specimens after water immersion at 40, 60, and 80 °C. Cured EVA rubber containing no glass beads gained only 1 weight percent water. This shows that the immense weight gain of other specimens is due to water absorption at the glass/elastomer interface. This was also verified by FTIR (Fourier Transform Infrared) spectroscopy.

The amount of water absorption in specimens containing glass beads is dramatically changed by the presence of the primer, particularly in the specimens immersed in water at the 80 °C temperature. Unprimed specimens aged in water at 60 °C for 2,000 hours showed a 2015percent increase in weight, whereas primed glass bead specimens gained only 35 percent. Reduction in tensile strength also followed primed and unprimed specimens, and increasing exposure times and temperatures.

Importantly, this method can provide semi-quantitative measurement of the bond resistance to water by measuring the time to the sudden increase in weight. This effect is due to degradation of the primer chemistry at the interface. The longer the time to weight increase, the more effective is the primer and the more water resistant is its chemical bond.

Duplicate specimens of those reported in the first table were dried to constant weight and retested for tensile strength. This was done to determine the recoverability of hydrothermal attack after water removal. Specimens containing primed glass microbeads recovered their initial properties almost entirely, whereas the unprimed glass beads, especially in the 80 °C water exposure, deteriorated too badly to be tested.

Table 2 presents the results as percentage retention of control values: This method offers a simplified way of (a) determining the effectiveness of primers used for bonding elastomers to glass, and perhaps other substrates, in environments with humidity or water exposure; (b) a semi-quantitative technique for determining bond stability; and (c) a method for determining the long term durability and onset of hydrolytic attack at the interface. This method has the advantages of requiring very few specimens and only a laboratory balance to perform measurements.

This work was done by Paul White of Caltech for NASA's Jet Propulsion Laboratory. Further information is contained in a TSP (see page 1). NPO-43912

## Complex Multifunctional Polymer/Carbon-Nanotube Composites CNTs are treated and incorporated into composites to obtain enhanced properties.

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A methodology for developing complex multifunctional materials that consist of or contain polymer/carbon-nanotube composites has been conceived. As used here, "multifunctional" signifies having additional and/or enhanced physical properties that polymers or polymer-matrix composites would not ordinarily be expected to have. Such properties include useful amounts of electrical conductivity, increased thermal conductivity, and/or increased strength. In the present methodology, these properties are imparted to a given composite through the choice and processing of its polymeric and CNT constituents.

The methodology involves utilization of CNTs in any or all of several ways:

- Coating the CNTs to impart desired properties — for example, coating them with electrically and/or thermally conductive polymers, which could be dissolved in solvents;
- Incorporating uncoated or coated CNTs into a polymeric matrix, possibly in such



This **Complex Composite** consists of a fabric of coated UHMWPE fibers in a matrix material consisting of an epoxy/CNT composite.

a manner as to improve the properties of the CNTs, the matrix, and/or the resulting composite; and/or

 Using a polymer/CNT composite as the matrix ingredient of a complex composite that includes any of a variety of other fibrous reinforcing materials.

The figure is a simplified illustration of an example of such a complex composite. In this case, a fabric made of coated ultra-high-molecular-weight polyethylene (UHMWPE) fibers is embedded in a matrix that is, itself, a composite of CNTs in an epoxy matrix. Typically, heretofore, such a composite would be designed and fabricated to obtain high strength, would not contain CNTs, and would be electrically insulating and, to some extent, thermally insulating. By incorporating a suitable quantity of CNTs, one can obtain enough electrical conductivity to drain off excess static electricity to prevent static discharge or to render the composite effective as a barrier against electromagnetic interference, and to obtain usefully large degrees of thermal conductivity and thermal stability, all without sacrificing mechanical strength.

This work was done by Pritesh Patel, Gobinath Balasubramaniyam, and Jian Chen of Zyvex Corp. for Marshall Space Flight Center. For further information, contact Sammy Nabors, MSFC Commercialization Assistance Lead, at sammy.a.nabors@nasa.gov. Refer to MFS-32355-1.