the 42 cells subjected to the immersion seal test.

E. Overcharge Test.

1. The overcharge tests were performed to determine the steady state voltage at specific rates. The test specified a series of constant current charges at c/20, c/10 and c/5 rates, for a minimum of 48 hours at each charge rate or until the increase of the "on-charge" voltage was less than 10 millivolts per day.

2. The cells were monitored hourly throughout the test. Charging was to be discontinued on cells which exceeded 1.50 volts while on charge. There was no need to remove any cells from the charging sequence.

3. The steady state voltage of each cell at the end of each 48-hour charge rate test is shown in Tables III and IV. Characteristic overcharge voltage curves are shown in Figures 4 and 5.

F. Internal Resistance Test of the Adhydrode.

1. This test was performed to determine the internal resistance of the adhydrode.

2. During the c/10 charge rate portion of the overcharge test; the voltage drop across the 51 ohm resistor connecting the adhydrode to the negative terminal was measured. The 51 ohm resistor was then shunted with a one ohm resistor for 5 to 10 seconds and the voltage drop across the two parallel resistors (0.9808 ohms) was measured. The internal resistance of the adhydrode in ohms was calculated according to the following formula:

$$R = \frac{V1 - V2}{I2 - I1}$$

where V1 = voltage drop in volts across the 51 ohm resistor.

V2 = voltage drop in volts across the 0.9808 ohm resistor.

I1 = current flow in amperes through the 51 ohm resistor.

I2 = current flow in amperes through the 0.9808 ohm resistor.

3. The internal resistance value for the adhydrode of each cell is shown in Tables V and VI. The values range from 5.46 to 14.04 ohms and from 4.39 to 12.32 ohms for the 12 and 20 ampere-hour cells respectively.

G. Internal Resistance Test of the Cell.

1. This test was performed to determine the internal resistance of the cell.

2. At the completion of the overcharge test, the cells were returned to the $c/20$ charging rate and given a short pulse (5-10 seconds) at the rate of c in amperes. The cell voltages, V_1 , immediately prior to the pulse; and V_2 , 5 milliseconds after the pulse, were read on a suitable recording instrument. A CEC high speed oscillograph recorder (28.8 inches of tape per second) was used. The internal resistance of the cell in ohms was calculated according to the following formula:

$$R = \frac{V_2 - V_1}{I_c - I_c/20}$$

V_1 and V_2 are in volts, I_c and $I_c/20$ are in amperes.

3. The internal resistance value for each cell is shown in Tables V and VI. The values range from 0.88 to 3.51 milliohms and from 1.58 to 2.11 milliohms for the 12 and 20 ampere-hour cells respectively.

TABLE I

GULTON 12 AMPERE HOUR CELLS

<u>Cell Number</u>	<u>Height (Inches)</u>	<u>Length (Inches)</u>	<u>Width (Inches)</u>	<u>Weight (Grams)</u>
167	4.636	0.891	2.965	567.1
168	4.634	0.890	2.975	564.6
169	4.616	0.887	2.972	565.8
170	4.634	0.891	2.968	564.6
171	4.629	0.890	2.980	577.4
172	4.636	0.890	2.980	569.4
173	4.630	0.890	2.975	574.8
174	4.620	0.892	2.976	572.1
175	4.612	0.897	2.976	576.1
177	4.640	0.896	2.970	567.8
178	4.627	0.885	2.982	564.2
179	4.618	0.903	2.980	580.5
180	4.620	0.886	2.975	570.7
181	4.630	0.888	2.969	577.7
182	4.600	0.891	2.984	565.9
184	4.628	0.888	2.976	566.1
186	4.614	0.888	2.975	565.0
191	4.633	0.890	2.982	563.8
192	4.610	0.886	2.970	562.9
196	4.616	0.900	2.971	576.2
198	4.589	0.884	2.972	562.9
199	4.639	0.894	2.971	573.8

TABLE II

GULTON 20 AMPERE HOUR CELLS

<u>Cell Number</u>	<u>Height (Inches)</u>	<u>Length (Inches)</u>	<u>Width (Inches)</u>	<u>Weight (Grams)</u>
121	7.130	0.895	2.989	931.5
122	7.109	0.889	2.988	922.0
123	7.128	0.885	2.980	921.3
124	7.128	0.889	2.988	918.2
125	7.104	0.894	2.991	928.7
126	7.100	0.884	2.986	921.7
127	7.118	0.890	2.986	929.5
128	7.112	0.894	2.993	920.0
129	7.125	0.885	2.986	920.4
130	7.108	0.889	2.975	935.5
131	7.120	0.889	2.977	923.1
132	7.125	0.883	2.985	927.2
133	7.107	0.890	2.984	926.7
134	7.129	0.880	2.983	917.3
135	7.135	0.889	3.000	931.3
136	7.141	0.889	2.982	931.0
137	7.110	0.884	2.980	932.2
138	7.136	0.885	2.994	923.8
139	7.104	0.894	2.992	923.7
140	7.123	0.884	2.990	925.6

GULTON 12 A.H. 3RD ELECTRODE

TABLE III

CELL NUMBER	END OF CHARGE WITH 51 OHM RESISTOR		CAPACITY NO. 1	END OF CHARGE WITH 24 OHM RESISTOR		CAPACITY NO. 2	END OF CHARGE WITH NO RESISTOR		CELL SHORT TEST	c/20 OVERCHARGE CELL VOLTAGE		c/10 OVERCHARGE CELL VOLTAGE		c/5 OVERCHARGE CELL VOLTAGE	
	Volts	Amps		Volts	Amps		Volts	Amps		Volts	Amps	Volts	Amps	Volts	Amps
167	1.42	0.0032	14.28	1.46	0.0178	16.02	1.44	0	1.19	1.40	0.340	0.0067	1.42	0.390	0.0076
168	1.42	0.0001	14.10	1.46	0.0040	15.72	1.44	0	1.19	1.40	0.377	0.0074	1.41	0.539	0.0106
169	1.42	0.0041	13.62	1.46	0.0153	15.42	1.44	0	1.19	1.40	0.410	0.0080	1.41	0.560	0.0110
170	1.42	0.0006	13.20	1.46	0.0175	15.12	1.44	0	1.18	1.40	0.540	0.0106	1.41	0.701	0.0137
171	1.42	0.0004	12.42	1.46	0.0167	15.60	1.44	0	1.18	1.40	0.560	0.0110	1.41	0.769	0.0151
172	1.41	0.0013	12.00	1.47	0.0160	15.78	1.45	0	1.19	1.40	0.410	0.0080	1.42	0.549	0.0108
173	1.41	0.0004	12.60	1.46	0.0144	16.02	1.44	0	1.19	1.40	0.400	0.0078	1.41	0.690	0.0135
174	1.42	0.0013	12.78	1.46	0.0158	15.78	1.44	0	1.19	1.40	0.450	0.0088	1.41	0.612	0.0120
175	1.43	0.0031	13.92	1.46	0.0166	15.48	1.45	0	1.19	1.40	0.584	0.0115	1.42	0.741	0.0145
177	1.43	0.0023	13.80	1.46	0.0171	15.48	1.44	0	1.17	1.40	0.510	0.0100	1.41	0.671	0.0132
178	1.43	0.0018	13.92	1.45	0.0153	14.82	1.44	0	1.19	1.40	0.430	0.0084	1.42	0.561	0.0110
179	1.40	0.0051	15.78	1.44	0.0150	15.48	1.43	0	1.20	1.41	0.399	0.0078	1.43	0.666	0.0131
180	1.44	0.0056	15.78	1.44	0.0155	15.30	1.43	0	1.18	1.41	0.474	0.0093	1.43	0.616	0.0121
181	1.40	0.0043	15.78	1.44	0.0154	16.08	1.44	0	1.18	1.41	0.460	0.0090	1.42	0.567	0.0111
182	1.40	0.0057	15.60	1.45	0.0162	15.78	1.44	0	1.20	1.41	0.547	0.0107	1.42	0.717	0.0141
184	1.40	0.0051	15.90	1.44	0.0100	16.38	1.44	0	1.19	1.41	0.510	0.0100	1.42	0.656	0.0129
186	1.44	0.0059	15.78	1.44	0.0168	15.18	1.43	0	1.17	1.41	0.446	0.0087	1.42	0.651	0.0128
191	1.44	0.0068	15.60	1.44	0.0186	14.58	1.43	0	1.20	1.41	0.660	0.0129	1.42	0.793	0.0155

GULTON 12 A.H. 3RD ELECTRODE

TABLE III (contd)

CELL NUMBER	END OF CHARGE WITH 51 OHM RESISTOR Volts Amps	CAPACITY NO. 1	END OF CHARGE WITH 24 OHM RESISTOR Volts Amps	CAPACITY NO. 2	END OF CHARGE WITH NO RESISTOR Volts Amps	CELL SHORT TEST	c/20 OVERCHARGE CELL VOLTAGE	THIRD ELECTRODE Volts Amps	c/10 OVERCHARGE CELL VOLTAGE	THIRD ELECTRODE Volts Amps	c/5 OVERCHARGE CELL VOLTAGE	THIRD ELECTRODE Volts Amps
192	1.43 0.0057	15.90	1.45 0.0168	15.60	1.44 0	14.82 1.18	1.41	0.528 0.0104	1.42	0.733 0.0144	1.42	0.820 0.0161
196	1.40 0.0054	16.20	1.44 0.0148	15.72	1.44 0	14.52 1.19	1.41	0.395 0.0077	1.42	0.626 0.0123	1.41	0.797 0.0156
198	1.40 0.0047	16.02	1.44 0.0164	15.72	1.44 0	15.00 1.19	1.41	0.459 0.0090	1.43	0.623 0.0122	1.42	0.748 0.0147
199	1.40 0.0045	16.02	1.45 0.0138	16.50	1.45 0	15.72 1.20	1.41	0.340 0.0067	1.43	0.455 0.0089	1.43	0.699 0.0137

GULTON 20 A.H. 3RD ELECTRODE

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TABLE IV

CELL NUMBER	END OF CHARGE WITH 51 OHM RESISTOR		CAPACITY NO. 1	END OF CHARGE WITH 24 OHM RESISTOR		CAPACITY NO. 2	END OF CHARGE WITH NO RESISTOR		CELL SHORT TEST	c/20 OVERCHARGE		c/10 OVERCHARGE		c/5 OVERCHARGE	
	Volts	Amps		Volts	Amps		Volts	Amps		CELL VOLTAGE	THIRD ELECTRODE Volts	CELL VOLTAGE	THIRD ELECTRODE Volts	CELL VOLTAGE	THIRD ELECTRODE Volts
121	1.41	0.0050	24.20	1.44	0.0180	25.80	1.45	0	1.20	1.42	0.426	1.44	0.556	1.44	0.664
122	1.42	0.0063	24.00	1.44	0.0181	25.50	1.43	0	1.20	1.42	0.427	1.43	0.535	1.43	0.634
123	1.43	0.0073	23.80	1.44	0.0185	24.20	1.43	0	1.20	1.42	0.459	1.44	0.587	1.43	0.725
124	1.44	0.0084	23.30	1.44	0.0176	22.50	1.44	0	1.20	1.42	0.487	1.43	0.603	1.43	0.749
125	1.41	0.0048	24.00	1.45	0.0183	26.70	1.44	0	1.20	1.42	0.467	1.43	0.604	1.42	0.732
126	1.42	0.0070	24.00	1.44	0.0188	24.80	1.44	0	1.20	1.42	0.460	1.43	0.553	1.43	0.685
127	1.41	0.0050	24.20	1.45	0.0183	26.50	1.45	0	1.20	1.42	0.455	1.43	0.562	1.42	0.700
128	1.44	0.0088	23.30	1.44	0.0172	22.50	1.43	0	1.20	1.42	0.507	1.43	0.631	1.43	0.760
129	1.44	0.0079	23.30	1.43	0.0192	23.50	1.44	0	1.20	1.42	0.500	1.43	0.627	1.42	0.762
130	1.41	0.0053	24.20	1.44	0.0208	25.70	1.44	0	1.20	1.42	0.500	1.44	0.633	1.43	0.771
131	1.40	0.0051	23.80	1.45	0.0183	27.30	1.44	0	1.20	1.41	0.365	1.43	0.490	1.43	0.602
132	1.40	0.0050	24.00	1.45	0.0175	27.50	1.44	0	1.20	1.41	0.377	1.42	0.497	1.43	0.592
133	1.40	0.0051	24.00	1.45	0.0183	27.50	1.44	0	1.20	1.41	0.374	1.42	0.522	1.42	0.666
134	1.43	0.0062	24.00	1.45	0.179	26.50	1.44	0	1.20	1.41	0.351	1.42	0.476	1.42	0.599
135	1.40	0.0045	23.70	1.45	0.0183	27.70	1.44	0	1.21	1.42	0.410	1.43	0.539	1.42	0.626
136	1.40	0.0050	24.00	1.45	0.0183	27.30	1.44	0	1.20	1.41	0.369	1.42	0.505	1.42	0.630
137	1.40	0.0050	24.00	1.46	0.0167	28.70	1.45	0	1.20	1.41	0.348	1.42	0.507	1.42	0.656
138	1.40	0.0049	24.00	1.45	0.0175	27.80	1.44	0	1.20	1.41	0.370	1.42	0.507	1.42	0.634
139	1.40	0.0052	24.00	1.44	0.0188	27.30	1.44	0	1.20	1.41	0.446	1.42	0.569	1.42	0.674
140	1.43	0.0053	24.00	1.45	0.0170	27.20	1.44	0	1.20	1.42	0.378	1.43	0.505	1.43	0.603

TABLE V
GULTON 12 AMPERE HOUR CELLS

<u>Cell Number</u>	<u>Auxiliary Electrode Resistance (ohms)</u>	<u>Cell Resistance (Milliohms)</u>
167	10.05	1.75
168	5.46	1.75
169	6.51	1.75
170	6.53	1.75
171	5.86	1.75
172	7.18	1.75
173	10.02	2.63
174	4.69	2.63
175	5.95	2.63
177	6.34	1.75
178	6.89	2.63
179	10.02	1.75
180	6.43	1.75
181	5.99	1.75
182	6.77	0.88
184	6.61	2.63
186	13.29	2.63
191	7.75	2.63
192	14.04	3.51
196	10.30	1.75
198	6.86	2.63
199	6.76	1.75

TABLE VI
GULTON 20 AMPERE HOUR CELLS

<u>Cell Number</u>	<u>Auxiliary Electrode Resistance (ohms)</u>	<u>Cell Resistance (Milliohms)</u>
121	12.32	2.11
122	6.23	2.11
123	6.49	2.11
124	5.52	2.11
125	5.38	1.58
126	5.13	2.11
127	5.46	1.58
128	6.07	1.58
129	6.45	1.58
130	6.26	1.58
131	5.61	2.11
132	4.71	1.58
133	4.84	2.11
134	4.39	2.11
135	5.25	2.11
136	4.70	2.11
137	4.55	2.11
138	4.79	2.11
139	5.69	2.11
140	5.15	2.11

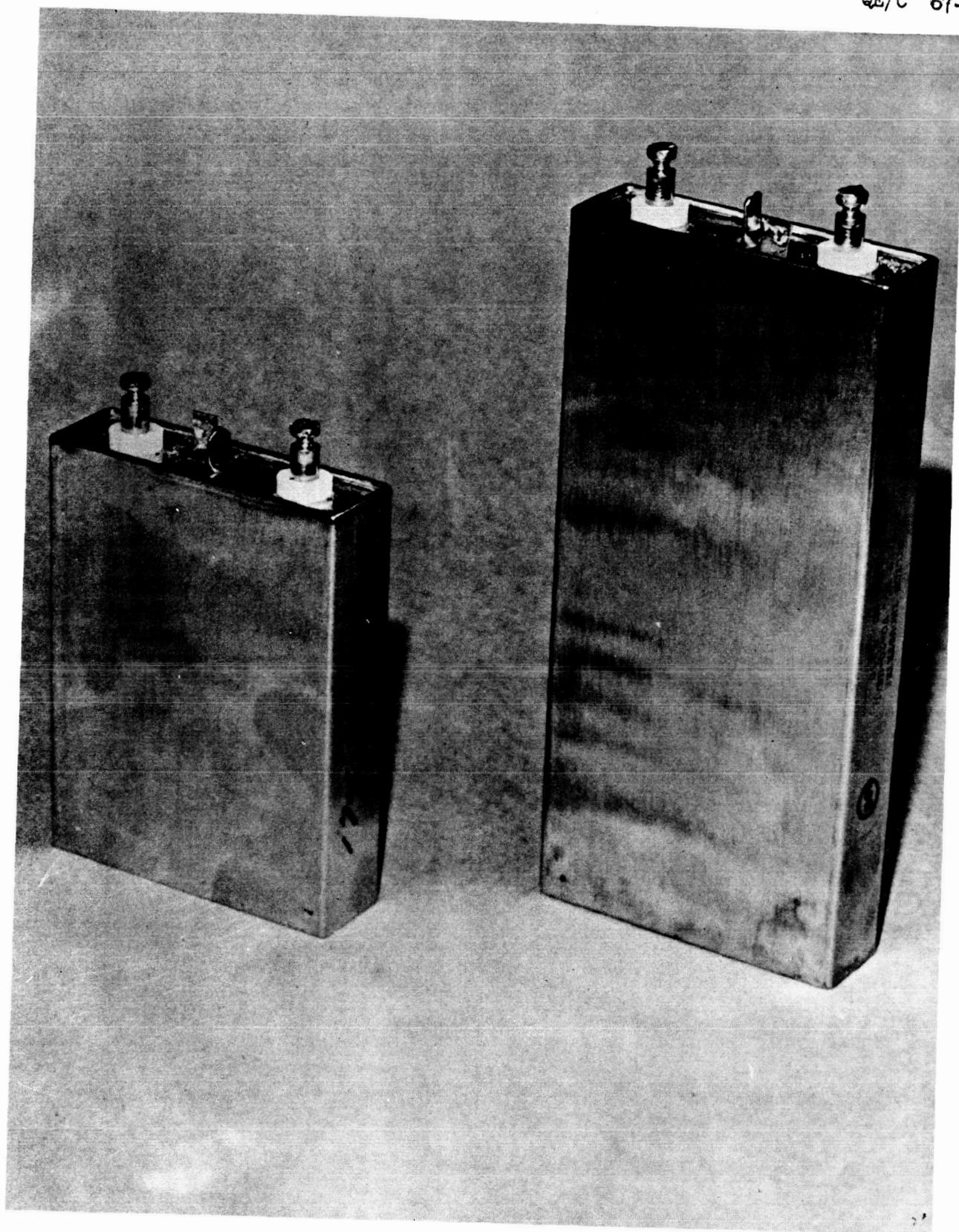
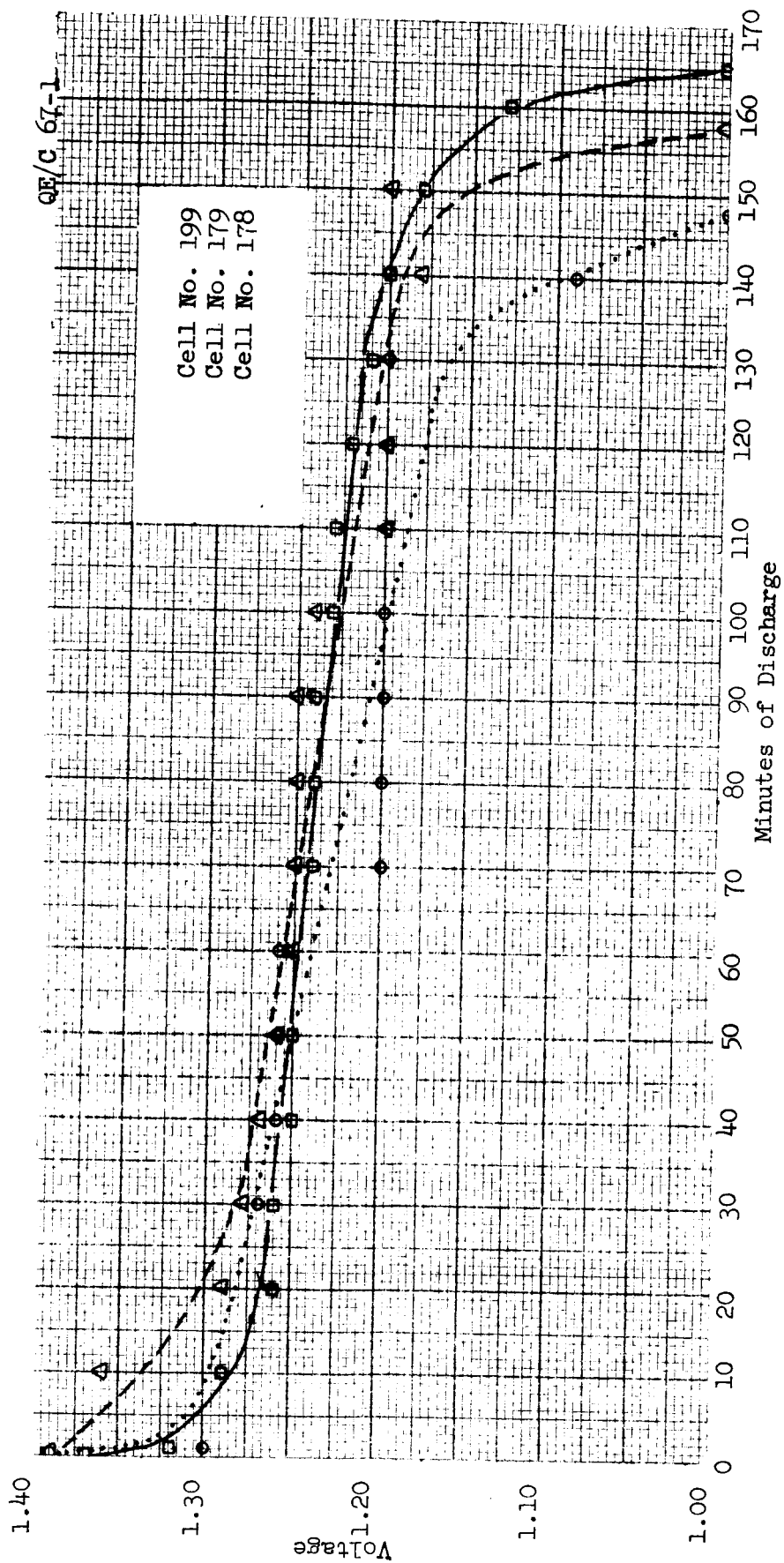


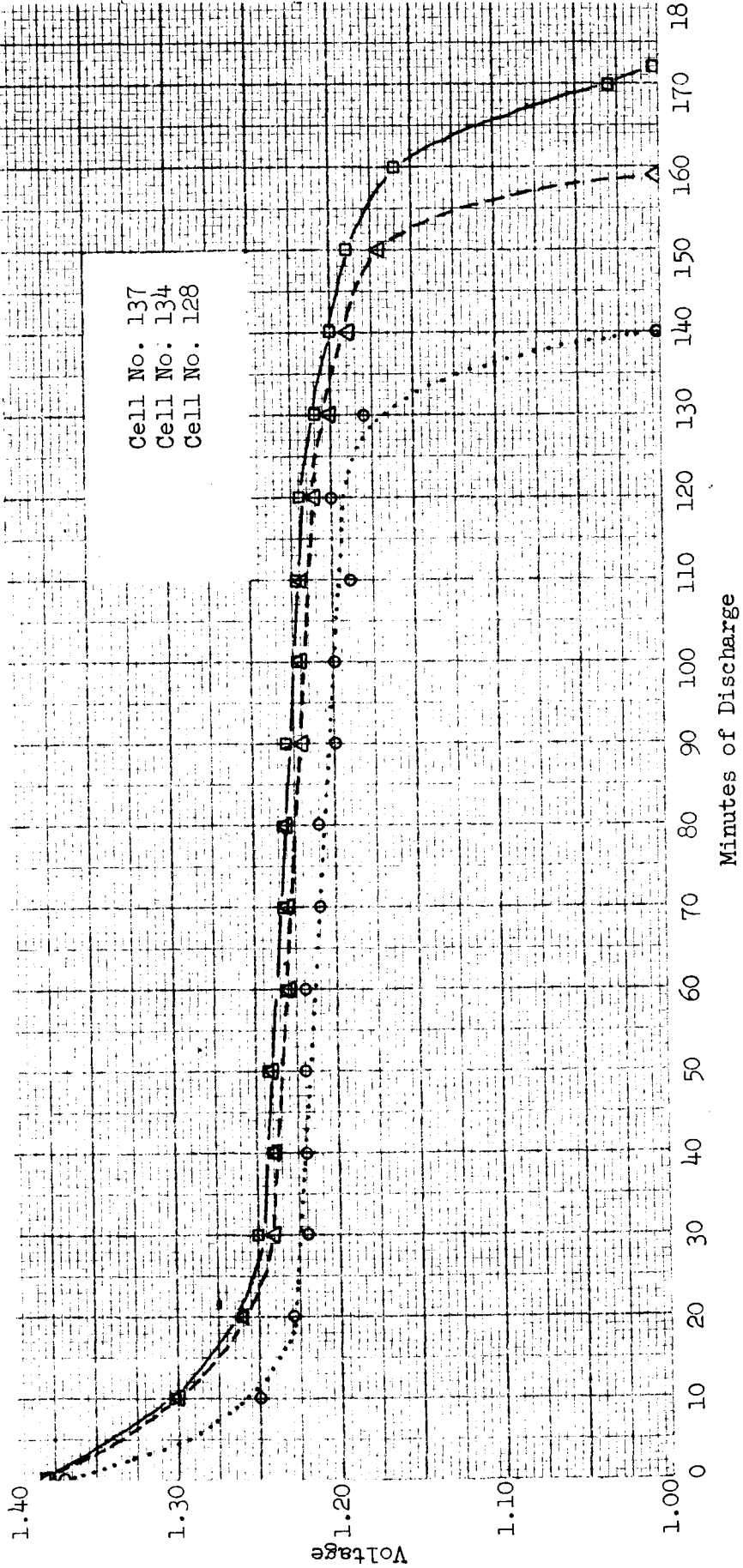
FIGURE 1



CHARACTERISTIC 2-HOUR RATE DISCHARGE

GULTON 12 AMPERE-HOUR ADHYDRODE NICKEL CADMIUM SEALED CELLS

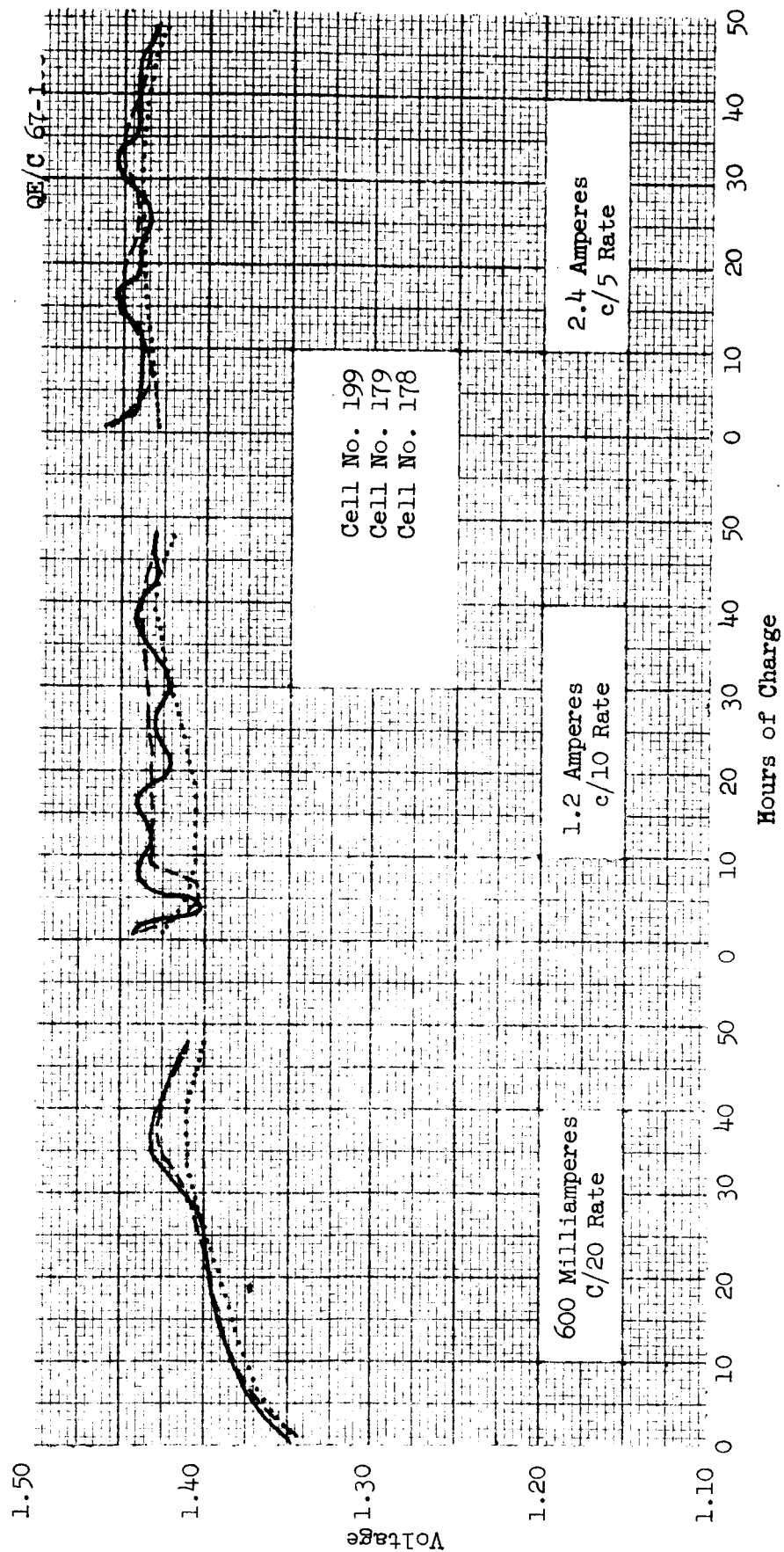
FIGURE 2



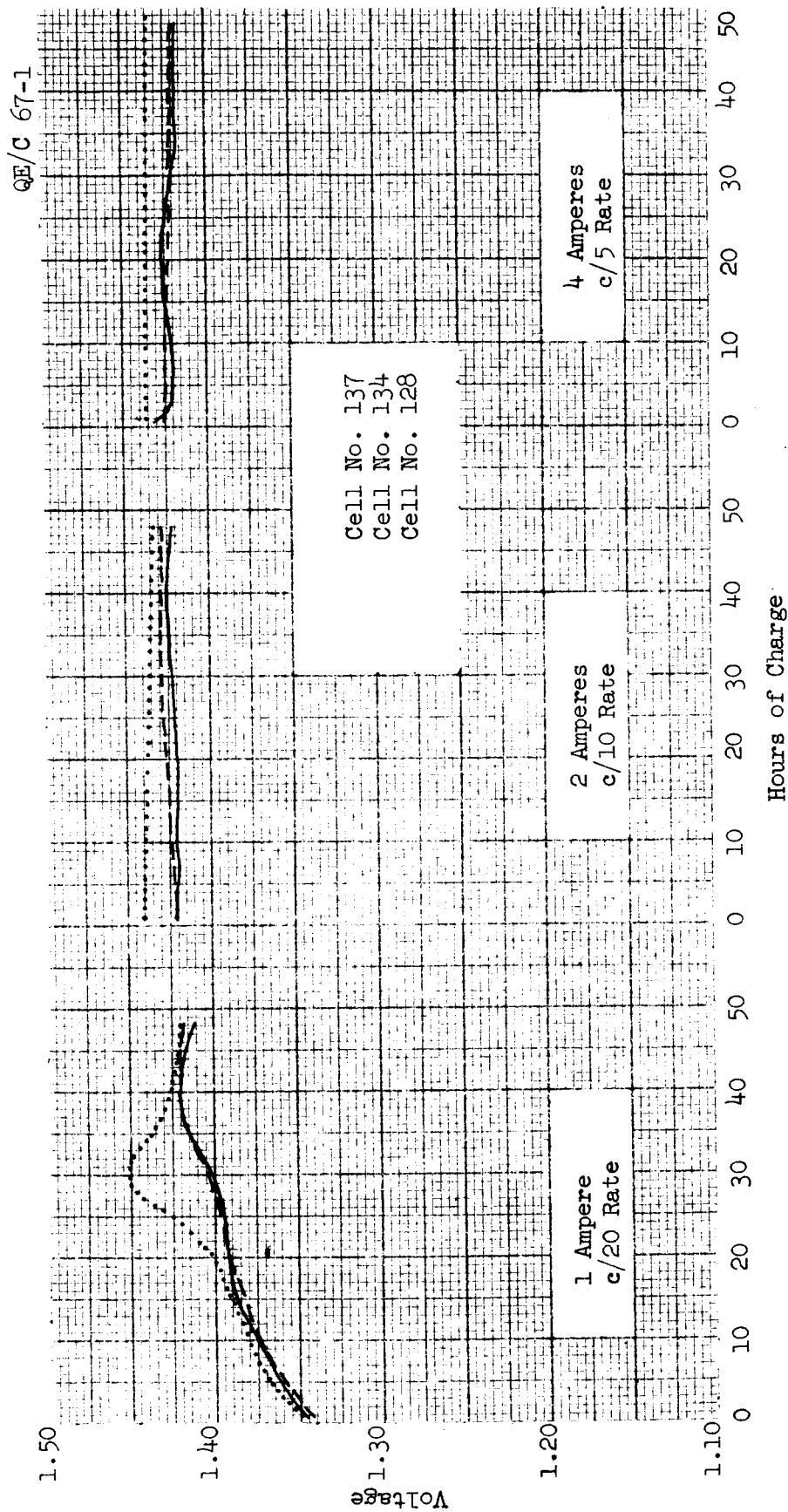
CHARACTERISTIC 2-HOUR RATE DISCHARGE

GULTON 20 AMPERE-HOUR ADHYDRODE NICKEL CADMIUM SEALED CELLS

FIGURE 3



CHARACTERISTIC 48-HOUR OVERCHARGE CURVES
GULTON 12 AMPERE-HOUR ADHYDRODE NICKEL CADMIUM SEALED CELLS



CHARACTERISTIC 48-HOUR OVERCHARGE CURVES

GULTON 20 AMPERE-HOUR ADHYDRODE NICKEL CADMIUM SEALED CELLS

FIGURE 5

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