

A Study of an Alternative Carbon Source to Improve Environmental Sustainability in Steel Production

Blake Stewart^{1,2}, Haley Doude¹, Terry Taylor³, Morgan Abney³, and Hongjoo Rhee^{1,2}

¹Center for Advanced Vehicular Systems, Mississippi State University, Mississippi State, MS 39762

²Department of Mechanical Engineering, Mississippi State University, Mississippi State, MS 39762

³Marshall Space Flight Center, National Aeronautics and Space Administration, Huntsville, AL 35812

October 2, 2019



MISSISSIPPI STATE UNIVERSITY™
CENTER FOR ADVANCED
VEHICULAR SYSTEMS

MS&T19
MATERIALS SCIENCE & TECHNOLOGY



Motivation

- One ton iron from blast furnace = 0.33~0.44 ton CO₂^[12]
- Global iron and steel CO₂ emissions = **1.7 gigatons**^[8]
- Environmental regulations are requiring manufacturers to reduce oxocarbon emissions.
- High quality steel alloys with smaller carbon footprint.
- Potential for production cost savings.



Benxi Steel Plant, China, 2000.
(Credit: Jon Bower Pollution/Alamy Stock Photo)

Overview

- Biomass coal limited in production and cost [3,10].
- ULCOS CO₂ capture method = 10~15% reduction^[11]
- U.K. recently funded decarbonization in steel industry^[9].
- Coal has been generated from CO₂ using liquid metals^[2].
- Two different ferrous alloys are manufactured with novel carbon source compared to conventional to determine viability as steel carbon source.
- Alternative carbon was produced via a carbon sequestration system.
- Mechanical and microstructural investigation revealed comparable metallurgical properties.

Comparison of Elemental Carbon

- Conventional and alternative carbon were evaluated before alloying.
- Investigation methods
 - X-Ray Powder Diffraction (XRD)
 - Allotropy
 - Field Emission Scanning Electron Microscopy (FE-SEM)
 - Morphology



Conventional Carbon

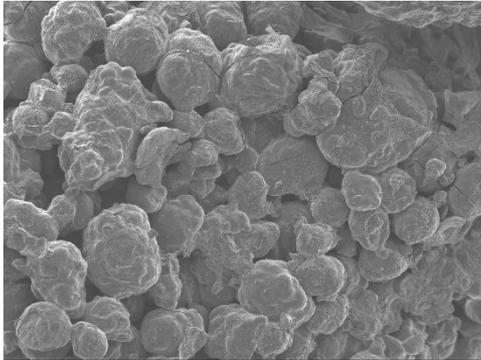


Alternative Carbon

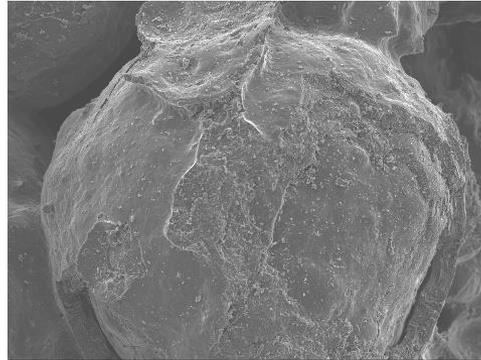
SEM reveals significant differences in morphology

Conventional C

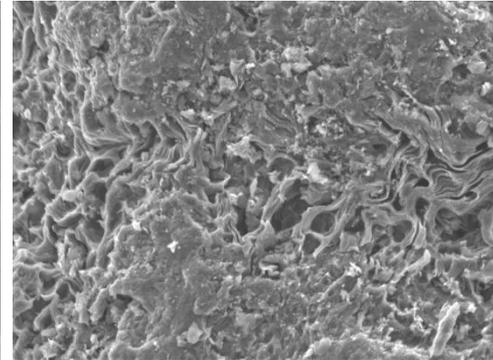
X100



X500

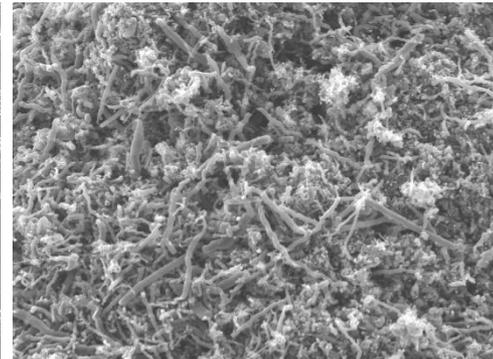
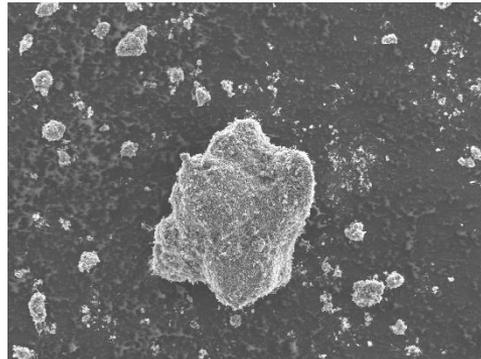
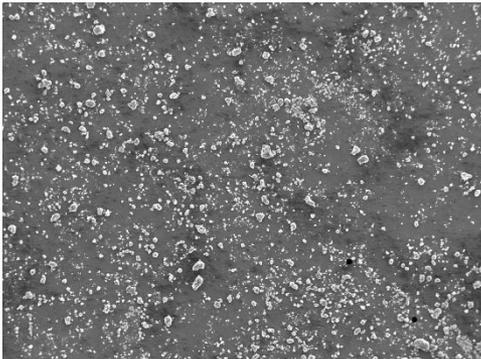


X5000



Clumps of spherical particles, 80-150µm

Alternative C



Tangled balls, 10-75µm

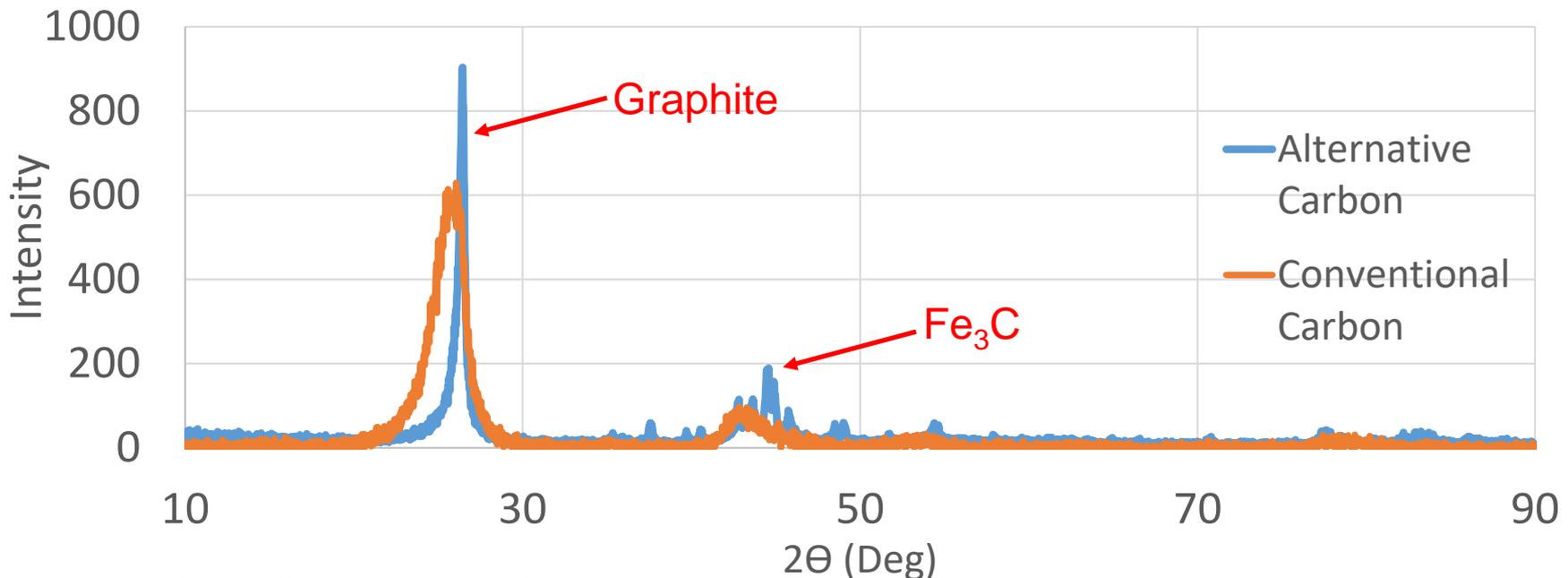
XRD shows similar crystalline structure

- Conventional Carbon

- Graphitic carbon
 - 26.1°
- Peak shift and broadening likely due to grinding prior to scan^[1]

- Alternative Carbon

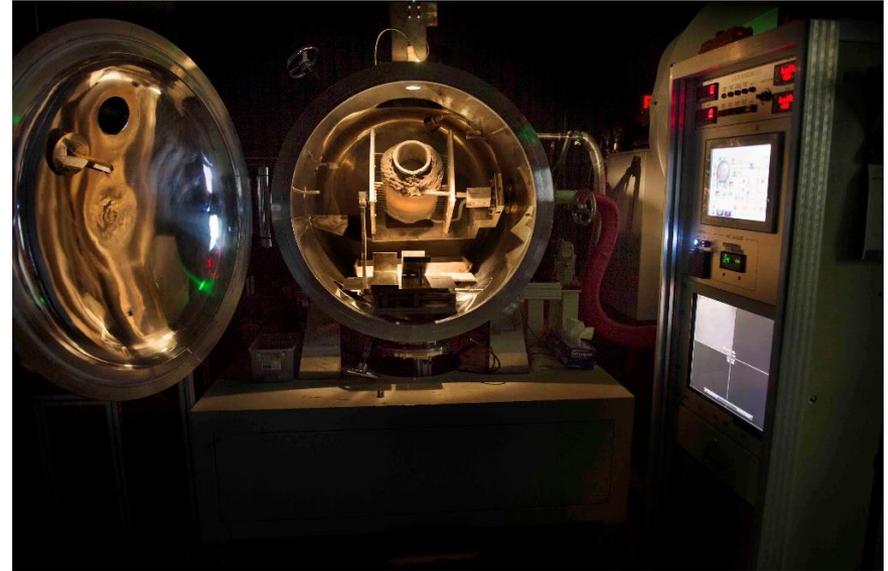
- Graphitic carbon
 - 26.4°
- Cementite Fe₃C
 - 44.6°



0.5mm glass slides, copper K-α wavelength, 10-90 degrees 2θ

Comparison of Produced Ingots

- Two ferrous alloys were produced using each carbon source
 - ▶ Low Carbon
 - AISI 1020
 - ▶ High Carbon
 - Gray Cast Iron
- Investigation methods
 - ▶ Chemical composition
 - ▶ Light microscopy
 - Phase fraction, grain size
 - ▶ Mechanical performance
 - Quasi-static tension/compression, 0.001/s until fracture
 - Brinell and Rockwell-B Hardness (additional HRC for cast iron)



Vacuum Induction Melting Furnace (VIMF)

Low Carbon Steels

- Cast in Vacuum Induction Melt Furnace (VIMF)
- Target composition
 - ▶ Carbon (C): 0.2 wt.%
 - ▶ Silicon (Si): 0 wt.%
 - ▶ Manganese (Mn): 0.45 wt.%
 - ▶ Iron (Fe): balance
- Hot rolled to 0.5"
 - ▶ 1250°C austenitizing
 - ▶ Air cooled
- **Analysis on as-rolled condition**



Hot Rolling using In-house Reversing Rolling Mill

Low carbon steel micrographs

3-D views of low carbon steel microstructures

Conventional C



Alternative C



Rolling Direction
↔

Low carbon steel grain sizes

- Average phase fractions

- ▶ Conventional:

- 82.6 ± 1.43 % ferrite

- 17.4 ± 1.43 % pearlite

- ▶ Alternative:

- 81.7 ± 0.627 % ferrite

- 18.3 ± 0.627 % pearlite

- Average grain sizes

- ▶ Conventional:

- 33.4 ± 4.95 μm (ASTM: G= 6.5-7.0)

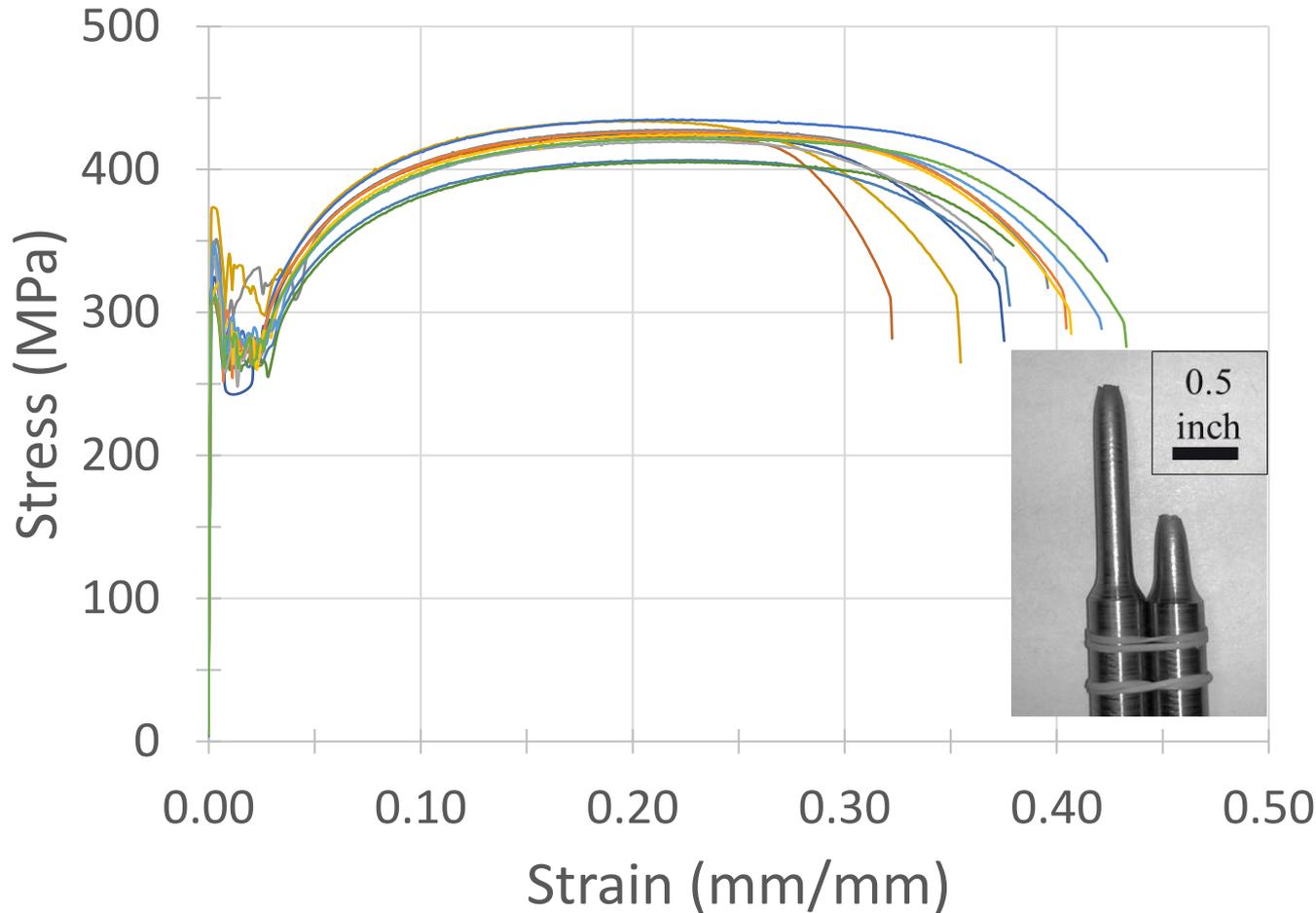
- ▶ Alternative:

- 36.6 ± 1.680 μm (ASTM: G= 6.5-7.0)



Grain size analysis of low carbon steel using conventional C by ASTM E112-13: Planimetric (Jefferies) Procedure

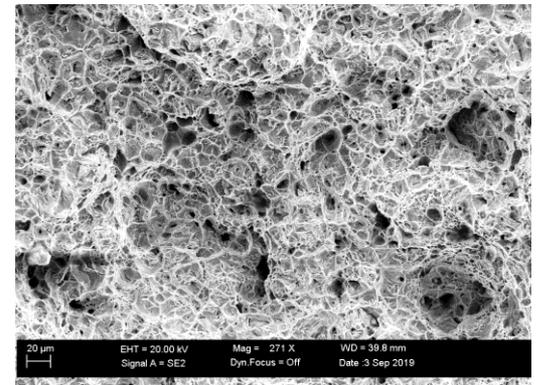
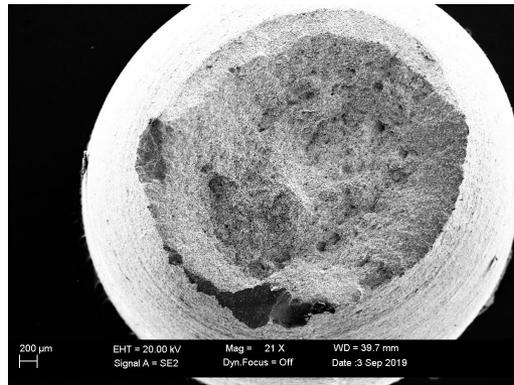
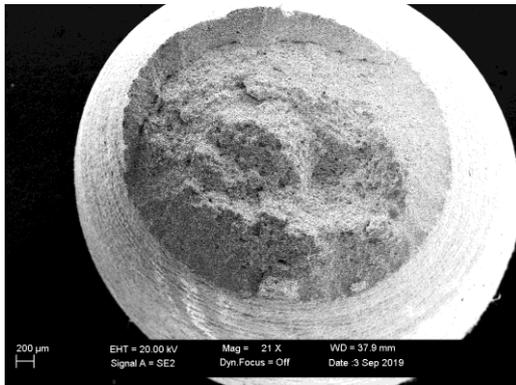
Low carbon steel tensile testing results



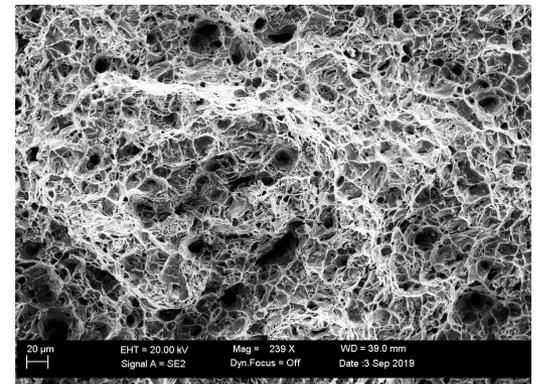
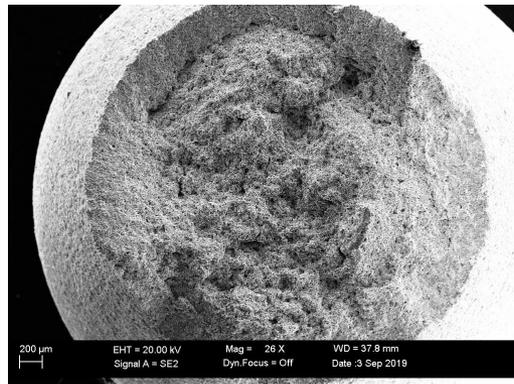
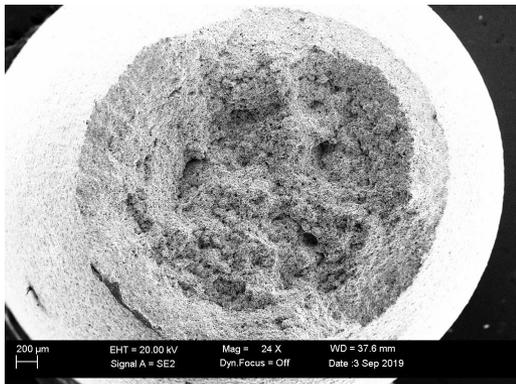
— L-C1 RD — L-C1 TD — L-C2 RD — L-C2 TD — L-C3 RD — L-C3 TD
— L-A1 RD — L-A1 TD — L-A2 RD — L-A2 TD — L-A3 RD — L-A3 TD

- L-C2 slightly higher σ_{uys} due to smaller grain size
 - Grain size reduced from colder rolling issue
- Similar to standard values for performance

Low carbon steel tensile fracture surfaces



Fracture surfaces of Conventional Carbon tensile specimen



Fracture surfaces of Alternative Carbon tensile specimen

Low carbon steel hardness

- Average hardness
 - ▶ HB, HRB, and
 - ▶ HRC (cast irons only)
- Industry standard for AISI 1020 around 111 HB

Alloy	Hardness	
	HB	HRB
L-C	116 ±	60.14 ±
	6.432	7.540
L-A	113 ±	60.92 ±
	5.481	3.103

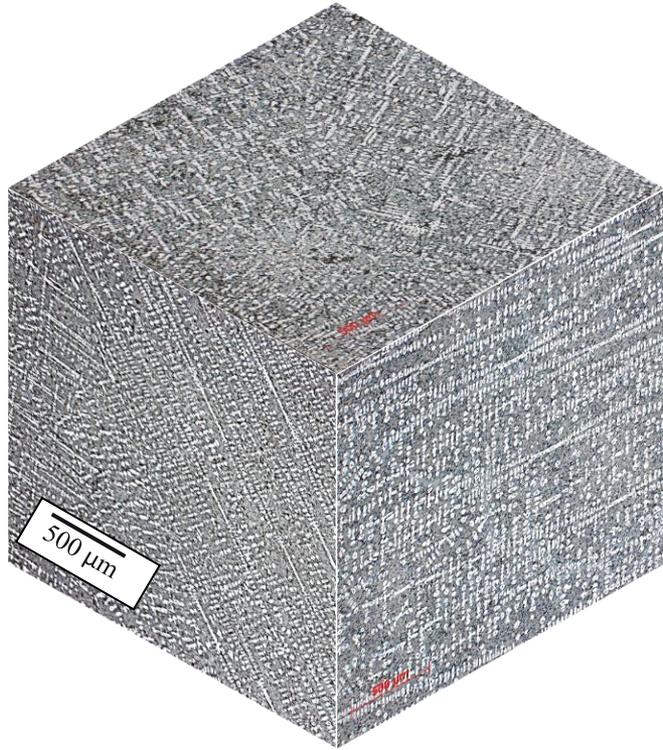
Cast Irons

- Cast in Vacuum Induction Melt Furnace (VIMF)
- Target composition
 - ▶ Carbon (C): 3.5 wt.%
 - ▶ Silicon (Si): 2.5 wt.%
 - ▶ Manganese (Mn): 0.45 wt.%
 - ▶ Iron (Fe): balance
- **Analysis on as-cast condition**



Cast iron micrographs

- Conventional C



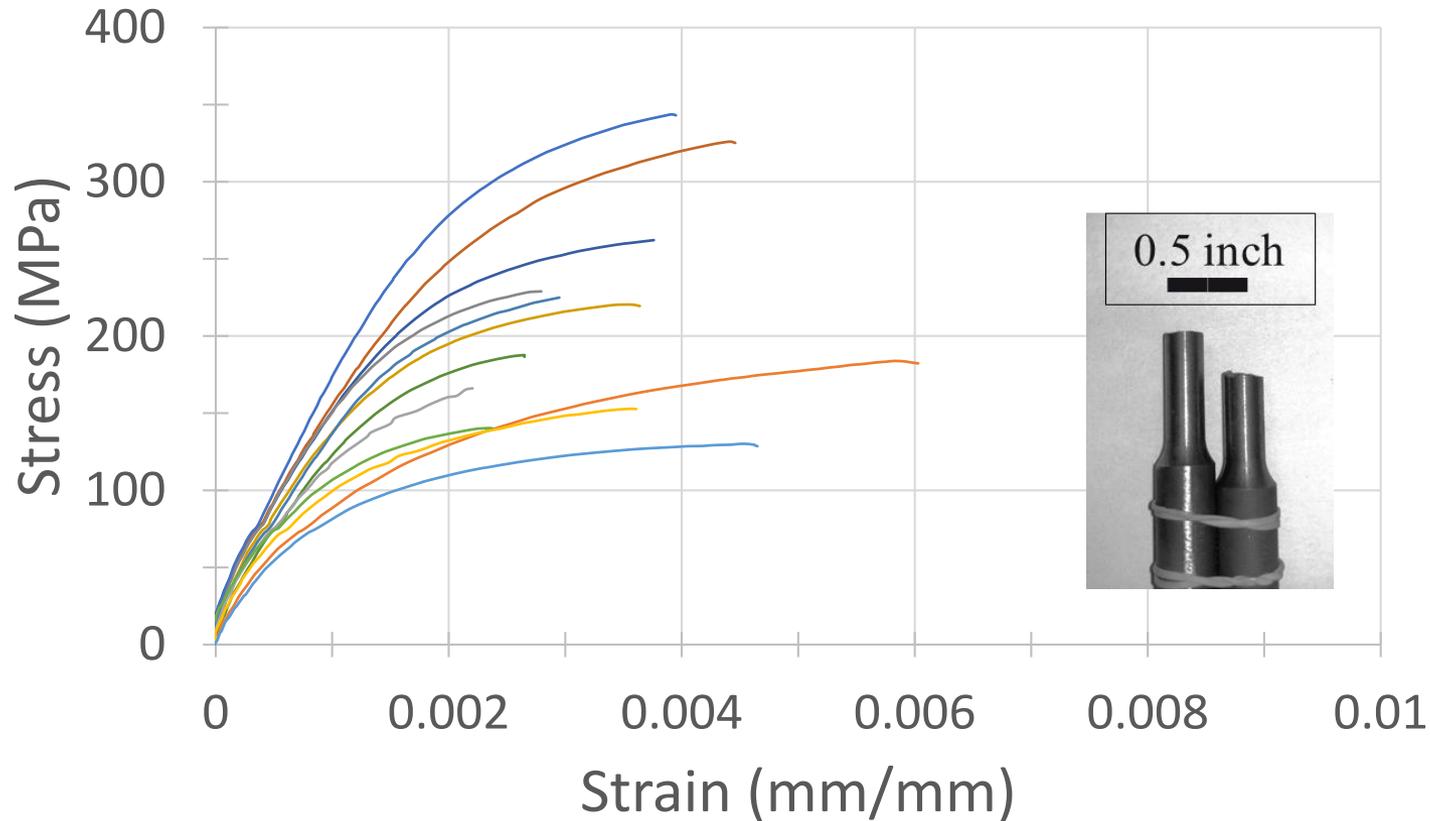
- Alternative C



Casting Direction ↑

Phase fraction / Grain structure:
Distribution D, Class 7

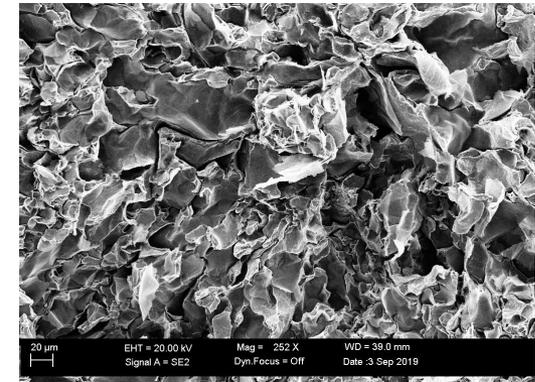
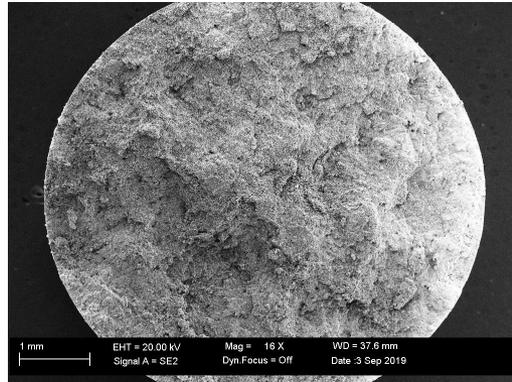
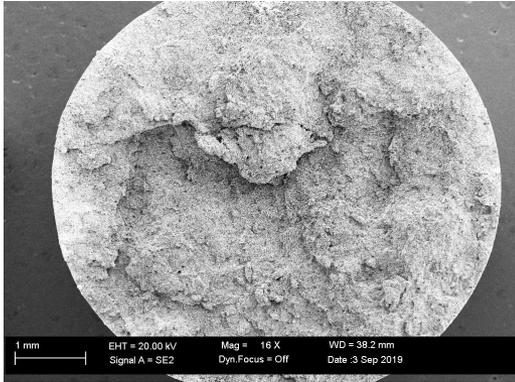
Cast iron tensile test results



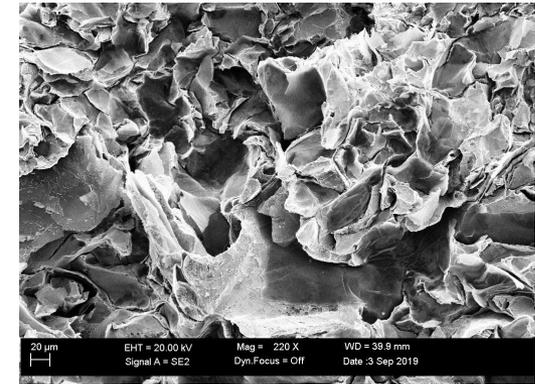
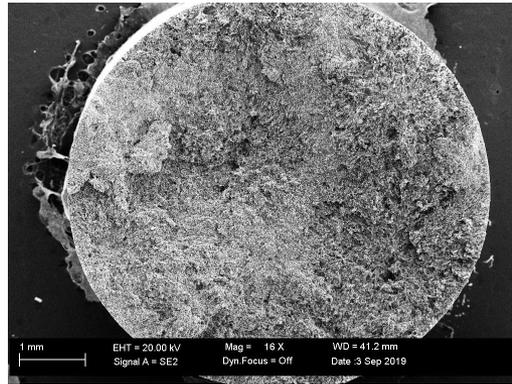
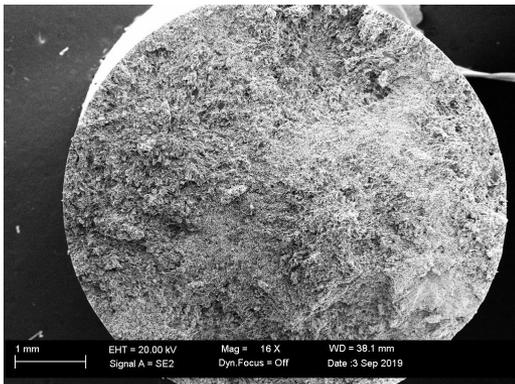
- Significant variability
- But, consistent when testing from the same ingot

— C-C1 RD — C-C1 TD — C-C2 RD — C-C2 TD — C-C3 RD — C-C3 TD
— C-A1 RD — C-A1 TD — C-A2 RD — C-A2 TD — C-A3 RD — C-A3 TD

Cast iron fracture surfaces



Fracture surface of Conventional Carbon tensile specimen



Fracture surface of Alternative Carbon tensile specimen

Cast iron hardness

- Average hardness
 - ▶ HB, HRB, and HRC
- Similar hardness for all cast iron with one exception
 - ▶ C-C1: ~250 HB
- Industry range from 120-550 HB

Alloy	Hardness		
	HB	HRB	HRC
C-C	193 ± 48.989	83.11 ± 1.925	4.47 ± 1.220
C-A	167 ± 6.763	82.84 ± 1.830	2.86 ± 1.150

- Similar in composition and hardness to SAE J431 automotive GCI
 - ▶ <187 HB

Summary / Discussion

- A novel carbon source was studied to determine if alternative carbon produces similar metallurgical results as conventional carbon
 - Two ferrous alloys, 1020 and grey cast iron, were manufactured
 - Low carbon alloys show comparable structure and properties for both carbon sources
 - Cast iron shows significant variance in properties
 - Believed to be caused by cooling rate inequalities throughout the ingot
 - Cooling significantly affects mechanical properties [6,7]
 - Implies alternative carbon could be used for numerous alloys and different solidification rates and heat treatments
 - Mechanical and microstructural investigation reveals comparable metallurgical properties
- ⇒ **The alternative carbon source showed it is possible to use as the elemental carbon source for steel making**

Acknowledgements

- Marshall Space Flight Center (MSFC), National Aeronautics and Space Administration (NASA)
- Environmental Control and Life Support System Group at MSFC, NASA
- Center for Advanced Vehicular Systems (CAVS), Mississippi State University (MSU)

References

- [1] B. Cullity and S. Stock, *Elements of X-ray diffraction*. [Reading]: Addison-Wesley Publishing Company, 1978, pp. 340-341.
- [2] D. Esrafilzadeh, A. Zavabeti, R. Jalili, P. Atkin, J. Choi, B. Carey, R. Brkljača, A. O'Mullane, M. Dickey, D. Officer, D. MacFarlane, T. Daeneke and K. Kalantar-Zadeh, "Room temperature CO₂ reduction to solid carbon species on liquid metals featuring atomically thin ceria interfaces", *Nature Communications*, vol. 10, no. 1, 2019 [Online]. Available: <https://doi.org/10.1038/s41467-019-08824-8>
- [3] E. Mousa, C. Wang, J. Riesbeck and M. Larsson, "Biomass applications in iron and steel industry: An overview of challenges and opportunities", *Renewable and Sustainable Energy Reviews*, vol. 65, pp. 1247-1266, 2016.
- [4] J. Davis, *ASM Specialty Handbook: Cast Irons*. Russel: ASM International, 1996.
- [5] J. Rodrigues, "Blast-Furnace Stoichiometry - PDF Free Download", *kundoc.com*, 2019. [Online]. Available: <https://kundoc.com/pdf-blast-furnace-stoichiometry-.html>. [Accessed: 12- Sep- 2019]
- [6] L. Collini, G. Nicoletto and R. Kocna, "Microstructure and mechanical properties of pearlitic gray cast iron," *Materials Science and Engineering: A*, vol. 488, issue 1-2, pp. 529-539, 2008.
- [7] M. Jabbari Behnam, P. Davami and N. Varahram, "Effect of cooling rate on microstructure and mechanical properties of gray cast iron", *Materials Science and Engineering: A*, vol. 528, no. 2, pp. 583-588, 2010.
- [8] Davis, S., Lewis, N., Shaner, M., Aggarwal, S., Arent, D., Azevedo, I., Benson, S., Bradley, T., Brouwer, J., Chiang, Y., Clack, C., Cohen, A., Doig, S., Edmonds, J., Fennell, P., Field, C., Hannegan, B., Hodge, B., Hoffert, M., Ingersoll, E., Jaramillo, P., Lackner, K., Mach, K., Mastrandrea, M., Ogden, J., Peterson, P., Sanchez, D., Sperling, D., Stagner, J., Trancik, J., Yang, C. and Caldeira, K. (2018). Net-zero emissions energy systems. *Science*, 360(6396), p.eaas9793.
- [9] "Steel News", *Aist.org*, 2019. [Online]. Available: <https://www.aist.org/news/steel-news/2019/august/26-30-august-2019/u-k-establishes-funds-to-support-steel-decarboniza>. [Accessed: 01- Sep- 2019]
- [10] T. Norgate, N. Haque, M. Somerville and S. Jahanshahi, "Biomass as a Source of Renewable Carbon for Iron and Steelmaking", *ISIJ International*, vol. 52, no. 8, pp. 1472-1481, 2012.
- [11] "ULCOS = Ultra Low CO₂ Steel Making", *Sustainableinsteel.eu*, 2019. [Online]. Available: https://www.sustainableinsteel.eu/p/532/ulcos_-_ultra_low_co2_steel_making.html. [Accessed: 06- Feb- 2019]
- [12] "United States Environmental Protection Agency", *Epa.gov*, 2019. [Online]. Available: <https://www.epa.gov/sites/production/files/2016-11/documents/iron-steel-ghg-bact-2012.pdf>. [Accessed: 10- Sep- 2019]