



Challenge and Opportunity of Advanced Materials and Chemistries for Electrochemical Energy Storages Development of NASA Future Missions

James J. Wu, Ph.D.

NASA Glenn Research Center (GRC)

**Seminar at Department of Chemistry and Biochemistry,
University of California at Santa Cruz, California, April 2, 2018**



Outline

- **Introduction/NASA energy requirements**
- **Advanced materials for high energy Li-Ion batteries**
- **Beyond Li-ion battery chemistries**
- **Research opportunities for academia on NASA energy storage technique development**
- **Facilities and capabilities at NASA GRC**
- **Closing remarks**



Energy Is an Important Business

- **Energy is ranked #1 in humanity's top problems for next 50 years**
- **Electrochemical energy conversion and storage, such as batteries, play vital roles for energy security**
- **General strategies to improve battery energy**
 - **Improve/extend existing battery technology today**
 - Optimization/maximum of existing materials and design
 - **Develop new/advanced battery technology**
 - New materials
 - New battery chemistries



Batteries: Important for NASA Missions

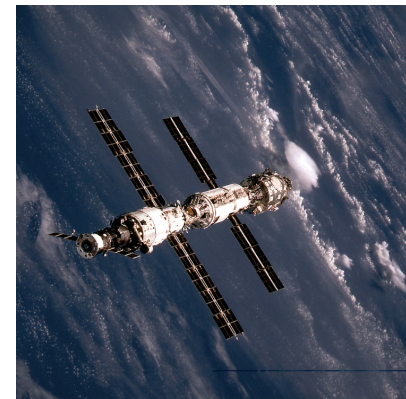
- **Batteries provide**

- a versatile, reliable, safe and portable energy source, and are an essential component of the power system of virtually all NASA missions



- Electrical energy storage options for NASA Space missions, such as:

- Power during spacecraft eclipses
 - Peaking power for sort high power needs





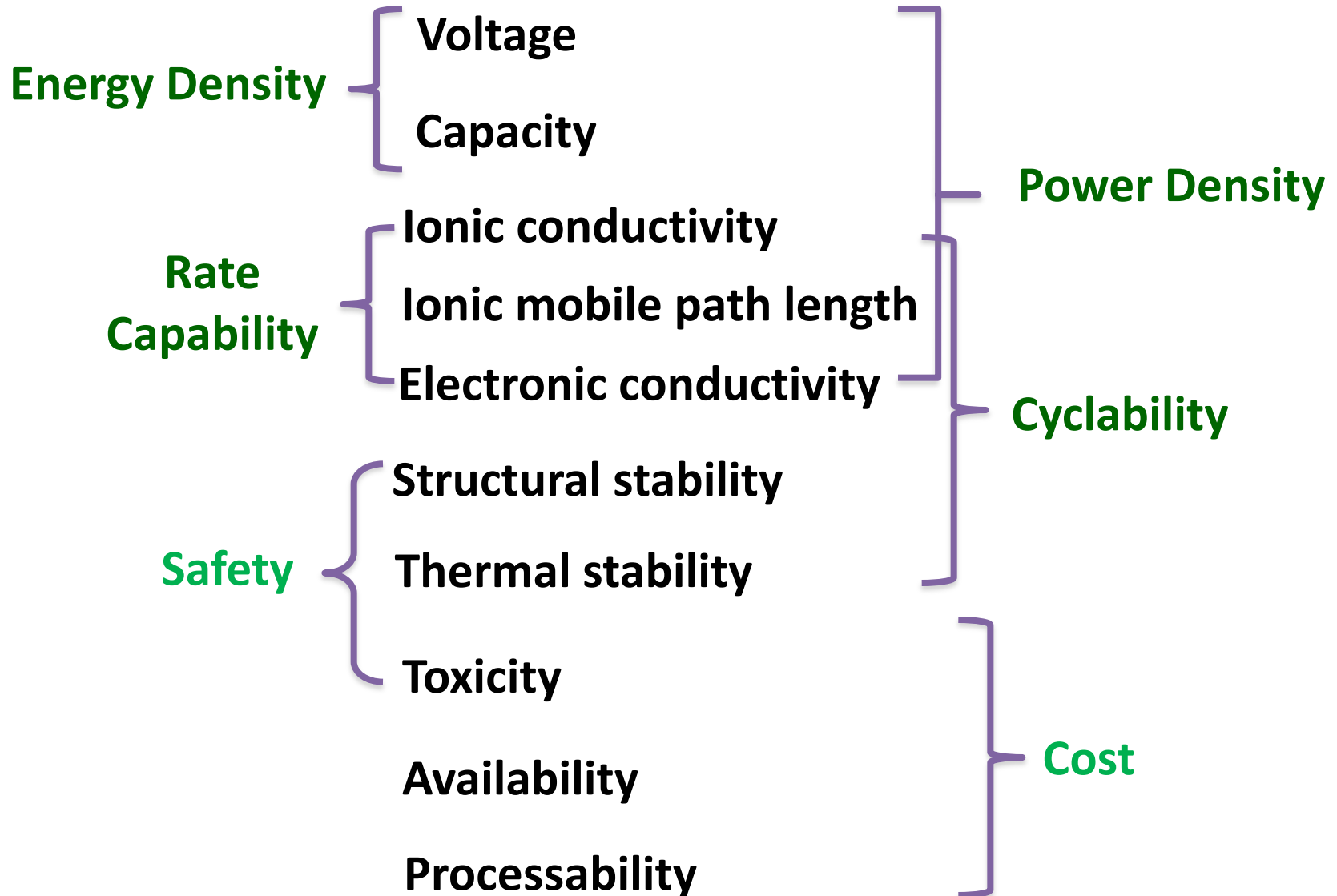
Desired Battery Properties for NASA Missions

- Safe
- High in specific energy
- Light in weight
- Compact in volume
- Long in shelf life
- Durable in wide temperature range and harsh environment
- Reliable in meeting mission requirements





Battery Material Design and Selection

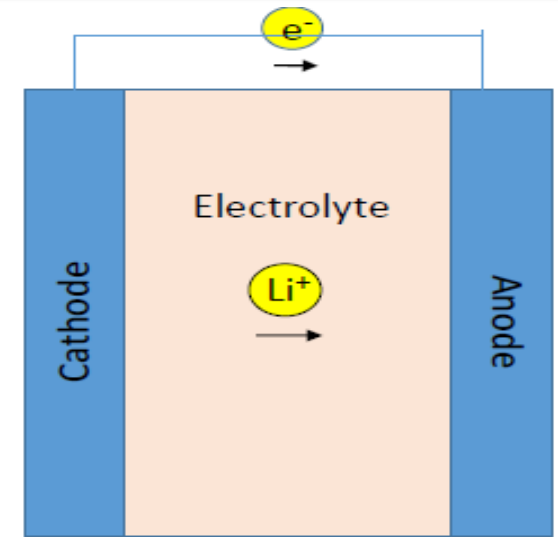




State-of-Art (SOA) Li-Ion Battery (LIB)

- **LIB Specs:**

- Specific energy: 180-200 Wh/kg
- Specific power: 300 W/kg
- Cycles: 1000s (excellent)
- Temp range: -20°C to 60°C
- LIB has excellent rechargability



Anode: Graphite;
Cathode: Metal Oxide;
Electrolyte: Carbonate Liquids

- **Limitations:**

- Maximum of energy density <250 Wh/kg
- Electrolyte flammable and fire hazards



NASA Demands Very High Specific Energy Batteries



Electric Aviation 500 – 750 Wh/kg

- Green aviation – Less noise, lower emissions, high efficiency
- Hybrid / All-electric aircraft – Limited by mass of energy storage system
- Commercial aviation – Safe, reliable, lightweight on-board electric auxiliary power unit



Extravehicular Activities (Spacesuit power) >400 Wh/kg

Required to enable untethered EVA missions lasting 8 hours within strict mass and volume limitations.

Astronaut life support
Safety and reliability are critical
100 cycles



Landers and Rovers, Robotic missions, In- space habitats >500 Wh/kg

Batteries are expected to provide sufficient power for life support and communications systems, and tools including video and lighting

- >100 cycles

NASA future mission requirements far exceed the capabilities of lithium-ion chemistries

➤ Progress in these areas requires advances in safe, very high energy batteries



Advanced Safe, High Energy Li Batteries

- **Advanced electrode materials to improve Li-ion battery energy**
 - Anode active materials
 - Cathode active materials
- **Advanced electrolyte to improve safety**
 - Non-flammable additives
 - Ionic liquids
 - Solid state electrolyte
- **Beyond Li-ion battery chemistries**
 - Li-S Chemistry
 - Metal air batteries chemistries such as Li/O_2



Advanced Anode Development

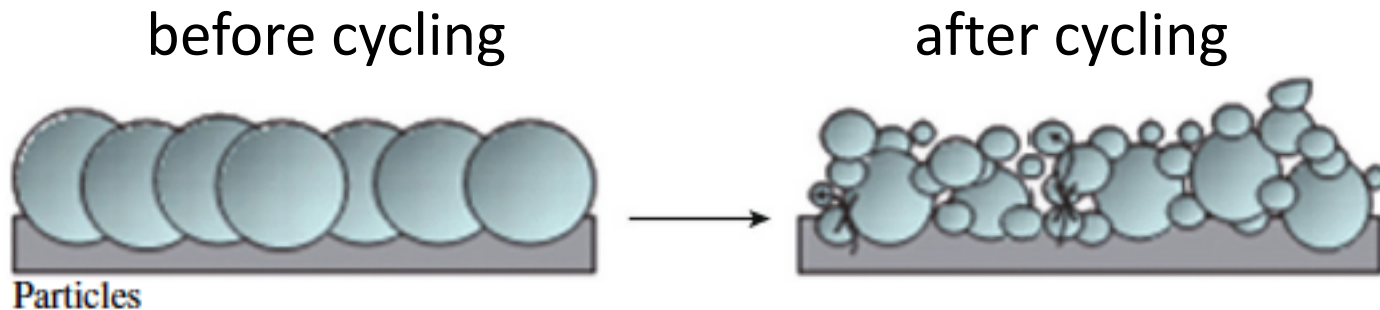
- **Silicon (Si) and lithium (Li) metal:**
promising as advanced anode materials
- **Very high theoretical capacity:**
 - Si (4200 mAh/g)
 - Li metal (3800 mAh/g)
 - 10x higher than SOA graphite (370 mAh/g) anode
- **Additional advantages:**
 - Si: abundant element on Earth
 - Li: lightest metal, 100% active material, no binder needed



Challenges: Si as Advanced Anode Materials

- **Challenges for Si:**

- Poor electrical conductivity
- Low diffusion coefficient of Li^+ in Si
- High volume changes (up to $\sim 400\%$) in Si particles upon lithiation and delithiation process, causing capacity fade

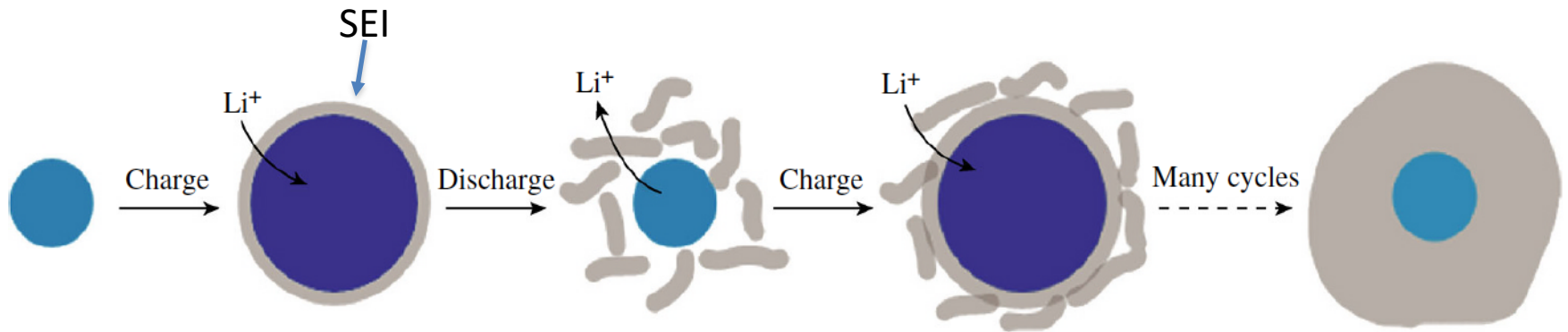


C.K. Chan et al Nat. Nanotechnol. 3(1), 31-35 (2008)



Approaches to Overcome Si Challenges

- with conductive materials (typical carbon) to form composite
- Nanosized Si particles/C composite, Si nanowire
- Nanostructured Si/CNF
- Improve Si electrode/electrolyte interphase: such as fast forming and stabilizing solid electrolyte interphase (SEI)

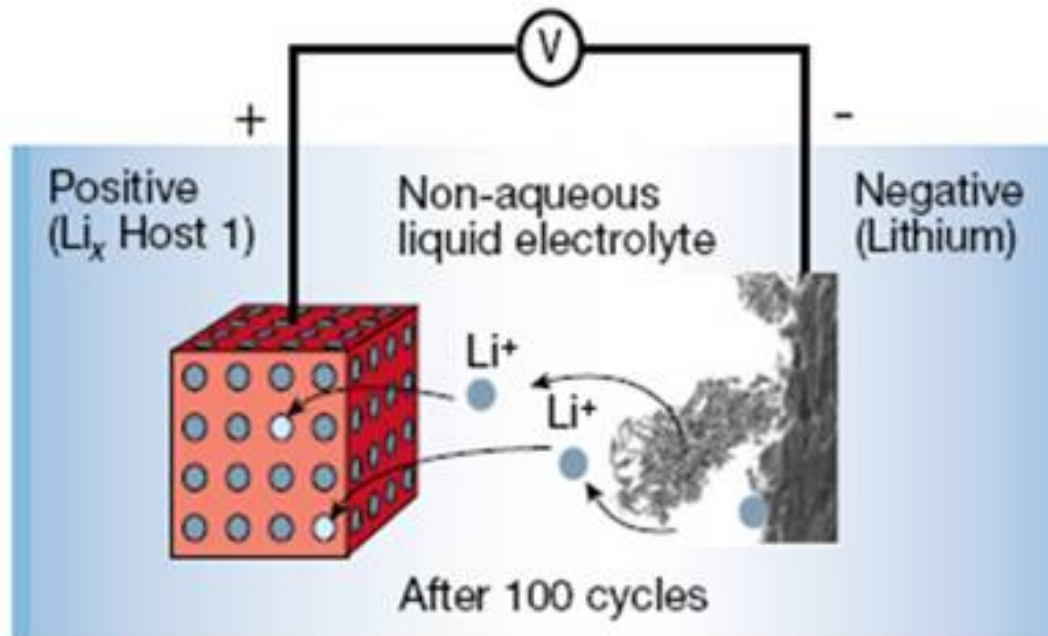


H. Wu et al Nat. Nanotechnol. 7(5), 310-315



Challenges: Li Metal as High Capacity Anode

- Very reactive, parasite reactions, resulting in low coulombic efficiency (CE)
- Shape changes during cycling, causing poor cyclicability
- Dendrite growth to cause safety concern



(Tarascon, J. M. et al, Nature, 2001, 414, 359-366)



Approaches for Li Metal Safe Cycling

- **Ways to overcome:**
 - Ionic liquid in electrolyte
 - High concentration Li salts
 - Solid state electrolyte
 - Ceramic
 - Solid polymer nanocomposite
- **Research on Li metal cycling at NASA**
 - NASA ARMD award to combine MD computational with experimental to study ionic liquid on Li metal (NASA Ames with GRC) John Larson in ARC, James Wu et al in GRC
 - CIF awards to solid polymer nanocomposite to make Li metal cycle safely (PI James Wu, 2016, 2018)



Advanced Cathode Development

- **Develop Li(NMC) Materials**
 - Offer higher capacity (>280 mAh/g) and enhanced thermal stability over conventional LiCoO_2 (170 mAh/g) cathodes
 - High voltage materials
- **Major Challenge:**
 - Voltage fade during cycling
- **Development Focus on:**
 - Increasing specific capacity
 - Improving rate capability
 - Stabilizing materials for higher voltage operation
 - Reducing irreversible capacity loss
 - Increasing tap density



Advanced Electrolyte Development

- **Develop advanced electrolytes with additives**
 - Non-flammable electrolytes and flame retardant additives
 - Stable at potentials up to 5V
 - Compatible with the NASA chemistries
 - **Development focus on:**
 - Reducing flammability
 - Stabilizing materials for higher voltage operation
 - Compatibility with mixed-metal-oxide cathodes and silicon composite anodes
- **Develop ionic liquids, solid polymer nanocomposite electrolyte and solid state electrolyte**



Solid State Electrolyte: Improve Safety

- **Solid state electrolyte enhances safety**
 - Ceramic type such as Garnet type
 - Solid polymer nanocomposite electrolyte
- **Challenges in Solid State Electrolyte Development**
 - Low ionic conductivity at room temperature ($<10^{-3}$ S/cm)
 - High contact/interfacial resistance issues)
 - Processing/fabrication issues
- **Approaches to Overcome**
 - 3D design of electrodes
 - Thin layer of electrolytes
 - Hybrid w/polymer, ionic liquid & nanomaterials





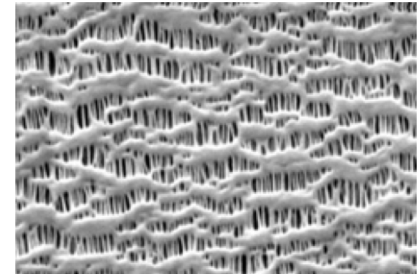
Advanced Separator Development

- **Significance**

- The function and reliability of the separator are critical for optimal cell/battery performance and safety
- Affects internal cell resistance, stability, cycle-life, operating temperature range and rate kinetics and intrinsic cell safety, especially under abuse or elevated-temperature conditions

- **Technology Challenges**

- Design optimization for high porosity and low ionic resistance to facilitate ionic conductivity while maintaining mechanical strength
- Shutdown separators must “shutdown” cell reactions at optimum temperature without shrinking or losing mechanical integrity

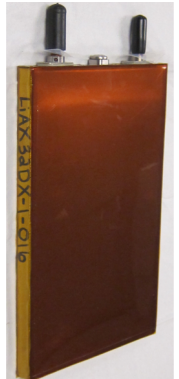




NASA Advanced Space Power System Program (ASPS)

(2008-2014)

- Partnered with University/Industry for Advanced components development
 - Si as advanced anode – Georgia Tech, Physical Science Inc.
 - NMC as advanced cathode – University of Texas at Austin
 - Low-flammable electrolyte: NASA JPL
- Although NMC cathode was difficult to scale up, NASA worked with industrial partners (PSI/Yardney) for scale-up Si production and the use of Si as anode in large format flight-type prismatic cells (35Ah)





Beyond Li-Ion Battery Chemistries

- **Li/S Battery Chemistry:**

- **Advantages**

- Very high theoretical specific energy: 2680 Wh/kg
 - abundance
 - Non-toxic
 - Environmentally benign

- **Challenges**

- Poor electronic conductivity
 - Low voltage
 - Short cycle life.
 - Li dendrite growth
 - Polysulfide shuttle causing short shelf life

- **Metal-Air Batteries Chemistries, e.g. Li/O₂**

- Very high theoretical specific energy (17000 Wh/kg)
 - Li at 13 kWh/kg, Carbon is light, O₂ from air or cylinder
 - Battery depends on environment e.g. O₂, humidity, air flow, temperature

 - air cathode is “open” to environmental;
 - Complex discharge products, high overpotential during cycling
 - Safety, Li dendrite growth

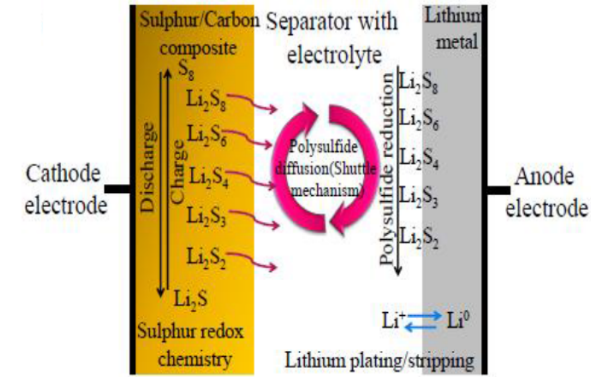
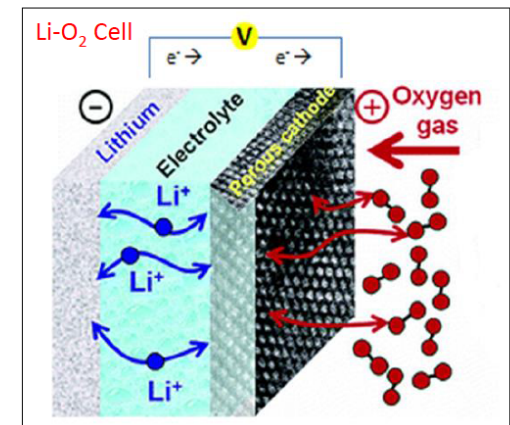


Image- courtesy of Storagenergy Technologies Inc





NASA Advanced Energy Storage System (AESS) Project under Game Change Program (2014-2017)

- **Phase I:** 8 month, 4 awards were given:
 - 1 award (Category I) on Si Anode based Li-ion battery (Amprius)
 - 3 awards (Category II) on Li/S battery development (JPL/CIT, IUPIU, University of Maryland (UMD))
- **Phase II:** 12 month, 2 award were given:
 - **Amprius: Silicon Anode Based Cells for High Specific Energy Systems**
 - Commercial standard cathode paired with Amprius' silicon anode
 - Phase I: Deliverables with >300 Wh/kg after 225 cycles (pouch cell)
 - Phase II: Scale-up cells (2X size in phase I) with >300 Wh/kg over 200 cycles
 - Additional temperature & safety evaluations at cell & battery levels
 - Battery pack brassboard delivering > 250 Wh/kg
 - **University of Maryland: Garnet Electrolyte-Based Safe Lithium-Sulfur Energy Storage**
 - All solid state battery with unique and scalable trilayer(porous-dense-porous) solid state electrolyte (SSE) structure.
 - Phase I: demonstrated the feasibility in lab cells (coin cell)
 - Phase II: optimize the parameters and scale up to 5cm x 6cm sizes with targeted energy density ~500 Wh/kg



Supercapacitor vs. Battery

- **Two major types of electrochemical-based energy storage devices**
 - **Battery:** Faradic/exothermal redox reaction
(many different varieties)
High energy density
Electrode degradation
Limited cycle life
 - **Capacitor:** Electrostatic/capacitive interaction
High power density
Electrode structural integration
Long cycle life



How to Improve Both Power Density and Energy Density of Battery

- **New materials with high specific capacity**
- **Novel architectures: 3D design of electrode**
 - **Thinner electrode (fast ionic transport)**
 - **High electronic conductivity (fast e^- transport)**
 - **High electrode/electrolyte interfacial area (fast charge transfer across the interface)**



How to Improve Both Energy Density and Power Density of Capacitor

- One approach is to hybrid the capacitor electrode with one battery electrode i.e. asymmetric supercapacitor
- One electrode (as cathode) from capacitor (i.e. active carbon w/high porosity and high surface area) undergoes electrostatic interaction
- The other electrode (as anode) from battery (i.e. silicon with high specific capacity) undergoes electrochemical redox reaction



NASA GRC Capabilities Related to Batteries

- ***To support the development advanced cells and batteries for aerospace applications:***
 - Development and characterization of materials, advanced battery components, and battery cells
 - Integration of advanced components and study of interactions
 - Development and application of modeling tools
 - Performance evaluations of cell and battery systems
 - Validation and verification of aerospace and flight cells and batteries
 - Li-ion battery/power system test bed development
 - Design support for battery/power systems for flight vehicles
 - Guidelines and requirements development and documentation for batteries for use in flight systems
- **Capabilities to address all levels of advanced battery design and development, ranging from component-level to cell-level to battery level**



NASA GRC Battery Energy Storage System – Background and Heritage

Overview

- Batteries provide a versatile, reliable, safe, modular, lightweight, portable source of energy for aerospace applications.
- Batteries have demonstrated the life and performance required to power current missions.
- Li-Ion batteries offer improvements in specific energy, energy density, and efficiency

Experience

- Lead battery development effort for Exploration Technology Development Program, Energy Storage Project
- Developed and validated advanced designs of Ni-Cd and Ni-H₂ cells adopted by NASA, cell manufacturers and satellite companies.
- Evaluated flight battery technologies for ISS
- Developed lightweight nickel electrodes, bipolar nickel hydrogen battery designs
- Jointly sponsored Li-ion battery development program with DoD that developed Li-Ion cells used on Mars Exploration Rovers
- Led NASA Aerospace Flight Battery Systems Steering Committee –agency-wide effort aimed at ensuring the quality, safety, reliability, and performance of flight battery systems for NASA missions.
- Conducted electric vehicle battery programs for ERDA/DOE

Products/Heritage

Beyond Li-ion: Li/S, Li/O₂

Li-Ion: Lithium-Ion

Ni-Cd: Nickel-Cadmium

Ni-H₂: Nickel-Hydrogen

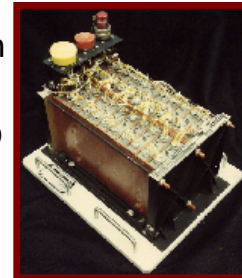
Ni-MH: Nickel-metal hydride

Ni-Zn: Nickel-Zinc

Ag-Zn: Silver-Zinc

Na-S – Sodium Sulfur

LiCFx: Lithium-carbon monofluoride





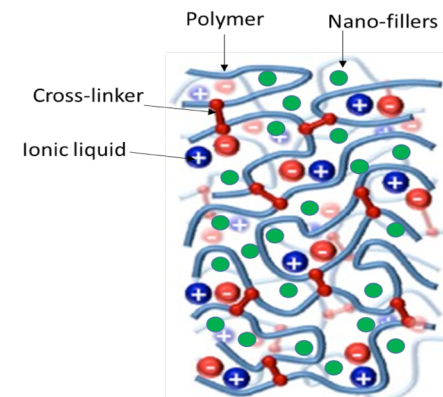
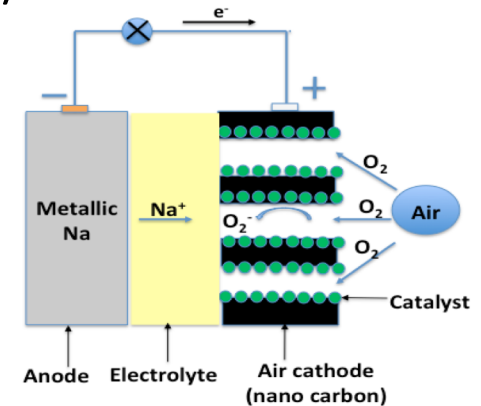
Advanced Battery Component/System Development

- **Convergent Aeronautics Solution (CAS) Projects**
 - High risk/high return projects
 - University/industry partners
- **Center Innovative Funds (CIF)**
 - seeding fund for support innovative ideas
- **Independent Research and Development (IRAD)**



NASA Center Innovative Fund (CIF)

- **CIF: seeding fund for support innovative ideas**
- **Previous/current awards:**
 - High Capacity Sulfur Cathode Development (PI James Wu, FY 12 CIF)
 - Safe and High Energy Density Na/O₂ Development w/Collaboration with University of Akron (PI James Wu, FY 15 CIF)
 - Novel catalyst on carbon cathode to make a superior air-cathode
 - New electrolyte for long cycle life
 - Solid Polymer Nanocomposite Electrolyte (PI James Wu, FY16, FY18 CIF)
 - Polymers as host matrix
 - Integrated with ionic liquid and Li salt
 - Incorporated with nanomaterials to
 - Further improve ionic conductivity
 - Mitigate interface with Li metal
 - Enhance electrochemical stability
 - Improve mechanical strength





Cell Integration

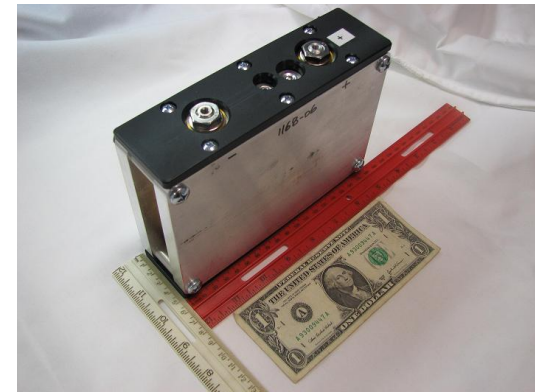
- **Assess performance of integrated components**
- **Predict full cell performance**
- **Determine optimum cycling parameters and cycling limitations**
- **Identify and understand performance and compatibility issues**



Cell Development

- **Assess NASA-developed components**
 - Build and test electrodes and screening cells
 - Provide manufacturing perspective from the start
- **Scale-up NASA-developed components**
 - Transition components from the lab to the manufacturing floor
- **Build and test evaluation cells:**
 - Determine component interactions
 - Determine cell-level performance
- **Design flightweight cells**
 - Identify high risk elements early

34PCell



DD Cells





Analytical and Thermal Safety Evaluations

- **Analytical studies to assess component structures, particle size and distribution, morphology, elemental composition, electrode purity, etc.**
- **Characterization of thermal behavior of cell components by Differential Scanning Calorimetry (DSC)**
 - **Separators**
 - **Electrolytes**
 - **Electrodes harvested from fully charged cells**
- **Characterization of thermal stability of cells and components by Accelerating Rate Calorimetry (ARC)**



Capabilities and Facilities for Batteries Research



Capabilities

- Fundamental electrochemical research - component development and characterization with state-of-the-art analytical test capability
- Cell/Battery Design
- Cell/Battery Performance and Life Testing
- Cell/Battery Safety Testing
- Battery Performance Modeling
- Environmental Testing

Facilities:

- Development Laboratories - SOA equipment for materials and component development, and analytical and electrochemical characterization
- 600 ft² Dry room with 1% relative humidity for handling moisture sensitive materials used in lithium based batteries
- State-of-the-art battery cycling facilities with >100 independent test channels, 1-200 Ahr, 1-50 V
- Environmental chambers to evaluate performance as a function of temperature (-75°C to +200°C)
- Accelerating Rate Calorimeter

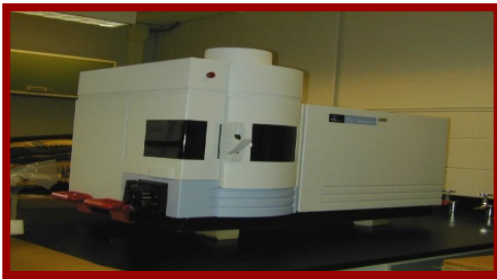


Branch Facilities



Imaging and Material Analysis Laboratory – Surface and Thermal Analysis Capability

- Inductively Coupled Plasma Optical Emission Spectrometer
- Scanning Probe Microscope
- Scanning Electron Microscope Energy Dispersive Spectrometer
- Stereomicroscope
- BET Surface Area Analyzer



Thermal and Material Analysis Laboratory

Molecular analysis, particle size distribution, thermal property analysis

- Differential Scanning Calorimeter
- Fourier Transform IR Spectrometer
- Thermogravimetric Analyzer (TGA)
- Raman Spectrometer
- Particle Size Analyzer





NASA/University Program - Faculty

- **SBIR/STTR (Small Business Innovation Research/ Small Business Technology Transfer) program**
- **NASA EPSCoR (Experimental Program to Stimulate Competitive Research)**
 - Solicits topic-specific proposals addressing high-priority NASA research and technology needs, awards are up to \$750K for a 3-year performance period
- **NASA MIRO program**

Minority university research and education program (MUREP) Institutional Research Opportunity, up to 1 million/year for 5 years
- **Small Spacecraft Technology Program, Smallsat Technology Partnership**

\$200K for 2 years with NASA researcher is also covered
- **NASA early faculty proposal call**
- **NASA Summer Faculty Research Opportunity**



NASA/University Program - Students

- **NASA Student Internship**
 - NASA Interns, Fellows and Scholars (NIFS), One Stop Shopping Initiative (OSSI) <https://intern.nasa.gov/solar/we/public/main>
- **NASA Space Technology Research Fellowship (NSTRF)**
- **Aeronautics Scholarship and Advanced STEM Training and Research (AS&ASTAR) Fellowship (AS & ASTAR)**



Closing Remarks

- **Summary Statement**

At NASA GRC, we have the capability to develop cell and battery technologies from fundamental research on battery components through the development, maintenance and management of flight battery systems

- **Area of Interest for Collaboration/Business Development**

- **Advanced battery/capacitor technologies**

- Component Level
- Cell Level

- **Next generation battery/capacitor technologies**

- **Cell and battery safety**

- **System level solutions to maximize energy**

- **Charge control**



Acknowledgements

- **Prof. Yat Li and Prof. Carrie Partch, University of California at Santa Cruz**
- **Battery Team Members (Thomas Miller, Lisa Kohout, Bill Bennett, Bri Demattia, Pat Loyselle, Concha Reid, Phillip Smith) at Photovoltaics and Electrochemical Systems Branch at NASA Glenn Research Center**



Thank you!

Any Questions?