Electrochemistry's Role in Current and Future Environmental Control and Life Support Systems

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What is ECLSS?



Current Electrolytic ECLSS Applications



Future Electrolytic ECLSS Applications

Environmental Control and Life Support Systems (ECLSS) keep astronauts alive and happy in space.





ECLSS is vital to current/future exploration missions by lowering Earth reliance by utilizing byproducts.





Electrochemistry is currently implemented onboard the N_2 ISS in the O₂ Generation Assembly (OGA).



The OGA electrolyzes water to produce oxygen required for crew.





OGA consist of nine Orbital Replacement Units (ORUs).





ORU	
1	Water
2	Inlet Deionizing Bed
3	Hydrogen
4	Recirculation Pump
5	Nitrogen Purge
6	Oxygen Outlet
7	Hydrogen Sensor
8	Power Supply Module
9	Process Controller

Over 16,184 lbm of O₂ and 2,023 lbm of H₂ has been produced by OGA as of April 21, 2019.¹





Applications of electrochemistry are currently under investigation for future CO₂ Reduction systems.











For future long duration missions, O_2 recovery rate must be a minimum of 75% with target goal of 90%.





Electrolytic technologies are being investigated to increase the recovery rate for long duration missions.





MICROFLUIDIC ELECTROCHEMICAL REACTOR: CO₂ to O₂ + ETHYLENE





ELECTRODEPOSITION OF CATALYSTS

ELECTROCHEMICAL SEPARATION/PURIFICATION OF H₂

The current exploration baseline O₂ recovery architecture includes SOA CRA with the addition of PPA.



PPA generates a H_2/CH_4 plasma to convert CH_4 to H_2 and C_2H_2 , ultimately requiring a H_2 separator for success.



An electrochemical H_2 separator, that separates H_2 and C_2H_2 from the PPA outlet stream, is under investigation.



H₂ is electro-oxidized to protons and electrons and the protons are electro-reduced in another chamber.



Alternative technologies, other than PPA, are being investigated to increase the recovery rate to >50%.





ELECTROCHEMICAL REACTORS: >70%

An alternative electrolytic CRA, would eliminate 3 pieces of baseline O_2 Recovery Architecture hardware.





Electrochemical reduction of CO_2 to ethylene (C_2H_4) using water as a proton source.





 $CO_2 + H_2O \rightarrow 0.5C_2H_4 + 1.5O_2$ Theoretical O_2 Recovery Rate: 73%





3D Multi-physics model developed to optimize the EDU design and operation.





Sabatier/PPA results in ~85% O₂ recovery and a MFECR results in 73%, but 100% recovery may be possible.



BOSCH TECHNOLOGY: 100%

100% O₂ recovery is possible with Bosch technology, but carbon fouling of catalysts is a major disadvantage.



Historical Bosch systems produce 1kg of C(s) per day. C(s) fouling would result in significant catalyst resupply.





Elimination of catalyst resupply can be achieved by a fully regenerable system that utilizes Ionic Liquids.



Electrodeposition of Fe onto catalyst substrates will eliminate catalyst resupply for Bosch technology.



Feasibility studies prove that catalyst substrates can be generated by electrodeposition utilizing ILs.



Electroplating Chamber



Substrate Pre-plating



Substrate Post-plating

Feasibility studies prove that IL can be used to extract Fe and regenerate the carbon fouled catalyst.



Current efforts are underway to build an IL regenerable carbon formation reactor.





CAD Model of IL Test Stand

Electrochemistry for ECLSS

Paving the way for future human exploration beyond our solar system.



References



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²Murdoch, K., Blanchard, R., Mukerjee,S., Stracensky, T., Sharma,M., Pavlicek,R., DeCastro,E., Greenwood,Z., "Closed Loop Hydrogen Recovery Enabled by Electrochemical Hydrogen Separation", ICES-2019-150, 49th International Conference on Environmental Systems, Boston, Massachusetts, July 2019.

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