Evaluating Material Flammability in Microgravity and Martian Gravity Compared to the NASA Standard Normal Gravity Test

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Drop tower tests are conducted at Martian gravity to determine the flammability of three materials compared to previous tests in other normal gravity and reduced gravity environments. The comparison is made with consideration of a modified NASA standard test protocol. Material flammability limits in the different gravity and flow environments are tabulated to determine the factor of safety associated with normal gravity flammability screening. Previous testing at microgravity and Lunar gravity indicated that some materials burned to lower oxygen concentrations in low gravity than in normal gravity, although the low g extinction limit criteria are not the same as 1g due to time constraints in drop testing. Similarly, the data presented in this paper for Martian gravity suggest that there is a gravity level below Earth’s at which materials burn more readily than on Earth. If proven for more materials, this may indicate the need to include a factor of safety on 1g flammability limits.

I. Introduction

Currently, NASA STD 6001 Test 1 is the major method used to assess flammability of materials intended for use in habitable environments of U.S. spacecraft.\textsuperscript{1} The method is a pass/fail upward flame propagation test conducted in the most severe flaming combustion environment (oxygen concentration, pressure) expected in the spacecraft (e.g. currently 30% oxygen at 10.2 psia which is the pre-breath environment on the ISS for spaceswalks). Materials that do not self-extinguish after six inches of burning must undergo other special considerations and/or tests if they are to be used on spacecraft. To better understand the actual spacecraft flammability limits, a modified Test 1 protocol has been evaluated with 3 materials in Martian gravity (0.38 g \textsubscript{Earth}) and compared to earlier related tests at 1g, Lunar g, and microgravity with forced convection. This modified Test 1 protocol relies on performing multiple tests in 1-g (refs. 2-4). For reduced gravity tests, the protocol is modified slightly (as described later) because of the number of drop tower tests which would be required.

This protocol, rather than qualifying materials as pass/fail in the worst-expected environments, measures the actual upward flammability limit (akin to the Limiting Oxygen Index (LOI)) for the material so that a more accurate assessment of the margin of safety of the material in the real use environment can be made. In broader application, the modified Test 1 protocol would provide the option of selecting better or best materials based on the best margin of safety relative to the use environment, as opposed to what would be considered just “passing” from a flammability point of view.

For future space missions, it is often desirable to reduce cabin pressure while maintaining oxygen partial pressure.\textsuperscript{5} Atmospheres are chosen based on several considerations including: material flammability, decompression sickness, hypoxia, and efficient crew transfers and excursions. Keeping oxygen partial pressure constant while reducing total pressure has the effect of increasing the concentration of oxygen which in turn leads to increased flammability of materials.

One of the major lessons learned from the Apollo 1 fire is that it is impossible to eliminate all ignition sources,\textsuperscript{6} so fire prevention is achieved in spacecraft through material controls near potential ignition sources and the use of fire resistant materials. After Skylab, the Space Shuttle and the International Space Station (ISS) have operated at normal sea-level conditions (air, 21% oxygen and 79% nitrogen by volume at 14.7 psia total pressure) except for brief pre-EVA activities when oxygen levels are increased for a short period of time to 30% oxygen while the total
pressure is lowered to 10.2 psia. Many of the materials which had been used in the Shuttle or which are currently used in ISS are therefore only tested and rated to 30% oxygen at 10.2 psia.

![Figure 1. Near limit probability that a material will burn or not, with the ULOI and MOC indicated.](image)

In the modified procedure, the oxygen concentration in Test 1 is successively reduced to identify the Upward Limiting Oxygen Index (1g ULOI) and the Maximum Oxygen Concentration (1g MOC) that consistently results in self-extinguishment of the material. The 1g ULOI is defined as the oxygen concentration at which a material passes the NASA STD 6001 Test 1 burn length criterion approximately half the time. The 1g MOC is defined as the oxygen concentration where at least five samples passed the burning criterion (according to the NASA STD 6001 Test 1 definition) and where at least one sample failed in the environment that contained 1 percent more oxygen by volume. This is shown schematically for an arbitrary material in Fig. 1. For reduced gravity tests, the definitions of MOC and ULOI are modified slightly (as described later) because of the number of drop tower tests which would be required.

II. Experiment Description

To better understand a material’s flammability limits at reduced gravity, we evaluate three materials in Martian-g using the centrifuge described below. These tests are combined with 1g tests performed at NASA’s White Sands Test Facility (WSFT), microgravity tests with forced convection, and Lunar-g tests at NASA’s Glenn Research Center.

The centrifuge was developed for use in a drop tower environment. The apparatus is subject to the usual constraints on size and volume for the Zero Gravity Facility at the NASA Glenn Research Center. The centrifuge consists of a flat circular base (turntable) and the chamber dome.

Most experiments examining fluid and gas flows require a chamber for containment and atmosphere, since the drop tower is evacuated before a drop. The chamber was designed to maximize the rotation radius possible on the drop bus. The strength of the chamber was built to permit tests up to 2 atm absolute pressure. A photograph of the chamber is shown in Fig. 2. More details are provided elsewhere.

The fuels used include Ultem® 1000, Nomex® HT90-40, and Mylar® G or Melinex® 515. These three materials were evaluated for microgravity, Lunar, and Martian flammability limits to compare with the 1g ULOI and 1g MOC.

Ultem® 1000 (fire retarded polyetherimide (PEI)) in 10 mil thick film is inherently flame-retarding, with charring characteristics, a very low smoke signature, very low smoke toxicity, and a low heat-release rate.
Nomex® HT90-40 is a 12 mil thick fire retarded aromatic nylon fabric which does not melt or drip as it burns. When exposed to a heat source, the Nomex fibers swell and seal the spaces between the fibers, stopping air movement through the fabric and thus inhibiting heat transfer through the fabric.

Mylar® G is a 5 mil thick plastic film made from polyethylene terephthalate. It is not fire-retarded, melts as it burns, and was selected primarily for its non-charring character. The WSTF comparison tests used Melinex® 515, a very similar Dupont Teijin film.

The solid fuel sample was mounted within the chamber as shown schematically in Fig. 3. The fuel sheet is mounted parallel to the turntable and near the outer periphery. It is sandwiched between two 0.005”-thick mica rectangular frames (5 cm x 15 cm opening) which are in turn sandwiched between 0.010”-thick stainless steel rectangular frames. The fuel is ignited with a hot wire at the furthest point from the center so that a concurrent-flow flame spread geometry is established.

The chamber is filled to the desired atmosphere. Prior to the drop, the chamber is rotated for at least five minutes to allow the interior gases to come up to speed (solid body rotation). To achieve Martian gravity levels, the rotation rate required is 30.8 RPM. Martian g-level is achieved at the radial distance of the igniter.

In order to maximize available drop time, the goal was to get the samples ignited right at or slightly before the drop. For the Ultem and Nomex samples, the igniter current was initiated 2.1 seconds before drop and was on for a
total of 3.5 seconds. For the Mylar samples, ignition was easier to achieve. The igniter current was initiated 1 second before the drop and was on for a total of 2 seconds. The Mylar sample ignitions were assisted using a small strip of thin tissue paper (Kimwipes, 0.5-cm width) which helped provide a uniform energy deposition to this melting fuel. For all cases, the average igniter power was 58 watts. All the Martian-g cases used the same igniter wire.

Experiments were then conducted to examine the maximum oxygen concentration at which three different materials burning in Martian gravity would self-extinguish. (Experiment materials and test conditions were chosen to match earlier tests performed at WSTF, described below.) Slightly different than the 1-g definitions of MOC and ULOI, the MOC* for these tests was defined as the maximum oxygen concentration at which the flame was observed to self-extinguish during the drop time. The ULOI* criterion was minimum oxygen concentration where the flame survived the full drop time. Images of the ULOI* conditions for each are shown in Fig. 4. The Mylar flame appeared stable (as shown later in Fig. 5) but the Nomex and Ultem flames were becoming quite dim and shrinking at the end of the drop (less than 1 cm in length and getting smaller). Since it is likely that these very weak flames might have extinguished given longer time in Martian g, the limits found are conservative.a

III. Results

Materials flammability tests were conducted for three materials at Martian-g. The results were compared to ULOI and MOC values in 1-g, Lunar-g, and 0g (with a concurrent flow of 30 cm/s, which is a reasonable maximum local spacecraft ventilation velocity9).

Representative flame images are shown in fig. 5 for Mylar burning at the Martian-g ULOI* condition (18% O2 at 10.2 psi). The igniter can be seen glowing on the left side of the frame until ignition. Igniter power is then turned off and the igniter cools. The flame starts out as dim blue but then brightens to yellow and actually appears to stabilize. While the flame does not span the 5-cm width of the sample, it survives for the duration of the drop test. When the oxygen is decreased in another test to 17%, the flame goes out.

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a The flames in these Martian-g centrifuge tests were relatively small compared to the sample size and spread only a short distance along the available fuel in the 5.18 sec-duration drop tower tests. Since the local centrifugal acceleration level is proportional to radius from the center of rotation, this means that the flame stays in a region of approximately constant g-level (lunar g). On the other hand, if the flame were to spread a significant way along the fuel, the local acceleration level experienced by the flame would change appreciably as it spread.
Figure 5. Flame images for Mylar at the Martian-g ULOI*. Images are shown at 10 Hz progressing left to right then top to bottom. The last frame shows flame extinction caused by impact of the experiment.

The limiting oxygen values (ULOI and MOC) for each fuel are summarized in Table I (includes earlier data from ref. 8). We reiterate that the definition of MOC* and ULOI* are slightly different in reduced gravity (compared to MOC and ULOI definitions in normal gravity) because of the number of drop tower tests which would be required. The near-limit flames are generally small and localized to the upstream edge of the material. We noticed that any distortion of the burned edge of the material, such as curling, swelling, or contracting, weakens the flame apparently by influencing the flow around the burned edge. Since oxygen transport is critical to low gravity flames, anything that reduces the free flow of oxygen past the sample will reduce the material’s flammability.
Table I: Limiting Oxygen Molar Concentrations and Oxygen Margin of Safety for Different Gravity Levels

<table>
<thead>
<tr>
<th>Fuel</th>
<th>Mylar® G</th>
<th>Ultem® 1000</th>
<th>Nomex® HT90-40</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>10.2 psia</td>
<td>10.2 psia</td>
<td>14.7 psia</td>
</tr>
<tr>
<td>1g ULOI</td>
<td>21.2</td>
<td>23.5</td>
<td>23.5</td>
</tr>
<tr>
<td>1g MOC</td>
<td>20.0</td>
<td>23.0</td>
<td>22.1</td>
</tr>
<tr>
<td>Martian ULOI*</td>
<td>18.0</td>
<td>22.0</td>
<td>19.9</td>
</tr>
<tr>
<td>Martian MOC*</td>
<td>17.0</td>
<td>21.1</td>
<td>19.0</td>
</tr>
<tr>
<td>Lunar ULOI*</td>
<td>15.6</td>
<td>21.0</td>
<td>21.0</td>
</tr>
<tr>
<td>Lunar MOC*</td>
<td>14.1</td>
<td>19.9</td>
<td>19.9</td>
</tr>
<tr>
<td>0g* ULOI</td>
<td>17.0</td>
<td>24.0</td>
<td>23.0</td>
</tr>
<tr>
<td>0g* MOC*</td>
<td>16.0</td>
<td>23.0</td>
<td>22.0</td>
</tr>
<tr>
<td>Martian ∆O₂%</td>
<td>-3.1</td>
<td>-1.7</td>
<td>-3.35</td>
</tr>
<tr>
<td>Lunar ∆O₂%</td>
<td>-5.75</td>
<td>-2.8</td>
<td>-2.35</td>
</tr>
<tr>
<td>0g ∆O₂%</td>
<td>-4.1</td>
<td>0.25</td>
<td>-0.3</td>
</tr>
</tbody>
</table>

From the limit data, we can evaluate the oxygen margin of safety of the material in low gravity relative to the 1g Test 1 data. The oxygen margin of safety, or ∆O₂%, is defined here as the mean 0g limit or the Lunar limit minus the mean 1g limit as follows:

\[
\Delta O_2\% = \left( \frac{ULOI^* + MOC^*}{2} \right) - \left( \frac{UROI_{1g} + MOC_{1g}}{2} \right)
\]

In Eq. (1), the lowest %O₂ at which the material burned in the drop test for the full test time was taken as the 0g Lunar-g, or Martian-g ULOI*; the maximum %O₂ where the material extinguished during the drop test was taken as the 0g, Lunar-g, or Martian-g MOC*. It should be noted that the low g extinction limit criteria are not the same as 1g due to time constraints in drop testing. A positive ∆O₂% means that the flame will burn in 1g at lower oxygen concentrations than in low gravity (Lunar or 0g). Conversely, a negative ∆O₂% means that the flame will propagate in Lunar or 0g at a lower oxygen concentration than in 1g and as reflected by the negative margin of safety. Values of ∆O₂% for both 0g and Lunar g (from ref. 8) are included in Table I.
As shown in Fig. 6 and Table I, the 1g flammability limits are generally not conservative for these materials as evidenced by the negative $\Delta O_2\%$, by up to $-5.75\%$ oxygen. For low gravity concurrent (upward) flow conditions, the absence of or reduction of buoyancy reduces convective heat loss from the flame zone. As such, the flame doesn’t have to release as much heat as in 1g to provide the same amount of heat flux to the unburned fuel. Thus, it is not entirely surprising that the low gravity flammability limits are lower than the 1g flammability limits, which is the general trend. The evaluation of the magnitude of this $\Delta O_2\%$ for other materials is continuing for concurrent flammability limits, to be followed by evaluation of the limits for opposed flow. If proven for more materials, this may indicate the need to include a factor of safety on 1g flammability limits.

Nomex and Ultem are charring fuels. Char layer formation is a known fire-retardant mechanism since it reduces the heat release rate of the material. For these charring materials, the oxygen margin of safety in 0g is very close to zero, indicating that any reduced charring for 0g may be offset by less heat release in the reduced convective environment. In Lunar gravity, the oxygen margin of safety for these charring fuels is $\sim 2.5\%$ lower than 1g. It may be that with adequate but not excessive convection, materials burn better, as predicted by flammability maps. If Mylar is a non-charring material. Just as for metals combustion\textsuperscript{11} the absence of gravitational-induced dripping keeps heated molten fuel in the flame zone rather than removing it, and the result is a significantly negative oxygen margin of safety ($-4.5\%$) in Martian, Lunar, and 0g. In fig. 7, limiting oxygen concentration is plotted for Mylar as a function of gravity level. There appears to be a range of g-levels within which a flame can survive. This is similar to the forced convective flammability boundary.\textsuperscript{10} The two u-shaped curves are presumed to bound the flammability boundary of this material. Fig. 7 includes additional two data points at sub-lunar g-levels to probe the behavior at low flow speeds, including one test at a g-level equaling that on Pluto. Tests were conducted using precision mixtures of gases (+/- 0.1% O$_2$) but at...
increments of 1% \( \text{O}_2 \). Given the sparcity of points, the exact shapes of the curves are not known, nor are the absolute minimums. However the curves indicated show the qualitative features.

IV. Conclusions

A series of drop tower tests has been completed to determine the upward flame spread oxygen concentration extinction limits of three materials at Martian buoyant flow gravity levels for comparison with upward normal gravity limits, Lunar gravity, and at microgravity (with forced convection). The limiting oxygen levels at Martian gravity were significantly lower than normal gravity (up to 5.75% \( \text{O}_2 \) lower), and typically between the normal gravity and Lunar gravity values. However, it should be noted that the low-g extinction limit criteria are not the same as 1g due to time constraints in drop testing. This consistent trend of negative margins of safety on normal gravity flammability screening tests indicates that normal gravity material flammability screening tests may not be conservative and some materials tested to be safe for use in space may actually be flammable. If proven for more materials, this may indicate the need to include a factor of safety on 1g flammability limits.

The limits are defined with consideration of a modified NASA STD-6001 Test 1 protocol. That modified protocol measures the actual upward flammability limit (akin to the Limiting Oxygen Index (LOI)) for the material so that a more accurate assessment of the margin of safety of the material in the real use space environment can be made rather than merely qualifying materials as pass/fail in the worst-expected environments.

The centrifuge apparatus is a useful tool to achieve gravitational acceleration levels in a range not easily attainable in simple drop tower or aircraft experiments. Additional research projects could make use of the chamber for their own specific objectives.

Figure 7. Limiting Oxygen Concentration as a function of g-level for Mylar. For a given oxygen concentration, there is range of g-levels which will permit the flame to be sustained.
Acknowledgments

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References