

## Progress Towards Modeling a Rapid Cycle Adsorption Pump for CO<sub>2</sub> Compression

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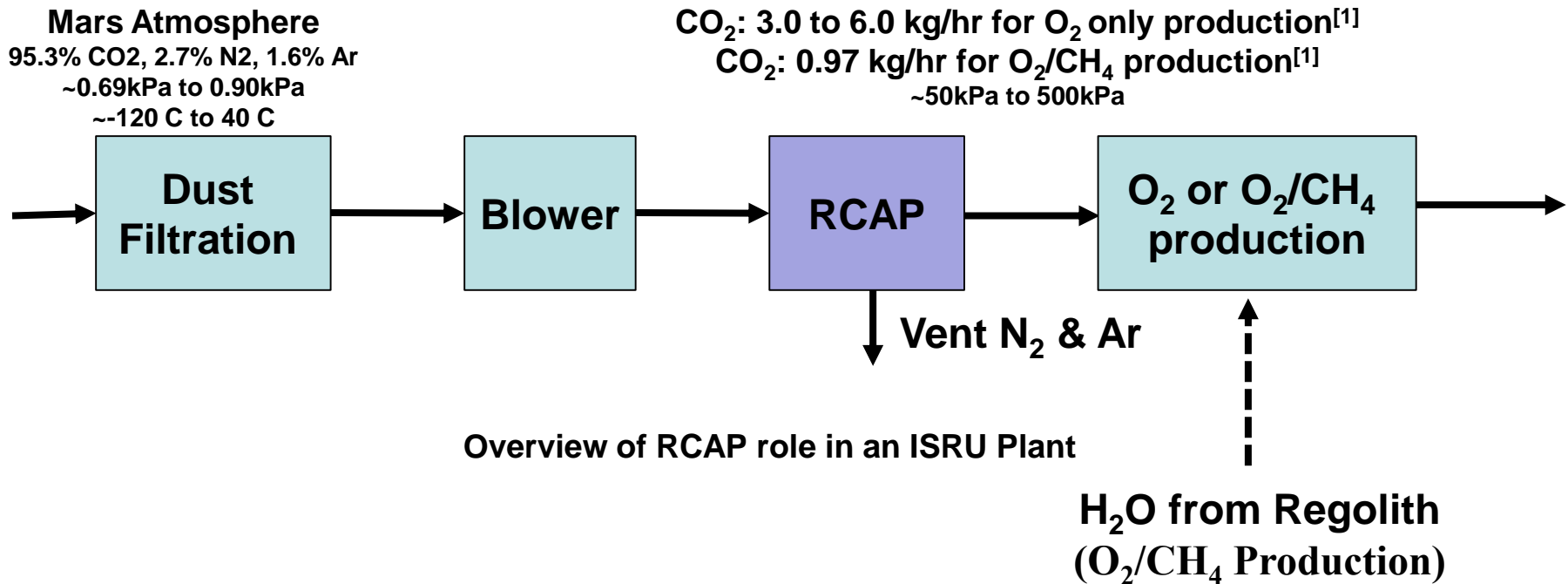


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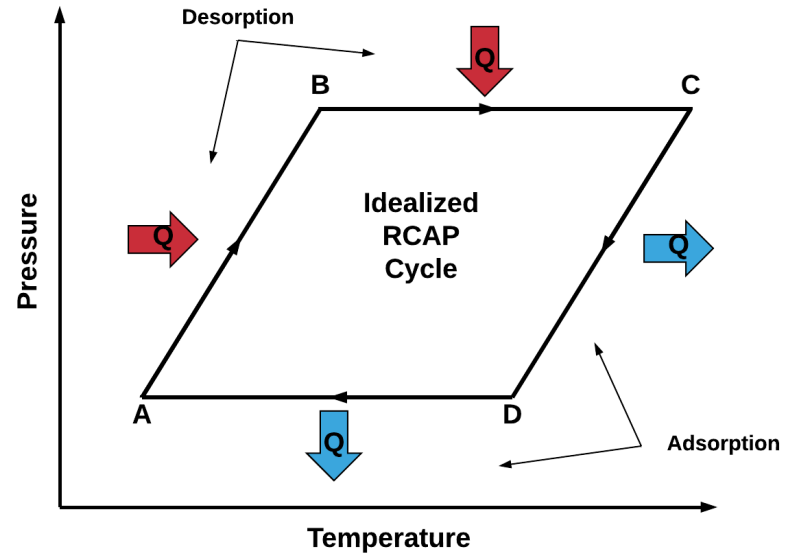
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Houston, TX

- **Rapid Cycle Adsorption Pump (RCAP) Purpose:**
  1. Separate carbon dioxide (CO<sub>2</sub>) from the Mars atmosphere.
  2. Send pressurized CO<sub>2</sub> to downstream processes in a Martian In-Situ Resource Utilization Plant.

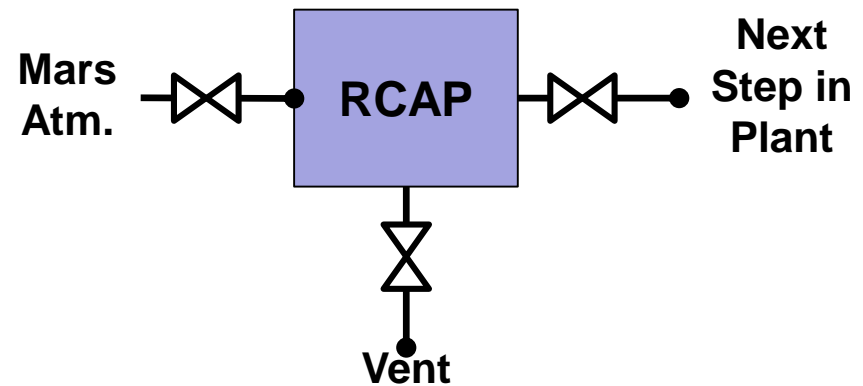


- RCAP fundamentals:
  - **Adsorption:** Cool to adhere  $CO_2$  particles to a material called an adsorbent. More  $CO_2$  adsorbed with *increasing pressure and colder temperatures*
  - **Desorption:** Heat the adsorbent to release the adhered particles.
  - Desorption in a closed volume will generate a pressurized product



Idealized RCAP pump processes operated via temperature swing represented on a Pressure-Temperature diagram.<sup>[3]</sup>

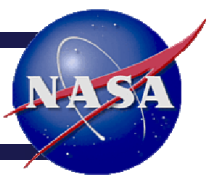
- Idealized 4 step cyclic process:
  - $A \rightarrow B$  : Constant Volume Heating of saturated bed (desorption)
  - $B \rightarrow C$ : Constant Pressure Heating (desorption)
  - $C \rightarrow D$ : Constant Volume Cooling
    - Cooling to restart cycle
  - $D \rightarrow A$ : Constant Pressure Cooling
    - Adsorption Mars atmosphere



RCAP Flow Configuration



# Modeling Approach – Adsorbent Properties



- The adsorbent is a porous material => Treat as a Finite Difference solid
  - Uniform bulk properties need to be determined for the thermal model – take into account the voids in the adsorbent:
    - $k_{bulk}$ , bulk conductivity
    - $\rho_{bulk}$ , bulk density
  - Follow approach in MSFC’s Four Bed Molecular Sieve (4BMS-X) model to calculate<sup>[4]</sup>
    1. How the void fraction ( $\phi = \frac{\text{total volume of voids in bed}}{\text{total bed volume}}$ ) varies with geometry
    2. Conductance at bed/wall interfaces
  - Account for enclosing the adsorbent with walls
  - Assumed free CO<sub>2</sub> in the bed to be static (e.g. no convection, fills voids).

where:

$\phi_{\infty}$  = far field bulk void fraction

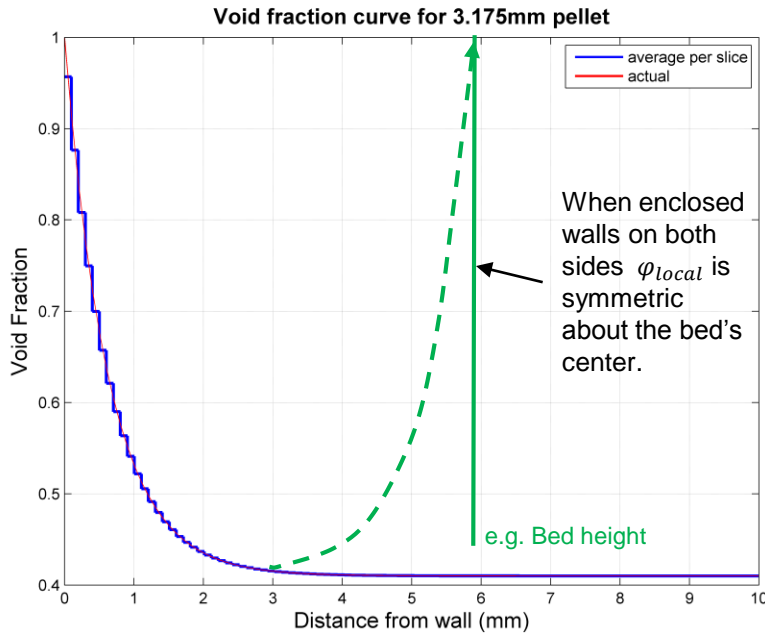
$y$  = distance from wall

$d_p$  = pellet diameter

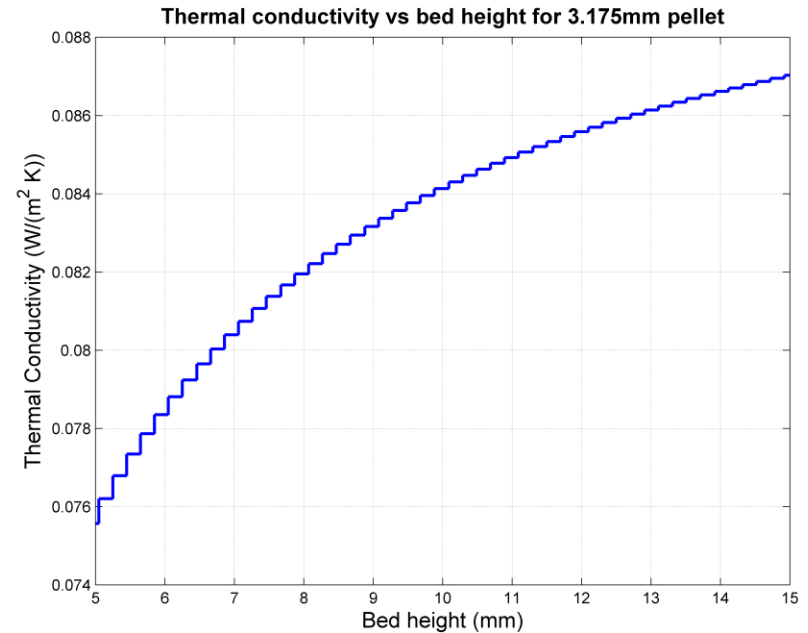
$C = (1 - \phi_{\infty}) / \phi_{\infty}$

$N$  = empirical constant  $\sim 5$ <sup>[5]</sup>

$$\phi_{local} = \phi_{\infty} \left[ 1 + C \exp \left( -N \frac{y}{d_p} \right) \right]$$



**Bulk local void fraction vs. distance from a flat wall.**  
Steps show average value at each 0.1mm slice.



**Bulk thermal conductivity vs bed height, used 0.1mm slices.**

- Bulk porosity:

- Bed is enclosed on both sides → bulk properties are for a specific height

$$\varphi_{bulk} = \frac{1}{y/2} \int_0^{y/2} \varphi_{local} dy$$

- Bulk Density:

- $\rho_{bulk} = \varphi_{bulk} \rho_{CO_2} + (1 - \varphi_{bulk}) \rho_{sorbent}$

- Bulk Conductivity:

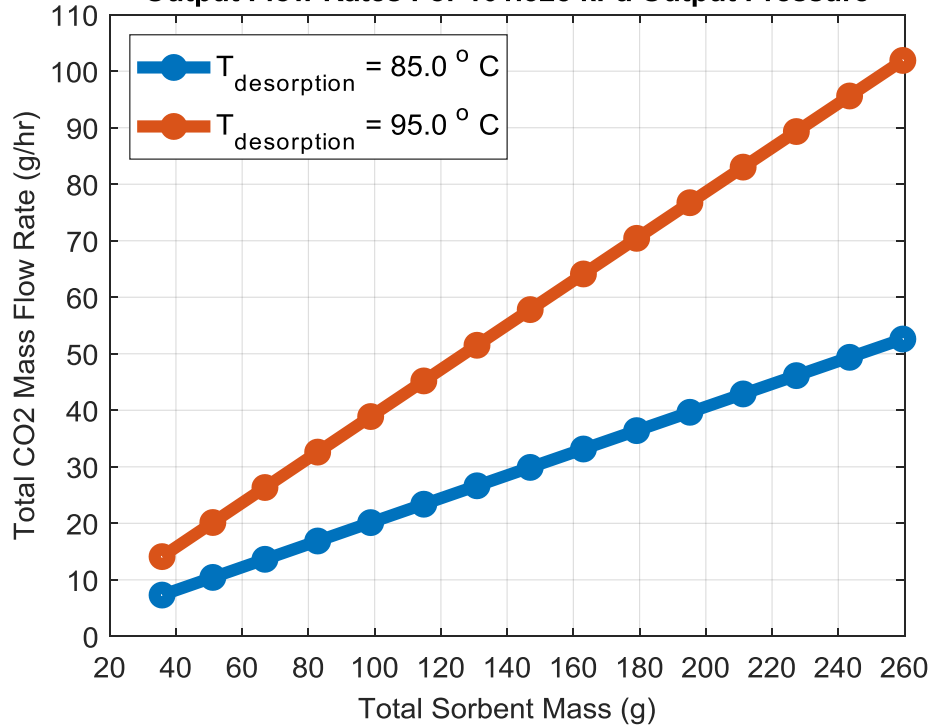
- Use upper bound thermal conductivity,  $k$ , estimate – treats heat flow as parallel through sorbent and gas.<sup>[5]</sup>
- $k_{bulk} = \varphi_{bulk} k_{CO_2} + (1 - \varphi_{bulk}) k_{sorbent}$

- Conductance at bed/wall interface:

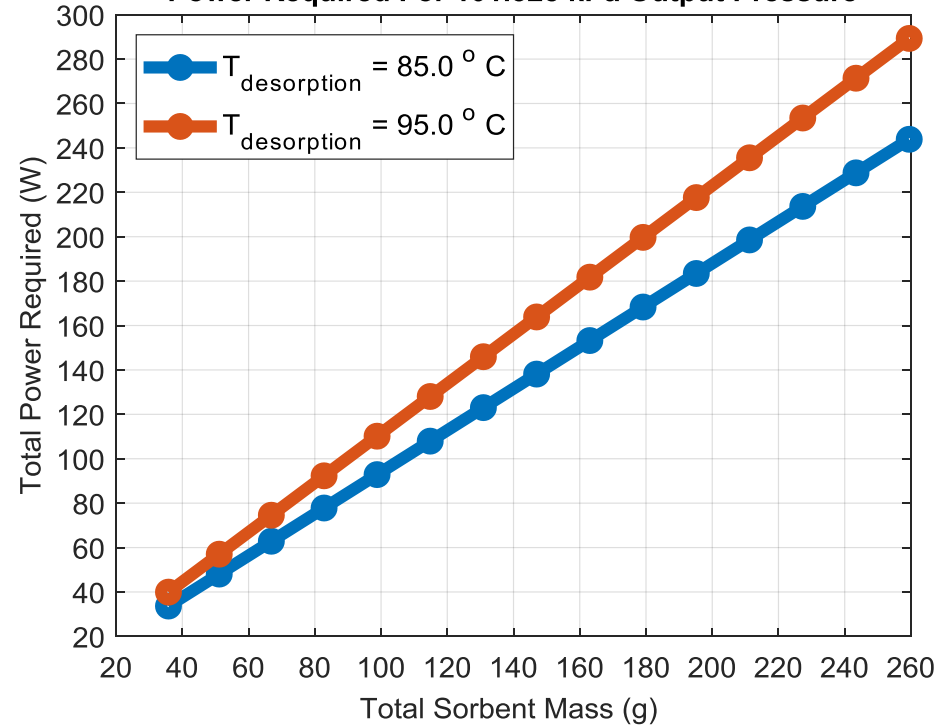
- Use conductance,  $G$ , across slice nearest the wall.<sup>[4]</sup>

- $G_{bed\ to\ flat\ surface} = \frac{k_{bed\ 1st\ slice}}{L_{slice}} \left[ \frac{W}{m^2 K} \right]$

Output Flow Rates For 101.325 kPa Output Pressure

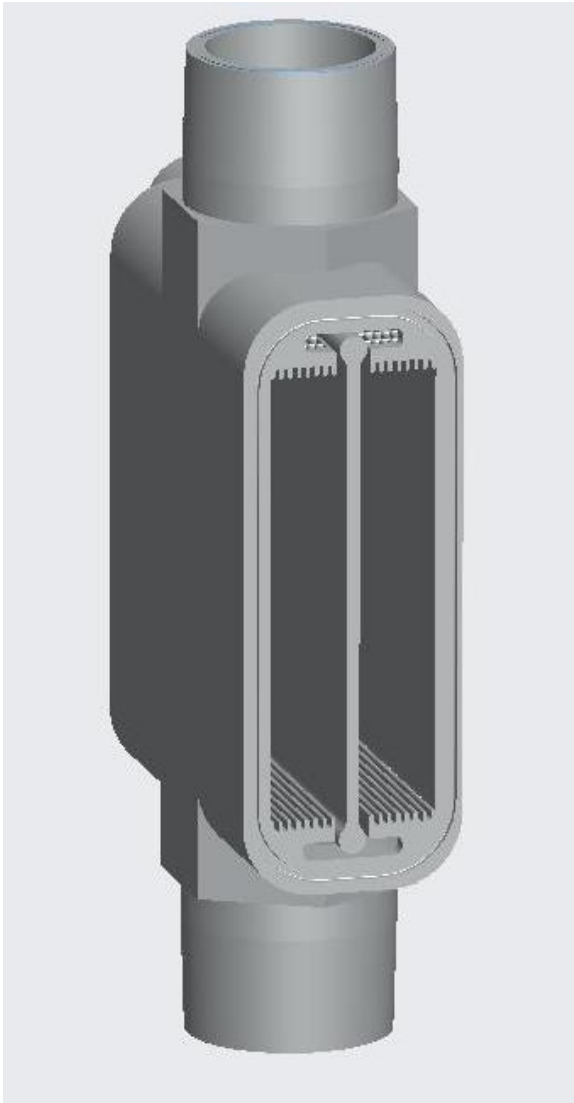


Power Required For 101.325 kPa Output Pressure

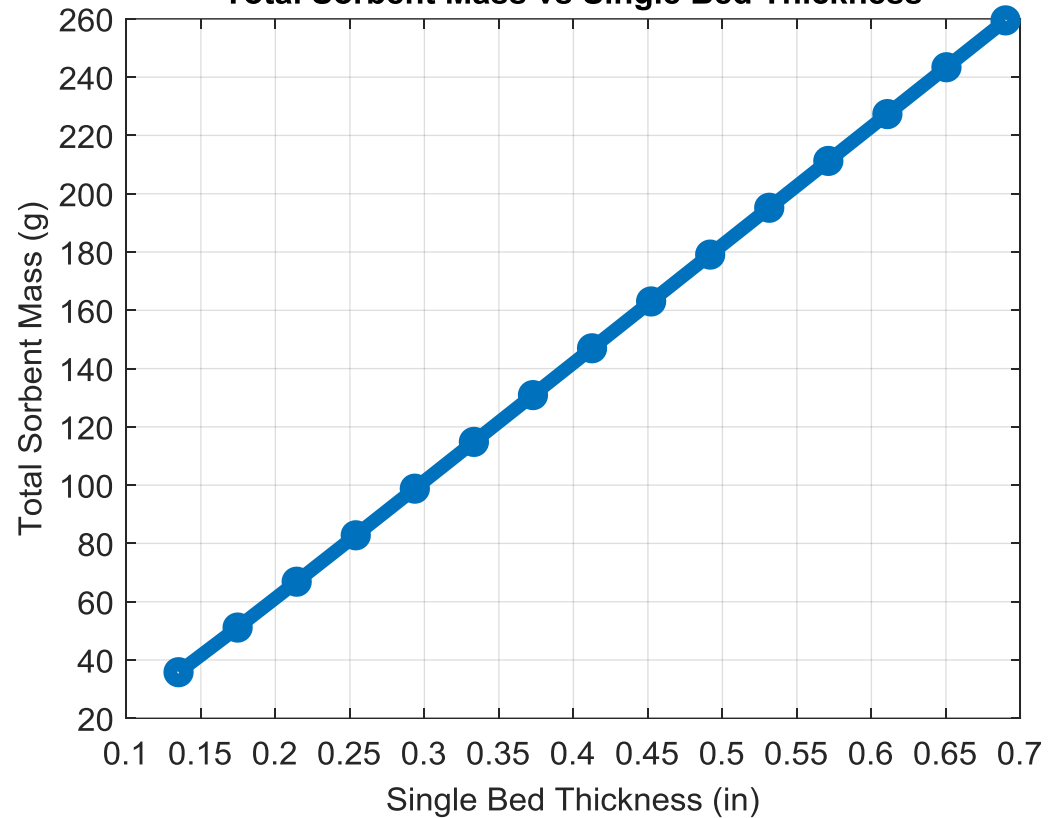


## Single-Stage RCAP Test Analysis parameters:

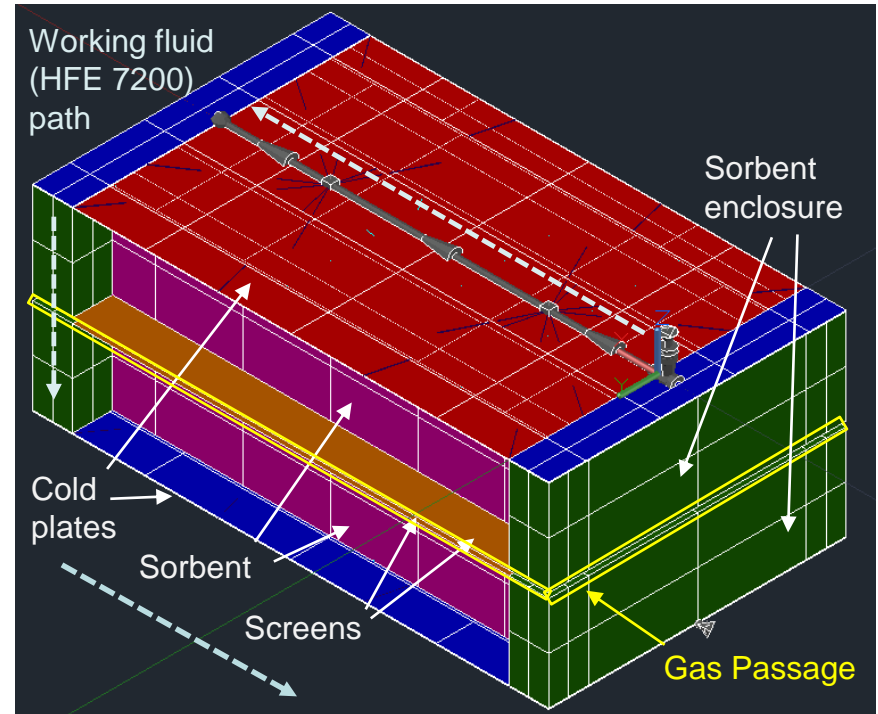
- Adsorption Temperature =  $-20\text{ }^{\circ}\text{C}$
- Inlet Pressure =  $0.933\text{ kPa}$
- Sorbent Heat Capacity [Ref 1]:  $0.96\frac{\text{kJ}}{\text{kg}-\text{K}}$



Total Sorbent Mass vs Single Bed Thickness



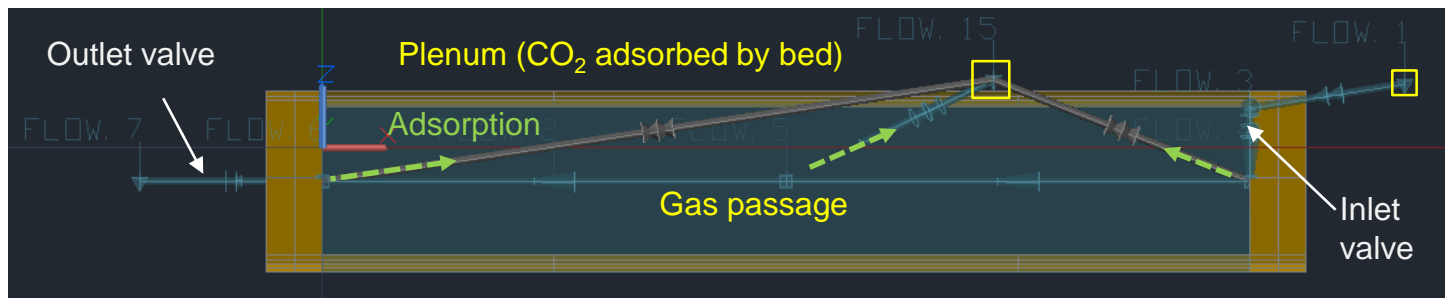
- Purpose
  - Given the heritage PNNL plate based design
  - Help size and choose a material for the NEW RCAP single-plate, single-stage test hardware (e.g. bed height).
- Overview:
  - 1 pump stage = 1 stack
    - 2 Cold plates
    - 2 Sorbent beds (Grace 544 13X)
    - 2 Metal screens
    - 1 Gas passage
  - Increasing the width of the thermal model is used to estimate real world configuration multiple-plates stacked.



Thermal Desktop model layout. Finite difference objects represent solid parts. FloCAD objects (e.g. paths, junctions) represent the process gas to be pressurized and the working fluid.



- Implement Adsorption/Desorption Via Set Mass Flow Rates Of CO<sub>2</sub> To/From A Plenum (Infinite Source/Sink). At each time step set:
  - Plenum's Temperature = Sorbent's Mass Average Temperature
  - Plenum's Pressure = Gas Passage's Average Pressure
- Toth Isotherm fit predicts adsorption / desorption
  - Adsorption Bed Loading at Equilibrium (as a Function of Temperature and Pressure)
  - Isothermic Enthalpy of Adsorption (Derived from Bed Loading)
- Linear Driving Force (LDF) Approximation Models:
  - Function of Toth loading isotherm
  - Lumped Constant from Experimental Results (provided by MSFC, Knox, J)
    - Included convection and mass transfer rates through the bed, Adsorption Kinetics
  - Sorbent Heating/Cooling Rates (for heat of adsorption / desorption)

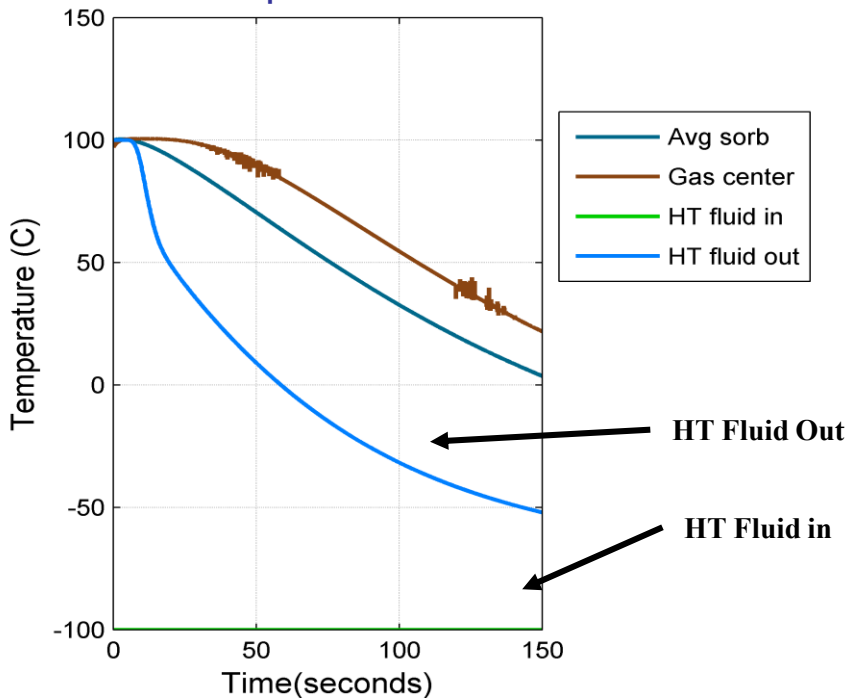


**Gas passage. Set flows are used to implement 1) inlet flow rate from Mars atmosphere, and 2) flow rates from of CO<sub>2</sub> adsorbing into or desorbing from the sorbent bed.**

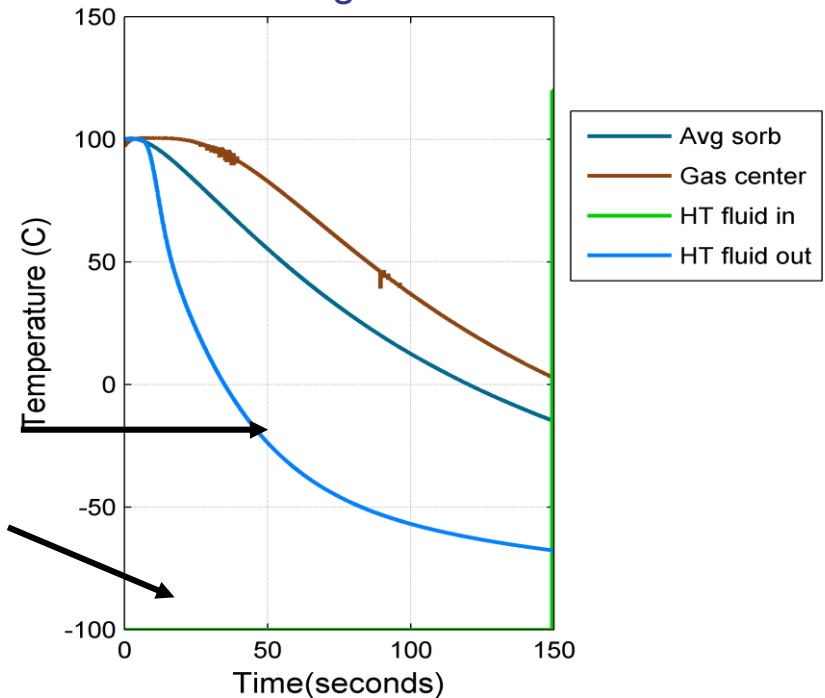
- Comparison Case:

- Cooled 1.72kg sorbent (ISRU Mission) from 100C in a sealed bed. (Thermal Effects Only)
- GRCoP-84 has a much higher thermal conductivity...
- AlSi10mg version had a much lower thermal mass and cooled faster.
- Use of AlSi10mg instead of GRCoP-84 resulted in **\$30,000 decrease** in 3D printing manufacturing costs

GRCoP-84 results:



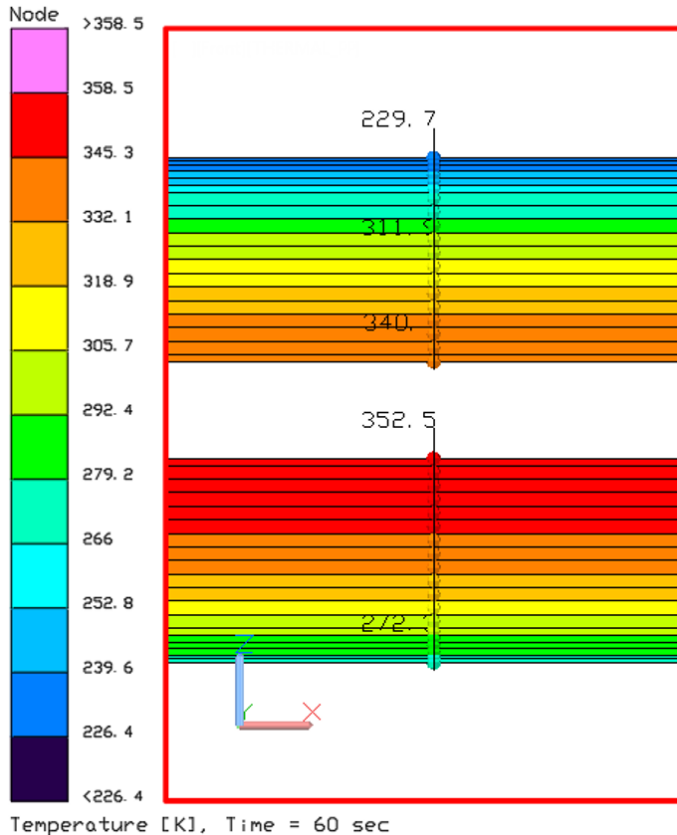
AlSi10mg results:



Cooling time with enclosure and cold plate set to GRCoP-84. Cold plate wall could be 0.75mm thin.

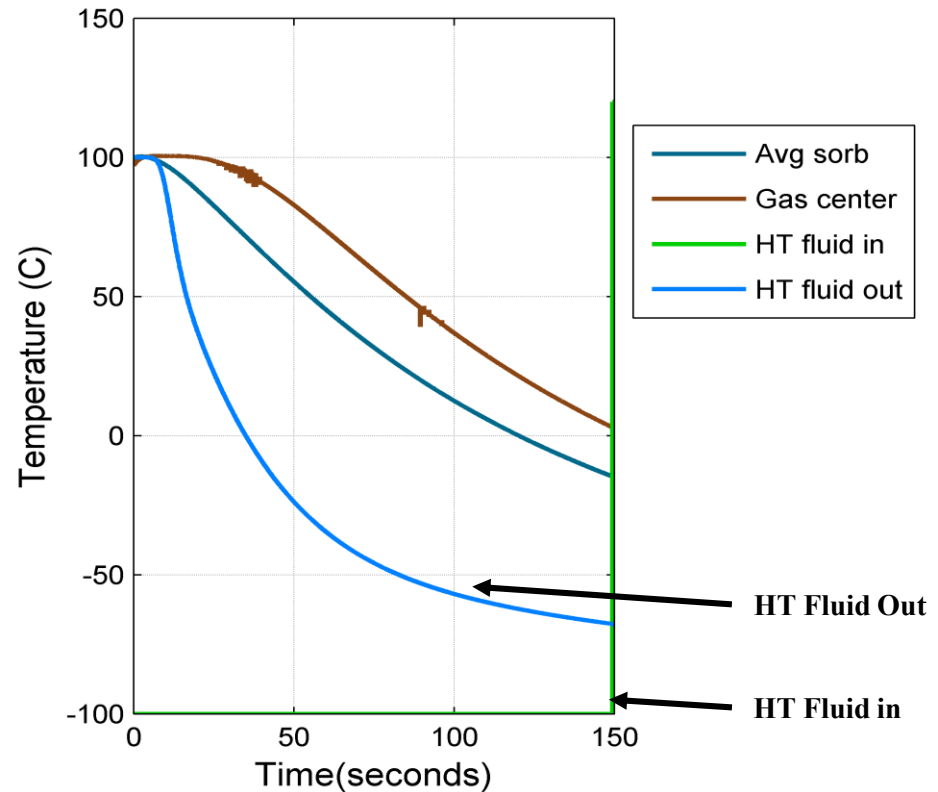
Cooling time with enclosure and cold plate set to AlSi10mg. Cold plate wall could be 0.55mm thin.

- Comparison Case:
  - Cooled 1.72kg sorbent from 100C in a sealed bed.
  - Observed large temperature gradients ( $\sim \Delta T$  110C) in the 5.23mm thick sorbent beds.



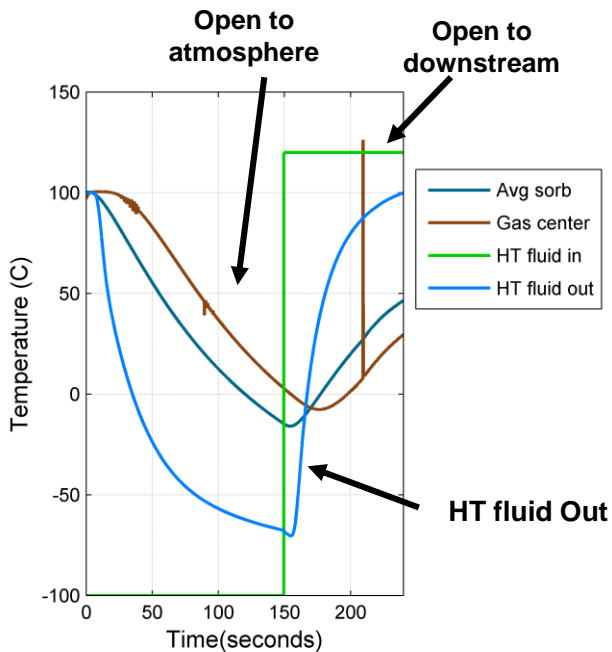
Temperature gradient in sorbent at 60s into cycle. At this point the gas passage center was at 74 C (347 K). The heat transfer fluid warmed from -100 C (173 K) to -34 C (239 K).

AlSi10mg results:

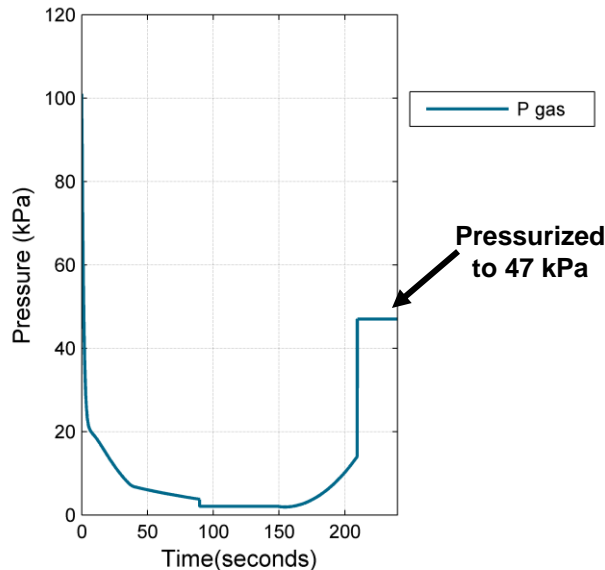


Cooling time with enclosure and cold plate set to AlSi10mg. Cold plate wall could be 0.55mm thin.

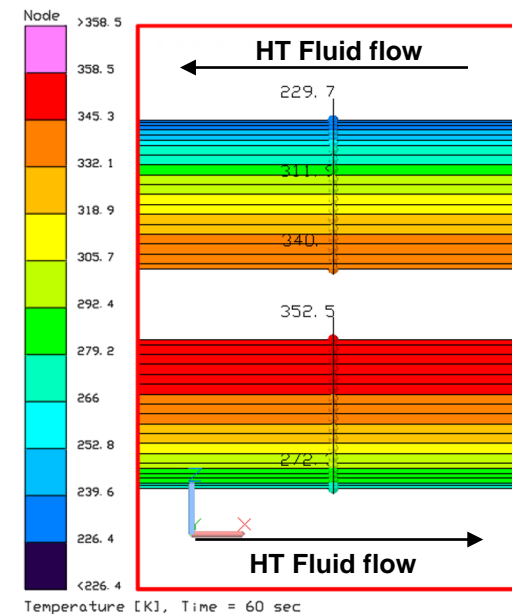
- Case:
  - 1.72kg sorbent heated/cooled with forced liquid HFE 7200 at 0.063kg/s
    - 1<sup>st</sup> stage of a 5.16kg pump.[3]
    - HFE 7200 mass flow rate approximately scale from on Linne (2013) experiment.[6]



Temperatures throughout one RCAP cycle



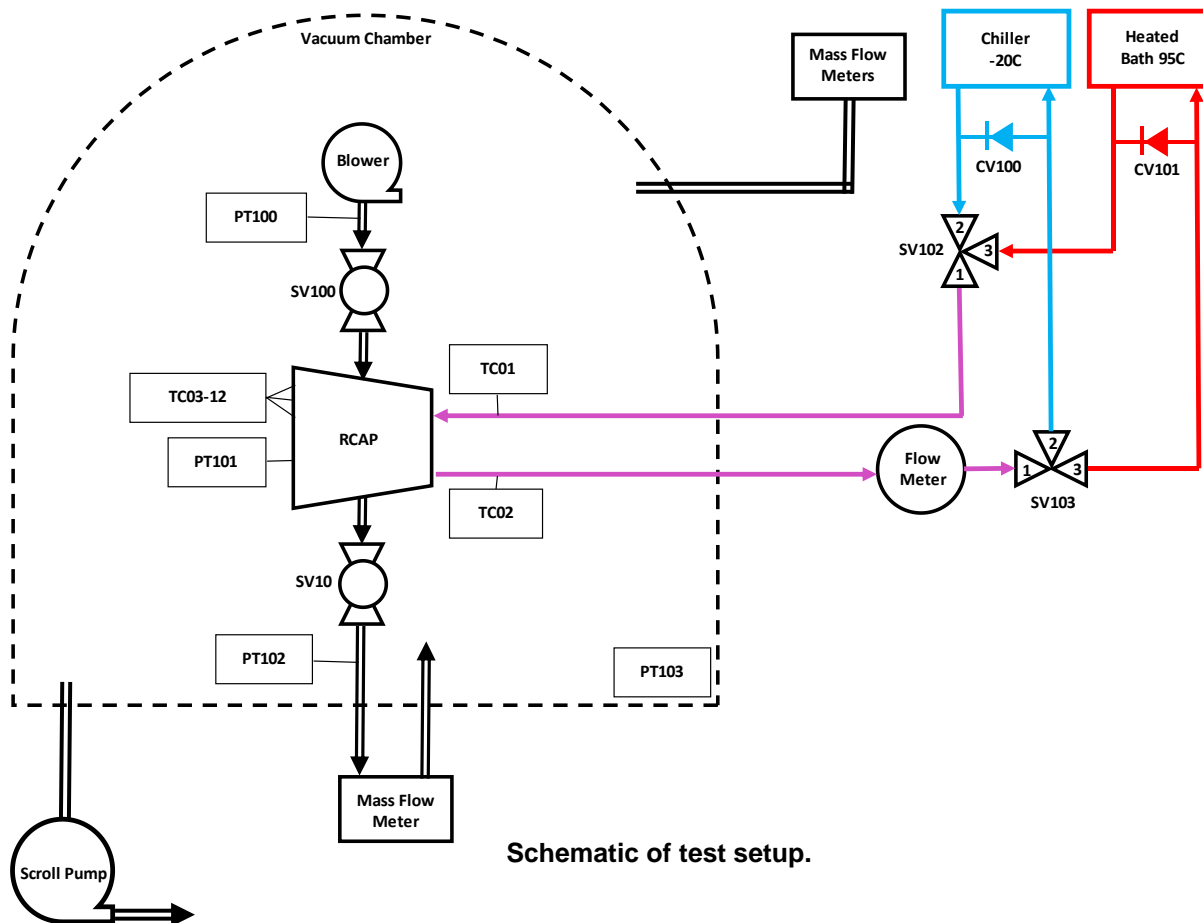
Average gas passage pressure throughout one RCAP cycle



Temperature gradient in sorbent at 60 s into cycle. At this point the gas passage center was at 74 C (347 K). The heat transfer fluid warmed from -100 C (173 K) to -34C (239 K).

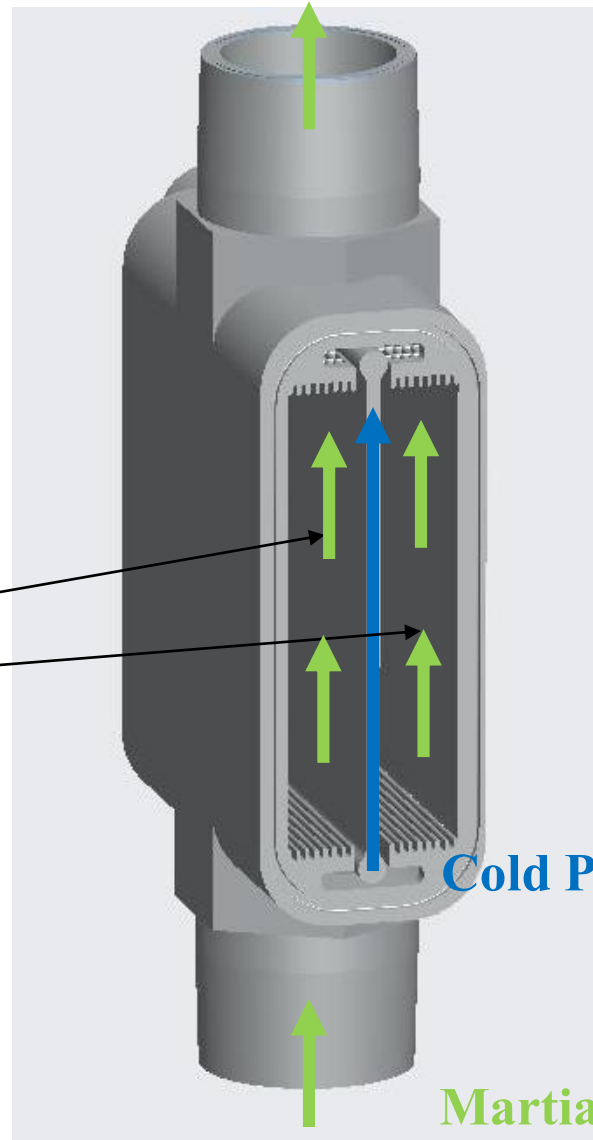
- Purpose

- Single-Plate, Single-Stage RCAP test article – scaled down RCAP.
  - Targeting 1/10<sup>th</sup> CO<sub>2</sub> production for full scale ISRU system. (O<sub>2</sub>/CH<sub>4</sub> Production)
  - Will be compared against testing results (Pressurization, Bed Temperature)



Schematic of test setup.

# Single-Plate, Single-Stage RCAP

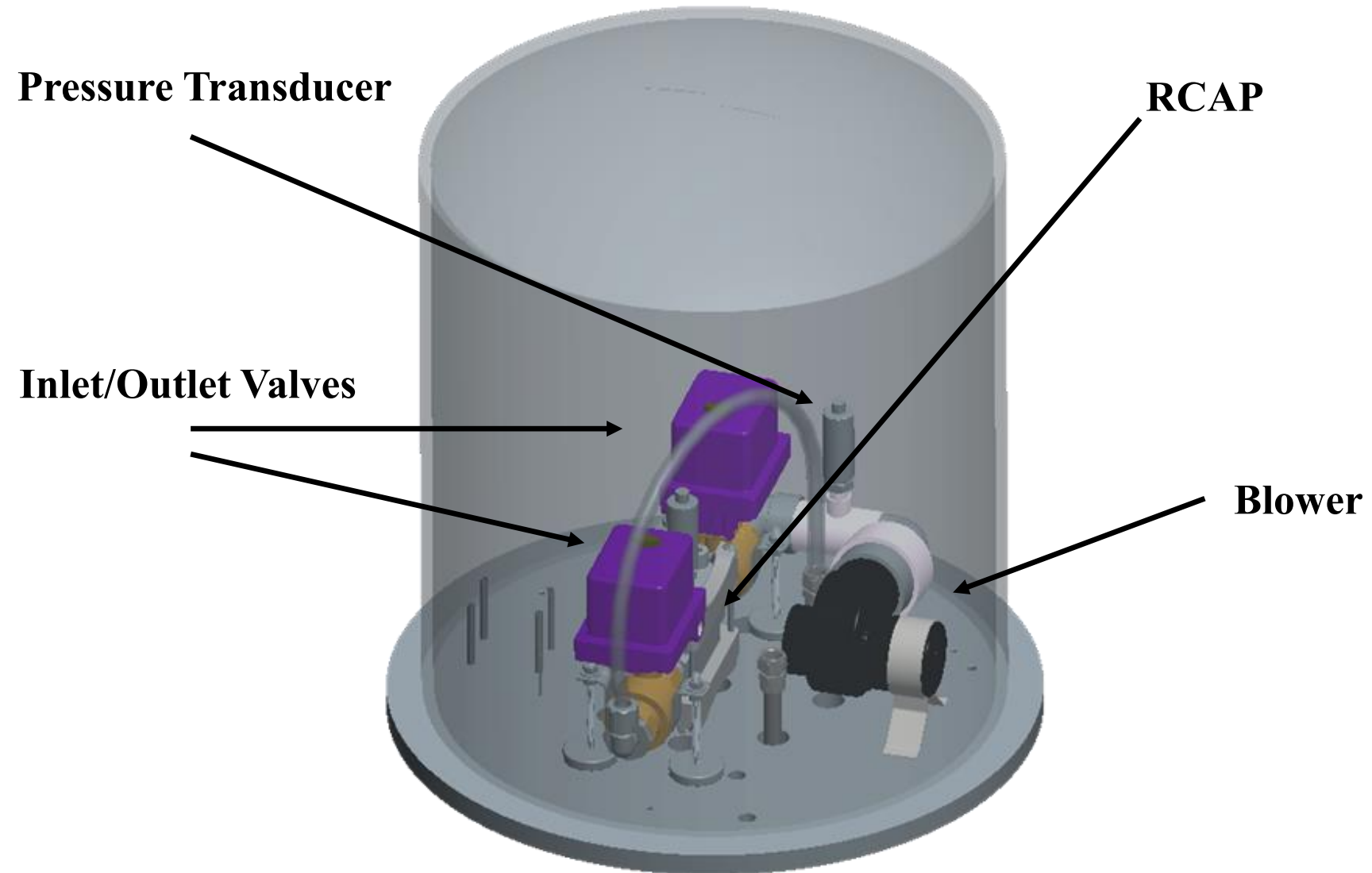


**Porous Bed**

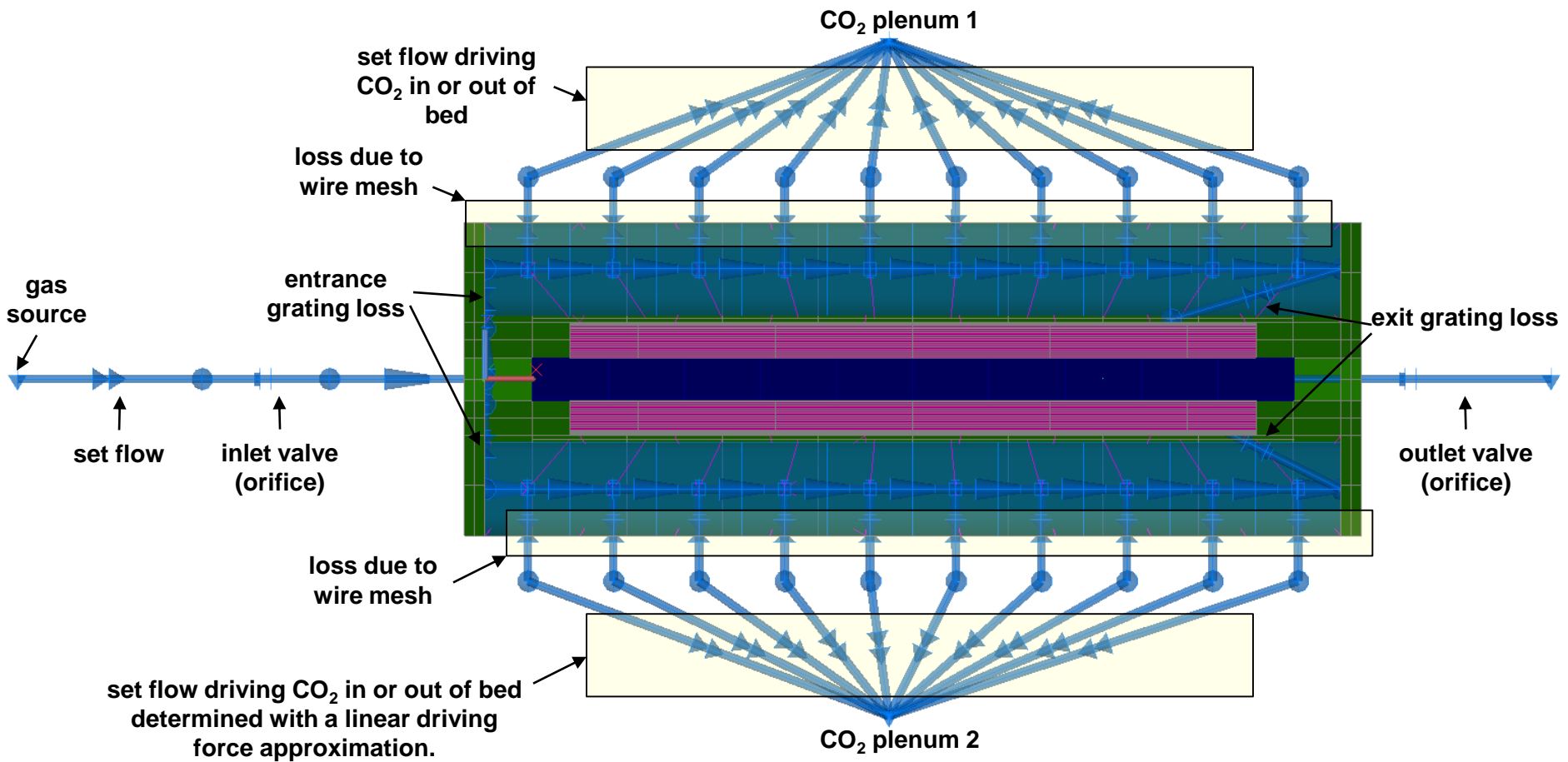
**Cold Plate, Heat transfer Fluid**

**Martian Atmosphere**

## RCAP in the Mars Air Chamber Simulator



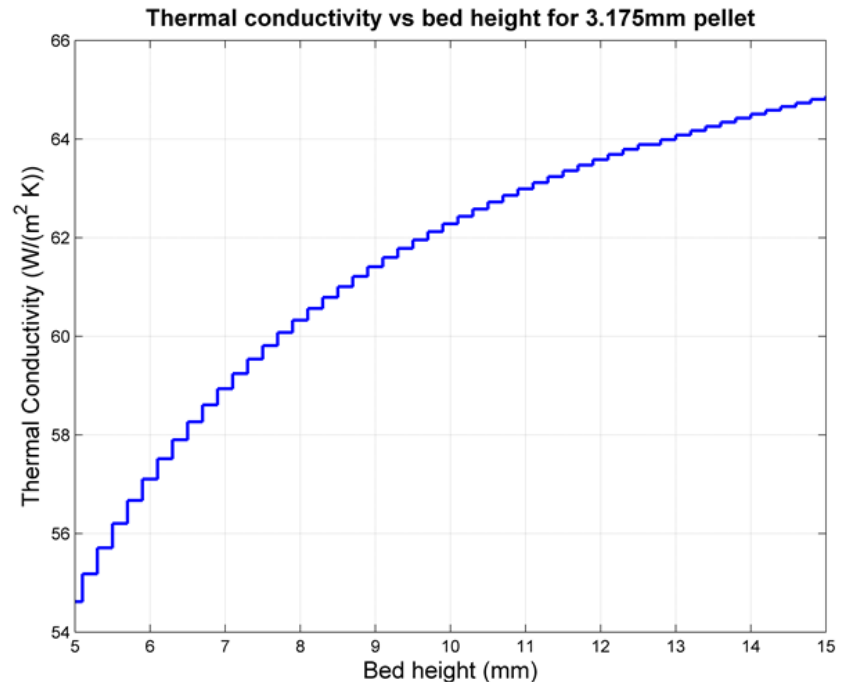
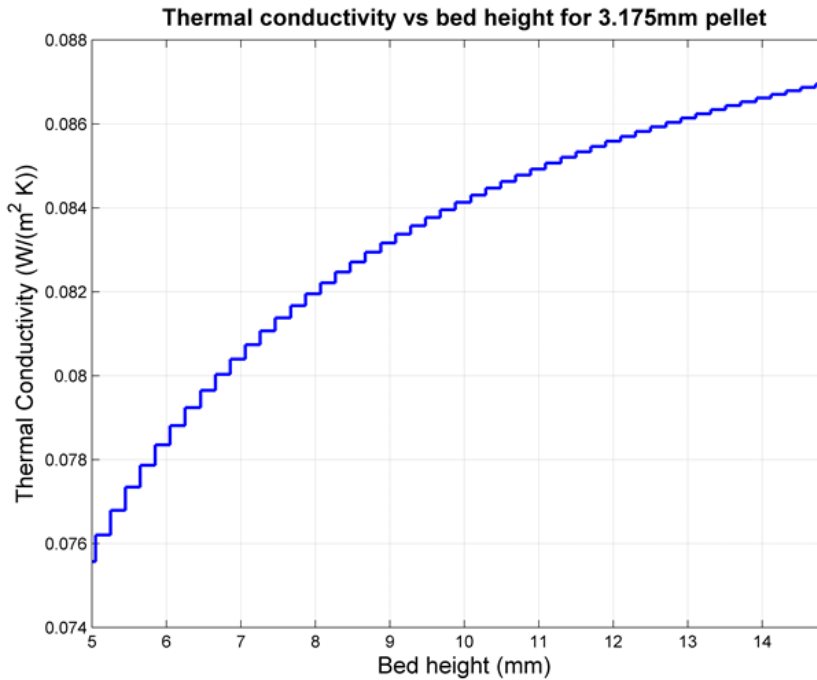
- Changes from Plate Based Design:
  - Updated geometry, increased divisions along gas flow path, added loss dues to wire mesh, update heating/cooling loop to reflect test setup.





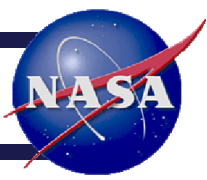
- Average Bed Temperature Takes Hours To Reach The - 20 C Average Bed Temperature
  - The insulating properties of Zeolite 13X make are not making it easy to transfer heat (via conduction) to layers more than ~2 mm from the cold plate
  - Cold plate mini-channel heat transfer performance is good, close layers respond within ~5 minutes.
  - Conduction through the porous media is the problem.
- Pressurization is dramatically lower (8kPa) versus Plate Based results (40KPa)
  - The Model is Sensitive to Voids Outside the Porous Media Bed.
  - While plastic fillers can be used in the actual testing model, we need to check the total model volume with respect to the adsorbent loading.

## Comparison of Porous Bed Thermal Conductivity Zeolite only (left) Compared against 50 % Aluminum Shot, 50 % Zeolite 13 X Adsorbent (By Volume) ~750X Improvement in Thermal Conductivity

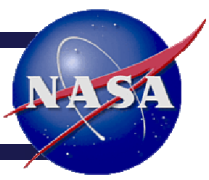




# Single-Plate, Single Stage Current Status

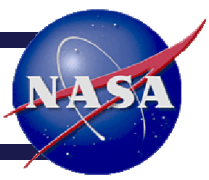


- Low Density Of The Martian Atmosphere Is Causing Compressibility Problems Within Thermal Desktop.
  - Mars Atmospheric Pressure is  $\sim 1/100$  Of STP Conditions
  - Valves Opening And Closing Specifically Cause Numerical Problems
    - In Progress Of Adding Rates To Opening And Closing Of RCAP Inlet And Outlet Valves (Rather than bang bang)
    - Is There A Way To Add Dampening Similar To CFD Simple Adsorption Layer Boundary Conditions?
      - Based On Sponge Effect (Layers of Source Terms Originally Derived from Porous Media CFD)
      - Not Method Of Characteristics



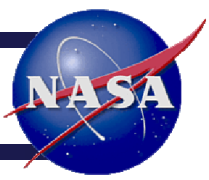
# Closing Remarks

- Thermal Desktop model for Plate-Based design (PNNL Heritage) performed more or less as expected.
- The Single-Plate, Single-Stage Thermal Desktop model did not achieve expected pressurization
- Adding Aluminum shot to the porous media bed appears to be a good idea.
- We may need a larger Hot / Cold Heat Transfer Fluid temperature swing for the RCAP
  - Model in current state can help us answer this question.
- Single-Plate, Single Stage experiment was specifically designed to measure the porous media bed Temperature
  - Results will be used to adjust constants in the 4BMX model.



# References

1. National Aeronautics and Space Administration, “NextSTEP-2 BAA Appendix D: In Situ Resource Utilization (ISRU) Technology - NNH16ZCQ001K-ISRU,” *FedBizOpps*. FedBizOpps. [Accessed: 30-Jul-2018].  
<https://www.fbo.gov/index?s=opportunity&mode=form&tab=core&id=34eb0ba219ff3a8d97c9c4b2c9302bf1>
2. R. T. Yang, *Gas Separation by Adsorption Processes*. Kent: Elsevier Science, 01.
3. Hasseeb, Hashmatullah and Iannetti, Anthony, “A System Level Mass and Energy Calculation for a Temperature Swing Adsorption Pump used for In-Situ Resource Utilization (ISRU) on Mars,” presented at the TFAWS, 2017.
4. R. Schunk, W. Peters, and J. Thomas, “Four Bed Molecular Sieve – Exploration (4BMS-X) Virtual Heater Design and Optimization,” Jul. 2017.
5. D. Nield and A. Bejan, *Convection in Porous Media*. New York, NY: Springer New York, 2006.
6. D. Linne *et al.*, “Demonstration of Critical Systems for Propellant Production on Mars for Science and Exploration Missions,” 2013.

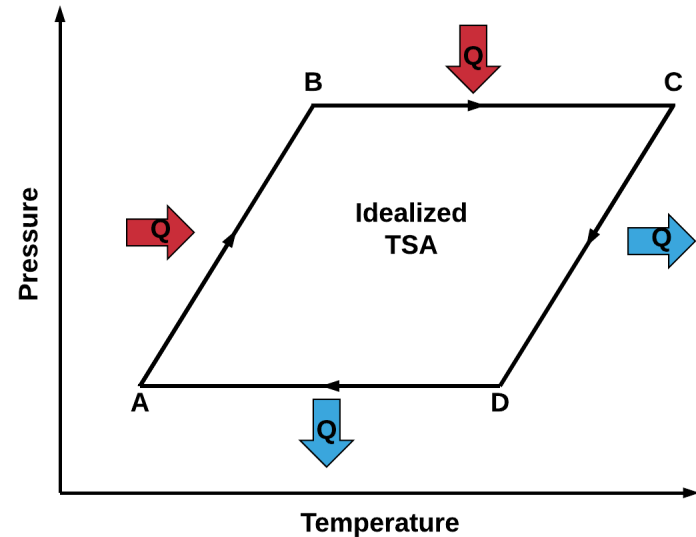


# Backup

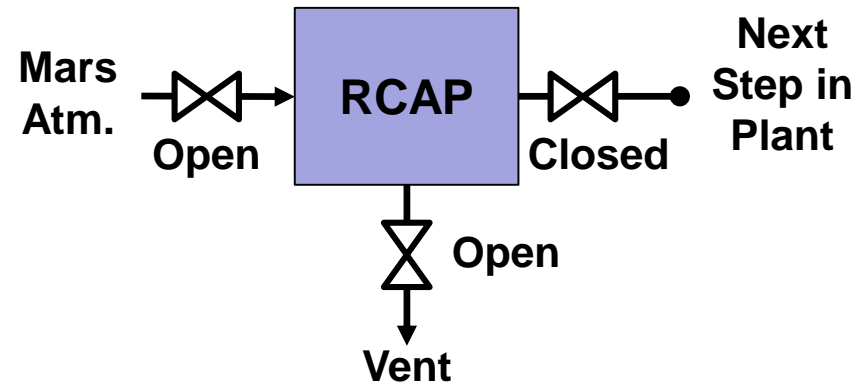
- 4 step cyclic process:

## D → A: Isobaric Adsorption

- Start with cooled sorbent bed.
- Upstream valves open.
- Blower directs flow through the bed. Non-CO<sub>2</sub> gasses vent.
- Adsorption occurs (CO<sub>2</sub> molecules bond to sorbent, e.g. zeolite) releasing heat.<sup>[2]</sup>
- Sorbent bed is externally cooled to the maintain a constant pressure.
- State A's temperature is called the adsorption temperature.

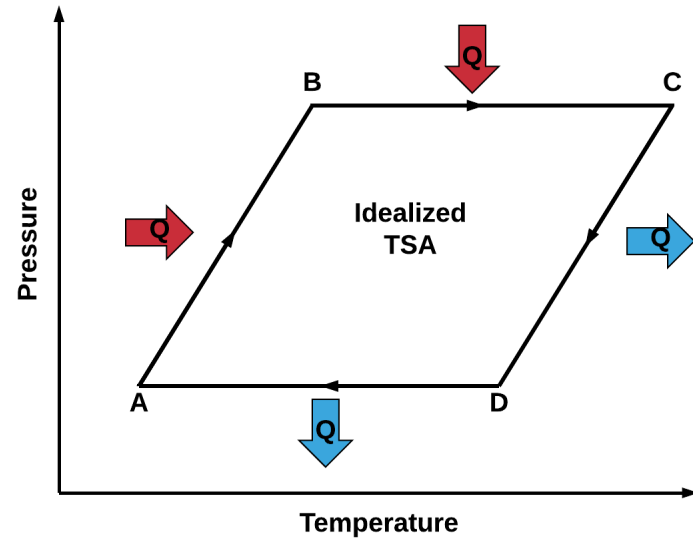


Idealized TSA pump processes represented on a Pressure-Temperature diagram.<sup>[3]</sup>

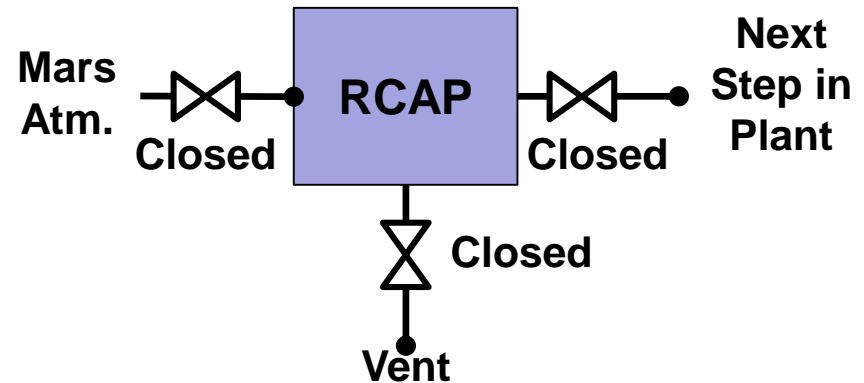


RCAP flow configuration during D to A transition.

- 4 step cyclic process:
- A → B: Const. Volume Compression
  - Start at adsorption temperature.
  - All system valves close.
  - Sorbent bed is externally heated until the target pressure is reached.
  - Desorption occurs:
    - “Pulls” heat from the system
    - And pressurizing the bed’s enclosure.



Idealized TSA pump processes represented on a Pressure-Temperature diagram.<sup>[3]</sup>



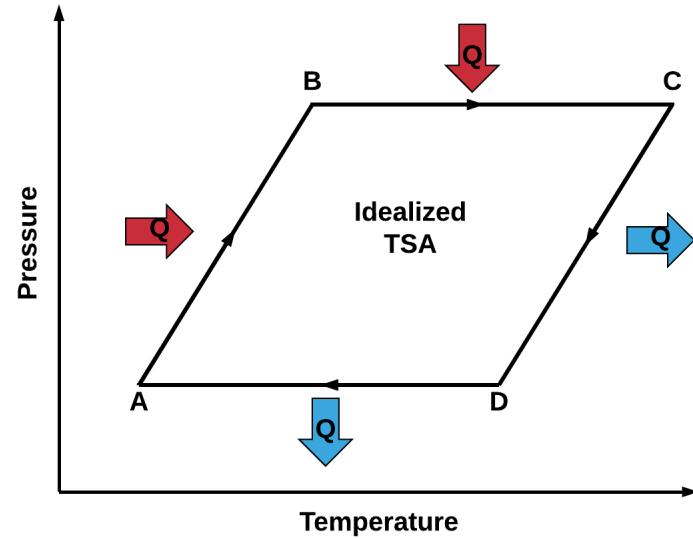
RCAP flow configuration during A to B transition.



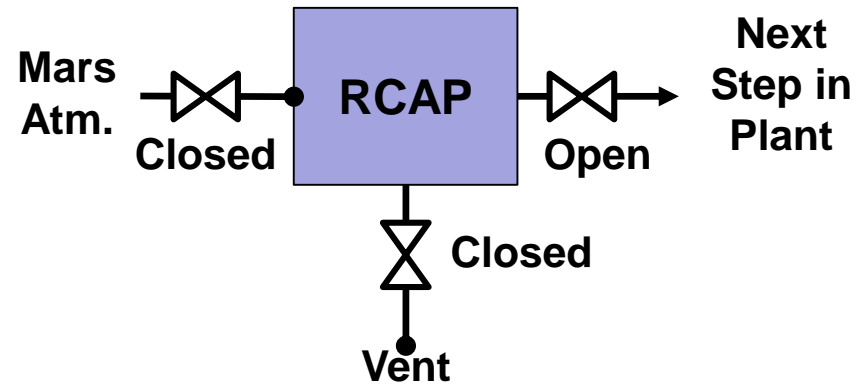
- 4 step cyclic process:

## B → C: Isobaric Desorption

- Downstream valves are opened to allow pressurized CO<sub>2</sub> to exit the bed.
- Desorption continues.
- Bed is heated to maintain constant pressure.
  - Desorption is endothermic.
  - Hotter beds hold less gas.
- State C's temperature is called the desorption temperature.

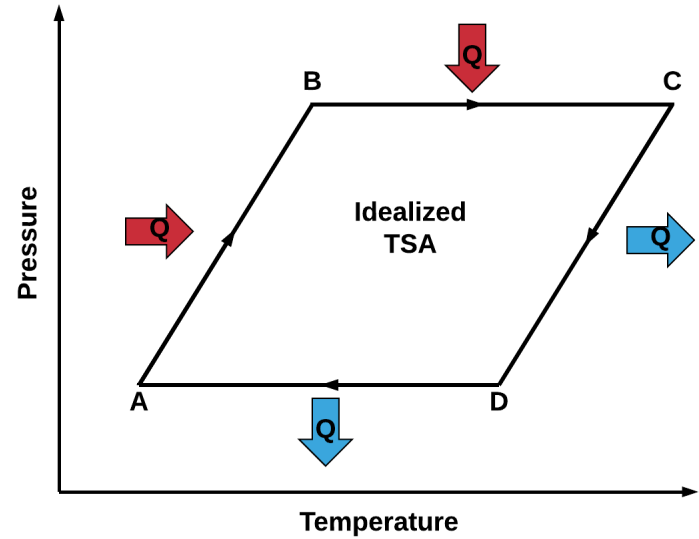


Idealized TSA pump processes represented on a Pressure-Temperature diagram.<sup>[3]</sup>

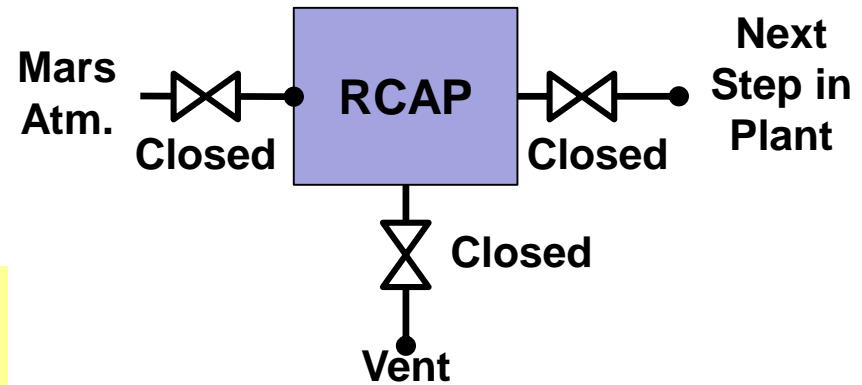


RCAP flow configuration during B to C transition.

- 4 step cyclic process:
- C → D: Const. Volume Cooling
  - Start at target desorption temperature.
  - Starts at max desorption temperature, and max bed pressure.
  - System valves are closed.
  - Bed is cooled in preparation for state D.
  - Remaining free CO<sub>2</sub> in the enclosure will start to adsorb, decreasing pressure in the enclosure.



Idealized TSA pump processes represented on a Pressure-Temperature diagram.<sup>[3]</sup>



RCAP flow configuration during C to D transition.

**To optimize required power & sorbent mass this cyclic process will be split into multiple stages.<sup>[3]</sup>**

- Bed properties with Al shot
  - Bed is enclosed on both sides → bulk properties are for a specific height
  - Use upper bound thermal conductivity,  $k$ , estimate – treats heat flow as parallel through sorbent and gas.<sup>[5]</sup>

$$k_{bed} = 0.5k_{al\ shot} + (1 - 0.5)k_{sorbent}$$

$$k_{slice} = \varphi_{slice} k_{CO_2} + (1 - \varphi_{slice})k_{bed}$$

$$k_{bulk} = \frac{\sum_{i=0}^{i=n_{slices}/2} k_{slice}}{\binom{n_{slices}}{2}}$$

- Density (volume averaged):

$$\rho_{bed} = 0.5\rho_{al\ shot} + (1 - 0.5)\rho_{sorbent}$$

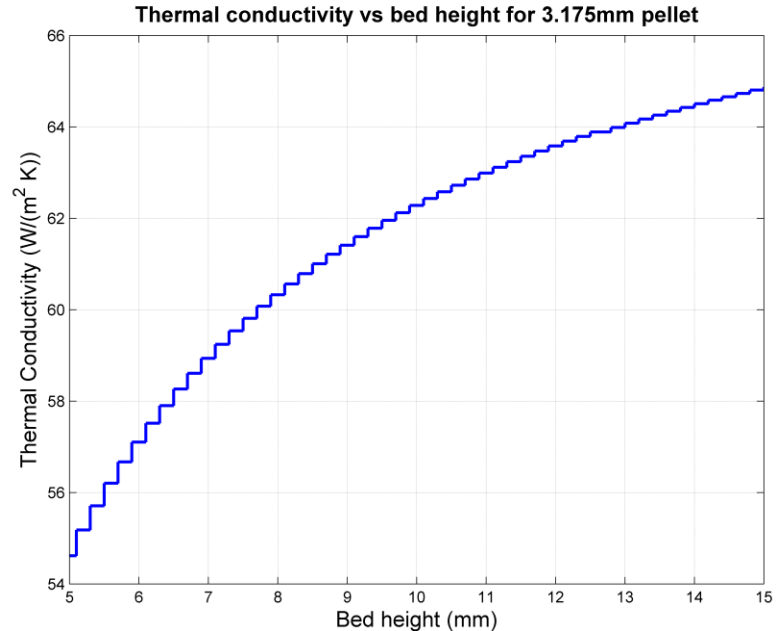
$$\rho_{slice} = \varphi_{slice}\rho_{CO_2} + (1 - \varphi_{slice})\rho_{bed}$$

$$\rho_{bulk} = \frac{\sum_{i=0}^{i=n_{slices}/2} \rho_{slice}}{\binom{n_{slices}}{2}}$$

- Conductance at bed/wall interface

- Use conductance,  $G$ , across slice nearest the wall.<sup>[4]</sup>

$$G_{bed\ to\ flat\ surface} = \frac{1}{\frac{1}{\frac{k_{bed\ 1st\ slice}}{L_{slice}} + \frac{1}{1000}}} \left[ \frac{W}{m^2K} \right]$$



**Bulk thermal conductivity vs bed height, used 0.1mm slices. The bed has 50% aluminum shot and 50% sorbent.**

$$mass_{sorbent} = 0.5 * \frac{m_{bed}}{\rho_{bed}} \rho_{sorbent}$$

**Dampen contact conductance with ~ Al to Al contact in vacuum. Values is ~990W/(m<sup>2</sup>K)**

To account for chiller’s cooling capacity or heater’s heating capacity (fake a loop):

- 1) Calculate chiller and heater reservoir temperatures using max heating or cooling capacities.
- 2) Set inlet plenums to calculated reservoir temperature unless the chiller or heater’s set point is exceeded. If a set point is surpassed, use the set point as the plenum’s temperature.

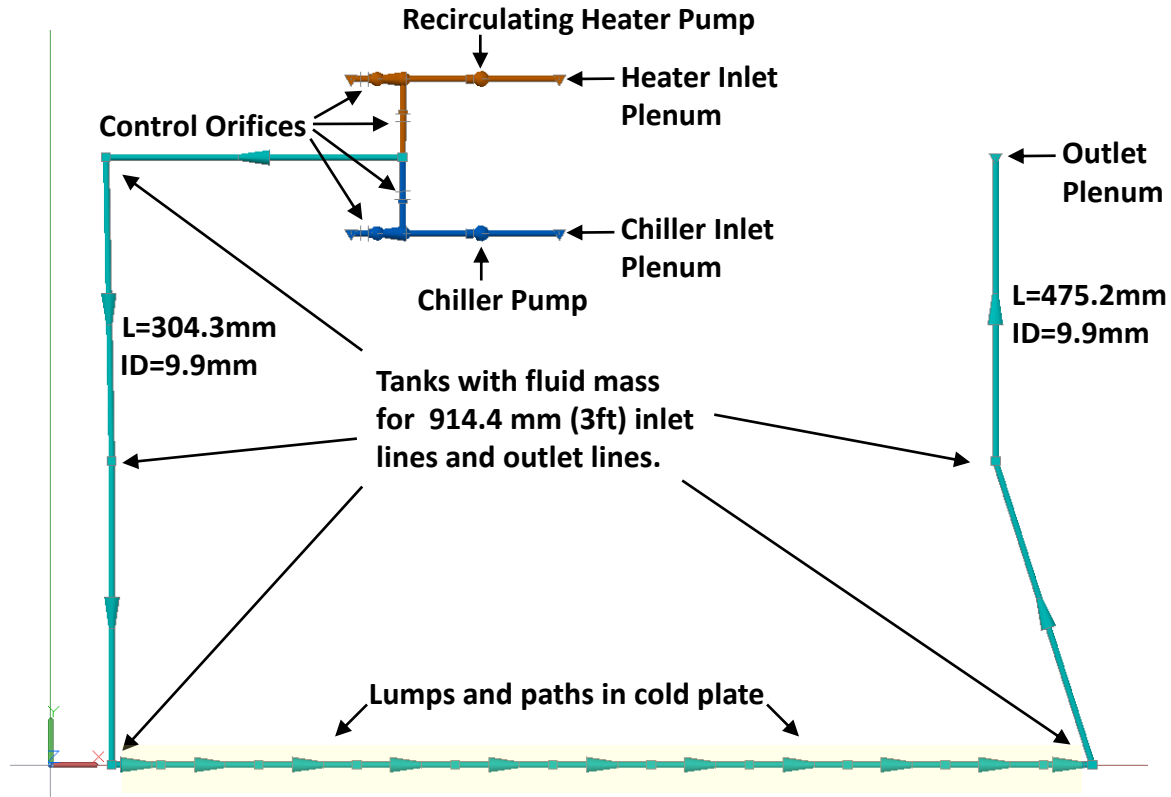
$$\frac{dU}{dt} = Q + \sum m_{in} h_{in} - \sum m_{out} h_{out}$$

$$m c_{p,res} \frac{dT}{dt} = Q + \dot{m} c_{p,avg} (T_{in} - T_{out})$$

$$m c_{p,res} \frac{\Delta T}{\Delta t} = Q + \dot{m} c_{p,avg} (T_{in,t-1} - T_{out,t-1})$$

$$\Delta T = \left( \frac{Q + \dot{m} c_{p,avg} (T_{in,t-1} - T_{out,t-1})}{m c_{p,res}} \right) \Delta t$$

$$T_{res} = T_{res,t-1} + \Delta T$$



# Why you check conductances

