

Phenomena and Material Property Requirements for a Combined Structural and Thermal Ablation Model

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Applications

Phenomena

Mathematical models

ATK code development (ITRAC and Hero)

Thermal property requirements

Structural considerations

On-going work



Applications

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External Insulation







In-Depth Modeling



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- Mass balance (gas phase)
 - Solid to gas conversion (pyrolysis)
 - Gas advection (permeation)
 - Storage (in the internal pores)
- Momentum balance
 - Balances friction with pressure gradient (Darcy's model)
- Energy balance
 - Conduction (in solid phase)
 - Storage (in solid and gas phases)
 - Pyrolysis energy
 - Advected energy (heat exchange with pyrolysis gases)



Control-volume for energy balance

In-Depth Modeling



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Surface Modeling

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Control-surface for energy balance

Unity Lewis number

$$q_{cond}'' = \rho_e u_e C_H [H_r - (1 + B')h_w + B_c'h_c + B_g'h_g] + \alpha_w q_{rad,inc}'' - \mathcal{E} T_w^4$$

Unequal diffusion coefficients

Control-surface for elemental balance

$$q_{cond}'' = \rho_e u_e C_H (H_r - h_w)_{f.e.g} + \rho_e u_e C_M \left[\left(\sum_i K_{ie} - \sum_i K_{iw} \right) h_i^{T_w} + B_c' h_c + B_g' h_g - B' h_w \right] + \alpha_w q_{rad,inc}'' - \varepsilon \sigma T_w^4$$

Equal diffusion coefficients

$$q_{cond}'' = \rho_e u_e C_H (H_r - h_w)_{f.e.g} + \rho_e u_e C_M \left[\left(\sum_i Z_{ie}^* - \sum_i Z_{iw}^* \right) h_i^{T_w} + B_c' h_c + B_g' h_g - B' h_w \right] + \alpha_w q_{rad,inc}'' - \varepsilon \sigma T_w^4$$

Surface ablation rate

$$\dot{s}_{chem} = \frac{\dot{m}_c''}{\rho_c} = \frac{B_c' \rho_e u_e C_M}{\rho_c}$$

Thermochemistry ("b-prime") tables from ACE code

ITRAC

- 1-D (planar, cylindrical, and spherical)
- Variable-grid finite-volume method
- Heat transfer, material pyrolysis, pore pressure, thermochemical ablation
- Various mechanical erosion models
- User-defined dynamic link libraries (DLLs)
- Ignition model

Hero (Heat Transfer and Erosion Analysis Program) (ATK)

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- Multi-D
- Variable-grid finite-element (hierarchical) method
- Heat transfer, material pyrolysis, pore pressure, thermochemical ablation, radiosity, structural
- Fluid-thermal-structural interaction (FTSI) capabilities (FEM Builder)
- Adaptive refinement
- Parallel processing

Hero Adaptive Refinement

Run-Time Improvement with Parallel Processing

Hero (Heat Transfer and Erosion Analysis Program) (ATK)

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Modeling of Complex Ablation Scenario

Nozzle Thermal Analysis Results

Radiosity with Solar Effects

Material Properties for In-depth Models

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$$\frac{\partial \rho_s}{\partial t} = -(\rho_v - \rho_c) \frac{\partial \alpha}{\partial t}$$

rate of solid density change

overall pyrolysis rate

Mass/Momentum Equation

$$(\rho_{v} - \rho_{c})\frac{\partial\alpha}{\partial t} + \nabla \cdot \left(\frac{\hat{\rho}_{g}}{\mu_{g}}\Gamma\nabla P\right) - \hat{\rho}_{g}(\phi_{c} - \phi_{v})\frac{\partial\alpha}{\partial t} - \frac{\phi\hat{\rho}_{g}}{k_{g}}\frac{\partial P}{\partial t} + \phi\beta_{g}\hat{\rho}_{g}\frac{\partial T}{\partial t} = 0$$

generation advection

advection (permeation)

storage

- Neglected storage (quasi-steady) $(\rho_v - \rho_c) \frac{\partial \alpha}{\partial t} + \nabla \cdot \left(\frac{\hat{\rho}_g}{\mu_c} \Gamma \nabla P \right) = 0$

generation

advection (permeation)

- 1-D simplification (with neglected storage) $\dot{m}_{g}''(x_{p}) = -\frac{1}{A} \int_{x_{p}}^{x_{b}} A \frac{\partial \rho_{s}}{\partial t} dx$

Energy Equation

 $(\underline{Q}_{s} - \underline{h}_{s} + \underline{h}_{g})(\underline{\rho}_{v} - \underline{\rho}_{c})\frac{\partial \alpha}{\partial t} + \hat{\rho}_{g}\phi\frac{\partial \underline{h}_{g}}{\partial t} + \underline{\rho}_{s}\frac{\partial h_{s}}{\partial t} + \hat{\rho}_{g}\mathbf{v}_{D}\cdot\nabla \underline{h}_{g} - \nabla\cdot\mathbf{K}\nabla T = 0$ pyrolysis energy storage advection conduction

$$\frac{d\alpha_i}{dt} = A_i e^{\left(-\frac{E_{a,i}}{RT}\right)} (1 - \alpha_i)^{m_i}$$

Insulator Properties Pyrolysis kinetics parameters x_i, E_i, A_i, m_i •Virgin and char densities ρ_v, ρ_c $\rho_s = \rho_v (1 - \alpha) + \rho_c \alpha$ •Virgin and char specific heat $c_{p,v}, c_{p,c}$ $h_v = \int_{T_{ref}}^T c_{p,v} dt \quad h_c = \int_{T_{ref}}^T c_{p,c} dt$ $h_{\rm s} = h_{\rm u}(1-\alpha) + h_{\rm c}\alpha$ •Virgin and char conductivity k.,, k. $k_s = k_v (1 - \alpha) + k_c \alpha$ •Elemental compositions (v & c) equilibrium $\widetilde{K}_{k,v}, \widetilde{K}_{k,c} \xrightarrow{analysis} h_a, MW_a$ Heats of formation $h_{f,v}^{o}, h_{f,c}^{o} \rightarrow Q_{s}$

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$$q_{cond}'' = \rho_e u_e C_H \left[(H_r - h_w)_{f.e.g.} + \frac{C_M}{C_H} \sum_i (K_{i,e} - K_{i,w}) h_i^{T_w} + B'_g h_g + B'_c h_c - B' h_w \right] + \alpha q''_{rad,inc} - \varepsilon \sigma T_w^4$$

Boundary Conditions

Transport coefficient
Enthalpy (recovery and frozen)
Incident radiation heat flux

Propellant Properties

•Stanton number ratio •Elemental composition

Surface Product Properties

- •Elemental composition
- Enthalpy of products

Insulator Properties

•Char enthalpy (specific heat) and pyrolysis gas enthalpy

•Radiation properties

(emissivity and absorptivity)

Property Summary

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Pyrolysis kinetic parameters

Density

• Virgin and char

Specific heat

- Virgin and char
- Versus temperature

Thermal conductivity

- Virgin and char
- Versus temperature

Elemental compositions

- Virgin and char
- Pyrolysis gas calculated

Heats-of-formation

- Virgin and char
- Pyrolysis gas value calculated
- Used to calculate heat-of-pyrolysis
- Used in surface thermochemistry

Radiation properties

• Emissivity and absorptivity

For pore pressure

- Porosity
- Permeability
- Pyrolysis gas molecular weight
- Pyrolysis gas viscosity

Structural modeling is often required for accurate assessment of structural integrity (including phenomena such as pocketing, ply-lifting, wedge-outs, and delamination)

- Phenomena can have significant impact on thermal protection
- Integrally tied to thermal responses
- Accurate modeling of heat transfer, material pyrolysis, pore pressure, thermochemical ablation, etc. is critical

Structural – High Temperature Issues

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Pocketing involves the expulsion of carbon-cloth phenolic (CCP) material due to a combination of stresses from thermal expansion and pore pressure driven stresses

Can significantly affect thermal protection

Structural – High Temperature Issues

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Ply lifting involves the lifting of a char cap of CCP material. Initial failure is caused by thermal expansion/contraction, lifting is caused by pore pressures

Structural – High Temperature Issues

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Ply lifting char caps can slough, causing loss of thermal protection

Structural modeling is dependent on thermal modeling, for example:

- Stiffness can be dependent on degree-of-char and presence of moisture
 - Charring causes material changes that affect the magnitude of the modulus
 - Moisture causes a plasticization of the matrix material

Structural Dependent on Thermal

Structural modeling is dependent on thermal modeling, for example:

- Structural behavior is also influenced by pressures
- Pressure magnitudes are highly dependent on structural loading
 - Permeability is affected by load
 - Pressure is affected by permeability

Property Summary

The following is a simple first principles equation that may be used for the structural model (complexity is usually added to address nonlinearities)

Material properties summary (dependent on thermal state)

Stiffness matrix (includes moduli and Poisson's ratios) S_{ijkl}

Pressure stress coupling, η_{ii}

Coefficient of thermal expansion, α_{ii}

Moisture expansion coefficient, β_{ij}

Decomposition coefficient, χ_{ii}

Failure criteria

Conjugate models are required for accurate simulation of thermal and structural behavior

Thermal

- 3-D surface ablation in Hero
- General thermal contact in Hero
- Advanced surface thermochemistry code (ACE replacement)
 - Alumina impingement
 - Chemical kinetics
- Material properties
- Comprehensive validations against historical nozzle data

Flow modeling

Advanced carbon-cloth phenolic CCP modeling

- Moisture
- Structural coupling

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