

Boron Nitride Nanotubes and Nanotube Mats

AVREY CARIFA¹, SANG-HYON CHU², AND CHEOL PARK³

¹DEPARTMENT OF MECHANICAL ENGINEERING, UNIVERSITY OF VERMONT, BURLINGTON, VT

²NATIONAL INSTITUTE OF AEROSPACE, HAMPTON, VA

³NASA LANGLEY RESEARCH CENTER, HAMPTON, VA

OCTOBER 11, 2019

Avrey Carifa

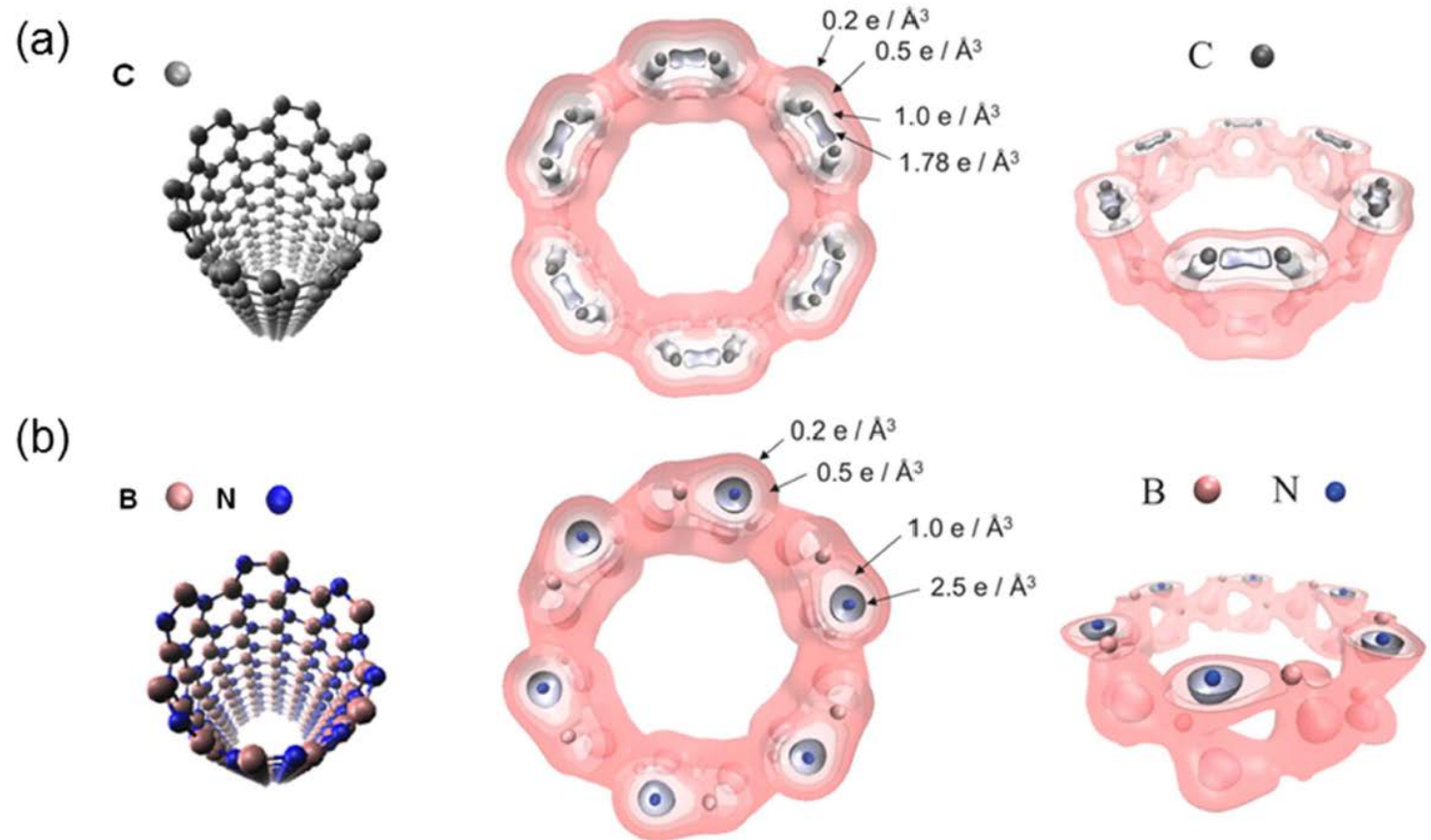
- 19 years old from Old Greenwich, CT
- Mechanical Engineering Major at University of Vermont
 - Computer Science Minor
- Interesting Facts:
 - Member of the Delta Delta Delta sorority
 - Play on the UVM Club Quidditch team
 - Read an average of 50 books a year



Introduction

Boron Nitride Nanotubes (BNNTs)

- Withstand high temperatures up to 900°C
- Flexible
- Lightweight
- Electrical Insulator
 - CNTs are conductive leading to galvanic corrosion
 - Beneficial for metal matrix composites used as conductors



(a) Carbon Nanotube (CNT)
(b) Boron Nitride Nanotube (BNNT)

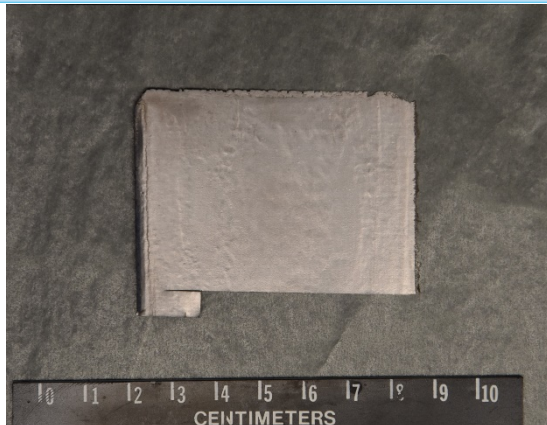
Objective

Boron Nitride Nanotube (BNNT) Nanocomposites

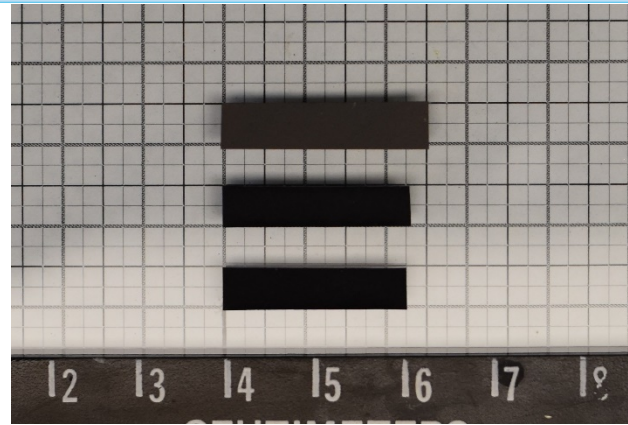


Thin, Dense, Flexible Mats

Metal Matrix Composites (MMC)



Ceramic Matrix Composites (CMC)



Polymer Matrix Composites (PMC)



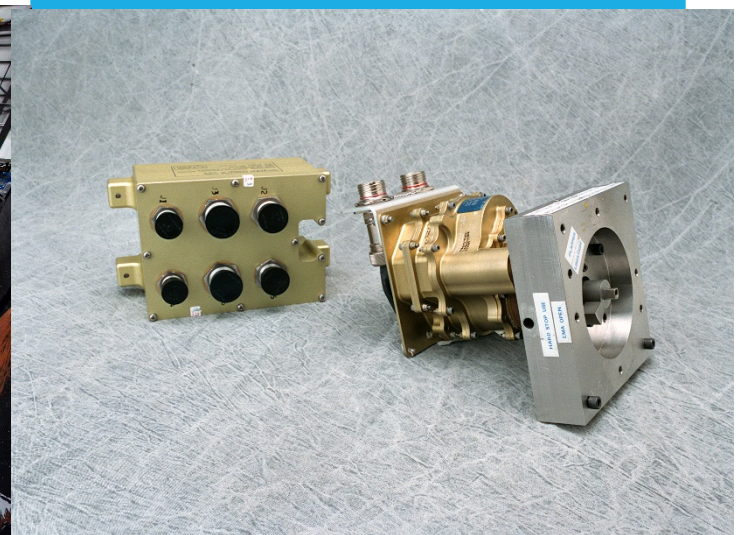
Potential Applications



High Temperature Applications

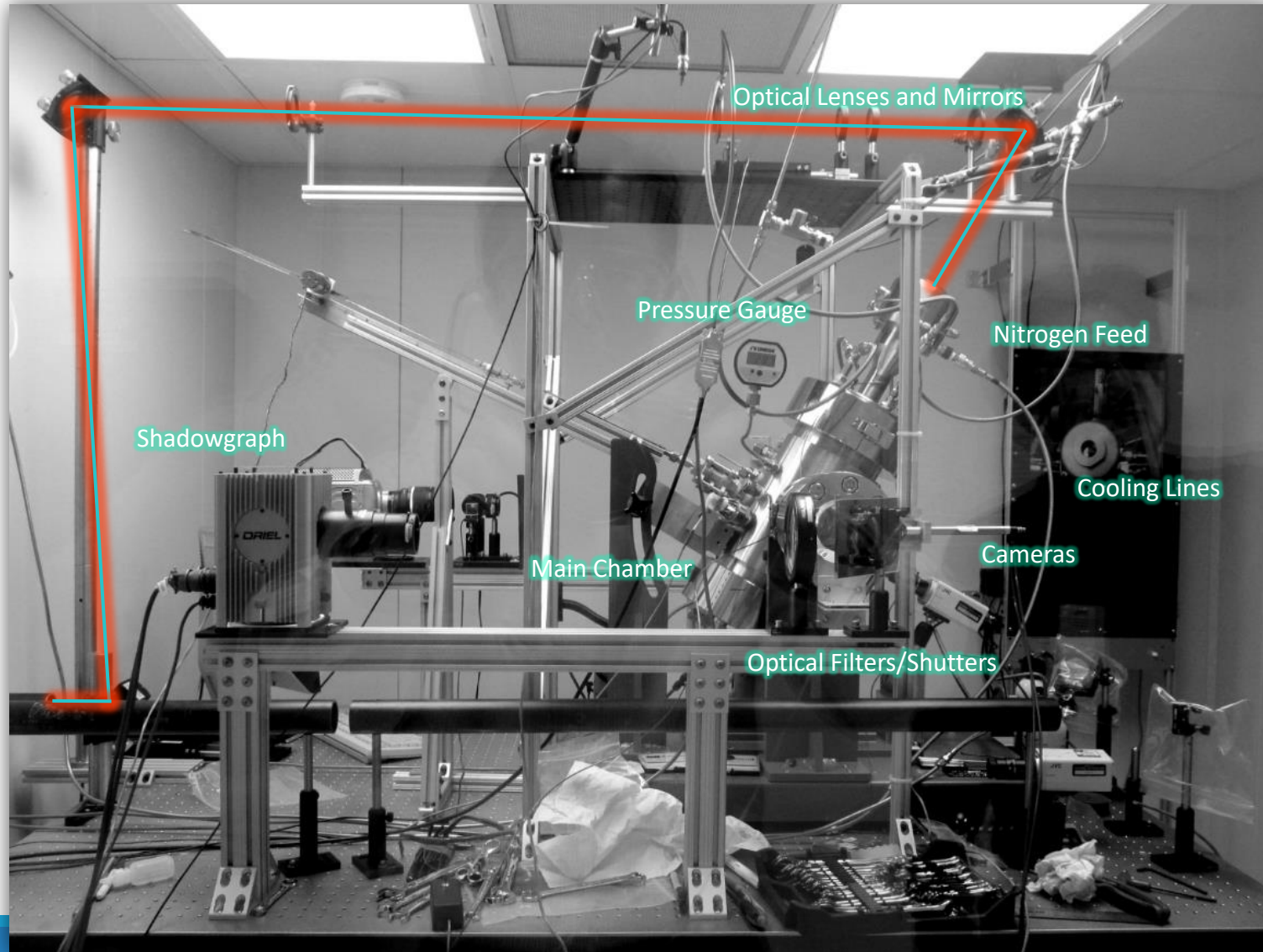


Space Radiation Shielding



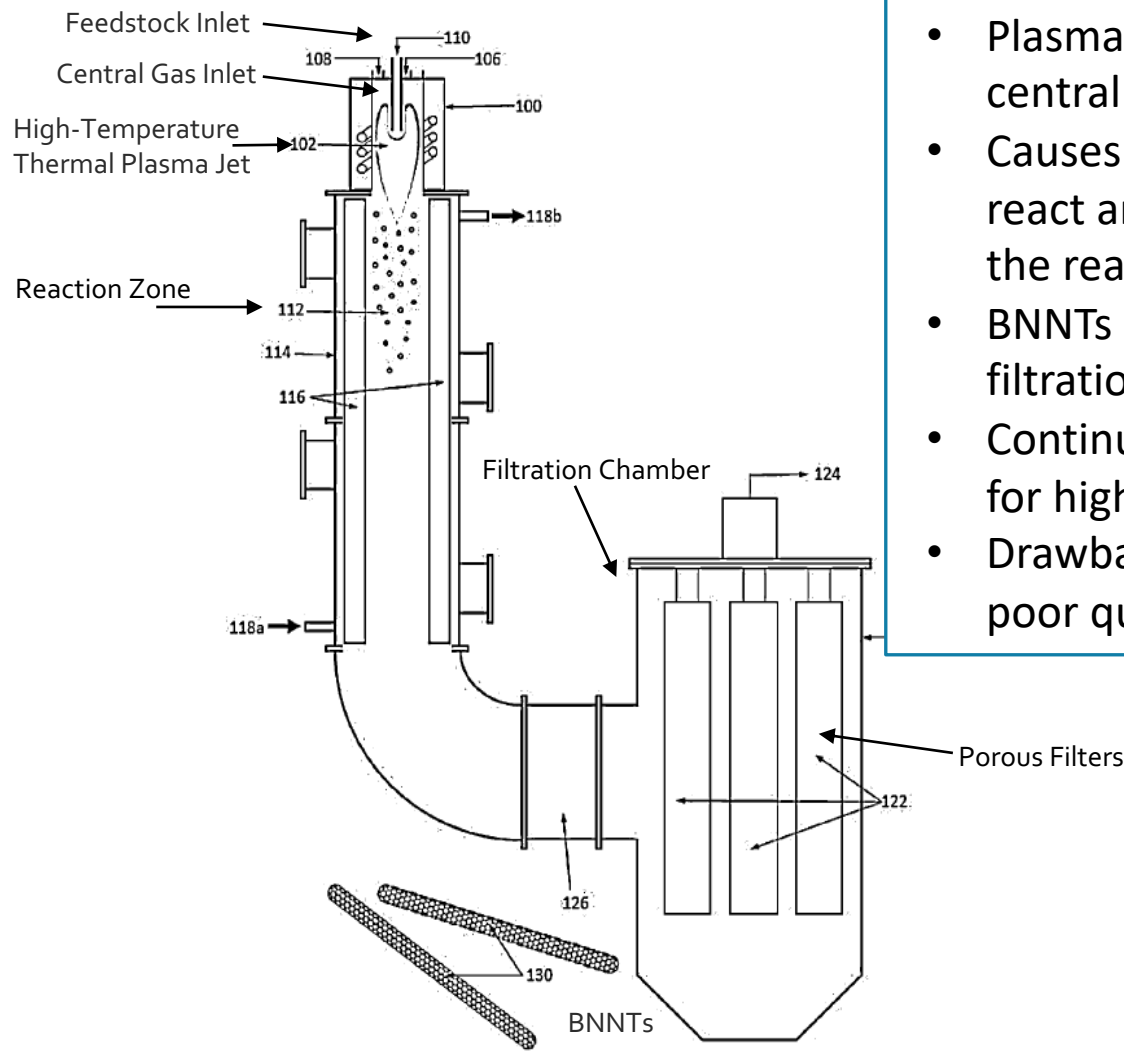
Sensors and Actuators

High Temperature-Pressure (HTP) Method



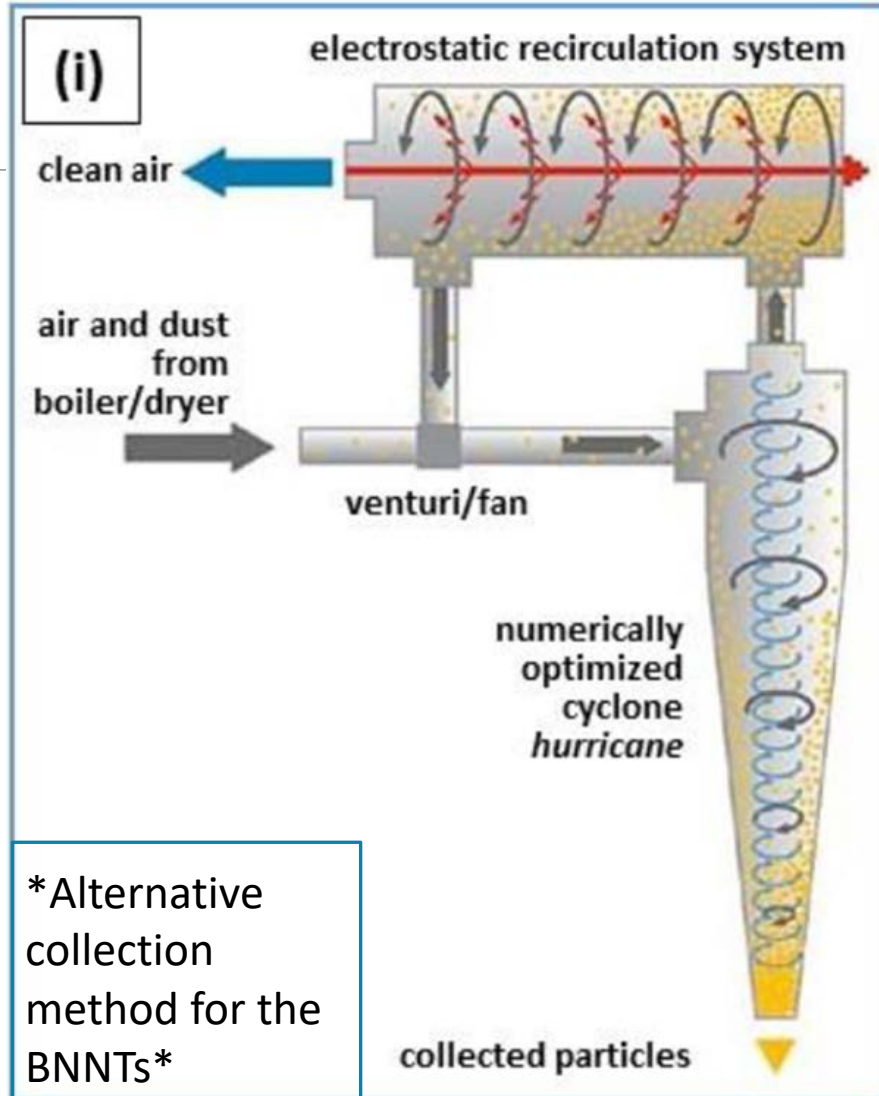
- CO₂ laser heats a boron nitride target
- Nitrogen and boron react and condense into BNNT
- Temperature: 4000K
- Pressure: 0.7-1.4 Mpa
- Produces high-quality BNNTS

Plasma Method

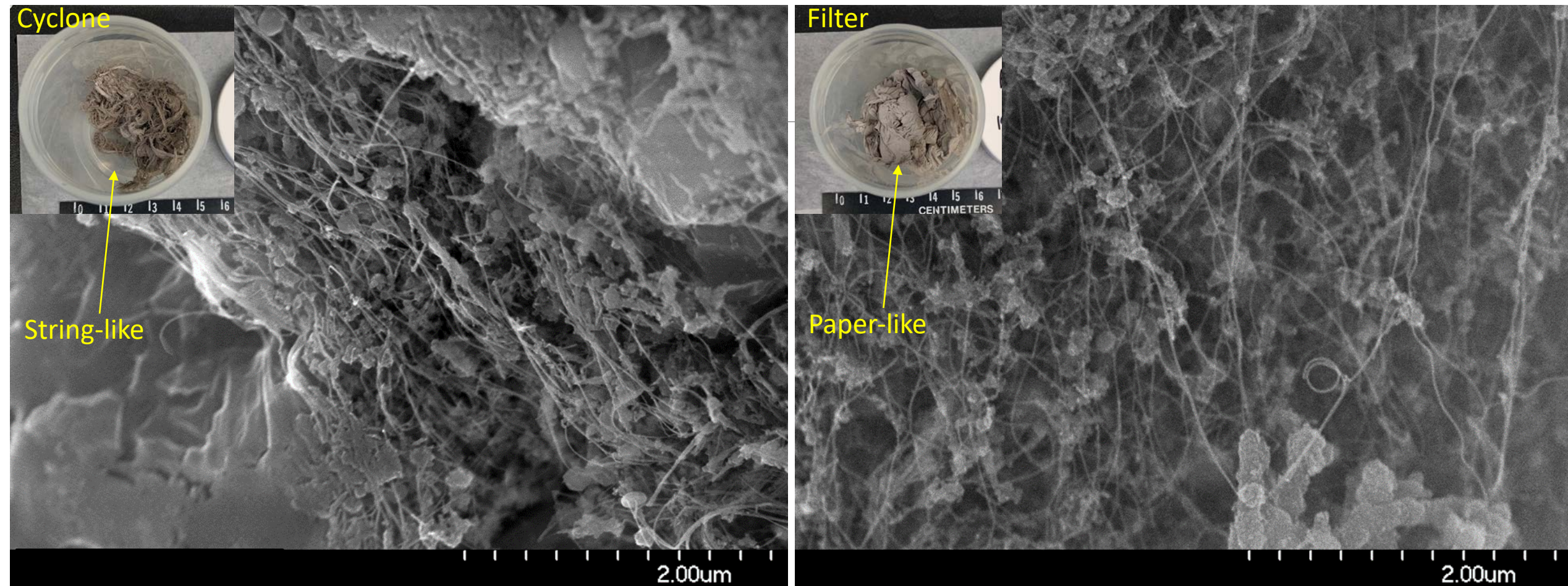


- Plasma torch heats a central gas to 8000K
- Causes h-BN feedstock to react and form BNNTs in the reaction chamber
- BNNTs collected in the filtration chamber
- Continuous process allows for high yield rate
- Drawback is the BNNTs are poor quality

Cyclone Filter:

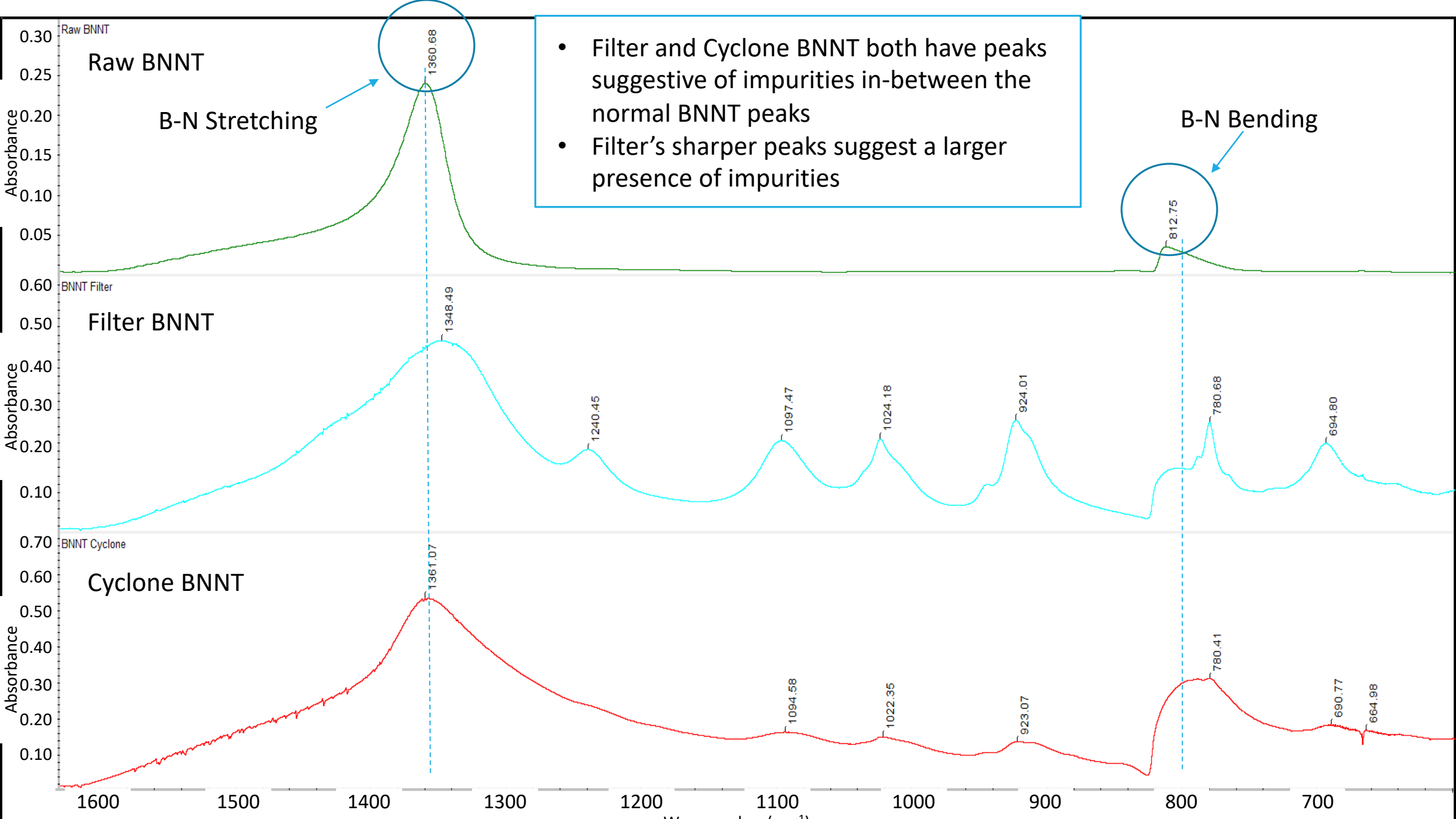


Cyclone and Filter Comparison (x 25k)



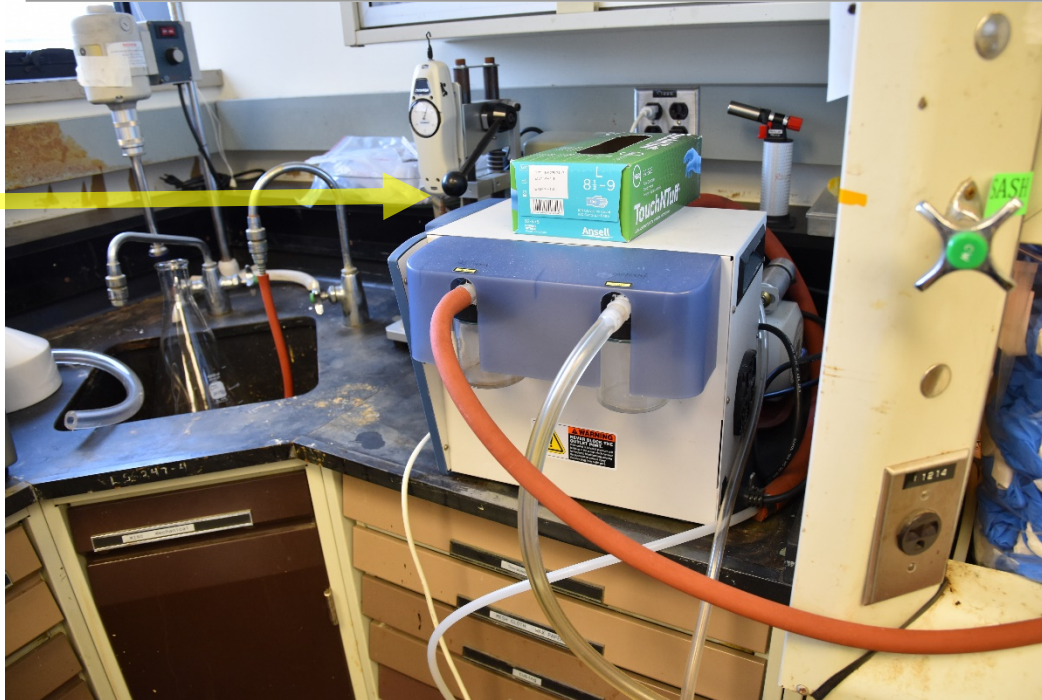
Observations:

- No apparent differences in the SEM images for the Cyclone and Filter BNNT samples
- Both have large amounts of impurities visible



Mat Making Set-Up

Vacuum



Glass Filter

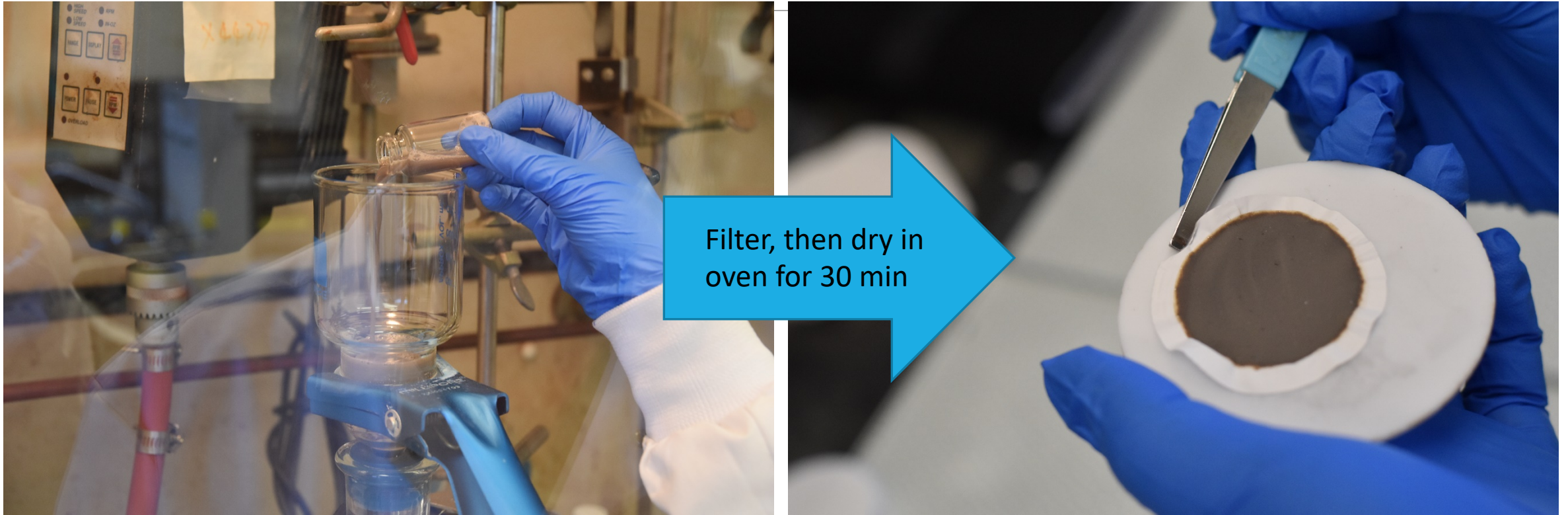
Filtration Apparatus



Clamp

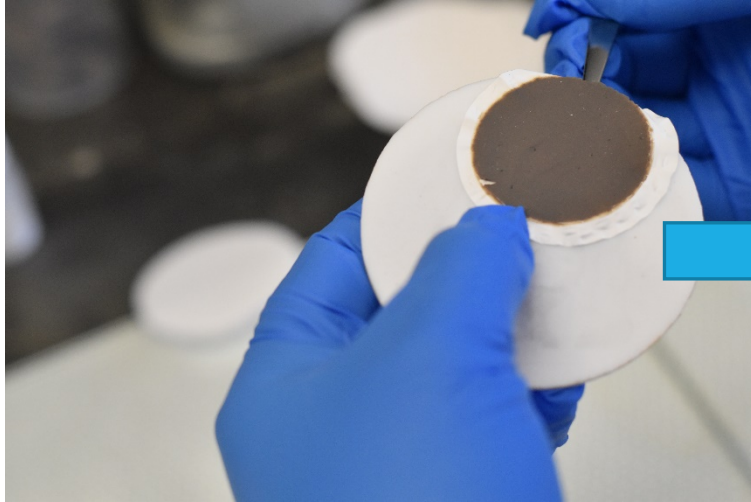
- A Teflon filter is placed on the glass filter with a funnel clamped on top
- Vacuum is used to filter out the excess water and surfactant in the BNNT solution

Mat Making Procedure



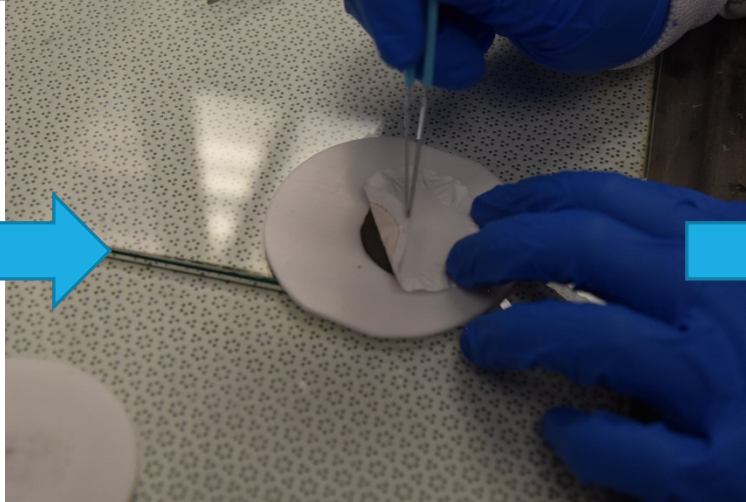
- BNNT solution is poured in the funnel and onto the filter, then the vacuum is turned on for filtering
- Once mat is barely dry, methanol is poured in to wash away excess surfactant
- When filtration is complete, mat and filter are placed in-between two Teflon plates and put in oven

Mat Peeling Procedure



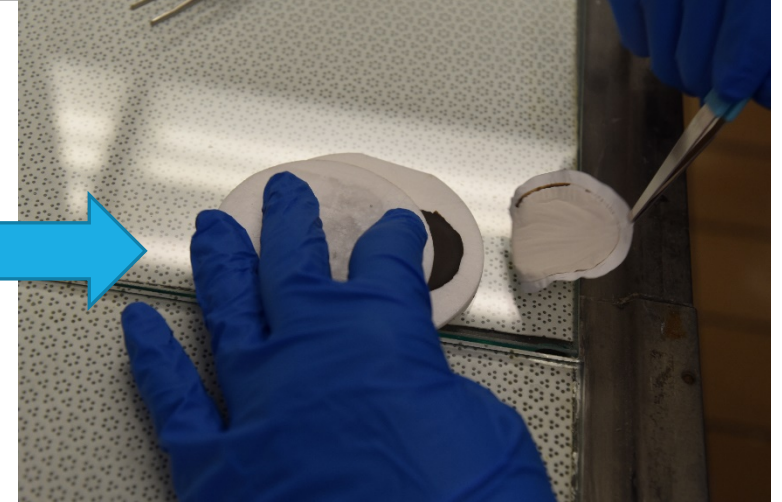
Original Procedure:

- Use the edge of the Teflon plate to free the edges by pulling on the filter with tweezers
- Repeat to gradually free the entire mat



Challenges faced:

- Time consuming (usually around an hour per mat)
- Inconsistent successes
- Damaged mats



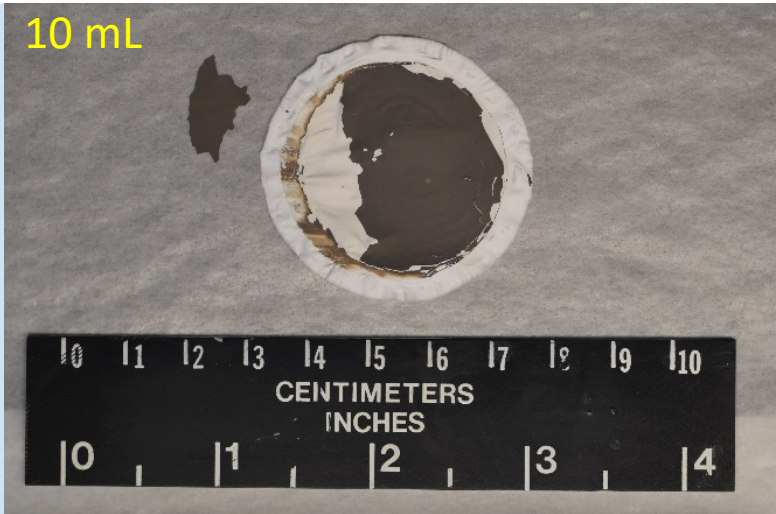
Solution:

- Free edges using original method
- Flip mat over and pull back filter to reveal freed edge
- Hold the freed mat in place with another plate and pull the filter off the mat

Unsuccessful mats

Filter Mats

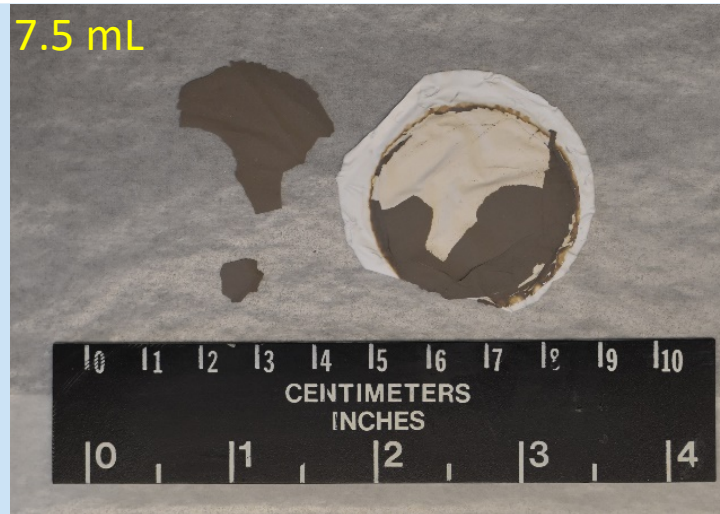
10 mL



- Crumbled as soon as peeling was attempted

Cyclone Mats

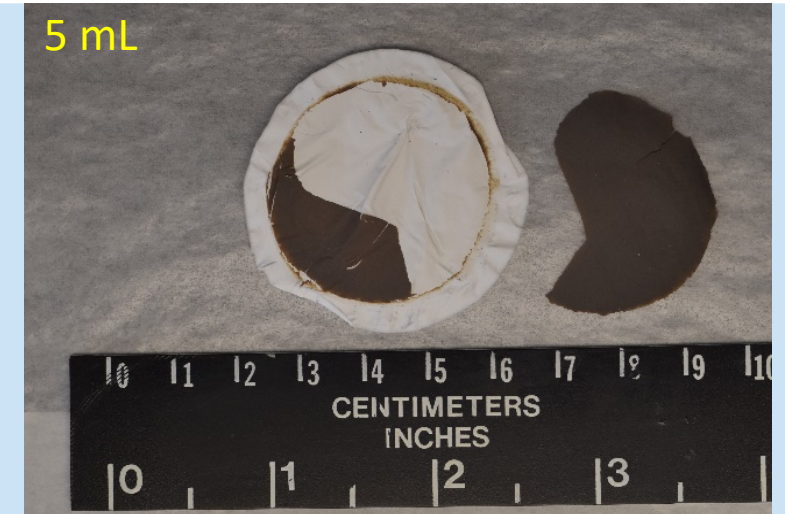
7.5 mL



- Small tear formed when freeing edges, caused the whole mat to rip

Q2B Mats

5 mL



- Extremely thin and fragile during peeling

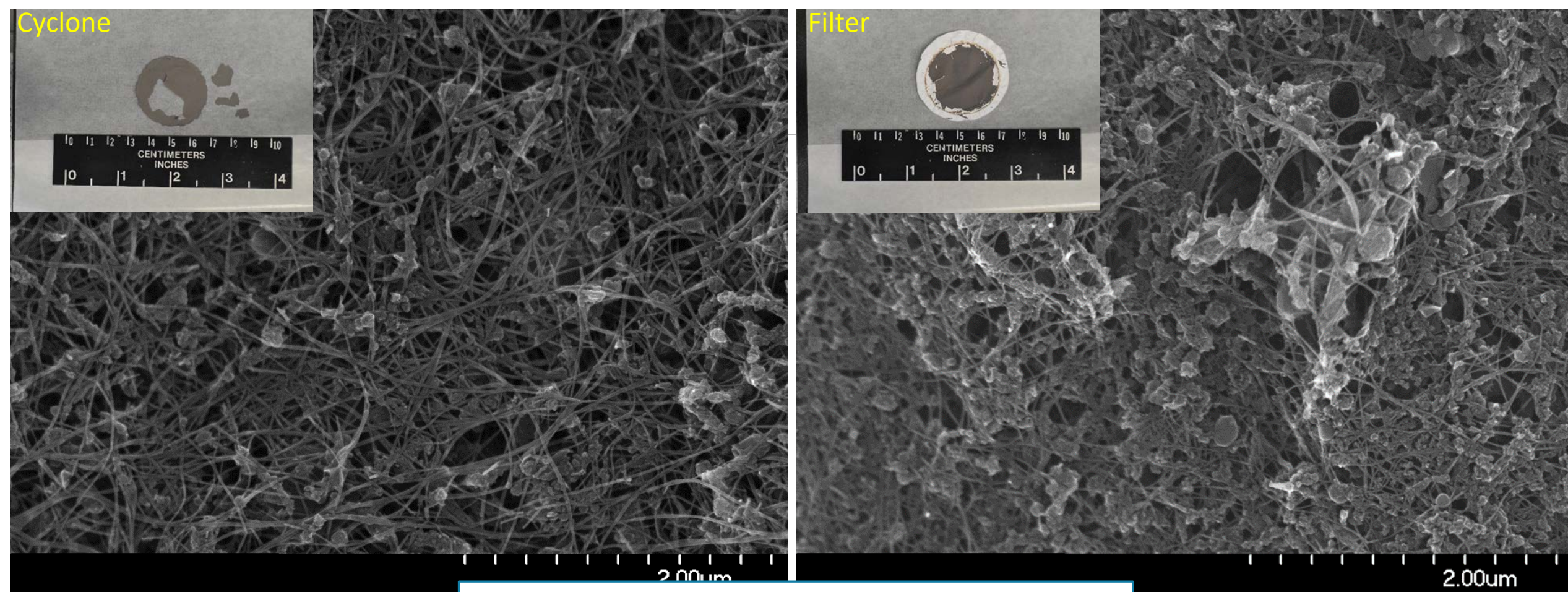
Goal: Thin Mats

- Mechanically superior
 - Flexible and tough

- More BNNTs present in thin films
 - Good for ceramic composites

- Works well for multilayered BNNT metal composites

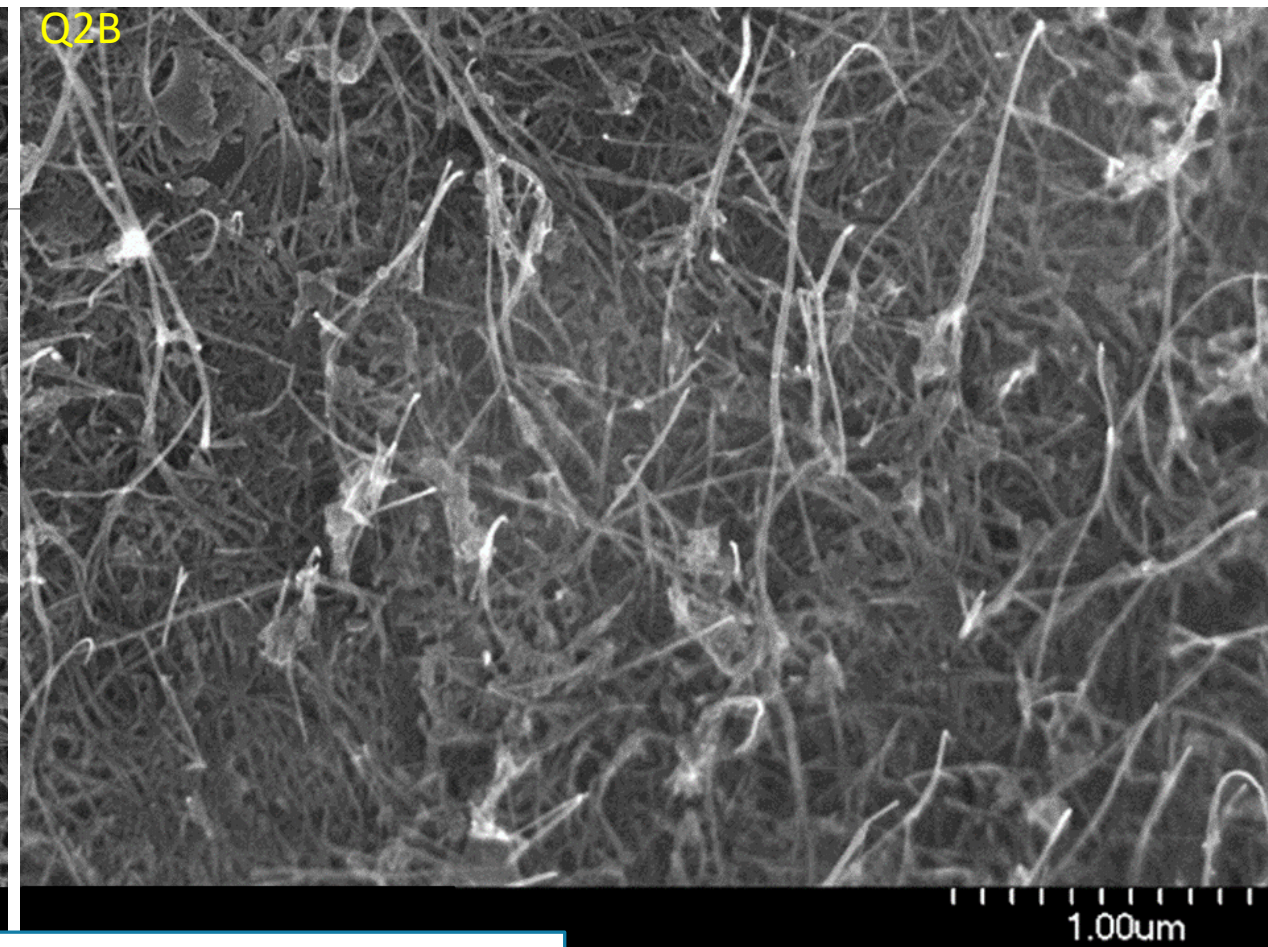
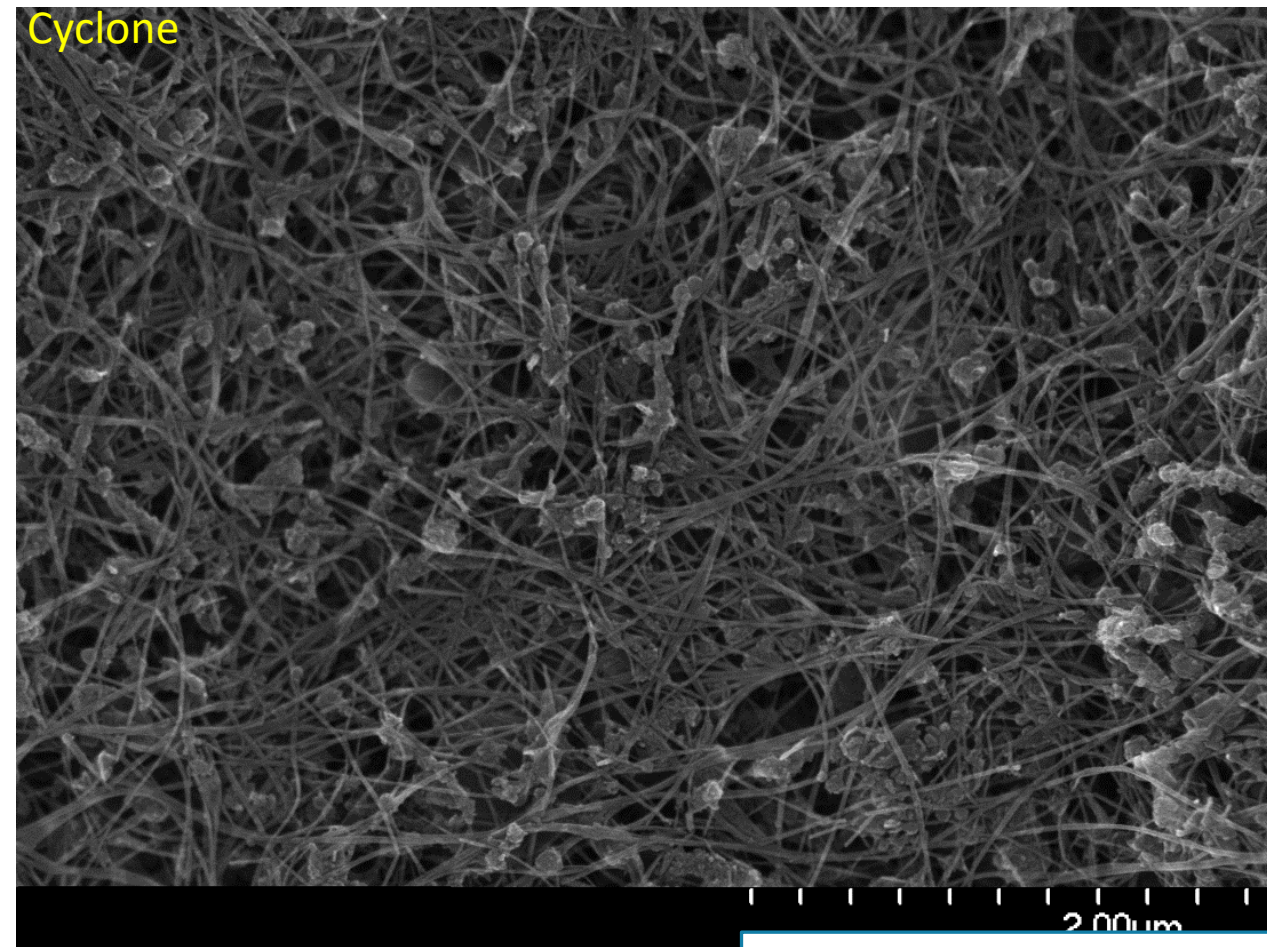
Cyclone vs. Filter 5 mL Mat SEM (x 25k)



Observations:

- Filter mat has significantly more impurities present
- Can clearly see long, densely packed BNNTs in the Cyclone image

Cyclone vs. Q2B Mat SEM



Observations:

- Q2B has less visible impurities than Cyclone
- Observed impurities could affect the creation of Cyclone and Filter mats as thin as the Q2B mats

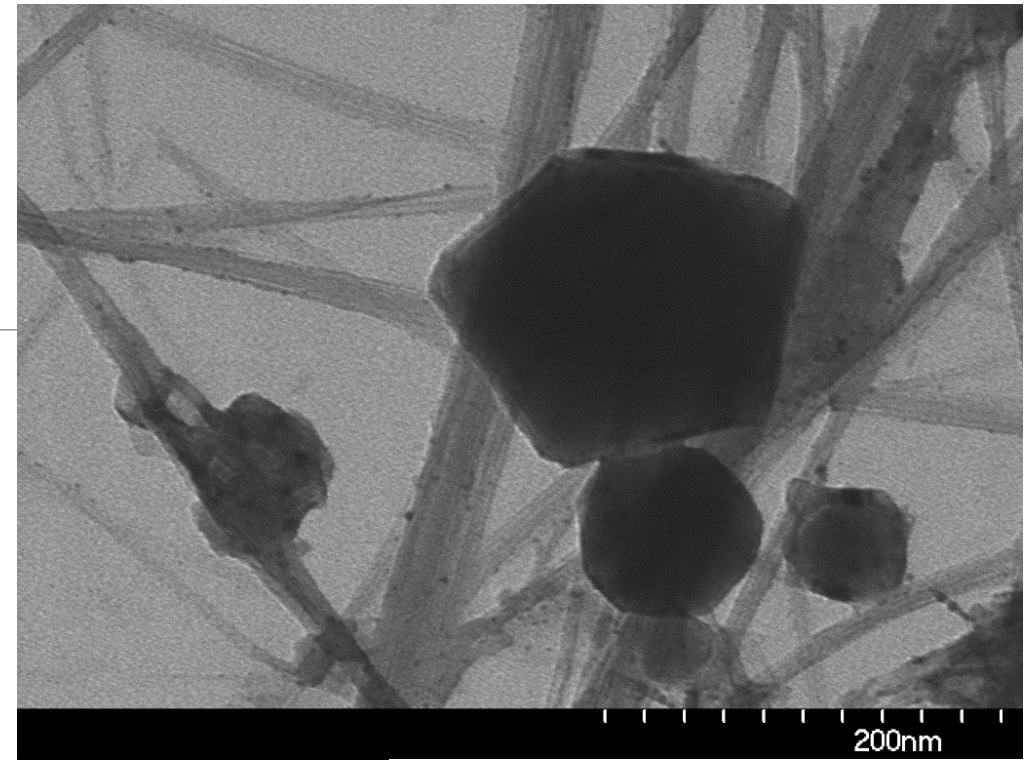
BNNT Impurities

Types of Impurities:

- h-BN shells filled with Boron
- h-BN solids
- Pure Boron particles

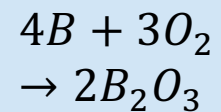
Disadvantages:

- Poor mechanical properties
 - Ex. Brittle
- Poor thermal stability
 - Boron oxide melts at ~500C



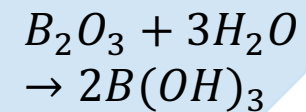
Boron

Reacts with: O_2



Boron
Oxides

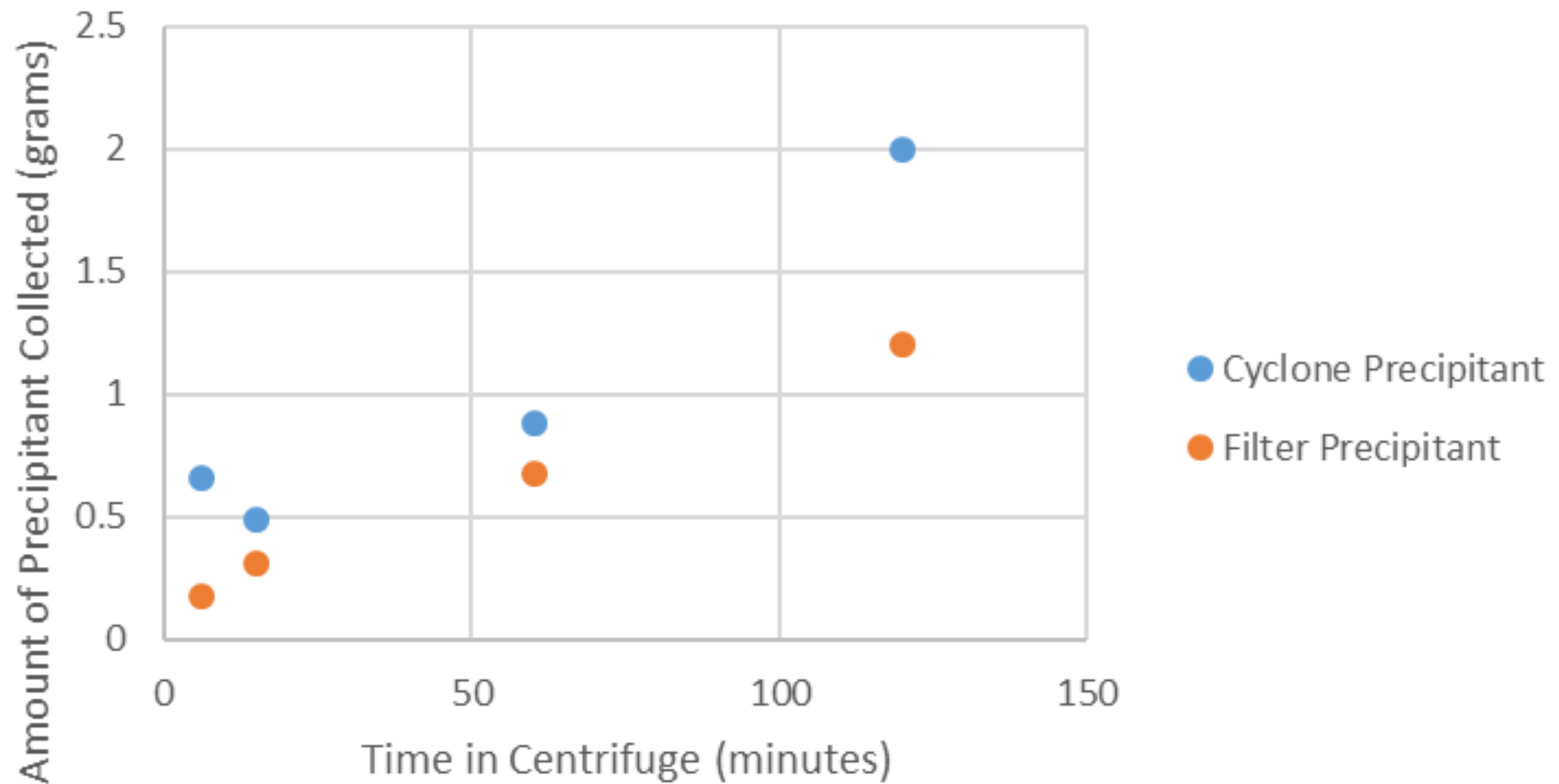
Reacts with: H_2O



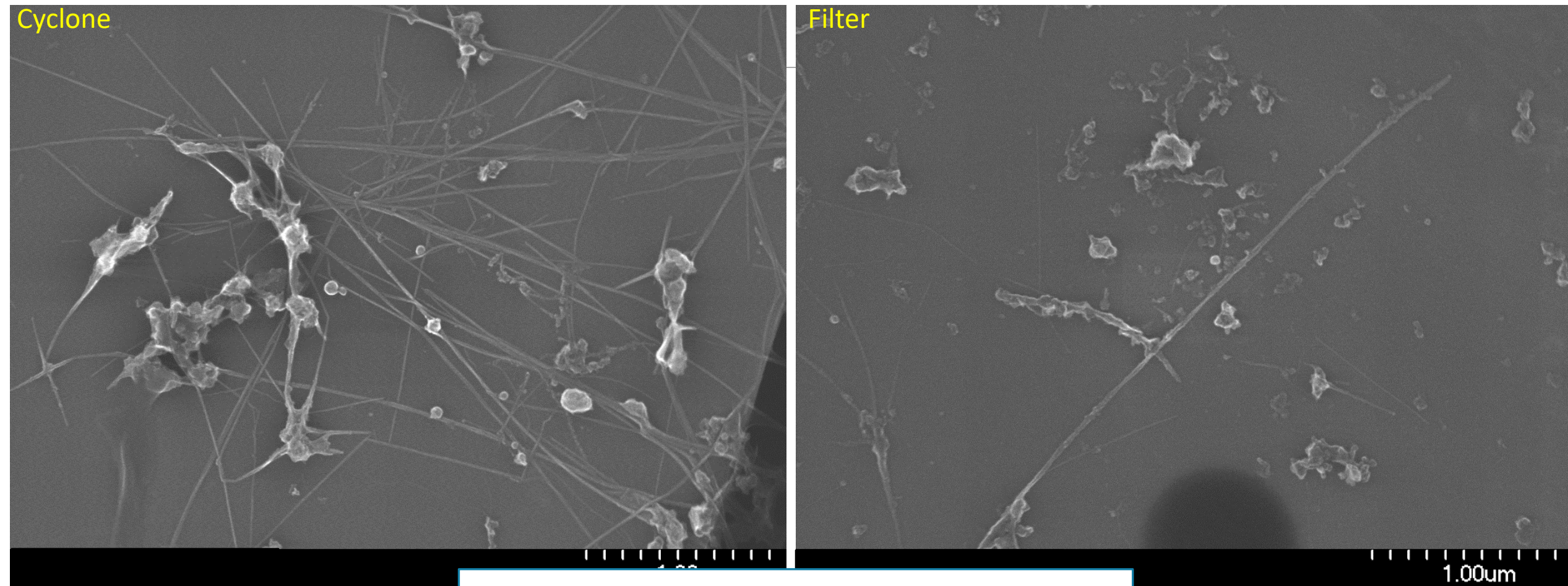
Boric
Acid

Dissolves
with: H_2O

Precipitant Collected vs. Time in Centrifuge



Cyclone and Filter Supernatant SEM (x 30k)

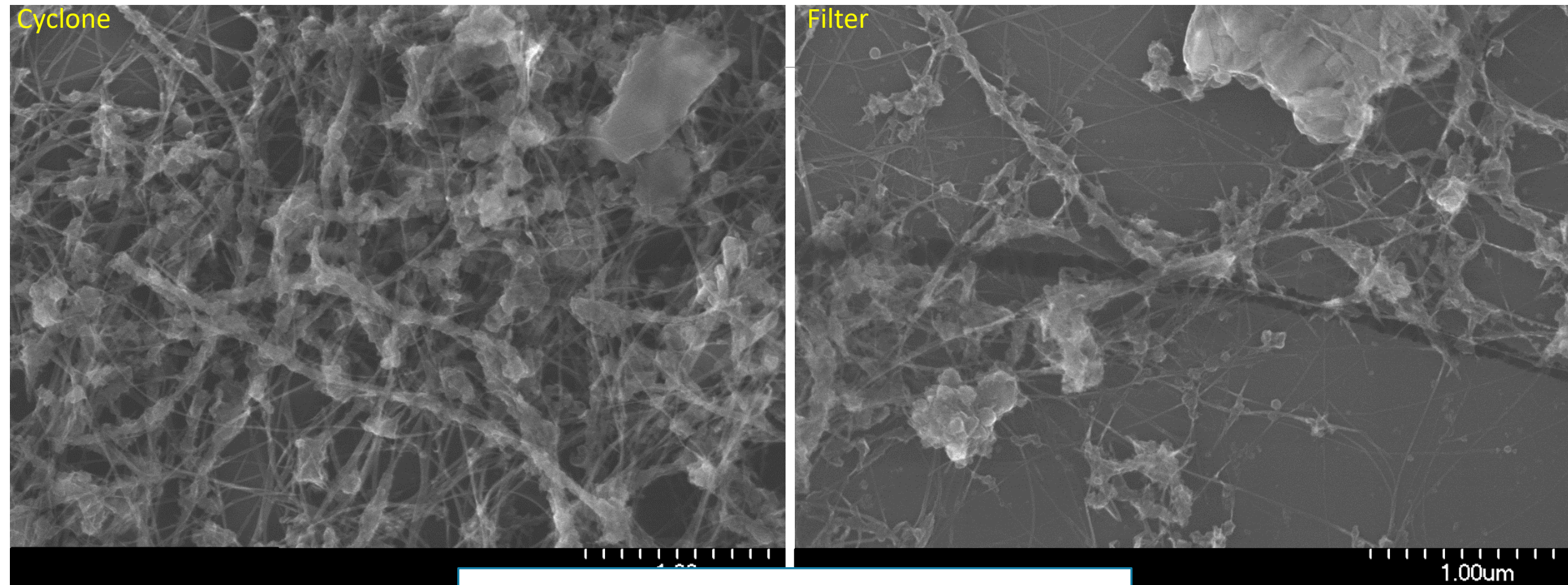


Looking for:

- Long tubes present with few impurities

Process still needs optimization

Cyclone and Filter Precipitant SEM (x 30k)



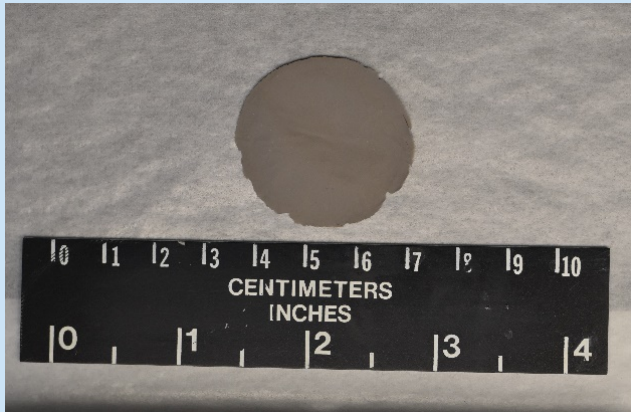
Looking for:

- Large particles that were collected with few tubes

Process still needs optimization

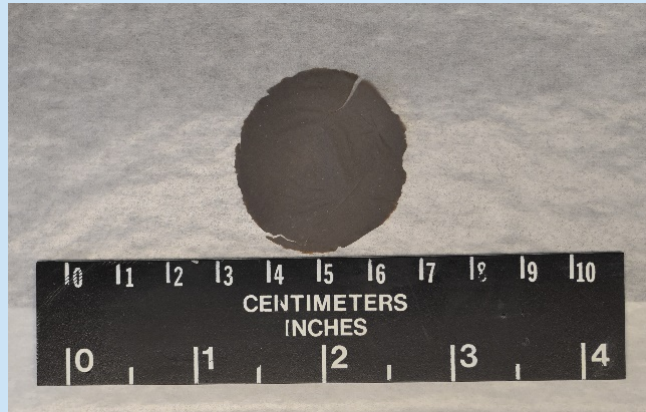
Successful Mats

10 mL Cyclone Mat



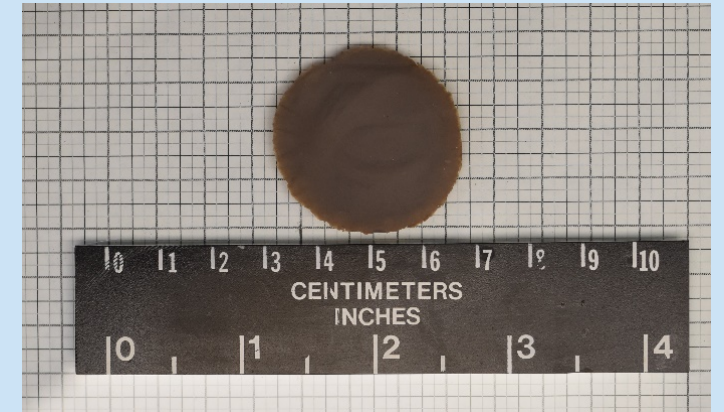
- Mass: 15.5 mg
- Density: 0.833 g/cm^3
- Thicker than previous Cyclone mat attempt, aided by improved peeling method

7.5 mL Q2B Mat



- Mass: 12.8 mg
- Density: 0.645 g/cm^3
- Cracks formed after peeling during extra drying

Diluted Q2B Mat (5:5)



Measured after being cut for composite use:

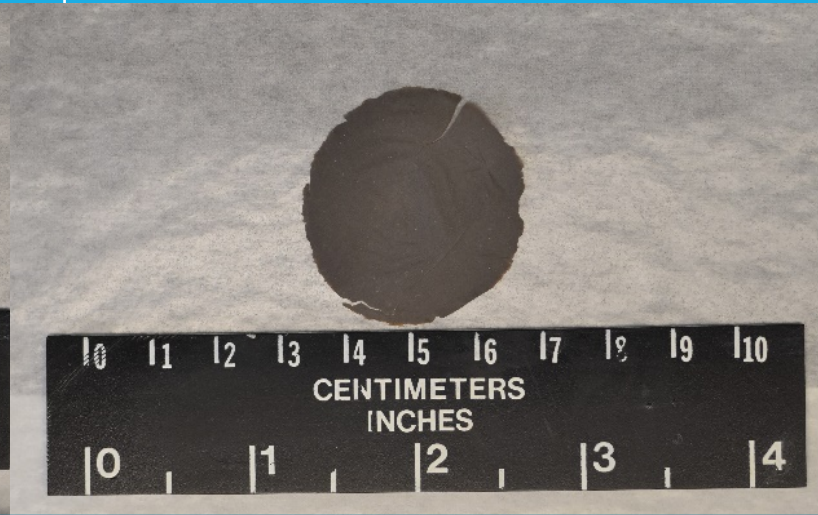
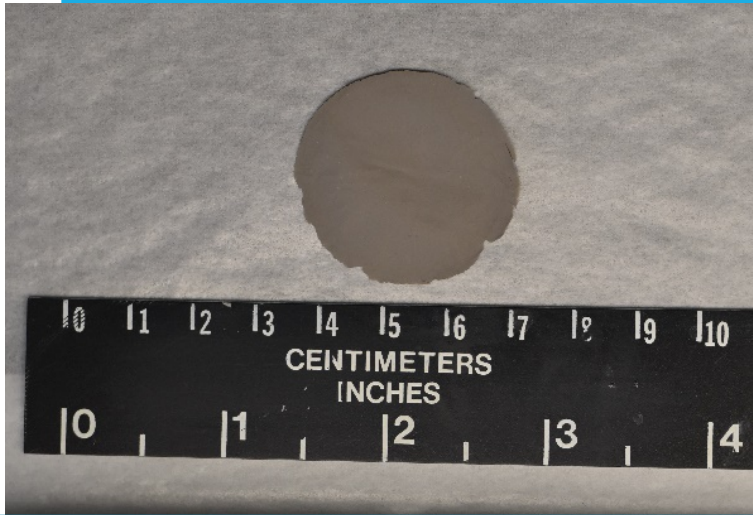
- Density: 1.10 g/cm^3
- Slow peeling process using new procedure
- Extremely difficult to reproduce

Successful Mats

10 mL Cyclone Mat

7.5 mL Q2B Mat

Diluted Q2B Mat (5:5)



Reasons for success:

- Improved peeling method – Elle's method
 - Freeing edges, then
 - Flipping mat over and peeling the filter back
- Worked with thicker mats
 - Gradually made thinner mats

* Some mats still were unsuccessful, even with improved techniques *

Acetone Bath Peeling Method

Method:

1. Free the edges of the mat slightly
2. Place in petri dish and fill dish with acetone
3. Leave to soak until the acetone is all evaporated
4. Wait until the filter is visibly dry (solid white again)
5. Free the edges a bit more
6. Flip over and continue with peeling method

Observations:

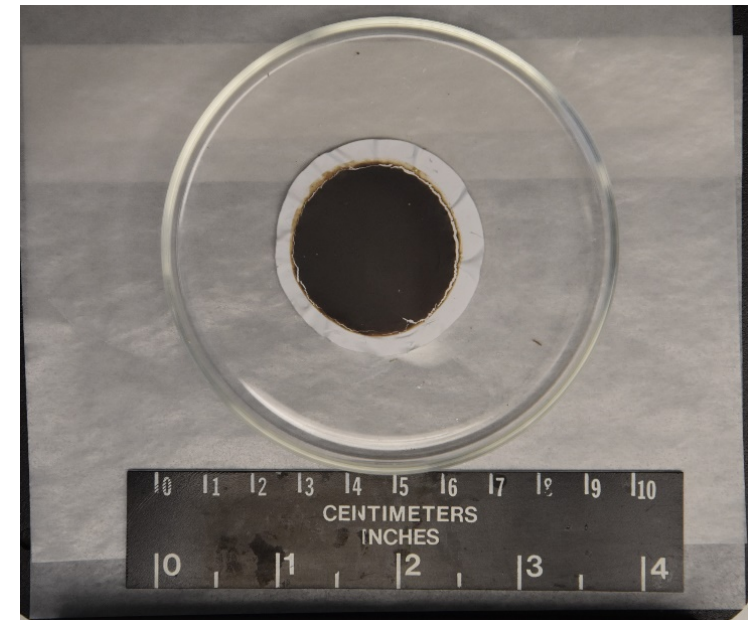
- Easier to peel
- Less fragile
- Slightly thicker
- Denser mats

Theory:

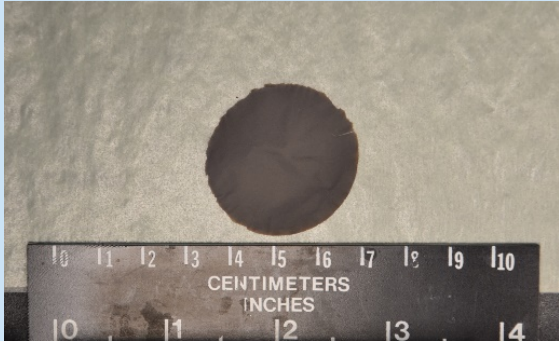
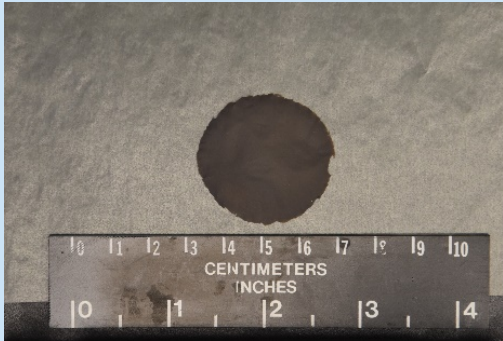
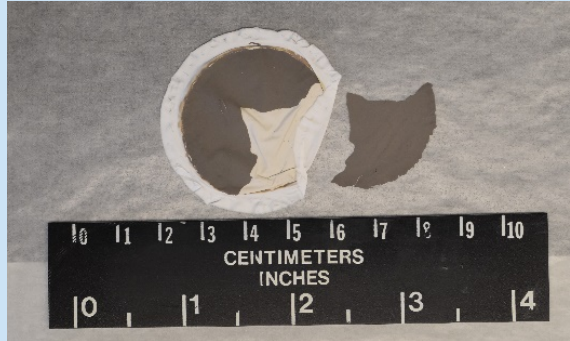
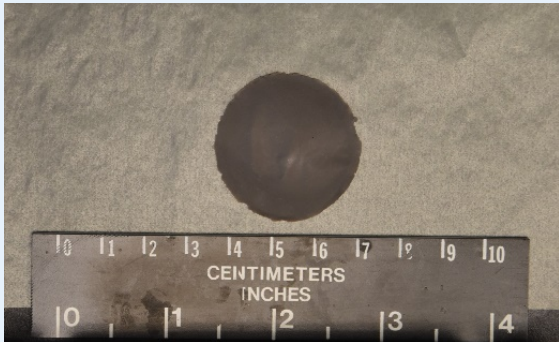
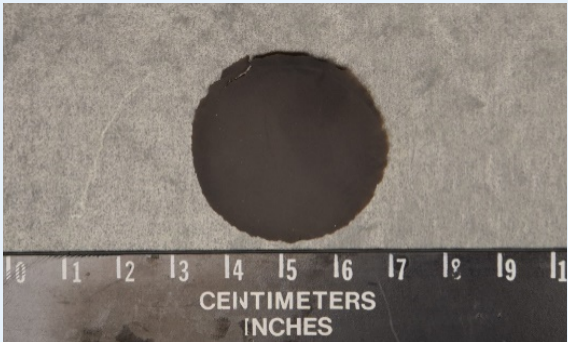
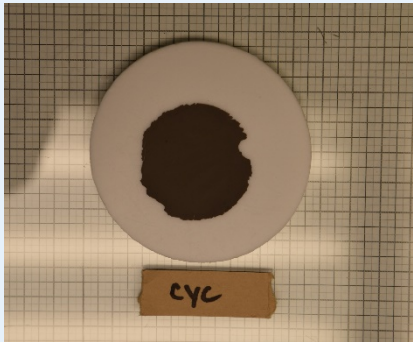
- Acetone will cause the BNNTs to shrink away and detach from the filter

Note:

- Also began trials with ethanol and methanol soaked mats



Acetone Bath Mats

	5.5 mL Q2B Mat	5 mL Q2B Mat	7.5 mL Cyclone Mat
Normal Peeled Mats	 A circular mat sample with a dark, irregular hole in the center. A ruler below shows the hole is approximately 2.5 cm in diameter.	 A circular mat sample with a dark, irregular hole in the center. A ruler below shows the hole is approximately 2.5 cm in diameter.	 A circular mat sample with a dark, irregular hole in the center. A ruler below shows the hole is approximately 2.5 cm in diameter.
Acetone Bath Mats	 A circular mat sample with a dark, irregular hole in the center. A ruler below shows the hole is approximately 2.5 cm in diameter.	 A circular mat sample with a dark, irregular hole in the center. A ruler below shows the hole is approximately 2.5 cm in diameter.	 A circular mat sample with a dark, irregular hole in the center. A ruler below shows the hole is approximately 2.5 cm in diameter.

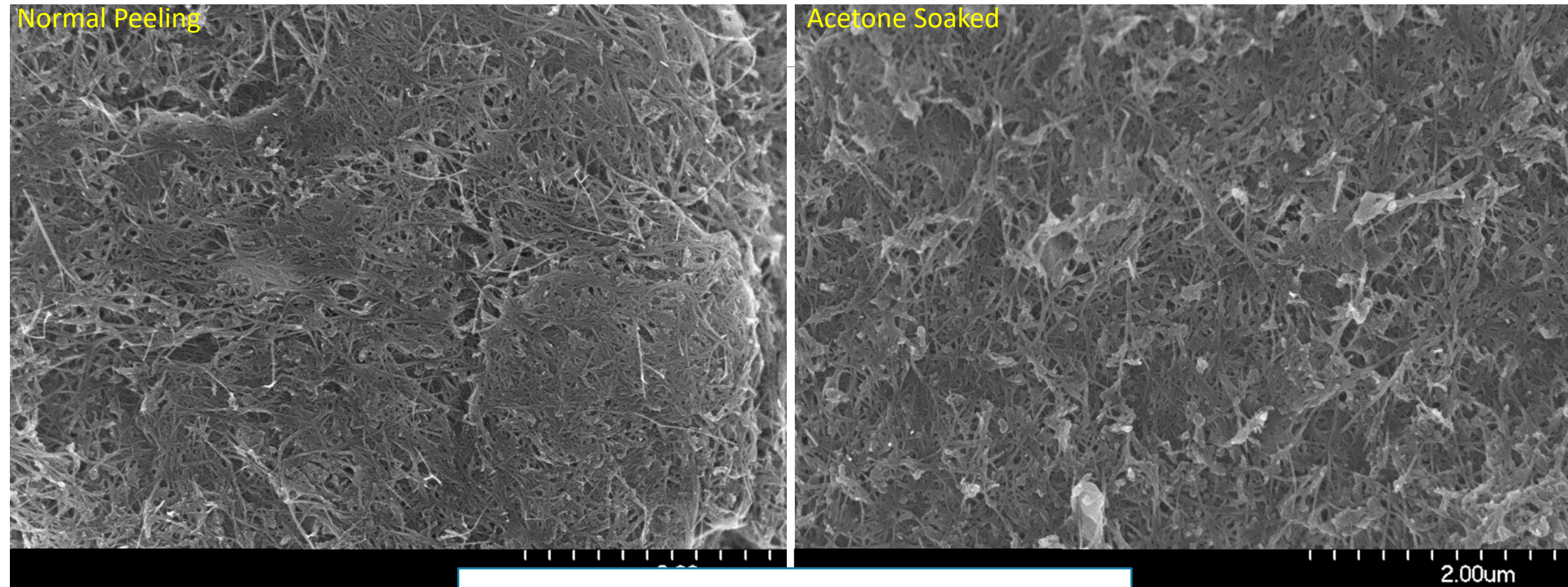
Observations:

- Mats had small tears that would normally have caused damage, but were able to be peeled successfully

Question:

- What effect did the acetone have on the morphology of the mat?

6 mL Q2B Mat SEM: Normal vs. Acetone Peeling (x 20k)

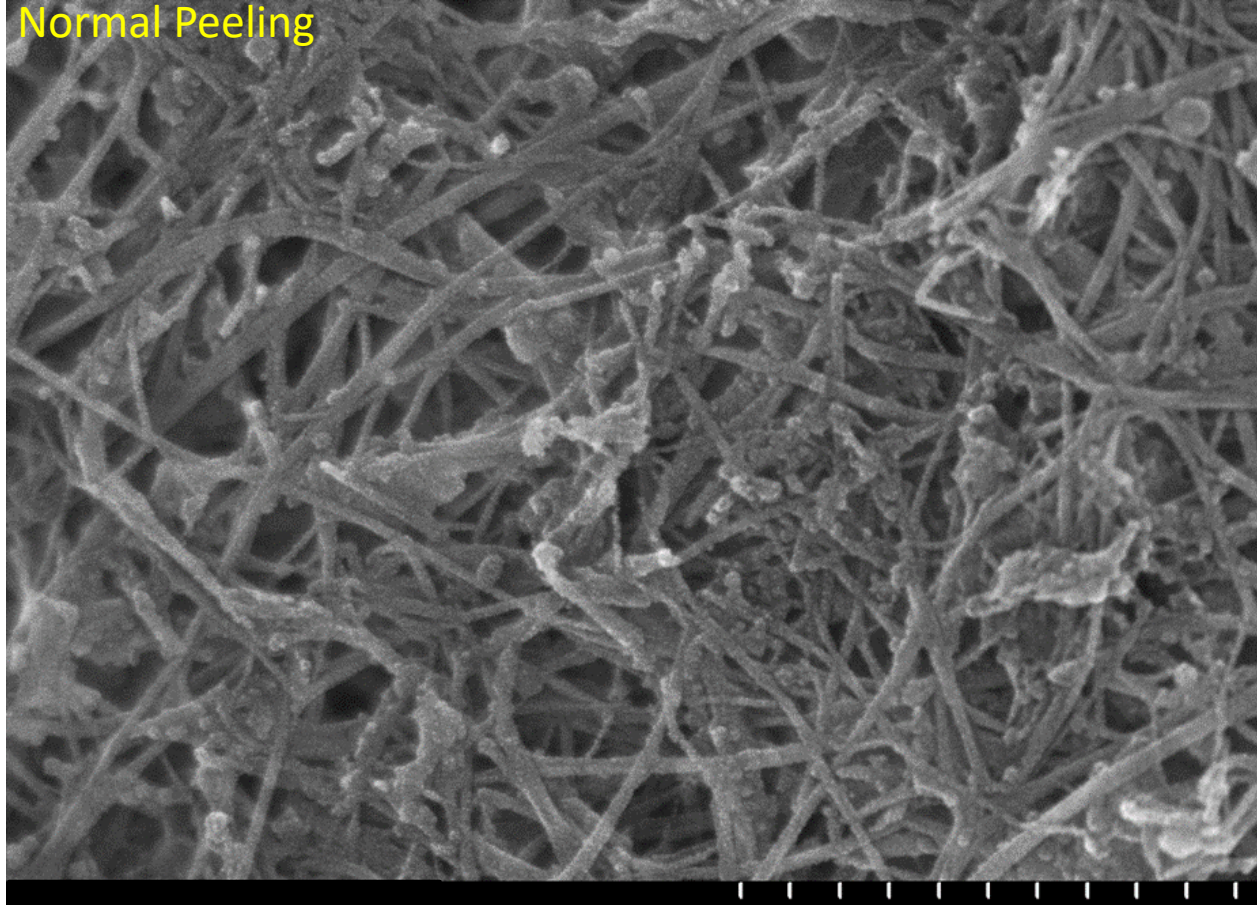


Observations:

- Slight change in the observable porosity of the mats

6 mL Q2B Mat SEM: Normal vs. Acetone Peeling (x 100k)

Normal Peeling

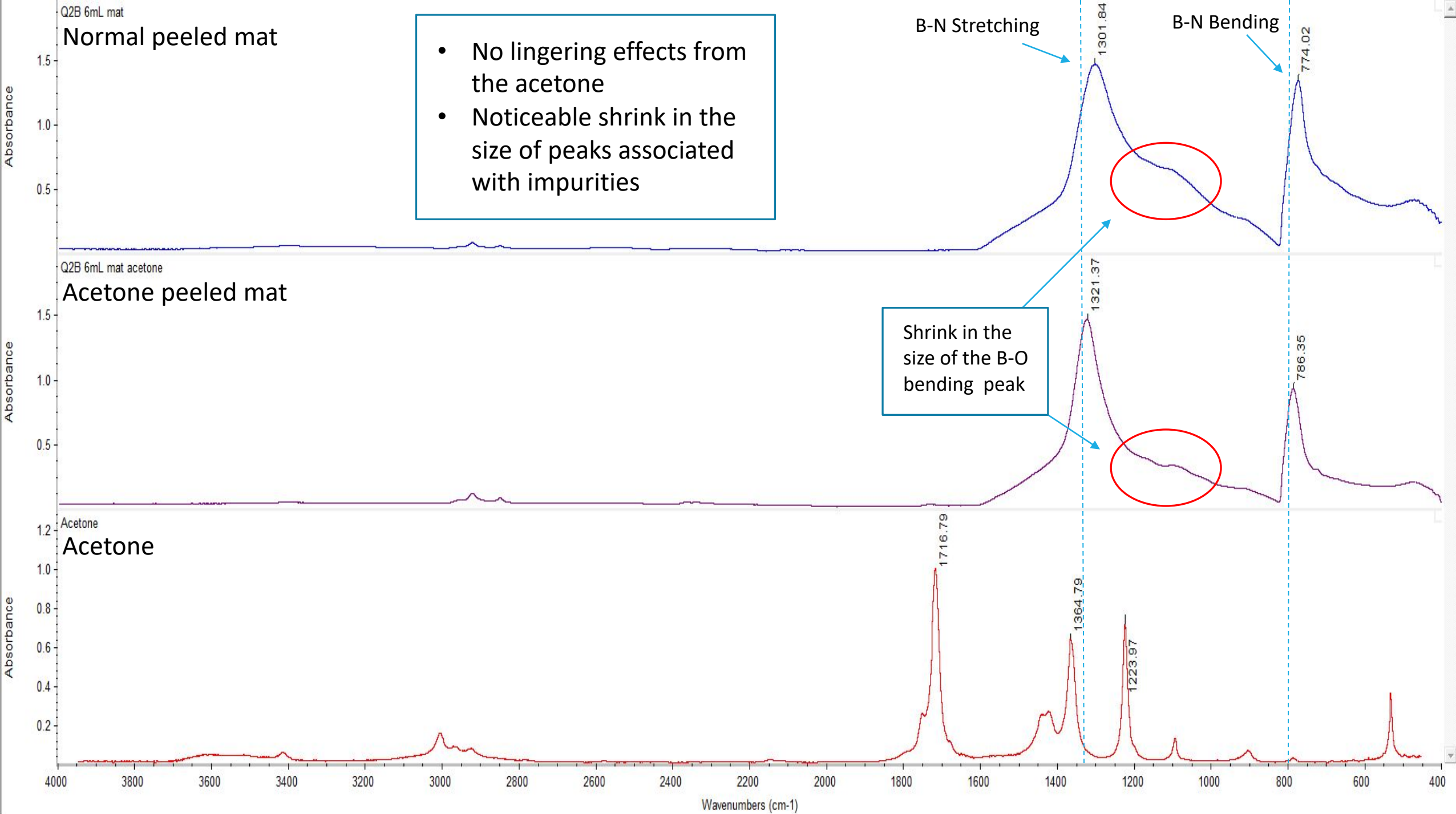


Acetone Soaked



Observations:

- Less impurities
- BNNTs appear thicker



Conclusions

- Plasma tubes may be too impure for use in applications without purification
 - Q2B works well for making thin, high-density mats
 - Good potential for use with ceramic and metal matrix composites
- Soaking mats in acetone may be a viable option for removing impurities
- Centrifuge process also shows potential for removal of impurities
- Future Work:
 - Further investigation of the effects of the centrifuge on the BNNT solution
 - Optimization of the centrifuge process
 - Exploration of soaking mats before peeling
 - Methanol
 - Ethanol
 - Develop new mat procedure to include acetone soaking

Acknowledgements

BNNT Research Group

- Dr. Sang-Hyon Chu
- Dr. Cheol Park
- Calista Lum
- Monroe Willis
- Pranav Iyer
- Dongwon Lee
- Justin Morales
- Dimitrius Khaladj
- Brendan Chabelal

Intern Coordinators

- Christine Dillard
- Valerie Ellis
- Patricia Sanchez

Advanced Materials Processing Branch

Vermont Space Grant Consortium

References

- Hales, S. J., Alexa, J. A., & Jensen, B. J. (2016). Radio Frequency Plasma Synthesis of Boron Nitride Nanotubes (BNNTs) for Structural Applications. Part I and II.
- Hong, X., Wang, D., & Chung, D. D. L. (2016). Strong viscous behavior discovered in nanotube mats, as observed in boron nitride nanotube mats. *Composites Part B: Engineering*, 91, 56-64.
- Kim, J. H., Pham, T. V., Hwang, J. H., Kim, C. S., & Kim, M. J. (2018). Boron nitride nanotubes: synthesis and applications. *Nano convergence*, 5(1), 17.
- Kim, K. S., Kim, M. J., Park, C., Fay, C. C., Chu, S. H., Kingston, C. T., & Simard, B. (2016). Scalable manufacturing of boron nitride nanotubes and their assemblies: a review. *Semiconductor Science and Technology*, 32(1), 013003.
- Kim, K. S., Kingston, C. T., & Simard, B. (2018). *U.S. Patent No. 9,862,604*. Washington, DC: U.S. Patent and Trademark Office.
- Smith, M. W., Jordan, K. C., Park, C., Kim, J. W., Lillehei, P. T., Crooks, R., & Harrison, J. S. (2009). Very long single-and few-walled boron nitride nanotubes via the pressurized vapor/condenser method. *Nanotechnology*, 20(50), 505604.
- Zhang, X., Chen, Y., & Hu, J. (2018). Recent advances in the development of aerospace materials. *Progress in Aerospace Sciences*, 97, 22-34.