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OBSERVATIONS OF COMETARY PARENT MOLECULES WITH THE IRAM RADIO TELESCOPE.

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Abstract

Several rotational transitions of HCN, H₂S, H₂CO and CH₃OH were detected in comets P/Brorsen-Metcalf 1989 X, Austin (1989c1) and Levy (1990c) with the IRAM 30-m radio telescope. This allows us to determine the production rates of these molecules and to probe the physical conditions of the coma.

OBSERVATIONS

Comets P/Brorsen-Metcalf 1989 X, Austin (1989c1) and Levy (1990c) were observed on September 2-7 1989, May 21-25 and August 26-31 1990, respectively, with the IRAM (Institut de Radio Astronomie Millimétrique) 30-m radio telescope at Pico Veleta (Spain). Three SIS mixer receivers were used simultaneously (85-115, 130-170 and 209-270 GHz). The spectrometers consisted in two banks of 128x100 kHz channels, two banks of 512x1 MHz channels, and an AOS (Acousto Optical Spectrometer) of 864 channels with a 505 MHz bandwidth.

RESULTS

The results concerning the detected species are summarized in the table. More details were (or will be) published by Bockelée-Morvan et al. (1990, 1991), Colom et al. (1990, 1992) and Crovisier et al. (1990, 1991).

HCN, H₂CO and H₂S production rates were derived from the observed line intensities using models treating the evolution of the excitation conditions from the collision dominated region (inner coma, collisions with H₂O, s = 10^{-14} cm², T_{kin} = 50 K) to the radiation dominated region (outer coma, IR excitation of the vibrational bands by the Sun). For CH₃OH, we assume LTE and used a rotational temperature of 30 K, in agreement with the observed relative line intensities. For the density distribution we assumed isotropic outflow from the nucleus at constant velocity (0.8 km s¹) and took into account the molecular lifetime against photodissociation.

Hydrogen cyanide

The J(1-0) 89 GHz and J(3-2) 266 GHz rotational transitions of HCN were marginally detected in comet P/Brorsen-Metcalf, whereas clear detections were obtained in comet Austin (1989c1) and Levy (1990c). HCN seems to be more abundant by at least a factor of two in periodic comets (P/Halley, P/Brorsen-Metcalf) than in non periodic comets (Wilson, Austin, Levy). This suggests a chemical difference between periodic and new comets. The very low upper limit obtained on the relative abundance of HC₃N ($5x10^{-5}$) shows that it is not the major lacking source of CN radicals.

Formaldehyde

The observations of the H₂CO 3₁₂-2₁₁ transition at 226 GHz in comet P/Brorsen-Metcalf gave only a marginal detection (S/N = 4). The 226 GHz line was easily detected in comet Austin (S/N = 10) and in comet Levy (S/N = 8). Observations of the 5_{15} - 4_{14} , 2_{12} - 1_{11} , 3_{03} - 2_{02} , 3_{22} - 2_{21} and 3_{21} - 2_{20} lines were negative, in agreement with excitation models (Bockelée-Morvan and Crovisier 1992). Production rates inferred in the assumption of release from the nucleus show that formaldehyde is a minor component of the nucleus with an abundance relative to water which ranges from 4×10^{-4} in Levy to 3×10^{-3} in P/Brorsen-Metcalf (Colom *et al.* 1992). These abundances are at least an order of magnitude less than the Vega IKS value for P/Halley (4%; Combes et al. 1988).

Hydrogen sulfide

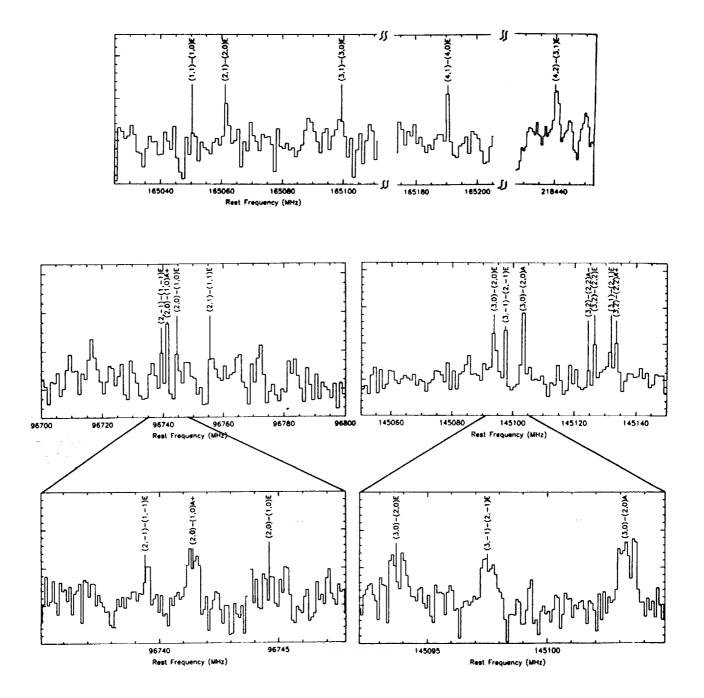
The observation of $H_2S \ 1_{10}-1_{01}$ at 169 GHz in comet Austin led to the first detection of hydrogen sulfide in a comet. In addition to the 169 GHz ortho line, the $2_{20}-2_{11}$ para line of H_2S at 217 GHz was detected in comet Levy. H_2S is a minor component, with a relative abundance of 2×10^{-3} . The other sulfur-bearing molecules observed (SO₂, OCS, H₂CS) are less abundant than hydrogen sulfide. (Crovisier et al. 1991.)

Comet and molecule	Date	Q a) [s-1]	Q/Q[H2O] b)
P/Brorsen-Metcalf (1989 X)			_
HCN J(1-0)	89/09/04-07	4.5×10^{26}	1.8×10 ⁻³
H ₂ CO 3 ₁₂ -2 ₁₁	89/09/04-07	7.6x10 ²⁶	3.0x10 ⁻³
Austin (1989cI)			
HCN J(1-0)	90/05/23	2.0×10^{25}	5.0x10 ⁻⁴
$H_2CO_{312}-2_{11}$	90/05/21-25	4.6×10^{25}	1.1x10 ⁻³
$H_2S 1_{10}-1_{01}$	90/05/24-25	1.1x10 ²⁶	2.7x10 ⁻³
CH ₃ OH (3,0)-(2,0)A	90/05/25	2.0x10 ²⁶	5.0x10 ⁻²
Levy (1990c)			
HCN J(1-0)	90/08/29	6.6x10 ²⁵	2.6x10 ⁻⁴
H ₂ CO 3 ₁₂ -2 ₁₁	90/08/26-30	1.0x10 ²⁶	4.0x10 ⁻⁴
H ₂ S 1 ₁₀ -101	90/08/30-31	5.0x10 ²⁶	2.0x10 ⁻³
CH ₃ OH (3,0)-(2,0)A	90/08/27	1.8x10 ²⁷	7.2x10 ⁻³
HC ₃ N J(24-23)	90/08/27	$< 1.2 \times 10^{25}$	< 5.0x10 ⁻⁵
SO2 717-606	90/08/29	< 6.0x10 ²⁶	< 2.5x10 ⁻³
OCS J(18-17)	90/08/28	< 5.0x10 ²⁶	< 2.0x10 ⁻³
H ₂ CS 414-313	90/08/28	$< 2.5 \times 10^{26}$	< 1.0x10 ⁻³

그는 그들 우리는 눈 Production rates and abundances.

a Assuming a parent distribution.

b Q[H₂O] from OH 18-cm observations: 2.5x10²⁹ s⁻¹ for P/Brorsen-Metcalf and Levy, 4.0x10²⁸ s⁻¹ for Austin.



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The spectra of methanol (CH₃OH) observed with the IRAM 30-m radio telescope in comet Levy (1990c). The upper panels show spectra observed with a 1-MHz spectral resolution. The lower panel shows parts of the spectra around 97 GHz and 145 GHz observed with a 100-kHz resolution.

Methanol

CH₃OH was detected in comet Austin through its $J(2-1) \Delta K = 0$ transitions at 97 GHz and its $J(3-2) \Delta K = 0$ transitions at 145 GHz. It was the first detection of methanol in a solar system body. A dozen of CH₃OH lines were detected in comet Levy, as is shown in the Figure. Methanol is a substantial component of the nucleus, with a relative abundance of the order of 1% in comets Austin and Levy.

Other molecules.

Limits on many other interesting lines of potential parent molecules were also obtained, either during dedicated searches or serendipitously. Here is a preliminary list (some of the corresponding limits on the production rates are given in the table; the other ones are presently under evaluation):

Hydrocarbons: CH₃CCH (propyne; several lines at 85.3 GHz); c-C₃H₂ (cyclopropenylidene: many lines).

OH species: HDO (deuterated water: 312-221 line at 225.897 GHz).

CHO species: many lines of HCOOH (formic acid), CH₃CHO (acetaldehyde), C₂H₅OH (ethanol). Nitrogen compounds: HC₃N (cyanoacetylene: 24-23 at 218.325 GHz); CH₃NH₂ (methylamine:

lines around 85.4 and 88.6 GHz); CH₂NH (methanimine: 1₁₀-0₀₀ at 225.5 GHz); HNCO (isocyanic acid); NH₂CHO (formamide).

Sulfur compounds: SO₂, OCS, H₂CS (see Crovisier et al. 1991).

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