

James I. Mueller



1917 - 1986

James I. Mueller



- Professor of Ceramic Engineering at the University of Washington
- Led team of scientists that helped develop insulating ceramic tiles used to protect the Space Shuttle
- Inspired future generations of materials scientists and engineers – one of whom, Bonnie Dunbar, became a NASA astronaut



Ultra High Temperature Ceramics A Journey

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Acknowledgements

- NASA Ames Research Center
 - Don Ellerby, Matthew Gasch, Michael Olson, Sarah Beckman, Tom Squire, John Lawson
- ELORET at NASA Ames Research Center
 - Mairead Stackpoole, Michael Gusman
- Glenn Research Center
- SRI International
- University of California at Berkeley
- Many colleagues throughout the world
- My family



Outline

- The Process of Discovery
- Education
- Experience
- Curiosity
- Opportunity
- Discovery
- New Frontiers
- Conclusion



Process of Discovery

Education

+

Experience

+

Interaction

+

Curiosity

+

Opportunity



Discovery



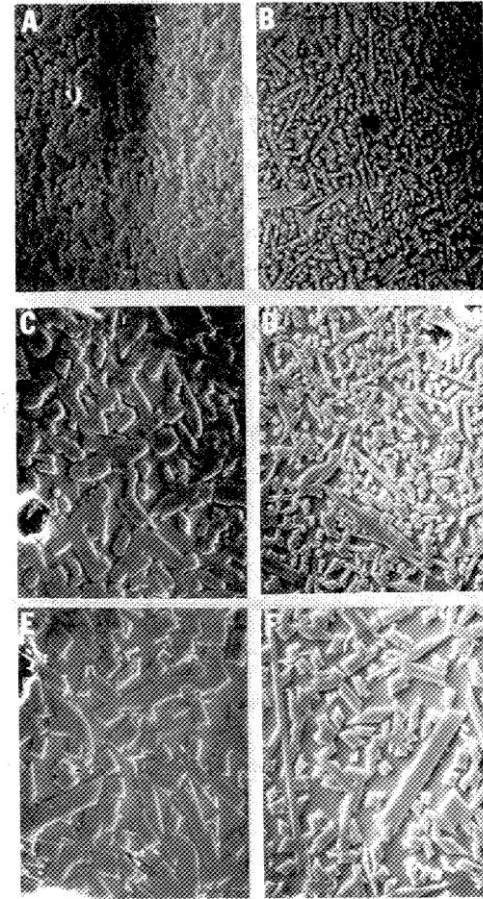
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UC Berkeley



Joseph A. Pask

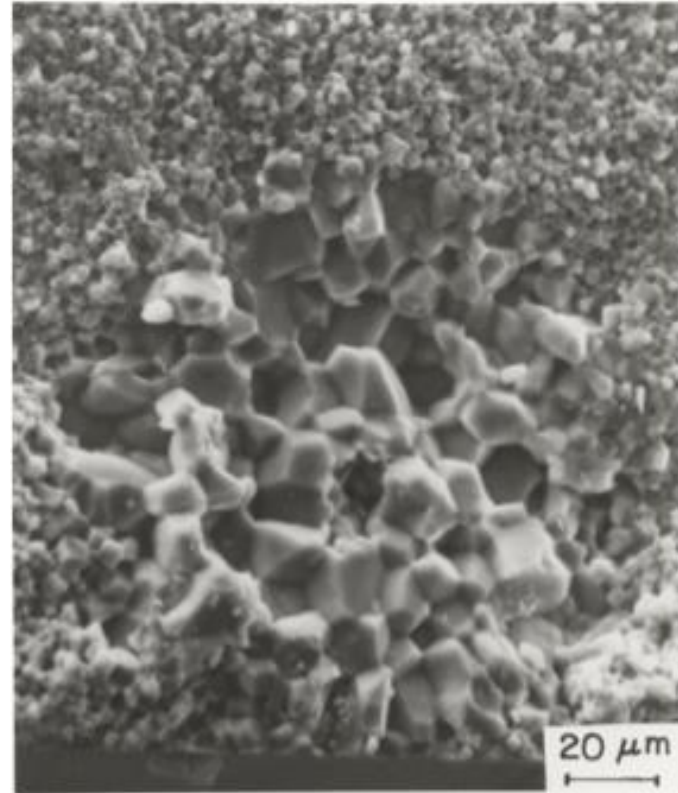


Effect of CaO on grain growth
in Al₂O₃ and kaolinite

UC Berkeley



Tony Evans



Effect of defects/large grains
on creep of Al_2O_3



Outline

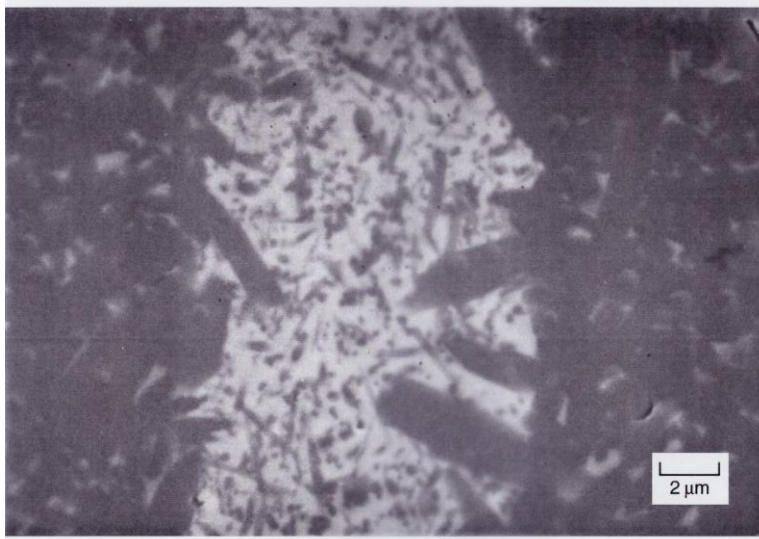
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Experience at SRI International

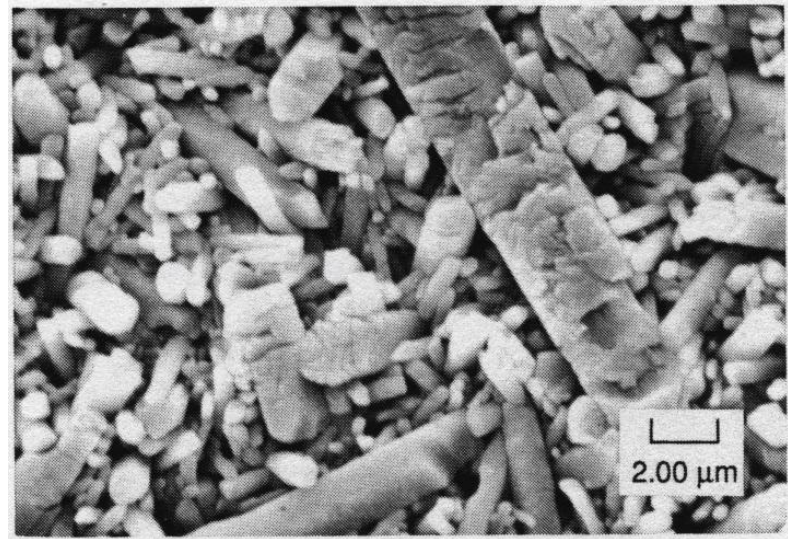
- Contract research: opportunities to solve problems and develop materials for many clients and applications
- Silicon nitride, powder synthesis, preceramic polymers, high temperature superconductors, oxide composites
- David Rowcliffe, Yigal Blum, Michael Gusman, Bob Lamoureaux, Don Hildenbrand

Grain Growth in Si_3N_4 Preceramic Polymer Mixtures

Silicon nitride joined with a
preceramic polymer

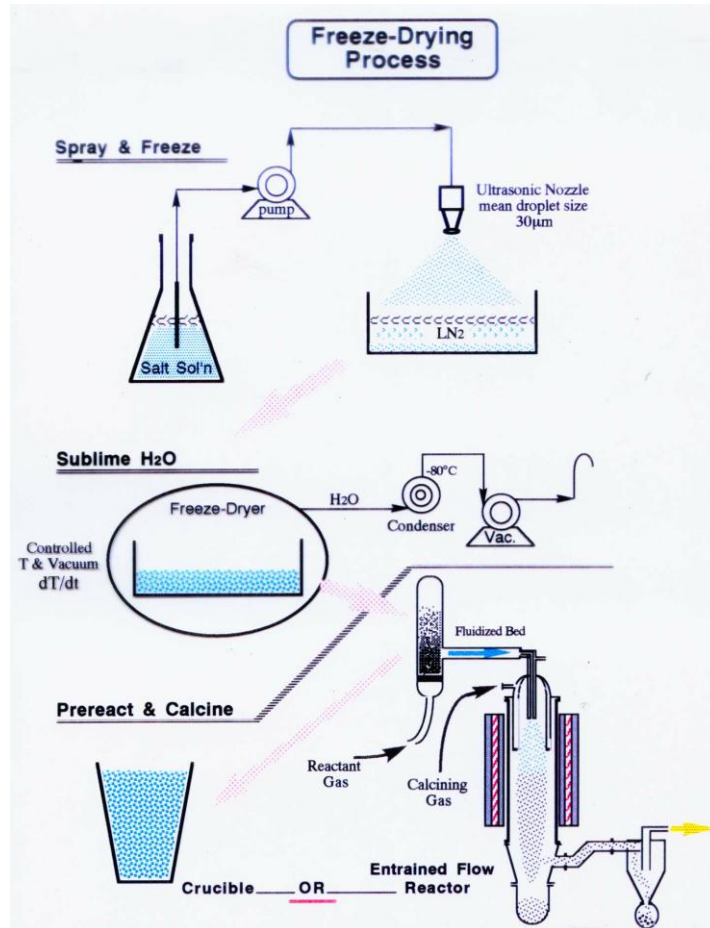


Elongated grains in sintered
silicon nitride /preceramic
polymer powder



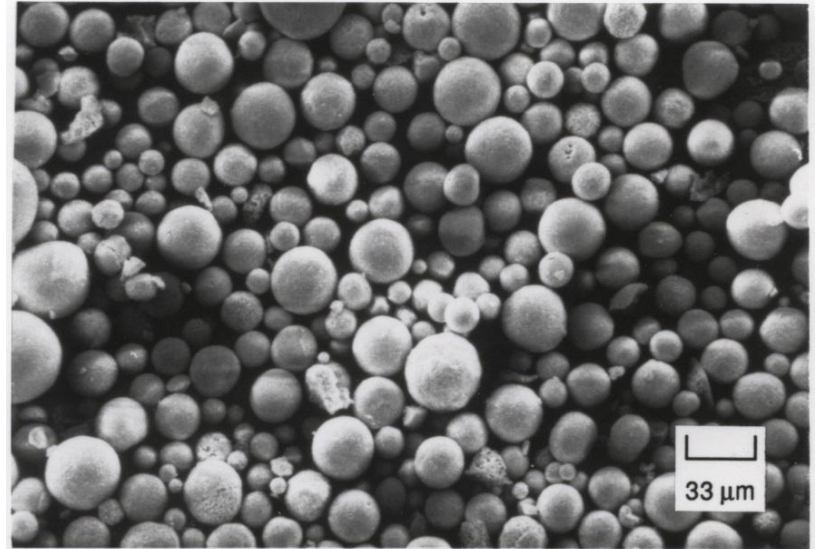
Images from work done at SRI International in 1980s/90s

Powder Preparation by Freeze-Drying



Patent: Cryochemical Method of Preparing Ultrafine Particles of High-Purity Superconducting Oxides - M. I. Gusman and S. M. Johnson. U.S. Patent 4,975,415

POLYMER/POWDER SPHERES



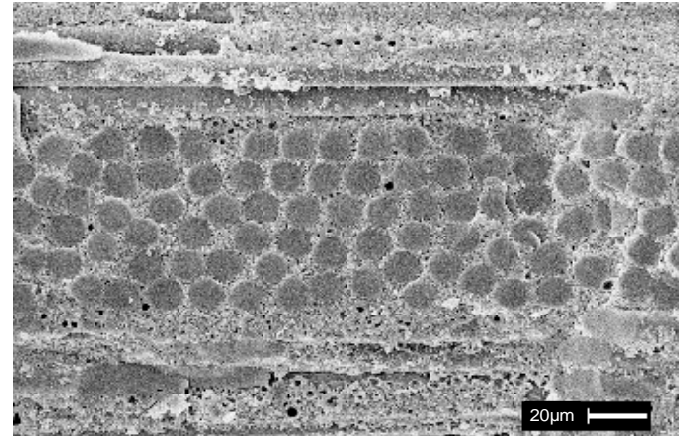
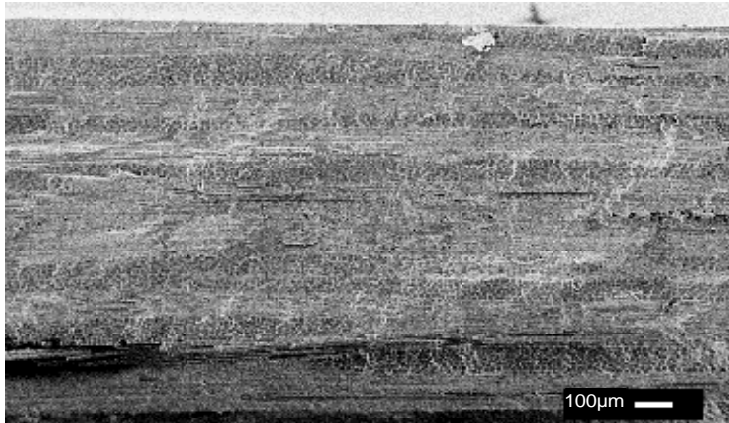
CP-340532-6

Si₃N₄/preceramic polymer powders

Images from work done at SRI International in 1990s

Oxide Composite Fabrication

- Consortium funded by DARPA
- SRI made matrix from Al/Al₂O₃/polymer slurry
- Rockwell demonstrated RTM
- 3M made large components by prepregging and by winding
- Large oxide-matrix structures made by winding



Al/Al₂O₃/polymer slurry (SRI matrix) infiltrates well into and between weaves in prepregged composite

Images from work done at SRI International in mid 1990s

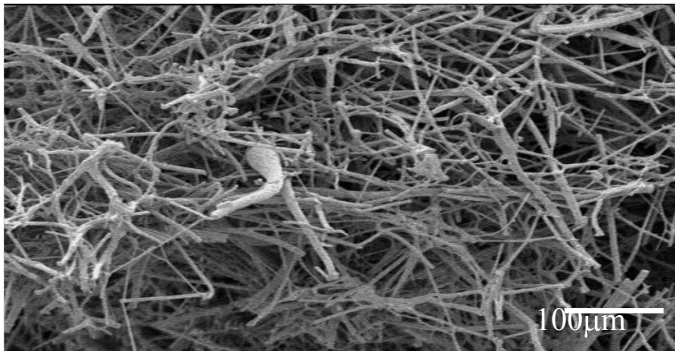


Thermal Protection Materials at NASA Ames

Two Types of Thermal Protection Systems (TPS)

Reusable systems reject as much heat as possible and conduct as little heat as possible

- Low thermal conductivity
- High emissivity



Microstructure of an alumino-silicate tile (AETB)

Ablative systems reject heat by material consumption

- Charring/pyrolyzing materials

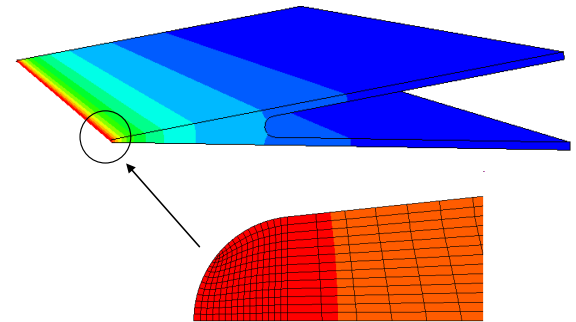


Charred ablator



Sharp Leading Edge Technology

- For enhanced aerodynamic performance
- Materials for sharp leading edges can be reusable but need different properties because of geometry and very high temperatures
- Require materials with significantly higher temperature capabilities, but for short duration
 - Current shuttle RCC leading edge materials: $T \sim 1650^{\circ}\text{C}$
 - Materials for vehicles with sharp leading edges: $T > 2000^{\circ}\text{C}$



High Temperature at Tip

Steep Temperature Gradient

Passive cooling is simplest option to manage the intense heating on sharp leading edges.

UHTCs are candidate materials



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Flight Hardware



SHARP-B1 May 21, 1997

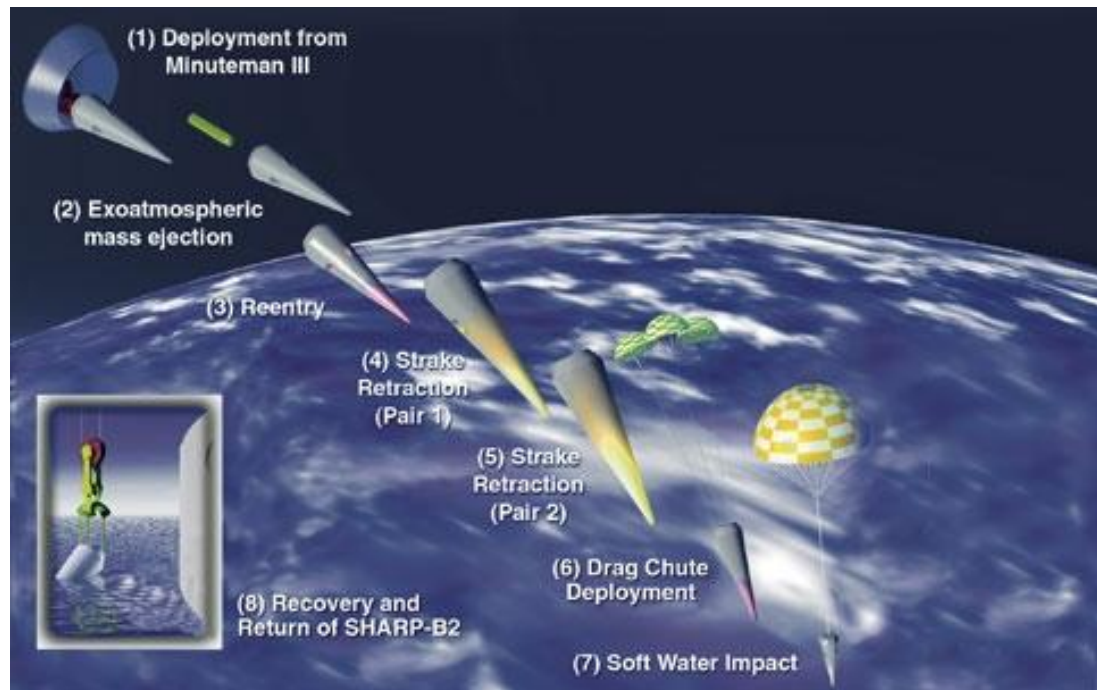


SHARP-B2 Sept. 28, 2000



SHARP-B2

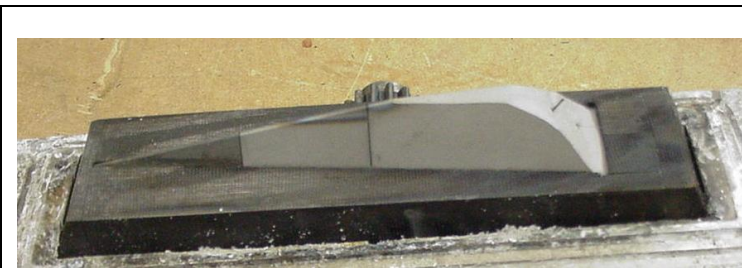
- Flight test designed to evaluate three different compositions of UHTCs in strake (fin) configuration exposed to ballistic reentry environment.
- Strakes exposed as vehicle reentered atmosphere, then retracted into protective housing.
- ***Material recovered. Led to new effort in UHTCs / decision to bring development in-house and improve processing.***



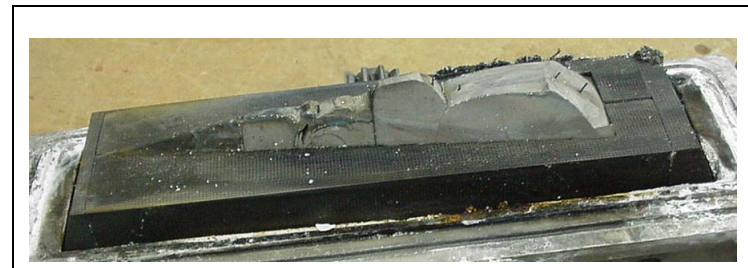
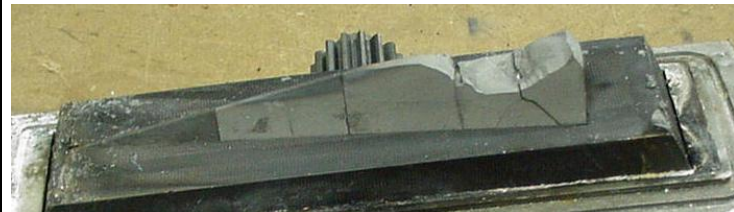


Recovered UHTC Strakes

- Post-flight recovery showed that all four $\text{HfB}_2\text{-SiC}$ aft-strake segments suffered similar, multiple fractures.
- No evidence of severe heating damage (for example, ablation, spallation, or burning) was observed.
- Defects inherent in material lot are present on fracture surfaces.
- Actual material properties exhibit wider scatter and greater temperature dependence than those assumed in design.



Pair 1 (47.9 km)

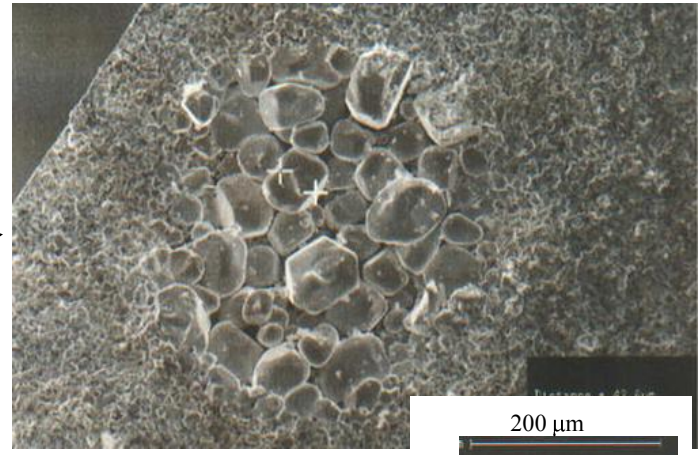
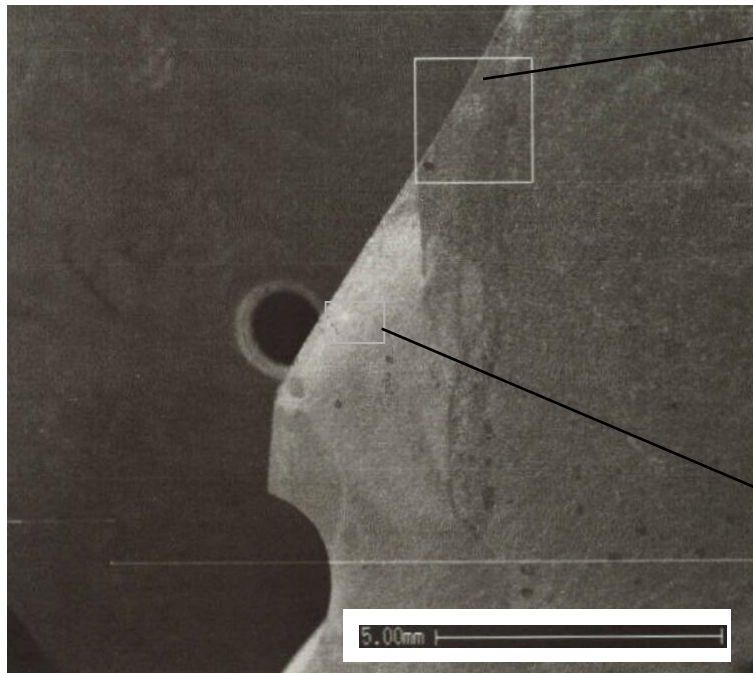


Pair 2 (43.3 km)

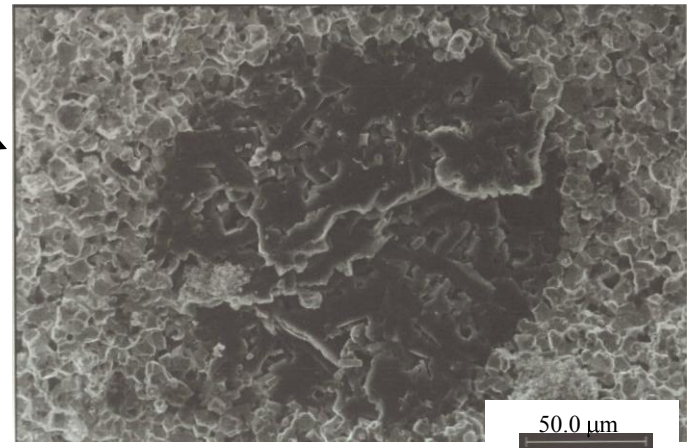




Processing Defects on Fracture Surface of Aft-Segment, Strake 2



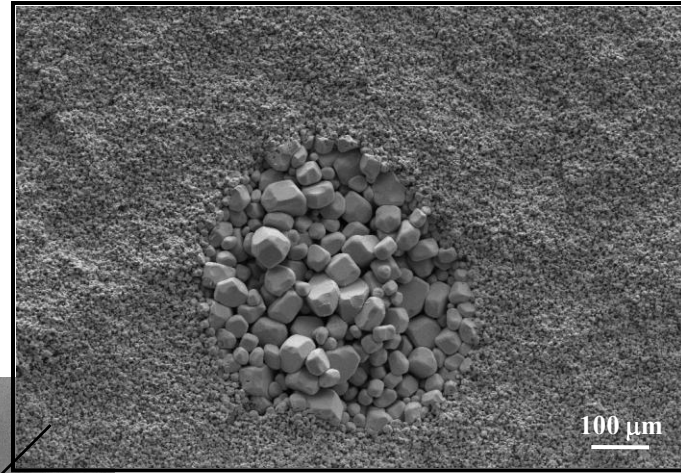
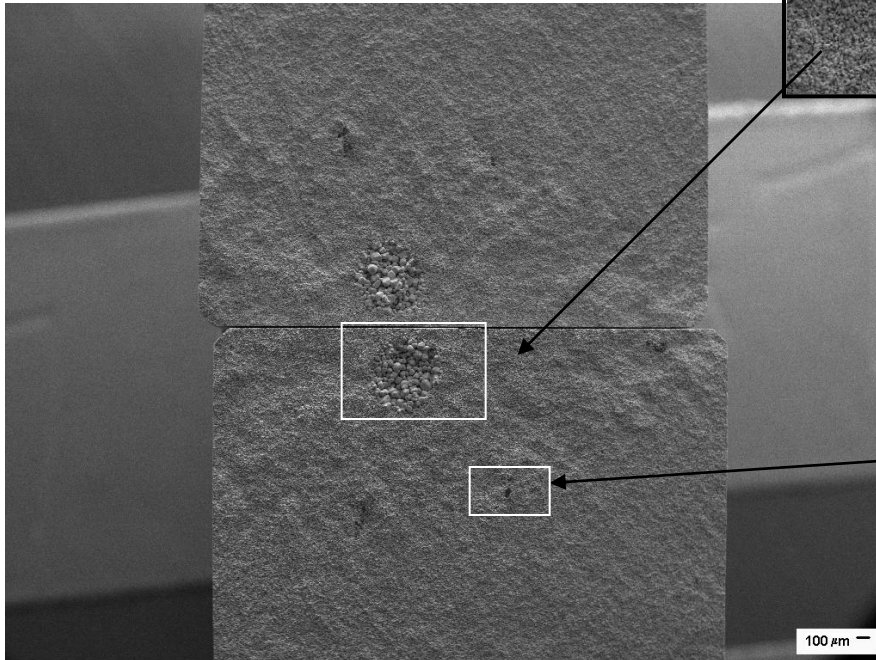
HfB₂ agglomerate



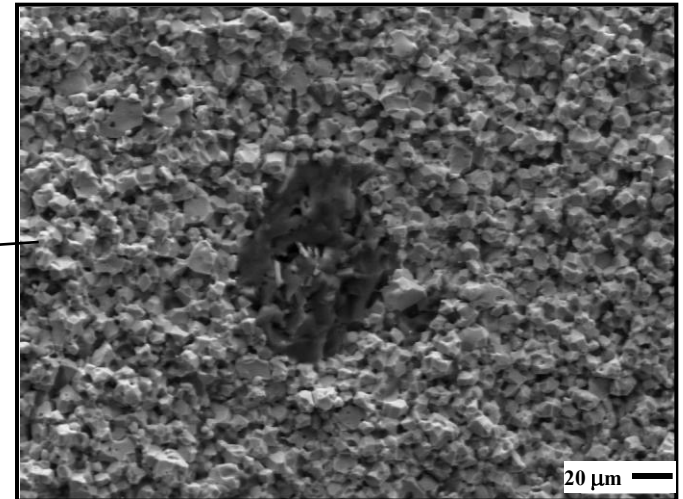
SiC agglomerate



Processing Defects in HfB_2 -SiC Flexure Specimens



HfB_2 agglomerate

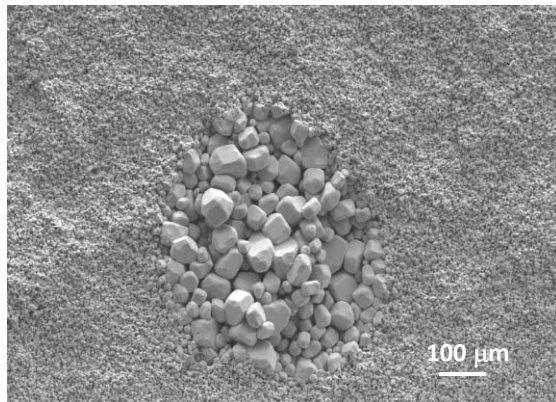


Grafoil™ agglomerate



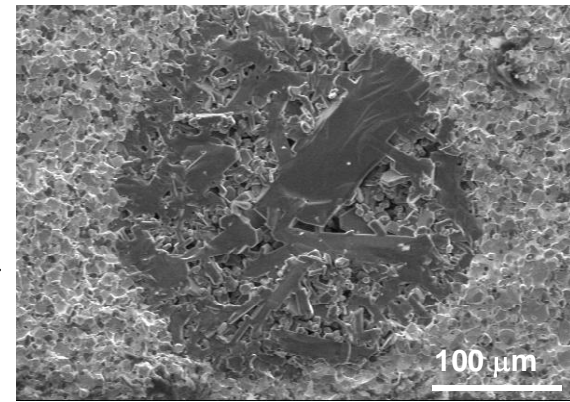
A Cautionary Tale

- Materials did not have expected fracture toughness, strength, or reliability (Weibull modulus).
- Unexpected fractures were due to poor materials processing by external vendor.
- SHARP B-2 underlined importance of controlling materials development, processing methodologies, and resulting material properties if we are to get the maximum value from an experiment.



Large HfB_2 agglomerate

Poorly processed
 HfB_2 20v%SiC



Large SiC-rich agglomerate

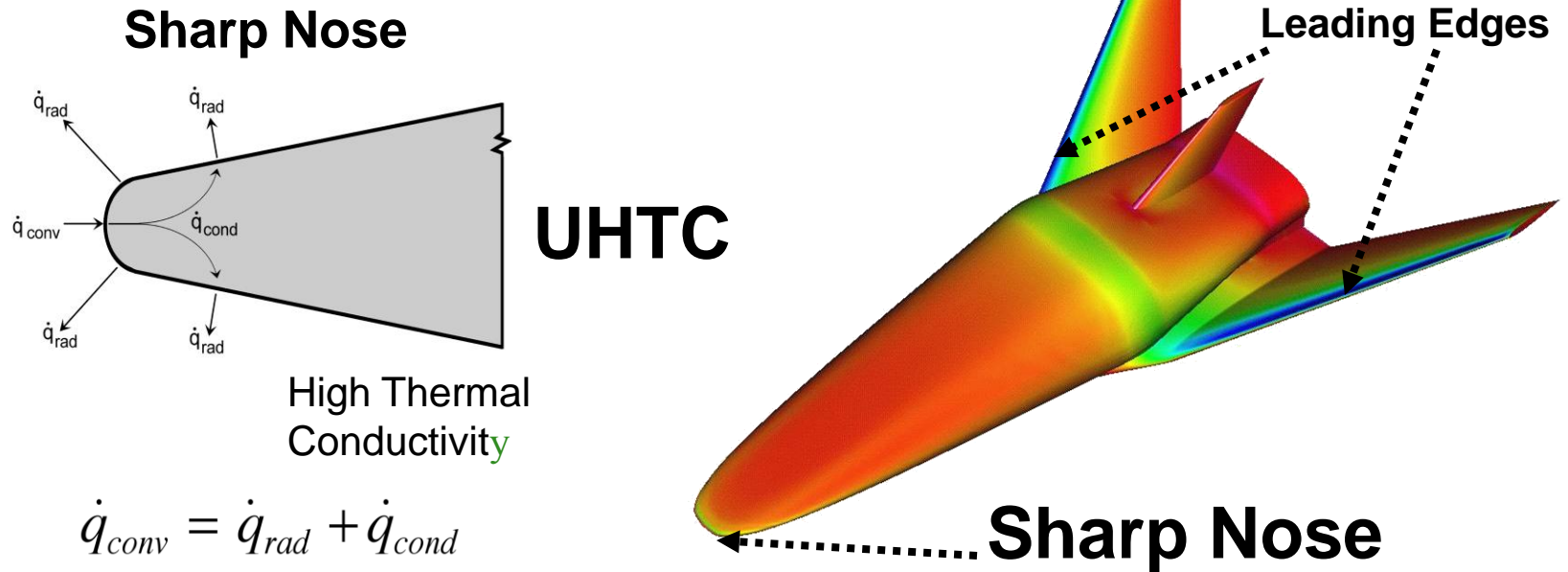


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Sharp Leading Edge Energy Balance



Insulators and UHTCs manage energy in different ways:

- Insulators store energy until it can be eliminated in the same way as it entered
- UHTCs conduct energy through the material and reradiate it through cooler surfaces



High Temperature Passive Material Options

- SiC-based coatings, as on Shuttle Orbiter leading edges, are applicable to $\sim 1650\text{ }^{\circ}\text{C}$ ($3000\text{ }^{\circ}\text{F}$).
- Above $\sim 1650\text{ }^{\circ}\text{C}$ ($3000\text{ }^{\circ}\text{F}$), different class of materials required:
 - Carbides, oxides, and diborides of Hf and Zr
 - Refractory metals such as iridium and rhenium
- Some of these materials can be used as a monolith or matrix; some are more appropriate as a coating.
- Thermal properties have a significant impact on surface temperatures.
- HfB_2 has highest melting temperature of borides
 - HfB_2 $T_m \sim 3380^{\circ}\text{C}$; ZrB_2 $T_m \sim 3250^{\circ}\text{C}$



UHTC billets, US quarter for scale



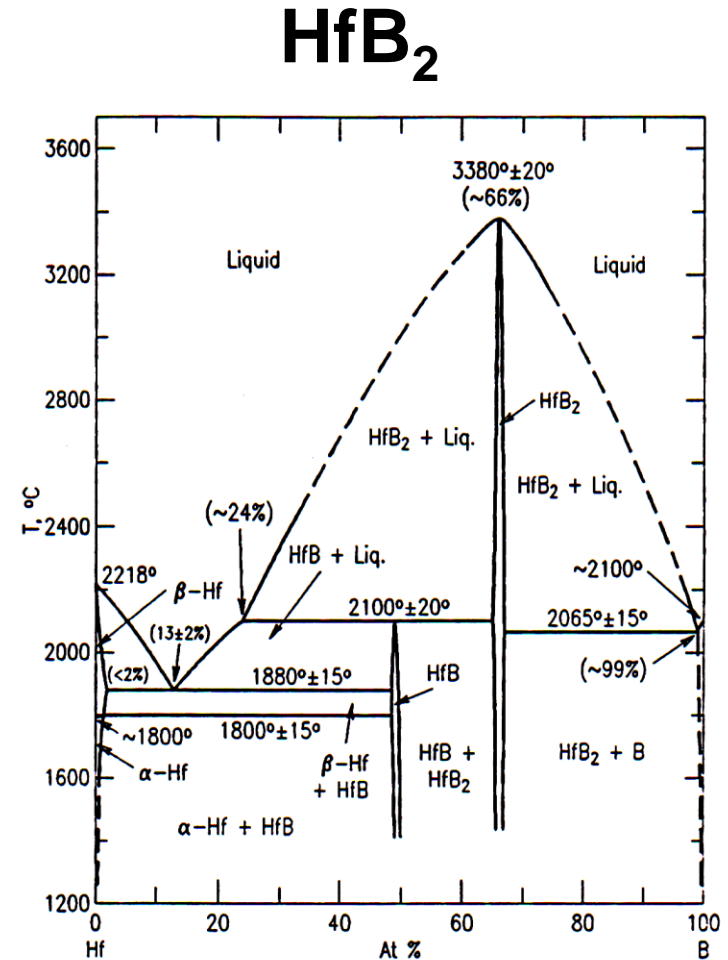
HfB₂-SiC

- HfB₂ has a narrow range of stoichiometry with a melting temperature of 3380°C

Density = 11.2 g/cm³

- Silicon carbide is added to boride powders
 - Promotes refinement of microstructure
 - Decreases thermal conductivity of HfB₂
 - 20v% may not be optimal but is common amount added
 - SiC will oxidize either passively or actively, depending upon the environment

Density = 3.2 g/cm³





UHTC Material Properties

Sharp leading edges require :

- High thermal conductivity (directional)
- High fracture toughness/mechanical strength/hardness
- Oxidation resistance (in reentry conditions)

Property	HfB ₂ /20vol%SiC	ZrB ₂ /20vol%SiC
Density (g/cc)	9.57	5.57
Strength (MPa) 21°C	356±97*	552±73*
1400°C	137±15*	240±79*
Modulus (GPa) 21°C	524±45	518±20
1400°C	178±22	280±33
Coefficient of Thermal Expansion (x10 ⁻⁶ /K) RT	5.9	7.6
Thermal Conductivity (W/mK) [#] RT	80	99

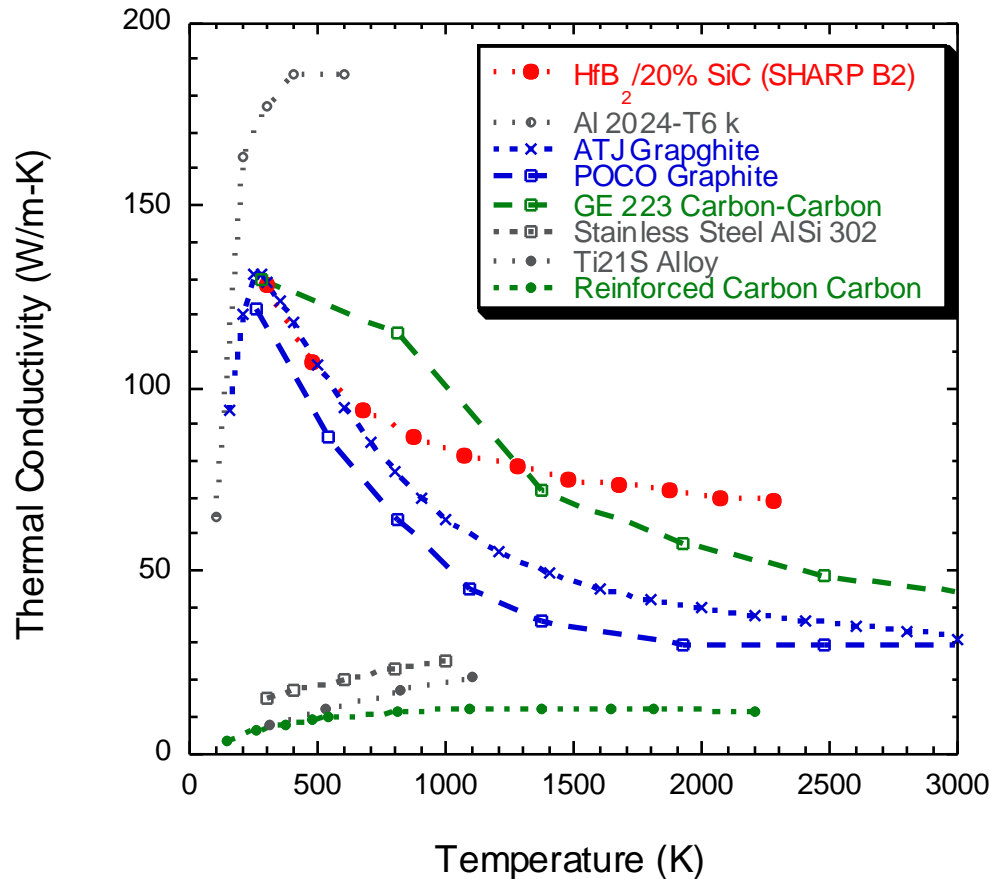
Source: ManLabs and Southern Research Institute

* Flexural Strength

R. P. Tye and E. V. Clougherty, "The Thermal and Electrical Conductivities of Some Electrically Conducting Compounds." Proceedings of the Fifth Symposium on Thermophysical Properties, The American Society of Mechanical Engineers, Sept 30 – Oct 2 1970. Editor C. F. Bonilla, pp 396-401.



Thermal Conductivity Comparison



HfB₂/SiC materials have relatively high thermal conductivity

- HfB₂/SiC material was measured on material from the SHARP B2 program.
- Thermal Diffusivity and Heat Capacity of HfB₂/SiC were measured using Laser Flash.



Some UHTC Development History

- Hf and ZrB_2 materials investigated in early 1950s as nuclear reactor material
- Extensive work in 1960s & 1970s (by ManLabs for Air Force) showed potential for HfB_2 and ZrB_2 for use as nosecones and leading edge materials (Clougherty, Kaufman, Kalish, Hill, Peters, Rhodes et al.)
- Gap in sustained development during 1980s and most of 1990s
 - AFRL considered UHTCs for long-life, man-rated turbine engines
- During late 1990s, NASA Ames revived interest in HfB_2/SiC , ZrB_2/SiC ceramics for sharp leading edges
- Ballistic flight experiments: Ames teamed with Sandia National Laboratories New Mexico, Air Force Space Command, and TRW
 - SHARP*-B1 (1997) UHTC nosetip & SHARP-B2 (2000) UHTC strake assembly
- Space Launch Initiative (SLI), NGLT, UEET programs: 2001-5
- NASA's Fundamental Aeronautics Program funded research until 2009
- Substantial current ongoing effort at universities, government agencies, & international laboratories

* Slender Hypervelocity Aerothermodynamic Research Probes



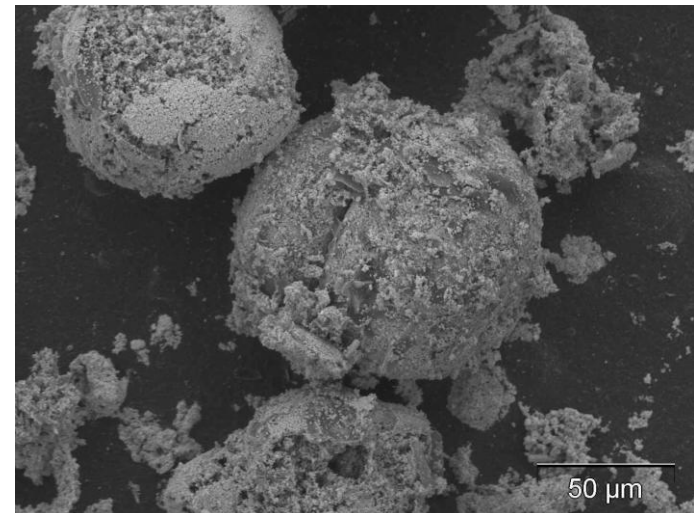
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Improving Processing and Microstructure

- Initial focus on improving material microstructure and strength
- $\text{HfB}_2/20\text{vol}\%\text{SiC}$ selected as baseline material for project constraints
- Major issue was poor mixing/processing of powders with different densities
 - Used freeze-drying to make homogenous powder granules
 - Developed appropriate hot pressing schedules

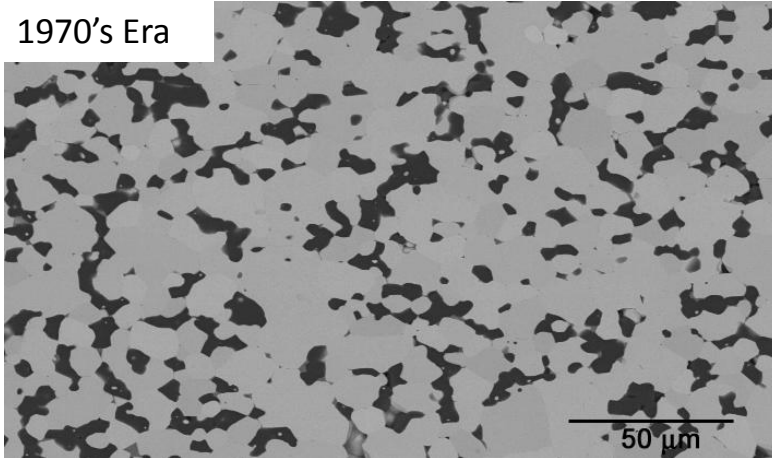


Granulated HfB_2/SiC Powder

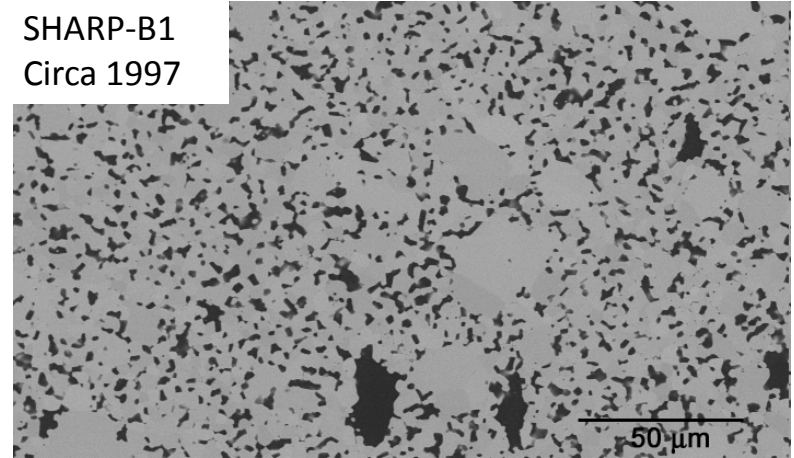


Early Progress in Processing of HfB_2 - 20% SiC Materials

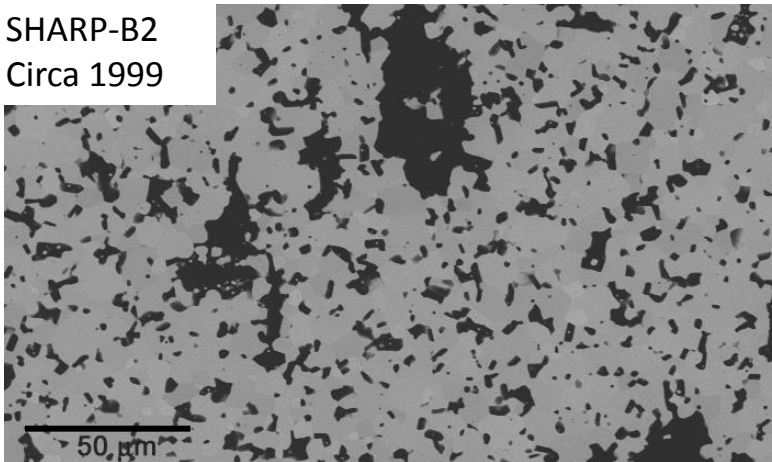
1970's Era



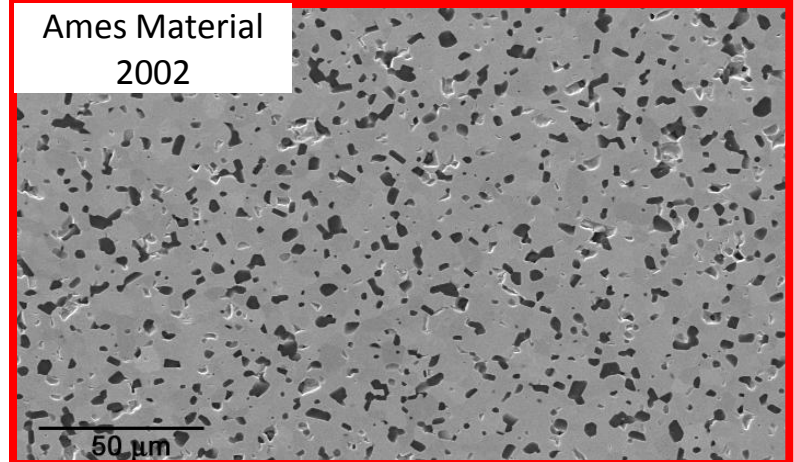
SHARP-B1
Circa 1997



SHARP-B2
Circa 1999



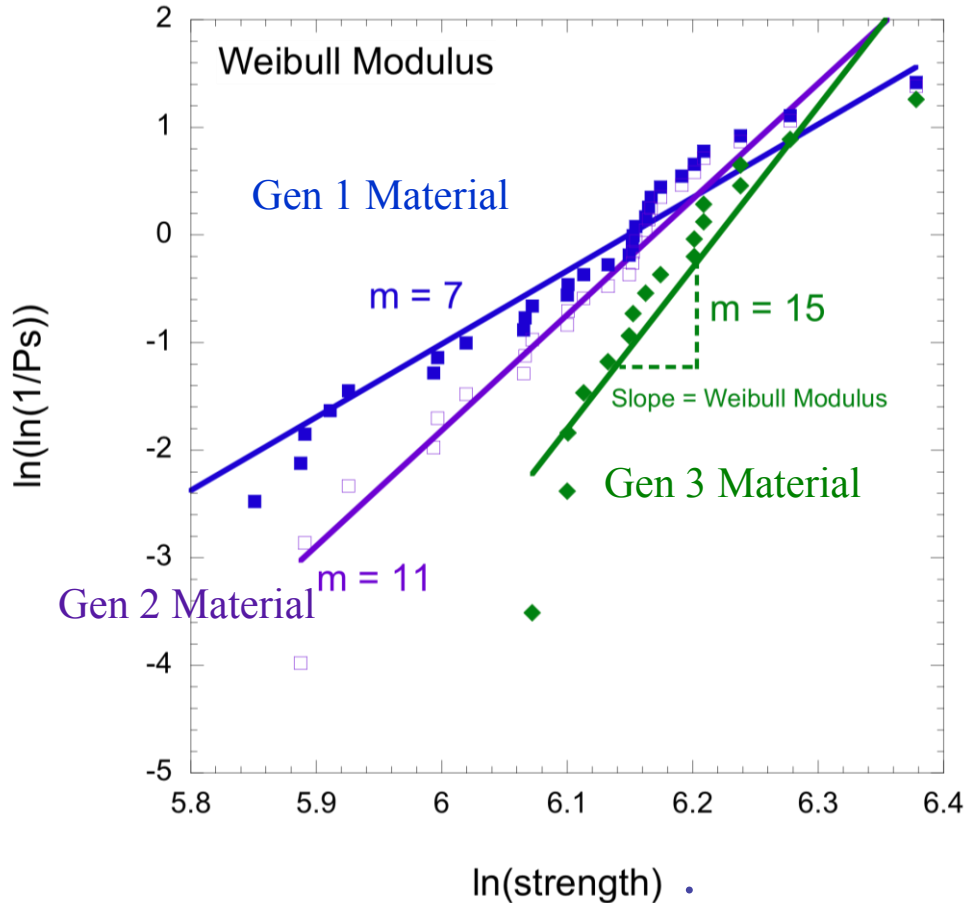
Ames Material
2002



- Early and SHARP materials made by an outside vendor
- Improvements in powder handling provide a more uniform microstructure



Weibull Modulus of ARC HfB₂/SiC Improved Compared to Previous Materials



Weibull Modulus SHARP B2 Materials ~4

Increased Weibull Modulus to ~15 with processing improvements



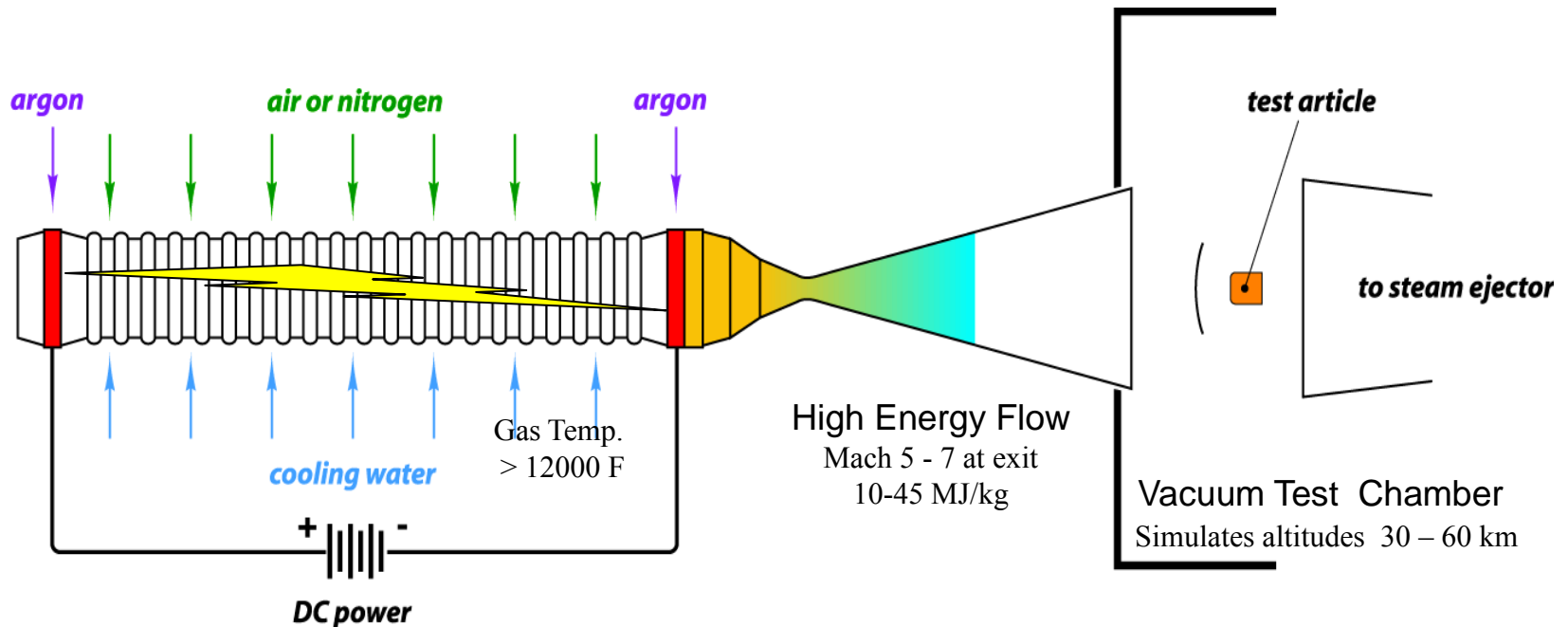
Need for Arc Jet Testing

- Arc jet testing is the best **ground-based method** of evaluating a materials oxidation/ablation response in re-entry environments
- A material's oxidation behavior when heated in static or flowing air at ambient pressures is likely to be significantly different than in a re-entry environment.
- In a re-entry environment:
 - Oxygen and nitrogen may be dissociated
 - Catalycity of the material plays an important role
 - Recombination of O and N atoms adds to surface heating
 - Stagnation pressures may be less than 1 atm.
 - Influence of active to passive transitions in oxidation behavior of materials
 - SiC materials show such a transition when the protective SiO₂ layer is removed as SiO



Arc Jet Schematic

Simulates reentry conditions in a ground-based facility



Method: Heat a test gas (air) to plasma temperatures by an electric arc, then accelerate into a vacuum chamber and onto a stationary test article

Stine, H.A.; Sheppard, C.E.; Watson, V.R. Electric Arc Apparatus. U.S. Patent 3,360,988, January 2, 1968.



UHTC Cone After 9 Arc Jet Exposures (89 minutes total run time)

Runs 4 and 5 lasted ~ 2 min. each

HSp-45
Pretest

Run 1
Post-Test

Run 2
Post-Test

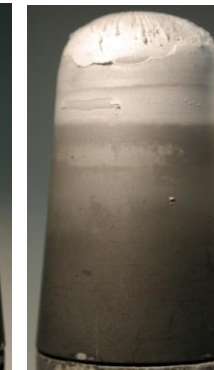
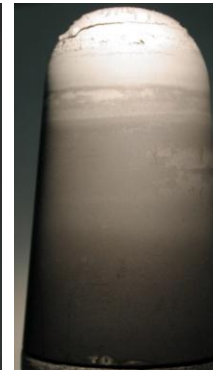
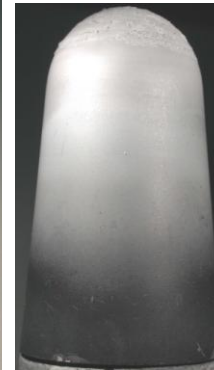
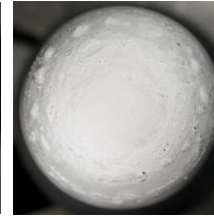
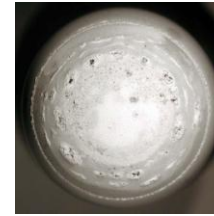
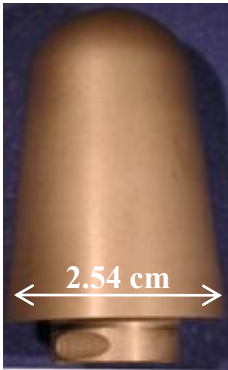
Run 3
Post-Test

Run 6
Post-Test

Run 7
Post-Test

Run 8
Post-Test

Run 9
Post-Test



300 sec
% \square wt = 0
 $T_{ss} = 1280^{\circ}\text{C}$

600 sec
% \square wt = 0
 $T_{ss} = 1220^{\circ}\text{C}$

600 sec
% \square wt = 0
 $T_{ss} = 1325^{\circ}\text{C}$

600 sec
% \square wt = -0.06
 $T_{ss} = 1970^{\circ}\text{C}$

1200 sec
% \square wt = -0.2
 $T_{ss} > 2000^{\circ}\text{C}$

1200 sec
% \square wt = -0.32
 $T_{ss} > 2000^{\circ}\text{C}$

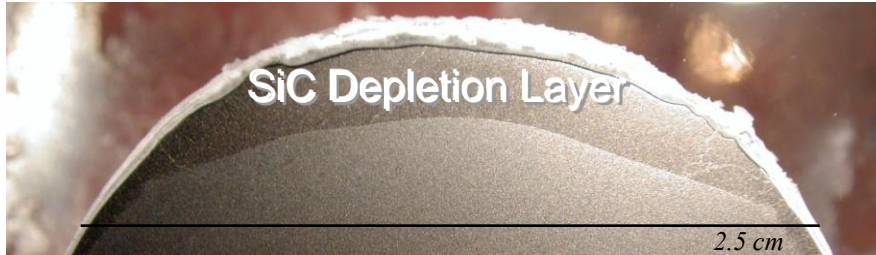
600 sec
% \square wt = -1.24
 $T_{ss} > 2000^{\circ}\text{C}$

Increasing heat flux



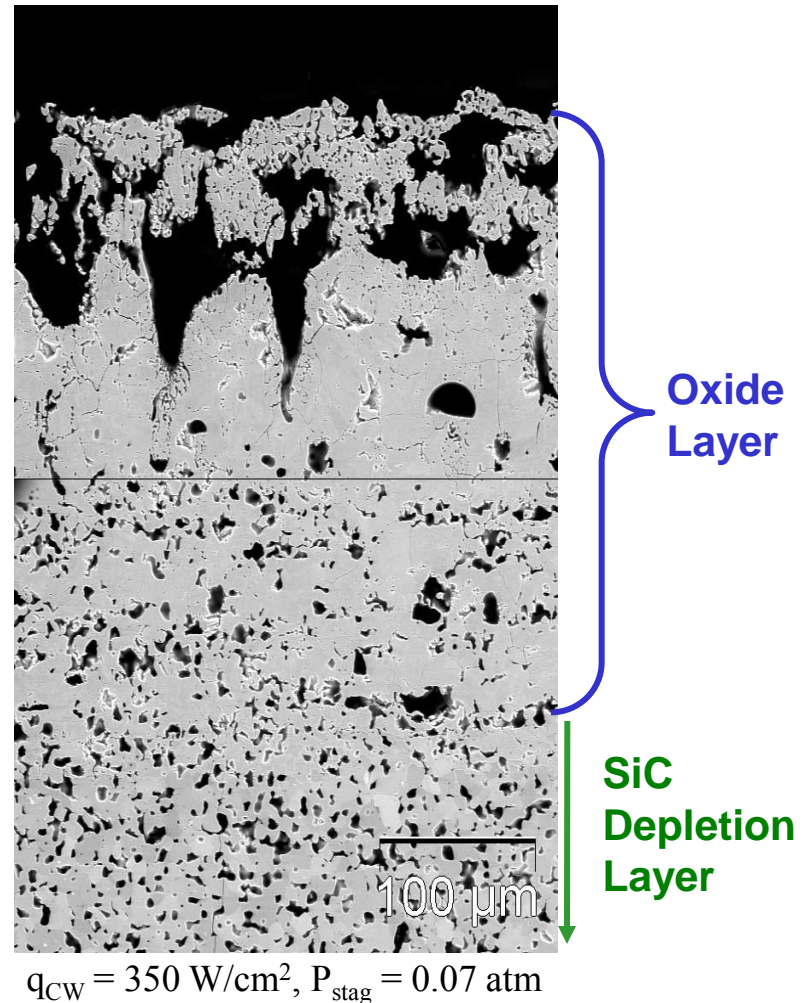


Reducing Oxide Formation



* Post-test arc jet nosecone model after a total of 80 minutes of exposure. Total exposure the sum of multiple 5 and 10 minute exposures at heat fluxes from $200\text{W}/\text{cm}^2$

- In baseline material:
 - SiC depleted during arc jet testing
 - Surface oxide is porous
- Potential solution: Reduce amount of SiC below the percolation threshold while maintaining mechanical performance



*Arc jet test data from Space Launch Initiative program



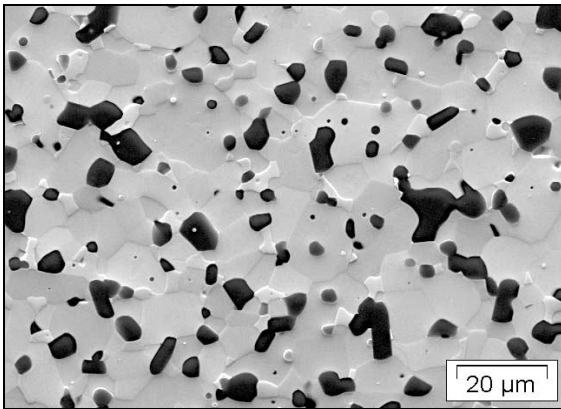
Controlling Microstructure & Composition

- Goal for UHTCs for TPS has been to improve:
 - **Fracture toughness**
 - Strength
 - Thermal conductivity
 - **Oxidation resistance — arcjet performance**
- Properties controlled by processing, microstructure, and composition
 - **Grain Size**
 - Additives (Ir additions)
 - Processing by field-assisted sintering (FAS)
 - **Grain Shape**
 - Addition of preceramic polymers
 - Particle coatings (Fluidized Bed CVD)
 - Purity (grain boundaries)
 - Addition of preceramic polymers
 - Processing (FB CVD)
 - Self-propagating reactions
 - **Oxide formation**
 - Increase oxide stability / emissivity (additives)
 - Reduce amount of SiC

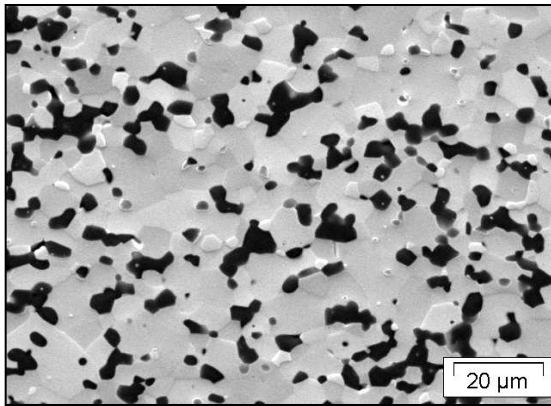


Control of Grain Size

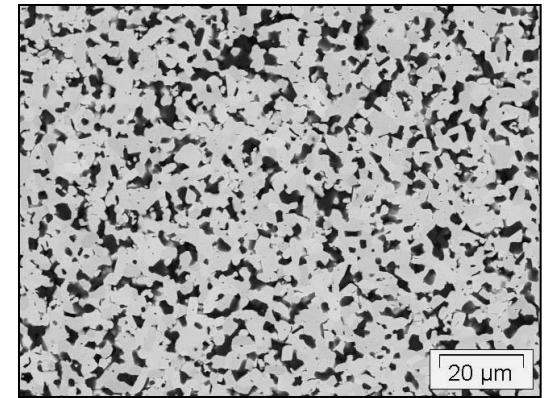
HfB₂/20v%SiC
Hot Pressed
(long process)



HfB₂/20v%SiC
Hot Pressed
(short process)



HfB₂/20v%SiC
Spark Plasma Sintered





Third-Phase Additions

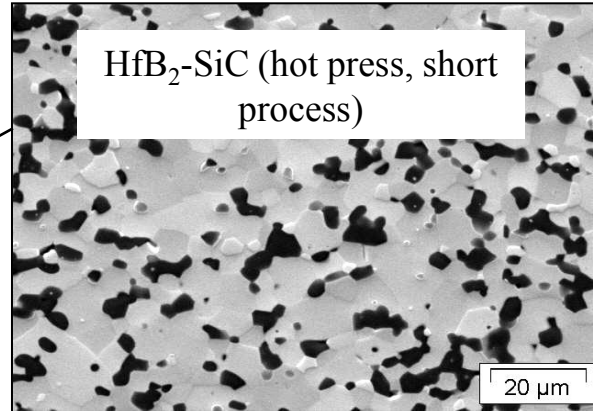
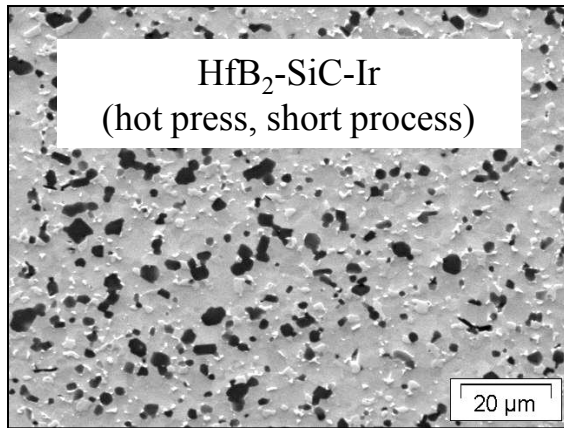
Explore effect of additional refractory phases on oxidation resistance / fracture toughness (ductile-phase toughening)

- Effect of additives on microstructure of baseline material (HfB₂-20 v% SiC):
 - Ir
 - Ir with TaSi₂
- Evaluation of thermal conductivity
- Evaluation of mechanical properties

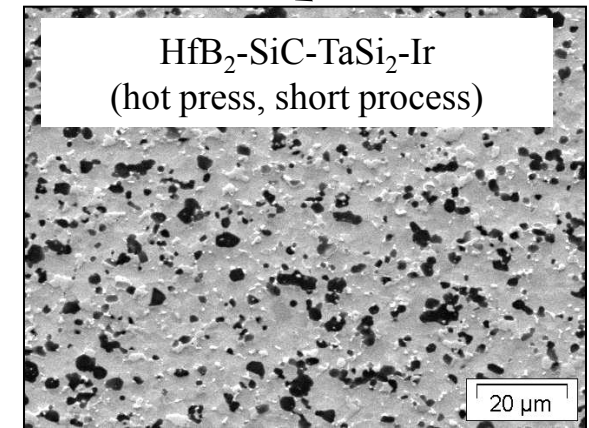


Effect of Additives on Microstructure

Addition of Ir
(short process)

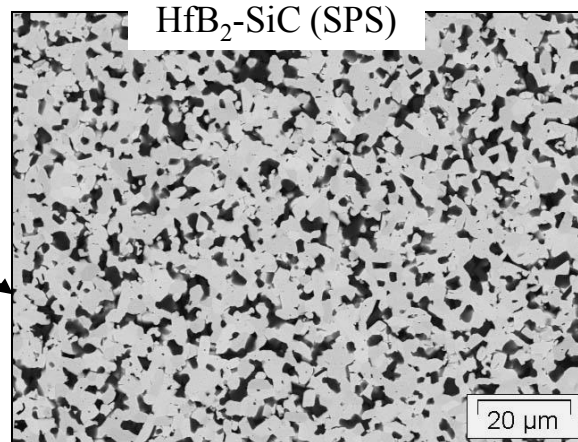


Addition of Ir and TaSi₂
(short process)



*Samples processed with
additional phases show
less grain growth*

Similar microstructure



Similar microstructure



Increasing Oxide Emissivity

HfB₂-SiC



HfB₂-SiC-TaSi₂



- Arcjet test: Performance of HfB₂/SiC-TaSi₂ **comparable** to HfB₂/SiC after testing for 5 minutes at $Q_{cw} \sim 300 \text{ W/cm}^2$
- HfB₂/SiC-TaSi₂ clearly has a higher post-test emissivity than HfB₂/SiC and demonstrated lower surface temperatures

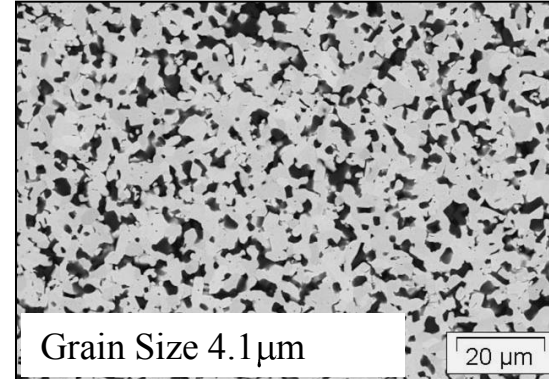
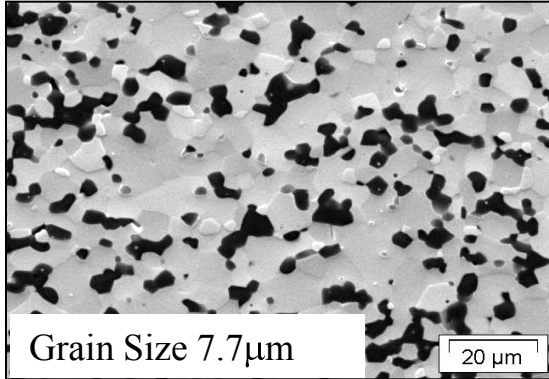
Opila, E. and Levine, S., "Oxidation of ZrB₂- and HfB₂-based ultra-high temperature ceramics: Effect of Ta additions," *Journal of Materials Science* 39 (2004) 5969–5977



Physical Characterization: Microstructure

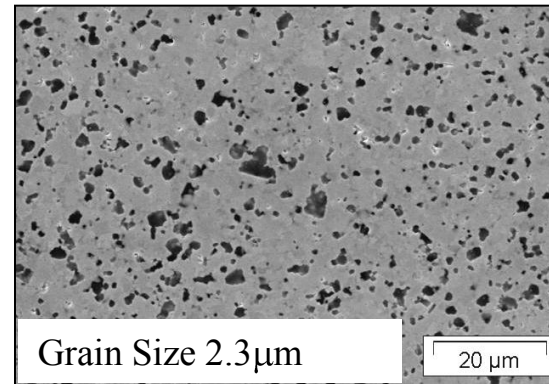
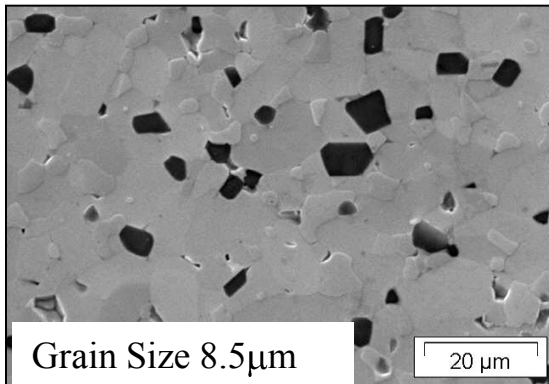
Hot Pressed

HfB₂-SiC
Baseline

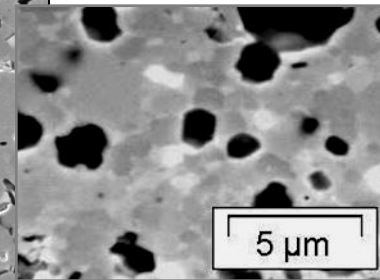
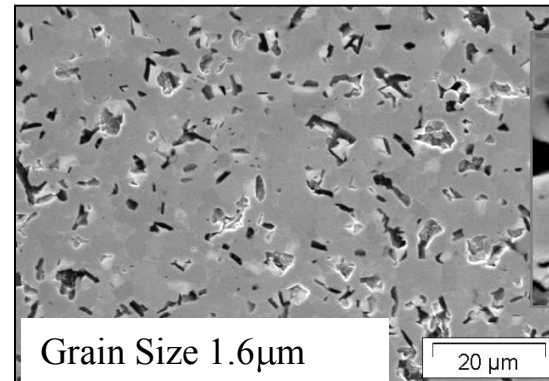
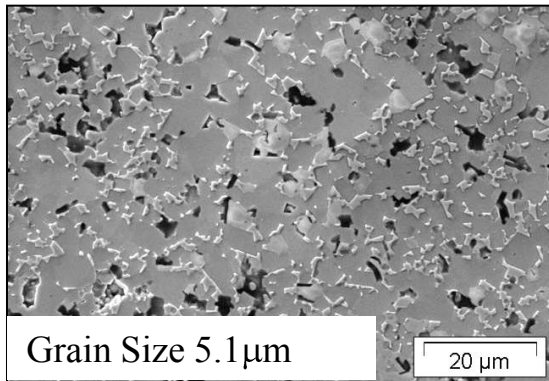


Spark Plasma
Sintered (SPS)

HfB₂-SiC-
TaSi₂

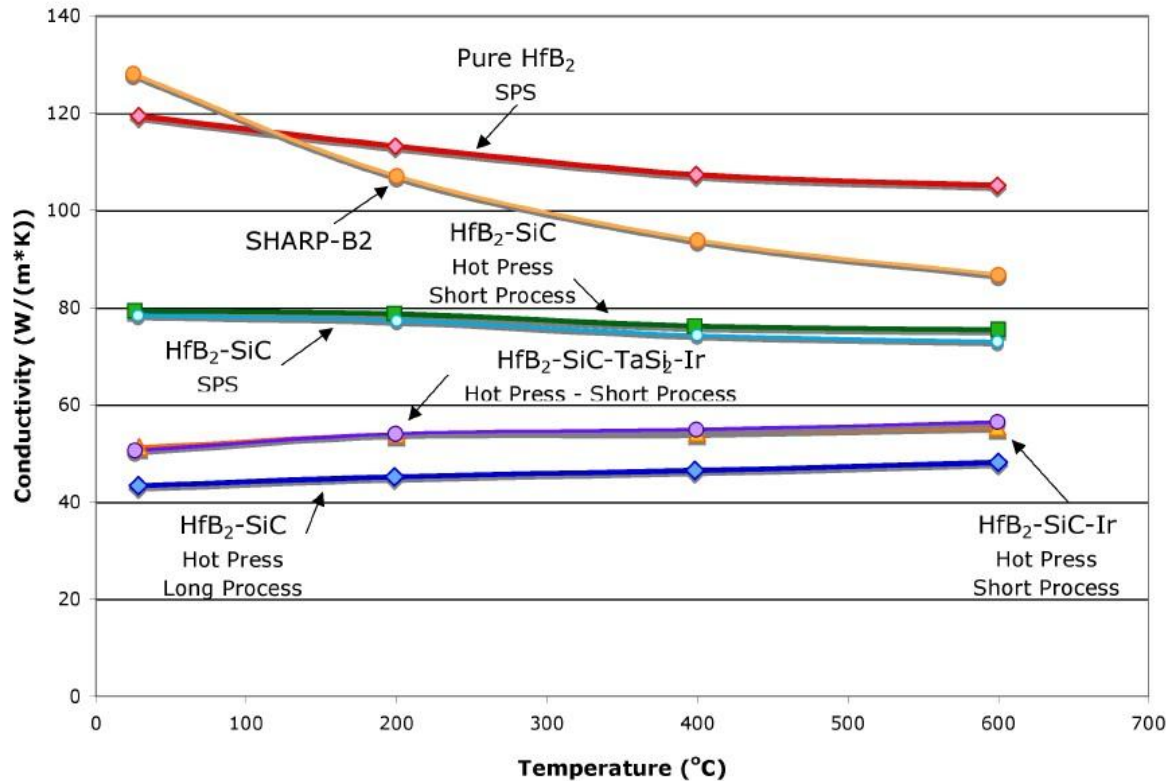


HfB₂-SiC-
TaSi₂-Ir





Thermal Conductivity



- Long process, hot pressing — lowest thermal conductivity
- Short process, hot pressing — significant increase in thermal conductivity
- SPS — similar increase in thermal conductivity to short process
- Addition of Ir or Ir and TaSi₂ to HfB₂/SiC (modified HP) — lowers thermal conductivity

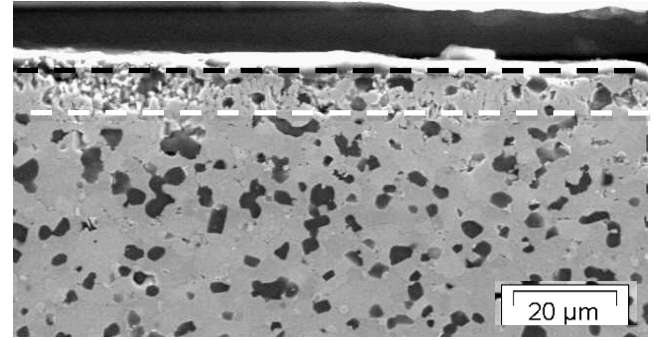
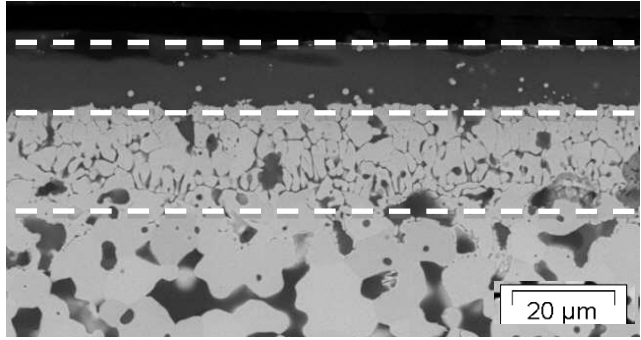


Arc Jet Characterization: Additives & Influence of Microstructure

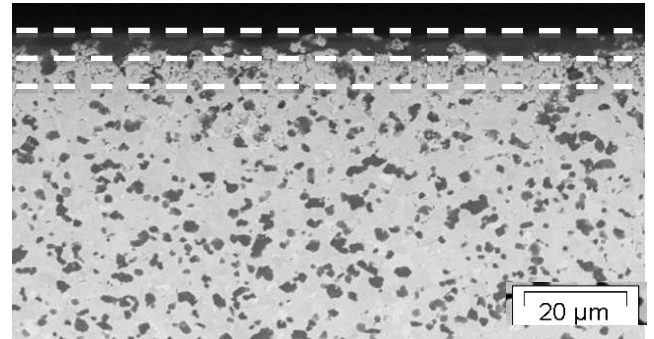
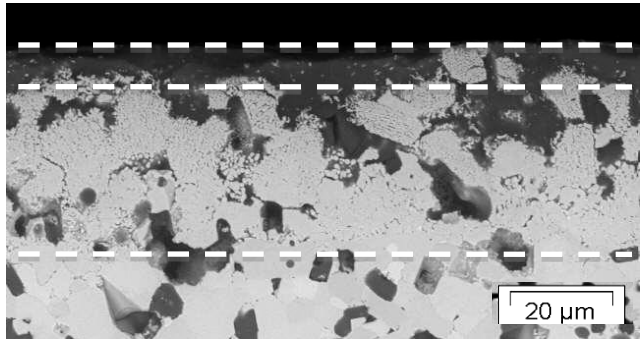
Hot Pressed

Spark Plasma Sintered

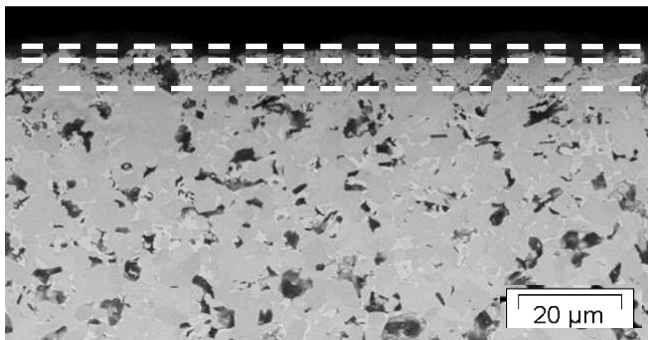
HfB₂-SiC
Baseline



HfB₂-SiC-
TaSi₂



HfB₂-SiC-
TaSi₂-Ir



**Both oxide scale and
depletion zone can
be reduced.**



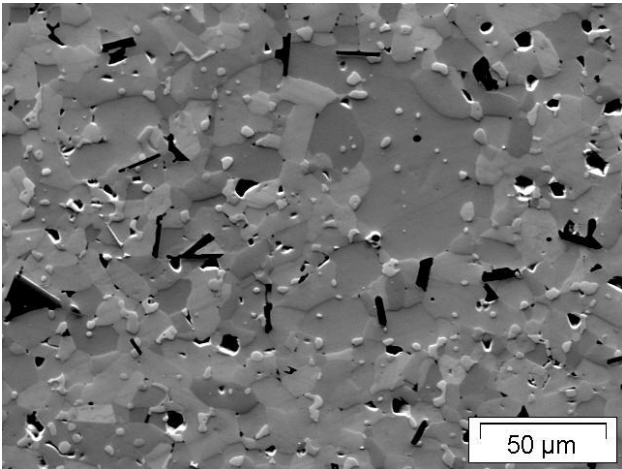
Controlling Grain Shape

Role of SiC Source

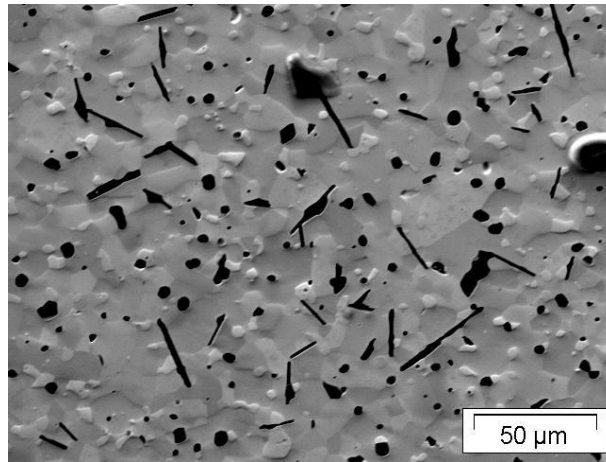
- Conventional source of SiC is powder.
- Preceramic polymer source:
 - Preceramic polymer will affect densification and morphology.
 - May achieve better distribution of SiC source through HfB_2 .
 - Previous work shows that preceramic polymers can enhance growth of acicular particles (for fracture toughness).
- Potential to improve mechanical properties with reduced amount of SiC and also potentially improve oxidation behavior.



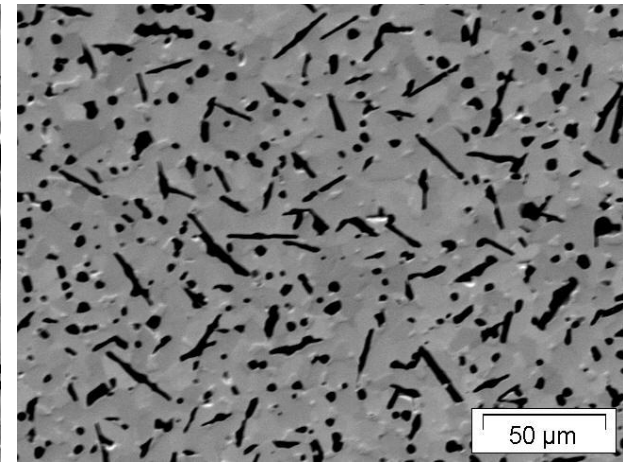
Growth of Elongated SiC Grains



5%* SiC



10%* SiC — Rod diameter $\sim 2\mu\text{m}$



15%* SiC — Rod diameter $\sim 5\mu\text{m}$

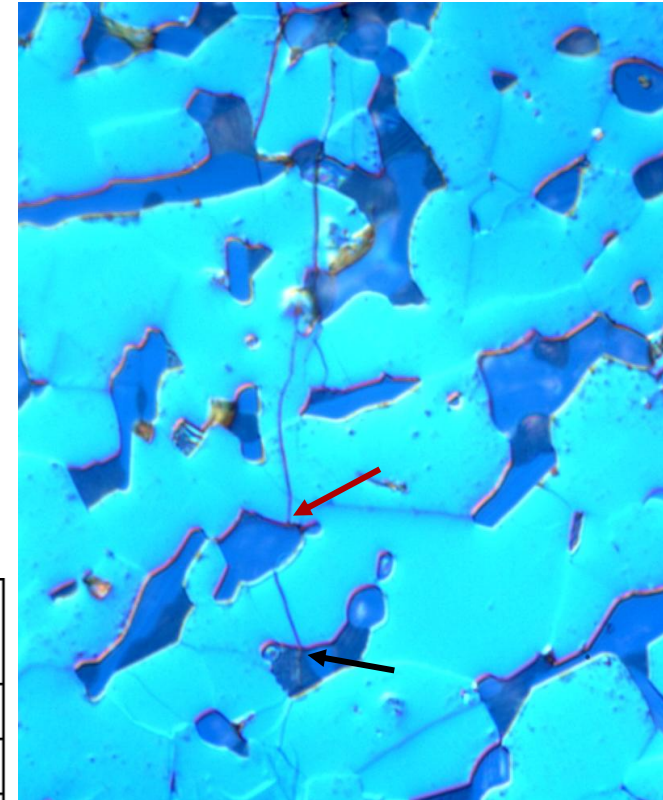
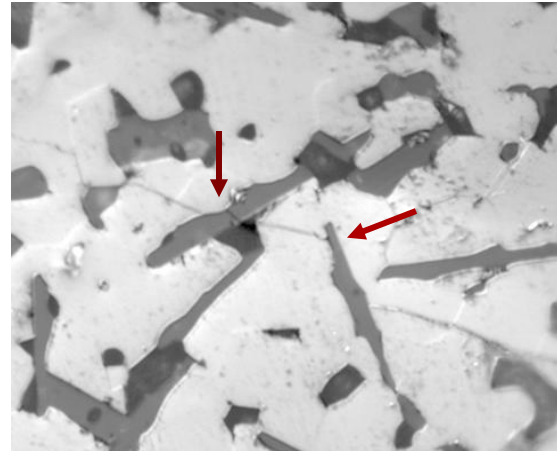
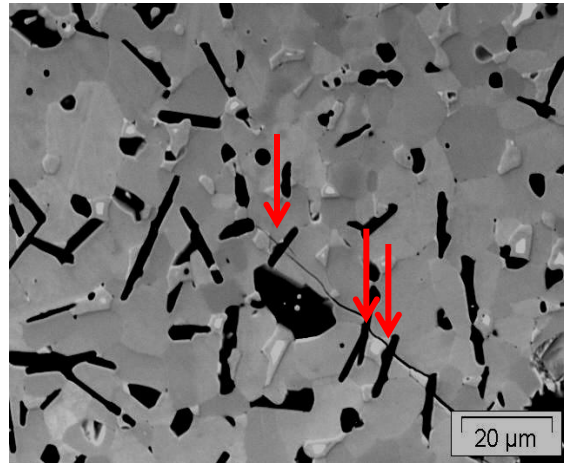
SiC Preceramic Polymer Promotes Growth of Acicular Grains

- Samples processed with 5 to >20 volume % SiC
- Can adjust volume of SiC in the UHTC without losing the high I/d architecture
- Amount of SiC affects number and thickness (but not length) of rods — length constant ($\sim 20\text{--}30\mu\text{m}$)
- Possible to obtain dense samples with high-aspect-ratio phase
- Hardness of high-aspect-ratio materials comparable to baseline material

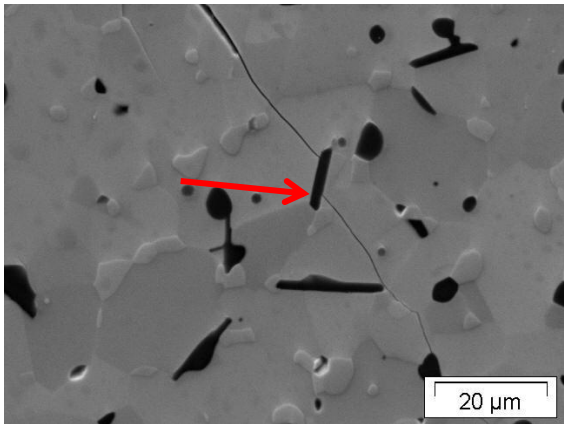
* Precursor added in amounts sufficient to yield nominal amounts of SiC



In Situ Composite for Improved Fracture Toughness



Oak Ridge National Laboratory

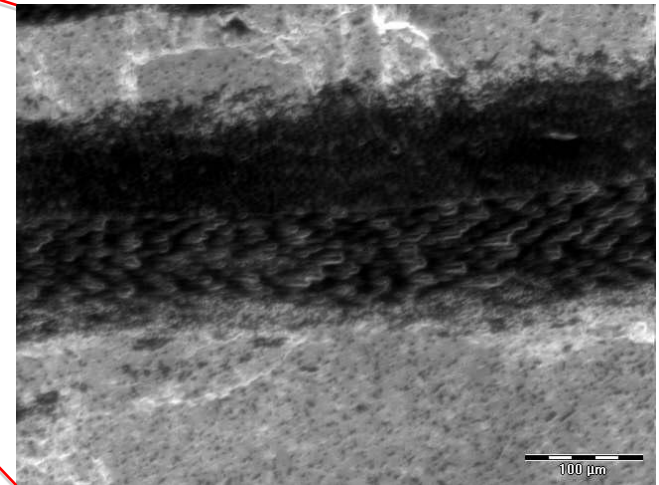
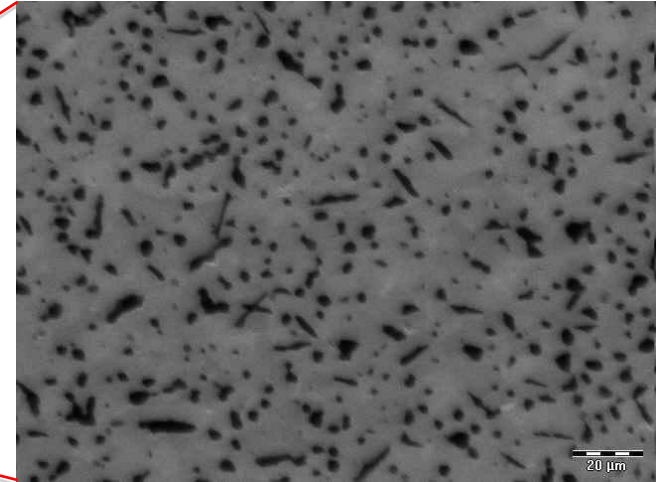
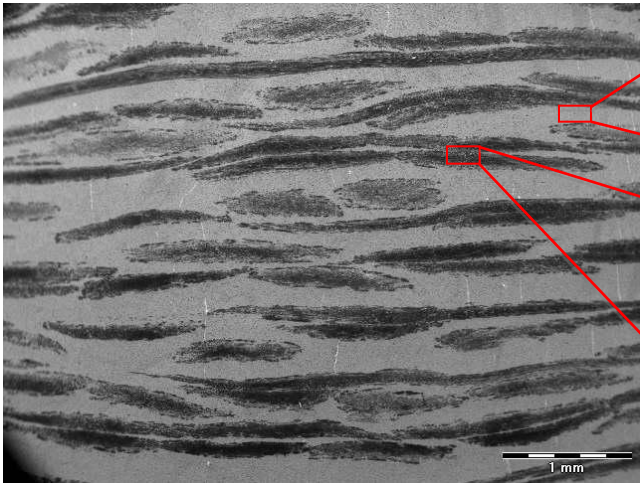


SiC Content	Fracture Toughness (MPam ^{1/2})
5%	3.61
10%	4.06
15%	4.47
Baseline UHTC (20%)	4.33

Evidence of crack growth along HfB₂-SiC interface, with possible SiC grain bridging



Ultra High Temperature Continuous Fiber Composites



- Image at top right shows dense UHTC matrix with indications of high aspect ratio SiC.
- Image at bottom right shows the presence of C fibers after processing.



Outline

- The Process of Discovery
- Education
- Experience
- Curiosity
- Opportunity
- Discovery
- **New Frontiers**
- Conclusion



UHTC Challenges

1. Fracture toughness

Composite approach is required

- Integrate understanding gained from monolithic materials
- Need high temperature fibers

2. Oxidation resistance in reentry environments/ hypersonic flight

Promising approaches but challenge is active oxidation of materials containing SiC

3. Modeling is critical

Shorten development time, improve properties, design



Outline

- The Process of Discovery
- Education
- Experience
- Curiosity
- Opportunity
- Discovery
- New Frontiers
- **Conclusion**



Some Recent Research Efforts in UHTCs: Materials and Properties

ZrB₂ Based Ceramics	Catalytic Properties of UHTCs
Missouri University of Science & Technology	PROMES-CNRS Laboratory, France
US Air Force Research Lab (AFRL)	CNR-ISTEC
NASA Glenn Research Center	CIRA, Capua, Italy
University of Illinois at Urbana-Champaign	SRI International, California
Harbin Institute of Technology, China	Imaging and Analysis (Modeling)
Naval Surface Warfare Center (NSWC)	University of Connecticut
NIMS, Tsukuba, Japan	AFRL
Imperial College, London, UK	NASA Ames Research Center
NASA Ames Research Center	Teledyne (NHSC-Materials and Structures)
CNR-ISTEC	Oxidation of UHTCs
HfB₂ Based Ceramics	AFRL
NASA Ames Research Center	NASA Glenn Research Center
NSWC—Carderock Division	Georgia Institute of Technology
Universidad de Extremadura, Badajoz, Spain	Missouri University of Science & Technology
CNR-ISTEC, Italy	Texas A & M University
Fiber Reinforced UHTCs	CNR-ISTEC, Italy
Chinese Academy of Sciences, Shenyang	University of Michigan, Ann Arbor, Michigan
University of Arizona	NSWC—Carderock
MATECH/GSM Inc., California	Harbin Institute of Technology, China
AFRL	University of Illinois at Urbana-Champaign

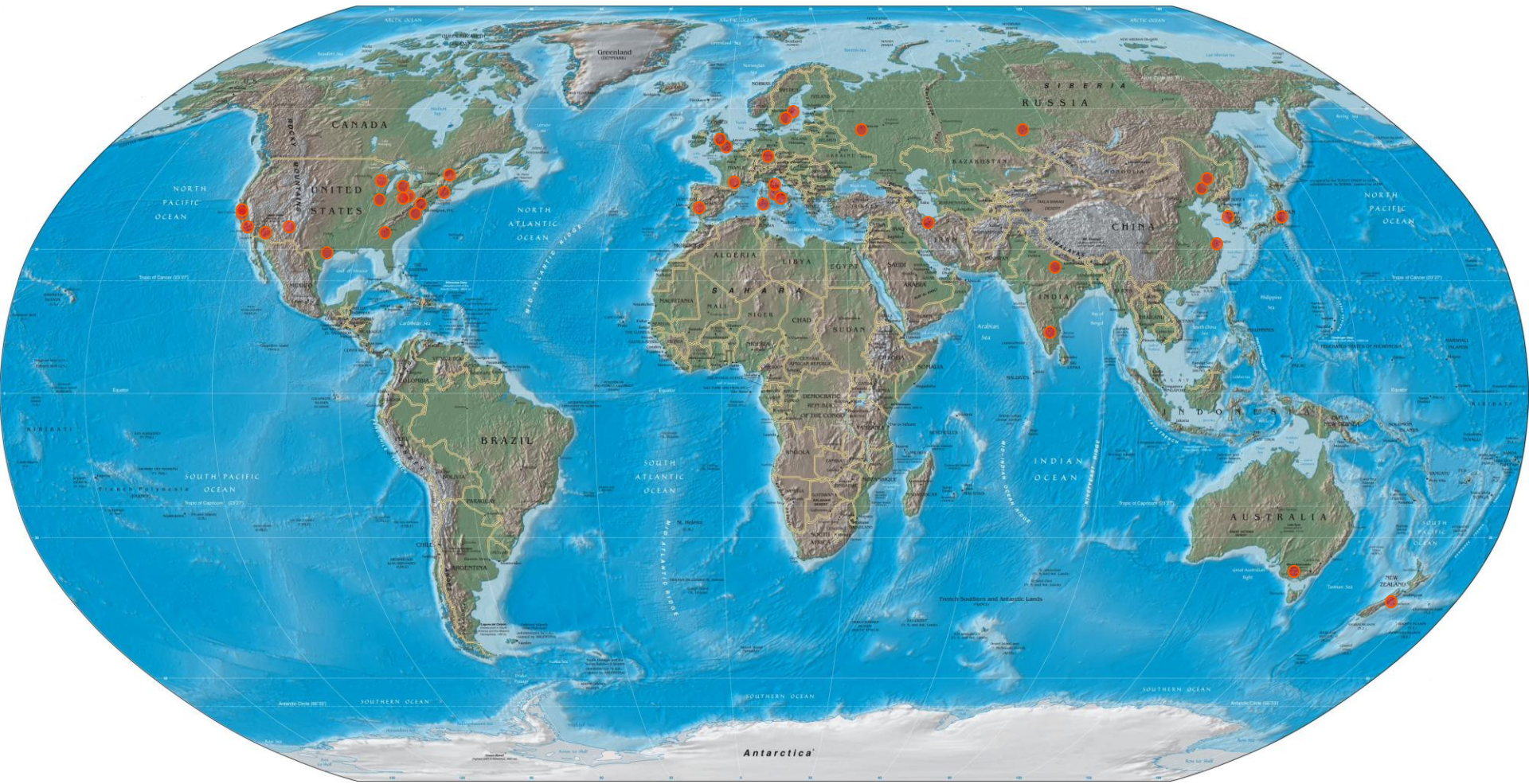


Some Recent Research Efforts in UHTCs: Processing

Field Assisted Sintering	UHTC Polymeric Precursors
University of California, Davis	SRI International, California
Air Force Research Laboratory (AFRL)	University of Pennsylvania
CNR-ISTEC, Italy	Missouri University of Science & Technology
Stockholm University, Sweden	MATECH/GSM Inc., California
NIMS, Tsukuba, Japan	Teledyne (NHSC)
Pressureless Sintering	
Missouri University of Science & Technology	
Reactive Hot-Pressing	
Shanghai Institute of Ceramics, China	
NASA Ames Research Center	
National Aerospace Laboratories, India	
Sandia National Laboratories, New Mexico	
McGill University, Montreal, Canada	



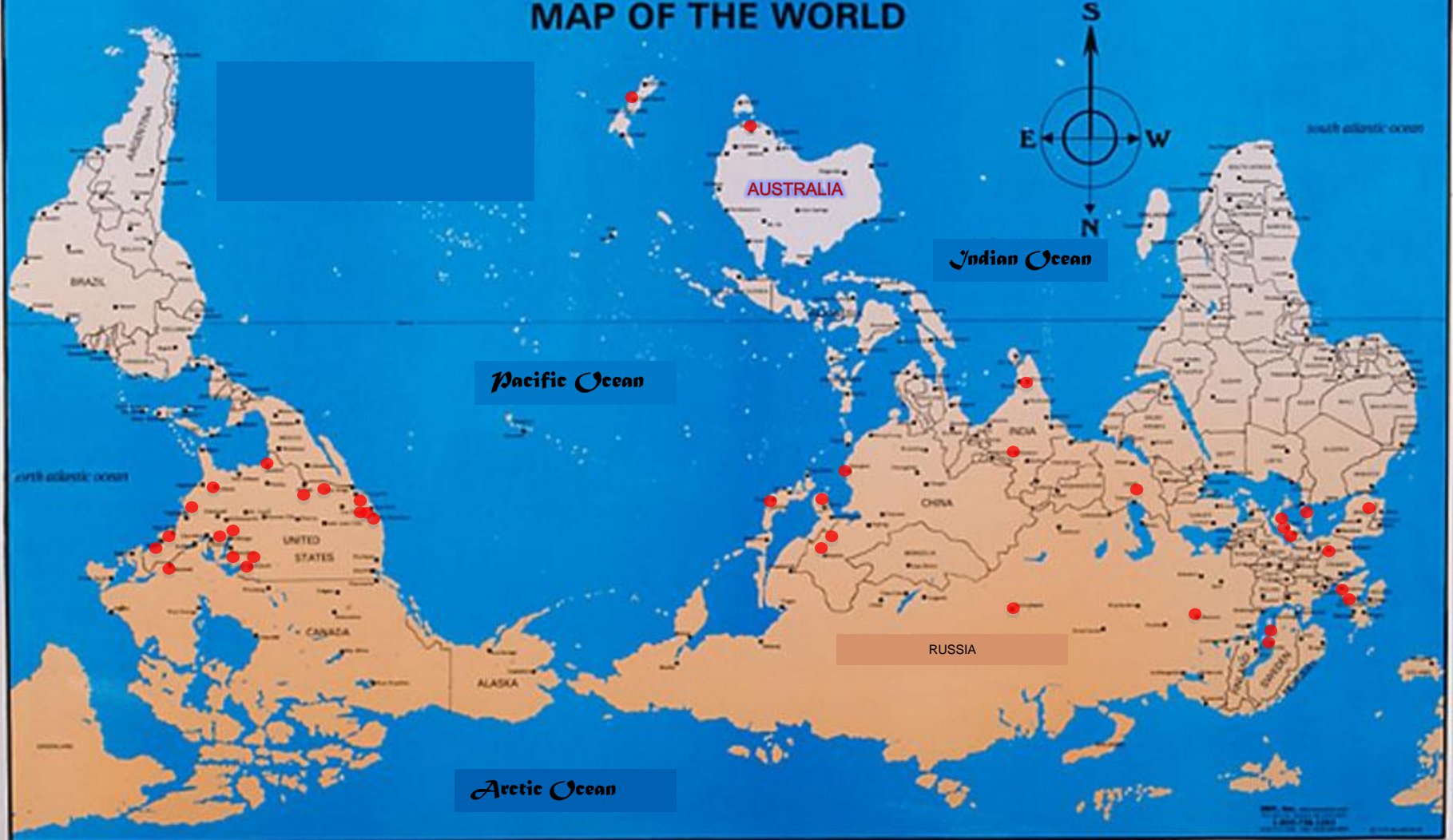
UHTC Researchers Throughout the World





Another Perspective.....

McARTHUR'S UNIVERSAL CORRECTIVE MAP OF THE WORLD



AUSTRALIA

Indian Ocean

Pacific Ocean

RUSSIA

Arctic Ocean

south atlantic ocean

south atlantic ocean

UNITED STATES

CANADA

ALASKA

CHINA

INDIA

FINLAND

SWEDEN

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