

ULTRA-HIGH TEMPERATURE CERAMIC COMPOSITES FOR LEADING EDGES

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ABSTRACT

Ultra-high temperature ceramics (UHTC) have performed unreliably due to material flaws and attachment design. These deficiencies are brought to the fore by the low fracture toughness and thermal shock resistance of UHTC. If these deficiencies are overcome, we are still faced with poor oxidation resistance as a limitation on UHTC applicability to reusable launch vehicles. We have been addressing the deficiencies of UHTC for the past two years via a small task at GRC that is in the Airframe part of the Next Generation Launch Technology Program. Our focus is on composite constructions and functional grading to address the mechanical issues and on composition modification to address the oxidation issue. The progress on approaches to improving oxidation resistance by alloying and functional grading will be reported. In particular, initial tests of tantalum additions have shown potential for major improvement. Results for additional tests at higher temperatures will be presented. These oxidation improvements are being incorporated in the composites approaches. Two fabrication approaches are being pursued to produce carbon fiber reinforced UHTC composites: prepregging and rigid perform infiltration. Fabrication procedures, microstructures, and initial mechanical property and oxidation results for composites will be reported.



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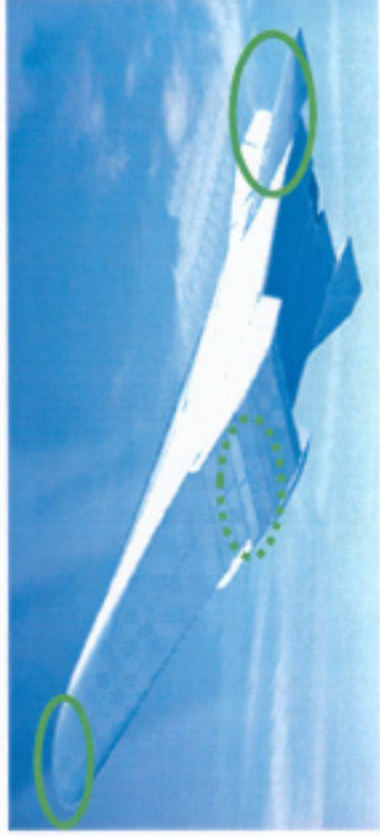
27th JANNAF Airbreathing Propulsion Subcommittee Meeting
December 1-5, 2003, Colorado Springs, CO

Ultra-High Temperature Ceramic Composites for Leading Edges

Outline

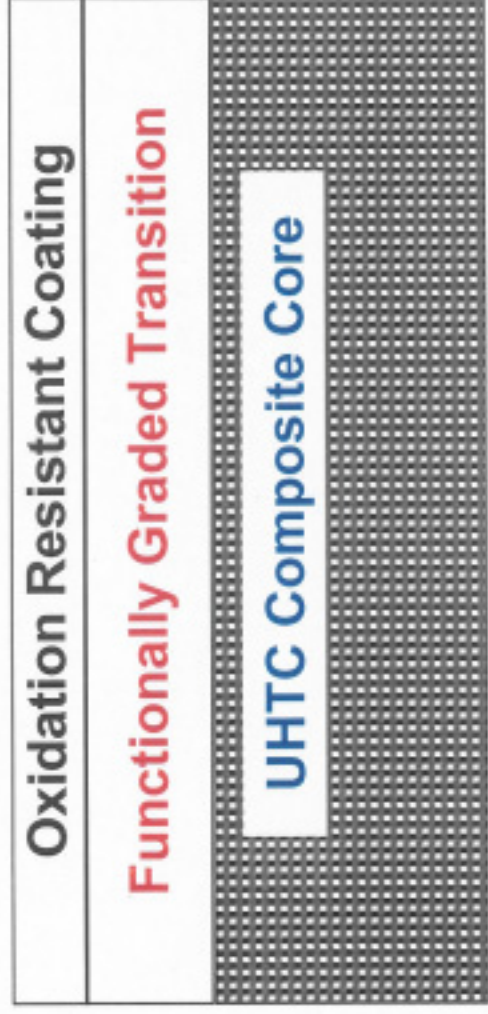
- Background
- Objectives
- Ta Modification of ZrB_2 - SiC
 - Furnace oxidation
 - Arc jet test
- UHTCC
 - Processing
 - Characterization of Starfire UHTCC
- Concluding Remarks

Ultra High Temperature Ceramic Composite (UHTCC) Leading Edge



Key Issues

- Thermal stress resistance
- Oxidation resistance
- Temperature capability
- Architecture optimization



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Mechanisms for ZrO₂ Protective Scale Enhancement via Ta₂O₅ Additions

- Ta₂O₅ as a glass modifier
 - Talmy et al
- Ta⁺⁵ as a dopant in ZrO₂ lattice
 - Fill O⁻² vacancies
 - Decrease O⁻² transport
- Ta₂O₅ as a major oxide scale constituent
 - Phase V
 - Low melting point

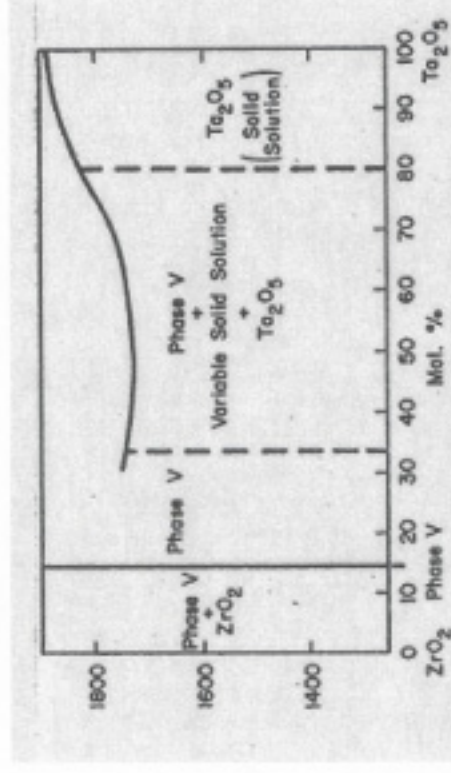


FIG. 374.—System Ta₂O₅-ZrO₂.

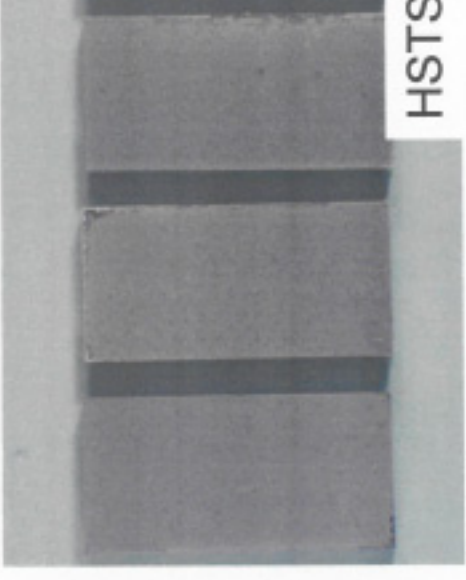
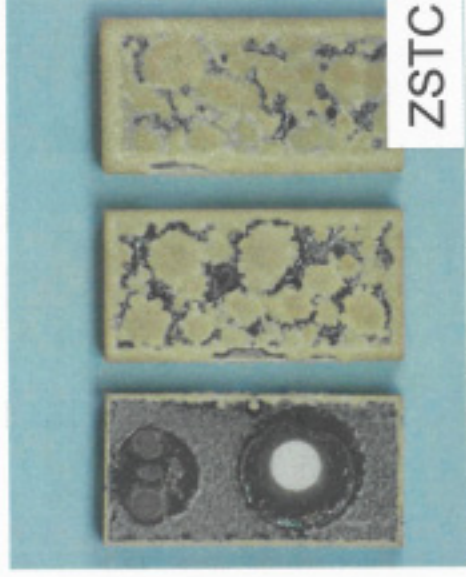
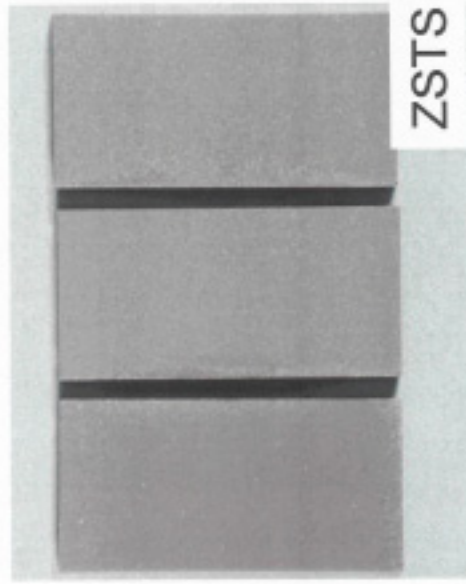
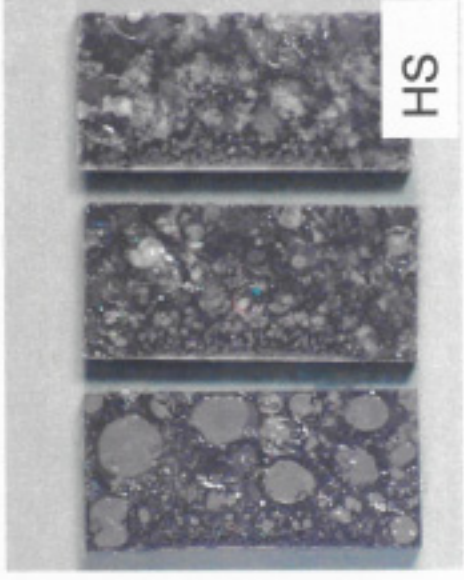
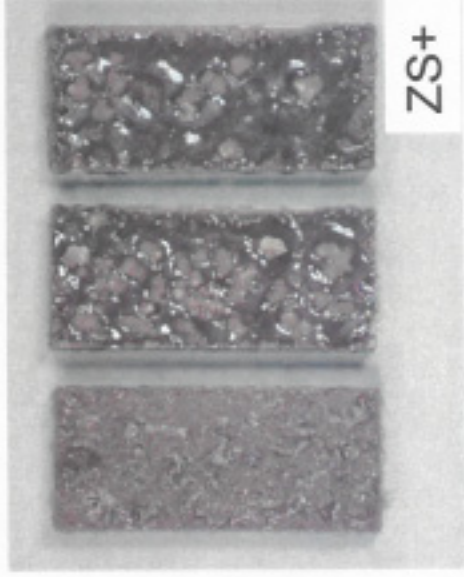
B. W. King, John Schultz, E. A. Durbin, and W. H. Duckworth, U. S. Atomic Energy Comm., BMI-1106, 15 (1956).

Materials Studied

composition	designation
ZrB ₂ 20v/o SiC	ZS
HfB ₂ 20v/o SiC	HS
ZrB ₂ 20v/o SiC 20v/o TaSi ₂	ZSTS
ZrB ₂ 33v/o SiC	ZS+
ZrB ₂ 20v/o SiC 20v/o TaC	ZSTC
HfB ₂ 20v/o SiC 20v/o TaSi ₂	HSTS

- powders ball milled, hot pressed ~ 2000°C, 10 ksi, 2h, vacuum
- 2.54 x 1.2 x 0.3 cm coupons machined from hot pressed plates
- coupons cleaned, weighed, surface area determined

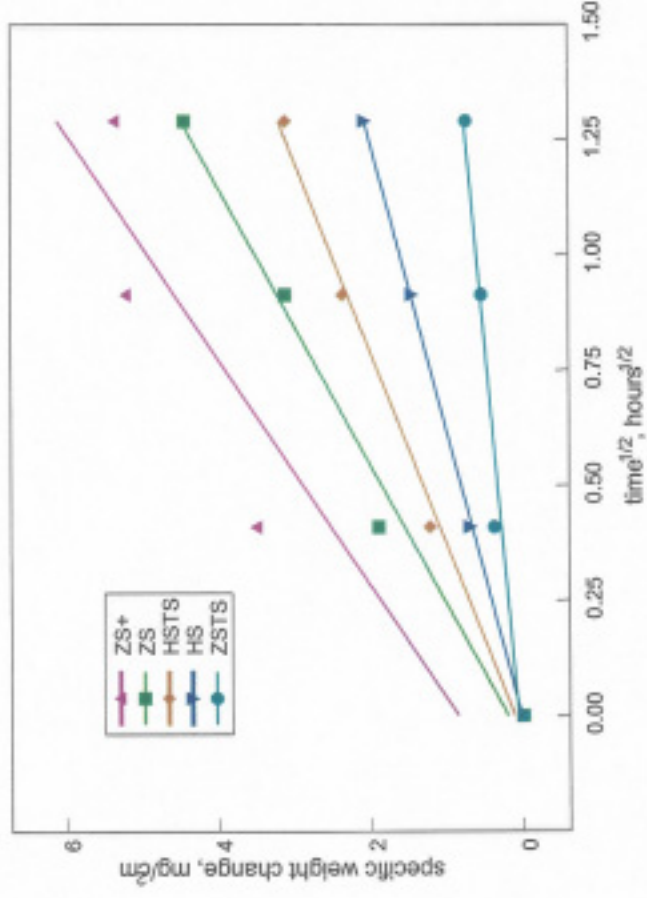
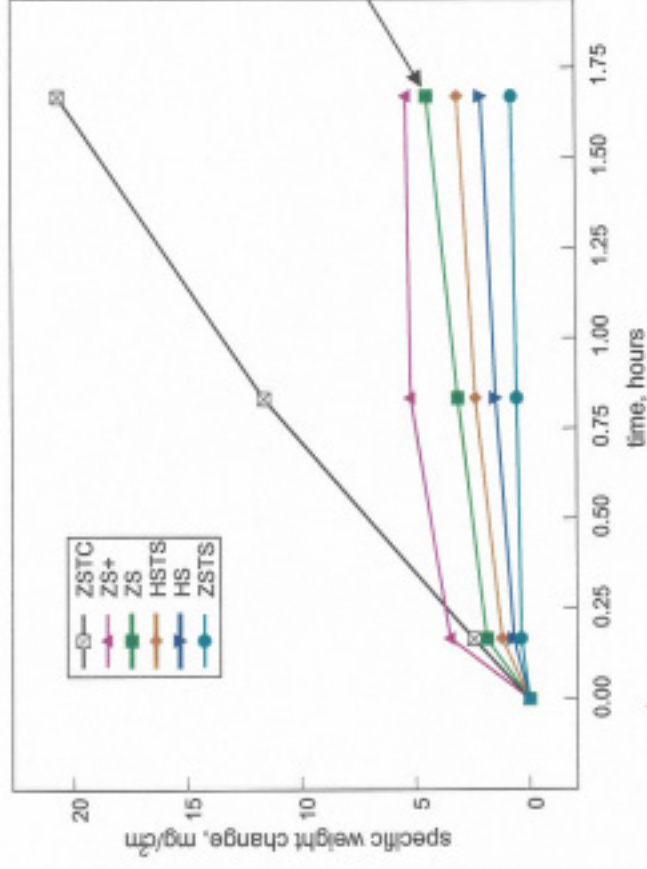
UHTC Coupon Appearance After Oxidation 1627°C, Stagnant Air



Left to right: 1, 5, and 10 ten-minute cycles

Ta-containing compositions show less glass formation

Weight Change of UHTC After Oxidation 1627°C, Stagnant Air



- Poor results for ZS+ relative to ZSTS show that Ta-addition, not additional Si, is responsible for improved oxidation resistance of ZSTS.
- TaC additions do not improve oxidation resistance at 1627°C.
- TaSi₂ additions do not improve HS.

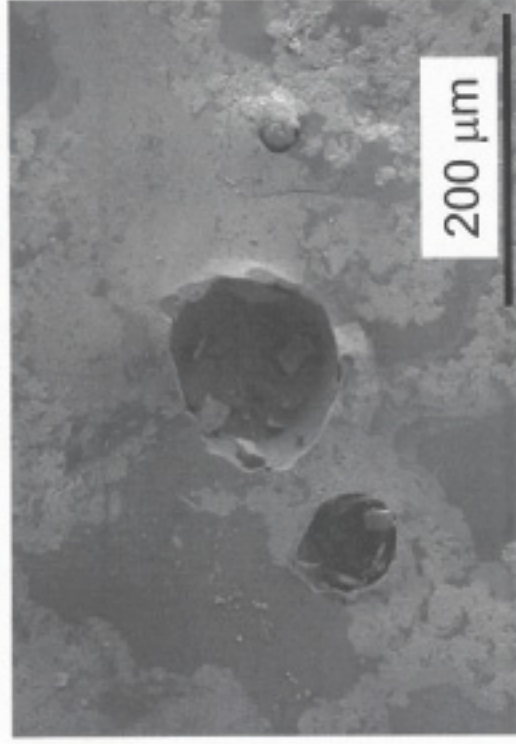
XRD Results for UHTC After Oxidation at 1627°C in Stagnant Air for 100 Min.

composition	phases present
ZS	ZrO ₂ (m), ZrO ₂ (c)
HS	HfO ₂ (m), HfO ₂ (c), HfSiO ₄
ZSTS	ZrO ₂ (m), ZrO ₂ (c)
ZS+	ZrO ₂ (m), ZrO ₂ (c)
ZSTC*	ZrO ₂ (t), ZrO ₂ (m)
HSTS	HfO₂(m), HfO₂(c), HfSiO₄

bold = major phase

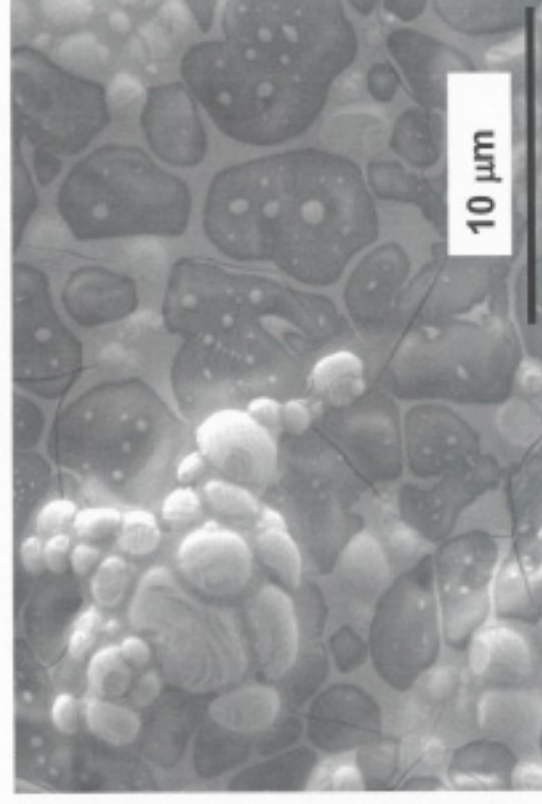
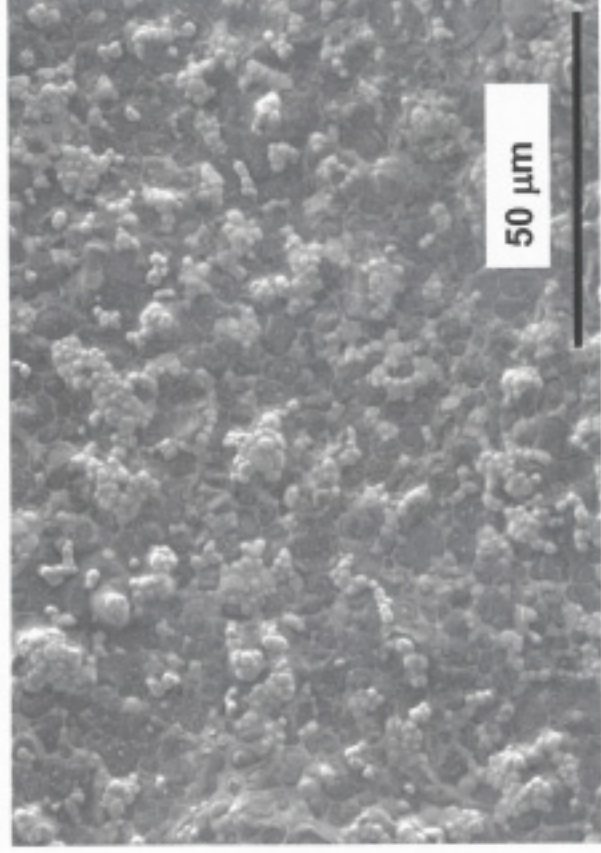
Ta₂O₅·6ZrO₂ (orthorhombic) difficult to distinguish from ZrO₂(c)

Surface Microstructure of ZS After Oxidation at 1627°C in Stagnant Air



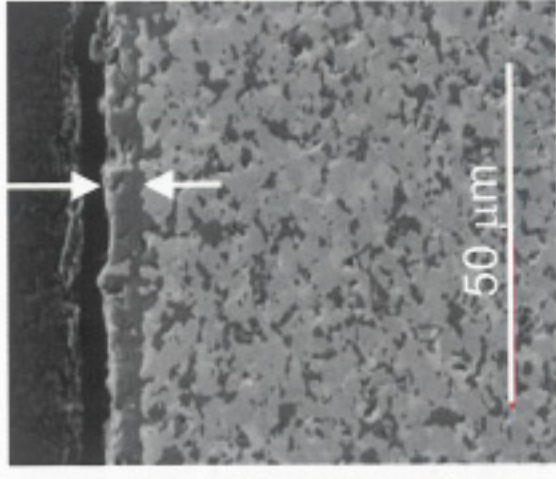
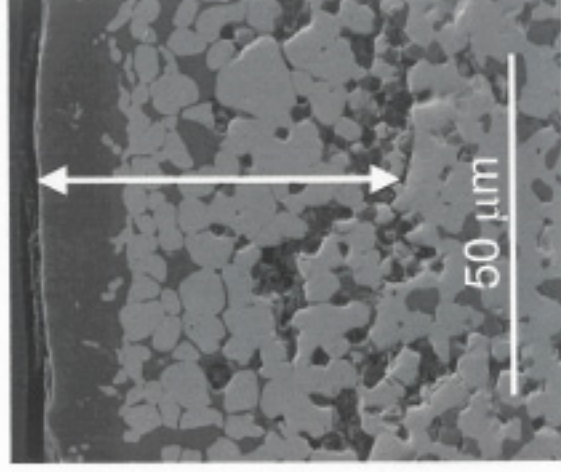
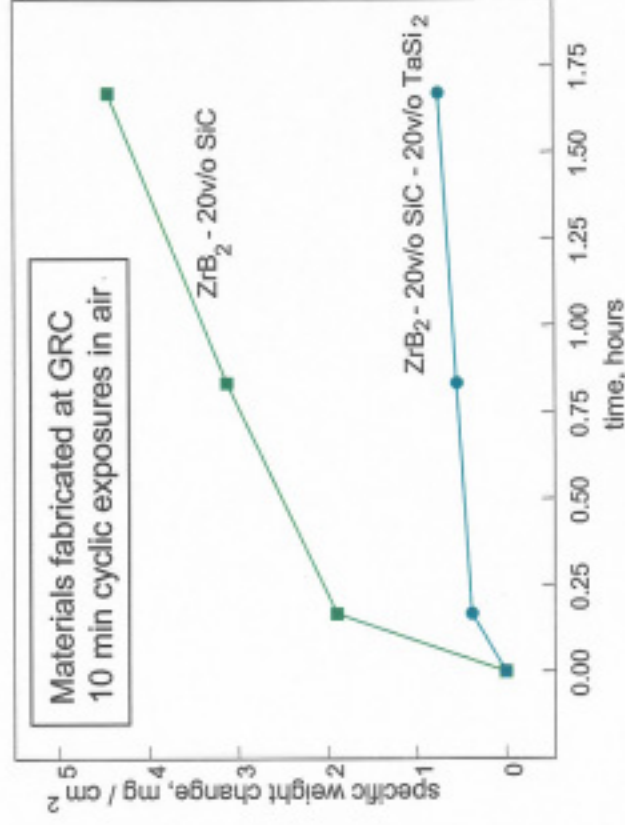
- Non-uniform dispersion of surface oxide phases.
- Fine dispersion of second phase in glassy areas.

Surface Microstructure of ZSTs After Oxidation at 1627°C in Stagnant Air



- Uniform dispersion of surface oxides
- Evidence of glass immiscibility

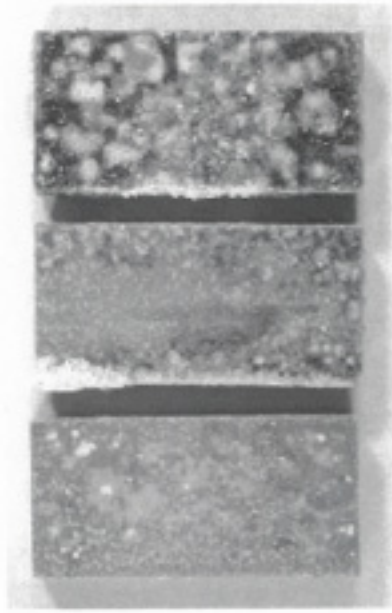
Tantalum Additions Improve Oxidation Resistance of ZrB_2 -20v/o SiC at 1627°C



UHTC Furnace Oxidized in Air

ZrB₂ + 20 v/o SiC

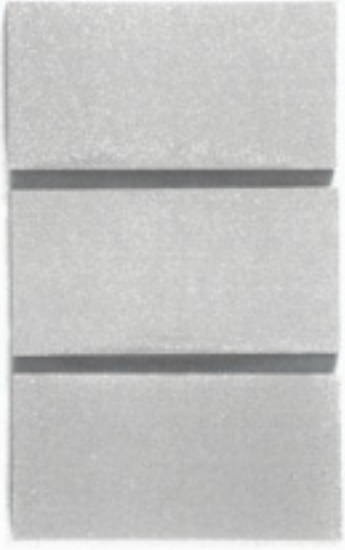
1* 5 10



1627°C

ZrB₂ + 20 v/o SiC + 20 v/o TaSi₂

1 5 10



1927°C



5 cycle specimen after attempted removal from setter

* # of cycles (1 cycle = 10 minutes hot & 10 minutes cool)

Arc Jet Test of ZrB_2 - 20 v/o SiC - 20 v/o TaSi_2



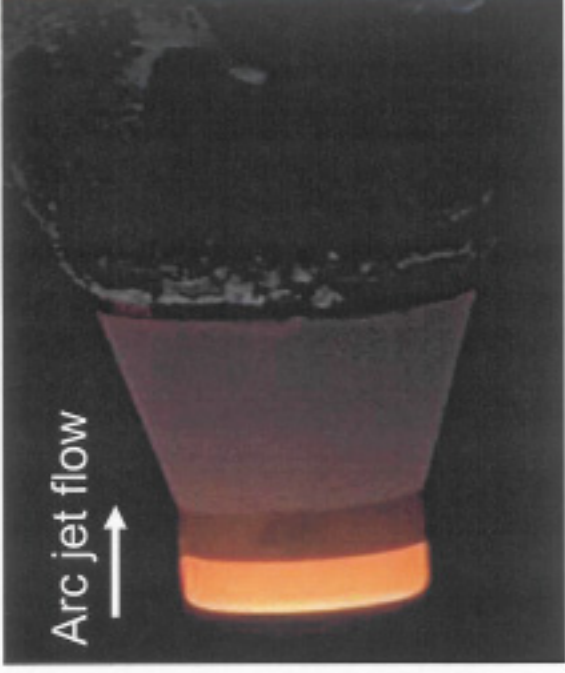
2.54 cm

Specimen



2.54 cm

Specimen mounted in
graphite holder



Arc jet flow

Specimen and graphite holder
mounted in arc jet sting

Test Conditions

350 W/cm²

0.07 atm.

600 seconds

% Δ wt = -1.4

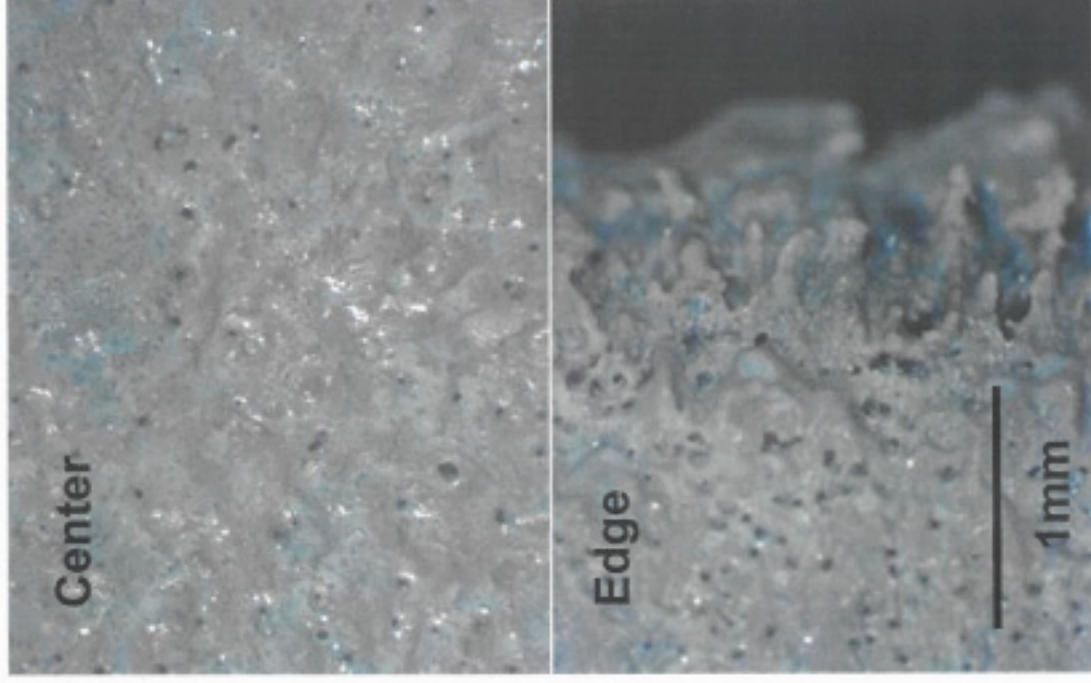
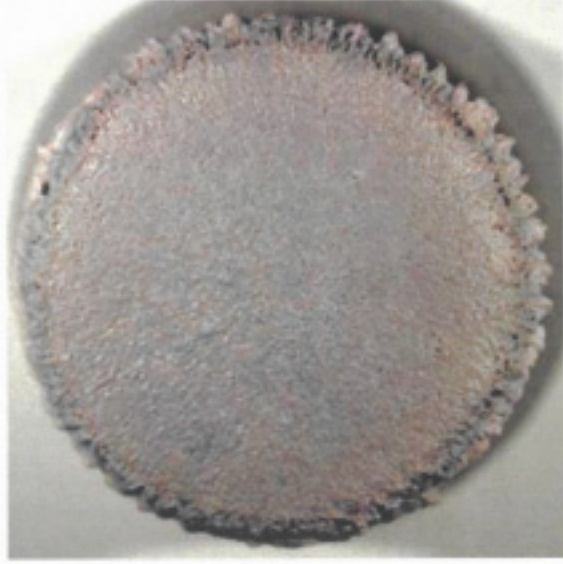
Temperatures and Heat Flux

Measured Temperature ~1800 °C

Edge Heat Flux Estimate ~ 600 W/cm²

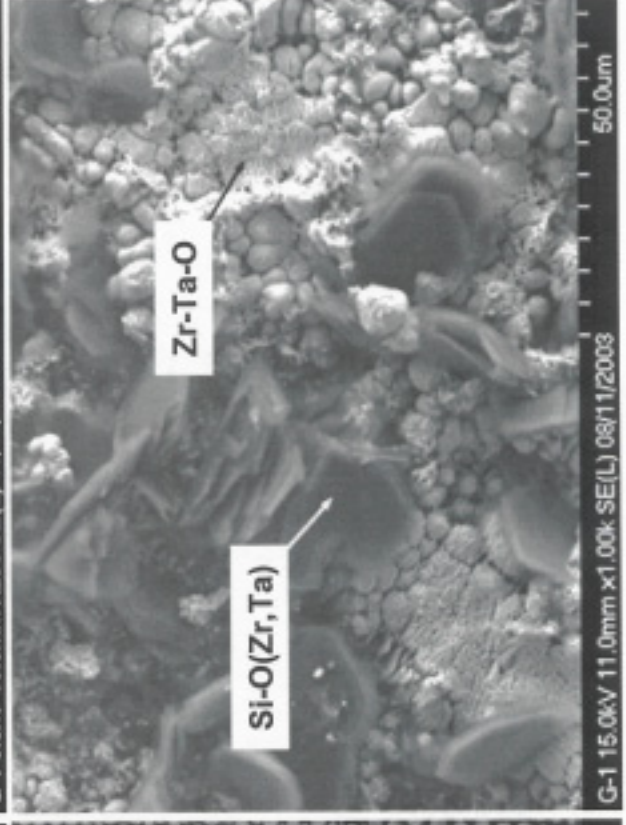
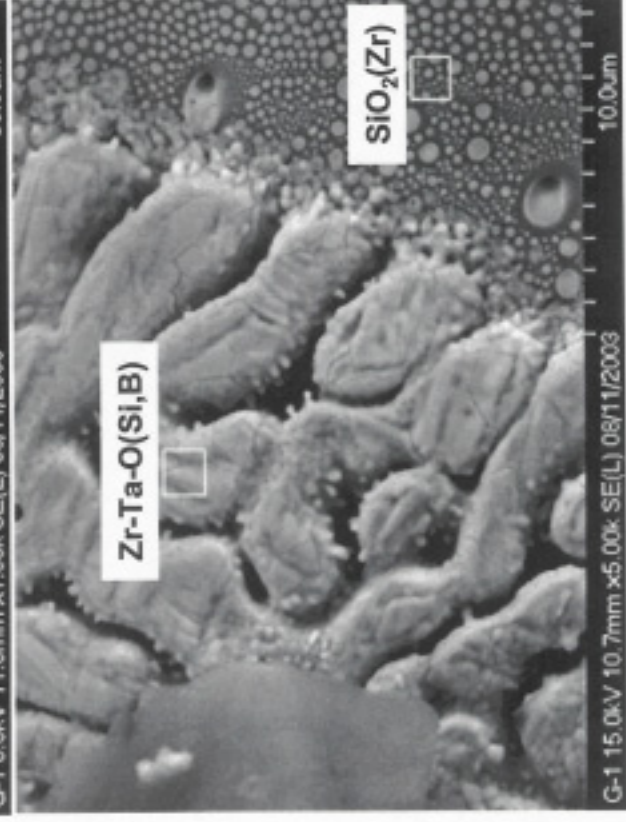
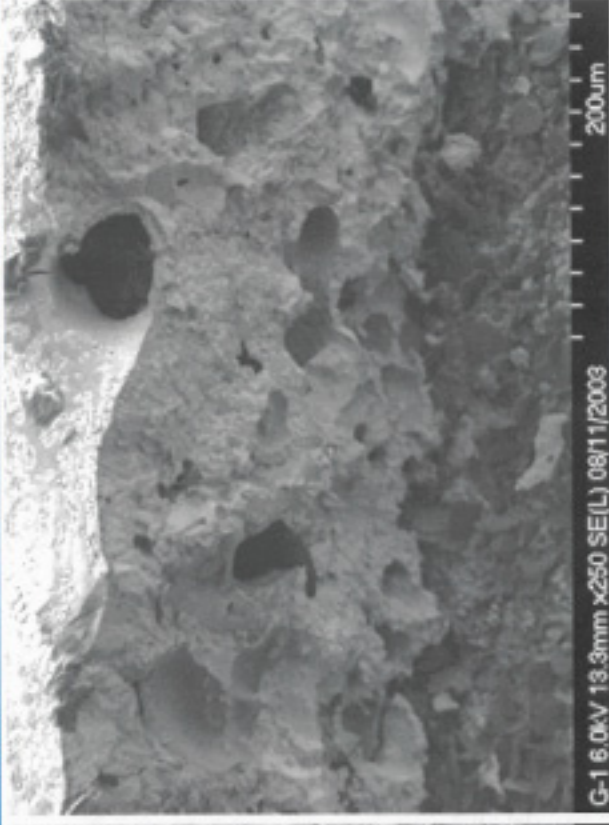
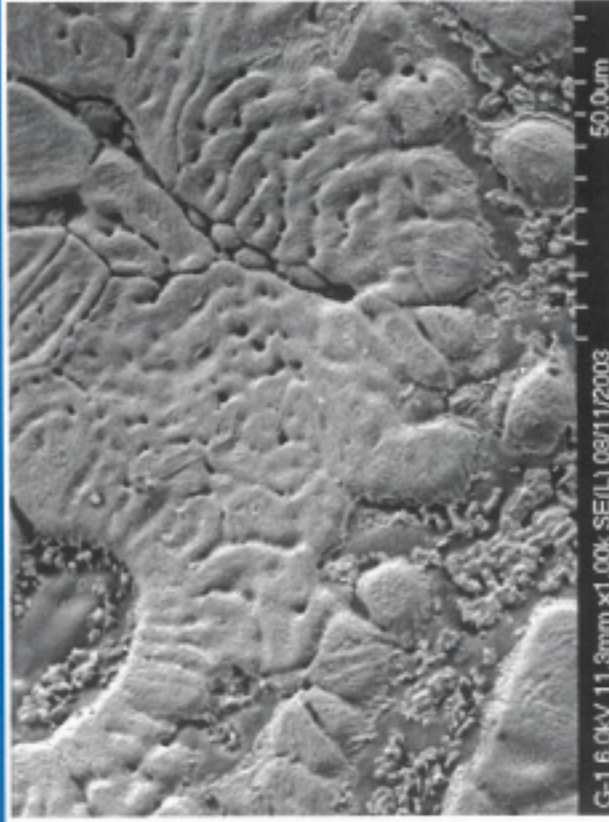
Edge Temp Estimate ~1950 to 2000°C

Images of GRC Model After Arc Jet Test



<u>XRD Results</u>	
•ZrO ₂ (m)	
•ZrO ₂ (c)	
•Ta ₂ O ₅	↑

SEM Images of Arc Tunnel Tested ZrB_2 -SiC-TaSi₂



Some Answers

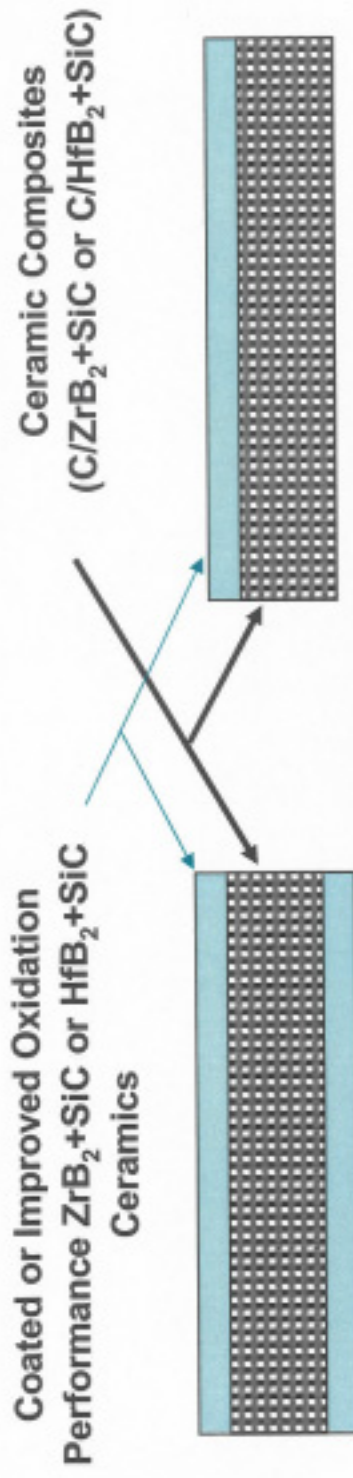
- Is Ta or Si in TaSi₂ responsible for the improved oxidation resistance of ZrB₂ – 20v/o SiC – 20v/o TaSi₂? Tantalum.
- Does the benefit of TaSi₂ additions observed at 1627°C extend to higher temperatures? No, not at 20v/o addition levels.
- Are TaSi₂ additions the best way to add Ta to ZrB₂-based UHTC? TaC additions are not effective at 1627°C.
- Can TaSi₂ additions improve the oxidation resistance of HfB₂-based UHTC which are already superior to ZrB₂-based materials? No, not at 20v/o addition levels.
- Can anything be learned about the mechanism by which Ta-additions improve the oxidation resistance of ZrB₂-SiC materials at 1627°C? Tantalum effects both glass phase and ZrO₂.

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UHTCMC Hybrid System



Advantages of UHTCMC hybrids

- Thickness and composition of UHTC layers can be changed to suit the application requirements
- Improved toughness
- Improved environmental durability
 - Tailored surface coatings
 - Crack healing matrices
 - Control of residual stresses
- Smooth composite surfaces
 - Low drag
 - Machinable without fiber damage
 - Easier to bond and attach sensors and other devices
 - Critical to good attachments/seals

Characterization of an Ultra-High Temperature Ceramic Composite

- **Objectives**
 - **Characterize a UHTC composite plate fabricated by Starfire Systems, Inc.**
 - **CAVEAT: Recognize that little or no development effort went into fabrication of this material. It was a best effort fabrication for NASA LaRC**
 - **Reveal some of the issues associated with the UHTCC concept**

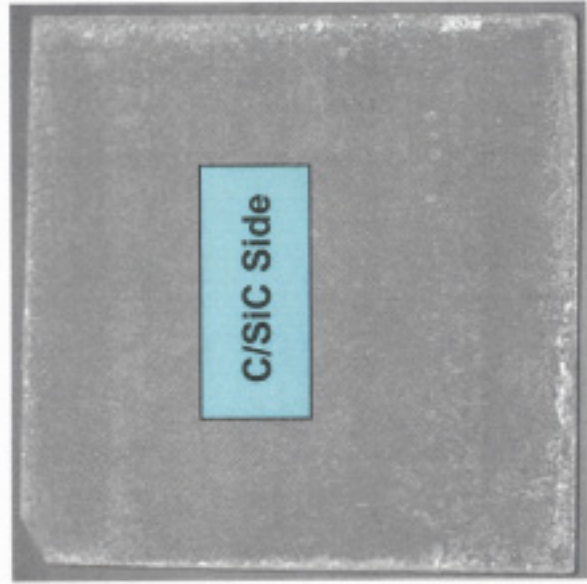
Starfire UHTCC Plate

Constituents

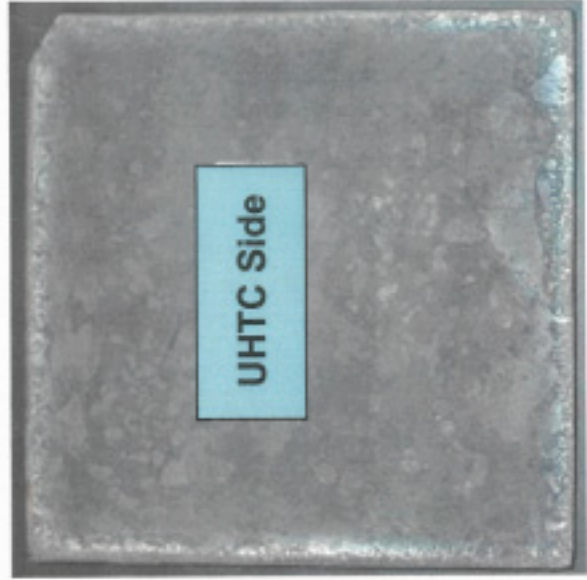
- Zoltek Panex® 30 Carbon Fabric PW06
- Starfire Systems' SP-Matrix Polymer (Allylhydridopolycarbosilane (AHPCS))
- HfB₂ Powder
- SiC Powder

Processing of Part Number 000928-6-64

- Initial Cycle:
 - For the initial lay up the **bottom 6 layers of cloth are coated with a SiC/AHPCS slurry** and the **top 5 layers are coated with a HfB₂ /AHPCS slurry** and assembled in a mold.
 - Cured to 400°C and fired to 850°C under inert gas to pyrolyze.
- Cycle 2:
 - Coat the **HfB₂ side of the plate with more HfB₂ /AHPCS slurry**.
 - Fire directly to 850°C under inert gas.
- Cycle 3: **Repeat cycle 2**
- Cycle 4 - 10:
 - Vacuum infiltrate with AHPCS only.
 - Pyrolyze directly to 850°C under inert gas, no clamping necessary.



C/SiC Side

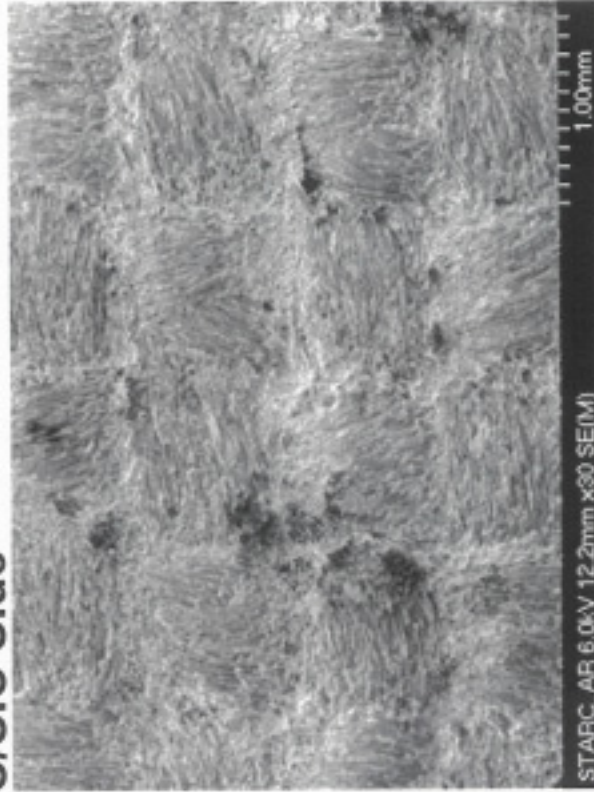


UHTC Side

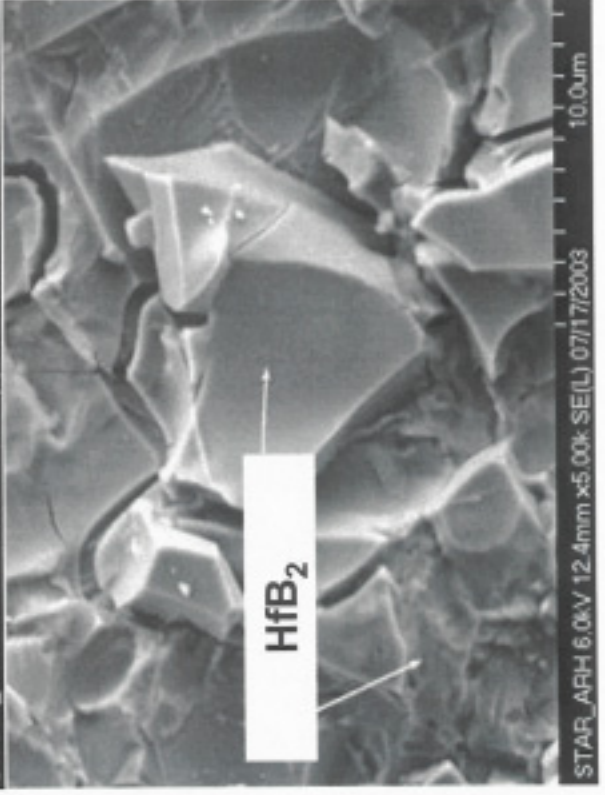
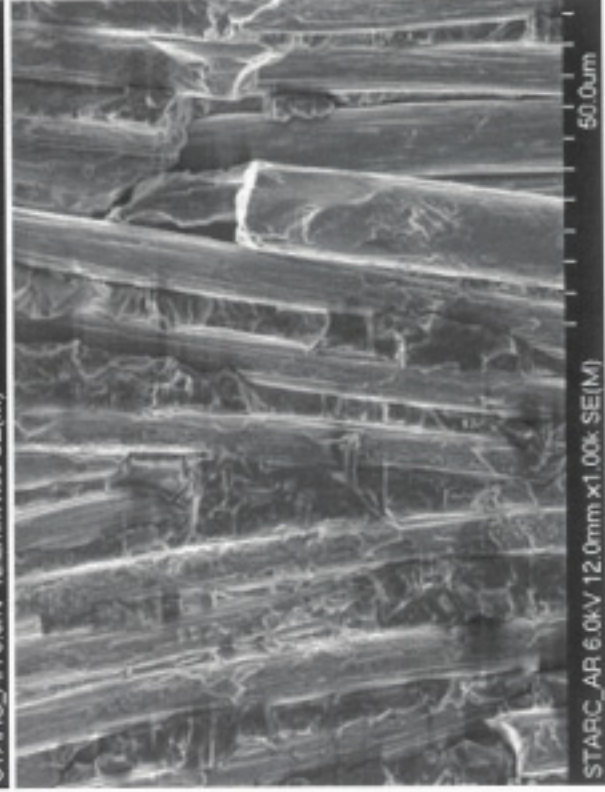
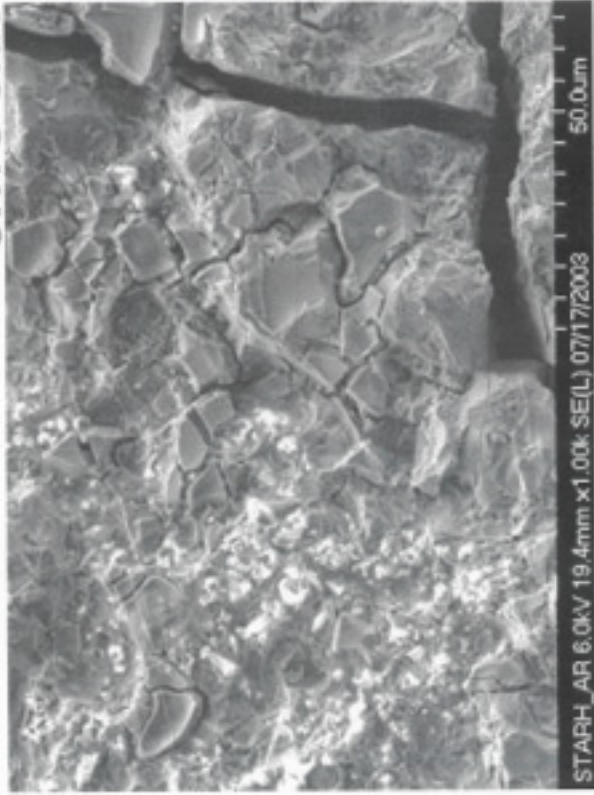


As -Received UHTCC

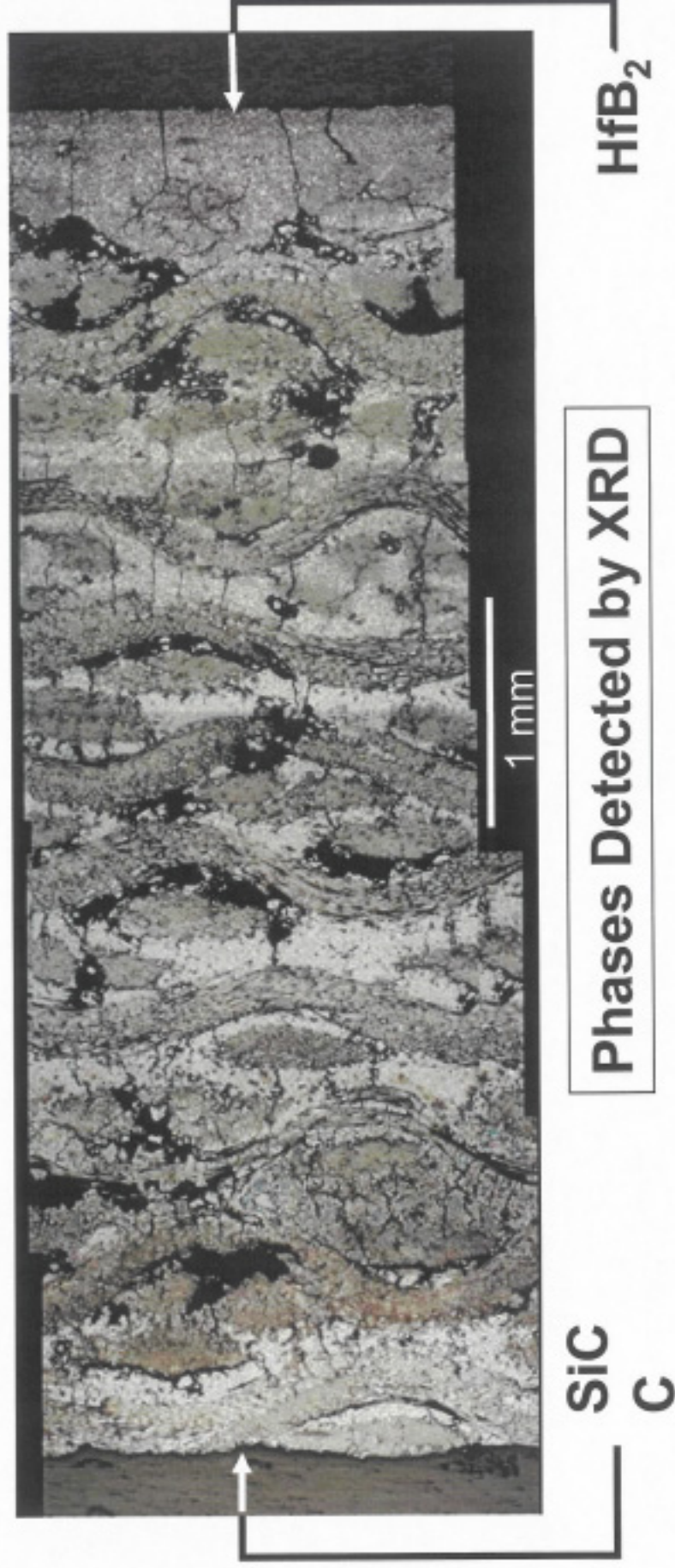
C/SiC Side



UHTC Side

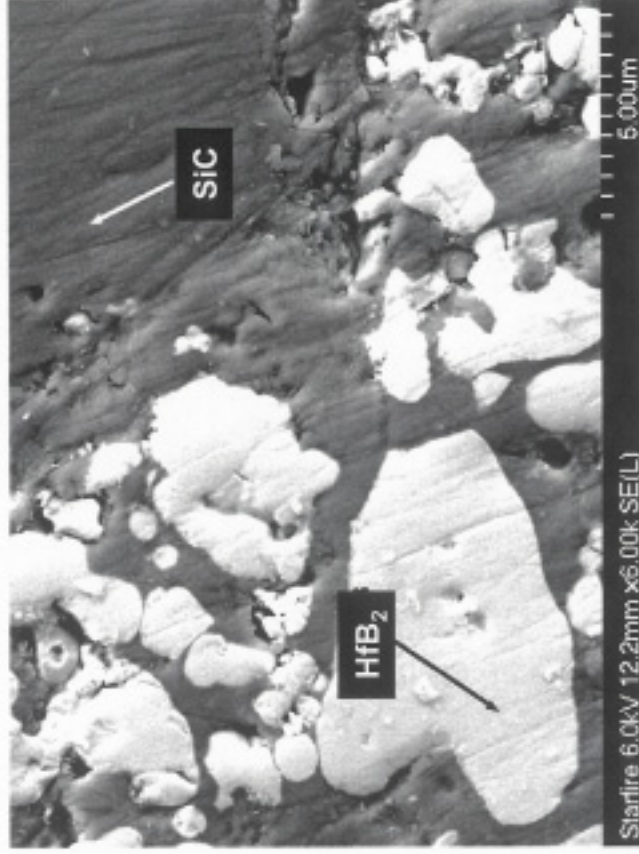


UHTCC Cross-section

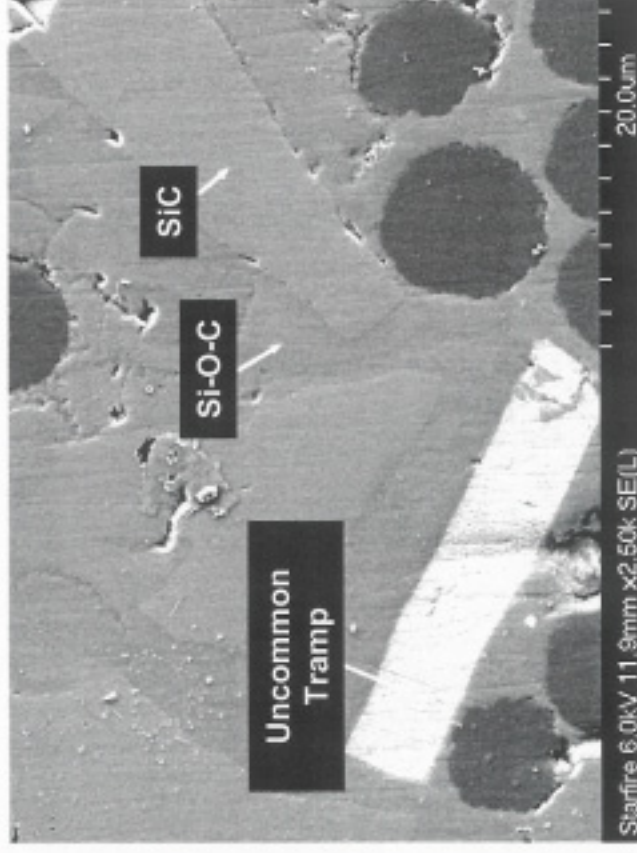


Constituents in As-Received UHTCC

HfB₂ Coating Layer



Matrix on C/SiC Side



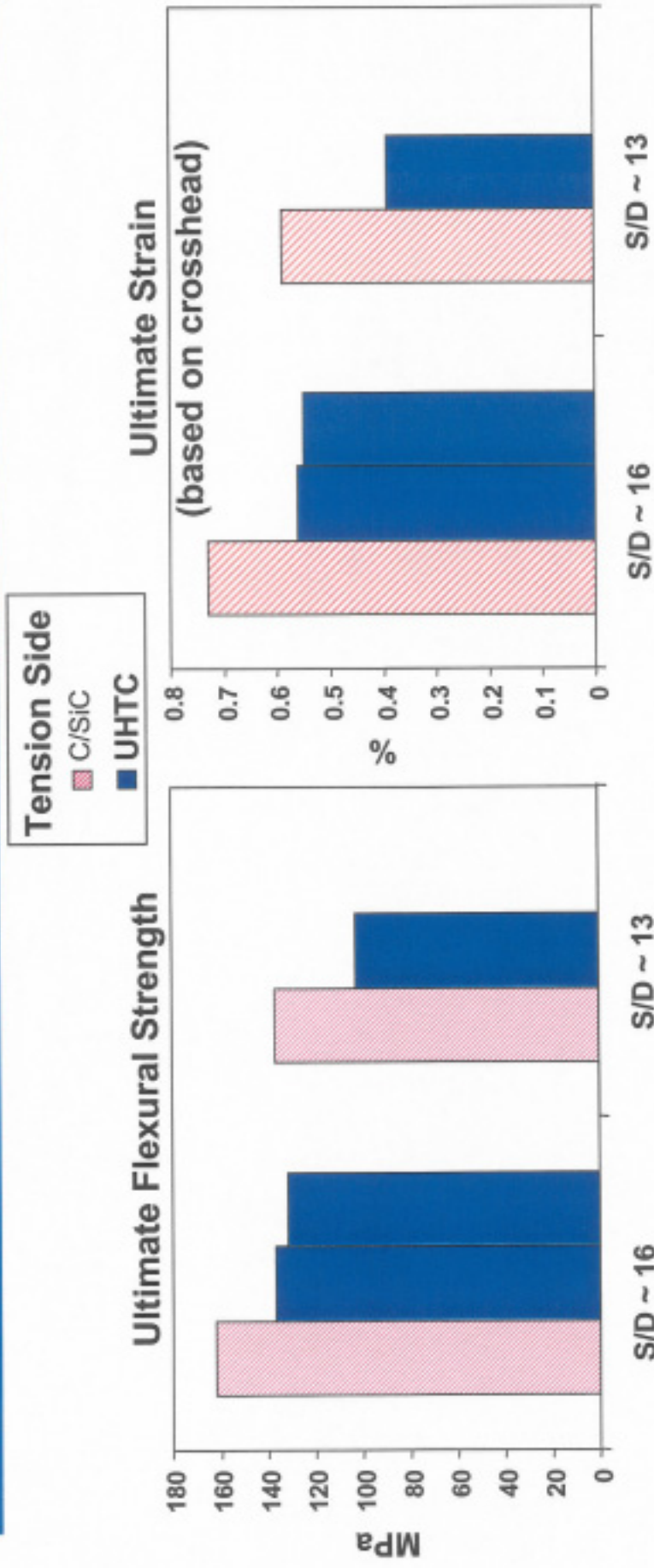
Flexural Strength Tests

Specimen		Test Fixture			Orientation		Ultimate Load	Calculated Load Based on Beam Theory*
		inner span	outer span/ depth	fracture	side down			
I	width thickness							
D								
	mm	mm	mm				N	N
A	12.7	40	80	13.8	under pin	C/SiC	972	1120
B	12.7	40	80	13.5	under pin	UHTC	757	
C	12.7	48	96	15.9	center	C/SiC	1025	
D	12.7	48	96	15.8	center	UHTC	811	1000
E	12.7	48	96	15.8	center	UHTC	855	

- Sample B remained intact. All other samples fractured into 2 pieces
- Samples B, D, and E retained an obvious permanent set

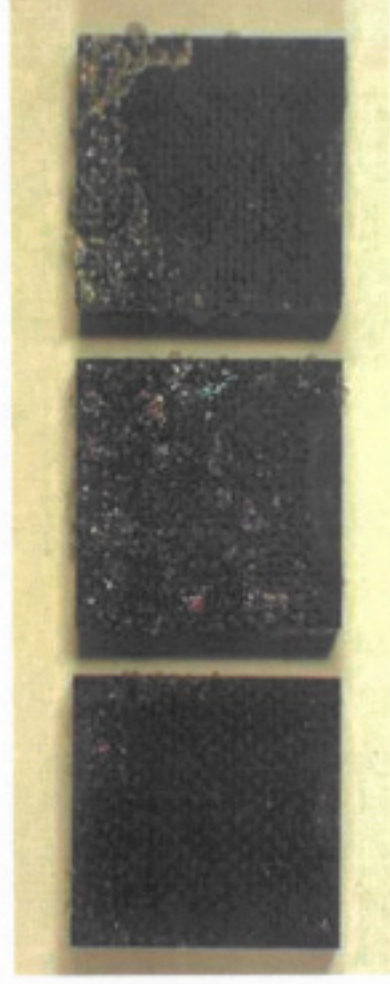
- *Calculated load at 0.7% strain
- Panex 30 minimum property: $E = 193 \text{ GPa}$
- Rule of mixtures with no matrix contribution

UHTCC Flexural Strength Results

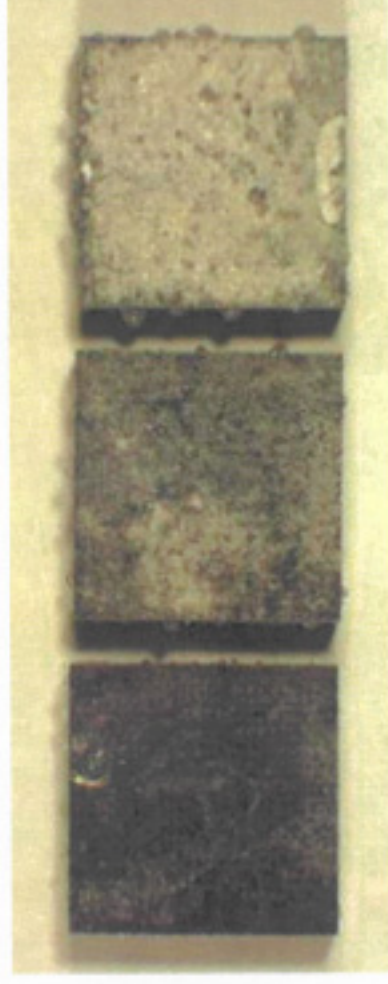


Note: S/D > 30 necessary for valid 4 point flexural test

UHTCC After 1627°C Furnace Oxidation



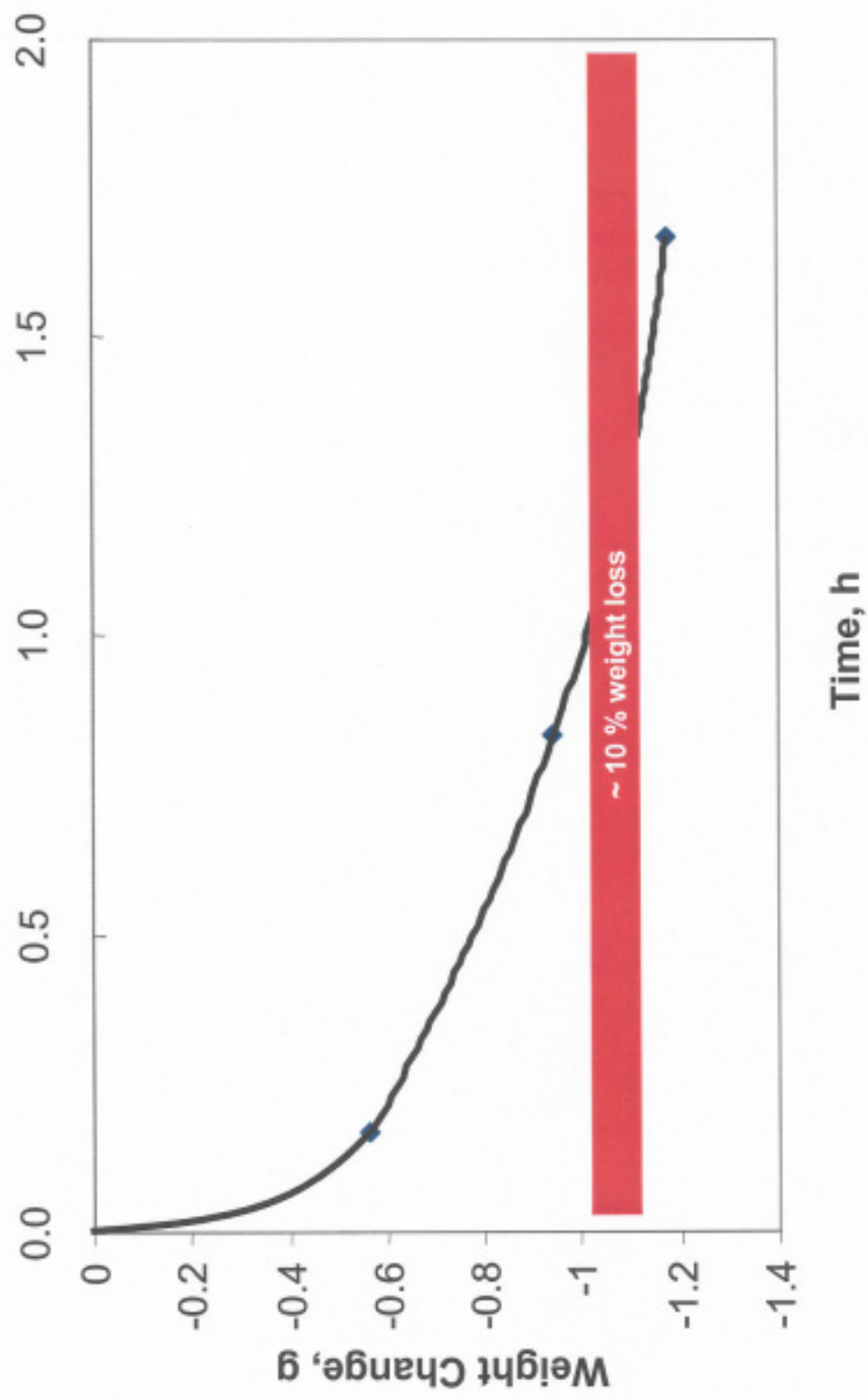
**C/SiC
Side**



**HfB₂
Side**

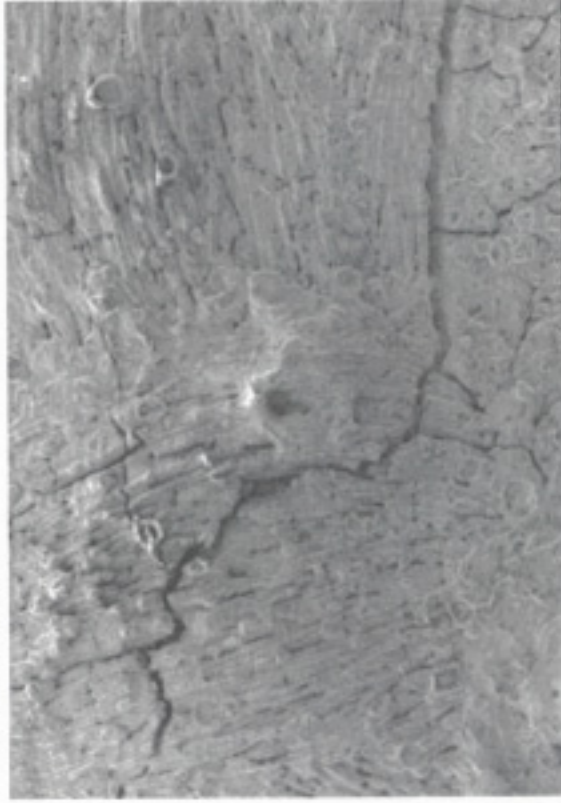
Cycles	1	5	10
Hours	0.167	0.833	1.667

Furnace Oxidation of UHTCC at 1627°C

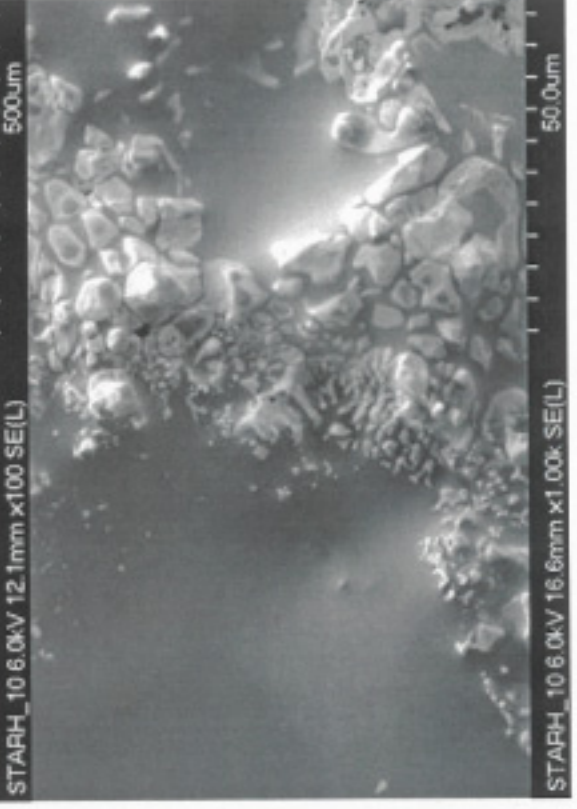
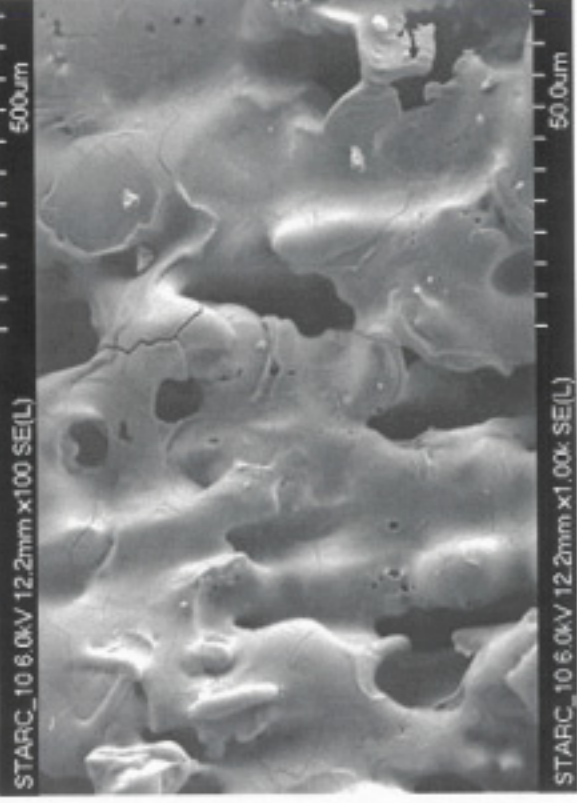
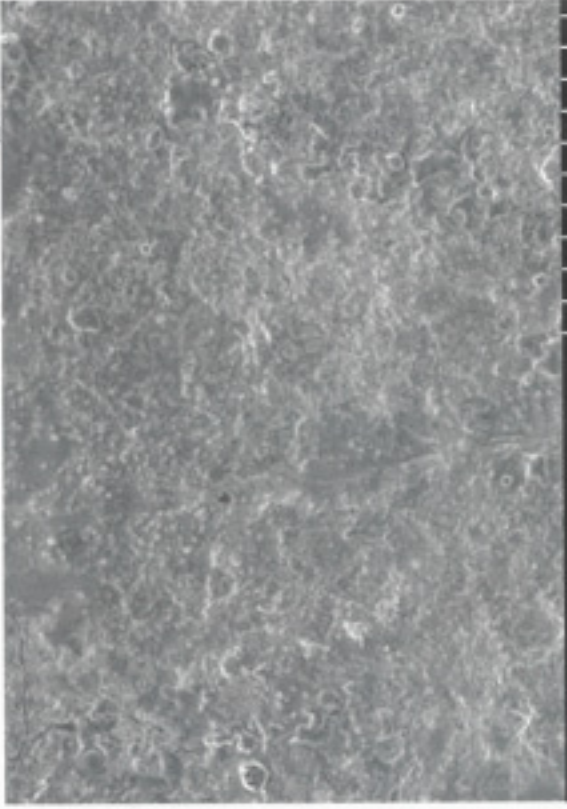


UHTCC After 10 Ten-minute Cycles in Air at 1627°C

C/SiC Side



HfB₂ Side

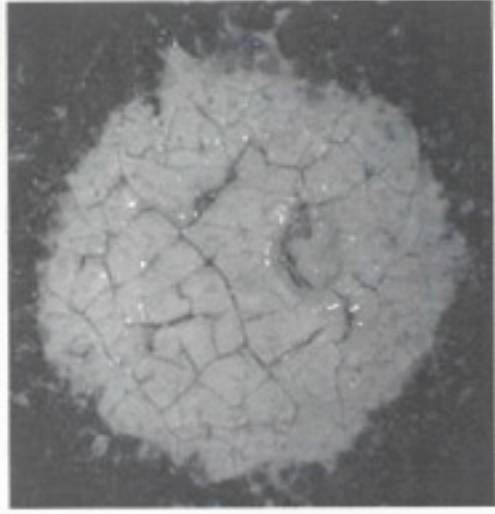
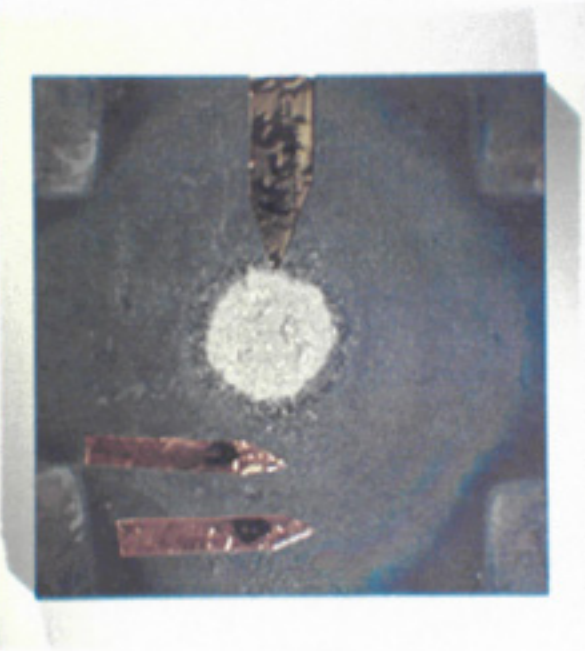


UHTCC Oxy-Acetylene Torch Test

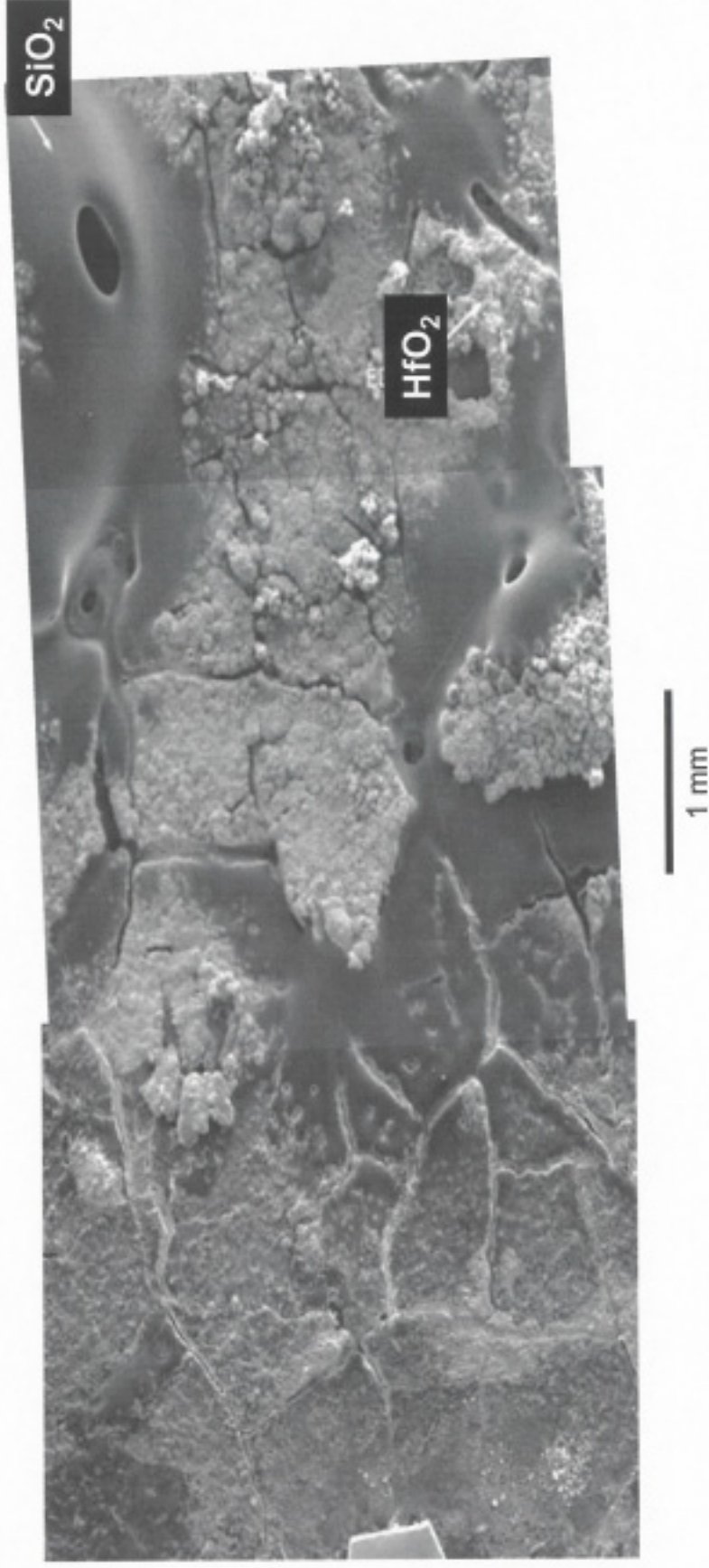
- **Sample O: One 4 min. cycle to 1805°C**
 - Photos on cool down
 - Temps with Ircon 2 color pyrometer, 980-1760°C range
 - Weight change
 - Initial weight 41.66 g
 - Final weight 40. 87 g
 - Weight loss 0.79 g, or 1.9%
- **Sample N: Three ~4 min. cycles**
 - Cycle 1 max temp 1815°C
 - Cycle 2 max temp 1915°C
 - Cycle 3 max temp 2015°C
 - Photos on heat up
 - Weight change
 - Initial weight 41.99
 - Final weight 40.04
 - Weight loss 1.95 g, or 4.6%

Sample O	
Time, min.	Temp, °C
0.5	1720
1.0	1750
1.5	1750
2.0	1755
2.5	1765
3.0	1775
3.5	1790
4.0	1805

UHTC Surface After Three 4-Minute Torch Cycles to 1815 to 2015°C



SEM of Center of UHTC Surface After Three Torch Cycles



Conclusions

- **TaSi₂ Additions**
 - Good results at 1627°C, but oxide scale too fluid at > ~ 1800°C
- **UHTCC**
 - **Processing**
 - Uniform and through thickness graded microstructure achieved
 - Matrix cracking due to thermal expansion mismatch between C fibers and matrix constituents is a concern
 - **Mechanical Properties**
 - Flexural strength was close to expected values based on rule of mixtures with no matrix contribution
 - Some evidence of composite behavior
 - **Furnace Oxidation**
 - Based on weight loss, carbon fiber oxidation occurred rapidly
 - **Torch Test**
 - Material withstood ~2000°C (~3600°F), severe heat-up and thermal gradients with no major visible distress
 - Based on observed temperature spikes during test, adherence of the HfO₂-rich scale is an area of concern

Recommendations

- Lower levels of Ta addition to achieve doping without low melting phase formation should be investigated
- The thermal stress response of early UHTCC makes the concept worthy of further study
- Fiber coatings need to be incorporated to address fiber oxidation issues
- Advanced SiC fibers need to be evaluated to address oxidation and thermal expansion mismatch issues

Future Work

- Complete oxidation studies focused on Ta additions
- Continue UHTCC development
- Continue UHTCC evaluation
 - Complete metallography on Starfire specimens
 - Evaluate other NASA and industry developed materials

Acknowledgements

- Thanks to QSS employees Terry R. McCue for scanning electron microscopy support and Ronald E. Phillips for assistance with testing at GRC.
- Thanks to David Glass of NASA Langley for providing the Starfire UHTCC plate.
- Thanks to Walt Sherwood of Starfire Systems Inc. for providing processing details and for permission to present this study.
- Thanks to Sarah E. Beckman and Jerome W. Ridge (Eloret Inc.) of NASA Ames Research Center for assistance with arc jet testing.