



# Applications of Atomic Layer Deposition in the Modification of Carbon Nanotubes

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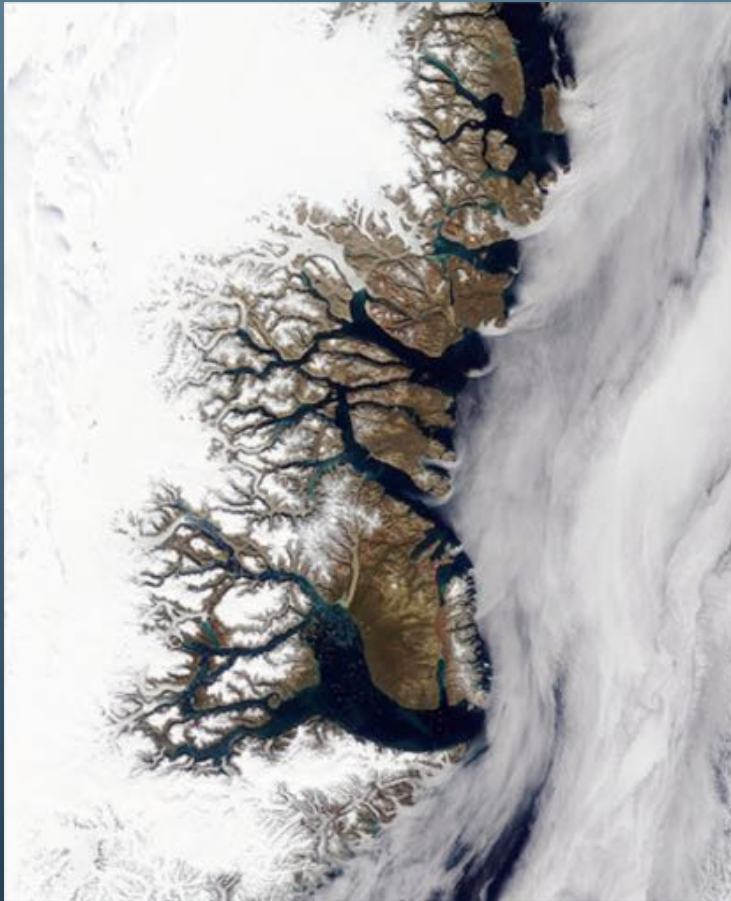


# Outline

- Motivation
- What is a Thin Film?
- Atomic Layer Deposition (ALD) – a primer
- ALD of NiO and *ex situ* reduction
- Results of Coated CNTs
- ALD deposition of Catalyst Layer for CNT growth
- Stray light suppression results
- Conclusions Q&A



# Motivation – Stray Light Control

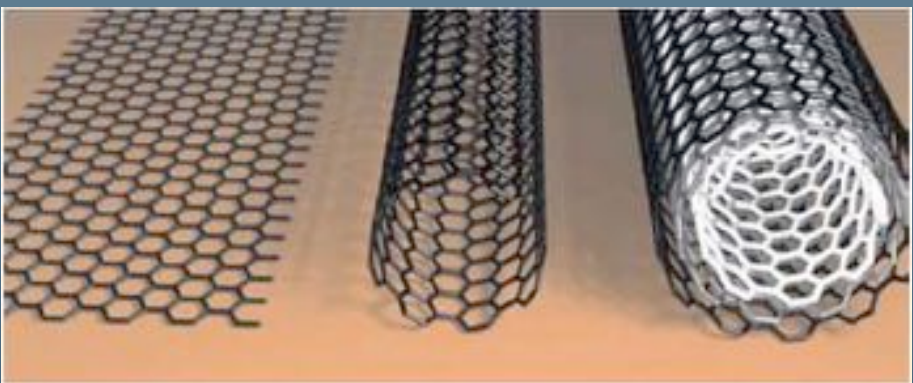


- Z306 absorbs 96%; our current piBlack nanotube formulation absorb > 99.5% from the visible to FIR
- Improving stray light performance of surface treatments can result in exponential stray light reduction at focal plane
  - Enabling new scientific observations with higher – Signal-Noise Ratio - S/N
  - Potentially doubling observational efficiencies in high contrast scenes such as those common in Earth science
  - Simplifying stray light designs by reducing number of controls required for equivalent performance

*High contrast scene; Greenland*

# Motivation – Stray Light Control

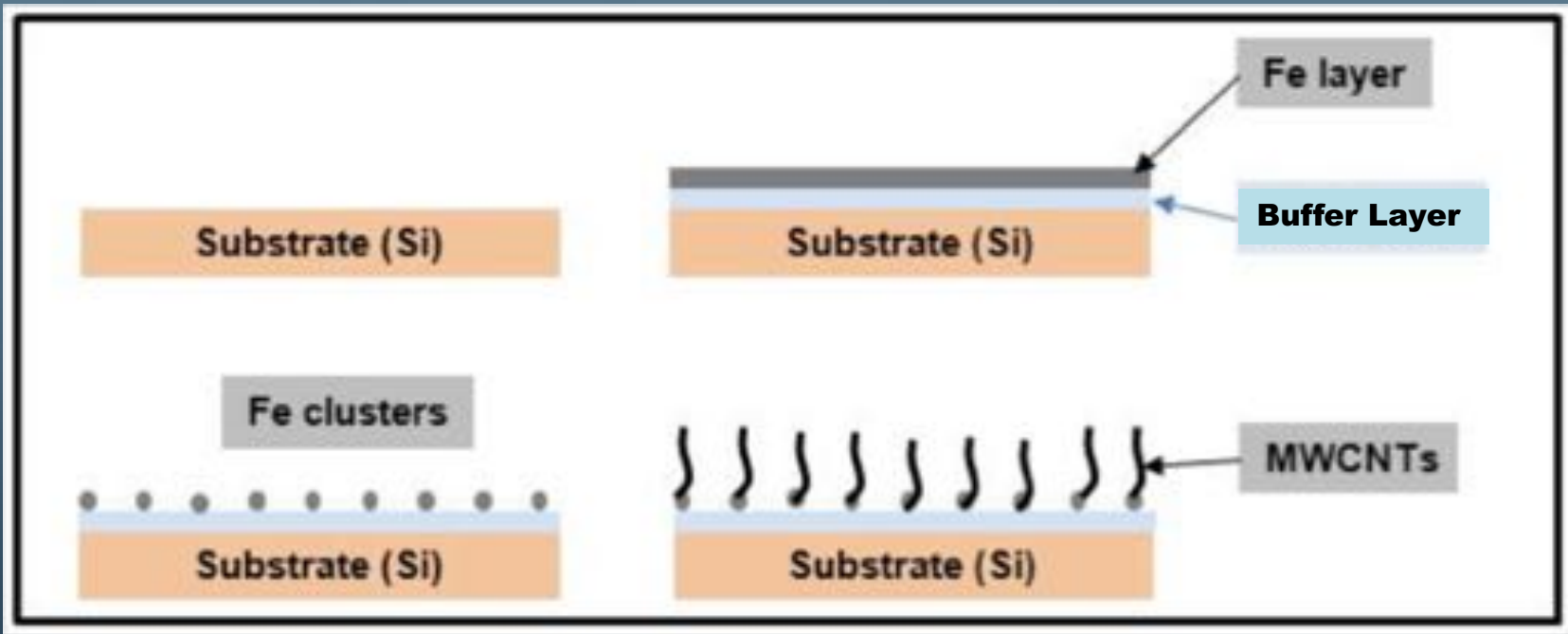
- Electron mobility in plane of carbon hexagonal matrix is high effectively creating electron gas behavior similar to a metal
- Light interacts with the electrons but confinement in lattice plane results in absorption and thermalization of energy
- This high electron mobility and confinement results in high absorption with little specular reflection



Formation of single and multi-walled carbon nanotubes from graphene. The graphene sheet on the left is rolled to form a single walled nanotube (middle) and multi-walled carbon nanotube (right).



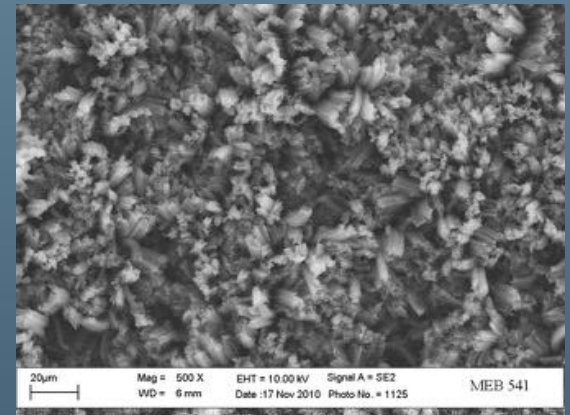
# Thermal CVD Growth of CNTs





# Stray Light Control – Design Parameters

- Parameters for optimization of light absorption
  - Low density; achieving an effective index of refraction approaching 1 vs graphite ( $n=4.4$ ) allows reflection at interface to be minimized
  - Minimization of amorphous carbon
  - Length; long enough to absorb light that gets into nanotube layer
  - Orientation; vertically aligned nanotubes are significantly better absorbers than randomly oriented formulations
  - Increasing surface roughness of substrate can improve absorption by decreasing interface reflectance as well
- These parameters can be tuned during the growth process by adjusting:
  - Catalyst thickness, gas flow and composition, substrate preparation



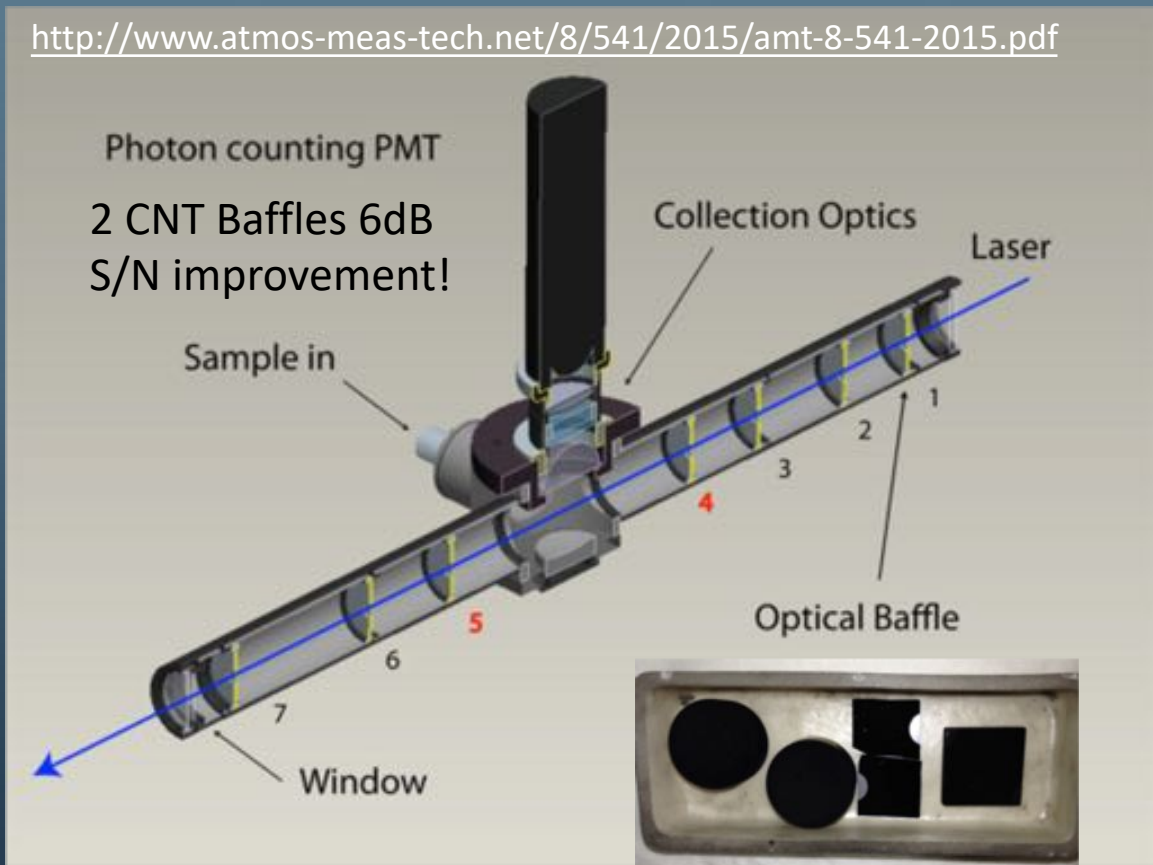
Electron Microscope



# Applications and Challenges

## Goddard in situ Formaldehyde Laser Induced Fluorescence Experiment (Tom Hanisco-PI)

<http://www.atmos-meas-tech.net/8/541/2015/amt-8-541-2015.pdf>



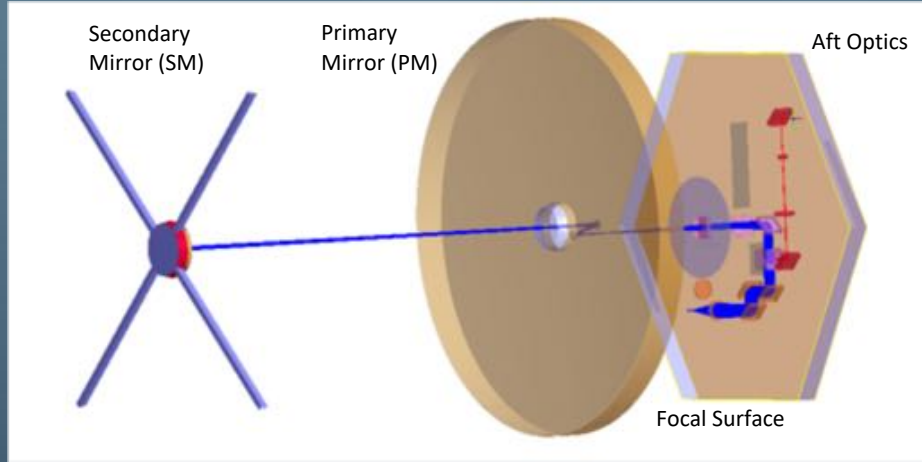
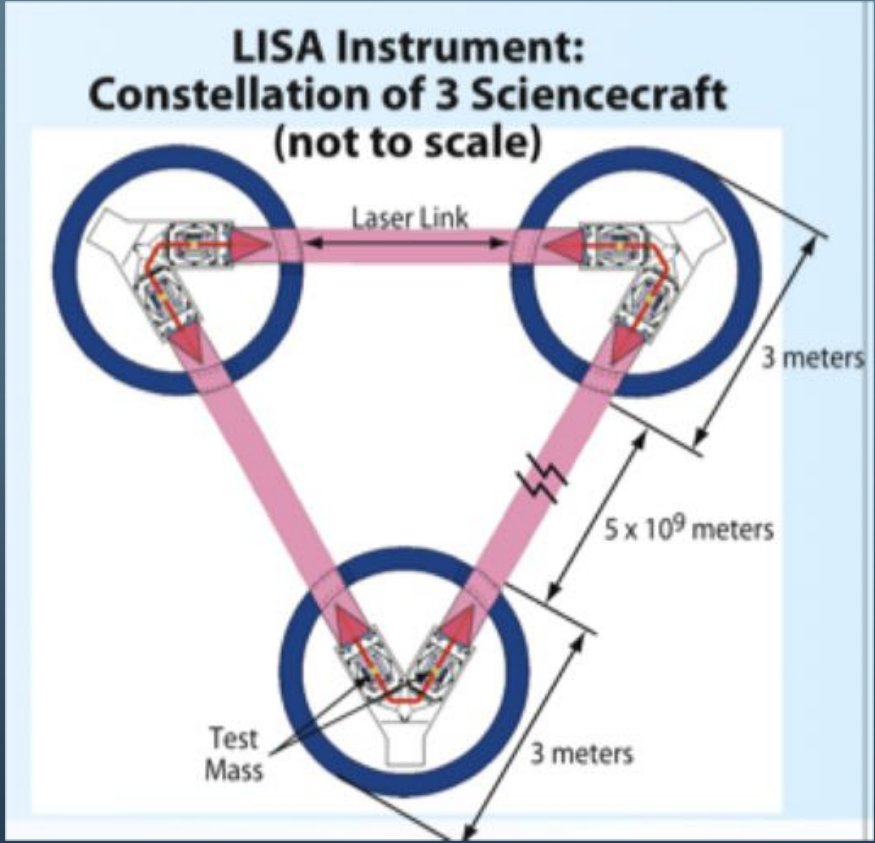
**Challenges:**  
Coating 3D baffles and the inside of cylindrical tubes with catalyst and controlling gas flow during CVD

The fluorescence detection cell of the Goddard in situ formaldehyde LIF experiment is shown with a cut-away view. The laser excites the sampled air in the center of the cell.



# Applications and Challenges

Laser Interferometer Space Array (LISA) Gravity Wave Sensor  
(Telescope PI – Jeffrey Livas)



- **LISA Stray Light Challenge**
  - Telescope used in duplex
  - Tx beam is 10<sup>9</sup>x intensity of Rx Beam
  - Tx beam reflected/scattered off of center of telescope secondary mirror (SM) must be suppressed by 10<sup>9</sup>

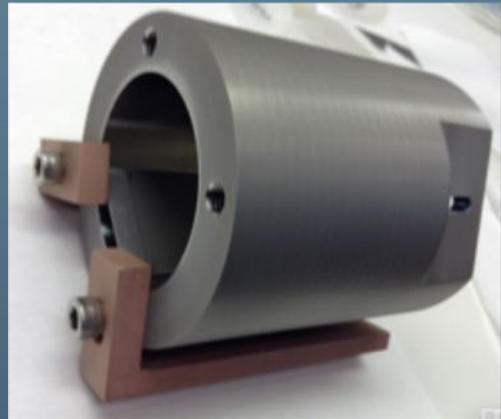
Constellation of 3 Sciencecraft with linked Telescopes



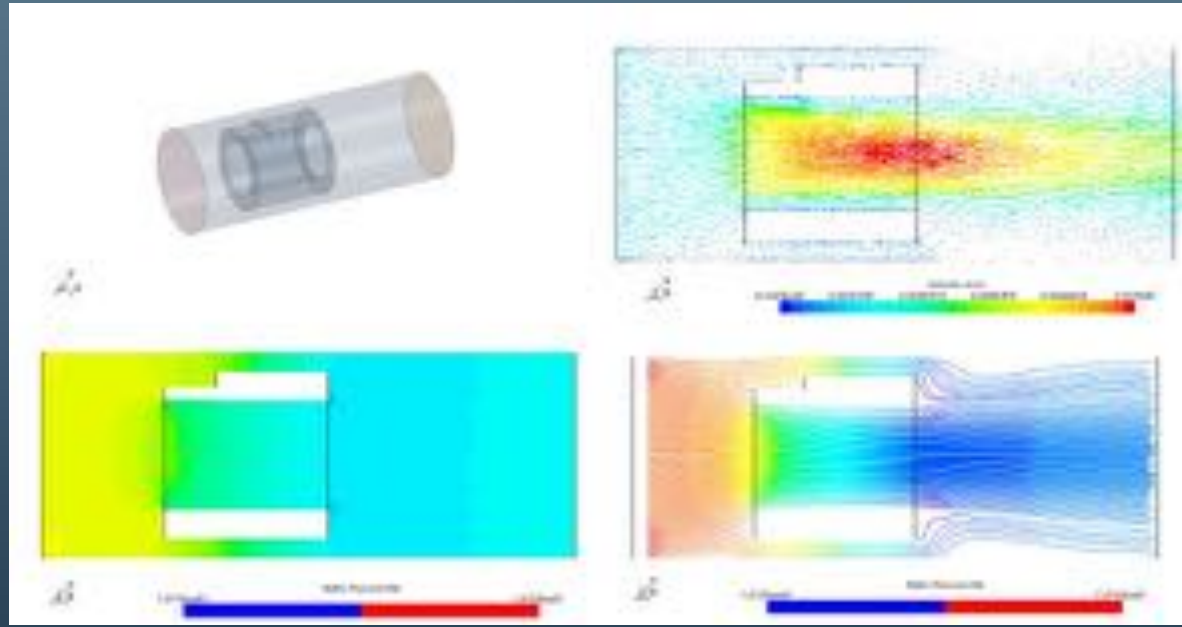


# Applications and Challenges

Solar Coronagraph – PI Doug Rabin



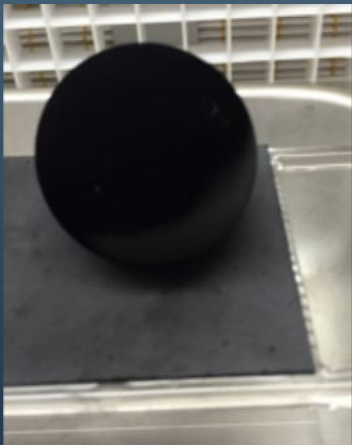
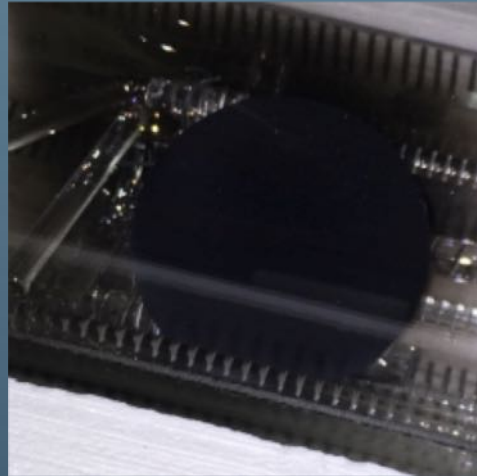
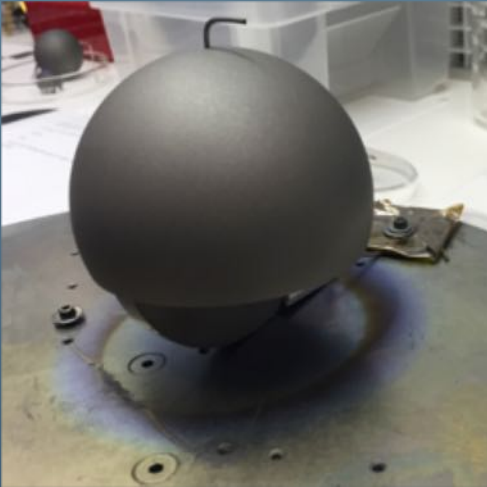
**Challenge:**  
**3D Catalyst Deposition, CVD flow control**  
**Solution: Catalyst sputtering, tapered boat**



[https://www.nasa.gov/content/goddard/nasa-ultra-black-nano-coating-to-be-applied-to-3-d-new-solar-coronagraph/#.Vj\\_a0Uar8ZM](https://www.nasa.gov/content/goddard/nasa-ultra-black-nano-coating-to-be-applied-to-3-d-new-solar-coronagraph/#.Vj_a0Uar8ZM)

# Applications and Challenges

Spherical Occulter Coronagraph CubeSat (SpOC Cube) – PI Phil Chamberlin



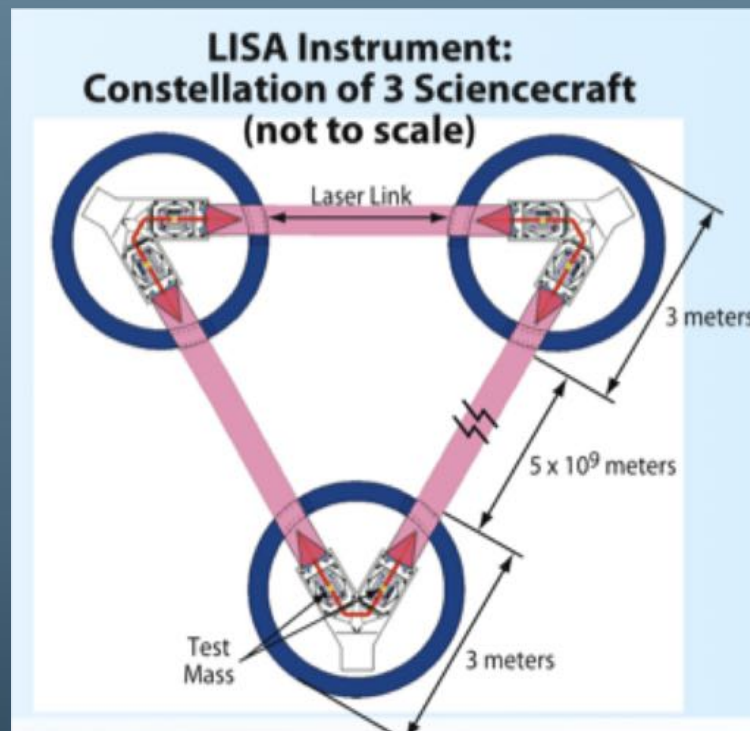
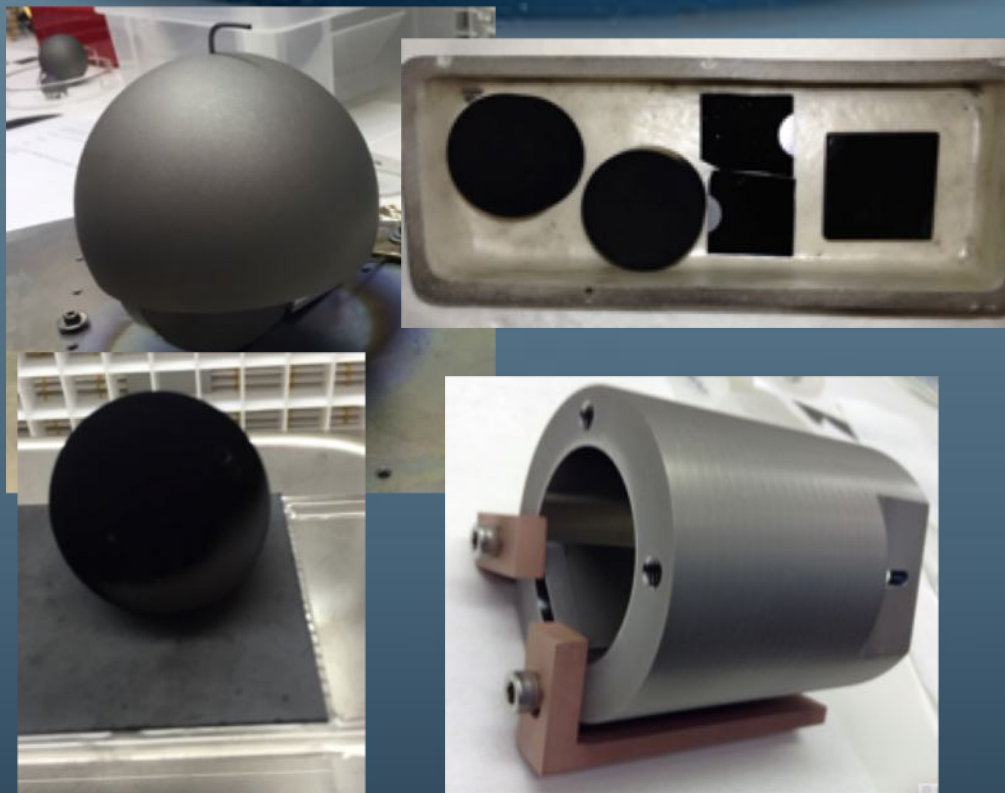
Three titanium spheres from ½" to 3" in diameter were fabricated to provide pathfinder elements for testing the theory

<https://gsfctechnology.gsfc.nasa.gov/Coronagraph.html>

**Challenges to fabricating spheres include:**

- 1. Uniformly depositing catalyst on a sphere during physical vapor deposition**
- 2. Controlling the gas flows during the chemical vapor deposition phase to prevent shadowing and uneven growth**

# Challenge



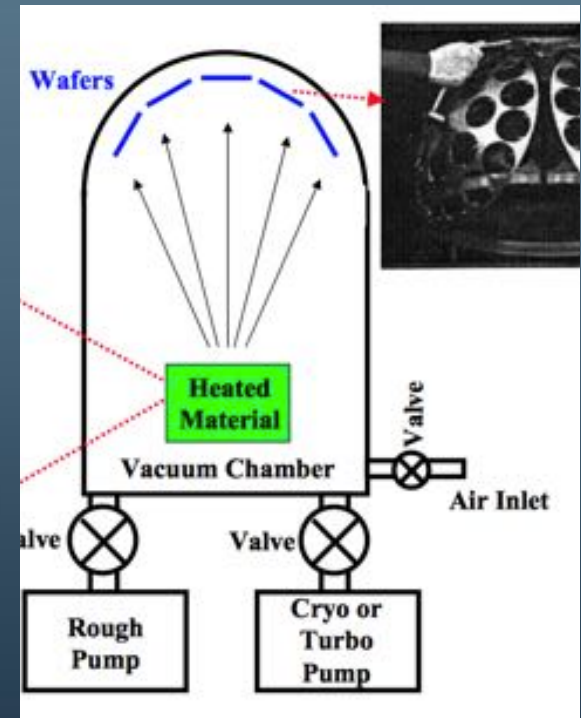
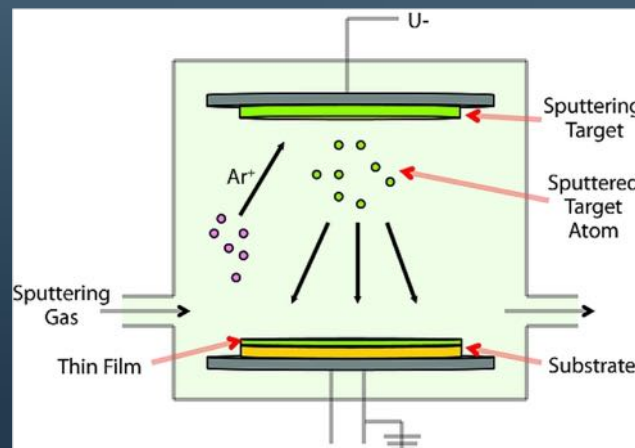
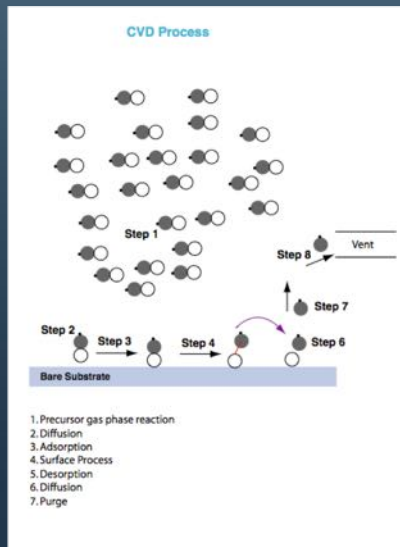
The deposition of the catalyst is in terms of: uniformity, thickness control, material regardless of geometry is paramount. Flat surfaces are easy!

# What is a Thin Film?

**Thin film:** thickness typically  $<1000\text{nm}$ .

**Special properties of thin films:** different from bulk materials, it may be –

- Not fully dense
- Under stress
- Different defect structures from bulk
- Quasi - two dimensional (very thin films)
- Strongly influenced by surface and interface effects

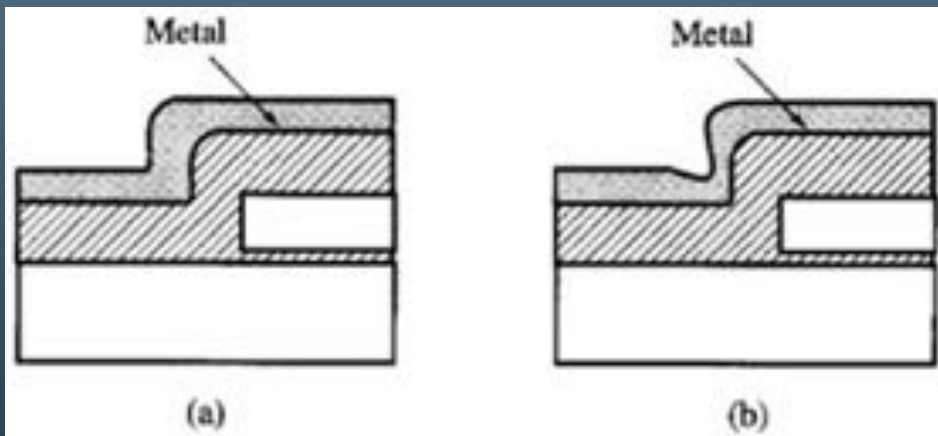


# Common Denominator



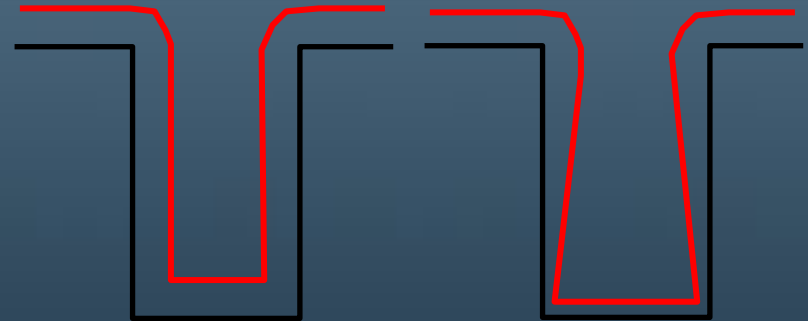
- Deposition only occurs on substrates that “see” the target.
- Plasma process can damage the substrate
- Poor thickness control
- Poor Step Control
- High Pressure High Temperature Environment

## Step Coverage Example



conformal

non-conformal



Step coverage of metal over non-planar topography.

(a) Conformal step coverage, with constant thickness on horizontal and vertical surfaces.

(b) Poor step coverage, here thinner for vertical surfaces.

# Atomic Layer Deposition



## Atomic Layer Deposition

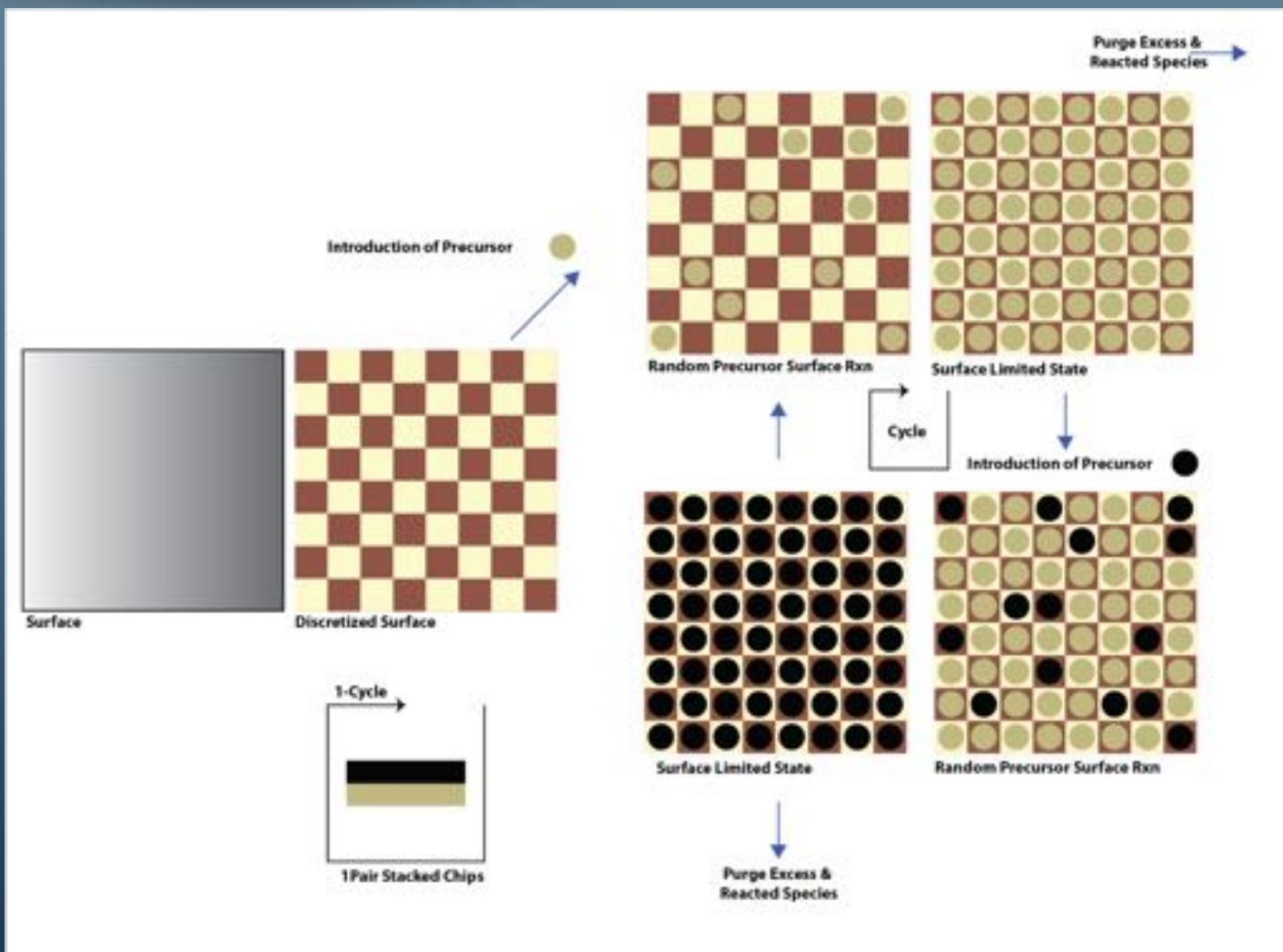


A thin film “nanomanufacturing” tool that allows for the conformal coating of materials on a myriad of surfaces with precise atomic thickness control.

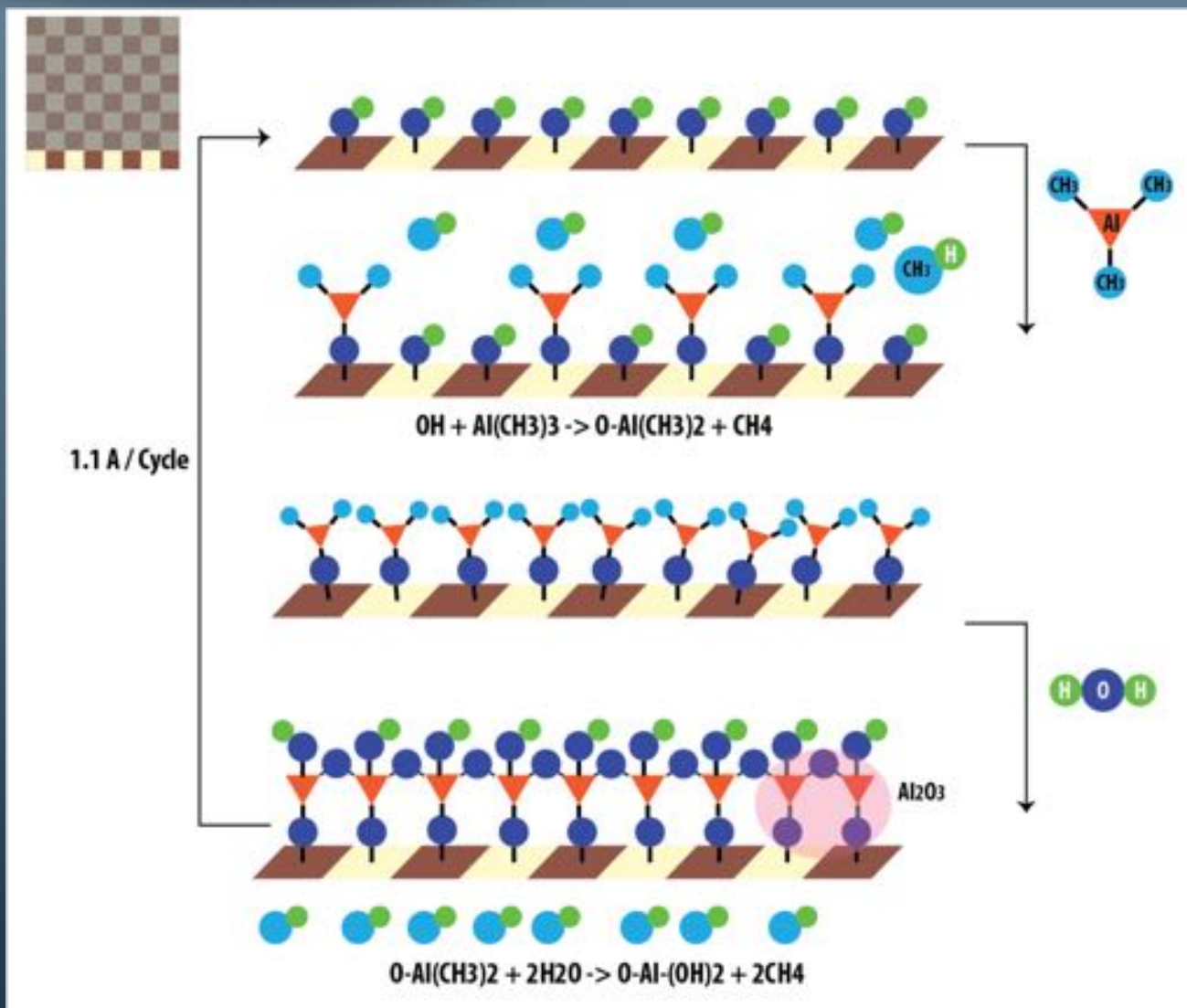
## Based on:

- Paired gas surface reaction chemistries
- Benign non-destructive temperature and pressure environment
  - Room temperature -> 250 °C (even lower around 45 °C)
  - Vacuum

# ALD Analogy (Checkers)



# ALD Analogy Chemistry



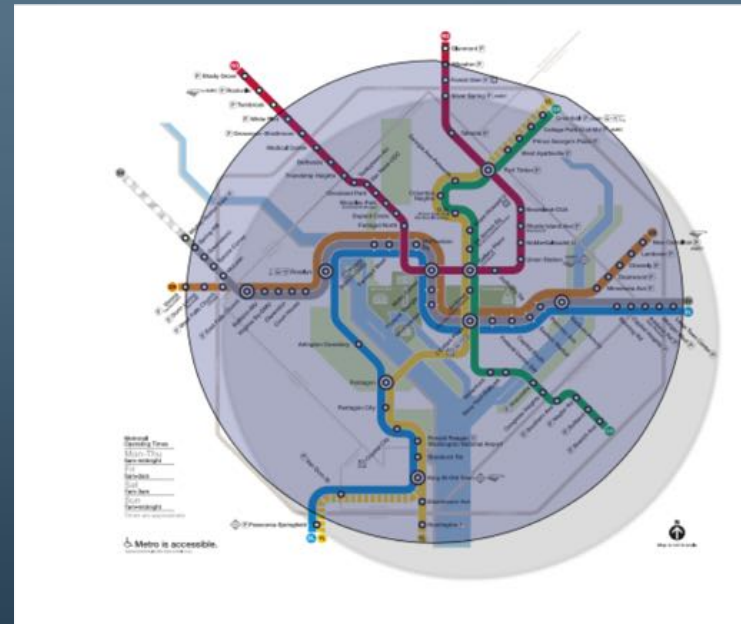
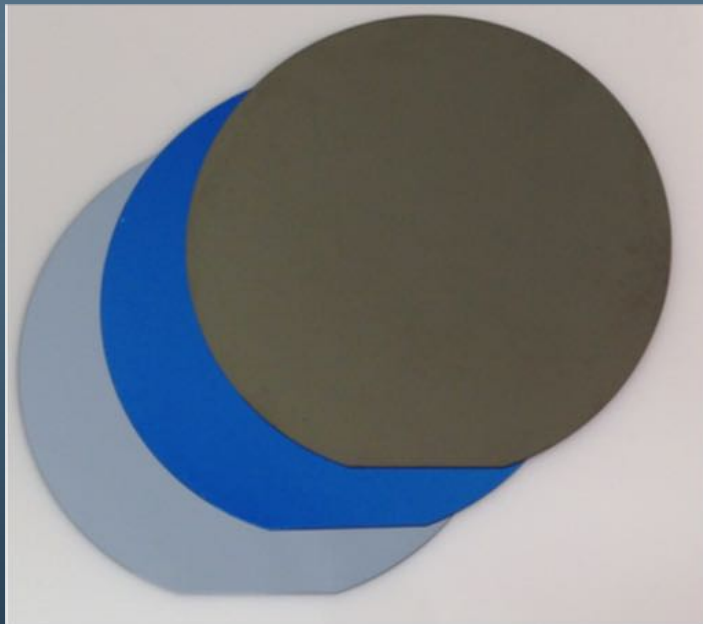


# ALD

Precursor A + Precursor B  $\rightarrow$  Solid film + Gas by-products

Cyclic operation: A  $\rightarrow$  purge  $\rightarrow$  B  $\rightarrow$  purge  $\rightarrow$  A  $\rightarrow$  purge  $\rightarrow$  ...

Atomic-level thickness control ...

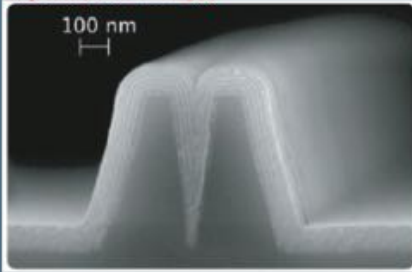


... equivalent to a 60  $\mu\text{m}$  layer over a city-sized wafer



# ALD Advantageous Property

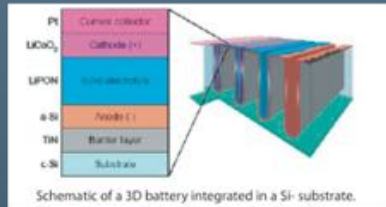
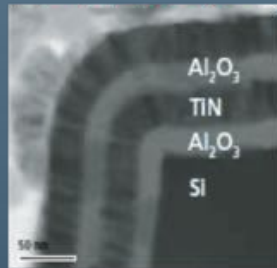
Artificial trench filled with an ALD nanolaminate  
Image courtesy of Austin University [20]



## Epitaxial Growth

Multilayer consisting of:  
Al<sub>2</sub>O<sub>3</sub> - 25 nm  
TiN - 20 nm  
Al<sub>2</sub>O<sub>3</sub> - 25 nm

Dr. Fred Roseboom, NXP Semiconductor Research and  
Dr. Edwin Kozuly, University of Technology, Eindhoven



Schematic of a 3D battery integrated in a Si-substrate.  
The cross-section shows the various functional layers  
in the battery stack as well as the candidate materials.

Kozuly, E.C. et al., *IEEE Trans.*, 26 (2009) pp. 232-244

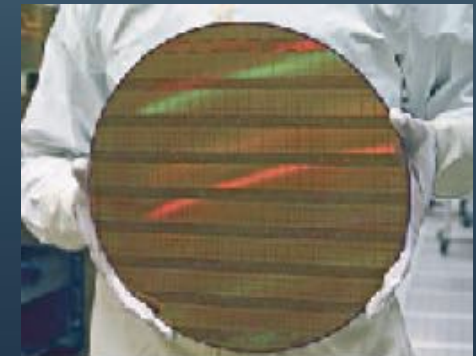
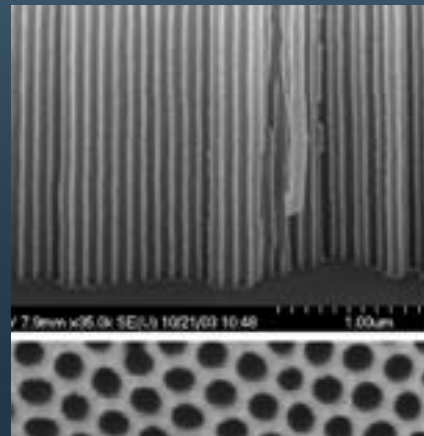
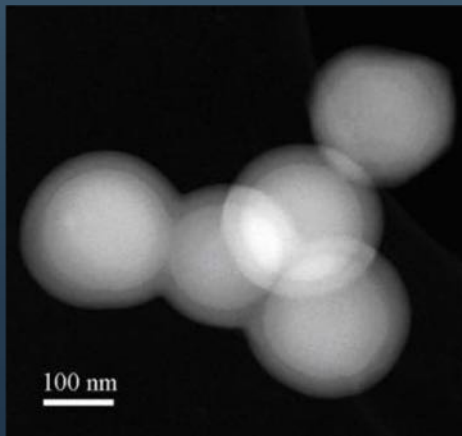
## Batch Process



Coating Silver with Aluminum Oxide  
<http://www.glassonweb.com>



## Substrate Independence



Goddard  
Space Flight Center

# ALD Material Systems

H 1																	He 2														
Li 3	Be 4																	B 5	C 6	N 7	O 8	F 9	Ne 10								
Na 11	Mg 12																	Al 13	Si 14	P 15	S 16	Cl 17	Ar 18								
K 19	Ca 20	Sc 21	Ti 22	V 23	Cr 24	Mn 25	Fe 26	Co 27	Ni 28	Cu 29	Zn 30	Ga 31	Ge 32	As 33	Se 34	Br 35	Kr 36														
Rb 37	Sr 38	Y 39	Zr 40	Nb 41	Mo 42	Tc 43	Ru 44	Rh 45	Pd 46	Ag 47	Cd 48	In 49	Sn 50	Sb 51	Te 52	I 53	Xe 54														
Cs 55	Ba 56	La 57	Hf 72	Ta 73	W 74	Re 75	Os 76	Ir 77	Pt 78	Au 79	Hg 80	Tl 81	Pb 82	Bi 83	Po 84	At 85	Rn 86														
Fr 87	Ra 88	Ac 89	Rf 104	Db 105	Sg 106	Bh 107	Hs 108	Mt 109																							
																		Ce 58	Pr 59	Nd 60	Pm 61	Sm 62	Eu 63	Gd 64	Tb 65	Dy 66	Ho 67	Er 68	Tm 69	Yb 70	Lu 71
																		Th 90	Pa 91	U 92	Np 93	Pu 94	Am 95	Cm 96	Bk 97	Cf 98	Es 99	Fm 100	Md 101	No 102	Lr 103

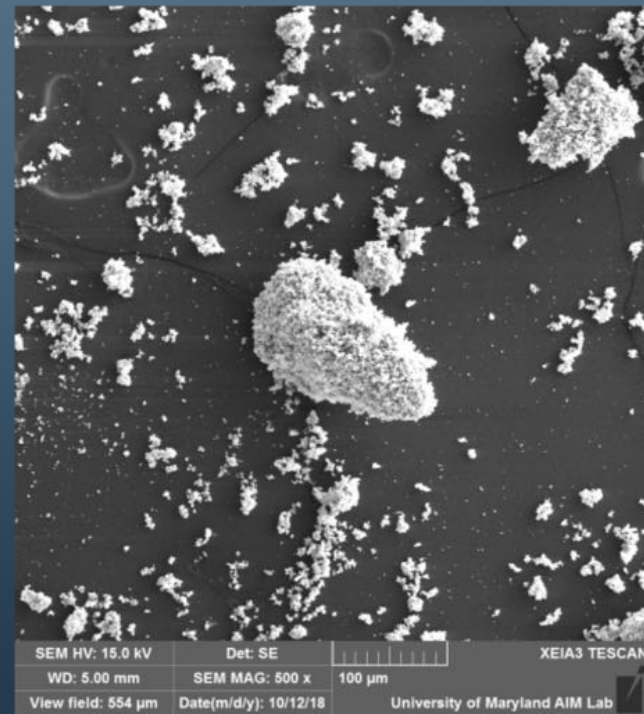
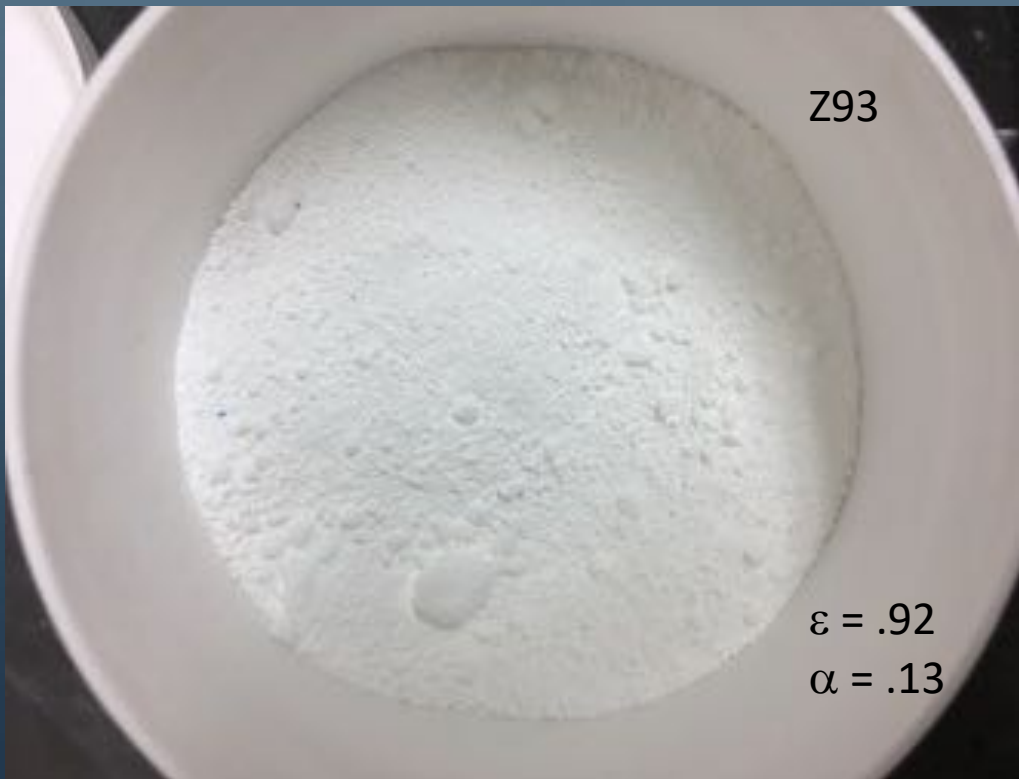
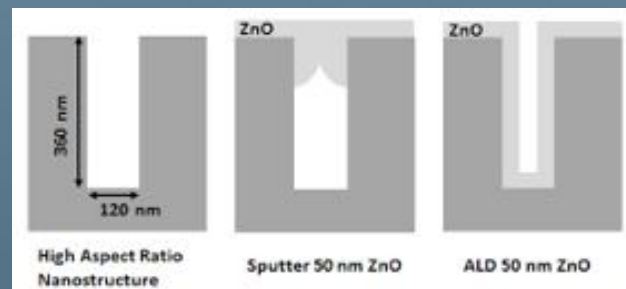
○ Oxide  
○ Carbide  
○ Nitride  
○ Fluoride  
○ Metal  
○ Dopant  
○ Phosphide/Arsenide  
○ Sulfide/Selenide/Telluride

○ Oxide of this element has been deposited by the ALD community  
○ Recipe for this material is available from CNT staff or customer base

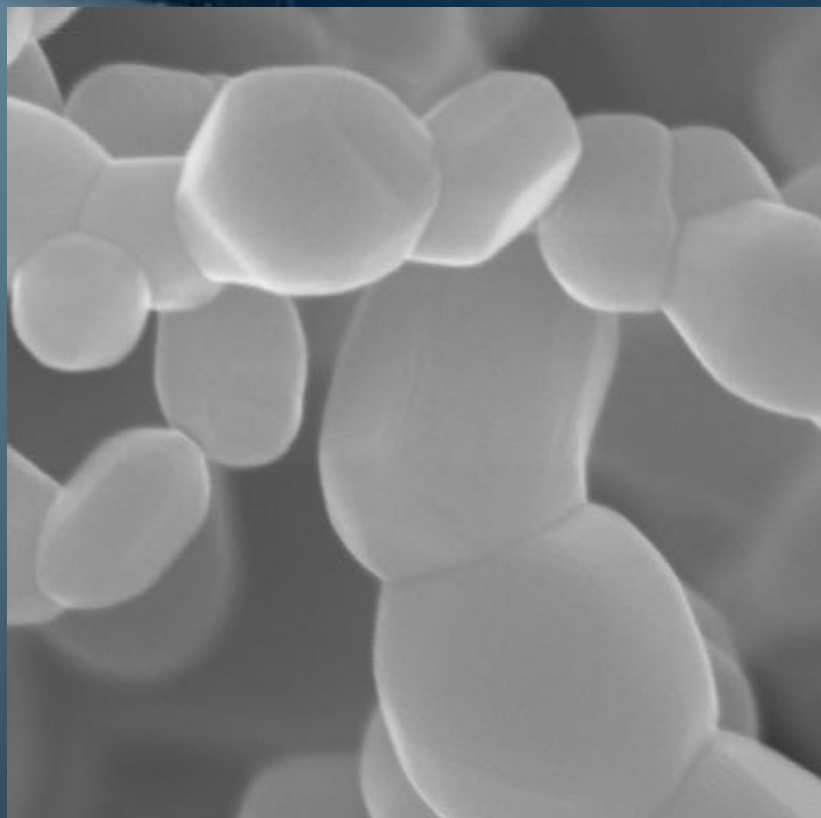
- Gordon, Roy (2008). Atomic Layer Deposition (ALD): An Enable for Nanoscience and Nanotechnology. PowerPoint lecture presented at Harvard University, Cambridge, MA.
- Elam, Jeffrey (2007). ALD Thin Film Materials. Argonne National Laboratory



# High Aspect Ratio (Radiation Pigments)

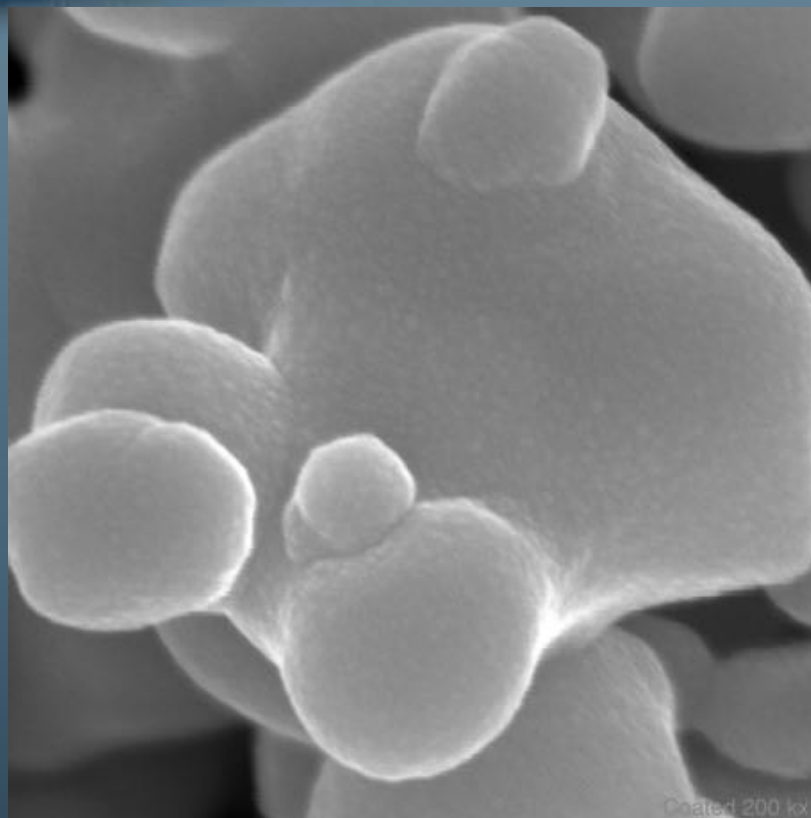


# Results



SEM HV: 15.0 kV	Det: In-Beam SE		XEIA3 TESCAN
WD: 4.93 mm	SEM MAG: 200 kx	200 nm	Uncoated 200 kx
View field: 1.38 $\mu$ m	Date(m/d/y): 10/12/18	University of Maryland AIM Lab	

**Uncoated Pigment**

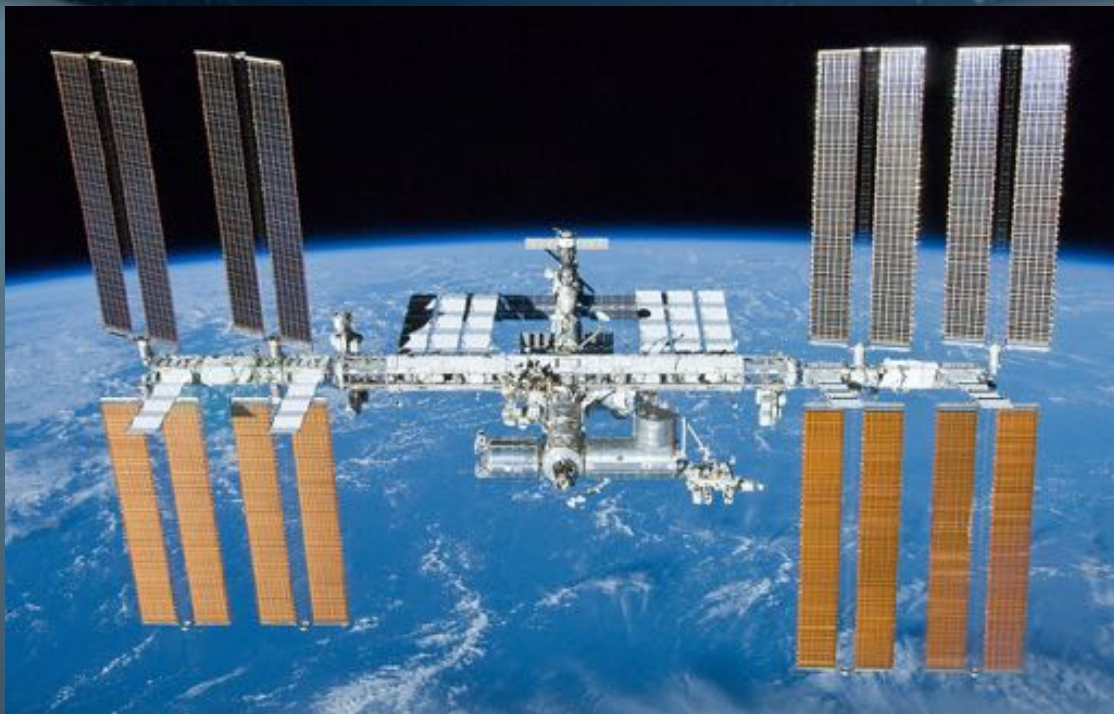


SEM HV: 15.0 kV	Det: In-Beam SE		XEIA3 TESCAN
WD: 5.00 mm	SEM MAG: 200 kx	200 nm	Coated 200 kx
View field: 1.38 $\mu$ m	Date(m/d/y): 10/12/18	University of Maryland AIM Lab	

**Coated Pigment**



# ISS Opportunity - MISSE-FF



The Materials ISS Experiment Flight Facility (MISSE-FF) with MISSE Sample Carriers (MSCs) in the fully open position exposing samples/experiments to the harsh environment of space in low-Earth Orbit (LEO). Image courtesy of Alpha Space.



An earlier MISSE mission



- Latest**
- Related**
- NASA TV Coverage Set for November Cygnus Launch to the International Space Station  
4 days ago
- Cygnus Dedicated to Astronaut John Young  
19 days ago
- Delings: The Little CubeSat That Could  
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- NASA Launching Mars Lander Parachute Test from Wallops Sep. 7  
2 months ago
- Launch of Orbital ATK Antares From NASA's Wallops Flight Facility  
6 months ago
- NASA Sends New Research on Orbital ATK Mission to Space Station  
8 months ago



Nov. 8, 2018

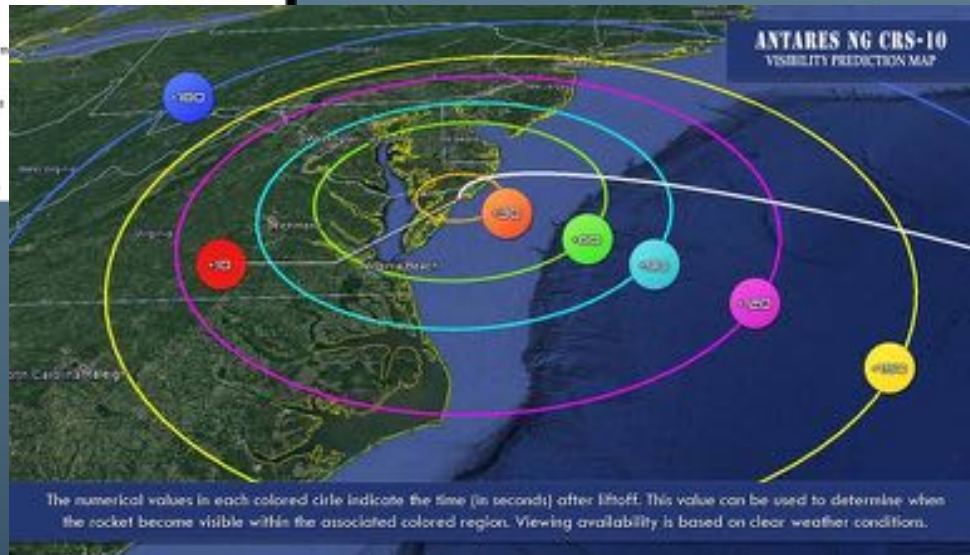
### Catch the Nov 17 Antares Launch from Wallops

Get up early Nov. 15 to view the Northrop Grumman's Antares rocket launch from the Mid-Atlantic Regional Spaceport at NASA's Wallops Flight Facility.

The NASA Wallops Flight Facility and Virginia's Mid-Atlantic Regional Spaceport are set to support the launch of the Antares rocket, carrying the company's Cygnus cargo spacecraft to the International Space Station at 4:49 a.m. EST, Nov. 15.

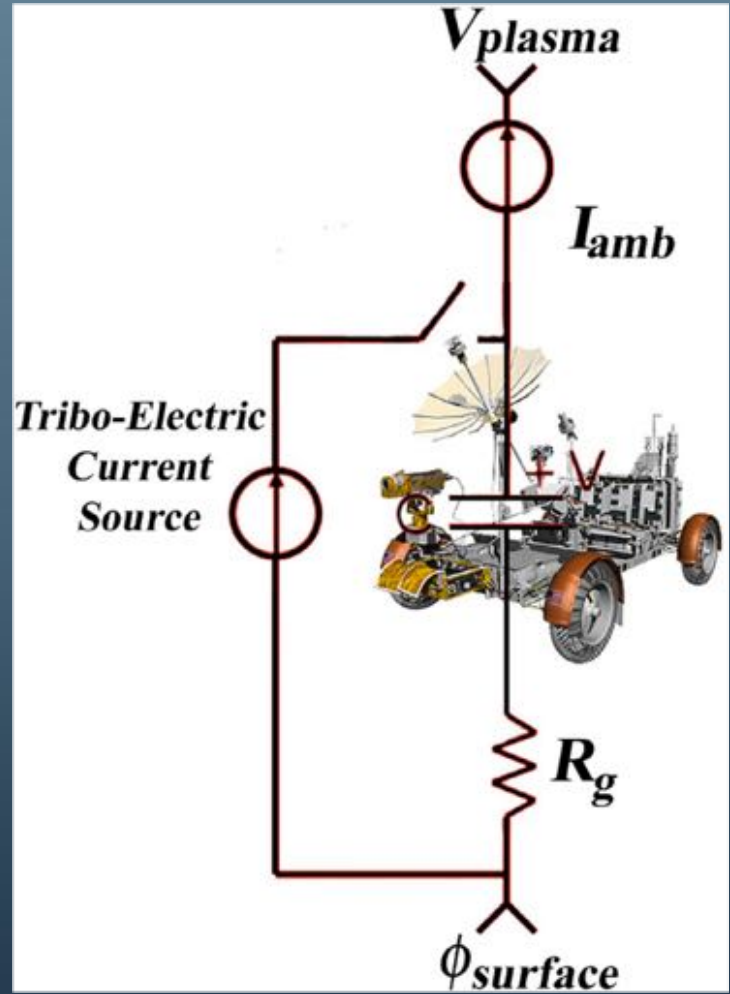
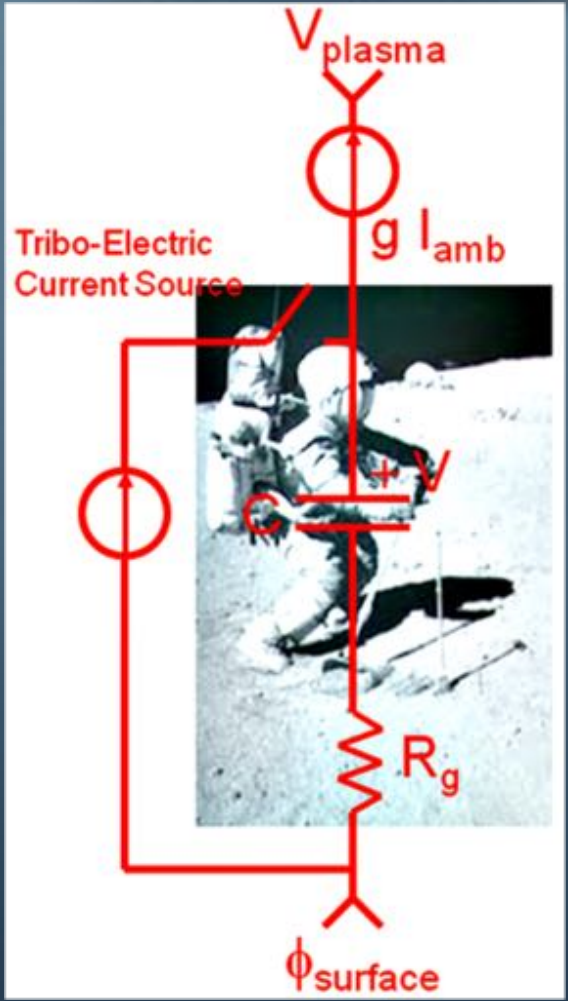
The launch may be visible, weather permitting, to residents throughout the East Coast of the United States.

The NASA Visitor Center at Wallops opens at 1 a.m. on launch day for public viewing. Additional locations for catching the launch are Robert Reed Park on Chincoteague Island or Beach Road spanning the area between Chincoteague and Assateague Islands. Assateague Island National Seashore/Chincoteague National Wildlife Refuge in Virginia will not be open for viewing the launch.





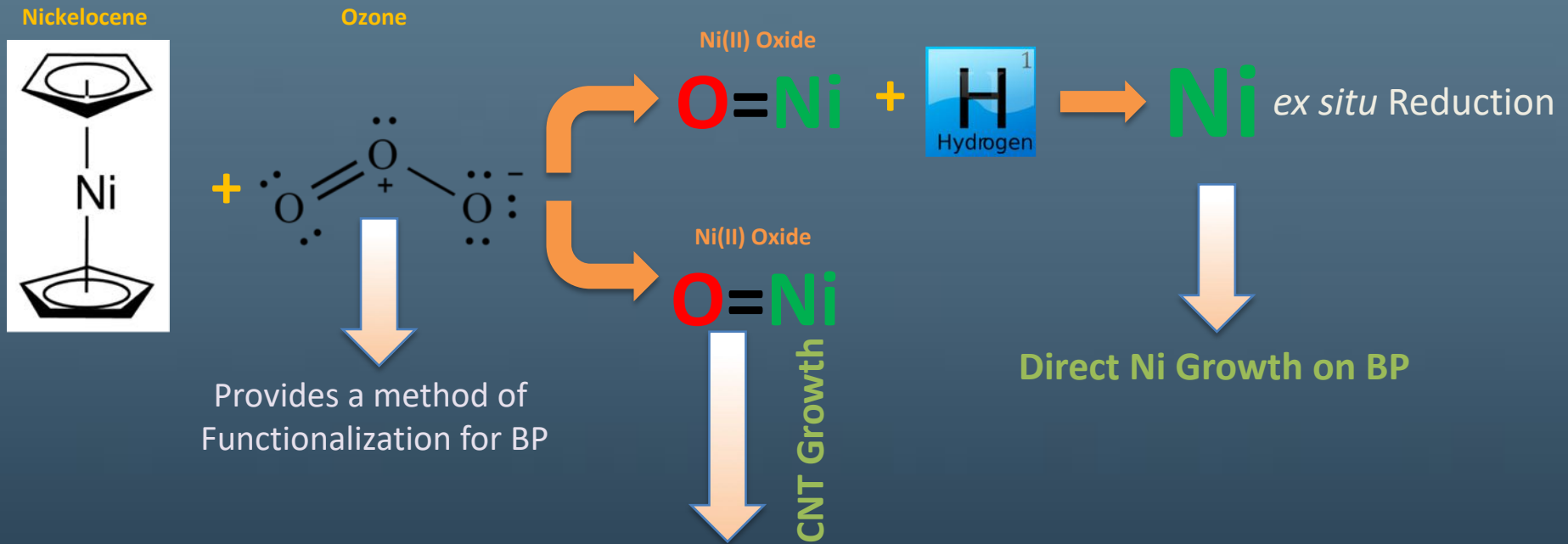
# Potential Applications





# Element of choice (Ni)

## Reaction Pathway



- During Growth Process Ni(II) Oxide reduces to Ni
- Ni (II) Oxide is Oxygen rich and reduction results in 40% decrease in thickness
- Resulting in Ni Cluster



# Experimental Procedures



## Reactor Parameters:

- Reactor Temperature: 245 C
- Ni Precursor Temperature: 90 C
- Argon Flow Rate: 20 SCCM
- Ozone Weight Percent: 8%
- Ozone Flow Rate: 20 SCCM

- Ni Pulse Time: .6 Sec (2\* .3 sec)
- Ni Residence Time: 15 Sec
- Ozone Pulse Time: .2 Sec
- Ozone Residence Time: 15 Sec

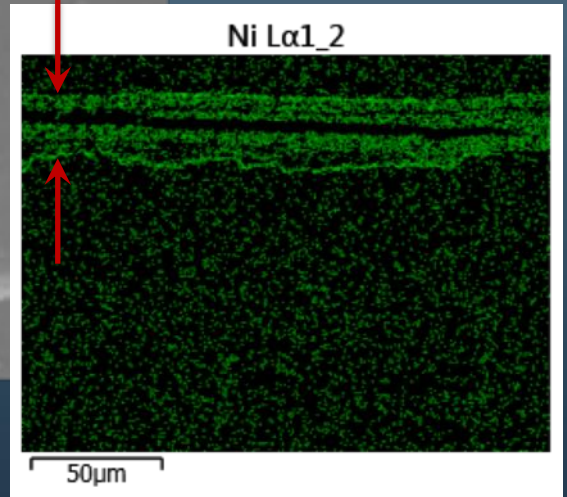
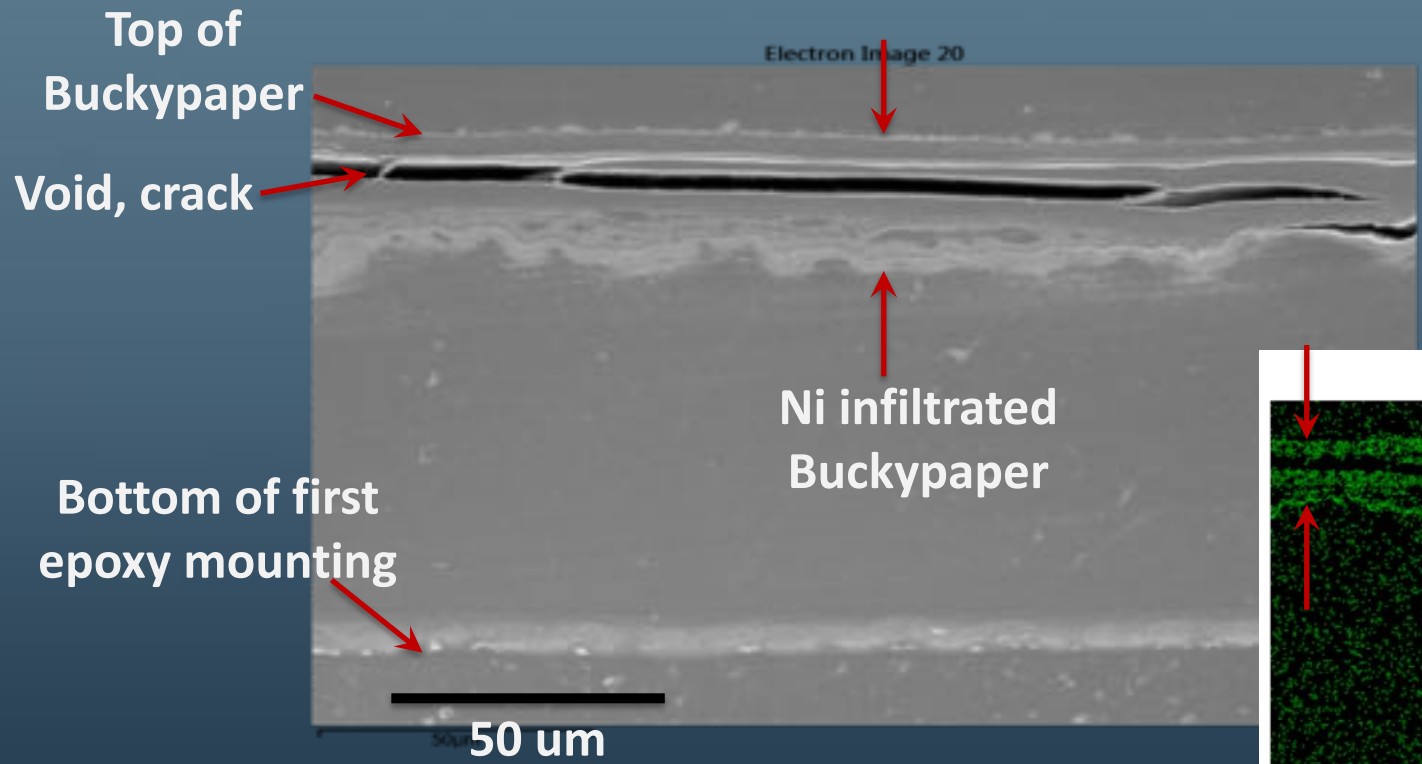


- Hydrogen Reduction:  
5 Hours  
450 C  
5% Hydrogen in Argon  
Flow Rate 550 sccm



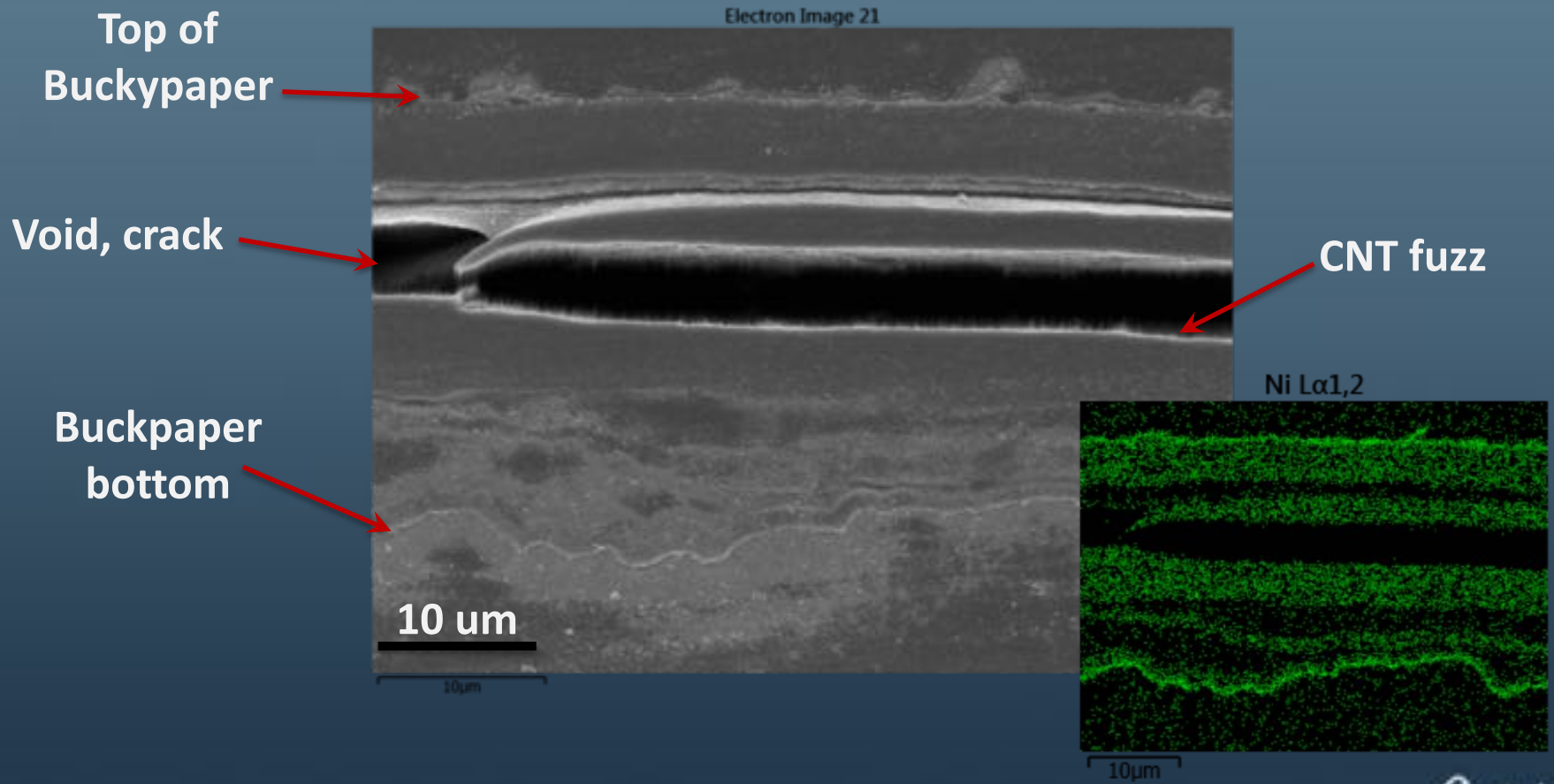
# Results - SEM & EDS of Ni coated buckypaper

Cross-sectional, edge view of the coat paper, 500 cycles of NiO – then reduced; mounted in metallographic epoxy: a crack frequently found in infiltrated region.



# Results - 500 cycles NiO-reduced

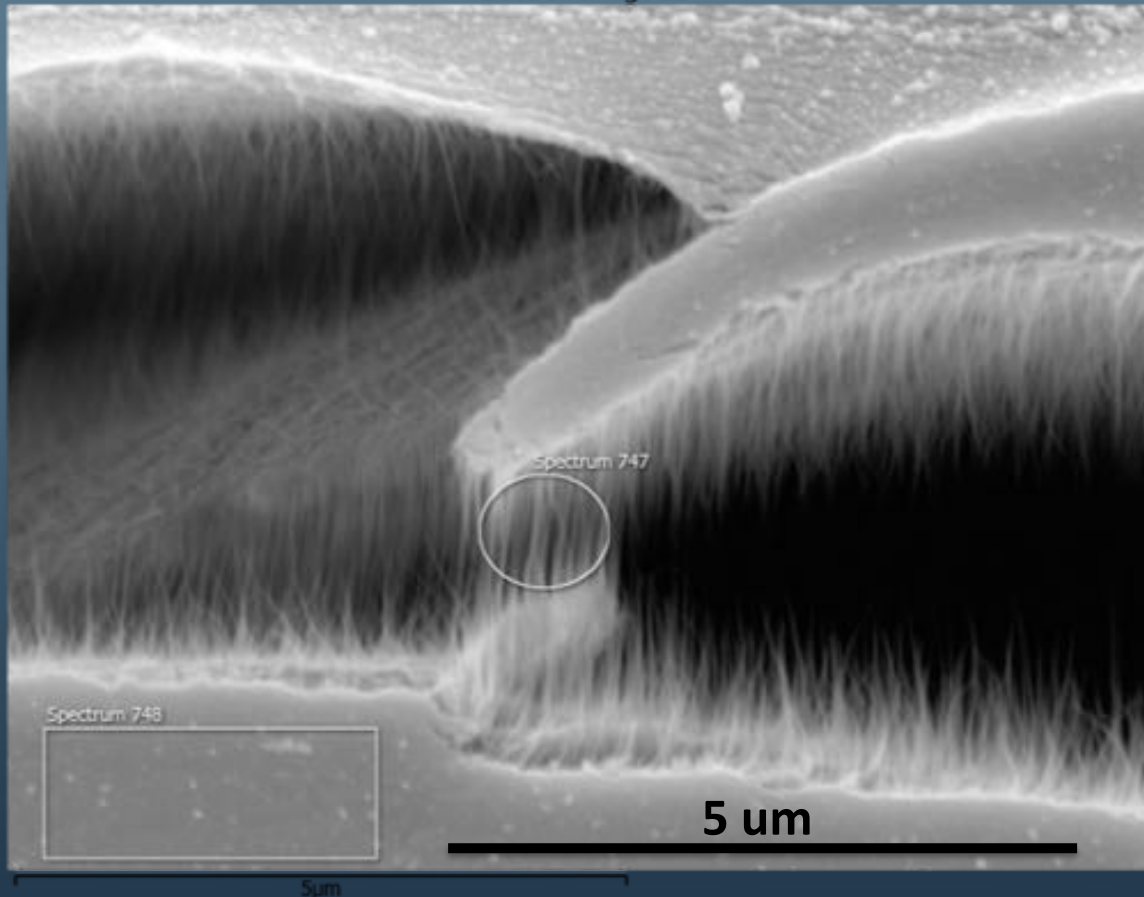
Closer view of the cracked area in the Buckypaper; Ni appears to be throughout the Buckypaper thickness.



# 500 cycles NiO-reduced

Closer view of the CNT in cracked area; showing less Ni than we would like, and more O than we would like.

Electron Image 22



CNT

Spectrum 747

	At%
C	94.1
O	3.9
Ni	2.0

Spectrum 748

	At%
C	92.2
O	5.7
Ni	2.1
N	0.0

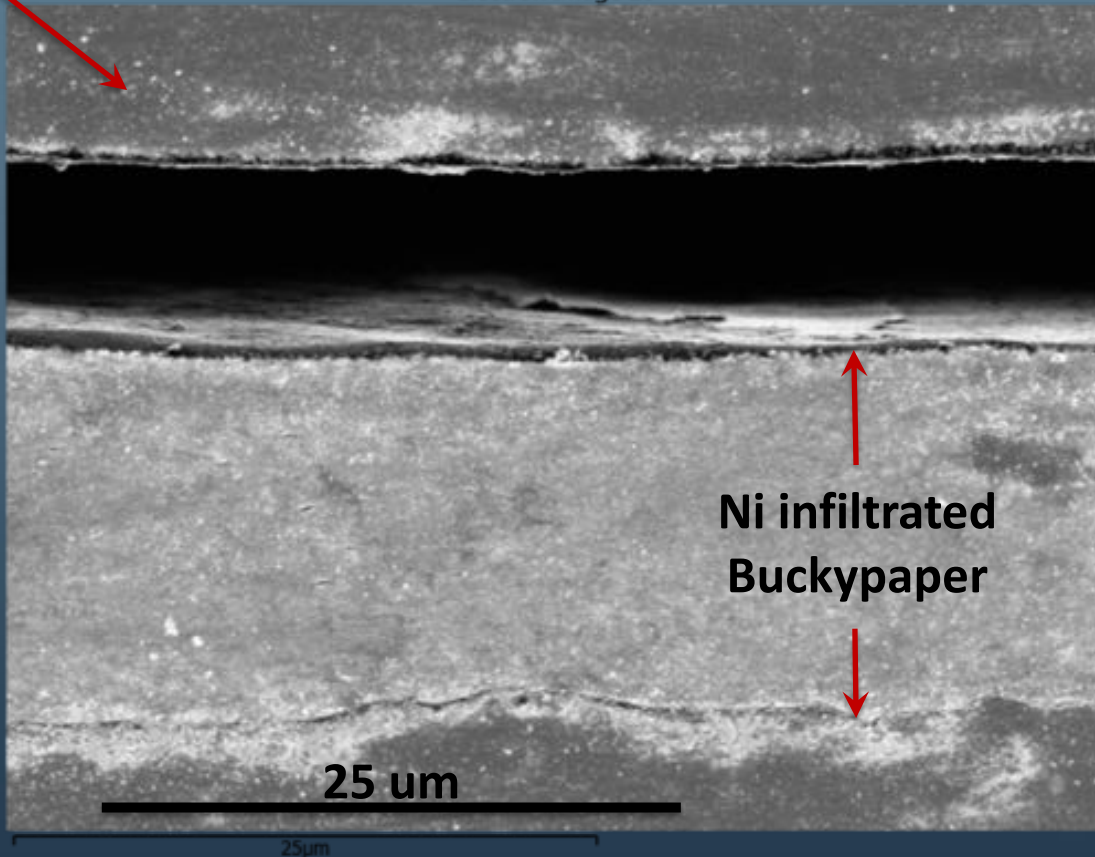
# 1000 cycles NiO-reduced



Buckypaper subjected to more deposition cycles showing good Ni penetration into the paper.

Epoxy

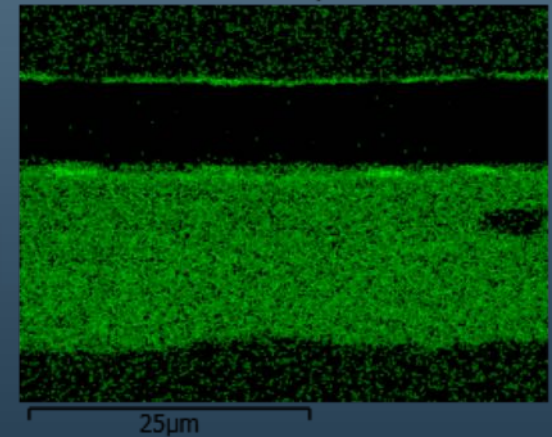
Electron Image 18



Crack

Ni infiltrated  
Buckypaper

Ni L $\alpha$ 1,2



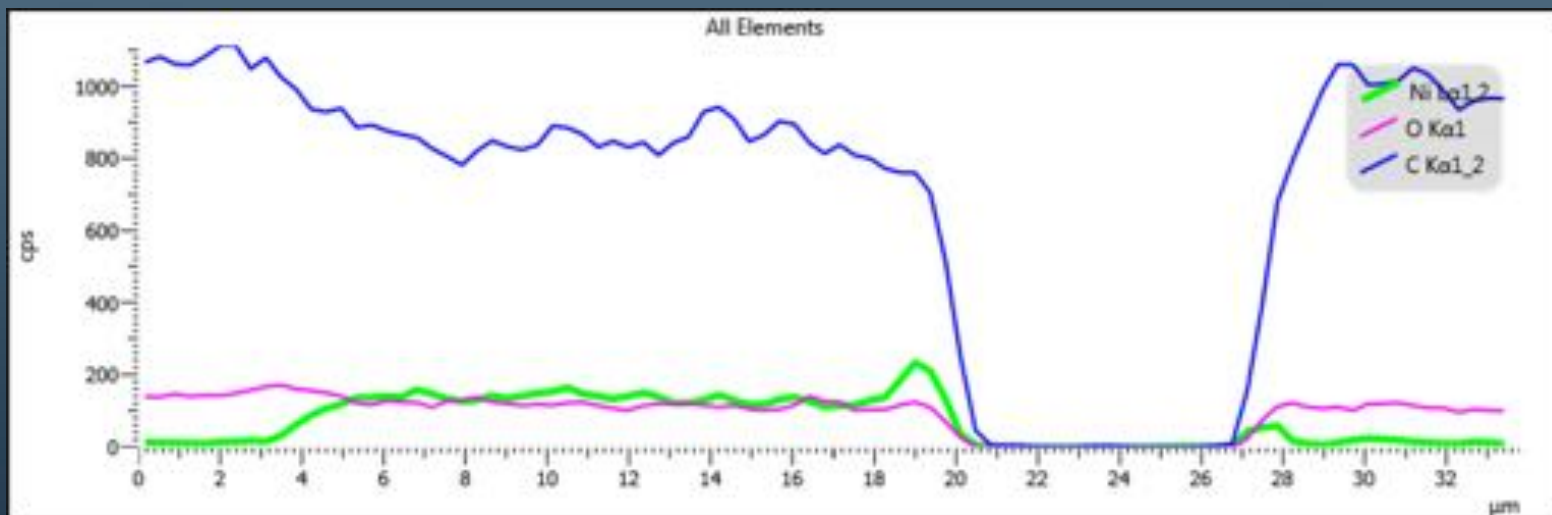
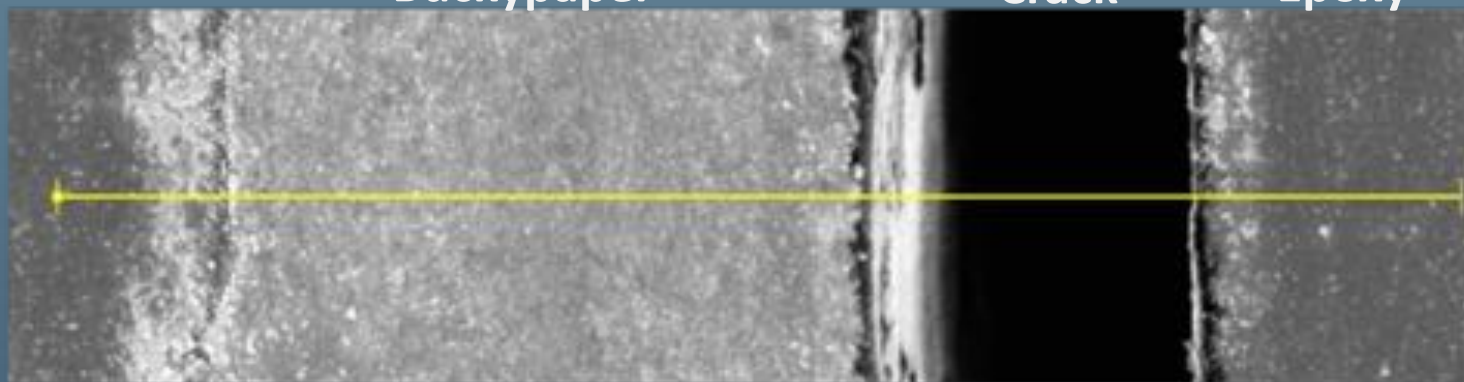
# Composition line scan across 1000 cycles NiO-reduced



Buckypaper

Crack

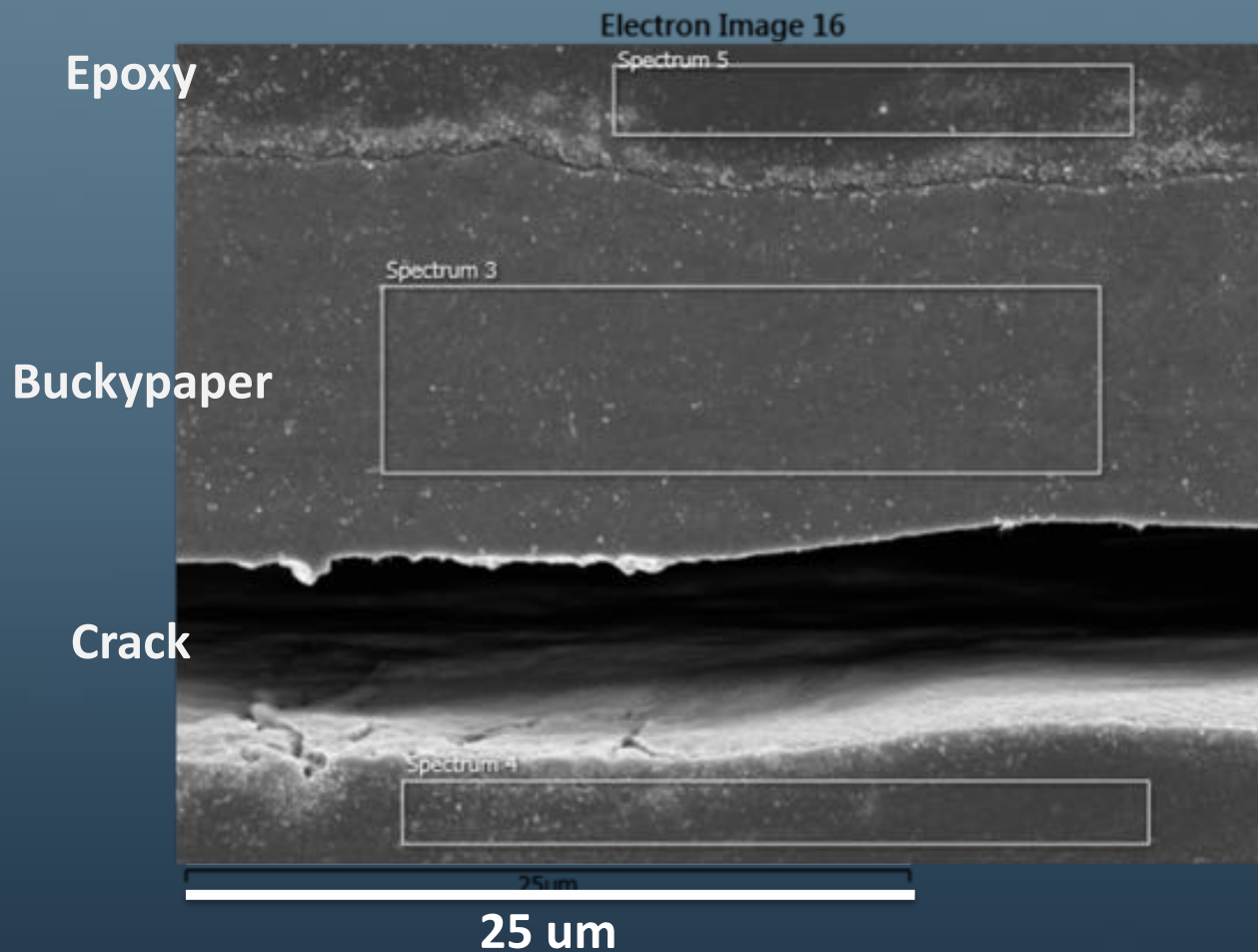
Epoxy





# 500 cycles NiO-NiO<sub>2</sub> reduced

Unreduced sample has twice the O content in the paper.



Spectrum 5	
	At%
C	90.4
O	9.6
Ni	0.0

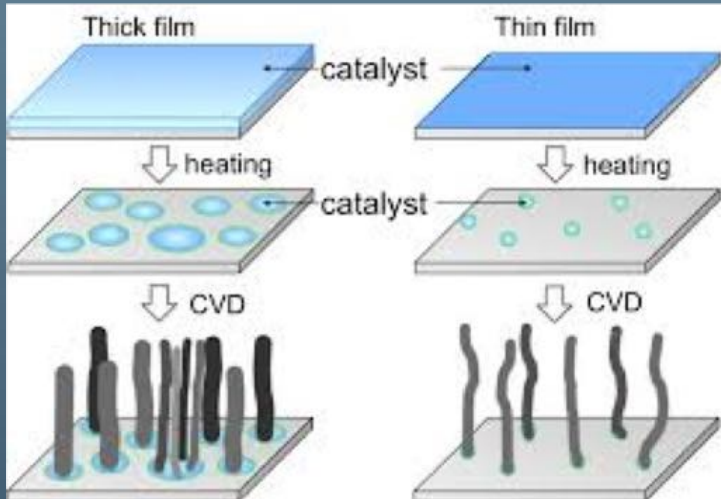
Spectrum 3	
	At%
C	87.1
O	10.2
Ni	2.7

Spectrum 4	
	At%
C	89.1
O	10.9
Ni	0.0

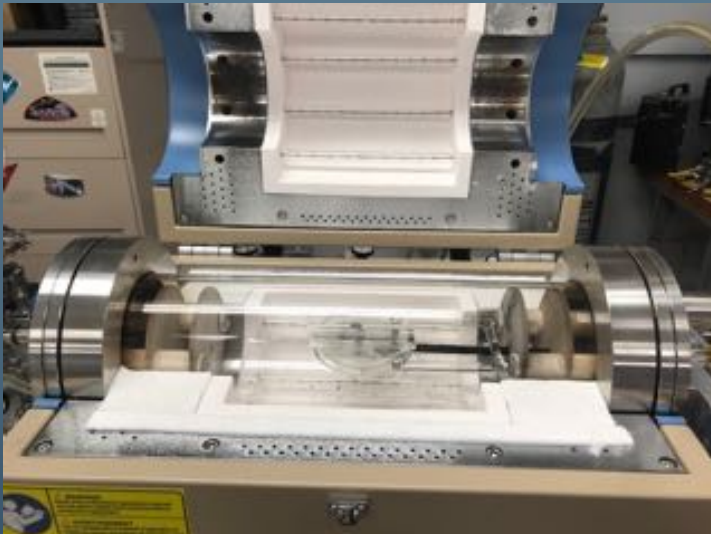


# Growth of CNT & ALD

## Typical Growth of CNTs



Substrate + Catalyst + Gas = CNNT  
Si,Ti, flat, 3d + Iron,Ni + Ethylene

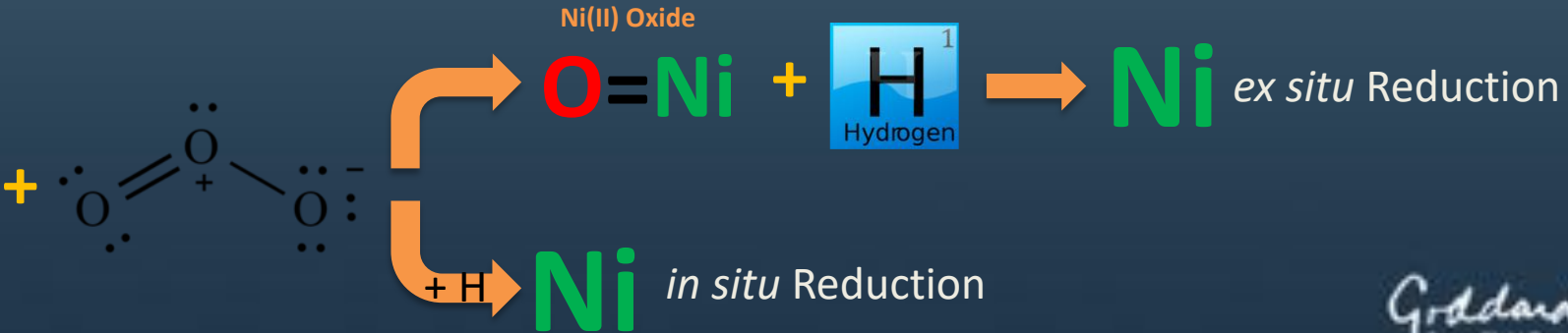
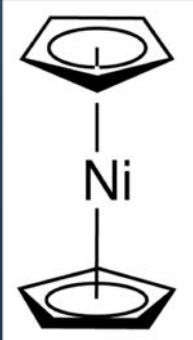


## ALD of Catalyst (Ni)

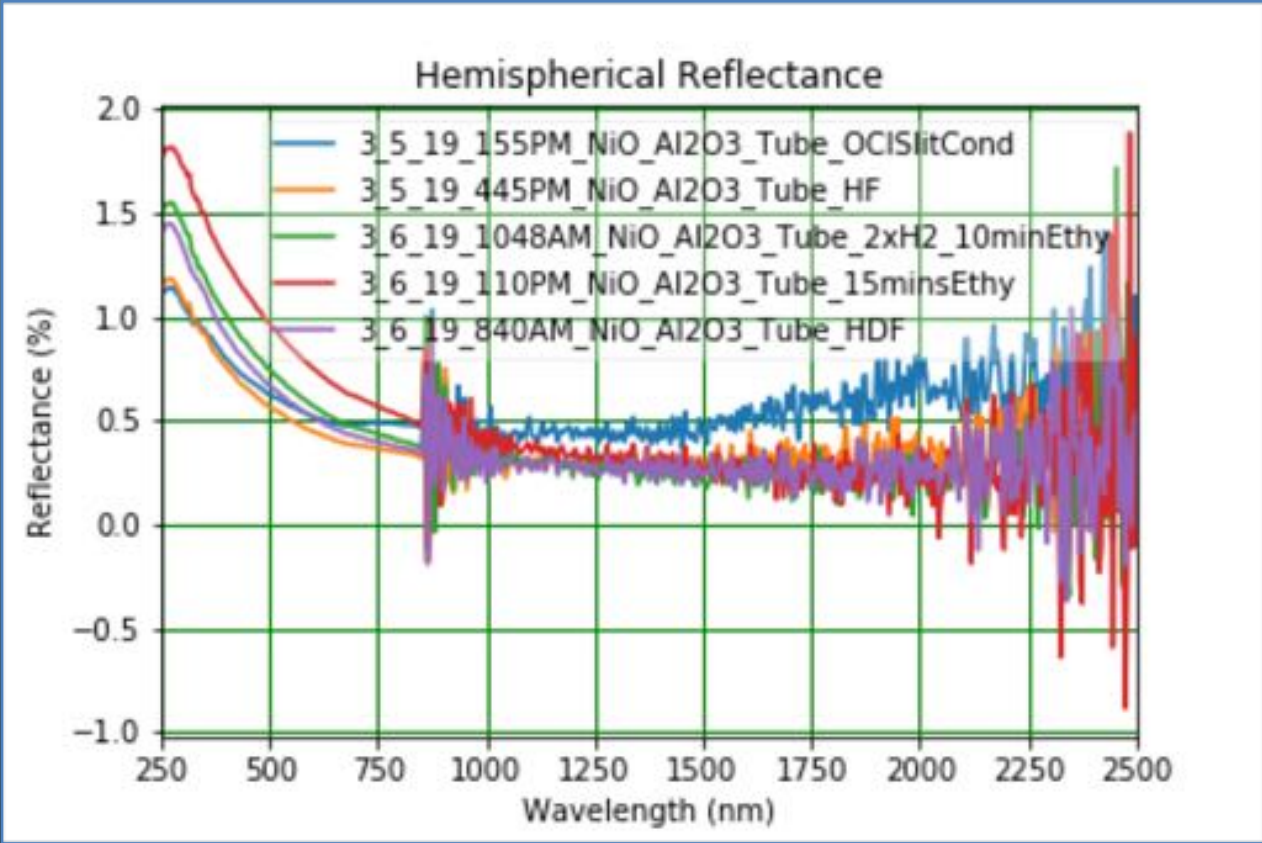
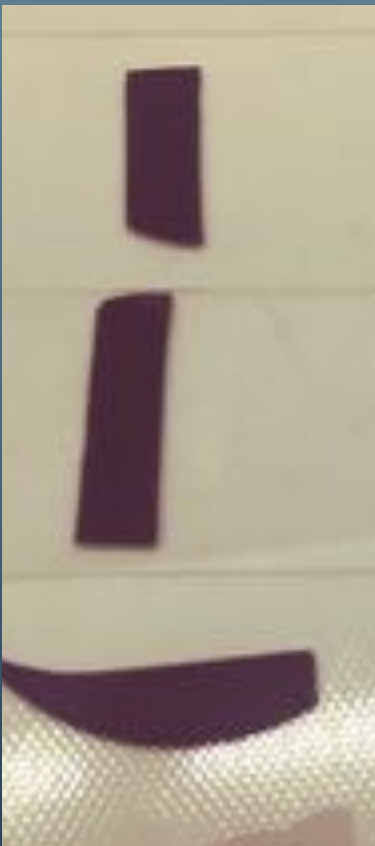
## Precursors and Reaction Pathway

Nickelocene

Ozone



# Initial Results





# Acknowledgments



Adomaitis Research Group