



Changes in Global Tropospheric OH Expected as a Result of Climate Change Over the Last Several Decades



Julie M. Nicely¹, R. J. Salawitch¹, T. Canty¹, A. R. Douglass², B. N. Duncan², C. Lang³, Q. Liang², L. D. Oman², J. M. Rodriguez², R. S. Stolarski³, S. Strode^{2,4}, D. W. Waugh³
¹University of Maryland, College Park, Md; ²NASA Goddard Space Flight Center, Greenbelt, Md; ³Johns Hopkins University, Baltimore, Md; ⁴Universities Space Research Association, Columbia, Md

1. Scientific motivation for studying tropospheric OH

Sensitivity of OH^{TROP} to climate change

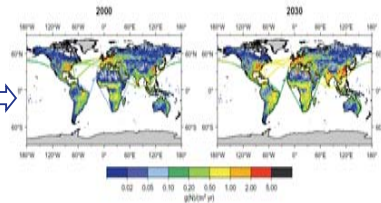
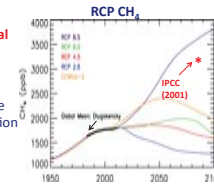
- IPCC (2001) predicted future OH would fall due to rising CH₄
- Best knowledge of global tropospheric OH (OH^{TROP}) comes from analysis of CH₃CCl₃ observations
- The CH₄ lifetime inferred from OH^{TROP} based on CH₃CCl₃ is 8.9 years; yet many sources use a CH₄ lifetime of ~12 years, which was found by modeling studies described in IPCC (2001)
- The 12 year CH₄ lifetime is central to subsequent IPCC reports and is called the "perturbation lifetime"
- Our work is motivated by understanding:
 - what factors other than rising CH₄ will affect OH^{TROP}
 - veracity of notion that future OH^{TROP} will decline

Importance of OH^{TROP}

- OH serves as the main sink for many species: CO, CH₄, CH₃CCl₃, all HCFCs and HFCs, biogenic VOCs (isoprene and terpene), anthropogenic VOCs (formaldehyde, benzene, toluene, etc.)
- HO_x is integral to O₃ formation and destruction pathways; alone, HO_x depletes O₃, but with NO_x and VOCs, HO_x creates O₃

Methane Oxidation and OH

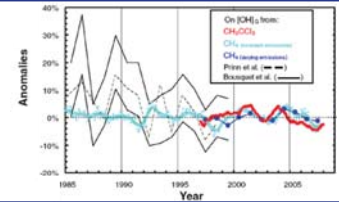
- Under low NO_x, the oxidation of CH₄ destroys HO_x
- Under high NO_x, the oxidation of CH₄ produces HO_x
- **Loss of methoxyperoxy radical CH₃O₂ with NO or HO₂ is critical**
- Future levels of CH₄ and NO_x are highly uncertain
 - CH₄ between now and 2100 varies dramatically among RCP scenarios
 - NO_x between now and 2100 will depend on whether the developing world implements selective catalytic reduction on coal power plants and catalytic converters on cars
- Future OH^{TROP} will also depend on overhead column O₃, local humidity, and biogenic emission of VOCs, etc.
- **Given this complexity, we expect CCMs to project a wide range of values for OH^{TROP} during the rest of this century**



3. Preliminary results for expected changes in OH^{TROP}

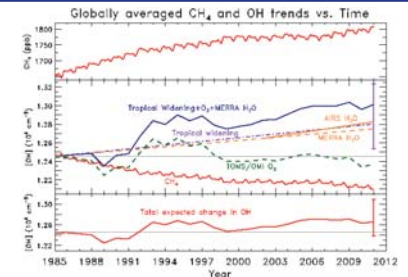
Previous Work on OH^{TROP} Trends

- Montzka *et al.*, 2011 found that OH^{TROP} does not vary interannually (from 1997 to present)
- Prior studies by Prinn *et al.*, 2001 and Bousquet *et al.*, 2005 suggest large interannual variability in OH^{TROP} (1985-2000)
- We suggest the OH^{TROP} behavior in all three studies may be physically possible, based on our preliminary results



Current Conclusions from OH^{TROP} Analysis

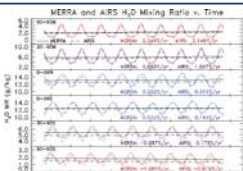
- Effect of CH₄ on OH^{TROP} is taken from IPCC (2001), Section 4.2.1.1, which states that "the feedback of CH₄ on tropospheric OH" found using contemporary chemical transport models is -0.32% for every 1% increase in CH₄ (red line, middle panel of figure to right)
- Primary effect of overhead O₃ is rise in OH^{TROP} following the 1991 eruption of Mount Pinatubo (green dashed line) due to enhanced removal of stratospheric O₃ by volcanic aerosol
- Rising H₂O from MERRA and AIRS increases OH^{TROP} (orange lines) by an amount comparable to the decrease expected from rising CH₄
- Widening of the tropics contributes an increase in OH^{TROP} (purple line) roughly equal to that due to rising H₂O, leading us to view this as a significant factor controlling OH^{TROP}
- Overall expected change in OH^{TROP} (bottom panel) shows higher level of interannual variability prior to ~1999 and lower variability thereafter



2. Methods for estimating changes in OH^{TROP}

Model OH

- Initial modern-day conditions for OH are taken from a time-slice run of GEOS CCM using 2005 emissions
- Monthly mean mixing ratios of OH and related species are provided on a 144 longitude, 91 latitude, 72 pressure level grid
- Changes in OH due to H₂O, overhead O₃ and tropical widening are found from analysis of initial OH field

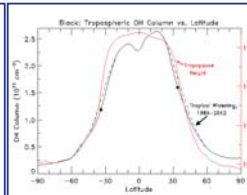


MERRA & AIRS H₂O

- Specific humidity files from the NASA Modern-Era Retrospective analysis for Research and Applications (MERRA) starting prior to 1985, &
- H₂O mixing ratio files from the NASA Atmospheric Infrared Sounder (AIRS) starting in 2002 were used to establish trends in H₂O by latitude
- OH^{TROP} is assumed to follow the square root of the change in H₂O using a steady-state assumption
- We plan to refine the effect of changing H₂O on OH^{TROP} by examining reaction rates from archived runs of GEOS CCM

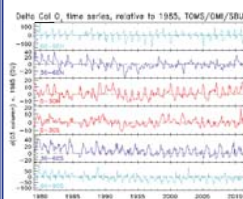
Overhead O₃

- Total column O₃ trends were obtained from the NASA merged O₃ data set, consisting of measurements from SBUV, TOMS, and Aura OMI instruments
- We then use our photolysis code to estimate the impact on J(O₃) → O^{(1)D} of decreasing initial GEOS CCM overhead O₃ columns by amount suggested by the NASA product
- OH^{TROP} is assumed to change by the square root of J(O₃) → O^{(1)D}



Tropical Widening

- Widespread evidence for expansion of the latitudinal circulation system (e.g. Seidel *et al.*, 2008)
- GCMs forced by GHGs cannot reproduce extent of tropical widening (Johanson and Fu, 2009)
- Allen *et al.* (2012) have suggested tropospheric O₃ and BC are responsible
- We simulate tropical expansion by increasing OH near the tropical boundary by a factor representative of 2° widening/decade (globally)



4. Future plans for refining estimates of ΔOH^{TROP}

Improve Estimate of d(OH^{TROP}) / d(H₂O)

- Reaction rates from recent runs of GEOS CCM are archived for reactions such as:

$$H_2O + O(^1D) \rightarrow 2OH$$
- Using these reaction rates we will determine the proportion of OH that is produced via reaction with H₂O
- The determined scaling factor would be used to calculate a new ΔOH^{TROP} based on the H₂O trends
- Estimate time- and pressure-varying values of d(OH^{TROP}) / d(H₂O)
- Evaluate discrepancies between MERRA and AIRS H₂O trends

Propagate Uncertainties

- Calculate uncertainties in AIRS and MERRA H₂O and NASA O₃ product

Evaluate CH₄/OH Feedback

- We will use a box model (details below) to probe relationship between CH₄ and OH^{TROP} and its dependence on NO_x

Box Model

- We will use the Dynamically Simple Model of Atmospheric Chemical Complexity (DSMACC) box model (provided by Mathew Evans) with chemical mechanistic information taken from the Master Chemical Mechanism, MCM v3.2 (Jenkin *et al.*, 1997; Saunders *et al.*, 2003) via website <http://mcm.leeds.ac.uk/MCM>. Using the MCM, we have the capability to consider:
 - the degradation of up to 135 VOCs
 - the influence of J(O^{(1)D}) and J(NO₂) on OH chemistry
 - the impact of NO_x on methane oxidation

- Evaluate standard deviation in average fraction of OH production occurring via H₂O + O^{(1)D}
- Estimate uncertainty in the box model evaluation of d(OH^{TROP}) / d(CH₄)

Assessing OH^{TROP} in CCMs

- Through our involvement with the IGAC / SPARC Chemistry-Climate Model Initiative, we have requested:
 1. hourly, instantaneous output from participating CCMs 1 day/season, 1 year/decade
 2. archival of all species, reaction rates, J-values, and physical parameters relevant to OH chemistry
 3. this output for both the REF-C1 (hindcast) and REF-C2 (future) runs

- We plan to assess the causes of differences between OH in the CCMs
- Use of the box model enables us to distinguish between OH differences due to **chemical mechanism** and those due to **differences in OH precursors**
- We can also use this output to predict future trends in OH^{TROP}, based on CH₄, H₂O, and overhead O₃ from the future CCM runs