

Hybrid reduced order model for N₂-N₂ interactions for application to dissociation and energy transfer processes

Oral Presentation

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Recent work in the aerothermodynamics community has focused on the development of reduced-order models for thermo-chemical non-equilibrium which avoid the restrictive assumptions of multi-temperature models and the prohibitive cost associated with State-to-State (StS) models. In the present work, this is accomplished by lumping energy states together and assuming groups of states are roughly in equilibrium. As a result, the non-equilibrium behavior of a gas can be captured at a reduced computational cost from a full StS simulation.

In this work, we present a hybrid grouping model for studying energy transfer and dissociation in a mixture of nitrogen molecules due to N₂-N₂ reactions. This is accomplished by making use of a grouping strategy informed by data from the N₂-N StS kinetic data.^[1] However, due to the massive computational cost associated with constructing StS data for the N₂-N₂ system, the kinetic data for the hybrid grouping model are calculated using the quasi-classical trajectory (QCT) method by sampling states for trajectory within the groups.^[2,3] This general framework is called the Maximum-Entropy Quasi-Classical Trajectory (ME-QCT) method. The primary challenge associated with this method is that rates for reverse grouped reactions cannot be obtained through detailed balance at a group level, due to the variation of group internal temperatures. To construct the full model for N₂-N₂ grouped kinetics using the ME-QCT method, detailed balance is invoked at the microscopic level, allowing for the calculation of the full kinetic data from QCT.

Results will be presented using the full ME-QCT model for the N₂-N₂ system in an isothermal and isochoric reactor simulation. In addition, simple CFD test cases for a one-dimensional standing shock and for a quasi-one-dimensional nozzle will be used for demonstration of the ME-QCT method. This method allows for the calculation of non-equilibrium behavior for the N₂-N₂ system without the prohibitive cost of a full StS simulation. Moreover, it enables the construction of a unified model for the dissociating and recombining non-equilibrium flows.

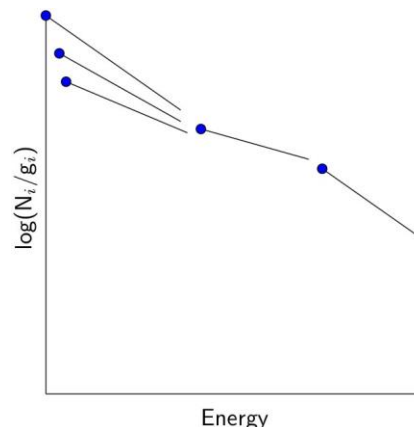


Fig.1: Schematic of hybrid grouping strategy where low lying energy states are grouped according to vibrational quantum number and high lying energy states are grouped according to energy.

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REFERENCES

- [1] A. Sahai, B. Lopez, C.O. Johnston, and M. Panesi, "Adaptive coarse-graining method for energy transfer and dissociation kinetics of polyatomic species," *J. Chem. Phys.*, **147**, 054107 (2017).
- [2] R.L. Macdonald, R.L. Jaffe, D.W. Schwenke and, M. Panesi, "Construction of a Maximum Entropy Quasi-Classical Trajectory Method. Part I: Theory and Application to N₂-N₂ System," *J. Chem. Phys.*, **148**, 054309 (2018).
- [3] R.L. Macdonald, M.S. Grover, T.E. Schwartzentruber, and M. Panesi, "Construction of a Maximum Entropy Quasi-Classical Trajectory Method. Part II: Comparison against the Direct Molecular Simulation Method," *J. Chem. Phys.*, **148**, 054310 (2018).