MATERIALS DATA HANDBOOK

Aluminum Alloy 7075 (2nd Edition)

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PREFACE

The revised edition of the Materials Data Handbook on the aluminum alloy 7075 was prepared by Western Applied Research & Development, Inc. under contract with the National Aeronautics and Space Administration, George C. Marshall Space Flight Center, Marshall Space Flight Center, Alabama. It is a revised and updated version of the Handbook originally prepared by the Department of Chemical Engineering and Metallurgy at Syracuse University, August 1966.

It is intended that this Handbook present, in the form of a single document, a summary of the materials property information presently available on the 7075 alloy.

The Handbook is divided into twelve (12) chapters. The scope of the information presented includes physical and mechanical property data at cryogenic, ambient and elevated temperatures, supplemented with useful information in such areas as material procurement, metallurgy of the alloy, corrosion, environmental effects, fabrication and joining techniques. Design data are presented, as available, and these data are complemented with information on the typical behavior of the alloy. The major source used for the design data is the Department of Defense document, Military Handbook-5A.

Information on the alloy is given in the form of tables and figures, supplemented with descriptive text as appropriate. Source references for the information presented are listed at the end of each chapter.

Throughout the text, tables, and figures, common engineering units (with which measurements were made) are accompanied by conversions to International (SI) Units, except in the instances where double units would over-complicate data presentation, or where SI units are impractical (e.g., machine tools and machining). In these instances, conversion factors are noted. A primary exception to the use of SI units is the conversion of 1000 pounds per square inch to kilograms per square millimeter rather than newtons, in agreement with the ASTM that this unit is of a more practical nature for worldwide use.

ACKNOWLEDGMENTS

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Sincere appreciation is tendered to the many commercial organizations and Government agencies who have assisted in the preparation of this document.

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TABULAR ABSTRACT

Aluminum Alloy 7075

TYPE:

Wrought, heat treatable aluminum alloy

NOMINAL COMPOSITION:

Al-5.6Zn-2.5Mg-1.6Cu-0.3Cr

AVAILABILITY:

Bare and clad sheet and plate, rod, bar, wir tube, extruded shapes, rolled rings, forgings, and forging stock

TYPICAL PHYSICAL PROPERTIES:

Density	2.80 g/cm ³ at room temperature
Thermal Conductivity (T6 temper)	0.31 cal/cm/cm ² / $^{\circ}$ C/sec at 25 $^{\circ}$ C
Av. Coeff. of Thermal Expansion	23.6 μ cm/cm/°C (20–100°C)
Specific Heat	$0.23 \text{ cal/g}^{\circ} \text{C} \text{ at } 100^{\circ} \text{C}$
Electrical Resistivity (T6 temper)	5.2 microhm-cm at 20°C

TYPICAL MECHANICAL PROPERTIES:

F _{tu} (O temper)	
(T6, T651 tempers)	$83.0 \text{ ksi} (50.4 \text{ kg/mm}^2)$
F'ty (O temper) (T6, T651 tempers)	$15.0 \text{ ksi} (10.5 \text{ kg/mm}^2)$
(T6, T651 tempers)	73.0 ksi (51.3 kg/mm ²)
e(2 inch, 50.8 mm) (O temper)	
(T6, T651 tempers)	
E (tension)	10.4×10^3 ksi (7.3 x 10^3 kg/mm ⁴)

FABRICATION CHARACTERISTICS:

Weldability	Fusion methods not recommended; resistance methods satisfactory in all heat treated tempers if proper procedures are employed.
Formability	Good in the annealed condition; difficult to form in heat treated tempers.
Machinability	Good in the O, F, or W tempers; more difficult in heat-treated and hardened tempers.

COMMENTS:

A very high strength aluminum alloy with good forming and machining characteristics. Alloy is resistant to stress-corrosion cracking in T73 temper.

SYMBOLS

a A Å AC AMS Ann ASTM Av or Avg	One-half notch section dimension Area of cross section; "A" basis for mechanical property values (MIL-HDBK-5A) Angstrom unit Air cool Aerospace Material Specifications Annealed American Society for Testing Methods Average
B bcc BHN br Btu	"B" basis for mechanical property values (MIL- HDBK-5A) Subscript "bending" Body centered cubic Brinell hardness number Subscript "bearing" British thermal unit(s)
°C	Degree(s) Celsius
CD	Subscript "compression"
CF	Cold drawn
cm	Cold finished
cp	Centimeter
CR	Specific heat
CR	Cold rolled
CW	Cold worked
CVM	Consumable vacuum melted
D or Dia	Diameter
DPH	Diamond pyramid hardness
e	Elongation in percent
E	Modulus of elasticity, tension
E _c	Modulus of elasticity, compression
e/D	Ratio of edge distance to hole diameter
E _s	Secant modulus
E _t	Tangent modulus
eV	Electron volt(s)
°F	Degree(s) Fahrenheit
f	Subscript "fatigue"
Fbru	Bearing ultimate strength
Fbry	Bearing yield strength

fcc	Face centered cubic
FC	Furnace cool
Fcy	Compressive yield strength
Fsu	Shear stress; shear strength
Ftu	Ultimate tensile strength
Fty	0.2% tensile yield strength (unless otherwise indicated)
cy	
g	Gram
g G	Modulus of rigidity
HAZ	Heat affected zone in weldments
hcp	Hexagonal close pack
hr	Hour(s)
HT	Heat treat
IACS	International annealed copper standard
in	Inch
ipm	Inches per minute
0	Derver (a) Kelmin
°K	Degree(s) Kelvin
K	Stress intensity factor; thermal conductivity
Кc	Measure of fracture toughness (plane stress) at point of
	crack growth instability
kg	Kilogram
KIc	Plane strain fracture toughness value
ksi	Thousand pounds per square inch
К _t	Theoretical elastic stress concentration factor
L	Longitudinal
1b	Pouvd
LT	Long transverse (same as transverse)
	Deng transverse (same as transverse)
Μ	Bending moment
m	Meter
М	Subscript "mean"
Max	Maximum
ml	Milliliter
MIL	Military
Min	Minimum
mm	Millimeter
N	Cycles to failure
NSR	Notch strength ratio
NTS	Notch tensile strength
OQ	Oil quench
-	-
ppm	Parts per million
pt	Point; part
-	

r	Radius
RA	Reduction in area; Rockwell hardness A scale
RB	Rockwell hardness B scale
RC	Rockwell hardness C scale
rpm	Revolutions per minute
RT	Room temperature
SA	Solution anneal
sec	Second
S-N	S = stress; N = number of cycles
Spec	Specifications; specimen
ST	Solution treat; short transverse
STA	Solution treated and aged
T	Transverse
t	Thickness; time
Temp	Temperature
typ	Typical
Var	Variable
VHN	Vickers hardness number
W	Width
WQ	Water quench

CONVERSION FACTORS

To Convert	То	Multiply By
angstrom units	millimeters	1×10^{-7}
Btu/lb/ ^o F	cal/g/ ^o C	1
Btu/ft ² /sec/ ⁰ F-inch	cal/g/cm ² /sec/ ⁰ C-cm	1.2404
circular mil	square centimeters	5,067 075 x 10 ⁻⁶
cubic feet	cubic meters	0.028 317
cubic feet/minute	liters/second	0.4720
cubic inches	cubic centimeters	16.387 162
feet	meters	0.304 800 609
foot-pounds	kilogram-meters	0.138 255
gallons (U.S.)	liters	3.785 411 784
inches	millimeters	25.4
ksi (thousand pounds per square inch	kilograms/square millimeter	0.70307
microns	millimeters	0.001
mils	millimeters	0.0254
ounces (avoir.)	grams	28.349 527
ounces (U.S. fluid)	milliliters	29.5729
pounds (avoir.)	kilograms	0.453 592 37
rounds/foot	kilograms/meter	1.488 16
pounds/cubic foot	grams/cubic centimeter	0.016 018 463
square feet (U.S.)	square meters	0.092 903 41
square inches (U.S.)	square centimeters	6.401 625 8
π	E 22) /5 /0)	

Temperature in ${}^{\circ}C = ({}^{\circ}F - 32) (5/9)$ Temperature in ${}^{\circ}K = {}^{\circ}C + 273.15$

Chapter 1

GENERAL INFORMATION

- 1.1 Aluminum alloy 7075 is a high strength, heat-treatable wrought alloy developed by the Aluminum Company of America in 1943. The alloy contains zinc, magnesium, chromium, and copper as hardeners plus small additions of other elements.
- 1.2 Aluminum 7075 responds to an age-hardening heat treatment that produces exceptionally high mechanical properties. This alloy, however, exhibits some degree of notch sensitivity. The alloy has good formability in the annealed and solution-treated conditions at ambient temperatures, and in the T6 condition at elevated temperatures. Alloy 7075 exhibits good machining qualities in the annealed state; little or no warpage occurs during the age-hardening treatments. Its resistance to corrosion is good and improves further with heat treatment and aging. The alloy is resistant to stresscorrosion cracking in the T73 temper. Alloy 7075 can be resistance welded, but fusion welding is generally not recommended. The 7075 alloy is available in a full range of commercial sizes for sheet and plate, extrusions, forgings, bar, rod, wire and tube. Alclad sheet and plate are also available (refs. 1.1 through 1.5).
- 1.3 Typical areas of applications for the 7075 alloy are in aircraft structures, piping systems, mobile equipment, and high pressure hydraulic units.

1.4 General Precautions

- 1.41 This alloy exhibits sensitivity to stress concentration (notch sensitivity), particularly at cryogenic temperatures, and this sensitivity should be recognized in the use of this material.
- 1.42 Overheated material exhibiting eutectic melting or material oxidized at high temperature should not be used and cannot be salvaged by reheat treating.
- 1.43 Quench operations should be performed as rapidly as possible to develop full hardening potential.
- 1.44 Prolonged heating or repeated heat treatments of clad material may cause diffusion of alloying elements into the coating and impair the resistance to corrosion.

Chapter 1 - References

- 1.1 <u>Alloy Digest</u>, "Aluminum 7075" (Filing Code Al-5), Engineering Alloys Digest, Inc., February 1953.
- 1.2 <u>Materials in Design Engineering</u>, Materials Selector Issue, Mid-October, 1964.
- 1.3 Reynolds Metals Co., "The Aluminum Data Book," 1965.
- 1.4 Aluminum Co. of America, "Alcoa Aluminum Handbook," 1962.
- 1.5 <u>Aluminum Standards & Data: 1970-71</u>, Aluminum Association, New York, New York.
- 1.6 <u>Aerospace Structural Metals Handbook</u>, J.G. Sessler and V. Weiss, Eds., AFML-TR-68-115, 1971 Edition.

Cha er 2

PROCUREMENT INFORMATION

- 2.1 General. Aluminum alloy 7075 is available commercially in a fuil range of sizes for sheet, strip, plate, bar, wire, seamless tube, forgings, shapes, and extrusions. Letailed tables of standard sizes and tolerances for the various products available are given in references 2.1 and 2.2.
- 2.2 Procurement Specifications. Specifications that apply to the 7075 alloy as of May 1971 are listed in table 2.2 for various products and tempers.
- 2.3 Comparison of Specifications. Federal procurement specifications are applicable to 7075 extruded bar, rod, shapes and tube; rolled or drawn bar, rod, wire and shapes; bare and clad sheet and plate (also clad one side only); forgings and rivet wire. Military specificatio s apply to sheet and plate (clad one side only), forgings and impact extrusions. ASTM specifications apply to all wrought products except roll-.apered clad sheet and plate, and clad sheet and plate, and clad one-side-only, forging stock and impact extrusions. AMS specifications cover all wrought products except rivet wire.
- 2.4 Major Producers of the Alloy (United States only)

Aluminum Company of America 1501 Alcoa Building Pittslurgh, Pennsylvania

Harvey Aluminum General Offices Torrance, California

Kaiser Aluminum and Chemical Corp. 919 North Michigan Avenue Chicago, Illinois

Olin-Mathieson Chemical Corporation 460 Park Avenue New York, New York

Reynolds Metals Company 6601 West Broad Street Richmond, Virginia

2.5 Available Forms, Sizes, and Conditions

2.51 The available forms, sizes, conditions, and tolerances for various 7075 alloy products are given in detail in references 2.1, 2,2, 2.5, and 2.11.

Source	Refs. 2.3, 2.4, 2.6, 2.7, 2.8, 2.9, 2.10					
Alloy	7075					
Product	Temper	Military	Federal	ASTM	AMS	NASA- MSFC
Sheet and plate	O T6 T651 F		QQ-A-250/12d QQ-A-250/12d QQ-A-250/12d QQ-A-250/12d	B209-71 B209-71	4045E	
Clad sheet and plate (both sides)	O T6 T651		QQ-A-250/13d QQ-A-250/13d QQ-A-250/13d	B209-71	4049F	
Clad sheet and plate (one side only)	T651 T6	MIL-A-8902 - MIL-A-8902	QQ-A-250/18d QQ-A-250/18d QQ-A-250/18d	B209-71		
Clad sheet and plate (roll taper)	те				4047B	
Impact extrusions		MIL-A-12545B MIL-A-12545B	1		4170	
Rivet wire	O, H13	-	QQ-A-430a	B316-70		
Die forgings	T652	MIL-A-22771B MIL-A-22771B	QQ-A-367g	B247-70	4139F	144B 144B 144B
Hand forgings	T652	MIL-A-22771B MIL-A-22771B MIL-A-22771B	QQ-A-367g			144B 144B 144B
Forging stock	Т6				4139F	
Bar, rod, shapes, tubes (extruded or CF)	O F T6 T651		QQ-A-200/11c QQ-A-200/11c	B221-71		389 389 389
	T6510 T6511 T73		QQ-A-200/11c QQ-A-200/11c			389
Bar, rod, wire, shapes (rolled or drawn)	∩ T6 T651 T73		QQ-A-225/9c QQ-A-225/9c QQ-A-225/9c QQ-A-225/9c	B211-71 B211-71 B211-71		

TABLE 2.2. - Procurement Specifications (a)

(a) Specified as of May 1971

Chapter 2 - References

- 2.1 <u>Aluminum Standards & Data: 1970-71</u>, 2nd Edition, The Aluminum Association, New York.
- 2.2 Aluminum Co. of America, "Alcoa Aluminum Handbook," 1962.
- 2.3 Aluminum Co. of America, "Alcoa Product Data Specifications," Section Al2A, July 1963.
- 2.4 1971 SAE Han look, Society of Automotive Engineers, 1965.
- 2.5 Harvey Aluminum, "Mill Products Alloys," HA Form B-667-3R.
- 2.6 <u>SAE Aerospace Materials Specifications</u>, Society of Automotive Engineers, latest Index, May 15, 1971.
- 2.7 Index of Specifications and Standards, Department of Defense, Part I, Alphabetical Listing, and Part II. Numerical Listing, July 1970, supplement May 1971.
- 2.8 ASTM Standards, Part 6, "Light Metals and Alloys, 1971.
- 2.9 MSFC-SPEC-144B, "Aluminum Alloy Forgings, Premium Quality, Heat Treated," August 13, 1963; Amendment 1, September 8, 1964; Custodian: NASA/Marshall Space Flight Center.
- 2.10 MSFC-SPEC-389, "Aluminum Alloy, Bars, Rods, Wire and Special Shapes, Rolled, Drawn, Extruded or Cold Finished, 7075," May 28, 1964; Custodian: NASA/Marshall Space Flight Center.
- 2.11 Olin Aluminum, "Olin Aluminum Mill Products and Casting Alloys," 1970.

Chapter 3

METALLURGY

- 3.1 Chemical Composition
- 3.11 Nominal chemical composition of 7075 alloy, in percent (ref. 3.1).
 - Zn
 5.6

 Mg
 2.5

 Cu
 1.6

 Cr
 0.3

 Al
 Balance
- 3.111 Sheet and plate are available in the Alclad condition. Cladding material is 7072 alloy; nominal composition, in percent:

Zn	1.0
Al	Balance

Cladding may be applied to both sides or to one side only. The nominal cladding thickness per side is 4 percent of the total composite thickness if the latter is 0.062 inch or below, 2.5 percent if the total thickness is between 0.062 and 0.187 inch and 1.5 percent if the total thickness is 0.188 inch or over. For thicknesses of 0.500 inch and over with 1.5 percent cladding, the average maximum thickness of cladding per side after rolling to specified plate thickness will be 3 percent of the plate thickness, as determined by averaging cladding thickness measurements taken at 100-diameter magnification on the cross section of transverse samples polished and etched for microscopic examination (refs. 3.1, 3.2). (Note: 1 inch = 25.4 mm.)

3.12 Chemical composition limits, in percent (ref. 3.1).

Zn Mg	5.1 - 6.1 2.1 - 2.9
Cu	1.2 - 2.0
Cr	0.18 - 0.35
Fe	0.5 max
Si	0.4 max
Mn	0.3 m a x
Ti	0.2 max
Others	
Each	0.05 max
Total	0.15 max
Al	Balance

Conformity with these composition limits is normally checked by spectrochemical analysis or in accordance with the procedure outlined in ASTM E34, "Standard Methods for Chemical Analysis of Aluminum and Aluminum Base Alloys" (ref. 3.1).

- 3.13 <u>Alloying Elements</u>. The principal alloying elements are zinc, magnesium, copper, and chromium, with lesser amounts of Fe, Si, Mn, and Ti. The aluminum-rich portions of the binary equilibrium diagrams for each of these principal elements are given in figures 3.1 and 3.2. The principal hardening constituent is a Zn-Mg phase. The particular combination of zinc, magnesium, and copper, with the addition of 0.3 percent chromium, was selected for this alloy to give very high strength and good resistance to stress-corrosion cracking (ref. 3.4). The amount of protection provided by cladding depends on the thickness and purity of the cladding material, and also on the heat treatments employed.
- 3.2 Strengthening Mechanism
- 3.21 <u>General</u>. The alloy is strengthened by precipitation hardening and cold work. Upon quenching from the solution temperature to room temperature, a Zn-Mg phase precipitation occurs in the form of submicroscopic particles which are obstacles to plastic flow and thus cause hardening. The major precipitating constituent has been identified as MgZn₂ (ref. 3.5).
- 3.22 Heat Treatment (refs. 3.1, 3.2, 3.6)
- 3.221 Anneal (O condition). All products: heat to 413° to 454° C, hold 2 to 3 hours, air cool, follow by heating to 232° C for about 6 hours (ref. 3.6).
- 3.2211 Anneal to remove cold work. All products: heat to 349°C. Time at temperature and cooling rate are not critical (ref. 3.2).
- 3.222 Solution Treatment (W condition) (ref. 3.6)

Rolled or drawn products: Heat to 460° to 499°C, hold 10 minutes to 1 hour in salt bath, or longer time in air; for heavy sections, quench in cold water. Sheet under 0.051 inch (1.295 mm) should be solution treated at 488° to 499°C.

Extruded products: Heat to 460° to 471°C, hold 10 minutes to 1 hour in salt bath, or longer time in air; for heavy sections, quench in cold water.

Forged products: Heat to 460° to 477°C, hold 10 minutes to 1 hour in salt bath, or longer time in air; for heavy sections, quench in cold water.

Recommended soaking times for solution heat treatment of all wrought products are given in table 3.1. Maximum allowable quench delay times range from 5 to 15 seconds, depending on thickness (ref. 3.6).

3.2221 Caution should be exercised in the control of the solution treating temperature. If the temperature is too high it may cause solid solution grain boundary melting which cannot be corrected by subsequent heat treatment operations. Low temperature may result in incomplete solution of the hardening constituents with a loss in hardening potential.

- 3.223 Precipitation Treatment (T6 condition). Heat solution treated material to 110° to 127°C, hold 22 hours minimum. Cooling rate is not critical (ref. 3.6).
- 3.224 Other Treatments. The alloy can also be hardened by cold work, but this procedure is not generally used to develop strength in commercial tempers except for rivet wire (H13 condition). Cold work, however, is employed for stress relief and for straightening. The available tempers and treatments employed for various products are listed in table 3.2.
- 3.3 Critical Temperatures. Melting range is approximately 477 to 638° C.
- 3.4 Crystal Structure. Face-centered-cubic matrix. The lattice parameter of aluminum decreases with the addition of zinc from $a_0 = 4.0410 \times 10^{-7}$ at 0% zinc to 4.030 x 10⁻⁷ mm at 12.2 atomicpercent zinc (ref. 3.4).
- 3.5 <u>Microstructure</u>. Figure 3.3 illustrates the grain structure of 7075-T6 rolled rod. The grain structure of thick plate is shown in figure 3.4.

Hot working causes a breaking up and distribution of constituents. For example, CrAl, segregation may occur if the Cr content is too high or the ingot casting temperature is too low. The segregation of this hard brittle compound can cause cracking, low strength, and difficulty in machining (ref. 3.5).

References 3.5 and 3.7 are recommended as excellent sources of information on the identification of constituents and phases in aluminum alloys.

3.51 Metallographic Procedures: In general, mechanical polishing is preferred to electropolishing, especially where larger microconstituents are present and the material is relatively soft, as objectionable relief effects produced by the electrolytic polishing technique may cause a misinterpretation of the microstructure (ref. 3.5). For homogeneous alloys, and for those conditions containing only finely dispersed particles, the electrolytic method is excellent. Preparatory polishing on metallographic polishing papers 0 to 000 should be performed wet with a solution of 50 g paraffin in 1 liter kerosene to keep the specimen bright and avoid imbedding of grinding compound particles into the soft specimen surface. Rough polishing on a "Kitten's Ear" broadcloth at 250 to 300 rpm with suspended 600 grade aluminum oxide and final polishing on a similar wheel at 150 to 200 rpm with heavy magnesium oxide powder is recommended (refs. 3.4, 3.7).

An alternate and popular method consists of the following steps:

- (a) Wet polishing (flowing water with 240 grit silicon carbide paper at approximately 250 rpm.
- (b) Wet polishing with 600 grit silicon carbide paper at approximately 250 rpm.
- (c) Polishing with 9-μm diamond paste on nylon cloth at 150 to 200 rpm using a mild soap solution for lubrication.
- (d) Final polish on a vibratory polisher using a microcloth containing a slurry of methyl alcohol and 0.1-μm aluminum oxide powder. A slurry of 0.1-μm aluminum oxide powder in a 10% solution of glycerine in distilled water may also be used for this step.

Etching reagents should be suited to the objective of the study. Keller's etch reveals microstructural details and grain boundaries satisfactorily. A 10% solution of NaOH gives better details of the microstructural constituents but does not delineate the grain boundares. Study of the "as polished" surface prior to etching may also give valuable information on the types of constituents present, especially when attention is paid to the colors of the various particles. Macroscopic studies of cracks, gross defects, forging lines, and grain structure should be made with the etching solutions given in table 3.3. Etching reagents for revealing microstructure are listed in table 3.4.

Source	Ref. 3.6			
	Soaking time, minutes (a)			
Thickness, inches (b,f)	Salt Bath		Air Furnace	
	min	max (alclad only) (e)	min	m ax (alclad only) (e)
0.016 and under 0.017 to 0.020 incl. 0.021 to 0.032 incl. 0.033 to 0.063 incl. 0.064 to 0.090 incl. 0.091 to 0.125 0.126 to 0.250 incl. 0.251 to 0.500 incl. 1.001 to 1.500 incl.	10 10 15 20 25 30 35 45 60 90	15 20 25 30 35 40 45 55 70 100	20 20 25 30 35 40 50 60 90 120	25 30 35 40 45 50 60 70 100 130
1.501 to 2.000 incl. 2.001 to 2.500 incl. 2.501 to 3.000 incl. 3.001 to 3.500 incl.	105 120 150 165	115 130 160 175	150 180 210 240	160 190 220 250
3.501 to 4.000 incl.	185	190	240	280

TARLE 3.1. - Soaking Time for Solution Treatment of 7075 Products

- (a) Longer soaking times may be necessary for specific forgings. Shorter soaking times are satisfactory when the soak time is accurately determined by thermocouples attached to the load.
- (b) The thickness is the minimum dimension of the heaviest section.
- (c) Soaking time in salt-bath furnaces should be measured from the time of immersion, except when, owing to a heavy charge, the temperature of the bath drops below the specified minimum; in such cases, soaking time should be measured from the time the bath reaches the specified minimum.
- (d) Soaking time in air furnaces should be measured from the time all furnace control instruments indicate recovery to the minimum 9th process range.
- (e) For alclad materials, the maximum recovery time (time between charging furnace and recovery of furnace instruments) should not exceed 35 minutes for gages up to and including 0.102 inch, and 1 hour for gages heavier than 0.102 inch.
- (f) 1 inch = 25.4 mm.

ſ											
bare and Clad (0/5 Products	Refs. 3.1, 3.2	Description of Treatment to Produce Indiated Temper		See chapter 3, section 3.221	See chapter 3, section 3.223	T6 + stress relief by stretching (a)	T6 + stress relief by compression	Cold worked to 3/8 hz d condition	Proprietary thermal treatment	Stress relief by stretching (b)	Stress relief by stretching (c)
e an(Forging Stock	٤ų						T73		
		Hand Forgings			T 6		T652		T73		
l'reatments lor		əduT (bəburtxə)		0	T6				T73	T6510	T6511
reatme		səqshi) Shapes	Ŀц	0	T6				T73	T6510	T6511
and		Rod and Bar (extruded)	۲ų	0	T6				T73	T6510	T6511
lempers		agnif bəllof			T6				T73		
I em		Rivet Wire and Rod		0				H13			
2.		Jube and Wire (drawn)		0	T 6						
IABLE 3.2.		Rod and Bar (rolled or CF)	ſщ	0	T6	T651			T73		
ЧT	eo.	Plate			T6*	T651*			T73		
	Source	təədz		* 0	T6*						

- Tempers and Treatments for Bare and Clad 7075 Products 0 3 TABLE

Also available as Alclad on both sides or on one side only (a) *

5 to 3 percent for sheet and plate
 1 to 3 percent for rod, bar, shapes, and tube
 0.5 to 3 percent for drawn tube
 No further straightening after stretching
 Minor straightening after stretching to comply with standard tolerances.

(a) (a)

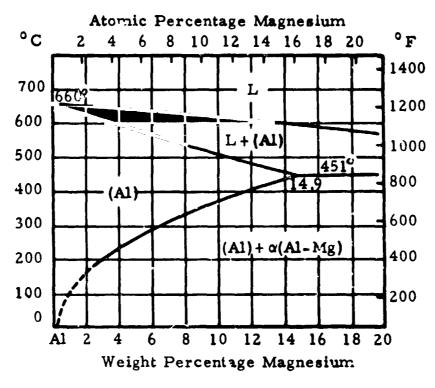
Source	R.of. 3.4				
Alloy	7075				
Solution	Concentration (a)		Specific Use		
Sodium Hydroxide	NaOH Water	10 g 90 ml	For cleaning surfaces, revealing un- soundness, cracks, and gross defects		
Tucker's	HCl (conc.) HNO ₃ (conc.) HF (48%) Water		For revealing structure of castings, forgings, etc.		
Modified Tucker's	HCl (conc.) HNO ₃ (conc.) HF (48%) Water	10 ml 10 ml 5 ml 75 ml	For revealing structure of all castings and forgings except high silicon alloys		
Flick's	HCl (conc.) HF (48%) Water	15 ml 10 ml 90 ml	alumin type alloys. Surface should be		

TABLE 3.3. - Etching Solutions for Revealing Macrostructure

(a) All of these solutions are used at room temperature

Source		Ref. 3.5		
Alloy		7075		
Composition		Uses	Remarks	
NaOH Water	1 g 99 ml	General microstructure	Swab with soft cotton for 10 seconds	
NaOH Water	10 g 90 ml	General microstructure (micro and macro)	Immerse 5 seconds at 71°C, rinse in cold water	
Keller's (conc.) HF (conc.) HCl (conc.) HNO ₃ (conc.) Water	10 ml 15 ml 25 ml 50 ml	General microstructure (micro and macro) for copper bearing alloys	Use concentrated for macroetching; dilute 9:1 with water for micro- etching	
Keller's (dilute) HF (conc.) HCl (conc.) HNO ₃ (conc.) Water	l ml 1.5 ml 2.5 ml 380 ml	General microstructure of 7075 alloy	Etch for 5 seconds	

TABLE 3.4. - Etching Reagents for Revealing Microstructure



Aluminum - Magnesium

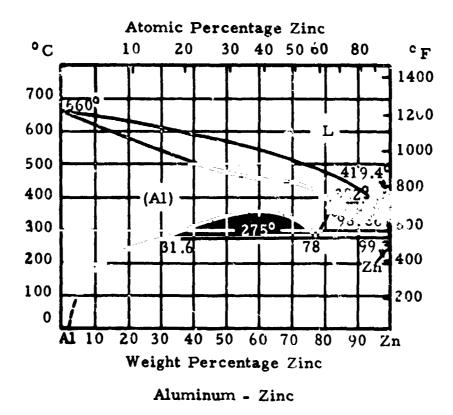
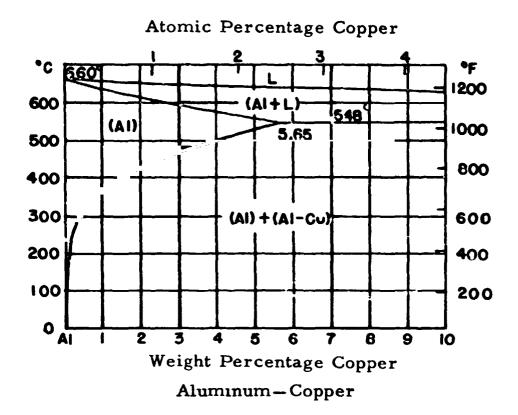
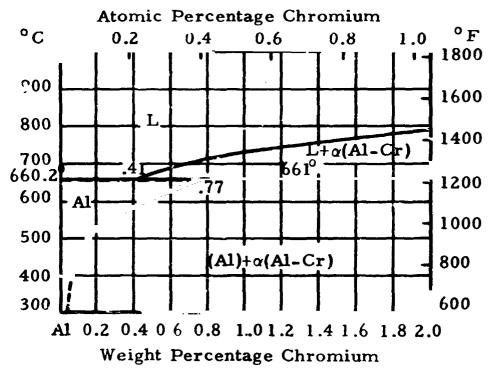


FIGURE 3.1. - Aluminum rich portion of binary equilibrium diagrams.

(Ref. 3.3)





Aluminum - Chromium

FIGURE 3.2. – Aluminum-rich portion of binary equilibrium diagrams.

(Ref. 3.3)

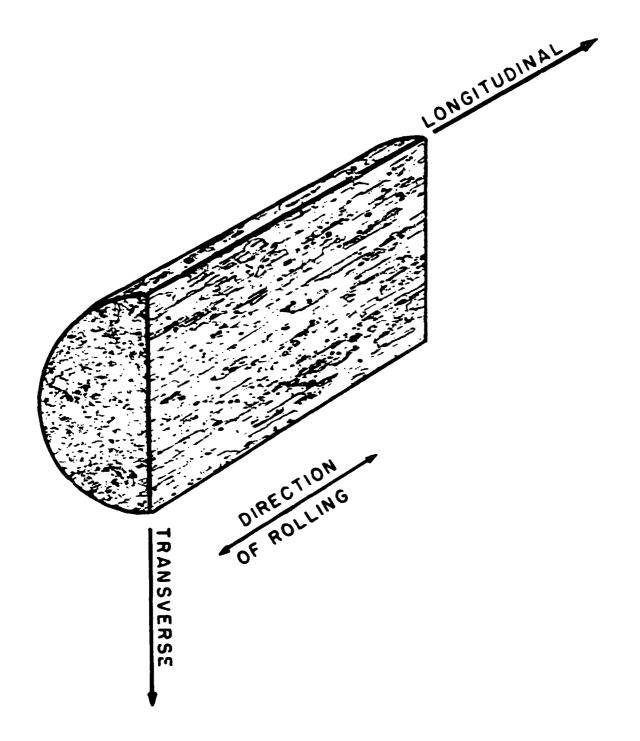


FIGURE 3.3. - Composite micrograph illustrating the grain structure of a 1-in. (25.4-mm) diameter rolled rod of 7075-T6 alloy. The relatively long and equiaxed crosssection grains are typical of rolled rod and bar of round, hex, or square cross section.

Etch: Keller's

Mag: 100X

(Courtesy Aluminum Co. of America)

(Ref. 3.8)

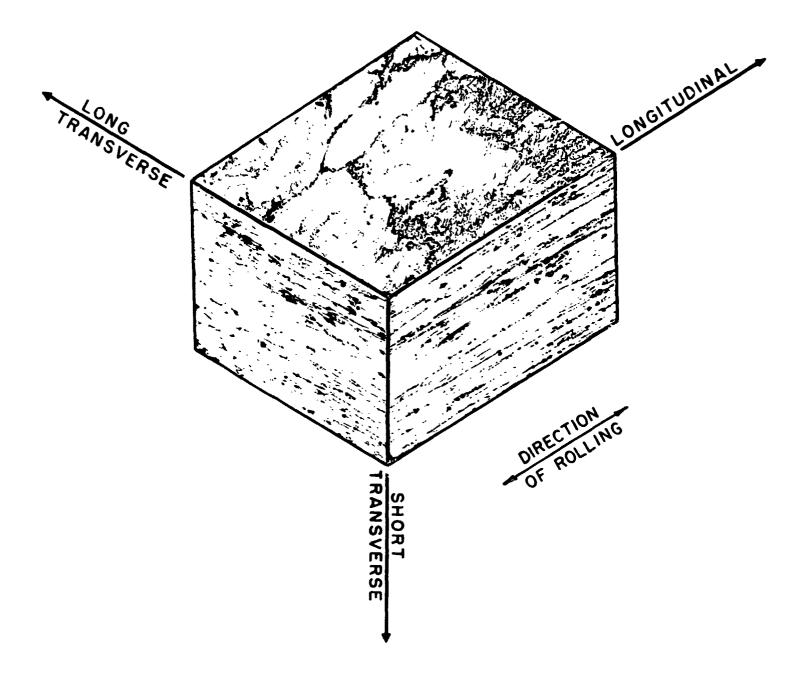


FIGURE 3.4. -Composite micrograph illustrating the grain structure of 11/2-in. (38.1 mm) thick plate of 7075-T6 alloy. The relatively long, wide, and thin unrecrystallized grains are typical of the grain structure of thick plate of other alloys also.

Etch: Keller's

Mag: 100X

(Courtesy Aluminum Co. of America) (Ref. 3.8)

Chapter 3 - References

- 3.1 <u>Aluminum Standards and Data: 1970-71</u>, 2nd Edition, The Aluminum Association, December 1969.
- 3.2 Reynolds Metals Co., "The Aluminum Data Book," 1965.
- 3.3 E.H. Wright and L.A. Willey, "Aluminum Binary Equilibrium Diagrams," Technical Paper No. 15, Aluminum Co. of America, 1960.
- 3.4 W.L. Fink et al., <u>Physical Metallurgy of Aluminum Alloys</u>, American Society for Metals, Cleveland, Ohio, 1958.
- 3.5 J.P. Vidosic, "Study of Phase Identification in Steel and Aluminum Alloys," Georgia Institute of Technology, Final Report, NASA Contract NAS8-5117, September 1963.
- 3.6 Military Specification, "Heat Treatment of Aluminum Alloys," MIL-H-6088E, February 1971.
- 3.7 F. Keller and G.W. Wilcox, "Identification of Constituents of Aluminum Alloys," Technical Paper No. 7, Aluminum Co. of America, 1942; revised 1958.
- 3.8 D.O. Sprowls, "Resistance of Wrought, High-Strength Aluminum Alloys to Stress Corrosion," Technical Paper No. 17, Aluminum Co. of America, 1962.

Chapter 4

PRODUCTION PRACTICES

4.1 General. In the United States, aluminum and its alloys are produced from an ore of impure hydrated aluminum oxide known as "bauxite." Important sources of bauxite are located in Arkansas, Dutch Guiana, and Jamaica. The impure ore is converted into pure aluminum oxide (alumina) through a series of chemical processes. Oxygen is removed from the alumina by smelting in carbon-lined electric furnaces known as reduction pots. Pure molten aluminum is deposited at the bottom of the pot, and is periodically siphoned off and poured into molds to form "pigs" and "sows." A separate furnace operation is used to form "alloy pig" from the pure aluminum by the addition of alloying elements, and this metal is cast into ingots for further processing (ref. 4.1).

For the 7075 alloy, the major alloying elements added are zinc, magnesium, and copper, plus a small addition of chromium. Generally, this phase of production practice involves carefully controlled melting, alloying, and casting of large 20,000- to 50,000-pound (\approx 9,000 to 23,000 kg) ingots. After the ingots are scalped and preheated in vertical electric soaking pits, they are ready for further processing to a particular form of product.

4.2 Manufacture of Wrought Products

- 4.21 Bar and rod are normally produced by hot rolling or extruding. Cold finished bar and rod are produced by hot working to a size slightly larger than specified and reducing to final dimensions by cold working. A better surface finish and closer dimensional tolerances are obtained in this manner (ref. 4.2).
- 4.22 A similar process is used to produce rolled structural shapes; special rolls are required. Finishing operations include rolleror stretch-straightening, and heat treatment.
- 4.23 Roll-form shapes are produced by passing strip through a series of roller dies. Each successive pair of rolls cause the work to assume a cross-section shape more nearly approaching that desired. The final desired shape is produced at the last pair of rolls.
- 4.24 Plate is produced by hot rolling of ingots to slabs (approximately 60 percent reduction), usually in a 4-high reversible mill. The slabs are then further reduced 50 percent in a reversible 2-high mill. The last stage of hot rolling is done in a hot reversing mill, where the plate is progressively rolled to the final hot mill

dimensions. Alloy plate may be subjected to "stress relief" stretching (about 2 percent permanent set) after solution treatment to improve flatness and reduce warpage upon machining. Plate is then sheared or sawed to the required dimensions (ref. 4.2).

- 4.25 Sheet is usually produced from plate by cold rolling to final sheet thickness, followed by trimming, tempering, heat treating, stretching, and other finishing operations.
- 4.26 Wire is produced by drawing rod through a series of progressively smaller dies to obtain the desired dimensions.
- 4.27 Extrusions are produced by subjecting reheated cast billets to enough pressure to force the metal to flow through a die orifice, forming a product whose cross-section shape and size conforms to that of the orifice. Speeds, pressures, and temperatures must be closely controlled to insure uniform quality of extruded products.
- 4.28 Tube is produced by extruding, by drawing, or by welding. Extruded tube is forced through an orifice as described in 4.27; a die and mandrel are used. Drawn tube is manufactured by a cold process which is similar to drawing bar and rod. A mandrel is used with one end fixed and a bulb attached to the other end; the tube is drawn over the mandrel bulb and through a die at the same time. Welded tube is produced by slitting coil stock into strips and passing the strips through a series of rolls to form tube; the longitudinal seam is welded as the tube leaves the last roll-forming station.
- 4.29 Forgings are made by pressing (press forging), hammering (drop forging), and impacting. Relatively heavy equipment is required since aluminum is not as plastic at its forging temperature as steel. Aluminum forgings compare favorably with structural steel in unit strength at about one-third the weight. With comparable strength and with a lower elastic modulus, aluminum alloys have a much higher impactenergy-absorbing capacity than a mild steel.

4.3 Casting of Alloy Ingots

4.31 Metal for wrought products is alloyed in large 10- to 25-ton double hearth furnaces, carefully controlled and instrumented. The direct chill (DC) method is generally used for casting these ingots. Molten metal is poured into a mold and a hydraulic piston descends slowly as the metal solidifies. Water is sprayed on the outside of the mold to promote rapid solidification. Additional processing may include scalping (machining of outside surfaces) or homogenizing (refs. 4.2 and 4.3).

Chapter 4 - References

- 4.1 Kaiser Aluminum and Chemical Sales, Inc., "Kaiser Aluminum Sheet and Plate Product Information," Second Edition, January 1958.
- 4.2 Reynolds Metals Co., "The Aluminum Data Book: Aluminum Alloys and Mill Products," 1958.
- 4.3 Aluminum Co. of America, "Alcoa Aluminum Handbook," 1962.

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Chapter 5

MANUFACTURING PRACTICES

5.1 General. This heat-treatable alloy, containing zinc, magnesium, copper, and chromium, has applications in highly stressed parts and air-frame structures. In general, 7075 is used where very high strength and good resistance to corrosion are required (ref. 5.6) and is available as the base alloy and in the Alclad condition. It is produced in most of the wrought forms with the exception of pipe, structural shapes, and foil; in the Alclad form, it is available as sheet and plate (ref. 5.1).

5.2 Forming

- 5.21 Sheet and plate. The relative formability of 7075 is not as good as many of the other heat-treatable alloys. Regular methods can be used; however, more care and precision fabricating techniques are required (ref. 5.4). Relative formability compared with other heat-treatable alloys can be noted in table 5.1.
- 5.211 Cold forming. The formability of alloy 7075 sheet and plate is directly related to the temper strength and ductility. In producing complex parts, the procedure is to use annealed sheet and to heat treat after forming. Because of its extra strength and hardness, 7075-T6 is relatively difficult to form. The simplest and most widely used forming method is probably that of bending. The ease of bending is indicative of most other forming operations. Table 5.2 indicates the ease of forming in terms of recommended minimum bend radii as a function of temper and sheet and plate thickness, using typical mechanical properties for 0.100-inch (2.54-mm) sheet. Aluminum sheets are normally formed using operations such as:
 - 1. Bending
 - 2. Flanging
 - 3. Rolling
 - 4. Drawing
 - 5. Pressing
 - 6. Stretching
 - 7. Embossing
 - 8. Coining

- 9. Stamping
- 10. Spinning
- 11. Contour Forming
- 12. Bulging
- 13. Beading and Roll Flanging
- 14. Necking
- 15. Curling

The factors that influence the bending of 7075 sheet also influence the fourteen other forming operations in the same general manner. Because of the lower modulus of elasticity of aluminum compared with steel, a much greater "springback" is encountered. Overforming is the common way of correcting the tendency. In addition, reducing the bend radius, increasing sheet thickness, and increasing the total amount of plastic deformation also decrease the extent of of springback. Alloy 7075 sheet can be formed to many shapes by drawing if care is used. This extensively employed mass production method can be used to produce simple parts in a single draw. In the case of more complex parts, the reduction and forming is accomplished by using successive draws with frequent intermediate anneals. By this practice, exhaustion of ductility and the introduction of cracks are avoided. Deep draws normally employ male and female metal dies. Forming in rubber (Guerin Process) for relatively shallow parts is a method where several thin layers of rubber are confined in a pod holder or retainer made of steel or cast iron. A descending ram on which this holder is mounted causes the aluminum sheet to be compressed against a form block to make the required part. If the aluminum is made to flow against a female die using fluid pressures behind a rubber diaphragm, the method is known as "hydroforming."

Alloy 7075 has been used for applications such as engine nacelle covers and contoured wing skin panels.

- 5.212 Hot forming. Although hot forming from 300° to 400° F (149° to 204°C) is used to ease forming for many aluminum alloys, 7075 which responds to artificial aging may actually require more power for forming because of the strengthening which takes place at the elevated temperature (ref. 5.4). Many successful forming methods have been developed where the metal is heated in the area to be formed. However, this practice can lead to undue softening of the material unless certain precautions are exercised. The maximum reheating periods which are recommended are given in table 5.3. The effect of forming temperature on springback of sheet in T6 condition is shown in figure 5.1. The effect of forming at elevated temperatures on room temperature tensile properties of plate in O condition prior to forming, and T6 condition after forming, is shown in figure 5.2. Figure 5.3 indicates the effect of forming temperature on bend factor in rubberforming of sheet.
- 5.22 Shapes and tubes. Either extrusion or rolling can be used to produce aluminum shapes. The relative formability of alloy 7075 as tubes or extrusions can be noted in table 5.4. This alloy, as has been pointed out, is one of the more difficult to form of the aluminum alloys. Sections in the O temper or W temper are bent and formed more easily than those in the T6 or T73 heat treated tempers.

Stretching, wiping, or rolling are general methods used to form shapes and tubes. Sheets, shapes, and tubes are stretch-formed by clamping at one end and pulling or stretching over a single male die so as to exceed the elastic limit. The metal section takes the shape of the die by stretching or elongating more in the heavier curvature areas than in the shallower ones. When working exceptionally thin-wall round, square, or rectangular tube on small radii, it is necessary to add a wiper and a flexible mandrel to provide extra support for the tube at the point of bending. Rolls can also be used for the forming, using dies to form the contour. 5.23 Forging. The very high strength 7075 is more difficult to forge than most other aluminum alloys. The high strength-to-weight ratio makes this alloy desirable for jet landing gear and other similar applications.

Forgings are made using either the open die or closed die methods and by impact or pressure. Prototype or other fevr-of-a-kind needs for aluminum parts usually do not warrant the cost of forging dies. Small runs are made using the hand-forging open-die techniques where the heated stock is worked between flat or simple dies that impose little or no lateral confinement on the material. Hand forgings over a ton in weight can be made. Hand forgings are provided in various tempers which are defined in table 5.5.

As in all forgings there is grain flow in 7075 which is cha acteristic of the forging process. The resultant grain pattern results in anisotropy of properties and this must be considered for property evaluations. The process for most production for aga starts with the stock which can vary from 3/8 inch to 4 inclus square stock, and rectangles from 3/8 inch for the minimum dimension to as r such as 10 inches on the maximum dimension. Conditioning to remove localized surface defects is permitted at this point. (1 in = 25.4 mm)

The stock is carefully heated in the reason of 600° to 900° F (316° to 482° C). The relative forgeability control is a function of the forging temperature can be compared to the start aluminum allows in figure 5.4. It can be seen that this alloy is the most difficult to forge. The mechanical properties are a function of the forging direction as well as the size of the hand forging. Figure 5.5 shows the test-bar orientation as a function of the principal directions.

Large production runs are made using closed dies. The cost of the die is prorated against the number of pieces contemplated. Either drop forging or press forging machinery is used. After preheating, the stock is formed in one step or in the case of complicated parts in several operations which may involve reheatings. Dies in the forging operation are heated with auxiliary gas or electric heaters. The flash resulting from excess metal overfilling the mold is removed by hot or cold trimming, sawing, or grinding.

Holes in the forging are pressed to produce "punchouts." Sometimes the punchout is combined with the trim operation. Very close tolerances can be met in a standard forging by die coining (colú) to precise dimensions, usually within a few thousands of an inch (tenths of a millimeter).

Straightening after heat treatment is often a required operation. Templates combined with indicators and other gages are used to determine the out-of-tolerances. Straightening ranges from hand straightening to "cold restrike" operations. The forgings are inspected for grain flow, mechanical properties, dimensions, and ultrasonic soundness. Design manuals for die forgings and aluminum impacts are available from the Aluminum Association (refs. 5.17, 5.18).

5.3 <u>Machining</u>

Conventional machining. The aluminum alloy 7075 has good machinability in all conventional machining operations. Hand forgings of 7075 which require a large amount of metal removal by roughing out before heat treatment should be machined in the F-temper. In those cases where hand forgings are to be machined to very close dimensions, with the additional requirement for a good surface condition, the W temper yields optimum results. Small hand forgings can be machined successfully in the T6 temper.

It is difficult to produce a precise tabulation of machining parameters for each of the different types of operations. However, table 5.6 is a compilation of typical fa^{tors} for many common machining operations. A wheel speed of 6000 ft/min and a table speed of 60 ft/min is typically used for grinding. The down feed will produce a rough finish if it is kept about 0.001 inch(0.025 mm) per pass. A fine finish will be produced if the down feed is kept to a maximum of 0.0005 inch (0.013 mm) per pass. The cross-feed is approximately one-third of the wheel width. The wheel type is A46KV with a water-base emulsion or chemical solution for the grinding wheel.

The 7075 alloy has been machined to produce contoured-wing skin panels by a huge three-gantry machine (ref. 5.13). Three-inch slabs (7.62-cm), 4 to 5 fect wide by 6 to 12 feet in length (1.2 to 1.5 m by 1.8 to 3.6 m), were first straightened by "stretcher-leveling" to the 7075-T6 condition. This treatment minimized distortion from the skin-milling. The entire operation of producing 2.5-inch (6.35-cm) deep cavities was numerically controlled. The vacuum held slabs produced the required section with excellent tolerances.

- 5.32 Electrochemical and chemical machining. Weight reductions such as the cavity machining by slab milling require large rigid machines. These weight reductions are important for space vehicle components, particularly in large boosters, where the fuel and oxygen tanks are fabricated from precurved cylindrical and spherical sections of high-strength aluminum alloys. The use of sections which are "integrally stiffened" by ribs which are left intact while the bulk of the metal stock is removed has been examined for both electrochemical and chemical methods. In some cases, chemical milling will allow early production of initial units without requiring the delay times inherent in the fixturing for production (ref. 5.14).
- 5.322 Electrochemical milling. Electrochemical machining for metal shaping subjects the chemically erodible workpiece to the action of anodic current flow in a suitable electrolyte. A second electrode which is the tool is provided for the cathodic action. The basic principles are the same as those generalized in Faraday's Law of Electrolysis. However, the

electrochemical machining, or ECM, process is the reverse of electrodeposition or electroplating. An exception is that the cathodic process involves the evolution of hydrogen, in most cases, rather than the electrodeposition of a metal. There are a number of tool workpiece configurations that may be employed in the ECM process depending upon the particular type of metal removal geometry desired. It is normally required that fresh electrolyte is supplied to the workpiece. Alloy 7075 is essentially pure aluminum as far as the rate of the electrochemical process is concerned. Hence, from the Faraday Law it is rather easily shown that 1.26 in^3 (20.64) cm³) of the metal can be removed per minute at 100,000 amperes (assuming 100% efficiency). In practice, efficiencies of 80% to 90% are encountered. An electrolyte of 5 to 10% NaCl solution has been found to yield excellent results, and the process can be carried out using voltages of 10 to 15 volts. The milling rate of the ECM process depends upon the current capacity of the power supply and the ability of the electrolyte system to provide irech electrolyte. High electrolyte pressure requirements of 100 to 250 psi (0.07 to 0.18 kg/mm²) provide even electrolyte flow and satisfactory cutting conditions. Temperatures of about 120° F (49° C) produce good quality finishes.

5.323 Chemical milling. The removal of metal stock by chemical dissolution or "chem-milling" has many potential advantages over convertional milling methods. The removal of metal by dissolving in an alkaline or acid solution is now routine for specialized operations on aluminum (ref. 5.6). For flat parts, on which large areas having complex or wavy peripheral outlines are to be reduced only slightly in thickness, chemical milling is usually the most economical method. A formed channel 23 feet (7 m) long made of 0.125-inch (3.175-cm) aluminum 7075-T6 had a scalloped edge and recessed pockets and a lengthwise selant groove. Chemical milling with caustic soda allowed production schedules for initial wing-beam unit: to be met before conventional router tooling could be fabricated (ref. 5.14). The metal (suitably masked) is immersed in an etching bath, which may be acidic or basic, to remove metal from specific areas so as to produce the desired configuration. Finally, the mask is stripped from the part. To produce a simple shape, the masking and milling procedure is only performed once. Complex designs are usually produced by repeating the masking and milling sequence until the desired shape is achieved.

Standard mechanical property tests indicate that chemical milling has appreciable effect on the compression, tension, or shear properties of aluminum alloy 7075 (ref. 5.11). Fatigue tests on 7075-T6, performed at high stress levels, show more favorable results for chemically-milled material than for machine-milled material.

Alloy	Alloys in Order of Decreasing Formability
Source	Ref. 5.2
Rating	Alloy
ここうせら ら て	No. 21 and No. 22 (Brazing Sheet) 6061 6066 2024 2014 2014 7178 7178

TABLE 5.1. - Relative Formability of Heat-Treatable

TABLE 5.2. .- Approximate Bend Radii for 90 Degree Cold Bend (a, c)

Source						4	Ĩ.	Ref. 5.1						
Alloy							7075	10						
Temper	Ftu,	ksi	F _{ty} ,	Ftu, ksi F _{ty} , ksi	e (b)				Thic	Thickness, t, inches	inche			
	Min	Max	Min	Max	Min		49	1/32	1/16	Min 1 /64 1/32 1/16 1/8	3/16	1/4	3/16 1/4 3/8 1/2	1/2
0	8	40	1	21	10			0	0-1t	0 0-1t 1/2t-11/2t 1t-2t 11/2t-3t 21/2t-4t 3t-5t	1t-2t	11/2t-3t	21/2t-4t	3t-5t
T6 (d)	77	1	66	•	80	2t-	4t	3t-5(2t-4t 3t-5t 4t-6t					7t i2t
							1							

(a) Radii for various thickness expressed in terms of thickness, t. (1 inch = 25.4 mm)
(b) Elongation, percent in 2 inch or 4D
(c) Mechanical properties are minimum or maximum for thickness below 0.200 inch
(1 ksi = 0.70307 kg/mm²
(d) Alciad sheet can be bent over slightly smaller radii than the corresponding tempers of the uncoated alloy

Source	R	lef. 5.4
Alloy	70	75-T6
Tempera ^o F	ture of Hold	Time in Indicated Units
300	149	10 - 12 hours
325	163	2 - 4 hours
350	177	l - 2 hours
375	191	30 - 60 minutes
400	204	5 - 10 minutes
425	218	to temperature
450	232	No
500	260	No

TABLE 5.3. - <u>Recommended Holding Times prior</u> to Forming, as a Function of Holding Temperature

Note: The above guide indicates maximum reheat periods; shorter heating times may give satisfactory results.

Under controlled conditions, strength loss due to reheating will seldom exceed 5 percent.

TABLE 5.4. - Relative Formability of Heat-

Treatable Alloys	(Extrusions	and Tube	3) in
Order of Dec			

Order	of	Decreasing	g Formability	7

Source Re	f. 5.2
Extrusions	Tubes
1. 6063, 6463 2. 6061, 6062 3. 2024 4. 2014	1. 6063 2. 6061, 6062 3. 2024 4. 2014
5. <u>7075</u> , 7079 6. 7178	5. <u>7075</u>

Source	Ref. 5.8
Alloy	7075
Temper	Treatment
F	As forged, no thermal treatment following fabrication operations
W	Solution heat-treated and quenched in water at 140° F (60° C)
Т6	Solution heat-treated, quenched in water at 140° F, and artificially aged
T652	Solution heat-treated, quenched in water at 140° F, stress relieved by cold com- pression, artificially aged

TABLE 5.5. - Heat Treat Tempers for Hand Forgings

Notes: Forgings in the T73 temper are also available. This temper is obtained by a proprietary thermal treatment (ref. 5.8)

> Premium strength forgings are available in the T736 temper (ref. 5.19)

		1					
Source	Ref.	5.9					
		Ξ	<u>High Speed</u>	I Tool	ΰ	Carbide Tc	ool
Operation	Cutting Conditions*	Speed ípm	Feed ipr	Tool mat'l	Speed	Feed ipr	Tool mat'l
Single point	0.250 inch depth of cut	0	0.015	T1, M1	1100	0.015	1
	050 inch	800	0.008	T1, M1	1400	.00	C-2
ool,	500 inch form tool	450	00.	T1, M1	1000	, 0035	1
turning	0.750 inch form tool width	450	.003	HSS	10001	103	C-2
	1.000 inch form tool width	450	0.003	HSS	1000	1.603	C-2 C-2
	inch form		000	HSS	1000	0.002	212 00
Boring	0.00 inch denth of cut	009	0.008	TI MI	1000	•	C-1, C-3
0	inch denth of	200	•	nec.	1050	•	
	inch depth of	540	0.015	HSS	1000	0.020	C-1, C-3
Planing	inch depth of	300	•	TI, MI	300	•	
	050 inch depth of	300	<u>.</u>	.	300	0.050	1
	0.010 inch depth of cut	300	3/4**	T1. M1	300	/4*	C-2
Face milling	250 inch depth of	800	•	T1, M1	тах	0.018*	C-2
	0.050 inch depth of cut	1000	0.022*	TI, MI	max	•	1
End milling	inct. cutter	200	0.006*	•	1200	0.005*	C2
(profiling)	1/2 inch cutter diameter	200	.009	M1, M10	1200	.008*	1
	inch cutter	1000	•	MI, MIO	1800	٠	
	8	1000	•	M1, M10	0	٠	ŧ
	3/4 inch cutter diameter	Ō	.007	MI, MI0		.006	1
	1 to 2 inch cutter diameter	1000	0.010*	M1. M10	1800	0,009*	C-2
Drilling	inch nominal hole	250	•	MI, MIO			
	inch nominal hole	2	.00	HSS			
	/2 inch nominal hole	7 2	•	HSS			
	3/4 inch nominal hole diameter	S	0.016	HSS			
		-	0	HSS			
	11/2 inch nominal hole diameter	2	Ö	HSS			
	2 inch nominal hole diameter	250	0.030	HSS			
	3 inch nominal hole diameter		0.030	HSS			
			1	,			

TABLE 5.6. - Machining Recommendations for Solution Treated and Aged 7075 Alloy

* Feed - inches per tooth

** Feed - 3/4 the width of square nose finishing tool

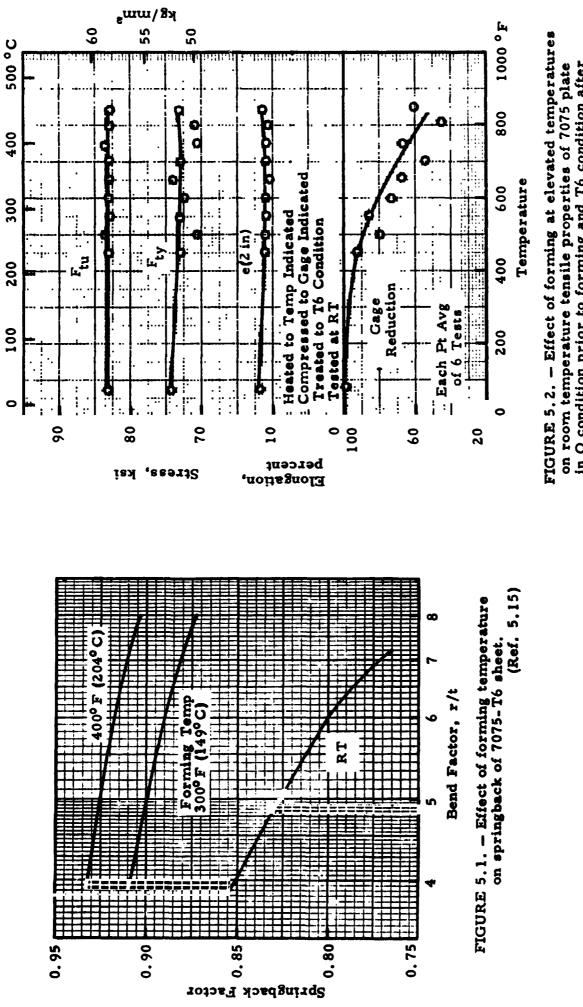


FIGURE 5.2. - Effect of forming at elevated temperatures on room temperature tensile properties of 7075 plate in O condition prior to forming and T6 condition after forming; thickness, 0.247 in (6.27 mm). (Ref. 5.16)



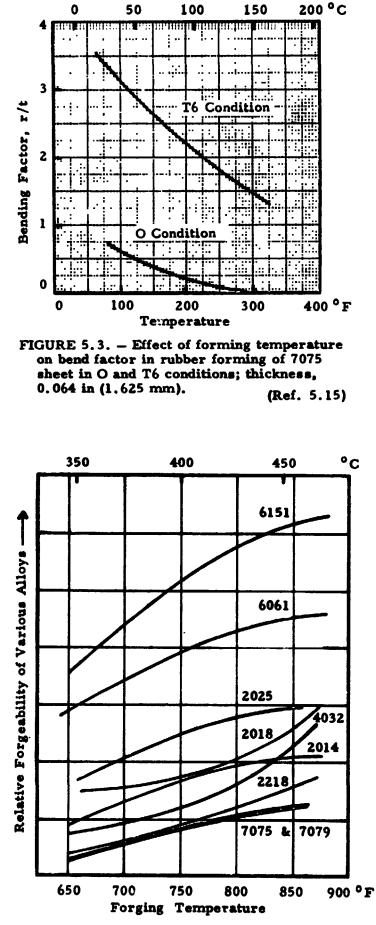


FIGURE 5.4. - Relative forgeability of various aluminum alloys. (Ref. 5.8)

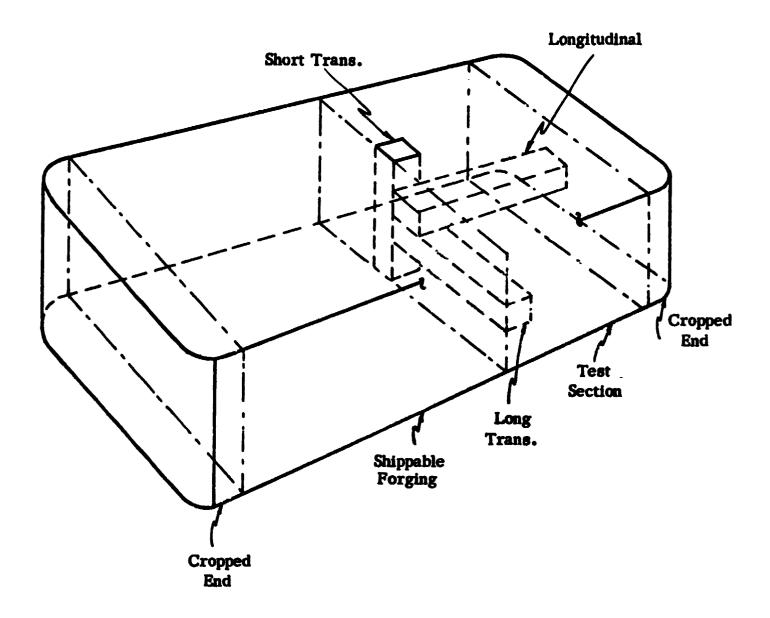


FIGURE 5.5. - Location of test bars for testing of hand forgings of rectangular or square cross sections.

(Ref. 5.8)

Chapter 5 - References

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- 5.10 J.B. Mohler, "Introduction to Chemical Milling," <u>Materials in</u> Design Engineering, April 1961, pp. 128-32.
- 5.11 G.H. Fox and H.H. Mueller, "Chemically Milled Structures," <u>Machine Design</u>, <u>33</u>, 126 (April 1961).
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Chapter 6

SPACE ENVIRONMENT EFFECTS

6.1 General. Aluminum alloys have been used in both structural and nonstructural applications in launch vehicles and spacecraft with excellent success since, in general, the aluminum alloys are relatively insensitive to degradation in typical space environment conditions. The vapor pressures of the structural aluminum alloys are sufficiently high (table 6.1) so that the combined temperaturevacuum effects generally are negligible. Structural alloys such as 7075 are sufficiently hardened so that nuclear and space indigenous radiation induced defects do not significantly affect mechanical and physical properties, at room ambient and elevated temperatures, below accumulated doses of about 10²² particles/cm². When irradiated at cryogenic temperatures, the 'hreshold may be lowered one or two decades, but the probabilities of experiencing doses on this order of magnitude are extremely remote except in the vicinity of nuclear reactors.

Elevated temperatures, hard vacuums, high energy radiations, and micrometeoroids can singularly and collectively influence surface characteristics of 7075 by desorption processes and erosion. These phenomena might be of great importance if optical properties, lubbrication, certain electrical properties, etc., were critical design parameters.

Sputtering of the surface by atomic or molecular particles can deteriorate surface finishes in a relatively short period. A 300-Å coating of aluminum (10^{-5} g/cm^2) can be destroyed in one month during a period of low intensity solar wind or in several hours during a solar storm, for example. The threshold energies of particles required to remove one or more atoms of the surface material they impinge are quite low, of the order of 6, 11, and 12 eV for O, N₂ and O₂ particles, respectively. Estimates of surface erosion by sputtering are given in table 6.2 for aluminum alloys.

Micrometeoroids can produce surface erosion similar to sputtering, although perhaps on a more macroscopic scale, as well as punctures. Micrometeoroids vary widely in mass, composition, velocity, and flux; generalizations about the rates of erosion and penetration, therefore, must be used with care. The predicted and measured frequency of impact as a function of meteoroid mass is given in figure 6.1. Data are given in figures 6.2 and 6.3 on the penetration and cratering of aluminum alloy skins of various thicknesses. Calculations of armor thickness required for protection of different structures and orientations are given in table 6.3. The design of bumper-hull meteoroid protection systems is discussed in reference 6.12. The surface erosion of aluminum alloys due to ccrpuscular radiation is probably insignificant, amounting to something of the order of 254 nanometers per year. Indigenous space radiation, however, will tend to accelerate the removal of surface films, which might result in loss of lubricity and an increased propensity to "cold weld." The interaction of indigenous radiation with descrption gases might cause some spurious, transient electrical conditions when aluminum alloys are used for electrical applications. The interaction of indigenous radiation with the alloys may produce some internal heating that might be significant for small items and may induce some radioactivity.

Source		Ref. 6.14					
		Evaporation	Rate, g/cm	² /sec			
Element	-100° C	0° C	100° C	250° C	500°C		
Aluminum	1.2×10^{-81}	$1.1 \ge 10^{-48}$	2.0×10^{-33}	$1.7 \ge 10^{-21}$	6.5×10^{-12}		
Titanium	<10-99	2.5×10^{-60}	4.1 x 10 ⁻⁴²	7.4×10^{-28}	2.0×10^{-16}		
Iron	<10-99	6.8 x 10 ⁻⁶⁴	2.4×10^{-44}	4.8×10^{-29}	9.1 x 10 ⁻¹⁷		
Nick ¹	<10-99	5.7 x 10^{-70}	1.3×10^{-48}	6.7×10^{-32}	1.7 x 10 ⁻¹⁸		
Copres	1.2×10^{-94}	1.4 x 10 ^{-F6}	6.2 x 10 ⁻³⁹	4.0×10^{-25}	4.7×10^{-14}		
Chromium	9.5 x 10^{-92}	$1.0 \ge 10^{-54}$	1.4×10^{-37}	$3.8 \ge 10^{-24}$	2.2×10^{-13}		
Vanadium	<10-88	$1.9 \ge 10^{-87}$	2.1 x 10 ⁻⁶¹	5.0×10^{-41}	1.2×10^{-24}		
Manganese	2.2 x 10-78	1.1×10^{-43}	6.5×10^{-28}	3.8 x 20-18	1.6 x 10 ⁻⁹		
Silicon	<10-99	1.9×10^{-62}	3.6×10^{-43}	4.3×10^{-28}	5.5 x 10 ⁻¹⁸		
Magnesium	2.9×10^{-36}	5.3 x 10^{-20}	1.8×10^{-12}	1.3 x 10 ⁻⁶	6.6×10^{-3}		
Zinc	3.5×10^{-30}	5.1×10^{-16}	1.8 x 10 ⁹	2.3×10^{-4}	2.80		

TABLE 6.1. – Evaporation Rates in Vacuum of Typical ElementsUsed in Aerospace Alloys (a, b)

(a) The actual evaporation rate of each element in combination with others will be lower.

(b) The values may be in error by several orders of magnitude as they have been extrapolated from high-temperature data. The rates a: low tempeatures will be considerably less than the values given in the table.

Source		Ref . 2		
	Orbiting Vehi	cie	Escaping V	chicle
Height, km	Rate, atom cm ⁻² sec ⁻¹	Time, sec/1x10 ⁻⁷ mm	Rate, atom cm ⁻² sec ⁻¹	Time, sec/lx10 ⁻⁷ mm
100 220 700 2500	3.1×10^{16} 2.0 x 10 ¹³ 2.2 x 10 ⁹ 4.3 x 10 ⁵	$\begin{array}{r} 1.9 \times 10^{-2} \\ 30 \\ 2.7 \times 10^{6} \\ 1.4 \times 10^{9} \end{array}$	3.4×10^{17} 2.0 x 10 ¹⁷ 3.4 x 10 ¹¹ 1.6 x 10 ⁸	1.8 x 10 ⁻⁸ 3.0 x 10 ⁻³ 1.8 x 10 ³ 3.8 x 10 ⁶

 TABLE 6.3. - Computed Thicknesses of Armor Required for Protection

 from Meteoroid Impact over a Period of 1000 Days

Source			Ref.	6.11			
Structure	Orientatior (a)		erable rea cm ²	Prob'y No Destructive Impact, %	Av. No. of Destructive Impacts per Mission		itical kness cm
Plane	i, leading	1000 500	92.9 46.5	99.5 99.75	0.005 0.0025	0.209 0.209	0.530 0.530
	i, trailing	1000 500	92.9 46.5	99.5 99.75	0.005 0.0025	0.109 0 7	0.278 0.278
	j, either side alone	2000 1000	185.8 92.9		0.01 0.005	0.232 0.232	0.590 0.590
	k, either side alone	2000 1000	185.8 92.9	99.0 99.5	0.01 9.005	0.197 0.197	0.500 0.500
Cylinder	i j k	2000 2000 2000	185.8 185.8 185.8	99.0 99.0 99.0	0.01 0.01 0.01	0.215 0.190 0.205	0.547 0.481 0.521
Sphere	(random)	2000	185.8	99.0	0.01	0.198	0.502

(a) i = direction of the apex of earth's movement

j = d rection within ecliptic plane, approximately away from sun, exactly perpendicular to apex of earth motion

.

k = direction p rpendicular to ecliptic plane, southward

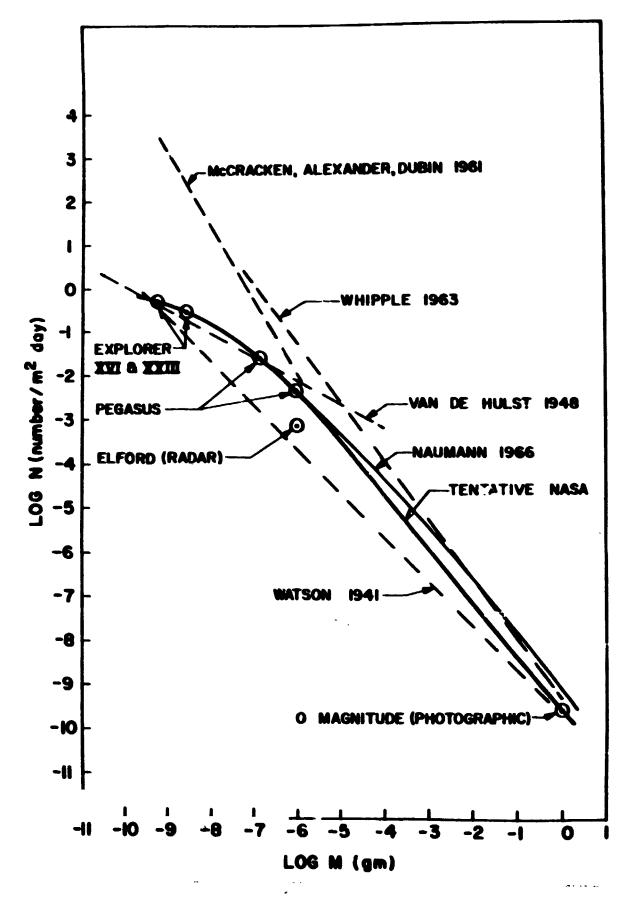


FIGURE 6.1. - Various estimates of meteoroid mass influx. (Ref. 6.3)

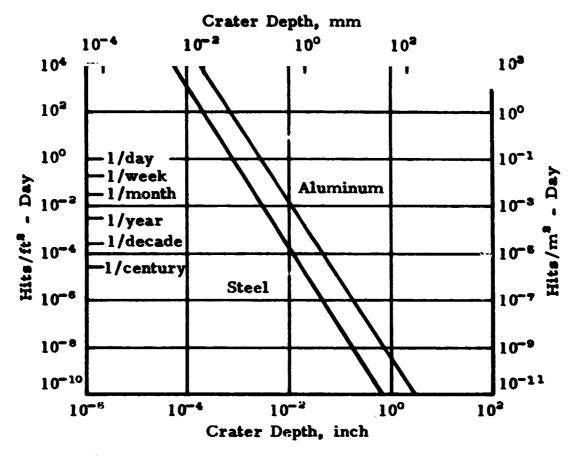


FIGURE 6.2. - Hit rate vs crater depth in the earth neighborhood but without earth shielding.

(Ref. 6.4)

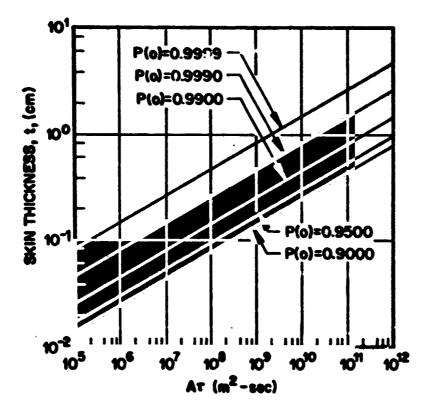


FIGURE 6.3. — Sheet thickness of Al as a function of the surface area—lifetime product required for various probabilities of no meteoroid puncture.

(Ref. 6.1)

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Chapter 7

STATIC MECHANICAL PROPERTIES

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Condition	Value	
0	60	
Т6	150	(ref. 7.4)

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Specification	NASA	A-MSFC-	SPEC-14	4B
Alloy		7075	(b)	
Max section thickness	3 inc	hes (7.0	62 cm)	
Temper	J	Г6	Т	73
Orientation	А	В	A	В
F _{tu} (min), ksi (a,c)	75.0	71.0	66.0	62.0
F _{ty} (min), ksi (a,c)	65.0	62.0	56.0	53.0
e (2 in or 4D)min, %	7	3	7	3

TABLE 7.111. - NASA Specified Mechanical Properties for Die Forgings and Separately Forged Test Bars

- A Lest specimen parallel to forging flow lines.
- B Test specimen not parallel to forging flow lines (die forgings only)
- (a) Tensile and yield strength test requirements may be waived for material in any direction in which the dimension is less than 2 inches because of the difficulty in obtaining a tension test specimen suitable for routine control testing. (2 in = 50.8 mm)
- (b) Die forgings in some configurations of this alloy can be purchased in the T652 temper conforming to the mechanical property requirements specified for the T6 temper.

⁽c) $1 \text{ ksi} = 0.70307 \text{ kg/mm}^2$

Specification	NAS	A-MSFC	-SPEC-1	44B (a)
Alloy		7075- T6		
Thickness, in (b)	Axis of test	F _{tu} , ksi	F _{ty} , ksi	e(2 in or 4D)
	specimen	min (c)	min (c)	min, %
≤ 2.000 (d)	L	74.0	63.0	9
	LT	73.0	61.0	4
2.001-3.000	L	73.0	61.0	9
	LT	71.0	59.0	4
	ST	69.0	58.0	3
3.001-4.000	L	71.0	60.0	8
	LT	70.0	58.0	3
	ST	68.0	57.0	2
4.001-5.000	L	69.0	58.0	7
	LT	68 0	56.0	3
	ST	66.0	56.0	2
5.001-6.000	L	68.0	56.0	6
	LT	66.0	55.0	3
	ST	65.0	55.0	2

TABLE 7.112. - NASA Specified Mechanical Properties forT6 Hand Forgings

(a) Maximum cross-sectional area is 256 in² (1.652 m²)

(b) Thickness is measured in the short transverse direction and applies to the 'as forged'' dimension before machining

(c) Tensile property requirements may be waived for directions in which the dimension is less than 2 inches.

(d) 1 inch = 25.4 mm; 1 ksi = 0.70307 kg/mm^2

Specification	NA	SA-MSF	C-SPEC-	144B
Alloy		7075-T	652 (a)	
Thickness, in (b)	Axis of test	F _{tu} , ksi	F _{ty} , ksi	e(2in or 4D)
	specimen	min (c)	min (c)	min, %
<u>≤</u> 2.000	L	74.0	63.0	9
	LT	73.0	61.0	4
2.001-3.000	L	73.0	61.0	9
	LT	71.0	59.0	4
	ST	69.0	57.0	2
3.001-4.000	L	71.0	60.0	8
	LT	70.0	58.0	3
	ST	68.0	56.0	2
4.001-5.000	L	69.0	58.0	6
	LT	68.0	56.0	3
	ST	66.0	55.0	1
5.001-6.000	L	68.0	56.0	6
	LT	66.0	55.0	3
	ST	65.0	54.0	1

TABLE 7.113. - NASA Specified Mechanical Properties ofT652. Hand Forgings

(a) Maximum cross-sectional area is 256 in^2 (1.652 m²)

(b) Thickness is measured in the short transverse direction and applies to the "as forged" dimension before machining.

(c) Tensile property requirements may be waived for directions in which the dimension is less than 2 inches.

(d) 1 inch = 25.4 mm; 1 ksi = 0.70307 kg/mm^2

TABLE 7.114	NASA Specifie	d Mechanical	Properties for

T73 Hand Forgings

Specification	NAS	A-MSFC-	SPEC-144	4 B
Alloy		7075-T73	(a)	-
T [↓] ~kness, in (b)	Axis of test	F _{tu} , ksi	F _{ty:} ksi	e(2in or 4D)
	specimen	min (c)	min (c)	r.in, %
≤ 3.000	L	66.0	56.0	7
	LT	64.0	54.0	4
	ST	61.0	52.0	3

(a) Maximum cross-sectional area is 256 in^2 (1.672 m²)

(b) Thickness is measured in the short transverse direction and applies to the "as forged" dimension before machining

(c) Tensile property requirements may be waived for directions in which the dimension is less than 2 inches.

(d) 1 inch = 25.4 mm; $\frac{1}{k}$ i = 0.70307 kg/mm²

Alloy			QQ-A-250/	12 (7075)		
•orm			Sheet and	1 Plat		
Condition	T73		T735	51		
Thickness, in. ^(a)	0.040 0.249	U.250 0.499	0.500 1.000	1.001 2.000	2.001 2.500	2.501 3.000
Sasis	S	S	S	S	S	S
Mechanical properties:						
F _{SU} , ksi:		1	ł			
L	67	69	69	69	66	64
LT	67	69	69	69	66	64
ST		·····		.	<u> </u> 	
F _{ty} , ksi:						
L	56	57	57	57	1 52	50
LT	56	57	57	57	52	49
F _{cv} , ksi:						
L	55	56	56	56	50	47
LT	58	58	58	59	54	51
ST				.	_	<u> </u>
<i>F_{5U}</i> , ksi	38	39	39	39	38	38
F _{bru} , ksi:						
(c/D = 1.5)	105	106	105	105	102	100
(c/D = 2.0)	134	137	135	135	131	128
F _{brv} , ksi:					!	
(c/D = 1.5)	84	86	86	86	82	79
(c/D = 2.0)	102	104	104	104	9 8	95
e, percent:						
L						
LT	8	7	7	6	6	6
ST	-	I		I]

TABLE 7.4111. - Design Tensile Properties for Sheet and Platein T73 and T7351 Tempers

Note: $1 \text{ ksi} = 0.70307 \text{ lg/mm}^2$; 1 inch = 25.4 mm.

TABLE 7.4112. - Design Tensile Properties for T6 and T651 Sheet and Plate

Alloy								6	QQ-A-250/12 (7075)	12 (70	75)							!
Form		She	Sheet								Pla	Plate						
Condition		L	T6								T651	51						
Thickness, in.	0.015-0.039	0.01 5 - 0.039	0.0 0.2	0.040- 0.249	0.250-0.499	0.250- 0.499	0.500-	0.500- 1.000	1.001-2,000	-10	2.001- 2.500	10	2.501- 3.000		3.001- 3.500		3.501- 4.000	40
Basis	¥	Ø	v	æ	Y	æ	4	æ	<	æ	×	m	~	B	<	æ	×	B
Mechanical properties: $F_{i,i}$, ksi:														 				
L	76	78	77	79	76	78	76	78	76	78	72	74	69	11	69	11	99	68
LT	76	78	77	19	77	79	77	79	77	79	73	75	20	72	20	72	67	69
51											65	67	63	65	63	65	09	62
T	99	69	67	70	68	70	68	70	68	70	64	66	62	64	59	61	55	57
	65	68	6 6	69	66	68	66	68	66	68	62	64	60	62	57	59	53	55
											55	57	53	55	51	53	47	49
	67	70	68	11	67	69	66	89	65	67	60	62	58	60	55	57	50	52
LT	2	73	11	74	20	72	70	72	70	72	66	89	64	66	60	62	56	58
ST F., ksi	46	47	46	47	43	44	44	45	45	46	63	65 44	61	63	58 42	60 44	54	56 42
Forn kai":		;	}	;	}			2	2	2	 }	;) }	2		р Р	4	2
11	114	117	116	119	117	120	117	120	117	120	111	114	106	109	106	109	102	105
(e/D = 2.0)	144	148	146	120	144	148	144	148	144	148	136	140	131	135	131	135	125	129
(e/D = 1.5)	92	97	94	8 6	97	100	98	101	100	103	96	66	94	97	91	94	86	68
(e/D = 2.0)	106	110	107	112	114	117	115	118	117	12	112	116	110	113	106	109	6 6	103
e, percent: L			00	-	00		ť						10		2			
LT	~ ~		000		00		9 10										~	
ST											-				-			
																•		

(a) See reference 7.5 for bearing reduction factors.
(b) Specific design properties are not shown for the T62 .emper because the properties of this temper may vary greatly dependent upon the particular comoination of original temper, amount and uniformity of cold working and subsequent thermal treatment. Although the design values for the T651 temper may be used as a general guide, it must be recognized that some properties of the T62 temper may be markedly lower.
Note: 1 ksi = 0.70307 kg/mm²; 1 inch = 25.4 mm.

																						ł
								ľ		QQ-A-250/13	250/1	3 (Clad	(Clad 7075)									
Form				ŝ	Sheet									Plate	te							1
Condition					-T6									-T651 ^c	اد							
Thickness, in	0.0	0.012- 0.039	0.040- 0.062	40- 62	0.063- 0.187	63- 87	0.188- 0.249	\$ 5 5	0.250- 0.499	9.9	0.500- 1.000 ⁶	తిరి	1.001- 2.000 ⁰	-i-9	2.001- 2.500 ^b	B	2.501- 3.000 ⁶		3.000 ⁶ 3.500 ⁶		3.501- 4.000 ⁶	ه∸ ا
Basis	¥.	B	۲	B	¥	8	<	æ	<	æ	<	æ	<	æ	•	m	<	<u>_</u>	<	8	<	m
Mechanical properties: E kei																			 			1
LT	70 70	73 73	72 72	74 74	73 73	75 75	75 75	77 77	74 75	76 77	74 75	76	74 75	76	71	72 73	67 68	69 70	67 68	69 70	65 65	68 67
ST												<u> </u>			65	67	63	65	63	65	60	62
LT	6 61	4 6	63 62	65 64	64 63	<u> </u>	64 64	69 66	66 66	89 89 89	% 4	89 89 99	68 64	66 66	62 60	64 62	58 58	62 62	57 55	59 57	53 51	55 53
ST															55	57	53	55	51	53	47	49
	62	65	64	66	65	67	66	68	65	67	64	66	63	65	58	60	56	58	53	55	48	50
LT ST	64	67	66	88	67	69	88	70	68	70	89	20	68	70	4 5	6 5	62	4 C	\$ \$ 8 8 8 8 8 8	0 0 0 0 0 0	54 54	56 56
Fat, ksi	42	4	43	4	44	45	45	46	42	43	\$	4	43	45	5	4 3	41	5	4	42	40	7
e/D = 1.5) (e/D = 2.0)	105 133	110 139	108 137	111	110	112 142	112 142	116 146	114	117	114	117	114	117 144	108 133	111	103 127	106	103	131	99 121	102
Fbry. ksf (e/D = 1.5)	35	06	88	16	06	92	16	94	94	67	95	98	97	00	93							86
e, percent	20	701	5	5	102	ŝ	5	10	011			<u> </u>	113	· · · · · · · · · · · · · · · · · · ·	201		<u>§</u>		_	6	g.	\$
LT	~ ~		∞ ∞		∞ ∞		∞ ∞		00 00		0 0		v 4		w w		ν ω		<u></u> v. m			
ST		1		:	:	:														· 		

TABLE 7.4113. - Design Tensile Properties for Clad T6 and T651 Sheet and Plate

(a) See reference 7.5 for bearing reduction factors.
(b) These values, except in the ST direction, have been adjusted to represent the average properties across the whole section, including the 1 1/2 percent per side nominal cladding thickness.
(c) Specific design properties are not shown for the T62 temper because the properties of this temper may vary greatly dependent upon the particular combination of original temper, amount and uniformity of cold working, and subsequent thermal treatment. Although the design values for the T651 temper may be used as a general guide, it must be recognized that some properties of the T62 temper may be markedly lower.
(Ref. 7.5)

Alloy			MIL-A-2	2771, Typ	e 7075			
Form	Die fe	orginga			Hand f	orgings		
Condition	-T6 and -T652	-T73 and -T7852		- T 6	and T65	2 b		-T78
Thickness, in.	$\frac{-1002}{\leq 8.000}$	<u>-17362</u> ≤8.000	≤ 2.000	2.001-	3.001 – 4.000	4.001-	5.001-	≤ 8.000
Basis	A	8	8	8	8	8	8	8
Mechanical properties;								
F _{tu} , ksi			1					
<i>L</i>	75	66	74	78	71	69	68	66
LT	*******		78	71	70	68	66	64
ST	71	62		69	68	66	65	61
F _{ty} , kai				1				
<i>L</i>	65	56	63	61	60	58	56	56
LT			61	59	58	56	55	54
ST	62	58		58+	570	563	550	52
F _{cy} , ksi			Į					
<i>L</i>	65	56	63	61				56
LT	•••••		61	59			*******	52
ST	58	51					*******	
F _{ent} ksi	45	89	44	44	48	41	41	89
F _{kra} , ksi		1						
(e/D-1.5)	*******	86		• • • • • • • • •				
(e/D-2.0)	•••••	119			******			••••••
F _{bre} , ksi					1			
(e/D=1.5)	*******	78						
(e/D-2.0)	•••••	84						
s, per cent								
<i>L</i>	7	7	9	9	8	7	6	7
LT			4	4	8	8	8	4
ST	8	8		85	23	21	26	8

TABLE 7.4114. - Design Tensile Properties for Hand and Die Forgings

• Maximum cross-sectional area 256 sq. in. • All properties are applicable to the -T652 temper except $F_{iy}(ST)$ as $i \in (ST)$, for which the values are as follows:

	2.0.1.	3.001	4.001-	5.001-
Thickness, in	8.06	4.000	5.900	6.000
F _{ty} (ST), kai	51	56	55	54
e(ST), per cent	3	1	1	1

."For die forgings, the L and ST values are for the directions parallel (within ± 15 degrees) and not parallel (as close as possible to the short transverse direction) respectively, to the forging flow lines "

Note: $1 \text{ ksi} = 0.70307 \text{ kg/mm}^2$; 1 inch = 25.4 mm.

Vire, and Shapes	
, Rod, W	in the second seco
Properties for Bar,	
ile Proper	
Design Tensile	
TABLE 7.4115	

Woy	Alloy						A-99	QQ-A-225/9 (7075)	(7075)			
шио,	Form				Bar, Roc	I, Wire a	nd Shape	s; Rolle	l, Drewn	Bar, Rod, Wire and Shapes; Rolled, Drewn or Cold-Finished	hished	
ondition	Condition				T6 or T651	51					T73 or T7351	1351
hickness, i	Thickness, in	Up to 1.000 ⁴	°000	1.001-	1.001-2.000	2.001-3.000	3.000	3.001-4.000	4.000	0.375 1.000	1.001- 2.000	2.001- 3.000
sis	Basis	×	B	۲	B	×	B	•	B	S	S	s
lechanical Lai	Mechanical Property:	F	ę		Ċ	Ę	ç	ţ	ģ	¢,	,	
[La. vai	LT	: [61 91	15	55	: 2	46	6 9	22	8	8	65 65
F _{fy} , ksi	L	66 66	6 8 8 8 8	8 8 8	68 68 68	66 63	68 65	% 80	68 62	56	- S6	56 52
F _{cy} , ksi	L	2 ¹	99	6 4	× 1	1 6	1 66	21	99 1	52 -	53	55 55
w. ksi	F ₂₄ , ksi	4	47	46	47	46	47	46	47	42	41	41
F _{bru} , ksi	(e/D = 1.5)	100	103 126	100 123	103 126	100 123	103 126	100 123	103 126	102 133	102	102
F _{bry} , ksi	(c/D = 1.5)	86 92	88 95	86 92	88 25	86 92	88 95	86 92	88 95	82 101	82 101	82 101
e, per cent	L	dr 4	11	r w	11	~ 7	ł į	r 1	11	10	10	- 10

(a) For rounds (rod) maximum diam is 4 in; for Jquare bar, maximum size is 3 1/2 in; for rectangular bar, maximum thickness is 3 in, with corresponding width of 6 in; for rectangular bar less than 3 in thick, maximum width is 10 in.
(b) Except for wire less than 0.125 inches in diam.
Note: 1 ksi = 0.70307 kg/mm²; 1 inch = 25.4 mm.

Alloy									8	QQ-A-200/11 (7075)	1 (70	15)		F					
Form .								Extrusio	v) suo	Extrusions (rod, bars, and shapes)	and s	hapes)						1	
Condition						Ĩ	5, T6	T6, T6510, T6511	11							1	T73, T73510, T/3511	10, T'	3511
Cross-sectional area, in. ²				VI	20						≤ 20		>20, ≤32	VI	: 32	V	20	VI	25
Thickness ^a (in.)	Up te	Up to 0.249 0.250-0.499	0.251	0-0.499		0.500-0.749	0.75(0.750-1.499	1.500	1.500-2.999	` `	499. د. 000-4	661	4.500	4.500-5.000	0.062 0.249	0.250 0.499	0.500 0.749	0.750 1.499
Basis	۷	B	۷	B	<	B	۲	B	×	B	V	B	s	V	в	s	s	s	s
Mechanical Properties: $F_{n,n}$ ksi:																			
	78	82	81	85	81	85	81	85	81	85	81	2	78	78	81	99	69	70	70
LT F ksi:	76	80	78	81	76	8	74	78	70	74	67	70	65	65	67	63	66	66	66
	70	74	73	11	72	76	72	76	72	76	11	74	70	68	71	58	61	61	61
LT	66	70	68	72	99	70	65	68	61	65	56	58	55	50	53	55	58	57	56
$F_{C_{T}}$ ksi: L	70	74	73	11	72	76	72	76	72	76	71	74	70	68	11	58	61	61	61
LT	72	76	4	78	72	76	71	74	67	11	61	64	60	56	58	59	62	61	60
Far ksi	42	44	43	45	43	45	42	44	41	43	40	41	38	37	39	36	37	38	38
Fbrur Ksi: (e/D = 1.5)	112		117	122	117		116		115	120	601	-	Ń	8	8	86	102	103	103
(e/D = 2.0)	141	148	146	· ~	146	153	145	152	4		142	147	136	135	140	127			133
Fbry, ksi: le/D = 1.5)	8	8	07		ક		š	ŝ	ö	ă	ő	6	8. 2	63	87	2	Š	Š	2
(e'D = 2.0)	110		115) ~	113	119	112	_	110		00	110	<u>5</u>	6) 03	10			5 <u>8</u>
e, percent:	(1		1		1							- <u>-</u>					
·······	2	80	~	80	2	00	~	00	~	80	-	80	9	9		7	-	2	1
<i>LT</i>	~	:	S		4		ę			:	-		1	-	:				
							-		-	-	-	-	T				-	1	

TABLE 7.4116. - Design Tensile Properties of T6- and T73-Type Extrusions

 ⁽a) For extrusions with outstanding legs, the load carrying ability of such legs shall be determined on the basis of the properties in the appropriate column corresponding to the leg thickness.
 Note: 1 ksi = 0.70307 kg/mm³; 1 inch = 25.4 mm.

Alloy	QQ-A-00200/15(ASG) Extrusions				
Form					
Condition	T76	Т	76, T76510, T76511	l	
Cross-sectional area, in ²		≤ 25			
Thickness, in ^a	0.125- 0.249	0.250- 0.499	0.500- 0.749	0.750- 1.000	
Basis	S	S	S	S	
Mechanical properties:					
F _{tu} , ksi					
<i>L</i>	74	75	75	75	
<i>LT</i>	71	72	71	70	
F_{ty} , ksi L	64	65	65	65	
<i>LT</i>	60	61	60	59	
F _{cy} , ksi					
⁷ L	64	65	65	65	
LT	65	66	65	64	
<i>F_{su},</i> ksi	39	40	40	40	
F _{bru} , ksi:					
(e/D = 1.5)	107	110	109	109	
(e/D = 2.0)	136	139	139	138	
Fbry, ksi:					
(e/D = 1.5)	86	89	88	87	
(e/D = 2.0)	102	105	105	104	
e, per cent:		-		_	
L	7	7	7	7	
LT		~	-	-	

FIGURE 7.4117. - Design Tensile Properties of T76-Type Extrusions

(a) For extrusions with outstanding legs, the load carrying ability of such legs shall be determined on the basis of the properties in the appropriate column corresponding to the leg thickness. Note: 1 ksi - 0.70307 kg/mm²; 1 inch = 25.4 mm.

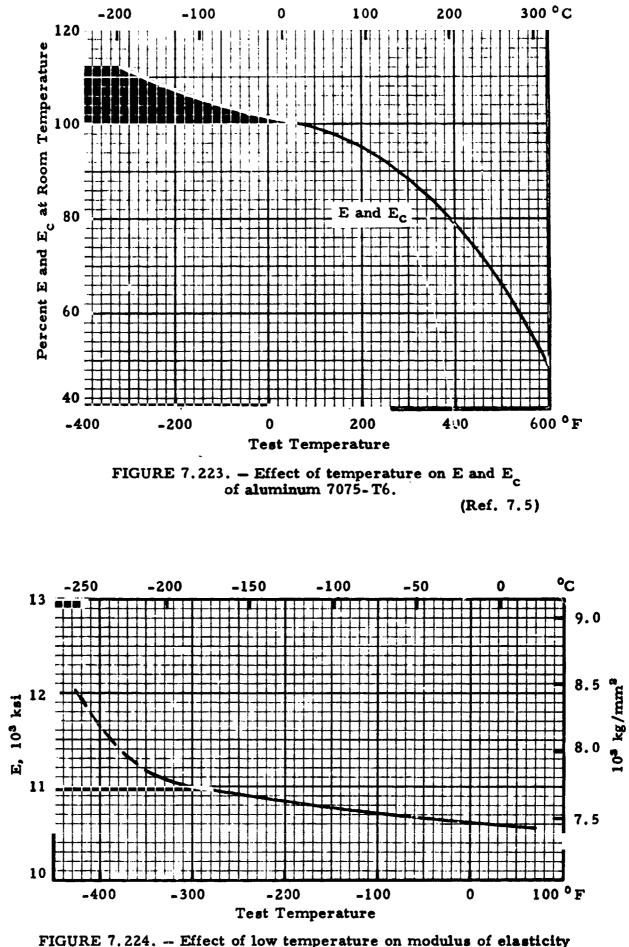


FIGURE 7.224. -- Effect of low temperature on modulus of elasticity of 0.250-in (0.635-cm) 7075-T6 sheet. (Ref. 7.7)

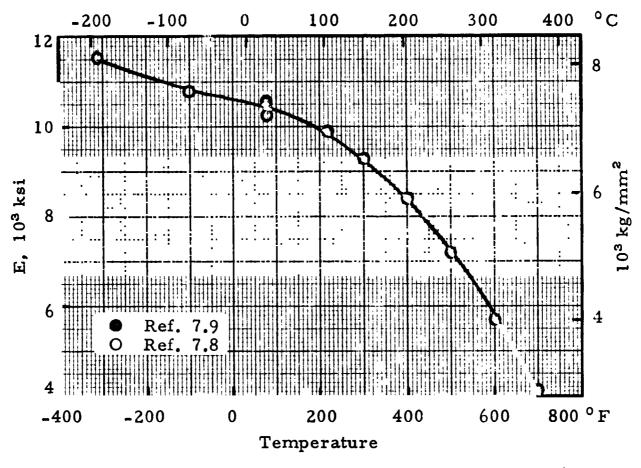
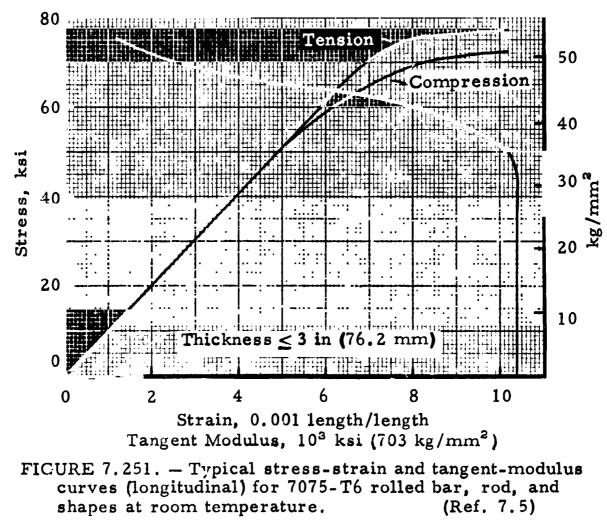


FIGURE 7.225. — Modulus of elasticity (static) of 7075-T6 bar at various temperatures.



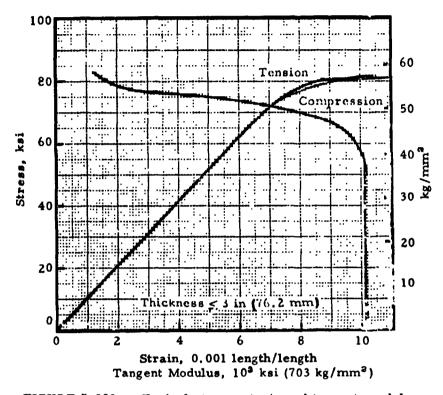
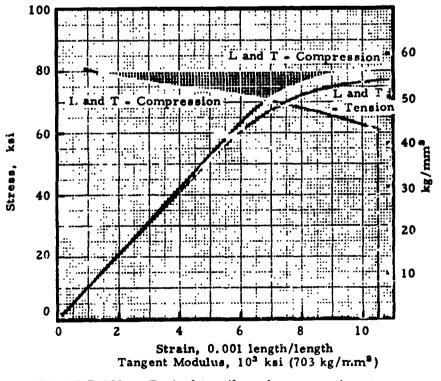
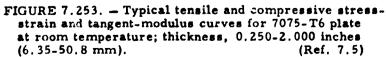


FIGURE 7.252. - Typical stress-strain and tangent-modulus curves (longitudinal) for 7075-T6 extrusions at room temperature. (Ref. 7.5)





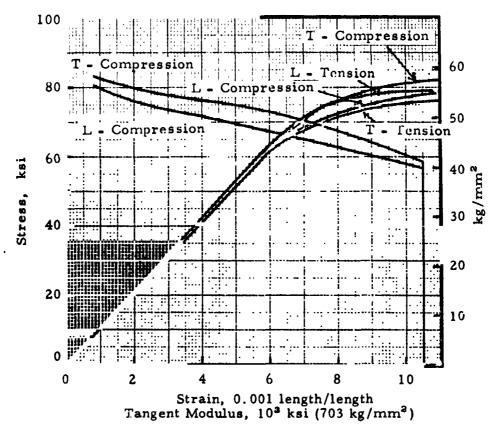
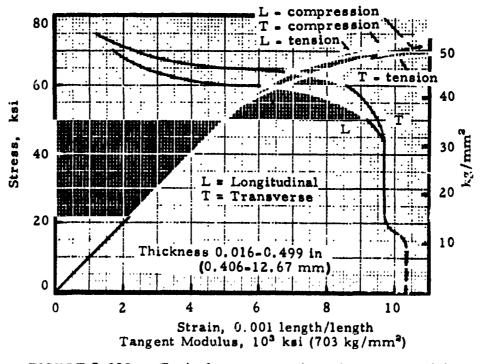
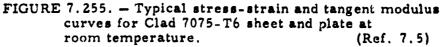


FIGURE 7.254. — Typical tensile and compressive stressstrain and tangent-modulus curves for 7075-T651 plate at room temperature; thickness, 0.250-2.000 inches (6.35-50.8 mm). (Ref. 7.5)





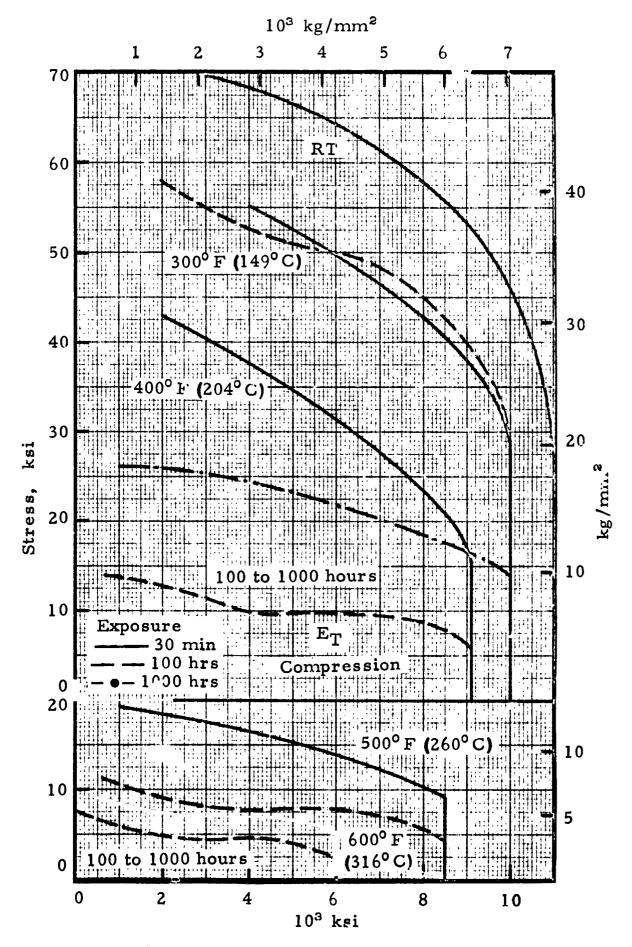
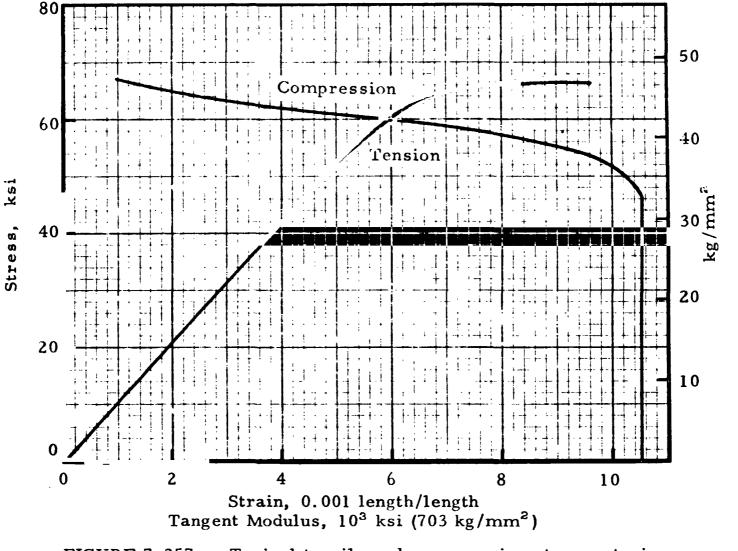
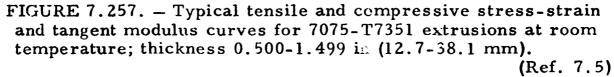


FIGURE 7.256. - Tangent modulus curves in compression for Clad 7075-T6 sheet at room and elevated temperatures; thickness, 0.064 in (1.625 mm).

(Ref. 7.10)





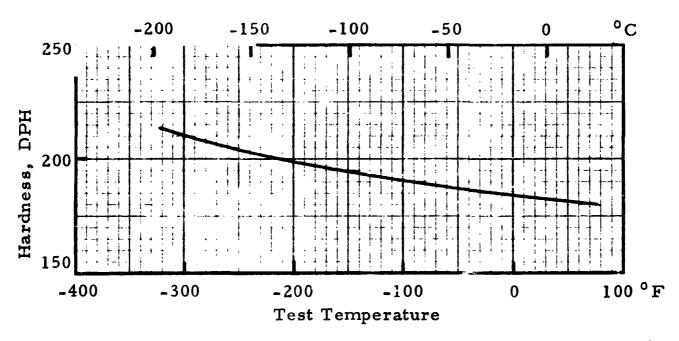
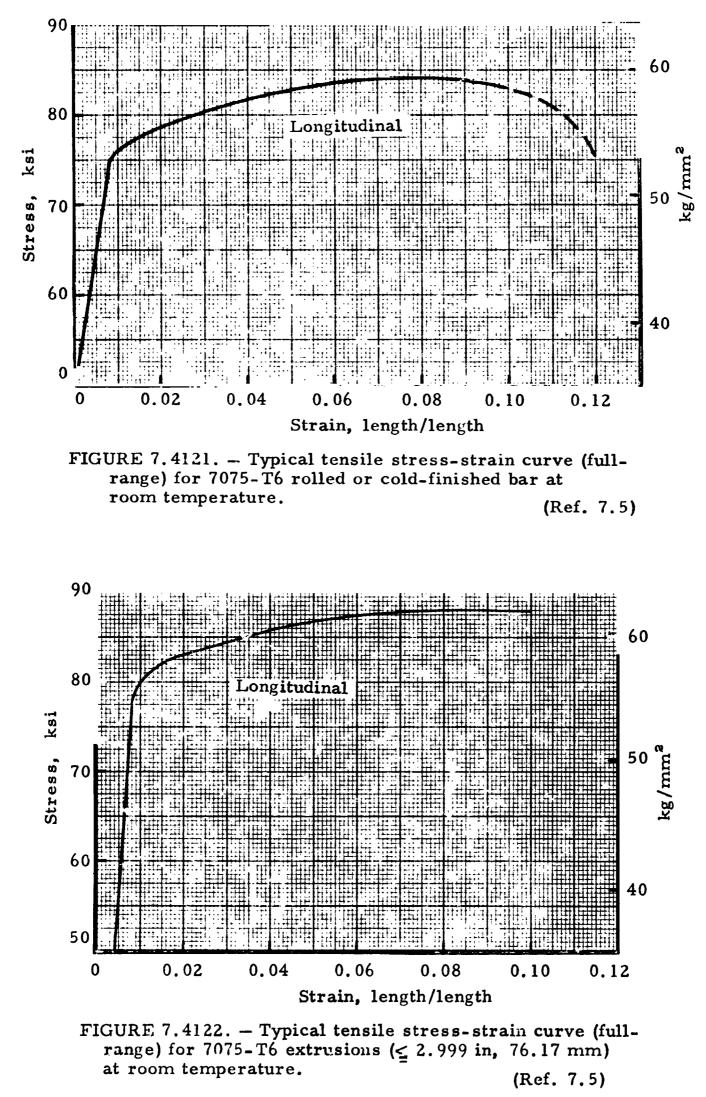
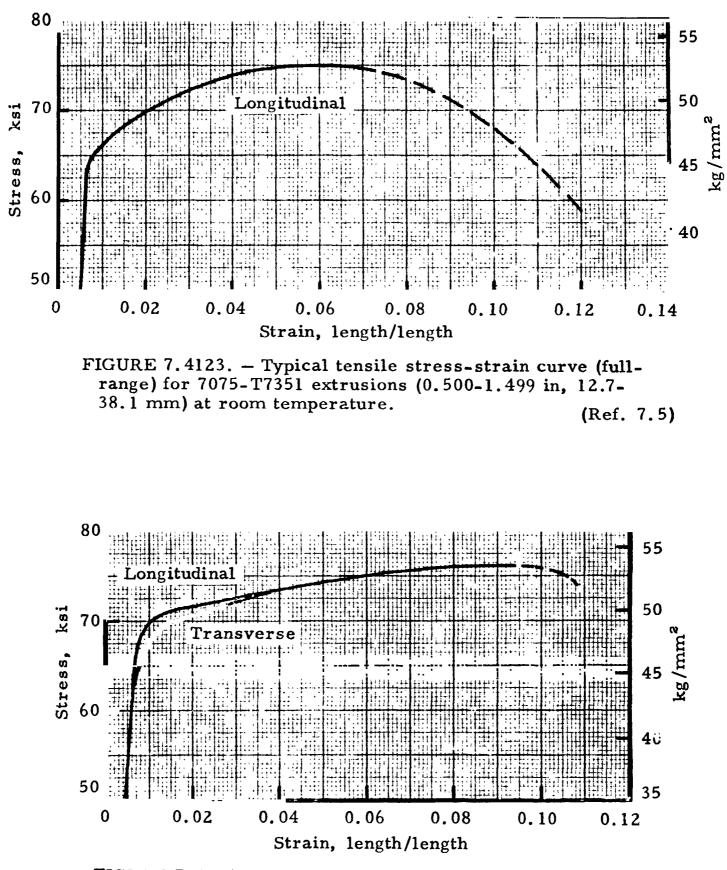
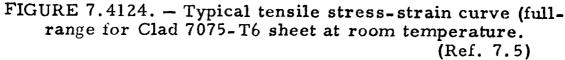


FIGURE 7.32. - Effect of low temperatures on hardness of 7075-T6 bar; thickness, 0.750 in (19.05 mm). (Ref. 7.11)







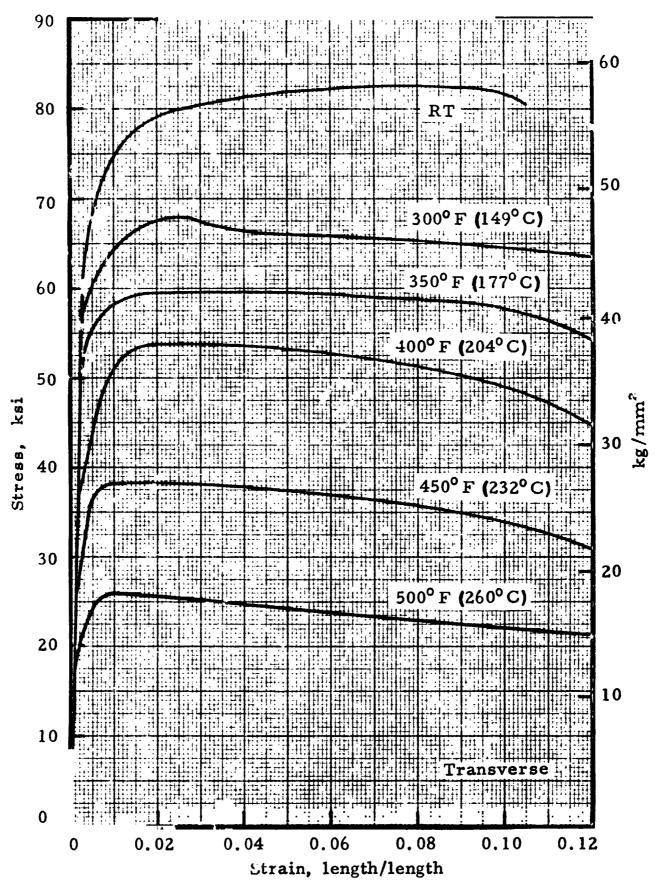
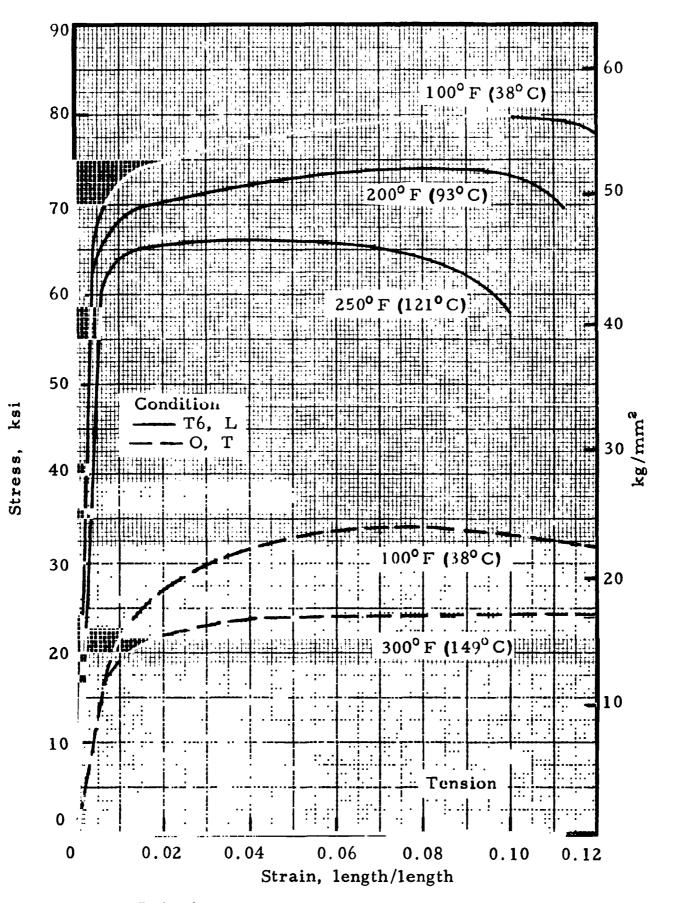
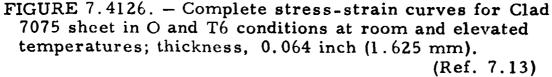
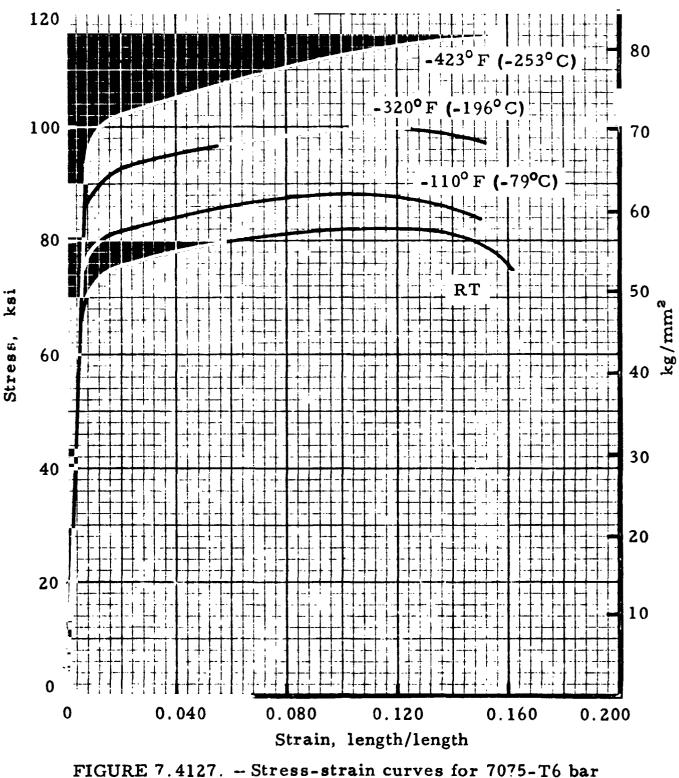


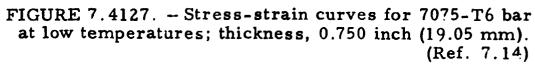
FIGURE 7.4125. — Complete stress-strain curves for 7075-T6 sheet at room and elevated temperatures.

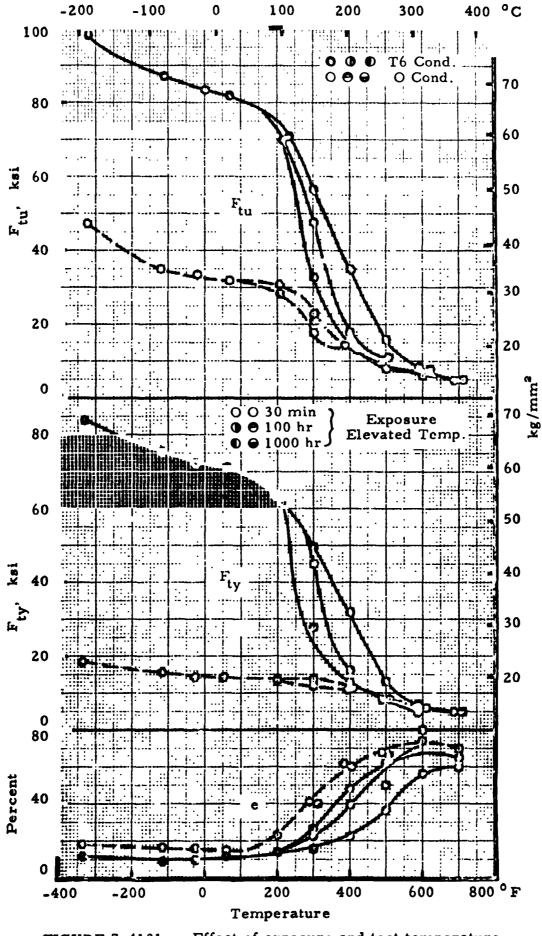
(Ref. 7.13)

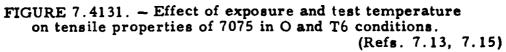












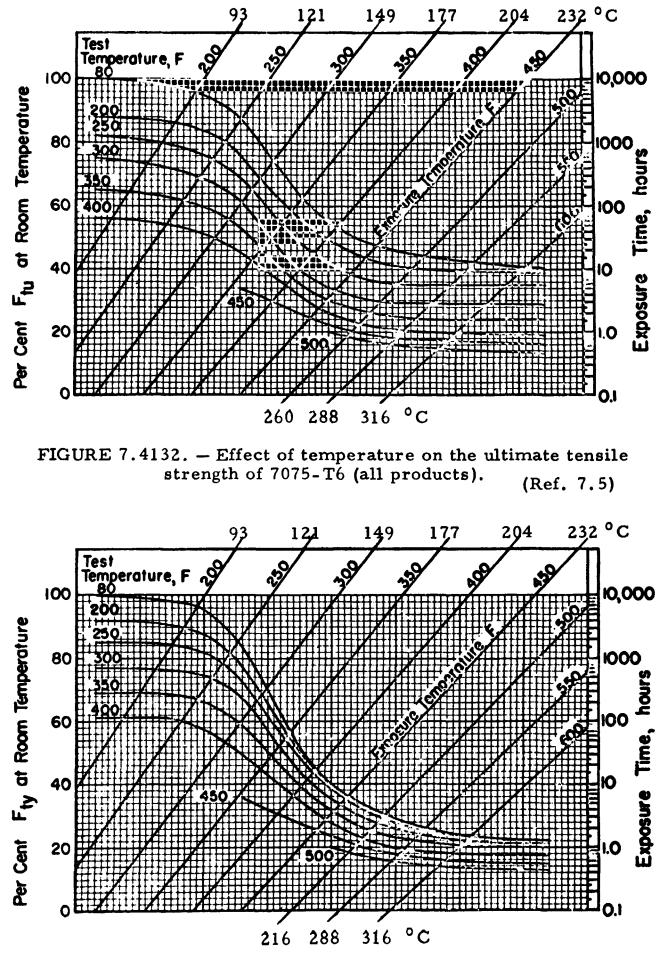


FIGURE 7.4133. - Effect of temperature on the tensile yield strength of 7075-T6 (all products). (Ref. 7.5)

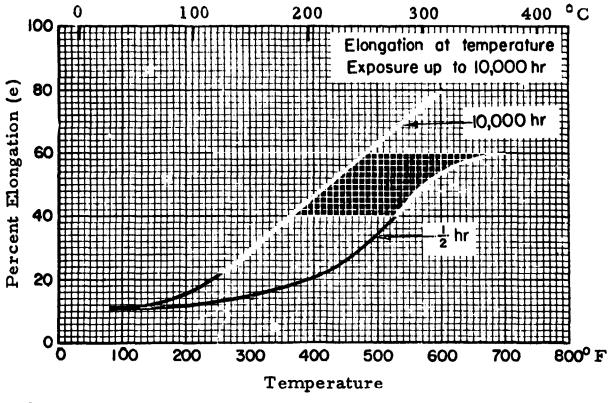


FIGURE 7.4134. - Effect of temperature on the elongation of 7075-T6 (all products except thick extrusions). (Ref. 7.5)

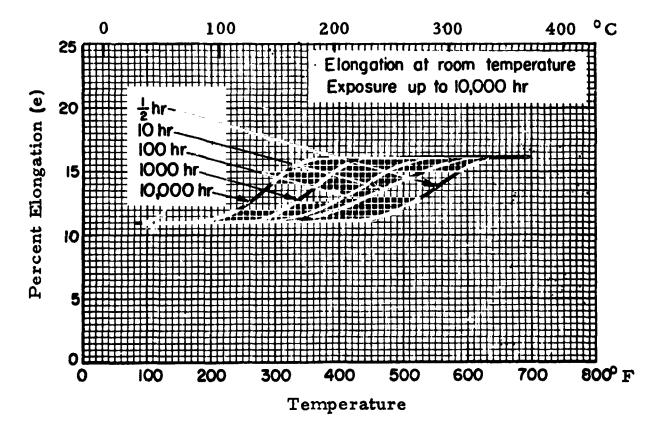
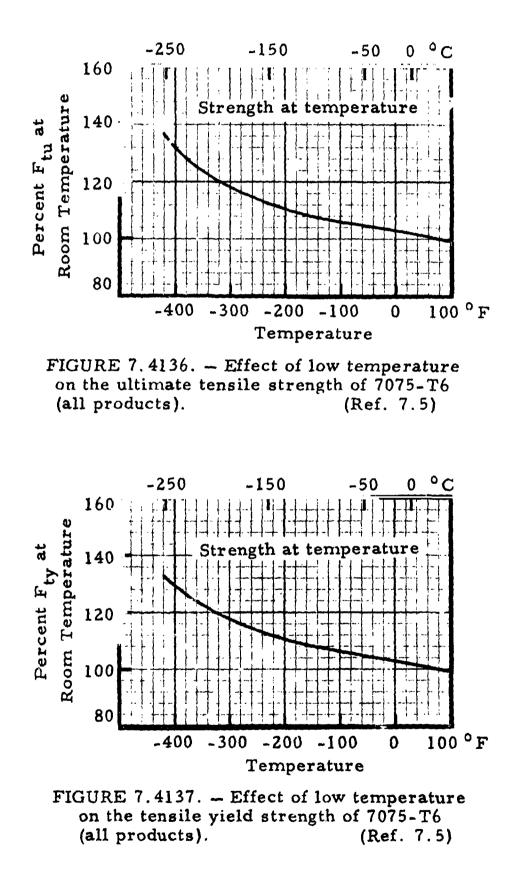


FIGURE 7.4135. - Effect of exposure at elevated temperatures on the elongation of 7075-T6 (all products except thick extrusions). (Ref. 7.5)



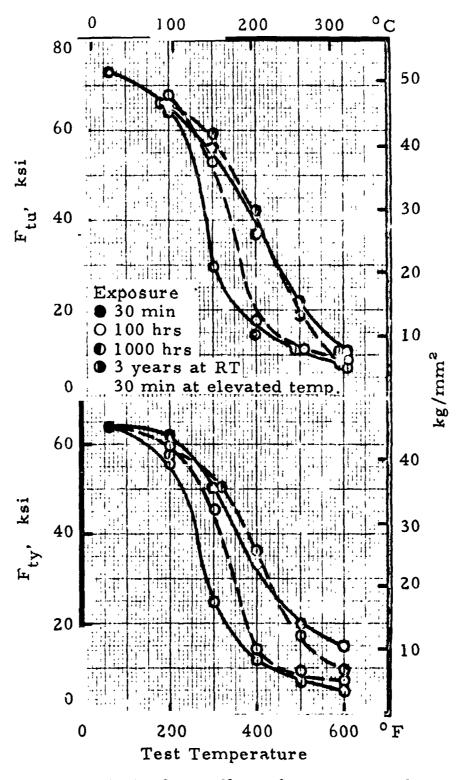
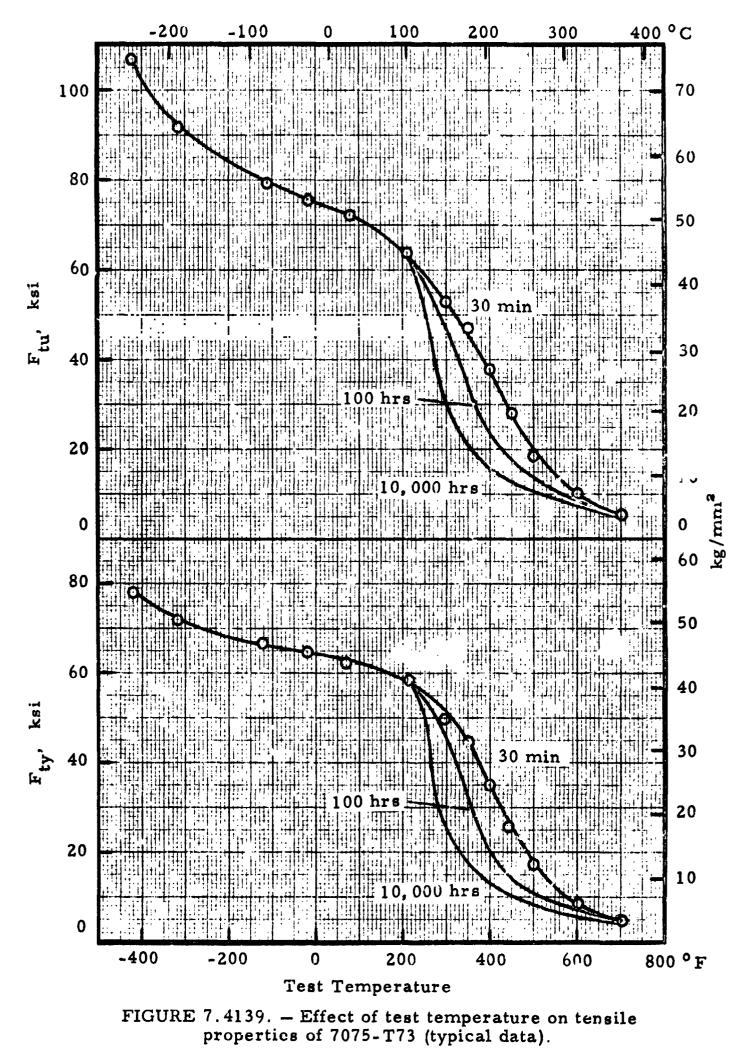
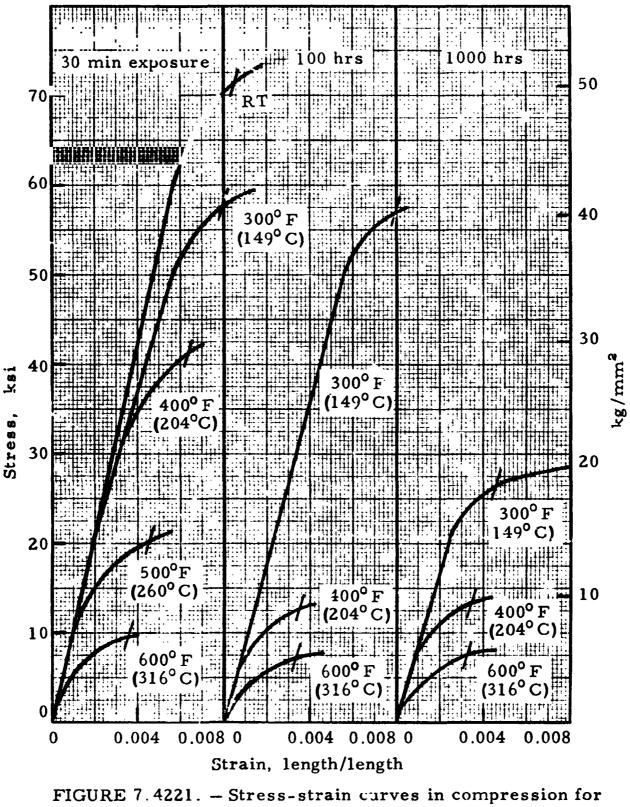
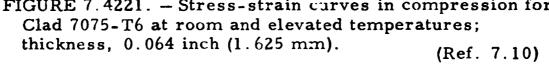


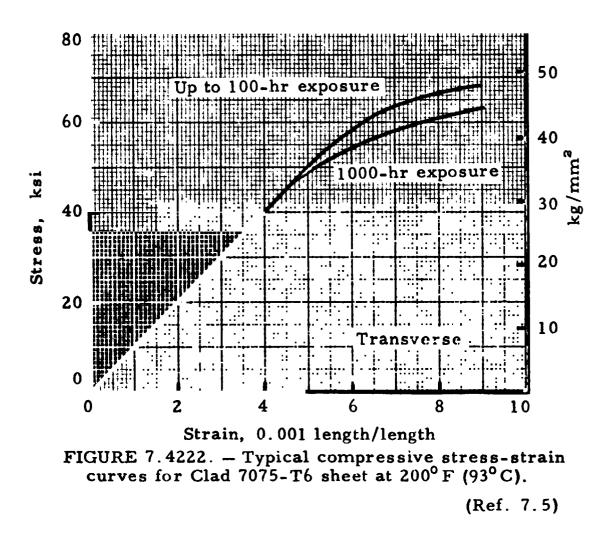
FIGURE 7.4138. - Effect of exposure and test temperature on tensile properties of Clad 7075-T6; thickness, 0.064 inch (1.625 mm). (Ref. 7.17)



(Ref. 7.16)







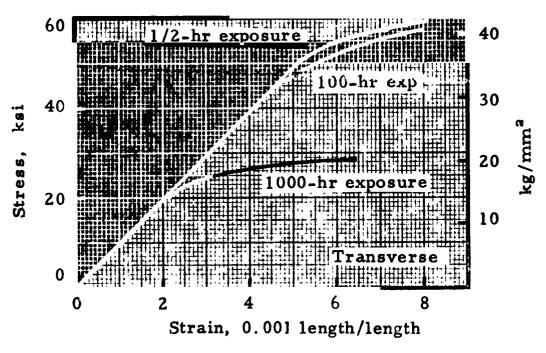
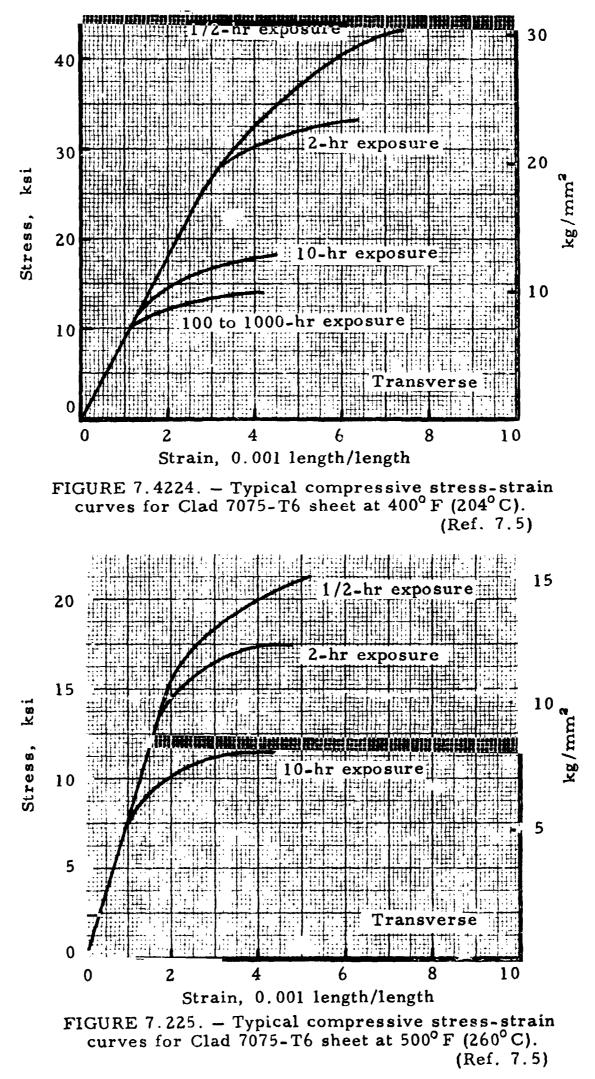


FIGURE 7.4223. - Typical compressive stress-strain curves for Clad 7075-T6 sheet at 300°F (149°C).

(Ref. 7.5)



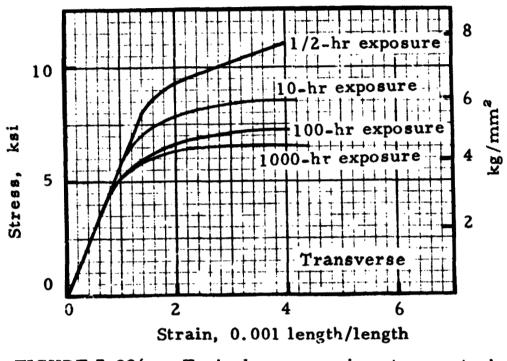


FIGURE 7.226. - Typical compressive stress-strain curves for Clad 7075-T6 sheet at 600°F (316°C). (Ref. 7.5)

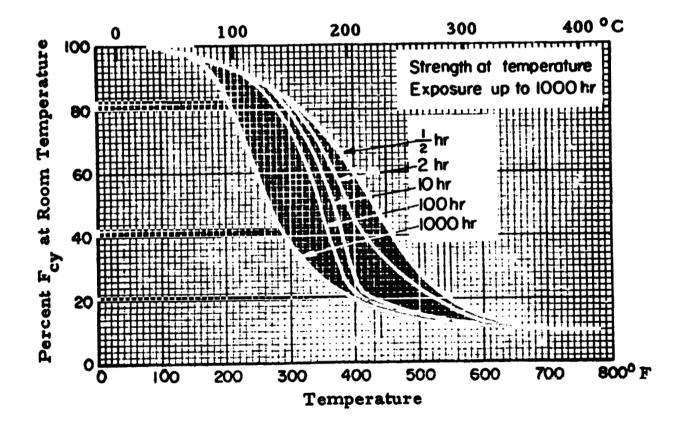


FIGURE 7.227. - Effect of temperature on the compressive yield strength of 7075-T6 (all products).

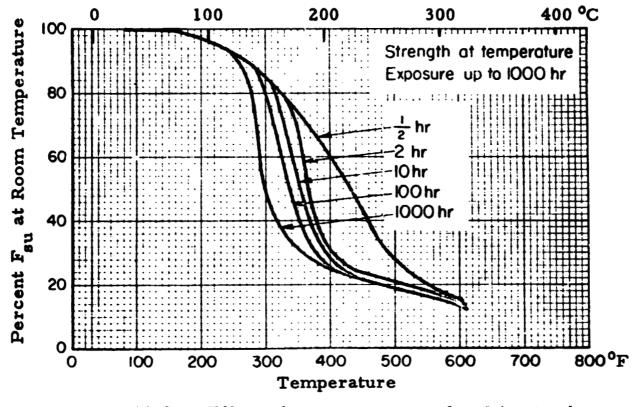
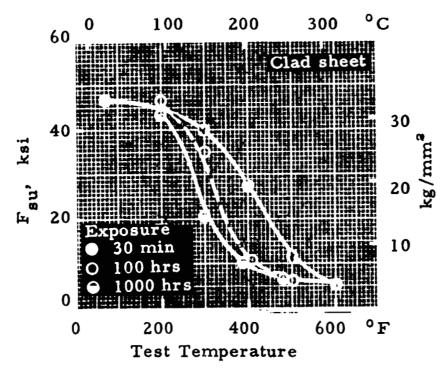
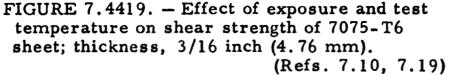
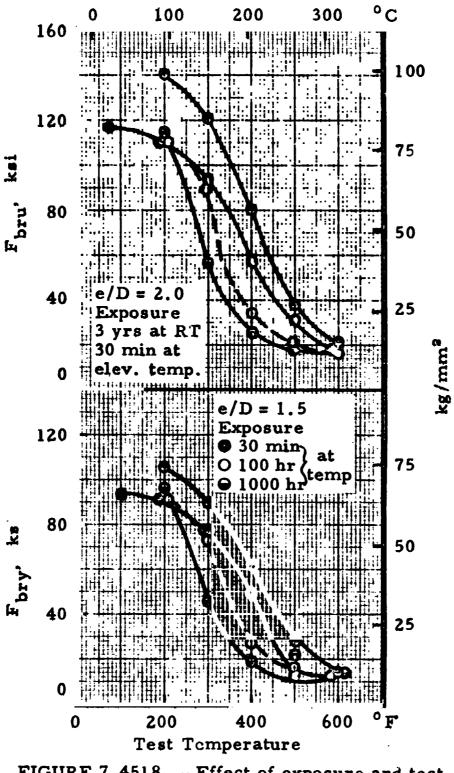


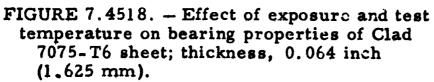
FIGURE 7.4418. - Effect of temperature on the ultimate shear strength of 7075-T6 (all products).

(Ref. 7.5)









(Refs. 7.10, 7.19)

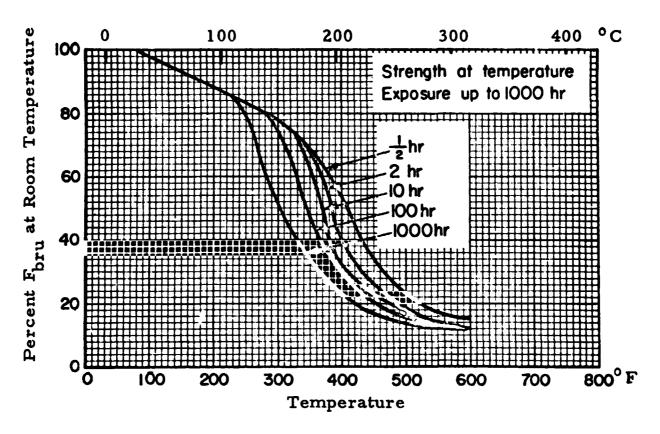


FIGURE 7.4519. - Effect of temperature on the ultimate bearing strength of 7075-T6 (all products).

(Ref. 7.5)

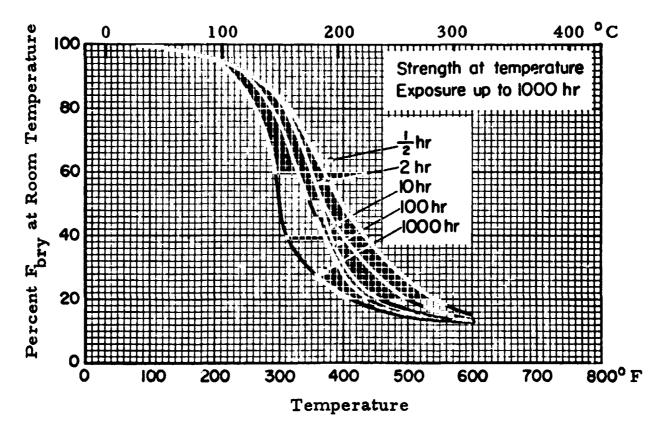
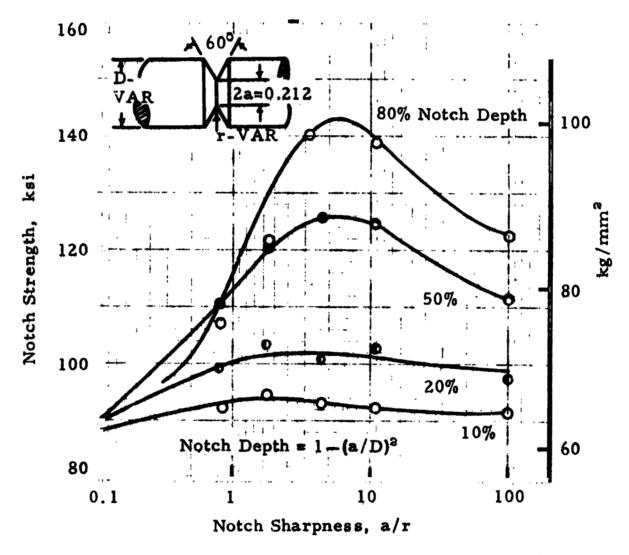
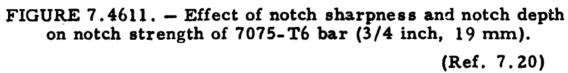
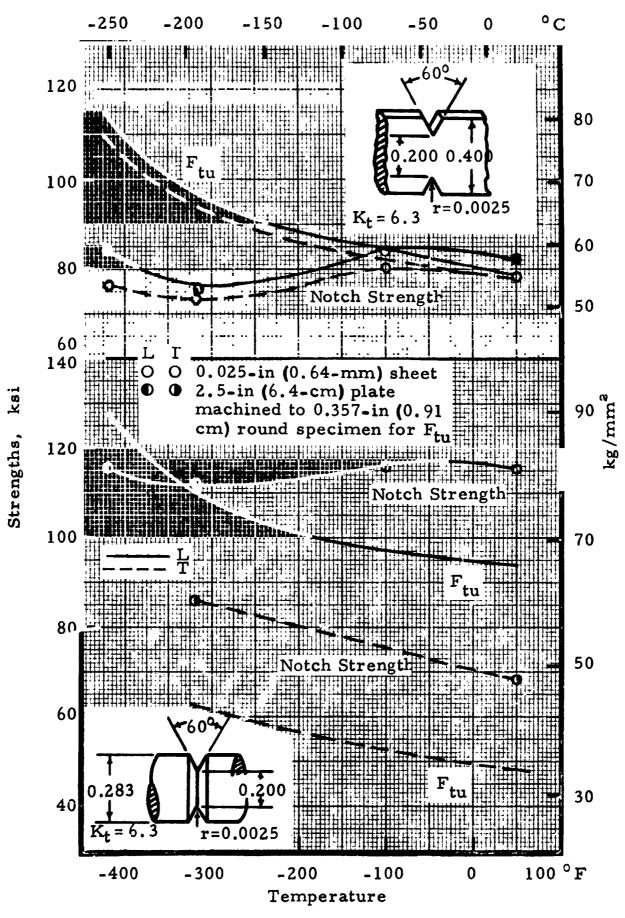


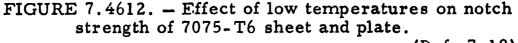
FIGURE 7.4520. - Effect of temperature on the bearing yield strength of 7075-T6 (all products).

(Ref. 7.5)

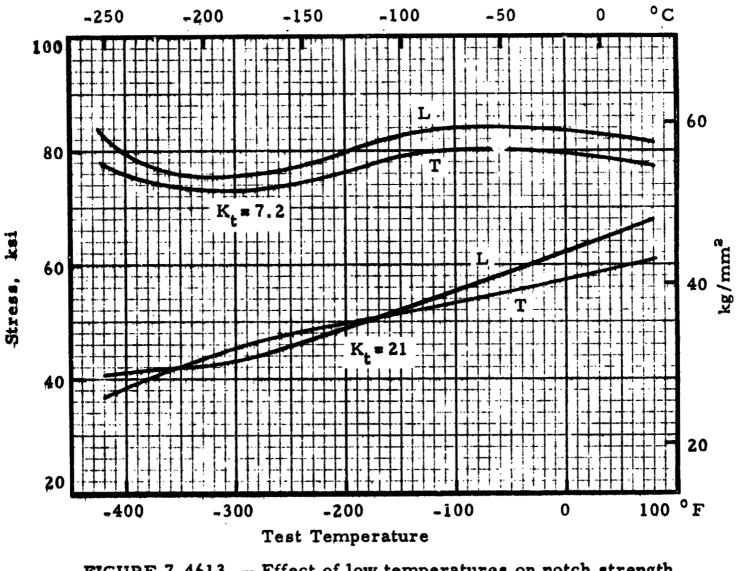


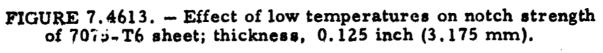




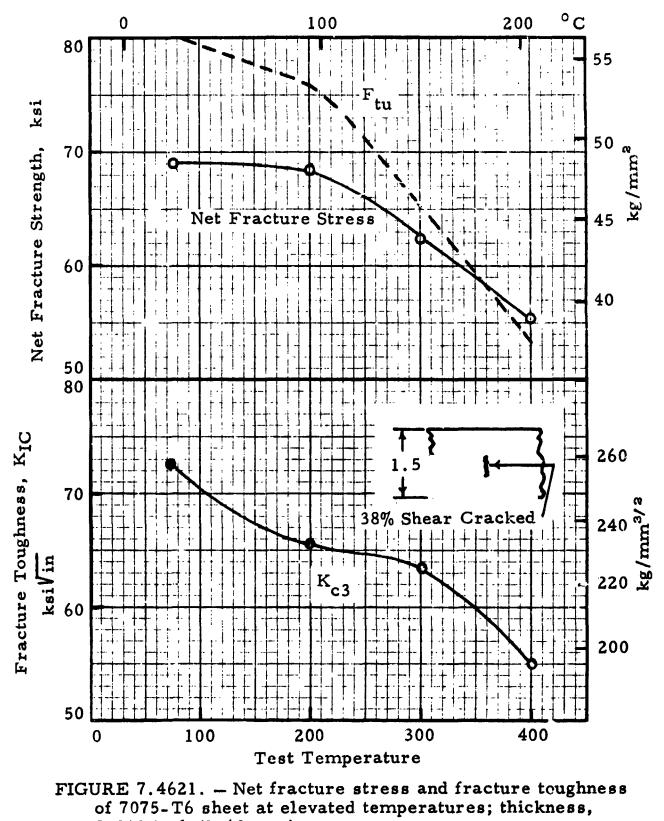


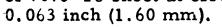
(Ref. 7.18)





(Ref. 7.11)





(Ref. 7.21)

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Chapter 8

DYNAMIC AND TIME DEPENDENT PROPERTIES

- 8.1 General. The room temperature strength of the 7075 alloy is among the highest attainable with aluminum alloys. Its elevated temperature strength, however, is inferior to other aluminumcopper alloys, such as 2014, 2024, and 2219.
- 8.2 Specified Properties
- 8.3 Impact
- 8.31 Low temperature impact strength of bar and rod in T6 condition, figure 8.31.
- 8.32 Effect of test temperature on impact strength of alloy in T6 condition, figure 8.32.
- 8.4 <u>Creep</u>
- 8.41 Creep-rupture
- 8.411 Creep and creep rupture curves for all products in T6 and T651 conditions (except extrusions and forgings), figure 8.411.
- 8,412 Creep and creep rupture curves for T6 and T6511 extrusions, figure 8.412.
- 8,42 Creep deformation
- 8.421 Short time total strain curves for clad sheet in T6 condition at 300° to 600° F (149° to 316° C), figure 8.421.
- 8.422 Isochronous stress-strain curves at 300° and 400° F (149° and 204° C), figure 8.422.
- 8.423 Master parameter curves for 0.5-percent total strain and creep rupture for clad sheet in T6 condition, figure 8.423.
- 8.5 Stability
- 8,51 Exposure effects
- 8.511 Effect of exposure to elevated temperatures on room temperature tensile properties of alloy in T6 condition, figure 8,5111.
- 8,6 Fatigue
- 8.611 Cryogenic fatigue strength data for bar and sheet, table 8.611.
- 8.612 S-N curves at low temperatures for sheet in T6 condition, figure 8.612,
- 8.613 S-N curves for extruded bar in T73 condition, figure 8.615.
- 8,614 Fatigue strength of smooth and notched bar at low temperatures, figure \$.614.
- 8,615 Rotating beam S-N fetigue data for plate, rod, and forgings in T73 condition, figure 8,617.
- 8.62 Stress range diagrams
- 8.621 Stress range diagram for bar and extrusions in T6 condition, figure 8.621.
- 8,622 Stress range diagram for smooth and notched bar and extrusions in T6 condition, figure 8.622.

Source				Ref. {	8.7					
Alloy				7075-T73	T73					
Form	Surface finish rms	Test Temperatu	perature °C	Fatigue test type	Fatigue str ksi (kg/mm ²) 10 ⁵ 10 ⁶	Fatigue strength (kg/mm ²) at cy 10 ⁶	rength at cycles 10 ⁷	Ratio, fatigue/ ultimate strength 10 ⁶ 10 ⁶ 10 ⁷	, fatigue, te streng 10 ⁶ 10	e/ lgth ()7
0.050-inch sheet (1.27-mm)	16	70 -320 -423	21 -196 -253	flexural	42 (29.5) 54 (38.0) 62 (43.6)	34 (23.9) 29 49 (34.5) 45 54 (38.0) -	29 (20.4) 45 (31.6)	0.57 0 0.57 0 0.60 0	0.46 0. 0.52 0. 0.52 -	33
	140	70 -320 -423	21 -196 -253	flexural	32 (22.5) 43 (30.2) 50 (35.2)	26 (18.3) 38 (26.7) 41 (28.8)	25 (17.6) 36 (25.3) 	0.44 0. 0.47 0. 0.51 0.	.36 0. .41 0.	35
7/8-inch bar (2.89-mm)	32	70 -320 -423	21 -196 -253	axial	52 (36.6) 44 66 (46.4) 59 78*(54.8) 66	44 (30.9) 42 (59 (41.5) 58 (66 (46.4) -	42 (29.5) 58 (40.8) 	0.69 0.73 0.74	0.59 0. 0.65 U. 0.62 -	. 56 . 64

TABLE 8.611. - Cryogenic Fatigue Strength Data for Sheet and Bar

*Maximum stress is equal to or above yield strength at temperature

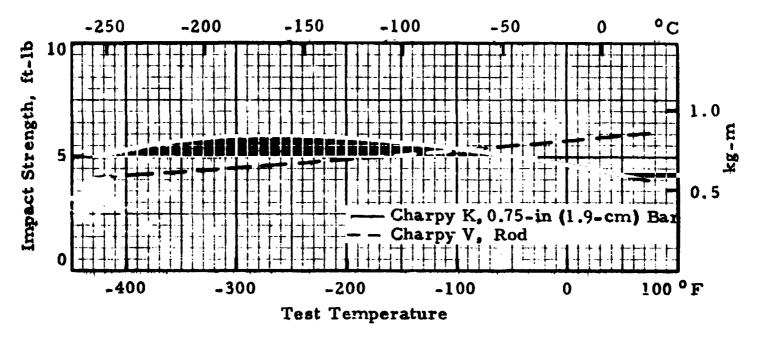


FIGURE 8.31. - Low temperature impact strength of 7075-T6 bar and rod.

(Ref. 8.2)

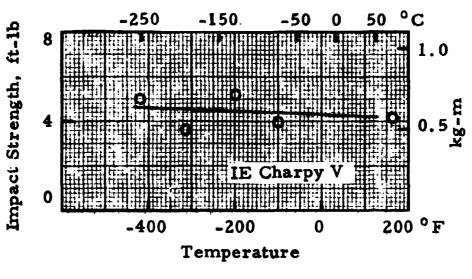


FIGURE 8.32. - Effect of test temperature on impact strength of 7075-T6. (Ref. 8.2)

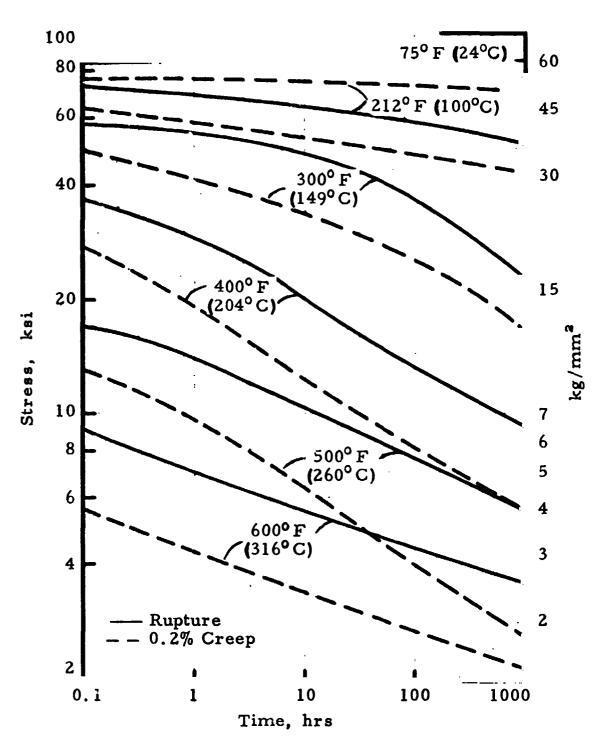


FIGURE 8.411. - Creep and creep rupture curves for all 7075-T6 and -T651 products (except extrusions and forgings).

(Ref. 8.13)

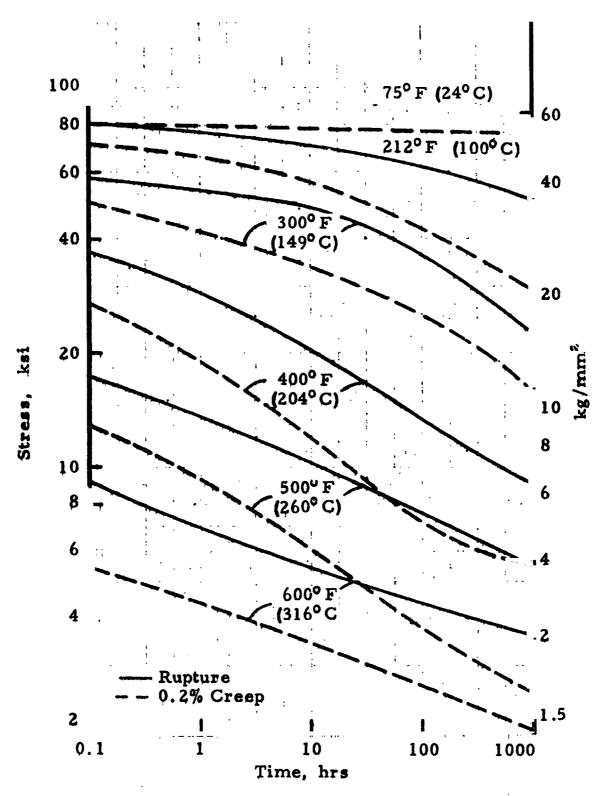
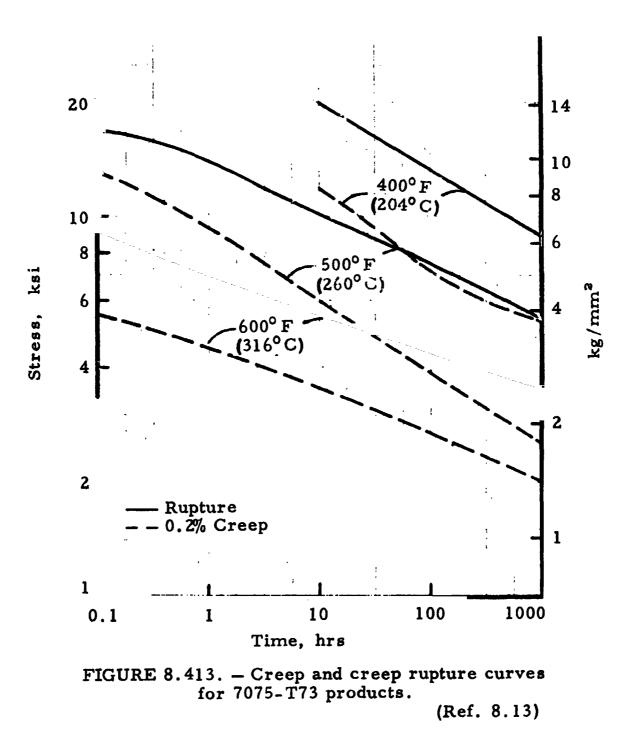
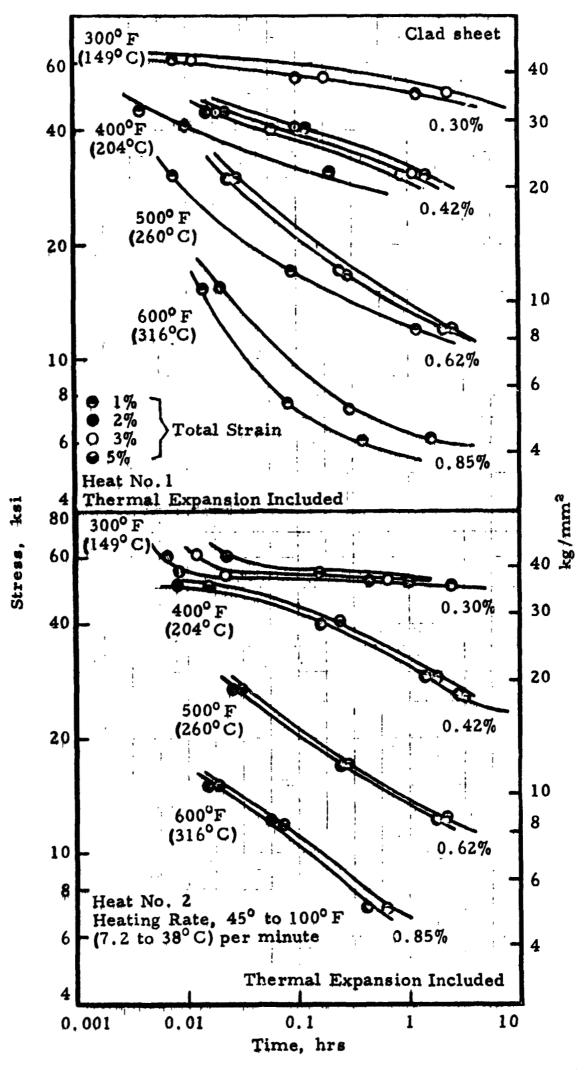


FIGURE 8.412. - Creep and creep rupture curves for 7075-T6 and -T6511 extrusions, 0.25 to 3.0 inches (6.35 to 76.2 mm) thick.

(Ref. 8.13)





F.GURE 8.421. - Short-time total strain curves for 7075-T6 sheet at elevated temperatures; thickness, 0.064 inch (1.625 mm). 95 (Ref. 8.5)

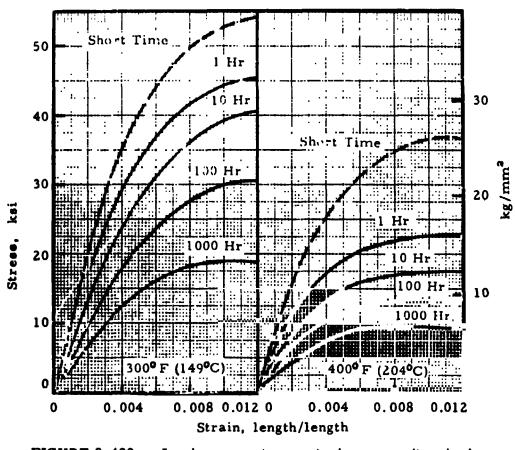
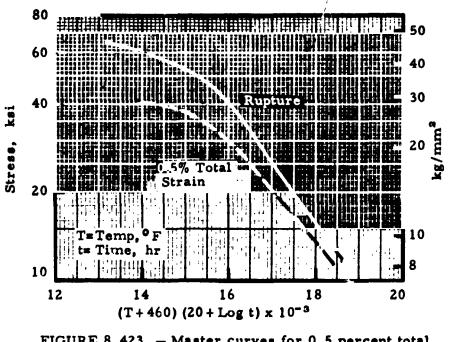
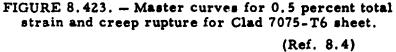
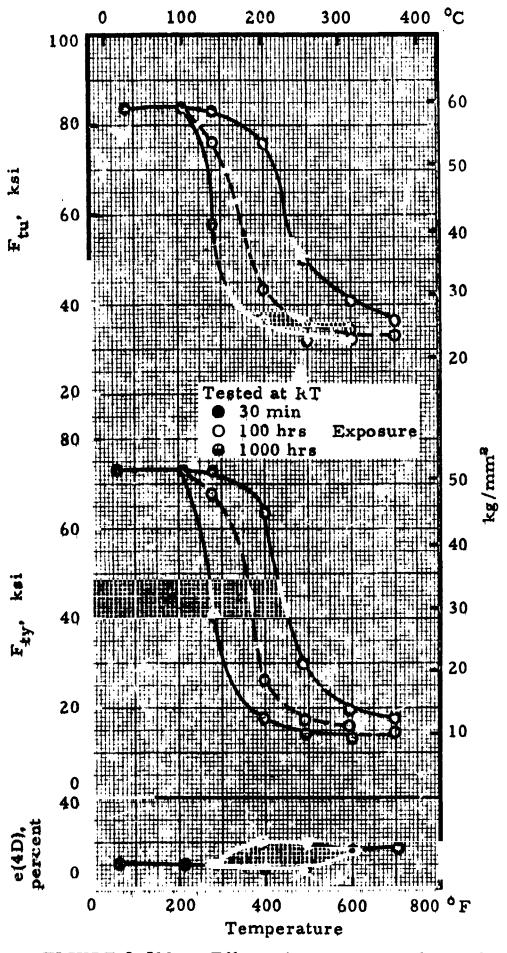


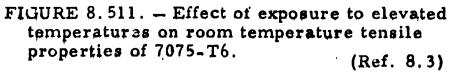
FIGURE 8.422. - Isochronous stress-strain curves (tension) at elevated temperatures for 7075-T6.

(Ref. 8.6)









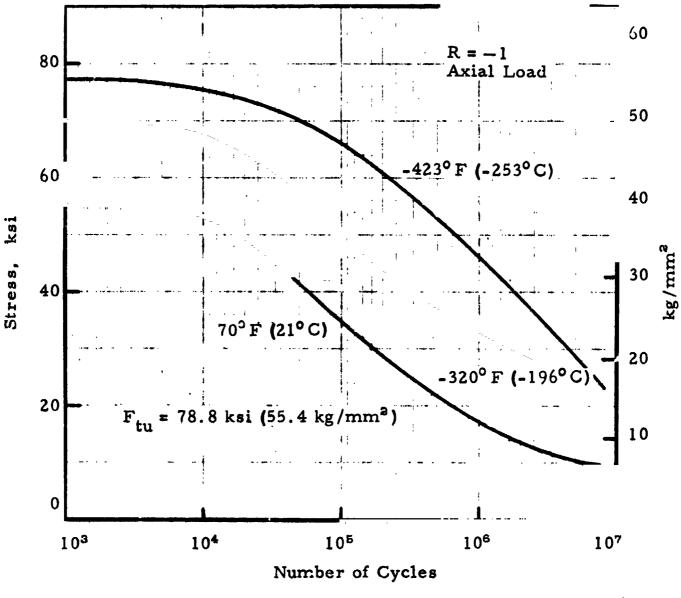
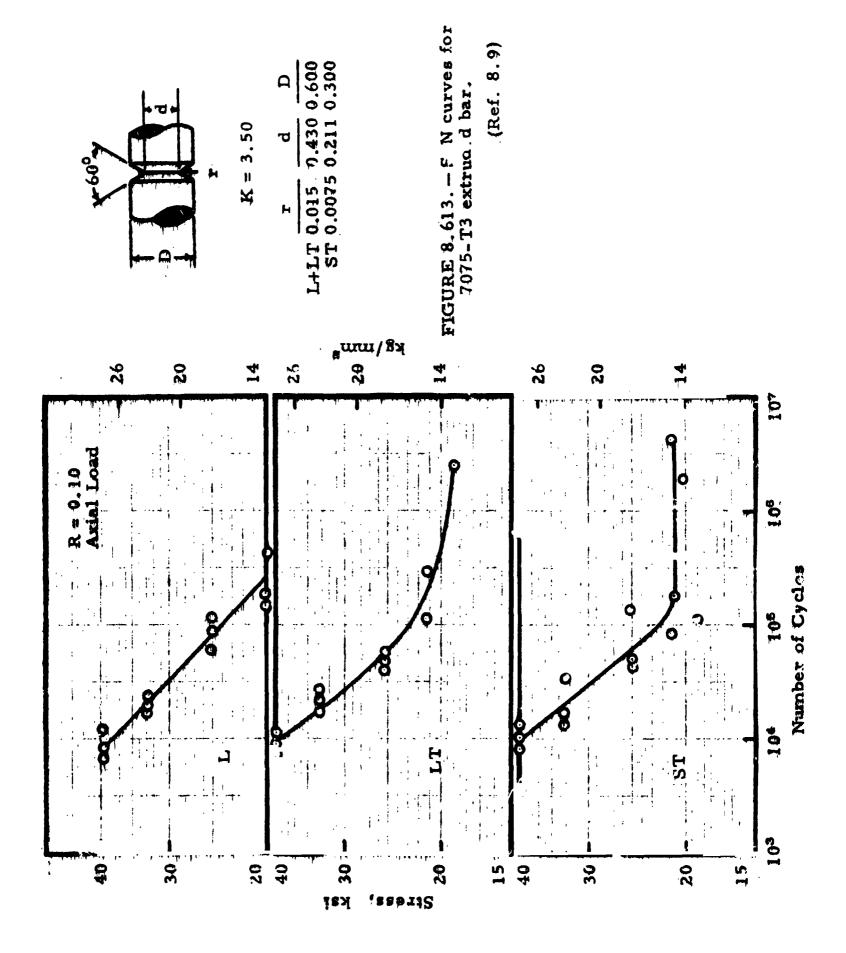
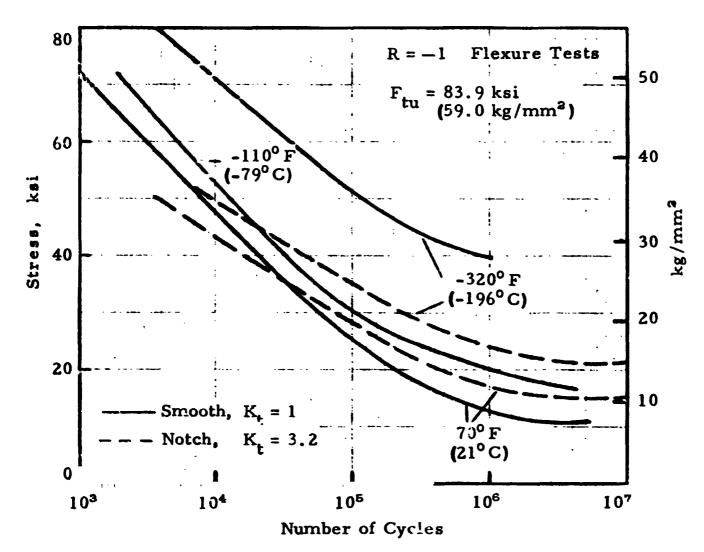
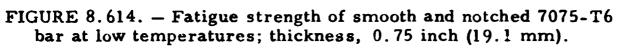


FIGURE 8.612. - S-N curves at low temperatures for 7075-T6 sheet; thickness, 0.100 inch (2.54 mm).

(Ref. 8.1)







(Ref. 8.1)

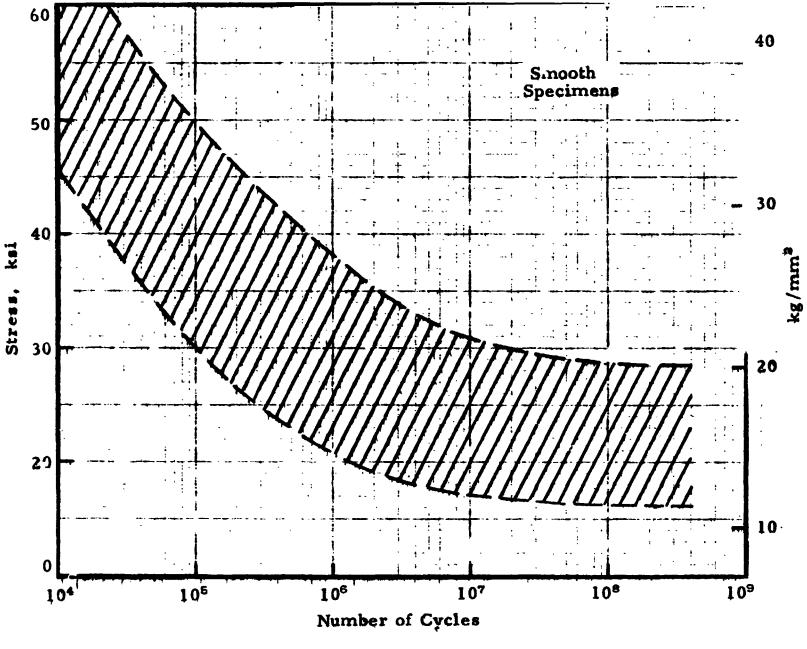
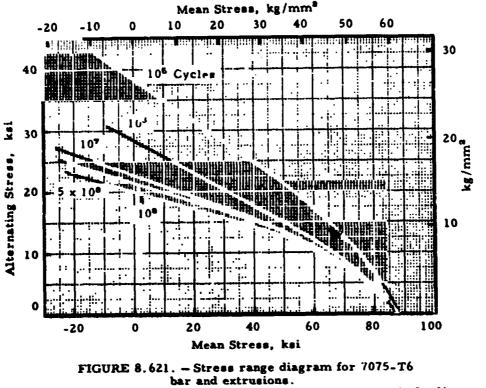
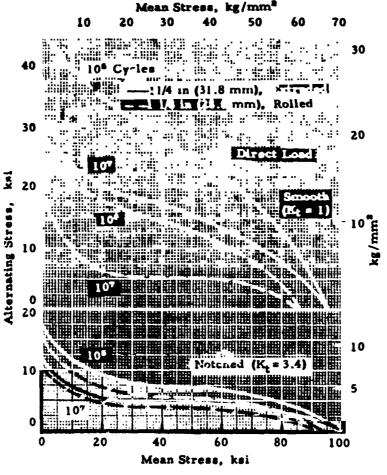


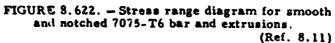
FIGURE 8.615. — Envelope of rotating beam S-N fatigue data points for 7075 plate, rod, and forgings in T73, T7351, and T6 conditions.

(Ref. 8.12)



(Ref. 8.10)





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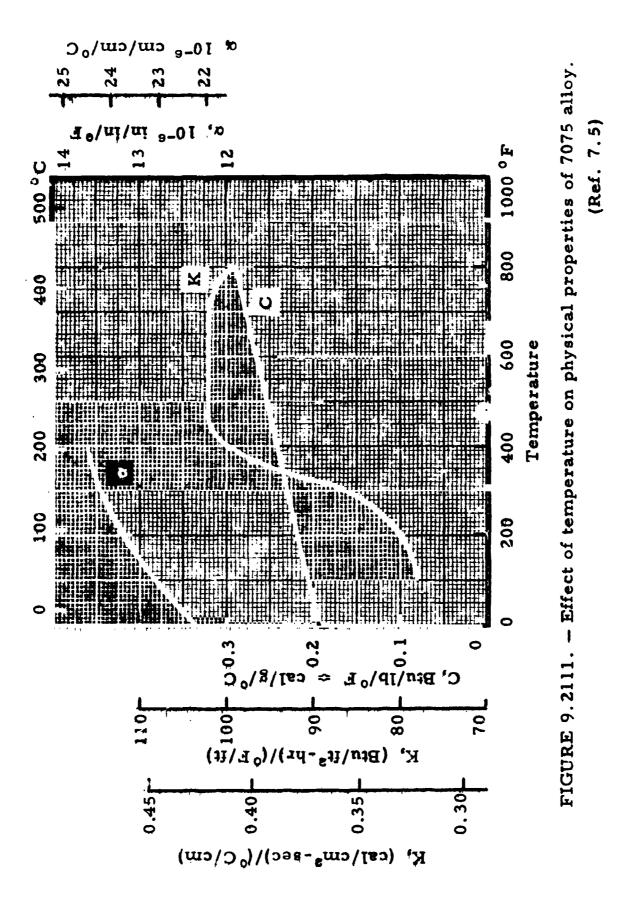
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Chapter 9

PHYSICAL PROPERTIES

- 9.1 Density $0.101 \text{ lb/in}^3 \text{ at } 68^{\circ} \text{ F} \text{ (ref. 9.2)}$ 9.11 **Specific Gravity** 2.80 g/cm³ at 20° C (ref. 9.2). 9.2 Thermal Properties Thermal conductivity (K) 9.21 T6 condition: 0.31 cal/cm/cm²/ $^{\circ}$ C/sec at 25 $^{\circ}$ C (ref. 9.2), 900 Btu/in/ft²/ $^{\circ}$ F/hr at 68 $^{\circ}$ F (ref. 9,2). 9.211 Thermal conductivity at various temperatures, figure 9.211. 9.22 Average coefficient of thermal expansion (α) 68° to 212° F 13.1 x 10⁻⁶ in/in/° F, 20° to 100° C $23.6 \times 10^{-6} \text{ cm/cm/}^{\circ}\text{C}$ (ref. 9.2). 9.221 Thermal expansion at various temperatures, figure 9.211. Specific heat (cp) 0.23 Btu/lb °F at 212°F, 9.23 0.23 cal/g °C at 100°C (ref. 9.3). 9.231 Specific heat at various temperatures, figure 9.211. Thermal diffusivity 9,24 9,3 **Electrical Properties** 9.31 Electrical resistivity T6 condition: 31 ohms-cir mil/ft at 68° F (ref. 9.2). 5.2 microhm-cm at 20° C (ref. 9.2), 9.32 Electrical conductivity T6 condition: 33% of IACS (equal volume) at 68° F (20°C), 105% of IACS (equal weight) $t 68^{\circ}$ F (20° C) (ref. 9.2). 9.4 Magnetic Properties Alloy is nonmagnetic. 9.5 Nuclear Properties
- 9.51 Aluminum alloys with high content of heavy metals such as zinc are not generally used in applications where a high neutron flux is present since certain isotopes of these heavier metals exhibit long "half-lives," leaving the part "hot" for extended periods.
- 9,52 Irradiation of 7075-T6 alloy with 5 x 10¹⁶ fast n/cm² at cryogenic temperatures affected tensile properties by only about five percent or less (ref. 9.4).

- 9.6 Other Physical Properties
- 9.61 Emissivity in air, 0.035 to 0.07 at 25°C (ref. 9.5).
 9.611 Emissivity is known to be a function of the wurface quality of a metal or alloy and the value is also influenced by environment.
- Damping capacity. 9.62
- 9.621 Damping capacity is a function of the hardness or temper of the alloy.



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Chapter 10

CORROSION RESISTANCE AND PROTECTION

10.1 General. Despite its high chemical reactivity and affinity for $\overline{0xygen}$, aluminum generally exhibits excellent corrosion resist, ance in most common environments because it passivates spontaneously under normal oxidizing conditions. The passive film is a hard, strongly adhering layer of aluminum oxide, estimated as 20-100 x 10⁻⁷ mm thick on aluminum exposed to air (ref. 10, 1), which protects the metal from direct attack. Thus, the corrosion rate of aluminum generally decreases with time, except under severe or specific exposure conditions which tend to disrupt the passive film.

> Outdoors, aluminum and its alloys weather to a pleasant gray color, with some initial superficial pitting which gradually ceases (ref, 10.2). Industrial soot, sulfur dioxide, sulfur trioxide, and marine spray tend to increase atmospheric corrosion, but hydrogen sulfide and carbon dioxide do not (ref. 10.3). Twenty-year tests at several marine, industrial and rural sites have shown that atmospheric attack on aluminum takes place principally in the first year and progresses very slowly beyond the second year (ref, 10.4). Even at high temperatures in dry atmospheres, aluminum is highly resistant to most common gases, except the halogens (ref. 10.2).

In aqueous environments, corrosion resistance of aluminum is greatest under neutral or slightly acidic or alkaline conditions, where the protective oxide film is most stable (pH 5.5-8.5 at room temperature, 4.5-7 at 95° C (refs. 10.1, 10.5). Strong alkalies and strong nonoxidizing acids destroy the oxide and greatly accelerate corrosion. Pitting attack occurs in waters containing chloride or other halogen ions, particulalrly at crevices or stagnant areas when passivity breakdown is accelerated by differential aerative effects. Traces of copper, iron, and mercury ions are also effective in promoting localized attack via galvanic cells formed between aluminum and meta. deposited by replacement reactions (ref. 10.1). Since aluminum is strongly anodic to most other common metals, galvanic coupling with them generally produces severe attack on the aluminum, especially in sea water (ref. 10, 2).

Aluminum and its alloys are rather resistant to most molten salts. However, molten metals generally attack aluminum, particularly zinc and tin, which form alloys (ref. 10.2). Even a small amount of mercury is especially harmful, since it breaks down passivity and amalgamates, causing rapid perforation of aluminum piping or sheet (ref. 10.1). Under some conditions, aluminum exhibits very poor resistance to chlorimated solvents and may even react explosively with them; however, such solvents, when properly inhibited, may be used for cleaning and degreasing without harm (ref. 10.6).

Aluminum purity significantly affects its corrosion resistance. High purity metal is more resistant than commercially pure aluminum, which in turn is generally more resistant than most alloys (ref. 10.1). Corrosion resistance of specific alloys is affected by composition, heat treatment, and stress conditions.

The anodic electrode potentials of aluminum alloys may cause them to corrode sacrificially when in contact with most other metals in corrosive environments. When possible, direct metallic contact with a more cathodic metal should be avoided. The 7075 alloy, containing both zinc and magnesium in solid solution in aluminum, has an electrode potential that is more anodic than that of pure aluminum. The electrode potentials of aluminum and some alloys are given in table 10.11.

The resistance to corrosion of the 7075 alloy and other aluminum alloys is affected by composition, heat treatment, and stress conditions, as discussed further below.

10.2 Resistance to Corrosion. The general resistance to corrosion of the 7075 alloy is good and its resistance is improved with heat treatment and artificial aging. Compared with other aluminum alloys, this alloy exhibits good corrosion resistance in rural atmospheres but is attacked by industrial and marine environments. The 7075 alloy, in general, is less resistant to corrosion in most other environments than the other wrought aluminum alloys. Corrosion resistance is improved by cladding; Alclad sheet and plate are available. The clad material normally used is a low-zinc alloy, 7072, which has a resistance to corrosion about equal to that of pure aluminum.

> A thermal treatment has recently been developed for the 7075 alloy which provides excellent resistance to corrosion. This new temper, designated T73, is highly resistant to stress-corrosion cracking (SCC), does not exfoliate, and is practically immune to intergranular corrosion. The general surface corrosion which occurs in severe environments is predominantly a pitting type for this temper (ref. 10.7). In die forgings, the T73 material is guaranteed to be capable of passing the accelerated stress-corrosion test specified in MIL-A-22771; SCC tests were conducted on 205 transverse specimens, from 66 lots, taken across the parting plane of 7075. T73 die forgings. The forgings were of various parts such as tubular fittings, hydraulic cylinders and landing gear sections of various sizes. The test employed was a 3.5 percent NaCl alternate immersion at a stress of 0.75 F_{ty} (42 ksi) for 84 days duration. No failures occurred 'n any of the 205 specimens (ref. 10.7).

In a seacoast atmosphere, 4 stressed specimens from T73 forgings did not fail after exposure for 46 months. However, 1 of 10 specimens exposed in an industrial environment failed in 28.6 months. Of the remaining 9 specimens, 4 did not fail in 47 months and 5 did not fail in 61 months. These results, and others, indicate that 7075-T73 has excellent resistance to stress corrosion in all directions with respect to grain orientation. It should be noted that the T6 temper exhibits high resistance in the longitudinal and long-transverse directions, but is susceptible to stress-corrosion cracking in the short-transverse direction.

The resistance to stress corrosion of 7075-T6 sheet and forgings, in various environments, is shown in table 10.21. Comparative lata for forgings and for commercial grade plate are presented in figures 10.21 and 10.22. Premium strength die forgings that provide mechanical properties equivalent to those of 7075-T6 and assure a stress-corrosion resistance approaching that of 7075-T73 have been developed and are offered in the 7075-T736 temper (ref. 10.14).

A study was made to ascertain the effect of short-time exposure of stressed tensile specimens of clad 7075-T6 sheet to a liquid fluorine environment. Specimens were tested in liquid nitrogen (-320°F, -196°C)to determine F_{tu} , F_{ty} , and elongation in a nonreactive environment. Similar determinations were made in an environment of liquid fluorine at -196°C, and the specimens were held at a stress equal to 0.9 F_{ty} for 2 hours before continuing the test to failure. A slight decrease in F_{tu} of about 3 percent and a decrease in clongation of about 24% were observed. It was believed that the indicated effect was due to contaminants in the fluorine environment (ref. 10.11).

The compatibility of engineering materials with cryogenic and noncryogenic propellants has been surveyed (ref. 10.12) and the report indicates that the 7075 alloy is compatible with the following propellants under most conditions for long-term applications:

Liquid oxygen (LOX); noncorrosive, but embrittlement may occur due to low temperature.

Aerozine-50 (50% hydrazine-50% UDMH) at 160° F (71°C),

Unsymmetrical Dimethylhydrazine (UDMH) at 160° F (71° C' maximum.

Hydrazine (N_2H_4) ; some authorities disagree, however.

Nitrogen tetroxide (N_2O_4); O and T6 conditions if less than 0.2 and 0.6% H₂O is present.

Pentaborane (B_5H_9) ; T6 condition.

The effect of neutron irradiation on the stress-corrosion of 7075-T6 in alternate-immersion tests (3.5% NaCl) is illustrated in figure 10.23.

The effect of shock-loading or the stress corrosion of 7075-T6 and -T73 in alternate-immersion tests (3.5% NaCl) is summarized in table 10.22.

Crack propagation rates for 7075-T6 in several environments are given in table 10.23.

10.3 Protective Measures. Anodic coatings are widely used for the corresion protection of aluminum alloys. These oxide coatings are hard and are abrasion- and corrosion-resistant. Cathodic protection has also proven effective in retarding both general dissolution and localized attack, although overprotection by this method should be availed to insure against narmful accumulation of alkali at the cathode surface (ref. 10.1).

It has been found that if 7075-T651 specimens are precorroded in 1 M NaCl (buffered to pH 4.7) and then dried thoroughly in a partial vacuum prior to corrosion testing under stress, the time to failure is greatly increased (ref. 10.15).

Painting, organic, and inorganic inhibitors have been applied with success in specific cases (refs. 10.2, 10.13).

The 7075 alloy is available as Alclad sheet and plate which consists of hare 7075 with a thin coating of 7072 alloy on one or both sumaces. The clad alloy is chosen to provide a surface having a high resistance to corrosion and sufficiently anodic to the 7075 core material to afford electrochemical protection.

The effects of various surface treatments on the resistance to corrosion of 7075-T6 forgings is summarized in table 10.31.

Surface treatments are discussed in greater detail in chapter 11.

Source	Ref. 10.8						
(Aqueous solu	(Aqueous solution of 53 g NaCl and 3 g H_2O_2 per liter)						
Al + Zn + Mg (4% Mg	Zn ₂ solid solution)	-1.07 Volt					
Al + Zn ', 70 Zn solid	solution	-1.05 Vclt					
Alclad 7075		-0.96 Vo::					
5456 Alloy		-0.87 Volt					
A1 (99.95+ %)		-0.85 Volt					
6961-76		-0.83 Volt					
7075-16		-0.81 Volt					
2014-T6		-0.78 Volt					
201-T4 and 2024-T4		-0.70 Volt					
fild Steel		-0.58 Volt					

TABLE 10.11. - Electrode Potentials of Aluminum and Some Alloys (a)

Source	urce Ref. 10.9					
Alloy	Environment (b)			Average Loss in Tensile Strength, Unstressed Stressed		
2014-T6	3-1/2% NaCl	84	0	42%	55%	
	Seacoast	365	0	18	28	
	Inland industrial	365	0	7	7	
22197 T81	3-1/2 % NaCl	84	0	21	26	
	Seacoast	365	0	6	8	
	Inland industrial	300	0	-	•	
2024 , T3	3-1/2% NaCl	84	0	33	40	
	Seacoast	365	0	16	20	
	Inland industrial	365	0	6	9	
7075- T 6	3-1,2% NaCl	84	6	13	22	
	Seacoast	365	10	7	10	
	Inland industrial	365	0	2	5	
7178-T6	3-1/2 % NaCl	84	0	14	24	
	Seacoast	368	20	8	18	
	Inland industrial	365	0	1	3	

TABLE 10.21, - Stress Corrosion Resistance of Aluminum Alloys (a)

(a) Specimens were taken from production sheet, 0.063-in (1,6-mm) thick.
(b) Specimens exposed to 3-1/2 % NaCl were alternately immersed.
(c) Stressed 75% of yield strength.

Source	Ref. 10.16						
Alloy	7075						
Billet	Heat-Treated Condition (a)	Time to Failur Unshocked (c)	re (av), days Shocked (b)				
2	solution-treated, 4-hr age at 250°F	10 (2)	2 (4)				
3	7075-T6	3 (4)	2 (4)				
3	7075-T6, overaged 1/2 hr at 350 ⁰ F	4 (4)	4 (4)				
3	7075-T6, overaged 1 hr at 350 ⁰ F	3 (4)	3 (4)				
3	7075-T6, overaged 1-1/2 hr at 350°F	6 (4)	3 (4)				
3	7075-T6, overaged 2-1/4 hr at 350° F	14 (4)	5 (3)				
2	7075-T6, overaged 3 hr at 350 ⁰ F	47 (1)	35 (1)				
2	7075-T6, overaged 5-1/2 hr at 350°F	71 (2)	32 (3)				
2	7075-T6, overaged 8-3/4 hr at 350° F	63 (2)	55 (2)				
1	7075-T73	51 (3)	40 (4)				

TABLE 10.22. - Effect of Shock Loading on Time to Failure of VariouslyAged Specimens in Alternate-Immersion Stress-Corrosion Tests

(a) 250° F = 121° C; 350° F = 177° C

(b) Shock-loaded at 204 Kb (about 2070 kg/mm^2)

(c) No. of tests in parentheses

Source	,	Ref. 10.13	1	··· · · · · ·
Alloy		7075-T6	ј 2001 — 4-11 - 1 - 1 	••••••••••••••••••••••••••••••••••••••
Stress Intensity,	Crack Prop	agation Rate,	Microinch	per Cycle (a)
ΔK, ksi Vin	Dry Air, <10% RH	Wet Air, >30% RH	Distilled Water	3.5% NaCl Solution
3.5	0.42	1.3	2,0	3,2
5.9	3.8 27	7.4 50	13 80	18 2100

TABLE 10.23. - Crack Propagation Rates ($\Delta 2a/\Delta N$) in Several Environments

(a) Axial loaded tests at 2 cycles/sec, R = 0.5

		مينا هم بال مربع بال يو يرب الم المرب الم المرب الم الم الم الم الم	
Source		Ref. 10,9	an in interpret of the number of the second
Alloy	7075-	T6 forgings (a)	an a state and a state a state In the state a s
Surface		and the second	Days
Treatment	Coating	3.5% NaCl, Alt. Im.	Ind. Atmos.
As machined	None	1, 5, 5, 17, 28	20, 37, 120, 161
Shot-blast (#230 steel shot)	None	ОК 365 (b), ОК 730 (b)	OK 3111
Grif, blasf (#25 steel grit)	None	5, 9, 11, 108, OK 182 (b)	1549, 1825, 2536
As machined	CrO ₃ anodic + paint (c)	OK 198 (b), OK 270 (b) OK 365, OK 1095 (b)	1493
Grit blast (#25 steel grit)	CrO_3 anodic + paint (d)	1395, OK 1825 (d)	<u>OK 34-24-945-244</u> (a)
Grit blast (#25 steel grit)	7072 metal spray (1 to 3 mils)	182, 1469, 2 OK 1095 (b), 1 OK 1825	268, 3 OK 3471
Grit blast (#25 steel grit)	7072 metal spray + paint (e)	ОК 806 (ђ)	OK 3471

TABLE 10.31. - Effect of Surface Treatment on Resistance to Stress Corrosion

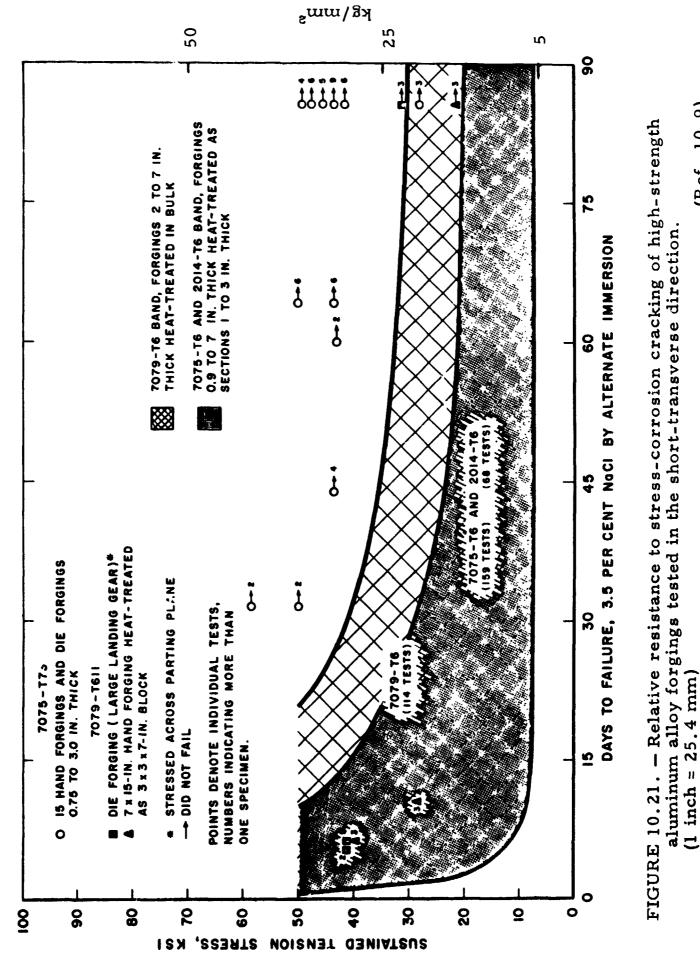
Linn das das de la construction de la const (a) 6x15x20 inches specimen; short transverse 0,437-in diameter tension bar; stress, 75% of yield strength. All grit and shot-blasting done prior to stressing. (15x38x51 cm spec; 1.1-cm tens bar.)

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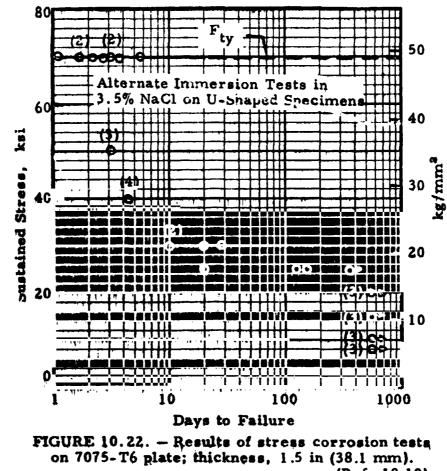
(b) Removed from test because specimen fractured in threaded grip,

(c) Zinc chromate primer plus coat of aluminum paint,

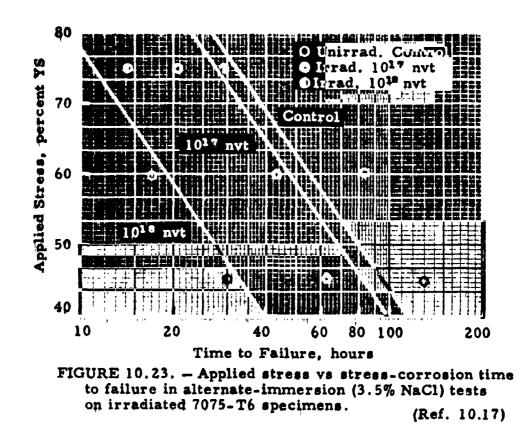
(d) Zinc chromate primer plus two coats of aluminum paint.







(Ref. 10.10)



Chapter 10 - References

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- 10.2 L.L. Shreir, <u>Corrosion</u>, John Wiley & Sons, New York, 1963, Vol. I, Sec.4.1.
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- 10.4 Symposium on Atmospheric Corrosion of Non-Ferrous Metals, ASTM STP-175, 1956.
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- 10.7 P. L Mehr et al., "Alcoa Alloy 7075-T73," Alcoa Green Letter, Aluminum Co. of America, August 1965.
- 10.8 <u>Meials Handbook</u>, Vol. 1, "Properties and Selection of Metals, 8th Edition, American Society for Metals, 1961.
- 10.9 D.O. Sprowls and R.H. Brown, "What Every Engineer Should Know About Stress Corrosion of Aluminum," Metal Progress, <u>81</u>(4), 1962; <u>81</u>(5), 1962.
- 10.10 H.R. Pritchard, "Stress-Corrosion Tests on Commercial and High Purity Grade 7075-T6 Aluminum Alloy," Frankford Arsenal, Memo Report M65-17-1, May 1965.
- 10.11 H. T. Richards and M. P. Hanson, "Influence of Fluorine Environment on the Mechanical Properties of Several Sheet Alloys," NASA TN D-1706, April 1963.
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- 10.13 D. N. Williams, "Environmental Corrosion-Fatigue Behavior of Aluminum Alloys," DMIC Memorandum 249, June 1970.
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- 10.16 A.J. Jacobs, "The Mechanism of Stress-Corrosion Cracking in 7075 Aluminum," presented at Conference on Fund. Aspects of Stress-Corrosion Cracking, Ohio State Univ., Columbus, Ohio, 11-15 September 1967; see also NASA Tech Brief 70-10527, October 1970.
- 10.17 A.J. Jacobs and G.G. Bentle, "The Effects of Point Defects and Dislocations on the Stress-Corrosion Susceptibility of Aluminum Alloys," Final Report under Contract NAS8-20471, 29 April 1969; see also NASA Tech Brief 70-10506, September 1970.

Chapter 11

SURFACE TREATMENTS

- 11,1 General. A wide variety of surface treatments can be applied to the 7075 alloy (and other aluminum alloys) to protect and improve the appearance of the surface. These include mechanical, chemical, and electrochemical finishes as well as organic, porcelain, and paint coatings Alclad forms of aluminum alloys have a very high inherent resistance to corrosion and may be used without benefit of protective coatings for some applications (refs. 11.1, 11.8). (See also table 10.31.)
- 11.2 Alclad Products. The 7075 alloy is available as Alclad sheet and plate, which consists of bare 7075 core material clad with a thin coating of 7072 alloy on one or both sides. The clad material is metallurgically bonded to the core material and is chosen to provide a surface having a high resistance to corrosion and sufficiently anodic to the 7075 core to afford electrochemical protection to it in corrosive environments. Consequently, any spot of attack can penetrate only as deep as the core alloy where further progress is stopped by cathodic protection. Corrosion is thus confined to the clad material only. The life of the cladding is a function of its thickness and the severity of the environment. Alclad products, therefore, limit corrosion to a relatively-thin clad surface layer (ref. 11.2).
- 11.3 Mechanical Finishes. Mechanical finishes are used to alter the texture of the alloy surface to provide a more decorative appearance or as a treatment prior to other finishing such as painting. Grinding, polishing and buffing result in smoother reflective surfaces. Abrasive blasting (sand or grit) gives a rough matte finish which is often used as a base for organic coatings. Scratch finishing, satin finishing, Butler finishing, and skin finishes are scratchedline finishes which remove minor surface defects and provide a decorative effect. Mechanical methods remove the original heavy oxide film. For this reason, mechanically finished parts are often given a protective coating by anodizing or lacquering. The possibility of generating an explosive mixture of fine powder and air during mechanical finish operations should be recognized (ref. 11.3).
- 11.4 <u>Anodizing</u>. Anodic coatings are hard, abrasion and corrosion resistant oxide coatings. The alloys can be anodically coated in a number of electrolytes, but most commerical anodizing is done by either the sulfuric acid or chromic acid process. The thickness of the coating is dependent upon the anodizing time. Coatings produced by the sulfuric acid process vary in thickness from 0.0001 to 0.001 inch (0.0025 to 0.025 mm). Coatings produced in chromic acid vary from 0.00001 to 0.00009 inch (0.00025 to 0.0023 mm). Anodic coatings provide good protection against corrosion and are excellent

bases for paint coatings (ref. 11.1). However, the chromic acid process does not provide as corrosion-resistant a coating as does the sulfuric acid process (ref. 11.10).

11.41 In recent years a number of new methods have been developed for producing heavier anodic coatings of from 0.001 to 0.010 inch (0.025 to 0.25 mm). These methods require electrolytes which enable the oxide growth process to continue until the desired coating is obtained.

Another recent development in coatings is that of hard anodizing, designated as "hardcoatings." Processes most suitable for a wide range of applications are Alumilite 226 [oxide coatings, 0.002 inch (0.050 mm) thick] and Martin Hardcoat [coating thicknesses up to 0.004 inch (0.10 mm)]. A flash hardcoat of a very thin film can also be applied by these methods by shortening the normal time cycle. The operating conditions for the two baths employed for these processes are given in table 11.1. The Martin process should be specified where maximum hardness and corrosion resistance are required along with thickness buildups to 0.004 inch (0.10 mm). Alumilite 226 is selected where hardness and corrosion resistance are required along with thickness and corrosion resistance are required and 0.002 inch (0.05 mm) is the acceptable maximum buildup. Further details of these processes are given in reference 11.9.

11.5 Chemical Finishes. Chemical finishes are of three main types. Finishes used for decorative effects include caustic etching, acid etching, and chemical polishing. Etched surfaces have a matte appearance while chemically polished surfaces are highly reflective and require protection by anodizing or lacquering.

Conversion coatings can be oxide, phosphate, or chromate types and are used primarily as base coatings prior to application of organic coatings. Miscellaneous special-purpose finishes include those produced by the Alrok process, modified Bauer-Vogel process, and processes for staining aluminum alloys.

- 11.6 <u>Electropolishing</u>. This process produces a highly reflective surface and is often used for surface preparation prior to microscopic examination of metallurgical structure.
- 11.7 <u>Electroplating</u> of aluminum alloys has gained increased commercial use in recent years. A commonly used finish consists of successive deposits of copper, nickel, and chromium. Other metals may be applied over the copper. A satisfactory base surface for electroplating is provided by immersing the aluminum part in a solution of sodium zincate of controlled composition. Brass, iron, silver, or chromium can be applied directly over this zinc immersion coating (ref. 11.4).

11.8 Painting. When severe conditions of exposure are to be encountered, it is frequently desirable to protect aluminum alloy surfaces with paint. Prior to painting, the surface should be properly prepared before priming. A clean and dry surface, one free from grease, dirt, dust, moisture, and foreign matter, is of prime importance. For adequate adhesion and maximum corrosion protection, a chemical conversion coating per Mil-C-5541 or anodic coating per Mil-A-8625 should be applied. The properly treated surfaces are then primed with a zinc chromate primer per Mil-P-8585; for severe conditions of exposure, both primer and joint compound should be used at joints. For less severe environments or wherever it is impractical to apply the pretreatment coatings, a chem, ical cleaning per Mil-M-10578B (phosphoric acid metal conditioner) or a mild mechanical cleaning are sometimes employed. These are followed by a chromate pigmented primer,

All surfaces except contacting surfaces may be given a second coat of paint consisting of two pounds of aluminum paste pigment (ASTM Spec. D962, Type II, Class B) per gallon (0.24 g/ml of varnish which meets Federal Spec. TT-V-86b, Type II or equivalent. The final assembled structure may be finished with one coat of aluminum paint. One or more coats of alkyd base enamel (pigmented to desired color) may be substituted for aluminum paint (ref, 11,5).

- 11.81 To minimize stress-corrosion cracking when the alloy is subjected to sustained surface stresses and corrosive environments, certain surface treatments and protective coatings are effective. The most effective protection is obtained by applying a topcoat of epoxypolyamide paint to shot-peened or metallized surfaces of the alloy. Satisfactory temporary protection is obtained by an electroplated galvanic coating (0.08 to 0.10 mm), or a topcoat of paint containing epoxy-polyamide or polyurethane resins. The former is preferred and can be used on unprimed surfaces. Care is necessary to prevent broaking or scratching the paint film. Shot peening alone will provide good surface protection (if all surfaces are treated) when corrosive environment is not severe. Anodic films and zinc-rich paints are the least effective coatings for prefenting stress-corrosion cracking (ref. 11.6).
- 11.9 Porcelain Enameling. The principal difference between porcelain enameling of aluminum alleys and other metals is the use of porcelain frits, which melt at lower temperatures. High-lead frits are commonly used and they can be formulated in a wide variety of colors and surface finishes. The enamel slip is sprayed onto chemaically cleaned and treated surfaces and then fired at temperatures of 950° to 1050° F (510° to 566° C) for a period of 4 to 8 minutes (ref. 11.7).

Source	Ref. 11.9		
Alloy	Aluminum Wrought Alloys		
Parameters	Pro	ocess	
Parameters	Martin (a)	Alumilite (b)	
Composition	15% H ₂ SO ₄	12% H ₂ SO ₄ , 1% H ₂ Cr ₂ O ₄	
Electrolyte Temp., [°] F , °C	25 to 32 4 to 0	48 to 52 9 to 11	
Current Density	25 asf	36 asf	

TABLE 11.1. - Baths for Hard Anodized Coatings

(a) Developed by the Martin Company

(b) Developed by the Aluminum Company of America

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Chapter 11 - References

- 11.1 <u>1971 SAE Handbook</u>, Society of Automotive Engineers, New York.
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Chapter 12

JOINING TECHNIQUES

12.1 General. The 7075 alloy is not considered to have good fusion weldability in any temper and fusion welding of this alloy is not normally recommended. The alloy is also difficult to weld in the annealed condition (O temper) by resistance welding techniques. Resistance welding of all heat treated tempers, however, is successfully accomplished if special practices are employed.

> Brazing, gas welding, or soldering of 7075 is not recommended, as satisfactory methods have not as yet been developed for this alloy. The alloy can be satisfactorily joined by riveting or bolting (refs. 12.4, 12.5, 12.7).

- 12.2 Welding
- 12.21 <u>Fusion Welding</u>. Fusion welding of the 7075 alloy is not normally recommended, as satisfactory methods have not been developed for this alloy.
- Electrical Resistance Welding. Resistance welding (spot welding 12.72 or seam welding) is a most useful and economical method of joining aluminum alloys. Satisfactory spot or seam welds are made in 7075 material in all heat treated tempers by resistance methods, but special practices are required. Mechanical or chemical cleaning of the contact surfaces is necessary to obtain consistent and sound welds. In aircraft construction, it is recommended that the contact resistance of the elements to be joined be continually checked to ensure surface cleanliness. For best results, the surface contact resistance should not exceed 50 microhms. Details on surface cleaning are given in reference 12.4. The choice of the type of resistance welding machine for spot or seam welding of aluminum alloys depends partly on the power supply, its voltage drop characteristics, demand limitations and other similar factors. A more detailed discussion of resistance welding equipment is given in references 12.4, 12.8,
- 12.23 Mechanical Properties of Spot Welds, The strength of spot welded joints depends to a large extent upon the static strength of each single weld spot. The statis strength of typical single spot welds in tension is given in figure 12.1 for clad 7075-T6 sheet of various thicknesses. The maximum static strength of spot welded joints, and corresponding maximum spot weld pitch for clad T6 sheet, are summarized in table 12.1.

The suggested minimum joint overlap and spacing of spot welds is given in table 12.2, and the minimum allowable edge distance for spot-welded joints is shown in table 12.3.

Spot weld maximum design shear strength in panels is presented in table 12.4 for bare and clad alloys. The efficiency of the parent metal in tension for various spot weld spacings is given in figure 12.2

The use of spot welds on military structural parts is governed by the requirements of the procuring or certificating agency (ref. 12.6). The requirements for equipment, mateirals, and production control of spot and seam welds in aluminum alloys is covered by military specification MIL-W-6858C.

- 12.3 Brazing. Brazing or soldering of the 7075 alloy is not recommended (ref. 12.5).
- 12.4 Riveting. Riveting is the most commonly used method for joining this alloy. Riveting methods are highly developed and are largely independent of the operator's skill. Thus, uniformity of riveted joints can be readily attained (ref. 12.10). Specifications for riveting of aluminum alloys are listed in table 12.5.
- 12.41 Aluminum alloy rivets are preferred for the fabrication of aluminum alloy structures, although cold-driven annealed steel rivets have been used successfully for certain applications. To determine the strength of riveted joints, it is necessary to know the strength of the individual rivet. In most cases, failure of such joints occurs by shearing, in bearing or tearing of the sheet or plate. Table 12.6 gives the average shear strength of driven rivets of various aluminum alloys. These values may be considered representative of properly driven rivets, although occasional driven rivets may fall below the average by 5 or 10 percent. It is customary to use a slightly larger factor of safety for the shear strength of rivets than is employed for other parts of an assembly. The design of joints where rivets are subjected to tensile loads should be avoided. Bolted connections may be used where high tensile stresses preclude the use of riveting. Information in greater detail on the riveting of aluminum alloys is given in references 12.10 and 12.11. Design data on mechanical joints using rivets or bolts may be found in Military Handbook-5A (ref. 12.6).

Source	Ref. 12.6					
Alloy		Clad 7075	5-T6			
Thickness of	Single Ro	w Joints	Multiple	Row Joints		
Thinnest Sheet, inch (b)	Strength 1b/in (c)	Pitch, inch	Strength lb/in (c)	Pitch/No. of Rows, inch		
0.010	288	0.167	438	0.110		
0.012	346	0.173	526	0.114		
0.016	461	0.191	701	0,126		
0.020	5 77	0.194	876	0.128		
0,025	721	0.205	1095	0.135		
0.032	923	0.225	1402	0,148		
0,040	1059	0.261	1752	0.158		
0,050	1230	0.302	2190	0.170		
0.063	1452	0.369	2759	0.194		
0.071	1589	0,415	3110	0.212		
0.080	1742	0.471	3504	0.234		
0.090	1913	0.525	3942	0.255		
0.100	2084	0,572	4380	0.272		
0.112	2289	0.622	4096	0.290		
0.125	2511	0.675	5475	0.310		

TABLE 12.1. - Maximum Static Strength of Spot-Welded Joints and Corresponding Maximum Spot-Weld Pitch (a)

(a) For multiple row joints, row spacing is at minimum and same pitch in all rows.

(b) 1 inch = 25.4 mm.

(c) 1 1b = 0.454 kg.

TABLE 12.2,	- Suggested	Minimum	Joint (Overlap

and	Spacing	of	Spot	Welds

Scurce	Ref. 12.4			
Alloy	Aluminum Alloys			
Thinnest Sheet in joint, inch (a)	Minimum joint overlap, inch	Minimum weld spacing, inch		
0,016 0.020 0.025 0.032 0.040 0.051 0.064 0.072 0.081 0.091 0.102 0.125	5/16 3/8 3/8 1/2 9/16 5/8 3/4 13/16 7/8 15/16 1 1 1/8	3/8 3/8 3/8 1/2 1/2 5/8 5/8 5/8 5/8 3/4 3/4 7/8 1 1 1/4		

(a) 1 inch = 25.4 mm

Source		Ref. 12.6				
Alloy	Alur	Aluminum Alloys				
Nominal thickness of the thinner sheet, inch Edge distance, E, in						
)16)20)25)32)36)40)45)50)63)71)80)90 100	3/16 3/16 7/32 1/4 1/4 9/32 5/16 5/16 3/8 3/8 13/32 7/16 7/16				
0.0 0.0 0.1 0.1)71)80)90 100	3/8 13/32 7/16				

TABLE 12.3. - Minimum Allowable Edge Distances for Spot-Welded Joints (a, b, c, d)

- (a) Intermediate gages will conform to the requirement for the next thinner gage shown.
- (b) Edge distances less than those specified above may be used provided there is no expulsion of weld metal or bulging of the edge of the sheet or damage to bend radii by electrode.
- (c) Values may be reduced for nonstructural applications or applications not depended on to develop full weld strength.
- (d) 1 inch = 25.4 mm.

Source	Ref. 12.6				
Alloy	Alumi num Alloys				
Nominal thickness	Mater	ial ultimat	e tensile stro	ength (d)	
of thinner sheet,	<u>></u> 56	20 to 56	19.5 to 28	< 19.5	
inch (e)	ksi	ksi	ksi	ksi	
0.010	48	40	_	_	
0.012	60	52	24	16	
0.016	88	80	56	40	
0.020	112	108	80	64	
0.025	148	140	116	88	
0.032	208	188	168	132	
0.040	276	248	240	180	
0.050	372	344	320	236	
0.063	536	488	456	316	
0,071	660	576	516	360	
0.080	820	684	612	420	
0.090	1004	800	696	476	
0.100	1192	936	752	540	
0.112	1424	1072	800	588	
0.125	1696	1300	840	628	
0.160	2496	1952	-	-	
Q.190	3228	2592	-	-	
0.250	5880	5120	-	-	

TABLE 12.4. -Spot Weld Maximum Design Shear Strength

in Panel for Bare and Clad Aluminum Alloys (a, b, c)

- (a) The reduction in strength of spotwelds due to cumulative effects of time - temperature - stress factors is not greater than the reduction in strength of the parent metal.
- (b) Strength based on 80 percent of minimum values specified in MIL-W-6858.
- (c) The allowable tensile strength of spotwelds is 25 percent of the shear strength.
- (d) 1 ksi = 0.70307 kg/mm[^]
- (e) 1 inch = 25.4 mm.

Source	Refs, 12.1, 12.3		
Products	Federal	Military	AMS
Rivets	FF-R-556a	MIL-R-1150A-1	7220C
	-	MIL-R-5674B-1 MIL-R-12221B	7222C 7223
Rivets, blind	-	MIL-R-7885A-1 MIL-R-8814-1	-
	-	MIL-R-27384	-
Rivet, wire	QQ-A-430-1	-	-

 TABLE 12.5. - Specifications for Aluminum Rivets

TABLE 12.6. – F_{su} (Average) for Driven Rivets (c)

Source	Ref. 12.11			
Alloy and Temper before Driving (a)	Driving Procedure	Alloy and Temper after Driving	F _{su} (av) ksi (1)	
1100-H14 2017-T4 2017-T4 2024-T4 2117-T4 5056-H32 6053-T61 6061-T4 6061-T4 6061-T6 7277-T4	Cold, as received Cold, as received Cold, immediately after quenching Cold, immediately after quenching Cold, as received Cold, as received Cold, as received Cold, immediately after quenching Hot, 990° to 1050° F (532°-566° C) Cold, as received Hot, 850° to 975° F (454°-524° C)	1100-F 2017-T3 2017-T31 2024-T31 2117-T3 5056-H321 6053-T61 6061-T31 6061-T43 6061-T43 6061-T6 7277-T41	11 39 34(b) 42(b) 33 30 23 24(b) 24(b) 30 38	

(a) These designations should be used when ordering rivets.

- (b) Immediately after driving, the shear strengths of these rivets are about 75% of the values shown. On standing at ambient temperatures, they age harden to develop full shear strength. This action takes about 4 days for 2017-T31 and 2024-T31 rivets. Values shown for 6061-T31 and 6061-T43 rivets are attained in about 2 weeks. Values of 26 ksi are attained by 6061-T31 rivets about 4 months after driving. Values shown for 7277-T41 rivets are attained in about one week.
- (c) These values are for rivets driven with core point heads. Rivets driven with heads requiring more pressure may be expected to develop slightly higher strengths.
- (d) 1 ksi = 0.70307 kg/mm^3 .

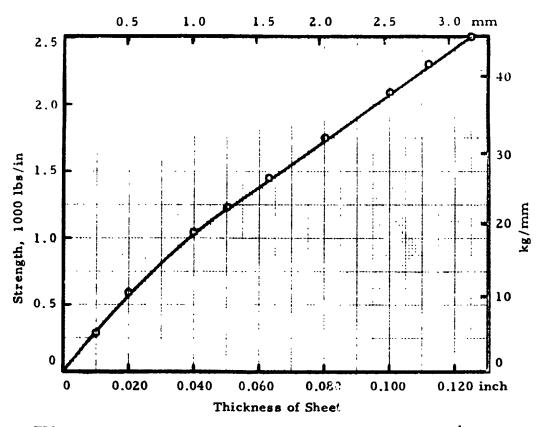
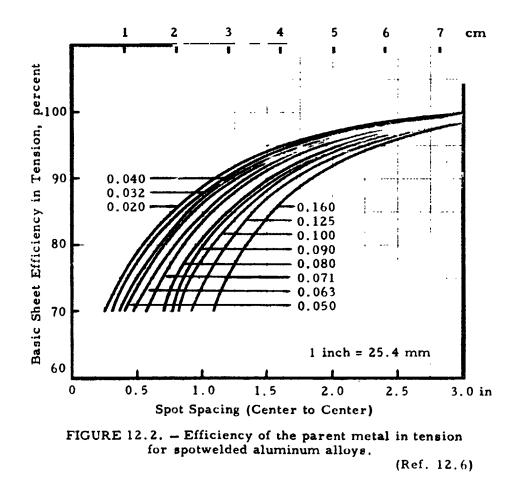


FIGURE 12.1. - Maximum static strength of typical single spotwelds in Clad 7075-T6 sheet of various thicknesses. (Ref. 12.6)



Chapter 12 - References

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- 12.11 Aluminum Co. of America, "Riveting Alcoa Aluminum," 1960.