Y3.N21/5:6/2097

NATIONAL ADVISORY COMMITTEE FOR AERONAUTICS

TECHNICAL NOTE 2097

IMPROVEMENT OF HIGH-TEMPERATURE PROPERTIES OF

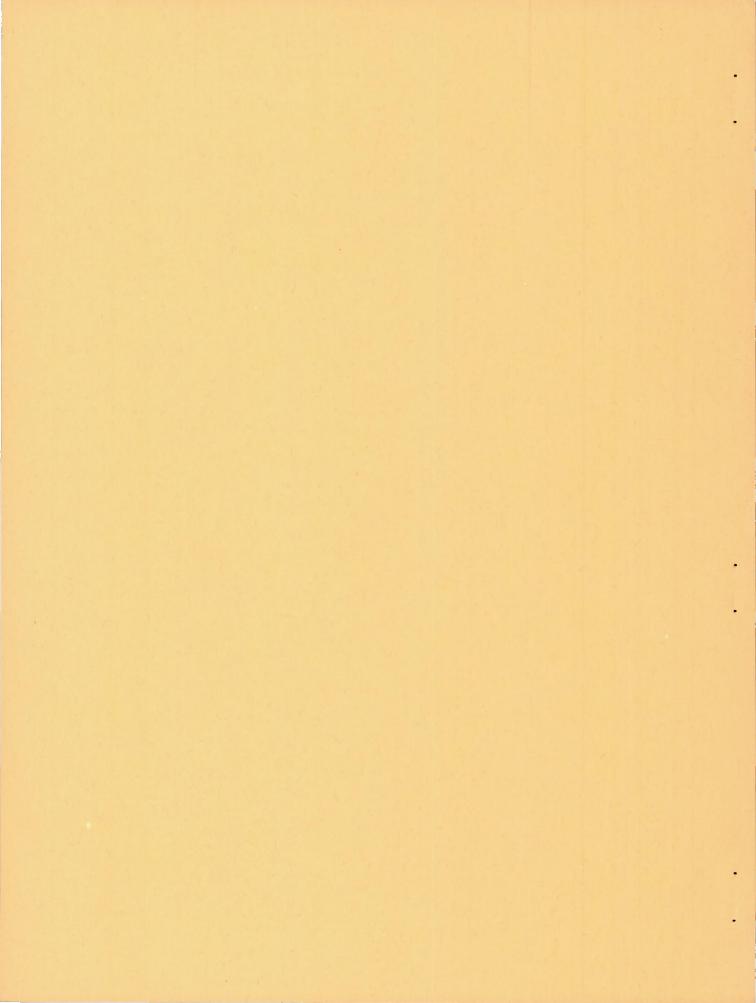
MAGNESIUM-CERIUM FORGING ALLOYS

By K. Grube, J. A. Davis, L. W. Eastwood, C. H. Lorig and H. C. Cross

Battelle Memorial Institute



Washington May 1950



NATIONAL ADVISORY COMMITTEE FOR AERONAUTICS

TECHNICAL NOTE 2097

IMPROVEMENT OF HIGH-TEMPERATURE PROPERTIES OF

MAGNESIUM-CERIUM FORGING ALLOYS

By K. Grube, J. A. Davis, L. W. Eastwood, C. H. Lorig and H. C. Cross

SUMMARY

An investigation was undertaken to obtain an improvement in the high-temperature properties and resistance to creep of magnesium-cerium forging alloys. From preliminary tests, it was decided that the best combination of tensile properties at room temperature and at 600° F, as well as maximum resistance to creep, was produced by the extrusion of a $2\frac{7}{8}$ -inch billet to 3/4 inch at 1000° F, at a speed of 4.11 inches per minute, followed by a solution heat treatment and an aging treatment. Also as a result of these tests, a magnesium-cerium-manganese alloy containing 6 percent cerium and 2 percent manganese was chosen as the base composition for the experimental alloys.

The experimental heats were made by minor element additions to this base composition. Tensile properties at 70° and 600° F were then obtained on these experimental alloys, and in addition most of the compositions were subjected to short—time creep tests at 600° F. The tensile properties were apparently improved by the addition of silver, barium, cadmium, nickel, sodium, strontium, or zirconium. Of these elements, nickel had the most outstanding beneficial effect. The most plausible method of improving creep resistance appeared to be the addition of a suitable alloying element. While some elements had a very harmful effect, a number of the fourth element additions, particularly aluminum, improved the creep properties somewhat.

INTRODUCTION

Available data prior to the initiation of this experimental program on magnesium—cerium forging alloys indicated that these alloys possess an unusual combination of low density and relatively high properties for elevated—temperature service up to at least 600° F. (A comparison of the high—temperature properties of magnesium—cerium wrought alloys

with those of two wrought aluminum alloys is given in the appendix.) While the magnesium—cerium—manganese alloys have not been widely used, they have been investigated experimentally, and forgings of an alloy similar to the EM-42 type had limited use in Germany during World War II. The EM-42 alloy contains a nominal 4 percent cerium and 2 percent manganese, while EM-62 contains a nominal 6 percent cerium and 2 percent manganese. The cerium content referred to in this report is actually rare—earth content of which approximately one—half is cerium. The objective of the research program was to obtain an improvement in these magnesium—cerium alloys in respect to their elevated—temperature tensile properties and resistance to creep.

The experimental work was concerned with:

- (1) The construction of indirect extrusion tools whereby a larger extrusion billet could be used with a given press capacity and a more uniform extruded product would be secured than is obtainable by the usual direct extrusion methods.
- (2) The conducting of preliminary tests to determine the proper extrusion temperature, extrusion rate, degree of reduction, and heat treatment to produce optimum high-temperature properties.
- (3) The processing and testing of a large number of experimental alloys, using the standard procedure developed under item (2). These experimental compositions were extruded and heat—treated, after which the tensile properties at room temperature and at 600° F were obtained. In addition, most of the compositions were also subjected to a screening—out creep test conducted at 600° F.

Although the magnesium—cerium alloys are fairly difficult to prepare and cast into sound ingots, no development of this phase of the project was required since a satisfactory melting procedure had been previously developed on a project for the Air Materiel Command, Wright—Patterson Air Force Base.

This program was conducted at the Battelle Memorial Institute under the sponsorship and with the financial assistance of the National Advisory Committee for Aeronautics.

PROCEDURE AND EQUIPMENT

Melting Procedure for Magnesium-Cerium Alloys

Most of the magnesium—cerium—base alloy compositions were prepared by the melting technique developed on the project previously mentioned.

By this method of melting, the fluid salt fluxes were replaced by dry nitrogen which protected the metal during the melting operation.

Specifically, this melting procedure consists of melting magnesiumcerium scrap and magnesium-base materials first in a steel crucible. If the charge so added to the steel crucible extends above the top, it is necessary to protect the charge, during the initial part of the melting, with Dow 181 reagent. As soon as the charge settles below the crucible top, the crucible is covered and dry nitrogen introduced over the surface of the charge. It is quite possible, of course, to perform the melting operation without loading the crucible above the top so that the entire melting operation can be conducted in a dry-nitrogen atmosphere. After the charge has been melted under the dry-nitrogen atmosphere and the temperature of about 1400° F is reached, the cover is removed and alloy additions made, with Dow 181 reagent used to provide protection against burning. After these alloy additions melt or dissolve and the crucible charge has been stirred to produce a uniform composition, the cover is replaced and the melt reheated to about 1400° F and held for about 30 minutes at 1400° F under the protective nitrogen atmosphere.

The metal is then poured into a preheated steel ingot mold 3 inches in diameter at the bottom, $3\frac{1}{2}$ inches in diameter at the top, and 14 inches deep, with Dow 181 again being used as a protective agent to prevent burning during the pouring operation. Since the melt temperature is high and the ingot mold is preheated to approximately the same temperature, it is possible to hold the melt for about 3 minutes before solidification begins. This provides an opportunity for certain nonmetallic particles to settle to the bottom of the ingot mold. Water is then introduced into the tank holding the ingot mold and, as the water level slowly rises, the ingot solidifies from the bottom upwards.

About 1/2 inch of the bottom of the ingot is sawed off, fractured, and examined to make certain that all the heavy nonmetallic particles are contained in a small volume near the bottom of the ingot. The top of the ingot is cropped off to produce a billet about 10 inches long. This is scalped to $2\frac{7}{8}$ inches in diameter preparatory to extruding.

Since most of the alloys contain 2 percent manganese, this is added in the form of A.S.T.M. Ml alloy melting stock. Accordingly, Ml alloy containing 2 percent manganese was prepared in a separate melting operation. To do this, 3 percent manganese in the form of manganese chloride is added to the commercially pure melt at 1650° F, using a protective covering of Dow 310 flux. After the manganese has been absorbed, the melt is cooled to 1450° F and poured off into ingot molds preparatory to remelting when preparing the experimental alloys. The use of any flux containing manganese chloride must be made in a separate melting operation.

If this is not done, the manganese chloride will react with the cerium and cause a serious loss.

After the scrap magnesium—cerium alloys and magnesium—base material have been melted, as described above, and the melt heated to 1400° F under a protective atmosphere of nitrogen, the cover is then removed, and the cerium as well as the minor elements are added as indicated. The minor element additions were made in a phosphorizer at the same time the cerium addition was made. However, a number of the higher melting—point metals were added to the Ml alloy composition at 1650° to 1750° F in a separate melting operation during which Dow 310 flux could be used for protection. These high melting—point metals could not be added to the magnesium—cerium alloy because the nitrogen atmosphere is not protective at the high melt temperatures required to obtain proper solution of such difficultly soluble elements.

Although a few experimental alloys containing beryllium as a minor element were melted under Dow 220 flux following the recommended commercial practice, this procedure was generally avoided because the Dow 220 did not always separate satisfactorily from the metal, and thus caused flux inclusions.

Extrusion Equipment

The extrusion equipment included a 250-ton press. This press was equipped with indirect extrusion tools and figure 1 shows an assembly drawing of them. In order to reduce the heat loss from the container through the bottom block, 1/8 inch of Transite insulation was placed between the container and the bottom block on which it rested.

The indirect extrusion method was chosen over the direct method because (1) lower maximum pressures are required, thereby permitting the use of a larger extrusion billet for a given press capacity, and (2) the extruded rod is reported to be more uniform from one end to the other because of the manner in which the metal flows from the container through the die.

A photograph of the hydraulic press with the extrusion equipment in place along with two scalped billets is presented as figure 2.

By the indirect extrusion method, the extrusion billet is placed in the preheated container and, as the die at the end of the ram moves downward, the metal is extruded through the hollow ram, and in this instance, up through the top of the press.

Creep Test

The equipment used to obtain the data on the creep properties of the experimental alloys is illustrated by figure 3. Each testing unit is equipped with a chromium-plated copper or steel shell furnace, 6 inches in diameter and 18 inches long, wound with 14-gage Chromel A wire and insulated with silica gel. These furnaces may be used for creep or stress-rupture tests up to 1800° F, but 600° F was the maximum temperature employed in this investigation. A small window is provided at both the front and back of the furnace for measuring the deformation of the test specimen by optical means.

A standard 0.505—inch test specimen and calibration specimen are shown by figure 4. The temperature gradient in the furnace can be controlled and changed by means of external shunts along the tap furnace windings. For this purpose the calibration specimen shown in figure 4 is used, and the furnace is shunted so that the maximum temperature variations measured by thermocouples at positions T_1 , C_1 , B_1 , T_0 , C_0 , and B_0 do not exceed $\pm 3^{\circ}$ F at the test temperature. In some cases, even smaller variations are obtained. All temperature variations are kept below the maximum allowed by reference 1. During the actual creep or stress—rupture tests, thermocouples are located at positions T_0 and B_0 and are used for adjusting the test temperature. Thermocouples are also placed in positions T and B, the control thermocouple at B, and the recorder thermocouple at T. These thermocouples can be replaced while the test is in progress.

The load is applied to the test specimens by means of a lever arm with a 9:1 ratio.

Although stress-rupture tests were not made on this project, the illustrated equipment is fitted with a switch mounted on the lever-arm stop under the lever arm in order to turn off the power to the furnaces and stop the clock when the test specimens break. The clock measures the duration of the test to one-tenth of an hour.

The test temperature of each furnace is maintained by Tag Celectray indicating controllers equipped with a throttling mechanism for closer control. Foxboro controllers with a heater loop anticipating device are also used for the control on some test units. For measurement of deformation, platinum strips are used which are attached to the shoulders of the specimen, one at each end of the gage length. Figure 4 shows the strips in position upon the test specimens. A series of very fine cross marks is ruled on each strip. Changes in length of the test specimen are measured by determining the change in distance between two chosen cross marks, one on each strip.

The microscope, with which deformation readings are made, has an eyepiece fitted with a filar micrometer, and is mounted on a graduated screw. Calibration shows the smallest division of the filar eyepiece to read 0.00005 inch which, on a gage length of about 2.3 inches, provides readings slightly over 0.00002 inch per inch or about 0.002 percent. Deformation readings are usually made daily by two observers.

High-Temperature Tensile Test

The only elevated temperature at which the experimental compositions were tested was 600° F. Briefly, the high-temperature tensile test consisted in placing a specimen in a specially calibrated furnace mounted on the testing machine. The furnace was calibrated so that the temperature variations between seven thermocouples were only ±5° F. Generally, an even closer temperature range was achieved. Upon reaching the desired temperature, the load was applied at the rate of 0.0l inch per minute per inch of gage length until the 0.2-percent yield strength was reached; the rate was then increased to 0.3 inch per minute per inch of gage length until rupture occurred. All of the bars were stabilized prior to the testing temperature. The stabilized and machined bars were inserted into the furnace, maintained at the testing temperature, and the furnace temperature again brought to equilibrium. This required 20 to 30 minutes after which time the test bar was broken. Stress-strain curves were obtained on each specimen, thus providing the data on proportional limit, modulus of elasticity, and yield strength.

PRELIMINARY TESTS

Before the work on the experimental compositions could be undertaken, it was necessary to determine the optimum extrusion speed, the extrusion temperature, the degree of reduction, and the heat treatment to be employed in the experimental work. It was also necessary to determine which magnesium—cerium—manganese alloy, EM-42 or EM-62, should be used as a base material. These preliminary tests were carried out on alloys containing 4, 6, or 10 percent cerium, with and without manganese, and 18 heats were prepared for this purpose.

The machined billets $2\frac{7}{8}$ inches in diameter and 10 inches long were prepared as described. The billets were then preheated in a furnace containing sulphur dioxide as a protective agent, brought to temperature, and then transferred to the preheated container on the extrusion equipment. This container was heated by the electrical-resistance method using a controller for proper operation. Thermocouples were inserted into the container near the top and bottom, and their temperatures were recorded along with the billet temperature, the speed of extrusion, and the pressure required.

Effect of extrusion rate and temperature.— Standard EM-62 alloy ingots were extruded through a 3/4-inch-diameter die at extrusion rates of 0.72 to 27.3 inches per minute, at extruding temperatures of 1000°, 900°, 850°, 800°, and 750° F. These bars were heat-treated for 3 hours at 1000° F and aged for 16 hours at 400° F; the tensile bars were machined, and room-temperature tensile properties obtained. The results of these tests are listed in table 1 and are graphically represented by figure 5.

As indicated by tables 1 and 2, for a given extrusion rate, the best tensile properties at room temperature and at 600° F were obtained at extrusion temperatures of 900° to 1000° F. The highest tensile and yield strengths and lowest elongations were obtained, for a given extrusion temperature, by the slowest extrusion rate; whereas, the most rapid extrusion rate produced the lowest tensile and yield strengths and the highest elongation values. For this reason, 4.11 inches per minute was the extrusion rate chosen for the balance of the experimental work.

Effects of degree of reduction and of heat treatment.— The effects of extrusion conditions and of heat treatment on the tensile properties at room temperature and the tensile and creep properties at 600° F were investigated by extruding standard EM-62 alloy at 800° and 1000° F through 1/2-inch- and 3/4-inch-diameter dies, at an extrusion speed of 4.11 inches per minute. Each set of specimens was tested after a complete heat treatment and after stabilization only. The solution heat treatment consisted of 16 hours at 1040° F, and the aging treatment was carried out 16 hours at 400° F. The stabilizing treatment performed on the heat-treated bars as well as the as-extruded bars tested at 600° F consisted of 24 hours at 650° F. This stabilizing treatment has been the standard one used on all magnesium-cerium alloys prior to testing at 600° F. The billets extruded through a 3/4-inch-diameter die were given a 93.75-percent reduction; whereas, those extruded through a 1/2-inch-diameter die were given a 97.22-percent reduction.

Samples for each extruding condition and each heat treatment were subjected to tensile tests at room temperature and at 600° F, as well as to creep tests at 600° F. The data obtained on this part of the experimental work are listed in the first 10 lines of table 2, heats A2985, A2986, A3056, and A3057.

While the creep data are not strictly comparable, as indicated by the data in table 2, it will be noted that the room-temperature tensile properties obtained by the higher degree of reduction are not markedly different from those obtained by extruding to 3/4-inch-diameter bar. This is probably contrary to ordinary experience where direct extrusion methods are employed. Had a direct extrusion method been used, it is

quite possible that some benefits may have been obtained by the higher reduction. However, with the experimental extrusion methods employed on this project, it is evident that little is gained by the higher reduction by the extrusion process. Furthermore, the use of a 1/2-inch-diameter bar would very seriously complicate the high-temperature tensile tests.

Figure 6 shows the microstructures at 100% of EM-62 alloy given a reduction by extrusions of 85.9, 93.75, and 97.22 percent. While these photomicrographs show that a fairly substantial improvement is obtained in the uniformity of the structure when the reduction is increased from 85.9 to 93.75 percent, a relatively slight increase in uniformity of structure has been obtained by increasing the reduction from 93.75 to 97.22 percent.

Heat treatment has a very marked effect upon the 600° F tensile strength and creep properties; however, the heat treatment has very little effect upon the room-temperature tensile properties. Thus, a comparison of two sets of bars from the same heat in which one set was given a solution, aging, and stabilization treatment, whereas the other was stabilized only, shows that an increase of about 1500 psi in tensile strength is obtained on the bars given the complete heat treatment at 600° F. In addition, the short-time creep tests at 600° F and a 1300-psi load show the minimum creep rate is 0.013 percent per hour for the stabilized material, whereas 0.0005 percent per hour is the minimum creep rate when the same alloy has been solution-heat-treated, aged, and then stabilized. It is obvious, on the basis of these results, that a full solution heat treatment followed by aging was employed on all the experimental compositions. Those bars which were tested either in tension or in creep at 600° F were also stabilized for 24 hours at 650° F.

It was initially thought that the best procedure would be to extrude the $2\frac{7}{8}$ -inch-diameter billets to 1-inch squares and then forge these 1-inch squares to 3/4-inch-diameter bar, from which the test specimens could be machined. It was soon discovered, however, that, while this procedure could be carried out, it was rather tedious and slow because the experimental alloys did not necessarily forge very satisfactorily even after the extrusion operation. Therefore, this procedure was abandoned and the $2\frac{7}{8}$ -inch-diameter billets were extruded to 3/4-inch-diameter rod which was then heat-treated and subjected to the routine tests.

Effect of cerium and manganese contents.— A heat of EM—42 alloy containing 4 percent cerium and 2 percent manganese was prepared in order to evaluate the desirability of using the EM—42 composition as a

base rather than EM-62. The data on heat A2980 in table 2 show the tensile properties at room temperature and at 600° F as well as some results of the creep tests. Again, a very marked improvement was obtained by the solution heat treatment followed by aging and then stabilization. This improvement is indicated not only by the markedly improved high-temperature tensile strengths but also by the very much lower creep rate. It will be noted from these data that EM-42 and EM-62, heat-treated and aged, produce about the same room-temperature tensile properties and resistance to creep at 600° F. However, at 600° F, the yield strength of EM-42 alloy appears to be about 500 psi lower and the tensile strength about 1000 psi lower than the strengths of the EM-62 composition. Therefore, the EM-62 composition was chosen as a base to which the various experimental additions were made.

A comparison of heats A2857 and A2838 in table 2 shows that an increase in cerium content to approximately 10 percent reduces the elongation, but does not increase the yield or tensile strength markedly at room temperature. At 600° F, both the yield strengths and the ultimate strengths are substantially higher than those obtained by 6 percent cerium. It is remarkable also that this improvement in tensile properties at 600° F is accompanied by a substantial increase in elongation. However, there is a good possibility that such a high cerium content may increase the difficulties of forging; therefore, this high cerium content was not selected as a base for the experimental alloy additions. These improved tensile and yield values warrant further consideration when developing the optimum composition of the magnesium-cerium type of alloy. Data on heats A3225 and A3226 indicate that, on the basis of the tensile data only, the manganese-free alloys have quite favorable tensile properties at room temperature and at 600° F. However, in the absence of manganese, the minimum creep rate of EM-60 alloy is about 16 times higher than when manganese is present. Consequently, no alloy development was based on the EM-60 composition.

Established procedure. On the basis of the results of the preliminary tests, the following procedure was established as standard:

- (1) Ingots were extruded by the indirect method from $2\frac{7}{8}$ inches in diameter to 3/4 inch at a ram speed of 1/4 inch per minute, corresponding to about 4 inches of extrusion per minute, and at a billet and container temperature of 900° to 1000° F.
- (2) Extruded 3/4—inch bar was heat—treated 16 hours at 1040° F and aged 16 hours at 400° F. The portion of extruded bar subjected to tests at 600° F was also given a stabilizing treatment consisting of 24 hours at 650° F.

- (3) With a few minor exceptions, all of the experimental alloy additions were made to a base composition consisting of 6 percent cerium and 2 percent manganese.
- (4) After heat treatment, all extrusions were machined, tensile properties obtained at room temperature and at 600° F, and creep data secured from tests conducted at 600° F and a 1300-psi load.

EXPERIMENTAL MAGNESIUM—CERIUM ALLOYS

One hundred and nine experimental magnesium heats were made, most of which had a base composition of 6 percent cerium and 2 percent manganese. Usually also the minor additions were made in quantities of 0.1, 0.5, and 2 percent. Lesser amounts were added when the solubility of the third element in the liquid was low. The single minor element additions included:

Aluminum	Cobalt	Molybdenum	Tantalum
Silver	Columbium	Sodium.	Tellurium
Arsenic	Copper	Nickel	Titanium
Boron	Chromium	Phosphorus	Thallium
Barium	Iron	Antimony	Vanadium
Beryllium	Germanium	Lead	Tungsten
Bismuth	Indium	Tin	Zinc
Calcium	Potassium	Silicon	Zirconium
Cadmium	Lithium	Strontium	

The composition of all the experimental alloys prepared and data on their tensile properties at room temperature and 600° F are listed in table 2. In addition, the results of the short—time creep tests are also listed on many of the alloy compositions, but this portion of the testing is not yet completed.

The experimental alloys containing minor element additions of silver, barium, cadmium, nickel, sodium, strontium, and zirconium had tensile properties somewhat higher than EM-62 alloy at 600° F. Additions of nickel resulted in the higher tensile properties at 70° and 600° F. Quantities of 0.1, 0.5, and 2.0 percent nickel produced a marked improvement in the tensile strengths and yield strengths at 600° F, and quite satisfactory tensile properties at room temperature were also obtained. On the basis of these data, the yield strength is raised about 2500 psi and the tensile strength about 3000 psi at 600° F. However, in view of the possibility of some experimental variations, these results need to be verified before complete acceptance. The effect of sodium was particularly interesting in that it caused extreme brittleness and low properties at room temperature, but fairly high tensile

properties were obtained at 600° F. Because of the increase in ductility of this alloy, the tensile properties were actually higher at 600° F than they were at room temperature.

As indicated by the comparative data obtained from the literature and on this project on the relative high-temperature properties of aluminum- and magnesium-base alloys, it is evident that the magnesiumcerium wrought alloys have excellent tensile properties, but in comparison with cast alloys have rather poor resistance to creep at 600° F. Minor additions of barium, bismuth, beryllium, calcium, cadmium, lead, tellurium, and aluminum may have produced a minor improvement in the creep properties of EM-62 extrusion at 600° F. Aluminum, at least in the initial tests, is outstanding so far in that it produced about a tenfold reduction in the average minimum creep rate in the standard creep test. Figure 7 illustrates typical time-deformation curves showing (1) a comparison of EM-42 and EM-62, (2) the effect of heat treatment, (3) a harmful (2.4-percent-cadmium) addition, and (4) a beneficial (0.5-percent-aluminum) alloy addition. Some elements, of which cadmium is outstanding, have a very harmful effect on resistance to creep.

It was thought possible that, since among wrought light alloys a fine—grain material has a markedly poorer resistance to creep than a similar composition having coarse grain, some effort should be made to produce a coarser—grain extrusion. Two methods were investigated in an effort to increase the grain size as follows:

- (1) Cold-rolling the extrusion a small amount prior to the solution heat treatment. Such small cold reductions followed by a high-temperature anneal frequently produce a relatively coarse grain.
- (2) Using a prolonged heat—treating period at 1040° F, at which temperature grain growth would normally be expected to occur, at least slowly.

Neither of these methods of producing an increase in grain size was particularly successful, and it is concluded that the improvement in creep properties must be brought about by alloy additions.

HIGH-TEMPERATURE OXIDATION OF MAGNESIUM-CERIUM ALLOYS

Creep tests on magnesium—cerium casting alloys conducted on another project (reference 2) showed that, at least during periods of high humidity, prolonged exposure to 700° F produced a very marked scaling of the alloy with preferential attack of the cerium—rich phase in the alloy. Such oxidation would, of course, seriously limit the use of

magnesium—cerium alloys, and it was also noted that this attack was much less though still serious after exposure for prolonged periods at 600° F.

There are, in general, two reasonably practical ways of improving the resistance of these alloys to oxidation. These methods are as follows:

- (1) By means of an alloy addition, a protective film may be formed, preventing further oxidation. While there is no obvious example whereby this principle is applied in light—alloy metallurgy, it is well known that silicon, chromium, or aluminum added to certain ferrous materials will greatly increase their resistance to oxidation by virtue of the surface film formed. Beryllium in both aluminum and magnesium melts is an example whereby the protective film produced by the beryllium reduces oxidation of the molten metal.
- (2) Chemical coatings applied to the magnesium—cerium alloy may be protective and, thereby, increase the resistance of the alloy to oxidation.

Both methods have been investigated. The possibility of using the first method was examined by testing all of the heat—treated, machined test bars of all experimental alloys fractured at room temperature, including all compositions up to heat A3275. These machined test bars were exposed to oxidation at 700° F for 900 hours. In addition to these, machined cast test bars were subjected to various types of coatings, and coated and uncoated samples were also exposed to the 700° F atmosphere at the same time. Data on the chemical coatings are listed in table 3, and the data on the experimental test bars and on the coated and uncoated cast samples are listed in table 4. On the basis of these results, the following conclusions are made:

- (1) Extrusions scale and oxidize more than castings of the EM-62 composition.
- (2) A reduction in cerium content, the elimination of manganese, or the addition of 0.5 or 2.0 percent aluminum or 1.0 percent silicon eliminates the scale formed on the EM-62 extrusions exposed 900 hours at 700°F, but does not adequately reduce oxidation of the cerium-rich constituent adjacent to the surface.
- (3) Only the fluosilicic—acid treatment has markedly reduced the oxidation attack of the cerium—rich phase at the surface of the cast EM-62 specimen. This coating has not been tried on an extrusion of this alloy.

SUMMARY OF RESULTS

The results from the investigation of magnesium—cerium alloys may be summarized as follows:

- 1. Eighteen heats of magnesium containing 4, 6, or 10 percent cerium, with and without manganese, were made to establish the best base compositions for the experimental work and to determine the best extrusion and heat—treating conditions to employ. As a result of this portion of the work, a base composition consisting of 6 percent cerium and 2 percent manganese was chosen for additions of the fourth element. The extrusion of a $2\frac{7}{8}$ —billet to 3/4 inch at 1000° F and a speed of 4.11 inches per minute followed by a solution heat treatment and an aging treatment produced the best combination of tensile properties at room temperature and at 600° F, as well as maximum resistance to creep. Alloys tested at 600° F were also subjected to a stabilizing treatment consisting of 24 hours at 650° F prior to testing. It was found that the heat—treating and aging operation greatly improved the resistance to creep as compared with that obtained by a stabilizing treatment only.
- 2. One hundred and nine experimental magnesium alloys containing 6 percent cerium, 2 percent manganese, and a fourth element addition were prepared. After the extrusion and heat—treating operations, tensile and creep bars were machined from the extrusion product, and tensile properties at 70° and 600° F were obtained on all the experimental heats. In addition, short—time creep tests at 600° F and a load of 1300 psi were carried out on over one—half of the experimental compositions. It was found that the tensile properties of the EM—62 composition at 600° F were apparently improved by the following single element additions: Silver, barium, cadmium, nickel, sodium, strontium, or zirconium. Of these elements, nickel had the most outstanding beneficial effect.
- 3. Since the creep resistance of the magnesium—cerium—manganese wrought alloys is markedly inferior to that of similar compositions in cast form, considerable emphasis was placed upon the attainment of better resistance to creep. Efforts to improve creep resistance by increasing the grain size of the extrusions were unsuccessful. As a result, the most plausible method of improving creep resistance appeared to be the addition of a suitable alloying element. While some elements had a very harmful effect, a number of the fourth element additions, particularly aluminum, improved the creep properties somewhat. It was found that aluminum reduces the creep rate in the standard test to about

one—tenth of the value which is obtained without aluminum. The effect of aluminum needs further investigation, however, to determine the optimum composition and the precise amount of the beneficial effects obtained.

4. Experimental work on cast alloys carried out on another project (reference 2) indicates that magnesium—cerium alloys are oxidized very severely when they are exposed to long periods at temperature of 700° F or even at 600° F. There is some indication that this oxidation is more severe when the humidity is high. This oxidation will, of course, seriously limit the application of magnesium-cerium alloys because of the surface attack. The two most promising methods of reducing this surface oxidation are by (a) alloy additions and (b) the application of chemical coatings, preferably those which do not markedly change the dimension. Both methods of reducing surface oxidation were investigated and it was found that, when exposed for 1000 hours at 700° F, EM-62 extrusions oxidize more readily than castings, and that 0.5 or 2.0 percent aluminum or 1.0 percent silicon added to EM-62 extrusions reduced the scaling somewhat. A fluosilicic—acid coating, described in table 3, markedly reduced the oxidation of the magnesium-cerium constituent in cast EM-62 alloy, but this coating has not yet been tried on extruded alloys.

Battelle Memorial Institute Columbus, Ohio, Aug. 12, 1947

APPENDIX

COMPARISON OF MAGNESIUM—CERIUM WROUGHT ALLOYS WITH TWO

WROUGHT ALUMINUM ALLOYS

A comparison of the magnesium—cerium wrought alloys with two wrought aluminum alloys, 18S—T and 32S—T, which are sometimes used for forgings for high—temperature applications, is given in the following discussion. Information on the high—temperature properties of aluminum alloys is taken from reference 3 and the high—temperature properties of magnesium—cerium wrought alloys are taken largely from reference 4. Figures 8, 9, and 10 show a comparison of the effects of temperature on the tensile strength, yield strength, and percentage of elongation, respectively, of the two wrought aluminum alloys, 32S—T and 18S—T, and the two magnesium—cerium wrought alloys, EM—62 and EM—42. Alloy EM—62 contains a nominal 6 percent cerium and 2 percent manganese, whereas alloy EM—42 contains a nominal 4 percent cerium and 2 percent manganese. The cerium, of course, refers to the rare—earth content rather than the cerium content which constitutes approximately one—half of the rare earths present.

In order to differentiate clearly between the data on the aluminum alloys and the magnesium alloys, the magnesium alloys are represented by solid lines; whereas, the aluminum alloys are represented by dashed lines in these figures. It will be noted that, at temperatures up to and including 300° F, the properties of the aluminum alloys are somewhat superior. At 400° F, the properties of the aluminum alloys drop quite rapidly with the result that, at this temperature and at 500° and 600° F, the tensile properties of the wrought magnesium—cerium alloys are substantially superior to the two aluminum alloys commonly used as forgings for high-temperature applications.

There are practically no comparable data available on the effect of temperature on the fatigue properties of the two aluminum and two magnesium—cerium alloys.

There are available some data on the effects of temperature on the creep properties of the four alloys being considered. It is quite difficult to make reliable comparisons because the creep tests are not always conducted under comparable conditions, particularly when the data are obtained in two different laboratories. Furthermore, in some instances, the initial deformation is subtracted from the elongation so that curves representing total elongation against load are not comparable. Some of the existing data on the creep properties of EM-62,

EM-42, 18S-T, and 32S-T are indicated in table 5. This table shows the creep rate of the various alloys at various temperatures and loads. The listed average creep rate is for the interval 48 to 96 hours of exposure to the test. This is a rather short time of exposure, and many alloys do not obtain their minimum creep rate during this interval. It is well known that grain size, at least in the wrought aluminum alloys, has a very large effect on the creep properties since the creep rate for a given loading condition will increase markedly with decreasing grain size. This is probably true of the magnesium-cerium alloys also, and this is one factor not usually reported for the creep data. Another factor is the effect of stabilization. It will be evident from the data in table 5 that a stabilized magnesium-cerium alloy has a higher creep rate than an unstabilized alloy exposed to the same conditions of temperature and load. This effect of stabilization is another factor which makes creep-rate data difficult to compare when they are obtained in different laboratories or under different conditions. If the magnesium-cerium alloys listed in table 5 had been heat-treated, their creep properties would have been better. In view of the effects of stabilization, grain size, heat treatment, and possibly other factors, the data in table 5 are not comparable to a high degree. However, there are some indications that the magnesium-cerium wrought alloys have somewhat higher creep rates under the listed conditions of load than the aluminum alloys, 18S-T or 32S-T.

Data obtained at Battelle show a comparison of creep rates of cast and wrought EM-62 as follows:

Alloy and conditions	Temperature (°F)	Load (psi)	Duration of test (hr)	Total deformation (percent)	Minimum creep rate (percent/hr)
EM-62 heat-treated, aged, and stabi- lized, cast		2500	100	0.18	0.0007
EM-62 heat-treated, aged, and stabi- lized, extruded	600	2500	113	.27	.0380

Table 6 shows a comparison of the creep properties of cast and wrought 142 and EM-62 alloys taken from references 3 and 4. It will be noted from these data that the wrought aluminum alloys were only slightly inferior to the same composition in cast form. On the other hand, at a load of 2200 psi, the wrought EM-62 alloy at 500° F has a creep rate 10 times that of the same composition in the cast form, but at a temperature 100° F higher.

In general, cast EM-62 in the heat-treated, aged, and stabilized condition at 600° F and a 2500-psi load has a creep rate similar to that of extruded EM-62 in the heat-treated, aged, and stabilized condition at 600° F, but at only a load of 1300 psi. The importance of improving the creep properties of magnesium-cerium alloys is therefore evident.

REFERENCES

- 1. Anon.: Recommended Practices for Conducting Long-Time High-Temperature Tensile Tests of Metallic Materials. Designation: E22-41.
 A.S.T.M. Standards, 1946, pt. 1A.
- 2. Grube, K., Eastwood, L. W., Lorig, C. H., and Cross, H. C.: The Development of Magnesium—Cerium Alloys for Elevated—Temperature Service. Summary rep., Contract NOa(s) 8957, Materials Branch, Airborne Equipment Div., Bur. Aero., Navy Dept., May 31, 1948.
- 3. Wyman, L. L.: High Temperature Properties of Light Alloys (NA-137). I Aluminum. OSRD No. 3607, Serial No. M-251, War Metallurgy Div., NDRC, April 15, 1944.
- 4. Wyman, L. L.: High Temperature Properties of Light Alloys (NA-137). II Magnesium. OSRD No. 4150, Serial No. M-292, War Metallurgy Div., NDRC, Sept. 18, 1944.
- 5. Kennedy, R. R.: Creep Characteristics of Aluminum Alloys. A.S.T.M. (Philadelphia), 1935.

TABLE 1.— EFFECT OF EXTRUSION RATE AND TEMPERATURE ON THE ROOM-TEMPERATURE TENSILE PROPERTIES OF EM-62 AND EM-42 ALLOYS

			Melt	ing data			-	Extrusion	data			Heat treat	ment		Room-tem	perature t	ensile prope	rties
Heat	Inte	nded compo	sition	Alloying	Holding	Pouring		Rate	Tem	perature	2	Solution	Aging	Grain size, average diameter	Elongation		Yield strength,	Tensile
Неас	Ce (percent) (a)	Mn (percent)	Other (percent)	temperature of Ce (°F)	time after Ce addition (min)	temperature (°F)	Reduction (percent)	(in./min of 3/4-in diam. rod)	Ingot (°F)	Container (°F)	Time (hr)	Temperature (°F)	at 400° F (hr)	(in.)	(percent in 2 in.)	of area (percent)	0.2-percent offset (psi)	strength (psi)
A2465	6	1.9	None				93.75	27.7	950	900	3	1000	16					
A2465-1	6	1.9	None				Sample cut	extruded		nd of bar	3	1000	16		3.0	3.5		32,500
A2465-3	6	1.9	None				Sample cut	t 22½ inches extruded		end of bar	3	1000	16		4.5	4.4	1	32,750
A2465-4	6	1.9	None				Sample cut	t 291 inches extruded		end of bar	3	1000	16		4.0	3.9		32,000
A2465 F	6	1.9	None				extrus	luced 93.75 ion and forg	ed to		3	1000	16		1.9	3.7		35,500
A2465-5	6	1.9	None				Sample cut	t 43 inches extruded		nd of bar	3	1000	16		4.5	5.1		29,500
A2465-6	6	1.9	None					t 50 inches es from end			3	1000	16		5.5	5.8		29,000
A2981	14	1.9	None	1410	22	1410	93.75	27.7	950	900	3	1000	16		7.1	7.6		33,750
A2980	4	1.9	None	1420	21	1420	097.22	27.7	950	910	3	1000	16		8.4	9.0		33,500
A2983-1	6	1.9	None	1400	20	1400	93.75	.72	1000	1000	3	1000	16		3.0		30,750	35,000
A2983-2	6	1.9	None	1400	20	1400	93.75	3.6	1000	1000	3	1000	16	0.0016	4.3		27,500	32,750
A2983-T3	6	1.9	None	1400	20	1400	93.75	8.9	1000	1000	3	1000	16		7.0		24,000	30,750
A2983-T4	6	1.9	None	1400	20	1400	93.75	27.3	1000	1000	3	1000	16		9.8		20,000	29,000
A2984-5	6	1.9	None	1400	20	1400	93.75	4.11	900	900	3	1000	16		5.0		27,500	33,250
A2984-T3	6	1.9	None	1400	20	1400	93.75	8.9	900	900	3	1000	16		5.0		22,750	30,500
A2984-T4	6	1.9	None	1400	20	1400	93.75	27.3	900	900	3	1000	16		8.5		18,500	28,250
A2982-5	6	1.9	None	1410	21	1410	93.75	4.11	850	840	3	1000	16		8.5		19,750	29,500
A2982-3	6	1.9	None	1410	21	1410	93.75	8.9	850	850	3	1000	16		8.0		17,250	28,500
A2982-4	6	1.9	None	1410	21	1410	93.75	27.3	850	840	3	1000	16		8.3		18,750	29,500
A3057-5	6	1.9	None	1400	33	1400	93.75	4.11	800	800	3	1000	16		6.0		17,500	27,750
A3057-T3	6	1.9	None	1400	33	1400	93.75	8.9	800	780	3	1000	16		5.5		16,750	26,250
A3057-4	6	1.9	None	1400	33	1400	93.75	27.3	800	800	3	1000	16		4.8		18,250	27,500
A2987-5	6	1.9	None	1400	33	1400	93.75	4.11	750	750	3	1000	16	.0014	6.8		16,750	27,500
A2987-T3	6	1.9	None	1400	33	1400	93.75	8.9	750	740	3	1000	16		6.0		17,000	27,000
A2985-4	6	1.9	None	1400	32	1400	93.75	27.3	750	750	3	1000	16		4.5		17,500	26,750

aRare—earth content.
bNo stabilizing treatment.

c_{1/2-in.-diam.' bar.}

TABLE 2.- EFFECT OF EXTRUSION TEMPERATURE, EXTRUSION REDUCTION, HEAT MAGNESIUM-CERIUM FORGING

				Melting	data					Extrusion	data			Не	at treatme	ent		1
	Inten	ded compo	sition	Alloying temper- ature of	Time minor element	Alloying temper-	Holding time	Pouring	Reduction	Rate (in./min	Ten	nperature	So	lution	Aging	Stal	bilizing (c)	Grain size,
Heat	Ce (percent) (b)	Mn (percent)	Other (percent)	minor element (°F)	held at tem- perature (min)	ature of Ce (°F)	after Ce addition (min)	temper- ature (°F)	(percent)	of 3/4-in diam. rod)	Ingot (°F)	Container (°F)	Time (hr)	Temper- ature (°F)	time at 400° F (hr)	Time (hr)	Temper- ature (°F)	diam. (in.)
A2985T	6	1.9				1400	32	1400	93.75	4.11	1000	1000				24	650	0.001
A2986	6	1.9				1400	32	1400	93.75	4.11	1000	1000				24	650	
A2986T	6	1.9				1400	32	1400	d97.22	4.11	1000	1000				2.4	650	
A2985TA	6	1.9				1400	32	1400	93.75	4.11	1000	1000	16	1040	16	24	650	.000
A2986A	6	1.9				1400	32	1400	,93.75	4.11	1000	1000	16	1040	16	24	650	
12986TA	6	1.9				1400	32	1400	d97.22	4.11	1000	1000	16	1040	16	24	650	
13057A	6	1.9				1400	33	1400	93.75	4.11	800	800	16	1040	16	24	650	
A3056	6	1.9				1400	30	1400	93.75	4.11	800	800				24	650	
A3056T	6	1.9				1400	30	1400	d97.22	4.11	800	800				24	650	
A3056TA	6	1.9				1400	30	1400	497.22	4.11	800	800	16	1040	16	24	650	
12980	4	1.9				1420	21	1420	93.75	4.11	950	950				24	650	
2980A	4	1.9				1420	21	1420	93.75	4.11	950	950	16	1040	16	24	650	
2316	e6.08	1.79	0.01 B	1400		1400	24	1400	93.75	4.11	995	990	16	1040	16	24	650	
3059	6	2	.1 Ba	1400		1400	38	1400	93.75	4.11	1000	1015	16	1040	16	24	650	
3060	6	2	.5 Ba	1400		1400	32	1400	93.75	4.11	1000	995	16	1040	16	24	650	
2455	e _{5.81}	2	.1 Bi	1400		1400	22	1420	93.75	4.11	1000	1005	16	1040	16	24	650	
3201	6	2	.5 Bi	1390		1400	26	1400	93.75	4.11	1000	975	16	1040	16	24	650	
2457	6.78	1.85	.89 Bi	1400		1400	18	1400	93.75	4.11	1000	1015	16	1040	16	24	650	
2473	e _{5.02}	2	e _{1.92} Cd	1430		1430	23	1390	93.75	4.11	1000	1005	16	1040	16	24	650	
3204	6	-2	1.5 Cd	1270		1400	33	1400	93.75	4.11	1000	970	16	1040	16	24	650	
2452	e _{5.72}	1.85	2.41 Cd	1450		1450	20	1330	93.75	4.11	1000	990	16	1040	16	24	650	
2451	e6.00	2	e.09 Cu	1460		1460	20	1320	93.75	4.11	1000	990	16	1040	16	24	650	
2448	6.19	1.93	.60 Cu	1430		1430	11	1320	93.75	4.11	1000	980	16	1040	16	24	650	
2458	5.82	1.90	.14 In	1440		1440	25	1410	93.75	4.11	1000	1010	16	1040	16	24	650	
2459	5.94	1.84	.61 In	1420		1420	24	1420	93.75	4.11	1000	945	16	1040	16	24	650	
3058	6	2	.1 Pb	1400		1400	40	1400	93.75	4.11	1000	980	16	1040 1040	16	24	650	
12461	e _{5.52}	2	e.54 Pb	1450		1400 1400	20	1400	93.75	4.11	1000	1010 940	16 16	1040	16	24	650 650	
2462	6.48 e _{5.43}	1.73	2.12 Pb e.05 Sb	1400 1460		1460	21 22	1410 1420	93.75 93.75	4.11	1000	970	16	1040	16 16	24 24	650	
2464	e5.06	2	e.10 Sb	1450		1450	21	1420	93.75	4.11	1000	925	16	1040	16	24	650	
2466	3.77	1.76	.10 Sb	1450		1450	22	1420	93.75	4.11	1000	990	16	1040	16	24	650	
2450	e _{5.92}	2	e.11 Sn	1430		1430	20	1420	93.75	4.11	1000	1005	16	1040	16	24	650	
2453	e _{5.86}	2	e.56 Sn	1430		1430	21	1420	93.75	4.11	1000	1010	16	1040	16	24	650	
3205	6	2	2.0 Sn	1420		1420	34	1420	93.75	4.11	1000	950	16	1040	16	24	650	
2445	e6.08	2	e.03 Si	1500		1430	10	1330	93.75	4.11	1000	985	16	1040	16	24	650	
2446	5.77	1.67	e.03 Si	1500		1430		1330	93.75	4.11	1000	975	16	1040	16	24	650	
3202	6	2	.1 Te	1400		1400	32	1400	93.75	4.11	1000	1020	16	1040	16	24	650	
3203	6	2	.5 Te	1400		1400	29	1400	93.75	4.11	1000	970	16	1040	16	24	650	
2467	e _{5.71}	2	e.09 Tl	1430		1430	23	1420	93.75	4.11	1000	980	16	1040	16	24	650	
2468	5.69	1.76	.44 Tl	.1380		1380	17	1380	93.75	4.11	1000	1000	16	1040	16	24	650	
2444	e _{5.94}	2	.11 Zn	1430		1430	21	1320	93.75 93.75	4.11	1000	1035	16	1040 1040	16 16	24 24	650 650	
3061	6	2 1.81	.5 Zn	1430 1390		1430 1390	40 22	1400 1400	93.75	4.11	1000	990	16	1040	16	24	650	
2449	5.73 5.69	1.69	1.49 Zn .14 Zr	1430		1430	20	1400	93.75	4.11	1000	1005	16	1040	16	24	650	
2478	e6.09	2	e <.01 Be	1630		1420	21	1420	93.75	4.11	1000	1010	16	1040	16	24	650	
2480	e _{5.95}	2	C 7.01 Be	1440		1440	21	1410	93.75	4.11	980	-980	16.	1040	16	24	650	
			e<.05 Cd	13				and You										
2630	e9.50	2	e<.01 Be	1430		1420	- 26	1420	93.75	4.11	980	970	16	1040	16	24	650	
2834	5.78	1.85	<.005 Be			1420	20	1420	93.75	4.11	1000	995	16	1040	16	24	650	
2835	5.73	1.88	<.005 Be			1420	20	1420	93.75	4.11	1000	1010	16	1040	16	24	650	
2836	5.56	1.93	<.005 Be			1420	21	1440	93.75	4.11	1000	1005	16 16	1040 1040	16 16	24 24	650 650	
2837	5.94	1.94 1.61	<.005 Be			1420	20	1420	93.75 93.75	4.11	1000	970	16	1040	16	24	650	
2838 2857	5.93 9.45	1.71	<.005 Be		in scrap	1410	20	1420	93.75	4.11	1000	965	16	1040	16	24	650	
3206	6	2	<.005 Be f.1 Li	1400		1400	28	1400	93.75	4.11	800	970	16	1040	16	24	650	
3207	6	2	fsti	1420		1420	30	1400	93.75	4.11	900	930	16	800	16	24	650	
3208	6	2	f2.0 Li	1400		1400	32	1400	93.75	4.11	800	Burned in	furna					
3209	6	2	f.1 Ca	1400		1400	37	1400	93.75	4.11	1000	975	16	1040	16	24	650	
2863	3.07					1400	26	1410	93.75	4.11	1000	980	16	1040	16	24	650	
2864	3.72	g.61		1450		1450	24	1420	93.75	4.11	1000	995	16	1040	16	24	650	
2865	3.80	g.80	1.34 Zn	1410		1410	20	1420	93.75	4.11	1000	990	16	1040	16	24	650	
2866	3.71	g.74	2.17 Zn	1420		1420	20	1400	93.75	4.11	800	880	16	800	16	24	650	
2977	4.24	g.60	3.11 Zn	1410		1410	23	1410	93.75	4.11	800	995	16	800	16	24	650	
2978	4.48	g.52	5.33 Zn	1430		1430	23	1410	93.75	4.11	1000	990		hot short;				
13210	6	1.9	f.5 Ca	1350		1350	44	1400	93.75	4.11	900	890	16	800	16	24	650	

aFor convenience in tabulation, data for creep rate and tensile properties at 600° F for heats of EM-42 and EM-62 alloys have been included in this table rather than in table 1.

CAll bars tested at room temperature were not stabilized.

d1/2-in.-diam, bars.

Actual analysis

fAdded.

Zirconium.

treatment, and alloy composition on the properties of experimental alloys at 70° and 600° $\rm F^{\,a}$

					Ta Chil					-	-				-
Room-	temperatu	re tensile pr	operties		Tensi	le properties	s at 600° F					Creep rat	e at 600° F		
Elon- gation percent n 2 in.)	Reduction of area (percent)	Yield strength, 0.2-percent offset (psi)	Tensile strength (psi)	Elon- gation (percent in 2 in.)	Reduction of area (percent)		strength osi) 0.2-percent offset	Tensile strength (psi)	Load (psi)	Dura- tion (hr)	Initial defor- mation (percent)	Minimum creep rate (percent/hr)	Final creep rate (percent/hr)	Final total defor- mation (percent)	Contraction or release of load (percent
5.0		20,250	30,500	136.0	91.6	4750	5500	9,750							
3.5		22,750	31,500						1300 J1300	170	0.018	0.013	0.018	2.60	0.01
5.8		24,000	32,750						2500	7.2	.105	4.0	4.0	13.56	
3.7		23,750 22,250	30,500	126.0 120.0	89.5	6500 6500	7000 7000	11,250	1300	402 884	.025	.0005	.00082	.320	.020
4.3		29,500	34,750 27,750						2500	113	.019	.038	.027	9.26	
6.0 5.4		17,500 20,750	29,500	131.0 158.0	93.5 97.3	5500 3750	6250 4500	9,750 7,250	1300	160	.026	.0006	.0006	.182	.018
7.0		19,000	29,250												
6.5 7.0		13,250 19,500	25,000 31,000	175.0	99.0	4000	4E00	7 750	1300	434	.019	.00070	.00110 4.0	.596 13.56	.008
					99.0		4500	7,750	2500	7.2	.105	.028	.040	8.26	
9.2		19,750	29,250	120.0	86.8	6000	6250	10,250	1300	619	.025	.00023	.00023	.209	.027
2.3	3.0 4.9	27,500 28,750	33,000 34,000	135.0 120.8	75.7 88.4	7250 7250	7500 7500	11,250 11,750	1300	159 169	.026	.0015	.0015	.275	.024
1.8	2.5	29,250	34,000	133.8	91.1	7750	8000	12,250	1300	144	.024	.00025	.00025	.084	.030
3.5 4.5	4.1	26,500 27,500	32,750 34,250	111.0 117.5	81.3 88.3	6750 6750	7250 7000	11,500 11,250	1300	145 263	.034	.0012	.0022	.239	.023
6.3	7.0	24,500	32,250	111.5	80.2	6500	6750	11,000	1300	170	.021	.0011	.0011	.243	.018
4.5	4.4	29,500	34,750	103.8	75.5	6250	6500	10,500	1300	166	.023	.0024	.0024	.504	.023
6.0	10.4	24,250 29,500	32,000 34,000	118.8 91.0	85.1 80.9	7250 7250	7750 7500	11,000	1300	170 121	.021	.00030	.00030	.908	.024
3.0	3.7	27,000	32,250	118.5	91.1	6750	7250	11,750	1300	118	.023	.0032	.0032	.438	.023
2.0	1.6 4.7	27,750 27,750	31,750 33,750	164.5 135.0	91.4 87.5	6500 6500	7000 7000	10,750 11,500	1300	123 168	.021	.0030	.0030	.408	.021
4.3	5.4	29,000	35,500	120.0	86.9	5250	5500	10,250							
3.5	4.7 3.3	30,500	35,000 32,500	122.5	74.5	7250 6250	7500	10,500	1300	192 171	.019	.00025	.00025	.116	.017
6.3	6.4	25,250 20,750	29,000	120.0 111.0	77.6 66.4	5500	6500 6000	10,250	1300	146	.023	.0013	.0013	.267	.019
5.0	4.7	25,000	32,000	133.5	83.5	6500	6750	10,750	1300	141	.032	.0044	.0044	.722	.028
7.0	8.1 7.3	21,000 20,250	29,500	149.5 88.0	82.1 67.4	6000 6250	6250 6750	10,000	1300	172	.028	.0010	.0010	.263	.019
5.0	5.1	27,000	33,750	121.8	93,1	6500	7000	10,750	1300	143	.030	.0010	.0010	.208	.024
3.5 6.0	4.7 8.9	27,500 23,500	34,500 32,000	109.0	90.6 85.7	6500 6000	6750 6500	10,750 9,750	1300	166	.028	.0011	.0011	.279	.024
3.0	3.5	26,250	32,750	139.5	91.7	6750	7000	11,500	1300	142	.030	.0024	.0024	.466	.025
3.0	3.9 2.0	25,750 28,250	32,500	144.0 118.5	92.1 84.0	6750 6500	7250 7000	11,500	1300	167 171	.023	.0014	.0014	.314	.022
4.5	*6.4	27,000	33,750	141.3	88.1	7250	7500	11,500 11,750	1300	238	.028	.00027	.00027	.142	.024
4.0	4.9	27,500	34,000	148.5	84.7	6250	6500	10,500	1300 1300	161 144	.036	.0022	.0022	.432	.034
2.0	2.5	31,000	36,750 34,250	105.0 117.0	80.6 92.6	5500 6750	6000 7000	11,250	1300	168	.029	.0034	.0040	.680	.017
1.8	3.1	. 32,750	35,000	160.0	88.9	6500	7250	11,000	1300	166	.026	.0011	.0011	.271	.015
3.3	4.1	29,000 27,250	33,000	140.5 106.8	89.4 84.8	6750 6500	7250 7000	12,250	1300	166	.030	.0025	.0025	.504	.006
2.5	2.7	27,750 23,250	34,500	118.8 126.3	79.9 73.6	6500 5750	7000 6250	11,500	1300 1300	142 169	.030	.0048	.0059 .0032	.856 .580	.017
1.8	2.1	30,000	34,000	105.0	71.6	6750	7250	12,500	1300	120	.028	.0014	.0014	.235	.021
4.3	6.8 5.4	27,500	33,750 36,250	117.5 110.5	92.5 82.6	7000 7000	7500 7250	11,000	1300 1300	172 161	.025	.0003	.0003	.114	.017
3.8	4.3	29,000	35,000	130.5	78.9	6250	6500	10,250	1300	119	.028	.0083	.013	1.25	.02
4.5	6.2	26,750	34,750	115.0	91.9	6750	7000	11,250	1300 1300	145 163	.028	.0049	.0068	.834	.024
4.0 1.3	2.7	29,500 30,250	35,000	118.5 167.5	93.4 95.1	7000 8250	7250 8750	11,250 13,250	1300	241	.049	.00035	.00075	.092	.021
3.8 2.5	4.9	28,000 31,750	33,250 35,250	101.3	89.9	7250 4750	7500 5500	11,500	1300	160	.032	.00035	.00035	.120	.015
5.0	6.8	26,500	33,000	95.0	67.8	7250	7750		1300	241	.028	.00020	.00020	.122	.032
8.0	9.3	24,250	31,750	122.5	72.5	5000	5250	10,750 8,750							
6.3	11.2	29,250	35,250	121.3	86.0	5000	5500	10,500	1900	169	026	0010	0010	250	024
0.0	11.2	24,000	32,250 36,500	117.5 173.8	85.0 98.9	5500 5250	6000 5750	11,000	1300	168	.026	.0019	.0019	.350	.024
8.0	10.0	28,000	36,250	178.8	95.4	4750	5500	7,750							
amaged	during ex	xtrusion 31,000	35,500	173.0		5500	5500	8,000	1300	97	.021	Elongated			



TABLE 2.- EFFECT OF EXTRUSION TEMPERATURE, EXTRUSION REDUCTION, HEAT MAGNESIUM-CERIUM FORGING ALLOYS

				Melting	data					Extrusion	data			Не	eat treatm	ent		Lan y
Hont	Inter	nded compo	osition	Alloying temper-	Time minor element	Alloying temper-	Holding time	Pouring	Deduction	Rate	Ten	perature	So	lution	Aging		ilizing (c)	Grain size avera
Heat	Ce (percent) (b)	Mn (percent)	Other (percent)	ature of minor element (°F)	held at tem- perature (min)	ature of Ce (°F)	after Ce addition (min)	temper- ature (°F)	Reduction (percent)	(in./min of 3/4-in diam. rod)	Ingot (°F)	Container (°F)	Time (hr)	Temper- ature (°F)	time at 400° F (hr)	Time (hr)	Temper- ature (°F)	diam (in.)
3211	6	1.9	f2.0 Ca	1400 1390		1400 1390	26 32	1400 1400	93.75 93.75	4.11 4.11	800 1000	835 1010	16 16	800 1040	16 16	24 24	650 650	0.000
3213	6	1.9	.5 A1	1410		1410	32	1400	93.75	4.11	1000	1010	16	1040	16	24	650	
3214	6	1.9	2.0 A1	1400		1400	24	1400	93,75	4.11	1000	975	16	1040	16	24	650	.00:
3215	6	1.9	.1 Cd	1400		1400	30	1400	93.75	4.11	1000	1005	16	1040	16	24	650	
3216	6	1.9	.5 Cd	1400		1400	32	1400	93.73	4.11	1000	1000	16	1040	16	24	650	
3217 3218	6	1.9	.1 Ag	1410 1410		1410 1410	31 30	1400 1400	93.75 93.75	4.11 4.11	1000	1005 1010	16 16	1040 1040	16 16	24 24	650 650	
3219	6	1.9	.5 Ag 2.0 Ag	1410		1410	24	1400	93.75	4.11	1000	955	16	1040	16	24	650	.00
3225	6	None	None			1430	34	1400	93.75	4.11	1000	1005	16	1040	16	24	650	.00
3226	4	None	None	1400		1400	36	1400	93.75 93.75	4.11	1000	995 970	16	1040	16	24	650	
3227	6	1.9	.75 Zr .50 Zr	1400 Charged	in caren	1400 1400	28 27	1400 1400	93.75	4.11	1000	995	16 16	1040 1040	16 16	24 24	650 650	.000
3229	6	1.9	.25 Zr	Charged		1420	23	1400	93.75	4.11	1000	940	16	1040	16	24	650	
3230	6	1.9	05 7r	Charged	in scrap	1400	38	1400	93.75	4.11	1000	1035	16	1040	16	24	650	
3231	6	1.9	f.1 Na	1400		1400	45	1400	93.75 93.75	4.11	1000	1000 1000	16 16	1040 1040	16 16	24 24	650	
3232	6	1.9	f.5 Na None	1400		1400 1400	26 27	1400 1400	93.75	4.11	1000	1030	16	1040	16	24	650 650	.00
3236	6	1.9	None			1410	34	1400	93.75	4.11	1000	1035	16	1040	16	24	650	
3236x	6	1.9	None			1410	34	1400	93.75	4.11	1000	1035	16	1040	None	24	650	
3237	4	1.9	None	1400	None	1410	31	1400	93.75 93.75	4.11 4.11	1000	1030 1010	16 16	1040 1040	16 16	24 24	650	
3238 3239	6	1.9	f.1 Sr f.5 Sr	1400 1300	None None	1400 1400	31	1400 1380	93.75	4.11	1000	1010	16	1040	16	24	650 650	
3240	6	1.9	f _{1.5} Sr	1380	None	1420	34	1400	93.75	4.11	1000	1000	16	1040	16	24	650	.000
3241	6	1.9	.1 Cd	1400	None	1410	31	1400	93.75	4.11	1000	1010	16	1040	16	24	650	
3242	6	1.9	.001 W	h ₁₇₀₀	120	1400	40	1390	93.75	4.11	1000	1020	16	1040	16	24	650	.00
3243 3244	6 6	1.9	.0005 W	Charged h ₁₇₀₀	120	1410 1400	47 37	1380 1390	93.75 93.75	4.11 4.11	1000	1035	16 16	1040 1040	16 16	24 24	650 650	.000
3245	6	1.9	.0005 Mo			1400	38	1380	93.75	4.11	1000	1040	16	1040	16	24	650	
3246	6	1.8	f.1 K	1375	None	1400	33	1380	93.75	4.11	1000	1000	16	1040	16	24	650	
3247	6 6	1.9	f.5 K 2.0 Ni	1370 h ₁₇₀₀	None 45	1400 1400	31 27	1380 1380	93.75 93.75	4.11	900	990 870	16 16	1040	16 16	24 24	650 650	.000
3249	6	1.9	.5 Ni	Charged		1410	39	1390	93.75	4.11	940	960	16	900	16	24	650	
3250	6	1.9	.1 Ni	Charged		1410	32	1390	93.75	4.11	1000	1020	16	1040	16	24	650	
3251	6	1.9	.01 B	1350	5	1400	29	1400	93.75	4.11	1000	1005	16	1040	16	24	650	.000
3253	6 6	1.9	.004 B	Charged h ₁₇₀₀	in scrap	1420 1400	30	1400 1400	93.75 93.75	4.11 4.11	1000	1020 1010	16 16	1040 1040	16 16	24 24	650 650	.00:
3255	6	1.9	.025 Fe	Charged		1410	32	1400	93.75	4.11	1000	1010	16	1040	16	24	650	
3256	6	1.9	1.0 Si	h ₁₇₀₀	120	1400	38	1380	93.75	4.11	1000	1020	16	1040	16	24	650 •	.002
3257	6	1.9	.5 Si	Charged		1410	30	1400	93.75	4.11	1000	1015	16	1040	16	24	650	
3258	6 6	1.8	.1 Si f.1 Ge	Charged h ₁₇₀₀	in scrap	1400 1430	33 25	1400 1400	93.75 93.75	4.11	1000	1015 990	16 16	1040 1040	16 16	24 24	650 650	.000
3260	6	1.9	.05 Ge	Charged		1400	28	1400	93.75	4.11	1000	1015	16	1040	16	24	650	
3264	6	1.9	.001 V	h1700	75	1420	29	1400	93.75	4.11	1000	1015	16	1040	16	24	650	.000
3265	6	1.9	.0005 V	Charged h ₁₇₀₀		1420 1400	32 40	1400 1410	93.75 93.75	4.11	1000	1015 1020	16 16	1040 1040	16 16	24	650 650	.001
3267	6	1.9	.005 Ti	Charged		1410	26	1410	93.75	4.11	1000	1020	16	1040	16	24	650	.00.
3271	6	1.9	1.0 Co	h1700	165	1400	33	1390	93.75	4.11	1000	1010	16	1040	16	24	650	
3272	6	1.9	.5 Co	Charged		1410	30	1420	93.75	4.11	1000	990 1010	16	1040	16	24	650	
3273	6 6	1.9	.1 Co .001 Cr	Charged h ₁₇₀₀	in scrap	1400 1420	28 36	1420 1390	93.75 93.75	4.11	1000	1010	16 16	1040 1040	16 16	24	650 650	
3275	6	1.9	.0005 Cr	Charged		1400	35	1400	.93.75	4.11	1000	1025	16	1040	16	24	650	
3276-1	6	1.9	None			1410	42	1400	185.9		1000	1030	16	1040	16	24	650	.00:
-2	E:		old-rolled 3		eiore								16	1040	16	24	650	.003
-3	E		heat-treatin old-rolled 5		efore								16	1040	16	24	650	.00
-4			heat-treating	ng									16	1040	16	24	650	.00:
	100		heat-treatin	ng		1410	0-	1460	00.55	4.11	1000	1015	1		1			
3277 3278	6 6	1.9	.5 Sb	1400 1300	15 25	1410 1420	35	1400	93.75 93.75	4.11 4.11	1000	1015 1015	16	1040 1040	16 16	24	650	
3278	6	1.7	1.0 Sb 6.0 Cd	1350	25	1350	35 30	1410 1310	93.75	4.11	1000	985	16	1040	16	24	650 650	
3282	None	1.9	4.0 Cd	1350	33			1410	93.75	4.11	1000	1030	16	1040	16	24	650	
3283	6	1.9	f.1 P	1250		1400	33	1410	93.75	4.11	1000	1015	16	1040	16	24	650	
3284 3285	6	1.9	f.1 As f.5 As	1350 1360		1390 1400	35 34	1420 1410	93.75 93.75	4.11 4.11	1000	1030 1010	16 16	1040 1040	16 16	24 24	650 650	
3287	6	1.9	.001 Cb	h1750	150	1400	33	1410	93.75	4.11	1000	980	16	1040	16	24	650	
3288	6	1.9	.0005 Cb	Charged	in scrap	1410	35	1410	93.75	4.11	1000	970	16	1040	16	24	650	
3289	6	1.9	.001 Ta	h1750		1400	30	1420	93.75	4.11	1000	975	16	1040	16	24	650	
3290	6	1.9	.0005 Ta	Charged	in scrap	1400	32	1430	93.75	4.11	1000	980	16	1040	16	24	650	

aFor convenience in tabulation, data for creep rate and tensile properties at 600° F for heats of EM-42 and EM-62 alloys have been included in this table rather than in table 1.
bRare-earth content.
CAll bars tested at room temperature were not stabilized.
fAdded.
hMinor element added to M alloy in a separate melting operation.
Extruded to 1 by 1 inch, then heat-treated and stabilized.



NACA TN 2097

TREATMENT, AND ALLOY COMPOSITION ON THE PROPERTIES OF EXPERIMENTAL

AT 70° AND 600° Fa - Concluded

Room-to	emperatur	re tensile pro	operties		Tensi	le propertie	s at 600° F			1		Creep.ra	te at 600° F	114	
Elon-	Reduc- tion of	Yield strength, 0.2-percent	Tensile	Elon- gation	Reduc- tion of		strength si)	Tensile strength	Load	Dura-	Initial defor-	Minimum creep rate	Final creep rate	Final total defor-	Contrac tion on release
ercent n 2 in.)	area (percent)	offset (psi)	(psi)	(percent in 2 in.)	area (percent)	0.1-percent offset	0.2-percent offset		(psi)	(hr)	mation (percent)		(percent/hr)	mation (percent)	of load (percent
1.0	2.1	30,250	33,750	140.0	96.8	4250	4750	8,250	1300	40	0.023		10 percent in		
4.0	6.0	32,750	37,000	90.0	92.3	7250	7500	11,500	1300	138 529	.028	.000040	.000040	.083	0.028
3.0	2.9	25,750	30,750	75.0	80.2	6500	7250	11,000	1300	258	.026	.00006	.00006	.045	.032
3.3	5.1	26,750 34,000	33,750	93.8 133.8	63.8 92.7	5000 8000	5500 8500	9,500	1300	117	.028	.00030	.000.30	.061	.028
3.0	3.7	30,750	35,000	120.0	91.4	7750	8250	12,250							
5.7	5.6	27,000	32,250	143.8	79.1	7750	8250	12,000	1300	144	.024	.0005	.0005	.139	.004
3.8	6.5 4.5	31,750 29,500	36,250 35,500	98.8 128.8	90.0	5500 5750	6250 6000	11,250	1300	165 142	.021	.0004	.0004	.130	.020
2.3	3.5	31,500	35,250	100.0	84.9	7500	8000	13,000	1300	142	.024	.0105	.0105	1.66	.02
2.7	3.5	32,750	37,000	93.5	89.2	7500	8000	12,750							
2.6	3.9	30,000	34,750	116.0	90.6	7750	8750	13,000	1300	160 142	.024	.0007	.0007	1.65 .301	.017
2.6	3.7	30,750	35,500 12,750	157.3	90.4	7750 7500	8250 8250	12,250	1300	162	.022	.0020	.0020	.452	.020
.2	.0		10,000	57.5	55.6	8000	8250	11,250	1300	135	.026	.00065	.00065	.165	.026
2.8	5.3	33,750	39,500	104.0	86.8	7500	7500	11.500	1300	191	.024	.00045	.00045	.152	.021
4.0	5.0 3.3 3.1	33,500 33,250	41,000	115.0	91.9 87.2	6750 7750	7250 8000	11,000	1300	261	.024	.00025	.00025	.140	.013
3.0	4.1	33,000 35,750	40,000	91.3	92.3	7500 7500	8000 7750	12,000	1300	169	.024	.0013	.0013	.314	.009
3.0	3.9	33,500	36,500	105.0	83.5	7750	8250	11,250 12,000	1300	185	.028	.00068	.00068	.189	.017
2.9	3.1	31,500	35,750	132.5	93.8	8000	8250	12,500	1300	185	.022	.00020	.00020	.128	.02
1.8	2.7	29,250	32,500	147.5	87.5	7500	8000	12,750	1300	166 142	.033	.0003	.0003	.121	.009
3.2	4.7 2.9	33,250 37,000	37,000 39,250	103.5	84.9	7750 8500	8000 8750	12,500 12,250	1300	207	.024	.00015	.00015	.093	.015
3.5	4.1	29,500	36,250	175.0	96.4	6750	6750	10,500							
2.0	4.1	35,500	38,500	101.3	68.6	6750	7000	10,750							
2.3	2.1	26,000 35,250	31,500	115.0 114.3	63.6 94.6	6750 7250	7250 7500	10,750							
1.3	1.1	33,500	34,750	106.3	61.7	6500	6750	10,750							
3.0	3.9	35,000	37,500	150.0	96.2	9000	9500	14,500	1300	149	.026	.0026	.0026	.672	.019
3.3	1.9	38,000 38,000	39,750 39,250	142.5 141.3	97.1 90.7	9000 9250	9750 9500	14,500 14,500	1300	120 190	.026	.024	.031	3.23	.03
2.8	3.1	34,000	37,000	137.3	95.5	7000	7250	11,250							
2.8	1.9	29,750	35,000	131.3	92.0	7000	7250	10,750							
3.0	1.9	34,500 34,500	36,750 37,250	128.8 132.5	93.2 93.4	6750 6500	7250 7000	11,250 10,750							
9.0	9.8	21,000	29,250	96.3	61.8	5000	5500	8,250							
4.5	5.3	29,750	34,500	78.8	63.9	6000	6500	8,250 9,750							
3.0 2.5	3.9 2.5	32,750 35,000	35,250 36,750	130.0 127.5	85.3 79.3	7000 6500	7500 7000	12,250							
3.3	3.3	35,500	38,000	128.5	94.6	7000	7250	11,000 10,750							
2.0	2.5	31,750	35,500	116.5	88.8	7250	7500	12,000							
2.5	6.4	31,250 29,500	37,000 36,750	106.5	86.9 75.2	7250 6750	7750	11,750							
1.8	2.3	32,250	35,750	110.0	89.3	7000	7500 7500	11,500 11,500							
3.3	5.3	33,000	36,000	171.5		7000	7500	12,250	1300	138	.025	.00072	.00072	.156	.025
3.8	5.2	31,750	34,500	120.0		7250	7750	12,500	1300	139 146	.030	.00066	.00066	.209	.012
3.5	3.1	34,750 33,750	36,000 34,500	119.0		7500 7500	8250 8250	12,500 11,500	1300	143	.026	.00030	.00030	.117	.019
2.8	3.7	33,750	34,250	126.0		7000	7500	11,500	1300	143	.024	.00170	.00170	.312	.030
				75.0		6250	7000		1300	167 143	.021	.00010	.00010	.067	.029
						47 8		11,000		88	.027			.106	Controlle
				72.0		5000	6500 5750	11,000	1300	138	.025	.0011	.0011	.081	failed .025
4.0	4.3	29,750	34,000	72.5		4500		9,750	1300	130	.025	.0002	.0002	.001	.045
4.8	5.0	34,000	35,250	115.0		6250	4750 6500	6,000					******		
5.3	5.3	31,000	34,500	106.0		5750	6250	9,250							
5.0	5.3	23,750	31,500	64.5		5000	5500	8,250							
6.8 5.0	8.3	32,000	34,500	91.0		5750 6250	6250 6750	10,250							
2.8	3.7	29,000	32,750	157.5		6250	7000	11,250							
3.8	4.7	34,000	36,500	91.0		7000	7500	11,250	1300	163	.028	.0043	.0066	1.01	.02
2.3	3.3 5.3	34,000	36,500 36,250	115.0		7250 7000	7500 7250	11,250 11,250	1300	142 141	.026	.0052	.0052	.788	.034
2.0	3.2	35,500 35,750	35,750	78.5		7000	7500	10,750	1300	144	.025	.0029	.00097	.220	.017



TABLE 3.- CHEMICAL COATINGS FOR HIGH-TEMPERATURE

PROTECTION OF MAGNESIUM—CERIUM ALLOYS

Sample	Coating	Application of coating and composition	Time and temperature
A3286A,B	Uncoated		
A3286C,D	Dow No. 9	A. Alkaline degreasing bath: Trisodium phosphate 240 g Sodium carbonate 240 g Distilled water 8.0 liters Cold-water rinse	Bath temperature of 180°-240° F for 5-15 min, using current density of 15 amp/sq ft
		B. Hydrofluoric-acid bath: Hydrofluoric acid 328 ml Distilled water 1.0 liter Cold-water rinse	Bath at room temperature, immersed 5 min
		C. Galvanic anodizing bath: Ammonium sulphate 180 g Sodium dichromate 180 g Ammonium hydroxide 17 ml Distilled water 6.0 liters Cold-water rinse, hot-water dip, dried in an air blast	Bath temperature of 120°-140° F, using current density of 6 amp/sq ft for 30 min
A3286E,F	Dow No. 10	A. Alkaline degreasing bath: Same as Dow No. 9	
		B. Chromium-nitric pickle: Sodium dichromate 85 g Nitric acid 59 ml Distilled water 1.0 liters Cold-water rinse, hot-water dip	Bath at room temperature, immersed 1 min
		C. Dichromate bath: Sodium dichromate 480 g Calcium fluoride 15 g Distilled water 4.0 liters Cold-water rinse, hot-water dip, dried in an air blast	Bath boiling, immersed 30 min
А3286С,Н	Treatment R	A. Alkaline degreasing bath: Same as Dow No. 9	
		B. Anodic bath: Sodium hydroxide, 5 percent 421.5 g Distilled water 8.0 liters Cold-water rinse	Bath temperature of 140°-160° F, using current density of 15 amp/sq ft for 30 min
		C. Seal treatment: Sodium chromate 277 g Distilled water 1.0 liter Cold-water rinse, hot-water dip, dried in an air blast	Bath temperature of 170°-180° F immersed 30 min
А3286І,Ј	Selenium treatment	A. Apply Dow treatment No. 10 complete	
		B. Selenium coating: Selenium dioxide 128 g Sodium hydroxide 92.6 g Distilled water 1.0 liter Cold-water rinse, hot-water dip, dried in an air blast	Bath temperature of 195°-212° F, immersed 40-90 min at pH 7.1 to 7.5, controlled by adding S _e O ₂
A3286K,L	Fluosilicic— acid treatment	A. Alkaline degreasing bath: Same as Dow No. 9	A partial for
		B. Fluosilicic—acid treatment: Hydrofluosilicic acid 250 ml Distilled water 750 ml Titanium potassium oxalate added to the saturation point Cold—water rinse, hot—water dip,	Bath at room temperature, immersed 10 min

Table 4.— surface oxidation of experimental alloys and coated and uncoated cast alloys after exposure at 700° f for 900 hours

1		B 113 B	Sw	rface
Heat	Intended composition	Coating	Visual observation of scale on surface	Microscopic observation, depth of penetration (mm)
A3235	CM-62	None	Very bad	
A3236	CM-62	None	Very bad	
A3237	CM-42	None	Bad	
A2316	EM-62 + 0.01B	None	Very bad	
A3059	EM-62 + 0.1Ba	None	Bad	
A3060	EM-62 + 0.5Ba	None	Very bad	The second second
A2455	EM-62 + 0.1Bi	None	Bad	
A3201	EM-62 + 0.5Bi	None	Bad	The second of the second
A2457	EM-62 + 0.9Bi	None	Some scale	
A2473	EM-62 + 1.9Cd	None	Very bad	
A3204	EM-62 + 1.5Cd	None	Very bad	
A2452	EM-62 + 2.4Cd	None	Worst	
A2451 A2448	EM-62 + 0.1Cu EM-62 + 0.6Cu	None None	Bad Vonr. bad	
A2446 A2458	EM-62 + 0.60u EM-62 + 0.1In	None	Very bad	
A2459	EM-62 + 0.6In	None	Very bad Bad	
A3058	EM-62 + 0.1Pb	None	Very bad	
A2461	EM-62 + 0.5Pb	None	Very bad	
A2462	EM-62 + 2.1Pb	None	Worst	
A2463	EM-62 + 0.05Sb	None	Very bad	
A2464	EM-62 + 0.1Sb	None	Very bad	
A2466	EM-62 + 0.4Sb	None	Very bad	
A2450	EM-62 + 0.1Sn	None	Very bad	
A2453	EM-62 + 0.6Sn	None	Very bad	
A3205	EM-62 + 2.0Sn	None	Some scale	
A2445	EM-62 + 0.3Si	None	Very bad	
A2446 A3202	EM-62 + 0.3Si Em-62 + 0.1Te	None None	Very bad	
A3203	EM-62 + 0.5Te	None	Bad Warm had	
A2467	EM-62 + 0.1T1	None	Very bad Very bad	
A2468	EM-62 + 0.4Tl	None	Very bad	
A2444	EM-62 + 0.1Zn	None	Worst	
A3061	EM-62 + 0.5Zn	None	Bad	
A2449	EM-62 + 1.5Zn	None	Bad	
A2479	EM-62 + 0.1Zr	None	Very bad	
A2478	EM-62 + < 0.01Be	None	Very bad	
A2480	EM-62 + <0.01Be	None	Very bad	
10/00	<0.050d	None		THE REPORT OF THE PARTY OF THE
A2630	EM-102 + < 0.01Be	None	Bad	
A2834 A2835	EM-62 + < 0.005Be EM-62 + < 0.005Be	None	Very bad	
A2836	EM-62 + <0.005Be	None None	Very bad Bad	
A2837	EM-62 + <0.005Be	None	Bad	
A2838	EM-62 + <0.005Be	None	Bad	
A2857	EM-62 + <0.005Be	None	Very bad	
A3206	EM-62 + 0.1Li	None	Very bad	
A3207	EM-62 + 0.5Li	None	Worst	
A3209	EM-62 + 0.1Ca	None	Bad	
A2863	3Ce	None	No scale	0.35
A2864	3Ce,0.6Zr	None	No scale	.25 to .85
A2865	3Ce,0.8Zr,1Zn	None	No scale	.40
A2866	3Ce,0.7Zr,2Zn	None	Some scale	
A2977	3Ce,0.6Zr,3Zn	None	No scale	.80
A3210	EM-62 + 0.5Ca	None	Very bad	
A3211	EM-62 + 2.0Ca	None	Very bad	
A3212	EM-62 + 0.1A1	None	Bad	
A3213	EM-62 + 0.5Al EM-62 + 2.0Al	None	No scale	.40
A3214	EN-OZ + Z. VAL	None	No scale	•50

TABLE 4.— SURFACE OXIDATION OF EXPERIMENTAL ALLOYS AND COATED AND UNCOATED

CAST ALLOYS AFTER EXPOSURE AT 700° F FOR 900 HOURS — Concluded

			Su	rface
Heat	Intended composition	Coating	Visual observation of scale on surface	Microscopic observation depth of penetration (mm)
A3215	EM-62 + 0.1Cd	None	Bad	
	EM-62 + 0.5Cd	None		
A3216			Very bad Worst	
A3217	EM-62 + 0.1Ag	None None	Worst	
A3218	EM-62 + 0.5Ag			
A3219	EM-62 + 2.0Ag	None	Worst	A 20
A3225	6Ce	None	No scale	0.30
A3226	4Ce	None	No scale	•35
A3227	EM-62 + 0.75Zr	None	Bad	
A3228	EM-62 + 0.50Zr	None	Bad	
A3229	EM-62 + 0.25Zr	None	Very bad	
A3230	EM-62 + 0.05Zr	None	Very bad	
A3231	EM-62 + 0.1Na	None	Very bad	
A3232	EM-62 + 0.5Na	None	Very bad	
A3238	EM-62 + 0.1Sr	None	Very bad	
A3239	EM-62 + 0.5Sr	None	Very bad	
A3240	EM-62 + 1.5Sr	None	Bad	
A3241	EM-62 + 0.1Cd	None	Very bad	
A3242	EM-62 + 0.001W	None	Very bad	
A3243	EM-62 + 0.0005W	None	Very bad	
A3244	EM-62 + 0.001Mo	None	Very bad	
A3245	EM-62 + 0.0005Mo	None	Bad	
A3246	EM-62 + 0.1K	None	Bad	
A3247	EM-62 + 0.5K	None	Bad	
A3248	EM-62 + 2.0Ni	None	Very bad	
A3249	EM-62 + 0.5Ni	None	Very bad	
A3250	EM-62 + 0.1Ni	None	Very bad	
A3251	EM-62 + 0.01B	None	Bad	
A3253	EM-62 + 0.004B	None	Bad	
A3254	EM-62 + 0.05Fe	None	Very bad	
A3255	EM-62 + 0.025Fe	None	Very bad	
A3256	EM-62 + 1.0Si	None	No scale	.25
A3257	EM-62 + 0.5Si	None	Bad	
A3258	EM-62 + 0.1Si	None	Very bad	
A3259	EM-62 + 0.1Ge	None	Bad	
A3260	EM-62 + 0.05Ge	None	Very bad	
A3264	HM-62 + 0.001V	None	Very bad	
A3265	EM-62 + 0.0005V	None	Very bad	
A3266	EM-62 + 0.01Ti	None	Very bad	
A3267	EM-62 + 0.005T1	None	Very bad	
A3271	EM-62 + 1.0Co	None	Bad	
A3272	EM-62 + 0.5Co	None	Very bad	
A3273	EM-62 + 0.1Co	None	Bad	
A3274	EM-62 + 0.001Cr	None	Bad	
A3275	EM-62 + 0.0005Cr	None	Bad	
A3286A	EM-62 Cast	None	No scale	
В	EM-62 Cast	None	No scale	•55
C	EM-62 Cast	Dow No. 9	No scale	•5
D	EM-62 Cast	Dow No. 9	No scale	
E	EM-62 Cast	Dow No. 10	No scale	•4
F	EM-62 Cast	Dow No. 10	No scale	
G	EM-62 Cast	Treatment R	No scale	1
H	EM-62 Cast	Treatment R	No scale	•55
I	EM-62 Cast }	Selenium	Some powder	•45
J	EM-62 Cast	treatment	on surface	2
K	EM-62 Cast]	Fluosilicic-	No scale	Deepest spot 0.2 mm;
L	EM-62 Cast	acid treatment	No scale	generally no penetration

TABLE 5.— AVERAGE CREEP RATE, PERCENT PER HOUR DURING 48 TO 96 HOURS OF TEST, OF 32S-T AND 18S-T ALUMINUM ALLOYS AND WROUGHT MAGNESIUM-CERIUM ALLOYS, EM-62 AND EM-42

					Testing ter	mperature			
			300° F	1	.00° F		500° F		600° F
Alloy	Condition	Load (psi)	Creep rate (percent/hr)	Load (psi)	Creep rate (percent/hr)	Load (psi)	Creep rate (percent/hr)	Load (psi)	Oreep rate (percent/hr)
EM-62	As—extruded	12,000 15,000 18,000	0.00004 .00008 .00014						
EM-62	Extruded and stabilized	12,000 15,000 18,000	.00012	4,000 6,000 8,000	0.00024 .00078 .00340	2200	0.0039		
EM-42	As—forged	15,000 18,000 23,000	.00008 .00010 .00038						
18s-T	Rod; medium grain; not stabilized			15,000 20,000 25,000	.00087 .00209 Failed, 72 hr			2,000	0.00114 .00139 Failed, 360 hr
a ₁₈₅ -T	Not stabilized			2,500 5,000 7,500 12,500	.00008 .00010 .00014 .00050			1,300 1,900 2,500 3,400	.00043 .00104 .00166 .01880
32 s —I	Forged; not stabilized					5000 7500	.0030 Discon- tinued, 10 hr	1,300 2,500	b.0004 Discon- tinued, 29 hrb
32S-T	Rod; not stabilized							2,000	.0002 Discon- tinued, 7 hr
a328-T	Not stabilized			5,000 7,500 10,000 15,000	.00014 .00026 .00040 .00064			1,300 2,500	.00004 .00028

^aFrom reference 5; other data from reference 3. bBars stabilized 83 days before testing.

TABLE 6.- COMPARISON OF CREEP PROPERTIES OF CAST AND WROUGHT 142 AND EM-62 ALLOYS [Data taken from references 3 and 4]

		Cas	tings	Wro	ught	
Condition	Load (psi)	Total defor- mation (percent)	Creep rate (percent/hr)	Total defor- mation (percent)	Average creep rate (percent/hr) (50-150 hr)	Test conditions
		142	2 - heat-treat	ed and aged		
Not stabilized	1,300	0.002	0.0005(?)			100 hr at 600° F
Not stabilized	2,000	.07		a0.06		100 hr at 600° F
Not stabilized	2,500	.11	.025	.13	0.085	100 hr at 600° F
Stabilized	2,500		•05.			100 hr at 600° F
Not stabilized	3,000	.16		a.3		100 hr at 600° F
Not stabilized	5,000	•45	•49	Failed		100 hr at 600° F
Not stabilized	10,000	.015	•005			100 hr at 400° F
Not stabilized	15,000	.025	•012	. 034	•015	100 hr at 400° F
Not stabilized	20,000	.065	•034	•15	•123	100 hr at 400° F
	The sec		EM-62			
Stabilized ^b	2,200	0.076	0.00036			100 hr at 600° F
Stabilizedb	3,000	.188	•00126			100 hr at 600° F
Stabilized ^b	4,000	1.630	.01800			100 hr at 600° F
Stabilized ^b	2,200			a _{0.40}	a0.0039	100 hr at 500° F
Stabilizedb	3,000	·		⁸ 2.70		100 hr at 500° F
Stabilized ^b	4,000	.076	•00050			100 hr at 500° F
Stabilized ^b	6,000	• 454	•00330			100 hr at 500° F

^aEstimated by interpolation. ^bExtruded alloys recrystallized 16 hr at 850° F. Cast alloys held 16 hr at 600° F prior to test. NACA

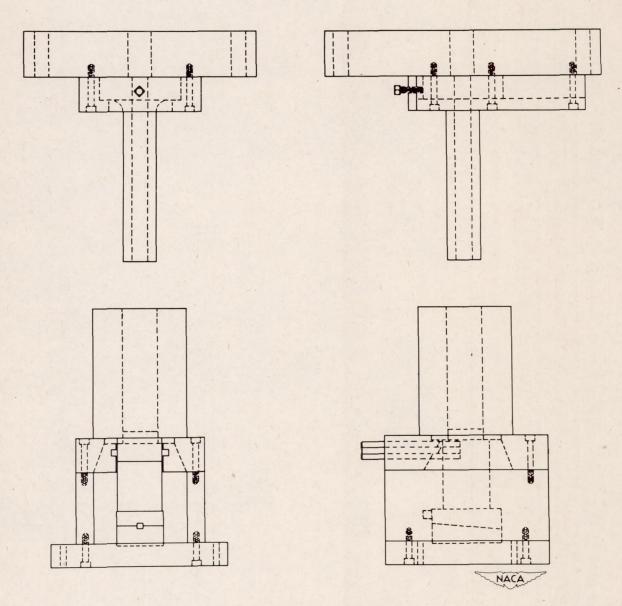
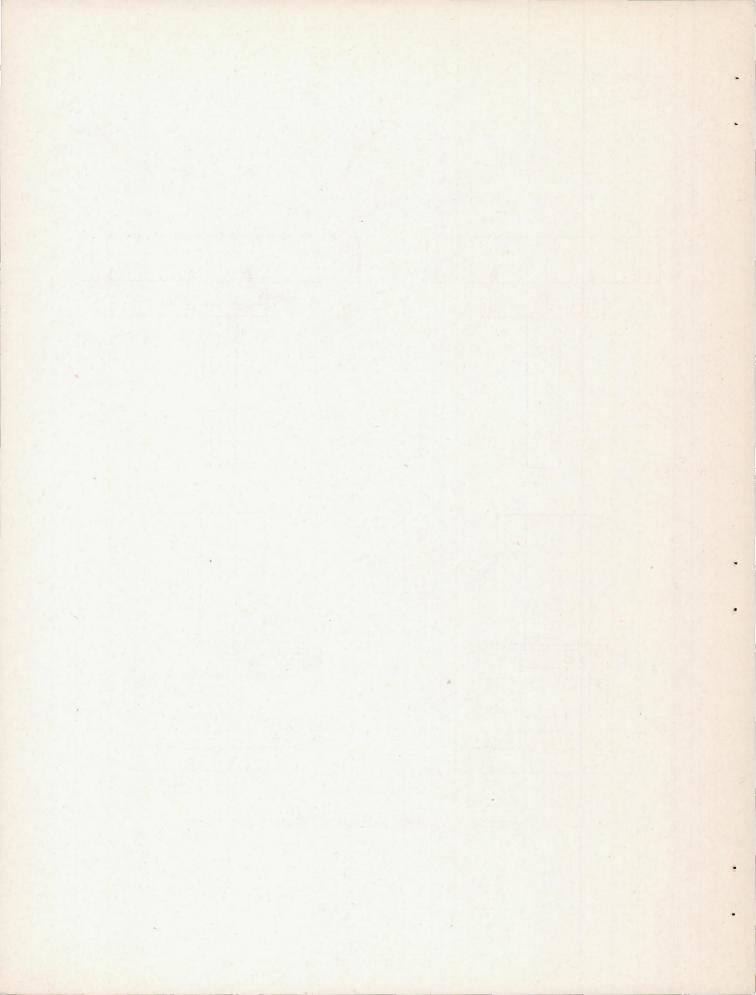


Figure 1.- Assembly drawing of extrusion tools.



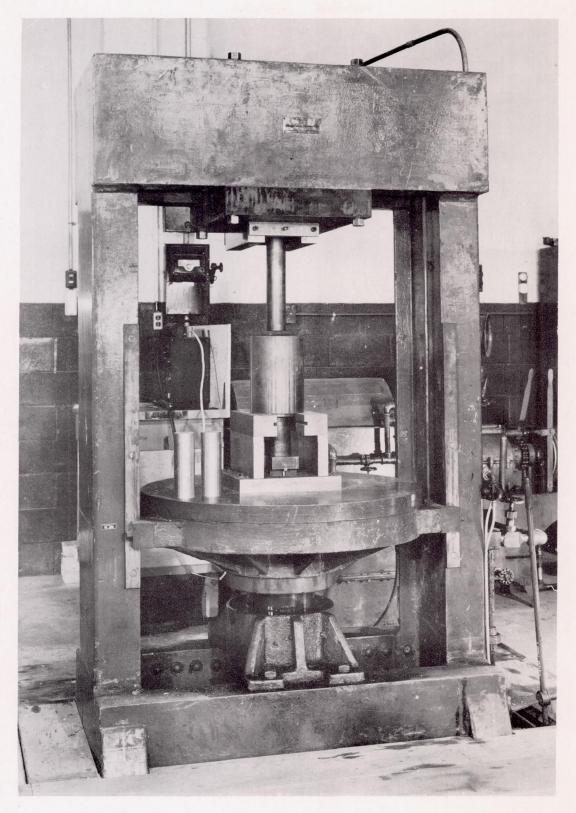
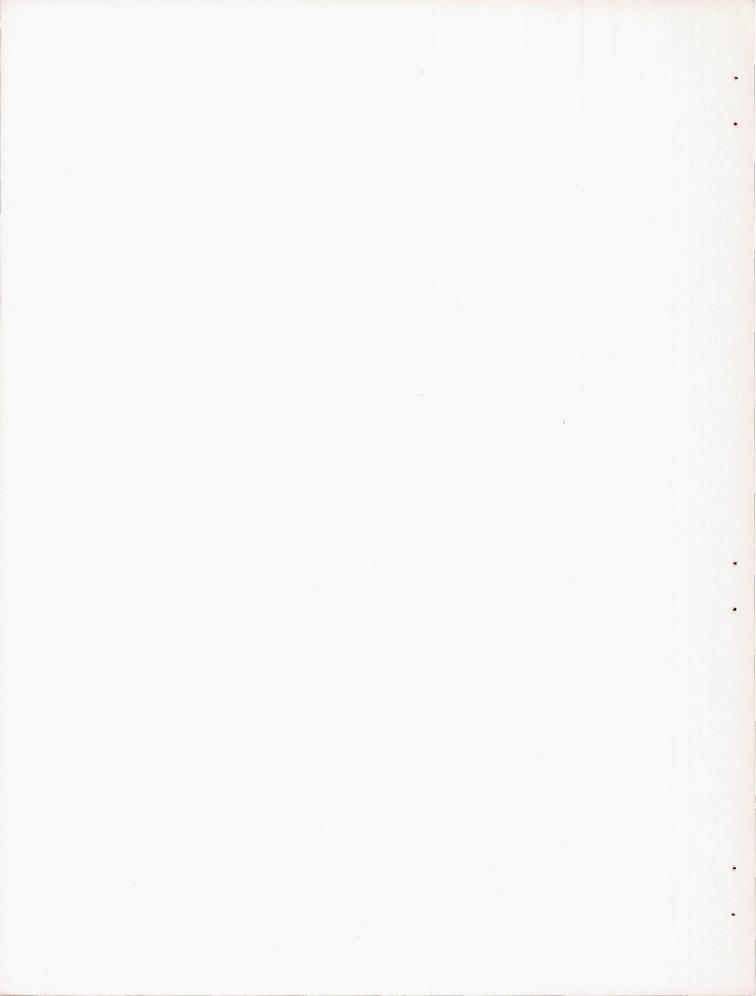


Figure 2.- General view of indirect extrusion press. NACA



33

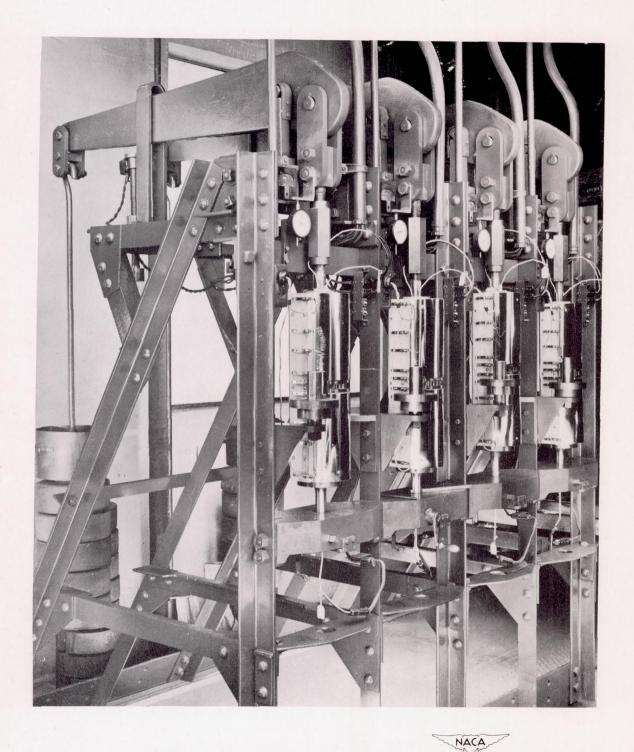


Figure 3.- Battelle creep and stress-rupture units.

				•
				•
				•
				*
			record burn to the last	

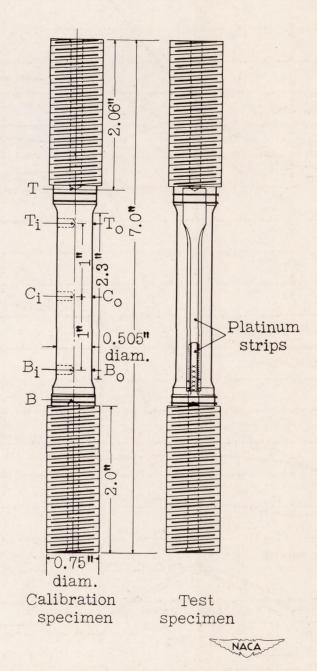


Figure 4.- Calibration and test specimens. Thermocouple positions denoted by T, B, T_i , C_i , B_i , T_o , C_o , and B_o .

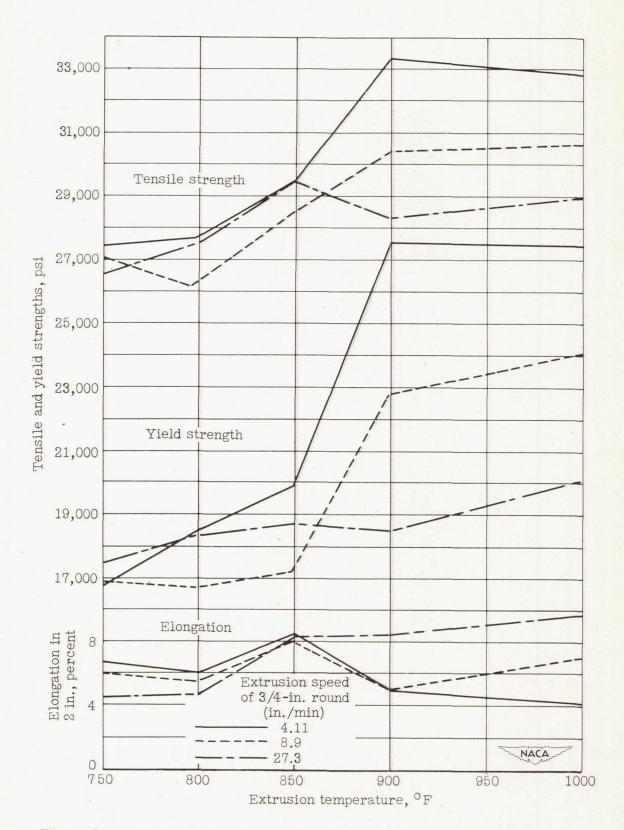
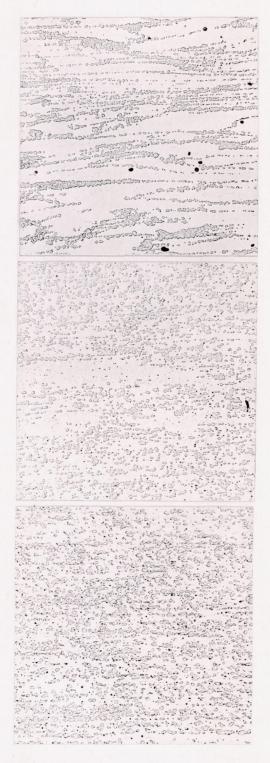


Figure 5.- Effect of extrusion speed and temperature on room-temperature tensile properties.

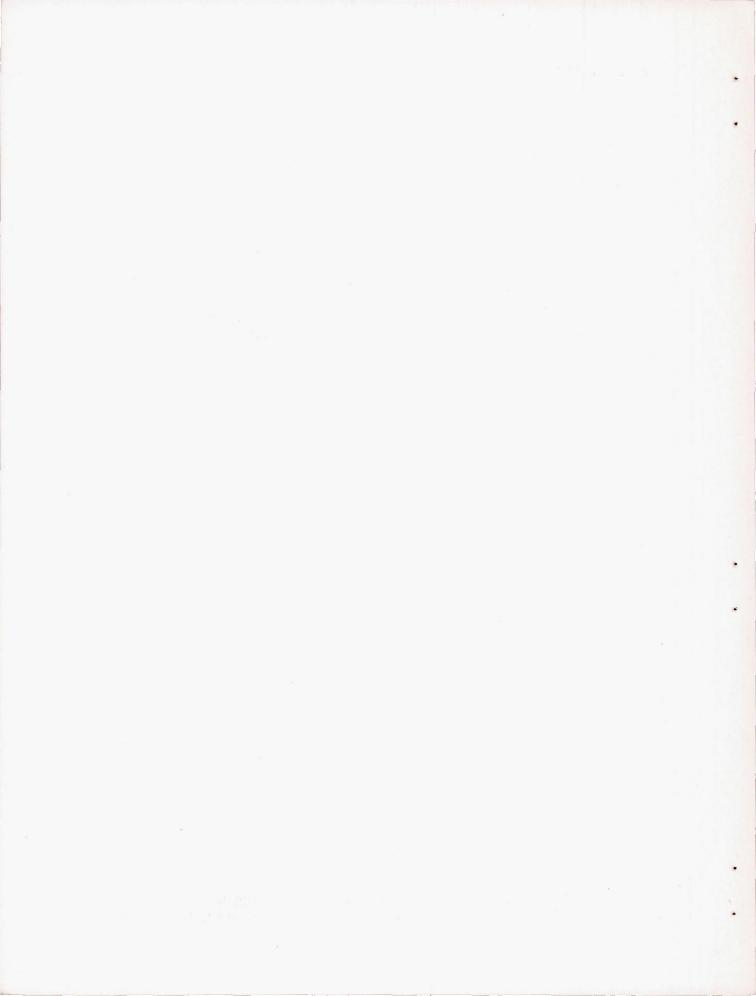


(a) Heat A2465; center of a longitudinal section from a 1- by 1-inch extrusion; reduction 7:1, 85.9 percent.

(b) Heat A2983; center of a longitudinal section from a 3/4-inch-diameter extrusion; reduction 16:1, 93.75 percent.

(c) Heat A2986; center of a longitudinal section from a 1/2-inch-diameter extrusion; reduction 37:1, 97.22 percent.

Figure 6.- Microstructure of EM-62 magnesium alloy after extrusion reductions of 85.9, 93.75, and 97.22 percent. Unetched, as-extruded. 100X.



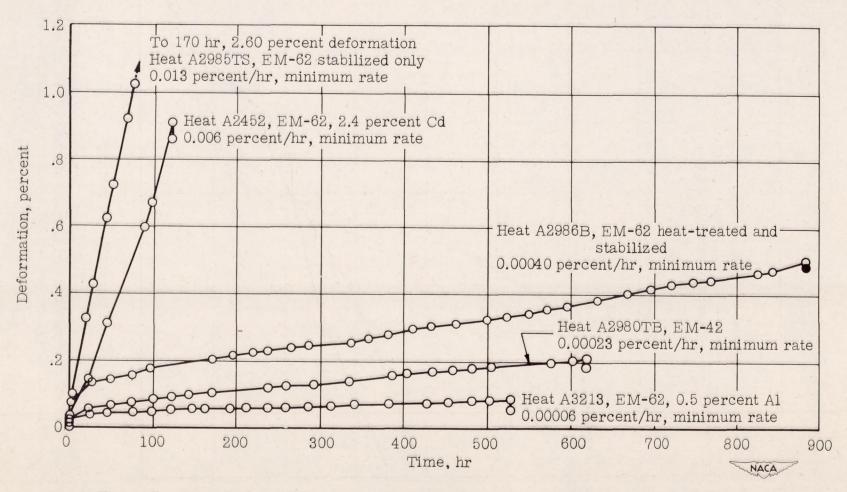


Figure 7.- Time-deformation curves of magnesium-cerium extruded alloys at 600° F and a 1300-psi stress.

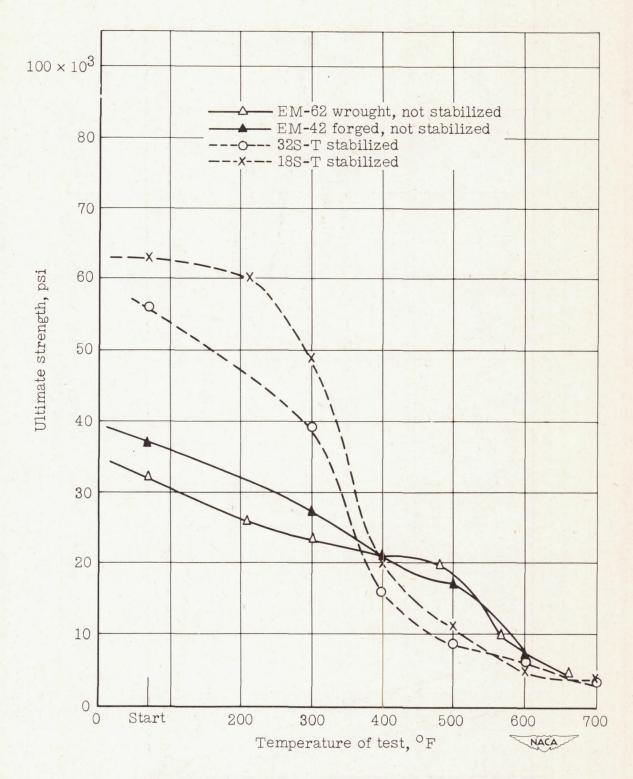


Figure 8.- Effect of temperature on ultimate strength of two magnesium-cerium alloys (solid lines) and two aluminum alloys (dashed lines).

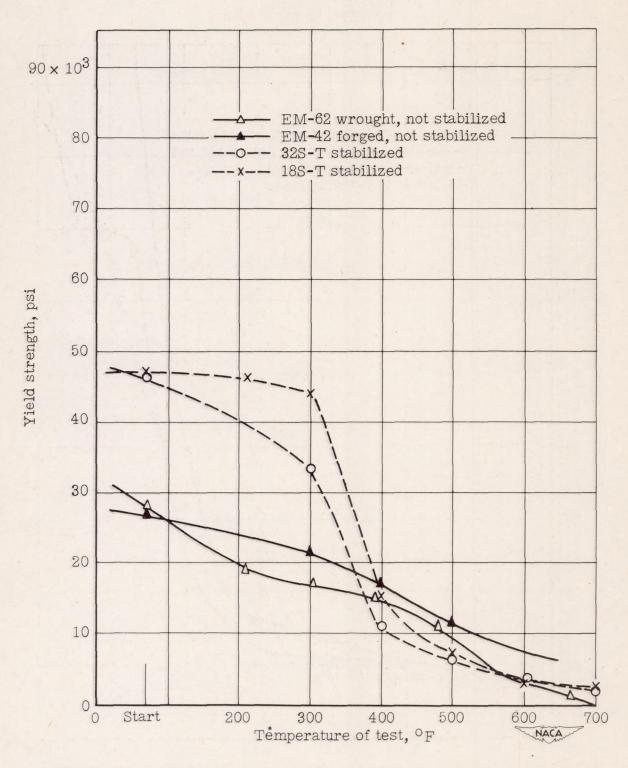


Figure 9.- Effect of temperature on yield strength of two magnesium-cerium alloys (solid lines) and two aluminum alloys (dashed lines).

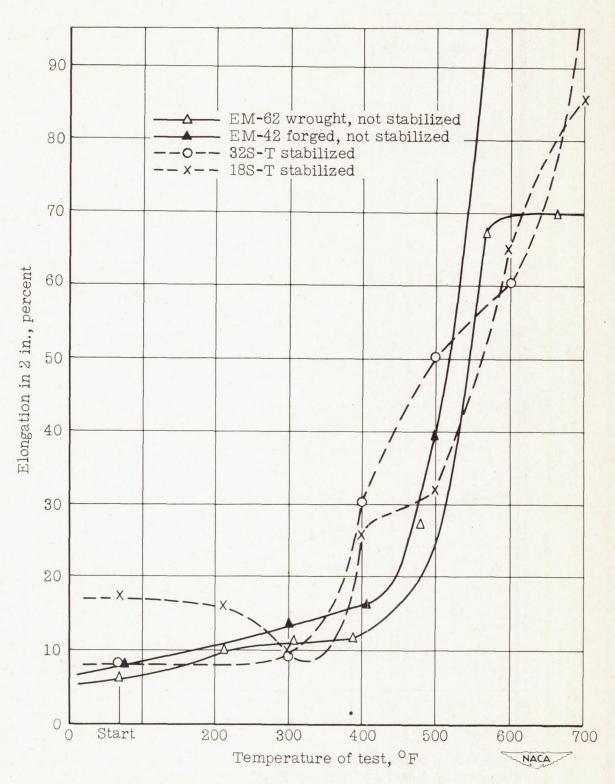


Figure 10.- Effect of temperature on elongation of two magnesium-cerium alloys (solid lines) and two aluminum alloys (dashed lines).