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COMPUTER PROGRAM FOR THE CALCULATION OF MULTICOMPONENT  
CONVECTIVE DIFFUSION DEPOSITION RATES FROM CHEMICALLY  
FROZEN BOUNDARY LAYER THEORY

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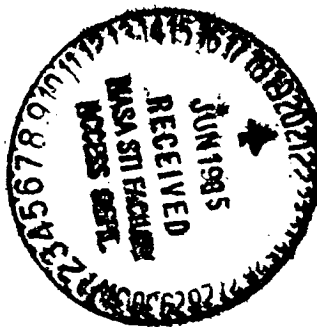
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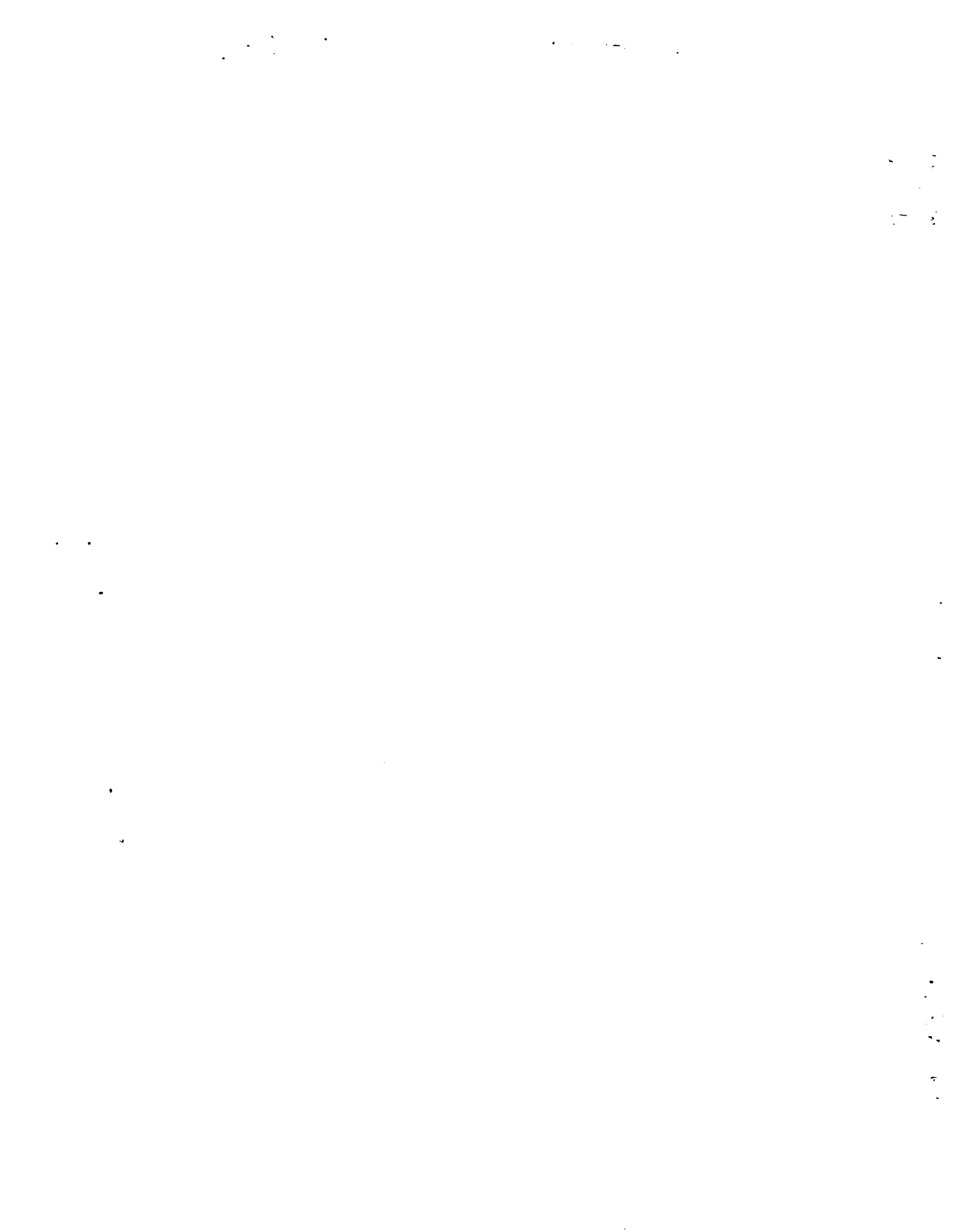


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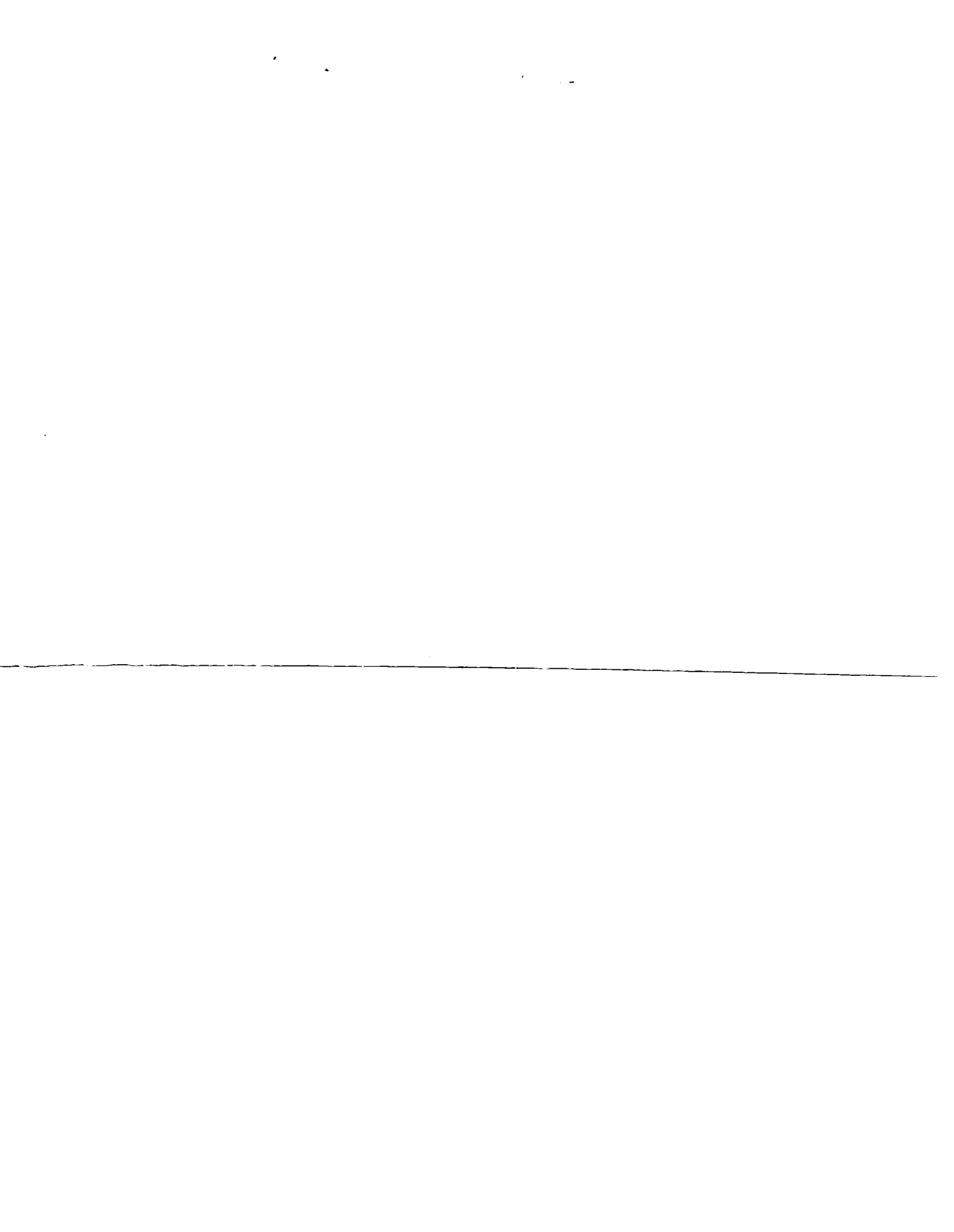
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COMPUTER PROGRAM FOR THE CALCULATION OF MULTICOMPONENT  
CONVECTIVE DIFFUSION DEPOSITION RATES FROM CHEMICALLY  
FROZEN BOUNDARY LAYER THEORY†

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SUMMARY

The computer program developed based on multicomponent chemically frozen boundary layer (CFBL) theory for calculating vapor and/or small particle deposition rates is documented. A specific application to perimeter-averaged  $\text{Na}_2\text{SO}_4$  deposition rate calculations on a cylindrical collector is demonstrated. The manual includes a typical program input and output for users.

I. INTRODUCTION

A wide variety of different engineering fields—including chemical coating of metals, solid state electronics device fabrication, filtration, aircraft icing, corrosion and fouling of gas turbine blades, heat exchangers, etc.—deal with vapor and particle capture, either to exploit or to suppress it, depending on the context. A comprehensive but tractable convective diffusion deposition rate theory has been developed at Yale University based on the assumption of a multicomponent "chemically frozen" boundary layer (CFBL) as outlined in Rosner et al. [1]. The theory which makes full use of available transport coefficient and property information is not only applicable under conditions of multi-component vapor transport (e.g., CVD applications), but it is also intrinsically capable of simultaneously dealing with particle transport provided the particles are small enough to be considered heavy molecules. The purpose of this manual is to document a simple but useful computer code based on CFBL

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theory for calculating alkali sulfate deposition rates. We hope that the manual will prevent further misapplications of CFBL theory as has been encountered in the literature [2]. Detailed description of this program, with specific application to perimeter-averaged  $\text{Na}_2\text{SO}_4$  deposition rate calculations on a cylindrical collector is given below. Typical program input and output is included with explanations. The manual contains the program listing in Appendix C.

## SYMBOLS

$B_T$	thermophoretic parameter, equation (35)
$C$	nozzle discharge coefficient
$C_{mh}$	coefficient defining relative sensitivity of mass transfer and heat transfer to variable properties, equation (37)
$c_p$	heat capacity per unit mass at constant pressure
$c_v$	heat capacity per unit mass at constant volume
$D$	Fick diffusion coefficient
$d$	collector (target) diameter
$d_j$	nozzle jet diameter
$F$	function
$f$	fuel to air mass flow rate ratio
$I$	mainstream turbulence intensity
$K$	coefficients of curve-fit in equation (11)
$k$	Boltzmann constant
$L$	mainstream turbulence length scale
$Le$	Lewis number (ratio of Fick to thermal diffusivity)
$M$	molecular weight
$M$	average molecular weight of the mixture
$\dot{m}''$	deposition mass flux
$Nu$	Nusselt number
$Nu$	Perimeter-averaged Nusselt number
$P$	pressure
$Pr$	Prandtl number (ratio of kinematic viscosity to thermal diffusivity)
$R$	universal gas constant
$Re$	Reynolds number
$S$	spin parameter, equation (33)
$s$	defined by equation (31)
$Sc$	Schmidt number (ratio of kinematic viscosity to Fick diffusivity)
$T$	absolute temperature
$U$	velocity
$v$	sonic velocity
$W_a$	air mass flow rate
$X$	mole fraction

### Greek Symbols:

$\alpha_T$	thermal diffusion factor
$\alpha_{T,\infty}$	thermal diffusion factor at infinite temperature
$\alpha_{T,-1}$	coefficient of $T^{-1}$ in power series expansion of $\alpha_T$
$\epsilon$	Lennard-Jones molecular interaction energy (well-depth) parameter
$\gamma$	specific heat ratio, $c_p/c_v$
$\lambda$	thermal conductivity
$\mu$	viscosity
$\nu$	number of Na atoms in species $i$

$\Phi$	Wilke-Wassiljewa coefficients, equation (21)
$\rho$	density
$\sigma$	Lennard-Jones molecular size parameter
$\omega$	angular speed of the collector (target)
$\Omega$	collision integral

Subscripts:

av	averaged quantity
cL	centerline
cp	constant properties
D	pertaining to Fick diffusion coefficient
h	heat transfer
i	species i
j	jet exit plane, location "j" in figure 1
k	species k
m	mass transfer
mix	gas mixture
o	stagnation conditions
t	mainstream turbulence
w	target surface (wall)
$\mu$	pertaining to viscosity
$\infty$	mainstream, location " $\infty$ " in figure 1

Miscellaneous:

CEC	chemical equilibrium code
CFBL	chemically frozen boundary layer
CVD	chemical vapor deposition

## II. DESCRIPTION OF PROGRAM WITH APPLICATION TO Na<sub>2</sub>SO<sub>4</sub> DEPOSITION

The CFBL computer code was developed for and is being used to interpret Na<sub>2</sub>SO<sub>4</sub> deposition experiments at NASA Lewis Research Center, Cleveland, Ohio [3] because of its importance in hot corrosion. Therefore, here we specifically concentrate on Na<sub>2</sub>SO<sub>4</sub> deposition rate prediction via the transport of many Na containing species to the collector, although the theory is general in that other salts can easily be incorporated. For a typical burner rig and collector configuration, and station nomenclature shown in figure 1, where any Na containing salt is seeded into the combustor, the deposition rate of Na<sub>2</sub>SO<sub>4</sub> can be shown to be:

$$\dot{m}_{Na_2SO_4}'' = \frac{1}{2} \cdot \frac{M_{Na_2SO_4}}{\bar{M}} \cdot F(\text{turb}) \cdot \frac{\rho_{\infty}}{d} \cdot \sum_1 \left\{ v_1 \cdot D_{1,\infty} \cdot \bar{Nu}_{m,1} \cdot F_1(\text{Soret}) \right. \\ \left. \cdot \left[ (X_{1,\infty} - X_{1,w}) - X_{1,w} \cdot \frac{B_{T,1}}{F_1(\text{Soret})} \cdot \left( \frac{Le_{1,w}}{Le_{1,\infty}} \right)^{0.6} \cdot \frac{c_{p,\infty}}{c_{p,w}} \cdot \frac{1}{C_{mh,1}} \right] \right\} \quad (1)$$

The multicomponent nature of the problem is reflected by the appearance of summations which include each Na carrier species i. The present program

considers four different possible Na carriers: Na, NaOH, Na<sub>2</sub>SO<sub>4</sub> and NaCl. Note the implicit assumption (checked below via the element flux ratio constraint) that once Na arrives at the surface of the collector by any of Na carriers, there is always enough sulfur available (e.g., supplied to the system by fuel) to form Na<sub>2</sub>SO<sub>4</sub>.

The mole fraction of each Na carrier species in the free stream and at the collector surface ( $X_{i,\infty}$  and  $X_{i,w}$ ) is supplied to the program by the user. The theory leads to simple algebraic flux relations for each species since it assumes that there is thermo-chemical equilibrium at stations " $\infty$ " and " $w$ ", although the boundary layer itself is chemically frozen. In the applications to Na<sub>2</sub>SO<sub>4</sub> deposition experiments at NASA LeRC, the Chemical Equilibrium Code (CEC) developed at NASA LeRC [4] is used to obtain the equilibrium compositions after combustion at specified temperatures and pressures. However, the CFBL code is by no means restricted to coupling with CEC, and the user may provide  $X_{i,\infty}$  and  $X_{i,w}$  values by any other means available.

Conditions prevailing at station " $w$ " require the imposition of one additional constraint other than vapor/condensate equilibrium [1,5]. Because Na<sub>2</sub>SO<sub>4</sub> is being deposited, the molar fluxes of Na and S must stand in the ratio 2:1 at the prevailing vapor compositions. The element mass fractions of the inorganic constituents characterizing the free stream are dictated by the experimental conditions; however, the situation at station " $w$ " is more complicated because transport across the boundary layer contributes to element "segregation." Therefore, the element mass fractions at the surface of the collector generally will not be the same as those in the free stream [5]. Hence, the CFBL program also calculates the total S molar flux to the surface with the same formulation given in equation (1). The dominant S carriers to the surface are chosen to be SO<sub>2</sub>, SO<sub>3</sub>, and H<sub>2</sub>S. Therefore, the user has to provide  $X_{i,\infty}$  and  $X_{i,w}$  values for SO<sub>2</sub>, SO<sub>3</sub>, and H<sub>2</sub>S as inputs to the program. For any particular set of  $X_{i,\infty}$  and  $X_{i,w}$  values, obtained from equilibrium calculations and input to the CFBL program, the output gives the calculated Na to S molar flux ratio. In order to satisfy both the thermodynamic (chemical equilibrium) and the transport (molar flux ratio) constraints, the user has to iterate on  $X_{i,w}$  values obtained from equilibrium calculations with clever choices of element mass fractions. Finally, the properly chosen element mass fractions will give the proper equilibrium compositions of Na and S carriers at station " $w$ " which in turn will satisfy the 2:1 Na to S molar flux ratio constraint. Experience with the CFBL program indicates that in cases where the system contains excess S, the predicted Na<sub>2</sub>SO<sub>4</sub> deposition rates are practically insensitive to the fulfillment of the transport requirement except close to the dew point temperature.†

Modification of the CFBL program for application to the multicomponent chemical vapor deposition of condensates other than Na<sub>2</sub>SO<sub>4</sub> is conceptually straightforward. It requires the proper choice of dominant carrier species and provision of relevant thermodynamic and transport parameters to the program. Generalized versions of the present CFBL code are being developed and applied to Na<sub>2</sub>SO<sub>4</sub> + K<sub>2</sub>SO<sub>4</sub> and Na<sub>2</sub>SO<sub>4</sub> + K<sub>2</sub>SO<sub>4</sub> + Li<sub>2</sub>SO<sub>4</sub> solution condensate deposition situations used at Yale University [5] and NASA LeRC.

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†For the discussion of the fact that "dew-point" is not a pure thermodynamic quantity, the reader is referred to references [5] and [6].



### III. EVALUATION OF THERMODYNAMIC AND TRANSPORT PROPERTIES

Description of the property calculations given below assumes that the following information is supplied to the program by the user as input for a typical configuration given in figure 1:

$d_j$	jet exit diameter, cm
$d$	diameter of cylindrical collector (target), cm
$f$	fuel to air mass flow rate ratio
$P_j$	jet exit (ambient) pressure, atm
$P_0$	stagnation pressure, atm
$T_0$	stagnation temperature, K
$T_w$	collector surface (wall) temperature, K
$W_a$	air mass flow rate, g/sec

Note that we are assuming a circular jet and a cylindrical collector. For other geometries proper characteristic dimensions must be supplied to calculate the jet exit area, the Reynolds number and mass transfer Nusselt number.

The combustion is assumed to be complete and dominated by the reaction:



Hence, starting with air of initial composition: 79.05 mole percent  $\text{N}_2$  and 20.95 mole percent  $\text{O}_2$ , we estimate the composition of combustion product gases by:

$$X_{\text{N}_2} = \frac{0.7905}{1 + 1.0331f} \quad (3)$$

$$X_{\text{O}_2} = \frac{0.2095 - 3.0993f}{1 + 1.0331f} \quad (4)$$

$$X_{\text{CO}_2} = X_{\text{H}_2\text{O}} = \frac{2.0662f}{1 + 1.0331f} \quad (5)$$

All other species present in the combustion product gas stream are assumed to be of trace amounts and are neglected in the property calculations.

#### A. Calculation of $\bar{M}$ , $\gamma$ , $T_j$ and $U_j$

Average molecular weight of the combustion gases is obtained from:

$$\bar{M} = X_{\text{N}_2} M_{\text{N}_2} + X_{\text{O}_2} M_{\text{O}_2} + X_{\text{CO}_2} M_{\text{CO}_2} + X_{\text{H}_2\text{O}} M_{\text{H}_2\text{O}} \quad (6)$$

where  $X$  is the mole fraction of each indicated species.

The ratio of heat capacity at constant pressure to heat capacity at constant volume of the mixture is calculated by:

$$\gamma = \frac{c_{p,mix}}{c_{v,mix}} = \frac{c_{p,mix}}{c_{p,mix} - (R/\bar{M})} \quad (7)$$

where  $R$  is the universal gas constant (1.9872 cal/mole - K), and the calculation of  $c_{p,mix}$  is described below. Assuming an isentropic jet, the temperature and the velocity at the jet exit plane is given by:

$$T_j = T_o \left( \frac{p_j}{p_o} \right)^{(\gamma-1)/\gamma} \quad (8)$$

and

$$U_j = \frac{RT_j}{p_j \bar{M}} \cdot \frac{W_a(1+f)}{\left( \pi \frac{d_j^2}{4} \right) \cdot C} \quad (9)$$

where  $R$  is the universal gas constant (82.057 atm-cm<sup>3</sup>/mole-K), and  $C$  is the discharge coefficient of the jet nozzle to be supplied by the user. The discharge coefficient depends on the geometric shape of the nozzle and the Reynolds number based on the nozzle diameter. If no information is available for the nozzle discharge coefficient, then the program automatically assigns unity as the value of  $C$ . Note that the velocity calculated from equation (9) based on the total mass flow rate and the effective area of the nozzle exit should agree with the velocity which can be calculated also from the prevailing Mach number based on the isentropic jet assumption as:

$$U_j = v \cdot \left[ \left( \frac{T_o}{T_j} - 1 \right) \frac{2}{\gamma - 1} \right]^{1/2} \quad (9a)$$

where  $v$  is the sonic speed in the gas mixture at  $T_o$ . In equation (9) it is assumed that the mixture obeys the ideal gas law, so that the density of the mixture is simply:

$$\rho_j = \frac{p_j \bar{M}}{RT_j} \quad (10)$$

#### B. Calculation of $c_p$ , $\lambda$ , $\mu$ , and $D_1$

The dimensionless heat capacity of each dominant species is computed from a curve fit:

$$\left( \frac{c_{p,i} M_i}{R} \right) = K_{5,i} T^4 + K_{4,i} T^3 + K_{3,i} T^2 + K_{2,i} T + K_{1,i} \quad (11)$$

where  $R$  is the universal gas constant (1.9872 cal/mole - K). Two different sets of constants,  $K$ , are used for  $T \geq 1000$  K and  $T < 1000$  K for a better fit.

Viscosity,  $\mu_i$ , and thermal conductivity,  $\lambda_i$ , of dominant species are calculated using Chapman-Enskog theory:

$$\mu_i = 2.6693 \times 10^{-5} \frac{\sqrt{M_i T}}{\sigma_i^2 \Omega_{\mu,i}} \quad (\text{g/cm-sec}) \quad (12)$$

$$\lambda_i = \frac{R}{M_i} \left[ \frac{15}{4} + 1.32 \left( \frac{c_{p,i} M_i}{R} - \frac{5}{2} \right) \right] \mu_i \quad (\text{cal/cm-sec-K}) \quad (13)$$

where  $\sigma_i$  is the Lennard-Jones molecular size parameter of the species and  $\Omega_{\mu,i}$  is the collision integral for viscosity which is a function of the dimensionless temperature  $kT/\epsilon_i$  where  $k$  is the Boltzmann constant and  $\epsilon_i$  the Lennard-Jones molecular interaction energy (well-depth) parameter for species  $i$ .

The binary Fick diffusion coefficient of trace species  $k$  in each of dominant species  $i$  is also calculated from Chapman-Enskog theory:

$$D_{kj} = 0.0018583 \frac{\sqrt{T^3 \left( \frac{1}{M_k} + \frac{1}{M_i} \right)}}{P \sigma_{ki}^2 \Omega_{D,ki}} \quad (\text{cm}^2/\text{sec}) \quad (14)$$

where

$$\sigma_{ki} = \frac{1}{2} \cdot (\sigma_k + \sigma_i) \quad (\text{\AA}) \quad (15)$$

and  $\Omega_{D,ki}$  is the collision integral for diffusion coefficient which (just like  $\Omega_{\mu,i}$ ) is also a function of  $kT/\epsilon_{ki}$  where:

$$\frac{\epsilon_{ki}}{k} = \frac{\sqrt{\epsilon_k \epsilon_i}}{k} \quad (\text{K}) \quad (16)$$

The dimensionless thermal diffusion factor for trace species  $i$  is conveniently expressed below after a curve-fit to calculations based on Chapman-Enskog theory.

$$\alpha_{T,i} = \alpha_{T,\infty,i} + \frac{\alpha_{T,-1,i}}{T} \quad (17)$$

The resulting least-squares constants,  $\alpha_{T,\infty,i}$  and  $\alpha_{T,-1,i}$  for a number of species are given in reference [7].

It should be noted that unless experimentally determined, the uncertainties involved in the estimation of  $\sigma$  and  $\epsilon/k$  are directly reflected in  $\mu$ ,  $\lambda$ ,  $D$ , and  $\alpha_T$  values [8]. Therefore the accuracy of prediction of the CFBL program is limited by the accuracy of such transport parameters.

### C. Calculation of Mixture Properties

Specific heat of the gas mixture is simply given by:

$$c_{p,mix} = \sum_i X_i c_{p,i} \quad (18)$$

where  $X_i$  is the mole fraction of species  $i$ . Viscosity and thermal conductivity of the mixture are calculated from similar formulae

$$\mu_{mix} = \sum_i \frac{X_i \mu_i}{\sum_k X_k \Phi_{ki}} \quad (19)$$

$$\lambda_{mix} = \sum_i \frac{X_i \lambda_i}{\sum_k X_k \Phi_{ki}} \quad (20)$$

where  $\Phi_{ki}$  are the Wilke-Wassiljewa coefficients given by:

$$\Phi_{ki} = \frac{1}{\sqrt{8}} \cdot \left(1 + \frac{M_i}{M_k}\right)^{-1/2} \cdot \left[1 + \left(\frac{\mu_i}{\mu_k}\right)^{1/2} \left(\frac{M_k}{M_i}\right)^{1/4}\right]^2 \quad (21)$$

The diffusion coefficient of species  $i$  in the mixture is calculated from:

$$D_{i,mix} = \left(\sum_k \frac{X_k}{D_{i,k}}\right)^{-1} \quad (22)$$

The reader is referred to reference [9] for the details of thermodynamic and transport property calculations.

### IV. DIMENSIONLESS NUMBERS

The dimensionless numbers given below are evaluated either at station " $\infty$ " or " $w$ " depending on relevance. The static temperature at station " $\infty$ ",  $T_\infty$ , is assumed to be equal to  $T_j$ , calculated in Section III.A. The collector surface temperature (or a representative average temperature if the collector is not isothermal) and the ambient pressure,  $P_\infty (= P_j)$ , are supplied as inputs by the user.

$$Pr = \frac{c_{p,mix} \nu_{mix}}{\lambda_{mix}} \quad (23)$$

$$Sc_i = \frac{\nu_{mix}}{\rho D_{i,mix}} \quad (24)$$

$$Le_i = \frac{Pr}{Sc_i} = \frac{D_{i,mix}}{[\lambda_{mix}/(\rho c_{p,mix})]} \quad (25)$$

$$Re = \frac{\rho U_{\infty} d}{\nu_{mix}} \quad (26)$$

Calculation of the Reynolds number requires knowledge of  $U_{\infty}$ . Although  $U_j$  is obtained in Section III.A,  $U_{\infty}$  is not necessarily the same as  $U_j$  depending on the jet velocity profile and the jet divergence angle. For jet diameters larger than the collector diameter (usually the case), one needs the centerline velocity at station " $\infty$ " for the proper calculation of  $Re$ . However,  $U_j$  calculated in Section III.A. is the average velocity at station "j". Therefore, two corrections are needed; first, to correct average  $U_j$  to centerline  $U_j$ ,  $F$  (shape), and second, to correct centerline  $U_j$  to centerline  $U_{\infty}$ ,  $F$  (div). These corrections take into account the boundary layer effects and the flow area variation from station "j" to station " $\infty$ " depending on jet divergence angle. Hence:

$$U_{\infty} = U_j \cdot F(\text{shape}) \cdot F(\text{div}) \quad (27)$$

where  $F(\text{shape})$  and  $F(\text{div})$  are defined as:

$$F(\text{shape}) \equiv \left( \frac{U_{cL}}{U_{av}} \right)_j \quad (28)$$

$$F(\text{div}) \equiv \frac{U_{cL,\infty}}{U_{cL,j}} \quad (29)$$

$F(\text{shape})$  and  $F(\text{div})$  are optional inputs to the program. If no information is available, the program automatically assigns unity for both. Based on available data for fully developed turbulent flow in smooth circular ducts, a suggested estimate for  $F(\text{shape})$  is:

$$F(\text{shape}) \cong \frac{(s+1)(2s+1)}{2s^2} \quad (30)$$

where

$$s \cong 6.349 \left( \frac{Re_i}{10^4} \right)^{0.0465} \quad (31)$$

The mass transfer Nusselt number correlations depend, among other things, on the geometry of the collector used. In the NASA LeRC experiment [3], a cylindrical collector was used. The example program listing given at the end of this manual also assumes a cylindrical collector. By making the replacement  $Pr \rightarrow Sc_i$ , the perimeter averaged mass transfer coefficient,  $Nu_{m,i}$ , becomes based on the heat transfer correlations recommended for stationary infinitely long cylinders [10]. The additional effects of variable properties and Mach number are approximated by incorporating a post-multiplier term based on the recommendations of references [11] and [12].

$$\overline{Nu}_{m,i} = (0.40 Re^{1/2} + 0.06 Re^{2/3}) (Sc_i^{0.4}) \left( \frac{T_w}{T_o} \right)^{0.04} \quad (32)$$

It should be noted that if a different collector geometry is used and/or the surface roughness of the collector can have an appreciable effect on mass transfer, the equation (32) may not be applicable, and the user has to modify the program to supply the best available corresponding mass transfer Nusselt number correlation from the literature.

In many experiments, uniformity of collector surface temperature is obtained by collector rotation and this requires additional modifications of the formulation given by equation (1). In the NASA LeRC deposition experiments the rotational speeds of the smooth collectors used have been neglected based on reference [13], which suggests that if one defines a "spin" parameter

$$S \cong \frac{\omega d}{2U_\infty} \quad (33)$$

where  $\omega$  is the angular speed of the target, then equation (1) (or eq. (32)) should be corrected by the multiplier

$$F(\text{spin}) \cong (1 + 2S^2)^{0.315} \quad (34)$$

#### V. THERMAL (SORET) DIFFUSION EFFECT, F (SORET)

Inclusion of thermal (Soret) diffusion in the mass transfer calculations introduces a nondimensional thermophoretic parameter, [14], which can be expressed as:

$$B_{T,i} \cong -\alpha_{T,i,w} \cdot (Le_{i,w})^{0.4} \cdot \left( \frac{T_o - T_w}{T_w} \right) \quad (35)$$

The effects of thermal (Soret) diffusion ("thermophoresis" for small particles) is thoroughly discussed and correlated in references [7, 14-17] for both vapor and small particle mass transfer. For the current applications of the CFBL program to NASA LeRC deposition experiments, the emphasis has been on vapor

deposition. Therefore, the "sink" effect of thermal diffusion is justifiably neglected, and only the "suction" effect is considered, i.e.:

$$F_i \text{ (Soret)} \cong F_i \text{ (suction)} \cong \frac{-B_{T,i}}{1 - \exp(B_{T,i})} \quad (36)$$

For small particle mass transfer applications where inertial effects are not important, the user should modify the program to include the "thermophoretic sink" effect as described in references [14] and [16]. For such applications a method for estimating small particle transport properties ( $D_i$ ,  $\alpha_{T,i}$ , etc.) is given in reference [17].

In the program the user also has the option of completely "turning off," if so desired, the thermal diffusion effect from inclusion in the calculations by specifying the input variable "FSORET" as "false". In this simple way we can demonstrate the relative importance of thermal diffusion mass transport in any multicomponent CVD situation.

It is known that variable properties across the boundary layer affect mass transfer rates in a somewhat different manner than they do the better known heat transfer rates. In case of thermal diffusion, this effect is in addition to the one already incorporated in the perimeter-averaged Nusselt number (eq. (32)). Certainly, the effect will be different depending on the temperature dependence of the transport properties of the species in question and the characteristics of the mass transfer boundary layer. Therefore, it is convenient to introduce a new dimensionless parameter,  $C_{mh,i}$ , defined as:

$$C_{mh,i} \equiv \frac{(Nu_m/Nu_{m,cp})_i}{(Nu_h/Nu_{h,cp})_i} \quad (37)$$

Simple engineering correlations to predict the effect of variable properties on mass transfer (the numerator in eq. (37)) and on heat transfer (the denominator in eq. (37)) are given in reference [12]. However, if the space reserved for  $C_{mh,i}$  in the input data is left blank, the program automatically assigns a value of unity for each  $C_{mh,i}$ .

## VI. MAINSTREAM TURBULENCE EFFECT, F (turb)

It is known that mainstream turbulence, apart from turbulence within the boundary layer, influences the transport rates to surfaces [18-19]. Although there is considerable quantitative disagreement among various researchers, the CFBL program uses the correlation given in reference [20] based on perimeter-averaged heat transfer measurements for cylinders in cross flow. If one has knowledge of free stream turbulence intensity,  $I_t$ , and axial macroscale of turbulence,  $L_t$ , then the correlation at a particular  $Re$  for a cylinder of diameter  $d$  is given in the separable form:

$$F \text{ (turb)} \cong 1 + fct_1(I_t Re) \cdot fct_2(L_t/d) \quad (38)$$

where  $fct_1$  and  $fct_2$  are presented in graphical form in reference [20]. The CFBL program uses the following curve fits for  $fct_1$  and  $fct_2$ .

$$fct_1(I_t Re) \cong \begin{cases} 12.375 \left[ 1 - \left( 1 - \frac{I_t Re}{10^4} \right)^{3/2} \right] & \text{if } I_t Re \leq 10^4 \\ 9.0 + 3.375 \times 10^{-4} I_t Re & \text{if } I_t Re > 10^4 \end{cases} \quad (39)$$

$$fct_2(L_t/d) \cong \begin{cases} 0.124 \times 10^{-3} [(L_t/d) - 11.0]^2 + 2.0 \times 10^{-3} & \text{if } L_t/d > 2 \\ -4.0 \times 10^{-3} [(L_t/d) - 1.75]^2 + 12.25 \times 10^{-3} & \text{if } L_t/d \leq 2 \end{cases} \quad (40)$$

F (turb) probably also depends on  $Sc_1$ , but the information is not yet sufficient in the literature to incorporate it into the present formalism. The user has three options in using the F (turb) factor:

1. If one chooses not to use the F (turb) factor, then leave the variable "TURB" blank in input data, in which case the program automatically assigns the value of unity for F (turb).
2. If the particular F (turb) to be used by the program is known, then that value should be read in as input for the variable "TURB", but leave the variables "TURIN" and "TURL" blank.
3. If one has knowledge of  $I_t$  and  $L_t$  and chooses to use the correlation supplied by the program based on equations (38) to (40), then input data for "TURB" is dummy, but "TURIN" (percent  $I_t = I_t/100$ ) and "TURL" input values must be read in.

## VII. ROUTINES

The names and functions of the subroutines used in the program are given below:

MAIN	calculates the deposition rate based on equation (1)
PARAM	provides $\epsilon/k$ , $\sigma$ , and molecular weight of dominant species in the gas mixture
CALESL	provides the dimensionless heat capacity of dominant species in the gas mixture from a curve fit
CALMIX	calculates $c_{p,mix}$ in cal/g-K
COLINT	evaluates $\Omega_\mu$ and $\Omega_D$ from a curve fit to the already supplied $kT/\epsilon$ versus $\Omega_\mu$ and $\Omega_D$ data [8]. Also calculates $\mu_1$ and $\lambda_1$ for dominant species
MIXPRO	calculates $\mu_{mix}$ and $\lambda_{mix}$
COEDIF	calculates $D_{ki}$ of trace species k in i
DIF	calculates $D_{i,mix}$
T MDF	calculates F (Soret)
NUM	calculates $Num_i$
TURBL	calculates F (turb)



## VIII. DESCRIPTION OF PROGRAM INPUT

The input data to the CFBL program is read in three different stages. The first stage reads in one card which contains the following information according to the format specified.

```
READ(5,1) IRUN, ITYPE, F, WA, TW, TO, PO, PJ, DC, WOBS, FSORET
1 FORMAT(2I2, 8F9.0, 3X, L1)
```

IRUN        The user can use this variable for his own purposes (e.g., to keep track of which experiment, which run, etc.).

ITYPE       The user can use this variable for his own purposes (e.g., to keep track of the type of salt tied to the combustor, etc.).

F           Fuel to air mass flow rate ratio

WA          Air flow rate, g/sec

TW          Collector (target) surface temperature, K. Give an average temperature if the collector surface is not isothermal.

TO          Stagnation temperature, K

PO          Stagnation pressure, atm

PJ          Jet exit (ambient) pressure, atm

DC          Discharge coefficient of jet nozzle. Optional input. Leave blank if no information is available. See Section III.A.

WOBS        Experimentally observed deposition rate. Used in calculating the percent error in the prediction, mg/hr.

FSORET     Logical variable. Specify as 'T' if thermal diffusion is desired to be included in the calculations, or as 'F' if otherwise.

The second stage reads in one card which contains the following information according to the format specified:

```
READ(5,998) DIAW, LW, DJ, TURB, TURIN, TURL, SHAPE, DIV
998 FORMAT(8F10.0)
```

DIAW        Diameter of the cylindrical target, cm

LW          Length (height) of the cylindrical target, cm

DJ          Diameter of circular jet nozzle exit plane, cm

TURB        Turbulence enhancement factor, F (turb). Optional input.  
            (a) Leave blank if no consideration is required.  
            (b) Give your own estimate if information is available, but then leave "TURIN" and "TURL" blank,

TURIN       Percent mainstream turbulence intensity ( $I_t/100$ ). Optional input.  
            (a) Leave blank if no consideration is required.  
            (b) Leave blank if you supply your own "TURB".  
            (c) Give your estimate if program consideration is desired, but then specify "TURL", too.

TURL        Axial macroscale of mainstream turbulence, cm. Optional input.  
            (a) Leave blank if no consideration is required.  
            (b) Leave blank if you supply your own "TURB".  
            (c) Give your own estimate if program consideration is desired, but then specify "TURIN", too.

SHAPE      F (shape). Optional input. Leave blank if no information is available. See Section IV.

DIV         F (div). Optional input. Leave blank if no information is available. See Section IV.

The third stage reads in seven cards, one for each species. The order is given by:

- 1 = NaOH
- 2 = Na
- 3 = Na<sub>2</sub>SO<sub>4</sub>
- 4 = SO<sub>2</sub>
- 5 = SO<sub>3</sub>
- 6 = NaCl
- 7 = H<sub>2</sub>S

Note that this input is specific to Na<sub>2</sub>SO<sub>4</sub> deposition only, and must be modified for other salts and/or salt solutions accordingly. Each card contains the following information according to the format specified.

```
READ(5,2) XW(I), XJ(I), CMH(I)
2 FORMAT(2E15.4, F10.0)
```

XW(I) Equilibrium mole fraction of species *i* at station "w"  
 XJ(I) Equilibrium mole fraction of species *i* at station "j" or equivalently at station "∞"  
 CMH(I) Ratio of effect of variable properties on mass transfer to heat transfer. Optional input. See Section V.

A typical input data set is given in Appendix A.

#### IX. DESCRIPTION OF PROGRAM OUTPUT

The program output contains all the information read into the program as input. Description of the additional variables printed out is given below:

D(I) Diffusion coefficient of species *i* in gas mixture,  $D_{i,mix}$ , in cm<sup>2</sup>/sec  
 NU(I) Mass transfer Nusselt number of species *i*,  $Nu_{m,i}$   
 SC(I) Schmidt number of species *i*,  $Sc_i$ , at station "∞"  
 TAU(I) Thermophoretic parameter of species *i*,  $B_T,i$   
 F(SORET)(I)  $F_i$ (Soret) of species *i*  
 M(I) Equivalent Na<sub>2</sub>SO<sub>4</sub> deposition rate of species *i* mass transfer rate in mg/hr  
 YNAJ Total Na mole fraction in flame  
 YNAW Total Na mole fraction at station "w"  
 YSJ Total S mole fraction in flame  
 YSW Total S mole fraction at station "w"  
 AM Average molecular weight of gas mixture (neglects trace species)  
 TURB  $F(turb)$   
 RE Reynolds number at station "∞"  
 SUM Total Na mass flux to the surface in g/cm<sup>2</sup>-sec  
 SMS Total S mass flux to the surface in g/cm<sup>2</sup>-sec  
 RATIO Na to S molar flux ratio at the surface  
 ERROR Percent error between experimentally observed and predicted Na<sub>2</sub>SO<sub>4</sub> deposition rates  
 GM  $\gamma = c_{p,mix}/c_v$   
 PR Prandtl number at station "∞"

X(I)	Mole fraction of dominant species in flame
TJ	Jet exit static temperature, $T_j = T_\infty$ , K
RHOJ	Density of gas mixture at station "j" or equivalently " $\infty$ " in g/cm <sup>3</sup>
UJ	Jet exit velocity, $U_j$ , in cm/sec
ETAMIX	$\nu_{mix}$ at station " $\infty$ " in g/cm-sec
LAMIX	$\lambda_{mix}$ at station " $\infty$ " in cal/cm-sec-K
CMIX	$c_{p,mix}$ at station " $\infty$ " in cal/g-sec

A typical output data set is given in Appendix B.

123456789012345678901234567890123456789012345678901234567890123456789012345678901234567890\*\*\*\* REFERENCE LINE

19	2	F	WA	TW	TO	PO	PJ	DC	WOBS	FSORET
		0.0481	20.0214	900.0	1885.5	1.068	1.000	1.000	27.00	T
		DIAW	LW	DJ	TURB	TURIN	TURL	SHAPE		DIV
		1.905	1.2649	2.54	0.0	0.0	0.0	0.0		0.0
		XW(I)		XJ(I)	CMH(I)					
		6.9810E-14		6.3810E-06	0.0					
		3.1280E-20		8.2740E-07	0.0					
		3.6060E-12		2.6170E-13	0.0					
		8.0742E-06		2.4994E-05	0.0					
		1.3345E-05		4.8236E-08	0.0					
		0.0		0.0	0.0					
		5.9210E-31		1.1600E-14	0.0					

12345678901234567890123456789012345678901234567890123456789012345678901234567890\*\*\*\* REFERENCE LINE

CFBL THEORY FOR SODIUM SULFATE DEPOSITION RATE

ALL GAS PROPERTIES PERTAIN TO TJ,PJ

IRUN =19  
 ITYPE = 2  
 F = 0.048100  
 TW (K) = 900.000  
 TO (K) = 1885.500  
 PO (ATM) = 1.068000  
 PJ (ATM) = 1.000000  
 DC = 1.000000  
 WOBS (MG/HR) = 27.0000  
 FSORET =T  
 WA (G/SEC) = 20.02139  
 DIAW (CM) = 1.90500  
 LW (CM) = 1.26490  
 DJ (CM) = 2.54000  
 TURIN = 0.0000 PERCENT  
 TURL (CM) = 0.00000  
 SHAPE = 0.0000  
 DIV = 0.0000

I=1=NAOH , I=2=NA , I=3=NA2SO4 , I=6=NACL

I	D(I)	NU(I)	SC(I)	X(I),J	X(I),W	TAU(I)	F(SORET),I	M(I)
1	0.29905E 01	0.78633E 02	0.10820E 01	0.63810E-05	0.69810E-13	0.73861E-01	1.03738	0.10364E 02
2	0.37881E 01	0.71538E 02	0.85418E 00	0.82740E-06	0.31280E-19	-0.23133E-01	0.98848	0.14756E 01
3	0.18316E 01	0.95670E 02	0.17666E 01	0.26170E-12	0.36060E-11	0.26800E 00	1.13998	-0.64711E-05
6	0.25178E 01	0.84235E 02	0.12851E 01	0.00000	0.00000	0.15291E 00	1.07840	0.00000

YNAJ	YNAW	YSJ	YSW
0.720840E-05	0.728181E-11	0.250422E-04	0.214192E-04

AM = 28.81070  
 TURB = 1.00000  
 RE = 12913.625  
 SUM (G/CM\*\*2/SEC) = 0.14063E-06  
 SMS (G/CM\*\*2/SEC) = 0.26371E-06  
 RATIO = 0.7437  
 PREDICTED NA2SO4 DEPOSITION RATE (MG/HR) = 11.83948  
 ERROR (%) = -56.1501  
 GM = 1.26601  
 PR = 0.689647  
 X(N2) = 0.753079  
 X(O2) = 0.057563  
 X(H2O) = 0.094679  
 X(CO2) = 0.094679  
 TJ (K) = 1859.6157

RHOJ(G/CM**3)	UJ(CM/S)	ETAMIX(POISE)	LAMIX(CAL/CM/K/S)	CMIX(CAL/G/S)
0.188806E-03	0.219344E 05	0.610923E-03	0.290277E-03	0.327682E 00

APPENDIX C

```

0000100 C
0000200 C*****
0000300 C
0000400 C
0000500 C
0000600 C
0000700 C
0000800 C
0000900 C
0001000 C
0001100 C ADT = TEMPORARY STORAGE VALUE FOR KT/E
0001200 C AII,AJJ = TEMPORARY STORAGE VALUE FOR DATOS(3,I)
0001300 C AJ = AREA OF JET NOZZLE (CM**2)
0001400 C ALPA(79) = KT/E AS INPUT PERMANENT DATA
0001500 C ALPHA(I) = THERMAL DIFFUSION COEFFICIENT FOR TRACE SPECIES I
0001600 C AM = MEAN MOLEC. WEIGHT OF GAS MIXTURE(NEGLECTS TRACE SPECIES)
0001700 C APHMIX = THERMAL DIFFUSIVITY OF MIXTURE AT STATION J
0001800 C APHW = THERMAL DIFFUSIVITY OF MIXTURE AT STATION W
0001900 C AW = AREA OF THE COLLECTOR(TARGET), (CM**2)
0002000 C
0002100 C C = CMIX INTERMEDIATE
0002200 C CJ = DIMENSIONLESS HEAT CAPACITY
0002300 C CMIX = MIXTURE HEAT CAPACITY AT STATION J (CAL/(G*K))
0002400 C CMH(I) = RATIO OF EFFECT OF VARIABLE PROP. ON MASS TO HEAT TRANS.
0002500 C OPTIONAL INPUT.LEAVE BLANK IF NO INFORMATION AVAILABLE.
0002600 C COEF1 = COEFFICIENTS FOR COMPUTING HEAT CAPACITY (1000K<T<5000K)
0002700 C COEF2 = COEFFICIENTS FOR COMPUTING HEAT CAPACITY (300K<T<1000K)
0002800 C CONDT = THERMAL CONDUCTIVITY (TEMPORARY) , (CAL/(CM*SEC*K))
0002900 C CP = MIXTURE HEAT CAPACITY AT STATION W (CAL/(G*K))
0003000 C
0003100 C D = TEMPORARY STORAGE FOR BINARY DIFF. COEF. (CM**2/SEC)
0003200 C DC = DISCHARGE COEFFICIENT OF JET NOZZLE. OPTIONAL INPUT.
0003300 C LEAVE BLANK IF NO INFORMATION IS AVAILABLE, IN WHICH
0003400 C CASE THE PROGRAM AUTOMATICALLY ASSIGNS UNITY AS ITS VALUE
0003500 C DA = DIFF. COEFF. OF TRACE SPECIES AT STATION J (CM**2/SEC)
0003600 C DATOS(1,I)= E/K FOR SPECIES I
0003700 C DATOS(2,I)= SIGMA FOR SPECIES I (COLLISION DIAMETER PARAMETER)
0003800 C DATOS(3,I)= MOLECULAR WEIGHT OF SPECIES I
0003900 C DEN = DENOMINATOR
0004000 C DIAW = DIAMETER OF THE CYLINDRICAL COLLECTOR (CM)
0004100 C DIFF = TEMPORARY STORAGE FOR BINARY DIFF. COEF. (CM**2/SEC)
0004200 C DIV = DIVERGENCE FACTOR FOR NOZZLE JET DEFINED AS THE RATIO
0004300 C OF CENTERLINE VELOCITY AT COLLECTOR LOCATION TO
0004400 C CENTERLINE VELOCITY AT JET EXIT. DEPENDS ON THE JET
0004500 C DIVERGENCE ANGLE. OPTIONAL INPUT, LEAVE BLANK IF NO
0004600 C INFORMATION AVAILABLE.
0004700 C DJ = DIAMETER OF JET NOZZLE EXIT PLANE (CM)
0004800 C DW = DIFF. COEF. OF TRACE SPECIES AT STATION W (CM**2/SEC)
0004900 C
0005000 C E/K = RATIO OF MOLECULAR INTERACTION ENERGY PARAMETER
0005100 C (LENNARD-JONES) TO BOLTZMANN CONSTANT (K)
0005200 C EOKI,EOKJ = TEMPORARY STORAGE FOR DATOS(1,I)=E/K
0005300 C EOVKIJ = E/K, SQUARE ROOT OF EOKI*EOKJ
0005400 C EPSOVK = DATOS(1,I)=E/K

```

0005500 C ERN = DIFFERENCE BETWEEN PREDICTED AND OBSERVED DEPO. RATES  
0005600 C ERROR = PERCENT ERROR BETWEEN PREDICTED AND OBSERVED DEP. RATE  
0005700 C ET = EXPONENTIAL(-TAU), SORET INTERMEDIATE  
0005800 C ETA = MIXTURE VISCOSITY AT STATION W (G/(CM\*SEC))  
0005900 C ETAMJ = TEMPORARY VALUE FOR MIXTURE VISCOSITY (G/(CM\*SEC))  
0006000 C ETAMIX = MIXTURE VISCOSITY AT STATION J (G/(CM\*SEC))  
0006100 C  
0006200 C F = FUEL TO AIR RATIO (BY MASS)  
0006300 C FSORET = LOGICAL VALUE PLACED INTO THE INPUT DATA FILE. IF IT IS  
0006400 C SET TO 'T' THEN THE SORET EFFECT WILL BE INCLUDED. IF IT  
0006500 C IS SET TO 'F' THEN THE SORET EFFECT WILL BE EXCLUDED.  
0006600 C  
0006700 C GM = GAMMA  
0006800 C GMR = (GAMMA-1.)/GAMMA  
0006900 C G = WASSILJEW-WILKE FACTOR  
0007000 C  
0007100 C ITYPE = NOT USED BY PROGRAM. OPTIONAL FOR USER PURPOSES.  
0007200 C IRUN = NOT USED BY PROGRAM. OPTIONAL FOR USER PURPOSES.  
0007300 C  
0007400 C LAMDA = MIXTURE THERMAL CONDUCTIVITY AT STATION W (CAL/(CM\*SEC\*K))  
0007500 C LAMP = TEMPORARY THERMAL CONDUCTIVITY (CAL/(CM\*SEC\*K))  
0007600 C LAMIX = MIXTURE THERMAL CONDUCTIVITY AT STATION J (CAL/(CM\*SEC\*K))  
0007700 C LEJ = LEWIS NUMBER FOR TRACE SPECIES AT STATION J  
0007800 C LOD = RATIO OF MAINSTREAM TURBULENCE LENGTH SCALE TO COLLECTOR DIAM.  
0007900 C LW = LENGTH OF THE CYLINDRICAL COLLECTOR (CM)  
0008000 C LEW = LEWIS NUMBER FOR TRACE SPECIES AT STATION W  
0008100 C  
0008200 C M = DATOS(3,1)=MOLECULAR WEIGHT OF SPECIES I  
0008300 C M(I) = EQUIVALENT NA2SO4 DEPOSITION RATE OF TRACE SPECIES MASS  
0008400 C TRANSFER RATE (MG/HR)  
0008500 C  
0008600 C NU = NUSSELT NUMBER FOR MASS TRANSFER  
0008700 C  
0008800 C OMETD,OVDT= COLLISION INTEGRALS  
0008900 C OMED(79) = COLLISION INTEGRAL FOR MASS DIFFUSIVITY  
0009000 C ONE (79) = COLLISION INTEGRAL FOR VISCOSITY OR THERMAL CONDUCTIVITY  
0009100 C  
0009200 C PHI = F(TURB) INTERMEDIATE RELATED TO TURBULENCE INTENSITY  
0009300 C PJ = JET EXIT PLANE PRESSURE (ATM)  
0009400 C PO = STAGNATION PRESSURE (ATM)  
0009500 C PR = PRANDTL NUMBER AT STATION J  
0009600 C PROV = INTERMEDIATE IN CALCULATION OF ETAMIX  
0009700 C PSI = F(TURB) INTERMEDIATE RELATED TO TURBULENCE LENGTH SCALE  
0009800 C  
0009900 C R = GAS CONSTANT=1.9872 CAL/(GMOLE\*K)  
0010000 C RATIO = SODIUM TO SULFUR MOLAR FLUX RATIO  
0010100 C RE = REYNOLDS NUMBER AT COLLECTOR LOCATION  
0010200 C REJ = REYNOLDS NUMBER AT OUTLET OF JET  
0010300 C RHOJ = DENSITY OF MIXTURE AT JET OUTLET  
0010400 C RHOW = DENSITY OF MIXTURE AT WALL  
0010500 C  
0010600 C SC = SCHMIDT NUMBER FOR TRACE SPECIES AT STATION J  
0010700 C SHAPE = SHAPE FACTOR (CENTERLINE VELOCITY)/(AVERAGE VELOCITY)  
0010800 C OPTIONAL INPUT. LEAVE BLANK IF NO INFORMATION AVAILABLE.

```

0010900 C SIGMA = MOLECULAR SIZE PARAMETER (LENNARD-JONES)
0011000 C SIGI,SIGJ = TEMPORARY STORAGE FOR DATOS(2,I)=SIGMA
0011100 C SIGK,SIGL = TEMPORARY STORAGE FOR DATOS(2,I)=SIGMA
0011200 C SIG2 = DATOS(2,I)=SIGMA
0011300 C SIG = SIGMA VALUE FOR SPECIES I
0011400 C SN(I) = MASS FLUX OF TRACE SPECIES I (G/CM**2/SEC)
0011500 C SMS = TOTAL SULFUR MASS FLUX TO THE SURFACE (G/CM**2/SEC)
0011600 C SUM = TOTAL SODIUM MASS FLUX TO THE SURFACE (G/CM**2/SEC)
0011700 C
0011800 C T = TEMPERATURE (K)
0011900 C TAU = SORET SUCTION PARAMETER FOR SPECIES I
0012000 C TO = STAGNATION TEMPERATURE (K)
0012100 C THERM = THERMAL DIFFUSION FACTOR FOR SPECIES I
0012200 C TJ = JET (STATIC) TEMPERATURE (K)
0012300 C TURB = TURBULENCE ENHANCEMENT FACTOR. OPTIONAL INPUT.
0012400 C A) LEAVE BLANK IF NO CONSIDERATION REQUIRED.
0012500 C B) GIVE YOUR OWN ESTIMATE IF INFORMATION AVAILABLE, BUT
0012600 C LEAVE 'TURIN' AND 'TURL' BLANK IF YOU DO SO.
0012700 C TURIN = PERCENT MAINSTREAM INTENSITY. OPTIONAL INPUT.
0012800 C A) LEAVE BLANK IF NO CONSIDERATION REQUIRED.
0012900 C B) LEAVE BLANK IF YOU SUPPLY YOUR OWN 'TURB'.
0013000 C C) GIVE YOUR ESTIMATE IF PROGRAM CONSIDERATION DESIRED,
0013100 C BUT THEN GIVE AN ESTIMATE FOR 'TURL',TOO.
0013200 C TURL = MAINSTREAM LENGTH SCALE (MACRO) , CM. OPTIONAL INPUT.
0013300 C A) LEAVE BLANK IF NO CONSIDERATION REQUIRED.
0013400 C B) LEAVE BLANK IF YOU SUPPLY YOUR OWN 'TURB'.
0013500 C C) GIVE YOUR ESTIMATE IF PROGRAM CONSIDERATION DESIRED,
0013600 C BUT THEN GIVE AN ESTIMATE FOR 'TURIN',TOO.
0013700 C TW = WALL TEMPERATURE (K)
0013800 C
0013900 C UJ = JET EXIT VELOCITY
0014000 C
0014100 C VISC(I) = VISCOSITY OF SPECIES (G/(CM*SEC))
0014200 C V(I) = NUMBER OF SODIUM ATOMS IN TRACE SPECIES I
0014300 C
0014400 C WA = AIR FLOW RATE (G/SEC)
0014500 C WI,WJ = TEMPORARY STORAGE VALUES FOR DATOS(3,I)
0014600 C WPRED = PREDICTED NA2SO4 DEPOSITION RATE (MG/HR)
0014700 C WOBS = EXPERIMENTALLY OBSERVED NA2SO4 DEPOSITION RATE (MG/HR)
0014800 C
0014900 C XJ(I) = MOLE FRACTION OF SPECIES I IN FLAME
0015000 C X1 = TURIN*RE/100,F(TURB) INTERMEDIATE
0015100 C XW(I) = MOLE FRACTION OF SPECIES I AT WALL
0015200 C X(4) = MOLE FRACTION OF DOMINANT SPECIES IN MIXTURE
0015300 C
0015400 C YNAJ = TOTAL SODIUM MOLE FRACTION IN FLAME
0015500 C YNAW = TOTAL SODIUM MOLE FRACTION AT STATION W
0015600 C YSJ = TOTAL SULFUR MOLE FRACTION IN FLAME
0015700 C YSW = TOTAL SULFUR MOLE FRACTION AT STATION W
0015800 C
0015900 C
0016000 C*****
0016100 C
0016200 C

```



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0016300 C SODIUM VERSION OF CHEMICALLY FROZEN BOUNDARY LAYER THEORY
0016400 C USED FOR SODIUM SULFATE DEPOSITION RATE CALCULATIONS
0016500 C
0016600 C MAIN PROGRAM
0016700 C
0016800 REAL LAMIX,LAMDA,LEW,LEJ,NU,LW
0016900 REAL M(7)
0017000 DIMENSION V(7),XW(7),XJ(7),SM(7),ALPHA(7),CMH(7)
0017100 COMMON/VA/DATOS(3,5)
0017200 LOGICAL FSORET
0017300 COMMON FSORET
0017400 COMMON/GFR/X(4)
0017500 DATA V/ 1.0 , 1.0 , 2.0 , 0.0 , 0.0 , 1.0 , 0.0 /
0017600 C
0017700 WRITE(6,490)
0017800 490 FORMAT(10X,'CFBL THEORY FOR SODIUM SULFATE DEPOSITION RATE',/)
0017900 WRITE(6,900)
0018000 900 FORMAT(10X,'ALL GAS PROPERTIES PERTAIN TO TJ,PJ',/)
0018100 C
0018200 C*****
0018300 C
0018400 C PAY SPECIAL ATTENTION TO THE FORMAT OF THE INPUT PARAMETERS *
0018500 C SEE 'VARIABLE NAME LIST' ABOVE FOR DESCRIPTION OF VARIABLES *
0018600 C *
0018700 C*****
0018800 C
0018900 C DOMINANT SPECIES IN THE MIXTURE
0019000 C 1=NITROGEN , 2=OXYGEN , 3=WATER , 4=CARBON DIOXIDE
0019100 C
0019200 C TRACE SPECIES IN THE MIXTURE
0019300 C 1=NAOH , 2=NA , 3=NA2SO4 , 4=SO2 , 5=SO3 , 6=NACL , 7=H2S
0019400 C
0019500 C
0019600 C INPUT : 1
0019700 C
0019800 READ(5,1)IRUN,ITYPE,F,WA,TW,TO,PO,PJ,DC,WOBBS,FSORET
0019900 1 FORMAT(2I2,8F9.0,3X,L1)
0020000 C
0020100 C INPUT : 2
0020200 C
0020300 READ(5,998) DIAW,LW,DJ,TURB,TURIN,TURL,SHAPE,DIV
0020400 998 FORMAT(8F10.0)
0020500 C
0020600 C INPUT : 3
0020700 C
0020800 DO 502 I=1,7
0020900 READ(5,2) XW(I),XJ(I),CMH(I)
0021000 2 FORMAT(2E15.4,F10.0)
0021100 502 CONTINUE
0021200 C
0021300 C OUTPUT OF INPUT PARAMETERS
0021400 C
0021500 WRITE(6,978) IRUN,ITYPE,F,TW,TO,PO,PJ,DC,WOBBS,FSORET
0021600 978 FORMAT(10X,'IRUN =',I2,/,10X,'ITYPE =',I2,/,10X,'F =',

```

```

0021700      *F10.6,/,10X,'TW (K) =',F10.3,/,10X,      -
0021800      *'TO (K) =',F10.3,/,10X,'PO (ATM) =',F10.6,/,10X,'PJ (ATM) =',F10.6-
0021900      *,/,10X,'DC =',F9.6/10X,'WOBS (MG/HR) =',F10.4,/,10X,'FSORET =',L1)
0022000      WRITE(6,988) WA, DIAW, LW, DJ, TURIN, TURL, SHAPE, DIV
0022100  988 FORMAT(10X,'WA (G/SEC) =',F10.5,/,10X,'DIAW (CM) =',F10.5,      -
0022200      *,/,10X,'LW (CM) =',F10.5,/,10X,'DJ (CM) =',F10.5,/,10X,      -
0022300      *'TURIN =',F8.4,' PERCENT',/,10X,'TURL (CM) =',F10.5,/,10X,      -
0022400      *'SHAPE =',F7.4,/,10X,'DIV =',F7.4)
0022500      WRITE(6,939)
0022600  939 FORMAT(/,10X,'I=1=NAOH , I=2=NA , I=3=NA2SO4 , I=6=NACL',/)
0022700 C
0022800 C  ASSIGN E/K,SIGMA AND MOLECULAR WEIGHT OF DOMINANT SPECIES
0022900 C
0023000      DATOS(1,1)=71.4
0023100      DATOS(2,1)=3.798
0023200      DATOS(3,1)=28.0134
0023300      DATOS(1,2)=106.7
0023400      DATOS(2,2)=3.467
0023500      DATOS(3,2)=31.9988
0023600      DATOS(1,3)=356.0
0023700      DATOS(2,3)=2.649
0023800      DATOS(3,3)=18.0152
0023900      DATOS(1,4)=195.2
0024000      DATOS(2,4)=3.941
0024100      DATOS(3,4)=44.0098
0024200 C
0024300 C  THERMAL DIFFUSION FACTOR OF TRACE SPECIES
0024400 C  I=1=NAOH,I=2=NA,I=3=NA2SO4,I=4=SO2,I=5=SO3,I=6=NACL,I=7=H2S
0024500 C
0024600      ALPHA(1)=0.1592*(1.-429.5/TW)
0024700      ALPHA(2)=-0.0607*(1.-551.2/TW)
0024800      ALPHA(3)=0.7828*(1.-476.2/TW)
0024900      ALPHA(4)=0.3131*(1.-198.2/TW)
0025000      ALPHA(5)=0.3877*(1.-184.3/TW)
0025100      ALPHA(6)=0.3462*(1.-420.0/TW)
0025200      ALPHA(7)=0.0592*(1.-79.38/TW)
0025300 C
0025400 C  COMPUTE THE AVERAGE MOLECULAR WEIGHT OF MIXTURE
0025500 C  (NEGLECT THE TRACE SPECIES)
0025600 C  1=NITROGEN,2=OXYGEN,3=WATER,4=CARBON DIOXIDE
0025700 C  MOLE FRACTIONS OF DOMINANT SPECIES AFTER COMBUSTION
0025800 C
0025900      DEN=1.+1.0331*F
0026000      X(1)=0.7905/DEN
0026100      X(2)=(0.2095-3.0993*F)/DEN
0026200      X(3)=2.0662*F/DEN
0026300      X(4)=X(3)
0026400      AM=DATOS(3,1)*X(1)+DATOS(3,2)*X(2)+DATOS(3,3)*X(3)+DATOS(3,4)*X(4)
0026500 C
0026600 C  CALCULATE JET EXIT TEMPERATURE,(K), AND VELOCITY,(CM/SEC)
0026700 C
0026800      CALL CALMIX(TO,AM,CMIX)
0026900      GM=CMIX/(CMIX-1.9872/AM)
0027000      GMR=(GM-1.)/GM

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0027100      AJ=3.14159*DJ*DJ/4.
0027200      TJ=TO*(PJ/PO)**GMR
0027220      IF(DC.EQ.0.00) DC=1.00
0027300      UJ=((82.057*TJ)/(PJ*AM))*(WA*(1.+F))/(AJ*DC)
0027400      C
0027500      C  CALCULATE MIXTURE PROPERTIES AT STATIONS J AND W
0027600      C
0027700      RHOJ=PJ*AM/TJ/82.057
0027800      RHOW=PJ*AM/TW/82.057
0027900      CALL CALMIX(TJ,AM,CMIX)
0028000      CALL CALMIX(TW,AM,CP)
0028100      CALL MIXPRO(TJ,ETAMIX,LAMIX)
0028200      CALL MIXPRO(TW,ETA,LAMDA)
0028300      APHMIX=LAMIX/CMIX/RHOJ
0028400      APHW=LAMDA/CP/RHOW
0028500      PR=CMIX*ETAMIX/LAMIX
0028600      C
0028700      C  CALCULATE REYNOLDS NUMBER
0028800      C
0028900      IF(SHAPE.EQ.0.0) SHAPE=1.0
0029000      IF(DIV.EQ.0.0) DIV=1.0
0029100      REJ=RHOJ*UJ*DIAW/ETAMIX
0029200      RE=REJ*SHAPE*DIV
0029300      C
0029400      C  GIVE VALUES OF E/K, SIGMA AND MOLEC. WEIGHT FOR TRACE SPECIES I
0029500      C
0029600      DO 10 I=1,7
0029700      25 IF(I.GT.1) GO TO 5
0029800      DATOS(1,5)=1962.0
0029900      DATOS(2,5)=3.804
0030000      DATOS(3,5)=39.9971
0030100      GO TO 11
0030200      5 IF(I.GT.2) GO TO 6
0030300      DATOS(1,5)=1375.
0030400      DATOS(2,5)=3.567
0030500      DATOS(3,5)=22.98977
0030600      GO TO 11
0030700      6 IF(I.GT.3) GO TO 7
0030800      DATOS(1,5)=2221.
0030900      DATOS(2,5)=5.00
0031000      DATOS(3,5)=142.0371
0031100      GO TO 11
0031200      7 IF(I.GT.4) GO TO 8
0031300      DATOS(1,5)=335.4
0031400      DATOS(2,5)=4.112
0031500      DATOS(3,5)=64.0588
0031600      GO TO 11
0031700      8 IF(I.GT.5) GO TO 9
0031800      DATOS(1,5)=431.4
0031900      DATOS(2,5)=4.207
0032000      DATOS(3,5)=80.0582
0032100      GO TO 11
0032200      9 IF(I.GT.6) GO TO 12
0032300      DATOS(1,5)=1989.

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0032400      DATOS(2,5)=4.186
0032500      DATOS(3,5)=58.4428
0032600      GO TO 11
0032700      12 CONTINUE
0032800      DATOS(1,5)=301.1
0032900      DATOS(2,5)=3.623
0033000      DATOS(3,5)=34.0758
0033100      11 CONTINUE
0033200 C
0033300 C      DIFFUSION COEFFICIENT OF TRACE SPECIES AT STATIONS J AND W
0033400 C
0033500      CALL DIF(TJ,PJ,DA)
0033600      CALL DIF(TW,PJ,DW)
0033700 C
0033800 C      LEWIS NUMBER OF TRACE SPECIES AT STATIONS J AND W
0033900 C
0034000      LEJ=DA/APHMIX
0034100      LEW=DW/APHW
0034200 C
0034300 C      SCHMIDT NUMBER OF TRACE SPECIES AT STATION J
0034400 C
0034500      SC=ETAMIX/RHOJ/DA
0034600 C
0034700 C      MASS TRANS. NUSSELT NO.(PERIMETER-AVERAGED,CYLINDRICAL COLLECTOR)
0034800 C
0034900      CALL NUM(TW,TO,RE,SC,NU)
0035000 C
0035100 C      COMPUTE F(SORET)
0035200 C
0035300      CALL TMDF(I,TW,TO,LEW,ALPHA,TAU,THERM)
0035400 C
0035500 C      COMPUTE F(TURB)
0035600 C
0035700      CALL TURBL(RE,TURB,TURIN,TURL,DIWA)
0035800 C
0035900 C      CALCULATE NA2SO4 DEPOSITION RATE CONTRIBUTED FROM SPECIES I
0036000 C      BY CFBL FORMULATION.
0036100 C
0036200      IF(CMH(I).EQ.0.0) CMH(I)=1.0
0036300      SM(I)=TURB*DA*RHOJ*THERM*((XJ(I)-XW(I))+XW(I)*TAU/THERM*
0036400      *(LEW/LEJ)**0.6*CMIX/CP/CMH(I))*NU*DATOS(3,5)/AM/DIAW
0036500      IF(V(I).EQ.0.0) GO TO 10
0036600      AW=3.14159*DIWA*LEW
0036700      M(I)=AW*3.600E6*0.5*142.0371/DATOS(3,5)*SM(I)*V(I)
0036800 C
0036900 C      PRINT INTERIM OUTPUT
0037000 C
0037100      IF(I.GT.1) GO TO 546
0037200      IF(FSORET) GO TO 100
0037300      WRITE(6,544)
0037400      544 FORMAT(/,10X,'CALCULATIONS DO NOT INCLUDE THERMAL DIFFUSION',/)
0037500      100 WRITE(6,545)
0037600      545 FORMAT(T2,'I',T11,'D(I)',T25,'NU(I)',T40,'SC(I)',T55,'X(I)',-
0037700      CJ',T70,'X(I),W',T85,'TAU(I)',T100,'F(SORET),I',T116,'M(I)')

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0037800 546 WRITE(6,550) I,DA,NU,SC,XJ(I),XW(I),TAU,THERM,M(I)
0037900 550 FORMAT(1X,I1,6E15.5,F15.5,2E15.5,/)
0038000 10 CONTINUE
0038100 C
0038200 C COMPUTE TOTAL SODIUM(NA) AND SULFUR(S) MASS FLUX
0038300 C SUM=J(NA), SMS=J(S)
0038400 C
0038500 SUM=SM(1)*22.98977/39.9971+SM(2)*1.0+SM(3)*2.*22.98977 -
0038600 //142.0371+SM(6)*22.98977/58.4428
0038700 SMS=SM(3)*32.06/142.0371+SM(4)*32.06/64.0588+SM(5)* -
0038800 *32.06/80.0582+SM(7)*32.06/34.0758
0038900 C
0039000 C COMPUTE SODIUM TO SULFUR MOLAR FLUX RATIO
0039100 C
0039200 RATIO=SUM/SMS*32.06/22.98977
0039300 C
0039400 C COMPUTE TOTAL NA AND S MOLE FRACTIONS AT STATIONS J AND W
0039500 C
0039600 YNAJ=XJ(1)+XJ(2)+(XJ(3)*2)+XJ(6)
0039700 YNAW=XW(1)+XW(2)+(XW(3)*2)+XW(6)
0039800 YSJ=XJ(3)+XJ(4)+XJ(5)+XJ(7)
0039900 YSW=XW(3)+XW(4)+XW(5)+XW(7)
0040000 WRITE(6,211)
0040100 211 FORMAT(//,T8,'YNAJ',T24,'YNAW',T40,'YSJ',T57,'YSW')
0040200 WRITE(6,212)YNAJ,YNAW,YSJ,YSW
0040300 212 FORMAT(4E16.6,/)
0040400 C
0040500 C CALCULATE NA2SO4 DEPOSITION RATE (MG/HR)
0040600 C
0040700 WPRED=M(1)+M(2)+M(3)+M(6)
0040800 C
0040900 C COMPUTE PERCENT ERROR BETWEEN M(PRED) AND M(OBS)
0041000 C
0041100 ERR=WPRED-WOBS
0041200 ERROR=ERR/WOBS*100.
0041300 C
0041400 C PRINT RESULTS
0041500 C
0041600 WRITE(6,4)AM,TURB,RE,SUM,SMS,RATIO,WPRED,ERROR,GM,PR,X(1),X(2),X(3), -
0041700 *X(4),TJ
0041800 4 FORMAT(/,10X,'AM =',F10.5,/,10X,'TURB =',F10.5,/,10X,'RE =',F10.3,/, -
0041900 *10X,'SUM (G/CM**2/SEC) =',E13.5,/,10X,'SMS (G/CM**2/SEC) =',E13.5,/, -
0042000 *10X,'RATIO =',F10.4,/,10X, -
0042100 *'PREDICTED NA2SO4 DEPOSITION RATE (MG/HR) =',F10.5,/,10X, -
0042200 *'ERROR (%) =',F10.4,/,10X,'GM =',F10.5,/,10X,'PR =',F10.6,/,10X, -
0042300 *'X(N2) =',F10.6,/,10X,'X(O2) =',F10.6,/,10X,'X(H2O) =',F10.6,/,10X, -
0042400 *'X(CO2) =',F10.6,/,10X,'TJ (K) =',F10.4,/)
0042500 WRITE(6,950)
0042600 950 FORMAT(/,T5,'RHOJ(G/CM**3)',T21,'UJ(CM/S)',T37,'ETAMIX(POISE) -
0042700 C',T52,'LAMIX(CAL/CM/K/S)',T73,'CMIX(CAL/G/S)')
0042800 WRITE(6,951)RHOJ,UJ,ETAMIX,LAMIX,CMIX
0042900 951 FORMAT(4E16.6,4X,E16.6,/)
0043000 STOP
0043100 END

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0043200 C
0043300 C*****
0043400 C
0043500     SUBROUTINE PARAM (J,EPSONK,SIG2,M)
0043600 C
0043700 C SUBROUTINE PARAM GIVES THE VALUES OF E/K,SIGMA AND MOLECULAR WEIGHT
0043800 C
0043900     REAL M
0044000     COMMON/VA/DATOS(3,5)
0044100     EPSONK=DATOS(1,J)
0044200     SIG2=DATOS(2,J)
0044300     M=DATOS(3,J)
0044400     RETURN
0044500     END
0044600 C
0044700 C*****
0044800 C
0044900     SUBROUTINE CALES1(J,T,CJ)
0045000 C
0045100 C THIS SUBROUTINE CALCULATES DIMENSIONLESS HEAT CAPACITY FROM A
0045200 C CURVE FIT VALID FOR 300K < T < 1000K AND 1000K < T < 5000K
0045300 C
0045400     DIMENSION COEF1(5,4),COEF2(5,4)
0045500     DATA COEF1/ 0.28532899E+01 , 0.16022128E-02 , -0.62936893E-06 , -
0045600     * 0.11441022E-09 , -0.78057465E-14 , -
0045700     * 0.36122139E+01 , 0.74853166E-03 , -0.19820647E-06 , -
0045800     * 0.33749008E-10 , -0.23907374E-14 , -
0045900     * 0.26340654E+01 , 0.31121899E-02 , -0.90278449E-06 , -
0046000     * 0.12673054E-09 , -0.69164732E-14 , -
0046100     * 0.44608041E+01 , 0.30981719E-02 , -0.12392571E-05 , -
0046200     * 0.22741325E-09 , -0.15525954E-13 /
0046300     DATA COEF2/ 0.37044177E+01 , -0.14218753E-02 , 0.28670392E-05 , -
0046400     * -0.12028885E-08 , -0.13954677E-13 , -
0046500     * 0.37837135E+01 , -0.30233634E-02 , 0.99492751E-05 , -
0046600     * -0.98189101E-08 , 0.33031825E-11 , -
0046700     * 0.41675564E+01 , -0.18106868E-02 , 0.59450878E-05 , -
0046800     * -0.48670871E-08 , 0.15284144E-11 , -
0046900     * 0.24007797E+01 , 0.87350957E-02 , -0.66070878E-05 , -
0047000     * 0.20021861E-08 , 0.63274039E-15 /
0047100     IF(T.GT.1000.) GO TO 10
0047200     CJ=COEF2(5,J)*T**4+COEF2(4,J)*T**3+COEF2(3,J)*T**2+COEF2(2,J)*T -
0047300     *+COEF2(1,J)
0047400     GO TO 20
0047500     10 CJ=COEF1(5,J)*T**4+COEF1(4,J)*T**3+COEF1(3,J)*T**2+COEF1(2,J)*T -
0047600     *+COEF1(1,J)
0047700     20 RETURN
0047800     END
0047900 C
0048000 C*****
0048100 C
0048200     SUBROUTINE CALMIX(T,AM,CMIX)
0048300 C
0048400 C SUBROUTINE CALMIX PROVIDES MIXTURE HEAT CAPACITY IN CAL/(G*K)
0048500 C

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0048600      COMMON/GFR/X(4)
0048700      C=0.
0048800      DO 8 J=1,4
0048900      CALL CALES(L,J,T,CJ)
0049000      C=C+CJ*X(J)*1.9872
0049100      8 CONTINUE
0049200      CMIX=C/AM
0049300      RETURN
0049400      END
0049500      C
0049600      C*****
0049700      C
0049800      SUBROUTINE COLINT(II,JJ,T,OVDT,ETA,CONDT,OMEDT)
0049900      C
0050000      C THIS SUBROUTINE GIVES COLLISION INTEGRALS,VISCOSITY AND THERMAL
0050100      C CONDUCTIVITY OF SPECIES USING CHAPMAN-ENSKOG THEORY
0050200      C
0050300      REAL M
0050400      DIMENSION OME(79),ALPA(79),OMED(79)
0050500      DATA ALPA/
0050600      *      0.300 , 0.350 , 0.400 , 0.450 , 0.500 , 0.550 , 0.600 , -
0050700      *      0.650 , 0.700 , 0.750 , 0.800 , 0.850 , 0.900 , 0.950 , -
0050800      *      1.000 , 1.050 , 1.100 , 1.150 , 1.200 , 1.250 , 1.300 , -
0050900      *      1.350 , 1.400 , 1.450 , 1.500 , 1.550 , 1.600 , 1.650 , -
0051000      *      1.700 , 1.750 , 1.800 , 1.850 , 1.900 , 1.950 , 2.000 , -
0051100      *      2.100 , 2.200 , 2.300 , 2.400 , 2.500 , 2.600 , 2.700 , -
0051200      *      2.800 , 2.900 , 3.000 , 3.100 , 3.200 , 3.300 , 3.400 , -
0051300      *      3.500 , 3.600 , 3.700 , 3.800 , 3.900 , 4.000 , 4.100 , -
0051400      *      4.200 , 4.300 , 4.400 , 4.500 , 4.600 , 4.700 , 4.800 , -
0051500      *      4.900 , 5.000 , 6.000 , 7.000 , 8.000 , 9.000 , 10.00 , -
0051600      *      20.00 , 30.00 , 40.00 , 50.00 , 60.00 , 70.00 , 80.00 , -
0051700      *      90.00 , 100.0 /
0051800      DATA OME/
0051900      *      2.7850 , 2.6280 , 2.4920 , 2.3680 , 2.2570 , 2.1560 , -
0052000      *      2.0650 , 1.9820 , 1.9080 , 1.8410 , 1.7800 , 1.7250 , -
0052100      *      1.6750 , 1.6290 , 1.5870 , 1.5490 , 1.5140 , 1.4820 , -
0052200      *      1.4520 , 1.4240 , 1.3990 , 1.3750 , 1.3530 , 1.3330 , -
0052300      *      1.3140 , 1.2960 , 1.2790 , 1.2640 , 1.2480 , 1.2340 , -
0052400      *      1.2210 , 1.2090 , 1.1970 , 1.1860 , 1.1750 , 1.1560 , -
0052500      *      1.1380 , 1.1220 , 1.1070 , 1.0930 , 1.0810 , 1.0690 , -
0052600      *      1.0580 , 1.0480 , 1.0390 , 1.0300 , 1.0220 , 1.0140 , -
0052700      *      1.0070 , 0.9999 , 0.9932 , 0.9870 , 0.9811 , 0.9755 , -
0052800      *      0.9700 , 0.9649 , 0.9600 , 0.9553 , 0.9507 , 0.9464 , -
0052900      *      0.9422 , 0.9382 , 0.9343 , 0.9305 , 0.9269 , 0.8963 , -
0053000      *      0.8727 , 0.8538 , 0.8379 , 0.8242 , 0.7432 , 0.7005 , -
0053100      *      0.6718 , 0.6504 , 0.6335 , 0.6194 , 0.6076 , 0.5973 , -
0053200      *      0.5882 /
0053300      DATA OMED/
0053400      *      2.6620 , 2.4760 , 2.3180 , 2.1840 , 2.0660 , 1.9660 , -
0053500      *      1.8770 , 1.7980 , 1.7290 , 1.6670 , 1.6120 , 1.5620 , -
0053600      *      1.5170 , 1.4760 , 1.4390 , 1.4060 , 1.3750 , 1.3460 , -
0053700      *      1.3200 , 1.2960 , 1.2730 , 1.2530 , 1.2330 , 1.2150 , -
0053800      *      1.1980 , 1.1820 , 1.1670 , 1.1530 , 1.1400 , 1.1280 , -
0053900      *      1.1160 , 1.1050 , 1.0940 , 1.0840 , 1.0750 , 1.0570 , -
0054000      *      1.0410 , 1.0260 , 1.0120 , 0.9996 , 0.9878 , 0.9770 , -
0054100      *      0.9672 , 0.9576 , 0.9490 , 0.9406 , 0.9328 , 0.9256 , -
0054200      *      0.9186 , 0.9120 , 0.9058 , 0.8998 , 0.8942 , 0.8888 , -

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0054000      *           0.8836 , 0.8788 , 0.8740 , 0.8694 , 0.8652 , 0.8610 , -
0054100      *           0.8568 , 0.8530 , 0.8492 , 0.8456 , 0.8422 , 0.8124 , -
0054200      *           0.7896 , 0.7712 , 0.7556 , 0.7424 , 0.6640 , 0.6232 , -
0054300      *           0.5960 , 0.5756 , 0.5596 , 0.5464 , 0.5352 , 0.5256 , -
0054400      *           0.5170 /
0054500      IF(JJ.EQ.0)GO TO 67
0054600      CALL PARAM(JJ,EOKJ,SIGL,AJJ)
0054700      CALL PARAM(II,EOKI,SIGK,AII)
0054800      M=(AII+AJJ)/2.
0054900      SIG2=(SIGL+SIGK)/2.
0055000      EOVKIJ=SQRT(EOKJ*EOKI)
0055100      ADT=T/EOVKIJ
0055200      GO TO 68
0055300      67 CALL PARAM(II,EP SOVK,SIG2,M)
0055400      ADT=T/EP SOVK
0055500      68 J=3
0055600      69 IF(ADT-ALPA(J))112,111,110
0055700      110 J=J+1
0055800      IF(J.LE.79)GO TO 69
0055900      GO TO 204
0056000      111 OVDT=OME(J)
0056100      OMEDT=OMED(J)
0056200      GO TO 200
0056300      112 B=(ADT-ALPA(J-1))/(ALPA(J)-ALPA(J-1))
0056400      OVDT=B*(B-1.)*OME(J-2)/2.+(1.-B**2)*OME(J-1)+B*(B+1.)*OME(J)/2.
0056500      OMEDT=B*(B-1.)*OMED(J-2)/2.+(1.-B**2)*OMED(J-1)+B*(B+1.)*OMED(J)/2-
0056600      *.
0056700      GO TO 200
0056800      204 OVDT=0.5882*((ADT/100.)**(-0.17))
0056900      OMEDT=0.5170*((ADT/100.)**(-0.15))
0057000      200 ETA=26.693*SQRT(M*T)/((SIG2)**2*OVDT)*1.E-06
0057100      CALL CALES1(II,T,CR)
0057200      CONDT=(1.9872/M)*(15./4.+1.32*(CR-5./2.))*ETA
0057300      RETURN
0057400      END
0057500 C
0057600 C*****
0057700 C
0057800      SUBROUTINE MIXPRO(T,ETAMIX,LAMIX)
0057900 C
0058000 C THIS SUBROUTINE PROVIDES MIXTURE VISCOSITY AND THERMAL CONDUCTIVITY
0058100 C
0058200      REAL LAMP,LAMIX
0058300      DIMENSION VISC(4),LAMP(4)
0058400      COMMON/VA/DATOS(3,5)
0058500      COMMON/GFR/X(4)
0058600      ETAMIX=0.
0058700      LAMIX=0.
0058800      I1=0
0058900      DO 40 J=1,4
0059000      N=J
0059100      CALL COLINT(N,I1,T,OVDT,ETAMJ,CONDT,OMEDT)
0059200      VISC(J)=ETAMJ
0059300      LAMP(J)=CONDT

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0059400 40 CONTINUE
0059500 DO 18 I=1,4
0059600 PROV=0.
0059700 DO 28 J=1,4
0059800 G=1./(2.*SQRT(2.))*((1.+DATOS(3,I)/DATOS(3,J))*(-0.5))*(1.
0059900 * +SQRT(VISC(I)/VISC(J))*SQRT(SQRT(DATOS(3,J)/DATOS(3,I))))**2
0060000 PROV=PROV+G*X(J)
0060100 28 CONTINUE
0060200 ETAMIX=ETAMIX+X(I)*VISC(I)/PROV
0060300 LAMIX=LAMIX+X(I)*LAMP(I)/PROV
0060400 18 CONTINUE
0060500 RETURN
0060600 END
0060700 C
0060800 C*****
0060900 C
0061000 SUBROUTINE COEDIF(T,P,I,J,D)
0061100 C
0061200 C THIS SUBROUTINE CALCULATES BINARY DIFFUSION COEFFICIENT
0061300 C
0061400 CALL COLINT(I,J,T,OVDT,ETA,CONDT,OMEDT)
0061500 CALL PARAM(I,EPSOVK,SIGI,WI)
0061600 CALL PARAM(J,EPSOVK,SIGJ,WJ)
0061700 SIG=1./2.*(SIGI+SIGJ)
0061800 D=0.0018583*SQRT((T**3)*(1./WI+1./WJ))/(P*OMEDT*(SIG**2))
0061900 RETURN
0062000 END
0062100 C
0062200 C*****
0062300 C
0062400 SUBROUTINE DIF(T,P,DA)
0062500 C
0062600 C SUBROUTINE DIF GIVES DIFFUSION COEFFICIENT OF TRACE SPECIES K
0062700 C IN GAS MIXTURE
0062800 C
0062900 COMMON/GFR/X(4)
0063000 DEN=0.0
0063100 DO 101 J=1,4
0063200 K=5
0063300 CALL COEDIF(T,P,K,J,DIFF)
0063400 DEN=DEN+X(J)/DIFF
0063500 101 CONTINUE
0063600 DA=1./DEN
0063700 RETURN
0063800 END
0063900 C
0064000 C*****
0064100 C
0064200 SUBROUTINE TMDF(I,TW,TO,LEW,ALPHA,TAU,THERM)
0064300 C
0064400 C SUBROUTINE TMDF CALCULATES THERMAL DIFFUSION EFFECT, F(SORET),
0064500 C LABELLED AS 'THERM'. IF FSORET IS FALSE, THERM=1.0 .
0064600 C
0064700 LOGICAL FSORET

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0064800      COMMON FSORET
0064900      REAL LEW
0065000      DIMENSION ALPHA(7)
0065100      IF(FSORET) GO TO 10
0065200      THERM=1.0
0065300      TAU=0.0
0065400      GO TO 20
0065500      10 TAU=ALPHA(I)*LEW**0.4*(TO-TW)/TW
0065600      ET=EXP(-TAU)
0065700      THERM=TAU/(1.-ET)
0065800      20 RETURN
0065900      END
0066000 C
0066100 C*****
0066200 C
0066300      SUBROUTINE NUM(TW,TO,RE,SC,NU)
0066400 C
0066500 C SUBROUTINE NUM CALCULATES PERIMETER AVERAGED NUSSELT NUMBER
0066600 C INCLUDING EFFECTS OF MACH NUMBER AND VARIABLE PROPERTIES
0066700 C
0066800      REAL NU
0066900      NU=(0.40*SQRT(RE)+0.06*RE**(2./3.))*(SC**0.4)*
0067000      1(TO/TW)**(-0.04)
0067100      RETURN
0067200      END
0067300 C
0067400 C*****
0067500 C
0067600      SUBROUTINE TURBL(RE,TURB,TURIN,TURL,DIAW)
0067700 C
0067800 C SUBROUTINE TURBL CALCULATES EFFECT OF MAINSTREAM TURBULENCE
0067900 C
0068000      REAL LOD
0068100      IF(TURIN.NE.0.0) GO TO 5
0068200      IF(TURB.EQ.0.0) GO TO 10
0068300      GO TO 30
0068400      5 LOD=TURL/DIAW
0068500      IF(LOD.GT.2.0) PSI=0.124*(LOD-11.0)**2.+2.
0068600      IF(LOD.LE.2.0) PSI=-4.0*(LOD-1.75)**2.+12.25
0068700      PSI=PSI/1000.
0068800      XI=TURIN*RE/100.
0068900      IF(XI-100.)1,2,2
0069000      1 PHI=12.375*(1.-(1.-XI/100.))*1.5)
0069100      GO TO 20
0069200      2 PHI=9.0+0.03375*XI
0069300      20 TURB=1.+PSI*PHI
0069400      GO TO 30
0069500      10 TURB=1.0
0069600      30 RETURN
0069700      END
0069800 C
0069900 C***** END OF PROGRAM *****

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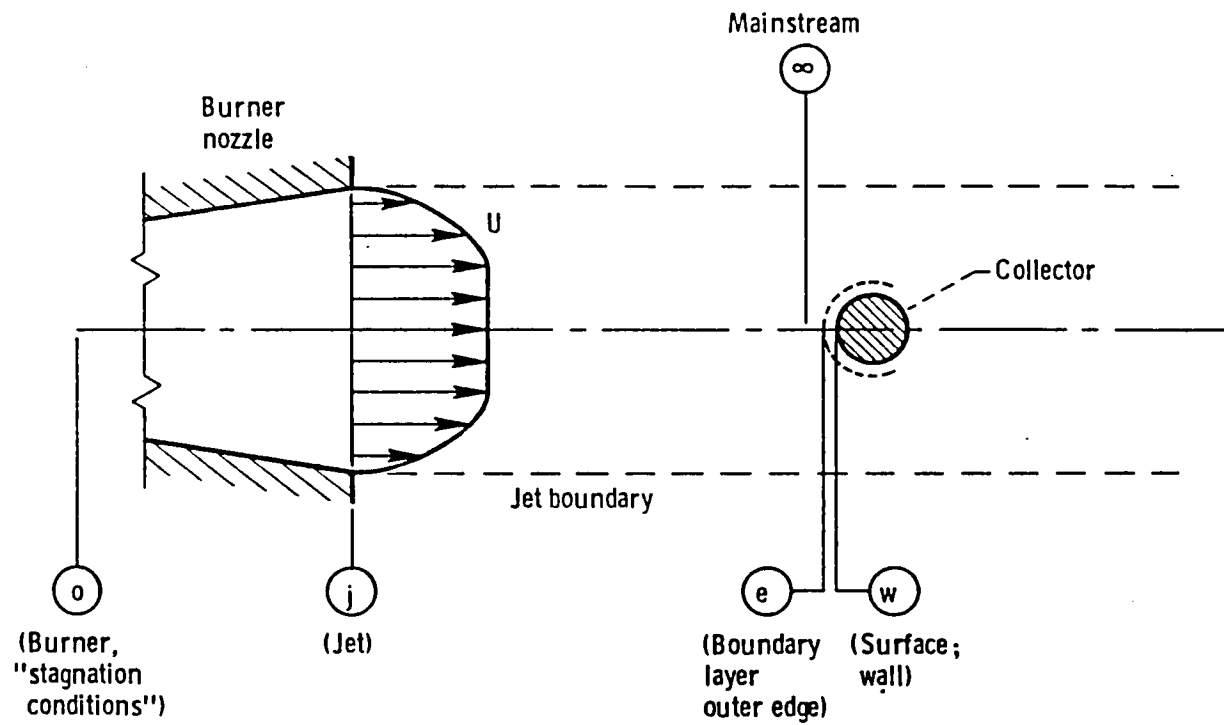


Figure 1. - Station nomenclature.

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16. Abstract The computer program developed based on multicomponent chemically frozen boundary layer (CFBL) theory for calculating vapor and/or small particle deposition rates is documented. A specific application to perimeter-averaged Na <sub>2</sub> SO <sub>4</sub> deposition rate calculations on a cylindrical collector is demonstrated. The manual includes a typical program input and output for users.					
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