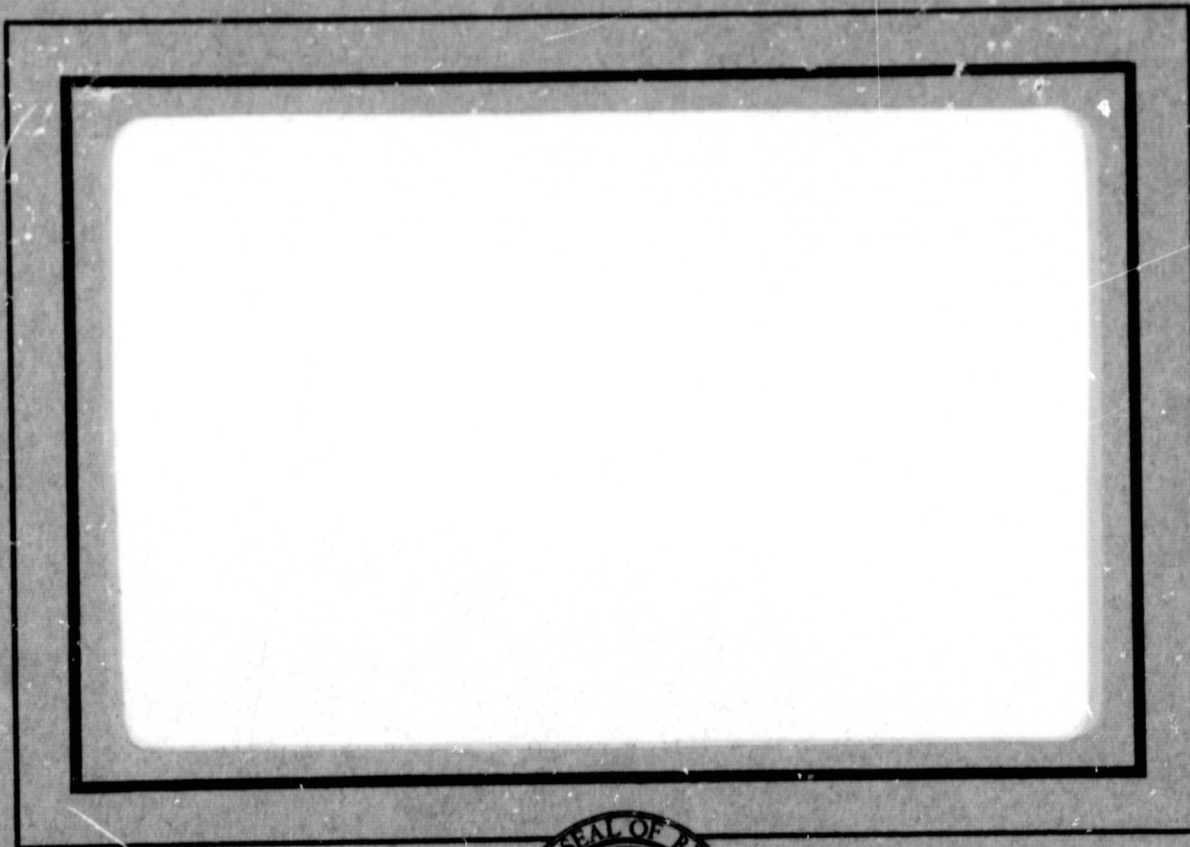


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FEASIBILITY STUDY
OF
LUNAR NEUTRON ALBEDO EXPERIMENT

Submitted to:

National Aeronautics and Space Administration
Washington, D. C. 20546

Submitted by:

Department of Space Science
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SUMMARY

Cosmic rays interacting with the surface nuclei of the moon produce neutrons, some of which emerge from the surface. The energy spectrum of this emergent albedo will depend strongly on the composition of the surface, especially upon concentrations of hydrogen. A measurement of the ratio of the slow to fast neutron counting rates will serve as an index of the hydrogen content in the lunar surface.

A feasibility study is made on the effect on the slow and fast neutron counting rates due to increases in water content. A lunar neutron simulator, consisting of neutron detectors and a fast neutron source, measures the effect of water concentration on the ratio using a basaltic material like that found on the moon by the Surveyors. A sand-water mixture was also used in order to demonstrate the effects of differing elemental compositions in the sample.

Data obtained using a 0.3% water content (by weight) in basalt indicate an increase of about 4% in the ratio over the 0.2% water content level. As expected from theory, the variation in ratio with water concentration is quite linear, for both samples tested. The only effect of variations in the elemental composition is to alter the slope of the line.

We conclude that the experiment as proposed is sufficiently sensitive to detect a variation in water concentration of 0.1% (by weight). Since the least concentration of water expected is $\sim 0.1\%$ (the amount found in stony meteorites) and the concentration may be $\sim$ the 1% found in volcanic rock, it appears that the lunar neutron albedo experiment will serve as a valuable tool in remotely prospecting the lunar surface for hydrogen matter such as water.

I. INTRODUCTION

A. Lunar Neutrons

Neutron production within the surface of the moon can be used as an indicator of the presence of hydrogen in the material of the lunar crust. The neutrons are produced by the interaction of cosmic rays and the surface material. The neutrons are scattered isotropically and some may escape the surface of the moon. This emergent flux is known as the lunar neutron albedo.

The production of neutrons with a large fast component (several Mev) in a medium of low neutron capture cross section creates a process of "thermalization". The fast neutrons lose energy due to collisions with the sub-surface nuclei until the neutrons have velocities characteristic of "slow" neutrons. Therefore, the lunar albedo consists of varying neutron energies dependent on the scattering nuclei of the material.

Elastic scattering has a strong dependence on the atomic weight of the scattering center, the loss being greatest for hydrogen. Consider the simple collision of a neutron with a nucleus in the center-of-mass coordinates as follows:

$$E_2 = (M-m)^2 / (M+m)^2 \times E_1$$

E_2 is the energy of the neutron after the collision with a nucleus of mass M ; E_1 is the energy before the collision and m is the mass of the neutron (Rusk, 1964). For a head on collision of a neutron with a proton at rest, the above

equation shows that the neutron loses all of its energy and momentum with the momentum and kinetic energy being transferred to the proton. However, since random collisions occur at varying angles, the neutron will lose on the average only 63 percent of its energy per collision. After two such collisions, the neutron would have only 14 percent of its original energy. A 2-Mev neutron can be reduced to thermal energies in about 18 collisions in hydrogen (Rusk, 1964). The average number of collisions needed to reduce the energy of a neutron from one value E_2 to any value E is given by

$$N = \ln(E_2/E)/\xi$$

where ξ is defined as the average decrease in the logarithm of energy per collision, i.e., $\xi = \ln(E_2/E)$. ξ for hydrogen has a value of 1; for carbon, $\xi = 0.16$ (Soodak, 1962). In the case of isotropic scattering in the CM system, the ξ for $A > 10$ is approximately $2/(A + .67)$. Hence, the neutrons that emerge from the surface of the moon depend strongly on the atomic weight of the scattering centers. Hydrogen will produce a high flux of slow neutrons if it appears in the sub-surface material.

B. Review of Neutron Scattering Theory

Elastic scattering is usually the only process which occurs with neutrons of a few Mev or less and elements of low A , i.e., $A \leq 25$. At higher energies the scattered neutron has less energy than the incident neutron; the difference is used in nuclear excitation. Usually a gamma ray is released in the de-excitation process. However, in the low A elements, the lowest excited state is very high (many Mev), hence, only elastic scattering occurs (Price, 1958).

As the fast neutrons traverse matter and lose energy, the process of thermalization is established. If the medium has a low absorption cross section, a Maxwellian distribution will arise from the motion of the scattering atoms and the thermal neutrons. This, however, is not exactly true since some absorption will prevail and there will be leakage before thermalization. For a low A material which weakly absorbs, the neutron energy distribution approximates an equilibrium distribution. This Maxwellian distribution can be represented as follows:

$$N(E)dE = 2\pi n / (\pi kT)^{3/2} e^{-E/kT}$$

The average energy is $(3/2)kT$ while the most probable velocity is $(2kT/m)^{1/2}$. At $T = 293.6K^\circ$, $E = kT = .0253$ ev; the most probable velocity is 2200 m/sec. The total flux is given by $\Phi = \int_0^\infty N(v)dv_v = n\bar{v} = \int \rho dv$ where ρ is the density of neutrons.

When the slow neutrons approach the nucleus they come within range of nuclear forces. If such a neutron is "attracted" by these forces it can be captured. Usually the excited nucleus emits a gamma ray; this is called radiative capture (Ishin, 1963). The presence of a neutron with an energy near the resonance energy increases the probability of capture. However, most of the resonance capture is for higher A elements. A medium with a low absorption cross section and a low atomic weight will mainly produce leakage. This leakage will be affected by capture mainly at slow neutron energies when the capture cross sections become important.

If a fast neutron source is placed near an absorbing-scattering material with the neutron beam essentially parallel, the transmitted neutron intensity I is related to the incident intensity I_0 by the exponential equation (Title, 1964).

$$I = I_0 e^{-N\sigma x}$$

where N is the number of atoms/cm³ in the absorber, σ is the total microscopic neutron cross section (cm²/atom) and x is the sample thickness. The energies of the incident and transmitted neutrons are the same. For more than one type of atom, a summation over N and σ is necessary. This summation is denoted by Σ_t , the total macroscopic neutron cross section.

The preceding provides many essentials for the problem of slowing down of neutrons. As the neutrons diffuse and scatter through the medium, the energies correspondingly

change. The number per unit volume whose energy has changed to E per unit time is the slowing down density q . The relation between q and the flux is as follows: (Title, 1964)

$$q = \phi \Sigma_s$$

where Σ_s is known as the "slowing down power" of the medium. ξ is the mean logarithmic energy loss per collision.

In a medium which produces thermalization, multi-group diffusion theory may be applied. In its simplest form, multi-group reduces to one-group where group refers to neutrons at the same energies. This is a low-order approximation of transport theory in that the neutrons of constant energy diffuse through a medium. The differential equation of diffusion theory is as follows (Glasstone and Edlund, 1952):

$$D \nabla^2 \phi - \frac{\phi}{\lambda_a} + S = \frac{\partial n}{\partial t}$$

Here ϕ is the neutron flux in neutrons/cm²-sec, λ_a the absorption mean free path, D the diffusion coefficient, S the source strength per unit volume, n the neutron density in neutrons per unit volume, and t is the time. The neutron flux is related to the neutron speed v , as follows:

$$\phi = nv$$

The neutron density in the steady state is constant in time, hence the right hand side of the above equation is equal to zero. Usually there is a point source; at all points except the origin S is zero. The above equation may then be written

$$\nabla^2 \phi - \frac{\phi}{L^2} = 0$$

The diffusion length L is defined as $(\lambda_a D)^{1/2}$. Thus

$$\nabla^2 \phi - \frac{\phi}{L^2} = 0$$

D is defined as $\frac{\lambda_{tr}}{3}$ where λ_{tr} is the transport mean free path given by $\frac{\lambda}{(1-\overline{\cos\theta})}$. λ is the total mean free path and $\overline{\cos\theta}$ is the average cosine of the scattering angle in the laboratory system. In the CM system, $\overline{\cos\theta} = 2/3(A)$, hence

$$\lambda_{tr} = \frac{\lambda}{(1-2/3[A])}$$

For a description of slowing down using one group diffusion theory, the "slowing down power" is used in place of the absorption mean free path. L then represents the slowing down length for epithermals, i.e., neutrons with energies from 0.1-100 ev. The solution of the one group equation for a point source in an infinite homogeneous medium is (Feld, 1953)

$$\phi = \frac{\phi_0}{4\pi D} \frac{e^{-r/L}}{r}$$

where ϕ is the epithermal or slowing down flux, or it is the thermal flux, depending on the energy.

C. Expected Lunar Intensities

The neutrons that are produced on the surface of the moon are a result of interactions of cosmic rays with the surface nuclei. Fig. 1 shows the observed solar (Fichtel, 1963) and galactic cosmic rays proton flux. Without a magnetic field or an atmosphere, the charged particles (mainly protons) of low energy are not deflected away from the moon, hence, there are direct interactions between the primaries and the surface nuclei. The flux may be like that found near the geomagnetic poles of the Earth. The nuclear reactions are complex which include meson production, evaporation, and knock-on mechanisms.

It should be noted that the production of neutrons at the surface of the moon is about five times that of the global average of the Earth (Lingenfelter, Canfield and Hess). Besides the high incident cosmic ray flux, the average atomic weight of the surface of the moon is higher than the Earth's atmosphere. Also, pions created in the primary interactions may interact before decaying. All of the preceding generate the increased production value. An approximation of the lunar albedo can be made by scaling the value of the terrestrial albedo (Haymes and Korff, 1960). The slow neutron density near 40 degrees geomagnetic latitude is 1×10^{-8} neutrons/cm³. Later measurements of the fast albedo revealed the flux to be approximately 0.1/cm²/sec at geomagnetic latitude of 41 degrees.

The neutrons produced on the lunar surface result from inelastic processes in excited nuclei. Gamma rays,

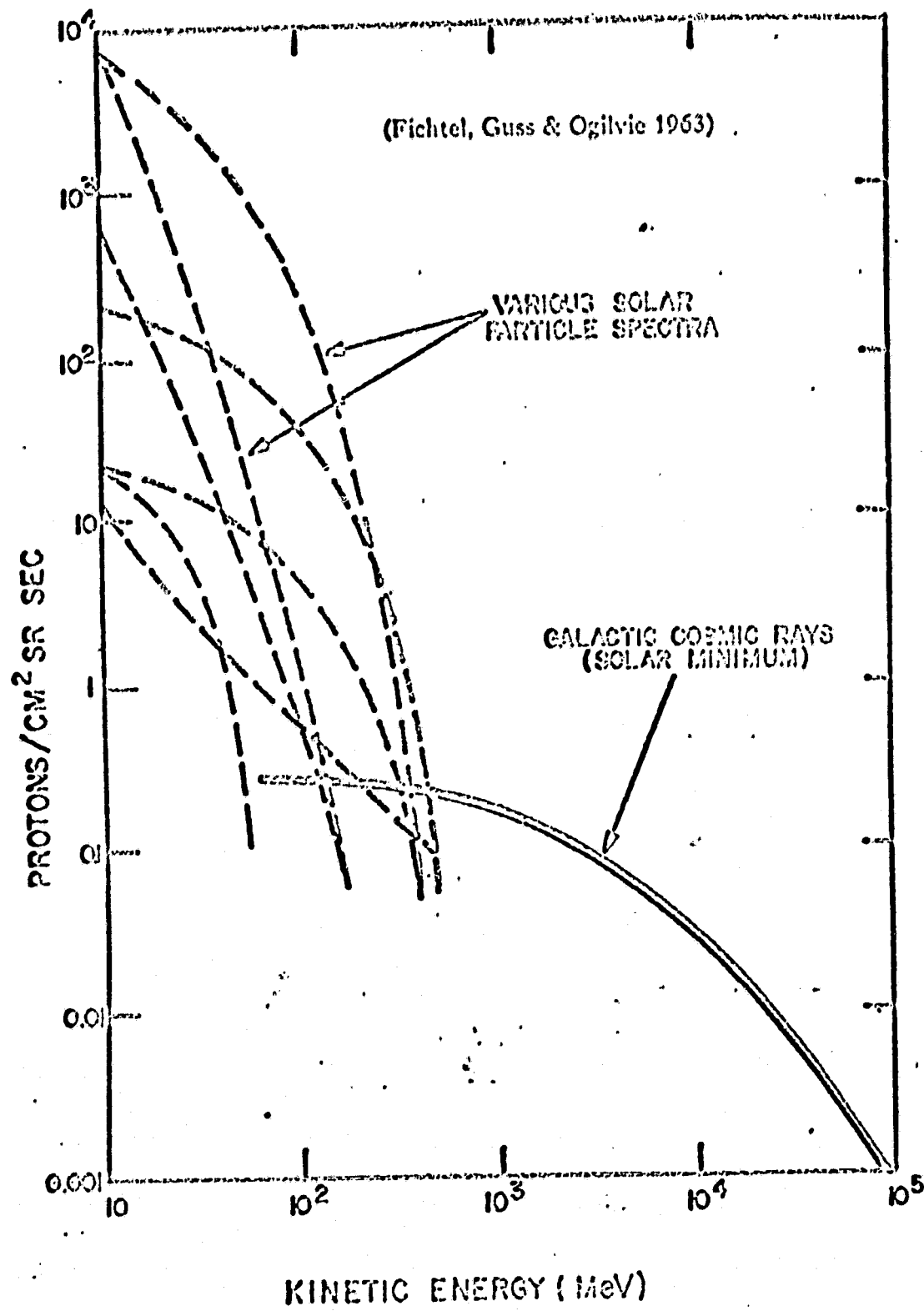


Fig 1

charged particles, and neutrons are a direct result of the interactions. The neutrons can be captured and produce gamma rays or escape the surface of the moon. Lingenfelter, Canfield and Hess (1962) used diffusion theory to estimate the lunar neutron albedo. The calculated equilibrium leakage spectrum produced is dependent on the assumed composition and very sensitive to the hydrogen concentration. The most significant result is that of the ratio of slow to fast neutrons varies with the variation of the ratio of hydrogen to silicon atoms. (Silicon is used as a basis because [Si] is the parameter in geological theories.)

In an experiment that would measure such a ratio of slow to fast neutrons from the lunar surface, the thermal neutron cross section would need to be known in order to interpret the results in terms of the absolute hydrogen concentration.

The lunar neutron flux will depend on the cross sections and on the hydrogen content. The hydrogen can be in the form of water of crystallization and/or in the form of ice.

Approximately 0.5% of the lunar surface is believed to be permanently shaded (Watson et al., 1961), and fossil ice would remain at such sites. Volcanic rocks have a water concentration of about 1% by weight while stony meteorites have about 0.1% by weight (Green, 1964). By correlating water concentration within the surface of the moon, and the composition of the surface, a geological connection can be made.

The experiment to be discussed herein employs a simulator of lunar neutrons. By using a material similar

to that of the surface of the moon and be having a source of fast neutrons, counting rate measurements are made of the slow and fast neutrons as a function of the water content in the material. A correlation is then made between the ratio of the slow to fast neutron counting rates and the water concentration.

Any significant statistical deviation in observed counting rates can be then used as an index of hydrogen concentration. Therefore, the purpose of this research and thesis was to test the ideas outlined above and establish quantitatively the sensitivity of the technique to water concentration.

II. LUNAR NEUTRON SIMULATION

A. Simulator

The lunar neutron simulator consists of (fig. 2) a cylindrical drum with dimensions of 5 feet in length and 3 feet in diameter. The drum is mounted so that it can be rotated around the horizontal axis and also placed in a vertical position. It is sufficiently large so that over a ton of material may be loaded into it.

The three neutron detectors are located in the center of the drum at a distance of approximately 9 and 1/2 inches from the neutron source. The detectors are 120 degrees apart.

The neutron source used was a plutonium-beryllium radioactive source with a neutron yield of approximately 10^6 neutrons/sec. The distribution of energies is from 1 Mev - 10 Mev with the peak energy yield centered at 5 Mev. This source was similar to the neutrons produced on the lunar surface since the lunar neutrons are initially fast neutrons of several Mev.

The media used were a basaltic material from West Texas (a volcanic rock) and a finely washed variety of sand. These two samples were used so that the effects of differing elemental composition could be evaluated.

The drum was filled to a level of 4 feet from the lower end to insure proper neutron thermalization. This also gave a geometry which had equal amounts of material from source to detectors and source to surface.

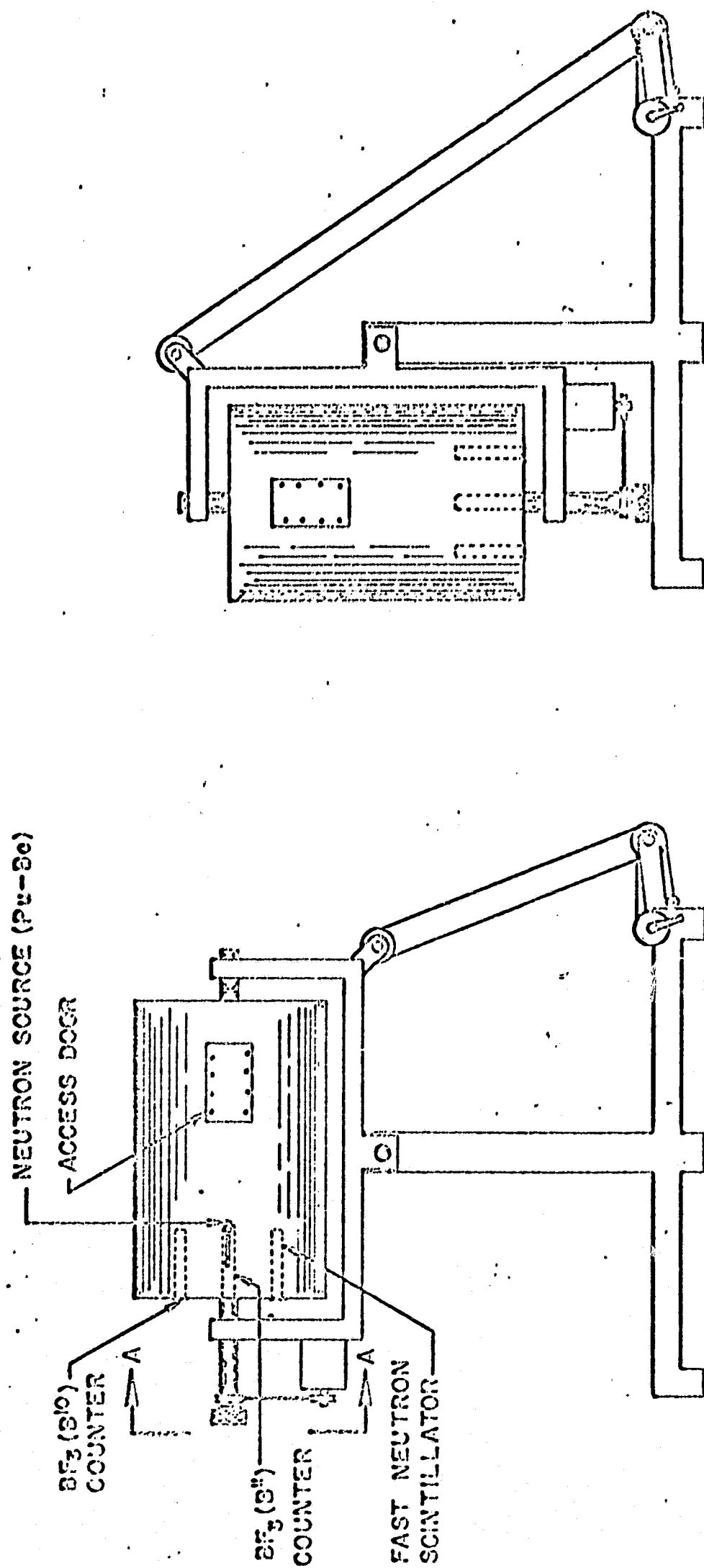


Fig 2

Any level less than 4 feet would allow epithermal neutrons to leak from the top surface thereby losing these neutrons for counting.

The procedure used during the experiment was to use the drum in a horizontal position for mixing the material and the water. The drum has curved vanes which allow proper mixing and uniformity of water content. After mixing, the drum is placed in the vertical position so that the material may cover the detectors and the neutron source. Counting rates were obtained in this mode of the simulator.

B. Neutron Detectors

Two unmoderated proportional counters filled with BF_3 were used as the slow neutron detector. The only difference between the counters is in the isotopic enrichment of the boron in the BF_3 gas. One counter has a concentration of 96% B^{10} while the other counter has a concentration of 11% B^{10} and 89% B^{11} . Boron-10 trifluoride detectors utilize the B^{10} (n- α) reaction to detect slow neutrons. B^{10} has a high cross section for absorption of slow neutrons ($1/v$ dependence; $\sigma_a = 4000$ barns). An alpha particle and lithium nucleus are products of this reaction; these charged particles produce the ionization in the counter. The ionizing power of the two particles is nearly the same so that the net pulse height is nearly equal to the pulse height due to alpha particles of energy 2.79 or 2.31 Mev. A 480 Kev gamma ray is liberated 94% of the time due to the decay of the lithium nucleus. The B^{11} isotope does not have an appreciable absorption cross section for slow neutrons; the preceding reaction does not take place (Blizard, 1962).

Both counters have a sensitive volume of 8 inches and a diameter of 2 inches. The fill pressure was 90 cm Hg with the measured plateau less than 2% per 100 volts (fig. 3). The operating voltage was found to be 3200 volts for both of the counters.

The difference of the counting rates in the two detectors is a measure of the slow neutron density.

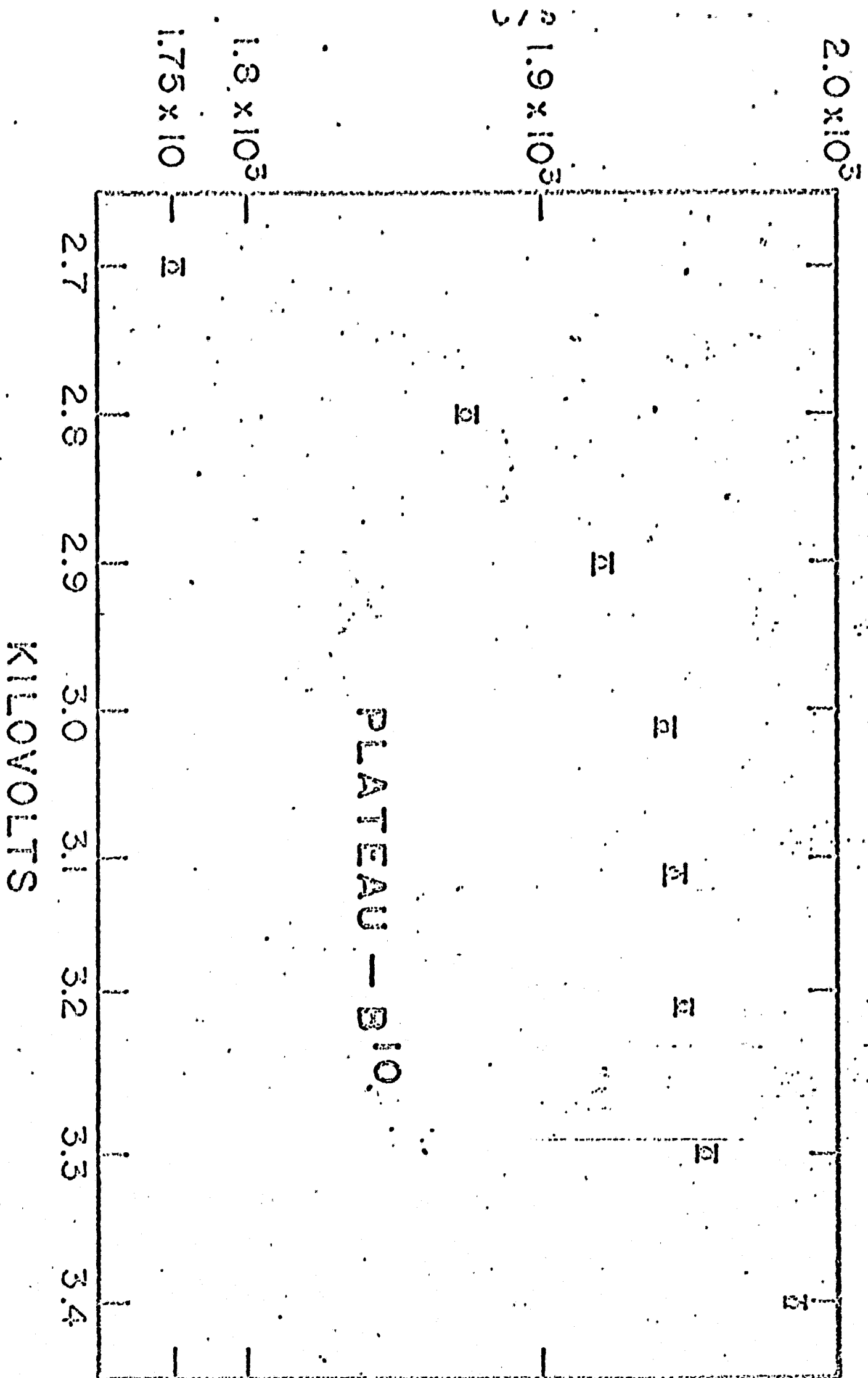


Fig 3

If n is the counting rate due to slow neutrons in a counter filled with 100% B^{10} , and b is the background counting rate due to charged particles, then

$$A = 0.96n + b$$

$$B = 0.11n + b$$

where A and B are the two measured counting rates of the two counters (Korff, 1945). These two equations may be solved for n and for the background counting rate b .

The fast neutron detector consists of an organic scintillator in which the neutron is detected through the (n,p) elastic scattering process. The scintillator has the property in that a neutron will lose some of its energy to the proton hence causing a pulse of light. A photomultiplier tube detects the light and the electronics acts upon this pulse (Haymes, 1965).

At the energies of fast neutrons, the scintillator also reacts to charged particles and gamma rays. Hence, a guard counter which consists of an inorganic scintillator surrounds the organic scintillator; this is called a phoswich system. Charged particles which traverse the two scintillators cause pulses to be generated in both layers. The pulses are differentiated from each other by pulse-shape discrimination. The gamma rays are also rejected through the pulse-shape discrimination. The lower level of the detector is set at 1 Mev with the range being up to 15 Mev. The resolving time is 10 μ sec.

C. Electronics

The two slow neutron detectors require a source of high voltage, an amplification system, and a counting system. The power supply used was a Fluke Corporation model 408A which is an extremely well regulated, low noise instrument. The regulation is 0.001% for 10% line change, with a ripple less than 1 mv RMS.

The amplification system consisted of an Ortec 103 preamp and 203 amplifier designed for use with solid state radiation detectors. This system has low noise and the high sensitivity needed for the detectors.

The fast neutron detector required two small power supplies for ± 8 volts and the counter.

The pulse from the (n, α) reaction is usually about a millivolt at the output of the proportional counter. Gamma rays generate fewer ions than do the neutrons. The pulses due to the gamma background are much smaller than the neutron pulses and are easily blocked by the proper selection of the bias in the amplifier system.

The B^{10} counter has a maximum counting rate of approximately 5×10^5 cps while the maximum counting rate observed in this experiment was on the order of 4×10^3 cps. Hence, the system was not being paralyzed by a high counting rate and capable of responding to successive events.

The efficiency of a neutron counter is defined as the fraction of the neutrons which, upon entering the counter, result in a count. The probability of the absorption of a neutron passing a distance d through the B^{10} is

$1 - \exp(-N\sigma d)$, assuming a collimated beam of monoenergetic neutrons (Price, 1958). N is the number of B^{10} atoms per cubic centimeter and σ is the appropriate cross section. In general, the efficiency depends on the size and shape of the counter and on the direction of incidence of the neutrons as well as the other properties of the counter. In fig. 4, the efficiency is depicted for the counter used in this experiment (96% enriched) for neutrons incident perpendicular to the counter and for those parallel to the counter. The efficiency for various energies has the form

$$\text{Efficiency} = 1 - \exp[-0.624d(0.025/E)^{\frac{1}{2}}]$$

where E is the neutron energy in electron volts.

The measured fast neutron scintillator efficiency is shown in fig. 5 (Haymes, 1965); it follows the shape of the elastic-scattering cross section on energy above the electronically-set threshold of 1 Mev.

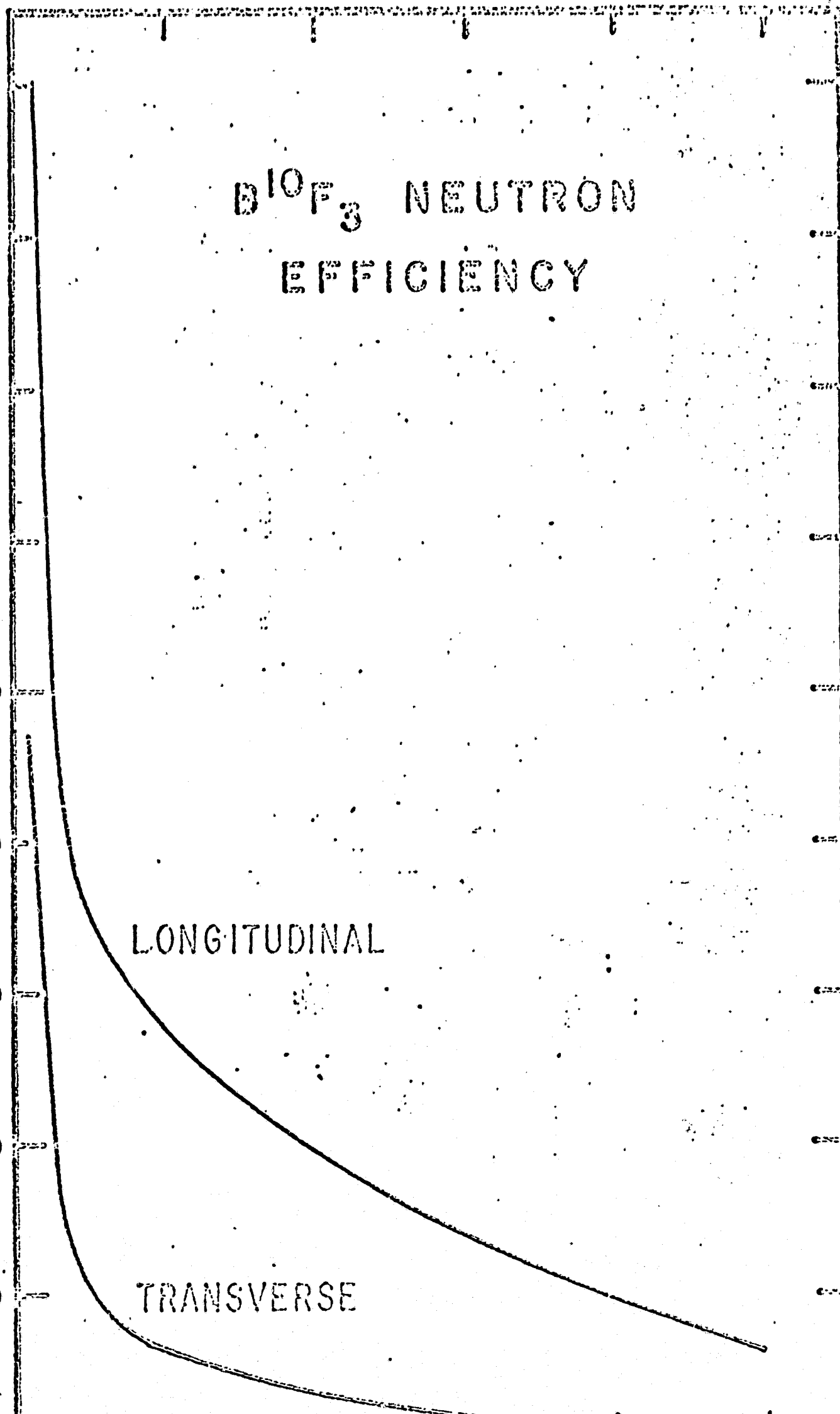
$B^{10}F_3$ NEUTRON
EFFICIENCY

CALCULATED PERCENT EFFICIENCY

90
80
70
60
50
40
30
20
10

LONGITUDINAL

TRANSVERSE



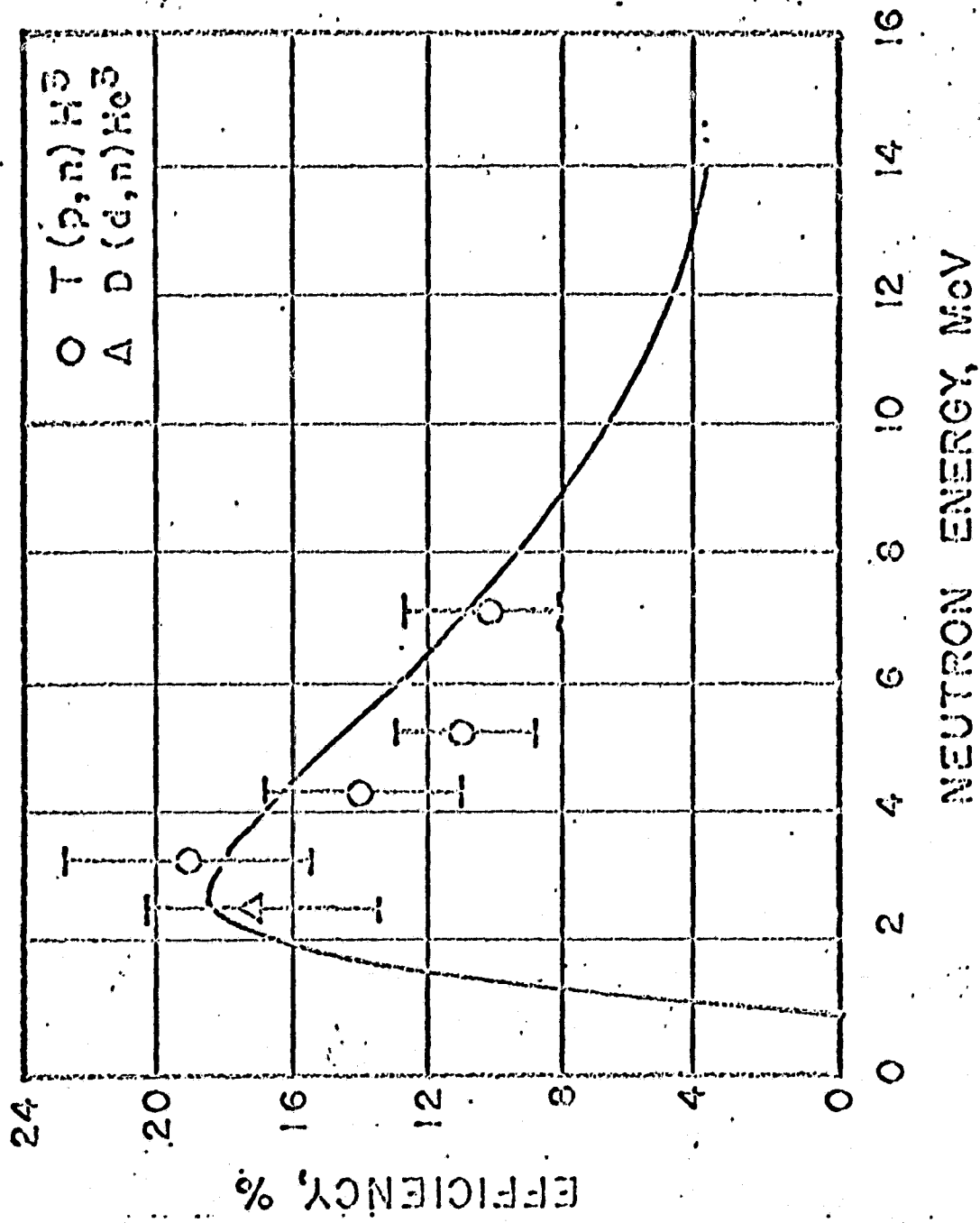


Figure 5. Calibration curve of the detector, obtained with monoenergetic neutrons. Calibration with a Pu-Be source yielded 16 ± 3 per cent, an average energy of 4 to 5 Mev.

D. Experimental Procedure

The objective of this experiment was to measure the slow neutron counting rate, the fast neutron counting rate, both at a given (and measured) water content in the material. As previously stated, the simulator was positioned in the vertical mode when obtaining the counting rates. This was to insure that the counters and the neutron source were entirely covered. Counting rates were obtained and normalized to five minute time segments. Due to the high counting rate, good statistics were obtained.

After the counting periods were made, the simulator was placed into the horizontal mode. Samples were then taken of the material at various positions in the simulator. These samples, ranging between 10-20 grams, were then weighed using an analytical balance. Each of the samples were weighed to the nearest milligram. The weighing was also ± 1 mg. Thus taking large samples reduced the relative errors to a minimum.

The samples were then heated to a temperature of approximately 600 degrees F for several hours until a constant weight was obtained. A dessicator was used to dry and cool the samples before reweighing. The weight loss then was recorded.

The determination of water content involves many factors such as particle size, chemical composition, and type of water binding. The basaltic material used had a chemical composition as depicted in fig. 16. The water contained in this material as indicated is water of

-17-

crystallization. The analysis given was based on 105 degrees centigrade. However, in an area where the relative humidity is high, this substance will have absorbed water (Skoog, 1962) which is easily driven off by heating to 105 C. There is also occluded water which occurs in microscopic pockets in the rock material.

At 600 degrees F, all of the absorbed water and water of crystallization will be driven out; some of the occluded water will also be lost. Since carbon dioxide is also driven out of the material, some correction might have to be made in the analysis through weight-loss. However, before the experiment began, the entire basaltic material was heated to 600 F in order to start from a low water content. Therefore, no corrections were required.

The other sample, sand, required no initial heating since it had a low water content ($< 0.1\%$). In fig. 6 the weight loss as a function of temperature is given (Nutting, 1943). In sand, the carbon dioxide driven off at 600 F is negligible compared to that of water.

The background which was present in the slow neutron counting system was primarily due to gamma rays produced in the fast neutron source. Cosmic rays add to the background; roughly one per minute will cross every square cm of horizontal sectional area near sea level (Korff, 1955). Natural contamination from the counter is negligible compared to the other sources. Finally, background due to local radioactive materials and the basaltic and sand materials give a small addition to the background. The background was less than 2% of the B^{10}

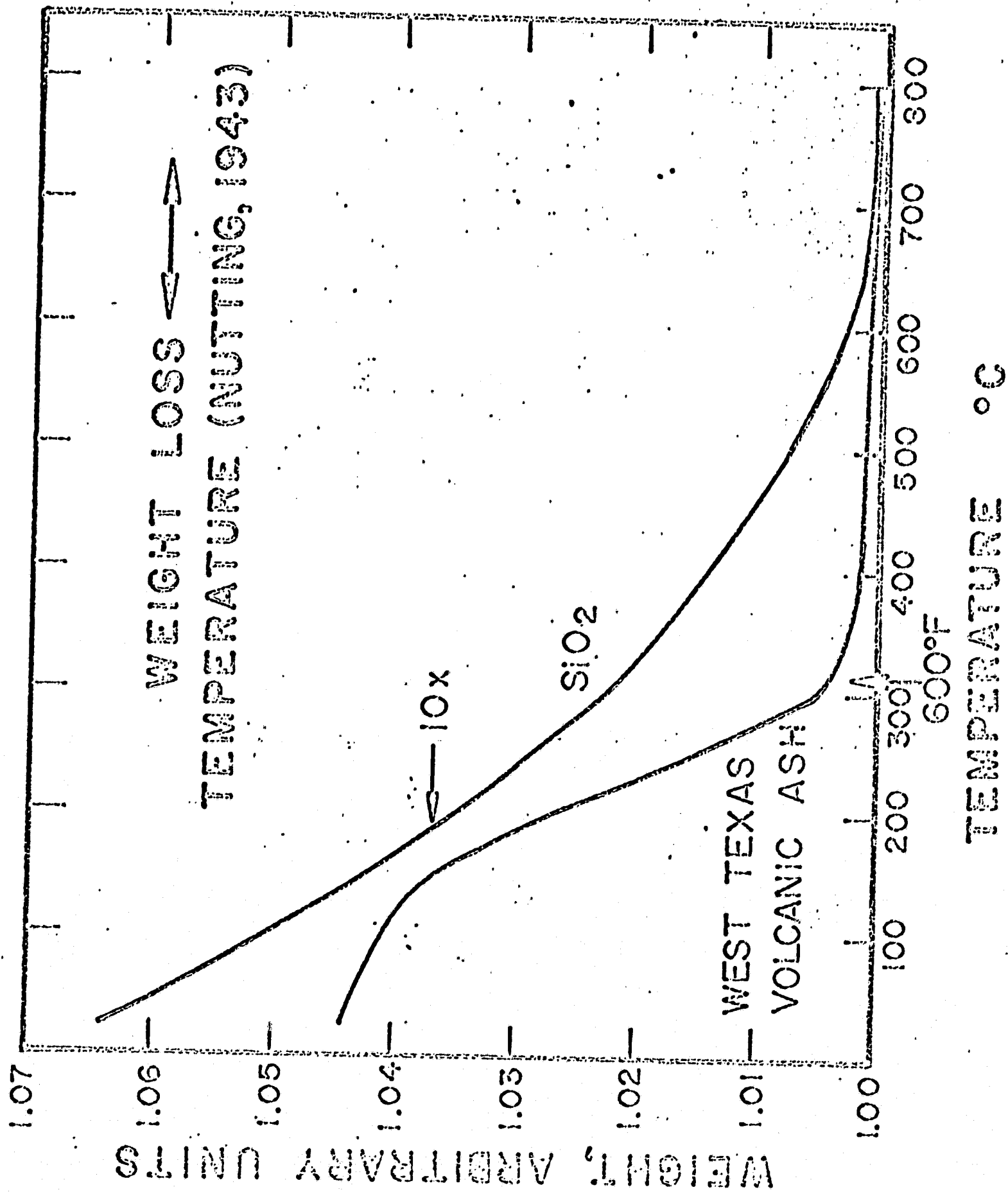


Fig 6

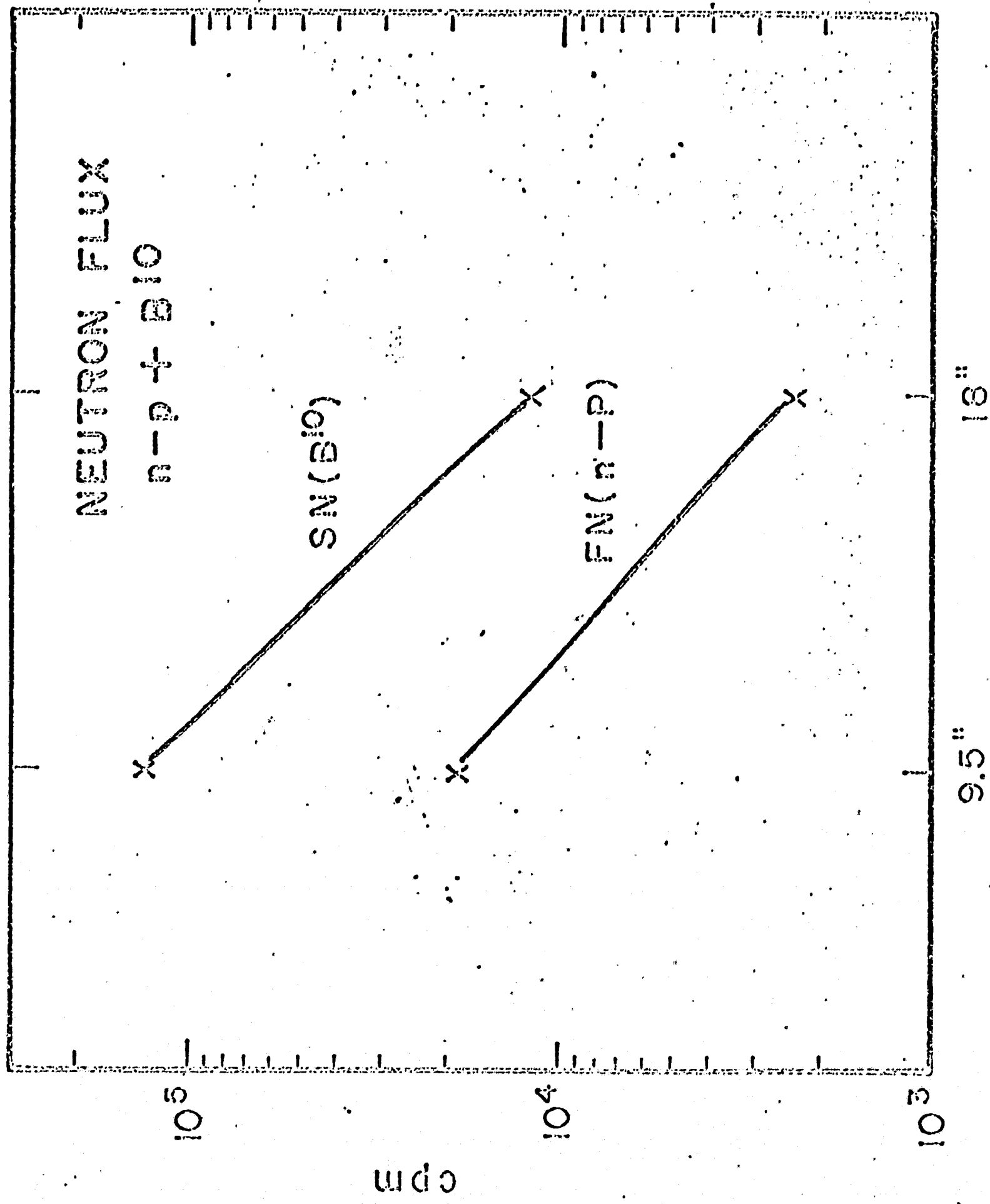
counting rate. Of this value, more than 90% was due to the neutron source inside the simulator.

The distribution of neutron energies and fluxes can be anticipated by use of the diffusion parameters. The slowing down length, defined as the average shortest distance that a neutron travels in slowing down from some energy to another (Isbin, 1964), is comparable to the source-detector distance. A value of approximately 9 inches in basalt indicates that in the energy range of the 4 Mev average [Beckurts, 1964] source energy to thermal energies, the neutrons are thermalized. However, the very high energy neutrons are not thermalized in this range, but are in the epithermal range. For closer source-detector distances, the counting rate will increase because the slow neutron density increases approximately as $1/r$ (Title, 1964).

As water is added to the material, the neutrons which are slowed down to the epithermal energies begin to be further slowed down, hence they are counted. This also will cause some of the faster neutrons to be slowed down so that they will not be counted by the fast counter.

In fig. 7 the counting rates for the source-detector distance are shown. Outside the simulator, the flux is approximately $1/10$ that of the 1.5" distance flux and represents the leakage from the simulator. The fluxes computed by diffusion theory and the resulting counting rates for thermal neutrons are comparable.

In fig. 8 the counting rate is plotted versus the depth below the top level of the medium. This indicates that the center of maximum slow neutron activity does



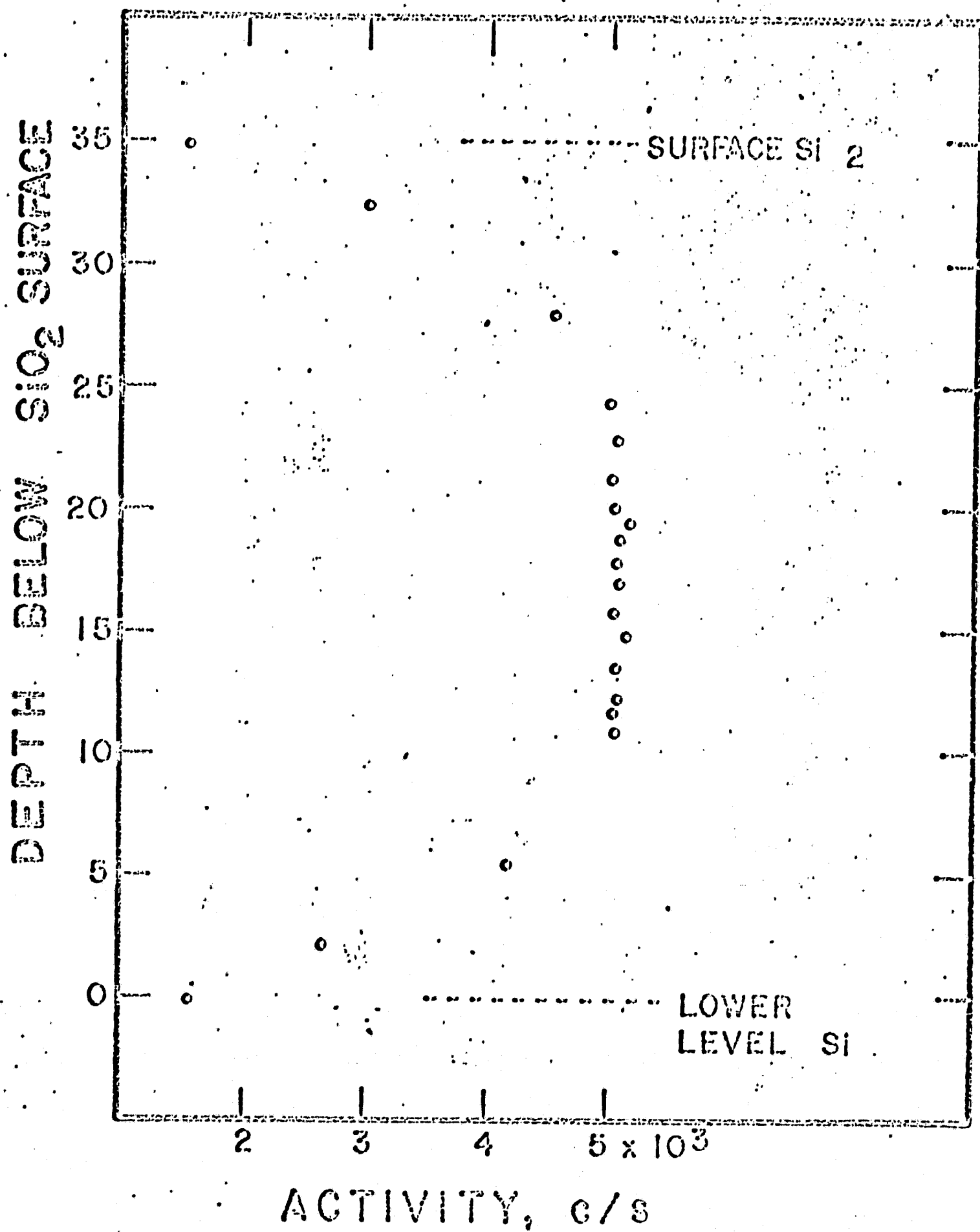


Fig 8

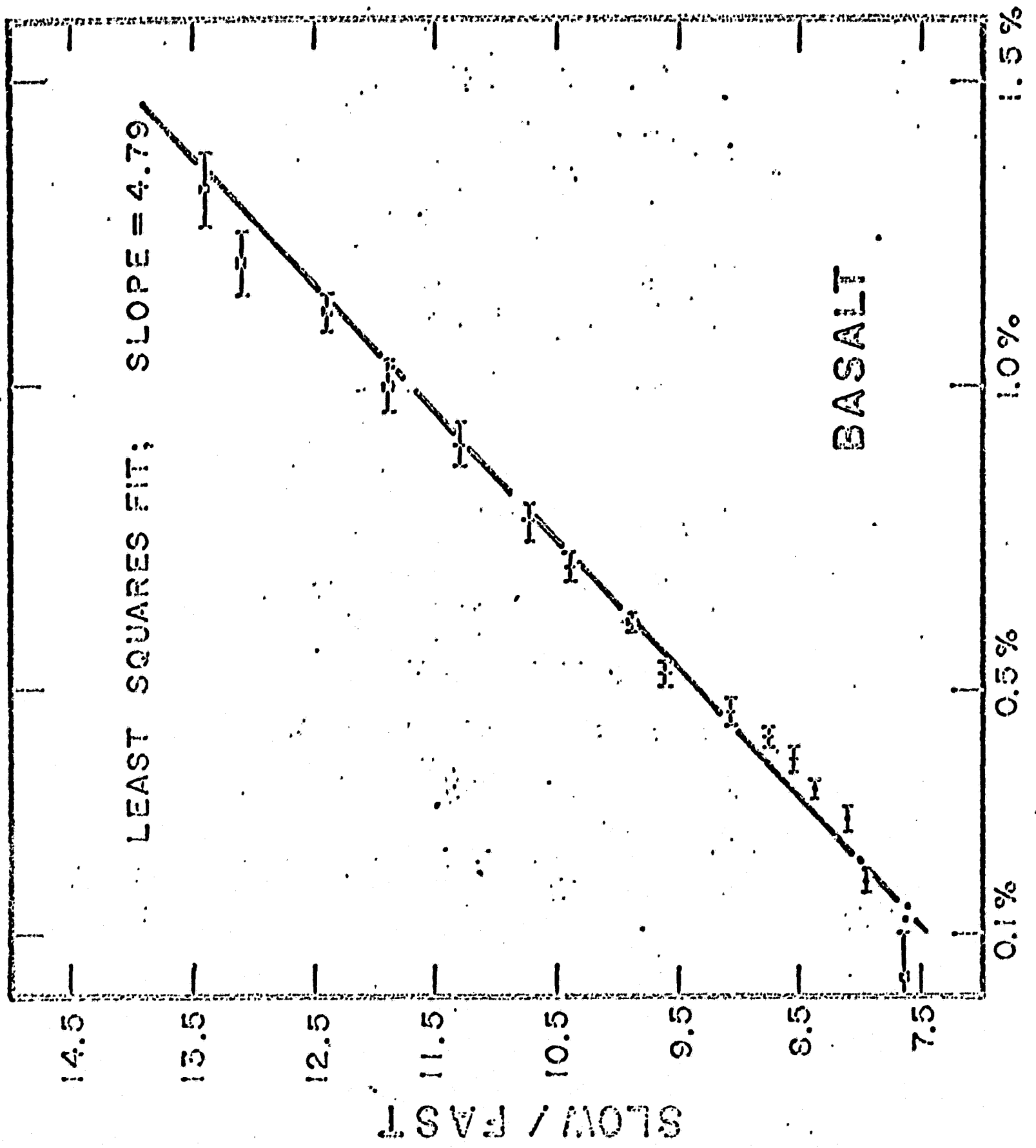
not vary appreciably over the ranges indicated. Hence, the levels of both the sand and the basalt are not required to exactly coincide, physically.

This is also an indication of the thermalization that is present; the small variations observed are not statistically significant. If the neutrons were not completely thermalized, then a small change in the source-detector distance would give an appreciable change in counting rates.

E. Results

In figs. 9 and 10, the ratio of slow to fast neutron counting rates corresponding to the equivalent water content are depicted. A least squares fit of a linear curve to the data gives a slope of $4.79/\%H_2O$ for basalt and $2.22/\%H_2O$ for sand. The standard deviation of the observed counting rates is very small due to high counting rates and long exposure times. The standard deviation of the water content values varies due to irregularities in the mixing process, such deviations for the sand are much lower than those measured for basalt. This is because the sand is much easier to mix with the water. The composite counting rates are plotted in figs. 11 and 12.

The difference in the slopes of the curves in figs. 9 and 10 can be attributed to two factors. One is that the slowing down length of the neutrons in the sand is longer than that of the basalt, so that there are fewer thermal neutrons to count. This is seen in the counting rates in fig. 12. There is a higher flux of epithermal neutrons that are then affected considerably by the addition of water. A more noticeable increase in the counting rate is consequently produced. The other factor is that the basalt has a higher cross section for absorption than the sand (see fig. 16 for chemical analysis). This will tend to cause more thermal neutrons to be extracted from the medium, hence, the counting rates are also affected.



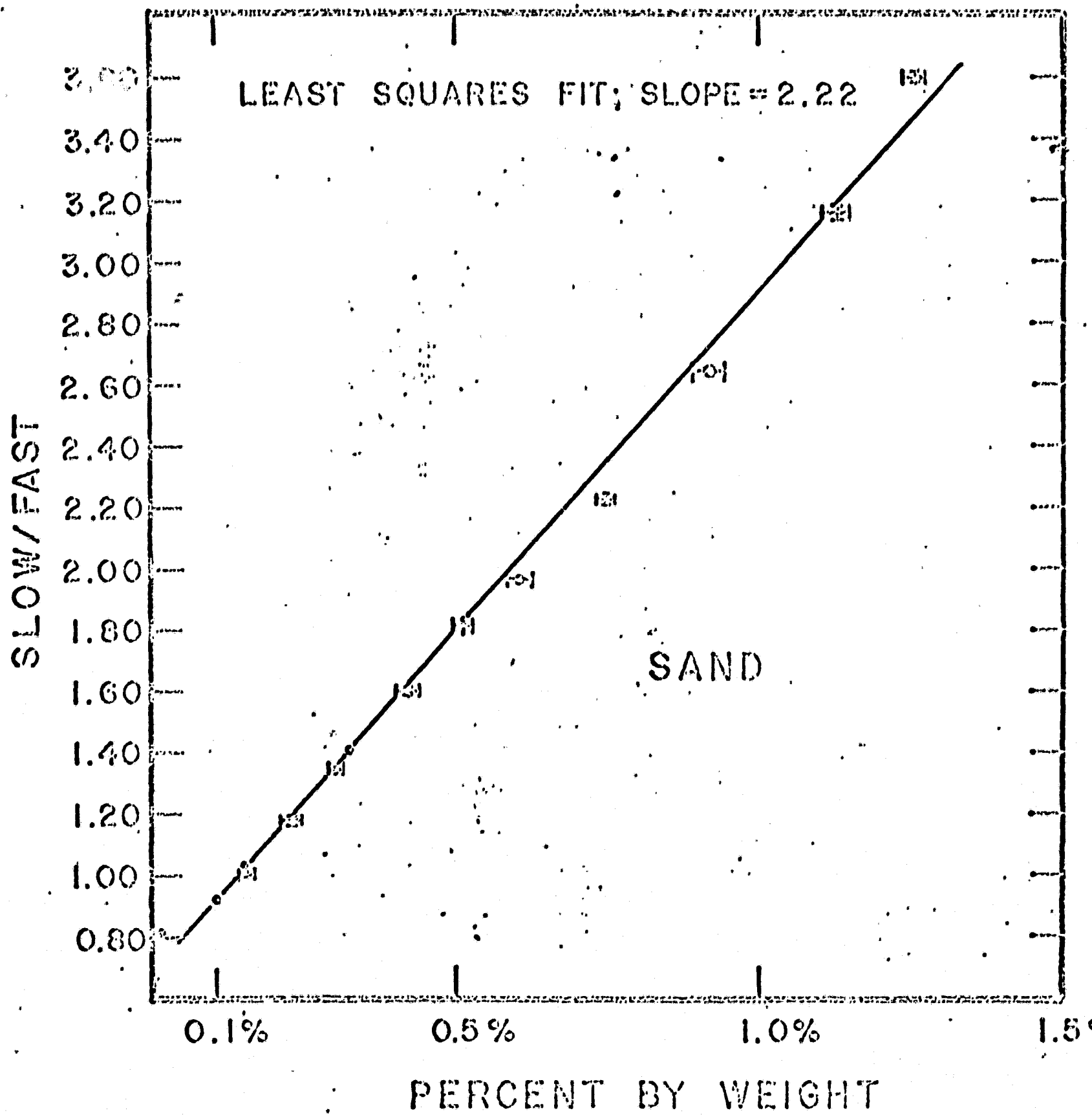
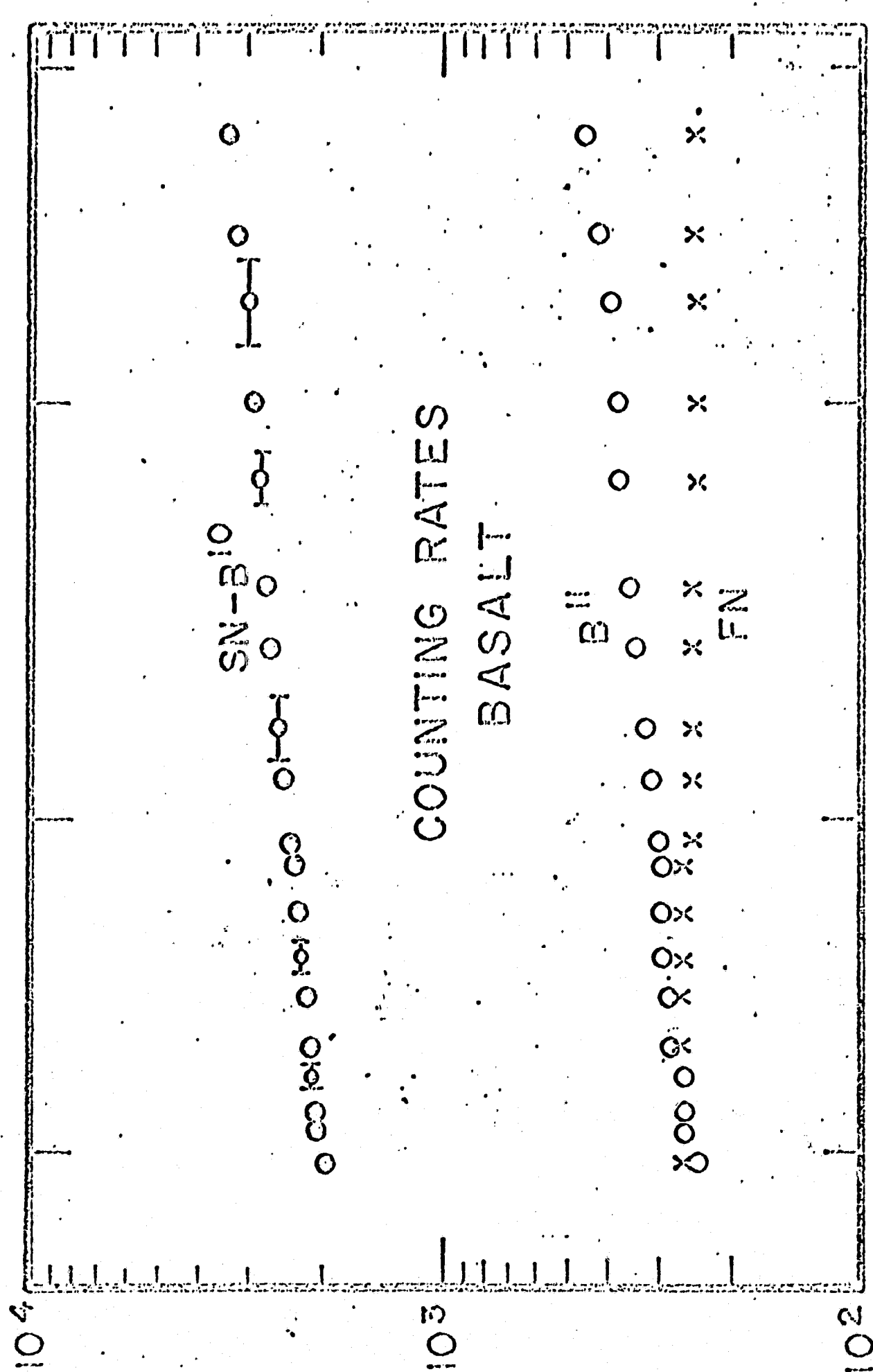
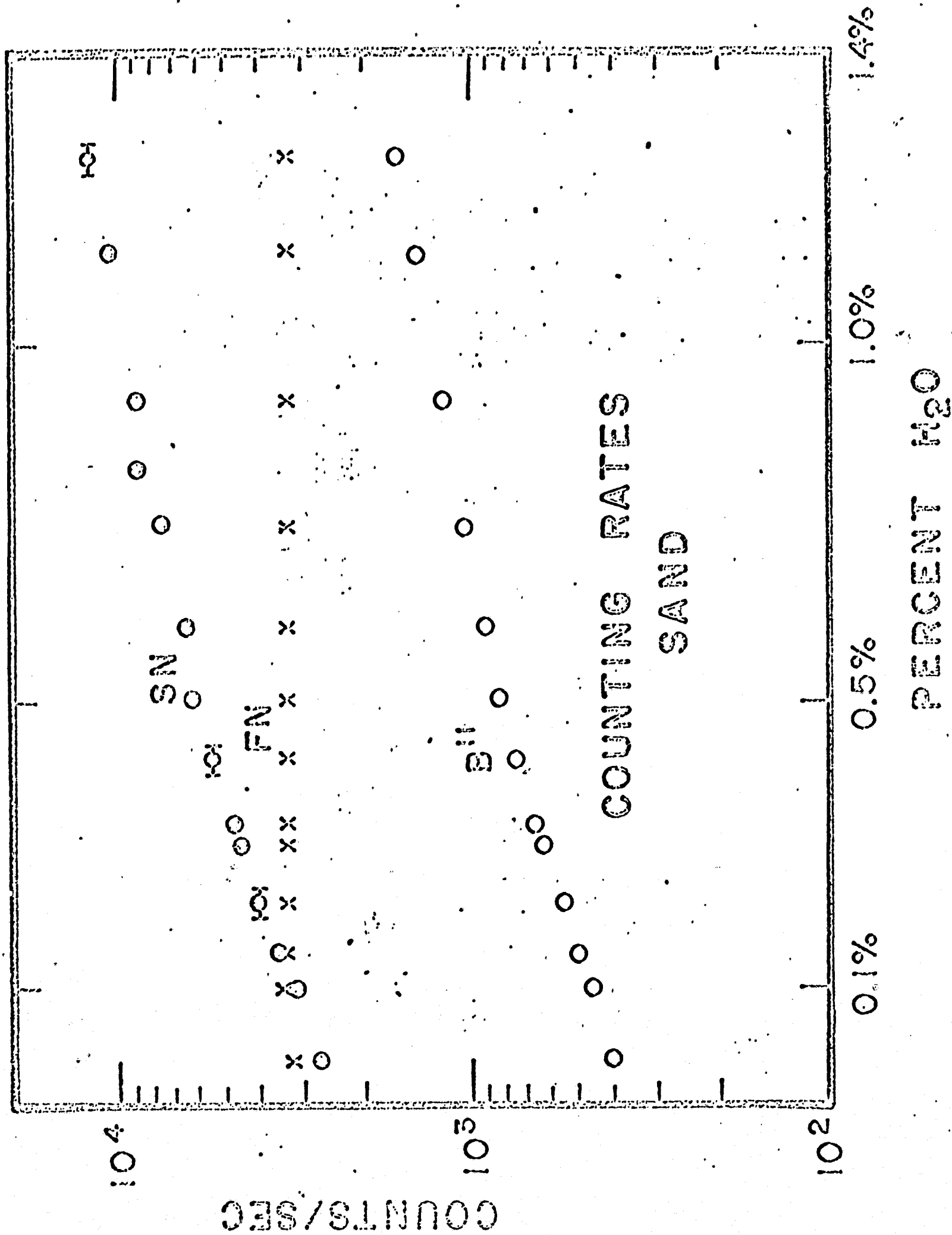


Fig 10



PERCENT H_2O

Fig 11



III. DISCUSSION

A. The Lunar Neutron Flux

Lingenfelter, Canfield and Hess (LCH) have calculated the cosmic ray induced neutron leakage spectrum for various assumed lunar compositions. The equilibrium spectrum is shown in fig. 13. According to this analysis, hydrogen may be detected if the ratio of $H/Si = 0.05$. This corresponds to chondrites of .3% water by weight. It was assumed by LCH that the surface of the moon was similar to chondritic material which has a composition similar to basic rocks of the Earth.

The ratio of $H/Si = 0.05$, which also is 0.25 gm/cm^2 in the upper 100 gm/cm^2 of the lunar crust, is dependent on the cross section of thermal absorbers and uncertain in LCH's calculation of $\pm 50\%$. The Surveyor missions have determined the chemical composition and hence the thermal-neutron cross section to $\pm 7\%$. With this value, the ratio of H/Si will be known to about $\pm 10\%$. (In general, the value of the H/Si ratio can be determined to nearly the same accuracy as the uncertainty in the total capture cross section for H/Si 0.1, LCH, 1961).

The following relation between mineral type and the fraction of neutrons which leak into space holds:

<u>Mineral Type</u>	<u>Neutron Leakage</u>
Chondrites	35.9%
Basalt	29.7
Granite	32.3
Tektites	40.0

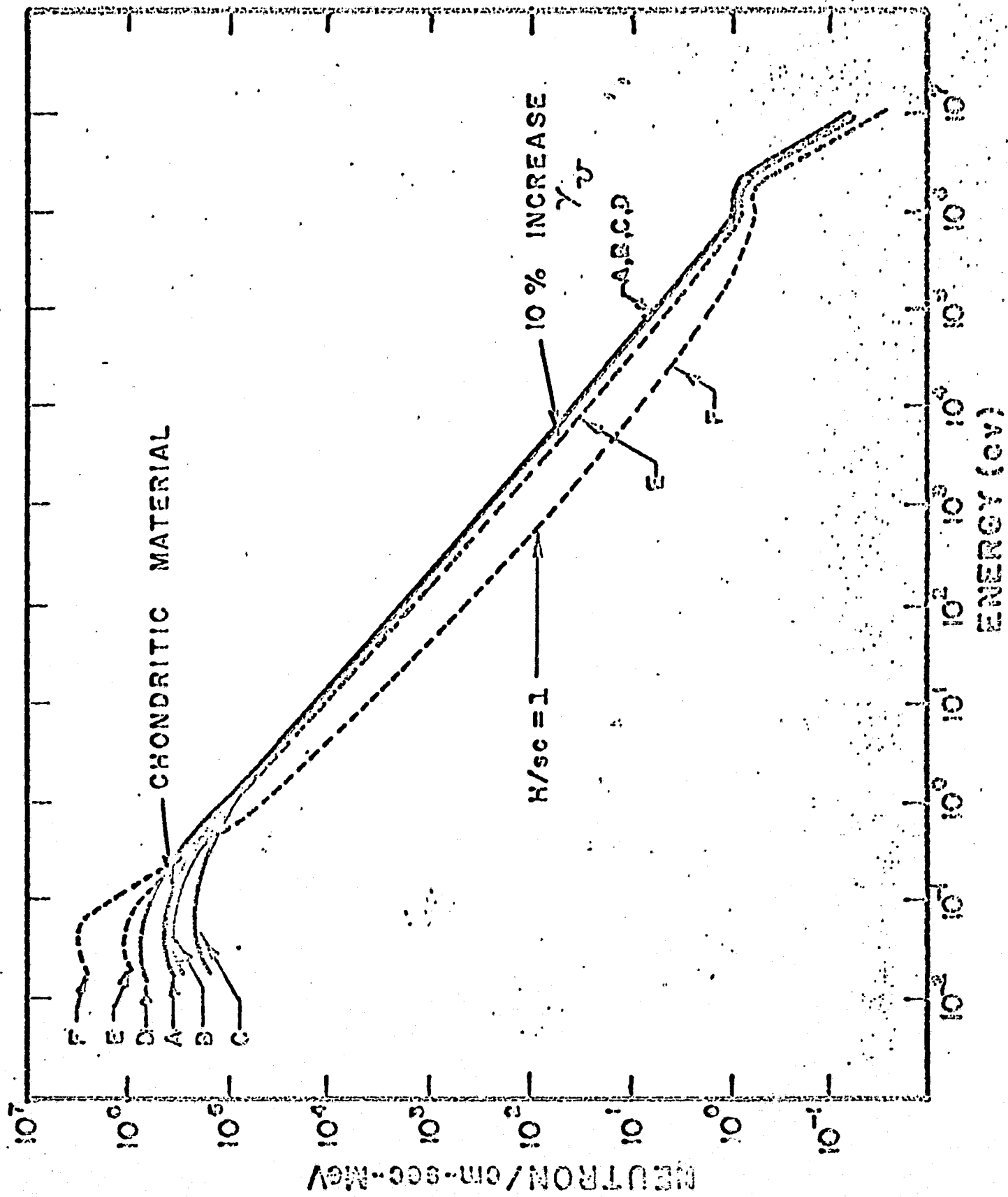
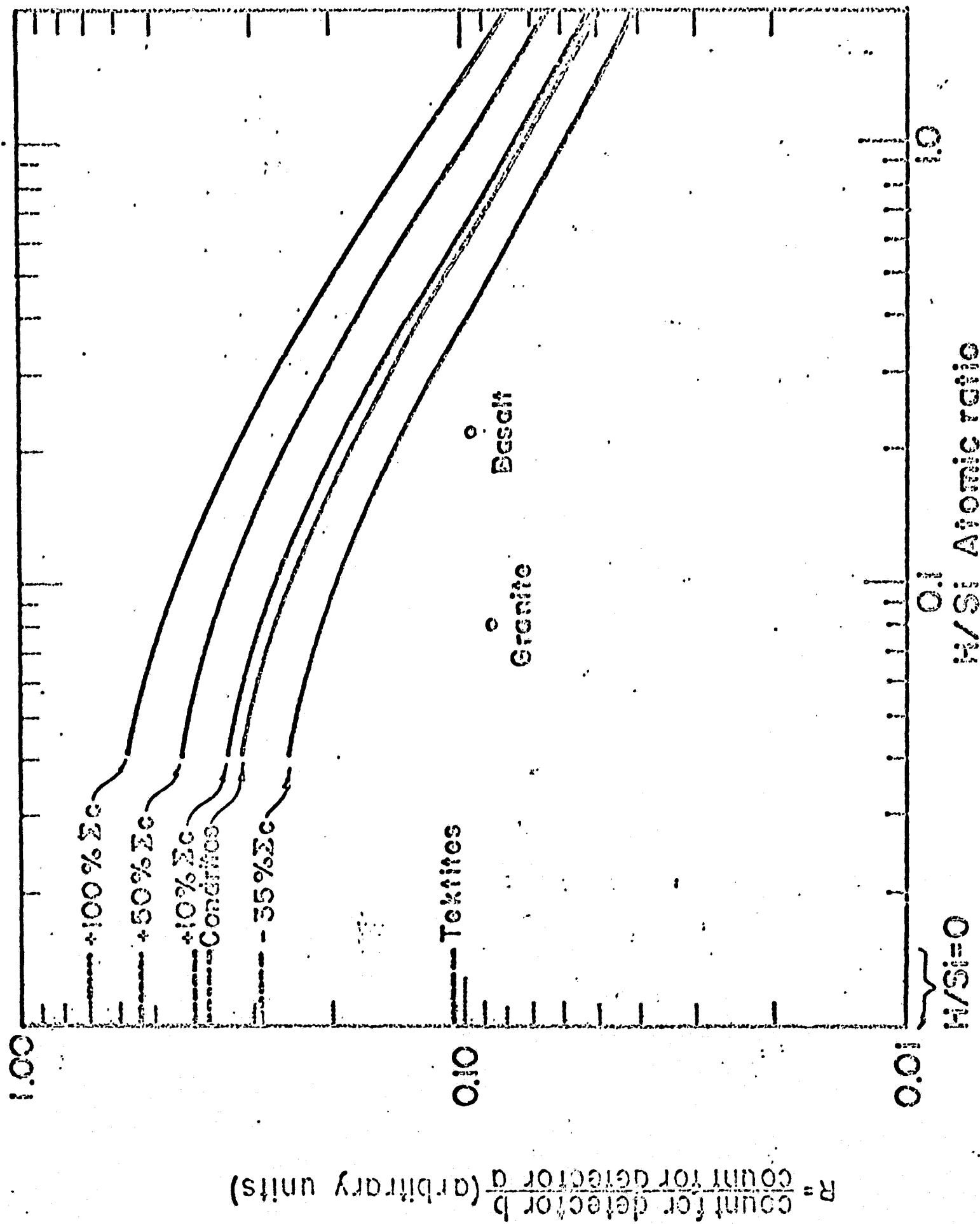


Fig 13



For chondritic material, the leakage decreases to 30.3% with the addition of 0.1 hydrogen atom per silicon, and 17.3% with 1.0 hydrogen per silicon (LCH, 1961). The leakage is quite insensitive to changes in the capture cross section; a 35% decrease in capture cross section yielding only a 0.6% increase in the leakage, and a 50% increase yielding only a 0.7% decrease.

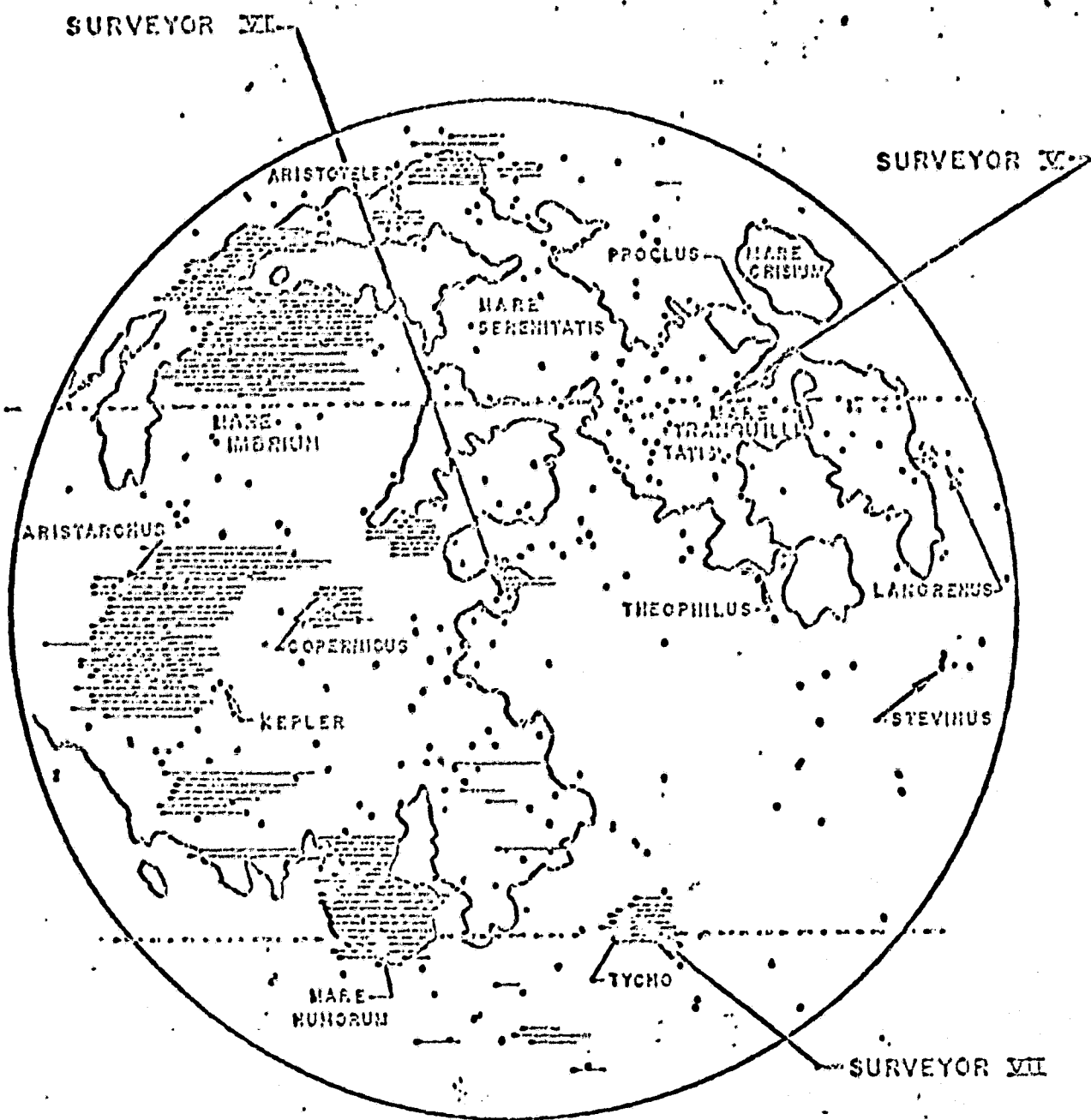
In fig. 14 the calculated neutron counting rate ratios for a $B^{10}F_3$ proportional counter and the same detector with a flat response at higher energies by using a moderator are shown. Two detectors are needed in the lunar experiment due to the uncertainty in the absolute source strength and the thermal cross section for absorption (LCH, 1961)

B. The Surveyor Experiments

The results of the alpha-scattering analyses of the top layers of the moon's surface (Turkevich, 1967) provides us with the chemical composition at the landing sites of Surveyors V, VI, and VII (fig. 15). The hydrogen concentration cannot be estimated from an alpha-scattering experiment. In fig. 17, the measurements of the lunar composition are shown, and in comparison, in fig. 18 the chemical analysis of the materials used in this experiment are shown.

The Surveyor V and VI landings were in the maria regions while Surveyor VII was performed in a highlands site. The results indicate that the composition is clearly different from that of chondrites and resembles that of terrestrial basalts. The VII analysis shows a marked difference from the other two analyses in that the iron content is distinctly less (the "Fe" peak in the Surveyor analyses also include neighboring elements).

LCH's calculations were primarily based on the chondritic theory of the moon. Chondrites, which are stony meteorites, were linked with a parent body, possibly the moon. Hence, this theory needs to be revised and LCH's calculations based on the basaltic composition.



SURVEYOR LANDING SITES

Fig 15

ANALYSIS

	<u>Wt. %, 105°C basis</u>	
	<u>Basaltic Rock</u>	<u>Sand</u>
SiO ₂	45.09	95.40
Al ₂ O ₃	5.87	3.22
CaO	13.94	0.06
MgO	13.64	0.03
FeO	6.87	-
Fe ₂ O ₃	3.28	0.11
TiO ₂	3.60	0.09
K ₂ O	1.13	0.83
Na ₂ O	2.66	0.12
BaO	0.09	0.03
SrO	0.13	0.002
NiO	0.05	-
CuO	0.008	0.005
ZnO	0.016	0.004
Nb ₂ O ₅	0.01	-
Rb ₂ O	0.002	-
ZrO	0.04	0.03
Y ₂ O ₃	0.004	-
H ₂ O ⁺	2.52	-
CO ₂	0.75	-
	<u>99.700</u>	<u>99.931</u>

Fig 16

SURVEYOR VII

SURVEYOR VI

SURVEYOR V

ELEMENT

CARBON	< 3 %		< 2 %		< 2 %
OXYGEN	58	± 5	57	± 5	58
SODIUM	< 2		< 2		
MAGNESIUM	3	± 3	3	± 3	4
ALUMINUM	6.5	± 2	6.5	± 2	8
SILICON	18.5	± 3	22	± 4	18.5
"CALCIUM"			6	± 2	6
"IRON"	13	± 3	5	± 2	2
					± 1

Measurements of Lunar Composition at Landing Sites of Surveyors V, VI, and VII

C. The Lunar Neutron Experiment

American Science and Engineering has prepared a report (ASE-1919) for the National Aeronautics and Space Administration for a proposed lunar surface exploration experiment. Part of this experiment will concern the measurement of the lunar neutron flux. This will be an orbital experiment in a near-moon orbit; the orbital velocity would be about 1.7 km/sec. In a thirty second time interval, an omnidirectional detector will scan a path on the lunar surface of about 50 km in length (the altitude is about 20 nautical miles).

For a basaltic lunar composition, a change in the water concentration from 0.2% - 0.3% would result in about a 4% change in the slow counting rate (assuming the fast component is constant). For the satellite experiment to see this change, one must look at the expected counting rates. Using the LCH value of a neutron strength approximately five times that of the Earth's atmosphere, and using (Haymes and Korff, 1960) the slow neutron density near the top of the mid-latitude atmosphere to be approximately 1×10^{-8} neutrons/cm³, a counting rate of 900 c/min is expected. A background of charged particles (using the values near the Earth's poles) is about 1000 c/min. These counting rates assume that the detectors employed are similar to those in this experiment. The values for the B¹⁰ counter would be about 1900 c/min and 1100 c/min for the B¹¹ counter.

IV. CONCLUSIONS

The principal results of the lunar neutron albedo feasibility study are shown in figs. 9 and 10. These data show that statistically-significant deviations in the ratio of the slow to fast neutrons result when the hydrogen concentration in the sample is only slightly changed.

One consequence of the Surveyor lunar alpha-scattering experiments is that basalt provides a good mockup of the lunar surface material. A change of 0.2 - 0.3% in water content (which represents the chondritic water content by weight) would produce a change in the slow counting rate of about 4%, using a basalt from West Texas (the fast component in this experiment was seen to be constant to a first approximation).

In a thirty second integration time, an area of ~50 km on the moon would be swept out. A proposed lunar orbiting experiment would see a neutron counting rate of approximately 1900 c/min for the slow B¹⁰ counter. A probability of 0.8 exists that a 4% change in the ratio would be detected, even in a single pass over an area this size.

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