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4	The Far Infrared Earth
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7	J. Harries <sup>1</sup> , B. Carli <sup>2</sup> , R. Rizzi <sup>3</sup> , C. Serio <sup>4</sup> , M. Mlynczak <sup>5</sup> , L. Palchetti <sup>2</sup> , T. Maestri <sup>3</sup> , H.
8	Brindley <sup>1</sup> , and G. Masiello <sup>4</sup> .
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 <sup>&</sup>lt;sup>1</sup> Blackett Laboratory, Imperial College, London, SW7 2AZ, UK
 <sup>2</sup> Istituto di Fisica Applicata "Nello Carrara", CNR, 50019 Sesto Fiorentino, Firenze, Italy
 <sup>3</sup> Atmospheric Dynamics Group, Dipartimento di Fisica, Universita di Bologna, 40127, Bologna, Italy
 <sup>4</sup> Dipartimento di Ingegneria e Fisica dell' Ambiente, Universita di Basilicata, Potenza, Italy
 <sup>5</sup> NASA Langley Research Center, Hampton, Virginia, USA

- 1 Abstract.
- 2

3 The paper presents a review of the far infrared (FIR) properties of the Earth's atmosphere, and 4 the role of these properties in climate. These properties have been relatively poorly 5 understood, and it is one of the purposes of this review to demonstrate that, in recent years, 6 we have made great strides in improving this understanding. Seen from space, the Earth is a 7 cool object, with an effective emitting temperature of about 255 K. This contrasts with a 8 global mean surface temperature of  $\sim 288$  K, and is due primarily to strong absorption of 9 outgoing longwave energy by water vapour, carbon dioxide and clouds (especially ice). A 10 large fraction of this absorption occurs in the FIR, and so the Earth is effectively a FIR planet. 11 The FIR is important in a number of key climate processes, for example the water vapour and 12 cloud feedbacks (especially ice clouds). The FIR is also a spectral region which can be used to 13 remotely sense and retrieve atmospheric composition in the presence of ice clouds. Recent 14 developments in instrumentation have allowed progress in each of these areas, which are 15 described, and proposals for a spaceborne FIR instrument are being formulated. It is timely to 16 review the FIR properties of the clear and cloudy atmosphere, the role of FIR processes in 17 climate, and its use in observing our planet from space.

18

19 [225 words]

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#### 1 1. Introduction

2 This paper reviews our understanding of the long wave portion of the spectrum of the Earth's 3 infrared emission to space. In all, this emission extends approximately from wavelengths of about 1 mm to about 4  $\mu m$  (or those wavenumbers between about 10 and 2500  $\text{cm}^{\text{-1}}$ ) and 4 5 includes the relatively unexplored part of the spectrum known as the far infrared (hereafter 6 abbreviated to "FIR") and the better known mid infrared, the "MIR". The key feature of the 7 FIR is that it is the spectral range characterised by transitions between pure rotational, or very 8 low vibrational energy states, and therefore it is a cool, low energy part of the spectrum. What 9 is our definition of the "FIR"? There is actually no precise definition of where the boundary 10 of the FIR occurs, nor do we need, or encourage one. In this review, we will accept whatever 11 definition the author has used in any particular piece of work that we are discussing, but for 12 the sake of orientation we might take a wavelength of about 15 µm, or a wavenumber of about 670 cm<sup>-1</sup>, as a rough boundary. The FIR, broadly, lies at longer wavelengths, or lower 13 14 wavenumbers than this boundary.

15

We argue that the FIR is actually highly characteristic of the Earth as a system seen from space, and that it is of fundamental significance for planetary energy balance considerations and therefore for understanding climate; it may also be of great practical value in application to remote sounding of the composition of the atmosphere from space, and the characterisation of clouds (cirrus in particular). Our review will, therefore, be focused on the peculiarities of the FIR in the broader context of the Earth's emission spectrum.

22

The Earth is a cool object in space: this is the prime reason why the FIR is so important to an understanding of the climate system. If we measure the brightness temperature of the Earth (that is the temperature of an equivalent black body), with an instrument in space, we would measure something like 255 K [Harries, 1995; Andrews, 2000]. Differentiation of the Planck

1 radiation law to produce the Wien Displacement law [Goody, 1964] tells us that the peak energy from a black body at 255 K occurs at about 500 cm<sup>-1</sup>. Bearing in mind that the 2 3 temperature of the Earth's upper troposphere falls as low as 230 K or so, when we do the 4 calculations, we find that (depending on the latitude) up to 35% of the energy escaping to space, so cooling the Earth, is at wavenumbers below 500 cm<sup>-1</sup> [Sinha and Harries, 1995]. If 5 the "boundary" of the FIR is taken to be 667 cm<sup>-1</sup> then this maximum fraction increases to 6 7 nearer 45%. So, it is very important that climate models include accurate details of the FIR 8 properties of the atmosphere if they are to produce accurate predictions of climate change. 9

10 Another area in which it is important to understand the FIR spectrum is in remote sounding of 11 the Earth and its atmosphere from space. Remote sounding at mid IR (MIR) wavelengths, 12 shorter than 15 µm, has been extensively used for measurements of temperature and 13 composition from space, as also has the microwave region at wavelengths longer than about 1 14 mm. The region in between has not been exploited, because it seemed to be at a disadvantage 15 in terms of low photon energy with respect to the mid IR, while not benefiting, for technical 16 reasons, from the multiplexing or heterodyne techniques usable in the mm region because of 17 the availability of suitable mixers and monochromatic sources. However, as we shall see, the 18 FIR, observed using modern techniques, offers the possibility of accurate remote sounding of 19 water vapour and other constituents, in both clear skies, and also (because of the relatively 20 high transparency of clouds in the FIR) in the vicinity of optically thin clouds.

21

It is timely to present a review of the scientific background to, and current understanding of, the FIR spectrum of the Earth for two primary reasons: first, because the need to make ever more accurate predictions of climate change demands a more profound understanding of important FIR processes; and, second, because future satellite instrumentation may include spectrometers and other devices that operate in this less familiar range of the electromagnetic

1	spectrum, between the mid infrared and the microwave region. This paper presents a review
2	of the FIR properties of the Earth's atmosphere, and of those FIR processes which are
3	significant to the planetary energy balance, and also describes the application of the FIR to
4	remote sounding in the presence of clouds. It describes a range of current research in the
5	field, in which basic science, instrument development, and measurements from aircraft and
6	balloons are all being carried out.
7	
8	[Insert Box 1 near here]
9	
10	
11	Box 1: Tutorial on infrared spectroscopy and radiative transfer in the atmosphere
12	
13	Spectroscopy
14	Two classic text books which provide comprehensive introductions to the spectroscopy of the
15	atmosphere are Penner [1959] and Goody [1964]. We provide a brief introduction to the
16	subject here. In the absence of an atmosphere, radiative transfer for the Earth would be
17	simple. Incoming solar radiation would be partly reflected and partly absorbed by the surface,
18	depending on the 'albedo', or reflectance of the local surface. The absorbed energy would
19	warm the surface, and emission of thermal radiation, according to the Planck radiation law
20	[Goody, 1964], would take place, transferring energy back to space. In balance, the two
21	fluxes of energy would be equal, on a globally averaged basis.
22	
23	The effect of the atmosphere is to moderate these fluxes. Ignoring clouds to begin with, the
24	atmosphere contains molecules such as H <sub>2</sub> O, CO <sub>2</sub> , CH <sub>4</sub> , O <sub>3</sub> , and many others that are excited
25	by the available energy to move (translational energy), to rotate, to vibrate, and to excite their
26	constituent electrons to higher energy states. Since we are concerned here with the FIR, we

1	can restrict our view to the thermal or infrared region, for which electronic transitions are not
2	relevant, being of too high an energy. The vibrational and rotational energy levels of the
3	molecules are quantised [Rae, 2002], which means that energy levels can take up only certain
4	discrete values, and energy exchange can only occur at discrete energies. Since energy, E, is
5	proportional to frequency ( $E = hf$ , where <i>h</i> is Planck's constant, and <i>f</i> is frequency), we can,
6	therefore, say that the atmosphere will absorb or emit infrared radiation only at discrete
7	frequencies. It is common to use alternative but entirely equivalent parameters to describe the
8	energy scale, including frequency, $f$ (in Hertz), wavelength $\lambda$ , (usually in $\mu$ m), or
9	wavenumber, $\nu$ (in cm <sup>-1</sup> ). The three are related via the speed of light, c, by
10	
11	$c = f\lambda;$ $v = 1/\lambda = f/c.$ {B1}
	$\succ$
12	$v = 1/\lambda = f/c \ . \qquad \qquad$
12 13	$v = 1/\lambda = f/c \ . \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ $
	$v = l/\lambda = f/c$ . {B1} In fact, energy transitions are not precisely monochromatic, but are broadened by inter-
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13 14 15 16 17	In fact, energy transitions are not precisely monochromatic, but are broadened by inter- molecular and other effects, such as collisions, and the Doppler effect, so that if we magnify a single absorption or emission line, we find that it has the shape of a single resonance maximum, and a definite half width at half peak height, both of which depend on parameters

depending on molecular structure as well as molecular density, and atmospheric temperature
and pressure, the real atmospheric spectrum is a highly complex merger of individual lines,
and groups (bands) of lines from all of the absorbing molecules. In some regions of the
spectrum, the atmosphere may be highly transparent; in others, it is strongly absorbing, and
photons have a small mean free path before being absorbed.

2

3 Figure B1 shows the total atmospheric optical depth at the centre of all the water vapour and 4 carbon dioxide absorption lines throughout the whole of the IR; similar information is 5 provided in Figure B2 for the remaining key absorbing species. Each absorption line is 6 represented by one point, the optical depth (see below for definition) of the atmosphere at the 7 central wavenumber. Note that the y-axis scale is logarithmic, so that each division represents 8 a factor of 10. This method of display allows a comparison of absorption strengths across the 9 whole IR spectral region with each band colour coded. The dominant effect of water vapour in the mid and far IR may be seen, as can the important bands of  $CO_2$  at 15  $\mu$ m (667 cm<sup>-1</sup>) and 10 4.3  $\mu$ m (2325 cm<sup>-1</sup>). The lower optical depths between 750 and 1250 cm<sup>-1</sup>, and between 2400 11 and 3000 cm<sup>-1</sup> represent regions of lower absorption, or higher transparency, commonly 12 referred to as "atmospheric windows". 13

14

15 The strength of absorption can be measured by a parameter known as the mass absorption 16 coefficient,  $\alpha_v$ , which is related to the fractional transmittance,  $\tau_v$ , of a path of length *l* 

17 (maximum path L) and absorber density,  $\rho$ , by  $\tau_{\nu} = \exp\left[-\int_{0}^{L} \alpha_{\nu}(l)\rho(l)dl\right]$ . The term

18 
$$\chi_{\nu} = \left[\int_{0}^{L} \alpha_{\nu}(l)\rho(l)dl\right]$$
 is known as the optical depth

19

In the FIR, the mechanism by which a molecule may absorb or emit radiative energy is either
through rotations or vibrations of the molecule in question, because the energies of such
processes correspond to the FIR. Each type of motion is quantised in terms of the allowed
energy states and transitions between them. Rotational spectral lines are lower in energy (i.e.
lower in characteristic frequency: see eq. B1 above) than vibrational energy transitions. The

1 frequencies of energy state transitions in the latter case depend on changes to the shape of the 2 molecule, and the atomic weights of the constituents, as well as on 'selection rules' that 3 determine whether transitions between particular energy states are allowed by quantum 4 mechanics or not. The vibrations of a molecule are classified broadly by whether the 5 vibration represents a stretching of a particular inter-atomic bond, or a bending of the 6 molecule. In this paper we refer to different vibration classes as  $v_1$ ,  $v_2$  or  $v_3$ : these refer, in 7 the case of tri-atomic molecules like CO<sub>2</sub> and H<sub>2</sub>O<sub>2</sub> to symmetric stretching of the two C-O or O-H bonds  $(v_1)$ , bending of the molecule about the 'pivot' of the central atom (C or O) of the 8 9 three  $(v_2)$ , and asymmetric stretching of the two C-O or O-H bonds  $(v_3)$ . Goody [1964] 10 provides a clear explanation of these definitions. 11 12 Radiative transfer 13 14 The transfer of radiative energy within and through the Earth's atmosphere is treated in a 15 number of classic text books, including Chandrasekhar [1960], and Goody [1964]. The 16 radiative transfer of the atmosphere is controlled by the spectra of atmospheric molecules, 17 described above. In the infrared, the transmittance of any given 1 metre of path length 18 depends on the molecular density, the temperature and the atmospheric pressure along that 19 path, but is principally determined by whether the chosen spectral wavelength (wavenumber) 20 coincides with an absorption line, or a collection or band of lines. 21 22 At some wavelengths, away from such absorptions, thermal radiation emitted upwards by the

23 surface has a high probability of escaping to space: at more highly absorbing wavelengths, a

24 photon has a very low probability of escaping to space, instead being absorbed by a higher

25 (and usually colder) layer of the atmosphere: subsequently, some of the absorbed energy may

26 be re-emitted, this time at the lower local temperature, and so on, until eventually a photon is

1 able to escape to space. This process is the physical basis of the greenhouse effect. In local 2 thermodynamic equilibrium (effectively the case for the Earth's atmosphere below  $\sim 60$  km), the emission at a given local temperature, T, is governed by the Planck law. The Planck law 3 4 may be written as:

5

$$B_{f}(T) = \frac{2hf^{3}}{c^{2} \{ \exp[hf/k_{B}T] - 1 \}}$$
 {B2}

7

8 where  $B_f(T)$ , the spectral radiance of a black body (unit emissivity), is the power per unit area, 9 per unit solid angle, per unit frequency interval at frequency, f, in a particular direction; in 10 addition to previously defined quantities,  $k_B$  is Boltzmann's constant, and h is Planck's 11 constant.

12

13 Radiative transfer within an atmosphere is described in terms of the Schwarzchild equation of 14 radiative transfer. This treatment is a little beyond the scope of this 'primer', but the 15 interested reader is referred to the texts by Chandrasekhar [1960], Goody, [1964], and Goody 16 and Yung [1989] for more information. This equation describes how radiation propagates 17 along a path through an atmosphere, where scattering, absorption and thermal emission may 18 all occur due to interactions with the molecules and cloud particles along the path. While 19 absorption (emission) will reduce (enhance) the intensity of the radiation as it propagates, 20 scattering processes can have either effect according to whether radiation is being scattered 21 into, or out of, the beam. 22 23

[Figure B3 near here in Box 1]

1	Figure B3 shows some calculations using a radiative transfer code <sup>6</sup> which illustrate some of
2	these points. Shown in Figure B3 are the upwelling, top of the atmosphere (commonly
3	abbreviated to TOA) spectra calculated for a warm, moist model atmosphere, and a cold, dry
4	model atmosphere (so-called "standard atmospheres", [Anderson et al, 1986]), both cloud-
5	free. The y-axis is in units of spectral radiance (power per unit area, per unit solid angle, per
6	unit wavenumber interval, in a particular direction). The black curve is the calculated
7	spectrum for each case. The red lines describe the shape of the Planck function for the
8	temperatures indicated. The line structure below about 500 cm <sup>-1</sup> is mainly due to water
9	vapour; the band centred at 667 $\text{cm}^{-1}$ is due to CO <sub>2</sub> ; the band centred at 1060 $\text{cm}^{-1}$ is due to
10	$O_3$ ; and the extended band above about 1250 cm <sup>-1</sup> is due primarily to H <sub>2</sub> O and other gases,
11	including CH <sub>4</sub> . In the window region, between about 800 and 1250 $cm^{-1}$ the brightness
12	temperature seen at the TOA is close to the assumed surface temperature. In the centre of the
13	intensely absorbing CO <sub>2</sub> band, the brightness temperature is that of the tropopause/lower
14	stratosphere, because below these heights the atmosphere is opaque. In the pure rotational
15	band of water vapour, below about 500 cm <sup>-1</sup> , the brightness temperature represents the
16	temperature at the level from which most of the radiation reaching the TOA is originating.
17	

<sup>18</sup> *Heating rates* 

<sup>&</sup>lt;sup>6</sup> [There are a number of powerful radiative transfer codes available, which allow the calculation of the transfer of electro-magnetic radiation through an atmosphere made up of gases, aerosols and clouds. A detailed treatment is beyond the scope of this work, but an excellent source in which to find references to and details of a wide range of codes is given in the Wikipedia website at:

http://en.wikipedia.org/wiki/List\_of\_atmospheric\_radiative\_transfer\_codes.

Specific publications include Cahalan et al. (2005), Edwards (1992), Rothman et al., (2005) and Stamnes K (1988)].

1 The concept of a heating rate follows on from the basic concepts of radiative transfer given

above. The heating (or cooling) rate of a volume of the atmosphere,  $\frac{dQ}{dt} = \hat{Q}$ , where Q is heat 2 3 energy, is a measure of the rate of gain (or loss) of energy by that volume to space. The 4 heating rate is defined as the energy gained per unit volume, per unit time, into a  $2\pi$  steradian 5 hemisphere centred on the vertical direction, and may be evaluated as a function of spectral 6 wavenumber,  $\nu$ , or as a spectrally integrated total rate. Unless otherwise indicated, we shall consider the heating rate to be a function of wavenumber. Thus, the heating rate per 7 wavenumber, as a function of altitude, may be written as  $\dot{Q}_{\nu}(z)$ , which, in units of Jm<sup>-3</sup>s<sup>-</sup> 8  $(cm^{-1})^{-1}$ , is equal to the vertical gradient of the net radiative flux<sup>7</sup>,  $F_N(z)$  [Andrews, 2000]. 9 10 For an atmosphere in hydrostatic equilibrium, with  $c_p$  being the specific heat at constant 11 pressure, and with the sign convention of net flux noted in the footnote:

12 
$$\dot{Q}_{\nu}(z) = \rho(z)c_{p}\frac{dT_{\nu}}{dt} = \frac{dF_{\nu,N}(z)}{dz}$$
, {B3}

13

though by common practice, at least in meteorology, an alternative heating rate is more
usually expressed as a rate of change of temperature for the volume of air concerned (units
Ks<sup>-1</sup>(cm<sup>-1</sup>)<sup>-1</sup>):

17 
$$h_{\nu}(z) = \frac{dT_{\nu}}{dt} = \frac{\dot{Q}_{\nu}}{\rho(z)c_{p}} = \frac{1}{\rho(z)c_{p}}\frac{dF_{\nu,N}(z)}{dz} \qquad .$$
 {B4}

18

19 The total heating rate may be obtained by integrating over v and/or z, as appropriate.

- 20 [End of Box 1]
- 21

 $<sup>^7</sup>$  Net radiative flux = downwelling flux – upwelling flux, or  $F_N$  =  $F^{\downarrow}$  -  $F^{\uparrow}$ 

- 1
- Background: Spectral properties and energy balance of the Earth in the FIR
   3
- 5

# 2.1. Clear sky spectral properties in the FIR

- Box 1 gives a "tutorial" on the basic principles and concepts of spectroscopy and radiative
  transfer which underlie a study of the IR properties of the Earth's atmosphere, and which
  appear later in the paper. We now turn to a more specific examination of the FIR in
  particular.
- 10

The spectral properties - that is the variation of the absorption coefficient with wavenumber of the cloud-free atmosphere at wavelengths longer than 15 µm are dominated by water vapour, and only at the high frequency end of this range does CO<sub>2</sub> absorb significantly. The absorption by these molecules in the FIR is so strong that from space the surface is entirely obscured, except in the coldest and driest conditions near the poles. Thus, the study of the FIR Earth from space is essentially a study of the atmosphere alone.

17

In the neutral atmosphere, in comparison to water vapour and CO<sub>2</sub>, other absorbing molecules 18 19 or atoms absorb an insignificant amount of energy. In the FIR the pure rotational band of 20 water vapour extends from the microwave region (weak transitions at wavenumbers as low as  $6 \text{ cm}^{-1}$ ), to the CO<sub>2</sub> v<sub>2</sub> vibration-rotational band centred at 667 cm<sup>-1</sup>. The centre of the pure 21 rotational band of water vapour is a broad feature around 100-150 cm<sup>-1</sup>, comprising many 22 23 hundreds of strongly absorbing individual pure rotational lines [Goody, 1964]. At higher wavenumbers, the intensities of transitions fall to lower values in the 400-600 cm<sup>-1</sup> region, 24 25 where a partially transparent "window" occurs, and where the lower frequency edge of the very strong CO<sub>2</sub> band begins, at about 500 cm<sup>-1</sup>. At wavenumbers below the peak of the 26

water vapour band, the intensity of transitions also decreases. Other molecular species (and
atomic oxygen), including O<sub>3</sub>, N<sub>2</sub>O, HDO, HNO<sub>3</sub>, and many others [Harries, 1977; Carli,
1984], have pure rotational or low energy vibrational transitions in the FIR, but these are
insignificant in terms of radiative energy transfer compared with water vapour. However, the
existence of spectral features due to these species does, of course, provide the possibility of
using the FIR for retrieval of atmospheric composition.

7

8 Following on from Figure B3, Figure 1 presents calculations of the transmittance of the 9 atmosphere in the FIR (Brindley, unpublished: see also Brindley and Harries [1998]) for the expanded region 0-1000 cm<sup>-1</sup>, which illustrate the dominant influence of water vapour on the 10 11 FIR spectrum. These calculations use a so-called line-by-line, or high resolution radiative 12 transfer code that calculates the atmospheric transmittance at very high spectral resolution so 13 that each individual spectral line shape is included. Spectrum (a) shows the fractional 14 transmittance from 12 km to the top of the atmosphere (TOA) for a moist, warm tropical 15 standard atmosphere [Anderson et al., 1986], and (b) shows the transmittance from the surface 16 to the TOA for the same model atmosphere. Figure 1(c) shows the transmittance from 8 km 17 to the TOA for a colder and drier polar standard atmosphere, while (d) shows, for this same 18 atmosphere, the transmittance from the ground to the TOA. This figure illustrates several 19 points:

20

The principal absorption features are the pure rotational band of water vapour between about 20 and 400 cm<sup>-1</sup>, and the CO<sub>2</sub> v<sub>2</sub> vibration band between about 600 and 720 cm<sup>-1</sup>.
For much of the FIR, the atmosphere becomes partially transparent only in the upper troposphere and stratosphere, due to the intense absorption strength of the pure rotational water vapour band. Over much of the FIR, the lower troposphere is totally opaque.

This means that the FIR is particularly useful for remote sensing of composition in the
stratosphere, where the water vapour absorption is weaker, and so does not block absorption
lines due to other absorbers [Carli, 1984].

It also means that much of the energy that is emitted and which cools the Earth to
space comes, not from the surface, but from the upper troposphere, because the lower
troposphere over most of the planet in the FIR is totally opaque, the surface is obscured, and
no energy can escape to space.

- 8
- 9

# [Figure 1 near here]

10

11 We can see from this evidence that in the FIR, the energy that the Earth emits to space as part 12 of the cooling processes which, in thermal steady-state, balance the heating due to the Sun, is 13 strongly modulated at different wavenumbers and different heights by water vapour. Bearing 14 in mind that the Earth is a FIR object in space, emitting at an effective temperature of about 255 K (we recall that at this temperature a Planck curve peaks at about 500 cm<sup>-1</sup>), we can see 15 16 that water vapour acts in the FIR as an important greenhouse gas: it can absorb energy from 17 the warm surface below, and re-emit it to space at a lower temperature in, say, the upper 18 troposphere. Therefore, water vapour exhibits a strong feedback, in which any greenhouse 19 warming due to increasing levels of  $CO_2$  is amplified if more water vapour enters the 20 atmosphere as a result. As CO<sub>2</sub> concentrations rise in the atmosphere, so the atmospheric and 21 surface temperatures rise: this rise in temperature in turn causes increased evaporation from 22 the oceans, and so more water vapour may be expected to enter the atmosphere. This 23 additional water vapour enhances the original greenhouse warming, and is a major 24 contribution to global climate change [IPCC, 2001 and 2007]. A corollary of these properties 25 is that from space in the FIR the surface is usually not visible, because of the very high 26 opacity of the atmosphere.

### 1

2

# 2.2 Continuum absorption by water vapour

3

4 The water vapour spectrum includes one aspect which is far from well understood theoretically. Whereas the absorption around the centre of a spectral line of water vapour is 5 6 very closely described by the Voigt line shape [Andrews, 2000] because both collisional broadening and Doppler broadening of the spectral lines are important in the FIR [Goody, 7 8 1964], the precise shape and properties of the absorption lines farther from line centre are not 9 well understood, despite having been the subject of considerable research over many years. It 10 is found that the absorption in these "far wings" of the lines is stronger than is predicted by 11 any of the current theories of collisional (pressure) or Doppler line broadening. 12 13 The ideas put forward to explain this feature fall into three classes: those which treat the 14 discrepancy as purely a matter of the wrong line shape in the far wings for molecule-molecule 15 interactions (so-called "continuum absorption": Ma and Tipping, [2002]; Ptashnik et al., 16 [2004]); those that invoke the creation of a new molecule as a result of collisions (a dimer, or 17 double molecule of water vapour, and even higher polymers) that is responsible for the 18 "missing" absorption [Vaida et al., 2001]; and those which adopt a more pragmatic approach, 19 [Clough et al., 1989] using an empirical model to describe the measured excess absorption: in 20 this approach, two terms have been invoked, one a self-broadening effect due to water-water 21 molecule collisions, and the other due to water-foreign gas (eg  $N_2$  or  $O_2$ ) broadening. 22 23 In this last approach, the water vapour continuum was defined as all absorption due to water

vapour not attributable to a Lorentz (collisional broadening) line [Goody, 1964], within 25

 $25 \text{ cm}^{-1}$  of each line centre [Clough et al. 1989]. The success of this approach, which employed a

26 common line shape for all spectral lines, led the authors to suggest that water vapour

1 continuum absorption was due to the intermediate and far wings of allowed water vapour 2 monomer transitions. Clough and colleagues claimed that their model effectively ended 3 speculation about the existence of discrete dimers or multimers. In [Mlawer et al 1999], 4 Clough and co-workers introduced a new formulation of the CKD approach to the water vapour continuum, designated MT CKD. The new model describes the contribution from 5 6 each spectral line as the sum of two terms: a sub-Lorentzian line shape, and a small, broad additional line shape that provides the needed super-Lorentzian absorption in the intermediate 7 8 line wings. The MT CKD version has been tuned to observations in the far infrared recorded 9 with an upward-looking ground-based Fourier transform spectrometer (Atmospheric Emitted 10 Radiance Interferometer) used in campaigns, for example, in cold dry Arctic conditions 11 [Tobin et al. 1999].

12

13 Estimates (Rizzi, unpublished) using the US standard atmosphere and the MT CKD model, 14 show that the continuum absorption due to water vapour reduces the outgoing thermal 15 radiation at the TOA in clear sky conditions by about 2.2%, equivalent to a change in radiance of about 2 Wm<sup>-2</sup>sr<sup>-1</sup>. The downwelling longwave radiation at the surface was found to 16 increase by almost 6 Wm<sup>-2</sup>sr<sup>-1</sup>, that is by about 5.8%. Recent work [Green, 2003; Green et al., 17 18 2007] has shown that continuum absorption, particularly water-foreign gas continuum 19 absorption is significant, even in the centre of the strong pure rotational band of monomer 20 water vapour, and so cannot be ignored in a complete treatment of the FIR. 21

22 2.3 Heating rates

23

The primer in Box 1 gives a definition of the parameter known as the heating (or cooling) rate. This is a measure of the rate of energy gain (or loss). It is interesting to calculate the heating rate as a function of altitude and of spectral wavenumber, as was done by Kiehl

1	[1983] and Charlock [1984], and later by Clough et al [1992], and this has been repeated for
2	the US Standard Atmosphere [Anderson et al., 1986] tropical and sub-arctic models, in Figure
3	2 [Brindley and Harries, 1998].
4	
5	[Figure 2 near here]
6	
7	Figure 2 makes clear the dominant effect that the pure rotational band of water vapour has on
8	the Earth's atmospheric radiative energy balance. Because the opacity of the pure rotational
9	band varies with wavenumber, and peaks in the region of 100-200 cm <sup>-1</sup> , we find that the
10	height at which maximum cooling (energy loss) to space occurs for a given wavenumber,
11	peaks in the mid- to upper troposphere, at altitudes that depend on the opacity and therefore
12	on the wavenumber. At any given wavenumber the cooling rate profile can be calculated
13	from the vertical gradient in net flux (equation B4). This quantity is itself determined by the
14	vertical gradient in the atmospheric transmission to space, which, for clear conditions in the
15	FIR, is effectively controlled by the vertical distribution of water vapour in the atmosphere.
16	The level of peak cooling occurs where the transmission gradient is a maximum. Below this
17	maximum the transmission to space tends to zero because the path to space at these lower
18	levels is blocked by the strongly absorbing layers above. Above the maximum the decay in
19	the amount of water vapour with height is sufficient to allow the transmission to tend to a
20	value of unity; when this occurs all of the radiation emitted from levels below this height
21	escapes directly to space.
22	
23	Other notable features that can be seen in Figure 2 include:
24	
25	• Strong absorption (local heating) in the lower stratosphere by the $v_3$ fundamental band
26	of ozone, O <sub>3</sub> centred at 1045 cm <sup>-1</sup> , which switches to strong cooling above.

2	• Very intense absorption by the $v_2$ fundamental band of CO <sub>2</sub> centred at 667 cm <sup>-1</sup> , so
3	intense that no radiation can escape to space except from stratospheric heights, where a strong
4	cooling to space occurs at the band centre;
5	
6	• Cooling to space in the atmospheric "window" region between about 800 and 1200
7	cm <sup>-1</sup> , due to weaker water vapour lines, and to continuum emission and absorption.
8	
9	2.4 Sensitivity of the emitted spectrum to changes in water vapour content and
10	temperature
11	
12	As suggested by section 2.3 the FIR spectrum at the top of the atmosphere is highly sensitive
13	to changes in atmospheric water vapour [Rizzi et al., 2002].
14	
15	[Figure 3 near here]
16	
17	Figure 3 shows the difference in outgoing longwave radiance between three simulations:
18	• Case A (Tropical standard atmosphere with water vapour concentration scaled by 0.9
19	and skin temperature decreased by 1K);
20	• Case B (Tropical standard atmosphere with only water vapour concentration scaled by
21	0.9, and no other change);
22	• Case C: control (Tropical standard atmosphere, with no scaling).
23	
24	Figure 3 shows the differences Case A-C, and Case B-C. Several features are apparent. First,
25	in the centre of the $CO_2$ band the radiance differences are zero in both cases, because the
26	atmosphere is so optically thick at these wavelengths that none of the changes to the

1 atmosphere are detected in the outgoing radiance: at these wavelengths the radiance 2 originates in the stratosphere (which remained unchanged in the simulation). The Case B-C 3 difference shows essentially the spectral profile of water vapour and continuum absorption. 4 The Case A-C difference shows the same difference as Case A-B in the pure rotational band below 600 cm<sup>-1</sup>, where the surface (reduced skin temperature) is not detected at the top of the 5 6 atmosphere because of the intense absorption by water vapour. However, in the more 7 transparent window region, the effect of the -1K skin temperature change in Case A is apparent between about 750 and 1300 cm<sup>-1</sup>. These differences indicate that spectral 8 9 differences are evident outside the optically thick spectral regions and therefore the spectral 10 signal provides information on the two separate physical processes. 11 12 Similarly, calculations for higher, including polar, latitudes (not shown, but see also Figure 2) 13 also show that the pure rotational band remains a strong absorber, even at the lower water 14 vapour concentrations. In fact, a very interesting partially transparent window in the 500-600 15 cm<sup>-1</sup> region opens up, raising the intriguing question of enhanced surface to space radiative 16 cooling, as reported by Shaw et al, [1999] on the basis of low water content measurements 17 from a high mountain. 18 19 Rizzi et al., [2002] also demonstrated that the FIR was particularly sensitive to water vapour 20 changes in the upper troposphere, and suggested that spectrally resolved measurements were 21 more sensitive to humidity changes at all levels in the atmosphere than broad-band (spectrally

23

22

24 **2.5** Cloudy sky properties in the FIR

integrated) measurements.

1 So far, we have reviewed the evidence for the FIR properties of a clear- i.e. cloud-free -2 atmosphere. However, atmospheric condensed phases of water vapour, especially the ice 3 phase, play an important role on the infrared atmospheric outgoing radiation and the energy 4 balance of atmospheric layers. The fundamental importance of the ice phase is simply that to replace a warm Earth surface with a very cold ice cloud surface has a larger energy balance 5 6 impact at the TOA than replacement by a low, warm liquid-phase cloud. Ice clouds in 7 particular are believed to be extremely important for climate in the tropics [Liou, 1986], 8 because of their prevalence and persistence, and it has been suggested that they represent a 9 significant feedback to climate forcing [Randall et al., 1989]. Bulk radiative studies of cirrus 10 clouds show that they may radiatively cool or heat the upper atmosphere in the thermal 11 infrared wavelengths depending upon height, geometrical and microphysical features [Slingo 12 and Slingo, 1988; Stephens et al., 1990; Stackhouse and Stephens, 1991].

13

14 Recent studies include the publication of the scattering and absorption properties of ice clouds 15 for the MIR and FIR [Yang et al (2005), and Baum et al, (2007)], and a demonstration of the 16 measurement of cloud phase using MIR and FIR observations [Turner et al (2003)]. Also, 17 Rizzi and Mannozzi [2000] quantified the importance of the FIR contribution to the spectrum 18 in outgoing top of the atmosphere fluxes, as a function of increasing cloudiness. They showed that under clear conditions the ratio of energy emitted between 10 and 600  $\text{cm}^{-1}$  to 19 total emission (computed from 10 to 2750 cm<sup>-1</sup>) lies between 0.38 for the tropical and 0.48 for 20 21 the sub-Arctic winter standard atmospheres. The effect of cloud is to make the range below 600 cm<sup>-1</sup> energetically more important than in the clear-sky case, with the same ratio 22 23 increasing up to about 0.56. This is because, at typical cloud-top temperatures, the cold 24 cloud emission is strongly dominated by emission at FIR wavelengths.

Further studies have clarified the importance of the FIR when addressing the problem of radiative cooling in the presence of clouds. Initial studies [Maestri and Rizzi, 2003] have shown that flux divergence, integrated over the whole cloud depth for a tropical cirrus cloud, reveals two well defined spectral regions: the FIR where net emission of radiation occurs, and the atmospheric window regions and near infrared where net absorption dominates. Therefore the integrated heat balance and the net diabatic effect on the cloud layer requires consideration of two large contributions of opposite sign.

- 8
- 9

#### [Figure 4 near here]

10

11 In the lower panels of Figure 4, examples of vertically propagating net fluxes in the FIR and 12 in the 8-14 µm window region, for three different tropical atmospheric conditions (clear sky, 13 semi-transparent cirrus and opaque cirrus) are reported. The cirrus cloud base is at 10.5 km, and has a thickness of 1.8 km. The ice water paths are zero, 26.7 g m<sup>-2</sup> and 890 g m<sup>-2</sup>, 14 respectively. In the upper panel the layer spectral energy balance is shown: across the layer 15 16 there is a FIR net emission (cooling) even for clear skies, because of the strong absorption by 17 the pure rotational band of water vapour. With ice cloud, this cooling is enhanced. In the 18 window region, the only significant changes of energy balance occur when cirrus is 19 introduced.

20

Cloud ice water content (IWC) strongly affects the proportion of energy absorbed and emitted by the layer, but once a cloud level and atmospheric profile are predefined, the crossover wavenumber between cloud net absorption and net emission is well defined, quite independently from cloud transmittance. The atmospheric temperature profile is recognized as the main factor in establishing the crossover wavenumber.

Maestri and Rizzi [2003] also point out that the presence of a cloud layer changes the diabatic heating of all layers below the cloud and observe that the atmospheric concentration of water vapour has an important impact on the cloud's energy balance, since the more the atmosphere below the cirrus is transparent, the more energy is absorbed by the cloud. The role of mid and lower troposphere water vapour in absorbing FIR and window downward radiation emitted by high level cirrus clouds is also highlighted, thus stressing another interaction between water vapour and clouds.

8

9 The calculation and the observation of cloud radiative interactions in the FIR, as at all 10 wavelengths, is difficult. The dynamical and transient nature of cloud formation and 11 destruction, the importance of knowledge of the cloud microphysical properties (especially 12 for ice clouds, where ice particle shape is important), and the problem of sampling the 13 changing, complex cloud field and associated radiation field, all conspire to make the problem 14 extremely difficult. Nevertheless, as has been demonstrated, the impact of clouds on the FIR 15 is very significant, and the work to date has impressed upon us the urgency of addressing the 16 cloud-radiation problem at these long wavelengths. Given all the uncertainties mentioned, it 17 is most unlikely that climate models include accurate descriptions of these processes, or of 18 how they affect the climate under changing conditions. Solving this problem is of the utmost 19 urgency.

20

#### 21

#### 3. The FIR and climate: water vapour and cloud feedback

22

As we have seen, both water vapour and clouds have strong FIR effects on the energy balance
of the Earth's climate system. Here we review work which has attempted to improve our
quantitative understanding of how the water vapour and the cloud radiative feedback
processes affect climate.

# 2

# 3.1 Water vapour feedback and the FIR

3

An excellent review of the physics of the water vapour feedback may be found in Held and
Soden [2000], in which they discuss the feedback in physical terms, and its dependence on
humidity and temperature. Also, they discuss the impact of dynamics on the feedback
process, and a controversy over the sign of the feedback (see later). Many of their ideas may
be extended to other processes than water vapour absorption.

9

10 Adapting the analysis provided by Held and Soden, we may consider a balance between the 11 globally averaged absorbed solar energy, or power into the system  $P_{in}$ , and the globally 12 averaged output emitted longwave radiation,  $P_{out}$ , as in

13

$$P_{in} = P_{out} + \Delta P \qquad , \qquad \{1\}$$

15 Where

16  

$$P_{in} = \frac{S}{4}(1 - A)$$
and
$$P_{out} = \sigma (1 - g)T_s^4$$

$$\left. \begin{cases} 2 \end{cases} \right\}$$

17

Here, S = solar constant,  $\sigma =$  Stefan-Boltzmann's constant, A is the global albedo,  $T_s$  is the surface temperature,  $\Delta P$  is discussed below, and g is the normalised greenhouse parameter:

21 
$$g = \frac{G}{\sigma T_s^4} = \frac{\sigma T_s^4 - \sigma T_e^4}{\sigma T_s^4} .$$
 {3}

22 Here

1	$G = \sigma(T_s^4 - T_e^4) $ <sup>{4</sup> }
2	
3	is the absolute greenhouse parameter.
4	
5	We may introduce the spectral variation of both $g$ and $G$ by inserting the Planck radiation law
6	(equation $\{B2\}$ ) in place of the Stefan-Boltzmann law in equations $\{3\}$ and $\{4\}$ . If we do
7	this, and calculate spectrally resolved values of $G_{\nu}$ (solid lines), and also the surface emission,
8	$E_{\nu}$ (broken lines: assumed to be a black body at the surface temperature, and to be described
9	by the Planck radiation law), we obtain the results shown in Figure 5 [Brindley and Harries,
10	1998]. Results are given for two model atmospheres, tropical and polar, in units of W $m^{-2}$
11	$(cm^{-1})^{-1}$ .
12	
13	[Figure 5 near here]
14	
15	Figure 5 clearly shows the dominating effect of the FIR on the water vapour greenhouse
16	effect.
17	
18	In equation $\{1\}$ above, the term $\Delta P$ was introduced, and in passing we might comment on its
19	significance in climate research. If $\Delta P$ is positive, it represents the energy stored in the
20	system, due to any processes, (an extreme example being deep ocean circulation), that causes
21	a delay in the temperature response to solar energy absorption. If $\Delta P$ is negative, this
22	represents energy lost to the system, for example, by scattering of sunlight back to space by
23	volcanic aerosol. The stored energy term depends principally on ocean parameters such as
24	deep ocean circulation rates and thermal mass, as well as transient parameters such as
25	atmospheric aerosol loading, and it has been suggested recently that $\Delta P$ may be measured

1 from broad band TOA net fluxes (from polar orbiting satellites, see: Wielicki et al., [2002]; 2 Wong et al., [2006]; and from geostationary satellite, see Harries et al. [2005]. It is possible 3 that such detections of stored energy may be better attempted using spectrally resolved 4 measurements (e.g. see Brindley and Harries [1998] and Harries et al. [2001]): this will be 5 discussed in a future paper. 6 7 The question of the magnitude of the contribution of the FIR to the total infrared greenhouse 8 effect in the climate system is important: if, as Figure 5 suggests, it is a significant 9 component, then it must be treated in detail and with accuracy in climate models. In a series 10 of studies, Sinha and Harries [1995; 1997] used tropical and sub-arctic standard atmospheres 11 and a band model to simulate the OLR spectrum at the TOA, and integrated the spectrum to estimate the FIR<sup>8</sup> and total fluxes leaving the TOA. The principal findings are shown in 12 13 Table 1, and refer back to Figure 5. 14 15 [Table 1 here] 16 The parameters shown in the Table for the two standard atmospheres represent the spectrally 17 resolved total greenhouse parameter,  $G_{\nu}$  and the normalised greenhouse parameter,  $g_{\nu}$ , as 18 19 defined above. The Table also shows the change in the total outgoing longwave radiation 20 (OLR) due to a doubling of  $CO_2$  in the band model; and the change in water vapour amount 21 required to give the same OLR change as the doubling of CO<sub>2</sub>. As we can see, the FIR contribution to the total water vapour radiative forcing at wavenumbers less than 500 cm<sup>-1</sup> 22

- amounts to between  $\frac{1}{5}$  and  $\frac{1}{2}$ : expressed as a fraction of the total greenhouse effect (for all
- 24 absorbers) of the atmosphere, the fraction below 500 cm<sup>-1</sup> lies between  $\frac{1}{4}$  and  $\frac{1}{3}$  Of course,
- 25 taking a different boundary for the FIR, for example 15 μm, would increase these fractions.

1								
2	This discussion of the contribution of the FIR to the total integrated OLR raises the issue of							
3	whether, in reverse, measurements of broadband fluxes by ERB radiometers in space could be							
4	analysed so as to throw light on the FIR radiative energy budget. At first it would seem a							
5	difficult prospect, since so many variables (surface temperature and emissivity, cloud type,							
6	height and extent, and atmospheric temperature and composition) all affect the balance							
7	between the MIR and FIR. Nevertheless, such a study would be worthwhile, since the							
8	increase in our understanding of the role of the FIR could be significantly enhanced.							
9								
10	Earlier in this section we mentioned a controversy over the sign of the water vapour feedback:							
11	is it positive or negative? The most obvious analysis would indicate a positive feedback:							
12	higher temperature means more evaporation and more water vapour, which means further							
13	warming. However, Lindzen [1990], controversially suggested that the feedback could be							
14	negative, depending on the drying of the upper troposphere by descending, desiccated air. A							
15	number of rebuttals of this view have since been published, which we will not review at							
16	length here, because it is some way from our main theme of the FIR. So, we refer to the							
17	valuable review of the water vapour feedback by Held and Soden [2000], where much more							
18	detail can be obtained. We also note in passing that the use of the Mt Pinatubo volcanic							
19	eruption as a means of demonstrating the positive feedback of water vapour, by Soden et al							
20	[2002] was a sophisticated riposte to the Lindzen view.							
21								
22	3.2 Cloud feedback and the FIR							
23								
24	The greenhouse effect of both water and ice clouds have been recognized to be of particular							
25	importance in the mid-infrared window and in the FIR regions [Maestri, 2000; Rizzi and							
	<sup>8</sup> Note that in the studies reported, Sinha and Harries chose the range 0-500 cm <sup>-1</sup> to represent the FIR.							

1 Maestri, 2001]. Simulations of the impact of ice clouds on the spectral greenhouse 2 parameters,  $g_{\nu}$  and  $G_{\nu}$ , are reported in Figure 6. In the plot a tropical standard atmosphere is considered. Calculations of the greenhouse parameters are repeated for various values of the 3 4 cloud optical depth (and hence transmittance), values of which are provided in the upper 5 panel. The upper panel shows the absolute effect: as the transmittance through the ice cloud gets smaller, so the radiance escaping to space decreases, and  $G_{\nu}$  gets larger (equation {4}). 6 A strong effect is obviously revealed in the 800-1200 cm<sup>-1</sup> window region, but a similar 7 amount of energy is also trapped in the atmosphere in the FIR region by the cirrus cloud. 8 9 The lower panel shows, for the same three cases, the normalised greenhouse parameter,  $g_{v}$ . This indicates that, relative to the emission from the surface, the greenhouse trapping 10 11 increases with wavenumber: however, in absolute terms (upper panel), the FIR is clearly very 12 important in the effect of ice clouds on the planetary radiation balance.

13

The upper panel of Figure 6 also shows that the spectral region near 400  $\text{cm}^{-1}$  is as sensitive as 14 the 750-1000 cm<sup>-1</sup> window to the presence of ice clouds in the high atmosphere. This occurs 15 16 because at the heights typical of cirrus clouds [Poore et al., 1995], even the strong FIR water vapour absorption is not strong enough to obscure the clouds when seen from space. This 17 18 would not be the case for lower, liquid water clouds: at the lower altitudes typical of liquid 19 water clouds, the FIR transmittance to space is very small, and they have little impact in the 20 FIR on the greenhouse effect. Further investigation [Rizzi and Maestri, 2001] of the results 21 shown in Figure 6 were able to demonstrate that the radiance to space depends in a complex 22 way on both the physical (height and thickness) and optical (effective radius, ice water mass) 23 ice cloud parameters.

- 24
- 25

[Figure 6 near here]

1 Rizzi and Maestri [2003] also studied the FIR contribution of clouds to the radiative balance 2 through the use of the parameter known as the cloud radiative forcing (CRF) [Ramanathan, 3 1987; Hartmann et al., 1986; Cess and Potter, 1988; Cess et al, 1990, 1996]. The CRF can also 4 be computed by differencing the clear-sky and total-sky radiative fluxes, and by using satellite 5 radiation budget data it is possible to estimate the effects of actual cloudiness on the TOA heat 6 balance and the dependence of these effects on location and season [Hartmann et al., 1986]. CRF is especially useful in quantifying the interactions between clouds, radiation and climate at 7 8 the TOA [Stuhlmann, 1995] since it provides an immediate picture of how much emitted 9 radiation is trapped and of how much solar radiation is reflected by clouds with respect to clear 10 sky conditions. Although cloud radiative effects do respond to greenhouse forcing as a 11 feedback, nevertheless, this view of a stand-alone forcing has its merits. A positive CRF 12 indicates that the clouds cause a warming of the overall Earth-atmosphere system, and vice 13 *versa* [Doelling et al., 2001]. The major limitation of this parameter is that the vertical 14 distribution of the cooling and warming is not dealt with. Nevertheless CRF is a fundamental 15 quantity needed to constrain the effects of clouds within short or long-term climate models and should be determined as accurately as possible [Doelling et al., 2001]. 16 17 18 Despite its theoretical simplicity, the calculation of the CRF from broadband Earth Radiation 19 Budget measurements is a very demanding task [Hartmann et al., 1986; Stuhlmann, 1995; 20 Tian and Ramanathan, 2002; Futyan et al, 2005]. For example, in many tropical convective 21 regions the large long-wave and short-wave cloud forcing more or less compensate each 22 other, so that it is extremely important to be able to measure very accurately the two 23 contrasting terms constituting the total CRF. Rizzi and Maestri [2003] were interested in the 24 FIR contribution to long-wave CRF (LCRF) in the presence of realistic clouds, focusing their

25 attention on the tropical and the sub Artic winter standard atmospheres. They studied the error

that can be introduced in the estimate of LCRF when radiation is measured by broadband

1 radiometers with an incomplete long-wave spectral coverage, since this is a common 2 instrumental problem. 3 Following these authors, the LCRF expressed in terms of radiance,  $R_t$  [W/m<sup>2</sup>sr], is given by 4 5  $F_t$ : 6  $F_t = R_t - R_t^*$ 7 {5} 8 9 where the asterisk indicates a cloudy scene, and the subscript, t, denotes a perfect 10 measurement (ie the actual LCRF), extending over the whole long-wave emission spectrum. By analogy with equation {5}, a similar expression  $(F_m = R_m - R^*_m)$  can be written for the 11 12 LCRF derived from a broadband radiometer which is insensitive to part of the FIR emission 13 spectrum, for example by the use of a spectral filter. Here the subscript *m* stands for 14 measured. The error in LCRF due to incomplete spectral coverage is given by  $F_t$  -  $F_m$  and the quantity  $D = 100^{*}(F_t - F_m) / F_t$  defines the percentage fraction of energy that cannot be 15 16 measured for technical limitations, weighted against the real LCRF. 17 18 Rizzi and Maestri [2003] showed that D is always positive (except possibly in some SAW 19 conditions when stratus clouds are embedded into a temperature inversion), so that 20 uncorrected measurements from sensors with incomplete spectral coverage produce a 21 systematic underestimation of the real LCRF. The value of D becomes particularly important 22 in presence of cirrus clouds (1 to 5% for the case considered by the authors), especially for the 23 highest cirrus cloud (15.4 km in the cited paper), not an uncommon situation in the tropics. 24 D is plotted in Figure 7 as a function of cloud transmittance at 900 cm<sup>-1</sup> for three cases of 25 26 tropical cirrus and one of sub-arctic cirrus. It was assumed that an instrument having a filter to

1	block all radiation below 350 cm <sup>-1</sup> was used. The Figure shows that for a given cloud type $D$
2	can be considered nearly constant with transmittance (that is for various values of ice water
3	content, IWC), and this same result (albeit with different numerical values) is obtained with
4	other types of filters and different cut-off values. The nearly constant behaviour is obtained
5	since the difference between the unmeasured part of the spectrum in clear and cloudy
6	condition, due to a particular type of cloud, increases almost linearly with decreasing
7	transmittance, and the rate of increase is almost the same as the rate of increase of the real
8	LCRF with transmittance.
9	
10	4. Retrieval of atmospheric composition and clouds in the FIR
11	
12	4.1 Retrieval of water vapour profiles using the FIR
13	
14	Water vapour vertical profiling from space using the inversion of emission features in the
15	outgoing longwave spectrum has been limited for the past 30 years mainly to measurements
16	in the $v_2$ water vapour vibrational-rotational band, centred at 1595 cm <sup>-1</sup> , for both operational
17	and research sensors. One early project which attempted to measure the water vapour profile
18	from space using the pure rotational band was the SIRS-B radiometer [Nithianandam et al.,
19	1993], and recent work described below has pointed to advantages that may exist using FIR
20	techniques.
21	
22	Recently, Rizzi et al. [2002] analysed the potential water vapour retrieval performance
23	attainable from spectral measurements with a resolution of 0.5 cm <sup>-1</sup> in the FIR, from 200 to
24	1000 cm <sup>-1</sup> . A technique known as the $\sigma$ -IASI methodology [Amato et al., 2002] was applied,
25	and it was found that the sensitivity of the rotational band to water vapour perturbations, as
26	measured by the matrix of the derivative of spectral radiance with respect to water vapour

1 amount (the Jacobian), may reach values 6 to 7 times greater than that exhibited using the  $v_2$ 2 band: this implies greater sensitivity of the radiance to changes in water vapour amount in the 3 FIR than in the mid-IR.

4

As an example, for the case of a tropical atmosphere standard model, Figure 8 shows a mesh
surface of the Jacobian derivative matrix, defined by

7

$$K(v,z) = \frac{\partial R(v)}{\partial q(z)}$$
(6)

9

10 where K is the Jacobian, v and z are wave number and altitude, respectively, R denotes the 11 spectral radiance and q(z) is the water vapour mixing ratio profile. In practice the kernel, K, is computed for a discrete set of  $v \equiv v_1$ ,  $v_M$  and  $z \equiv z_1$ ,  $z_N$ , so that the size of the resulting 12 13 matrix, **K** is  $M \times N$ . Figure 8 illustrates the magnitude of K(v, h) for the pure rotational band (top) and the  $v_2$  vibrational-rotational band (bottom), illustrating the much greater magnitude 14 15 of the Jacobean in the FIR. Of course, the Jacobian does not take account of the fact that 16 photon energies, and therefore radiances, R, are intrinsically larger at higher wavenumbers, an 17 effect which must also be taken into account. Nevertheless, to demonstrate a greater intrinsic 18 sensitivity is of interest.

- 19
- 20

#### [Figure 8 near here]

21

Since it is not immediately possible to capture all information present in a Jacobian computation because of its 2-dimensional complexity, the problem of water vapour retrieval capability using information content analysis has been addressed by a number of authors [Tarantola, 1987, Rodgers, 1976, Serio et al 2007]. This method also properly addresses the

1	problem of the	he accuracy	of the	observations,	which,	in t	turn	depends	on	the	detector
2	technology.										

4 In order to simulate a FIR sensor retrieval performance, we need to define the following:

5

6	1. A suitable set of atmospheric states to be used as reference in all the calculations.
7	This basic set is used to compute an a-priori covariance matrix [Rodgers, 1976].
8	2. A minimum set of parameters which define a given technology, i.e. in this case an
9	IR/FIR spectrometer, used to record the spectral observations. The set used includes:
10	- the radiometric noise (used to build up the observational covariance matrix);
11	- the Instrumental Response Function, $S(\nu)$ , of the measuring instrument
12	(convolved with the infinite resolution spectrum to simulate the finite
13	resolution of the real spectrometer), assumed to be a Fourier transform
14	spectrometer.
15	3. A reference inverse methodology [Rodgers, 1976].
16	
17	Item 1: Atmospheric states.
18	
10	A get of state meeting which are representative of the widest streagehouis conditions

A set of *state vectors*, which are representative of the widest atmospheric conditions,
has been used. This set has been developed at ECMWF by Chevalier [2001].

21

22 Item 2: Instrument parameters

23

We assume the use of a Fourier Transform Spectrometer (FTS), which is defined by the spectral range or bandwidth, S(v), and the radiometric noise budget. The spectral coverage is

assumed to be 200 to 1000 cm<sup>-1</sup>; S(v) has been assumed to be a pure sinc function [Bell, 1 2 1972]:

3

4

$$S(\nu) = 2L \frac{\sin(2\pi\nu L)}{2\pi\nu L}$$
<sup>{7}</sup>

5

6 where the maximum path difference of the FTS spectrometer L is chosen in such a way to be consistent with a spectral resolution of  $\Delta v = 0.5 \text{ cm}^{-1}$ , that is  $L = \frac{1}{2\Lambda v} = 1 \text{ cm}$ . 7 8

9 The radiometric noise budget applies at the level of a single detector, whose footprint at nadir 10 is assumed to be 10 km [Palchetti et al., 2006].

11

12 Item 3: Reference inverse methodology

13

19

To provide an extensive description of the inversion theory used would be beyond the scope 14 15 of this paper, and since references already exist, it is not necessary to go into details. 16 Basically, we move in the context of Statistical Regularization [Rodgers 1976] and derive the 17 information content tool by transforming the inverse solution to Principal Components space 18 via Singular Value Decomposition [Amato et al, 1995, Carissimo et al 2005, Serio et al 2007]

20 To make a comparison to the retrieval performance achieved with a realistic sounder in the 21 mid infrared (IR), we have used the results from a recent analysis in which the performance of 22 the mid infrared radiometer sounder (IR sounder) for the Meteosat Third Generation (MTG), observing the  $v_2$  vibrational band of water vapour between 1210 and 2000 cm<sup>-1</sup>, was assessed 23 [Grieco et al., 2004]. 24

2

# 3

# [Figure 9 near here]

The comparison shows that the H<sub>2</sub>O retrieval capability in the far infrared is comparable to 4 5 and sometimes slightly better than that obtained for the MTG IR sounder. This is illustrated 6 in Figure 9, which shows the simulated H<sub>2</sub>O retrieval accuracies for the case of the FIR rotational band, and for the MIR vibrational band. Apart from the tropopause level, the 7 8 expected H<sub>2</sub>O retrieval performance is better than 10% in the troposphere for both 9 instruments. The FIR retrieval shows a slightly better performance than the MIR sounder in 10 the tropopause/lower stratosphere region. This is to be expected because of the stronger intensity of the rotational band lines by comparison with the  $v_2$  band. The water vapour 11 12 accuracy can be improved, of course, by considering FIR sounders with a larger FOV 13 (reduced horizontal spatial resolution), so that more photons reach the detector and the signal-14 to-noise is improved. A study by Serio et al [2007] provides quantitative information on the 15 relation between horizontal and vertical spatial resolution and the number of independent 16 pieces of information that are available in the retrieval.

17

# 18 **4.2** Water vapour retrieval in the presence of clouds (and vice-versa).

19

Infrared data are frequently affected by clouds. Thus, observations must be processed for
operational data assimilation and inversion for geophysical parameters (temperature and water
vapour), either by screening to remove cloud-contaminated soundings or by a so-called
process of cloud clearing. It should be stressed that, for meteorological applications,
operational numerical weather prediction centres currently use cloud-screened or cloud
cleared data. This is mostly because of time processing constraints. Forward calculations in
the presence of clouds remain time-consuming, making real time applications difficult. Cloud

screening or cloud detection is the first step in the assimilation chain of infrared radiances and
 it has been proved that high spectral resolution infrared observations in the atmospheric
 window (800 to 900 cm<sup>-1</sup>) can be effectively used for this task [Serio et al., 2000, Masiello et
 al., 2002].

5

The cloud clearing process can increase the uncertainty in the observed spectral radiances.
However, the spectral radiance uncertainty is typically confined below 1 K (in terms of
brightness temperature) for most of the cases [Rizzi, 1994, Rizzi et al., 1994, Li et al., 2005],
and the retrieval accuracy is only slightly degraded with respect to that shown in Figure 9 for
clear sky.

11

Another approach which is now becoming popular is to retrieve water vapour only above the cloud [Li et al, 2007], which again takes us back to the clear sky case discussed in the previous section. Crucial to this method is the correct determination of the cloud top height and temperature. The most powerful method to perform this task is the so-called CO<sub>2</sub> slicing method, which exploits the altitude sensitivity of narrow-band CO<sub>2</sub> channels located in the CO<sub>2</sub> absorption band at 15 µm [Chahine, 1974].

18

19 For water vapour retrieval in the upper part of the atmosphere, the limb sounding 20 measurement mode is commonly used. In this mode, we observe the atmosphere with a long 21 horizontal optical path, and with this geometry of observation even optically thin (in the 22 vertical) clouds can significantly attenuate the atmospheric signal, and jeopardise the 23 sounding of the atmospheric composition. However, this effect is not the same at all 24 wavelengths. Figure 10 illustrates this by showing the variation of ice particle extinction 25 coefficient (the sum of both absorption coefficient and scattering coefficient) as a function of 26 ice particle size and wavenumber. The variations shown in Figure 10 suggest the possibility

1	that the FIR might in some cases be used to penetrate deeper into a cloud that at MIR
2	wavelengths. Indeed, it is possible that more detailed studies may reveal that the FIR and the
3	MIR have sufficiently complementary properties of absorption and scattering that they may
4	advantageously be used in combination to detect the presence of clouds [Raspollini et al.,
5	2006; Reburn et al., 1998]. Further work is required.
6	
7	[Figure 10 near here]
8	
9	Turning to much longer wavelengths, in the millimetre wavelength region, a very useful
10	region of high transparency is observed even for the most dense ice cirrus clouds. In a case
11	study performed for the MARSCHALS instrument [Oldfield et al., 2001], which observes the
12	atmospheric emission in the millimetre wave region with a spectral resolution of 200 MHz ,
13	Del Bianco et al. [2007] have quantified the loss of information caused by ice clouds when
14	making atmospheric composition measurements with this instrument. They showed that the
15	effect of most clouds is adequately described with a model of atmospheric continuum
16	absorption. In the case of optically thick clouds, retrieval errors can occur because of
17	scattering effects and the problem of detecting sharp discontinuities in a cloud, when
18	observing with limited vertical resolution. When cloud altitude and scattering are correctly
19	modelled, a limited loss of information is caused by the presence of cloud. In the worst case it
20	has been shown that there is a loss of about 7% in terms of information content, relative to
21	cloud free conditions [Carli et al, 2007].
22	
23	[Figure 11 near here]
24	
25	Obviously, whatever the observation geometry, there is a complementary issue, which is the
26	characterisation of cirrus clouds from observations of the real atmosphere in the presence of

1 the interfering effects due to water vapour, carbon dioxide and other absorbers. Del Bianco 2 [private communication, 2007] has carried out calculations of how the FIR and MIR spectra 3 are affected by ice crystals. Figure 11 shows some examples of such calculations for a variety 4 of parameters including cloud altitude, ice water path, crystal habit and phase. Frame (a) shows clear sky, water cloud and ice cloud spectra. Frames (b) to (f) show a series of 5 6 difference spectra, identifying how the difference signals between clear sky and cloudy skies depend on water phase, altitude, ice content, particle radius and shape. Clearly, there is a 7 8 large amount of information contained in these difference spectra, which can be used to 9 characterise clouds from spectral measurements. Yang et al (2003) were able to show that 10 FIR spectral signatures of ice are useful in the retrieval of ice cloud properties: for example, they show that the brightness temperature near 400 cm<sup>-1</sup> is sensitive to ice crystal size, and 11 that the brightness temperature difference between 250 and 560 cm<sup>-1</sup> is sensitive to optical 12 13 thickness. However, these signals are not unambiguous, and more detailed studies are needed to determine precisely how much information on an unknown cloud can be extracted from 14 15 FIR spectral measurements. 16 17 5. **Observational studies of the FIR spectrum of the atmosphere and Earth system** 18 19 In this section we review experimental work on the properties of the FIR spectrum of the 20 Earth's atmosphere. Following a short historical perspective, we describe work using 21 commercial interferometers to understand the continuum absorption of water vapour, and new 22 instrument developments to measure the FIR spectrum from aircraft and balloons (TAFTS, 23 REFIR and FIRST), and refer to efforts to develop a design for a future satellite FIR

24 spectrometer (REFIR).

25

### 26 **5.1 Early submillimetre and FIR investigations**

2	With the development of the energy-efficient technique of Fourier transform spectroscopy
3	[Bell, 1972] in Europe (pioneered by workers such as Jacquinot and Fellgett), the weak
4	photon energies typical of the FIR became accessible when ambient temperature detectors,
5	such as the Golay cell, and the pyroelectric detector were used. Groups in the UK (National
6	Physical Laboratory; Queen Mary College), France (Observatoire de Meudon; Office
7	National d'Etudes et Recherches Aeronautique), Canada (U. Calgary), Italy (U. of Florence;
8	the National Research Council, CNR), and the USA (Smithsonian Astrophysical Institute;
9	National Center for Atmospheric Research) began to develop Fourier spectrometers,
10	detectors, and Fourier transform analytical techniques, to study the lower atmosphere (a
11	hindrance to astronomy at these wavelengths), and the stratosphere (which, because of the
12	very high line strengths in the FIR, and very low water vapour mixing ratios, proved to be an
13	ideal part of the atmosphere to be probed in the FIR).

14

15 The earliest work established the existence of transparent windows in the stratospheric 16 spectrum at submillimetre wavelengths, and a partly transparent troposphere from the surface 17 to space at millimetre wavelengths. Using the FIR to measure the humidity of the 18 stratosphere (still a controversial question at the time: was the stratosphere wet or dry?) 19 proved very successful [Gebbie et al., 1968, Burroughs and Harries, 1970, Harries, 1973, 20 1976]. With the advent of helium-cooled solid state detectors, the capability of FIR systems 21 increased dramatically, and measurements at higher resolutions (down to a few hundredths of 1 cm<sup>-1</sup>) on aircraft and balloons became possible, introducing the possibility of measuring not 22 23 just the principal spectral features, due to H<sub>2</sub>O, and O<sub>2</sub> and O<sub>3</sub>, but many other minor 24 constituents, through their pure rotational spectra. Numerous experiments to study this 25 spectral region from aircraft [Harries et al., 1972; Bussoletti and Baluteau, 1974; Marten and 26 Chauvel, 1975; Mankin, 1975; Carli et al., 1977] were used to study the FIR spectrum and the

composition of the upper troposphere and lower stratosphere. Other systems were developed
 for the more exposed environment of high altitude (40 km) balloon flights, when mid stratosphere water vapour and other constituents were measured [Harries , 1977; Clark and
 Kendall, 1976; Carli et al, 1980].

5

The science of FIR spectroscopy in the laboratory also developed fast, with measurements of
the quantitative positions, strengths and half-widths of spectral lines becoming available,
essential, of course, for making quantitative determinations of atmospheric concentrations.
Work on many species took place, including polar molecules like H<sub>2</sub>O, O<sub>3</sub> and more exotic
species such as nitric acid HNO<sub>3</sub> [Harries et al., 1971], as well as the collision-induced dipole
absorption in symmetric molecules such as CO<sub>2</sub> and N<sub>2</sub> at high pressures [Harries, 1970].

12

These measurements confirmed the spectroscopic properties of the atmosphere in the FIR. 13 For example, the long wavelength region from 4 to 200 cm<sup>-1</sup> was studied by Baldecchi et al, 14 15 [1984, 1988]: this region was where most of the rotational spectra of those atmospheric 16 constituents with a permanent or transient dipole moment can be observed. Figure 12 shows 17 an example of the high resolution atmospheric emission spectrum measured with the IBEX 18 balloon experiment [Carli et al., 1984] from an altitude of 38 km at the limb with a zenith angle of 91.6°. A portion of the spectrum between 45 and 65 cm<sup>-1</sup> is displayed. The upper 19 panel shows a synoptic view of this 20 cm<sup>-1</sup> spectral interval where it is possible to note some 20 21 general characteristics. Water vapour is present with some prominent spectral features that are 22 irregularly distributed (since the water molecule is a so-called asymmetric rotor [Herzberg, 23 1959]). A regular pattern of triplet lines is caused by molecular oxygen. Ozone is present with 24 some Q-branches that appear with a constant period and with many other ubiquitous lines. When the fine structure of the spectrum is analysed (see lower panel for the expanded 61-69 25 cm<sup>-1</sup> region), the contribution of several minor constituents can be observed. In this spectral 26

1 interval emission lines due to HCl (with its two isotopes), CO and OH can be observed. Other 2 interesting contributions are those of some isotopes of water vapour (HDO and  $H_2O^{17}$ ) and of 3 ozone (<sup>18</sup>OOO and O<sup>18</sup>OO) as well as some rotational transitions of vibrationally excited 4 ozone, identified with their vibrational quantum numbers.

- 5
- 6

### [Figure 12 near here]

7

Several other features have been observed in the spectral region between 4 and 200 cm<sup>-1</sup>, in 8 9 addition to those already mentioned: these include HNO<sub>3</sub>, N<sub>2</sub>O, NO<sub>2</sub>, HCN, ClO, HF, HBr, 10 HOCl, H<sub>2</sub>O<sub>2</sub>, HO<sub>2</sub> and atomic oxygen, O. Notably missing in the FIR are the features of CO<sub>2</sub> and N<sub>2</sub> which do not have a permanent dipole moment, and therefore no rotational spectrum: 11 these gases absorb in the FIR only under high pressures, when transient dipole moments are 12 13 induced in the symmetric molecules by forceful inter-molecular collisions [Harries, 1970]. 14 Annex 1 provides a list of atmospheric constituent species which have been detected by FIR 15 spectroscopy in the Earth's atmosphere. Since all molecules possessing a dipole moment 16 have rotational transitions in the FIR, this list is by now extensive.

17

18 Measurements in the FIR region have the disadvantage that the signal is intrinsically weak, 19 because the Planck function falls to low values at long wavelengths. However, there are also 20 advantages in the FIR. The atmospheric emission is observed in the Rayleigh-Jeans region of the black-body distribution where the dependence on temperature, T, is only linear<sup>9</sup>, so that 21 22 any errors in our knowledge of the atmospheric temperature cause less error in composition 23 retrieval than in other regions of the black body distribution, where the dependence of 24 radiance on temperature goes as a much higher power of T. Also, the line strength of the 25 rotational features can be accurately determined from dipole moment measurements, so that

1	measurements made with rotational spectra are intrinsically of high accuracy, and can be used
2	as a validation tool for the more difficult measurements made in other spectral regions
3	[Mencaraglia et al., 2006].
4	
5	The typically weak signal in the FIR may be efficiently detected either using cryogenic
6	detectors or (where suitable local oscillators exist) heterodyne techniques. The latter
7	technique is currently exploited for spaceborne observation of minor atmospheric constituent
8	in the millimetre and submillimetre spectral region (e.g. the Microwave Limb Sounder
9	experiments: see Waters et al. [2006]).
10	
11	5.2 Ground based FIR observations and studies of continuum emission.
12	
13	5.2.1 Background
14	
15	The continuum absorption due to water vapour was described in section 2.2, and continues to
16	be an un-resolved problem, meriting further work. With the advent and success of the
17	empirical CKD model [Clough et al 1989] for the water vapour continuum, a number of
18	experimental determinations of this absorption in the laboratory and the atmosphere have
19	taken place. This has led to a series of revisions to the empirical CKD model. A number of
20	the field campaigns have been based on an instrument whose basic design is due to ABB
21	Bomem Inc (ABB Bomem MR-100 series interferometer). [Knuteson et al, 2004a, 2004b].
22	The instrument is now deployed in many of the American ARM (Atmospheric Radiation
23	Measurement program) sites, where it is known as AERI (Atmospheric Emitted Radiance
24	Interferometer) [Ackerman and Stokes, 2003]. Most notable among these observations are
25	those taken with the version extended in the far infrared from SHEBA (Surface Heat Budget

<sup>9</sup> In equation B2, if  $hf << k_BT$ , then the exponential function can be simplified to  $1 + hf/k_BT$ 

of the Artic Ocean) and the North Slope of Alaska ARM Atmospheric Radiation
 Measurement) site [Tobin et al. 1999].

3

#### 4 5.2.2. Instrument Design

5

6 In its basic version, the ABB-Bomem MR100 (hereafter MR100) is an FTS with a liquid nitrogen cooled sandwich MCT/InSb detector to cover the 500 to 5000 cm<sup>-1</sup> spectral region. 7 More precisely the detector provides sensitivity for 5 - 20 µm (MCT) and 2 - 5.5 µm (InSb) in 8 two separate channels. The spectral sampling is  $0.4822 \text{ cm}^{-1}$  for double-sided interferograms. 9 10 The absolute infrared spectral radiance of the sky is measured and calibrated using two 11 accurately (±0.01 K) stabilized infrared black body sources. A typical measurement cycle 12 consists of a sky dwell period followed by two dwell periods, one for each of the blackbodies. 13 The instrument field of view is about 40 mrad. The mirror movement is effected by means of 14 a flexi-pivot, which minimises wavefront tilt and shear.

- 15
- 16 5.2.3. Campaign measurements
- 17

18 The MR-100 instrument has been used during the Italian phase of the European AQUA 19 Thermodynamic Experiment (EAQUATE) [Taylor et al 2007] to acquire nearly continuous observations of atmospheric emitted spectral radiance in the range 500 to 2000 cm<sup>-1</sup> [Esposito 20 21 et al, 2007a]. During the campaign, atmospheric temperature and water vapour were profiled 22 by radiosondes, with a time sampling rate of approximately 2 hours. In addition, a microwave 23 radiometer was operated continuously with a time sampling rate of 5 minutes and a 24 ceilometer was used to detect the presence of clouds, with a time sampling rate of 20 seconds. 25 These additional data were used to provide the atmospheric state input to the forward model to yield synthetic spectral radiance for comparison with the observations. 26

-	
2	The comparison was aimed at testing the reliability of state-of-art line and continuum water
3	vapour absorption parameters in the far infrared. An example of this comparison is provided
4	in Figure 13, where various editions of the HITRAN database [Rothman et al., 1998, 2003,
5	2005], and two version of the newest MT_CKD Clough continuum model are compared
6	against the observations. While allowing us to check the quality of state-of-art line and
7	continuum absorption parameters, Figure 13 also highlights that there are still appreciable
8	differences among the various data base versions, which, in turn, lead us to conclude that
9	there is still need of research efforts towards the understanding of a portion of the spectrum
10	which plays a fundamental role in the Earth's radiation budget.
11	
12	[Figure 13 near here]
13	
14	A MR-100 spectrometer has recently been extended by the DIFA group and colleagues at
15	IMAAA <sup>10</sup> in Basilicata, Italy, to operate at longer wavelengths by the replacement of the
16	usual zinc selenide beamsplitter by a silicon one, and by adding a deuterated tri-glycine
17	sulphate (DTGS) detector. Further studies of the continuum using this extended instrument
18	are planned.
19	
20	<b>5.3</b> Measurement of FIR heating rates with TAFTS – The Tropospheric Airborne
21	Fourier Transform Spectrometer
22	
23	5.3.1 Instrument concept

<sup>&</sup>lt;sup>10</sup> IMAAA: Istituto di Metodologia per l'Analisi Ambientale

A novel radiometer, the Tropospheric Airborne Fourier Transform Spectrometer (TAFTS)<sup>11</sup>,
has been developed at Imperial College to operate specifically in the FIR on aircraft
platforms, with the capability to observe nadir and zenith radiances, and to derive heating
rates. The aim is to study the in-situ radiative properties of the upper troposphere. A general
description of the TAFTS instrument and performance may be found in Murray et al [2007].
An example of early data may be found in Cox et al, [2007].

7

### 8 5.3.2 Instrument design

9

10 TAFTS is a differential, dual-input, polarising Fourier Transform Spectrometer (FTS), of the 11 Martin-Puplett kind [Martin and Puplett 1969]. This type of interferometer makes 12 simultaneous differential measurements of spectrally resolved radiance from two input ports. 13 TAFTS employs a sampling system that is particularly simple, and which also yields 14 diagnostic information on the vibration environment of the instrument as it affects the optical 15 path (vibration is often a problem on aircraft deployments of a FTS, and can give rise to 16 'ghosts' of real spectral features appearing in shifted spectral positions in the measured 17 spectra). TAFTS covers a wide spectral range at high spectral resolution: currently it operates across two spectral bands covering 80 to 320 cm<sup>-1</sup> and 320 to 600 cm<sup>-1</sup>, but could operate up 18 to 1000 cm<sup>-1</sup> with other detectors and optical filters. The instrument has a un-apodised 19 resolution of 0.1 cm<sup>-1</sup> (see Bell [1972] for an explanation of 'apodisation', a process for 20 21 removing unwanted oscillations in the spectra measured by a FTS). The optical and detector 22 system employed in TAFTS provides two differential signals within each spectral band. The 23 cryogenic optical design makes use of high-efficiency non-imaging components such as light 24 pipes and concentrators. TAFTS is controlled by an on-board computer that employs a real-

<sup>&</sup>lt;sup>11</sup> For extensive information about TAFTS, see: <u>http://www.sp.ph.ic.ac.uk/tafts/</u>

1	time Linux operating system. The instrument is designed to be fully autonomous or capable of
2	remote control via the internet. Table 2 details the instrument specifications.
3	
4	[Figure 14 and Table 2 near here]
5	
6	TAFTS comprises four units. These are: the pointing optics (scene selection including
7	calibration targets); the interferometer (dual input combiner and beamsplitter, sampling laser,
8	moving mirror and associated drive); the cryostat (cooled detector housing); and the
9	electronics control box. Figure 14 shows a simple schematic diagram of the optical layout of
10	TAFTS, showing the first three of these units. Each of the two inputs can be steered to view
11	either a hot or ambient temperature blackbody or a sky view. The radiation from these views
12	are steered into the interferometer housing and combined at the polariser beam-combiner P1.
13	P2 is a second polariser at 45° to the first, which acts as the beamsplitter. L1, L2 and L3 are a
14	combination of off-axis parabola and spherical mirrors which define the field of view and a
15	virtual aperture cold stop. LN2 and He represent low-pass filters at 79 K and 4.2 K
16	respectively, mounted on the radiation shields of the cryostat. P3 is a third polariser, which
17	acts as an analyser to separate the two complementary outputs of the interferometer [Martin
18	and Puplett, 1969]. Dichroic filters are used to transmit/reflect the radiation to the separate
19	long-wave and short-wave detectors. These detector blocks consist of an off-axis mirror and
20	hyperbolic concentrator, designed to reject off-axis radiation, and an integrating cavity which
21	houses the photoconductor. Further details of each of the four units are given in Murray et al.
22	[2007], and may be found via the web site quoted in the footnote.
23	

**5.3.3 Laboratory performance** 

1	In order to measure the instrument spectral sensitivity and ground-based performance, TAFTS
2	was placed within a N <sub>2</sub> -purged polystyrene box with free-standing liquid nitrogen sources and
3	water vapour traps. The up-welling blackbody temperature was maintained at 337 K, and the
4	down-welling blackbody at 304 K. Interferograms were recorded, and the corresponding
5	spectra computed by complex Fourier transformation. The input blackbody temperature
6	difference was used to convert the raw spectral signal (a voltage) to a brightness temperature
7	scale, which was used to derive noise equivalent brightness temperature as a function of
8	frequency from the voltage noise fluctuations: this is shown in Figure 15. Both the longwave
9	and shortwave spectra are the sums of 30 2-second interferometer scans, equivalent in total to
10	a 60-second integration. The apodized spectral resolution is approximately 0.12 cm <sup>-1</sup> . The
11	broad absorption features apparent in the short-wave channel (panel b) are caused by the
12	window and polariser substrate absorption (polypropylene and mylar). The increase in noise
13	equivalent brightness temperature towards the edges of the bands is due to falling instrument
14	throughput at the edges of filters and other components.
14 15	throughput at the edges of filters and other components.
	throughput at the edges of filters and other components. [Figure 15 near here]
15	
15 16	
15 16 17	[Figure 15 near here]
15 16 17 18	[Figure 15 near here] 5.3.4 Flight performance
15 16 17 18 19	[Figure 15 near here] 5.3.4 Flight performance TAFTS has now been flown on a number of aircraft missions. As an example, in November
15 16 17 18 19 20	[Figure 15 near here] 5.3.4 Flight performance TAFTS has now been flown on a number of aircraft missions. As an example, in November and December 2002 TAFTS participated in the EMERALD II (Egrett Microphysics
15 16 17 18 19 20 21	[Figure 15 near here] 5.3.4 Flight performance TAFTS has now been flown on a number of aircraft missions. As an example, in November and December 2002 TAFTS participated in the EMERALD II (Egrett Microphysics Experiment with Radiation Lidar and Dynamics in the Tropics) campaign based in Darwin,
<ol> <li>15</li> <li>16</li> <li>17</li> <li>18</li> <li>19</li> <li>20</li> <li>21</li> <li>22</li> </ol>	[Figure 15 near here] 5.3.4 Flight performance TAFTS has now been flown on a number of aircraft missions. As an example, in November and December 2002 TAFTS participated in the EMERALD II (Egrett Microphysics Experiment with Radiation Lidar and Dynamics in the Tropics) campaign based in Darwin, Australia. Atmospheric spectra were observed during an aircraft ascent from 4.6 km to 13.3
<ol> <li>15</li> <li>16</li> <li>17</li> <li>18</li> <li>19</li> <li>20</li> <li>21</li> <li>22</li> <li>23</li> </ol>	[Figure 15 near here] <b>5.3.4 Flight performance</b> TAFTS has now been flown on a number of aircraft missions. As an example, in November and December 2002 TAFTS participated in the EMERALD II (Egrett Microphysics Experiment with Radiation Lidar and Dynamics in the Tropics) campaign based in Darwin, Australia. Atmospheric spectra were observed during an aircraft ascent from 4.6 km to 13.3 km, and measurements of the on-board TAFTS black body calibration targets were also taken.

spaced flight levels. In order to derive the heating rate (see Box 1), the net flux in the vertical
is obtained by differencing the up- and down-welling spectra, and the vertical gradient of this
net flux then calculated, using the observations at the different flight levels. This vertical flux
divergence is then used to derive the heating rate in the local atmosphere, using equations
{B3} and {B4}.

6

7 Some of the calibrated net radiance spectra measured during this campaign are reproduced in 8 Figure 16, which shows the net radiance measured at a number of flight levels between 7.5 and 13.55 km on the 19<sup>th</sup> November 2002: the heights are shown on the Figure. The 9 10 background slopes of these radiance spectra, of course, follow the shape of the Planck 11 function at the local air temperature. The increasing transparency of the many windows 12 between the strong water vapour lines as the aircraft flies at higher levels is clear. Bearing in 13 mind that the recording time for the spectra shown varies between 2 and 10 s, the signal to 14 noise ratio is excellent. The heating rates have been derived from these spectra in the manner 15 described in Box 1 and section 2.3, and an example of early data is shown in Figure 17 16 [Straine, 2005]. Initial comparisons with simulated heating rates show these measurements to 17 be similar, except that the effect of variations with height in the vertical distribution of water 18 vapour causes very significant variability in the heating rates. For this reason, direct 19 observations are essential to establish that we understand this sensitivity. Further work on 20 heating rate determination from TAFTS is being prepared for publication. 21 22 [Figure 16 near here] 23 24 [Figure 17 near here] 25

26

5.4

### **REFIR - Radiation explorer in the far infrared**

1	
2	5.4.1 Instrument concept
3	
4	A European consortium has proposed the flight of a FIR spectrometer plus ancillary
5	instrumentation in low Earth orbit. The concept of the REFIR space mission was developed
6	to address several important issues relating the climate and the hydrological cycle, exploiting
7	the FIR [Rizzi et al., 2000; Rizzi et al., 2001].
8	
9	In the initial REFIR design, the FTS was a polarizing interferometer with a novel optical
10	scheme with double-port configuration (Figure 18) that is capable of resolving the upwelling
11	radiation from the Earth, seen from the low Earth orbit, with sufficient spectral resolution (0.5
12	cm <sup>-1</sup> ) and signal-to-noise ratio (SNR>100) in broadband operations (100-1100 cm <sup>-1</sup> ), to allow
13	scientific investigations of the atmosphere and climate.
14	
15	[Figure 18 near here]
16	
17	Figure 18 shows the optical configuration, described in more detail in Carli et al., [1999],
18	which optimises the reliability of the spectrometer in particular for long lifetime space
19	applications, and its performance for ambient temperature operations. Two prototypes of the
20	REFIR FTS have been developed for laboratory testing and use in field operations [Serio et
21	al.,, 2002]: REFIR-BB (Breadboard) for ground-based campaigns [Palchetti et al., 2005] and
22	REFIR-PAD (Prototype for Applications and Development) for air-borne campaigns
23	[Bianchini, 2006 a and b].
24	
25	5.4.2 Prototype characteristics and performance
26	

1	The two prototypes have been used to study different solutions. The main design requirement
2	of wideband coverage with uncooled operation can be met by using efficient wideband beam
3	splitters (BS) and high-performance room-temperature detectors. The detectors chosen were
4	doped tri-glycine sulphate (DLATGS) pyroelectric effect detectors, with low noise equivalent
5	spectral radiance (NESR). Polariser BS were developed and tested, but efficiency was found
6	to be poor, especially above 700 cm <sup>-1</sup> . Best performances were obtained with amplitude BS
7	(using a thin germanium (Ge) film deposited on Polyethylene Terephthalate (PET)
8	membrane) in combination with the pyroelectric detectors. In this case, the input/output
9	polariser (the one at 45° in Figure 18) is removed and the interferometer configuration
10	becomes the Mach-Zehnder scheme [Jenkins and White, 1981]. The other optical
11	characteristics, such as the double-port configuration, and the full tilt compensation of the
12	moving mirror are maintained.
13	
14	Table 3 provides a summary of the main parameters that characterise the prototype design.
15	
16	[Table 3 near here]
17	
18	Both prototypes were characterised under vacuum and in laboratory conditions. The noise was
19	dominated by the detector component and was found to be constant under different working
20	conditions. Figure 19 shows, for instance, the NESR measured under vacuum with REFIR-
21	PAD (operated with Ge-coated PET BS and 30 s acquisition time) and REFIR-BB (operated
22	with polarising BS and 120 s acquisition time). This performance was also obtained during
23	the field operations described in the following section.
24	
25	[Figure 19 near here]
26	

- 1 5.4.3 Ground based and stratospheric balloon observations
- 2

3 The REFIR-BB instrument was tested in June 2004 in a ground-based field campaign held at 4 Toppo di Castegrande at 1285 m altitude in South Italy [Esposito, 2007a and b]. The 5 downward atmospheric emission spectrum was measured and compared with that measured by a commercial FTS (BOMEM MR100) measuring above 500cm<sup>-1</sup>, and is shown in Figure 6 20. This shows the atmosphere to be opaque below about  $450 \text{ cm}^{-1}$ , so that radiance curve 7 follows the Planck radiation law for some local temperature. Between 450 and 620 cm<sup>-1</sup> the 8 9 partially transparent window between the water vapour pure rotational band and the  $CO_2 v_2$ band shows some transparency, before the intense centre of the  $v_2$  band again causes the 10 atmosphere to be opaque. On the high frequency side of the  $v_2$  band the downwelling 11 12 radiation rapidly falls to much lower intensity in the atmospheric window. The lower panel of 13 Figure 20 shows that the two measured spectra agree to within about  $\pm$  5% error. 14 15 [Figure 20 near here] 16 17 The measurement was simulated for clear sky using the Line-by-Line Radiative Transfer 18 Model (LBLRTM: See Clough et al, [1992]; Clough and Iacono, [1995]), which was used to 19 compute spectra based on water vapour and temperature vertical profiles measured by 20 radiosonde observations during the mountain observations. 21 22 [Figure 21 near here] 23 24 Figure 21 shows the result expressed in terms of brightness temperature (BT), for the expanded spectral band 500-650 cm<sup>-1</sup>. This result demonstrates that REFIR-BB is able to 25 26 accurately measure the spectral structure in the water vapour rotational band from groundbased observations. Thus, this first attempt seems to demonstrate that the REFIR concept,
 using un-cooled components is feasible. This is a realisation of great importance to the
 feasibility of a future FIR satellite experiment.

4

5 REFIR-PAD was integrated in 2004 on board a gondola belonging to the Laboratoire de 6 Physique Moleculaire pour l'Atmosphere et l'Astrophysique (LPMAA-CNRS, France), which 7 also carried the balloon-borne simulator of the Infrared Atmospheric Sounding Interferometer 8 (IASI) instrument. The instruments were flown for the first time on the 30th of June, 2005 9 from the Timon airfield, near Teresina, located in the North-East Brazil (5°5'S, 42°52' W). 10 The flight reached the mean floating altitude of 34 km for about 8 h, and a total of 1080 11 spectra of the upwelling atmospheric emission spectrum were measured. 12 13 Figure 22 shows a typical result and the comparison with the measurement of the IASIballoon prototype sharing the same platform and covering the spectral range above 650 cm<sup>-1</sup>. 14 15 The results [Palchetti et al., 2006; Bianchini et al., 2006a and b] indicate agreement within a 16 few percent between the observation and the simulation, and therefore confirm that the 17 different components of the Earth's OLR, including the FIR, can be detected with a single uncooled instrument using uncooled detectors, with a NESR of about 2 mW ( $m^2$  sr cm<sup>-1</sup>)<sup>-1</sup> 18 (see Figure 19), and an absolute calibration error which is less than 0.5 K at 280 K. 19 20 21 [Figure 22 near here] 22 23 These developments and results for REFIR clearly indicate that the feasibility is very high of developing a FIR spectrometer with moderately high spectral resolution (0.25 cm<sup>-1</sup>: see Table 24 25 3) for accurate measurements of the FIR spectrum of the Earth from space. 26

1

5.5

# The Far-Infrared Spectroscopy of the Tropsophere (FIRST) Instrument

### 3 5.5.1 Instrument Concept

4

The FIRST instrument was developed by the NASA Langley Research Center, USA, in 5 6 partnership with the Space Dynamics Laboratory (Utah State University) and the Harvard 7 Smithsonian Center for Astrophysics, both also from the USA: the project was funded by 8 NASA. FIRST is a Michelson Fourier Transform Spectrometer (FTS), designed to demonstrate the ability to measure the spectrum between 1000 and 100 cm<sup>-1</sup> (10 to 100  $\mu$ m) 9 10 globally on a daily basis at high spatial resolution (10 km from low earth orbit), high spectral resolution (0.625 cm<sup>-1</sup>), and high temporal resolution (one spectrum every 1.4 sec), on a 11 12 single focal plane. A critical aspect of the FIRST design is the ability to measure the spectrum 13 between 10 and 15 µm in addition to the far-IR spectrum from 15 to 100 µm. The former 14 interval allows for direct comparison with conventional mid-infrared spectral sensors for the 15 purpose of calibration and validation, which is essential for confidence in the calibration of 16 the far-IR part of the spectrum for which accurate radiometric standards are not as well-17 developed.

18

# 19 5.5.2 Instrument Design

20

The FIRST instrument consists of a scene select mirror, the FTS, aft optics, a detector
assembly, and associated electronics [Mlynczak et al., 2005]. The FTS and aft optics are
cooled to ~180 K by liquid nitrogen, to reduce background and simulate spacecraft
conditions. The detectors are cooled to 4.2 K, and the rest of the instrument is at ambient
temperature. Thin polypropylene windows isolate the cold FTS optics from the scene select
mirror and from the detector cryostat. During balloon flights the scene select mirror alternates

1 between a nadir view of the Earth, a high elevation angle space view that is used to estimate 2 instrument background, and an ambient-temperature blackbody calibration source. The FTS is 3 a compact plane mirror Michelson interferometer that achieves very high throughput (0.47  $cm^2 sr$ ) with a modest 7 cm diameter beam. Broadband response (50-2000 cm<sup>-1</sup>) is made 4 5 possible by a bilayer thin-film beamsplitter. FTS scanning and detector sampling are 6 controlled by a separate metrology laser interferometer that monitors the position of the scan 7 mirror. Interferometer alignment (for both the infrared and laser interferometers) can be 8 adjusted if necessary during the balloon flight by remotely controlling the tip and tilt of the 9 non-scanning interferometer mirrors.

10

11 The FTS scans over optical path differences of  $\pm 0.8$  cm to achieve the nominal unapodized resolution of  $0.625 \text{ cm}^{-1}$ ; the scan time varies from 1.4 to 8.5 s, depending on the detector 12 13 sample interval. Trimming and centering the interferograms reduces the realized unapodized resolution to 0.643 cm<sup>-1</sup>, which is the resolution of all the FIRST data presented in this paper. 14 15 The aft optics focus the collimated FTS output onto an array of Winston cone concentrators, 16 coupled to discrete silicon bolometers in individual integrating cavities. The focal plane is 17 37.5 x 37.5 mm, large enough for a 10x10 array of cones and detectors, although for the 18 demonstration flight the focal plane was populated with a total of 10 cones (2 in each corner 19 and 2 in the center.) The focal plane is sized to demonstrate the technology required to obtain 20 daily global coverage from a cross-track scanning instrument on an orbiting satellite. The 100 21 detector focal plane, with each detector recording a spectrum every 1.4 seconds, and scanned 22  $\pm$  48 deg from nadir, would achieve daily global coverage, approximately 5 million spectra 23 per day, from a low-earth orbiting satellite.

24

To simulate the space observation configuration, FIRST was designed to operate from a
gondola on a high altitude balloon. The instrument first flew on June 7, 2005, and achieved an

1	altitude of 28 km, lifted by an 11 million cubic foot helium-filled balloon. The results from
2	this flight are reported by Mlynczak et al. [2006]. A second flight occurred on September 18,
3	2006, and achieved an altitude of 33 km. Both flights were successful in recording
4	approximately 6 hours of data at the balloon float altitude, simulating operation in and
5	observations from a space-like environment. The flights occurred from the NASA Columbia
6	Scientific Balloon Facility located in Ft. Sumner, New Mexico. In addition, both flights
7	occurred with a simultaneous overpass of the NASA Aqua satellite at approximately 2:00
8	p.m. local time, offering the opportunity to compare directly spectra measured by the AIRS
9	instrument in the mid-infrared part of the spectrum common to both sensors, and to compare
10	the measured far-IR spectrum with those computed from AIRS temperature and moisture
11	profiles. With the exception of the latter part of the 2006 flight, both flights observed clear-
12	sky conditions.
13	
-	
14	[Figure 23 near here]
	[Figure 23 near here]
14	[Figure 23 near here] Figure 23 shows a radiance spectrum recorded by FIRST from the June, 2005 flight. The
14 15	
14 15 16	Figure 23 shows a radiance spectrum recorded by FIRST from the June, 2005 flight. The
14 15 16 17	Figure 23 shows a radiance spectrum recorded by FIRST from the June, 2005 flight. The spectrum clearly shows the complex spectral structure of the far-IR (see earlier in this paper)
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<ol> <li>14</li> <li>15</li> <li>16</li> <li>17</li> <li>18</li> <li>19</li> <li>20</li> <li>21</li> <li>22</li> </ol>	Figure 23 shows a radiance spectrum recorded by FIRST from the June, 2005 flight. The spectrum clearly shows the complex spectral structure of the far-IR (see earlier in this paper) with alternating regions of strong and weak lines, evidenced by the variation in radiance (or equivalently, in the brightness temperature) of the lines. The spectrum covers the spectral region from approximately 2000 cm <sup>-1</sup> (5 $\mu$ m) to wavenumbers less than 100 cm <sup>-1</sup> . There are some gaps in the spectrum which correspond to regions in which the FIRST beamsplitter exhibits significant absorption, and whose effects have not yet been corrected. Otherwise,
<ol> <li>14</li> <li>15</li> <li>16</li> <li>17</li> <li>18</li> <li>19</li> <li>20</li> <li>21</li> <li>22</li> <li>23</li> </ol>	Figure 23 shows a radiance spectrum recorded by FIRST from the June, 2005 flight. The spectrum clearly shows the complex spectral structure of the far-IR (see earlier in this paper) with alternating regions of strong and weak lines, evidenced by the variation in radiance (or equivalently, in the brightness temperature) of the lines. The spectrum covers the spectral region from approximately 2000 cm <sup>-1</sup> (5 $\mu$ m) to wavenumbers less than 100 cm <sup>-1</sup> . There are some gaps in the spectrum which correspond to regions in which the FIRST beamsplitter exhibits significant absorption, and whose effects have not yet been corrected. Otherwise, Figure 23 confirms not only the complexity of the far-IR structure, but also shows the ability

1 [Figure 24 near here] 2 3 Figure 24 presents a comparison of a FIRST brightness temperature spectrum and a spectrum 4 from the AIRS instrument recorded during the June 2005 flight. The agreement between the 5 two spectra, typically to within 1-2 K, demonstrates remarkable consistency between the two 6 sensors, especially considering that from balloon flight altitude the FIRST footprint is about 7 200 meters diameter while the AIRS sensor is 14 km diameter, nearly 5000 times larger than 8 FIRST. The comparison illustrated here validates the calibration of the FIRST sensor. 9 10 [Figure 25 near here] 11 12 In Figure 25 we show far-IR spectra recorded about 2 p.m. local time from the 2005 and 2006 13 flights. The radiances recorded in 2006 are clearly smaller than in 2005 from 400 to 600 cm<sup>-1</sup>, 14 by as much as 15% (see black difference spectrum). Based on the brightness temperatures in 15 Figure 23, the radiation in this interval comes from the lower troposphere between 600 and 16 850 hPa. The temperature and moisture profiles recorded by the AIRS instrument 17 simultaneously with the FIRST spectra indicate that the lower troposphere between 700 and 18 850 hPa is 5 to 25 K colder during 2006 than in 2005, while the mass mixing ratios of water vapor are slightly larger. At 600 cm<sup>-1</sup>, a decrease of 15 K (from 270 K) results in a decrease in 19 20 radiance of approximately 20%. We therefore attribute the smaller radiances observed in 2006 21 primarily to a cooler lower troposphere. 22 23 The spectra from both FIRST flights have been processed and approximately 16,000 nearly 24 contiguous, nadir view spectra of the Earth are now available (approximately 9000 spectra 25 from the 2005 flight and 7000 spectra from the 2006 flight). The spectra may be obtained 26 from one of the authors (MGM).

## 2 **6.** The future

3

4 We have seen from this review that the Earth is, indeed, a far infrared planet. Because of the 5 heavily absorbing atmosphere at FIR wavelengths, the energy that is emitted to space comes 6 from layers well above the surface, which are colder than the surface. Because the planet 7 presents a colder emission temperature to space, so the peak black body energy is emitted at 8 longer wavelengths, further into the FIR, than the emission from the surface. Conversely, the 9 opacity of the atmosphere in the FIR is a major contributory process in keeping our planet's 10 surface far more comfortable for mankind than otherwise. While it is unnecessary to define a 11 specific arbitrary border between the FIR and elsewhere in the IR, it is vital to recognise that 12 to fully understand the energy budget of the Earth, and to fully exploit the spectrum for 13 remote sounding, a better knowledge of the longwave part of the spectrum, the FIR, is 14 mandatory.

15

16 Up to recent times, our knowledge of the FIR, and our ability to explore the region, in the 17 laboratory, in the field and in space, has been limited. However, as this review has shown, the 18 last decade or two has been a period of major progress, in spectroscopy, in radiometry, in 19 technology, and in measurements on the real atmosphere. We are now achieving 20 measurements in the FIR with differing configurations (balloon, aircraft and ground based) 21 that have a spectral resolution suitable for the observation of the atmospheric features, over a 22 wide spectral range, and that are suitable for the characterisation of the radiation budget in 23 clear skies, and in the presence of clouds. Preliminary comparisons of the measurements with 24 models show that we have a good understanding of the radiative transfer in this spectral 25 region.

1 Now, as we have seen in this review, we have several new instruments (TAFTS, REFIR, 2 FIRST) that are capable of observing the far-IR from suborbital (aircraft and balloon) 3 platforms. These instruments have also operated from the ground, observing in the zenith 4 view the downwelling radiation from the atmosphere. Further experiments are planned from aircraft, balloons, and the surface (especially at cold sites, near the poles, or on mountains). 5 6 We are also using these instrumental developments to set the scene for future FIR spectrometers that can be flown as part of all-IR instruments, in space. 7 There are still 8 experimental difficulties, of course. We need more efficient, un-cooled detectors for the FIR, 9 to remove the inconvenience of using liquid cryogens; more use of advanced cooling engines, 10 when cooling is unavoidable; beam-splitter technology has some way to go to maximise 11 efficiency; and there is always the hope of new, more efficient spectrometer designs.

12

Of course, scientific problems also remain. One of the most obvious spectroscopic problems 13 that needs much further work is that of the water vapour continuum: this effect is still the 14 15 major source of uncertainty in gaseous radiative transfer in the FIR. Because models for 16 continuum absorption are constrained by laboratory and atmospheric measurements, it is 17 important to improve the quantity and quality of spectrally resolved observation in the H<sub>2</sub>O rotational band. Along this line there are programmes in both Italy (the ECOWAR<sup>12</sup> project) 18 and the UK (the CAVIAR<sup>13</sup> project) which seek to make a concerted effort though laboratory 19 20 and field measurements, theory and modelling, to put our understanding of this phenomenon 21 on a much firmer basis.

<sup>&</sup>lt;sup>12</sup> ECOWAR: Earth Cooling by Water Vapour Radiation: see: www.difa.unibas.it

<sup>&</sup>lt;sup>13</sup> CAVIAR: Continuum Absorption at Infrared wavelengths and its Atmospheric Relevance: see: <u>http://www.met.rdg.ac.uk/~sws01ivp/</u>

1 A further scientific and observational problem is that we need to disentangle the interference 2 between gas and cloud properties through a chain of validation activities. With a well 3 calibrated instrument it will be possible to complete the on-going assessment of the 4 spectroscopy of the clear atmosphere (line strengths; pressure broadening coefficients; line mixing effects; water vapour continuum). While cloud properties have been mainly derived 5 6 from the mid infrared, Evans et al. [1999] have demonstrated that complementary information on cloud properties can be obtained in the very FIR (sub-millimetre) region. Spectrally 7 8 resolved observations of the cloud in a broad spectral band that includes the FIR could 9 provide exhaustive information on cloud composition, phase, altitude, IWP and equivalent 10 particle size. The problem of cirrus cloud studies is twofold: to characterise their climatology, 11 and to verify the mechanisms that govern the radiative exchange of the different types of cirri. 12 In both cases field measurements are needed and observations in the FIR can provide a new and powerful measurement capability which fills a long due observation requirement. 13

14

15 We have shown through this review that, as a result of the research described, we now know 16 far more than even just a decade ago about how to make measurements on the real 17 atmosphere, and how to start the job of disentangling the absorption and emission by 18 molecules, and the corresponding processes in clouds. We know more about how the FIR 19 contributes to the TOA energy balance of the Earth, and the role that water vapour and ice 20 clouds play in that balance. Knowing about these processes means that we are closer to being 21 able to understand more accurately how global warming might affect the Earth. Finally, we 22 have a much better idea about how to build the sensitive instruments that we need, and can 23 now start to look forward with more expertise than before to the prospect of making 24 measurements of our planet in the FIR from space.

Turning to the future, we note that it has been approximately 30 years since the last spectra of the Earth at wavelengths longer than 15 μm were recorded from an instrument on an orbiting satellite [Spankuch and Dohler, 1985], and nearly four decades since the Nimbus IV satellite produced such amazing Earth IR emission spectra [Hanel et al., 1970; 1971]. In that time the importance of the far-IR has been fully realized, especially with respect to the role it plays in climate, and also in offering the possibility for the remote sensing of cirrus optical properties from space.

8

9 For space based observations it now seems obvious that we must monitor our planet's state in 10 the FIR: a far-IR instrument should be flown as soon as possible. The recent Decadal Survey 11 [National Research Council, 2007] conducted by the U.S. National Research Council 12 recommends as its highest priority a mission with an instrument capable of covering 200 to 2000 cm<sup>-1</sup> (5 to 50 µm) with a 1 cm<sup>-1</sup> spectral resolution. The instrument must be 13 14 exceptionally well-calibrated (and stable) to detect small changes with time in the spectral 15 distribution of outgoing longwave radiation. After many years of investment we are on the 16 doorstep of a far more penetrating understanding of the FIR properties of our Earth, and of 17 realizing a measurement from space that is destined to revolutionize our knowledge of the 18 Earth's climate system.

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- 22

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2

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1	Figure Captions	
2		
3	B1: Optical depth of lines due to water vapour and carbon dioxide throughout the IR. (T.	
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5		
6	B2: Optical depth of lines due to remaining key absorbers throughout the IR. (T. Maestri,	
7	PhD Thesis)	
8		
9	B3: Simulated upwelling emission spectrum at the TOA, for (a) Tropical, and (b) Sub-Arctic	
10	Winter standard atmospheres. Dashed red lines show the equivalent black-body emission at	
11	the given temperatures (H. Brindley, unpublished)	
12		
13	1: Transmittance in FIR: (a) Tropical standard, 12 km to TOA; (b) Tropical standard: surface	
14	to TOA; (c) Sub-Arctic Winter standard, 8 km to TOA; (d) Sub-Arctic Winter standard:	
15	surface to TOA (H. Brindley, unpublished)	
16		
17	2: Cooling rate across the infrared and from the surface to TOA, in units of K day <sup>-1</sup> cm <sup>-1</sup> . (a)	
18	Tropical (TRP) and (b) Sub-Arctic Winter (SAW) standard atmospheres (Figure originally	
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20	'The impact of far ir absorption on clear sky greenhouse forcing: Sensitivity studies at high	
21	spectral resolution', by Brindley and Harries, pp 151-180, Copyright Elsevier (1998))	
22		
23	3: Radiance difference at the TOA in W m <sup>-2</sup> sr <sup>-1</sup> (cm <sup>-1</sup> ) <sup>-1</sup> . Dashed line is the difference	
24	between case A (Tropical standard atmosphere with water vapour concentration scaled by 0.9	
25	and skin temperature decreased by 1K) and control case C (Tropical standard atmosphere).	

Solid line is the difference between case B (Tropical standard atmosphere with only water 1 2 vapour concentration scaled by 0.9) and control case C (T. Maestri and R. Rizzi, unpublished) 3 4: (a) Energy balance in mW m<sup>-2</sup> (cm<sup>-1</sup>)<sup>-1</sup> versus wave-number in cm<sup>-1</sup> for a 1.8 km thick 4 tropical cirrus; ice water path (IWP) is expressed in  $g m^{-2}$ . Lower panels: vertical profiles of 5 net flux in W m<sup>-2</sup> for the same clouds and for clear sky: (b) FIR band (10-600 cm<sup>-1</sup>); (c) 6 Window (WIN) band (750-980 and 1100-1250 cm<sup>-1</sup>) (Adapted from Maestri and Rizzi, 2003, 7 copyright 2003, American Geophysical Union) 8 9 5: Variation of surface emission,  $E_{\nu}$ , at surface temperature indicated, and greenhouse 10 11 parameter,  $G_{\nu}$  as function of wavenumber for (a) Tropical and (b) Sub-Arctic Winter 12 standard atmospheres (Figure originally published in Journal of Quantitative Spectroscopy 13 and Radiative Transfer, Vol. 60., entitled: 'The impact of far ir absorption on clear sky 14 greenhouse forcing: Sensitivity studies at high spectral resolution', by Brindley and Harries, 15 pp 151-180, Copyright Elsevier (1998)) 16 6: Top panel: Spectral greenhouse parameter,  $G_{\nu}$ ; Bottom panel: Normalized spectral 17 greenhouse parameter,  $g_{\nu}$ , for a set of tropical cirrus clouds comprised of aggregates and 18 placed between 9-12 km in altitude. The cloud optical depth, OD at 900 cm<sup>-1</sup> is varied as 19 20 indicated. The legend in the top panel applies to both plots. (T. Maestri, adapted from PhD 21 thesis) 22 23 7: D as a function of the cloud transmittance. Three tropical cirri (top altitudes are 15.4 (h), 24 12.3 (m) and 9.7 (l) km) and a sub-arctic one (top altitude at 8.3 (a) km) are considered with a filter cut-off at 350 cm<sup>-1</sup> (From Rizzi and Maestri, 2003, Copyright, 2003, American 25 26 Geophysical Union)

2	8: Absolute value of the water vapour Jacobian ( $W/m^2$ -cm <sup>-1</sup> -sr-(g/kg)) for a tropical
3	atmosphere. The upper panel refers to the rotational band. For comparison the same
4	computation, shown in the lower panel, has been performed for the vibrational band at 6.7 $\mu$ m
5	(C. Serio, unpublished)
6	
7	9: Comparison of the accuracy of the $H_2O$ retrieval of FIR rotational band 30 km aggregate
8	resolution (solid line) with that of a MIR sounder (dash-dotted curve) for a standard tropical
9	atmosphere (C. Serio, unpublished)
10	
11	10: Extinction coefficient for spherical ice particles of different radii as a function
12	of wavenumber for a constant IMC (Ice Mass Content) equal to1.E-8 g cm <sup>-3</sup> (B. Carli,
13	unpublished)
14	
15	11: Changes in the wide-band TOA emission spectrum, including both the FIR and the MIR
16	spectral regions, due to the different properties of clouds.
17	(a) simulated TOA spectra for a clear atmosphere and for two types of clouds with different
18	phase (water and ice), but equal particle size (radius 10 microns) and equal ice water path
19	(IWP=4.0 g m <sup>-2</sup> ) in the same atmospheric layer from 5 to 6 km.
20	(b) Differences between the clear and cloudy spectra shown in (a). The ice cloud produces a
21	larger difference in the MIR and a smaller difference in the FIR.
22	(c) Differences between clear and cloudy spectra for ice clouds located at different altitudes
23	(for equal particle size of 30 microns and equal IWP (= $4.0 \text{ g m}^{-2}$ ).
24	(d) Differences between clear and cloudy spectra for ice clouds with different IWP (for equal
25	particle size of 30 microns and the same altitude layer from 7 to 8 km).

1	(e) Differential effect of particle size for changes of $\pm 10\%$ in the particle radius for an ice
2	cloud between 7 and 8 km with an IWP of $4.0 \text{ g m}^{-2}$ .
3	(f) Difference in observed radiance for the case of different particle shapes when the other
4	variables are kept constant (ice cloud between 7 and 8 km with an IWP of 4.0 g $m^{-2}$ and equal
5	mass of the single particle).
6	(Private communication from S. Del Bianco, 2007)
7	
8	12: Emission spectrum of the atmosphere observed in the far infrared from an altitude of 38
9	km with a zenith angle of 91.6°. The upper panel shows a synoptic view of a 20 cm <sup>-1</sup> spectral
10	interval and the lower panel shows an expanded view of a 2 cm <sup>-1</sup> spectral interval. The
11	observed features are due to the rotational spectra of the atmospheric constituents. (Carli,
12	unpublished in this form)
13	
14	13: (a) Comparison of MR-100 spectral observation against various computations. H-00, H-04
15	and H-06 stand for HITRAN 2000 [Rothman et al, 1998], 2004 [Rothman et al, 2003] and
16	2006 [Rothman et al, 2005], respectively. Computations have been obtained with LBLRTM

v9.4 [Clough et al. 1992, Clough and Iacono 1995], with the water vapour continuum model
MT\_CKD [Tobin et al. 1999]. The version 1.0 of this model is used in conjunction with H-00,

19 while for H-04 and H-06 the version 1.2 has been used. (b) Observed-calculated differences 20 and  $\pm 1 \sigma$  error bars (Serio and Masiello, unpublished).

21

22 14: Schematic of the TAFTS optical system. The instrument is, optically, divided into three

23 sections; the pointing optics box (POB), main interferometer unit (MI) and cryostat (CRYO).

24 The MI and CYRO sections are evacuated. The passage of the radiation is described in the

25 main text. The labeled elements are three polarizers, P1 (the beamcombiner, BC), P2 (the

26 beamsplitter, BS) and P3, the power optics, L1, L2 and L3, and the four detectors, either in

2 one on each of the liquid nitrogen (LN2) and Helium (He) jackets (P.Green and J.Murray, 3 unpublished) 4 15: Noise equivalent temperature for single output (a) longwave and (b) shortwave TAFTS 5 channels. The integration time is 60 s at 0.12 cm<sup>-1</sup> resolution. Blackbody target temperatures 6 7 were at 337 K and 304 K in the up-welling and down-welling arms respectively (J. Murray, 8 unpublished) 9 10 16: Net radiance spectra measured by TAFTS during EMERALD II at altitudes between 7.50 and 13.55 km (marked on figure). Data taken on 19/11/02. The spectral resolution is 1 cm<sup>-1</sup>. 11 12 The growth of the transparent windows between individual water vapour absorption lines is 13 shown. It is in these transparent regions that heating rates can be calculated from the vertical 14 divergence of the net flux measured at each altitude. (G. Straine, unpublished) 15 16 17: Atmospheric cooling rate derived from the spectra in Figure 15, in units of K (day cm<sup>-1</sup>)<sup>-1</sup> x  $10^3$ . The band structure is due to the positions of the water vapour lines 17 18 (G. Straine, PhD thesis) 19 20 18: Conceptual optical scheme of the REFIR interferometer. Scanning of the optical path 21 difference is achieved in an efficient double sided Roof-top mirror unit (RTMU); Pol. 22 Indicates polarisers, with polarisation angle in brackets (From L. Palchetti, C Serio and B. 23 Carli, unpublished) 24 25 19: Noise Equivalent Spectral Radiance (NESR) as a function of wavenumber measured

the longwave (LW) and shortwave(SW) for channels 0-3 (CHi). Two long pass filters exist,

1

26 under vacuum on the two REFIR prototypes (L. Palchetti, C Serio and B. Carli, unpublished)

2	20: Outgoing Longwave Radiation (OLR) measurements in clear sky conditions from Toppo
3	di Castelgrande, Italy, on June 10-11 <sup>th</sup> , 2004. The comparison between REFIR-BB (blue line)
4	and BOMEM MR100 (red line) is shown in the top panel, and the difference in the bottom
5	panel. (L. Palchetti, C Serio and B. Carli, unpublished)
6	
7	21: Brightness temperature (BT) measured in clear sky conditions from Toppo di
8	Castelgrande, Italy, on June 10-11 <sup>th</sup> , 2004. The comparison between REFIR-BB (blue line),
9	BOMEM MR100 (red line), and LBLRTM (green line) is shown in the top panel. Differences
10	between REFIR-BB and BOMEM (labelled Diff RB. on y-axis), and between REFIR-BB
11	and LBLRTM (labelled Diff RL. on y-axis) are shown in the central and bottom panels,
12	respectively. (L. Palchetti, C Serio and B. Carli, unpublished)
13	
14	22: Measurement of OLR from 34 km altitude in clear sky conditions in tropical atmosphere
15	(Brazil, June 30 <sup>th</sup> , 2005). The comparison between REFIR-PAD (blue line) and IASI-balloon
16	(red line) is shown in the top panel, and difference in the bottom panel (L. Palchetti, C Serio
17	and B. Carli, unpublished)
18	
19	23: A single spectrum recorded by the FIRST instrument on June 7 2005, from an altitude of
20	28 km. The entire energetically significant portion of the Earth's thermal emission spectrum is
21	recorded by a single detector. (After Mlynczak et al., 2006. Copyright 2006 American
22	Geophysical Union. Reproduced by permission of the American Geophysical Union.)
23	
24	24: FIRST spectrum (black line) and two spectra (red, green lines) from the AIRS instrument
25	recorded June 7, 2005. The spectral range covers the entire span measured by the AIRS

1	instrument in the mid-infrared. This figure illustrates the agreement between FIRST and AIRS
2	and confirms the calibration of the FIRST instrument. (M. Mlynczak, unpublished)
3	
4	25: FIRST measurements of the far-IR portion of the Earth's emission spectrum from June 7,

5 2005 (blue) and September 18, 2006 (red). The black line shows the difference between the
6 June and September cases. The higher values in 2005 are due primarily to a much warmer
7 lower troposphere (M. Mlynczak, unpublished)

- 1 Annex 1: Examples of atmospheric constituents for which a vertical profile has been
- 2 measured using FIR spectroscopy

<ul> <li>"Heavy ozone distribution in the stratosphere from far infrared observations", Geophys. Res. 92 D11, 13231.</li> <li>M. Carlotti, A. Barbis, B. Carli, 1989, "Statospheric ozone vertical distribution from far-infrared balloon spectra and statistical analysis of the errors", J. Geophys. Res. 94, 16,365 - 16,372.</li> <li>HCl, HF</li> <li>Traub W.A., and Chance K.V., "Stratospheric HCl and HF observations" Geophys. Res. Lett. 8, 1075-1077 (1981).</li> <li>OH</li> <li>B. Carli, M. Carlotti, B.M. Dinelli, F. Mencaraglia, J.H. Park, 1989, "The mixing ratio of stratospheric hydroxyl radical from far-infrared emission measurements" J. Geophys. Res. 94, 11049 - 11058.</li> <li>CO</li> <li>M.M. Abbas, M.J. Glenn, I.G. Nolt, B. Carli, F. Mencaraglia, M. Carlotti, 198 "Far infrared measurements of stratospheric carbon monoxide", J. Geophys. Res. Lett. 15, 140.</li> <li>HNO<sub>3</sub></li> <li>F. Mencaraglia, G. Bianchini, A. Boscaleri, B.Carli, S. Ceccherini, P.Raspolli A. Perrin, J-M Flaud, "Validation of MIPAS satellite measurements of HNO3 using comparison of rotational and vibrational spectroscopy" J.G.R. 111, D19305, doi: 1029/2005JD006099, 2006.</li> </ul>	Species	Reference
distribution of H2 <sup>16</sup> O, H2 <sup>17</sup> O, H2 <sup>18</sup> O, and HDO from far-infrared spectra",         Geophys. Res. 96, 7509-7514.         O3         M.M. Abbas, J. Guo, B. Carli, F. Mencaraglia, M. Carlotti, I.G. Nolt, 198         "Heavy ozone distribution in the stratosphere from far infrared observations",         Geophys. Res. 92 D11, 13231.         M. Carlotti, A. Barbis, B. Carli, 1989, "Statospheric ozone vertical distribution         from far-infrared balloon spectra and statistical analysis of the errors", J.         Geophys. Res. 94, 16,365 - 16,372.         HCl, HF       Traub W.A., and Chance K.V., "Stratospheric HCl and HF observations"         Geophys. Res. Lett. 8, 1075-1077 (1981).         OH       B. Carli, M. Carlotti. B.M. Dinelli, F. Mencaraglia, J.H. Park, 1989, "The         mixing ratio of stratospheric hydroxyl radical from far-infrared emission         measurements" J. Geophys. Res. 94, 11049 - 11058.         CO       M.M. Abbas, M.J. Glenn, I.G. Nolt, B. Carli, F. Mencaraglia, M. Carlotti, 198         "Far infrared measurements of stratospheric carbon monoxide", J. Geophys.         Res. Lett. 15, 140.       F. Mencaraglia, G. Bianchini, A. Boscaleri, B.Carli, S. Ceccherini, P.Raspolli         M. Perrin, J-M Flaud, "Validation of MIPAS satellite measurements of HNO3       using comparison of rotational and vibrational spectroscopy" J.G.R. 111,         D19305, doi: 1029/2005JD006099, 2006.       HO6		
Geophys. Res. 96, 7509-7514.         O3       M.M. Abbas, J. Guo, B. Carli, F. Mencaraglia, M. Carlotti, I.G. Nolt, 199         "Heavy ozone distribution in the stratosphere from far infrared observations",         Geophys. Res. 92 D11, 13231.         M. Carlotti, A. Barbis, B. Carli, 1989, "Statospheric ozone vertical distribution         from far-infrared balloon spectra and statistical analysis of the errors", J.         Geophys. Res. 94, 16,365 - 16,372.         HC1, HF       Traub W.A., and Chance K.V., "Stratospheric HC1 and HF observations"         Geophys. Res. Lett. 8, 1075-1077 (1981).         OH       B. Carli, M. Carlotti. B.M. Dinelli, F. Mencaraglia, J.H. Park, 1989, "The         mixing ratio of stratospheric hydroxyl radical from far-infrared emission         measurements" J. Geophys. Res. 94, 11049 - 11058.         CO       M.M. Abbas, M.J. Glenn, I.G. Nolt, B. Carli, F. Mencaraglia, M. Carlotti, 198         "Far infrared measurements of stratospheric carbon monoxide", J. Geophys.         Res. Lett. 15, 140.         HNO3       F. Mencaraglia, G. Bianchini, A. Boscaleri, B.Carli, S. Ceccherini, P.Raspolli         A. Perrin, J-M Flaud, "Validation of MIPAS satellite measurements of HNO3         using comparison of rotational and vibrational spectroscopy" J.G.R. 111,         D19305, doi: 1029/2005JD006099, 2006.	$H_2O$	
O3       M.M. Abbas, J. Guo, B. Carli, F. Mencaraglia, M. Carlotti, I.G. Nolt, 198         "Heavy ozone distribution in the stratosphere from far infrared observations", Geophys. Res. 92 D11, 13231.       M. Carlotti, A. Barbis, B. Carli, 1989, "Statospheric ozone vertical distribution from far-infrared balloon spectra and statistical analysis of the errors", J. Geophys. Res. 94, 16,365 - 16,372.         HCl, HF       Traub W.A., and Chance K.V., "Stratospheric HCl and HF observations" Geophys. Res. 124, 1075-1077 (1981).         OH       B. Carli, M. Carlotti. B.M. Dinelli, F. Mencaraglia, J.H. Park, 1989, "The mixing ratio of stratospheric hydroxyl radical from far-infrared emission measurements" J. Geophys. Res. 94, 11049 - 11058.         CO       M.M. Abbas, M.J. Glenn, I.G. Nolt, B. Carli, F. Mencaraglia, M. Carlotti, 198         "Far infrared measurements of stratospheric carbon monoxide", J. Geophys. Res. Lett. 15, 140.         HNO3       F. Mencaraglia, G. Bianchini, A. Boscaleri, B.Carli, S. Ceccherini, P.Raspolli A. Perrin, J-M Flaud, "Validation of MIPAS satellite measurements of HNO3 using comparison of rotational and vibrational spectroscopy" J.G.R. 111, D19305, doi: 1029/2005JD006099, 2006.		distribution of $H_2^{16}O$ , $H_2^{17}O$ , $H_2^{18}O$ , and HDO from far-infrared spectra", J.
<ul> <li>"Heavy ozone distribution in the stratosphere from far infrared observations", Geophys. Res. 92 D11, 13231.</li> <li>M. Carlotti, A. Barbis, B. Carli, 1989, "Statospheric ozone vertical distribution from far-infrared balloon spectra and statistical analysis of the errors", J. Geophys. Res. 94, 16,365 - 16,372.</li> <li>HCl, HF</li> <li>Traub W.A., and Chance K.V., "Stratospheric HCl and HF observations" Geophys. Res. Lett. 8, 1075-1077 (1981).</li> <li>OH</li> <li>B. Carli, M. Carlotti. B.M. Dinelli, F. Mencaraglia, J.H. Park, 1989, "The mixing ratio of stratospheric hydroxyl radical from far-infrared emission measurements" J. Geophys. Res. 94, 11049 - 11058.</li> <li>CO</li> <li>M.M. Abbas, M.J. Glenn, I.G. Nolt, B. Carli, F. Mencaraglia, M. Carlotti, 198 "Far infrared measurements of stratospheric carbon monoxide", J. Geophys. Res. Lett. 15, 140.</li> <li>HNO<sub>3</sub></li> <li>F. Mencaraglia, G. Bianchini, A. Boscaleri, B.Carli, S. Ceccherini, P.Raspolli A. Perrin, J-M Flaud, "Validation of MIPAS satellite measurements of HNO3 using comparison of rotational and vibrational spectroscopy" J.G.R. 111, D19305, doi: 1029/2005JD006099, 2006.</li> </ul>		Geophys. Res. 96, 7509-7514.
<ul> <li>Geophys. Res. 92 D11, 13231.</li> <li>M. Carlotti, A. Barbis, B. Carli, 1989, "Statospheric ozone vertical distribution from far-infrared balloon spectra and statistical analysis of the errors", J. Geophys. Res. 94, 16,365 - 16,372.</li> <li>HCl, HF</li> <li>Traub W.A., and Chance K.V., "Stratospheric HCl and HF observations" Geophys. Res. Lett. 8, 1075-1077 (1981).</li> <li>OH</li> <li>B. Carli, M. Carlotti. B.M. Dinelli, F. Mencaraglia, J.H. Park, 1989, "The mixing ratio of stratospheric hydroxyl radical from far-infrared emission measurements" J. Geophys. Res. 94, 11049 - 11058.</li> <li>CO</li> <li>M.M. Abbas, M.J. Glenn, I.G. Nolt, B. Carli, F. Mencaraglia, M. Carlotti, 198 "Far infrared measurements of stratospheric carbon monoxide", J. Geophys. Res. Lett. 15, 140.</li> <li>HNO<sub>3</sub></li> <li>F. Mencaraglia, G. Bianchini, A. Boscaleri, B.Carli, S. Ceccherini, P.Raspolli A. Perrin, J-M Flaud, "Validation of MIPAS satellite measurements of HNO3 using comparison of rotational and vibrational spectroscopy" J.G.R. 111, D19305, doi: 1029/2005JD006099, 2006.</li> </ul>	O <sub>3</sub>	M.M. Abbas, J. Guo, B. Carli, F. Mencaraglia, M. Carlotti, I.G. Nolt, 1987,
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COM.M. Abbas, M.J. Glenn, I.G. Nolt, B. Carli, F. Mencaraglia, M. Carlotti, 198"Far infrared measurements of stratospheric carbon monoxide", J. Geophys. Res. Lett. 15, 140.HNO3F. Mencaraglia, G. Bianchini, A. Boscaleri, B.Carli, S. Ceccherini, P.Raspolli A. Perrin, J-M Flaud, "Validation of MIPAS satellite measurements of HNO3 using comparison of rotational and vibrational spectroscopy" J.G.R. 111, D19305, doi: 1029/2005JD006099, 2006.		mixing ratio of stratospheric hydroxyl radical from far-infrared emission
"Far infrared measurements of stratospheric carbon monoxide", J. Geophys. Res. Lett. 15, 140.HNO3F. Mencaraglia, G. Bianchini, A. Boscaleri, B.Carli, S. Ceccherini, P.Raspollin A. Perrin, J-M Flaud, "Validation of MIPAS satellite measurements of HNO3 using comparison of rotational and vibrational spectroscopy" J.G.R. 111, D19305, doi: 1029/2005JD006099, 2006.		measurements" J. Geophys. Res. 94, 11049 - 11058.
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HOCl	Chance K.V., Johnson D.G., and Traub W.A., (1989), Measurements of
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H <sub>2</sub> O <sub>2</sub>	Chance K.V., Johnson D.G., Traub W.A., and Jucks K.W., (1991),
	Measurement of stratospheric hydrogen peroxide concentration profile using far
	IR thermal emission spectroscopy, Geophys. Res. Lett. 18, 1003-1006
HO <sub>2</sub>	Chance KV, Park K, Evenson KM, Zink LR, and Stroh F, (1995), Far-Infrared Spectrum of HO2, J. Mol Spec., 172, 407-420.

## 1 Tables

## Table 1. Variation in Greenhouse Parameters between the TRP and SAW Atmospheres

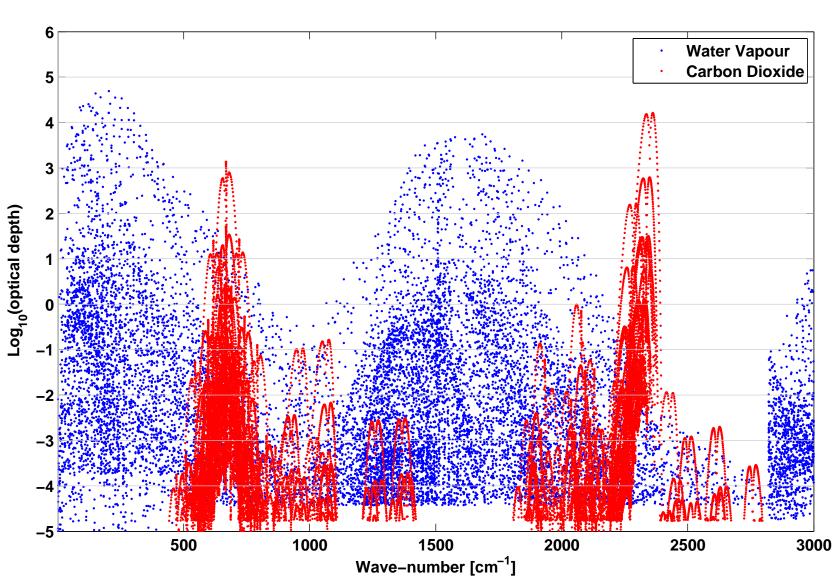
	TRP	SAW
Greenhouse effect, $G$ (Wm <sup>-2</sup> )	170	50
Proportion of G in the far infra-red	27 %	35 %
Normalised greenhouse effect, g	0.372	0.202
Change in OLR upon doubling CO <sub>2</sub> (Wm <sup>-2</sup> )	-3.76	-1.86
Water vapour change giving same forcing as CO2 doubling	+17 %	+43 %
Fraction of forcing from water vapour change in far infra-red	19 %	53 %
$s_{\rm fir} ({\rm m}^2{\rm kg}^{-1}) (0 < v < 500 {\rm cm}^{-1})$	$1.00 \ge 10^{-3}$	7.26 x 10 <sup>-3</sup>
$s_{\rm win}  ({\rm m}^2  {\rm kg}^{-1})  (800 < \nu < 1250  {\rm cm}^{-1})$	2.15 x 10 <sup>-3</sup>	0.59 x 10 <sup>-3</sup>
$s_{\text{rest}} (\text{m}^2 \text{kg}^{-1}) (500 \le \nu \le 800; \nu \ge 1250 \text{ cm}^{-1})$	1.00 x 10 <sup>-3</sup>	4.67 x 10 <sup>-3</sup>

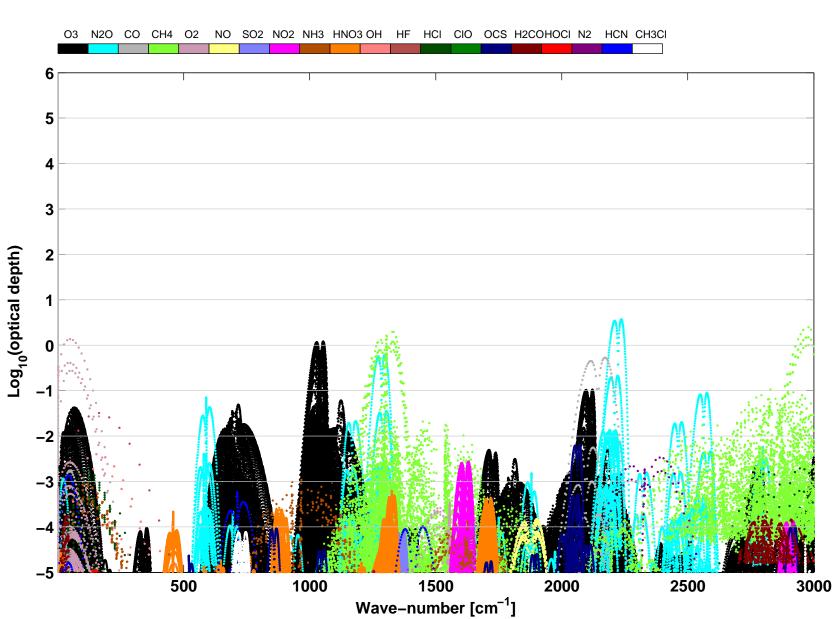
- 1 Table 2. TAFTS instrument specifications and performance

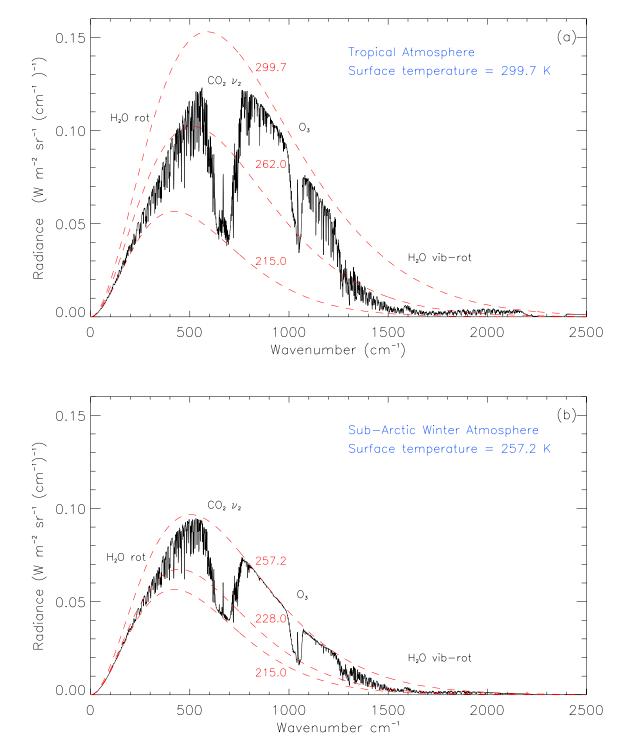
Interferometer	Martin-Puplett
Spectral range	80 - 600 cm <sup>-1</sup>
Spectral resolution	0.1 cm <sup>-1</sup> unapodised
Radiometric sensitivity	$1 \text{ mW m}^{-2} \text{ sr}^{-1} (\text{cm}^{-1})^{-1}$
Integrated sensitivity	0.10 W m <sup>-2</sup> (80 - 600) cm <sup>-1</sup>
Field of view	$\pm 0.8^{0}$
Aperture diameter	25 mm nominal
Optical input ports	2: polarisation coded
Output signals	4: two true differential pairs
Detectors	Ge:Ga and Si:Sb photo-conductors
Detector cooling	Liquid helium, 4.2 K
Cryostat hold time	12 hours measured
Scan sampling	HeNe laser (2.5312 µm spatial interval)
Signal sampling rate	4 channels, 20 kHz per channel
Digital Converters	2 stereo $\Sigma$ - $\Delta$ ADCs, 20 bit resolution
Data volume	560 kBytes per scan, 806 Mbytes per hour
Data storage	Twin 18 Gbyte SCSI-2 hard disks
Control PC	AMPRO P5V, 266 MHz Pentium
Software/operating system	GNU C/ real time Linux
Power Consumption	< 1 kW from 220V ac, 400 Hz
Dimensions	900 x 415 x 550 mm
Mass (main instrument/control box)	90 kg/30 Kg

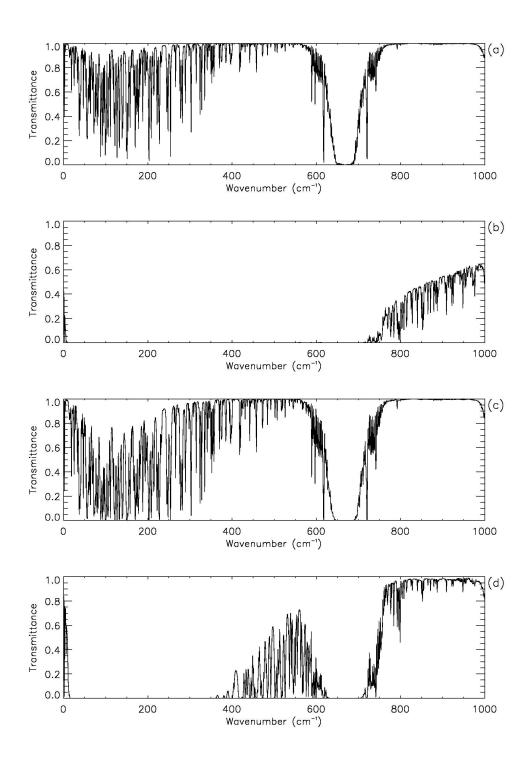
1	Table 3. Main	characteristics	of REFIR	FTS prototypes
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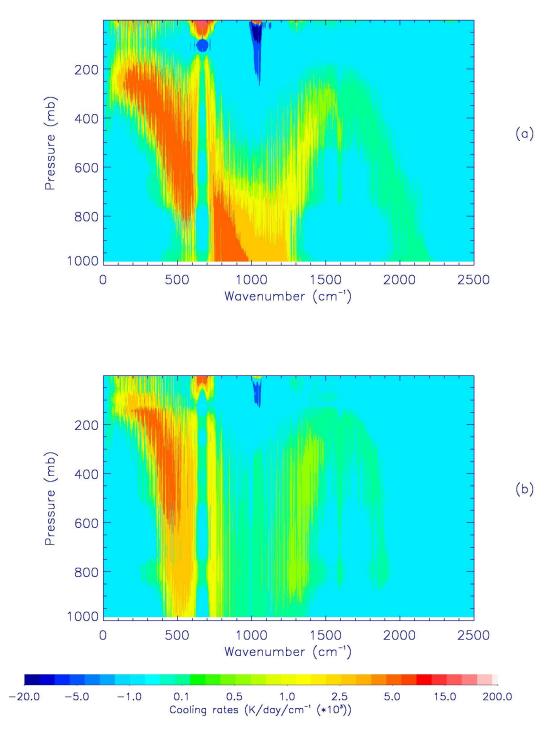
Interferometer	Mach-Zehnder scheme
BS type	Amplitude multilayer, Ge-coated PET film
Spectra coverage	100-1400 cm <sup>-1</sup>
Spectral resolution (max)	$0.25 \text{ cm}^{-1}$
Optical throughput	$0.01 \text{ cm}^2 \text{ sr}$
Field of view	133 mrad
Detectors	2 room-temperature pyroelectrics (DLATGS)
Reference laser	monomode laser diode at 780 nm
Control PC	PC-104, 486DX100 CPU, Linux operating system
Acquisition time	30-120 s
Acquisition frequency	5-20 kHz
NESR	$0.3-10 \text{ mW/(m}^2 \text{ sr cm}^{-1})$ , see Figure 18
Absolute calibration error	< 0.5 K at 280 K
Size	62 cm diam., 26 cm height
Weight/power consumption	55 kg / 50 W











Variation of longwave cooling rates with pressure and wavenumber: (a) TRP cntrl, (b) SAW cntrl

