## N92-13669 THE 2.5-5.0 $\mu$ M SPECTRA OF IO: EVIDENCE FOR H<sub>2</sub>S AND H<sub>2</sub>O FROZEN IN SO<sub>2</sub>

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The techniques of low temperature spectroscopy are applied here to identify the constituents of the ices covering the surface of Io, a satellite of Jupiter.

Infrared spectra of lo in the 4000-2000 cm<sup>-1</sup> region, including new observational data, are analyzed using laboratory studies of plausible surface ices.

Besides the well-known absorption bands attributable to sulfur dioxide frosts, four unidentified infrared spectral features of lo are pointed out. Two are at 2597 cm<sup>-1</sup> and 2558 cm<sup>-1</sup> and the second pair fall at 3367 cm<sup>-1</sup> and 3175 cm<sup>-1</sup>. These absorptions fall close to the fundamental X-H stretching modes in H<sub>2</sub>S and H<sub>2</sub>O respectively. The infrared absorption spectra of mixed molecular ices ranging from pure materials, to binary mixtures of H<sub>2</sub>S and SO<sub>2</sub> (either mixed at different concentrations or layered), to H<sub>2</sub>O/H<sub>2</sub>S/SO<sub>2</sub> mixtures are discussed. The effects of ultraviolet irradiation (120 and 160 nm) and temperature variation (from 9 K to 130 K) on the infrared spectra of the ices are also examined. The comparative study shows that: (1) Io most likely contains  $H_2S$  and  $H_2O$  mixed with SO<sub>2</sub>. The 2597 cm<sup>-1</sup> and 2558 cm<sup>-1</sup> bands in the lo spectra can be accounted for by the absorption of the S-H stretching vibration ( $v_1$ ) in H<sub>2</sub>S aggregates and isolated molecules in an SO<sub>2</sub> matrix. The weak 3367 cm<sup>-1</sup> and 3175 cm<sup>-1</sup> bands which vary spatially and temporally in the lo spectra coincide with the  $v_3$ and  $v_1$  O-H stretching vibrations of clusters of H<sub>2</sub>O complexed with SO<sub>2</sub>. (2) The observations are well matched by SO2 matrices containing about 3% H2S and 0.1% H<sub>2</sub>O and which have been formed by the condensation of a mixture of the gases onto a 100 K surface. (3) In the comparison of the spectra using the mixed molecular ice samples versus the layered ice samples only the former can explain the shifts and splitting of the absorption bands in the lo spectrum and account for the fact that solid H<sub>2</sub>S is observed in the surface material of lo at temperatures and pressures above the sublimation point of pure H<sub>2</sub>S. In addition to pointing out the presence of H<sub>2</sub>S and H<sub>2</sub>O on Io, the originality of this study comes from the fact that it is the first to consider mixed solids in carrying out laboratory simulations of planetary surfaces providing a realistic simulation of the "dirty" ices covering the surfaces of many satellites.