SURVEY OF DUST ISSUES FOR LUNAR SEALS AND THE RESOLVE PROJECT

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Challenges of Future Lunar Exploration

"Dust represents the single largest technical challenge to prolonged human presence on the Moon."

Harrison Schmidtt, Apollo 17 Astronaut March 2005

- The extent and duration of planned lunar surface activities is much higher than prior Apollo experiences.
- Systems and components will be exposed to environmental factors for periods of time orders of magnitude longer than those previously addressed.

Dust mitigation strategies:

- 1. Design systems tolerant of dust properties
- 2. Develop techniques to clean or remove dust from surfaces
- 3. Active abatement methods to minimize or eliminate deposition and/or adhesion of dust



The Apollo Lunar Sample Return Container (ALSRC) maintained a lunarlike vacuum around the samples until they were opened in the Lunar Receiving Laboratory. In practice, substantial leakage was detected in 4 of the **12** ALSRC's returned from the moon due to pieces of equipment or dust interfering with the seals.

¹ Stansbery, E.K., Kaplan, D.I., Allton, J.H. and Allen, C.C.(1997) Planetary Protection Issues for a Mars Sample Return Mission. Report to the NASA Planetary Protection Officer. October 1997





- 1. Carson, M.A., Rouen, M.N., Lutz, C.C. and McBarron, J.W.(1975) Biomedical Results of Apollo. Chapter 6 Extravehicular Mobility Unit. NASA SP-368
- 2. Young, L.A. and Young, A.J. The Preservation, Storage, and Display of Spacesuits. Smithsonian National Air and Space Museum Collection Care, Reprot Number 5, December 2001.
- 3. Apollo Operations Handbook (1971) Extravehicular Mobility Unit. Volume I. System Description. CSD-A-789-(1). Revision V, March 1971.
- 4. Gaier, J.R. (2005) The Effects Of Lunar Dust On Eva Systems During The Apollo Missions. NASA TM 2005-213610, March 2005.





- 1. Young, L.A. and Young, A.J. The Preservation, Storage, and Display of Spacesuits. Smithsonian National Air and Space Museum Collection Care, Reprot Number 5, December 2001.
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Characteristic	Description
Size	90% < 1000 μm, 70% < 100 μm
Shape	Angular/subangular sharp
Bulk Density (0-30 cm)	$1.58 \pm g/cm^3$
Hardness	5-7 (Mohs scale)
Porosity (0-15 cm)	52% <u>+</u> 2%
Cohesion (0-15 cm)	0.52 KPa (.0053 kg/cm ^{2;} .0754 psi)
Toxicity	Primarily non-toxic
Corrosiveness	Not active in vacuum
Electrostatic	Highly charged
Magnetic	<20 µm high ferromagnetic susceptibility
Thermal Conductivity	1.72-2.95 x 10 ⁻⁴ W/cm °K (Apollo 17)
Compressibility (loose)	0.3 (compression index)

Size 1000 µm=1 mm=.04"

Bulk density - property of particulate materials. <u>mass</u> particles divided by the <u>volume</u> they occupy. The volume includes the space between particles as well as the space inside the <u>pores</u> of individual particles.

Hardness - resistance to permanent deformation (Diamond=10, Quartz=7, Gypsum=2)

Porosity of a <u>porous medium</u> (such as <u>rock</u> or <u>sediment</u>) describes how densely the material is packed. It is the proportion of the non-solid volume to the total volume of material. 15 cm depth measurement?

Cohesion-particles stick together

Electrostatic- forces exerted by a static (i.e. unchanging) electric field upon charged objects

Magnetic- Susceptibility of Soil Particles Increases as Grain Size Decreases; Effects of Vapor-Deposited Nanophase Feo are a Direct Function of Surface Area and Most Pronounced in the Finest Grain Sizes; Virtually All <10 μ m Particles are Easily Attracted by a Simple Hand-held Magnet

Thermal Conductivity- ability to conduct heat. Apollo 17 heat flow probes 2.36 m

Copper is 385 W/m-K, 3.85 W/cm-K

Compressibility is a geological term used to quantify the ability of a soil to reduce in volume with applied pressure



Composition from Apollo mission samples at specific sites

Other

Titanium 2%????

Sodium	0.2 %
Manganese	0.2 %
Potassium	0.1 %
Chromium	0.1 %
Nickel	300 ppm
Hydrogen	100 ppm
Carbon	100 ppm
Nitrogen	80 ppm
Copper	5 ppm
Thorium	1 ppm

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Equatorial radius	1738.1 km
Surface area	37.8 x 10 ⁶ km ²
Mass	7.35 x 10E ²² kg
Density	3.34 g/cm ³
Surface gravity	1.63 m/sec ²
Orbital Period around earth	27.32 Earth days
Atmospheric Pressure	3 x 10 ⁻¹³ KPa (2 x 10 ⁻¹² torr)
Measured Surface Temps. (Apollo)	Min -181 C (92 K) Max 111 C (384 K)
Atmosphere (%)	Helium (25);Neon (25);Hydrogen(23);Argon(20); Trace: Methane; Ammonia; Carbon Dioxide
Lunar Radiation Sources	Galactic cosmic rays (GCR)
	Solar particle events (SPE) wind, cosmic rays

Summary of radiation found on page 48 of:

Heiken, G.H., Vaniman, D.T. And French, B.M., "LUNAR SOURCEBOOK, A User's Guide to the Moon," Cambridge University Press 1991.

Estimated lunar surface temperature of 40K from Lunar Study Group in 1972 based on Earth-based observations and need to be updated with actual measurements on the Lunar surface.

Dalton, C. and Hoffman, E. (1972) "Conceptual Design of a Lunar Colony" NASA Grant Rpt. NGT 44-005-114, NASA, Washington, D.C.

Shakleton Crater

Due to this almost constant illumination, the crater rim is considered a preferable location for a future <u>lunar outpost.[9]</u> The light could be converted into <u>electricity</u> using <u>solar panels</u>. The temperature at the location is also more favorable than on most of the surface, and does not experience the extremes along the lunar equator where it rises to 100 °C when the Sun is overhead, to as low as -150 °C during the lunar night. The continuous shadows in the south polar craters cause the floors of these formations to maintain a temperature that never exceeds about -173 °C, or 100 K.

http://www.answers.com/topic/shackleton





¹ Sanders, Gerald B., et. al., "Regolith & Environment Science, and Oxygen & Lunar Volatile Extraction (RESOLVE) for Robotic Lunar Polar Lander Mission," Lunar International Conference 2005.

Lunar regolith is loaded into the crusher, goes to heating chamber for volatiles characterization, and hydrogen reduction process. Off-gases from the oven pass through anhydrous salt bed for water capture. Water then goes through electrolysis to produce oxygen and hydrogen. For hydrogen reduction process, hydrogen is supplied to the reaction chamber (oven). Again the off-gases pass through the anhydrous salt bed for water capture. Remaining gases pass through a metal hydride bed for hydrogen capture.









This is a cross sectioned view of a concept design being considered for the Volatiles Characterization Oven. The yellow oven rotates relative to the brown lid to the fill opening for loading regolith into the oven. Heating of the regolith occurs in the position shown in this chart. To dump the regolith the yellow oven and brown lid rotate ~180 degrees to the right and the yellow container rotates to align with the fill opening. Some type of o-ring would ride in the seal groove. The o-ring would be subject to sliding motion. The lid would shield the o-ring during the fill operation. However, it may still be necessary to provide a means to provide a wiper to clear any dust that may fall onto the o-ring as it slides past the fill opening.

