

Phenolic Polymer Interactions with Water and Ethylene Glycol Solvents

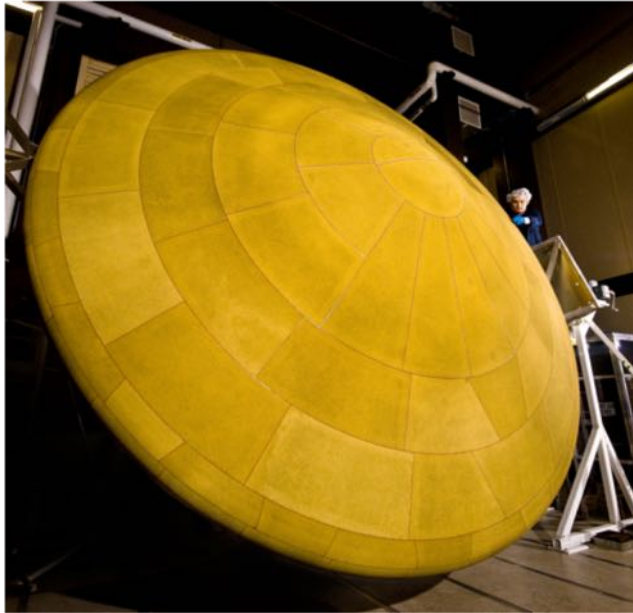


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¹AMA, Inc., Thermal Protection Materials Branch, NASA Ames Research Center

²EAP, Thermal Protection Materials Branch, NASA Ames Research Center

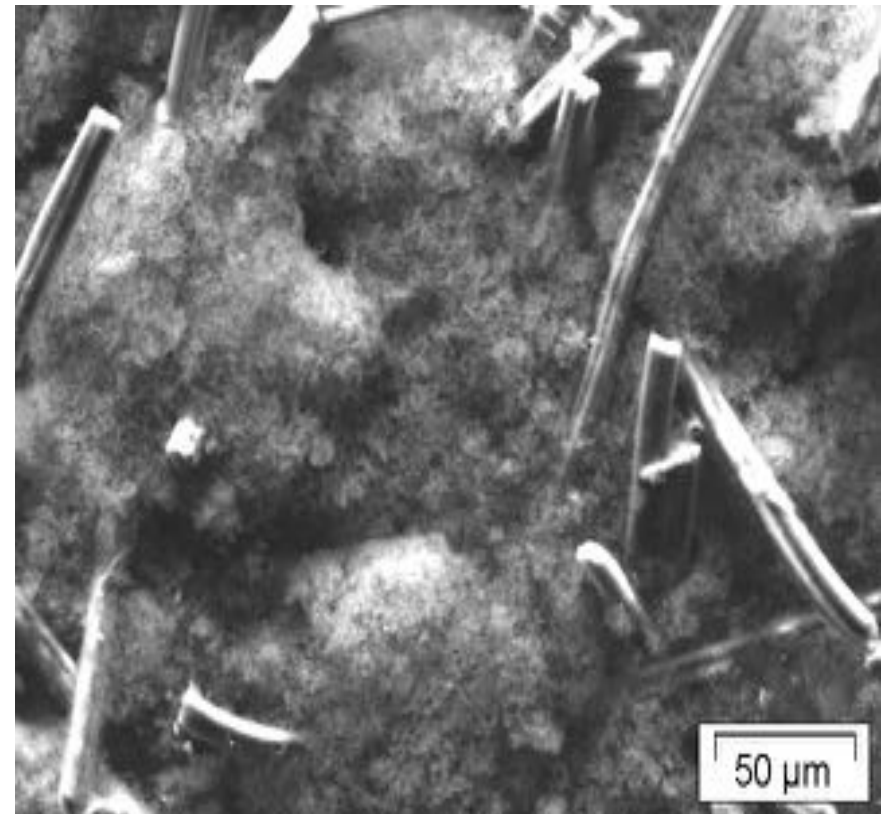
³Thermal Protection Materials Branch, NASA Ames Research Center



Mars Science Lander



Stardust



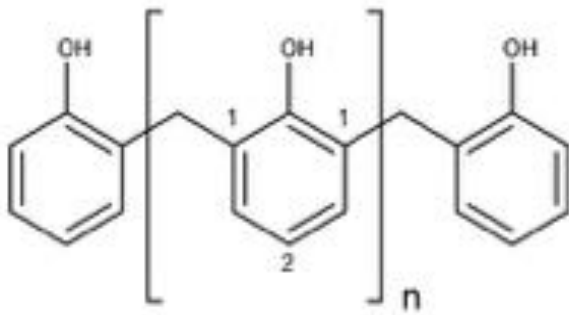
Ablative Composites for Re-entry
(carbon fiber/phenolic matrix)



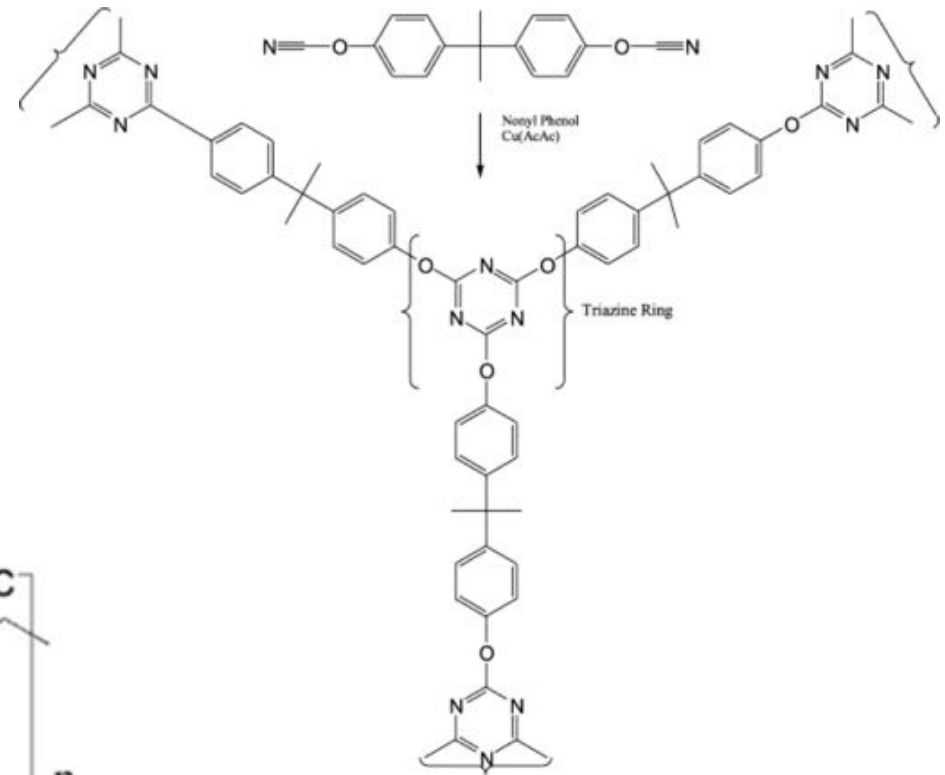
Next Generation Resins for Heat Shields



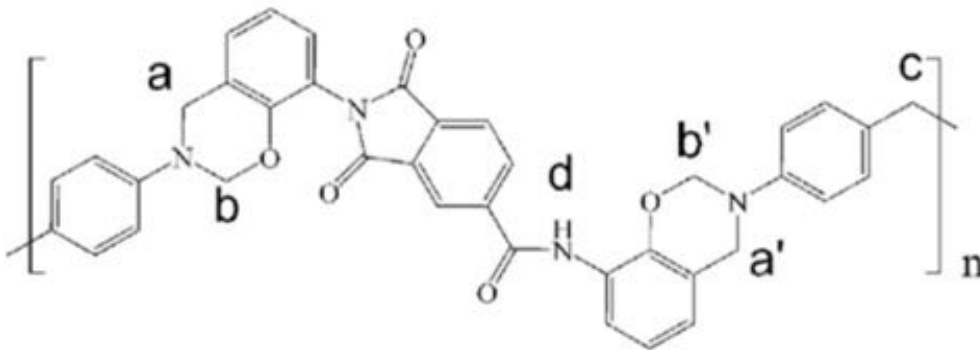
Phenolic (SOA)



Cyanate Esters



Polyimides



New resin chemistries for heat shields require different solvents for processing

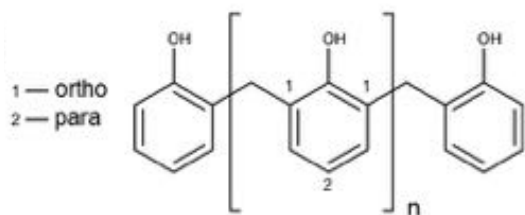


Phenolic Polymers

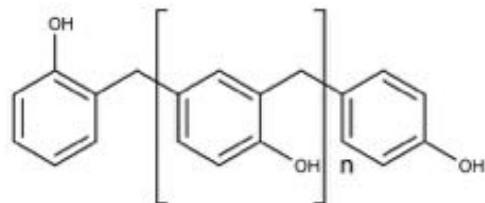


Polymers

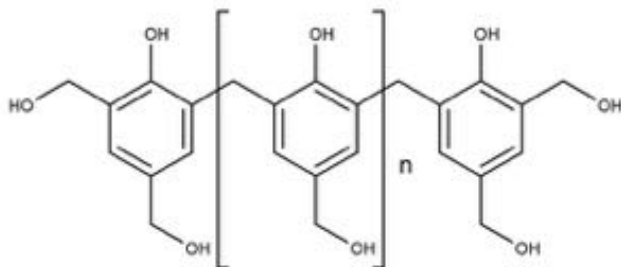
ortho-ortho novolac



ortho-para novolac

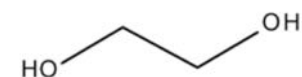


ortho-ortho resole



Solvents

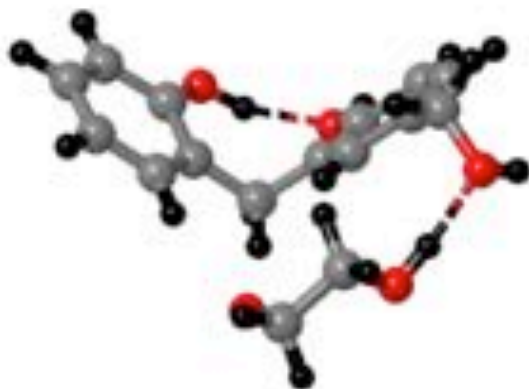
ethylene glycol



water

H₂O

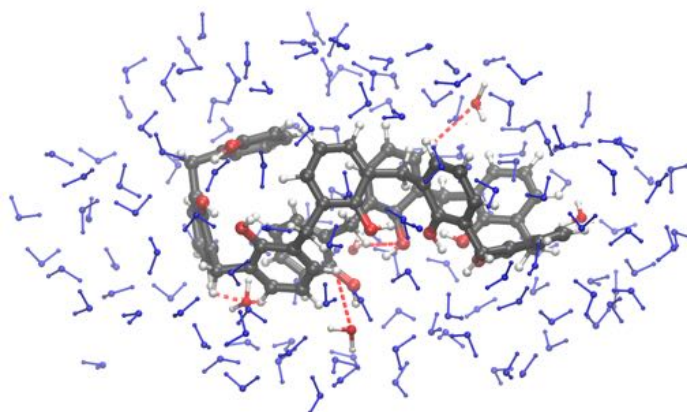
Design rules for SOA polymer and solvents



Ethylene glycol-phenolic dimer

Quantum Chemical Calibration: understand basic polymer-solvent interactions and benchmark MD models

- combination of DFT, MP2, CCSD(T)
- water and ethylene glycol dimers
- solvent-monomer dimers



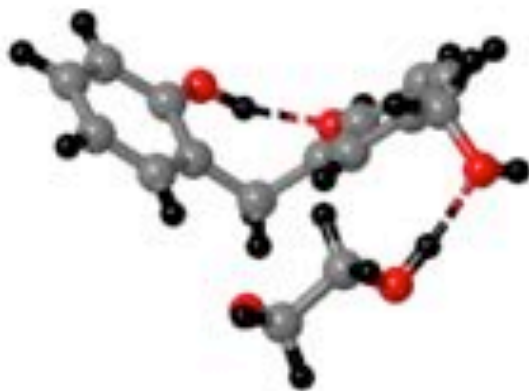
Phenolic in Water

Molecular Dynamics Simulation: characterize polymer solubility in solvents

- OPLS-AA-SEI force field
- single polymers in large solvent boxes
- 500-100,000 solvent molecules
- polymers with 3-27 units



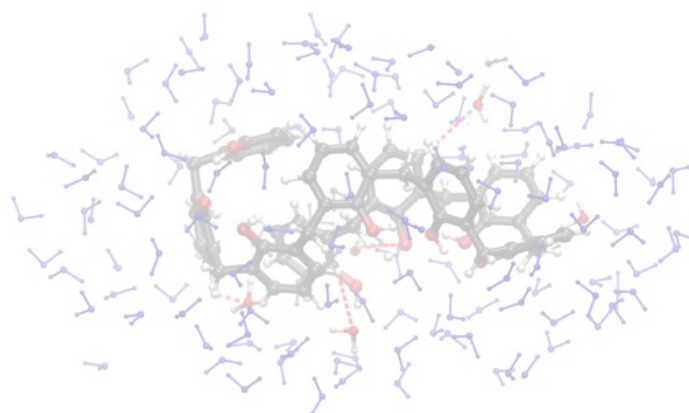
Outline



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Phenolic in Water

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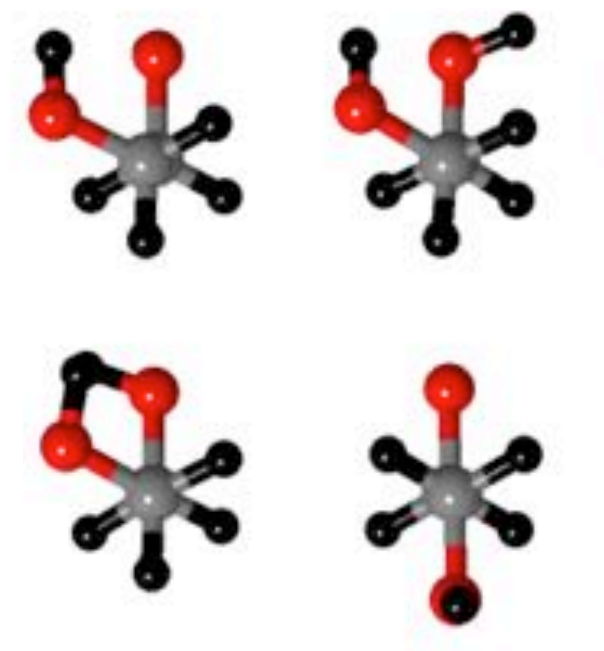
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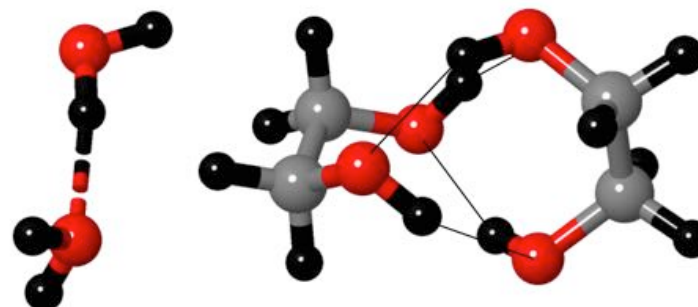
Water and Ethylene Glycol Interactions



Conformers of Ethylene Glycol



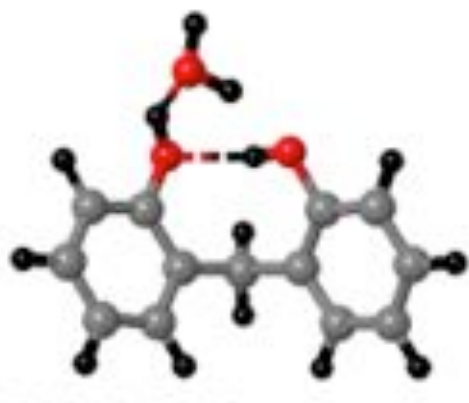
Water and Ethylene Glycol Dimers



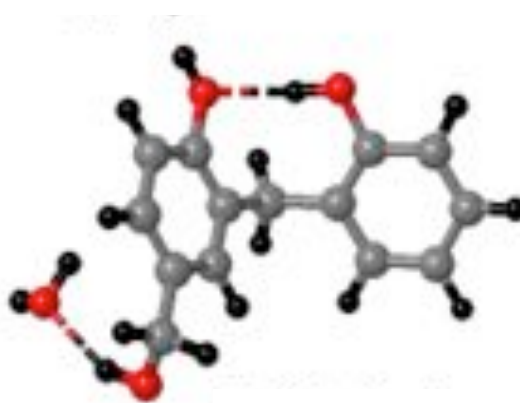
OPLS energetics within 2 kcal/mol of CCSD (T)

H = black; O = red; C = gray

2 Unit Novolac



2 Unit Resole



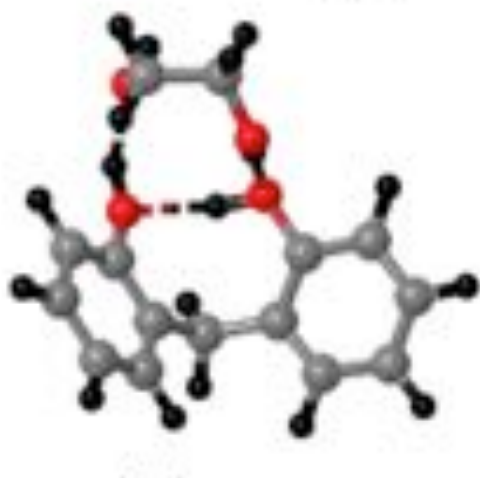
3 Unit Novolac



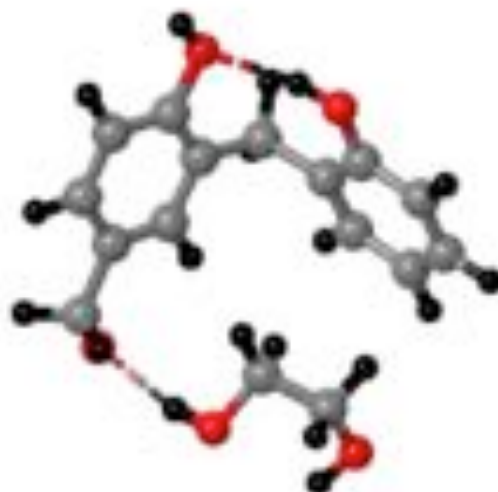
OPLS interactions within 2 kcal/mol of MP2/CBS

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2 Unit Novolac



2 Unit Resole



3 Unit Novolac

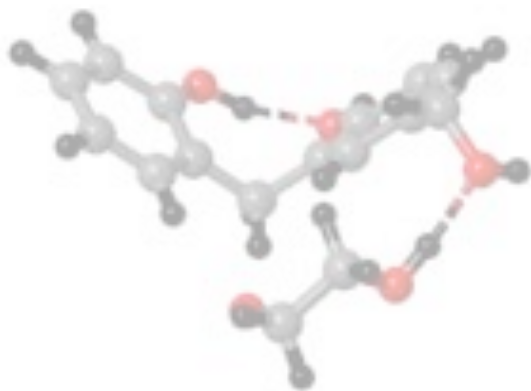


OPLS interactions within 3 kcal/mol of MP2/CBS

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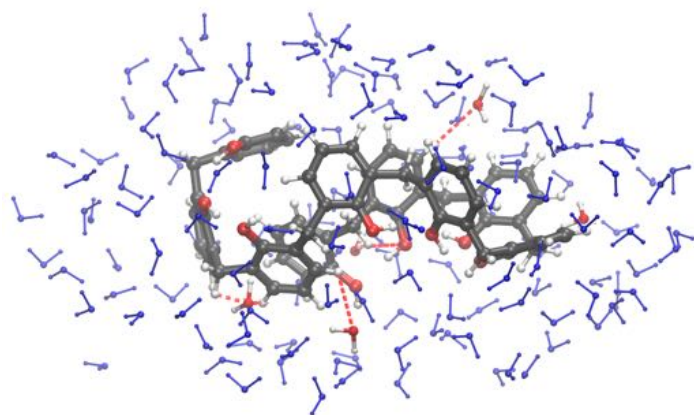
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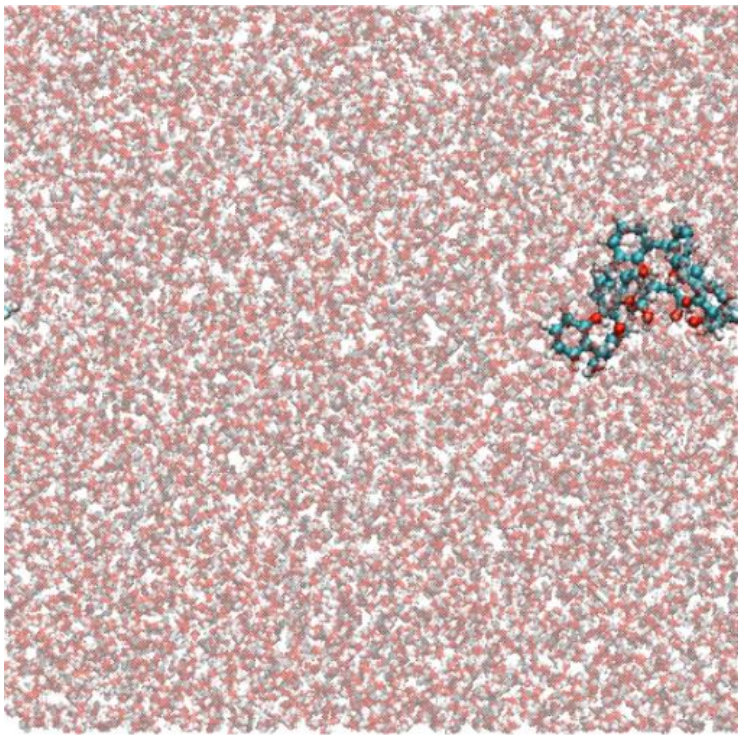


Phenolic in Water

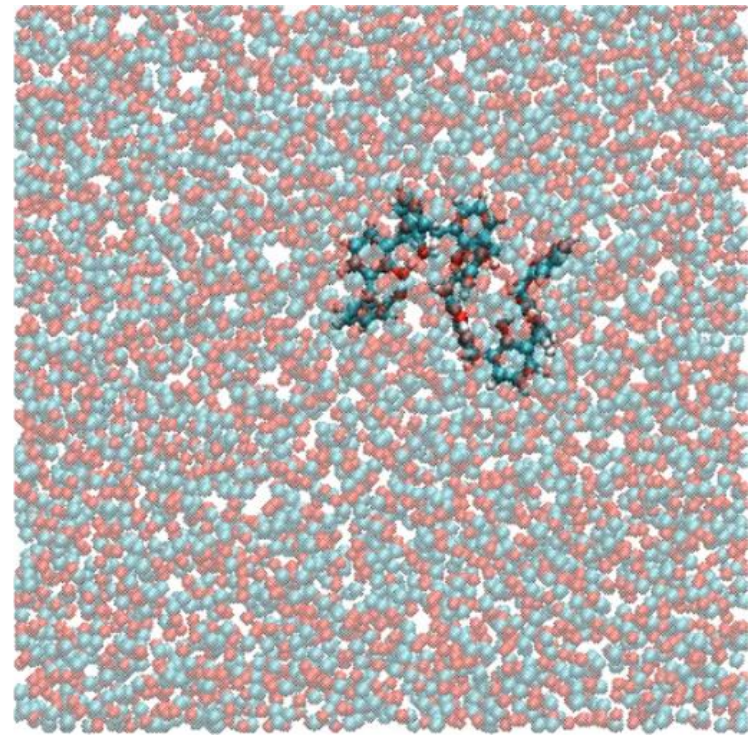
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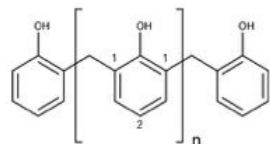
Phenolic in Water



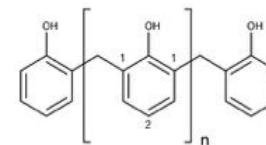
Phenolic in Ethylene Glycol



Diffusion and viscosity of solvent strongly affect polymer dynamics

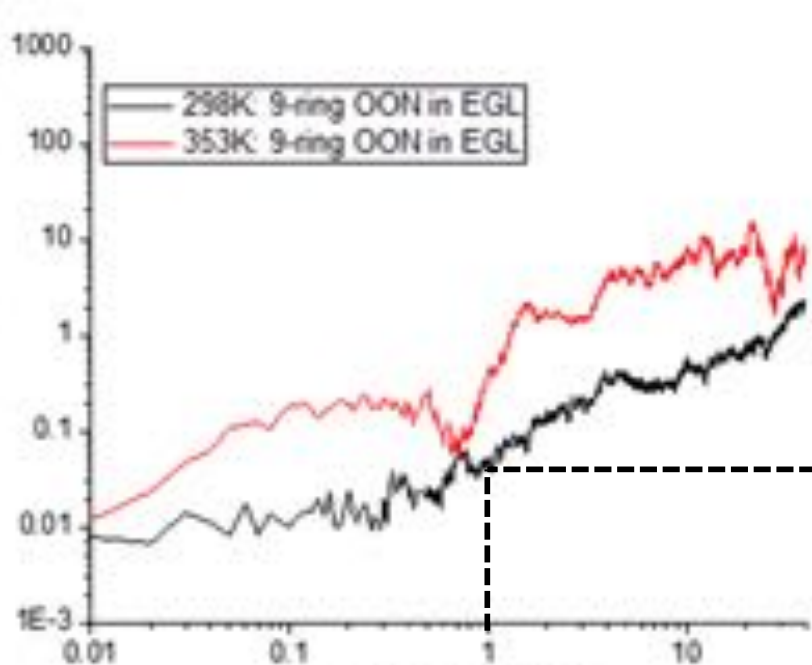


ortho-ortho novolac

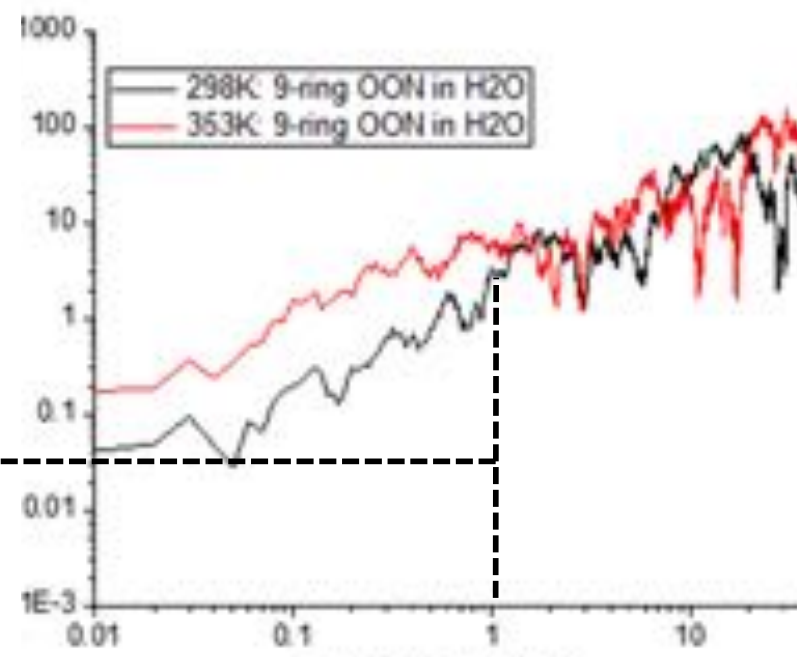


ortho-ortho novolac

Mean Square Displacement (nm²)



Simulation Time (ns)



Simulation Time (ns)

Larger diffusion coefficients in water

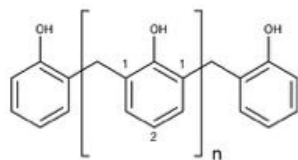


Solvation Free Energy

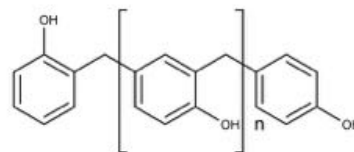


Solvation Free Energy (kcal/mol)

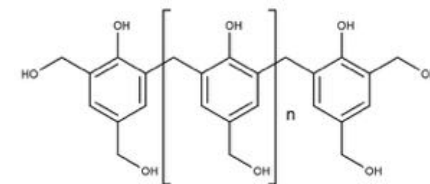
Ethylene Glycol	-27.1	-53.1	-84.0
Water	-8.5	-23.8	-46.5



ortho-ortho novolac



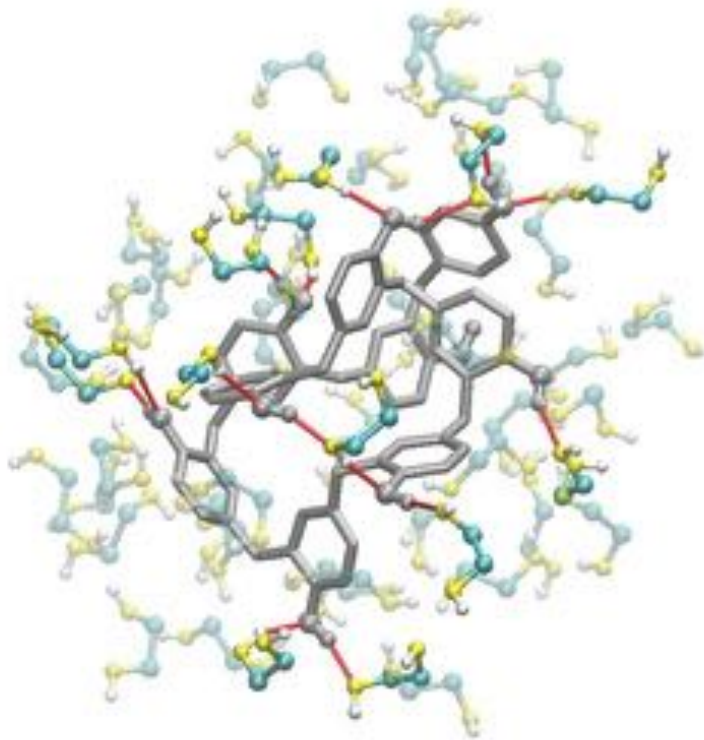
ortho-para novolac



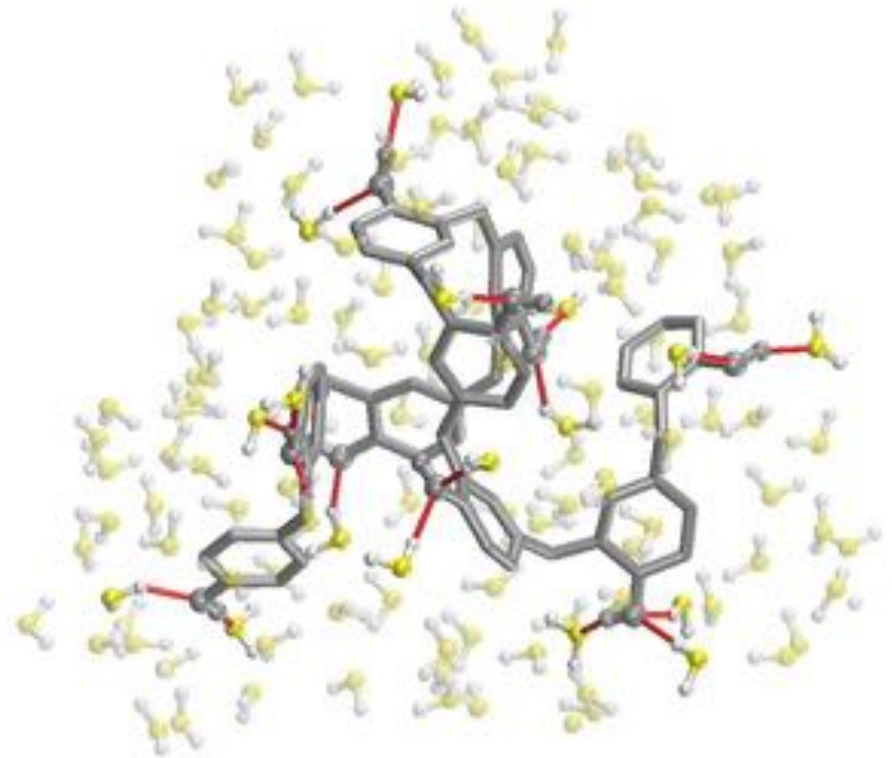
ortho-ortho resole

Polymers more soluble in ethylene glycol
Resole most soluble polymer

Ethylene Glycol



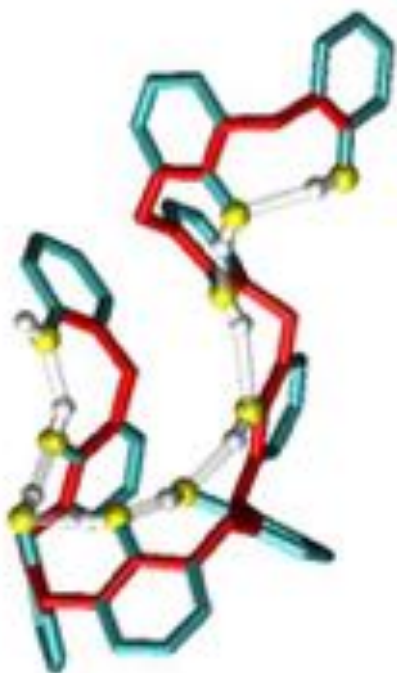
Water



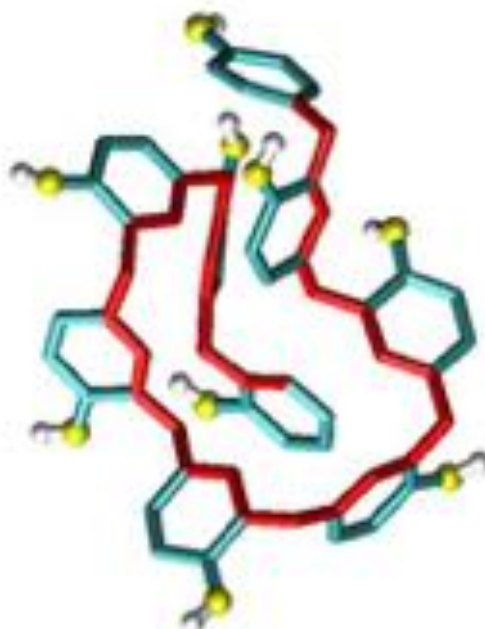
Solvation structure governs properties

polymer = gray; O = yellow; C = green; H = white

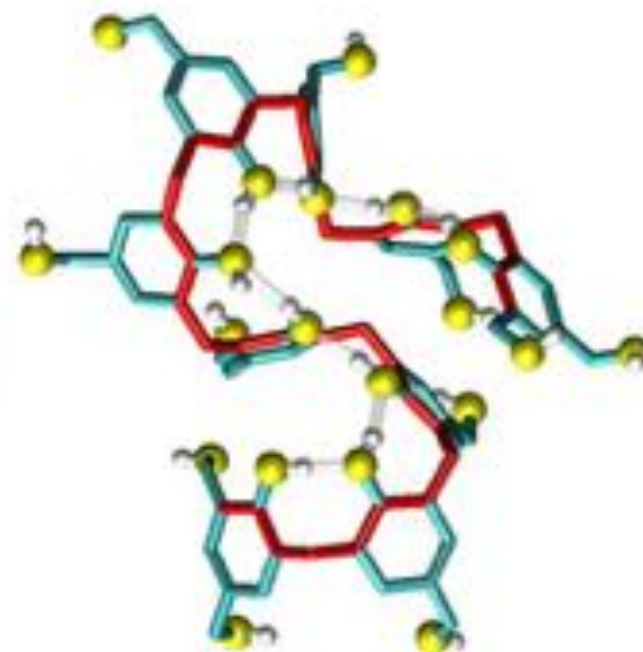
ortho-ortho novolac



ortho-para novolac



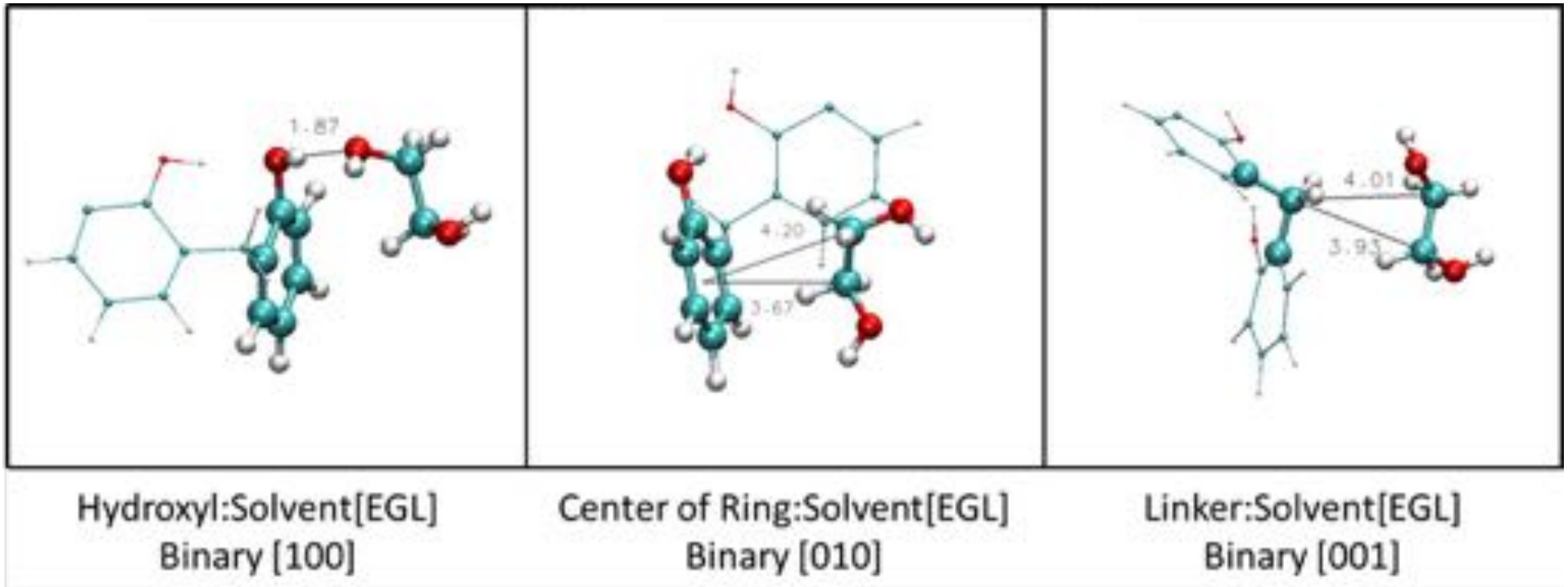
ortho-ortho resole



Self hydrogen bonding in ortho-ortho systems
Ortho-para novolac and ortho-ortho resole have free –OH groups



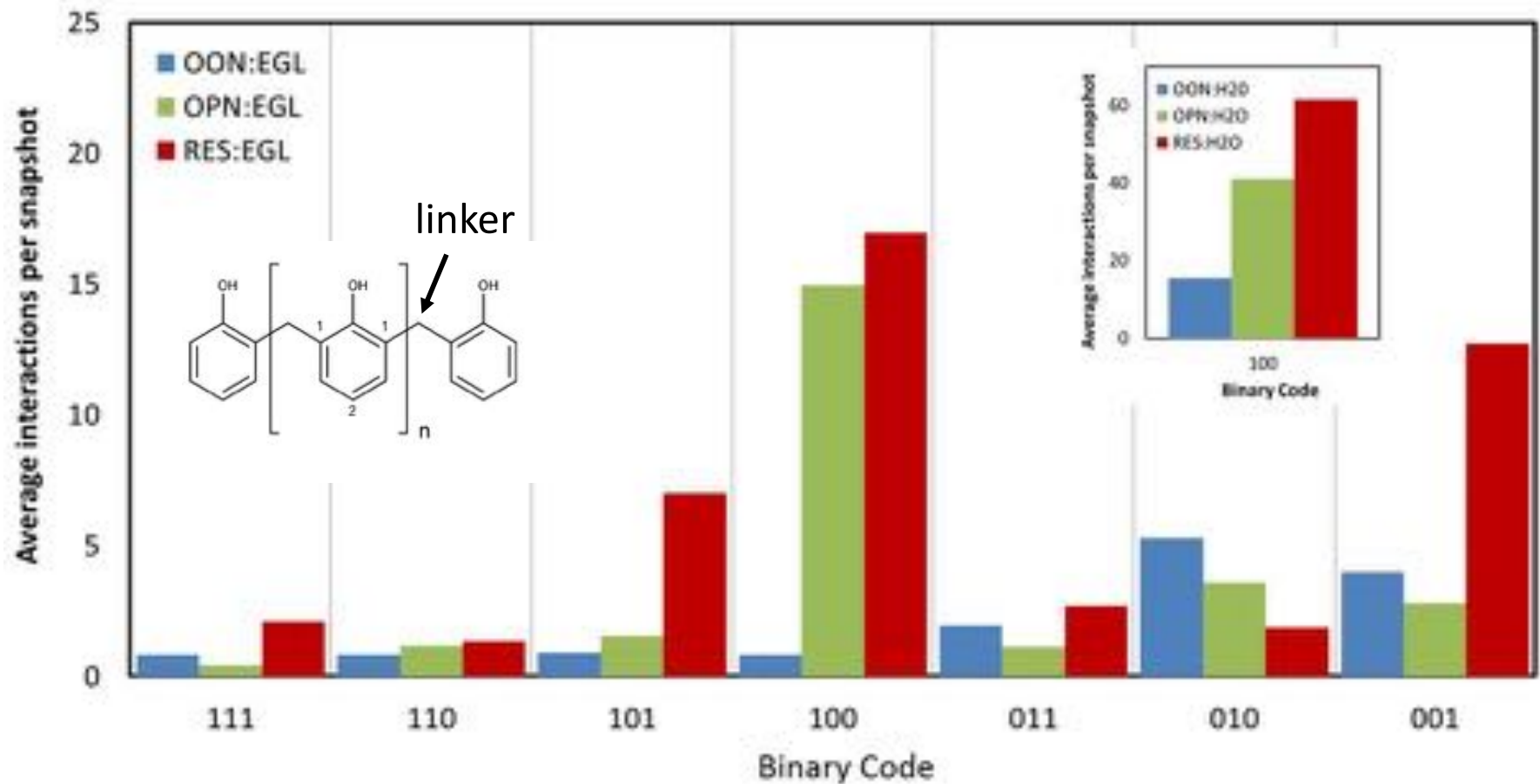
Specific Types of Polymer-Solvent Bonding



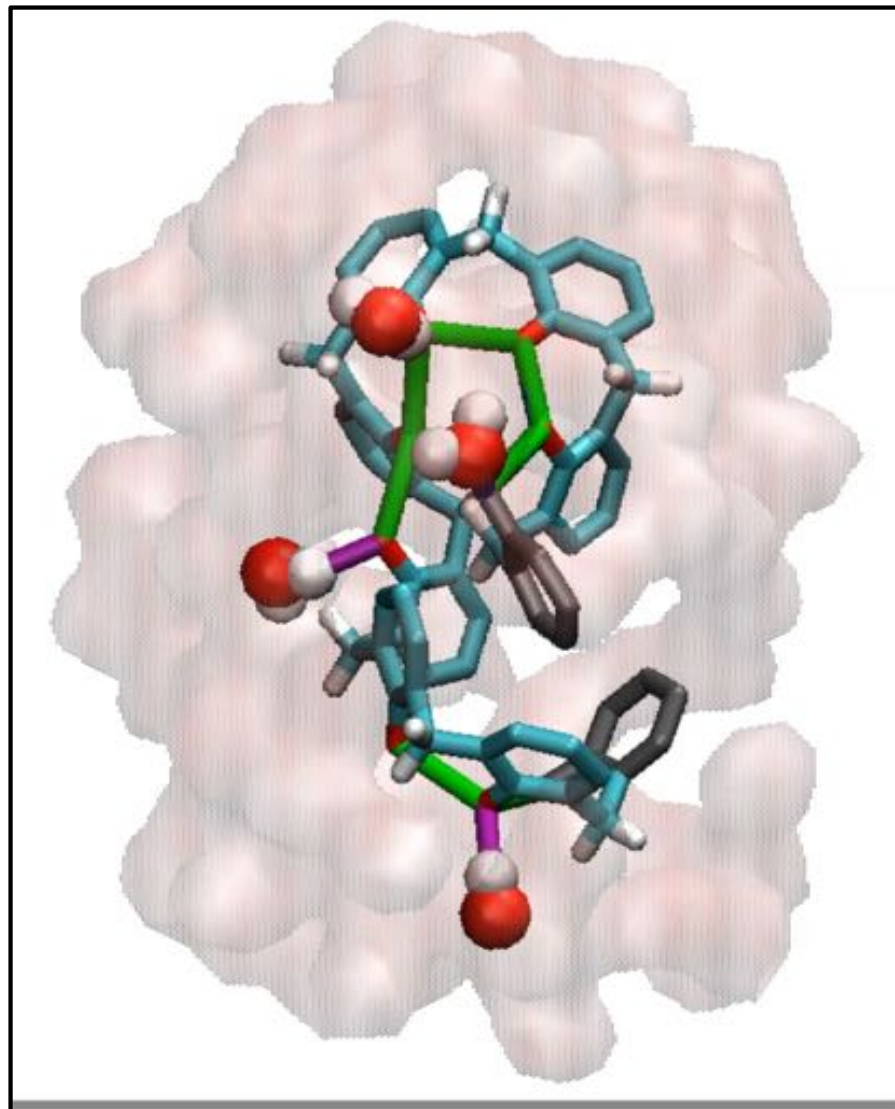
Three primary interactions found in polymer-ethylene glycol solvation; one type in polymer-water solvation



Specific Types of Polymer-Solvent Bonding



Hydrogen bonding and carbon linker interactions dominate

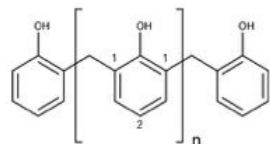


Hydrogen bonding common to both solvents and most prevalent bonding

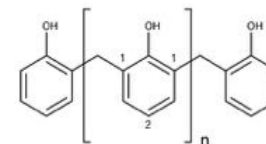
O = red; C = green; H = white



Polymer-Solvent Hydrogen Bonds

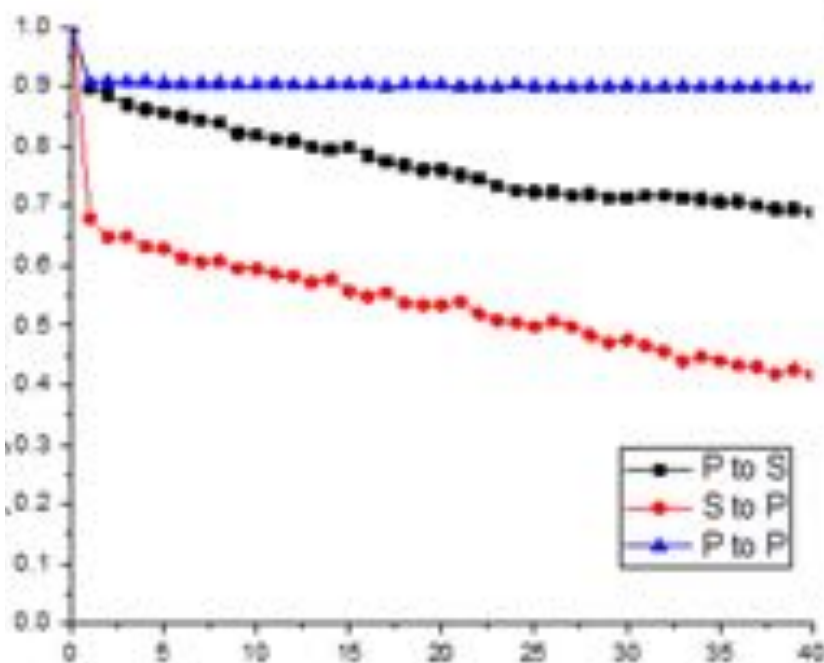


ortho-ortho novolac

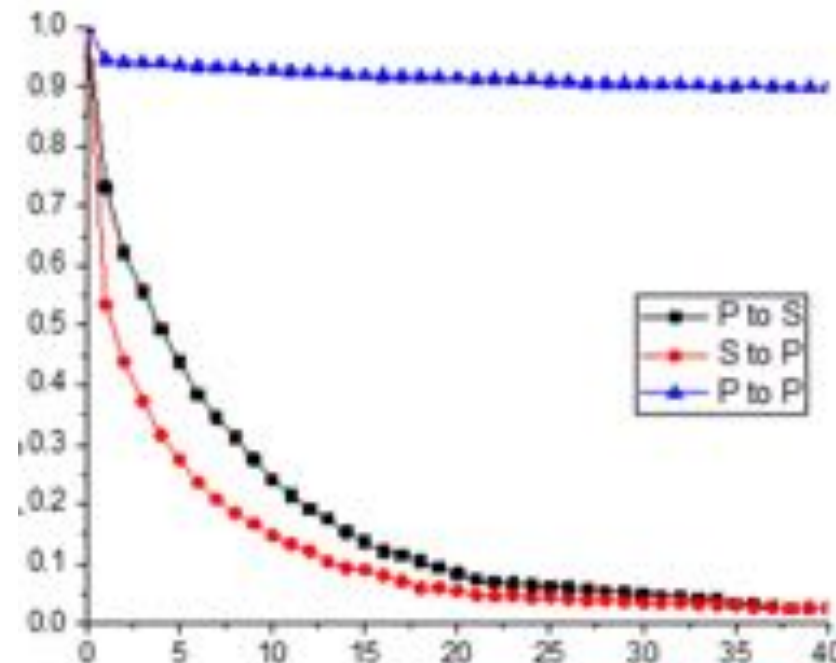


ortho-ortho novolac

Hydrogen Bond Autocorrelation



Simulation Time (ns)



Simulation Time (ns)

Hydrogen bonding more persistent in ethylene glycol



Conclusions



- **OPLS-AA-SEI energetics agree with high accuracy CCSD(T) solvent computations**
- **OPLS-AA-SEI polymer-solvent interactions within a few kcal/mol of MP2/CBS computations**
- **Ethylene glycol more readily solvates the polymers because of more, longer-lived hydrogen bonding than water**
- **Resole is more soluble than the novolac polymers: more hydrogen bonding and hydrophobic-hydrophobic interactions**

Bucholz, et al. *JPCB* **121**, 2839 (2017)

Bauschlicher, et al. *JPCB* **121**, 2852 (2017)



Questions?

Bucholz, et al. *JPCB* **121**, 2839 (2017)

Bauschlicher, et al. *JPCB* **121**, 2852 (2017)