

# The use of the Molecular Adsorber Coating technology to mitigate vacuum chamber contamination during Pathfinder testing for the James Webb Space Telescope

## ■ *Nithin S. Abraham*

*Presented on Wednesday, August 31st, 2016*

### **SPIE Paper 9952-11**

*Session 3: Space Mission Contamination: Technology Application, Testing and Flight Measurements*

**Nithin S. Abraham, Mark M. Hasegawa, Eve M. Wooldridge**

*NASA Goddard Space Flight Center, Greenbelt, Maryland*

**Kelly A. Henderson-Nelson**

*Stinger Ghaffarian Technologies, Inc., Greenbelt, Maryland*

**SPIE Optics + Photonics: Optical Engineering + Applications**

*Systems Contamination: Prediction, Control, and Performance 2016 (9952)*

*San Diego Convention Center, San Diego, California, United States of America*



# Abstract

As a coating made of highly porous zeolite materials, the Molecular Adsorber Coating (MAC) was developed to capture outgassed molecular contaminants, such as hydrocarbons and silicones. For spaceflight applications, the adsorptive capabilities of the coating can alleviate on-orbit outgassing concerns on or near sensitive surfaces and instruments within the spacecraft. Similarly, this sprayable paint technology has proven to be significantly beneficial for ground based space applications, in particular, for vacuum chamber environments. This paper describes the recent use of the MAC technology during Pathfinder testing of the Optical Ground Support Equipment (OGSE) for the James Webb Space Telescope (JWST) at NASA Johnson Space Center (JSC). The coating was used as a mitigation tool to entrap persistent outgassed contaminants, specifically silicone based diffusion pump oil, from within JSC's cryogenic optical vacuum chamber test facility called Chamber A. This paper summarizes the sample fabrication, installation, laboratory testing, post-test chemical analysis results, and future plans for the MAC technology, which was effectively used to protect the JWST test equipment from vacuum chamber contamination.

- **Keywords:** Molecular Adsorber Coating, zeolite, molecular adsorber, adsorber, adsorption, outgassing, molecular contamination, spaceflight applications, vacuum applications, James Webb Space Telescope, JWST, Chamber A, DC-704, diffusion pump oil, silicones, sprayable paint technology, coatings, getters, passive getter



# Table of Contents

## ■ Introduction

■ Molecular Adsorber Coating .....	5
■ Vacuum Applications .....	8
■ Chamber A .....	10
■ Molecular Contaminants .....	14

## ■ Application Efforts

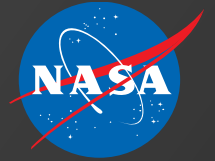
■ Fabrication & Installation .....	18
■ MAC Barn Door Panels .....	19
■ MAC Plenum Samples .....	23

## ■ Testing Efforts

■ Molecular Capacitance Testing .....	28
■ Chemical Analysis	
■ <i>Vacuum Desorption Bake-Out Method</i> .....	31
■ <i>Sample Solvent Rinse Method</i> .....	34
■ Closing Remarks .....	41

## ■ Conclusions

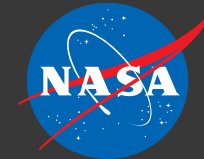
■ Future Plans .....	43
■ Acknowledgements .....	44
■ References .....	46
■ Contact Information .....	47



# Introduction

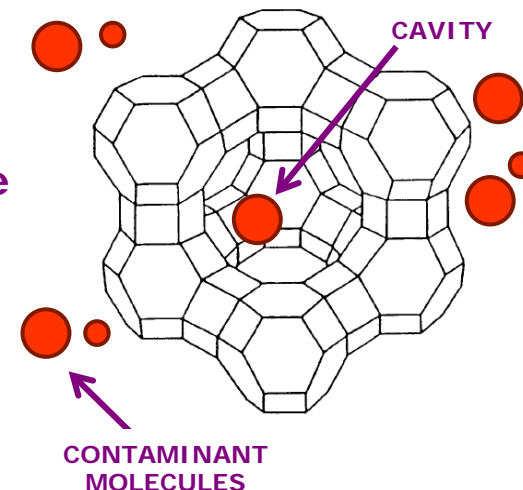
- *Molecular Adsorber Coating*
- *Vacuum Applications*
- *Chamber A*
- *Molecular Contaminants*

# Molecular Adsorber Coating



- Developed by NASA Goddard Space Flight Center (GSFC)
- Sprayable, zeolite based and highly porous coating technology that was designed to **passively capture** outgassed contaminants

- Available in **white** and **black** coating variations
  - *White Molecular Adsorber Coating, GSFC MAC-W*
  - *Black Molecular Adsorber Coating, GSFC MAC-B*



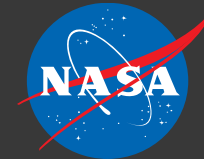
- Examples of molecular contaminants include high molecular weight chemical species, such as:

- *Hydrocarbons*
- *Phthalates*
- *Palmitates*
- *Esters*
- *Silicones*

- Sources of contaminants are products of outgassing from materials found within the spacecraft, such as:

- *Plastics*
- *Adhesives*
- *Lubricants*
- *Epoxies*
- *Potting Compounds*

# Molecular Adsorber Coating



- Designed to be used as a contamination control **mitigation method** to address **material outgassing** concerns on or near sensitive surfaces and instruments:
  - *Inside instrument cavities, electronics boxes, detectors, and baffles*
  - *Near components such as, telescopes, cameras, lasers, mirrors, and optics*
- Reduces the risk of on-orbit molecular contamination from degrading the performance of spaceflight hardware
- Through GSFC's Internal Research and Development (IRAD) program, significant testing and demonstration efforts were performed in relevant environments (i.e. vacuum) for spaceflight applications
  - *Adsorptive Capabilities*
  - *Thermal/Optical Properties*
  - *Adhesion Performance*
  - *Thermal Stability*
  - *Particulate Characteristics*

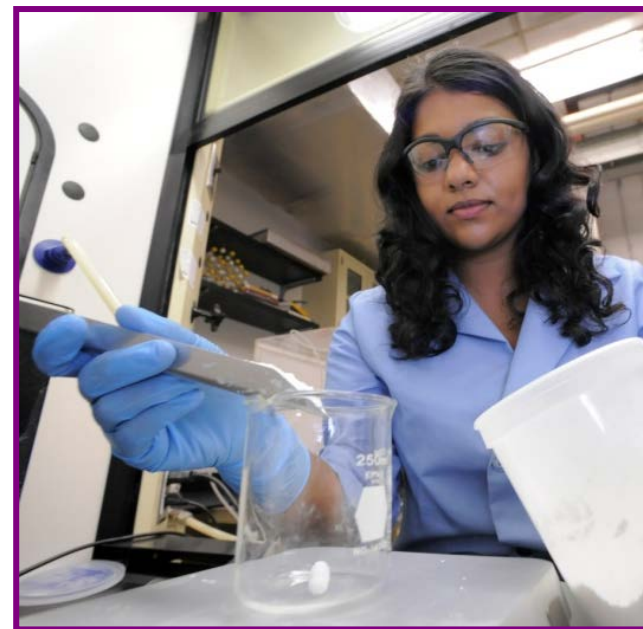
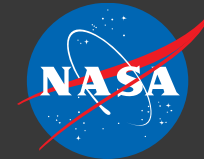


Photo Credit: NASA/Pat Izzo

# Molecular Adsorber Coating



- MAC provides several advantages as an innovative coating technology
  - Serves as a multi-purpose contamination control coating (*thermal, straylight*)

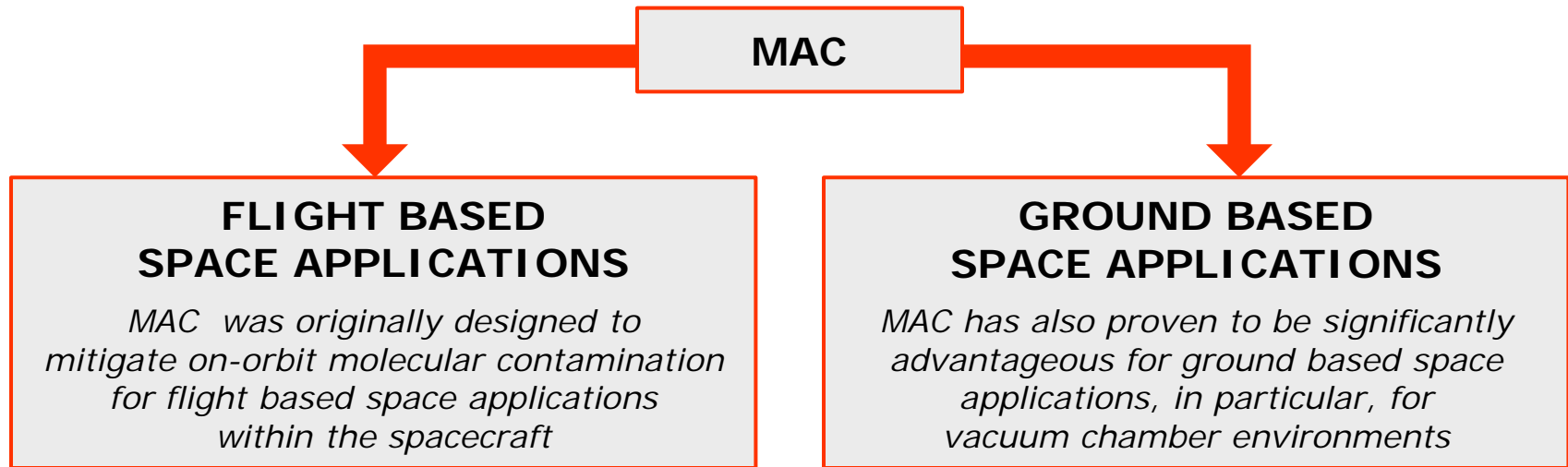
	WHITE THERMAL CONTROL COATING	WHITE MAC	BLACK MAC	BLACK THERMAL CONTROL COATING
Molecular Contamination Control		✓	✓	
Thermal Control Properties	✓	✓	✓	✓
Optical Stray Light Control			✓	✓

MAC can provide  
**THERMAL CONTROL**  
for internal surfaces  
*(White MAC, Black MAC)*

MAC can provide  
**STRAYLIGHT CONTROL**  
for baffles and optical surfaces  
*(Black MAC)*

- Low mass (*adds very little additional mass to the spacecraft*)
- Cost effective (*made from low cost materials*)
- Ease of sprayability onto most substrates (*with optimal adhesion performance*)
- Tailorable adsorption characteristics (*thickness dependent*)
- Low outgassing properties (*made from inorganic materials*)
- Limited particulation effects (*with cleaning mitigation techniques available*)

# Vacuum Applications



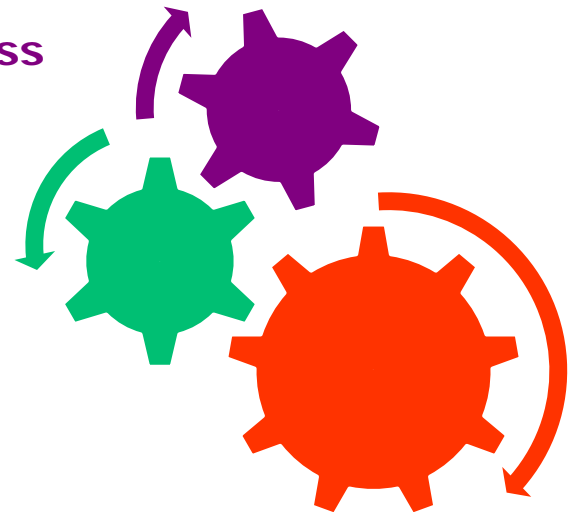
## *Use of MAC in Vacuum Chamber Environments*

- As a **passive getter material** and as an **additional vacuum pump**
  - In industry, the use of getters is a common practice within vacuum systems, such as chambers or hermetically sealed containers
    - *Getters are often used to improve and maintain vacuum efficiency by scavenging molecular contaminants from the evacuated space by absorption, adsorption, or chemical binding*
  - Although there are no mechanical moving elements to the coating, MAC may be described to be analogous to a vacuum pump
    - *A vacuum pump removes molecules from an enclosed volume*



# Vacuum Applications

- Some advantages of using MAC in vacuum chamber environments include:
  - **Reducing outgassing rates**
    - During ground based vacuum testing, such as thermal cycle tests, bake-out runs, and other Thermal Vacuum (TVAC) tests
  - **Limiting the use of cryogenic scavenger panels**
    - Cryogenic scavenger panels are normally used to trap outgassed contaminants that otherwise could condense on critical surfaces
  - **Helpful in reducing the pump down process**
    - When the chamber is being evacuated by the vacuum pump
  - **Achieving high vacuum and lower pressures more efficiently**
    - Than a pump could achieve on its own



# Chamber A

- Originally built in 1965 as part of the Space Environment Simulation Laboratory at NASA Johnson Space Center (JSC) in Houston, Texas
- Best known for space environmental testing of the space capsules and equipment for NASA's Apollo missions with & without the mission crew

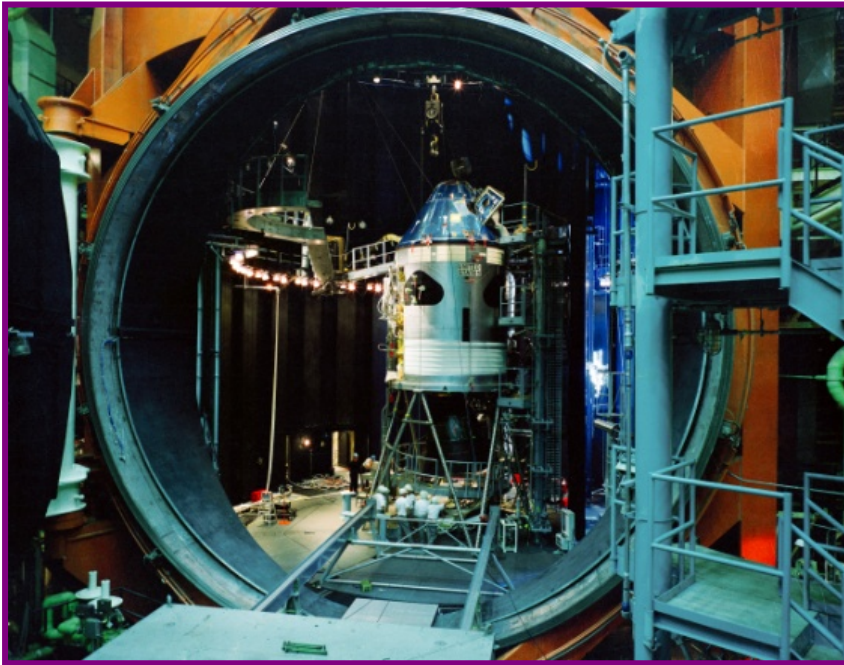


Photo Credit: NASA

*Apollo command and service module 2TV-1 in Chamber A  
for a full mission duration vacuum test in 1968*



Photo Credit: NASA

*Astronaut Buzz Aldrin on the surface  
of the moon during Apollo 11*

# Chamber A

## *Dimensions:*

- 55 ft diameter by 90 ft tall vacuum chamber
- Has an interior volume of 400,000 cubic ft
- Has a hydraulically controlled 40 ft diameter 40 ton door

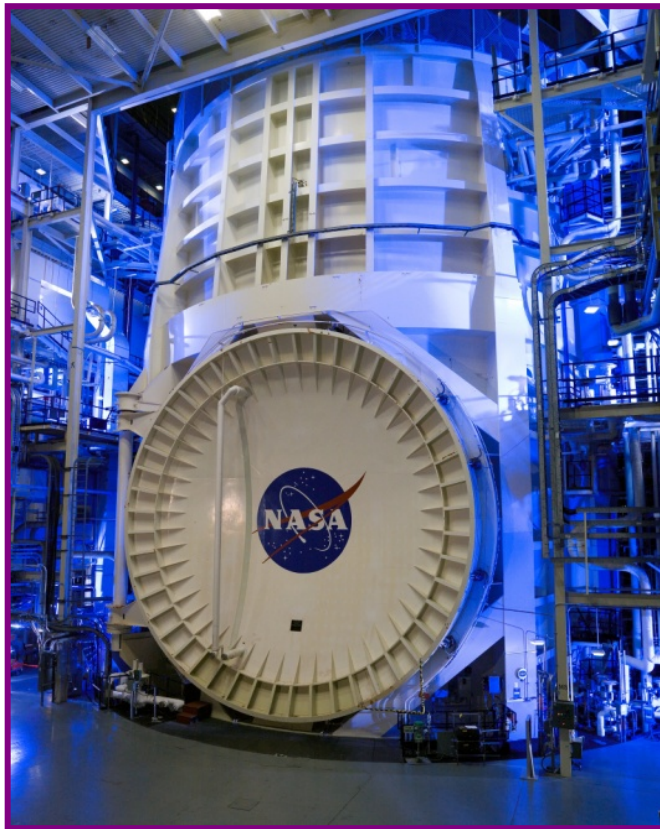


Photo Credit: NASA

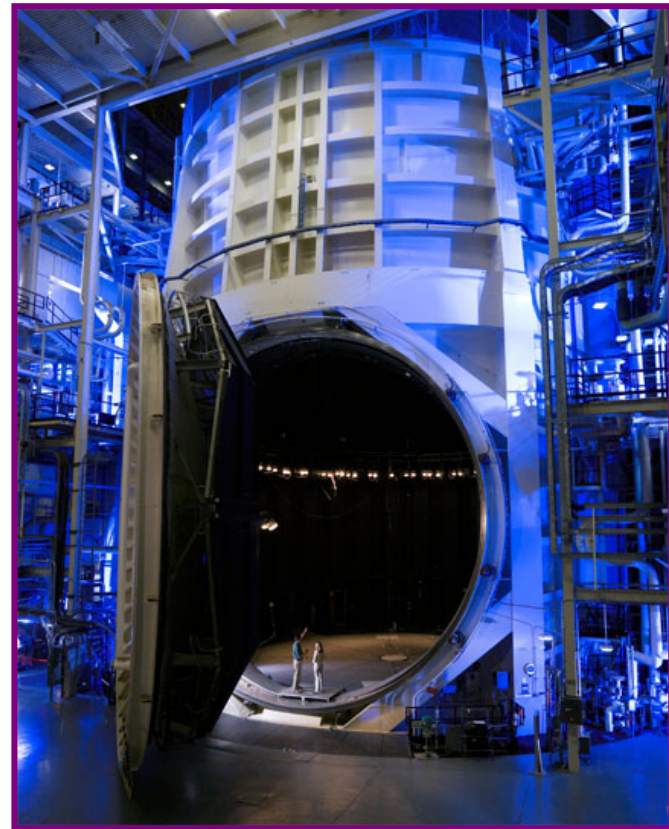


Photo Credit: NASA



# Chamber A

- Over the past several years, Chamber A has experienced significant upgrades to accommodate the arrival and testing of JWST in a space simulation environment

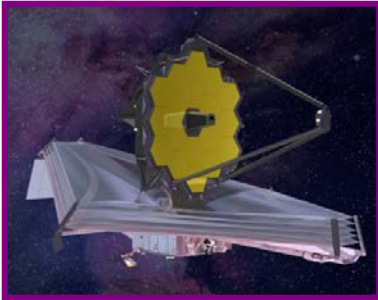


Photo Credit: NASA

## James Webb Space Telescope (JWST)

- Successor to the Hubble Space Telescope
- Considered one of the most powerful infrared space telescopes ever to be built
- Has a 21.3 ft diameter primary mirror
- Has a tennis court sized five layer sunshield
- Will experience a cryogenic environment near Sun–Earth L2 Lagrangian point (about 1 million miles from Earth)

- Chamber A upgrades include:

- **Liquid Helium Shroud**

- Capable of reaching cryogenic temperatures as low as  $-262^{\circ}\text{C}$  to simulate the extremely cold environment that the telescope will be exposed to

- **Clean Room**

- Retrofitted to the test facility

- **Pumping Systems**

- Ultra-clean hydrocarbon-free high vacuum pumping systems



Photo Credit: NASA/Robert Markowitz & Bill Stafford

# Chamber A

- With these impressive upgrades, Chamber A is now categorized as one of the largest high vacuum, cryogenic optical test chambers in the world!
- The Pathfinder model for JWST has been used for practicing ambient and vacuum testing that will be performed on the flight telescope
- Recently, MAC was implemented during the Pathfinder testing of the JWST **Optical Ground Support Equipment (OGSE)**
  - *MAC was used as a mitigation technique to capture molecular contamination within the upgraded Chamber A vacuum test facility*

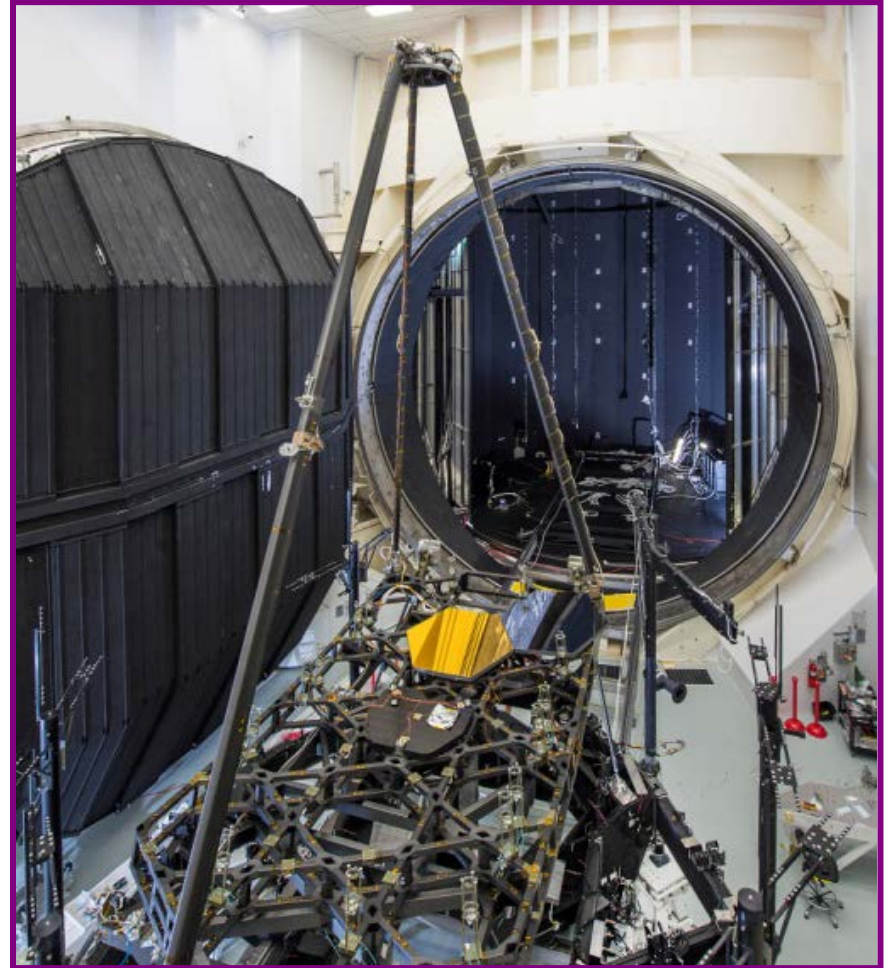
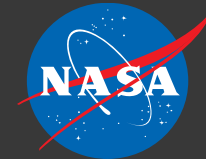


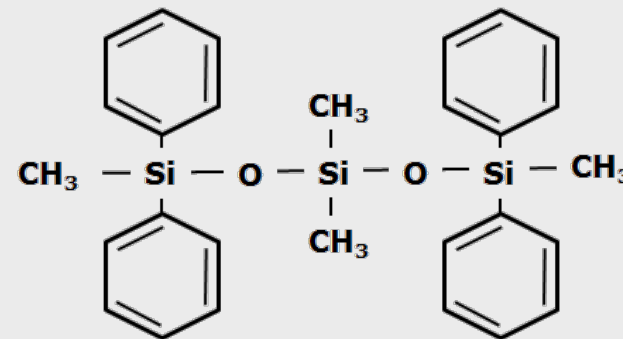
Photo Credit: NASA/Chris Gunn

*Full-scale model of JWST Pathfinder shown entering the recently upgraded Chamber A for cryogenic testing in 2015*

# Molecular Contaminants



- Due to Chamber A's history prior to JWST, molecular contaminants, such as silicones, still remain within the **chamber plenum**
- In particular, one of the main contaminant sources was the residual silicone from **Dow Corning® 704**
  - Also known as DC-704
  - Single component, silicone based diffusion pump oil
  - Commonly used for high vacuum systems
  - Designed to work well with diffusion pumps due to its properties
    - *Low vapor pressure*
    - *Low volatility*
  - Frequently used in Chamber A for Apollo mission testing

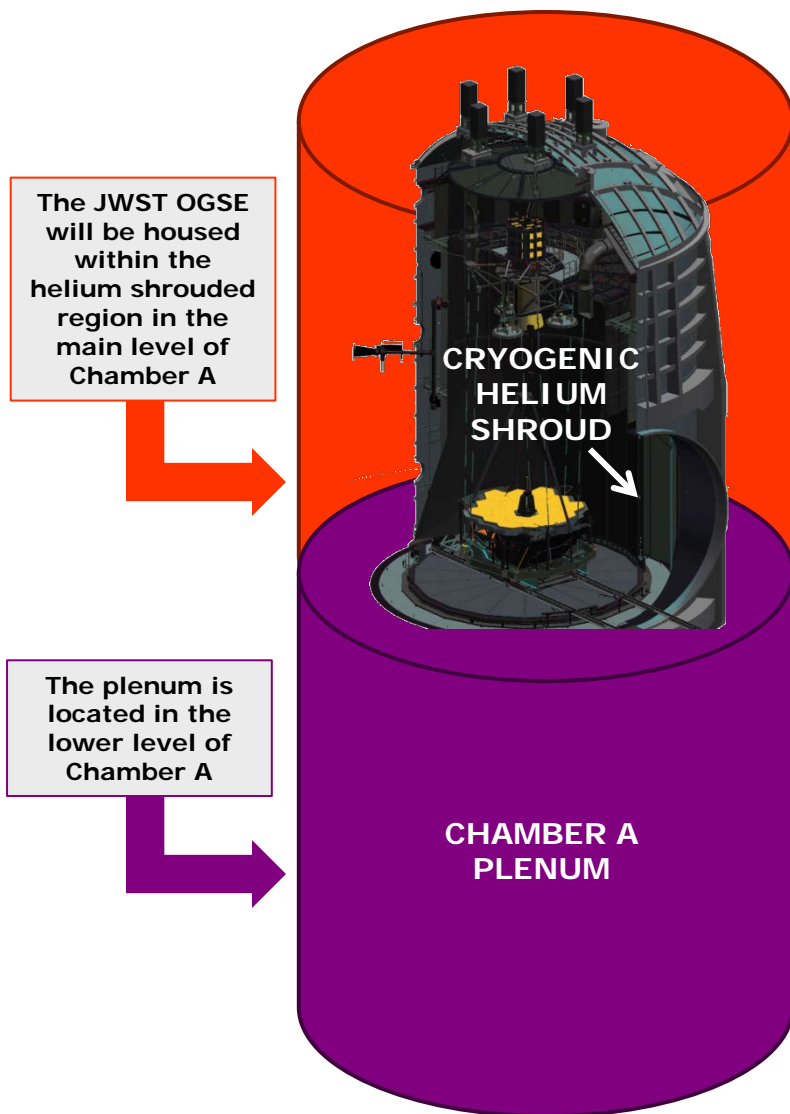


***Molecular Structure of DC-704***

TETRAMETHYL TETRAPHENYL TRISILOXANE

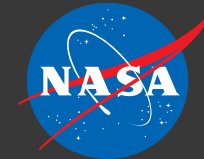


# Molecular Contaminants



- Silicone based contaminants are known to outgas at ambient temperatures, and are extremely difficult to remove
  - *If not properly mitigated, the outgassing effects of DC-704 can accumulate on contamination sensitive surfaces during vacuum testing*
- As a result, many cleaning efforts were performed by the JWST Contamination Control team to remove DC-704 from the plenum
  - *Although these cleaning efforts reduced the silicone levels significantly, there was still some residual DC-704*
- Additionally, MAC was proposed as an innovative contamination mitigation method to be placed in strategic locations during OGSE tests scheduled in 2015
  - *MAC added an extra level of precaution by cost effectively lowering the contamination risk and preventing harmful outgassed species originating from within the chamber environment from migrating and depositing onto JWST's highly sensitive OGSE surfaces*

# Molecular Contaminants



## ***"Proof of Concept" MAC Demonstration***

- Occurred prior to its first large scale application during JWST's Chamber A **Commissioning Test** in Oct 2014
  - Four 1 ft by 1 ft white MAC aluminum panel samples were placed throughout the chamber at various locations to detect sources of contamination
    - *In particular, the migration of DC-704 from the plenum to the main level where the test equipment would be located during Pathfinder testing of the OGSE*
- Chemical analysis of contaminated MAC samples
  - Demonstrated adsorption of various chemical species from within Chamber A, such as:
    - *Hydrocarbons*
    - **DC-704 Silicone Based Diffusion Pump Oil**
    - *Methyl Silicones*
  - Consequently, continued use of MAC for the OGSE tests in 2015 was planned to mainly capture DC-704, among other contaminants



Photo Credit: NASA

**Pristine White MAC**

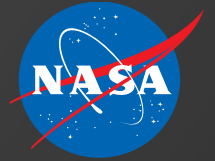
The discoloration of MAC is a visual indication for the collection of chemical species to the pores of the coating



Photo Credit: NASA

**Contaminated White MAC**

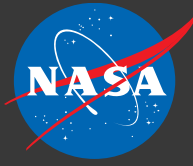




# Application Efforts

- *Fabrication & Installation*
- *MAC Barn Door Panels*
- *MAC Plenum Samples*

# Fabrication & Installation



## *Fabrication Efforts*

- NASA GSFC custom designed 98 MAC samples in various sizes and substrate materials. This included a 6 month fabrication effort:
  - *December 2014 - April 2015 (~4 months)*
  - *August 2015 - September 2015 (~2 months)*
- Fabricated two sets of MAC samples
  - **MAC Barn Door Panels:** *planned for use in the main level*
  - **MAC Plenum Samples:** *planned for use in the plenum*

Both sets will be positioned in two different strategic locations within Chamber A to capture vacuum chamber contamination and prevent them from entering the test environment where the OGSE is housed

## *Installation Efforts*

- Installed prior to the start of the two JWST Pathfinder testing efforts of the OGSE
  - **OGSE-1:** *First OGSE test from May 2015 – June 2015*
  - **OGSE-2:** *Second OGSE test from September 2015 – October 2015*
- MAC samples were exposed to the vacuum chamber environment
  - *During installation activities prior to the test*
  - *Throughout the duration of the two tests*
- MAC samples were removed and shipped back to NASA GSFC after the completion of the OGSE tests for chemical analysis and further testing

On average, the two JWST OGSE tests in Chamber A lasted for about  $34 \pm 3$  days

# MAC Barn Door Panels

## Main Level of Chamber A

### ■ “Barn Door” – Cryogenic Helium Shroud

- During Pathfinder testing of the OGSE, the cryogenic helium shroud reaches temperatures:
  - As cold as  $-241^{\circ}\text{C}$
  - As warm as room temperature
- Internal wall of shroud
  - Painted with a thermal /optical black coating
- External wall of shroud
  - Made of an aluminum finish

### ■ Proposed MAC Location

- Against the base of the external wall on the shroud to cover some of the exposed gaps near the perimeter along the barn door of Chamber A
  - To capture vacuum chamber contaminants that may have migrated from the plenum and prevent them from depositing on the sensitive JWST test equipment housed internal to the cryogenic helium shroud



Cryogenic  
Helium Shroud  
& Barn Door

Photo Credit: NASA/Chris Gunn

Base perimeter along barn  
door on external wall of  
cryogenic helium shroud

# MAC Barn Door Panels

## Substrate Information

Thickness	0.0625 in
Material	6061-T6 Aluminum Alloy
Height	6 in
Width	11 in - 46 in (varies)
Border Edge	0.50 in - 0.75 in (varies)

6 in by 12 in Black MAC and  
White MAC Barn Door Panels



Photo Credit: NASA



Photo Credit: NASA

*Border was implemented on samples to reduce possible coating damage due to handling and installation activities*



Photo Credit: NASA/Chris Gunn

Installation of a white  
MAC barn door panel on  
the external wall of the  
cryogenic helium shroud  
covering the gap along  
the base perimeter of  
Chamber A





# MAC Barn Door Panels

## *Fabrication Efforts*

- Fabricated a total of 65 samples (~ 57 ft<sup>2</sup>)

Molecular Adsorber Coating	Sample Dimensions (in x in)	Sample Border (in)	Sample Quantity	Coating Area (in <sup>2</sup> )	Total Coating Area (in <sup>2</sup> )	Total Coating Thickness (mils)
MAC-W (White)	6 x 11	0.50	15	50	750	8
	6 x 12	0.75	4	47	189	7
	6 x 15	0.50	2	70	140	8
	6 x 20	0.50	1	95	95	7
	6 x 20	0.75	4	83	333	7
	6 x 31	0.50	10	150	1500	9
	6 x 33	0.50	1	160	160	6
	6 x 36	0.50	1	175	175	7
	6 x 46	0.50	19	225	4275	6
MAC-B (Black)	6 x 12	0.75	4	47	189	2 to 3
	6 x 20	0.75	4	83	333	2 to 3
Totals		-	65	-	~ 57 ft <sup>2</sup>	-



# MAC Barn Door Panels

## Installation Efforts

- Installed a total of 53 samples (~ 50 ft<sup>2</sup>) for both tests

OGSE-1 Installation Date: May 2015			OGSE-2 Installation Date: September 2015		
Sample Dimensions (in x in)	Quantity	Total Coating Area (in <sup>2</sup> )	Sample Dimensions (in x in)	Quantity	Total Coating Area (in <sup>2</sup> )
6 x 11	9	450	6 x 12	5	236
6 x 15	1	70	6 x 15	1	70
6 x 20	1	95	6 x 20	7	583
6 x 31	6	900	6 x 31	5	750
6 x 33	1	160	6 x 46	6	1350
6 x 46	11	2475	-	-	-
Totals	29	~ 29 ft <sup>2</sup>	Totals	24	~ 21 ft <sup>2</sup>



# MAC Plenum Samples

## Lower Level of Chamber A

### Plenum of Chamber A

- Encompasses a large volume
- Located beneath the chamber
- Classified as a confined space area
- During Pathfinder testing of the OGSE, plenum contamination (DC-704) may migrate towards the main level cryogenic helium shroud, and ultimately find a path into the chamber where the sensitive equipment will be placed



Photo Credit: NASA/Chris Gunn

### Proposed MAC Location

- MAC samples were placed against the walls of the plenum to capture contamination at its source



As shown, NASA GSFC installed MAC samples on the walls of the plenum in the Chamber A test facility prior to the start of the two OGSE tests

The original plan was to “wallpaper” the plenum with MAC as much as possible. However, due to the massive surface area to cover, it was not feasible to completely “wallpaper” the plenum within the given time frame

# MAC Plenum Samples



- Variety of flexible substrates were explored for the plenum samples

- **Aluminum Foil**

- Used primarily during OGSE-1 tests
- At times, handling was a challenge due to the low tear resistance of the material

- **Kapton**

- Limited use during OGSE-1 tests

- **Aluminum Laminate Materials**

- Used primarily during OGSE-2 tests
- NepTape® 1026
  - Typically used in industry as a second shield in multi-shielded Local Area Network (LAN) coaxial cables
  - Its layered structure allows the material to exhibit a higher tear resistance, and consequently is more flexible and easier to handle than aluminum foil coated MAC samples



Photo Credit: NASA/Chris Gunn

*A thin border less than 0.25 in was implemented on the plenum samples to avoid direct contact with the coating during handling and installation. This border provided a location to adhere Kapton tape to during its placement on the plenum wall.*

## Construction of NepTape® 1026

1 mil Aluminum Foil

0.92 mil Polyester Film

1 mil Aluminum Foil





# MAC Plenum Samples

## *Fabrication Efforts*

- Fabricated a total of 33 samples (~ 60 ft<sup>2</sup>)

Molecular Adsorber Coating	Flexible Substrate Material	Sample Dimensions (in x in)	Sample Quantity	Coating Area (in <sup>2</sup> )	Total Coating Area (in <sup>2</sup> )	Total Coating Thickness (mils)
MAC-W (White)	5 mil Aluminum Foil	19 x 12	12	228	2736	4 to 7
		18.5 x 13.5	10	250	2498	
	Kapton Heaters	10 x 5	2	50	100	
		15 x 5	2	75	150	
	NepTape® 1026	24.5 x 18.5	4	453	1813	
MAC-B (Black)	NepTape® 1026	24.5 x 18.5	3	453	1360	2 to 3
Totals	-	-	33	-	~ 60 ft <sup>2</sup>	-

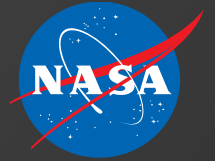


# MAC Plenum Samples

## Installation Efforts

- Installed a total of 33 samples (~ 60 ft<sup>2</sup>) for both tests

OGSE-1 Installation Date: May 2015				OGSE-2 Installation Date: September 2015			
Flexible Substrate Material	Sample Dimensions (in x in)	Quantity	Total Coating Area (in <sup>2</sup> )	Flexible Substrate Material	Sample Dimensions (in x in)	Quantity	Total Coating Area (in <sup>2</sup> )
5 mil Aluminum Foil	19 x 12	12	2736	NepTape® 1026	24.5 x 18.5	7	3173
	18.5 x 13.5	10	2498		-	-	-
Kapton Heaters	10 x 5	2	100		-	-	-
	15 x 5	2	150		-	-	-
Totals	-	26	~ 38 ft <sup>2</sup>	Totals	-	7	~ 22 ft <sup>2</sup>



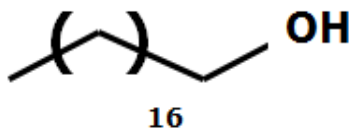
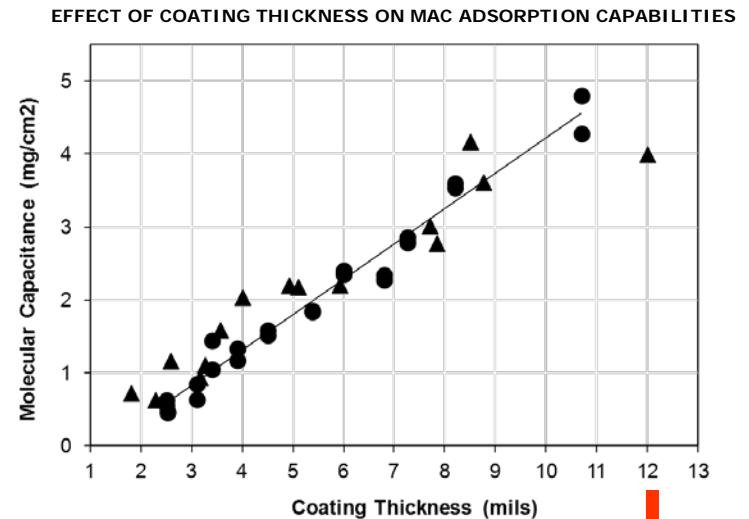
# Testing Efforts

- *Molecular Capacitance Testing*
- *Chemical Analysis*
- *Closing Remarks*

# Molecular Capacitance

## Molecular Adsorption Capacity

- Also referred to as “molecular capacitance”
- Defined as the measure of the coating’s capability to **adsorb or entrap outgassed materials** (i.e. molecular contaminants)
- Dependent on parameters, such as:
  - Coating Thickness
  - Surface Area Coverage
  - Type of Contaminant
  - Duration of Exposure to Contaminant
- Calculated **gravimetrically**, based on mass changes in the coating due to exposure to the contaminant source
- Past test efforts confirmed that the molecular adsorption capacity of MAC is directly proportional to coating thickness
  - *This set of experimental data used **STEARYL ALCOHOL** at 45 °C as a model contaminant source at exposures between 88 and 160 hours*

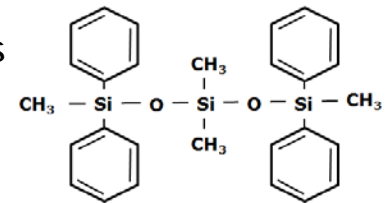


- Also known as 1-octadecanol
- An organic fatty alcohol commonly found in consumer products, such as lubricants, resins, cosmetics, shampoos, perfumes, emulsifiers, and ointments
- Long chain hydrocarbon that is representative of commonly outgassed materials found in spaceflight applications

Stearyl alcohol was selected for these molecular capacitance tests because it is a highly volatile condensable material with a constant vapor pressure for a given temperature, which in turn results in a constant contaminant source rate for baseline comparison purposes

# Molecular Capacitance

- Prior to MAC fabrication, an additional molecular capacitance test was performed to determine if the coating specifically **adsorbs DC-704**
- DC-704 is a more complex high molecular weight chemical species than the previously tested model contaminant (stearyl alcohol)
  - *Has a higher molecular weight; about 1.8 times greater*
  - *Has a lower vapor pressure and lower volatility properties*



Type of Contaminant	Contaminant Source	Chemical Name	Chemical Formula	Molecular Weight	Vapor Pressure at 25 °C
Long Chain Hydrocarbon	Stearyl Alcohol	1-octadecanol	$C_{18}H_{38}O$	270.5 g/mol	$2.7 \times 10^{-6}$ Torr
Silicone Based Compound	DC-704 Diffusion Pump Oil	tetramethyl tetraphenyl trisiloxane	$C_{28}H_{32}O_2Si_3$	484.8 g/mol	$2.0 \times 10^{-8}$ Torr

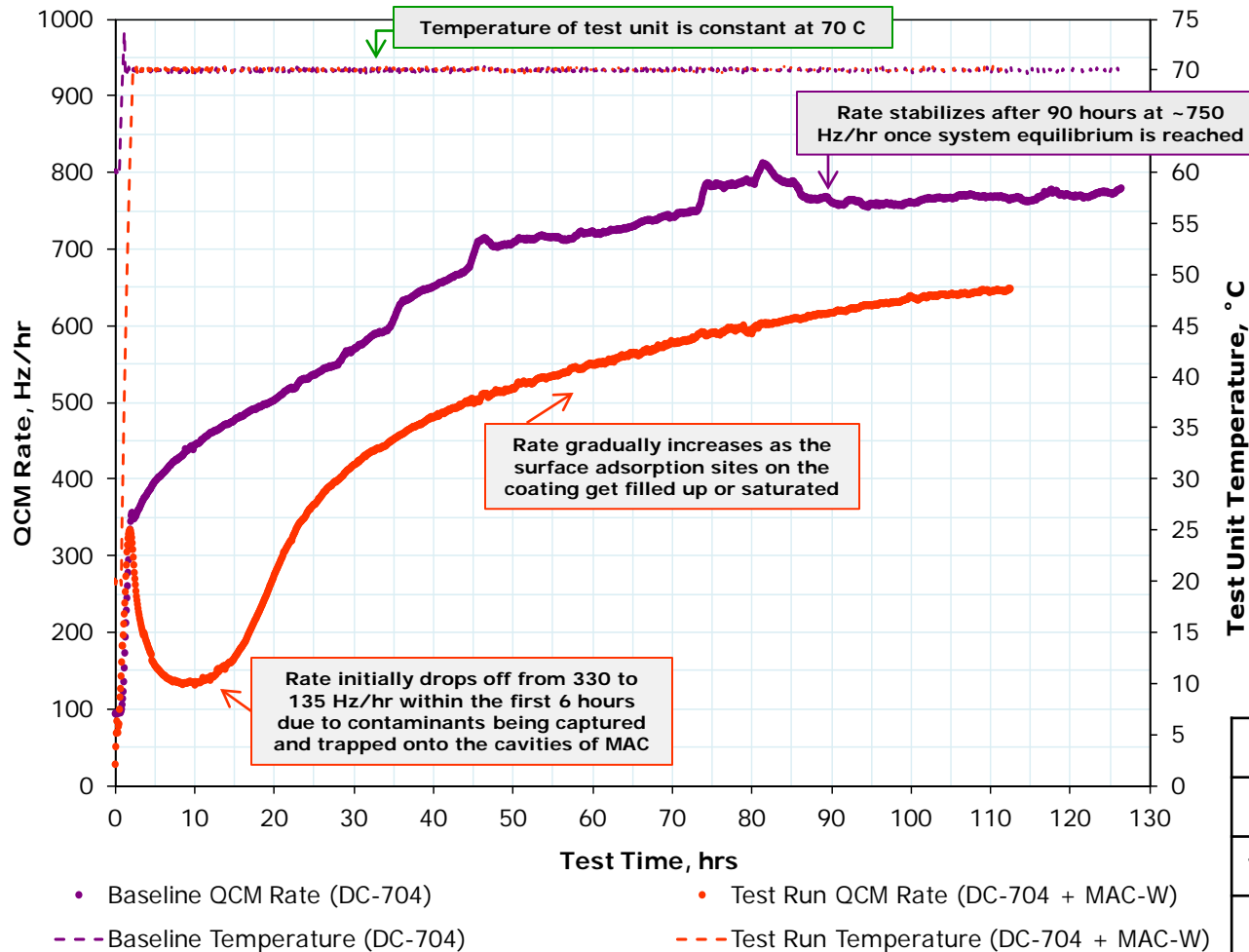
- Due to the complexity of the silicone based single component fluid, test parameters such as temperature, were modified to reduce the test duration
  - *18 white MAC samples with a total coating thickness of 6.1 mils*

Contaminant Source	Source Temperature	Exposure Time	Coating Thickness	Adsorption Capacity
Stearyl Alcohol	45 °C	88 hrs – 160 hrs	6.1 mils	~ 2.3 mg/cm <sup>2</sup>
DC-704 Diffusion Pump Oil	70 °C	115 hrs	6.1 mils	~ 1.2 mg/cm <sup>2</sup> *

\* The molecular adsorption properties reported for DC-704 is not its maximum capacity because the samples were not completely saturated during testing

# Molecular Capacitance

**Quartz Crystal Microbalance (QCM)**  
**Collection Rate Response and Temperature Data**  
*using DC-704 Diffusion Pump Oil as the Contaminant Source*



## "Ideal" Experimental Scenario:

- The test run is expected to approach the baseline when the pore sites on the coating become saturated and can no longer easily adsorb the flux of contaminants. When this state is achieved, the test system behaves as if MAC samples are no longer present during the test run.

## Test Run Status:

- Due to time constraints, the test run was brought to an end at ~115 hours. As a result, it did not approach the baseline. This indicated that the MAC samples have not yet reached saturation. If the test was continued for a longer duration, the MAC samples will continue to adsorb until the majority of the pore sites are filled.

	Baseline Run	Test Run
DC-704	✓	✓
MAC		✓
Temperature	70 °C	70 °C
Duration	~127 hrs	~115 hrs

# Chemical Analysis

## Vacuum Desorption Bake-Out Method

- In industry, a common practice used to regenerate microporous materials, such as zeolite, involve **high temperature vacuum bake-outs** between 175 - 315 °C
- Performed on a single aluminum foil white MAC sample that had been deployed in the Chamber A plenum environment during OGSE-1 (May 2015)
  - Small piece (4 in by 3.5 in) was cut from Sample PM 13
  - Total coating thickness of 7 mils
- Constructed a bake-out box
  - Substrate Type: 6061-T6 aluminum alloy
  - Substrate Thickness: 0.050 in
  - Dimensions of 216 cubic in
- Baseline run without the MAC plenum sample was performed to determine the chamber background contamination that would deposit on the Liquid Nitrogen (LN2) cold plate
- Test configuration was jacketed with aluminum foil during testing to limit the chamber background deposition on the cold plate

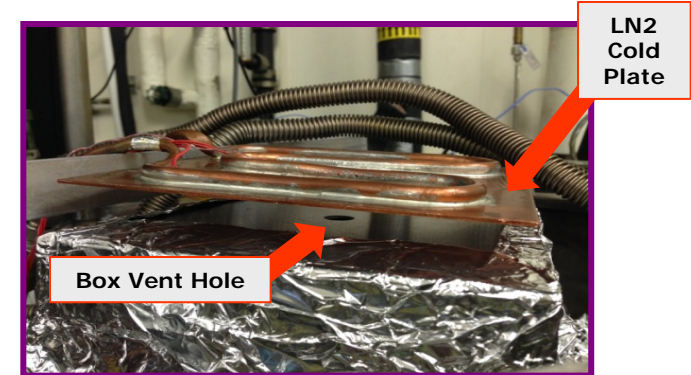
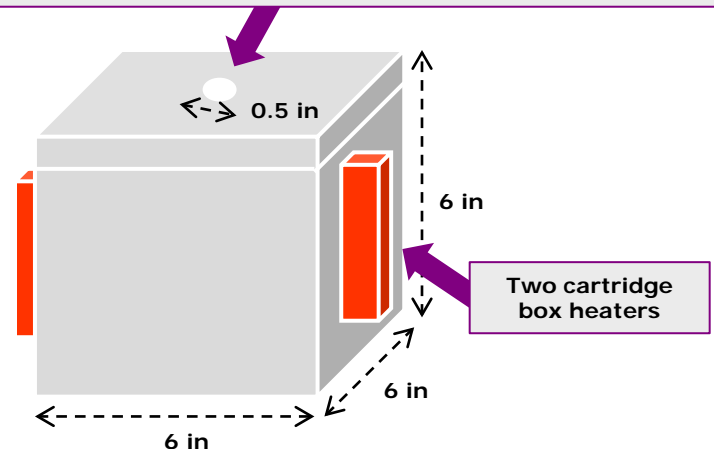


Photo Credit: NASA

Vent hole was made for desorbed contaminants to exit the box and directly deposit on the LN2 cold plate above

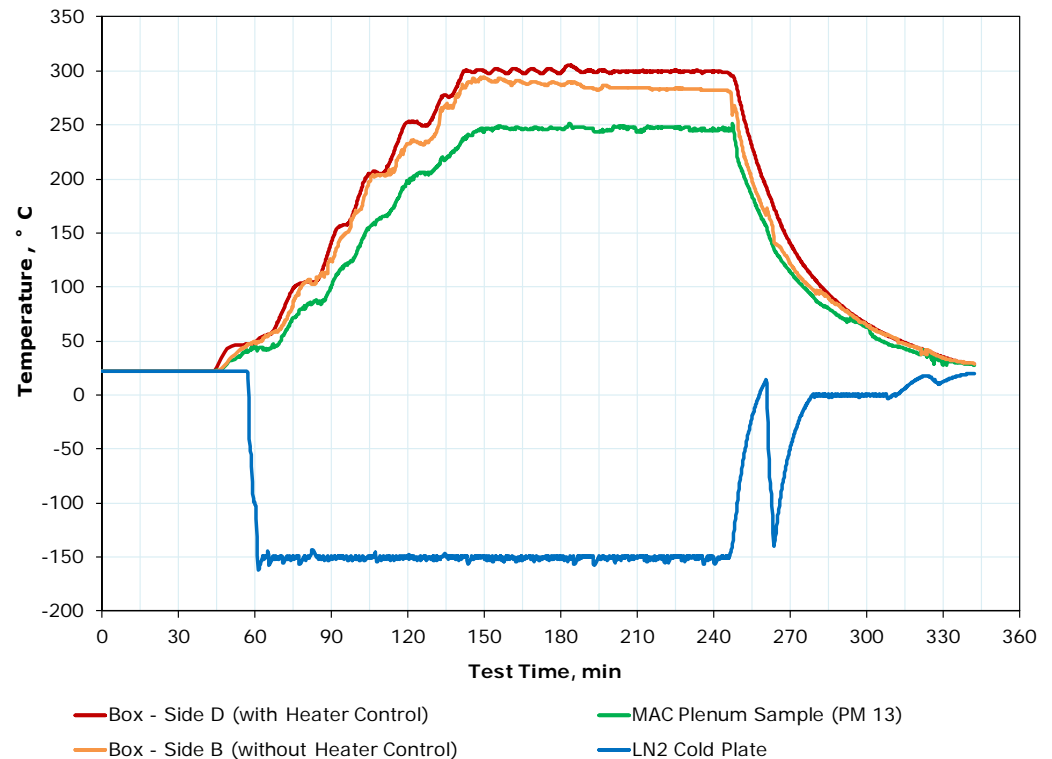


# Chemical Analysis

- Cold plate was immediately rinsed twice after repressing the chamber
  - First rinse was with Isopropyl Alcohol (IPA)
  - Second rinse was with Chloroform
- Rinsates were transferred to separate pre-weighed dishes and allowed to evaporate to dryness
- Remaining Non-Volatile Residue (NVR) was weighed and analyzed using two methods:
  - Fourier Transform Infrared Spectroscopy (FTIR)
  - Pyrolysis-Gas Chromatography /Mass Spectrometry (GC/MS)

## Temperature Profile for Vacuum Desorption Test

Sample PM 13 was tested inside the bake-out box for a duration of 5 hours in high vacuum around  $3.0 \times 10^{-5}$  Torr. The box was heater controlled at 300 °C, and the sample reached 250 °C for ~100 minutes.



Rinse	Solvent	NVR (mg)	GC/MS Analysis
First	Isopropyl Alcohol	$3.58 \pm 0.04$	<ul style="list-style-type: none"> <li>DC-704 diffusion pump oil (80%)</li> <li>Hydrocarbons (20%)</li> </ul>
Second	Chloroform	$19.78 \pm 0.04$	<ul style="list-style-type: none"> <li>Hydrocarbons (97%)</li> <li>Other (3%)</li> </ul>

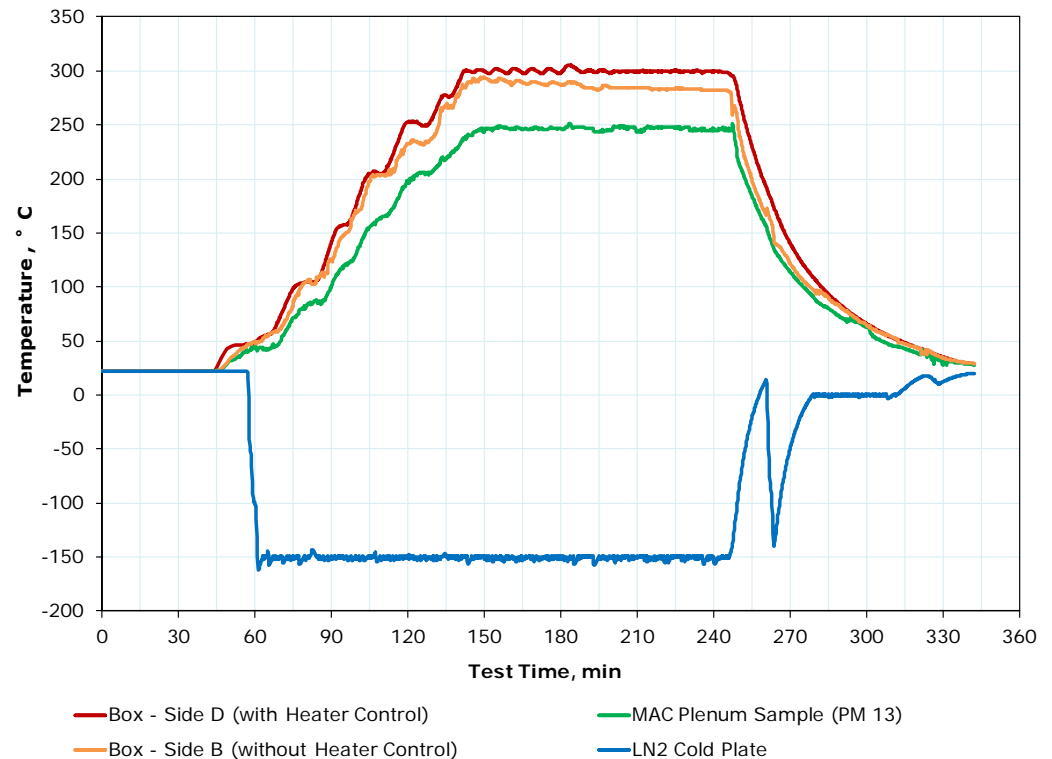


# Chemical Analysis

- Results demonstrate that chloroform removed over 5.5 times the amount removed from the initial rinse with IPA
  - *This suggests that IPA does not sufficiently remove the cold plate contaminants that were collected from the sample bake-out at 250 °C*
- IPA rinse results show DC-704 diffusion pump oil as the most prevalent NVR species
  - *This suggests that most of the DC-704 from the cold plate was removed from the initial rinse*
- Chemical species found during the chamber background and from the solvent itself were subtracted from the results shown below:

## Temperature Profile for Vacuum Desorption Test

Sample PM 13 was tested inside the bake-out box for a duration of 5 hours in high vacuum around  $3.0 \times 10^{-5}$  Torr. The box was heater controlled at 300 °C, and the sample reached 250 °C for ~100 minutes.



Rinse	Solvent	NVR (mg)	GC/MS Analysis
First	Isopropyl Alcohol	$3.58 \pm 0.04$	<ul style="list-style-type: none"> <li>DC-704 diffusion pump oil (80%)</li> <li>Hydrocarbons (20%)</li> </ul>
Second	Chloroform	$19.78 \pm 0.04$	<ul style="list-style-type: none"> <li>Hydrocarbons (97%)</li> <li>Other (3%)</li> </ul>

# Chemical Analysis

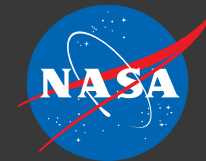
## *Sample Solvent Rinse Method*

- Involves directly rinsing the MAC surface with a solvent
  - *Destructive test that damages the coating surface*
- Qualitatively provides a **general approximation** of the chemical species that were bound to the surface of the coating and can be dissolved using the selected solvent
  - *Does not remove all the contaminants that are entrapped on the porous structure of MAC*
  - *Does not provide a complete representation to quantitatively assess the exact amount and types of contaminants that were collected on MAC*
- Samples from each OGSE test were rinsed with chloroform
  - *4 OGSE-1 and 4 OGSE-2 barn door panels (Total: 8 samples)*
  - *4 OGSE-1 and 4 OGSE-2 plenum samples (Total: 8 samples)*
- Similar to the vacuum desorption method:
  - *Rinsates from the samples were collected and allowed to evaporate to dryness in separate pre-weighed dishes. Remaining NVR was weighed and analyzed using FTIR and pyrolysis-GC/MS*

### **SAMPLE SOLVENT RINSE PROCEDURE:**

*Rinsing the textured coating surface with solvent produced fine particles that were dispersed in the rinsates. Consequently, collected NVR was placed in a micro-vial inside a liner, which was heated in the GC inlet at a high rate of 30 °C/sec to a high temperature of 600 °C for pyrolysis. The volatile and semi-volatile compounds that evolved were then introduced to the GC column interface with the MS for a typical GC/MS run. The non-volatile compounds remain in the micro-vial to avoid inlet contamination.*

# Chemical Analysis



- NVR results of a **single rinse of chloroform** on MAC samples from the JWST OGSE tests in Chamber A

- **Barn Door Panels:** *On average, NVR/area increased by a factor of 2 on the second set of panels*

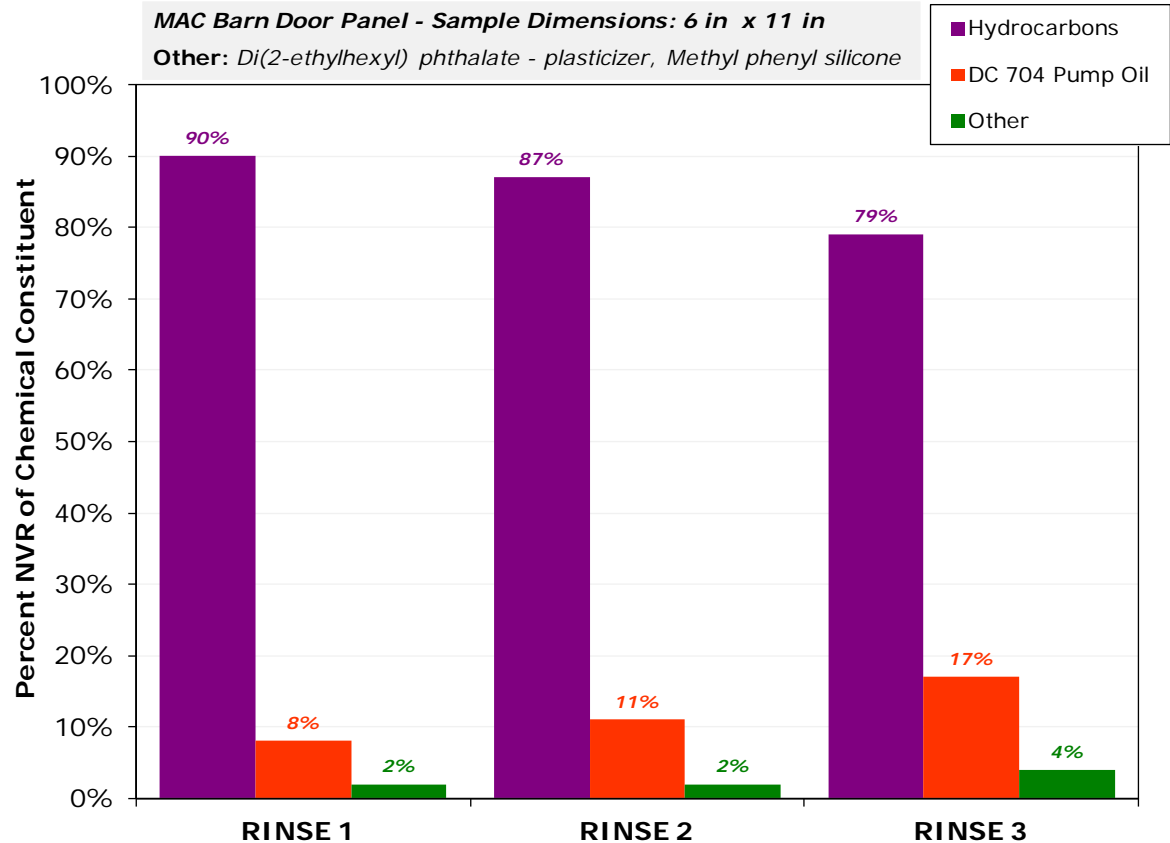
Sample Location/Type	OGSE-1 Installation Date: May 2015 Sample Size: 6 in by 11 in		OGSE-2 Installation Date: September 2015 Sample Size: 6 in by 12 in	
	Sample ID	NVR/Area ( $\mu\text{g}/\text{cm}^2$ )	Sample ID	NVR/Area ( $\mu\text{g}/\text{cm}^2$ )
MAC Barn Door Panels	BD 42	$13.48 \pm 0.09$	BB 20	$31.11 \pm 0.09$
	BD 41	$15.62 \pm 0.09$	BW 03	$13.56 \pm 0.09$
	BD 43	$13.74 \pm 0.09$	BW 12	$24.35 \pm 0.09$
	BD 19	$6.46 \pm 0.09$	BB 06	$26.18 \pm 0.09$
Average	$\sim 12 \mu\text{g}/\text{cm}^2$		$\sim 24 \mu\text{g}/\text{cm}^2$	

- **Plenum Samples:** *On average, NVR/area slightly decreased by about 12 % for the second set of samples*

Sample Location/Type	OGSE-1 Installation Date: May 2015		OGSE-2 Installation Date: September 2015	
	Sample ID	NVR/Area ( $\mu\text{g}/\text{cm}^2$ )	Sample ID	NVR/Area ( $\mu\text{g}/\text{cm}^2$ )
MAC Plenum Samples	PM 04	$47.93 \pm 0.09$	PW 03	$21.70 \pm 0.09$
	PM 05	$84.73 \pm 0.09$	PW 05	$58.90 \pm 0.09$
	PM 12	$20.66 \pm 0.09$	PB 01	$47.16 \pm 0.09$
	PM 22	$45.57 \pm 0.09$	PB 04	$49.82 \pm 0.09$
Average	$\sim 50 \mu\text{g}/\text{cm}^2$		$\sim 44 \mu\text{g}/\text{cm}^2$	

# Chemical Analysis

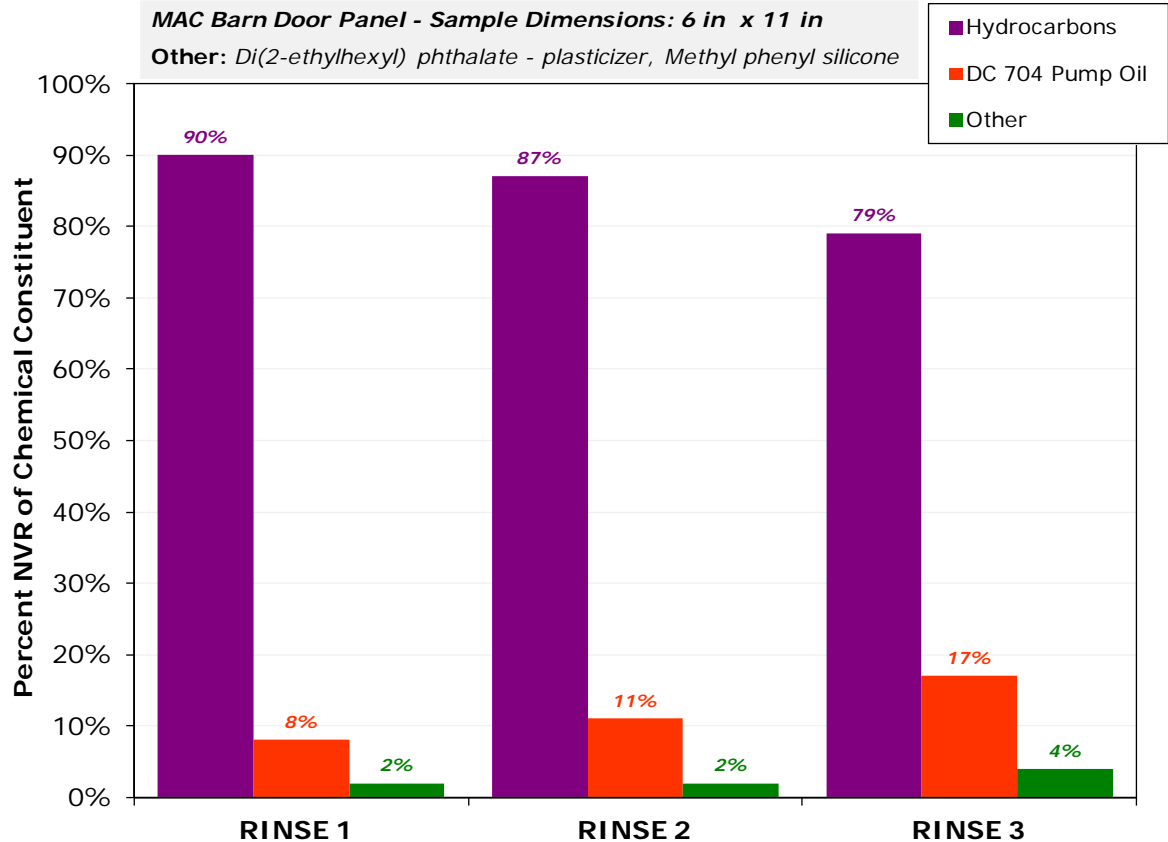
- One MAC barn door panel (Sample BD 42) from OGSE-1 was **rinsed three times** to determine how much additional NVR will be removed with multiple rinses of chloroform
- Results confirm that using this method does not fully remove the chemically adsorbed contaminants
  - *Two consecutive rinses remove the same amount of contaminant each time*
  - *A third rinse showed a **64 % reduction** in NVR than the first two rinses*



Solvent Rinse	NVR (mg)	NVR/Area ( $\mu\text{g}/\text{cm}^2$ )
1	$5.74 \pm 0.04$	$13.48 \pm 0.09$
2	$5.55 \pm 0.04$	$13.03 \pm 0.09$
3	$2.05 \pm 0.04$	$4.81 \pm 0.09$

# Chemical Analysis

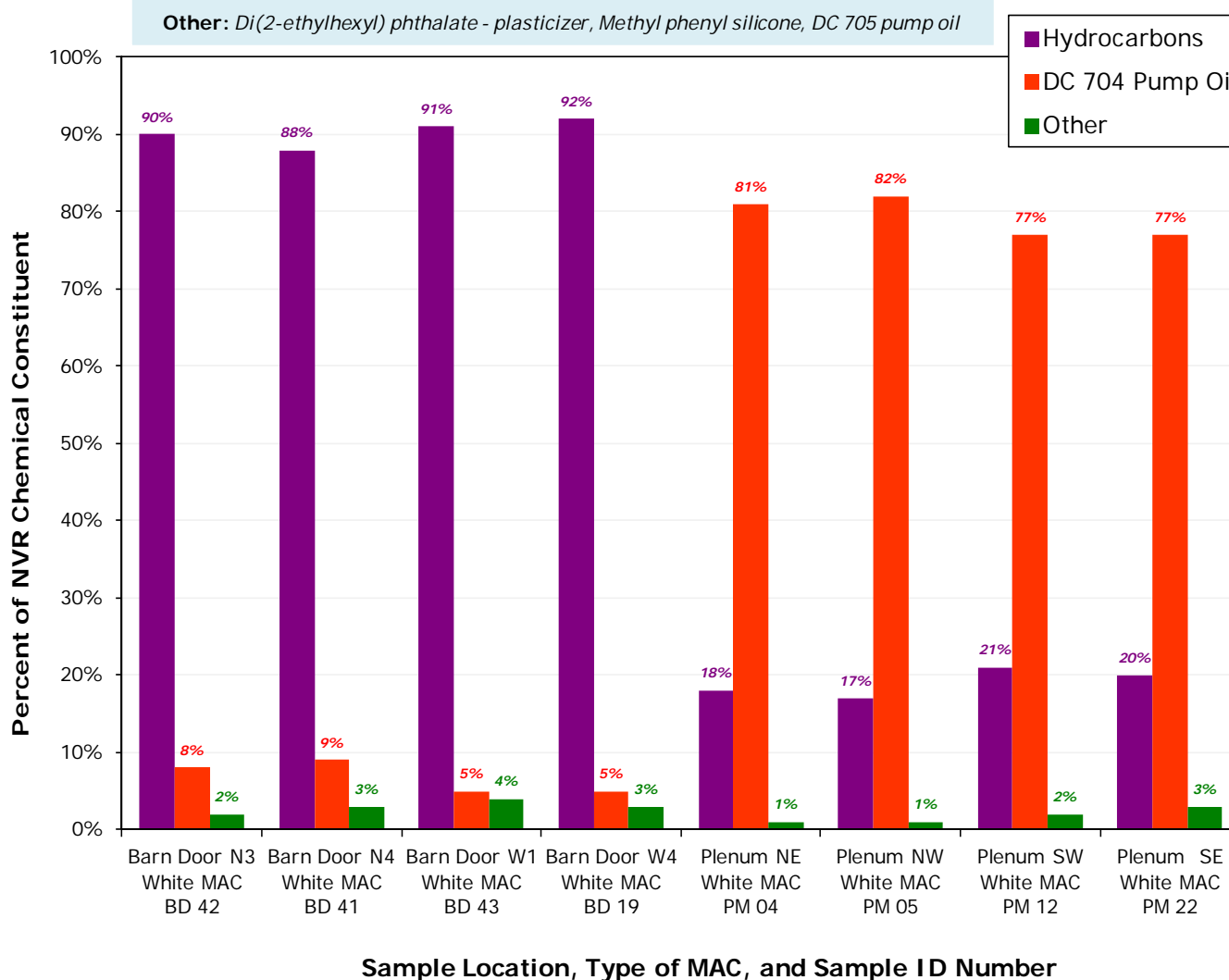
- Regardless, the significance of this method is to **qualitatively** determine the types and relative amounts of chemical species that were detected from a solvent rinse of the coating surface
  - *Each consecutive rinse displays the same chemical species:*
    - Hydrocarbons (most prominent)
    - DC-704
  - *With each repeated rinse, there is a:*
    - Gradual reduction of hydro carbons from 90 % to 79 %
    - Gradual increase of DC-704 present from 8 % to 17 %



Solvent Rinse	NVR (mg)	NVR/Area ( $\mu\text{g}/\text{cm}^2$ )
1	5.74 $\pm$ 0.04	13.48 $\pm$ 0.09
2	5.55 $\pm$ 0.04	13.03 $\pm$ 0.09
3	2.05 $\pm$ 0.04	4.81 $\pm$ 0.09

# Chemical Analysis

## Single Solvent Rinse Analysis Results from the OGSE-1 Test in May 2015



### MAC BARN DOOR PANELS

**High levels of hydrocarbons between 88 - 92 %**

**Low levels of DC-704 between 5 - 9 %**

*As expected, results show migration of DC-704 to the main level of Chamber A near the helium shroud and barn door*

### MAC PLENUM SAMPLES

**High levels of DC-704 between 77 - 82 %**

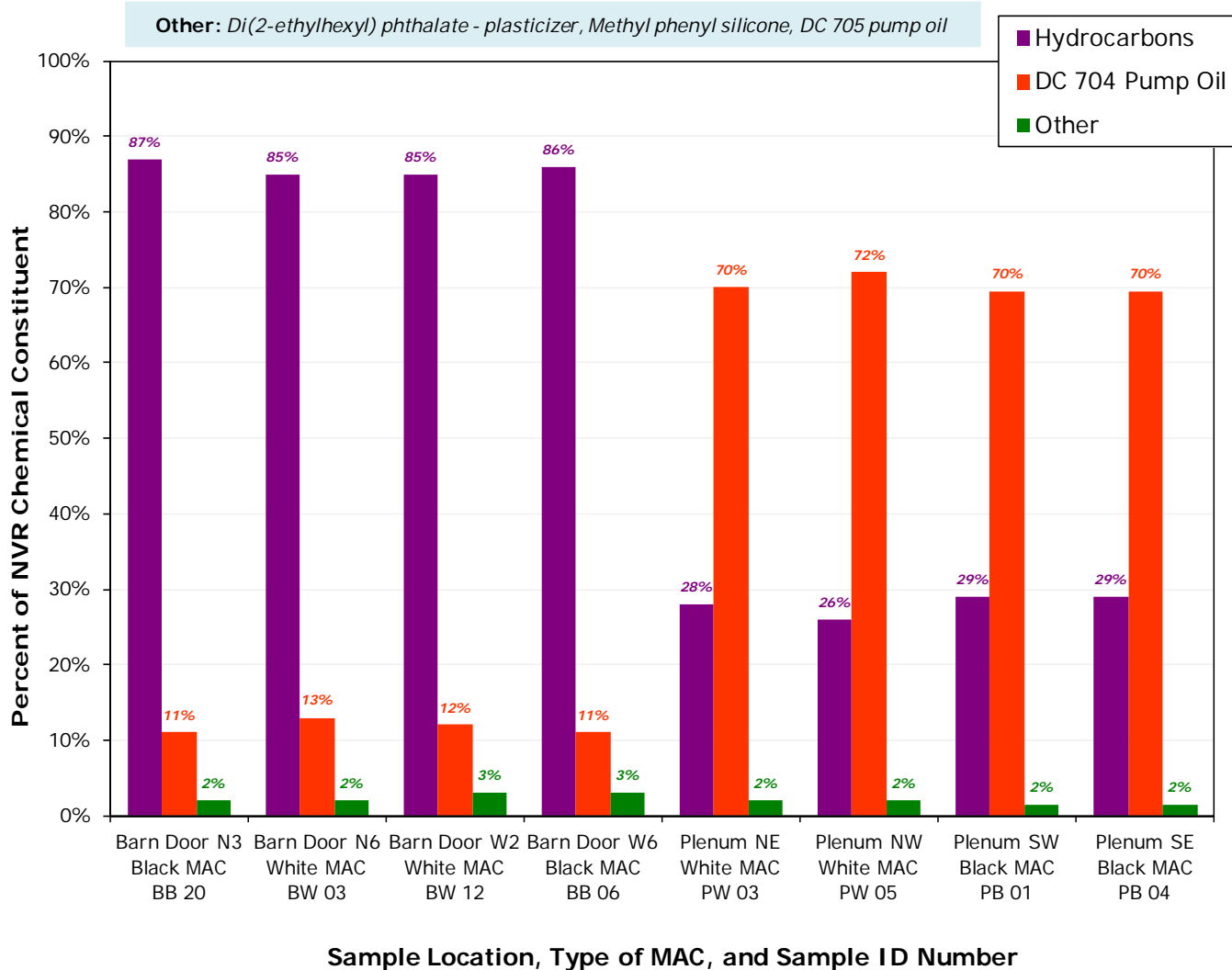
**Low levels of hydrocarbons between 17 - 20 %**

*This is predictable considering that the plenum is the source of the silicone contamination*

# Chemical Analysis

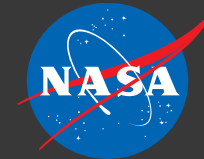


## Single Solvent Rinse Analysis Results from the OGSE-2 Test in September 2015

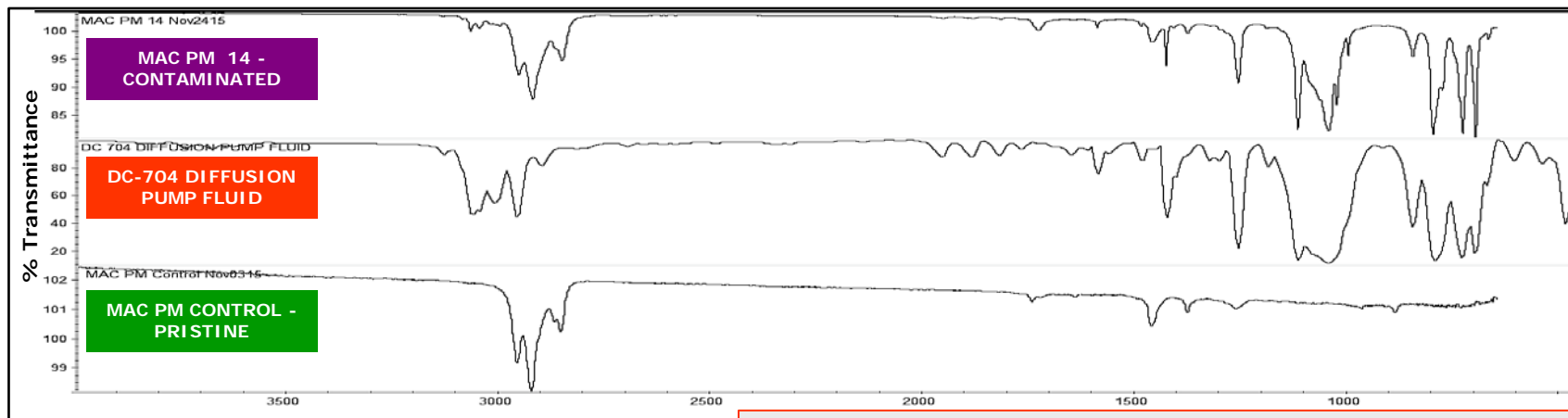


- Same trends were observed for the MAC samples from the OGSE-2 test
- Main contaminant adsorbed by the barn door panels were hydrocarbons
- Main contaminant adsorbed by the plenum samples were DC-704
- Reduction of 11-12 % of DC-704 adsorbed for OGSE-2 plenum samples compared to OGSE-1 plenum samples
- Relative amounts of hydrocarbons adsorbed on barn door panels were slightly lower, ranging between 85 - 87% for OGSE-2 test

# Chemical Analysis

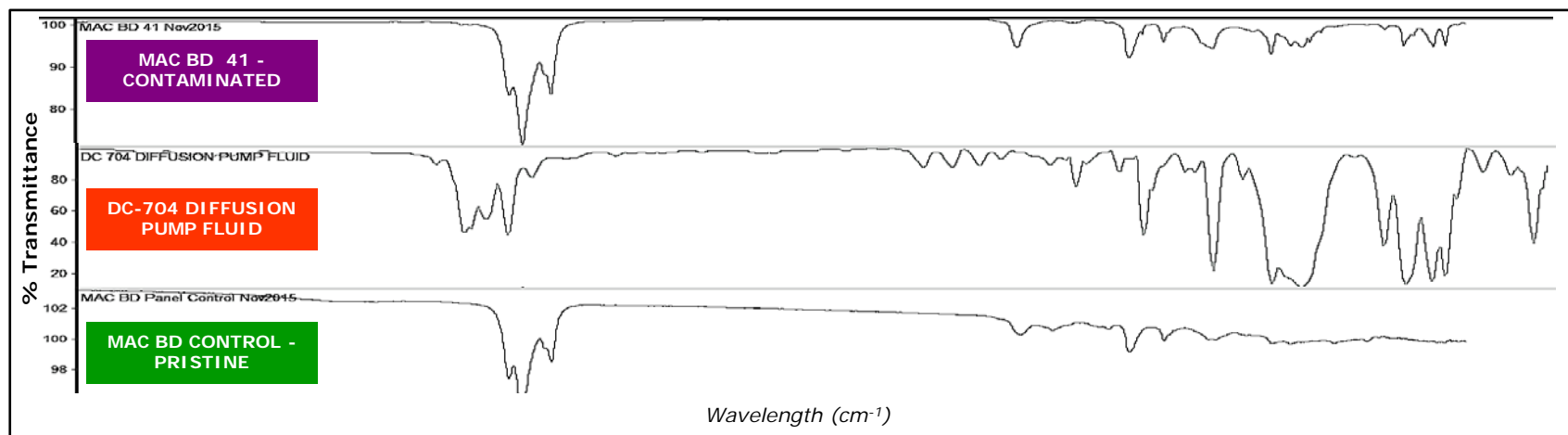


## FTIR Spectra Comparison for MAC Plenum Sample 14



The spectrum illustrates that the contaminated plenum sample is a better match to DC-704 when compared to the contaminated barn door panel spectrum, particularly in the 500 to 2000 cm<sup>-1</sup> wavelength range

## FTIR Spectra Comparison for MAC Barn Door Panel 41







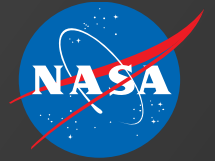
# Closing Remarks

## *Mitigator and Indicator*

- MAC can serve not only as a **contaminant risk mitigator**, but also as a **contaminant indicator** by identifying the molecular contamination risks in the chamber that may not be collected on post vacuum witness foils
  - *For instance, molecular species that strike the coating surface are captured and less likely to be released during warm-up to ambient conditions*
- Results from laboratory testing and chemical analysis methods have proven that MAC will continue to:
  - *Collect outgassed silicone based diffusion pump oil*
  - *Reduce the risk of molecular contamination from the chamber to test equipment*

## *Coating Particulation*

- Particulation related anomalies from MAC were not observed during post test chamber inspections



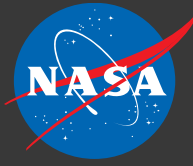
# Conclusions

- *Future Plans*
- *Acknowledgements*
- *References*
- *Contact Information*



# Future Plans

- Continued use of MAC is planned for upcoming tests due to the successful application during the JWST OGSE tests in Chamber A
  - Currently planned for use for the **JWST Thermal Pathfinder Test**
    - *Tentatively scheduled for Fall 2016*
- Future work include:
  - Fine tuning the chemical analysis methods for determining the amount of contaminants adsorbed onto the coating
  - For example:
    - *Exploring different solvents for rinsing*
    - *Improving vacuum desorption tests for greater test efficiency*
  - Investigating other tear resistant substrates
  - Performing more analysis on future samples
  - Continuing to expand upon the benefits of using MAC for vacuum chamber applications



# Acknowledgements

- A summary of the authors of this SPIE paper, their affiliations, and support roles for this JWST/MAC effort are described below:

## Nithin S. Abraham

- *NASA Goddard Space Flight Center, Code 546*
  - MAC Research and Development Lead Coatings Engineer

## Mark M. Hasegawa

- *NASA Goddard Space Flight Center, Code 546*
  - Thermal Coatings Application and Development Group Lead

## Eve M. Wooldridge

- *NASA Goddard Space Flight Center, Code 546*
  - JWST Project Support Contamination Engineer Lead

## Kelly A. Henderson-Nelson

- *Stinger Ghaffarian Technologies, Inc., Code 546*
  - JWST Project Support Contamination Engineer



# Acknowledgements

- The authors of this paper would like to thank the **JWST Program Office at NASA GSFC** for funding this coatings application and testing effort. The authors would also like to acknowledge some talented individuals who have contributed to this effort.

## *A special thanks to:*

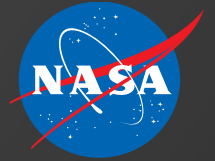
NAME	AFFILIATION	NASA GSFC CODE	DESCRIPTION
Joseph Ward	SGT, Inc.	Code 546	<i>Chamber A Contamination Support</i>
Niko Stergiou	SGT, Inc.	Code 546	<i>Chamber A Contamination Support</i>
Jason Durner	SGT, Inc.	Code 546	<i>Chamber A Contamination Support</i>
John Petro	NASA GSFC	Code 546	<i>Coatings Application Support</i>
Alfred Wong	SGT, Inc.	Code 546	<i>Coatings Application Support</i>
Grace Miller	SGT, Inc.	Code 546	<i>Coatings Application Support</i>
Kenneth O'Connor	SGT, Inc.	Code 546	<i>Sample Fabrication Coatings Testing &amp; Measurement Support</i>
Alexson Harris-Kirksey	Telophase	Code 504	<i>Coatings Testing &amp; Measurement Support</i>
Griffin Jayne	SGT, Inc.	Code 546	<i>Coatings Testing &amp; Measurement Support</i>
George Meadows	SGT, Inc.	Code 546	<i>Coatings Testing &amp; Measurement Support</i>
Doris Jallice	NASA GSFC	Code 541	<i>Chemical Analysis Support</i>
Paul Pless	SGT, Inc.	Code 541	<i>Chemical Analysis Support</i>
Jeremy Knipple	SGT, Inc.	Code 541	<i>Chemical Analysis Support</i>



# References

1. Abraham, N. S., Hasegawa, M. M., and Straka, S. A., "Development and Testing of Molecular Adsorber Coatings", Proc. SPIE 8492, Optical System Contamination: Effects, Measurements, and Control 2012, 849203 (October 2012)
2. Abraham, N. S., Hasegawa, M. M., and Straka, S. A., "Black Molecular Adsorber Coatings for Spaceflight Applications", Proc. SPIE 9196, Systems Contamination: Prediction, Measurement, and Control 2014, 91960F (September 2014)
3. Abraham, N.S. "NASA Applications of Molecular Adsorber Coatings", Contamination, Coatings, Materials Science, and Planetary Protection Workshop (July 2015)
4. Abraham, N.S. "NASA Applications of Molecular Adsorber Coatings", Thermal and Fluids Analysis Workshop (August 2015)
5. Jousten, K. Handbook of Vacuum Technology. Weinheim: Wiley-Blackwell, (2008)
6. Hablanian, M. H. High-Vacuum Technology: A Practical Guide, 2nd ed. Revised and Expanded, New York: M. Dekker, (1997)
7. O'Hanlon, J. F., A User's Guide to Vacuum Technology, New York: Wiley, (1980)
8. National Aeronautics and Space Administration, "NASA Readies Famous "Chamber A" to Welcome the James Webb Space Telescope", 1 June 2016. < [http://www.nasa.gov/mission\\_pages/webb/news/chamber-a.html](http://www.nasa.gov/mission_pages/webb/news/chamber-a.html)>
9. Pearlman, R., "NASA Upgrades Historic Giant Vacuum Chamber for Space Telescope", 1 June 2016. <http://www.space.com/20535-nasa-vacuum-chamber-space-telescope.html>
10. National Aeronautics and Space Administration, "Chamber A", 1 June 2016. [http://www.nasa.gov/centers/johnson/engineering/integrated\\_environments/altitude\\_environmental/chamber\\_A/](http://www.nasa.gov/centers/johnson/engineering/integrated_environments/altitude_environmental/chamber_A/)
11. National Aeronautics and Space Administration, "James Webb Space Telescope", 1 June 2016. <<http://www.jwst.nasa.gov/>>
12. Dow Corning Corp., "Product Information: Information about Dow Corning® 702 Diffusion Pump Fluid, Dow Corning® 702 Diffusion Pump Fluid, Dow Corning® 702 Diffusion Pump Fluid", Form No. 10-838-98 (1998)
13. Neptco Inc., "NepTape® 1026 Data Sheet: Shielding Tape - Standard Foil/Film/Foil Laminates NepTape® 1026" (August 2000)
14. "Final Report on the Safety Assessment of Stearyl Alcohol, Oleyl Alcohol, and Octyl Dodecanol", Journal of the American College of Toxicology, Volume 4 - Number 5, Mary Ann Liebert, Inc., Publishers, (1985)
15. National Institutes of Health, Pub Chem: Open Chemistry Database, "1-octadecanol" , 1 June 2016. < <https://pubchem.ncbi.nlm.nih.gov/compound/1-octadecanol#section=Top>>
16. National Institutes of Health, Pub Chem: Open Chemistry Database, "Tetraphenyl-1,3,3,5-tetramethyltrisiloxane" , 1 June 2016. < <https://pubchem.ncbi.nlm.nih.gov/compound/19882#section=Top>>
17. Sigma-Aldrich. "Molecular Sieves - Technical Information Bulletin." 1 June 2016. <http://www.sigmaaldrich.com/chemistry/chemical-synthesis/learning-center/technical-bulletins/al-1430/molecular-sieves.html>
18. Jallice, D., Materials Engineering Branch Chemical Analysis Report: MEB13659, "Chemical Analysis of Sample Rinses from Baseline Empty Box Bake and Bake of MAC Foil from JWST JSC OGSE-1 Plenum" (February 2016)
19. Jallice, D., Materials Engineering Branch Chemical Analysis Report: MEB13258 Rev A, "MAC JWST OGSE-1 Coating Rinse Analysis" (December 2015)
20. Jallice, D., Materials Engineering Branch Chemical Analysis Report: MEB13526, "MAC JWST OGSE-2 Coating Rinse Analysis" (January 2016)





# Contact Information

***Nithin S. Abraham***

**Thermal Coatings Engineer**

NASA Goddard Space Flight Center, Code 546

Contamination and Coatings Engineering Branch

E-mail: [nithin.s.abraham@nasa.gov](mailto:nithin.s.abraham@nasa.gov)

Phone: (301) 614-7070