### Full-scale Mars Science Laboratory Tiled Heatshield Material Response

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### **Overview** – Geometry from literature





### **Overview** – Aerothermal environment



### **Overview** – Material response











### Aerothermal environment computed from DPLR\*

#### **DPLR** assumptions

- Laminar boundary layer
- Non-blowing & smooth wall
- Chemical and thermal non-equilibrium
- Radiative equilibrium
- Super-catalytic wall
- Mars atmosphere:  $y_{CO2} = 0.97$ ,  $y_{N2} = 0.03$
- 12 reactions & 8 species [12]



BLAYER calculates the boundary layer edges using a curvature-based method

Surface pressure  $p_w$ , heat transfer coefficient  $C_H$  and enthalpy  $h_e$  at the boundary layer edges are used as inputs in the material response code: PATO

\* DPLR = Data Parallel Line Relaxation [6]

### Computational domain of the material response

0.125 m



Mass and momentum conservation

\* PATO = Porous material Analysis Toolbox based on OpenFOAM [3] Open Source Release: <u>http://pato.ac</u>

Mass and momentum conservation

$$\boldsymbol{v}_{g} = -\frac{1}{\epsilon_{g}} \left( \frac{1}{\mu} \left( \overrightarrow{\overline{R}} + \frac{1}{p} \, \overline{\overline{\beta}} \right) \cdot \boldsymbol{\partial}_{x} p_{g} \right)$$
  
H  
Permeability [2,11]

$$\partial_t \epsilon_g \rho_g - \partial_x \cdot (\epsilon_g \rho_g \boldsymbol{v}_g) = \Pi$$

\* PATO = Porous material Analysis Toolbox based on OpenFOAM [3] Open Source Release: http://pato.ac

Mass and momentum conservation

$$\boldsymbol{v}_{g} = -\frac{1}{\epsilon_{g}} \left( \frac{1}{\mu} \, \overline{\overline{K}} + \frac{1}{p} \, \overline{\overline{\beta}} \right) \cdot \, \boldsymbol{\partial}_{x} p_{g}$$
Klinkenberg correction [2,11]

$$\partial_t \epsilon_g \rho_g - \partial_x \cdot (\epsilon_g \rho_g \boldsymbol{v}_g) = \Pi$$

\* PATO = Porous material Analysis Toolbox based on OpenFOAM [3] Open Source Release: <u>http://pato.ac</u>

Mass and momentum conservation

$$\boldsymbol{\nu}_g = -\frac{1}{\epsilon_g} \Big( \frac{1}{\mu} \, \overline{\boldsymbol{K}} + \frac{1}{p} \, \overline{\boldsymbol{\beta}} \Big) \cdot \, \boldsymbol{\partial}_{\boldsymbol{X}} p_g \, .$$

$$\partial_t \epsilon_g \rho_g - \partial_x \cdot (\epsilon_g \rho_g v_g) = \prod_{i=1}^{n}$$

Total pyrolysis-gas production rate [3]

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Mass and momentum conservation

$$\boldsymbol{v}_g = -\frac{1}{\epsilon_g} \left( \frac{1}{\mu} \, \overline{\boldsymbol{K}} + \frac{1}{p} \, \overline{\boldsymbol{\beta}} \right) \cdot \, \boldsymbol{\partial}_{\boldsymbol{x}} p_g$$

 $\partial_t \epsilon_g \rho_g - \partial_x \cdot (\epsilon_g \rho_g \boldsymbol{v}_g) = \Pi$ 

#### **Pyrolysis**

$$\partial_t \chi_{i,j} = (1 - \chi_{i,j})^{m_{i,j}} T^{n_{i,j}} A_{i,j} \exp\left(\frac{-E_{i,j}}{RT}\right)$$



$$\Pi = \sum_{i=1}^{N_p} \sum_{j=1}^{P_i} \sum_{k=1}^{N_g} \zeta_{i,j,k} \,\epsilon_{i,0} \,\rho_{i,0} \,F_{i,j} \,\partial_t \chi_{i,j}$$



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#### **Energy conservation**

$$\sum_{i=1}^{N_p} \left[ \left( \epsilon_i \rho_i c_{p,i} \right) \partial_t T \right] - \boldsymbol{\partial}_{\boldsymbol{x}} \cdot \left( \overline{\boldsymbol{k}} \cdot \partial_{\boldsymbol{x}} T \right) = \sum_{i=1}^{N_p} \left[ h_i \partial_t (\epsilon_i \rho_i) \right] - \partial_t \left( \epsilon_g \rho_g h_g - \epsilon_g p_g \right) + \boldsymbol{\partial}_{\boldsymbol{x}} \cdot \left( \epsilon_g \rho_g h_g \boldsymbol{v}_g \right)$$

## Energy conservation $\sum_{i=1}^{N_p} \left[ \left( \epsilon_i \rho_i c_{p,i} \right) \partial_t T \right] - \partial_x \left( \overline{k} \cdot \partial_x T \right) = \sum_{i=1}^{N_p} \left[ h_i \partial_t (\epsilon_i \rho_i) \right] - \partial_t \left( \epsilon_g \rho_g h_g - \epsilon_g p_g \right) + \partial_x \left( \epsilon_g \rho_g h_g v_g \right)$

Solid phases storage - implicit in T

#### **Energy conservation**

$$\sum_{i=1}^{N_p} \left[ \left( \epsilon_i \rho_i c_{p,i} \right) \partial_t T \right] \left( - \partial_x \cdot \left( \overline{k} \cdot \partial_x T \right) \right) = \sum_{i=1}^{N_p} \left[ h_i \partial_t (\epsilon_i \rho_i) \right] - \partial_t \left( \epsilon_g \rho_g h_g - \epsilon_g p_g \right) + \partial_x \cdot \left( \epsilon_g \rho_g h_g v_g \right)$$

Conduction – implicit in T

#### **Energy conservation**

$$\sum_{i=1}^{N_p} \left[ \left( \epsilon_i \rho_i c_{p,i} \right) \partial_t T \right] - \partial_x \cdot \left( \overline{k} \cdot \partial_x T \right) \left( = \sum_{i=1}^{N_p} \left[ h_i \partial_t (\epsilon_i \rho_i) \right] - \partial_t \left( \epsilon_g \rho_g h_g - \epsilon_g p_g \right) + \partial_x \cdot \left( \epsilon_g \rho_g h_g v_g \right) \right]$$

Solid mass loss by pyrolysis and heterogeneous reactions - explicit

#### **Energy conservation**

$$\sum_{i=1}^{N_p} \left[ \left( \epsilon_i \rho_i c_{p,i} \right) \partial_t T \right] - \partial_x \cdot \left( \overline{\overline{k}} \cdot \partial_x T \right) = \sum_{i=1}^{N_p} \left[ h_i \partial_t (\epsilon_i \rho_i) \right] - \left( \partial_t \left( \epsilon_g \rho_g h_g - \epsilon_g p_g \right) + \partial_x \cdot \left( \epsilon_g \rho_g h_g v_g \right) \right]$$
  
Gas storage – explicit

#### **Energy conservation**

$$\sum_{i=1}^{N_p} \left[ \left( \epsilon_i \rho_i c_{p,i} \right) \partial_t T \right] - \partial_x \cdot \left( \overline{k} \cdot \partial_x T \right) = \sum_{i=1}^{N_p} \left[ h_i \partial_t (\epsilon_i \rho_i) \right] - \partial_t \left( \epsilon_g \rho_g h_g - \epsilon_g p_g \right) \left( \partial_x \cdot \left( \epsilon_g \rho_g h_g v_g \right) \right) \right]$$
Convection – explicit

#### **Energy conservation**

$$\sum_{i=1}^{N_p} \left[ \left( \epsilon_i \rho_i (c_{p,i}) \partial_t T \right] - \partial_x \cdot \left( \overline{k} \cdot \partial_x T \right) = \sum_{i=1}^{N_p} \left[ h_i \partial_t (\epsilon_i \rho_i) \right] - \partial_t \left( \epsilon_g \rho_g h_g - \epsilon_g p_g \right) + \partial_x \cdot \left( \epsilon_g \rho_g h_g v_g \right) \right]$$

#### **Isotropic TACOT properties**



#### **Energy conservation**

$$\sum_{i=1}^{N_p} \left[ \left( \epsilon_i \rho_i (\overline{c_{p,i}}) \partial_t T \right] - \partial_x \left( \overline{\overline{k}} \right) \partial_x T \right] = \sum_{i=1}^{N_p} \left[ h_i \partial_t (\epsilon_i \rho_i) \right] - \partial_t \left( \epsilon_g \rho_g h_g - \epsilon_g p_g \right) + \partial_x \left( \epsilon_g \rho_g h_g v_g \right)$$

#### **Isotropic TACOT properties**



#### **Energy conservation**

$$\sum_{i=1}^{N_p} \left[ \left( \epsilon_i \rho_i (c_{p,i}) \partial_t T \right] - \partial_x \left( \overline{\overline{k}} \right) \partial_x T \right) = \sum_{i=1}^{N_p} \left[ \overline{h_i} \partial_t (\epsilon_i \rho_i) \right] - \partial_t \left( \epsilon_g \rho_g h_g - \epsilon_g p_g \right) + \partial_x \left( \epsilon_g \rho_g h_g v_g \right)$$

#### **Isotropic TACOT properties**



#### **Boundary Conditions**

 $\begin{array}{c} \text{mass balance } [7] \\ \hline \text{mass transfer flux} & \text{advection flux} & \text{environment} \\ \hline \hline \hline \hline \hline \hline \hline \hline \hline \uparrow \uparrow \uparrow \\ (m_{pg} + m_{ca}) z_{k,w} \\ \hline \hline \hline \hline m_{pg} z_{k,pg} & \hline m_{ca} z_{k,ca} \\ \hline \hline \hline \hline \end{array} \\ \begin{array}{c} \hline m_{pg} z_{k,pg} \\ \hline \hline \end{array} \\ \hline \hline \end{array} \\ \begin{array}{c} \hline m_{ca} z_{k,ca} \\ \hline \hline \end{array} \\ \hline \end{array} \\ \begin{array}{c} \hline m_{cs} z_{k,ca} \\ \hline \hline \end{array} \\ \hline \end{array} \\ \begin{array}{c} \hline \hline \end{array} \\ \begin{array}{c} \hline m_{cs} z_{k,ca} \\ \hline \end{array} \\ \hline \end{array} \\ \begin{array}{c} \hline \end{array} \\ \begin{array}{c} \hline \end{array} \\ \begin{array}{c} \hline m_{cs} z_{k,ca} \\ \hline \end{array} \\ \hline \end{array} \\ \begin{array}{c} \hline \end{array} \\ \end{array} \\ \begin{array}{c} \hline \end{array} \\ \end{array} \\ \begin{array}{c} \hline \end{array} \\ \end{array} \\ \begin{array}{c} \hline \end{array} \\ \begin{array}{c} \hline \end{array} \\ \begin{array}{c} \hline \end{array} \\ \end{array} \\ \begin{array}{c} \hline \end{array} \\ \end{array} \\ \begin{array}{c} \hline \end{array} \\ \begin{array}{c} \hline \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \hline \end{array} \end{array} \\ \end{array} \\ \begin{array}{c} \hline \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \hline \end{array} \end{array} \\ \end{array} \\ \end{array}$ 

porous material

Surface mass balance [7]

Enthalpy at the wall  $h_w$ Char ablation rate  $\dot{m}_{ca}$ 

pyrolysis gas (pg) flux char ablation (ca) flux

#### **Boundary Conditions**



Surface mass balance [7]

Enthalpy at the wall $h_w$ Char ablation rate $\dot{m}_{ca}$ 

Surface energy balance [8,9]



Temperature at the wall  $T_w$ 

### Temporal and spatial interpolations

**Temporal interpolation** 

**Spatial interpolation** 



11 **discrete** times (50s to 100s of MSL entry) **linear** interpolation



Separate mesh regions are numerically **connected** by the **Arbitrary Mesh Interface** (**AMI**) utility using local **Galerkin projection** [10] implemented in **OpenFOAM** 

### "Fencing" effect at tile interfaces

Post-test arcjet coupons [5]



Pre Test Sample

85 W/cm<sup>2</sup>, 0.33 atm

m 175 W/cm<sup>2</sup>, 0.28 atm

270 W/cm<sup>2</sup>, 0.27 atm

### "Fencing" effect at tile interfaces

Post-test arcjet coupons [5]



MSL heatshield front surface at the nose



### **Temperature from PATO**



### **Recession from PATO**





### Velocity inside the porous material



IVI (m/s) 0 0.05 0.1 0.15 0.2

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### Tiled configuration changes the flow inside the material

#### Monolithic configuration

#### **Tiled configuration**



### 1D and 3D material response comparison – nose





### 1D and 3D material response comparison – shoulder







#### Hypersonic environment (DPLR)

- Laminar
- Super-catalytic wall
- Non-blowing
- 8 species & 12 reactions

# Porous material response (PATO) Pyrolysis CMA-type BL approx. No finite-rate Equilibrium

#### Hypersonic environment (DPLR)

- Laminar
- Super-catalytic wall
- Non-blowing
- 8 species & 12 reactions ۲



#### **Porous material response** (PATO)

- Pyrolysis
- CMA-type BL approx.
- No finite-rate
- Equilibrium •

#### Hypersonic environment (DPLR)

- Laminar
- Super-catalytic wall
- Non-blowing
- 8 species & 12 reactions

#### Soft coupling

Linear in time Conservative in space by local Galerkin projection

#### Porous material response (PATO)

- Pyrolysis
- CMA-type BL approx.
- No finite-rate
- Equilibrium

### Outputs

- Monolithic & tiled
- Temperature 1D & 3D
- Recession 1D & 3D
- Internal velocity

### Future work



### Future work

Strong coupling

Linear in time

Conservative in space by local Galerkin projection

**Future work includes** 

blowing from pyrolysis & moving mesh from recession

#### Hypersonic environment (DPLR)

- Laminar
- Super-catalytic wall
- Non-blowing
- 8 species & 12 reactions

### New technology

#### • 2020 mission

- Non-uniform thickness
- Transient turbulent
- MMOD & micro-scale

#### New technology

PATO is capable of massively parallel computing for material response

#### Porous material response (PATO)

- Pyrolysis
- CMA-type BL approx.
- No finite-rate
- Equilibrium

### Outputs

- Monolithic & tiled
- Temperature 1D & 3D
- Recession 1D & 3D
- Internal velocity

### References

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## Questions ?

### 9<sup>th</sup> Ablation Workshop

Montana State University, August 30th - 31st, 2017

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