

# **Direct Measurements of the Convective Recycling of the Upper Troposphere**

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1 We present a statistical representation of the aggregate effects of deep convection on the  
2 chemistry and dynamics of the Upper Troposphere (UT) based on direct aircraft  
3 observations of the chemical composition of the UT over the Eastern United States and  
4 Canada during summer. These measurements provide new and unique observational  
5 constraints on the chemistry occurring downwind of convection and the rate at which air  
6 in the UT is recycled, previously only the province of model analyses. These results  
7 provide quantitative measures that can be used to evaluate global climate and chemistry  
8 models.

1 Deep convection is a highly efficient mechanism for the vertical transport of air from  
2 near the Earth's surface (0-2 km) to the Upper Troposphere (UT) (6-12 km) (1-5).  
3 Typical convective storms have spatial scales of tens of kilometers and vertical velocities  
4 as large as  $15 \text{ m sec}^{-1}$  (6), making their local influence in the UT extremely strong. The  
5 rapid upward flow is balanced by much slower descending flow that occurs over a larger  
6 spatial scale (7). Convection is also associated with lightning a significant source of  $\text{NO}_x$   
7 ( $\text{NO}_x \equiv \text{NO} + \text{NO}_2$ ) in the UT (8, 9). The source strength and spatial distribution of  
8 lightning  $\text{NO}_x$  emissions are not well known, with estimates ranging from 2-20  $\text{Tg(N) yr}^{-1}$   
9 for the global average (10), compared to 25  $\text{Tg(N) yr}^{-1}$  from fossil fuel combustion (11).  
10 Although there have been a number of case studies of the chemical effects of individual  
11 storms (12), studies of the aggregate effects of convection on the chemical composition  
12 and radiative forcing of the UT have been largely the province of modeling and theory  
13 (13, 14). Here we describe measurements that provide a direct link between an  
14 observable property and the ensemble of convective events.

15  
16 The chemical and radiative consequences of convection and lightning are known to be  
17 large (2, 3, 15). Upper tropospheric  $\text{O}_3$ , either transported directly from the boundary  
18 layer via convection or formed in situ following detrainment of convectively lofted  $\text{O}_3$   
19 precursors ( $\text{NO}_x$ , odd hydrogen radicals ( $\text{HO}_x$ ) and hydrocarbons) in the outflow region,  
20 directly impacts climate through a positive radiative forcing (15). Additionally, deep  
21 convection accounts for a significant fraction of the net flux of moisture from near the  
22 Earth's surface to the UT (16). Thus, the rate at which the UT is turned over by

1 convection has important implications on the hydrological cycle and the magnitude of the  
2 water vapor feedback on global temperature (17).

3  
4 In this study we describe a method for calculating the time air spends in the upper  
5 troposphere following convection from in situ measurements of the chemical composition  
6 of the UT and discuss the chemistry occurring in the outflow region as a function of time  
7 since convection. We use measurements of  $\text{NO}_2$  (18, 19) ( $\text{NO}_x$  is calculated from  $\text{NO}_2$ ,  
8  $\text{O}_3$ ,  $\text{HO}_2$  and photolysis rates),  $\text{HNO}_3$  (20, 21), OH and  $\text{HO}_2$  (22),  $\text{O}_3$  (23), aerosol number  
9 density (24), actinic flux (25), CO (26) and  $\text{CO}_2$  (27) obtained during the Intercontinental  
10 Chemical Transport Experiment – North America (INTEX-NA) aboard the NASA DC-8  
11 (28). Measurements were made at altitudes between the surface and 12.5 km, over a wide  
12 area of the US and Canada, west of 40° W and between 30 and 50° N. There were a large  
13 number of vertical profiles allowing a reasonably unbiased statistical sampling of air over  
14 this region during July and August of 2004.

15  
16 We use the deviation of the observed  $\text{NO}_x$  to  $\text{HNO}_3$  ratio from steady-state as an indicator  
17 of convective influence. The  $\text{NO}_x$  to  $\text{HNO}_3$  ratio is reset to near infinity in moist  
18 convection as a result of preferential wet scavenging of  $\text{HNO}_3$  relative to  $\text{NO}_x$  (i.e., the  
19 Henry's Law Constant for  $\text{HNO}_3$  is  $\sim 10^8$  times larger than for  $\text{NO}_x$ ) (29). Further,  
20 lightning initiated  $\text{NO}_x$  production, often coincident with convection, dramatically  
21 enhances  $\text{NO}_x$  in the outflow region. The coupling of these processes makes the  $\text{NO}_x$  to  
22  $\text{HNO}_3$  ratio in the UT an effective indicator of convective influence, where  $\text{NO}_x/\text{HNO}_3$   
23  $\gg 1$  is indicative of recent cloud outflow (30, 31). In the days following convection, the

ratio decays toward steady-state providing a chemical clock that marks the time an air-mass has spent in the UT following convection (32). A number of previous studies have used species which have no upper tropospheric source (e.g. CH<sub>3</sub>I) (33), or alternative chemical ratios to provide estimates of age of air in the UT (34, 35). Our study is unique because of the availability of high time resolution NO<sub>2</sub>, OH and HNO<sub>3</sub> measurements that allow us to build a much more extensive data set than previous studies. After the initial turbulent mixing in the near field of the convection, mixing is slow, thus the time evolution of NO<sub>x</sub>/HNO<sub>3</sub> following convection depends largely on the partitioning of NO<sub>x</sub> (between NO and NO<sub>2</sub>), the concentration of OH and the actinic flux.

### Reactive Nitrogen Partitioning in the UT

The only significant chemical sinks of UT NO<sub>x</sub> are reaction with OH to produce HNO<sub>3</sub> (Equation 1) and nighttime loss through NO<sub>3</sub> (Equation 2a-b) followed by hydrolysis of N<sub>2</sub>O<sub>5</sub> to produce HNO<sub>3</sub> (36). NO<sub>x</sub> is regenerated by nitric acid photolysis (and subsequent NO<sub>3</sub> photolysis to NO<sub>2</sub>) and reaction of OH with HNO<sub>3</sub> (Equations 3 and 4).



Including the altitude dependent rain-out rate for HNO<sub>3</sub> ( $k_{\text{rain-out}}$ ) as derived by Giorgi and Chameides (37), the expected steady-state NO<sub>x</sub>/HNO<sub>3</sub> is:

$$\left( \frac{[NO_x]}{[HNO_3]} \right)_{Steady-State} = \frac{J_{HNO_3} + k_{HNO_3+OH}[OH] + k_{HNO_3\ rainout}}{\left( k_{NO_2+OH}[OH] + 2k_{N_2O_5+H_2O} \frac{[N_2O_5]}{[NO_2]} \right) x \left( \frac{[NO_2]}{[NO_x]} \right)} \quad (3)$$

Our observations show the  $NO_x$  to  $HNO_3$  ratio to be much higher than the ratio described by equation 3 at altitudes greater than 6 km (c.f. Fig. 2). The difference between the observed ratio and that predicted by equation 3 grows with altitude, reaching a maximum at 10 km. Previous observations of  $NO_x$  and  $HNO_3$  (either measured directly or calculated from observations of  $NO_x$ , PAN and  $NO_y$ ) have shown the  $NO_x/HNO_3$  ratio to be significantly larger than the steady-state prediction in the UT (30, 31, 38-42). This has been shown to be primarily a result of convection and lightning reinitializing the system before steady-state is achieved (30, 31). Although a series of other hypotheses have been put forth (40-42), we (like Jaegle et al.) find no evidence for a mechanism other than convection responsible for holding  $NO_x/HNO_3$  out of steady-state in the UT.

### Chemical Signatures of Convection

Fig. 3 depicts one of many convectively influenced air-masses sampled in the UT during INTEX-NA. Three distinct convective events (40–80 km wide) are identified by enhancements in  $NO_x/HNO_3$  in Fig. 3a. Coincident enhancements are present in  $SO_2$ , an indicator of a recent boundary layer source for this air, and Ultra-fine Cloud Condensation Nuclei (UCN) ( $3 \leq D_p \leq 10$  nm), an indicator of cloud detrainment (Fig. 3b) (43, 44). Sharp decreases in  $CO_2$  also indicate the convective lofting of boundary layer air depleted in  $CO_2$ , a result of photosynthetic activity (Fig. 3c) (44). Enhancements in CO,  $CH_2O$  and various hydrocarbons, relative to the surrounding UT

1 air, were also observed in these plumes, further indicating that these parcels originate  
2 from the Planetary Boundary Layer (PBL). Backward air trajectories, initialized along  
3 the flight track, coupled to the spatial and temporal distribution of cloud-to-ground  
4 lightning strikes, indicate that the sampled air-mass was recently influenced by lightning  
5 approximately one day prior to DC-8 sampling (c.f. Fig. 3 bottom panel) (45). Such  
6 features with high  $\text{NO}_x/\text{HNO}_3$  were observed throughout the UT during INTEX-NA.

7  
8 To assess the extent to which the UT over the Eastern U.S. and Canada during the  
9 summer of 2004 was influenced by convection and describe the chemical evolution of  
10 convective outflow, we use a constrained time-dependent photo-chemical box model to  
11 map the observed  $\text{NO}_x/\text{HNO}_3$  to the time since the ratio was last reinitialized. The model  
12 is described in the supplemental information included with this article. It is initialized  
13 with observations at 1km vertical intervals from 6 to 12 km. The derived timing indicator  
14 for the convectively influenced air sampled on 11 August 2006 is shown in Fig. 3d. The  
15 properties of the ensemble of our measurements are shown in Figs. 4-6.

16  
17 The aerosol size distribution provides an independent indicator of air recently detrained  
18 from clouds. Cloud-processed air is depleted of aerosol surface area permitting new  
19 particle formation in the outflow region (43, 44). Fig. 4a depicts the fraction of  
20 condensation nuclei found in the 3-10 nm bin as a function of time since convective  
21 influence. The fraction of particles in this ultra-fine mode is largest during the first few  
22 days confirming that the  $\text{NO}_x$  to  $\text{HNO}_3$  ratio, and the timing indicator derived from it, is  
23 reinitialized in the UT by cloud processing. Strong enhancements in  $\text{CH}_3\text{OOH}/\text{H}_2\text{O}_2$  (not

shown), also an indicator of recent cloud processing (33, 46), were observed during the first two days after cloud processing.

As expected, both elevated  $\text{NO}_x$  and suppressed  $\text{HNO}_3$  are observed at short times (c.f. Figs. 4b-c). Enhancements in  $\text{NO}_x$  during the first few days is indicative of convection of boundary layer and/or lightning  $\text{NO}_x$  (47, 48). The suppression of  $\text{HNO}_3$  at short times is clear indication of  $\text{HNO}_3$  scavenging during convection. Fig. 4d confirms that reactive nitrogen ( $\text{NO}_y \equiv \text{NO}_x + \text{total peroxy nitrates} + \text{HNO}_3$ ) is conserved during the chemical processing following convection, a fact which provides further support for the use of  $\text{NO}_x/\text{HNO}_3$  as a marker representing time since convection.

### **Chemical Processing in Convective Outflow**

Mapping the ensemble of observations made throughout the UT onto the coordinate of time since convection allows us to assess the chemical and dynamical processes occurring following convection, without attempting a Lagrangian convection study. In this analysis we concentrate on the time evolution of CO and  $\text{O}_3$ .

The time evolution of CO, following detrainment into the UT, is set by the abundance of OH and the rate at which the convective plume entrains air from the background UT (c.f. Fig. 5a). Due to the direct dependence of the chemical clock on  $\text{HO}_x$ , we constrained both OH and  $\text{HO}_2$  to the observations as a function of  $\text{NO}_x$  and pressure in the time-dependent model used to generate time. As a result, we can iterate the model to determine the proper mixing rate of the convective plume by matching the modeled and

1 observed time evolution of CO following convection. Using this approach for a series of  
2 long lived species (e.g. CO, CH<sub>4</sub>, CH<sub>3</sub>OH and others), we calculate an average mixing  
3 rate of  $0.05 \pm 0.02 \text{ day}^{-1}$  following detrainment into the UT. This is in good agreement  
4 with the upper limit of  $0.06 - 0.1 \text{ day}^{-1}$  determined by Ray et al. from observations of  
5 convective plumes observed in the stratosphere during the CRYSTAL-FACE mission  
6 (49), however it is slower than the 2 day dilution time-scale determined by Wang et al.  
7 from observations during the SONEX Experiment (32). Since the DC-8 did not routinely  
8 sample in the turbulent environment directly surrounding convective outflow, this mixing  
9 rate likely reflects diffusive and shear induced mixing subsequent to the initial turbulent  
10 mixing occurring during detrainment from the convective system. Entrainment of UT air  
11 either during convective lofting or cloud detrainment is discussed in the next section.

12  
13 The O<sub>3</sub> mixing ratio as a function of time since convection is shown in Fig. 5b. We find  
14 that on average, convectively lofted air-masses contain less O<sub>3</sub> than the background UT.  
15 This result is consistent with the observed vertical gradient in O<sub>3</sub> observed over the  
16 continental US during INTEx-NA, with lower O<sub>3</sub> in the PBL than above (50). Rapid  
17 changes in the O<sub>3</sub> mixing ratio are observed during the first two days following  
18 detrainment, with the observed O<sub>3</sub>  $15 \text{ nmol mol}^{-1}$  above the initial value by the end of day  
19 two. The observed rate of increase slows exponentially with an asymptote at long time  
20 approaching zero and the O<sub>3</sub> mixing ratio approaching a constant value of  $85 \text{ nmol mol}^{-1}$ .  
21 This is a surprising result, as our model of the O<sub>3</sub> rate of change never approaches zero,  
22 but continues to predict a net increase of  $3 \text{ nmol mol}^{-1} \text{ O}_3 \text{ day}^{-1}$  at the end of day five  
23 (51).

## Constraints on the Convective Turnover Rate of the UT

The convective turnover rate of the upper troposphere is critical for accurately describing  $\text{NO}_x$ ,  $\text{HO}_x$  and  $\text{O}_3$  chemistry in the UT (52). However, at present there is a paucity of observation based constraints available (either meteorological or chemical) to test the aggregate effects of convection in the current generation of global chemistry and climate models. To determine the convective turnover rate of the UT from the observations presented here, both the extent to which the UT is influenced by convection and the fraction of PBL air in the convectively influenced air-masses must be known with high confidence.

To determine the fraction of PBL air contained in fresh convective outflow, we use observations of insoluble long-lived species made throughout the INTEX-NA campaign over the continent. Assuming that we conducted a statistically unbiased sampling of both the boundary layer and free troposphere during INTEX-NA, we can calculate the fraction of PBL air present in fresh convection ( $f$ ) through the following equation:

$$[X]_{UT(t=0)} = f [X]_{\text{surface}} + (1 - f) [X]_{UT} \quad (4)$$

where  $[X]_{UT(t=0)}$  is the mean mixing ratio of species  $X$  in fresh convective outflow (as identified using our timing indicator),  $[X]_{UT}$  is the mean mixing ratio of species  $X$  in the UT (7.5-11.5 km) and  $[X]_{\text{surface}}$  is the mean mixing ratio of species  $X$  between 0-1.5 km. Using observations of  $\text{CO}$ ,  $\text{CO}_2$ ,  $\text{CH}_3\text{OH}$ ,  $\text{CH}_4$  and  $\text{C}_2\text{H}_6$  we calculate the fraction of PBL

1 air in fresh convection to be  $0.19 \pm 0.05$ ,  $0.11 \pm 0.03$ ,  $0.26 \pm 0.05$ ,  $0.15 \pm 0.05$ , and  $0.34 \pm$   
2  $0.09$ , respectively. We calculate a weighted average for the fraction of PBL air in  
3 convective outflow of  $0.17 \pm 0.02$  by weighting each value by the inverse square of its  
4 uncertainty. This implies that convectively lofted PBL air rapidly entrains the  
5 surrounding air either during ascent or in the turbulent environment of the detraining  
6 flow. These results are consistent with: i.) the observations of Ray et al., who determined  
7 the fraction of tropospheric air in convective plumes sampled in the stratosphere to be  
8 between 0.1 and 0.4 (49), ii.) the observations of Cohan et al., who calculate the fraction  
9 of BL air in fresh convection outflow to be between 0.32 and 0.64 from observations of  
10  $\text{CHBr}_3$ ,  $\text{CH}_3\text{OOH}$  and  $\text{CH}_3\text{I}$  in fresh convection (33) and iii.) the modeling studies of  
11 Mullendore et al., who calculate the fraction of PBL air present in the convective outflow  
12 region of a supercell storm to be 0.26, 10 hours after storm initialization (53).

13  
14 Fig. 6a shows the normalized frequency distribution of the observed time since  
15 convection based on the ratio of  $\text{NO}_x$  to  $\text{HNO}_3$ . We find that 54% of the air sampled  
16 between 7.5 and 11.5km had been influenced by convection during the past two days.  
17 The convective outflow was strongest between 9.5 and 10.5 km, where the fraction of  
18 sampled air that is less than two days old exceeds 69%. The vertical distribution  
19 presented here is consistent with previous observations and model analyses of convective  
20 outflow to the UT from individual storms (4, 54) and the vertical distribution of  
21 convectively influenced laminae observed in  $\text{O}_3$  sonde data from the summer of 2004  
22 over the northeastern U.S. (55). The shift toward longer times between 10.5 and 11.5 km  
23 suggests that either convective cloud tops on average do not extend higher than 10.5 km

1 over the mid-latitude during the summer (56) or that transport of stratospheric air, rich in  
2  $\text{HNO}_3$ , contributes to keeping the  $\text{NO}_x$  to  $\text{HNO}_3$  ratio low at altitudes greater than 10.5  
3 km (57).

4  
5 To constrain the turnover rate of the UT from the ensemble statistics generated from our  
6 calculated time since convective influence (Fig. 6a), we constructed a two dimensional  
7 model of the UT. We assume that it takes 4 days for any individual box to pass through  
8 the sampling region and that each box has not been influenced by convection upon  
9 entering the sampling window. Every six hours we: i.) represent convection by  
10 randomly reinitializing the age of x% of the boxes in the sampling domain to 0 (the value  
11 of x is determined by the turnover rate (varied between 0.05 and 0.2  $\text{day}^{-1}$ ) and the  
12 fraction of PBL air contained in fresh convection (assumed to be a constant at 0.17)) and  
13 ii.) dilute each box with the mean value of the adjacent 8 boxes at the rate of 0.05  $\text{day}^{-1}$ .

14  
15 Fig. 6b depicts the observed and three calculated normalized frequency distribution of  
16 time since convective influence between 7.5 and 11.5 km. The shape of the distribution  
17 suggests that UT air sampled during INTEX-NA was largely influenced by convection,  
18 and that convectively lofted plumes did not have sufficient time to either mix or age prior  
19 to sampling by the DC-8, but instead were transported to the East out of the domain.  
20 Frequency distributions of time since convection, using three different convective  
21 turnover rates, in the Eastern half of the 2-D UT model analysis (where we sampled most  
22 frequently) are also shown in Fig. 6b. Assuming the DC-8 made a statistically unbiased  
23 sampling of the continental UT during summer, the best match between the model and

1 observations would imply a convective turnover rate between  $0.1 - 0.2 \text{ day}^{-1}$ . However, if  
2 we assume the DC-8 had a positive bias toward sampling fresh convection in accordance  
3 with Fuelberg et al., our observed frequency distributions are consistent with a convective  
4 turnover rate closer to  $0.1 \text{ day}^{-1}$  (45, 58).

5  
6 For comparison, the model detrainment cloud mass flux between 400 and 200 hPa  
7 (approximately 7.2 - 11.8 km based on standard atmosphere) for the INTEX-NA  
8 sampling domain ( $80^{\circ} - 100^{\circ}\text{W}$  for  $30^{\circ} - 35^{\circ}\text{N}$  and  $70^{\circ} - 100^{\circ}\text{W}$  for  $35^{\circ} - 50^{\circ}\text{N}$ ) between  
9 July 1<sup>st</sup> and August 15<sup>th</sup> 2004 was  $0.0085 \text{ kg m}^{-2} \text{ sec}^{-1}$ . This corresponds to a turnover  
10 rate of  $0.37 \text{ day}^{-1}$  (using a column mass of  $1.9 \times 10^3 \text{ kg m}^{-2}$  between 7.2 - 11.8 km). As  
11 expected, the observed and modeled detrainment rates for the extratropics are slower than  
12 in the tropics. For example, Folkins and Martin determine a maximum convective  
13 detrainment rate of  $0.4 \text{ day}^{-1}$  at 12.5 km from calculation of the clear-sky radiative and  
14 precipitation induced evaporative cooling rates using a one-dimensional model  
15 constrained by observations of temperature and water vapor (59).

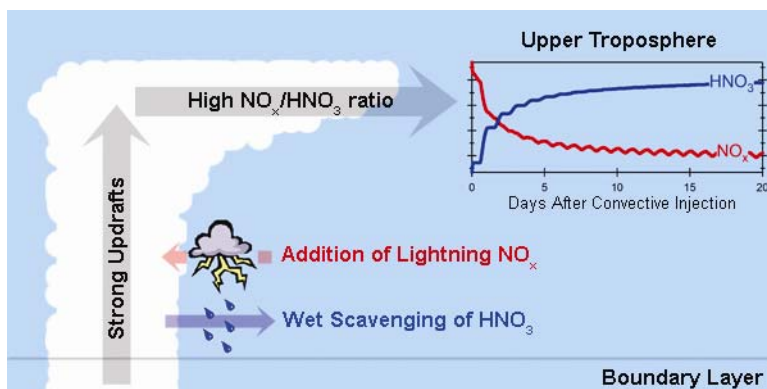
## 16 17 **Conclusions**

18 We present a statistical representation of the aggregate effects of convection on the  
19 chemistry and dynamics of the upper troposphere using in situ measurements taken  
20 aboard the NASA DC-8 during the summer of 2004 over the Eastern U.S. and Canada.  
21 These observations provide a new and unique constraint on: i.) the extent to which  
22 convection perturbs the continental UT during summer, ii.) the fraction of boundary layer  
23 air present in convective outflow, and iii.) the convective overturn rate of the upper

1 troposphere. In addition, the chemical clock described here defines a coordinate that can  
2 be used to assess the chemistry occurring down-wind of convective injection. These  
3 direct measures of atmospheric rates present a new opportunity for quantitative tests of  
4 model representations of processes governing UT ozone, convection, lightning and their  
5 impact on climate.

6

1

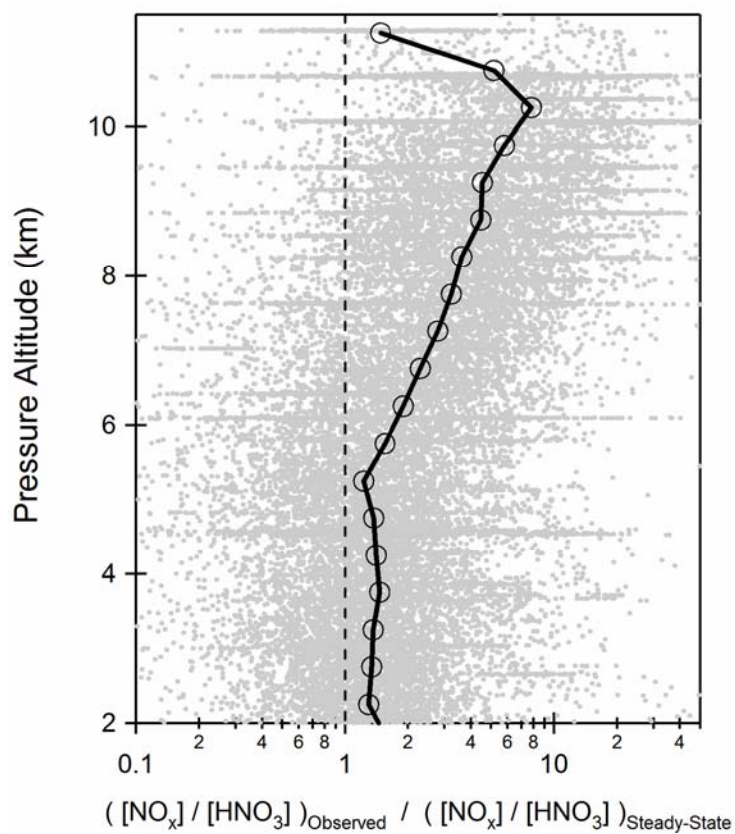


2

3

4 **Figure 1:** In moist convection, air from near the Earth's Surface is rapidly transported  
 5 upwards and detrained into the Upper Troposphere. In this process, Nitric Acid (highly  
 6 soluble) is efficiently scavenged, while  $\text{NO}_x$  (insoluble) remains.  $\text{NO}_x$  is dramatically  
 7 elevated by concurrent lightning  $\text{NO}$  production, resulting in high  $\text{NO}_x$  to  $\text{HNO}_3$  ratios in  
 8 the convective outflow region. Following detrainment into the UT,  $\text{NO}_x$  is converted to  
 9  $\text{HNO}_3$  by  $\text{OH}$  during the day and via  $\text{NO}_3/\text{N}_2\text{O}_5$  at night. The chemical evolution of the  
 10  $\text{NO}_x/\text{HNO}_3$  ratio provides a unique indicator of the time a sampled air-mass has been in  
 11 the UT following convection.

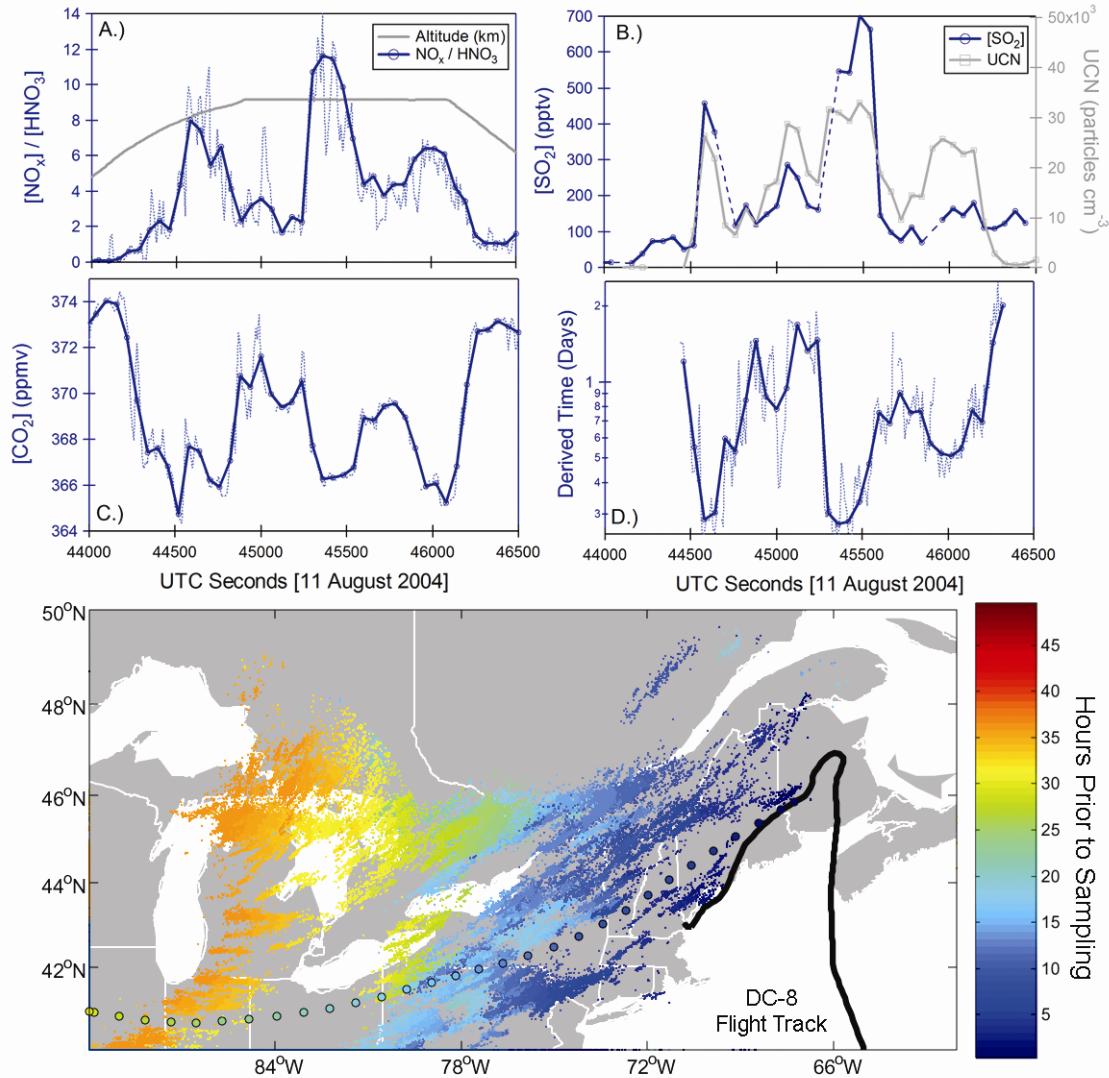
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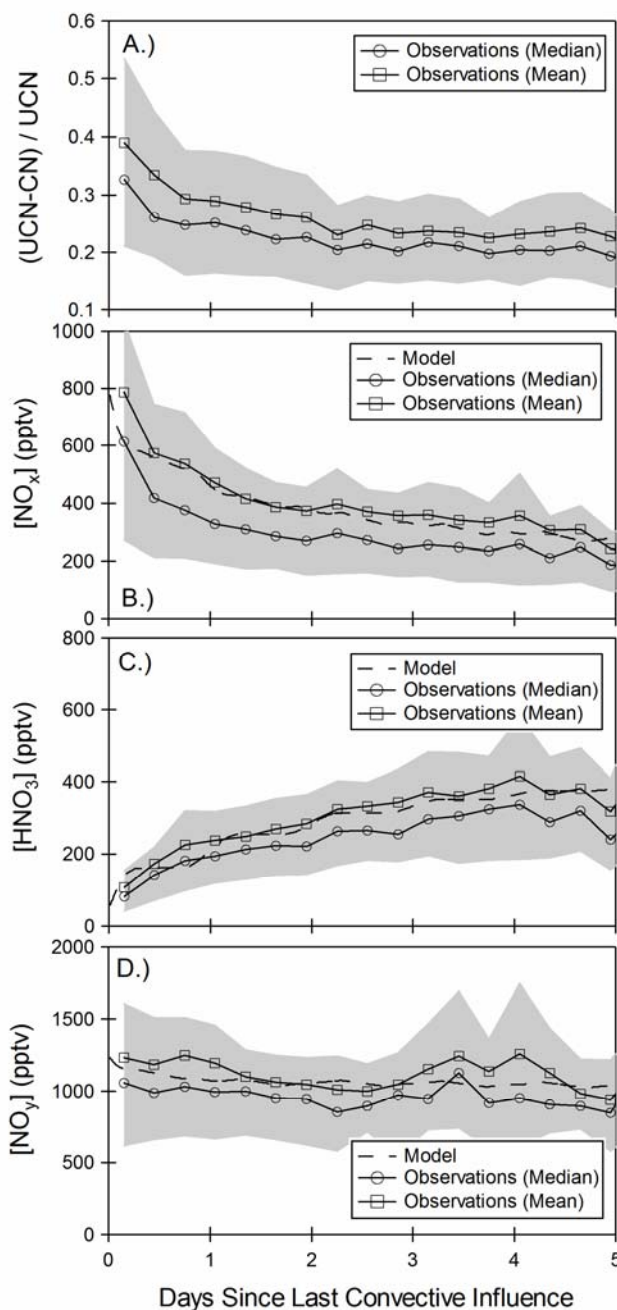
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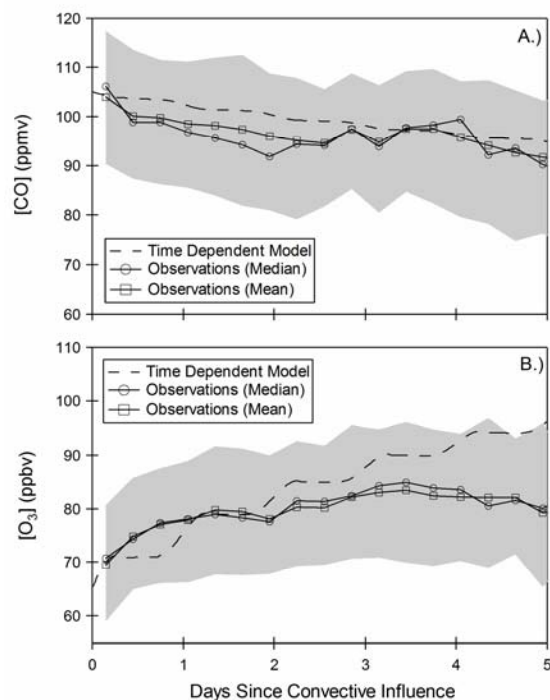
4 **Figure 2:** The observed deviation of the  $\text{NO}_x$  to  $\text{HNO}_3$  ratio from steady-state as a  
 5 function of altitude in the UT. The mean values within 500 m vertical bins are shown  
 6 with circles ( $\circ$ ). The steady-state  $\text{NO}_x$  to  $\text{HNO}_3$  ratio was calculated from measured  $\text{NO}_x$ ,  
 7  $\text{OH}$  and  $J_{\text{HNO}_3}$  and includes the rain out parameterization of Giorgi and Chameides  
 8 (1985).



**Figure 3:** *top panel* Time series of measurements taken in the vicinity of recent convective activity on 11 August 2004 between 5 and 9 km. Panel A suggests the sampling of a series of fresh convective plumes, indicated by a sharp increase in the  $NO_x/HNO_3$  ratio. Panels B & C depict coincident enhancements in  $SO_2$  and UCN ( $3nm > D_p > 10nm$ ) and coincident sharp drops in  $CO_2$ , indicative of the convective lofting of boundary layer air depleted in  $CO_2$ . The derived time since the sampled air-mass had been influenced by convection is shown in Panel D. *bottom panel* NLDN lightning hits (small dots) on the 10<sup>th</sup> and 11<sup>th</sup> of August. The color-code represents the time of the hit (hours) prior to aircraft sampling. The DC-8 sampling location corresponding to measurements shown in Figure 1 is located on the Maine – New Brunswick border [46°N 67°W]. The two day back trajectory [ $\bullet$ ] (initialized at the point of the second convective plume shown in Panel A) is also color-coded by time prior to DC-8 sampling (dots with black edges).

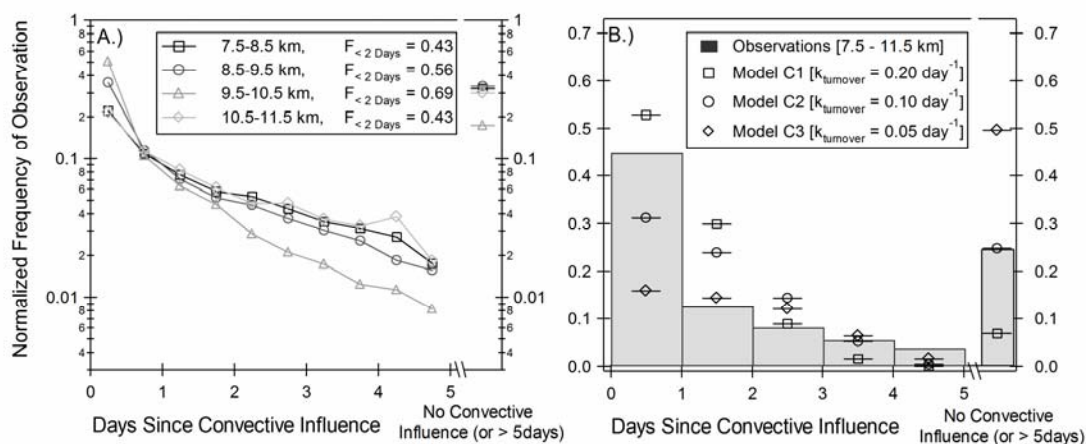


**Figure 4:** Observations of the fraction of ultra-fine condensation nuclei [number density of aerosol (3-10 nm) / total aerosol number density] (Panel A),  $\text{NO}_x$  (Panel B),  $\text{HNO}_3$  (Panel C) and  $\text{NO}_y$  (Panel D) as a function of modeled time since convective influence. The mean ( $-\square-$ ) and median ( $-\circ-$ ) of the observations, within 8 hour bins, is shown along with interquartile range (shaded region). Results from the time-dependent box model, initialized at 10 km and 12PM, are shown with dashed lines for the gas phase species (B-D).



**Figure 5:** Observations of CO (Panel A) and Ozone (Panel B) as a function of modeled time since cloud processing in the UT. The mean (-□-) and median (-○-) of the observations, within 8 hour bins, is shown along with the interquartile range (shaded region). Results from the time-dependent box model, initialized at 10 km and 12PM, are shown with dashed lines.

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**Figure 6:** *left panel* Normalized frequency distribution in the time since convective influence, as calculated from observations of the  $\text{NO}_x$  to  $\text{HNO}_3$  ratio made during the summer of 2004. Calculations are separated into 1 km altitude bins (ranging from 7.5-11.5 km). The fraction of air that had been influenced by convection within the past two days ( $f_{< 2 \text{ days}}$ ) is included in the figure legend. *right panel* Comparison of observed frequency distribution (7.5-11.5 km) with various modeled representations of the convective turnover rate.

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51. Net  $\Delta\text{O}_3$  of  $0 \text{ nmol mol}^{-1} \text{ day}^{-1}$  could be achieved if the air parcel: i.) subsided to where  $\text{H}_2\text{O}$  abundances are large enough to provide a sink of  $\text{O}_3$  through  $\text{O}^1\text{D}$  that balanced production from  $\text{NO}+\text{HO}_2$  ( $\sim 6 \text{ km}$ ), ii.) entrained air containing lower  $\text{O}_3$  mixing ratios or iii.) contained additional  $\text{O}_3$  loss terms beyond  $\text{NO}_x$ ,  $\text{HO}_x$ ,  $\text{H}_2\text{O}$  (via  $\text{O}^1\text{D}$  removal). To match the deviation between the model and measurement, we would require an additional  $2\text{-}3 \text{ nmol mol}^{-1} \text{ day}^{-1}$  of chemical ozone loss. In order for mixing to explain the deviation, air of lower  $\text{O}_3$  would need to be mixed into the air parcel. As shown in Fig. 5b, the only air in the UT containing significantly less  $\text{O}_3$  is that pumped directly from the PBL. While mixing fresh and aged outflow could help to explain the discrepancy in  $\text{O}_3$ , it is inconsistent with the observed decay in CO at long time (2-5 days).
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past 2 days. In Fuelberg et al., the authors determine that when considering the entire INTEX-NA sampling domain (both in space and time), convection was present in 12.5% of the grid points. This is substantially smaller than the percent of observations within 6 hours of convection (21.4%), suggesting that the DC-8 had a positive bias toward sampling fresh convection. This bias is reflected in the sharp drop in population between day 1 and 2 as shown in Fig. 6a. Correcting for this bias has little effect on our assessment of the fraction of air less than 2 days old, lowering our results from 0.43, 0.56, 0.69, and 0.43 to 0.38, 0.50, 0.62, and 0.39 at 8, 9, 10, and 11 km respectively.

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60. The authors thank the flight and ground crews of the NASA DC-8 Aircraft and the entire INTEX-NA science team for their contributions during the 2004 intensive field campaign. We thank A.M. Thompson, I. Folkins, M.G. Lawrence and D. Allen for helpful discussions, T. Kucsera for help with the GEOS-4 calculations and W.H. Brune and X. Ren for OH and HO<sub>2</sub> data. NLDN data was collected by Vaisalia-Thunderstorm and provided to the INTEX Science Team by the Global Hydrology Resource Center (GHRC) at NASA Marshall Space Flight Center (MSFC). Work at U.C. Berkeley was supported under NASA grants NNG05GH196 and NAG5-13668. The INTEX-NA field program was supported by the NASA-ESE Tropospheric Chemistry Program.

## Supplemental Online Information

### 1. INTEX-NA Experiment Description and Instrument Descriptions

The Intercontinental Chemical Transport Experiment – North America (INTEX-NA) took place between 1 July and 14 August 2004. Research flights were conducted out of Dryden Flight Research Center (Edwards AFB, CA), Mid-America Airfield (Mascoutah, IL); and PEASE International Trade-Port (Portsmouth, NH). Figure S1 depicts the vertical and horizontal extent of research flights conducted aboard the NASA DC-8 during INTEX-NA (1). DC-8 flight tracks are shown in the left panel of Figure 1 and the number of samples (10 second averaging time) in 1km vertical bins are shown in the right panel. *In situ* observations relevant to this study include; NO<sub>2</sub>, HNO<sub>3</sub>, OH, O<sub>3</sub>, CO, CO<sub>2</sub>, SO<sub>2</sub> and Ultra-fine Condensation Nuclei (UCN). Table S1 describes the detection threshold, uncertainty and time response for each measurement used in this analysis.

**NO<sub>2</sub> LIF Instrument** The NO<sub>2</sub> instrument flown aboard the DC-8 was described in detail by Thornton et al. (2), with specifics of the jet-expansion described by Cleary et al. (3). Briefly, NO<sub>2</sub> fluorescence is detected at 1Hz following excitation of a specific jet-cooled rovibronic transition in NO<sub>2</sub> at 585 nm. Red-shifted fluorescence is imaged at 90° onto an air cooled photomultiplier tube that is both optically and temporally filtered to remove laser scatter. NO<sub>2</sub> mixing ratios are calculated directly from fluorescence counts following calibration to NO<sub>2</sub> gas standards and measurements of the instrument zero from compressed air mixtures containing zero NO<sub>2</sub>. Selectivity to NO<sub>2</sub> is demonstrated by tuning on and off of a specific NO<sub>2</sub> resonance, where the difference in observed fluorescence at the two different frequencies is attributed solely to NO<sub>2</sub>. We calculate a

NO<sub>2</sub> detection threshold of 8 pptv in 10 seconds at the surface and 25 pptv in 10 seconds at the aircraft ceiling (12.5 km).

## **2. 0-D Time Dependent Model**

The chemical evolution of convective outflow was modeled using a 0-D time dependent model. The model was initialized with chemical conditions, altitudes and detrainment times consistent with observations of fresh convection made during INTEX-NA. As time propagates in the model, we calculate the production and loss of O<sub>3</sub>, CO, NO, NO<sub>2</sub>, NO<sub>3</sub>, N<sub>2</sub>O<sub>5</sub>, PAN, HO<sub>2</sub>NO<sub>2</sub>, HNO<sub>3</sub>, OH, HO<sub>2</sub>, RO<sub>2</sub>, H<sub>2</sub>O<sub>2</sub>, CH<sub>3</sub>OOH, H<sub>2</sub>CO and C1-C6 Hydrocarbons for 20 days following cloud detrainment. The conversion of NO<sub>x</sub> to HNO<sub>3</sub> in the outflow region is used as an indicator of time since convection. Figure S2 depicts the results of a single run initialized at 10km with a noon detrainment time. Initial conditions correspond to [NO<sub>x</sub>]<sub>i</sub> = 800 pptv, [O<sub>3</sub>]<sub>i</sub> = 65 ppbv and [CO]<sub>i</sub> = 105 ppbv. Rapid conversion of NO<sub>x</sub> to HNO<sub>3</sub> is observed during the first few days as the system approaches steady-state. In this analysis we assume: i.) HNO<sub>3</sub> is scavenged with unit efficiency in deep convection, ii.)  $\gamma_{\text{N}_2\text{O}_5} = 0.01$  and iii.) HNO<sub>3</sub> is not scavenged by aerosols (or rain) following injection into the UT. All kinetic rates used in this analysis were taken from the NASA JPL Chemical Kinetics and Photochemical Data for Use in Atmospheric Studies, Evaluation Number 14 (4).

### **2.1 Treatment of OH and HO<sub>2</sub>**

The calculated time since convective detrainment is directly coupled to the HO<sub>x</sub> budget through the daytime NO<sub>x</sub> sink to HNO<sub>3</sub> via reaction with OH. As in other model

descriptions of the UT during INTEX-NA (5, 6), our unconstrained model over-estimates OH by nearly a factor of two in the UT and under-estimates HO<sub>2</sub> by a similar amount. Due to the direct dependence of our timing indicator on HO<sub>x</sub>, we constrain the mixing ratios of OH and HO<sub>2</sub> to the observed values as a function of NO<sub>x</sub> and altitude. Figure S3 depicts the modeled mixing ratios of OH and HO<sub>2</sub> (lines), constrained to the observations (dots), as a function of NO<sub>x</sub> and SZA at 10km. The observed OH is a strong function of NO<sub>x</sub>, while observations of HO<sub>2</sub> remain insensitive to NO<sub>x</sub>. Constraints for OH and HO<sub>2</sub> were derived independently for each 1km altitude bin. Constraining OH and HO<sub>2</sub> to the observations increases the time required for the NO<sub>x</sub>-HNO<sub>3</sub> system to reach steady-state (by slowing the rate of OH + NO<sub>2</sub>) and enhances the modeled O<sub>3</sub> production in the outflow region (by speeding up the rate of HO<sub>2</sub>+NO).

## 2.2 Calculation of Time since Convection

The time since a sampled air-mass had been cloud processed is calculated by applying the mapping of time to NO<sub>x</sub>/ HNO<sub>3</sub> derived in the box model to the observed NO<sub>x</sub> to HNO<sub>3</sub> ratio. Figure S4 depicts the best-fit relation between the modeled NO<sub>x</sub> to HNO<sub>3</sub> ratio and time since cloud processing at 10km. This function is calculated at 1km increments from 6-12km and applied to the measured NO<sub>x</sub> to HNO<sub>3</sub> ratio.

## 2.3 Model Assumptions and Uncertainty

In order to access the uncertainty in the calculated time, we ran the time-dependent model under various different conditions encountered during INTEX-NA (e.g. [NO<sub>x</sub>]<sub>i</sub> (0.2-3.0 ppbv), [O<sub>3</sub>]<sub>i</sub> (40-100 ppbv), [CO]<sub>i</sub> (80-150 ppbv), detrainment time (noon, 4PM,

midnight), altitude (6-12km) and time of year (June-September). As illustrated in figure S4, the NO<sub>x</sub> to HNO<sub>3</sub> ratio has good resolution (i.e. large rate of change per unit time) during the first five days following convection. Beyond five days small changes in NO<sub>x</sub>/HNO<sub>3</sub> correspond to larger changes in the derived time. From the variance in the calculated time of individual model runs, we estimate the uncertainty in our modeled time to be ±6 hours at 1 day, ±12 hours at 2 days and ±1 day at 4 days. In addition, the INTEX-NA sampling domain did not permit frequent measurement of aged convection (>5 days). For these reasons we limit our analysis to the first five days following convection.

## 2.4 Treatment of Mixing

The mixing rate was determined by iterating the model until we had closure between the observed and modeled time evolution of a suite of long-lived species (e.g. CO, CH<sub>4</sub>, CH<sub>3</sub>OH and others). The modeled time rate of change of species X is determined as:

$$\frac{d[X]}{dt} = P(X) - L(X) - k_{\text{dilution}} ([X] - [X]_{\text{Background}})$$

where P(X) represents the chemical production of species X, L(X) represents the chemical loss of species X and k<sub>dilution</sub> is the mixing rate of the convective plume with background UT. We find this mixing term to be on average 0.05 ± 0.02 day<sup>-1</sup>. That is, after 5 days, the plume still has 75% of its original contents. While individual convective plumes may mix faster (or slower) than this, the aggregate mixing rate of all the sampled plumes can be described by this rate. This rate supports the conclusion that over the course of 5 days, the convective plume remains relatively isolated from the background

UT. Background mixing ratios used in the dilution calculation were taken as the mean observed UT mixing ratios outside of fresh convective plumes.

Due to subsidence of convectively lofted air parcels following injection, our calculated time represents a lower bound for age as the chemical clock speeds up (due to  $\text{NO}_x$  repartitioning) as the parcel descends in altitude. However, this is a relatively small effect as calculated subsidence rates are approximately  $35 \text{ hPa day}^{-1}$  (7).

### **3.0 Comparison of Chemical and Meteorological Convective Influence Calculations**

The results presented here provide a chemical constraint on the rate at which the UT over the continental US is influenced by convection during summer. In addition to the meteorological analysis of Fuelberg et al., presented in this manuscript, Thompson et al. assessed the effects of convection on the  $\text{O}_3$  budget in the UT, over eastern North America, during the summer of 2004. Using results from the INTEX Ozone Sonde Network Study (IONS), the authors conclude that 10-15% (lower-limit) of the below tropopause  $\text{O}_3$  can be attributed to the interaction of regional pollution with convection and lightning (8).

The INTEX-NA sampling period (June-August) and region (Eastern North America) is characterized by intense lightning activity. Cooper et al. calculated that 13% of the global annual lightning  $\text{NO}_x$  emissions occurred between  $108^\circ \text{ W} - 18^\circ \text{ E}$  and  $18^\circ - 72^\circ \text{ N}$  the between June 21 and August 15, 2004 (9). In terms of lightning flash counts, Hudman et al., conclude that 2004 was typical (within 20% of the mean) of the past 5

years (6). Persistent frontal passages prevented the formation of stagnant high pressure systems, typically observed during the summer over the northeastern United States (10). These frequent passages led to both record low temperatures and number of O<sub>3</sub> exceedances in the northeast (11). In contrast UT/LS O<sub>3</sub>, as observed from the IONS network, was comparable to the climatology (11).

#### 4.0 Measurement Uncertainties

In this analysis we calculate NO<sub>x</sub> from observations of NO<sub>2</sub>, O<sub>3</sub>, HO<sub>2</sub> and photolysis rates measured directly on the DC-8. NO was measured directly on the DC-8 via a commercial grade chemiluminescence detector. The sensitivity of the chemiluminescence instrument (detection threshold > 50 pptv) and long integration time (1 minute) prohibited its use in these calculations. NO was calculated from steady-state using the following equation:

$$[NO]_{Steady-State} = \frac{J_{NO_2} [NO_2]}{k_{NO+O_3} [O_3] + k_{NO+HO_2} [HO_2]}$$

We calculate the accuracy of the derived NO to be better than  $\pm 30\%$  based on the propagation of the individual errors used on the calculation. Observations of Nitrogen Dioxide, Ozone and JNO<sub>2</sub> made during the INTEX-NA campaign were compared directly with measurements made aboard the NOAA WP-3D during a series of in-flight comparisons. During these experiments the principal individual components (NO<sub>2</sub>, O<sub>3</sub> and JNO<sub>2</sub>) showed agreement to within their stated instrumental uncertainty.

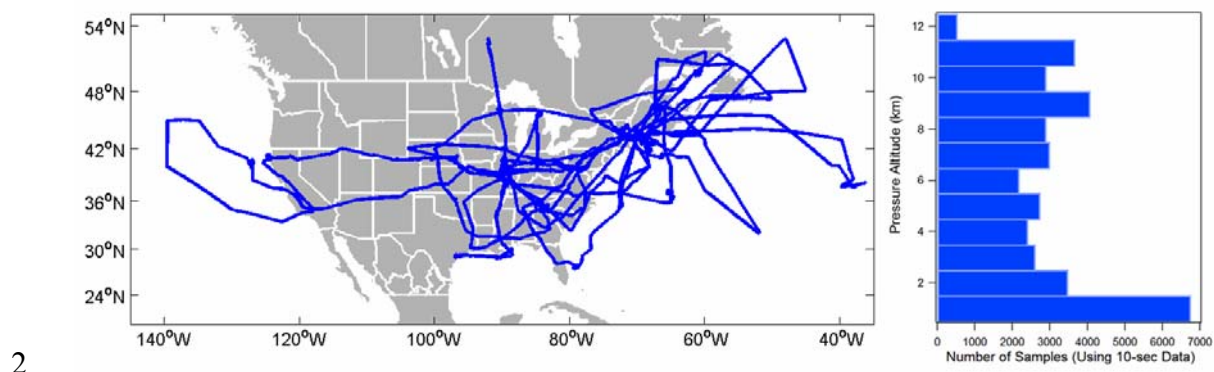
1 In Figure S6 we compare the measured  $\text{NO}_x$  (using the chemiluminescence and the Laser  
2 Induced Fluorescence measurements) and  $\text{NO}_x$  calculated from steady-state for the entire  
3 INTEx-NA campaign. The steady-state  $\text{NO}_x$  agrees with the measured  $\text{NO}_x$  to within the  
4 calculated uncertainty when averaged to 1-minute and divided into 1km vertical bins.  
5 The observed upper tropospheric  $\text{NO}_x$  concentrations during INTEx-NA are on average  
6 much higher than previous reported on intensive aircraft field campaigns over North  
7 America. During the SUCCESS campaign, Jaegle et al. report mean NO concentrations  
8 of  $0.030 \pm 0.022$  ppbv and  $0.061 \pm 0.045$  ppbv, for the altitude ranges of 8-10 km and 10-  
9 12 km, respectively (12). However, it must be noted that the scientific objective of  
10 SUCCESS was the sampling of aircraft exhaust and contrails, thus the values reported in  
11 Jaegle et al., were filtered to exclude fresh aircraft exhaust ( $\text{CO}_2 > 368$  ppmv and  $\text{NO} >$   
12  $300$  pptv). Crawford et al., report mean NO concentrations of 0.1 ppbv for observations  
13 made between 6-12 km during the SONEX campaign during the fall of 1997 over the  
14 North Eastern United States and the North Atlantic (13). Neither of these studies  
15 provides a direct comparison to the INTEx-NA data-set as SUCCESS was conducted in  
16 the spring and SONEX in the fall, while peak lightning and convection occurs over the  
17 continental North America during summer. The best comparison comes from  $\text{NO}_x$   
18 measured aboard a commercial passenger aircraft during the NOXAR program between  
19 1995 and 1997. Brunner et al. report a mean UT  $\text{NO}_x$  concentration of 0.4 ppbv during  
20 June, July and August over North America (14). As seen in Figure S6, our observations  
21 during the summer of 2004 are consistent with this.

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1 We use the Caltech CIMS  $\text{HNO}_3$  due to its fast time response (5 seconds as compared to  
2 105 seconds for the UNH Mist Chamber Technique) and the UNH MC results when the  
3 fast  $\text{HNO}_3$  was unavailable. To account for the systematic bias between the two  
4 observations in the UT ( $[\text{HNO}_3]_{\text{UNH}} = 0.6 \times [\text{HNO}_3]_{\text{Caltech}}$ ), we scale both the CIMS and  
5 MC observations to split the difference between the two measurements (i.e. we increase  
6  $[\text{HNO}_3]_{\text{UNH}}$  by 20% and decrease  $[\text{HNO}_3]_{\text{CIT}}$  by 20%).

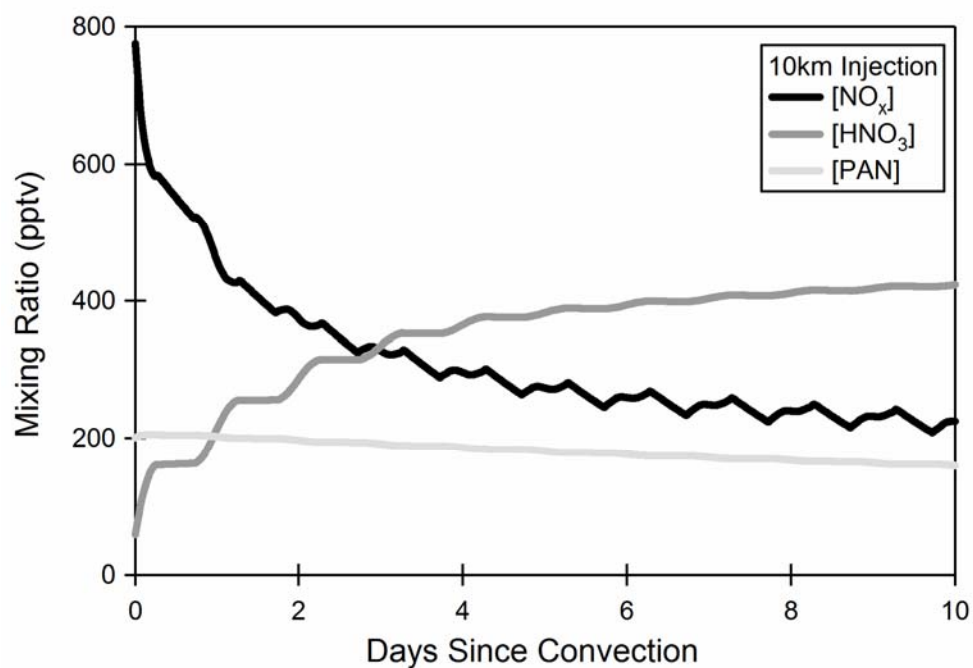
7  
8 Due to the observed systematic bias, the choice of which  $\text{HNO}_3$  measurement to use in  
9 the analysis has the potential to complicate our conclusions. To address these effects we  
10 have conducted the analysis using Nitric Acid as measured from: i.) the Mist Chamber –  
11 Ion Chromatography Instrument, ii.) the Chemical Ionization Mass Spectrometer and iii.)  
12 the scaled difference between the two techniques. The results are compared in Figures  
13 S7 and S8. Figure S7 shows the normalized frequency distribution in the time since  
14 convective influence, as calculated from observations of the  $\text{NO}_x$  to  $\text{HNO}_3$  ratio made  
15 during the summer of 2004. Figure S8 depicts the fraction of air that had been influenced  
16 by convection within the past two days ( $f < 2 \text{ days}$ ) as a function of altitude. Calculations  
17 derived from the Mist Chamber – Ion Chromatography Instrument result in a higher  
18 fraction of convectively influenced air when compared with calculations made using  
19 measurements from the CIMS instrument. When compared with the results shown in  
20 Figure 6B of the manuscript, on which our conclusions regarding the convective overturn  
21 rate are drawn from, calculations using either the CIMS, MC-IC or the scaled difference  
22 result in a convective overturn rate between 0.1 and 0.2  $\text{day}^{-1}$ .

1 **2. Supplemental Figures**



3 **Figure S1:** *left panel* INTEx-NA flight tracks made between 1 July 2004 and 14 August  
4 2004 aboard the NASA DC-8. *right panel* Number of samples (using 10-sec averaged  
5 data) within 1km altitude bins between 0-12 km during the entire campaign.

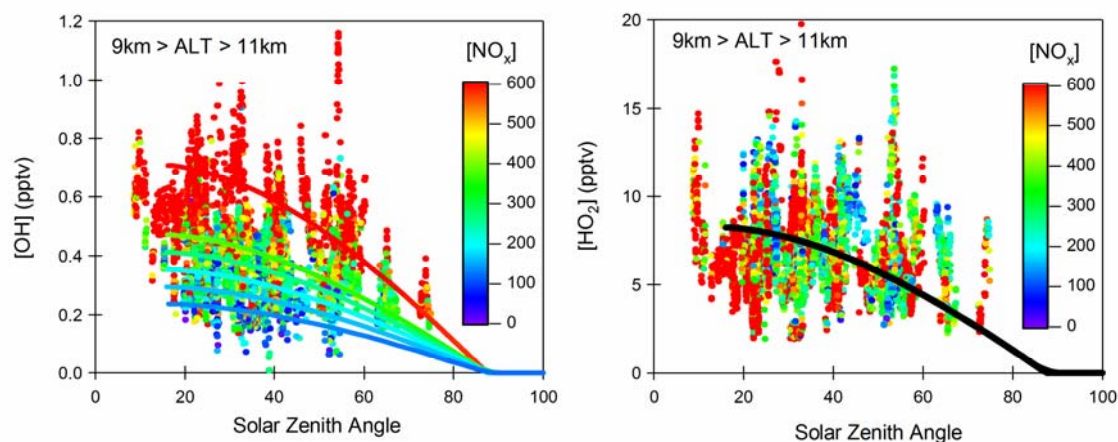
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3 **Figure S2:** Time-dependent model illustrating the conversion of NO<sub>x</sub> to Nitric Acid in  
 4 the days subsequent to a cloud processing event occurring at 10km. The above model  
 5 was initialized at 12PM local time at 30°N in August using [NO<sub>x</sub>]<sub>i</sub> = 800 pptv, [CO]<sub>i</sub> =  
 6 105 ppbv at [O<sub>3</sub>]<sub>i</sub> = 65 ppbv.

1

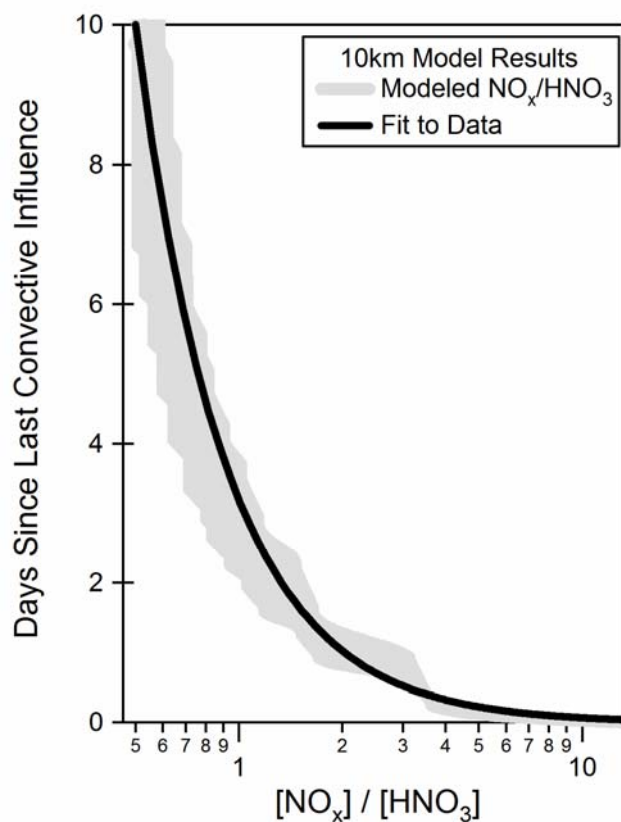


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4 **Figure S3:** Model representation of OH (left panel) and HO<sub>2</sub> (right panel) as a function  
 5 of SZA and  $[NO_x]$ . Model results (solid lines) are shown on top of the *in situ*  
 6 observations (dots). The model was initialized at noon at 10km with  $[NO_x]_i = 800$  pptv,  
 7  $[CO]_i = 105$  ppbv at  $[O_3]_i = 65$  ppbv. Observations shown were taken aboard the DC-8  
 8 between 9 and 11 km.

1

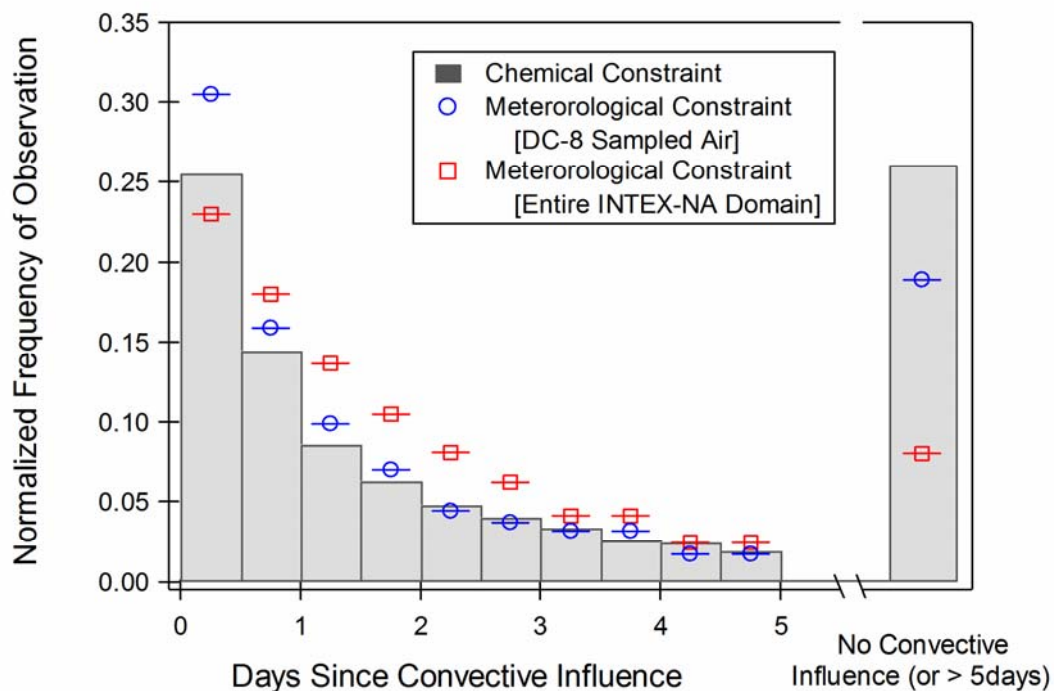


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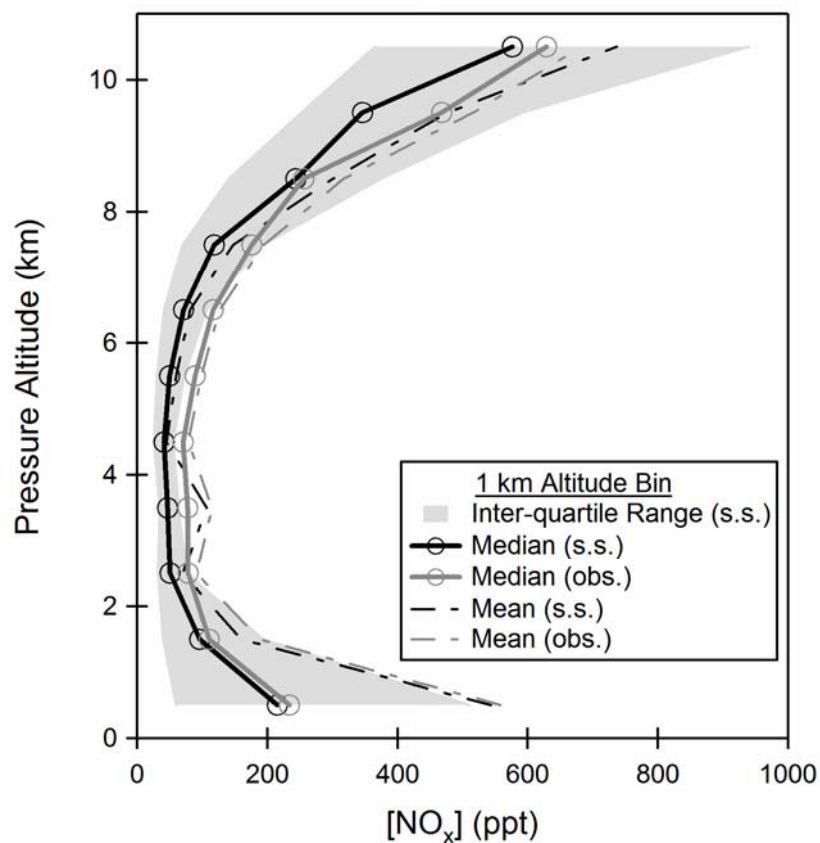
3

4

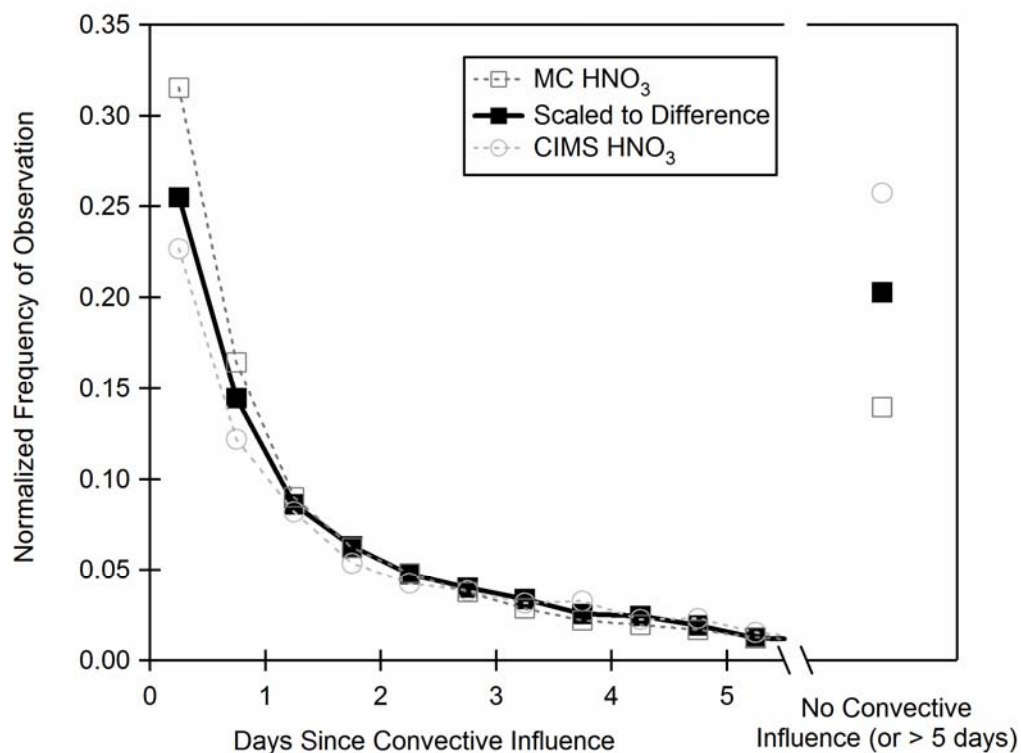
5 **Figure S4:** Observed  $\text{NO}_x$  to  $\text{HNO}_3$  ratios are converted to a time since last convective  
 6 influence using the best fit equation relating the  $\text{NO}_x/\text{HNO}_3$  ratio to time as calculated  
 7 using the time-dependent model in 1km altitude bins from 7.5-11.5 km. The above  
 8 equation is valid for pressure altitudes between 9.5 and 10.5 km.



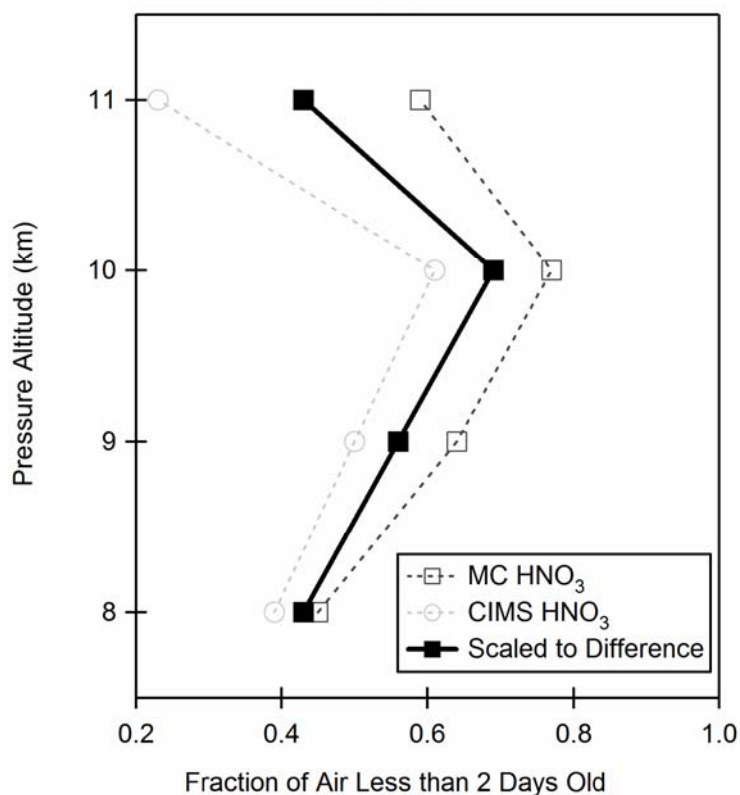
**Figure S5:** Comparison of chemical (grey bars) and meteorological constraints (-○-, -□-) on convective influence during INTEX-NA. Convective influence on air *sampled* by the DC-8 is shown with blue circles (-○-), while convective influence on the entire INTEX-NA domain is shown with red squares (-□-).



**Figure S6:** Comparison of observed (grey lines) and steady-state (black lines) NO<sub>x</sub> for the entire INTEX-NA field campaign. The shaded region represents the interquartile range of the calculated NO<sub>x</sub>.



**Figure S7:** Normalized frequency distribution in the time since convective influence, as calculated from observations of the  $\text{NO}_x$  to  $\text{HNO}_3$  ratio made during the summer of 2004. Calculations were conducted using Nitric Acid as measured from: i.) the Mist Chamber – Ion Chromatography Instrument ( $-\square-$ ), the Chemical Ionization Mass Spectrometer ( $-\circ-$ ) and the scaled difference ( $-\blacksquare-$ ).



**Figure S8:** Fraction of air that had been influenced by convection within the past two days ( $f_{<2 \text{ days}}$ ) as a function of altitude. Calculations were conducted using Nitric Acid as measured from: i.) the Mist Chamber – Ion Chromatography Instrument ( $\square$ ), the Chemical Ionization Mass Spectrometer ( $\circ$ ) and the scaled difference ( $\blacksquare$ ).

1

Species	Measurement Technique	Detection Threshold	Accuracy	Time Response	Reference
NO <sub>2</sub>	LIF <sup>1</sup>	8 pptv / 10 sec	± 10% 1σ	1 Hz	(2, 3)
HNO <sub>3</sub>	CIMS <sup>3</sup>	10 pptv / 0.5s	± 30%	0.5s sample every 5s	(15)
	Mist Chamber – IC <sup>4</sup>	5 pptv / 105 sec		105 sec	(16)
OH	LIF <sup>1</sup>	0.01 pptv	± 32% 2σ - 1 min	20 sec	(17)
O <sub>3</sub>	Chemiluminescence	Precision = ±0.8pptv, ±1% of reading	± 2 ppbv, ±3%	1 Hz	(18)
CO	IR-Absorption	Precision = ±1ppbv, ±1.5% of reading	± 1.4 ppbv, ± 2.6% 2σ	1 Hz	(19)
CO <sub>2</sub>	IR-Absorption	Precision < 0.07 ppmv	± 0.25 ppmv	1 Hz	(20)
UCN	TSI CN counter <sup>5</sup>	± 10%	± 10%	1 Hz	(21)
J <sub>NO2</sub>	Actinic Flux Spectroradiometer	4.1 x 10 <sup>-7</sup> Precision = ±4.5%	± 8%	1 Hz	(22)

2

<sup>1</sup>LIF – Laser Induced Fluorescence

3

NO<sub>2</sub> detection threshold is 8 pptv / 10 sec at 760 Torr (ground) and 20 pptv / 10 sec at 200 Torr (10 km)

4

<sup>3</sup>CIMS – Chemical Ionization Mass Spectrometry

5

<sup>4</sup>IC – Ion Chromatography

6

<sup>5</sup>UCN (Ultra-fine Condensation Nuclei) was obtained by the difference of the UCN (D<sub>p</sub>>3nm, TSI 3025) and CN (D<sub>p</sub>>10nm, TSI 3010) Condensation Nuclei (CN) instruments.

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**Table S1:** Detection thresholds, measurement uncertainty and time response of the *in situ* measurements used in this study. (Note: ppmv = μmol mol<sup>-1</sup>, ppbv = nmol mol<sup>-1</sup> and pptv = pmol mol<sup>-1</sup>)

10

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