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Invited review

**The Renazzo-like carbonaceous chondrites as resources to understand the origin, evolution, and exploration of the Solar System**

N. M. Abreu<sup>a\*</sup>, J. C. Aponte<sup>2b,c</sup>, E. A. Cloutis<sup>d</sup>, A. N. Nguyen<sup>e</sup>

<sup>a</sup>Earth Science, Pennsylvania State University – DuBois Campus, DuBois, PA, 15801, USA

<sup>b</sup>Solar System Exploration Division, Code 691, NASA Goddard Space Flight Center, Greenbelt, Maryland 20771, USA.

<sup>c</sup>Department of Chemistry, Catholic University of America, Washington, DC 20064, USA.

<sup>d</sup>Department of Geography, University of Winnipeg, 515 Portage Avenue, Winnipeg, MB R3B 2E9, Canada.

<sup>e</sup>Jacobs Engineering Group Inc., NASA Johnson Space Center, Houston, Texas 77058, USA

\*Corresponding author.

*E-mail address:* [abreu@psu.edu](mailto:abreu@psu.edu) (N. M. Abreu).

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4 ABSTRACT  
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8 We present here a review of the characteristics of CR carbonaceous chondrite meteorites.  
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10 Over the past three decades, our knowledge and understanding of the scientific value of  
11 the CR chondrites have increased dramatically, as more samples from cold and hot  
12 deserts have become available for analysis. Based on a variety of compositional,  
13 mineralogical, isotopic, and spectroscopic studies, we have come to understand that CR  
14 chondrites are excellent samples of asteroidal meteorites to look for virtually unaltered  
15 solar nebula material and to observe asteroidal processes in progress. This paper  
16 summarizes these investigations, their similarities, and differences with other chondritic  
17 groups, their relationships to asteroids, and the questions yet to be addressed.  
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26 **1. Introduction**  
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30 Carbonaceous chondrites are undifferentiated meteorites. Their principal  
31 constituents, namely chondrules, Fe-Ni metal, sulfides, matrix material, and refractory  
32 inclusions, accreted from the solar nebula 4.56 billion years ago to form asteroids (e.g.,  
33 Weisberg et al., 1993; Zolensky et al., 1993). Chondrules are millimeter-sized spherical  
34 silicate droplets. Between the chondrules, there is dark, fine-grained, silicate-rich,  
35 opaque-bearing assemblage. In addition to these materials, matrices contain dust particles  
36 that originated in the gaseous outflows or ejecta of dying stars, as well as a wide variety  
37 of organic materials.  
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44 Residence in asteroids modified the texture, composition, and mineralogy of  
45 chondritic components through aqueous alteration, thermal, and impact metamorphism.  
46 However, not all chondrite groups underwent the same level of processing while residing  
47 in their parent asteroids. The abundance and origin of alteration products are codified in  
48 the petrologic sequence (i.e., Van Schmus and Wood, 1967; Weisberg et al. 2006), in  
49 which chondrites that have not been affected by asteroidal processes are assigned type  
50 3.0. Petrologic types 1–2 denote increasing replacements of solar nebula mineral and  
51 organic assemblages via aqueous alteration. Water, as a solvent and mass transport agent,  
52 is the primary source of low-temperature chemical alteration in some carbonaceous  
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4 chondrites. Secondary phases in carbonaceous chondrites formed by aqueous alteration  
5 include layered silicates (e.g., serpentine), sulfide-hydroxide (tochilinite), sulfates,  
6 oxides, carbonates, and hydroxides, which initially replace matrix materials and  
7 eventually affect coarser-grained components (e.g., Zolensky et al., 2018 and references  
8 therein). Increasing effects of thermal metamorphism are assigned increasing petrologic  
9 types 3-6. The consequences of thermal metamorphism are textural, compositional, and  
10 mineralogical homogenization and loss of volatiles.  
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17 The abundance variations and chemical compositions of presolar silicates can be  
18 used to infer the degree of processing on the parent body. While some presolar phases are  
19 resilient to secondary processing, presolar silicates are susceptible to alteration or  
20 destruction by radiation, shock, collisions in the interstellar medium (ISM), and by  
21 asteroidal processes (Bringa et al., 2007; Floss and Haenecour, 2016a; Jones and Nuth,  
22 2011; Jones et al., 1994; Tielens et al., 1994; van Dishoeck, 2004). Parent body aqueous  
23 alteration can destroy presolar silicates, giving rise to abundance variations among  
24 chondrites. Other presolar phases, such as SiC and oxides, are not destroyed by aqueous  
25 alteration, and the abundance variations of these phases likely point to the heterogeneous  
26 distribution of presolar grains in the early solar nebula (Leitner et al., 2012a; Davidson et  
27 al., 2014a). Alternatively, these distributions may reflect destruction by significant  
28 thermal alteration. Thermal metamorphism can destroy presolar silicate grains and also  
29 alter their chemical compositions, particularly their Fe contents (Floss and Stadermann,  
30 2012; Nguyen et al., 2016; Nguyen and Zinner, 2004).  
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43 CR chondrites, in particular, are comprised of abundant chondrules, small and few  
44 refractory inclusions, and a variety of fine-grained materials (e.g., Fig. 1). In average  
45 CRs, chondrules occupy ~64.6 area%, fine-grained materials ~35 area% (including  
46 matrix and dark inclusions), and refractory inclusions ~0.4 area% (Schrader et al., 2011).  
47 Fe-Ni metal grains are commonly associated with chondrules (e.g., Kong and Palme,  
48 1999; Krot et al., 2002). These grains are more abundant than in other hydrated  
49 chondrites, which led investigators to group these meteorites with the admittedly much  
50 more metal-rich CB and the CH chondrites as part of the CR clan (e.g., Weisberg et al.,  
51 1995; Krot et al., 2002). These authors noted that, despite their petrologic differences,  
52 metal-rich carbonaceous chondrites shared depletions in moderately volatile elements,  
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4 bulk O and N isotope trends, and similar matrix mineral assemblages.  
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6 Multiple lines of evidence indicate that most CR chondrites preserve a wealth of  
7 records from solar nebular processes (e.g., Kong and Palme, 1999; Krot et al., 2002;  
8 Floss et al., 2014; Schrader et al., 2015). CR chondrites have solar abundances for all but  
9 the most volatile elements, which are extremely depleted (e.g., Krot et al., 2002). They  
10 contain a variety of presolar phases of diverse origins (e.g., Davidson et al., 2014a; Zhao  
11 et al., 2013; Floss and Stadermann, 2009a,b). Preservation of presolar materials is  
12 consistent with our understanding that the CR group is particularly pristine. CR  
13 chondrites record all stages of aqueous alteration, but essentially no evidence of thermal  
14 metamorphism (e.g., Harju et al., 2014; Schrader et al., 2015; Abreu, 2016a).  
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22 Here, we summarize the characteristics of all components of CR chondrites, their  
23 formational histories, intra- and inter-group relationships, the effect that residence in  
24 asteroidal parent bodies and terrestrial environments have had on these materials, and  
25 how to identify CR parent bodies via spectroscopic analysis.  
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## 35 **2. Survey and provenance of CR chondrites**

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### 38 *2.1. Falls*

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40 There are only two CR observed falls, the type meteorite for the group – Renazzo –  
41 and an anomalous (e.g., unusually matrix-rich) chondrite, Al Rais (Table 1). Renazzo was  
42 the second carbonaceous chondrite observed to fall (Mason, 1962). It was collected near  
43 Ferrara (Italy) on January 15, 1824 (Cordier, 1827). Renazzo is made up of several  
44 stones, of which at least three were recovered (Mason, 1962). The largest Renazzo  
45 fragment was reported to weigh ~5 kg, but only ~1.1 kg of this mass was in museum  
46 collections by 1962. The main mass is located at the University of Bologna (Mason,  
47 1962). Early workers recognized the importance of this meteorite in elucidating  
48 hypotheses about carbonaceous chondrites (e.g., Urey and Craig, 1953; Wiik, 1956;  
49 Mason, 1960a,b, 1962; Mason and Wiik, 1962; Wood, 1962, 1963a,b). Wood (1962)  
50 suggested that Renazzo was more pristine than CI chondrites. These early descriptions  
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4 identified the predominant minerals (i.e., olivine, clino and orthopyroxenes, serpentine,  
5 Fe-Ni metal, Fe-sulfides, magnetite, and chromite). Based on its high abundance of  
6 metallic Fe, Renazzo was initially thought to be an intermediate between the  
7 carbonaceous and the enstatite chondrites (Mason and Wiik, 1962; Mason, 1971). Later  
8 authors grouped Renazzo with the CV chondrites, because they have chondrules of  
9 similar diameters, but noted its unusual characteristics (e.g., McSween, 1977a). As more  
10 extensive studies were conducted and more CR chondrites became available, it became  
11 evident that Renazzo belonged to a different, new group of carbonaceous chondrites (e.g.,  
12 McSween, 1979a; Weisberg et al., 1993; Zolensky et al., 1993). The second CR fall, Al  
13 Rais, was observed to fall near Medina (Saudi Arabia) on December 10, 1957. The main  
14 mass of this meteorite was originally 162 g. Other stones were reported to exist (Mason,  
15 1962). Starting with early studies, Al Rais was grouped with Renazzo based on its high  
16 metallic Fe abundance (e.g., Mason, 1962, 1971).  
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## 29 *2.2. Antarctic finds*

### 30 *2.2.1. Provenance and mass of Antarctic finds*

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37 The impact that the Antarctic suite has had on our understanding of CR chondrites  
38 is difficult to overstate. In terms of mass, variety of features, and low degrees of  
39 terrestrial weathering, Antarctic CRs provide unique scientific opportunities. As of  
40 February of 2019, a total of 101 stones belonging to the CR group had been recovered  
41 from Antarctica. Of these 101 stones, 18 were recovered by Japanese expeditions to the  
42 Asuka and Yamato Mountains. In fact, the first Antarctic CR chondrites were collected in  
43 1979 at Yamato. The American Antarctic Search for Meteorites (ANSMET) program  
44 collected the remaining Antarctic CR chondrites.  
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52 Despite the large number of stones, many Antarctic CR chondrites belong to a  
53 large pairing group (i.e., were initially part of a single meteor). This group, EET 87711  
54 pairing group, includes 50 stones. These samples were collected during the 1987 and  
55 1992 field seasons. GRO 17063 and GRO 17064 also form a small pairing group. After  
56 accounting for pairing, 50 different CR meteors appear to be represented in the Antarctic  
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4 collections. It is possible that other pairing groups exist and will be identified as we learn  
5 more about these samples.  
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8 The current total mass of Antarctic CR chondrites is 10.77 kg. However, many  
9 Antarctic CR chondrites of the highest scientific interest (e.g., QUE 99177, MET 00426)  
10 have small masses (Richter et al., 2017). Of the 50 unpaired Antarctic CRs, 19 have  
11 masses <10 g. Based on their scarcity, NASA's Antarctic CRs have been assigned to the  
12 list of protected samples. The 12 most massive CRs and the EET 87711 pairing group  
13 represent >90% of total Antarctic CR mass. One meteorite alone, MIL 090001, represents  
14 >58% of the total Antarctic CR mass. As discussed below, whether MIL 090001 is a CR  
15 is debated (Alexander and Bowden, 2018). The EET 87711 pairing group has a combined  
16 mass of 1.34 kg or 12.4% of the total Antarctic CR mass.  
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19 The largest Antarctic CR chondrite that has been recovered to date is MIL  
20 090001, weighing 6.29 kg (Table 1). The classification of this meteorite has been a  
21 subject of controversy. It was initially classified as a CV chondrite (Keller, 2011). Soon  
22 after, the meteorite was reclassified based on whole-rock oxygen isotope studies (Keller  
23 et al., 2012; Alexander et al., 2012). However, Alexander and Bowden (2018) have  
24 recently argued that the bulk H content and  $\delta D$  from MIL 090001 plot below the trend  
25 established by other CR chondrites. They argued that MIL 090001 might actually be an  
26 ungrouped petrologic type 2 carbonaceous chondrite with affinities to ungrouped C2  
27 LEW 85332. Authors have undertaken a detailed petrologic comparison between these  
28 meteorites. However, results from this work are beyond the scope of this review.  
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### 46 *2.2.2. Effect of weathering on Antarctic finds*

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50 Antarctic finds are prone to chemical interaction with the terrestrial environment.  
51 To indicate the degree to which meteoritic minerals have been affected by their  
52 interaction with the ice, collection curators use weathering grades A, B, C, and e (e.g.,  
53 Gooding, 1981, 1986; Velbel et al., 1988; Losiak and Velbel, 2011). Grades A-C denote  
54 increasing elemental mobilization and mineral replacement. Grade e is independently  
55 assigned to samples with evaporite deposits.  
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4 Weathering ubiquitously affected Fe-Ni metal and Fe-oxides in CR chondrites  
5 (e.g., Abreu 2016a). Abreu (2016a) described that the effect of Antarctic weathering on  
6 Fe-Ni metal included thick rust rims, oxide sub-grains, and even complete  
7 pseudomorphic replacement in some cases. The effects of these processes are shown in  
8 Fig. 2, where the metal from stones of the same pairing group is shown. Most Fe-Ni  
9 metal grains are located inside the chondrules or on the margins of chondrules adjacent to  
10 fine-grained matrix. Abreu (2016a) observed that there are no systematic differences  
11 between the weathering features in Fe-Ni grains in chondrule interiors versus those in  
12 direct contact with the matrix.  
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21 Abreu (2016a) also reported the presence of Fe-hydroxide veins on the periphery of  
22 metal grains. In a grade C CR chondrite, she observed Fe-hydroxide veins that cut across  
23 the length of a petrographic thin-section. Ferric Fe phases, such as those found in the  
24 veins described in weathered CRs, are generally assumed to be terrestrial in origin.  
25 However, Le Guillou et al. (2015a) suggested that some pure ferric Fe phases (e.g.,  
26 hematite and ferrihydrite) might, in fact, have formed as matrix silicates. As they  
27 progressively altered, their  $\text{Fe}^{3+}/\sum\text{Fe}$  ratios decreased. It is not known if or how much  
28 hematite and ferrihydrite are present in the most aqueously altered CR fall, Al Rais. If  
29 significant amounts of these preterrestrial ferric Fe minerals are present in Al Rais, the  
30 origin of ferric Fe minerals in Antarctic chondrites ought to be revisited.  
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39 The effect of terrestrial alteration on the silicate mineralogy of CR chondrites remains  
40 mostly unexplored. Ascertaining the effect is a challenging question to address because  
41 terrestrial alteration products, in particular those formed under Antarctic conditions, may  
42 resemble aqueous alteration occurring at the low temperature, low water:rock ratios,  
43 thought to also be prevalent in the CR parent body. Abreu (2016a) observed that in  
44 weathering grade B/C MIL 07525, matrix regions adjacent to altered Fe-Ni metal are  
45 unusually Fe-rich. Amorphous Fe-Mg matrix silicates in MIL 07525 contain ~17 wt.%  
46 more FeO than similar materials in regions that did not show textural evidence of  
47 terrestrial weathering. In MIL 07525, amorphous Fe-Mg silicates are associated with the  
48 ferroan serpentine and hisingerite. More in-depth studies are necessary because the effect  
49 of terrestrial weathering of CR silicates is not easily resolved.  
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4 2.3. *North African finds*  
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8 2.3.1. *Provenance and mass of North African finds*  
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11 As of February 2019, 72 individual stones from hot deserts have been classified as  
12 CR chondrites. The current total mass of Saharan CRs is over twice the total mass of  
13 Antarctic CR chondrites (27.05 kg Saharan vs. 10.77 kg Antarctic CRs – based on the  
14 Meteoritical Bulletin Database; Table 1). The Saharan CR suite includes eight of the ten  
15 most massive CRs (combined mass = 15.1 kg). Like Antarctic finds, some North African  
16 finds are also paired. Four pairing groups have been proposed: (1) El Djouf 001, Acfer  
17 059, 087, 097, 114, 139, 186, 187, 209, 270; (2) Acfer 270 and 324; (3) Acfer 394-400;  
18 and (4) NWA 801 and NWA 1776 (Bischoff et al., 1993a,b; Ruzicka et al., 2014, 2015).  
19 As with the Antarctic finds, it is possible that there are more paired samples. One  
20 challenge that is unique to this CR sample suite is that little field information is available  
21 for most of these stones, which makes pairing more difficult.  
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32 The North African CR suite has significantly grown over the last two decades –  
33 59 of the 72 stones were recovered since 2000, many of which have not been studied.  
34 Although several studies have described the general characteristics of some of the  
35 Saharan CR chondrites (e.g., Bischoff et al., 1993a,b; Harju et al., 2014), less is known  
36 about their sub-micron mineralogy.  
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41 In addition to the 72 North African CR chondrites, seven highly equilibrated (i.e.,  
42 petrologic types 6 and above) samples have been reported to have affinities with the CR  
43 group, most remarkably, Tafassasset. Gardner-Vandy et al. (2012) presented a wide array  
44 of analyses and reclassified Tafassasset as a primitive achondrite. Bunch et al. (2013)  
45 argued that NWA 7531 is a polymict breccia composed of both ordinary chondritic and  
46 CR chondritic clasts formed during a collisional event. Only a brief description from the  
47 Meteoritical Bulletin Database was available for the newest equilibrated sample, NWA  
48 11561. They described the sample as having affinities with Tafassasset and NWA 7531.  
49 Ruzicka et al. (2015) reported that NWA 6921 and 7317 are paired with NWA 2994,  
50 NWA 3250, and NWA 6901, which are not currently classified as CR chondrites.  
51 Instead, these meteorites were classified as primitive achondrites. Further studies are  
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4 needed to determine if and how these highly equilibrated meteorites are related to the CR  
5 group.  
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10 *2.3.2. Terrestrial weathering of North African finds*  
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13 Meteorites recovered from hot deserts are generally affected by extensive  
14 terrestrial weathering (e.g., Jull et al., 1991; Wlotzka, 1993; Wlotzka et al., 1995; Al-  
15 Kathiri et al., 2005; Bland et al., 2006 and references therein). The dominant effects that  
16 terrestrial weathering have on meteorites – Fe oxidation and mobilization – might affect  
17 CR chondrites more extensively than other hydrated meteorites, because the CRs are  
18 comparatively rich in Fe–Ni metal. Saharan meteorites are classified according to a  
19 different scheme than Antarctic meteorites, ranging from W0-W2 for progressive  
20 weathering of Fe-Ni metal, W3-W4 for progressive weathering of troilite, and W5-W6  
21 for weathering of mafic silicates (e.g. Bland et al., 2006 and references therein). Although  
22 some Saharan CR chondrites have been studied in detail, there have not been systematic  
23 studies of their weathering features. In fact, only 38 of the 66 known Saharan CRs have  
24 been assigned to a terrestrial weathering grade – representing 58% of the total number  
25 and only 43% of the total mass available for study.  
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39 **3. Bulk compositional and isotopic characteristics of the CR chondrites**  
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42 *3.1. Bulk chemical compositions*  
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46 The bulk composition of several CR chondrites is presented in Table 2. Elements  
47 more volatile than Cr show increasing depletions with respect to CI chondrites (Weisberg  
48 et al., 1993; Bischoff et al., 1993a; Kallemeyn et al., 1994; Huss, 2004; Braukmüller et  
49 al., 2018). For elements more refractory than Cr, lithophile and siderophile elements are  
50 unfractionated with respect to each other and show ~40% with respect to CI chondrites  
51 for most elements (e.g., Huss, 2004). Braukmüller et al. (2018) noted that major elements  
52 were slightly below CI values and that REEs were flat.  
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### 3.2. Bulk hydrogen, nitrogen, and carbon contents

Although CR chondrites show a broad range of aqueous alteration, several studies have shown that their H, N, and C contents are low relative to other hydrated carbonaceous chondrites. Alexander et al. (2012) measured the bulk composition of these three elements for Renazzo, Al Rais, EET 92042, EET 96286, GRA 95229, GRO 95577, LAP 02342, LAP 04720, MET 00426, PCA 91082, and QUE 99177. It is noteworthy that they found no differences between falls and finds, indicating that terrestrial weathering did not modify the total H content measurably. Bonal et al. (2013) also studied the hydrogen isotopic composition of water in QUE 99177, MET 00426, EET 92042, GRA 95229, Renazzo, and Al Rais. Alexander et al. (2013) analyzed three other Antarctic CRs, anomalous MIL 090001, and heated GRA 06100 and GRO 03116.

Alexander et al. (2012) measured the average bulk H content to be 0.6 wt.%. Hydrogen contents for individual CRs ranged from 0.4 wt.% in CR3 MET 00426, to 0.9 wt.% in anomalous, matrix-rich CR2 Al Rais, to 1.3 wt.% in CR1 GRO 95577. Both CR minerals and organic matter contain H – structural OH and H in minerals are termed “water.” To estimate the amount of this structural water, Alexander et al. (2013) subtracted the H measured in organics. The average structural water in CRs is ~4.2 wt.%, with a range from 2.7-10.9 wt.%. They noted that both the hydrogen contents and the structural water contents of bulk CR chondrites are low compared with CM chondrites (H ~1.2 wt.%; 0.9-1.5 wt.% range) and maximum structural “water” (9.4 wt.% and ~7.1-12.2 wt.% range). At values of ~0.1 wt.% H, the structural water content for GRA 06100 and GRO 03116 are disproportionately low compared to the average H content in CRs (Alexander et al., 2013). This is to be expected since GRA 06100 experienced some heating, which is not the case for other CR chondrites.

Alexander et al. (2012) determined that the nitrogen and carbon contents of CR chondrites are also low compared to CM chondrites. The average nitrogen of the CR chondrites is 0.08 wt.% versus 0.10 wt.% in the CMs, and the average C in the CRs is 1.3 wt.% versus 2.0 wt.% in the CMs. As with hydrogen, heated CR chondrites have low carbon contents (Alexander et al., 2013).

### 3.3. Bulk hydrogen, nitrogen, and carbon isotopic compositions

The bulk H, N, and C isotopic compositions of CRs are distinct from other groups.  $\delta D$  values are high for CRs compared to other carbonaceous chondrites (e.g., Kung and Clayton, 1978; Robert and Epstein, 1982; Grady et al., 1991; Pearson et al., 2006; Alexander et al., 2012, 2013; Bonal et al., 2013). Alexander et al. (2012) measured the isotopic compositions for Renazzo, Al Rais, EET 92042, EET 96286, GRA 95229, GRO 95577, LAP 02342, LAP 04720, MET 00426, PCA 91082, and QUE 99177. They determined that the average  $\delta D$  content for unheated CR chondrites is 639.0‰ (compared with CMs:  $\delta D \sim -57.7\%$ ). These high values were also reported by Bonal et al. (2013), which measured  $\delta D$ s as high as 1600‰ and high variability in isotopic composition at the micron scale. Furthermore, Bonal et al. (2013) found no  $\delta D$  trends along the CR aqueous alteration sequence. In contrast, heated CR2s GRA 06100 and GRO 03116 have very low  $\delta D$  contents ( $\delta D = -80.3 \pm 0.0$  and  $-96.6 \pm 0.0$  respectively; Alexander et al., 2013). The average  $\delta^{13}C$  for the CR chondrites is  $-4.6 \%$ , which is comparable with values for the CMs ( $-2.9 \%$  - Alexander et al., 2013). Finally, CR chondrites have a high average  $\delta^{15}N = 174.3\%$ , compared with the  $\delta^{15}N$  of the CM chondrites, which is  $56.3\%$  (e.g., Kung and Clayton, 1978; Robert and Epstein, 1982; Grady et al., 1991; Alexander et al., 2012). The high positive  $\delta^{15}N$  value is a characteristic that CR chondrites share with other members of the CR clan (e.g., Krot et al. 2002).

### 3.4. Bulk oxygen isotopic composition

The bulk oxygen-isotopic composition of CR chondrites lies on a line below the terrestrial fractionation line and has a distinct slope of  $\sim 0.7$  (Weisberg et al., 1993; Weisberg et al., 1995; Clayton and Mayeda, 1999; Schrader et al., 2011, 2014; Harju et al., 2014). Schrader et al. (2011) refined the value of the CR slope line to be  $0.70 \pm 0.04$  ( $2\sigma$ ) and determined that the  $\delta^{17}O$ -intercept is  $-2.23 \pm 0.14$  ( $2\sigma$ ) (Fig. 3). These authors noted some spread around the line. This bulk CR chondrite oxygen line has been explained by mixing between  $^{16}O$ -poor phyllosilicates and a  $^{16}O$ -rich anhydrous silicate component, which are thought to represent at least two distinct reservoirs. Since the CR

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4 line represents mixing between a hydrous and an anhydrous reservoir, the relationship  
5 between oxygen isotopic composition of CR chondrites and their degree of aqueous  
6 alteration has been the subject of extensive investigations. Schrader et al. (2011) found  
7 that the oxygen isotopic composition of QUE 99177 is  $^{16}\text{O}$ -rich, and GRA 06100 is  $^{16}\text{O}$ -  
8 poor. However, the correlation between aqueous alteration and O-isotopic compositions  
9 remains an area of active research.  
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#### 15 16 17 **4. Chondrules in CR chondrites** 18 19

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21 CR chondrules are made up of coarse-grained ferromagnesian silicates, nano-  
22 crystalline and glassy mesostasis, and Fe-Ni opaque nodules (Weisberg et al., 1993;  
23 Zolensky et al., 1993; Kallemeyn et al., 1994; Wasson, 1996; Kong and Palme, 1999;  
24 Krot et al., 2002; Schrader et al., 2008, 2011, 2013, 2015, 2018a; Wasson and Rubin,  
25 2010; Berlin et al., 2011). Most CR chondrules are porphyritic olivine (PO) and  
26 porphyritic olivine/pyroxene (POP) (e.g., Weisberg et al., 1993; Zolensky et al., 1993;  
27 Kallemeyn et al., 1994; Kong and Palme, 1999; Krot et al., 2002; Schrader et al., 2008,  
28 2011, 2015) (Fig. 4). However, other mineralogies and textures are also represented,  
29 including barred olivine (BO), compound PO/BO, compound BO/POP, compound  
30 porphyritic olivine/radial olivine (PO/RO), and compound porphyritic olivine/granular  
31 olivine (PO/GO). Authors inferred that the preponderance of PO and POP chondrules  
32 indicates that few of the chondrule precursors were totally melted or that the CR region  
33 of chondrule formation was possibly dusty (e.g., Hewins et al., 2005; Schrader et al.,  
34 2015).  
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46 CR chondrules have a variety of textures, including well-rounded, extensively melted  
47 chondrules with large silicate phenocrysts and opaque grains, and irregular, slightly  
48 melted, fine-grained objects. Some of these objects have multiple concentric igneous  
49 rims, made up of micro-porphyritic and granular silicates, as well as Fe-Ni metal grains  
50 (e.g., Weisberg et al. 1992; Huss et al. 1996; Krot et al., 2002; Ebel et al., 2008; Wasson  
51 and Rubin, 2010). Figures 4a-b show an example of this texture. Wasson and Rubin  
52 (2010) estimated the average thickness of CR igneous rims to be ~270 microns, which is  
53 thick compared with rims around CK chondrules (190 microns), ordinary chondrules  
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4 (160 microns), and CM, CO, EH, and EL chondrules (30-60 microns). The compositional  
5 differences in Fe-Ni metal nodules in igneous rims and the implications of these  
6 differences for chondrule formation are discussed in sections below. Using on 3D  
7 tomography, Ebel et al. (2008) imaged the olivine-pyroxene layering sequence in  
8 chondrules with igneous rims. They argued that because olivine condenses at higher  
9 temperatures than pyroxene, the olivine-pyroxene layering sequence is consistent with  
10 high-temperature equilibration of chondrule melts with a solar gas. Weisberg et al. (1993)  
11 suggested that both the chondrule cores and igneous rims originated from the same  
12 reservoirs. CR chondrites also contain compound chondrules (e.g., Wasson and Rubin,  
13 2010). These objects consist of two chondrules, either in contact or nested within each  
14 other. Finally, some CR chondrules have agglomeratic, or partially melted grains, ranging  
15 in sizes from 1-80 microns (e.g., Schrader et al., 2018a).

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Chondrule sizes have been used both as criteria to classify chondrites and as a  
constraint for chondrule formation conditions. Like CV and CK chondrites, chondrules in  
CR chondrites are large compared with CM and CO chondrules (e.g., Bischoff et al.,  
1993a; Kallemeyn et al., 1994; Wasson and Rubin, 2010; Friedrich et al., 2015). For a  
relatively small sample, Kallemeyn et al. (1994) reported that the average radius of CR  
chondrules was ~ 700  $\mu\text{m}$ . Based on the textural complexity of CR chondrules, Friedrich  
et al. (2015) cautioned that it is possible that different authors used different delineation  
criteria for chondrule edges and introducing additional uncertainty in the measurements.  
Campbell et al. (2005) hypothesized that the grain size of silicate phenocrysts increases  
with the degree to which chondrule precursors melted in the nebula. Along with CV and  
CK chondrites, Wasson and Rubin (2010) attributed the large chondrule diameters of CR  
chondrules to re-melting in slow-cooling, dust-rich environments, in which chondrules  
accumulated dust layers on their surfaces.

In addition to mineralogical and textural classifications, CR chondrules are  
subdivided into three types based on their chemical compositions, which are thought to  
have been established under different nebular conditions. Type I and type II chondrules  
are classified based on the iron to magnesium ratio of their olivines and pyroxenes (Fig.  
4; Tables 3, 4). Type I chondrules are reduced, MgO-rich, and FeO-poor ( $\text{Fe}/(\text{Fe}+\text{Mg})$   
<10%). Type II chondrules are oxidized, MgO-poor, and FeO-rich ( $\text{Fe}/(\text{Fe}+\text{Mg})$  >10%).

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4 In addition to ferromagnesian chondrules, CR chondrites contain objects dominated by  
5 high-Ca, low-Ca pyroxenes, and Mg-bearing anorthitic plagioclase termed Anorthite-  
6 Rich Chondrules or ARCs, which have affinities somewhat intermediary between types I  
7 and II chondrules and CAIs (e.g., Krot and Keil, 2002; Table 5).  
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11 In the CR chondrites, there is a sharp gap in iron to magnesium ratios between type I  
12 and type II chondrules. This gap is more prominent than in other carbonaceous chondrite  
13 groups. Schrader et al. (2017) interpreted this paucity in composition as evidence that  
14 chondrules in CR chondrites formed under distinct nebular conditions and that no  
15 chondrules from other groups mixed with the CRs.  
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21 Of the three compositional types of chondrules, only type I chondrules are abundant  
22 in CR chondrites. These MgO-rich chondrules account for ~63.1% (range 30-74%) of  
23 the total CR chondrites' areas, as studied by Schrader et al. (2011). Type II chondrules  
24 comprise approximately 2.3 area % (range 0.1 to 5.6%), and ARCs are the least common,  
25 accounting for ~0.5% (range 0-1.6%) of the total CR areas (Schrader et al. 2011).  
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#### 30 31 *4.1. Type I chondrules* 32 33

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35 Type I chondrules vary in size from ~400-1000  $\mu\text{m}$  and may reach up to 3 mm in  
36 apparent diameter (e.g., Fig. 4a,c). They are relatively rich in Fe-Ni metal and are often  
37 multilayered, with cores dominated by forsteritic olivine and in some cases low-Ca  
38 pyroxene phenocrysts, and silicate-rich layers (e.g., Weisberg et al., 1993; Kallemeyn et  
39 al., 1994; Krot et al., 2002; Schrader et al., 2008, 2013, 2015; Tenner et al., 2013, 2015;  
40 **Table 3**). Igneous rims have silica-rich outer layers. In addition, these chondrules have  
41 mesostasis, containing nanophase crystals, which in some cases are partially to entirely  
42 replaced by phyllosilicates, most commonly serpentine and chlorite (e.g., Weisberg et al.,  
43 1993; Keller, 2011). An example of mesostasis containing predominantly anhydrous  
44 nanocrystalline phases is shown in Fig. 5 a-d. Type I mesostases have higher albite and  
45 orthoclase contents than mesostases in type II chondrules (e.g., Krot et al., 2002).  
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51 Whereas Fe-sulfides are common in type I chondrules in other chondrite groups, they  
52 are scarce in type I CR chondrules (e.g., Wasson and Rubin, 2010). The low abundance  
53 of sulfur in type I chondrules has been interpreted as evidence of multiple heating events  
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4 (e.g., Schrader et al., 2008; Wasson and Rubin, 2010; Jacquet et al., 2013). However, the  
5 mechanism and conditions under which S losses occurred is an active area of research.  
6 For example, Schrader et al. (2008) argued that type I chondrules formed by heating gas  
7 and dust in a reducing and non-sulfidizing (low  $fO_2$  and  $fS_2$ ) region. In contrast, Wasson  
8 and Rubin (2010) suggested that the low S contents of type I CR chondrules resulted  
9 from slow cooling during re-melting events, which allowed migration of S to chondrule  
10 surfaces and eventual evaporative losses. Jacquet et al. (2013) also argued for at least two  
11 nebular heating events affecting type I CR chondrules, with the first event lasting > 1 day  
12 and resulting in equilibration between chondrules and Fe-Ni metal and in evaporative loss  
13 of Fe from the chondrule. This first heating event would be followed by accretion of a  
14 fine-grained mantle on the chondrule surface. Like Wasson and Rubin (2010), Jacquet et  
15 al. (2013) suggested that a second heating event would explain the formation of igneous  
16 rims. However, unlike Wasson and Rubin (2010), who suggested for the second heating  
17 event to be slow, Jacquet et al. (2013) argued that the second heating event was rapid, at  
18 least compared to element diffusion and limited to the outside of the chondrule.  
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#### 32 33 *4.1.1. Fe-Ni metal in Type I CR chondrules* 34 35 36

37 Numerous studies have focused on the characteristics and origins of Fe-Ni metal  
38 nodules in CR chondrites and used these characteristics to constrain conditions for  
39 chondrule formation (e.g., Wood, 1963a,b; Lee et al., 1992; Kong et al., 1999; Connolly  
40 et al., 2001; Ebel et al., 2008; Jacquet et al., 2013). These nodules are the reservoir of  
41 approximately a third of the total iron in CR chondrites (Campbell et al., 2005). Fe-Ni  
42 metal nodules are found in the interiors of chondrule phenocrysts and mesostasis (~45-  
43 50%), chondrule margins (~15%), and embedded in fine-grained matrix (e.g., Lee et al.,  
44 1992; Connolly et al., 2001; Ebel et al., 2008). Ebel et al. (2008) used X-ray  
45 microtomography to determine that Fe-Ni grains make up to 37% of the volume of type I  
46 chondrules. The distribution, as well as the composition and textural characteristics of  
47 these nodules, appear to vary with petrologic setting and degree of chondrule melting  
48 (e.g., Lee et al., 1992; Weisberg et al., 1993; Connolly et al., 2001; Humayun et al., 2002;  
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4 Campbell et al., 2005). Fe-Ni metal is concentrated in porphyritic type I chondrules,  
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6 whereas non-porphyritic types do not generally contain any.  
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8 The petrologic setting of Fe-Ni metal nodules places constraints on their chemical  
9 compositions. Those in chondrule interiors have higher contents of refractory siderophile  
10 elements than Fe-Ni metal on margins (e.g., Co, Ni, Re, Os, W, Ir, Ru, Mo, Pt - Jacquet  
11 et al., 2013). Conversely, Fe-Ni metal located on chondrule margins has relatively higher  
12 abundances of moderately volatile elements (e.g., Fe, Cu, Au). Campbell et al. (2005)  
13 noted that small Fe-Ni metal grains that are poikilitically enclosed in MgO-rich silicates  
14 have high, correlated concentrations of Ni and P and low Fe and Cr. Moreover, Wasson  
15 and Rubin (2010) observed that Fe-Ni metal composition is uniform within individual  
16 concentric shells in chondrules. Fe-Ni metal grain sizes are also smaller in igneous rims  
17 around chondrules compared with chondrule interiors and margins (e.g., Fig. 4b).  
18 However, as Ebel et al. (2008) found, the two-dimensional arrangement of Fe-Ni metal  
19 grains in chondrules – on which these correlations rely – may not be representative of the  
20 three-dimensional arrangement of metal.  
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32 Based on the shapes and sizes of silicate phenocrysts, Zanda et al. (2002) determined  
33 that the texture of Fe-Ni metal grains varies with the degree to which their CR chondrule  
34 hosts underwent melting. They reported that Fe-Ni metal is stochastically distributed  
35 throughout type I chondrules with small silicate phenocryst sizes. These chondrules with  
36 finer-grained phenocrysts would have formed by less extensive melting episodes than  
37 chondrules with round outlines and large silicate phenocrysts, which have well-developed  
38 shells made of coarse-grained Fe-Ni metal grains. In extensively melted chondrules,  
39 olivines have higher forsterite contents and Fe-Ni metal grains have higher Ni and P.  
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46 Several studies argued for a common origin for Fe-Ni metal nodules in all chondrules  
47 and matrix (e.g., Zanda et al., 1994; Kong et al., 1999). In fact, Kong et al. (1999)  
48 suggested that Fe-Ni metal grains outside chondrules probably were also derived from  
49 chondrules. How these grains formed, however, has been the subject of debate. To test  
50 the validity of Fe-Ni metal formational hypotheses, their petrologic textures, settings, and  
51 minor and trace elements were studied. Formational hypotheses for metal grains include:  
52 (1) reduction during thermal metamorphism in the CR parent body(s); (2) direct  
53 condensation from a nebular gas; (3) evaporation and re-condensation during chondrule  
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4 formation; and (4) segregation of immiscible metallic Fe melts from chondrule melts in  
5 the nebula.  
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8 While current discussions center around a nebular setting, early workers suggested  
9 that Fe-Ni metal formed by reduction of  $\text{Fe}^{2+}$  during thermal metamorphism (e.g., Lee et  
10 al., 1992). This hypothesis is inconsistent with newer evidence that thermal  
11 metamorphism of the CR parent body was minimal (e.g., Wasson and Rubin, 2010;  
12 Schrader et al., 2015; Abreu, 2016a). Also, Wasson and Rubin (2010) noted that the  
13 negative correlation between Ni in the metallic shell and olivine Fa contents in the host  
14 silicates precludes in-situ, thermally-driven reduction.  
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21 A second hypothesis that has fallen out of favor is the direct condensation of Fe-Ni  
22 metal from a nebular gas. Early studies appeared to indicate that Fe-Ni metal could have  
23 condensed directly from the nebula before silicate formation, based on their abundances  
24 of Ni, Cr, and P (e.g., Wood, 1963a,b, 1967; Grossman and Olsen, 1974). However, later  
25 work indicated that some CRs show a positive, linear correlation between Co and Ni and  
26 an approximately solar ratio of  $\sim 0.046$  (Lee et al. 1992; Weisberg et al. 1993; Connolly  
27 et al. 2001; Jacquet et al. 2013). Weisberg et al. (1993) interpreted the correlation  
28 between Co and Ni as a result of thermodynamically-controlled condensation under  
29 nebular conditions. It is noteworthy that several studies show that the Co/Ni ratios are not  
30 constant within individual or across different minimally heated CRs (e.g., Zanda et al.,  
31 1994, 2002; Wasson and Rubin, 2010; Schrader et al., 2015). Researchers have argued  
32 against the direct nebular condensation hypothesis, based on the presence of a graphite  
33 layer on Fe-Ni metal surfaces, thermodynamic equilibrium of metallic Fe melts under  
34 nebular conditions, and trace siderophile element abundances (e.g., Kong et al., 1999;  
35 Connolly et al., 2001; Humayun et al., 2002; Jacquet et al., 2013). Kong et al. (1999)  
36 observed that Fe-Ni metal grains have low-Ni contents and were embedded in graphite  
37 layers. They argued that scenarios for nebular condensation could either reproduce the  
38 low-Ni composition of Fe-Ni metal or the presence of the graphite layer, but not both  
39 features. The Ni contents would require isolation of metal from the nebular gas at high  
40 temperature. In contrast, graphite could only be stable at temperatures of 600K (at  $10^{-5}$   
41 bar) in the nebula, requiring long exposure of Fe-Ni metal to the nebular gas at low  
42 temperatures. Finally, Humayun et al. (2002) found a linear correlation between Ni/Fe  
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4 and Pd/Fe ratios in the CR2 Renazzo, which cannot be explained by the order of  
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6 condensation from a gas of solar composition.  
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8 The two remaining hypotheses suggest that Fe-Ni metal formed in the aftermath or as  
9 a part of the chondrule-forming events. Wasson and Rubin (2010) suggested that  
10 separation of Ni and Co from Fe-metal occurred during volatilization at low ambient  
11 nebular temperatures associated with chondrule formation. Some argued that large Fe-Ni  
12 metal grains formed by reduction of FeO and separation from the silicate fraction of  
13 chondrule melts (Connolly et al., 1994, 2001; Zanda et al., 1994; Campbell et al. 2002,  
14 2005). This hypothesis was initially based on the depletion of volatile siderophile  
15 elements. Connolly et al. (2001) suggested that FeO reduction was assisted by the  
16 presence of carbon. Graphite-assisted reduction is supported by the fact that this mineral  
17 is spatially associated with Fe-Ni metal (Kong et al., 1999). Some studies explained the  
18 Co-Ni trend observed in some CR metal in terms of oxidation/reduction reactions, based  
19 on the observation that Fe-Ni metal and silicates appear to be in redox equilibrium with  
20 each other within individual chondrules (Lee et al., 1992; Zanda et al., 1994). Because Ni  
21 and Co are slightly more refractory than Fe, the initial metal composition would have  
22 been enriched in Ni and Co with respect to the last condensates.  
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35 The last two hypotheses are not mutually exclusive. In fact, Connolly et al. (2001)  
36 suggested that they were both at play, depending on whether Fe-Ni metal grains were  
37 located in rims or interiors of chondrules. They suggest that rim metal grains formed by  
38 evaporation/re-condensation. Metal grains in chondrule interiors would have formed by  
39 the separation of immiscible metal-silicate chondrule melt. Connolly et al. (2001) and  
40 others also argued that at least some of the Fe-Ni metal formed by devolatilization,  
41 concurrent with the formation of immiscible metal-silicate chondrule melts and  
42 subsequent recondensation from the resulting vapor, which would be depleted in  
43 refractory siderophiles (Kong et al., 1999; Kong and Palme, 1999). This scenario can  
44 explain both the fact that rim Fe-Ni metal grains are depleted in refractory siderophiles as  
45 well as the complementarity between the Au, As, Sb, and Ga contents of coarse Fe-Ni  
46 metal and non-magnetic Fe-Ni matrix phases. However, Campbell et al. (2005) argued  
47 that such a scenario is unlikely because elements such as Au and Cu may not be easily  
48 mobilized from the melt under the kind of reducing conditions needed to form Fe-Ni  
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4 metal. Jacquet et al. (2013) presented additional evidence against the re-condensation of  
5 metal vapor based on their observations of inconsistent depletion of refractory siderophile  
6 elements in metal along chondrule margins compared to chondrule interiors.  
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10 A variety of different types of Fe-Ni metal precursors have been proposed. Some  
11 authors have suggested that decomposition of pre-existing Fe-Ni sulfides resulted in the  
12 formation of Fe-Ni metal grains (e.g., Zanda et al., 1997, 2002; Hewins et al., 1997).  
13 Zanda et al. (2002) argued that the correlation between the degree of melting and the  
14 composition of silicates and Fe-Ni metal supports this hypothesis. In contrast, Jacquet et  
15 al. (2013) suggested that Fe-Ni metal precursors resembled the least melted chondrules.  
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20 As stated above, Fe-Ni metal nodules are commonly found in concentric layers within  
21 and along the periphery of chondrules. Multiple hypotheses have been proposed to  
22 explain the arrangement of Fe-Ni metal. Wood and McSween (1977b) explained the  
23 shell-like distribution of Fe-Ni metal by addition and/or condensation of successive  
24 layers in the nebula. Alternatively, multiple authors have suggested that metallic Fe  
25 migrated to the periphery of chondrules via centrifugal force density differential, acting  
26 on immiscible metallic Fe versus forsteritic melts (e.g., Grossman and Wasson, 1985;  
27 Kong et al., 1999; Kong and Palme, 1999; Humayun et al., 2002; Tsuchiyama et al.,  
28 2009). Wasson and Rubin (2010) argued that centrifugal forces could not reproduce the  
29 shell-like, concentric structures that Fe-Ni globules formed within many chondrules.  
30 Instead, they suggested that surface tension resulted in the mobilization of the metallic Fe  
31 melt across channels. The metallic melt then redistributed itself on the surface of the  
32 silicate melt via interfacial tension. If the rate of cooling of the mobilized metallic Fe  
33 melt was slow compared to Ni diffusion, this would result in the uniform composition of  
34 individual concentric shells, but this model is inconsistent with the trace element data  
35 obtained by Connolly et al. (2001). Wasson and Rubin (2010) argued that the variations  
36 in PGE contents within shell reported by Connolly et al. (2001) were artifacts of ion  
37 microprobe analysis. However, similar PGE patterns were independently reproduced  
38 using ICP-MS (Humayun et al., 2002; Campbell et al., 2002).  
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#### 57 *4.1.2. Mesostasis of Type I chondrules*

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4           Only a few studies of CR mesostasis are currently available (e.g., Richardson, 1981;  
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6 Ichikawa and Ikeda, 1995; Burger, 2005; Harju et al., 2014; Tenner et al., 2015). It is  
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8 noteworthy that some studies did not specify the type of chondrule where mesostases  
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10 were described. In both type I and type II chondrules, mesostasis generally appears non-  
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12 isotropic or made up of minute crystals too small for reliable resolution and identification  
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14 via optical microscope. Isotropic glass in type I chondrules was only identified in a few  
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16 pristine CRs (Harju et al., 2014; Burger, 2005; Abreu and Brearley, 2010). Not all  
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18 members of each pairing group contain clear mesostasis, revealing heterogeneities in  
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20 their asteroidal and/or terrestrial histories.

21           Mesostasis anisotropies have been generally attributed to the presence of asteroidal  
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23 phyllosilicates or excessive thin section thickness (cf., Harju et al., 2014). However,  
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25 conclusive identification of phyllosilicates in CR mesostasis has only been carried out for  
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27 MIL 090001 and GRO 95577 (Keller, 2011). In a MIL 090001 thin section, emerald-  
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29 green to slightly reddish materials are found in chondrule interiors, consistent with  
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31 phyllosilicate replacing the glass. Mesostasis in GRO 95577 underwent both  
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33 compositional and textural integration with chondrule phenocrysts, which makes it  
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35 challenging to determine which phyllosilicates replaced which materials. Tenner et al.  
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37 (2015) described the mineralogy of mesostasis in 48 type I chondrules from MET 00426  
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39 and QUE 99177 as heterogeneous in texture, typically consisting of fine-to-coarse  
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41 grained pyroxene and plagioclase. Tenner et al. (2014) reported silica excesses in  
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43 chondrule plagioclase.

44           Harju et al. (2014) measured the composition of the mesostasis in POP two  
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46 chondrules in LAP 02342 and EET 92062. In LAP 02342, they found that the boundary  
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48 between mesostasis and fine-grained matrix was enriched in Na and depleted in Ca, Ti,  
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50 and Cr. They attributed these trends to parent body aqueous alteration. Tenner et al.  
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52 (2015) used EDS to measure the composition of the mesostasis in one region of  
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54 recrystallized mesostasis (SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, FeO, MgO, CaO, and Na<sub>2</sub>O contents of 56, 24, 1,  
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56 3, 12, and 2.5 wt.%, respectively) and one region of glassy mesostasis (SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, FeO,  
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58 MgO, CaO, and Na<sub>2</sub>O contents 68, 15, 4, 1.5, 6.5, and 3.5 wt.%, respectively).  
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4 *4.1.3. Oxygen isotopes in type I chondrules*  
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8 Multiple authors have analyzed the oxygen isotopic compositions of type I  
9 chondrules in CRs (e.g., Weisberg et al., 1993; Krot et al., 2006a-b; Schrader et al., 2013,  
10 2014, 2017; Tenner et al., 2015). Krot et al. (2006b) measured the O-isotopic  
11 compositions of all major phases in nine chondrules from three CRs and found that they  
12 plot near the carbonaceous chondrite anhydrous mineral (CCAM) line. They observed  
13 that, except for relict grains, chondrules are isotopically homogeneous, with all minerals  
14 and mesostasis from each chondrule plotting in small clusters at  $\Delta^{17}\text{O} \sim \pm 2 \text{ ‰}$ .  
15 Interchondrule variations range from  $\Delta^{17}\text{O} = 0$  to  $-5 \text{ ‰}$ . Krot et al. (2006b) suggested that  
16 the oxygen isotopic composition of CR chondrules was established during melting and  
17 isotopic homogenization of precursor materials with a  $^{16}\text{O}$ -poor gas. Schrader et al.  
18 (2013) studied olivine grains in eight type I chondrules from pristine CRs QUE 99177  
19 and GRA 95229 and heated GRA 06100 using secondary ion mass spectrometry. They  
20 concluded that individual chondrules acted as independent igneous systems with their  
21 own isotopic composition. They also argued that aqueous alteration of the meteorites  
22 studied had not modified the oxygen isotopes of their chondrules.  
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35 Schrader et al. (2014) analyzed 7 barred olivine (BO) chondrules and 3 silica-bearing  
36 porphyritic type I chondrules from Gao-Guenie (b), GRA 95229, PCA 91082, and Shişr  
37 033. They found that the oxygen-isotope compositions of olivine plotted along a slope-1  
38 line. Oxygen isotopic variations of BO chondrule olivines are small (i.e.,  $\Delta^{17}\text{O} \sim 2.5 \text{ ‰}$ ).  
39 They noted that olivine grains in type I BO chondrules have similar major and minor  
40 elements and O-isotope compositions as those of olivine in porphyritic chondrules. Based  
41 on these similarities in isotopic compositions, they argued against the chondrule  
42 formation model in which porphyritic chondrules formed in transient planetesimals.  
43 Using secondary ion mass spectrometry, Tenner et al. (2015) measured the O-isotopic  
44 composition of olivine, pyroxene, and plagioclase in 45 type I chondrules from pristine  
45 CRs QUE 99177 and MET 00426. Consistent with prior studies, they observed that  $\Delta^{17}\text{O}$   
46 values are nearly homogeneous for all minerals analyzed.  
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57 The question of whether the O isotopes in chondrule melts equilibrated with the  
58 nebular gas is an area of active research. Schrader et al. (2014) concluded that even  
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4 barred olivine chondrules, which were at one point completely melted, only record partial  
5 equilibration with the nebular gas. In contrast, Tenner et al. (2015) argued that the  
6 chondrule melts and the surrounding nebular gas exchanged oxygen isotopes efficiently.  
7 They found an inverse correlation between  $\Delta^{17}\text{O}$  and Mg-numbers, which they suggested  
8 had implications for the processes forming type II chondrules. They used the trend  
9 between  $\Delta^{17}\text{O}$  and Mg-numbers to constrain the amount of dust and/or  $\text{H}_2\text{O}$   
10 enhancements needed for type I chondrules to form. They proposed a 100-200x dust  
11 enrichment and that these dust precursors had 0-0.8 x CI chondrite water abundance.  
12 Schrader et al. (2017), however, noted that their study of seven other CR chondrules did  
13 not reproduce the correlation between  $\Delta^{17}\text{O}$  and Mg-numbers.  
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#### 26 *4.1.4. Radiogenic isotopes in Type I chondrules*

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30 Several systems have been investigated for these chondrules, including the Al-Mg,  
31 Hf-W, and Mn-Cr geochronometers. Studies of the Al-Mg systematics of CR chondrules  
32 show few  $^{26}\text{Mg}$  excesses (e.g., Krot and Keil, 2002; Hutcheon et al., 2009; Nagashima et  
33 al., 2014; Olsen et al., 2016). Schrader et al. (2017) and Budde et al. (2018)  
34 independently found the same average age for CR chondrite chondrules using different  
35 isotopic systems.  
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40 Schrader et al. (2017) chose CRs that had not been sufficiently affected by thermal  
41 metamorphism to ensure that their isotopic signatures had not been altered and measured  
42 the oxygen and Al-Mg systematics for seven type I porphyritic chondrules. They  
43 estimated that only approximately 32% of chondrules show resolvable excesses of  $^{26}\text{Mg}$ .  
44 Furthermore, based on the probability density of the Al-Mg ages, they suggested that CR  
45 chondrites contain three different generations of chondrules. Some authors have  
46 interpreted the scarcity of chondrules with  $^{26}\text{Mg}$  excess as evidence that CR chondrules  
47 are younger than their counterparts from other chondritic groups. Other researchers  
48 attributed this paucity to spatial variations of the  $^{26}\text{Al}/^{27}\text{Al}$  ratio in the protoplanetary  
49 disk. Olsen et al. (2016) interpreted the variability in  $^{26}\text{Mg}$  as the result of Mg-isotope  
50 heterogeneity of the chondrule precursors.  
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4 Budde et al. (2018) applied the short-lived  $^{182}\text{Hf}$ - $^{182}\text{W}$  chronometer to chondrule Fe-  
5 Ni metal and obtained ages that are  $3.6 \pm 0.6$  million years (Ma) after the formation of  
6 CAIs, which corresponds to  $\sim 1$ – $2$  Ma later than chondrules found in most other groups of  
7 chondrite. The younger chondrule ages support the hypothesis of late CR chondrule  
8 formation. Along with the low variability in  $^{54}\text{Cr}$  observed in CR chondrules, Olsen et al.  
9 (2016) argued that these chondrules must have formed in a spatially constrained region,  
10 possibly located in the outer solar system.  
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#### 19 *4.2. Type II chondrules*

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22 Type II chondrules (e.g., Fig. 4 c,d; Table 4) are much less abundant than type I  
23 chondrules, and they often occur as fragments (41%; Schrader et al., 2015). Schrader et  
24 al. (2008) suggested that the preferential breakup of type II chondrules resulted either  
25 from formation in a more dynamically active or denser nebular region or because these  
26 chondrules are more brittle and/or porous than type Is.  
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31 Several authors have explored the relationship between type I and type II chondrules.  
32 Connolly et al. (2008) argued that type II chondrules formed by flash-heating of both dust  
33 and pre-existing type I chondrules under higher than canonical  $f\text{O}_2$  and  $f\text{S}_2$ . Connolly and  
34 Huss (2010) tested this hypothesis by measuring the oxygen isotopic compositions of  
35 type II CR chondrules (see below in the discussion of isotopes). Schrader et al. (2013)  
36 studied the relationship between the FeO contents of olivines and their minor element  
37 contents (MnO,  $\text{Cr}_2\text{O}_3$ ,  $\text{Na}_2\text{O}$ ) to ascertain whether type I chondrules are the precursor  
38 materials for type II chondrules. They argued that this relationship is unlikely. Schrader  
39 et al. (2015) observed compositional overlaps between FeO and minor elements in type I  
40 silicate phenocrysts compared with FeO-poor silicate relicts located in type II chondrules.  
41 Overlaps suggest that while the origin of type II chondrules from type I chondrule  
42 materials is possible, it is also possible that both types originated from similar precursors.  
43 In either case, additional precursors were invoked to account for compositional  
44 differences between these two types of chondrules.  
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57 To establish if a relationship among type II chondrules from different chondritic  
58 groups exists, the Fe-Mg-Mn systematics of silicates in CR type II chondrules were  
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4 studied: Berlin et al. (2011) argued that chondrules from different chondritic groups  
5 could be differentiated based on the trends defined by their Fe/Mn versus Fe/Mg ratios.  
6 However, Schrader et al. (2015) collected a much larger dataset that showed significant  
7 overlap among groups. They suggested that a variety of factors might explain the Fe-Mg-  
8 Mn systematics, including (1) individual type II chondrules behaving as individual  
9 igneous systems formed under distinct oxygen fugacities (Schrader et al., 2013); (2)  
10 formation from variable precursors (e.g., Berlin et al., 2011); (3) formation at different  
11 cooling rates (e.g., Jones and Lofgren, 1993; Berlin et al., 2011); or (4) experiencing  
12 complete/incomplete condensation.  
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#### 22 *4.2.1. Opaque assemblages in Type II chondrule*

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26 Compared with type I chondrules, the mineralogy of opaques is more complex in type  
27 II chondrules (Schrader et al., 2008, 2015). Opaques in type II chondrules are Fe-Ni  
28 metal, troilite, pentlandite, magnetite, and phosphates (e.g., Schrader et al., 2008, 2015).  
29 Although Schrader et al. (2008) had indirectly identified tochilinite in these assemblages,  
30 a later study determined that these occurrences were instead sub-micron mixtures of  
31 magnetite, pyrrhotite, and pentlandite (Schrader et al., 2015).  
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37 Schrader et al. (2008) proposed a two-stage process for the formation of these opaque  
38 assemblages. In the first stage, Fe-Ni metal and troilite would form by flash-  
39 heating/cooling, condensation and corrosion reactions during chondrule formation in the  
40 solar nebula. They envisioned that this was a dynamically violent environment,  
41 explaining the fragmentation of chondrules described above (Schrader et al., 2008). In the  
42 second stage, magnetite and phosphates formed via replacement of Fe-Ni metal in the CR  
43 parent body at temperatures ~ 300°C, which they supported with subsequent experimental  
44 studies (Schrader and Lauretta, 2010). They reproduced the mineralogy of the type II  
45 chondrule opaque assemblages by reacting Co, Cr, P-bearing, Fe-Ni alloys with H<sub>2</sub>-H<sub>2</sub>O-  
46 CO-CO<sub>2</sub>-H<sub>2</sub>S gas mixtures at 1000°C and cooling them at a 3000°C/hour rate.  
47 Subsequent TEM observations showed that pentlandite in type II chondrules might be  
48 nebular (Schrader and Lauretta, 2010; Schrader et al., 2015). Schrader et al. (2018b)  
49 further constrained the subsolidus cooling rates for 230-600 °C to be 10<sup>0</sup>-10<sup>1</sup>°C/hour  
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4 using type II chondrule pyrrhotite and pentlandite. These authors argued that the rapid  
5 cooling rates are consistent with chondrule formation by shocks triggered by the  
6 accretion of Jupiter or other planetary embryos.  
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#### 10 11 12 13 *4.2.2. Mesostasis in Type II chondrules* 14 15

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17 Few studies have described the mesostasis in type II CR chondrules. Tenner et al.  
18 (2015) characterized the mesostasis in 3 type II chondrules in MET 00426 and QUE  
19 99177 as glassy, and they used EDS to measure the composition in one region of glassy  
20 mesostasis to be 60 SiO<sub>2</sub>, 8-14 Al<sub>2</sub>O<sub>3</sub>, 13, FeO, 2 MgO, 5 CaO, and 0.5 Na<sub>2</sub>O (all in  
21 wt.%). Burger (2005) collected EPMA point analyses of mesostases in type II chondrules  
22 from paired EET 87770 and EET 92106. He measured lower Al<sub>2</sub>O<sub>3</sub> and significantly  
23 higher Na<sub>2</sub>O contents (53-61 SiO<sub>2</sub>, 5-7 Al<sub>2</sub>O<sub>3</sub>, 12-16, FeO, 2-6 MgO, 2-3 CaO, and 7-10  
24 Na<sub>2</sub>O (all in wt.%). Burger (2005) identified small areas of glassy mesostasis. Aqueous  
25 alteration appears to have affected mesostasis in type II chondrules, but these effects  
26 appear to be spatially limited (e.g., Burger, 2005; Brearley and Burger, 2009).  
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#### 37 *4.2.3. Oxygen isotopes in Type II chondrules* 38 39 40

41 Although individual type II CR chondrules show larger  $\Delta^{17}\text{O}$  values than type I  
42 chondrules, their oxygen isotopic compositions are generally similar (e.g., Krot et al.,  
43 2006b; Connolly and Huss, 2010; Schrader et al., 2013). Krot et al. (2006b) measured the  
44 O-isotopic composition of olivine and mesostasis in one type II chondrule from MAC  
45 87320, which generally plotted between the CCAM and the terrestrial fractionation line.  
46 Connolly and Huss (2010) analyzed the oxygen isotopic composition of 11 type II  
47 porphyritic chondrules from Renazzo, EET 92011, and MAC 87320. Similar to type I CR  
48 chondrules, minerals in individual chondrules are nearly isotopically homogenous within  
49 each chondrule, and each chondrule is distinct from others. Connolly and Huss (2010)  
50 found that type II chondrules tend to be more  $\delta^{17}\text{O}$ - and  $\delta^{18}\text{O}$ -rich than type I chondrules  
51 and have  $\Delta^{17}\text{O}$  ranging from  $-2.0\pm 0.7$  to  $0.5\pm 0.7\text{‰}$ . Schrader et al. (2013) measured  
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4 olivines in 8 relict-free, type II chondrules from weakly alternated QUE 99177 and GRA  
5 95229 and heated GRA 06100. These analyses included POP and compound BO/POP  
6 and radial olivine/POP chondrules. For individual chondrules, olivines have O-isotope  
7 compositions ranging in  $\delta^{17}\text{O} = 1.7\text{-}3.9\text{‰}$ , in  $\delta^{18}\text{O} = 5.7\text{-}8.0\text{‰}$ , and in  $\Delta^{17}\text{O} = -1.8\text{-}0.9\text{‰}$ .

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12 Connolly and Huss (2010) observed that relict grains in type II chondrules have a  
13 lighter oxygen isotopic composition than their hosts. They hypothesized that these relict  
14 grains originated from type I CR chondrules. They also envisioned that type II chondrule  
15 formation was aided by the combination of dust enhancements and the addition of  $^{16}\text{O}$ -  
16 poor ( $^{17}\text{O}$  and  $^{18}\text{O}$ -rich) ice to precursors. Schrader et al. (2013) found that relict-grain-  
17 bearing type II chondrules have much larger ranges in  $\delta^{17}\text{O}$  and  $\delta^{18}\text{O}$  ( $\sim 14\text{‰}$ ) and in  
18  $\Delta^{17}\text{O}$  ( $\sim 7\text{‰}$ ) compared to relict-free ones. They calculated that relict-grain-free type II  
19 chondrules formed under  $\text{H}_2\text{O}/\text{H}_2$  ratios between 230 and 740 times solar, whereas, relict  
20 grain-bearing type-II chondrules formed under a narrower range of  $\text{H}_2\text{O}/\text{H}_2$  ratios, 350  
21 and 510 times solar. They suggested that the type II CR chondrule dust precursor had a  
22 higher S content than the type I chondrule precursor, accounting for the paucity of  
23 sulfides in type I chondrules.

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Tenner et al. (2015) studied three type II chondrules from pristine QUE 99177 and MET 00426. As discussed above, they used type I chondrule trends between  $\Delta^{17}\text{O}$  and Mg-numbers to support the hypothesis of Connolly and Huss (2010), in which type I chondrules were the precursors for type II chondrules.

Schrader et al. (2014) studied four barred olivine and one porphyritic olivine pyroxene chondrule from Gao-Guenie (b), GRA 95229, PCA 91082, and Shişr 033. They found that the oxygen-isotope compositions of olivines in type II BO chondrules span a slightly larger range in  $\Delta^{17}\text{O} \sim 2.1 \text{‰}$  compared with olivines in BO type I chondrules. Based on their observations, they suggested that there was isotopic exchange, but not homogenization between chondrules and the surrounding nebular gas.

#### 4.2.4. Radiogenic isotopes in Type II chondrules

Studies of radiogenic isotopes from type II chondrules are scarce. Nagashima et al. (2014) found excesses of  $^{26}\text{Mg}$  at the 2-sigma level in one out of two mineralogically

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4 pristine type II chondrules from GRA 95229. Schrader et al. (2017) observed a  $^{26}\text{Mg}$   
5 excess in plagioclase while simultaneously measuring the oxygen isotopes and the Al-Mg  
6 systematics of one type II porphyritic CR chondrule from heated GRO 03116, suggesting  
7 that at least some type II CR chondrules might have formed before the formation of most  
8 type I CR chondrules.  
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#### 10 11 12 13 14 15 *4.3. Anorthite-rich chondrules* 16

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19 Anorthite-rich chondrules (ARCs) are the least abundant chondrule type (<1 vol.%) in  
20 CR chondrites. Some ARCs have regions that resemble type I chondrules. In particular,  
21 ARCs are dominated by pyroxene and olivine phenocrysts, which generally have high  
22 Mg-numbers and also contain Fe-Ni metal nodules (Krot and Keil, 2002; Table 5). ARCs  
23 have crystalline mesostases made up of silica, anorthite, and high-Ca pyroxene.  
24 However, anorthite-rich chondrules are mineralogically distinct from other types (e.g.,  
25 Krot et al., 2002). Unlike type I chondrules, ARCs have interstitial anorthite and Al-Ti-  
26 Cr-rich low-Ca and high-Ca pyroxenes, and in some cases, contain refractory inclusions  
27 (Krot and Keil, 2002). The association with refractory inclusions is the basis of the  
28 argument that ARCs formed by melting of the spinel–anorthite–pyroxene CAIs mixed  
29 with ferromagnesian precursors compositionally similar to type I chondrules (Krot and  
30 Keil, 2002; Krot et al., 2006b). The high abundances of moderately-volatile elements in  
31 ARCs suggest that these chondrules could not have been formed by the volatilization of  
32 type I chondrule precursors or by the melting of refractory materials only (e.g., Cr, Mn,  
33 Si; Krot and Keil, 2002).  
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46 Krot et al. (2006b) measured the isotopic composition of different minerals in five  
47 ARCs. Unlike type I and type II chondrules, which they found to be essentially  
48 isotopically homogeneous, they reported that two out of five ARCs were heterogeneous,  
49 with large variations in  $\Delta^{17}\text{O}$  (-6 to -15‰ and -2 to -11‰). They attributed these  
50 variations to the formation from incomplete melting of CAI-like precursors.  
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55 Using SIMS, Nagashima et al. (2014) measured seven ARCs. Not all these  
56 chondrules showed evidence of in-situ decay of  $^{26}\text{Al}$ . They noted that  $^{26}\text{Mg}$  excesses in  
57 CR chondrules were small compared with excesses in other pristine chondrites, such as  
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4 LLS, COs, or unique carbonaceous chondrite Acfer 094. These small  $^{26}\text{Mg}$  excesses were  
5 attributed to CR chondrule formation occurring about a million years after chondrules  
6 from other pristine chondrites.  
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## 10 11 **5. Refractory inclusions in CR chondrites** 12 13 14

15 Refractory inclusion is an umbrella term for both calcium-aluminum-rich refractory  
16 inclusions (CAIs) and amoeboid olivine aggregates (AOAs) (e.g., McPherson et al.  
17 2014). Refractory inclusions are not very common in CRs (CAIs ~0.4-1 vol.%; Aléon et  
18 al., 2002; Schrader et al., 2011).  
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24 *CAIs* in CR chondrites are relatively small (<500  $\mu\text{m}$ ), irregularly shaped, some  
25 have igneous textures, and most are fragmented (Aléon et al., 2002; Krot et al., 2017;  
26 Table 6). Based on their mineralogy and petrography, Aléon et al. (2002) divided the  
27 CAIs in CR chondrites into grossite +/- hibonite-rich, melilite-rich, pyroxene-anorthite-  
28 rich, and spinel-pyroxene-melilite aggregates. They measured the oxygen isotopic  
29 compositions of minerals in 27 CAIs from seven CRs and determined that they are  $^{16}\text{O}$   
30 rich. The  $\Delta^{17}\text{O}$  for hibonite, melilite, spinel, pyroxene, and anorthite is < -13 ‰. It is  
31 noteworthy that most CAIs appear to have escaped both aqueous alteration and thermal  
32 metamorphism (Aléon et al., 2002).  
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42 *Amoeboid olivine aggregates* are assemblages of anhedral forsterite grains and  
43 refractory inclusions of spinel, pyroxene, and anorthite, with some with Fe-Ni metal  
44 grains (Aléon et al., 2002; Weisberg et al., 2004). The composition of Fe-Ni metal in  
45 AOAs is similar to that in type I chondrules. Texturally, these AOAs have irregular  
46 shapes, can be fluffy, and lack the mesostasis present in chondrules (Weisberg et al.,  
47 2004). Aléon et al. (2002) measured  $-24 \text{‰} < \Delta^{17}\text{O} < -20 \text{‰}$  in forsterite, spinel,  
48 anorthite, and pyroxene in AOAs. According to Weisberg et al. (2004), AOAs record a  
49 complex nebular history, beginning with gas-solid condensation, reactions between  
50 minerals and nebular gas, small degrees of melting, and sintering of the assemblage. They  
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4 also described a textural association between an AOA and a type A CAI, suggestive of  
5 heating during the same nebular event.  
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## 9 10 **6. Fine-grained materials in CR chondrites**

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13 As in other chondrites, the bulk of CR matrix was probably interstellar dust that was  
14 processed in a variety of ways in the solar nebula and, hence, can be considered to be of  
15 local origin (e.g., Nuth et al., 2005). The oxygen isotopic compositions of matrices are  
16 reported in Fig. 7. Matrices are a complex mixture of fine-grained materials whose  
17 origins have been the subject of considerable debate. Matrices are volumetrically  
18 important components of CRs, only second to chondrules in abundance. CR matrices are  
19 the repository of organic and presolar materials. Their grain size is generally  $< 1\mu\text{m}$ .  
20 Fine-grained materials are texturally variable, occurring as fine-grained rims,  
21 interchondrule matrix, or within dark inclusions (Fig. 6; Table 7), “feathered” matrix, and  
22 smooth rims (e.g., Abreu and Brearley, 2010; Harju et al., 2014; Wasson and Rubin,  
23 2014; Abreu, 2016a).  
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33 The mineralogy of CR matrices is highly dependent on the degree to which they were  
34 affected by aqueous alteration. The least altered CR matrices are dominated by  
35 amorphous Fe-Mg silicates, Fe-nanophase sulfides, and minor olivine and pyroxene  
36 (Abreu, 2007; Abreu and Brearley, 2010; Howard et al., 2015a; Le Guillou et al., 2015a-  
37 b; Abreu, 2016a; Table 8). Examples of matrices from both unheated and weakly heated  
38 CRs are shown in Figs. 8 and 9. Matrices have been extensively hydrated, even in the  
39 most pristine CR that were initially classified as type 3 based on the absence of  
40 phyllosilicates (Alexander et al., 2013; Bonal et al., 2013; Beck et al., 2014; Garenne et  
41 al., 2014; Le Guillou et al., 2015). Hydration only resulted in extensive phyllosilicate  
42 formation in the most altered CRs, in which pre-accretionary phases have been replaced  
43 to varying degrees by Fe-Mg phyllosilicates, magnetite, partly oxidized Fe-sulfides, and  
44 tochilinite (Abreu, 2007; Abreu and Brearley, 2010; Harju et al., 2014; Le Guillou et al.,  
45 2015; Abreu, 2016a).  
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### 59 *6.1. Mineralogy and petrology of fine-grained materials*

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6 One of the key characteristics of matrices in weakly and moderately altered CR  
7 chondrites is that they contain variable amounts of Fe- and Mg-rich, Si-bearing  
8 amorphous material, referred to hereafter as amorphous Fe-Mg silicates (Figs. 8, 9).  
9 According to Abreu (2016a), the composition of these silicates is highly variable,  
10 containing 30-53 FeO, 8-20 MgO, 33-58 SiO<sub>2</sub>, and minor Al<sub>2</sub>O<sub>3</sub> (0.2-3.9) and S (0.1-1.6),  
11 all in wt.%. Amorphous Fe-Mg silicates are not separate units with clearly defined  
12 boundaries, such as the ones observed in glass with embedded metal and sulfides  
13 (GEMS) in interplanetary dust particles (e.g., Bradley and Dai, 2004; Keller and  
14 Messenger, 2011). Instead, they are diffuse regions of silicate material with no  
15 crystallographic structure (Fig. 8a and, e.g., Abreu and Brearley, 2010; Le Guillou and  
16 Brearley, 2014; Le Guillou et al. 2014, 2015a; Abreu, 2016a). Using coordinated  
17 scanning transmission X-ray microscopy (STXM) and TEM, Le Guillou et al. (2015a)  
18 determined that amorphous Fe–Mg silicates are hydrated and have high Fe<sup>3+</sup>/ΣFe ratios  
19 (68-78%) in weakly to moderately altered CRs. They reported that the major element  
20 composition is intermediate between serpentine and saponite in weakly to moderately  
21 altered CRs, whereas altered and extensively altered CRs were richer in Mg and Si.  
22 Intriguingly, amorphous Fe–Mg silicates were predicted to be the most reactive phases in  
23 these matrices, yet they are also one of the last ones to be fully replaced. The matrix of  
24 the most altered CR identified to date, GRO 95577, contains some amorphous Fe–Mg  
25 silicates along with a few relict metal, olivine, and pyroxene phenocrysts (Abreu, 2016a;  
26 Howard et al., 2015b).

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44 Fe-Mg phyllosilicates are the second most common type of silicate in CR matrices  
45 (e.g., Abreu, 2007, 2016a; Abreu and Brearley, 2010; Le Guillou et al., 2015a). By far,  
46 the most common phyllosilicates have compositions between serpentine and saponite.  
47 Most phyllosilicates are generally only a few unit cells (e.g., Abreu, 2007, 2016a).  
48 However, Le Guillou et al. (2014a) identified both fine-grained Fe-Mg phyllosilicates  
49 and coarser-grained (~100 nm) Fe-rich cronstedtite in the matrix of Renazzo. Very few  
50 areas contain only amorphous Fe-Mg silicates or only phyllosilicates (e.g., Abreu,  
51 2016a). Although there is significant compositional overlap between amorphous Fe-Mg  
52 silicates and phyllosilicates, the latter are richer in MgO and SiO<sub>2</sub> and contain FeO (8-  
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4 41), MgO (6-31) and SiO<sub>2</sub> (40-58, all in wt.%), and, like the amorphous material, contain  
5 small amounts of Al<sub>2</sub>O<sub>3</sub> (0.1-4.6 wt.%). The texture and composition (oxides and water  
6 contents) of these materials form a continuum between amorphous Fe-Mg silicates and  
7 Fe-Mg phyllosilicates, which suggests that phyllosilicates formed by replacement of  
8 amorphous silicates (e.g., Abreu, 2007, 2016a; Abreu and Brearley, 2010; Le Guillou et  
9 al., 2015a).

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15 There are a variety of opaque phases distributed within the amorphous Fe-Mg  
16 silicates and phyllosilicates (Fig. 8), with nanophase pyrrhotite and pentlandite being the  
17 most common (Lee et al., 1992; Abreu and Brearley, 2010; Abreu, 2016a). Fe sulfides  
18 with a range of Ni contents are generally thought to have formed in the solar nebula (e.g.,  
19 Schrader et al., 2008). Sub-micron and nanophase sulfides have been partially oxidized in  
20 moderately altered CRs (Abreu, 2016a). Authors have identified nanophase tochilinite in  
21 some of the most pristine CR matrices (e.g., Le Guillou and Brearley, 2014; Abreu,  
22 2016a). The latter observed spatial relationships among pentlandite, tochilinite, and  
23 serpentine, including layers of tochilinite on sulfide surfaces. Based on their occurrence  
24 as overgrowths of magnetite and association with siderite grains, Tyra (2013) argued that  
25 at least some of the coarse-grained Fe-sulfides in GRO 95577 are asteroidal.

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35 The second most common group of opaques in CR matrices are Fe-oxides. The  
36 oxygen isotopic composition of these oxides ranges from  $\delta^{17}\text{O} = -9.7\text{-}3.8\text{‰}$ , in  $\delta^{18}\text{O} = -$   
37  $18\text{-}5\text{‰}$  (Jilly-Rehak et al., 2018). In weakly to moderately altered CRs, Fe-oxides are  
38 generally fine-grained (<50  $\mu\text{m}$ ) and scarce (Abreu and Brearley, 2010). However, in  
39 some weakly altered CRs, rounded Fe-oxides are found decorating the periphery of fine-  
40 grained rims around chondrules. Some Fe-oxides occur as independent rounded or  
41 platelet grains disseminated throughout the matrix, whereas others form framboidal  
42 aggregates (e.g., Kallemeyn et al., 1994; Abreu and Brearley, 2010; Tyra 2013; Abreu,  
43 2016a). The abundance of Fe-oxides increases with aqueous alteration, as surmised from  
44 their high abundance in altered GRO 95577, where Fe-oxides are comparatively large  
45 (100  $\mu\text{m}$  - 1 mm), and texturally and mineralogically complex (e.g., Weisberg and Huber,  
46 2007; Tyra, 2013). Morlok and Libourel (2013) explained the presence of Fe-oxides in  
47 GRO 95577 by alteration of Fe-Ni metal nodules instead of primary matrix phases.  
48 Kallemeyn et al. (1994) argued that Fe-oxides formed during hydrothermal alteration at  
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4 temperatures between 350-540 K. They proposed that, as precursor phases were replaced  
5 during aqueous alteration, Fe was released and entered magnetite and phyllosilicates.  
6 According to their model, the preferential formation of either phyllosilicates or magnetite  
7 depended on the relative thermodynamic stabilities of these phases.  
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11 Finally, although most CR matrices some contain carbonates, reports of significant  
12 presence of these minerals are limited to the most altered meteorites (e.g., Weisberg et al.  
13 1993; Endress et al., 1994; Abreu and Brearley, 2010; Brunner and Brearley, 2011;  
14 Abreu, 2016a). However, Alexander et al. (2015) found no systematic differences in the  
15 abundances of C in carbonates, which ranged from 0.04 wt.% in heated GRA 06100 to  
16 0.19 wt.% in matrix-rich, aqueously altered Al Rais (Alexander et al. 2015). Both weakly  
17 altered GRA 95229 and extensively altered GRO 95577 contained 0.16 wt.% C in  
18 carbonate.  
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26 The composition of matrix carbonates is variable. Carbonates in weakly to  
27 moderately altered CR chondrites are predominantly Ca-rich, although Mg-bearing  
28 siderite was found in both Acfer 059/El Djouf 001 and in pristine MET 00426 (Weisberg  
29 et al. 1993; Endress et al., 1994; Brunner and Brearley, 2011). Tyra (2013) also reported  
30 the presence of calcite, “siderite,” and minor dolomite in GRO 95577.  
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35 Carbonates are distributed either throughout the matrix or along the periphery of  
36 chondrule pseudomorphs in GRO 95577 (Tyra, 2013). Calcite was also reported in a  
37 GRO 95577 chondrule interior as a replacement product for mesostasis (Jilly-Rehak et  
38 al., 2018). Matrix carbonates are texturally associated with one another, with  
39 phyllosilicates, Fe-oxides, and Fe-sulfides.  
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44 Several groups have measured the isotopic compositions of carbonates in CR  
45 matrices (Weisberg et al. 1993; Alexander et al., 2015; Jilly-Rehak et al., 2018).  
46 Alexander et al. (2015) established two populations in terms of  $^{13}\text{C}$  contents: weakly  
47 altered CRs had  $\delta^{13}\text{C} \sim 30\text{-}40\text{‰}$  and extensively altered CRs had  $\delta^{13}\text{C} \sim 65\text{-}80\text{‰}$ . Jilly-  
48 Rehak et al. (2018) found that the  $\delta^{18}\text{O}$  of calcite ranges from 9 to 35‰, and the  $\delta^{17}\text{O}$  of  
49 calcite ranges from  $\sim 11$  to 14‰. These compositions plot on a line with a slope of  $\sim$   
50 0.64. They found no clear relationship with the degree of aqueous alteration recorded by  
51 each meteorite. Extensively altered GRO 95577 experienced little to no fluid evolution  
52 during carbonate precipitation, which Jilly-Rehak et al. (2018) attributed to the high  
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4 water-to-rock ratio of the sample. They used the isotopic fractionation in carbonate-  
5 magnetite assemblages in Al Rais to estimate that their precipitation temperature was  
6 approximately 60°C and that the global aqueous alteration temperature of the CR parent  
7 body was between 55-88°C. Also, they identified dolomite in dark inclusions, with  $\delta^{18}\text{O}$   
8 ranging from ~ 23 to 27‰.  
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13 Jilly-Rehak et al. (2017) measured the  $^{53}\text{Mn}$ - $^{53}\text{Cr}$  chronometer of carbonates for three  
14 different CR-chondrite lithologies in moderately altered Renazzo and extensively altered  
15 GRO 95577. Renazzo carbonates formed approximately 4.3–5.3 Myr after CAIs in CV  
16 chondrites, which overlaps with the time for aqueous alteration of Tagish Lake, CI and  
17 CM chondrites. On the other hand, calcite grains in GRO 95577 are younger, having  
18 formed approximately 12.6 Myr (Jilly-Rehak et al., 2017) and  $\geq 24$  Myr (Tyra, 2013) after  
19 CAI formation.  
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## 30 *6.2. CR interchondrule matrix*

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34 Abreu (2007, 2016a) observed that the majority of fine-grained materials in CR  
35 chondrites show no clear spatial association with chondrules (i.e., it does not form rims),  
36 but simply occupies the space between them. The size of interchondrule matrix regions is  
37 very variable; some areas are occupied by relatively large (500-800  $\mu\text{m}$ ) units, whereas in  
38 other areas, chondrules are barely separated from each other.  
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43 Abreu (2007) distinguished texturally distinct types of interchondrule matrix,  
44 including exclusively fine-grained (crystals  $< 5 \mu\text{m}$ ), matrix clasts, and distinct Fe-rich  
45 regions. The interchondrule matrix contains a mixture of the sub-micron materials  
46 described above (i.e., amorphous Fe-Mg silicates, phyllosilicates, Fe-sulfides, Fe-oxides  
47 and in some cases calcite), mineral fragments (generally forsteritic olivine),  
48 polymineralic chondrule fragments, Fe-Ni metal, and coarser prismatic to rounded Fe-  
49 sulfides (e.g., Zolensky et al., 1993; Abreu and Brearley, 2010; Harju et al., 2014;  
50 Wasson and Rubin, 2009, 2014; Abreu, 2016a).  
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## 59 *6.3. CR fine-grained rims*

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6 Fine-grained rims are significantly less abundant in CRs compared to CM chondrites  
7 (e.g., Abreu, 2007, 2016a; Abreu and Brearley, 2010; Tyra, 2013; Wasson and Rubin,  
8 2009, 2014). In pristine and weakly altered CR chondrites, continuous, well-defined rims  
9 are < 200  $\mu\text{m}$  in thickness and consist of homogeneous, fine-grained materials as  
10 described above, decorated with magnetite and sulfide crystals (Abreu, 2007; Abreu and  
11 Brearley, 2010). In addition to fine-grained rims, Harju et al. (2014) identified smooth  
12 rims around chondrules in pristine LAP 02342. These rims are featureless in texture in  
13 back-scattered electron images. Harju et al. (2014) argued that these smooth rims are  
14 dominated by phyllosilicates formed as a consequence of aqueous alteration. However,  
15 Abreu (2016a) performed direct mineralogical characterization of one of these LAP  
16 02342 rims using TEM and found that amorphous Fe-Mg silicates dominated it,  
17 suggesting that they might have a similar origin as the rest of CR matrix.  
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### 29 *6.3. CR dark inclusions*

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33 Most CR chondrites contain numerous, matrix-like, extensively aqueously altered  
34 dark inclusions (DI). They are distinct, fine-grained clasts that differ from their host  
35 meteorites because they have lower abundances of chondrules and finer-grained matrix  
36 material that makes them appear dark in plane-polarized light. They are generally <3 mm  
37 along their maximum dimensions (e.g., Scott et al., 1988). Dark inclusions are more  
38 abundant in the CR group compared to other chondrites, comprising up to ~ 20 vol.% of  
39 the fine-grained materials in some weakly and moderately altered CRs (e.g., Bischoff et  
40 al., 1993a,b; Endreß et al., 1994; Krot et al., 2002). Dark inclusions are unusually large in  
41 CR1 GRO 95577 (Weisberg and Huber, 2007; Tyra, 2013).  
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50 Despite their textural differences and sharp boundaries with their hosts, the  
51 mineralogy of DIs is similar to that of chondritic matrix, resembling particularly that in  
52 CI chondrites. Dark inclusions are made of fine-grained silicates, Fe-oxides and sulfides,  
53 roughly in order of decreasing abundance. These clasts contain variable amounts of  $\mu\text{m}$ -  
54 sized opaques, which generally occur as framboidal masses (up to 50  $\mu\text{m}$  in length) or  
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4 rounded (<20  $\mu\text{m}$ ) Fe-oxides, probably magnetite. Other textures are large, elongated,  
5 feathery, and filamentous. Opaques in some DIs are oriented.  
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## 9 10 **7. Presolar grains in CR chondrites**

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13 A small portion of presolar grains survived secondary processing in nebular and  
14 asteroidal environments and still retained the isotopic signatures of their parent stars. As  
15 such, they are identified by their highly anomalous isotopic compositions compared to  
16 solar system materials. Numerous presolar phases have been identified in chondritic  
17 matrices, including nanodiamonds, SiC and other carbides, graphite,  $\text{Si}_3\text{N}_4$ , oxides, and  
18 silicates (e.g., Zinner, 2014). These nanometer- to micrometer-sized grains originated  
19 from red giant and asymptotic giant branch (AGB) stars, supernovae (SNe), and novae.  
20 Aside from nanodiamonds, silicate grains are the most abundant presolar phase (Zinner,  
21 2014). They are also the most mineralogically and chemically diverse (Nguyen et al.,  
22 2016; Floss and Haenecour, 2016a).  
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32 Presolar silicates, oxides, and SiC grains have been identified in sixteen CR  
33 chondrites of all degrees of aqueous alteration. Their abundances are shown in Fig. 10.  
34 We note that reported presolar silicate and oxide abundances obtained by raster ion  
35 imaging of thin sections or dense grain separates are lower limits due to beam overlap  
36 onto adjacent, isotopically normal grains (Nguyen et al., 2003, 2007).  
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### 42 *7.1. Abundance of presolar SiC*

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46 Although most studies suggest that the abundance of silicon carbides is higher in  
47 CR chondrites than in most other chondritic groups, there are some disagreements,  
48 depending on the analytical techniques used. Noble gas analyses indicate that Renazzo  
49 has a very low abundance of presolar SiC (1.86 ppm). Huss et al. (2003) interpreted this  
50 low abundance as having resulted from moderate temperatures during nebular processing  
51 of CR precursors, which destroyed SiC grains. Conversely, in situ isotopic studies of  
52 Renazzo matrix and extracted insoluble organic matter (IOM) revealed much higher  
53 presolar SiC abundances of 35 – 55 ppm (Davidson et al., 2014a; Floss and Stadermann,  
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4 2005; Leitner et al., 2012b). In fact, with few exceptions, the presolar SiC abundances in  
5 CR chondrites are ~30-50 ppm, substantiating the view that CRs are among the most  
6 primitive chondrites (Fig. 10). Pristine MET 00426, and moderately altered NWA 852  
7 and GRV 021710 have unusually high presolar SiC abundances of 90 ppm, 160 ppm, and  
8 182 ppm, respectively (Floss and Stadermann, 2009b; Leitner et al., 2012a; Zhao et al.,  
9 2013). These high abundances could be a consequence of presolar grain clustering in the  
10 matrix (e.g., Floss and Stadermann, 2009a), or of the heterogeneous distribution of  
11 presolar grains in the CR parent body forming region. Davidson et al. (2014a, 2015)  
12 determined the presolar SiC abundances in the insoluble organic matter (IOM) extracted  
13 from primitive meteorites of various groups and degrees of alteration. They found that the  
14 abundances among seven CR chondrites of petrographic types 1-3 range from 24-47  
15 ppm, with an average of 32 ppm. Primitive chondrites of other groups also have  
16 abundances ~30 ppm. Davidson et al. (2014a) concluded that the relatively uniform  
17 concentration in CRs of different petrographic types indicates any aqueous and thermal  
18 processing experienced in the nebula or in the CR parent body was not extensive enough  
19 to destroy SiC grains, and that the distribution of presolar grains in the chondritic forming  
20 region was homogenous.  
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### 37 *7.2. Abundance of presolar silicates and oxides as indicators of asteroidal processing*

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41 The abundances of presolar oxide grains vary widely among different chondritic groups.  
42 In CR chondrites, they are <10 ppm, except NWA 852, which has an abundance of 38  
43 ppm (Leitner et al., 2012a). It is unclear whether this variation is due to asteroidal  
44 processes, the differential degree to which these processes might have affected different  
45 chondritic components, or to the heterogeneous distribution of presolar grains in the solar  
46 nebula. A recent study of a carbon-rich clast (CRC) in LAP 02342 thought to originate in  
47 a comet (Nittler et al., 2019). This CRC contains an abundance of SiCs that is similar to  
48 CR values, but more numerous O-bearing presolar grains. A higher abundance of these  
49 presolar materials suggests that the distribution of presolar grains indeed heterogeneous.  
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57 CR chondrites have a wide range of concentrations of presolar silicates, reflecting  
58 the degree of alteration experienced in the solar nebula and on their parent body (Fig. 10).  
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4 The primitive QUE 99177 has among the highest abundance of presolar silicates (150 –  
5 220 ppm) observed in meteorites (Floss and Stadermann, 2009a; Nguyen et al., 2010).  
6 This abundance is surpassed only by that of the chondrites DOM 08006 (CO3), MET  
7 00526 (L/LL3.05), and RBT 04133 (CV3), primitive interplanetary dust particles (IDPs),  
8 and returned samples from comet Wild 2 (Busemann et al., 2009; Davidson et al., 2014b;  
9 Floss and Haenecour, 2016b; Floss et al., 2006, 2013; Nittler et al., 2018). Note that the  
10 abundance determined for RBT 04133 is based on only 3 grains and is therefore uncertain  
11 (Davidson et al., 2014b), and the calculated abundance for Wild 2 is based on an  
12 estimated presolar grain loss rate from sample collection (Floss et al., 2013). Some  
13 analyzed regions of QUE 99177 display clusters of presolar grains and have abundances  
14 that are analogous to primitive IDPs (375 ppm) (Floss and Stadermann, 2009a). These  
15 matrix regions either accreted a higher concentration of presolar grains in the solar nebula  
16 or were not altered as much as presolar grain-poor areas. The ratios of presolar silicates to  
17 presolar oxides, which have been used to infer the degree of parent body processing, are  
18 also very high in pristine QUE 99177 and MET 00426, and in CR2 GRV 021710 (Floss  
19 and Stadermann, 2009a; Zhao et al., 2013). These ratios are similar to those observed in  
20 primitive IDPs and to the theoretical value from AGB stars (Leitner et al., 2012a), clearly  
21 attesting to the primitive nature of these CRs. These samples could, therefore, be a  
22 measure of the initial abundances of presolar silicate and oxide grains in the region of the  
23 solar nebula where CR parent bodies formed. Conversely, the CR2 NWA 852 has a low  
24 presolar silicate/oxide ratio and lower presolar silicate abundance, indicating preferential  
25 destruction of presolar silicate grains by aqueous and thermal alteration (Leitner et al.,  
26 2012a). Renazzo and GRA 06100 have experienced extensive asteroidal processing and  
27 consequently have the lowest abundances of presolar silicate grains among CR chondrites  
28 (Leitner et al., 2012b, 2015).

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50 Recent studies suggest that the petrologic setting of presolar grains influences  
51 their abundances. Leitner et al. (2016b) conducted a survey of presolar grain abundances  
52 in the matrix and fine-grained rims of chondrules of moderately altered NWA 801 and  
53 Renazzo, weakly altered EET 92161, GRA 95229, MIL 07525, and pristine MET 00426.  
54 The presolar SiC abundances are similar for all the CRs studied. Except for EET 92161  
55 and MET 00426, the presolar silicate and oxide abundances in the rims and the matrices  
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4 are the same within  $2\sigma$  error. However, the combined data for CR2 chondrites indicate  
5 that the abundances of presolar O-rich phases (i.e., silicates and oxides) are  $\sim 3$  times  
6 greater in the chondrule rims than in the interchondrule matrix. In contrast, a higher  
7 concentration of presolar O-rich grains was found in the matrix of MET 00426. Overall,  
8 the presolar O-rich grain abundances in the fine-grained chondrule rims are the same for  
9 all of the samples analyzed. Abundances in the matrix are similar, except for MET  
10 00426, which has a significantly higher abundance of presolar silicates and oxides than  
11 the CR2 chondrites. Leitner et al. (2016b) suggested that the fine-grained rims accreted  
12 onto chondrules in the solar nebula before parent body formation and were less affected  
13 by aqueous alteration than the interchondrule matrix material. However, it is noteworthy  
14 that petrologic observations do not show systematic differences in the degree of aqueous  
15 alteration between interchondrule matrix and rims (e.g., Abreu, 2016a). The much lower  
16 abundance of presolar O-rich phases in the fine-grained rims compared to the  
17 interchondrule matrix in the pristine CR chondrite must also be accounted for. Haenecour  
18 et al. (2018) studied three CO3 chondrites and found that the presolar silicates are more  
19 abundant in the matrix than in chondrule rims, whereas the abundances of the presolar  
20 SiC are the same. These authors concluded that matrix and rim materials originated from  
21 the same nebular reservoir, but pre-accretionary aqueous alteration of the fine-grained  
22 rim material destroyed a fraction of the presolar silicates. The possibility that this  
23 scenario could be extended to CR chondrites is a subject of current debate.

### 42 7.3. *Chemical compositions of presolar silicate grains*

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46 The chemical properties and mineralogy of circumstellar dust grains have  
47 classically been inferred by astronomical infrared spectroscopy. Observations of  
48 circumstellar O-rich dust indicate an abundance of amorphous silicate grains of olivine-  
49 like compositions (Tielens et al., 1998). These grains are Fe-bearing with estimated Fe  
50 contents of  $\sim 50\%$  (Kemper et al., 2001). In contrast, the crystalline silicates are extremely  
51 Mg-rich, with pyroxene being more abundant than olivine (Molster et al., 2002). The  
52 mineralogical and chemical analyses of presolar silicate grains in the laboratory can bring  
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4 ground truth to these remote observations, as well as trace the effects of alteration in the  
5 solar nebula and on the parent body.  
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8 Due to the small size of presolar silicate grains (typically < 300 nm), their  
9 chemical compositions are best characterized by Auger spectroscopy (Floss and  
10 Stadermann, 2009a; Leitner et al., 2016b; Zhao et al., 2013) and energy-dispersive X-ray  
11 spectroscopy (EDX) in the TEM. Presolar silicates from several CR chondrites have been  
12 analyzed by these techniques: QUE 99177 (Floss and Stadermann, 2009a; Nguyen et al.,  
13 2016, 2017a), MET 00426 (Floss and Stadermann, 2009a; Nguyen et al., 2017b; Stroud  
14 et al., 2009), GRV 021710 (Zhao et al., 2013), GRA 95229 and EET 92161 (Leitner et  
15 al., 2016b). Most of the grains have stoichiometries consistent with olivine or pyroxene,  
16 but about a third have non-stoichiometric elemental compositions in between these  
17 minerals. In addition, some grains are Si-poor or Si-rich, and two grains are tentatively  
18 identified as SiO<sub>2</sub> (Floss and Stadermann, 2009a; Zhao et al., 2013).  
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28 The majority of presolar silicate grains in CRs are ferromagnesian with a small  
29 portion containing Ca and/or Al, similar to findings in other primitive chondrites. The Fe-  
30 contents are high and vary widely, with up to 31 at.%. The pyroxene grains and those  
31 with intermediate compositions are more Fe-rich than olivine grains (Floss and  
32 Stadermann, 2009a; Zhao et al., 2013). Presolar silicates from the pristine QUE 99177  
33 and MET 00426 and weakly altered GRA 95229, EET 92161, and GRV 021710 share  
34 similar ranges of elemental compositions. However, presolar silicates from GRV 021710  
35 have lower Fe-contents than those from QUE 99177 and MET 00426. This, coupled with  
36 the high presolar silicate abundance and matrix petrology, led Zhao et al. (2013) to  
37 suggest that GRV 021710 is probably as pristine as QUE 99177 and MET 00426.  
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46 The high concentration of Fe in presolar silicates was somewhat unexpected,  
47 given the primitive nature of the CR chondrites studied. However, isotopically normal  
48 matrix silicates in pristine CR chondrites are also Fe-rich compared to silicates in other  
49 chondritic groups, including those that have undergone much more significant aqueous  
50 alteration such as the CM chondrites (e.g., Abreu, 2007, 2016a; Abreu and Brearley,  
51 2010). Grain condensation models predict Fe-rich silicates formed under non-equilibrium  
52 conditions and Mg-rich silicates under equilibrium conditions (Ferrarotti and Gail, 2001).  
53 Thus, Fe could have been incorporated into the presolar silicates during condensation  
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4 under non-equilibrium conditions. The silicate matrix in thermally altered chondrites is  
5 Fe-rich, and this Fe could have been assimilated into presolar silicate grains during parent  
6 body thermal annealing (e.g., Brearley, 1991; Floss and Stadermann, 2012). The fact that  
7 the matrix grains in the CR chondrites were not extensively annealed suggests thermal  
8 metamorphism was not responsible for their Fe-contents. However, presolar silicates  
9 from the CR chondrites have a wider range of elemental compositions and higher Fe-  
10 contents than presolar silicates from IDPs (Busemann et al., 2009; Keller and Messenger,  
11 2011; Messenger et al., 2005). These IDPs are more primitive than CR chondrites, and  
12 their chemical compositions of the presolar silicates better represent the original presolar  
13 compositions. It is thus likely that for many of the CR presolar silicates, most of the Fe  
14 was acquired during primary condensation, and secondary alteration enhanced the Fe-  
15 content. Of all the presolar phases identified, presolar silicates have the most varied  
16 elemental compositions, a consequence of both primary condensation and subsequent  
17 alteration.

#### 31 *7.4. Mineralogy of presolar silicate and oxide grains*

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35 Nine presolar silicates and one large presolar oxide from CR chondrites have been  
36 studied. All originated in AGB stars except for one supernova silicate. The mineralogy of  
37 presolar SiC grains in CRs have not been characterized.

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41 The mineralogical characterization of presolar silicate grains attests to their highly  
42 diverse nature. Nguyen et al. (2016) studied five presolar silicate grains with sizes < 500  
43 nm from QUE 99177, with one being a euhedral olivine crystal with composition  $Fe_{0.80}$ .  
44 The Fe in this grain could have been acquired from parent body alteration, but presolar  
45 Fe-bearing olivine crystals have also been identified in anhydrous IDPs that have largely  
46 escaped parent body processes (Busemann et al., 2009; Messenger et al., 2005).

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4 silicates formed first at ~1200 K under equilibrium conditions, and the amorphous  
5  
6 silicate mantle formed around these grains at a greater radial distance and lower  
7  
8 temperature of ~900 K. Their Fe was likely incorporated during condensation, as  
9  
10 extensive secondary processing would likely have altered the small crystalline inclusions.  
11  
12 Two non-stoichiometric, very Fe-rich amorphous silicates, one of which is a supernova  
13  
14 grain, were also identified (Nguyen et al., 2017a). The Fe is uniformly distributed  
15  
16 throughout the grains and is likely a primary condensation feature, as QUE 99177 did not  
17  
18 undergo thermal metamorphism. These two grains formed under similar non-equilibrium  
19  
20 conditions at low temperatures or in an Fe-rich gas. Nguyen et al. (2017b) also analyzed  
21  
22 by TEM three presolar silicates from MET 00426 – one is a crystalline Mg-rich olivine  
23  
24 and is unusually large for a presolar silicate (2000×750 nm). Strain contrast and  
25  
26 dislocations suggest parent body compaction and lithification. A grain comprised of  
27  
28 amorphous silicate and nanocrystalline olivine has varying elemental composition on a  
29  
30 10 nm scale, and a presolar silicate of similar structure has a more homogeneous  
31  
32 elemental composition (Stroud et al., 2009).

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34 Evidence for radiation damage in space is rare. One presolar silicate from QUE  
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36 99177 was found to be an amorphous silicate of stoichiometric enstatite composition with  
37  
38 a crystalline enstatite core (Nguyen et al., 2016). The fact that the entire grain has the  
39  
40 composition of stoichiometric enstatite suggests it originally formed as a single enstatite  
41  
42 crystal but later was partially amorphized by high-energy radiation in space. A portion of  
43  
44 the grain is rimmed with Fe, which Nguyen et al. (2016) argued was likely acquired from  
45  
46 surrounding grains during parent body alteration.

47  
48 In addition to the silicates, Leitner et al. (2012a) studied a large (1750×300 nm<sup>2</sup>)  
49  
50 presolar Al-rich hibonite crystal (CaAl<sub>12</sub>O<sub>19</sub>) from NWA 852. The grain has micro-cracks  
51  
52 caused by stress from grain-grain collisions. The hibonite contains a central <100 nm  
53  
54 perovskite-like sub-grain with a Ca/Ti ratio ~1, which formed at high temperature in a Ti-  
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56 enriched gas before being enclosed in the host crystal.

## 57 **8. Carbonaceous materials in CR chondrites**

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4 The carbonaceous material in CR chondrites, with a bulk carbon content between  
5 0.3-2.5 wt.% (Bischoff et al., 1993a; Cody et al., 2008), has been of particular interest, as  
6 it may shed light on the chemical activity of organics from their formation in interstellar  
7 environments to their incorporation into asteroidal bodies.  
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### 10 11 12 13 *8.1. Soluble organic materials in CR chondrites* 14 15

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17 The molecular diversity of soluble organic compounds in CRs has been studied  
18 using high-field nuclear magnetic resonance and ultrahigh-resolution mass spectrometry,  
19 resulting in an estimated millions of unique chemical structures of organic compounds  
20 that remain free and that are readily extractable using water and other polar and non-polar  
21 solvents. Given the impossibility of identifying each individual molecular structure,  
22 chondritic soluble organic compounds are traditionally extracted and analyzed as groups  
23 of compounds sharing the same chemical functionalities (Glavin et al., 2011, 2018). The  
24 total numbers of molecular species that have been extracted from CRs to date are listed in  
25 Table 9 (concentrations in ppm or  $\mu\text{g/g}$  of meteorite).  
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#### 34 35 *8.1.1. Amino acids* 36 37 38

39 Aliphatic primary monoamino, monocarboxylic, nonheteroatom (or “simple”) amino acids are some of the building blocks for proteins and enzymes in all terrestrial life. Based on their astrobiological importance, these compounds have been extensively studied in carbonaceous chondrites (Burton et al., 2012a-b; Cobb and Pudritz, 2014; Elsila et al., 2016).  
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48 Concentrations of amino acids in some pristine and weakly altered Antarctic CRs are the highest ever recorded from any group of carbonaceous chondrite (Glavin et al., 2011). The total amino acid concentrations in CR chondrites range from a few to over 300 ppm (Table 9). Two other independent laboratories confirmed the high abundance of amino acids in CRs (Pizzarello et al., 2008, 2012; Pizzarello and Holmes, 2009; Glavin et al., 2011). Although the total concentrations of amino acids in CR2 Renazzo are similarly low as those in the CIs Orgueil and Ivuna (Botta et al., 2002), amino acids in weakly  
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4 altered CRs EET 92042 and GRA 95229 are at least 10 times higher than those found in  
5  
6 CM2 chondrites such as Murchison and Murray (Martins et al., 2007).  
7

8 Amino acids in CR chondrites appear to be primarily dominated by  $\alpha$ -amino  
9  
10 isomers, though their complete molecular diversity has not been fully investigated. The  
11  
12 high abundance is of the same order of those in moderately aqueously altered CM2  
13  
14 chondrites (Glavin et al., 2011) and differs from what is seen in heavily aqueously altered  
15  
16 and thermally metamorphosed carbonaceous chondrites, which are richer in  $\beta$ - and other  
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18  $n$ - $\omega$ -amino acid isomers (Ehrenfreund et al. 2001; Burton 2012a-b; Elsila et al. 2016).  
19  
20 The high abundance of  $\alpha$ -amino acids in CR chondrites suggests the Strecker-  
21  
22 cyanohydrin synthesis as one of the most active mechanisms generating amino acids in  
23  
24 the CR parent body (Peltzer et al., 1984; Lerner et al., 1993; Matthews and Minard,  
25  
26 2008).  
27

28 One significant similarity between amino acids in CR and those in CI and CM  
29  
30 chondrites is the L-enantiomeric excess of isovaline in CR1 GRO 95577 ( $ee_{L\text{-isovaline}} =$   
31  
32  $11.0 \pm 7.2\%$ ; Glavin et al., 2011). Isovaline is an  $\alpha$ -methyl- $\alpha$ -amino acid, and thus it is not  
33  
34 susceptible to the loss of enantiomeric enrichment or racemization through aqueous or  
35  
36 radiogenic processing on geological timescales (Pollock et al., 1975; Bonner et al., 1979).  
37  
38 In addition, isovaline is not very common in the terrestrial biosphere and has shown large  
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40 heavy isotope enrichments (D/H,  $^{13}\text{C}/^{12}\text{C}$ ,  $^{15}\text{N}/^{14}\text{N}$ ) in CR2 and CM2 chondrites (Elsila et  
41  
42 al., 2012). Isovaline can be a product of the Strecker-cyanohydrin synthesis through  
43  
44 aqueous alteration inside a parent body, using an achiral ketones (2-butanone) as starting  
45  
46 material. Although the exact mechanisms for the break in symmetry and amplification of  
47  
48 a small enantiomeric excess are subject of ongoing debate (Takano, 2007; De Marcellus  
49  
50 et al., 2011; Meinert et al., 2014; Modica et al., 2014), a correlation between the L-  
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52 enantiomeric excesses and degree of aqueous alteration experienced in their parent bodies  
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54 has been observed for CR, CM, and CI chondrites (Glavin and Dworkin, 2009; Glavin et  
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56 al., 2011).  
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### 8.1.2. Amines

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4 Acyclic aliphatic monoamines (“amines”) are compounds that are structurally  
5 related to amino acids and, therefore, their study provides insights about the synthesis of  
6 chondritic amino acids. Pizzarello et al. (2008) were the first to report the presence of  
7 amines in CRs. They found that, similar to amino acids, amines were more abundant in  
8 CRs than in CM chondrites (Pizzarello and Holmes, 2009; Pizzarello et al., 2012).  
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There are both similarities and differences between amines in CR and CM chondrites. For example, there is a trend in both CM2s and CR2s for amines to have the amino group ( $-NH_2$ ) on a secondary carbon, compared with having the  $-NH_2$  in a primary or tertiary carbon. Contrary to amines in CM chondrites, the abundance of some of the homologous series of the CR2 amines does not decline with increasing length of chain. This observation was confirmed by Aponte et al. (2016), who also found that isopropylamine is the most abundant amine species in CR2 chondrites, while methylamine is the most abundant in CM2 and CM1/2 chondrites. The ratios of amino acids to amines concentration are different for CR and CM chondrites. In CRs, the concentration of amino acids is higher than that of amines. In CM chondrites, however, amines are more abundant than amino acids. Aponte et al. (2016, 2018) interpreted these differences as either original signatures of distinct parent bodies, or as evidence that aqueous alteration, which was more active in the CM parent body, was more destructive to amino acids than to amines.

Like amino acids, some amines can be chiral and, thus, might potentially exhibit enantiomeric excesses, such as those reported for isovaline. The structural analog of isovaline is *sec*-butylamine, and a large range of enantiomeric compositions has been reported for this  $\alpha$ -methyl-amine. Pizzarello and Yarnes (2016) reported enantiomeric distributions of (*S*)-*sec*-butylamine ranging from 0% to 66% in five Antarctic CRs. In contrast, Aponte et al. (2016) reported only racemic compositions for *sec*-butylamine in three Antarctic CR chondrites, which include GRA 95229 and LAP 02342 previously investigated by Pizzarello and Yarnes (2016). Aponte et al. (2016) explained the differences in enantiomeric compositions reported for *sec*-butylamine extracted from the same meteorites as the result of sample heterogeneities. However, a thorough examination of the methodology used for extraction and derivatization of amines might not rule out the potential for sample contamination and deracemization. It remains likely

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4 that the different enantiomeric compositions might originate from the different  
5 methodologies used. Regardless, studies of the anisotropy of aliphatic amines and amino  
6 acids suggest that ultra-violet circularly polarized light (UVCPL) might induce  
7 enantiomeric excesses in aliphatic amino acids, but not in aliphatic amines (Meinert et al.,  
8 2012, 2016; Myrgorodska et al., 2017).  
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### 15 8.1.3. Monocarboxylic acids

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19 Until recently, the abundances of monocarboxylic acids (MCAs) in CR chondrites  
20 were believed to be lower than those in Murchison. Antarctic CR2s GRA 95229  
21 (Pizzarello et al., 2008, 2012) and EET 87770 (Aponte et al., 2011) were the first CR2  
22 chondrite analyzed for acyclic aliphatic monocarboxylic acids. Both meteorites have  
23 substantially lower contents of MCAs than CM chondrites. However, a new study  
24 reevaluated the abundance of MCAs in GRA 95229 and quantified the abundances in  
25 LAP 02342, MIL 090657, and MIL 090001 (Aponte et al., 2019). These measurements  
26 suggested that the total abundances of MCAs, just like those of amino acids and amines,  
27 are higher in CRs than in all other investigated chondrites. The disagreement in  
28 measurements was attributed to instrumental causes. MCAs measurements require  
29 preparatory and analytical methods (i.e., multi-stepped extraction, purification, and  
30 derivatization methods), as well as the use of different analytical instrumentation in  
31 different laboratories, which might introduce significant bias and contribute to the  
32 difficulty in cross-comparison of meteoritic organics (Glavin et al., 2011). Indeed, MCAs  
33 in CRs are the highest among all chondritic soluble organics and approach those for  
34 chondritic carbonates and insoluble organic matter (IOM; Aponte et al., 2019).  
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48 Like amino acids, but contrary to aliphatic amines, the abundance of MCAs  
49 decreases with increasing molecular weight – acetic acid being the most abundant in all  
50 CRs. Also, the abundance of MCAs decreases linearly with increasing aqueous alteration  
51 and thermal metamorphism of the host meteorite (Aponte et al., 2019).  
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55 Similar to aliphatic amino acids and amines, MCAs can be chiral, 2-  
56 methylbutanoic acid being the structural analog of isovaline and *sec*-butylamine. The  
57 enantiomeric composition of 2-methylbutanoic acid has been determined for CM and CR  
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4 chondrites using a variety of methods, and in some cases, they are racemic (Pizzarello et  
5 al., 2008, 2012; Aponte et al., 2014, 2019). The racemic composition of chondritic 2-  
6 methylbutanoic acid is consistent with that of *sec*-butylamine, as anisotropy studies of  
7 both molecules have suggested that UVCPL may not be able to induce enantiomeric  
8 excesses in them (Meinert et al., 2016). Thus, no amplification of an original imbalance is  
9 expected to occur inside the asteroid parent body. The lack of enantiomeric excess in 2-  
10 methylbutanoic acid and *sec*-butylamine suggests that these compounds may have been  
11 racemic before their incorporation to the parent body or that the processes that resulted in  
12 the L-enantio-enrichments in its structurally analogous amino acid, isovaline, did not  
13 have a significant effect on them.  
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#### 24 *8.1.4. Other soluble organic compounds*

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28 Analyses of other aliphatic and aromatic soluble organics reported from CR  
29 chondrites include aldehydes and ketones, hydroxy monoaldehydes, polyols, hydroxy  
30 monocarboxylic acids, dicarboxylic acids, hydrocarbons, polycyclic aromatic  
31 hydrocarbons (PAHs), and purines (Table 9; -Pizzarello et al., 2008, 2012; Cooper and  
32 Rios, 2016; Callahan et al., 2011). Compared to amino acids and MCAs concentrations in  
33 CRs, these compounds constitute trace species, and in some cases, their presence has  
34 been questioned. The detection and quantification of purines in CRs is an important case  
35 study. As discussed above, CR chondrites contain the largest amount of amino acids,  
36 amines, and MCAs among carbonaceous chondrites, including the C-rich CM group.  
37 Additionally, ammonia (NH<sub>3</sub>), which is thought to be a key starting material for the  
38 abiotic synthesis of purines (Callahan et al., 2011), has been reported as one of the most  
39 abundant inorganic species present in CRs. Further, only trace amounts of ammonia were  
40 detected in the CM2 Murchison (Pizzarello et al., 2011; Pizzarello and Williams, 2012).  
41 Purines, however, unlike other nitrogen-bearing molecules, were found in low  
42 concentrations in CR chondrites (Callahan et al., 2011).  
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#### 57 *8.2. Insoluble organic materials in CR chondrites*

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4 Based on their exotic isotopic signatures, extracts of insoluble organic material  
5 (IOM) and solid carbonaceous material from CR chondrite matrices may contain a  
6 component of interstellar organics (e.g., Ash and Pillinger, 1995; Messenger et al., 2000;  
7 Cody and Alexander, 2005; Busemann et al., 2006, 2007; Pizzarello et al., 2006; Alexander  
8 et al., 2007).

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13 The complexity of carbonaceous material in carbonaceous chondrites is most likely  
14 the result of multistage processing through a variety of mechanisms in several diverse  
15 environments. Organic compounds may have formed by a combination of some or all the  
16 mechanisms that operated in the interstellar medium, solar nebula, and asteroidal parent  
17 bodies (Gilmour, 2003). A component of the organics in meteorites probably formed in  
18 diffuse interstellar regions and was later injected into the molecular cloud from which the  
19 solar nebula originated (Sanford, 1996; Sanford et al., 2001). This material likely  
20 experienced further synthesis and processing in the solar nebula and planetesimals by  
21 aqueous alteration and thermal processing (e.g., Kerridge, 1999; Gilmore and Pillinger,  
22 1994).

### 23 24 25 26 27 28 29 30 31 32 33 *8.3. Carbonaceous materials and their relationship with CR matrices*

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37 A variety of microbeam techniques has been used to characterize organic matter  
38 *in situ*. TEM observations showed that carbonaceous materials are distributed at the  
39 nanoscale in a variety of textural occurrences (e.g., Abreu, 2007; Floss et al., 2014; Le  
40 Guillou and Brearley, 2014; Le Guillou et al., 2014). Le Guillou and Brearley (2014)  
41 documented the presence of nanoglobules, elongated veins, and networks of sub-rounded  
42 grains connected by cracks. They attributed these textural differences to variations in the  
43 degree of aqueous alteration of organics, which had originally accreted into the CR parent  
44 body as ice-organics particles. Organic materials show a very low degree of  
45 graphitization (e.g., Abreu, 2007; Le Guillou and Brearley, 2014). These organics occur  
46 both as isolated masses and in association with nano-sulfides, tochilinite, carbides,  
47 magnetites, and phyllosilicates (e.g., Abreu, 2007; Le Guillou and Brearley, 2014). Le  
48 Guillou et al. (2014) observed that CR2 Renazzo contains individual grains of  
49 molecularly homogeneous organic materials, but in some cases have heterogeneous  
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4 morphologies and D/H ratios. They speculated that these grains probably contained  
5 insoluble organic compounds. Consistent with TEM observations, Auger microscopy  
6 showed that organic grains had both nanoglobule and irregular morphologies. TEM  
7 analysis revealed a high degree of crystallographic disorder (Floss et al., 2014). They  
8 found no correlations between isotopic composition and morphology, petrographic  
9 association, or elemental composition and suggested that the diffuse regions formed by  
10 redistribution of organics during aqueous alteration. Based on comparative studies with  
11 CI (Orgueil) and CM (Murchison) chondrites, Le Guillou et al. (2014) also argued that  
12 organic materials were altered during aqueous alteration.  
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21 Abreu (2007) suggested that some of the organic materials in CR chondrites must  
22 have formed or been reprocessed after interstellar synthesis, favoring Fischer-Tropsch  
23 type synthesis as a mechanism playing a role in the formation of organics in the solar  
24 nebula based on the mineral-organic spatial relationships. She suggested that the  
25 association between nanophases and carbonaceous materials resulted from Fischer-  
26 Tropsch type reactions. Catalysis would have occurred below ~ 300°C, probably during  
27 nebular cooling after the formation of sulfides and amorphous Fe-Mg silicates during  
28 high-temperature evaporation and recondensation events. This solar carbonaceous  
29 material would later be mixed with an interstellar component, giving rise to the exotic  
30 isotopic signatures characteristic of CRs.  
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39 Based on coordinated NanoSIMS, Auger, and FIB-TEM, organics in CR  
40 chondrites are isotopically anomalous with respect to solar system values (e.g., Floss and  
41 Stadermann, 2009a,b; Floss et al., 2014). Floss and Stadermann (2009a) reported the  
42 presence of both <sup>13</sup>C-enriched and <sup>13</sup>C-depleted organic regions in CRs. Additionally,  
43 Floss et al. (2014) found that organic matter has solar carbon isotopic compositions, but  
44 significant enrichments in <sup>15</sup>N compared to solar (average ~ 195‰).  
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## 54 **9. Processes affecting the CR parent asteroid(s)**

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### 59 *9.1. Aqueous alteration of CR chondrites*

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6 The interest in secondary minerals in CRs stems from the fact that this group  
7 contains some of the least and the most aqueously altered carbonaceous chondrites in our  
8 collections. A summary of the effects of aqueous alteration in different petrologic CR  
9 types is given in Fig. 11. Even the earliest studies of CR chondrites recognized that they  
10 contained hydrous silicates (e.g., Mason and Wiik, 1962). The presence of these silicates  
11 placed most CRs in between petrologic types 2 and 3 but considered them unusual as  
12 they preserved metallic Fe (e.g., Mason 1971).  
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19 Compared with the other hydrated carbonaceous chondrites, CRs are not as water-  
20 rich. However, some CRs have affinities with both the CM and CI chondrites (e.g.,  
21 Bischoff et al., 1993a,b; Zolensky et al., 1993; Kallemeyn et al., 1994; Abreu 2016a).  
22 Alteration of CIs resulted in the formation of phyllosilicates, oxides, carbonates, sulfides,  
23 and sulfates (Kerridge et al., 1979; Tomeoka and Buseck, 1985; Tomeoka et al., 1988;  
24 Zolensky et al., 1993). CR, CI, and some CM chondrites contain framboidal and platelet  
25 magnetite. Alteration of CM matrices resulted in the formation of phyllosilicates,  
26 tochilinite, carbonates, Fe-sulfides, sulfates, oxides, and hydroxides (e.g., Browning et  
27 al., 1996; Chizmadia and Brearley, 2008; de Leuw et al., 2010; Hanowski and Brearley,  
28 2001; Hewins et al., 2014; McSween Jr, 1979a,b; Rubin et al., 2007; Tomeoka and  
29 Buseck, 1985; Zega and Buseck, 2003; Zolensky et al., 1993). Most of these minerals are  
30 also present in the CRs, though in lesser abundances.  
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#### 42 *9.1.1. Effects of aqueous alteration on CR chondrite components*

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46 Differences in the degree of alteration originated from the fact that CR chondrites are  
47 made up of materials that are not uniformly susceptible to aqueous alteration. Alteration  
48 affected chondrule materials to a lesser extent than CR matrices.  
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51 Petrologic observations suggest that chondrule phenocrysts and Fe-Ni metal nodules  
52 in all but the most altered CR largely escaped aqueous alteration. In extensively altered  
53 GRO 95577, most silicate phenocrysts in chondrules were replaced by phyllosilicate  
54 pseudomorphs, ranging in sizes from less than 0.05-2 mm (e.g., Weisberg and Huber,  
55 2007; Tyra, 2013; Harju et al. 2014; Abreu, 2016a). Unaltered Fe-Ni metal, occurring as  
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4 individual crystals and associated with chondrules, has also been preserved (Krot et al.,  
5 2002). However, chondrule mesostasis has been affected by aqueous alteration even in  
6 weakly altered CRs (e.g., Ichikawa and Ikeda, 1995; Weisberg et al., 1995; Burger, 2005;  
7 Keller, 2011). In these CRs, anhydrous mesostasis silicates have been partially to  
8 completely altered to serpentine, saponite, and chlorite (Ichikawa and Ikeda, 1995;  
9 Weisberg et al., 1995). Weisberg et al. (1993) reported the mobilization of Ca from a Ca-  
10 carbonate rim on the curved surface of a chondrule fragment in Y 790112. Burger (2005)  
11 observed extensive replacement, most notably when mesostases were in direct contact  
12 with fine-grained matrix in two type II chondrules in paired EET 87770 and EET 92105.  
13 He found that whereas unaltered mesostasis regions were albite normative, Na and K  
14 were leached from altered mesostasis. In contrast, altered mesostasis and adjacent matrix  
15 had high Ca abundance. Burger (2005) added that Ca and P mobilization occurred early  
16 in the alteration process. This assertion was supported by later FIB/TEM observations,  
17 showing that even in the glassy mesostasis, alteration proceeded by the removal of Ca  
18 and P through the formation of Ca-phosphate veins (Brearley and Burger, 2009).

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Matrices in CRs have been replaced to different extents by Fe-Mg phyllosilicates  
(predominantly serpentine), framboidal magnetite, and minor calcite (e.g., Weisberg et  
al., 1993; Zolensky et al., 1993; Kallemeyn et al., 1994; Abreu and Brearley, 2010; Le  
Guillou et al., 2015a,b; Abreu, 2016a).

#### 9.1.2. *Effects of aqueous alteration on the oxygen isotopes of CR chondrite components*

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The oxygen isotopic composition of CR chondrule and matrices has been  
evaluated to place constraints on their aqueous alteration histories. CR chondrules show  
high  $\delta^{17}\text{O}$  and  $\delta^{18}\text{O}$ , which have been interpreted as evidence for aqueous alteration (Krot  
et al., 2002). However, enrichments are not evenly distributed throughout all chondrule  
components. In contrast, Jilly-Rehak et al. (2018) argued that the variations in oxygen  
isotopic compositions of matrix minerals in CR chondrites resulted from aqueous  
alteration processes affecting the CR parent body in a very heterogeneous manner. They

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4 suggested that their observations are inconsistent with carbonates in CR chondrites  
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6 forming from a single, uniform fluid. They concluded that O-isotopic compositions of the  
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8 altering fluids evolved from approximately the Al Rais water composition of  $\Delta^{17}\text{O} \sim 1\text{‰}$   
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10 and  $\delta^{18}\text{O} \sim 10\text{‰}$ , to become increasingly  $^{16}\text{O}$ -enriched toward a final fluid composition  
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12 of  $\Delta^{17}\text{O} \sim -1.2\text{‰}$  and  $\delta^{18}\text{O} \sim -15\text{‰}$ . These water compositions may not be consistent with  
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14 alteration by extremely heavy water compositions such as those recorded from cosmic  
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16 symplectite found in Acfer 094.  
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### 21 *9.1.3. Petrologic sub-types of aqueously altered CR chondrites*

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24 A large gap in samples exists between CR chondrites that record intermediate and  
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26 those that record very extensive aqueous alteration (e.g., Alexander et al., 2013; Harju et  
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28 al., 2014; Howard et al., 2015a; Abreu, 2016a). With the exception of CR1 GRO 95577,  
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30 CR chondrites fall between to petrologic types 2 and 3 (e.g., Zolensky et al., 1993;  
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32 Kallemeyn et al., 1994; Weisberg and Prinz, 2001; Perronnet and Zolensky, 2006; Abreu  
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34 and Brearley, 2010; Wasson and Rubin, 2014; Le Guillou and Brearley, 2014; Le Guillou  
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36 et al., 2015a,b; Abreu 2016a).

37 Three scales have been devised to quantify the degree of aqueous alteration of CRs,  
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39 with particular emphasis on assigning sub-types to the CR2 chondrites. The first scheme  
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41 is based on the observation that there is a linear relationship between the bulk H, C, and  
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43 N elemental and isotopic compositions of CM and, to a lesser extent, CR chondrites  
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45 (Alexander et al. 2013). The authors used this correlation to assign petrologic subgroups.  
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47 The second scheme is based on petrologic criteria. Harju et al. (2014) postulated to  
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49 classify the CRs based on (1) presence/abundance of chondrules containing altered  
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51 mesostasis, (2) presence/abundance of magnetite, (3) replacement of primary silicate  
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53 phenocrysts by phyllosilicates, (4) replacement of Fe-Ni metal with oxides, (5)  
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55 abundance of phyllosilicates in matrix and chondrule rims and (6) S-content of the  
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57 matrix. Finally, Howard et al. (2015a) used the ratio of phyllosilicates to anhydrous  
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59 silicates (i.e., olivine and pyroxene) to assign petrologic sub-types. Although the  
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61 classification schemes proposed by Alexander et al. (2013) and Howard et al. (2015a,b)  
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4 are generally consistent with each other, these schemes often disagree with petrologic  
5 classifications (Harju et al., 2014) and sub-micron observations (e.g., Abreu 2016a).  
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8 Petrology-based classifications require understanding the sequence of mineral  
9 replacement and how these replacements relate to aqueous alteration. Abreu (2016a)  
10 suggested that differences in classifications might arise from the fact that CRs have very  
11 fine-grained matrices dominated by amorphous Fe-Mg silicates, by their abundant Fe-Ni  
12 metal, and by extensive brecciation. Identifying evidence of incipient mineral  
13 replacement is particularly challenging for CR chondrites because secondary phases are  
14 generally very fine-grained. Furthermore, primary and secondary matrix phases are  
15 mixed at the sub-micron scale (e.g., Abreu and Brearley, 2010). Abreu (2016) suggested  
16 that a multi-technique approach is needed to determine the degree of aqueous alteration  
17 of these meteorites. While the bulk elemental and isotopic compositions of CRs provide  
18 valuable insights on how prevalent the changes observed at the nanometer scale are,  
19 Abreu (2016a) suggested that several characteristics of the CRs limit their applicability as  
20 indicators of aqueous alteration. First, these bulk measurements are affected by terrestrial  
21 weathering and impact processes. Second, CRs contain variable amounts of clasts from  
22 different regions of the CR parent body or from altogether different planetary bodies (i.e.,  
23 Hiyagon et al., 2016; Abreu, 2013; MacPherson et al., 2009); consequently gram-size  
24 samples may contain numerous clasts recording diverse aqueous alteration histories.  
25 Third, intra- and inter-sample variations in CR matrix and dark inclusion abundances are  
26 too large to allow to resolve CR sub-groups based on bulk compositional and isotopic  
27 data alone (Abreu, 2016a). Overcoming these obstacles requires bulk data to be woven  
28 together with sub-micron observations.  
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#### 50 *9.1.4. Constraints on the timeline for CR aqueous alteration*

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54 The duration of aqueous alteration, the effect on individual components, and the  
55 classification of aqueous alteration features have been studied to improve our  
56 understanding of the history of water in the CR parent body. Some have used  $^{53}\text{Mn}$ - $^{53}\text{Cr}$   
57 systematics of carbonates to constrain the duration of aqueous alteration in the CR parent  
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4 body (Tyra, 2013; Jilly-Rehak et al., 2017). Jilly-Rehak et al. (2017) found that whereas  
5 alteration of CR2 Renazzo was contemporaneous with unique carbonaceous chondrite  
6 Tagish Lake, CI, and CM chondrites, they found that CR1 GRO 95577 preserved much  
7 younger carbonates. They argued that the older carbonates found in most CR chondrites  
8 (such as calcite and dolomite measured in Renazzo) likely formed via parent body  
9 aqueous alteration driven by the presence of liquid water heated by  $^{26}\text{Al}$  decay, similar to  
10 carbonates in CM, CI, and CO chondrites. However, the ages that both Jilly-Rehak et al.  
11 (2017) and Tyra (2013) obtained for GRO 95577 carbonates are too young to be easily  
12 explained by decay-driven hydrothermal alteration in a typical chondritic parent body.  
13 Jilly-Rehak et al. (2017) proposed two scenarios. First, that the CR parent body was large  
14 enough (30-50 km radius) for heat from radiogenic decay to be retained even 8 Myr after  
15 accretion. Second, that episodic formation of secondary phases ensued as a consequence  
16 of impact-driven metamorphism.

#### 30 *9.1.5. Stages of CR aqueous alteration*

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33 Authors argued that aqueous alteration of CR chondrites occurred at low temperatures  
34 and involved multiple stages (e.g., Le Guillou et al., 2015a; Jilly-Rehak et al., 2018). Le  
35 Guillou et al. (2015a) suggested that aqueous alteration of CR matrices took place on the  
36 CR parent body and was a two-step process that began with hydration and oxidation of  
37 nebular amorphous  $\text{Fe}^{2+}$ -Mg silicates to produce metastable, predominantly  $\text{Fe}^{3+}$   
38 amorphous Fe-Mg silicates, followed by formation of Fe-Mg phyllosilicates with lower  
39  $\text{Fe}^{3+}/\sum\text{Fe}$  ratios (e.g., 55% in extensively altered Al Rais and GRO 95577). Le Guillou et  
40 al. (2015b) carried out experimental studies that suggest that these early replacement  
41 reactions took place at approximately 90°C. Consistent with these experiments, Jilly-  
42 Rehak et al. (2018) measured the oxygen isotopic fractionation between carbonates and  
43 magnetites and calculated the precipitation temperatures of these phases to be ~ 60°C. For  
44 closed-system alteration of the CR parent body, Jilly-Rehak et al. (2018) estimated that  
45 its global temperature during alteration was ~55-88°C, which is generally consistent with  
46 the temperatures for phyllosilicate formation.

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6 9.2. *Thermal metamorphism of CR chondrites*  
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10 Mild effects of heating have been consistently reported in CR chondrites. Heating  
11 mechanisms include high-temperature reduction, giving rise to low Co and Ni contents at  
12 the edges of Fe-Ni metal grains (Lee et al., 1992). Some redistribution of Cr<sub>2</sub>O<sub>3</sub> in type II  
13 chondrule olivines has been reported (Schrader et al., 2015; Abreu, 2016b).  
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16 Most CR chondrites show metamorphism consistent with petrologic types <3.2  
17 (Briani et al., 2013; Schrader et al., 2015). There are two known exceptions, GRA 06100  
18 and GRO 03116, which appear to have been heated to temperatures corresponding to  
19 petrologic types as high as 3.6 (Abreu and Singletary, 2011; Abreu and Bullock, 2013;  
20 Briani et al., 2013; Schrader et al., 2015). These studies argued that the mechanism  
21 driving heating in GRA 06100 and GRO 03116 had a short duration and consistent with  
22 an impact event. Briani et al. (2013) studied the effect of thermal metamorphism on the  
23 Ni content of Fe-Ni metal nodules, the hydration state of matrix minerals, and the  
24 structure and composition of organic matter of fifteen meteorites. Using infrared spectra,  
25 Briani et al. (2013) found that, unlike other CR chondrite matrices, the matrices of GRA  
26 06100 and GRO 03116 show significant signs of dehydration. Their organic materials  
27 have low carbonyl abundances and high CH<sub>2</sub>/CH<sub>3</sub> ratios compared to CRs that were not  
28 affected by heating. They concluded that the thermal metamorphism style of the CRs is  
29 different from that of other chondrites, in which heating has been attributed to the decay  
30 of short-lived radioisotopes. Schrader et al. (2015) measured the effects of thermal  
31 metamorphism on the major and minor element abundances of olivine, pyroxene, metal,  
32 and sulfides from 15 unpaired CR chondrites. They found that the least thermally altered  
33 CR chondrite is EET 96259, followed by QUE 99177, and the most heated is Y-793495,  
34 followed by NWA 801, GRA 06100, and EET 87770. However, they cautioned that  
35 because type II chondrules are scarce in CR chondrites, some variations in the ordering of  
36 these meteorites might be possible.  
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55 Abreu and Bullock (2013) studied the effects of thermal metamorphism on  
56 opaque minerals in one of the two heated CRs, GRA 06100. They concluded that opaques  
57 in GRA 06100 formed through shock metamorphism, followed by shock-driven  
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4 hydrothermal alteration of chondrule Fe-Ni metal and chondrule and matrix Fe-(Ni)  
5 sulfides. Opaque assemblages in GRA 06100 are aggregates of exotic, high-temperature  
6 minerals found in type I and type II chondrules. Therefore, they differ from type II  
7 chondrule Fe-Ni metal-sulfide assemblages (e.g., Schrader et al., 2015). Iron-nickel metal  
8 in GRA 06100 is rare and dominated by Ni-poor and Ni-rich intergrowths, in which Co  
9 appears to have preferentially partitioned into kamacite. Kimura et al. (2008, 2011)  
10 observed similar compositional trends in other carbonaceous chondrites and attributed  
11 them to thermal metamorphism. Abreu and Bullock (2013) found kamacite-taenite  
12 assemblages in GRA 06100, surrounded by fine-grained (<50 microns) phases, including  
13 suessite and a diaplectic glass that observed in shocked ureilites (e.g., Rubin, 1988).  
14 Other accessory phases include Ti-Cr-Fe-Ni alloys, as well as the high-temperature  
15 hydrothermal phosphate-sulfate scorzalite-lazulite and the carbonate ankerite (Abreu and  
16 Bullock 2013). Furthermore, these authors noted that opaque assemblages in GRA 06100  
17 are similar in texture and mineralogy to those formerly known as Fremdlinge located in  
18 CAIs in CV chondrites (e.g., Blum et al., 1988, 1989). In fact, major, minor, and trace  
19 element siderophile element patterns spanning a broad range of volatilities show very  
20 similar trends. It is generally accepted that Fremdlinge formed through a complex  
21 sequence of high-temperature asteroidal processes. Such processes are consistent with  
22 other evidence observed in GRA 06100 opaques, further supporting the hypothesis that  
23 this meteorite was processed at temperatures hundreds of degrees higher than other CR  
24 chondrites.

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Based on observations from GRA 06100, Abreu and Bullock (2013) argued that the CR parent body records at least one impact event in which the initial shock heating resulted in the formation of Ni-poor and Ni-rich subgrains, mobilization of Fe from precursor Fe-Ni metal that created ephemeral shock veins, formation of suessite, and overall comminution and deformation. Subsequent aqueous alteration gave rise to the Fe sulfides and Fe oxides that dominate the mineralogy of opaque assemblages and cracks throughout the meteorite. Exchange of shock metamorphic gases and fluids probably occurred on a centimeter-scale, based on the characteristics of partly filled cracks and low abundances of sulfur in the adjacent matrix.

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4 *9.3. Regolith processes affecting CR chondrites*  
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8 Physical mixing by asteroidal impacts, which has operated throughout the history  
9 of asteroids, modified their composition and mineralogy at scales ranging from individual  
10 grains to whole bodies (e.g., Bischoff et al., 2006). Impacts resulted in brecciation,  
11 regolith gardening, and derivative deformation and mechanical mixing of chondritic  
12 fragments sampling a broad range of degrees of thermal and/or aqueous alteration. CR  
13 chondrites record evidence of mixing of materials formed in different locations of an  
14 asteroid or in altogether different parent bodies. CR chondrites contain two types of  
15 foreign lithic fragments: water-rich dark inclusions, and high-pressure xenoliths.  
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18 Dark inclusions are juxtaposed against hosts with often widely different degrees of  
19 aqueous alteration. The relationship between the host CR chondrites and dark inclusions  
20 has not been clearly established. In early studies, Endreß et al. (1994) suggested that CR  
21 matrix and DIs bear no genetic relation, based on isotopic evidence, and may have been  
22 altered in completely different environments. However, Zolensky et al. (1993) argued  
23 that despite the differences in texture and grain size, CR matrices and dark inclusions are  
24 genetically related. They suggested that the matrix originated from secondary processing  
25 of dark inclusion-material. Subsequent petrologic studies of DIs and matrix show that the  
26 abundance of secondary minerals is larger in DIs, indicating that these materials cannot  
27 be the precursor of CR matrices but placing no constrains on whether these occurrences  
28 are otherwise related (e.g., Abreu, 2007). Jilly-Rehak et al. (2018) observed that  
29 carbonates in interchondrule matrix and dark inclusions plot along a single line with  
30 slope  $\sim 0.7$ , which they interpreted as both lithologies recording alteration by a single  
31 evolving fluid on the CR parent body. In addition, the dolomite from dark inclusions in  
32 Renazzo yields the same Mn-Cr ages as the calcite in the Renazzo's matrix (Jilly-Rehak  
33 et al., 2017), supporting contemporaneous alteration. Finally, it is also possible that some  
34 DIs might have formed in the CR parent body, while others had a different origin.  
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54 *9.4. High-pressure xenoliths from planetary-sized bodies*  
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4 Two CR chondrites - QUE 99177 and NWA 801 - contain xenolith clasts formed  
5 under high pressures, possibly inside large planetary bodies (Abreu, 2007, 2013; Kimura  
6 et al., 2013; Hiyagon et al., 2016). These clasts share unusual mineralogical  
7 characteristics, most prominently the presence of omphacite, a high-pressure pyroxene  
8 that has not been identified in any other meteorite group. Images from the omphacite-  
9 bearing clast in QUE 99177 are shown in Fig. 12.

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15 Omphacite clasts have distinct petrographic fabrics. Abreu (2007, 2013) found a  
16 clast in pristine QUE 99177 that consists of matrix (~48 vol.%) that is partially integrated  
17 with chondrule-like objects that have poorly defined boundaries. The matrix in this clast  
18 contains coarse (micron-sized), interlocking, compositionally equilibrated crystals,  
19 including graphite laths. The chondrules are extensively recrystallized. They contain no  
20 mesostasis, metal, or graphite. Kimura et al. (2013) found clasts that contained omphacite  
21 in NWA 801. Unlike the clast in QUE 99177, Kimura et al. (2013) reported that the clasts  
22 in NWA 801 did not have a chondrite-like lithology. They described the clasts as  
23 assemblages of a finer-grained “graphite-bearing lithology” and coarse-grained “graphite-  
24 free lithology.” The graphite-bearing lithology represented ~10 vol.% and was dominated  
25 by micron-sized silicates. They did not identify chondrule relicts in these clasts.

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35 Clasts in QUE 99177 and NWA 801 shared some mineralogical features. The  
36 dominant minerals in clasts in both QUE 99177 and NWA 801 are olivines and  
37 pyroxenes. Olivines in the three clasts in NWA 801 are more equilibrated and have  
38 slightly higher MgO contents compared to those in the clast in QUE 99177. For the clast  
39 in QUE 99177, the olivine compositions range from Fo<sub>59-64</sub> and Fo<sub>62.6</sub> average (Abreu,  
40 2013) versus Fo<sub>66-68</sub> and Fo<sub>67.2</sub> average for NWA 801 clasts (Kimura et al., 2013).  
41 Pyroxenes in the clasts in both meteorites have very similar compositions (En<sub>74.0</sub>Fs  
42 <sub>24.4</sub>Wo<sub>1.5</sub> average for QUE 99177 clast; and En<sub>73.2</sub>Fs<sub>26.1</sub>Wo<sub>0.7</sub> average for NWA 801  
43 clasts; Kimura et al., 2013). In both meteorites, graphite crystals in all clasts are  
44 interspersed within the fine-grained silicates. However, the graphite crystals in the clast  
45 in QUE 99177 are coarser-grained (up to 70 μm) and more abundant than in the clasts in  
46 NWA 801 (Abreu, 2007, 2013).

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57 The accessory mineralogy is also similar in all these clasts. In both meteorites, the  
58 clasts contain sulfides (pentlandite and troilite) and hydrous phases. The clast in QUE  
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4 99177 contains amphiboles (pargasite and pargasitic hornblende), whereas the ones in  
5  
6 NWA 801 have small amounts of the phyllosilicate phlogopite. Those in NWA 801  
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8 contain chloroapatite and metallic Fe, which have not been identified in the clast in QUE  
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10 99177 (Abreu, 2007, 2013; Kimura et al., 2013; Hiyagon et al., 2016). Unlike the clast in  
11  
12 QUE 99177, those in NWA 801 contain garnets with large pyrope and almandine  
13  
14 components (Kimura et al., 2013). Coexisting omphacite and garnets indicate that these  
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16 clasts formed in an eclogite-like environment (Kimura et al., 2013).

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18 The characteristics of omphacite indicate that these clasts had different  
19  
20 formational histories (Abreu, 2007, 2013; Kimura et al., 2013). The crystal structure of  
21  
22 omphacites has been used as a proxy for pressure in terrestrial rocks. The P2/n  
23  
24 polymorph indicates high cation ordering, whereas the C2/c corresponds to a higher  
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26 temperature, higher disordered structure (Fleet et al., 1978; McNamara, 2012; Moghadam  
27  
28 et al., 2010; Oberti and Caporuscio, 1991). In slow exhumation in terrestrial tectonic  
29  
30 settings, disordered C2/c transitions to the P2/n polymorph (Fleet et al., 1978). An  
31  
32 omphacite crystal in the QUE 9177 clast was indexed as P2/n, which corresponds to high  
33  
34 cation order. In contrast, omphacites in the clasts in NWA 801 have C2/c structure. These  
35  
36 differences might indicate that: (1) omphacite in the QUE 99177 clast formed at a slower  
37  
38 rate than in the NWA 801 clasts; (2) these xenoliths were excavated at different rates  
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40 (highly unlikely), or (3) the clast in QUE 99177 underwent a later metamorphic event  
41  
42 that is not recorded by the clasts in NWA 801.

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44 The chemical composition of omphacite suggests that the clasts in NWA 801  
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46 formed under higher pressures than the one in QUE 99177, as suggested by its higher  
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48 Al<sub>2</sub>O<sub>3</sub> contents. Hiyagon et al. (2016) reported that omphacites in the NWA 801 clasts  
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50 are more equilibrated than in the clast in QUE 99177, Na-rich, Ca-poor and more ferroan  
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52 compared with omphacites in the QUE 99177 clast (i.e., 5.8-7.3 wt.% FeO; Kimura et  
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54 al., 2013 vs. 5.2 wt.% FeO; Abreu, 2013) and higher in Al<sub>2</sub>O<sub>3</sub> (7-9.4 wt.% vs. 6.6 wt.%).

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56 Calculations of minimal overburden pressures for these clasts were performed by  
57  
58 very different methods and with very different results. Kimura et al. (2013) used  
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60 terrestrial geothermobarometers because clasts in NWA 801 have coexisting olivine,  
61  
62 pyroxenes, and garnet (garnet is not present in the clast in QUE 99177). They estimated  
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64 that 2.8–4.2 GPa and 940–1080 °C were required to form the clasts in NWA 801. In  
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4 contrast, the compositional and crystallographic features of the QUE 99177 clast indicate  
5 that omphacite formed at pressures in excess of 6 GPa (Abreu, 2007, 2013). This pressure  
6 corresponds to an overburden depth of ~760 km if typical B-type asteroid densities are  
7 assumed.  
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11 There is general agreement that these clasts must have formed over long periods  
12 of time (i.e., not during impact events), by overburden in planetary-sized, largely  
13 undifferentiated bodies that were later catastrophically disrupted (Abreu, 2007, 2013;  
14 Kimura et al., 2013; Hiyagon et al., 2016). Based on the diffusion rates for Mg–Fe in  
15 olivine and orthopyroxene at a temperature of 1000°C, Hiyagon et al. (2016) estimated  
16 that clasts in NWA 801 formed over timescales of the order of  $10^2$ – $10^3$  years. However,  
17 two very different geological settings have been proposed for the formation of these  
18 clasts. Abreu (2007, 2013) proposed that all mineral phases in the QUE 99177 clast  
19 formed and experienced transitions in the solid-state during multiple heating events in the  
20 interior of a large asteroid. In contrast, Hiyagon et al. (2016) argued that localized  
21 igneous processes generated clasts in NWA 801.  
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32 Abreu (2013) favored a metamorphic origin in a large chondritic asteroid that was  
33 catastrophically disrupted at least once and whose materials re-accreted in the CR-  
34 forming region. She argued that omphacite and graphite were the first minerals to form  
35 and that a high abundance of carbon clasts resulted from the graphitization of a carbon-  
36 rich asteroidal precursor, such as a C-rich carbonaceous chondrite. Amphibole formation  
37 postdated graphite formation and occurred by the reaction of olivine, clinopyroxene, and  
38 omphacite in the presence of an oxidizing, Na-rich fluid at high temperatures ( $> 700^\circ\text{C}$ ).  
39 Even if omphacites in the clast in QUE 99177 originally formed at high-temperature, a  
40 later high-temperature amphibole-forming event could have been sufficient for an order-  
41 disorder transition into the P2/n polymorph to occur. The absence of Fe-Ni metal in the  
42 QUE 99177 clast was interpreted as evidence that the altering fluid was also rich in  
43 sulfur. Abreu (2007, 2013) suggested that amphibole formation may have occurred at  
44 initially high temperature ( $\sim 760^\circ\text{C}$ ) and that sulfide ( $< 650^\circ\text{C}$ ) and graphite ( $> 550^\circ\text{C}$ )  
45 formation occurred as temperature decreased. In this scenario, fluid evolution played a  
46 central role in the final mineralogy. Although the conditions under which the clast formed  
47 were reducing, interaction with strongly oxidizing fluids could have produced  $\text{Fe}^{3+}$ -  
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4 bearing amphiboles. Further evolution into an H<sub>2</sub>S-rich fluid could have sulfidized  
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6 chondritic Fe-Ni alloys. Formation of omphacite and retention of carbon and sulfur  
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8 during high-temperature metamorphism is indicative of formation at high depth, instead  
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10 of during collisional, near-surface conditions.

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12 Hiyagon et al. (2016) suggested a very different sequence of events for clasts in  
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14 NWA 801. They argued that the presence/absence of graphite in the otherwise similar  
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16 lithologies was due to the formation of these lithologies at different depths. In this model,  
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18 the two lithologies were mixed, buried, and sintered over multiple collisional events. A  
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20 subsequent large collision would have given rise to a Moon-sized planetary object, and  
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22 the clasts in NWA 801 would be emplaced near the center of this ephemeral planet. In  
23  
24 these locations, the following processes would have taken place over 10<sup>2</sup>-10<sup>3</sup> years:  
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26 formation of the eclogitic phases (i.e., omphacite and garnet), homogenization of Fe-Mg  
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28 olivine and pyroxenes, and redistribution of P and REEs between two lithologies. Finally,  
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30 a collision would have catastrophically disrupted the Moon-sized body, quenching the  
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32 clasts and transporting them to the region where they accreted to form the CR parent  
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34 body. The origin of the single hydrous phase, phlogopite, was not discussed in detail in  
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36 this model.

37  
38 It is unclear if clasts in QUE 99177 and NWA 801 came from precursors with  
39  
40 widely different histories. Their different mineralogical assemblages, in particular the  
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42 high abundance of graphite and absence of garnet in the clast in QUE 99177, suggest that  
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44 the precursors' composition was indeed different (e.g., Hiyagon et al., 2016). More  
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46 detailed comparative studies are needed to determine if these clasts had indeed the  
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48 different histories as current investigations suggest. It is important to note that the  
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50 graphite-free lithology in the NWA 801 clasts (Kimura et al., 2013) closely resembles the  
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52 objects that Abreu (2013) identified as relict chondrules in the QUE 99177 clast. The  
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54 graphite-bearing lithology in the NWA 801 clasts also resembles the matrix materials in  
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56 the QUE 99177 clast. The absence of graphite in relict chondrules in the QUE 99177  
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58 clast is easily explained by the fact that chondrule precursors would have been essentially  
59  
60 carbon-free (Abreu, 2007, 2013). In contrast, the formation of the graphite-bearing and  
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62 the graphite-free lithologies at different depths would require a very complex sequence of  
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64 excavation events and mixing, as suggested by Hiyagon et al. (2016).  
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10 **10. Physical properties of CR chondrites**  
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13       Measurements of the physical properties of CR chondrites are scarce. Little is  
14 known, for example, about electrical and thermal properties of either bulk samples or  
15 their components. Studies of physical properties generally involve destructive analyses,  
16 which are hampered by the small masses of many CRs, the largest of which are  
17 extensively weathered finds from hot deserts (e.g., Lee and Bland, 2004; Bland et al.,  
18 2006).  
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21       Terrestrial weathering might affect physical properties, which might, in turn, limit  
22 the value of their analyses. However, it is difficult to determine what the effects of  
23 weathering might be specifically for CR chondrites because there are only two observed  
24 falls that can be used for comparison (Macke et al., 2011). Some researchers have argued  
25 that weathering of carbonaceous chondrite finds does not produce discernible changes in  
26 their physical properties (Flynn et al., 2017). Perhaps because all changes occur very  
27 early during the terrestrial weathering process, it is challenging to find samples that were  
28 not affected by them. If most changes in physical properties of CRs occur after mild  
29 terrestrial weathering, as it is the case for ordinary chondrites (e.g., Flynn et al. 2017),  
30 patterns are not likely to be observable in extensively weathered CR chondrites from hot  
31 deserts.  
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33       Macke et al. (2011) measured several physical properties of falls Al Rais and  
34 Renazzo and finds Acfer 097, Acfer 270, Dhofar 1432, El Djouf 001, and Tafassasset.  
35 They found the average bulk porosity to be  $9.5\% \pm 2.7$ . In comparison, Corrigan et al.  
36 (1997) obtained lower porosities for Al Rais and Renazzo matrices (2-8%). Macke et al.  
37 (2011) also determined the average bulk and grain densities of  $3.11 \pm 0.14 \text{ g/cm}^3$  and  
38  $3.42 \pm 0.08 \text{ g/cm}^3$ , respectively. Zolensky et al. (2018) proposed that these values should  
39 be adjusted because Gardner-Vandy et al. (2012) reclassified Tafassasset as a primitive  
40 achondrite. The adjusted values then are for average porosity  $11.23\% \pm 7.4$ , for average  
41 bulk and grain densities of  $2.99 \pm 0.35 \text{ g/cm}^3$  and  $3.36 \pm 0.16 \text{ g/cm}^3$ .  
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4 The largest variability in the averages for the physical properties was observed for  
5 Al Rais and Tafassasset. Macke et al. (2011) explained intra-sample heterogeneities as  
6 because most CR chondrites are breccias. Brecciation would result in different sets of  
7 physical properties for different lithic fragments. Furthermore, variability in the  
8 properties of Al Rais was attributed to the fact that this was the only fall analyzed by  
9 Macke et al. (2011). Zolensky et al. (2018) also noted that, aside from being a fall, Al  
10 Rais is also an anomalous CR2, having low chondrules:matrix abundance ratios with  
11 respect to other CR chondrites. Schrader et al. (2011) found that chondrule:matrix ratios  
12 are highly variable in CR chondrites and, for Al Rais is 0.46, compared to 1.87 for  
13 average CR chondrites. If fine-grained chondritic materials are more porous than  
14 chondrules, Zolensky et al. (2018) hypothesized that the high average porosity of Al Rais  
15 might have resulted from the fact that it has a low chondrule:matrix ratio. In light of the  
16 re-classification of Tafassasset, it is not surprising that the physical properties of this  
17 meteorite are so different from other CR chondrites.  
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30 Rochette et al. (2008) analyzed 16 CR chondrites including falls and cold and hot  
31 desert finds, and obtained average magnetization of  $\log \chi = 5.04 \pm 0.12$  (in  $10^{-9} \text{ m}^3/\text{kg}$ ;  
32 range 4.51-5.17). These values include Tafassasset and Kaidun, both of which have since  
33 been reclassified. If the values from these two chondrites are removed from the CR  
34 chondrite average, the result is  $\log \chi = 4.93 \pm 0.23$  (in  $10^{-9} \text{ m}^3/\text{kg}$ ; range 4.48-5.17).  
35 Removing Tafassasset, the average magnetic susceptibility obtained by Macke et al.  
36 (2011) is  $\log \chi = 4.94 \pm 0.18$ . Finally, Ibrahim (2012) determined the mean  
37 compressional wave velocity for CR chondrites to be  $2770 \pm 993 \text{ m/s}$  and the average  
38 shear wave velocity to be  $1751 \pm 611 \text{ m/s}$  and noted that porosity and velocities are  
39 anticorrelated. However, it is not known if differences in the mechanisms that lower the  
40 porosities (e.g., aqueous alteration versus impact metamorphism) might influence the  
41 compressional wave and shear wave velocities.  
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52 Compared with CM chondrites, average CR chondrite densities are slightly  
53 higher, and porosities are approximately half (Macke et al., 2011). Higher Fe-Ni metal  
54 abundances in CR chondrites probably resulted in their higher density. Differences in  
55 porosity could be the results of lower chondrule:matrix ratios in CMs compared to CRs.  
56 The magnetic susceptibility of CR chondrites is the highest for all the hydrated  
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4 carbonaceous chondrites and only lower than that of enstatite chondrites and the very  
5 metal-rich CH and CB chondrites. The average magnetic susceptibility values for CRs  
6 are comparable/intermediate between CO and CV chondrites (Rochette et al., 2008;  
7 Macke et al., 2011; Flynn et al., 2017). However, Rochette et al. (2008) cautioned that the  
8 magnetic susceptibilities of CRs should be interpreted as a lower limit because metallic  
9 minerals in CR chondrites have been subjected to significant terrestrial weathering and  
10 replacement.  
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## 19 **11. Spectral studies of CR chondrites:**

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22 For a number of reasons, the spectroscopic properties of CR chondrites are  
23 diverse. The petrologic characteristics that affect spectroscopic observations are: (a)  
24 presence of both pre- and post-terrestrial Fe oxyhydroxides; (b) the fact that they record  
25 the full range of aqueous alteration processes; and (c) some CRs might have been  
26 thermally metamorphosed after being aqueously altered. For example, spectral  
27 differences between falls and finds were found for CM chondrites, which are  
28 mineralogically most similar to type 2 CR chondrites (Cloutis et al., 2011).  
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### 37 *11.1. VNIR reflectance spectra of CR chondrites*

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41 The most comprehensive study to date of the spectral reflectance properties of CR  
42 chondrites was presented by Cloutis et al. (2012). Since then, the spectral reflectance  
43 properties of other CR chondrites have been studied (e.g., Abreu et al., 2018; Kiddell et  
44 al., 2018). Cloutis et al. (2012) found that a combination of low reflectance (4-16% near  
45 0.75  $\mu\text{m}$ ), red spectral slope, and weak silicate absorption bands (<5% deep) seem to  
46 characterize CR chondrites. It also appears that CR spectra range from neutral to slightly  
47 red-sloped, which may distinguish them from some other carbonaceous chondrite groups.  
48 Diagnostic absorption features associated with silicates are weak because of their low  
49 FeO contents. However, this weakness may, in fact, be somewhat diagnostic. Overall, the  
50 reflectance is also generally low, but not uniquely so.  
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4 In terms of uniqueness, CR reflectance spectra do not exhibit particularly  
5 diagnostic features (Cloutis et al., 2018b). Some of this stems from the fact that CRs are  
6 petrologically diverse, ranging from mafic silicate- to phyllosilicate-dominated. Beyond  
7 this, the overprinting effects of terrestrial oxyhydroxides can obscure any diagnostic  
8 absorption features below  $\sim 1 \mu\text{m}$ . The consequences of these effects are discussed below.  
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### 15 *11.1.1. Effects of terrestrial alteration on spectral features*

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19 One factor that affects VNIR reflectance spectra of CR chondrites is the presence  
20 of Fe oxyhydroxides. There are only two CR fresh falls, Al Rais and Renazzo, and only  
21 Renazzo has been spectrally characterized (one spectrum of one subsample) (Cloutis et  
22 al., 2012). Compared to the CR finds, Renazzo has extremely weak absorption features in  
23 the  $1 \mu\text{m}$  region (Cloutis et al., 2012). The weakness of absorption features in Renazzo is  
24 attributed to the fact that most anhydrous CR silicates have very low Fe-content (e.g.,  
25 Kallemeyn et al., 1994; Weisberg et al., 1993).  
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32 Features consistent with the presence of Fe-oxyhydroxides dominate reflectance  
33 spectra of CR finds (Cloutis et al., 2012). The spectra of Fe-oxyhydroxides have very low  
34 reflectance below  $\sim 0.5 \mu\text{m}$ , and an  $\text{Fe}^{3+}$ -associated absorption band in the  $0.9 \mu\text{m}$  region  
35 (e.g., Sherman et al., 1982; Morris et al., 1985). As discussed above, both terrestrial and  
36 preterrestrial origins have been proposed for Fe-oxyhydroxides (e.g., Le Guillou et al.,  
37 2015a; Abreu, 2016a). Fe-oxyhydroxides could have also formed via incipient aqueous  
38 alteration in the CR parent body. The reason for the suggestion that Fe-oxyhydroxides  
39 formed early in asteroidal history is the same as for their formation in cold terrestrial  
40 deserts: CR matrices and Fe-Ni are highly reactive and likely to be affected by even small  
41 amounts of water, reacting at low temperatures. This is the reason why Abreu (2016a)  
42 noted that terrestrial weathering effects are likely to affect the CR chondrites more  
43 strongly than for other groups of carbonaceous chondrites. It also means that these two  
44 origins are not mutually exclusive. The preponderance of Fe-oxyhydroxides in CR  
45 reflectance spectra is consistent with the alteration of phases prone to reactions with  
46 water.  
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4           Kiddell et al. (2018) studied the effects of Fe-oxyhydroxides on the spectra of  
5 various carbonaceous chondrites, including CR2 NWA 801. As is the case with many hot  
6 desert finds, this meteorite is brown/orange in color, with pervasive veins of rust  
7 throughout (Connolly et al., 2007). Kiddell et al. (2018) found that “derusting” NWA 801  
8 samples by exposing the samples to a thiol-bearing compound reacts preferentially with  
9 hydrated ferric iron resulted in complete or partial removal of the Fe oxyhydroxides,  
10 restoring some, but not all, carbonaceous chondrite spectra to their unaltered state.  
11 Whereas derusting may mitigate the spectrum-altering effects of Fe oxyhydroxides, some  
12 of these phases may be of preterrestrial origin, and hence “derusted” CR spectra may not  
13 accurately reflect their preterrestrial condition. In addition, CR chondrites have high  
14 ferric to total iron ratios, suggesting abundant preterrestrial Fe-oxyhydroxides. Despite  
15 these caveats, “derusting” can mitigate the spectrum-altering effects of Fe-  
16 oxyhydroxides. Removing Fe-oxyhydroxides allows diagnostic absorption features from  
17 hydrated and anhydrous mafic silicates to be more clearly discerned.  
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### 32 *11.1.2. Effects of asteroidal processes on spectral features*

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35           CR chondrites show the full range of aqueous alterations, and there are CRs that  
36 appear to have undergone thermal metamorphism subsequent to aqueous alteration (e.g.,  
37 GRA 06100 and GRO 03116). Cloutis et al. (2012) studied CR chondrites of all  
38 petrologic types, including a CR1 chondrite (GRO 95577), a number of CR2 chondrites,  
39 and a CR3 chondrite (MET 00426). Fig. 13a shows the reflectance spectra of powders of  
40 three different petrologic grade CR chondrites. Absolute differences in reflectance  
41 spectra are probably attributable, at least in part, to grain size variations, although for  
42 such fine-grained powders, the effects are expected to be small (Cloutis et al., 2018a).  
43 The steepness of the spectra shortward of ~0.5  $\mu\text{m}$  for MET 00426 and GRO 95577  
44 compared to Renazzo is probably due in part to the Fe oxyhydroxides resulting from  
45 terrestrial weathering of these two finds (both GRO 95577 and MET 00462 are  
46 weathering grade B or B/C). However, some mineralogical information can be derived  
47 even from the CR spectra of weathered samples.  
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4 Differences in the degree of aqueous alteration result in changes in CR spectral  
5 reflectance properties because this process eventually mobilizes a large fraction of the Fe  
6 from Fe-Ni metal nodules into silicates and oxides. CR3s have abundant, nearly intact  
7 type I chondrules (MgO-rich and FeO-poor), Fe-Ni metal nodules, and matrices that are  
8 rich in amorphous Fe-Mg silicates, resulting in higher fractions of FeO-poor silicates than  
9 other petrologic types, if chondrule to matrix ratios are held constant. If the first stage of  
10 CR aqueous alteration resulted in the formation of Fe-oxyhydroxides, then this ought to  
11 be the largest spectral change observed in type CR2. Along with this process, amorphous  
12 Fe-Mg silicates give rise to crystalline Fe-Mg silicates, and amorphous Fe-Mg silicates  
13 give rise to Fe-Mg phyllosilicates (dominantly serpentine and saponite). Abreu (2016a)  
14 found that the FeO and MgO contents of these phases did not systematically change in  
15 most CRs. Furthermore, they observed no changes in the FeO contents of chondrule  
16 phenocrysts or mesostases (Abreu, 2016a). These observations suggest that, at least in the  
17 early stages of aqueous alteration, there were no changes in the FeO contents of CR  
18 silicates. In contrast, FeO mobilization is particularly marked in CR1 GRO 95577, where  
19 type I chondrules have been extensively replaced by Fe-Mg phyllosilicates (e.g., Morlok  
20 and Libourel, 2013). Increasing aqueous alteration also leads to alteration of metal to  
21 magnetite and formation of carbonates (e.g., Tyra, 2013). Consequently, GRO 95577 has  
22 the largest abundance of FeO in silicates among the CRs.  
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39 Cloutis et al. (2012) found that reflectance declines with increasing aqueous  
40 alteration, consistent with higher opaque abundances in lower petrologic grades  
41 (Perronnet et al., 2007). Overall, CR spectral reflectance properties in the 0.3-2.5  $\mu\text{m}$   
42 region show evidence of pyroxene  $\pm$  olivine and/or glassy Fe-Mg silicates (CR3) to  
43 phyllosilicate  $\pm$  olivine/pyroxene (CR2), to fully dominated by Fe-bearing phyllosilicates  
44 (CR1). Because FeO contents of the silicates are uniformly low (with the exception of  
45 matrix silicates), any associated absorption bands are expected to be weak. Also, it is  
46 likely that spectral slopes become less red (bluer). The overall red spectral slope of CRs  
47 is likely due to contributions of Fe-Ni metal, poorly graphitized carbon, and sulfides (all  
48 of which are red-sloped), but not magnetite (which imparts a blue slope). However, as  
49 noted, only one CR chondrite fall has been spectrally characterized (Renazzo), and it is  
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4 possible that fresh CRs may have a range of spectral slopes. Continuum removal applied  
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6 to the 1  $\mu\text{m}$  region can allow observing some silicate absorption bands.  
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8 Cloutis et al. (2012) determined the characteristics of CRs from different  
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10 petrologic types as follows. For MET 00426 and other CR3 chondrites, identifying  
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12 absorption bands associated with chondrule silicates is difficult largely because  
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14 absorption bands of silicates with low Fe-contents are weak (e.g., King and Ridley, 1987;  
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16 Clark et al., 1990; Cloutis and Gaffey, 1991). CR3s contain abundant pyroxene (McBride  
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18 et al., 2001a,b; McSween, 1977b; Perronnet et al., 2007), in large enough amounts that  
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20 they could affect the reflectance spectra (Cloutis et al., 1986). The spectrum of MET  
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22 00426 shows an absorption feature in the 0.9  $\mu\text{m}$  region, consistent with pyroxene, but  
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24 also highly overlapped by Fe oxyhydroxide absorption bands (Sherman et al., 1982;  
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26 Morris et al., 1985). Its strength would also be reduced, because pyroxenes,  
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28 predominantly from type I chondrules, have low FeO contents (McSween, 1977a;  
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30 McBride et al., 2001a,b; Perronnet et al., 2007; Schrader et al. 2015). The 1.8  $\mu\text{m}$  region  
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32 shows stronger evidence for pyroxene: a broad, but shallow absorption feature centered  
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34 near 1.81  $\mu\text{m}$ , consistent with low-Fe pyroxene (Cloutis and Gaffey, 1991); but again,  
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36 this spectral region can also be affected by Fe oxyhydroxides, and this is suggested by a  
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38 broad absorption feature near 1950 nm. Thus, spectral evidence for low-FeO mafic  
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40 silicates is ambiguous, but largely because of terrestrial weathering effects.

41 The reflectance spectrum for Renazzo shows a weak ( $\leq 2\%$  deep) absorption  
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43 feature in the 1  $\mu\text{m}$  region, consistent with both Fe-bearing serpentine (band near 1.1  $\mu\text{m}$ )  
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45 and magnetite (broad feature in this region), both of which are abundant in this meteorite  
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47 (Mason and Wiik, 1962; Zolensky et al., 1993). However, the weakness of this absorption  
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49 feature makes phase identification problematic. The overall slightly red spectral slope  
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51 may be due to the small amounts of metal and/or sulfides.

52 CR1 GRO 95577 contains various phyllosilicates and little or no anhydrous  
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54 silicates, consistent with pervasive aqueous alteration. Spectrally, the broad 1  $\mu\text{m}$  region  
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56 shows strong evidence of Fe-bearing phyllosilicates: an absorption band near 1.1  $\mu\text{m}$ ,  
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58 which is not overlapped by Fe oxyhydroxide bands. A small absorption feature just  
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60 longwards of 2.3  $\mu\text{m}$  is also most consistent with Fe-Mg phyllosilicates (Cloutis et al.,  
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62 2012), again in agreement with its dominant mineralogy.  
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6 *11.1.3. Thermal metamorphism effects*  
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10 The major effects of heating are to (slightly – a few percent) increase the FeO  
11 content in the silicates, and dehydrated pre-existing phyllosilicates (Briani et al., 2013;  
12 Schrader et al., 2015; Abreu, 2016a). Spectrally, the thermally metamorphosed CRs for  
13 which we have spectral reflectance data (Fig. 13a), show intermediate reflectance (~8%  
14 at 0.75  $\mu\text{m}$  versus 4-16% for all CRs; Cloutis et al., 2012), suggesting that thermal  
15 metamorphism may not affect overall reflectance. However, the 1  $\mu\text{m}$  region shows clear  
16 evidence of this metamorphism: specifically, olivine-associated absorption features: an  
17 absorption band near 1.05  $\mu\text{m}$  and a shoulder near 1.2  $\mu\text{m}$ ; and a possible pyroxene  
18 absorption band in the 1.8  $\mu\text{m}$  region.  
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28 *11.2. Infrared spectra of CR chondrites:*  
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32 A number of CR chondrites have been spectrally characterized by various infrared  
33 spectroscopies and have focused on silicates and organics. For example, Miyamoto and  
34 Zolensky (1994) measured the 2.5-25  $\mu\text{m}$  reflectance spectra of a number of  
35 carbonaceous chondrites, including Renazzo. They reported that the OH/H<sub>2</sub>O-associated  
36 absorption feature in the 2.7-3.5  $\mu\text{m}$  region is most similar to CM chondrites in shape, but  
37 of lower intensity, consistent with a lower content of hydrated minerals. CRs and other  
38 carbonaceous chondrites exhibit a general trend of increasing band intensity with  
39 increasing H content. Renazzo also shows a carbonate absorption band in the 7  $\mu\text{m}$  region  
40 that is most similar in intensity to those of CM2 carbonaceous chondrites. Osawa et al.  
41 (2001) measured diamond cell infrared transmission spectra of two CRs over the range  
42 2.5-16  $\mu\text{m}$ . They found that the transmission spectra are most similar to that of the  
43 Murchison CM2 chondrite in terms of shape and position of O-H stretching bands in the  
44 3  $\mu\text{m}$  region, and Si-O stretching bands in the 10  $\mu\text{m}$  region. Osawa et al. (2005)  
45 measured 2.5-13  $\mu\text{m}$  infrared transmission spectra of carbonaceous chondrites using KBr  
46 and diamond press methods. Their sample suite included 3 Antarctic CR2s. They  
47 measured two OH/H<sub>2</sub>O-stretching absorptions in the 3  $\mu\text{m}$  region which are less intense  
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4 than the same bands in CM spectra. The CR spectra also exhibited a series of absorption  
5 bands in the 6-7.5  $\mu\text{m}$  region and a stronger double band in the 9.5-11  $\mu\text{m}$  region, sharing  
6 some similarities with CI and CM chondrites.  
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10 Briani et al. (2013) presented infrared transmission spectra of insoluble organic  
11 matter (IOM) extracted from four CR chondrites, as well as spectra of matrix material  
12 from six CR chondrites. The spectra of the IOM from two CR chondrites that  
13 experienced thermal metamorphism (as determined by multiple criteria; GRO 03116 and  
14 GRA 06100) differ from those of unmetamorphosed CR chondrites. They showed that  
15 both metamorphosed CRs display Si-O stretching bands dominated by crystalline olivine.  
16 No spectral evidence of hydroxyl was found (in contrast to CI and CM chondrites), but  
17 variable evidence of molecular water was noted. Orthous-Daunay et al. (2013) measured  
18 2.5-25  $\mu\text{m}$  infrared absorption spectra of IOM from a suite of carbonaceous chondrites,  
19 including three CR chondrites. They found that the absorption bands are similar to  
20 petrologic grade 2 carbonaceous chondrites, and variations between meteorites are  
21 associated with a high-temperature short-duration heating event.  
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32 Beck et al. (2014) measured KBr disk 2-25  $\mu\text{m}$  infrared transmission spectra of a  
33 suite of 40 carbonaceous chondrites, including six CR chondrites. They noted that the CR  
34 group (petrologic grade 1-2.5) shows the most spectral variability of any of the groups.  
35 Spectral types range from CM-like to CV-like. The CV-like spectrum is for GRA 06100,  
36 which, as noted, appears to have been thermally metamorphosed. They noted that the 10  
37  $\mu\text{m}$  region of the CR spectra is more complex than for other carbonaceous chondrites,  
38 likely because of contributions from both olivine and pyroxene. They also noted that the  
39 CR1 chondrite, GRO 95577, shows infrared spectral features similar to those of heavily  
40 altered CI and CM chondrites, and to saponite.  
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48 Infrared reflectance spectra of powders of four CR chondrites are presented in  
49 Figure 13c-d (the CR2s MAC 87320 and PCA 91082; and pristine MET 00426 and QUE  
50 99177). At shorter wavelengths, all four spectra show a somewhat featureless absorption  
51 feature in the 3  $\mu\text{m}$  region that suggests contributions from hydroxyl (absorption near 2.8  
52  $\mu\text{m}$ ) and water (absorption near 3.1  $\mu\text{m}$ ). The shape of this feature differs somewhat  
53 between the various spectra, but all show contributions from hydroxyl and molecular  
54 water. They all also show minor spectral contributions from aliphatic organics: weak  
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4 absorption bands in the 3.4-3.5  $\mu\text{m}$  region. At longer wavelengths, reflectance  
5 progressively decreases, and the spectra all show further evidence of molecular water  
6 (absorption band near 6.1  $\mu\text{m}$ ), and various Si-O stretching bands in the 8-12  $\mu\text{m}$  region.  
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8 Differences, if any, between these spectra are minor.  
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## 10 11 12 13 **12. CR chondrites and exploration of their parent asteroids** 14

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17 Because of their limited availability and scientific importance, NASA's Antarctic  
18 CRs have been assigned to the list of protected samples. It is unlikely that these CRs will  
19 be widely available for experiments that require masses larger than a few grams or for  
20 exploration-driven activities, such as simulant production or comparative studies. Sample  
21 mass limitations will likely place demands on the North African CR sample suite.  
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### 24 25 26 27 28 *12.1. Implications of matrix studies for the distribution of water and organics in the solar* 29 *system, in carbonaceous asteroids, and for ISRU of asteroidal materials* 30 31

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33 Exploration of CR parent bodies holds great promise for both science and *in-situ*  
34 resource utilization (ISRU). Petrologic studies can address the ISRU-relevant questions,  
35 including (a) How are Fe-Ni metal nodules affected by aqueous and hydrothermal  
36 alteration in asteroids?; (b) What is the distribution of water-bearing minerals in heated  
37 and unheated CR-like materials?; and (c) What is the effect of thermal metamorphism on  
38 the distribution of organics? Several studies proposed the future use of H<sub>2</sub> and O<sub>2</sub> from  
39 water-bearing minerals for in-situ production of fuel and propellants (e.g., Anand et al.  
40 2012). In CRs, the mineralogical hosts of OH and H<sub>2</sub>O are amorphous Fe-Mg silicates  
41 and phyllosilicates. We are only beginning to characterize these materials. Understanding  
42 the effect of asteroidal processes on Fe-Ni metal will constrain the availability,  
43 distribution, and mineral host of precious metals for ISRU demonstrations. Extraction of  
44 volatiles and metals will involve material processing: removing and mechanically  
45 processing the regolith (e.g., drilling, scooping, crushing, sorting, etc.) and chemically  
46 processing the regolith (e.g., heat of volatilization, reaction mechanisms). Designing tools  
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4 to accomplish ISRU goals should be informed by the distribution, chemical composition,  
5 and mineralogical characteristics of analog materials.  
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## 10 *12.2. Key spectral characteristics for identification of CR parent bodies*

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14 Spectral matches between CRs and potential parent bodies are hindered by the  
15 effect of terrestrial weathering on the mineralogy of these meteorites. We face multiple  
16 challenges to address this uncertainty. Spectrally, CR chondrites differ from other  
17 carbonaceous chondrites in ways that relate to their mineralogies. CR chondrites include  
18 different petrologic grades, ranging from ~3 (relatively unaltered) to ~1 (pervasively  
19 aqueously altered), as well as two CRs that show evidence of subsequent thermal  
20 metamorphism (suggestive of petrologic grade up to ~3.6). These petrologic differences  
21 translate into mineralogical properties that vary from abundant amorphous Fe-Mg  
22 silicates (petrologic grade 3) to increasing amounts of Fe-Mg phyllosilicates (toward  
23 petrologic grade 1), and abundant mafic silicates (above petrologic grade 3). The  
24 anhydrous silicates are characterized by, on average, very low FeO contents, and those  
25 above petrologic grade 3 include abundant pyroxene.  
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36 Collectively we expect the parent asteroids of all CR chondrites to have the  
37 following characteristics: (1) weak silicate-associated absorption bands in the 0.3-2.5  $\mu\text{m}$   
38 region (<5% deep); (2) pyroxene-associated absorption bands in the 0.9 and 1.8  $\mu\text{m}$   
39 regions (for petrologic grade  $\geq 3$ ); (4) red-sloped spectra; and (4) low overall reflectance  
40 (between 4 and 15% in the visible region).  
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45 At longer wavelengths, we expect (1) an absorption feature in the 3  $\mu\text{m}$  region  
46 with contributions from both hydroxyl and molecular water; (2) possible weak aliphatic  
47 organic absorption bands in the 3.4-3.5  $\mu\text{m}$  region, a molecular water absorption band in  
48 the 6  $\mu\text{m}$  region, and (3) Si-O stretching bands in the 9-12  $\mu\text{m}$  region that are consistent  
49 with olivine (petrologic grade above ~2) and pyroxene (petrologic grade above ~3).  
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54 No one spectral property will uniquely identify CR chondrite parent bodies, but  
55 multiple lines of spectral evidence can strengthen the case for linking specific asteroids to  
56 CR chondrites. Further spectral analysis of CR chondrites with minimal terrestrial  
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4 alteration products, or “derusted” in the laboratory, should yield additional insights into  
5 potentially diagnostic spectral parameters.  
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### 9 10 **13. Conclusions**

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13 CR-like parent bodies are scientifically important and resource-rich targets for  
14 exploration. These meteorites contain some of the most pristine solar nebular materials.  
15 In many CRs, post-assembly heating and aqueous alteration were too minor and/or  
16 localized to homogenize the material. These highly pristine samples have matrices that  
17 are dominated by amorphous, water-rich, Fe-Mg-rich silicates. In other groups of  
18 carbonaceous chondrites that record more extensive asteroidal processes, these materials  
19 have been progressively converted to hydrous and anhydrous crystalline silicates. The  
20 primitive nature of CRs is supported by the presence and high relative abundance of pre-  
21 solar grains from a wide variety of stellar and interstellar sources, as well as from the  
22 presence of exotic organics.  
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32 Parent bodies vary widely in terms of size, gravity, physical state, composition,  
33 and volatile abundance (Hutchison, 2004). Imposed on these distinctions is an equally  
34 variable history of impacts, alteration, differentiation, and igneous fractionation.  
35 Consequently, understanding how these asteroidal processes impact the spectral  
36 characteristics of these parent bodies is a fundamental first step in exploration.  
37 The record of aqueous alteration in CRs is one of the most complete among the  
38 carbonaceous chondrites. Our collections include very pristine samples and those that  
39 have been entirely converted into water-bearing minerals and oxides (i.e., GRO 95577).  
40 In addition to these materials that reveal important information about nebular and  
41 presolar processes, the CR Suite includes both monomict and polymict breccias. CR  
42 polymict breccias potentially come from bodies larger than any present-day asteroids.  
43 Finally, spectral analyses of CR chondrites are our baseline for asteroidal observations  
44 and exploration of their asteroidal parent bodies.  
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16  
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20  
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22  
23  
24  
25

## 26 **References**

- 27  
28 Abreu, N. M., 2007. Fine-scale mineralogical study of the matrices of CR carbonaceous  
29 chondrites: Insights on early solar system processes Proquest Dissertations And  
30 Theses 2007. Section 0142, Part 0372 225 pages; [Ph.D. dissertation]. The  
31 University of New Mexico, United States, New Mexico, Publication Number:  
32 AAT 3273431. Source: DAI-B 68/07.  
33  
34  
35  
36 Abreu, N. M., 2011. Petrographic evidence of shock metamorphism in CR2 chondrite  
37 GRO 03116 (abstract). Meteorit. Planet. Sci. Suppl., id.5211.  
38  
39  
40 Abreu, N. M., 2013. A unique omphacite, amphibole, and graphite-bearing clast in Queen  
41 Alexandra Range (QUE) 99177: A metamorphosed xenolith in a pristine CR3  
42 chondrite. Geochim. Cosmochim. Acta 105, 56-72.  
43  
44  
45 Abreu, N. M., 2016a. Why is it so difficult to classify Renazzo-type (CR) carbonaceous  
46 chondrites? - Implications from TEM observations of matrices for the sequences  
47 of aqueous alteration. Geochim. Cosmochim. Acta 194, 91-122.  
48  
49  
50 Abreu, N. M., 2016b. Are phyllosilicates in CR chondrite matrices generated by  
51 hydrothermal alteration? (abstract). 47<sup>th</sup> Lunar Planet. Sci. Conf. LPI Contribution  
52 No. 1903, p.1926.  
53  
54  
55 Abreu, N. M., Brearley, A. J., 2005. Carbonates in Vigarano: Terrestrial, preterrestrial, or  
56 both? Meteorit. Planet. Sci. 40, 609-625.  
57  
58  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 Abreu, N. M., Brearley, A. J., 2010. Early solar system processes recorded in the  
5 matrices of two highly pristine CR3 carbonaceous chondrites, MET 00426 and  
6 QUE 99177. *Geochim. Cosmochim. Acta* 74, 1146–1171.  
7  
8  
9 Abreu, N. M., Bullock, E. S., 2013. Opaque assemblages in CR2 Graves Nunataks  
10 (GRA) 06100 as indicators of shock-driven hydrothermal alteration in the CR  
11 chondrite parent body. *Meteorit. Planet. Sci.* 48, 2406-2429.  
12  
13  
14 Abreu, N. M., Singletary, S., 2011. Alteration history of CR2 chondrite GRA 06100: FE-  
15 EPMA and TEM analysis (abstract). LPI Contribution No. 1608, p.2659.  
16  
17  
18 \*Abreu, N. M., Cloutis, E. A., Hamilton, V. E. 2018. Understanding the effects of  
19 Antarctic weathering on the petrologic and spectral characteristics of pristine CR  
20 carbonaceous chondrites (abstract). 49<sup>th</sup> Lunar Planet. Sci. Conf. LPI Contribution  
21 No. 2083, id.1297.  
22  
23  
24 Aléon, J., Krot A. N., McKeegan, K. D., 2002. Calcium-aluminum-rich inclusions and  
25 amoeboid olivine aggregates from the CR carbonaceous chondrites. *Meteorit.*  
26 *Planet. Sci.* 37, 1729-1755.  
27  
28  
29 Alexander, C. M. O.'D., Bowden, R., 2018. Lewis Cliff (LEW) 85332 and Miller Range  
30 (MIL) 090001, a new Grouplet that is distinct from the CR chondrites? (abstract)  
31 81<sup>st</sup> Ann. Meet. Meteoritical Society. LPI Contrib. No. 2067, 2018, id.6063.  
32  
33  
34 Alexander, C. M. O.'D., Bowden, R., Fogel, M. L., Howard, K. T., 2015. Carbonate  
35 abundances and isotopic compositions in chondrites. *Meteorit. Planet. Sci.* 50,  
36 810-833.  
37  
38  
39 Alexander, C. M. O. D., Howard, K. T., Bowden, R., Fogel, M. L., 2013. The  
40 classification of CM and CR chondrites using bulk H, C and N abundances and  
41 isotopic compositions. *Geochim. Cosmochim. Acta* 123, 244-260.  
42  
43  
44 Alexander, C. M. O.'D., R. Bowden, Fogel M. L., Howard K. T., Herd, C. D. K., Nittler,  
45 L. R., 2012. The provenances of asteroids, and their contributions to the volatile  
46 inventories of the terrestrial planets. *Science* 337, 721-723.  
47  
48  
49 Alexander, C. M. O'D., Fogel, M., Yabuta, H., Cody, G. D., 2007. The origin and  
50 evolution of chondrites recorded in the elemental and isotopic compositions of  
51 their macromolecular organic matter. *Geochim. Cosmochim. Acta* 71, 4380–  
52 4403.  
53  
54 Al-Kathiri, A., Hofmann, B. A., Jull, A. J. T., and Gnos, E., 2005. Weathering of  
55 meteorites from Oman: Correlation of chemical/mineralogical weathering proxies  
56 with <sup>14</sup>C terrestrial ages and the influence of soil chemistry. *Meteorit. Planet. Sci.*  
57 40, 1215–1239.  
58  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 Anand, M., Crawford, I. A., Balat-Pichelin, M., Abanades, S., van Westrenen W.,  
5 Pe´raudeau, G., Jaumann ,R., Seboldt, W., 2012. A brief review of chemical and  
6 mineralogical resources on the Moon and likely initial in situ resource utilization  
7 (ISRU) applications. *Planet. Space Sci.* 74, 42–48.  
8  
9  
10 Aponte, J. C., Woodward, H. K., Abreu, N. M., Elsila, J. E., Dworkin, J. P., 2019.  
11 Molecular distribution, <sup>13</sup>C-isotope, and enantiomeric compositions of  
12 carbonaceous chondrite monocarboxylic acids. *Meteorit. Planet. Sci.* 54, 415-430.  
13  
14  
15 \*Aponte, J. C., Abreu, N. M., Keller, L. P., Elsila, J. E., Dworkin, J. P., 2018. Soluble  
16 amines in anomalous CR chondrite Miller Range (MIL) 090001 (abstract). 49<sup>th</sup>  
17 Lunar Planet. Sci. Conf., LPI Contrib. No. 2083, id.1543.  
18  
19  
20 Aponte, J. C., McLain, H. L., Dworkin, J. P., Elsila, J. E., 2016. Aliphatic amines in  
21 Antarctic CR2, CM2 and CM1/2 carbonaceous chondrites. *Geochim. Cosmochim.*  
22 *Acta* 189, 296-311.  
23  
24  
25 Aponte, J. C., Tarozo, R., Alexandre, M. R., Alexander, C. M. O’D., Charnley, S. B.,  
26 Hallmann, C., Summons, R., Huang, Y., 2014. Chirality of meteoritic free and  
27 IOM-derived monocarboxylic acids and implications for prebiotic organic  
28 synthesis. *Geochim. Cosmochim. Acta* 131, 1-12.  
29  
30  
31 Aponte, J. C., Alexandre, M. R., Wang, Y., Brearley, A. J., Alexander, C. O’D., Huang  
32 Y., 2011. Effects of secondary alteration on the composition of free and IOM-  
33 derived monocarboxylic acids in carbonaceous chondrites. *Geochim. Cosmochim.*  
34 *Acta* 75, 2309-2323.  
35  
36  
37 Ash, R. D., Pillinger, C. T., 1995. Carbon, nitrogen and hydrogen in Saharan chondrites:  
38 The importance of weathering. *Meteoritics* 30, 85-92.  
39  
40  
41 Beck, P., Garenne, A., Quirico, E., Bonal, L., Montes-Hernandez, G., Moynier, F.,  
42 Schmitt, B., 2014. Transmission infrared spectra (2-25 µm) of carbonaceous  
43 chondrites (CI, CM, CV-CK, CR, C2 ungrouped): Mineralogy, water, and  
44 asteroidal processes. *Icarus* 229, 263-277.  
45  
46  
47 Berlin, J., Jones, R. H., Brearley, A. J., 2011. Fe–Mn systematics of type IIA chondrules  
48 in unequilibrated CO, CR, and ordinary chondrites. *Meteorit. Planet. Sci.* 46, 513–  
49 533.  
50  
51  
52 Bischoff, A., Palme, H., Ash, R. D., Clayton, R. N., Schultz, L., Herpers, U., Stoffler, D.,  
53 Grady, M. M., Pillinger, C. T., Spettel, B., Weber, H., Grund, T., Endreß, M.,  
54 Weber, D., 1993a. Paired Renazzo-type (CR) carbonaceous chondrites from the  
55 Sahara. *Geochim. Cosmochim. Acta* 57, 1587–1603.  
56  
57  
58 Bischoff, A., Palme, H., Schultz, L., Weber, D., Weber, H. W., Spettel, B., 1993b.  
59 ACFER 182 and paired samples, an iron-rich carbonaceous chondrite -  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 Similarities with ALH85085 and relationship to CR chondrites. *Geochim.*  
5 *Cosmochim. Acta* 57, 2631-2648.  
6  
7  
8 Bischoff, A., Scott, E. R. D., Metzler, K., Goodrich, C. A., 2006. Nature and Origins of  
9 Meteoritic Breccias. *In Meteorites and the Early Solar System II* (eds. D.S.  
10 Lauretta and H.Y. McSween, Jr.). University of Arizona Press. pp. 679-712.  
11  
12  
13 Bland, P. A., Zolensky, M. E., Benedix, G. K., Sephton, M. A., 2006. Weathering of  
14 chondritic meteorites. *In Meteorites and the Early Solar System II* (eds. D.S.  
15 Lauretta and H.Y. McSween, Jr.). University of Arizona Press. pp. 853- 867.  
16  
17  
18 Blum, J. D., Wasserburg, G. J., Hutcheon, I. D., Beckett, J. R., Stolper, E. M., 1988.  
19 “Domestic” origin of opaque assemblages in refractory inclusions in meteorites.  
20 *Nature* 331, 405–409.  
21  
22  
23 Blum, J. D., Wasserburg, G. J., Hutcheon, I. D., Beckett, J. R., Stolper, E. M., 1989.  
24 Origin of opaque assemblages in C3V meteorites—Implications for nebular and  
25 planetary processes. *Geochim. Cosmochim. Acta* 53, 543–556.  
26  
27  
28 Bonal L., Alexander C. M. O'D., Huss G. R., Nagashima K., Quirico, E., Beck, P., 2013.  
29 Hydrogen isotopic composition of the water in CR chondrites. *Geochim.*  
30 *Cosmochim. Acta* 106, 111-133.  
31  
32  
33 Bonner, W. A., Blair, N. E., Lemon, R. M., 1979. The radioracemization of amino acids  
34 by ionizing radiation: geochemical and cosmochemical implications. *Origin of*  
35 *Life* 9, 279-290.  
36  
37  
38 Botta, O., Glavin, D. P., Kminek, G., Bada, J. L., 2002. Relative amino acid  
39 concentrations as a signature for parent body processes of carbonaceous  
40 chondrites. *Origins of Life Evolution of Biosphere* 32, 143-163.  
41  
42  
43 Bradley, J. P., Dai, Z. R., 2004. Mechanism of formation of glass with embedded metal  
44 and sulfides. *Astrophys. J.* 617, 650-655.  
45  
46  
47 Braukmüller, N., Wombacher, F., Hezel, D. C.; Escoube, R., Münker, C., 2018. The  
48 chemical composition of carbonaceous chondrites: Implications for volatile  
49 element depletion, complementarity and alteration. *Geochim. Cosmochim. Acta*  
50 239, 17-48.  
51  
52  
53 Brearley, A. J., 1991. Mineralogical and chemical studies of matrix in the Adelaide  
54 meteorite, a unique carbonaceous chondrite with affinities to ALH A77307 (CO3)  
55 (abstract). 22<sup>nd</sup> Lunar Planet. Sci. Conf., # 133.  
56  
57  
58 Brearley, A. J., Burger, P. V., 2009. Mechanisms of aqueous alteration of type IIA  
59 chondrule glass in the CR chondrite EET 92105: Insights from FIB/TEM analysis  
60 (abstract). *Meteorit. Planet. Sci., Suppl.*, 5148.  
61  
62  
63  
64  
65

- 1  
2  
3  
4  
5  
6 Briani, G., Quirico, E., Gounelle, M., Paulhiac-Pison, M., Montagnac, G., Beck, P.,  
7 Orthous-Daunay, F.-R., Bonal, L., Jacquet, E., Kearsley, A., Russell, S. S., 2013.  
8 Short duration thermal metamorphism in CR chondrites. *Geochim. Cosmochim.*  
9 *Acta* 122, 267-279.
- 10  
11 Bringa, E. M., Kucheyev, S. O., Loeffler, M. J., Baragiola, R. A., Tielens, A. G. G. M.,  
12 Dai, Z. R., Graham, G., Bajt, S., Bradley, J. P., Dukes, C. A., Felter, T. E., Torres,  
13 D. F., Breugel, W. V., 2007. Energetic processing of interstellar silicate grains by  
14 cosmic rays. *Astrophys. J.* 662, 372.
- 15  
16  
17  
18 Browning, L. B., McSween, H. Y., Jr., Zolensky, M. E., 1996. Correlated alteration  
19 effects in CM carbonaceous chondrites. *Geochim. Cosmochim. Acta* 60, 2621-  
20 2633.
- 21  
22  
23 Brunner, C. E., Brearley, A. J., 2011. Microstructural investigation of the crystalline  
24 component of matrix in the pristine CR chondrite MET 00426: Implications for  
25 diversity in nebular dust (abstract). 42<sup>nd</sup> Lunar Planet. Sci. Conf. LPI Contribution  
26 No. 1608, p.1815.
- 27  
28  
29 Budde G., Kruijer T. S., Kleine T., 2018. Hf-W chronology of CR chondrites:  
30 Implications for the timescales of chondrule formation and the distribution of <sup>26</sup>Al  
31 in the solar nebula. *Geochim. Cosmochim. Acta*, 222, 284-304.
- 32  
33  
34 Bunch, T. E., Wittke, J. H., Irving, A. J., Rumble, D., 2013. Unique polymict breccia  
35 Northwest Africa 7531 composed of recrystallized LL clasts associated with CR  
36 metachondrite material: Evidence for highly equilibrated ordinary chondritic  
37 impactors onto the CR chondrite parent body (abstract). 44<sup>th</sup> Lunar Planet. Sci.  
38 Conf., LPI Contribution No. 1719, p.2214
- 39  
40  
41 Burger, P. V., 2005. Incipient aqueous alteration of meteorite parent bodies: Hydration,  
42 mobilization, precipitation and equilibration. M.S. thesis, University of New  
43 Mexico, Albuquerque, NM, USA.
- 44  
45  
46 Burton, A. S., Elsila, J. E., Callahan, M. P., Martin, M. G., Glavin, D. P., Johnson, N. M.,  
47 Dworkin, J. P., 2012a. A propensity for n- $\omega$ -amino acids in thermally altered  
48 Antarctic meteorites. *Meteorit. Planet. Sci.* 47, 374-386.
- 49  
50  
51 Burton, A. S., Stern, J. C., Elsila, J. E., Glavin, D. P., Dworkin, J. P., 2012b.  
52 Understanding prebiotic chemistry through the analysis of extraterrestrial amino  
53 acids and nucleobases in meteorites. *Chem. Society Rev.* 41, 5459-5472.
- 54  
55  
56 Busemann, H., Nguyen, A. N., Cody, G. D., Hoppe, P., Kilcoyne, A. L. D., Stroud, R.  
57 M., Zega, T. J., Nittler, L. R., 2009. Ultra-primitive interplanetary dust particles  
58 from the comet 26P/Grigg-Skjellerup dust stream collection. *Earth Planet. Sci.*  
59 *Lett.* 288, 44-57.
- 60  
61  
62  
63  
64  
65

- 1  
2  
3  
4  
5  
6 Busemann, H., Alexander, C. M. O'D., Nittler, L. R., 2007. Characterization of insoluble  
7 organic matter in primitive meteorites by microRaman spectroscopy. *Meteorit.*  
8 *Planet. Sci.* 42, 1387–1416.  
9
- 10 Busemann, H., Young A. F., Alexander, C. M. O'D., Hoppe P., Mukhopadhyay S.,  
11 Nittler L. R. 2006. Interstellar chemistry recorded in organic matter from  
12 primitive meteorites. *Science* 312, 727-730.  
13  
14
- 15 Callahan, M. P., Smith, K. E., Cleaves, II H. J., Ruzicka, J., Stern, J. C., Glavin, D. P.,  
16 House, C. H., Dworkin, J. P., 2011. Carbonaceous meteorites contain a wide  
17 range of extraterrestrial nucleobases. *Proc. Natl. Academy Sci. USA* 108, 13995-  
18 13998.  
19  
20
- 21 Campbell, A. J., Zanda, B., Perron, C., Meibom, A., Petaev, M. I., 2005. Origin and  
22 thermal history of Fe-Ni metal in primitive chondrites. *In* Chondrites and the  
23 protoplanetary disk, edited by Krot A. N., Scott E. R. D., and Reipurth B.  
24 *Astronomical Society of the Pacific Conference Series*, vol. 341. pp. 407–431.  
25  
26
- 27 Campbell, A. J., Humayun, M., Zanda, B., 2002. Partial condensation of volatile  
28 elements in Renazzo chondrules. *Geochim. Cosmochim. Acta* 66, A117.  
29  
30
- 31 Chizmadia, L. J., Brearley, A. J., 2008. Mineralogy, aqueous alteration, and primitive  
32 textural characteristics of fine-grained rims in the Y-791198 CM2 carbonaceous  
33 chondrite: TEM observations and comparison to ALHA81002. *Geochim.*  
34 *Cosmochim. Acta* 72(2), 602–625.  
35  
36
- 37 Clark, R. N., King, T. V. V., Klewja, M., Swayze, G. A., 1990. High spectral resolution  
38 reflectance spectroscopy of minerals. *J. Geophys. Res.* 95, 12653-12680.  
39  
40
- 41 Clayton, R. N., Mayeda, T. K., 1999. Oxygen isotope studies of carbonaceous chondrites.  
42 *Geochim. Cosmochim. Acta* 63, 2089-2104.  
43  
44
- 45 Cloutis, E. A., Pietrasz, V. B., Kiddell, C., Izawa, M. R. M., Vernazza, P., Burbine, T. H.,  
46 DeMeo, F., Tait, K. T., Bell III, J. F., Mann, P., Applin, D. M., Reddy, V., 2018a.  
47 Spectral reflectance “deconstruction” of the Murchison CM2 carbonaceous  
48 chondrite and implications for spectroscopic investigations of dark asteroids.  
49 *Icarus*, 305, 203-224.  
50  
51
- 52 Cloutis, E. A., Izawa, M. R. M., Beck, P., 2018b. *In* Primitive Meteorites and Asteroids,  
53 Chapter 4 – Reflectance Spectroscopy of Chondrites. *Physical, Chemical and*  
54 *Spectroscopic Observations Paving the Way to Exploration* (N. Abreu, editor);  
55 Elsevier, Amsterdam; <https://doi.org/10.1016/B978-0-12-813325-5.00004-5> pp.  
56 273-343.  
57  
58  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 Cloutis, E. A., Hudon, P., Hiroi, T., Gaffey, M. J., 2012. Spectral reflectance properties  
5 of carbonaceous chondrites: 3. CR chondrites. *Icarus* 217, 389-407.  
6  
7  
8 Cloutis, E. A., Hudon, P., Hiroi, T., Gaffey, M. J., Mann P., 2011. Spectral reflectance  
9 properties of carbonaceous chondrites: 2. CM chondrites. *Icarus* 216, 309-346.  
10  
11 Cloutis, E. A., Gaffey M. J., 1991. Pyroxene spectroscopy revisited: Spectral-  
12 compositional correlations and relationship to geothermometry. *J. Geophys. Res.*  
13 96, 22, 809-22,826.  
14  
15  
16 Cloutis, E. A., Gaffey, M. J., Jackowski, T. L., Reed, K. L., 1986. Calibrations of phase  
17 abundance, composition, and particle size distribution for olivine-orthopyroxene  
18 mixtures from reflectance spectra. *J. Geophys. Res.* 91, 11,641-11,653.  
19  
20  
21 Cobb, A. K., Pudritz, R. E., 2014. Nature's starships. I. Observed abundances and relative  
22 frequencies of amino acids in meteorites. *Astrophys. J.* 783, 140 (12pp).  
23  
24  
25 Cody G. D., Alexander C. M. O'D., 2005. NMR studies of chemical structural variation  
26 of insoluble organic matter from different carbonaceous chondrites groups.  
27 *Geochim. Cosmochim. Acta* 69, 1085–1097.  
28  
29  
30 Cody G. D., Alexander C. M. O'D., Yabuta H., Kilcoyne A. L. D., Araki T., Ade H.,  
31 Dera P., Fogel M., Militzer B., Mysen B. O., 2008. Organic thermometry for  
32 chondritic parent bodies. *Earth Planet. Sci. Letters* 272, 446-455.  
33  
34  
35 Connolly, Jr., H. C., Huss, G. R., 2010. Compositional evolution of the protoplanetary  
36 disk: Oxygen isotopes of type-II chondrules from CR2 chondrites. *Geochim.*  
37 *Cosmochim. Acta* 74, 2473–2483.  
38  
39  
40 Connolly, H. C., Huss, G. R., Nagashima, K., Weisberg, M. K., Ash, R. D., Ebel, D. S.,  
41 Schrader, D. L., Lauretta, D. S., 2008. Oxygen isotopes and the nature and origins  
42 of type-II chondrules in CR2 chondrites (abstract). 39<sup>th</sup> Lunar Planet. Sci. Conf.,  
43 LPI Contribution No. 1391, p.1675.  
44  
45  
46 Connolly Jr., H. C., Smith, C., Benedix, G., Folco, L., Righter, K., Zipfel, J., Yanaguchi,  
47 A., Chennaoui Aoudjehane, H., 2007. *The Meteoritical Bull.* No. 92, Meteorit.  
48 *Planet. Sci.* 42, 1647-1694.  
49  
50  
51 Connolly, Jr., H. C., Huss, G. R., Wassenburg, G. J., 2001. On the formation of Fe–Ni  
52 metal in Renazzo-like carbonaceous chondrites. *Geochim. Cosmochim. Acta* 65,  
53 4567–4588.  
54  
55  
56 Connolly, Jr., H. C., Hewins, R. H., Ash, R. D., Zanda, B., Lofgren, G. E., Bourot-  
57 Denise, J., 1994. Carbon and the formation of reduced chondrules. *Nature* 371,  
58 136–139.  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 Cooper, G., Rios, A. C., 2016. Enantiomer excesses of rare and common sugar  
5 derivatives in carbonaceous meteorites. *Proc. Natl. Academy Sci. USA* 113,  
6 E3322-E3331.  
7  
8  
9 Cordier, L. 1827. Rapport fait 'a l'Academie des Sciences, sur une pierre meteorique  
10 tombee pres de Ferrare en 1824. *Annales de Chimie et de Physique*, 34, 132-139.  
11  
12  
13 Corrigan, C. C., Zolensky, M. E., Dahl, J., Long, M., Weir, J., Sapp, C., 1997. The  
14 porosity and permeability of chondritic meteorites and interplanetary dust  
15 particles. *Meteorit. Planet. Sci.* 32, 509-515.  
16  
17  
18 \*Das, J. P., Meshik, A. P., Pravdivtseva, O. V., Hohenberg, C. M., 2011. A first test of a  
19 new analyte. 193 laser ablation system: in-situ helium, neon and argon  
20 compositions of chondrule zones and surrounding matrix in NWA 801 CR2  
21 chondrite (abstract). 42<sup>nd</sup> Lunar Planet. Sci. Conf., LPI Contribution No. 1608, p.  
22 2238.  
23  
24  
25 Daulton, T. L., Bernatowicz, T. J., Lewis, R. S., Messenger, S., Stadermann, F. J., Amari,  
26 S., 2003. Polytype distribution of circumstellar silicon carbide: Microstructural  
27 characterization by transmission electron microscopy. *Geochim. Cosmochim.*  
28 *Acta* 67, 4743-4767.  
29  
30  
31 \*Davidson, J., Alexander, C. M. O. D., Schrader, D. L., Nittler, L. R., Bowen, R., 2015.  
32 Miller Range 090657: A very pristine Renazzo-like (CR) carbonaceous chondrite  
33 (abstract). 46<sup>th</sup> Lunar Planet. Sci. Conf., #1603.  
34  
35  
36 Davidson, J., Busemann, H., Nittler, L. R., Alexander, C. M. O'D., Orthous-Daunay, F.-  
37 R., Franchi, I. A., Hoppe, P., 2014a. Abundances of presolar silicon carbide grains  
38 in primitive meteorites determined by NanoSIMS. *Geochim. Cosmochim. Acta*  
39 139, 248-266.  
40  
41  
42 Davidson, J., Schrader, D.L., Alexander, C.M.O.D., Lauretta, D.S., Busemann, H.,  
43 Franchi, I.A., Greenwood, R.C., Connolly Jr., H.C., Domanik, K.J., Verchovsky,  
44 A., 2014b. Petrography, stable isotope compositions, microRaman spectroscopy,  
45 and presolar components of Roberts Massif 04133: A reduced CV3 carbonaceous  
46 chondrite. *Meteorit. Planet. Sci.* 49, 2133-2151.  
47  
48  
49 de Leuw, S., Rubin, A., Wasson J., 2010. Carbonates in CM chondrites: Complex  
50 formational histories and comparison to carbonates in CI chondrites. *Meteorit.*  
51 *Planet. Sci.* 45, 513–530.  
52  
53  
54 De Marcellus, P., Meinert, C., Nuevo, M., Filippi, J.-J., Danger, G., Deboffle, D., Nahon,  
55 L., d'Hendecourt, L. Le S., Meierhenrich, U. J., 2011. Non-racemic amino acid  
56 production by ultraviolet irradiation of achiral interstellar ice analogs with  
57 circularly polarized light. *Astrophys. J. Lett.* 727, L27-L32.  
58  
59  
60  
61  
62  
63  
64  
65



- 1  
2  
3  
4 Ebel, D. S., Weisberg, M. K., Hertz, J., Campbell, A. J., 2008. Shape, metal abundance,  
5 chemistry, and origin of chondrules in the Renazzo (CR) chondrite. *Meteorit.*  
6 *Planet. Sci.* 43, 1725–1740.  
7  
8  
9 Ehrenfreund, P., Glavin, D. P., Botta, O., Cooper, G., Bada, J. L., 2001. Extraterrestrial  
10 amino acids in Orgueil and Ivuna: Tracing the parent body of CI type  
11 carbonaceous chondrites. *Proc. Natl. Academy Sci. USA* 98, 2138-2141.  
12  
13  
14 Elsilá, J. E., Charnley, S. B., Burton, A. S., Glavin, D. P., Dworkin, J. P., 2012.  
15 Compound-specific carbon, nitrogen, and hydrogen isotopic ratios for amino  
16 acids in CM and CR chondrites and their use in evaluating potential formation  
17 pathways. *Meteorit. Planet. Sci.* 47, 1517-1536.  
18  
19  
20 Elsilá, J. E., Aponte, J. C., Blackmond, D. G., Burton, A. S., Dworkin, J. P., Glavin, D.  
21 P., 2016. Meteoritic amino acids: Diversity in compositions reflects parent body  
22 histories. *ACS Cent. Sci.* 2, 370-379.  
23  
24  
25 Endreß, M., Keil, K., Bischoff, A., Spettel, B., Clayton, R. N., Mayeda, T. K., 1994.  
26 Origin of dark clasts in the ACFER 059/El Djouf 001 CR2 chondrite. *Meteoritics*  
27 29, 26-40.  
28  
29  
30 Ferrarotti, A.S., Gail, H.-P., 2001. Mineral formation in stellar winds II: Effects of Mg/Si  
31 abundance variations on dust composition in AGB stars. *Astron. Astrophys.* 371,  
32 133-151.  
33  
34  
35 Fleet, M. E., Herzberg, C. T., Bancroft, G. M., Aldridge, L. P., 1978. Omphacite studies;  
36 I, The P2/n- > C2/c transformation. *Amer. Mineral.* 63, 1100–1106.  
37  
38  
39 Floss, C., Haenecour, P., 2016a. Presolar silicate grains: Abundances, isotopic and  
40 elemental compositions, and the effects of secondary processing. *Geochem. J.* 50,  
41 3-25.  
42  
43  
44 Floss, C., Haenecour, P., 2016b. Meteorite Hills (MET) 00526: An unequilibrated  
45 ordinary chondrite with high presolar grain abundances (abstract). 47<sup>th</sup> Lunar  
46 *Planet. Sci. Conf.*, #1030.  
47  
48  
49 Floss, C., Le Guillou, C., Brearley, A., 2014. Coordinated NanoSIMS and FIB-TEM  
50 analyses of organic matter and associated matrix materials in CR3 chondrites.  
51 *Geochim. Cosmochim. Acta* 139, 1-25.  
52  
53  
54 Floss, C., Stadermann, F. J., Kearsley, A. T., Burchell, M. J., Ong, W. J., 2013. The  
55 abundance of presolar grains in comet 81P/Wild 2. *Astrophys. J.* 763, 140-150.  
56  
57  
58 Floss, C., Stadermann, F. J., 2012. Presolar silicate and oxide abundances and  
59 compositions in the ungrouped carbonaceous chondrite Adelaide and the K  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 chondrite Kakangari: The effects of secondary processing. *Meteorit. Planet. Sci.*  
5 47, 992-1009.  
6  
7  
8 Floss, C., Stadermann, F. J., 2009b. High abundances of circumstellar and interstellar C-  
9 anomalous phases in the primitive CR3 chondrites QUE 9917 and MET 00426.  
10 *Astrophys. J.* 697, 1242-1255.  
11  
12 Floss, C., Stadermann, F. J., 2009a. Auger nanoprobe analysis of presolar ferromagnesian  
13 silicate grains from primitive CR chondrites QUE 99177 and MET 00426.  
14 *Geochim. Cosmochim. Acta* 73, 2415-2440.  
15  
16  
17 Floss, C., Stadermann, F. J., Bradley, J. P., Dai, Z. R., Bajt, S., Graham, G., Lea, A. S.,  
18 2006. Identification of isotopically primitive interplanetary dust particles: A  
19 NanoSIMS isotopic imaging study. *Geochim. Cosmochim. Acta* 70, 2371-2399.  
20  
21  
22 Floss, C., Stadermann, F. J., 2005. Presolar (circumstellar and interstellar) phases in  
23 Renazzo: The effects of parent body processing (abstract). 36<sup>th</sup> Lunar Planet. Sci.  
24 Conf., #1390.  
25  
26  
27 Flynn, G. J., Consolmagno, G. J., Brown, P., Macke, R. J., 2017. Physical properties of  
28 the stone meteorites: Implications for the properties of their parent bodies. *Chem.*  
29 *Erde* 78, 269-298.  
30  
31  
32 Friedrich, J. M., Weisberg, M. K., Ebel, D. S., Biltz, A. E., Corbett, B. M., Iotzov, I. V.,  
33 Khan, W. S., Wolman, M. D., 2015. Chondrule size and related physical  
34 properties: A compilation and evaluation of current data across all meteorite  
35 groups. *Chem. Erde* 75, 419-443.  
36  
37  
38 Gardner-Vandy, K. G., Lauretta, D. S., Greenwood, R. C., McCoy, T. J., Killgore, M.,  
39 Franchi, I. A., 2012. The Tafassasset primitive achondrite: Insights into initial  
40 stages of planetary differentiation. *Geochim. Cosmochim. Acta* 85, 142-159.  
41  
42  
43 Garenne, A., Beck, P., Montes-Hernandez, G., Chiriac, R., Toche, F., Quirico, E., Bonal,  
44 L., Schmitt, B., 2014. The abundance and stability of “water” in type 1 and 2  
45 carbonaceous chondrites (CI, CM and CR). *Geochim. Cosmochim. Acta* 137, 93-  
46 112.  
47  
48  
49 Gilmour, I., 2005. Structural and isotopic analysis of organic matter in carbonaceous  
50 chondrites. *In* *Treatise on Geochemistry; Meteorites, Comets and Planets* (ed.  
51 Davis, A. M.), pp. 269-290. Elsevier, Amsterdam.  
52  
53  
54 Gilmour, I., Pillinger, C. T., 1994. Isotopic compositions of individual polycyclic  
55 aromatic hydrocarbons from the Murchison meteorite. *Mon. Not. R. Astron. Soc.*  
56 269, 235-240.  
57  
58  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 Glavin, D. P., Alexander, C. M. O'D., Aponte, J. C., Dworkin, J. P., Elsila, J. E., Yabuta,  
5 H., 2018. The origin and evolution of organic matter in carbonaceous chondrites  
6 and links to their parent bodies. *In* Primitive Meteorites and Asteroids, 1<sup>st</sup> Edition,  
7 N. Abreu (Ed.), pp 205-271, Elsevier, Amsterdam, Netherlands.  
8  
9  
10 Glavin, D. P., Callahan, M. P., Dworkin, J. P., Elsila, J. E., 2011. The effects of parent  
11 body processes on amino acids in carbonaceous chondrites. *Meteorit. Planet. Sci.*  
12 45, 1948-1972.  
13  
14  
15 Glavin, D. P., Dworkin, J. P., 2009. Enrichment of the amino acid L-isovaline by aqueous  
16 alteration on CI and CM meteorite parent bodies. *Proc. Natl. Academy Sci. USA*  
17 106, 5487-5492.  
18  
19  
20 Gooding, J. L., 1986. Clay-mineraloid weathering products in Antarctic meteorites.  
21 *Geochim. Cosmochim. Acta* 50, 2215-2223.  
22  
23  
24 Gooding, J. L., 1981. Mineralogical aspects of terrestrial weathering effects in chondrites  
25 from Allan Hills, Antarctica. *Proc. 12<sup>th</sup> Lunar Planet. Sci. Conf.*, 1105–1122.  
26  
27 \*Grady, M. M., Ash, R. D., Pillinger, C. T. EET 87770: A light element stable isotope  
28 study of a new Renazzo-like carbonaceous chondrite. (abstract). 22<sup>nd</sup> Lunar  
29 Planet. Sci. Conf., # 471.  
30  
31  
32 Greshake, A., 1997. The primitive matrix components of the unique carbonaceous  
33 chondrite ACFER 094: A TEM study. *Geochim. Cosmochim. Acta* 61, 437-452.  
34  
35  
36 Grossman, J. N., Wasson, J. T., 1985. The origin and history of the metal and sulfide  
37 components of chondrules. *Geochim. Cosmochim. Acta* 49, 925–939.  
38  
39  
40 Grossman, L., Olsen, E., 1974. Origin of the high-temperature fraction of C2 chondrites.  
41 *Geochim. Cosmochim. Acta* 38, 175-187.  
42  
43 \*Haenecour, P., Floss, C., Zega, T. J., Croat, T. K., Wang, A., Jolliff, B. L., Carpenter, P.,  
44 2018. Presolar silicates in the matrix and fine-grained rims around chondrules in  
45 primitive CO3.0 chondrites: Evidence for pre-accretionary aqueous alteration of  
46 the rims in the solar nebula. *Geochim. Cosmochim. Acta* 221, 379-405.  
47  
48  
49 Hanowski, N. P., Brearley, A. J., 2001. Aqueous alteration of chondrules in the CM  
50 carbonaceous chondrite Allan Hills 81002: Implications for parent body  
51 alteration. *Geochim. Cosmochim. Acta* 65, 495-518.  
52  
53  
54 Haramura, H., Kushiro I., Yanai, K., 1983. Chemical compositions of Antarctic  
55 meteorites: I. *Proc. Eighth Symp. Antarct. Meteor. NIPR.* 109-121.  
56  
57  
58 Hashiguchi, M., Yurimoto, H., 2016. Hydrogen isotopic compositions and chemical  
59 structures of organic materials in Northwest Africa 801 CR2 chondrite:  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 Implications for metamorphism histories of extraterrestrial organic materials  
5 (abstract). 47<sup>th</sup> Lunar Planet. Sci. Conf. No. 1903, p. 1216.  
6  
7
- 8 Hashiguchi, M., Kobayashi, S., Yurimoto, H. 2013. In situ observation of D-rich  
9 carbonaceous globules embedded in NWA 801 CR2 chondrite. *Geochim.*  
10 *Cosmochim. Acta* 122, 306-323.  
11
- 12 Harju, E. R., Rubin, A. E., Ahn, I., Choi, B.-G., Ziegler, K., Wasson, J. T., 2014.  
13 Progressive aqueous alteration of CR carbonaceous chondrites. *Geochim.*  
14 *Cosmochim. Acta* 139, 267-292.  
15  
16
- 17 Hertkorn, K., Harir, M., Schmitt-Kopplin, Ph., 2015. Nontarget analysis of Murchison  
18 soluble organic matter by high-field NMR spectroscopy and FTICR mass  
19 spectrometry. *Magnetic Resonance Chem.* 53, 754-768.  
20  
21
- 22 Hewins, R. H., Bourot-Denise, M., Zanda, B., Leroux, H., Barrat, J.-A., Humayan, M.,  
23 Göpel, G., Greenwood, R.C., Franchi, I.A., Pont, S., Lorand, J.-P., Cournède,  
24 Gataceca, J., Rochette, P., Kuga, M., Marrocchi, Y., Marty, B., 2014. The Paris  
25 meteorite, the least altered CM chondrite so far. *Geochim. Cosmochim. Acta* 124,  
26 190–222.  
27  
28
- 29 Hewins, R. H., Connolly, Jr., H. C., Lofgren, G. E., Libourel, G., 2005. Experimental  
30 constraints on chondrule formation. *In Chondrites and the Protoplanetary Disk*  
31 (eds. A. N. Krot, E. R. D. Scott and B. Reipurth), pp. 286–316. ASP Conference  
32 Series 341. Astronomical Society of the Pacific, San Francisco.  
33  
34
- 35 Hewins R. H., Yu Y., Zanda B., Bourot-Denise M., 1997. Do nebular fractionations,  
36 evaporative losses, or both, influence chondrule compositions? 21<sup>st</sup> Symp.  
37 Antarctic Meteorites, NIPR, 275-298.  
38  
39
- 40 Hiyagon, H., Sugiura, N., Kita, N. T., Kimura, M., Morishita, Y., Takehana, Y., 2016.  
41 Origin of the eclogitic clasts with graphite-bearing and graphite-free lithologies in  
42 the Northwest Africa 801 (CR2) chondrite: Possible origin from a Moon-sized  
43 planetary body inferred from chemistry, oxygen isotopes and REE abundances.  
44 *Geochim. Cosmochim. Acta* 186, 32–48.  
45  
46
- 47 Howard, K. T., Alexander, C. M. O'D., Schrader, D. L., Dyl, K. A., 2015a. Classification  
48 of hydrous meteorites (CR, CM and C2 ungrouped) by phyllosilicate fraction:  
49 PSD-XRD modal mineralogy and planetesimal environments. *Geochim.*  
50 *Cosmochim. Acta* 149, 206-222.  
51  
52
- 53 Howard, K. T., Davidson, J., Alexander, C. M. O'D., Abreu, N. M., 2015b. The glue that  
54 holds worlds together: Abundance, origin and significance of amorphous Fe, Mg  
55 silicate in carbonaceous chondrites (abstract). 46<sup>th</sup> Lunar Planet. Sci. Conf. No.  
56 1832, p .2244.  
57  
58  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 Humayun, M., Campbell, A. J., Zanda, B., Bourot-Denise, M., 2002. Formation of  
5 Renazzo chondrule metal inferred from siderophile elements (abstract). 33<sup>rd</sup> Lunar  
6 Planet. Sci. Conf., No. 1965.  
7  
8  
9 Huss, G. R., Meshik, A. P., Smith, J. B., Hohenberg, C. M., 2003. Presolar diamond,  
10 silicon carbide, and graphite in carbonaceous chondrites: Implications for thermal  
11 processing in the solar nebula. *Geochim. Cosmochim. Acta* 67, 4823-4848.  
12  
13  
14 \*Huss, G. R., Weisberg, M. K., Wasserburg, G. J. 1996. Magnesium and silicon isotopes  
15 in layered chondrules from EL Djouf (CR2). (abstract). In Abstracts of the Lunar  
16 Planet. Sci. Conf. 27, 575-576.  
17  
18  
19 Hutcheon, I. D., Marhas, K. K., Krot, A. N., Goswami, J. N., Jones, R. H., 2009. <sup>26</sup>Al in  
20 plagioclase-rich chondrules in carbonaceous chondrites: Evidence for an extended  
21 duration of chondrule formation. *Geochim. Cosmochim. Acta* 73, 5080-5099.  
22  
23  
24 Hutchison, R., 2004. *Meteorites*. pp. 520. Cambridge University Press, Cambridge, UK.  
25  
26 Ibrahim, E.-M.I., 2012. The elastic properties of carbonaceous chondrites M.Sc. Thesis.  
27 University of Calgary, Alberta, Canada.  
28  
29  
30 Ichikawa, O., Ikeda, Y., 1995. Petrology of the Yamato-8449 CR chondrite. *Proc. NIPR*  
31 *Symp. Antarctic Meteorites, National Institute of Polar Research* 8, 63-78.  
32  
33  
34 Jacquet, E., Paulhiac-Pison, M., Alard, O., Kearsley, A. T., Gounelle, M., 2013. Trace  
35 element geochemistry of CR chondrite metal. *Meteorit. Planet. Sci.* 48, 1981-  
36 1999.  
37  
38  
39 Jilly-Rehak, C. E., Huss, G. R., Nagashima, K., Schrader, D. L., 2018. Low-temperature  
40 aqueous alteration on the CR chondrite parent body: Implications from in situ  
41 oxygen-isotope analyses. *Geochim. Cosmochim. Acta* 222, 230-252.  
42  
43  
44 Jilly-Rehak, C. E., Huss, G. R., Nagashima, K., 2017. <sup>53</sup>Mn-<sup>53</sup>Cr radiometric dating of  
45 secondary carbonates in CR chondrites: Timescales for parent body aqueous  
46 alteration. *Geochim. Cosmochim. Acta* 201, 224-244.  
47  
48  
49 Jones, A. P., Nuth, J. A., III, 2011. Dust destruction in the ISM: a re-evaluation of dust  
50 lifetimes. *Astron. Astrophys.* 530, A44-A55.  
51  
52  
53 Jones, A. P., Tielens, A. G. G. M., Hollenbach, D. J., McKee, C. F., 1994. Grain  
54 destruction in shocks in the interstellar medium. *Astrophys. J.* 433, 797-810.  
55  
56  
57 Jones, R. H., Lofgren, G. E., 1993. A comparison of FeO-rich, porphyritic olivine  
58 chondrules in unequilibrated chondrites and experimental analogues. *Meteoritics*  
59 28, 213-221.  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 \*Jull et al., 1991. Jull A. J. T., Wlotzka F., and Donahue D. J. (1991) Terrestrial ages and  
5 petrologic description of Roosevelt County meteorites (abstract). *In Abstracts of*  
6 *the Lunar Planet. Sci. Conf.* 22, 667–668.  
7  
8  
9 Kallemeyn, G. W., Rubin, A. E., Wasson, J. T., 1994. The compositional classification of  
10 chondrites: VI. The CR carbonaceous chondrite group. *Geochim. Cosmochim.*  
11 *Acta* 58, 2873-2888.  
12  
13  
14 Kallemeyn, G. W., Wasson, J. T., 1982. The compositional classification of chondrites-  
15 III. Ungrouped carbonaceous chondrules. *Geochim. Cosmochim. Acta* 46, 2217-  
16 2228.  
17  
18  
19 Keller, L. P., McKeegan, K. D., Sharp, Z. D., 2012. The oxygen isotopic composition of  
20 MIL 090001: A CR2 chondrite with abundant refractory inclusions (abstract). 43<sup>rd</sup>  
21 Lunar Planet. Sci. Conf. LPI Contribution No. 1659, id.2065.  
22  
23  
24 Keller, L. P., 2011. Mineralogy and petrography of MIL 090001, a highly altered CV  
25 chondrite from the reduced sub-group (abstract). 42<sup>nd</sup> Lunar Planet. Sci. Conf.,  
26 LPI Contrib. No. 1608, p. 2409.  
27  
28  
29 Keller, L. P., Messenger, S., 2011. On the origins of GEMS grains. *Geochim.*  
30 *Cosmochim. Acta* 75, 5336-5365.  
31  
32  
33 Kemper, F., Waters, L. B. F. M., de Koter, A. Tielens, A. G. G. M., 2001. Crystallinity  
34 versus mass-loss rate in Asymptotic Giant Branch stars. *Astron. Astrophys.* 369,  
35 132-141.  
36  
37  
38 Kerridge J. F., Mackay A. L., Boynton W. V., 1979. Magnetite in C1 carbonaceous  
39 meteorites: Origin by aqueous activity on a planetesimal surface. *Science* 205,  
40 395-397.  
41  
42  
43 Kerridge, J. F., 1985. Carbon, hydrogen and nitrogen in carbonaceous chondrites:  
44 Abundances and isotopic compositions in bulk samples. *Geochimica et*  
45 *Cosmochimica Acta* 49, 1707-1714.  
46  
47  
48 Kerridge, J. F., 1999. Formation and processing of organics in the early solar system.  
49 *Space Science Reviews* 90, 275-288.  
50  
51  
52 Kiddell, C. B., Cloutis E. A., Dagdick B. R., Stromberg J. M., Applin D. M., Mann J. P.,  
53 2018. Spectral reflectance of powder coatings on carbonaceous chondrite slabs:  
54 Implications for asteroid regolith observations. *J.Geophys. Research – Planets*  
55 123, 2803-2840; doi: <https://doi.org/10.1029/2018JE005600>.  
56  
57  
58 King, T. V. V., Ridley, I. W., 1987. Relation of the spectroscopic reflectance of olivine to  
59 mineral chemistry and some remote sensing implications. *J. Geophys. Res.* 92,  
60 11,457-11,469.  
61  
62  
63  
64  
65

- 1  
2  
3  
4  
5  
6 Kimura, M., Sugiura, N., Mikouchi, T., Hirajima, T., Hiyagon, H., Takehana, Y., 2013.  
7 Eclogitic clasts with omphacite and pyrope-rich garnet in the NWA 801 CR2  
8 chondrite. *Amer. Mineral.* 98, 387-393.  
9
- 10 Kimura, M., Grossman, J. N., Weisberg M. K., 2011. Fe-Ni metal and sulfide minerals in  
11 CM chondrites: An indicator for thermal history. *Meteorit. Planet. Sci.* 46, 431–  
12 442.  
13  
14
- 15 Kimura, M., Grossman, J. N., Weisberg M. K., 2008. Fe-Ni metal in primitive chondrites:  
16 Indicators of classification and metamorphic conditions for ordinary and CO  
17 chondrites. *Meteorit. Planet. Sci.* 43, 1161–1177.  
18  
19
- 20 Koch, I., Floss, C., 2017. Abundances and compositions of presolar grains in CR2  
21 chondrite EET 92042 (abstract). 48<sup>th</sup> Lunar Planet. Sci. Conf., #2984.  
22  
23
- 24 Kong, P., Palme, H., 1999. Compositional and genetic relationship between chondrules,  
25 chondrule rims, metal and matrix in the Renazzo chondrite.  
26 *Geochim. Cosmochim. Acta* 63, 3673–3682.  
27  
28
- 29 Kong, P., Ebihara, M., Palme, H., 1999. Distribution of siderophile elements in CR  
30 chondrites: Evidence for evaporation and recondensation during chondrule  
31 formation. *Geochim. Cosmochim. Acta* 63, 2637–2652.  
32  
33
- 34 Krot, A. N., Nagashima, K., van Kooten, E. M. M., Bizzarro, M., 2017. High-temperature  
35 rims around calcium-aluminum-rich inclusions from the CR, CB and CH  
36 carbonaceous chondrites. *Geochim. Cosmochim. Acta* 201, 155-184.  
37  
38
- 39 Krot, A. N., Libourel, G., Goodrich, C. A., Petaev, M. I., 2004. Silica-rich igneous rims  
40 around magnesian chondrules in CR carbonaceous chondrites: evidence for  
41 fractional condensation during chondrule formation. *Meteorit. Planet. Sci.* 39,  
42 1931–1955.  
43  
44
- 45 Krot, A. N., Keil, K., 2002. Anorthite-rich chondrules in CR and CH carbonaceous  
46 chondrites: genetic link between calcium–aluminum-rich inclusions and  
47 ferromagnesian chondrules. *Meteorit. Planet. Sci.* 37, 91–111.  
48  
49
- 50 Krot, A. N., Yurimoto, H., McKeegan, K. D., Leshin, L., Chaussidon, M., Libourel, G.,  
51 Yoshitake, M., Huss, G. R., Guan, Y., Zanda, B., 2006a. Oxygen isotopic  
52 compositions of chondrules: implications for evolution of oxygen isotopic  
53 reservoirs in the inner solar nebula. *Chemie der Erde - Geochemistry* 66, 249–  
54 276.  
55  
56
- 57 Krot, A. N., Libourel, G., Chaussidon, M., 2006b. Oxygen isotope compositions of  
58 chondrules in CR chondrites. *Geochim. Cosmochim. Acta* 70, 767–779.  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 Krot, A. N., McKeegan, K. D., Leshin, L. A., MacPherson, G. J., Scott, E. R. D., 2002.  
5 Existence of an  $^{16}\text{O}$ -rich gaseous reservoir in the solar nebula. *Science* 295, 1051–  
6 1054.  
7  
8  
9 Kung, C. -C., Clayton, R. N., 1978. Nitrogen abundances and isotopic compositions in  
10 stony meteorites. *Earth Planet. Sci. Lett.* 38, 421-435.  
11  
12  
13 Lee, M. R., Bland, P. A., 2004. Mechanisms of weathering of meteorites recovered from  
14 hot and cold deserts and the formation of phyllosilicates. *Geochim. Cosmochim.*  
15 *Acta* 68, 893–916.  
16  
17  
18 Lee, M. S., Rubin, A. E., Wasson, J. T., 1992. Origin of metallic Fe-Ni in Renazzo and  
19 related chondrites. *Geochim. Cosmochim. Acta* 56, 2521-2533.  
20  
21  
22 Le Guillou, C., Changela, H. G., Brearley, A. J., 2015a. Widespread oxidized and  
23 hydrated amorphous silicates in CR chondrite matrices: Implications for alteration  
24 conditions and H<sub>2</sub> degassing of asteroids. *Earth Planet. Sci. Lett.* 420, 162-173.  
25  
26  
27 Le Guillou, C., Dohmen, R., Rogalla, D., Müller, T., Vollmer, C., Becker, H.-W. 2015b.  
28 New experimental approach to study aqueous alteration of amorphous silicates at  
29 low reaction rates. *Chemical Geology* 412, 179-192.  
30  
31  
32 Le Guillou, C., Brearley, A. J., 2014. Relationships between organics, water and early  
33 stages of aqueous alteration in the pristine CR3.0 chondrite MET 00426.  
34 *Geochim. Cosmochim. Acta* 131, 344-367.  
35  
36  
37 Le Guillou, C., Bernard, S., Brearley, A. J., Remusat, L., 2014. Evolution of organic  
38 matter in Orgueil, Murchison and Renazzo during parent body aqueous alteration:  
39 In situ investigations. *Geochim. Cosmochim. Acta* 131, 368-392.  
40  
41  
42 Leitner, J., Hoppe, P., Zipfel, J., 2016a. The presolar grain inventory of the CR chondrite  
43 Elephant Moraine 92161 (abstract). 47<sup>th</sup> Lunar Planet. Sci. Conf., #1873.  
44  
45  
46 Leitner, J., Vollmer, C., Floss, C., Zipfel, J., Hoppe, P., 2016b. Ancient stardust in fine-  
47 grained chondrule dust rims from carbonaceous chondrites. *Earth Planet. Sci.*  
48 *Lett.* 434, 117-128.  
49  
50  
51 Leitner, J., Hoppe, P., and Zipfel, J., 2015. Distribution and abundance of presolar silicate  
52 and oxide stardust in CR chondrites (abstract). 46<sup>th</sup> Lunar Planet. Sci. Conf.,  
53 #1874.  
54  
55  
56 Leitner, J., Vollmer, C., Hoppe, P., Zipfel, J., 2012a. Characterization of presolar material  
57 in the CR chondrite Northwest Africa 852. *Astrophys. J.* 745, 38.  
58  
59  
60  
61  
62  
63  
64  
65



- 1  
2  
3  
4 Leitner, J., Hoppe, P., Zipfel, J., 2012b. The stardust inventories of Graves Nunataks  
5 95229 and Renazzo: Implications for the distribution of presolar grains in CR  
6 chondrites (abstract). 43<sup>rd</sup> Lunar Planet. Sci. Conf., #1835.  
7  
8  
9 Leitner, J., Hoppe, P., Zipfel, J., 2011. The stardust inventory of the CR chondrites GRA  
10 95229 and GRA 06100 assessed by NanoSIMS (abstract). 42<sup>nd</sup> Lunar Planet. Sci.  
11 Conf., #1713.  
12  
13  
14 Lerner, N. R., Peterson, E., Chang, S., 1993. The Strecker synthesis as a source of amino  
15 acids in carbonaceous chondrites–Deuterium retention during synthesis. *Geochim.*  
16 *Cosmochim. Acta* 57, 4713-4723.  
17  
18  
19 Losiak, A., Velbel, M. A., 2011. Evaporite formation during weathering of Antarctic  
20 meteorites-A weathering census analysis based on the ANSMET database.  
21 *Meteorit. Planet. Sci.* 46, 443-458.  
22  
23  
24 Macke, R., Consolmagno, G., Britt, D., 2011. Density, porosity, and magnetic  
25 susceptibility of carbonaceous chondrites. *Meteorit. Planet. Sci.* 46, 1842-1862.  
26  
27  
28 MacPherson, G. J., 2014. Calcium-aluminum-rich inclusions in chondritic meteorites. *In*  
29 *Volume 1 of Treatise on Geochemistry (Second Edition), Meteorites and*  
30 *Cosmochemical Processes*, Elsevier (Eds: Andrew M. Davis), p.139-179.  
31  
32  
33 MacPherson, G. J., Mittlefehldt, D. W., Lipschutz, M. E., Clayton, R. N., Bullock, E. S.,  
34 Ivanov, A. V., Mayeda, T. K., Wang, M.-S., 2009. The Kaidun chondrite  
35 breccias: Petrology, oxygen isotopes, and trace element abundances. *Geochim.*  
36 *Cosmochim. Acta* 73, 5493-5511.  
37  
38  
39 Martins, Z., Alexander, C. M. O. D., Orzechowska, G. E., Fogel, M. L., Ehrenfreund, P.,  
40 2007. Indigenous amino acids in primitive CR meteorites. *Meteorit. Planet. Sci.*  
41 42, 2125-2136.  
42  
43  
44 Mason, B., 1971. The carbonaceous chondrites--a selective review. *Meteoritics* 6, 59-70.  
45  
46  
47 Mason, B., 1962. The classification of chondritic meteorites. *Amer. Museum Novitates*  
48 2085, 1-20.  
49  
50  
51 Mason, B., 1960a. The origin of meteorites. *J. Geophys. Res.* 65, 2965-2970.  
52  
53  
54 Mason, B., 1960b. Origin of chondrules and chondritic meteorites. *Nature* 186, 230-231.  
55  
56  
57  
58  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 \*Matsuda, S., Nakashima, D., Iio, H., Bajo, K., Nagao, K., 2009. Laser Microprobe  
5 Noble Gas Analysis of Chondrules in the NWA 801 CR2 Chondrite (abstract).  
6 40<sup>th</sup> Lunar Planet. Sci. Conf. LPI Contribution id.1628.  
7  
8  
9 Matthews C. N., Minard, R. D., 2008. Hydrogen cyanide polymers connect  
10 cosmochemistry and biochemistry. Proc. IAU Symp. Series 4, 453-457.  
11  
12 McBride, K., McCoy, T., Welzenbach, L., 2001a. MET 00426. Antarctic Meteorite  
13 Newslett. 24, 18.  
14  
15 McBride, K., McCoy, T., Welzenbach, L., 2001b. QUE 99177. Antarctic Meteorite  
16 Newslett. 24, 14.  
17  
18  
19  
20 McNamara, D. D., 2012. Omphacite—a mineral under pressure! *Geology Today* 28, 71–  
21 75.  
22  
23  
24 McSween, H. Y., Jr., 1979a. Are carbonaceous chondrites primitive or processed? A  
25 review. *Rev. Geophys. Space Phys.* 17, 1059-1078.  
26  
27  
28 McSween, H. Y., Jr., 1979b. Alteration in CM carbonaceous chondrites inferred from  
29 modal and chemical variations in matrix. *Geochim. Cosmochim. Acta* 43, 1761-  
30 1770.  
31  
32  
33 McSween, H. Y., Jr., 1977a. Petrographic variations among carbonaceous chondrites of  
34 the Vigarano type. *Geochim. Cosmochim. Acta* 41, 1777-1790.  
35  
36  
37 McSween, H. Y., Jr., 1977b. On the nature and origin of isolated olivine grains in  
38 carbonaceous chondrites. *Geochim. Cosmochim. Acta*, 41, 411-418.  
39  
40  
41 Meinert, C., Jones, N. C., Hoffmann, S. V., Meierhenrich, U. J., 2016. Anisotropy  
42 spectroscopy of chiral alcohols, amines, and monocarboxylic acids: Implications  
43 for the analyses of extraterrestrial samples. *J. Photochem. Photobiol. A* 331, 130-  
44 137.  
45  
46  
47 Meinert, C., Hoffmann, S. V., Cassam-Chenai, P., Evans, A. C., Giri, C., Nahon, L.,  
48 Meierhenrich, U. J., 2014. Photonenergy-controlled symmetry breaking with  
49 circularly polarized light. *Angew. Chem. Int. Ed.* 53, 210-214.  
50  
51  
52 Meinert, C., Bredehöft, J. H., Filippi, J. J., Baraud, Y., Nahon, L., Wien, F., Jones, N. C.,  
53 Hoffmann, S. V., Meierhenrich, U. J., 2012. Anisotropy spectra of amino acids.  
54 *Angew. Chem., Int. Ed.* 51, 4484-4487.  
55  
56  
57 Messenger, S., Keller, L. P., Lauretta, D. S., 2005. Supernova olivine from cometary  
58 dust. *Science* 309, 737-741.  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 Messenger, S., 2000. Identification of molecular-cloud material in interplanetary dust  
5 particles. *Nature* 404, 968-971.  
6  
7  
8 Miyamoto, M., Zolensky, M. E., 1994. Infrared diffuse reflectance spectra of  
9 carbonaceous chondrites: Amount of hydrous minerals. *Meteoritics* 29, 849-853.  
10  
11 Modica, P., Meinert, C., De Marcellus, P., Nahon, L., Meierhenrich, U. J., d'Hendecourt,  
12 L. Le S., 2014. Enantiomeric excesses induced in amino acids by ultraviolet  
13 circularly polarized light irradiation of extraterrestrial ice analogs: A possible  
14 source of asymmetry for prebiotic chemistry. *Astrophys. J. Lett.* 788, 79-89.  
15  
16  
17 Moghadam, R. H., Trepmann, C. A., Stockhert, B., Renner, J., 2010. Rheology of  
18 synthetic omphacite aggregates at high pressure and high temperature. *J.*  
19 *Petrology* 51, 921-945.  
20  
21  
22 Molster, F. J., Waters, L. B. F. M., Tielens, A. G. G. M., Koike, C., Chihara, H., 2002.  
23 Crystalline silicate dust around evolved stars. III. A correlations study of  
24 crystalline silicate features. *Astron. Astrophys.* 382, 241-255.  
25  
26  
27 Morlok, A., Libourel, G., 2013. Aqueous alteration in CR chondrites: Meteorite parent  
28 body processes as analogue for long-term corrosion processes relevant for nuclear  
29 waste disposal. *Geochim. Cosmochim. Acta* 103, 76-103.  
30  
31  
32 Morris, R. V., Lauer Jr., H. V., Lawson, C. A., Gibson Jr., E. K., Nace, G. A., Stewart,  
33 C., 1985. Spectral and other physicochemical properties of submicron powders of  
34 hematite ( $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>), maghemite ( $\gamma$ -Fe<sub>2</sub>O<sub>3</sub>), magnetite (Fe<sub>3</sub>O<sub>4</sub>), goethite ( $\alpha$ -  
35 FeOOH), and lepidocrocite ( $\gamma$ -FeOOH). *J. Geophys. Res.* 90, 3126-3144.  
36  
37  
38 Myrgorodska, I., Meinert, C., Hoffmann, S. V., Jones, N. C., Nahon, L., Meierhenrich, U.  
39 J., 2017. Light on chirality: Absolute asymmetric formation of chiral molecules  
40 relevant in prebiotic evolution. *ChemPlusChem* 82, 74-87.  
41  
42  
43 Nagashima, K., Krot, A. N., Huss, G. R., 2014. <sup>26</sup>Al in chondrules from CR2 chondrites.  
44 *Geochemic. J.* 48, 561-570.  
45  
46  
47 Nagashima, K., Krot, A. N., Yurimoto, H., 2004. Stardust silicates from primitive  
48 meteorites. *Nature* 428, 921-924.  
49  
50  
51 Nakashima, D., Matsuda, S., Iio, H., Bajo, K., Ebisawa, N., Nagao, K., 2009. Noble gases  
52 in the NWA 852/801 CR2 chondrites (abstract). 40<sup>th</sup> Lunar Planet. Sci. Conf., LPI  
53 Contribution, id. 1661.  
54  
55  
56 Nguyen, A. N., Keller, L. P., Messenger, S., Rahman, Z., 2017a. Mineralogical  
57 characterization of Fe-bearing AGB and Supernova silicate grains from the Queen  
58 Alexandra Range 99177 meteorite (abstract). 48<sup>th</sup> Lunar Planet. Sci., #2371.  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 Nguyen, A. N., Keller, L. P., Rahman, Z., Messenger, S., 2017b. Microstructural study of  
5 an  $^{18}\text{O}$ -poor presolar silicate grain from the Meteorite Hills 00426 CR3 chondrite  
6 (abstract). *Meteorit. Planet. Sci.* 80, A6332.  
7  
8  
9 Nguyen, A. N., Keller, L. P., Messenger, S., 2016. Mineralogy of presolar silicate and  
10 oxide grains of diverse stellar origins. *Astrophys. J.* 818, 51-67.  
11  
12 Nguyen, A. N., Nittler, L. R., Stadermann, F. J., Stroud, R. M., Alexander, C. M. O. D.,  
13 2010. Coordinated analysis of presolar grains in the Allan Hills 77307 and Queen  
14 Elizabeth Range 99177 meteorites. *Astrophys. J.* 719, 166-189.  
15  
16  
17 Nguyen, A. N., Stadermann, F. J., Zinner, E., Stroud, R. M., Alexander, C. M. O. D.,  
18 Nittler, L. R., 2007. Characterization of presolar silicate and oxide grains in  
19 primitive carbonaceous chondrites. *Astrophys. J.* 656, 1223-1240.  
20  
21  
22 Nguyen, A. N., Zinner, E., 2004. Discovery of ancient silicate stardust in a meteorite.  
23 *Science* 303, 1496-1499.  
24  
25  
26 Nguyen, A., Zinner, E., Lewis, R. S., 2003. Identification of small presolar spinel and  
27 corundum grains by isotopic raster imaging. *Astronom. Society of Australia* 20,  
28 382-388.  
29  
30 L. R. Nittler, Stroud, R. M., Trigo-Rodríguez, J. M., De Gregorio, B. T., Alexander, C.  
31 M. O'D., Davidson J., Moyano-Camero, C. E., Tanbakouei, S., 2019. A  
32 cometary building block in a primitive asteroidal meteorite. *Nature Astronomy* 3,  
33 659-666.  
34  
35  
36 Nuth, J. A., III, Brearley, A. J., AND Scott, E. R. D., 2005. Microcrystals and  
37 amorphous material in comets and primitive meteorites: Keys to understanding  
38 processes in the early solar system. *In Chondrites and the Protoplanetary Disk*  
39 (eds. A. N. Krot, E. R. D. Scott and B. Reipurth), pp. 675-700. ASP Conference  
40 Series 341. Astronomical Society of the Pacific, San Francisco.  
41  
42  
43 Nittler, L. R., Alexander, C. M. O. D., Davidson, J., Riebe, M. E. I., Stroud, R. M.,  
44 Wang, J., 2018. High abundances of presolar grains and  $^{15}\text{N}$ -rich organic matter  
45 in CO3.0 chondrite Dominion Range 08006. *Geochim. Cosmochim. Acta* 226,  
46 107-131.  
47  
48  
49 Oberti, R., Caporuscio, F. A., 1991. Crystal-chemistry of clinopyroxenes from mantle  
50 eclogites—a study of the key role of the M2 site population by means of crystal-  
51 structure refinement. *Amer. Mineral.* 76, 1141–1152.  
52  
53  
54 Olsen, M. B., Wielandt, D., Schiller, M., Van Kooten, E. M. M. E., Bizzarro, M., 2016.  
55 Magnesium and  $^{54}\text{Cr}$  isotope compositions of carbonaceous chondrite chondrules  
56 – Insights into early disk processes. *Geochim. Cosmochim. Acta* 191, 118-138.  
57  
58  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 Orthous-Daunay, F.-R., Quirico, E., Beck, P., Brissaud, O., Dartois, E., Pino, T., Schmitt,  
5 B., 2013. Mid-infrared study of the molecular structure variability of insoluble  
6 organic matter from primitive chondrites. *Icarus* 223, 534-543.  
7  
8  
9 Osawa, T., Kagi, H., Nagao, K., 2001. Mid-infrared transmission spectra of individual  
10 Antarctic micrometeorites and carbonaceous chondrites. *Antarctic Meteorite*  
11 *Research* 14, 71-88.  
12  
13  
14 \*Osawa, T., Kagi, H., Nakamura, T., Noguchi, T., 2005. Infrared spectroscopic taxonomy  
15 for carbonaceous chondrites from speciation of hydrous components. *Meteorit.*  
16 *Planet. Sci.* 40, 71-86.  
17  
18  
19 Pearson, V. K., Sephton, M. A., Franchi, I. A., Gibson, J. M., Gilmour, I., 2006. Carbon  
20 and nitrogen in carbonaceous chondrites: Elemental abundances and stable  
21 isotopic compositions. *Meteorit. Planet. Sci.* 41, 1899-1918.  
22  
23  
24 Peltzer, E. T., Bada, J. L., Schlesinger, G., Miller, S. L., 1984. The chemical conditions  
25 on the parent body of the Murchison meteorite: some conclusions based on  
26 amino, hydroxy and dicarboxylic acids. *Adv. Space Research* 4, 69-74.  
27  
28  
29 Perronnet, M., Berger, G., Zolensky, M. E., Toplis, M. J., Bajagic, M., 2007. The  
30 aqueous alteration of CR chondrites: Experimental and geochemical modeling  
31 (abstract). 38<sup>th</sup> Lunar Planet. Sci. Conf., #1110.  
32  
33  
34 Perronnet, M., Zolensky, M. E., 2006. Characterization and quantification of metallic and  
35 mineral phases in the highly hydrated Grosvenor Mountains 95577 CR1 chondrite  
36 (abstract). 37<sup>th</sup> Lunar Planet. Sci. Conf., no.2402.  
37  
38  
39 Pizzarello, S., Yarnes, T., 2016. Enantiomeric excesses of chiral amines in ammonia-rich  
40 carbonaceous meteorites. *Earth Planet. Sci. Lett.* 443, 176-184.  
41  
42  
43 Pizzarello, S., Schrader, D. L., Monroe, A. A., Lauretta, D. S., 2012. The chiral  
44 composition of primitive CR meteorites and the diverse effects of water in  
45 cosmochemical evolution. *Proc. Natl. Academy Sci. USA* 109, 11949–11954.  
46  
47  
48 Pizzarello, S., Williams, L. B., 2012. Ammonia in the early solar system: An account  
49 from carbonaceous meteorites. *Astrophys. J.* 749, 161 (6pp).  
50  
51  
52 Pizzarello, S., Williams, L. B., Lehman, J., Holland, G. P., Yarger, J. L., 2011. Abundant  
53 ammonia in primitive asteroids and the case for a possible exobiology. *Proc. Natl.*  
54 *Academy Sci. USA* 108, 4303-4306.  
55  
56  
57 Pizzarello, S., Holmes, W., 2009. Nitrogen-containing compounds in two CR2  
58 meteorites: <sup>15</sup>N composition, molecular distribution and precursor molecules.  
59 *Geochim. Cosmochim. Acta* 73, 2150-2162.  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 Pizzarello, S., Huang, Y., Alexandre, M. R., 2008. Molecular asymmetry in  
5 extraterrestrial chemistry: Insights from a pristine meteorite. *Proc. Natl. Academy*  
6 *Sci. USA* 105, 3700-3704.  
7  
8  
9 Pizzarello, S., Cooper, G., Flynn, G., 2006. The nature and distribution of the organic  
10 material in carbonaceous chondrites and interplanetary dust particles. *In*  
11 *Meteorites and the Early Solar System II*. Univ. Arizona Press (Eds: D. Lauretta  
12 and H. McSween), p.p. 625-651.  
13  
14  
15 Pollock, G. E., Cheng, C.-N., Cronin, S. E., Kvenvolden, K. A., 1975. Stereoisomers of  
16 isovaline in the Murchison meteorite. *Geochim. Cosmochim. Acta* 39, 1571-1573.  
17  
18  
19 Richardson, S. M., 1981. Alteration of mesostasis in chondrules and aggregates  
20 from three C2 carbonaceous chondrites. *Earth Planet. Sci. Lett.* 52, 67-75.  
21  
22  
23 Righter, K., Satterwhite, C., Funk, R., Harrington, R. 2017. Historical trends in US  
24 Antarctic meteorite allocations, with a close look at CR chondrites (abstract). 80<sup>th</sup>  
25 Ann. Meet. Meteoritical Soc., LPI Contribution No., 1987, id. 6303.  
26  
27  
28 Robert, F., Epstein, S., 1982. The concentration and isotopic composition of hydrogen,  
29 carbon and nitrogen in carbonaceous meteorites. *Geochim. Cosmochim. Acta* 46,  
30 81-95.  
31  
32  
33 Rochette, P., Gattacceca, J., Bonal, L., Bourot-Denise, M., Chevrier, M., Clerc, J.-P.,  
34 Consolmagno, G., Folco, L., Gounelle, M., Kohout, T., Pesonen, L., Quirico, E.,  
35 Sagnotti, L., Skripnik, A., 2008. Magnetic classification of stony meteorites: 2.  
36 Non-ordinary chondrites. *Meteorit. Planet. Sci.* 43, 959–980.  
37  
38  
39 Rubin, A. E., 1988. Formation of ureilites by impact-melting of carbonaceous chondritic  
40 material. *Meteoritics* 23, 333-337.  
41  
42  
43 Rubin, A. E., Trigo-Rodríguez, J. M., Huber, H., Wasson, J. T., 2007. Progressive  
44 aqueous alteration of CM carbonaceous chondrites. *Geochim. Cosmochim. Acta*  
45 71, 2361-2382.  
46  
47  
48 Russell, S. S., Zipfel, J., Grossman, J. N., Grady, M. M. (2002). The Meteoritical  
49 Bulletin, No. 86, 2002 July. *Meteorit. Planet. Sci.* 37, A157-A184.  
50  
51  
52 Ruzicka, A., Grossman, J. N., Bouvier, A., Herd, C. D. K., Agee, C. B., 2015. The  
53 Meteoritical Bull. No. 102. *Meteorit. Planet. Sci.* 50, 1662-1662.  
54  
55  
56 Ruzicka, A., Grossman, J. N., Garvie, L., 2014. The Meteoritical Bulletin, No. 100.  
57 *Meteorit. Planet. Sci.* 49, E1-E101.  
58  
59  
60 Sandford, S. A., 1996. The inventory of interstellar materials available for the formation  
61 of the solar system. *Meteorit. Planet. Sci.* 31, 449-476.  
62  
63  
64  
65

- 1  
2  
3  
4  
5 Sandford, S. A., Bernstein, M. P., Dworkin, J. P. 2001. Assessment of the interstellar  
6 processes leading to deuterium enrichment in meteoritic organics. *Meteorit.*  
7 *Planet. Sci.* 36, 1117-1133.  
8  
9  
10 Schmitt-Kopplin, P., Gabelica, Z., Gougeon, R. D., Fekete, A., Kanawati, B., Harir, M.,  
11 Gebefuegi, I., Eckel, G., Hertkorn, N., 2010. High molecular diversity of  
12 extraterrestrial organic matter in Murchison meteorite revealed 40 years after its  
13 fall. *Proc. Natl. Academy Science USA* 107, 2763-2768.  
14  
15  
16 Schrader, D. L., Lauretta, D. S., 2010. High-temperature experimental analogs of  
17 primitive meteoritic metal-sulfide-oxide assemblages. *Geochim. Cosmochim.*  
18 *Acta* 74, 1719-1733.  
19  
20  
21 Schrader, D. L., Nagashima, K., Waitukaitis, S. R., Davidson, J., McCoy, T. J., Connolly,  
22 H. C., Lauretta, D. S., 2018a. The retention of dust in protoplanetary disks:  
23 Evidence from agglomeratic olivine chondrules from the outer Solar System.  
24 *Geochim. Cosmochim. Acta* 223, 405-421.  
25  
26  
27 Schrader, D. L., Fu, R. R., Desch, S. J., Davidson, J. 2018b. The background temperature  
28 of the protoplanetary disk within the first four million years of the Solar System.  
29 *Earth Planet. Sci. Lett.* 504, 30-37.  
30  
31  
32 Schrader, D. L., Nagashima, K., Krot, A. N., Ogliore, R. C., Yin, Q.-Z., Amelin, Y. A.,  
33 Stirling, C. H., Kaltenbach, A., 2017. Distribution of  $^{26}\text{Al}$  in the CR chondrite  
34 chondrule-forming region of the protoplanetary disk. *Geochim. Cosmochim. Acta*  
35 201, 275–302.  
36  
37  
38 Schrader, D. L., Connolly, Jr. H. C., Lauretta, D. S., Zega, T. J., Davidson, J., Domanik,  
39 K. J., 2015. The formation and alteration of the Renazzo-like carbonaceous  
40 chondrites III: Toward understanding the genesis of ferromagnesian chondrules.  
41 *Meteorit. Planet. Sci. Suppl.* 50, 15-50.  
42  
43  
44 Schrader, D. L., Nagashima, K., Krot, A. N., Ogliore, R. C., Hellebrand, E., 2014.  
45 Variations in the O-isotope composition of gas during the formation of chondrules  
46 from the CR chondrites. *Geochim. Cosmochim. Acta* 132, 50–74.  
47  
48  
49 Schrader, D. L., Connolly, H. C. Jr., Lauretta, D. S., Nagashima, K., Huss, G. R.,  
50 Davidson, J., Domanik, K. J., 2013. The formation and alteration of the Renazzo-  
51 like carbonaceous chondrites II: Linking O-isotope composition and oxidation  
52 state of chondrule olivine. *Geochim. Cosmochim. Acta* 101, 302–327.  
53  
54  
55 Schrader, D. L., Franchi, I. A., Connolly, Jr. H. C., Greenwood, R. C., Lauretta, D. S.,  
56 Gibson, J. M., 2011. The formation and alteration of the Renazzo-like  
57 carbonaceous chondrites I: Implications of bulk-oxygen isotopic composition.  
58 *Geochim. Cosmochim. Acta* 75, 308-325.  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4  
5  
6 Schrader, D. L., Connolly, Jr. H. C., Lauretta, D. S., 2008. Opaque phases in type-II  
7 chondrules from CR2 chondrites: Implications for CR parent body formation.  
8 *Geochim. Cosmochim. Acta* 72, 6124-6140.  
9
- 10 Sherman, D. M., Burns, R. G., Burns, V. M., 1982. Spectral characteristics of the iron  
11 oxides with application to the Martian bright region mineralogy. *J. Geophys.*  
12 *Research* 87, 10169-10180.  
13  
14
- 15 Scott, E. R. D., Barber D. J., Alexander, C. M., Hutchinson, R., Peck, J. A., 1988.  
16 Primitive material surviving in chondrites – Matrix. *In Meteorites and the early*  
17 *solar system*, (eds Kerridge J. F. and Matthews M. S.). University of Arizona  
18 Press, 718-745.  
19  
20
- 21 Stroud, R. M., Floss, C., Stadermann, F. J., 2009. Structure, elemental composition and  
22 isotopic composition of presolar silicates in MET 00426 (abstract). 40<sup>th</sup> Lunar  
23 Planet. Sci. Conf., #1063.  
24  
25
- 26 Takano Y., Takahashi J., Kaneko T., Marumo K., Kobayashi K., 2007. Asymmetric  
27 synthesis of amino acid precursors in interstellar complex organics by circularly  
28 polarized light. *Earth Planet. Sci. Lett.* 254, 106-114.  
29  
30
- 31 Tenner, T. J., Nakashima, D., Ushikubo, T., Kita, N. T., Weisberg, M. K., 2015. Oxygen  
32 isotope ratios of FeO-poor chondrules in CR3 chondrites: Influence of dust  
33 enrichment and H<sub>2</sub>O during chondrule formation. *Geochim. Cosmochim. Acta*  
34 148, 228–250.  
35  
36
- 37 Tenner, T. J., Ushikubo, T., Nakashima, D., Kita, N. T., Weisberg, M. K., 2013. <sup>26</sup>Al in  
38 chondrules from the CR3.0 chondrite Queen Alexandra Range 99177: A link with  
39 O isotopes (abstract). 44<sup>th</sup> Lunar Planet. Sci. Conf., LPI Contrib. No. 1719, 2010.  
40  
41
- 42 Tielens, A. G. G. M., McKee, C. F., Seab, C. G., Hollenbach, D. J., 1994. The physics of  
43 grain-grain collisions and gas-grain sputtering in interstellar shocks. *Astrophys. J.*  
44 431, 321-340.  
45  
46
- 47 Tielens, A. G. G. M., Waters, L. B. F. M., Molster, F. J., Justtanont, K., 1998.  
48 Circumstellar silicate mineralogy. *Astrophys. Space Sci.* 255, 415-426.  
49  
50
- 51 Tomeoka, K., Buseck, P., 1988. Matrix mineralogy of the Orgueil CI carbonaceous  
52 chondrite. *Geochim. Cosmochim. Acta* 52, 1627-1640.  
53  
54
- 55 Tomeoka, K., McSween, Y. H., Jr., Buseck, R. P., 1988. Aqueous alteration of CI and  
56 CM carbonaceous chondrites: A review. 13<sup>th</sup> Symp. Antarctic Meteorites, NIPR,  
57 160.  
58  
59  
60  
61  
62  
63  
64  
65



- 1  
2  
3  
4 Tomeoka, K., Buseck, P. R., 1985. Indicators of aqueous alteration in CM carbonaceous  
5 chondrites: Microtextures of a layered mineral containing Fe, S, O and Ni.  
6 Geochim. Cosmochim. Acta, 49, 2149-2163.  
7  
8  
9 \*Treiman, A. H., Gross, J. 2015. The CR2 chondrite NWA 801: Petrography and  
10 petrology (abstract). 78<sup>th</sup> Ann. Meet. Meteoritical Soc., LPI Contribution No.  
11 1856, p. 5077.  
12  
13  
14 Tsuchiyama, A., Mashio, E., Imai, Y., Noguchi, T., Miura, Y., Yano, H., Nakamura, T.,  
15 2009. Strength measurement of carbonaceous chondrites and micrometeorites  
16 using micro compression testing machine (abstract). Meteorit. Planet. Sci. Suppl.,  
17 5189.  
18  
19  
20 Tyra, M. A., 2013. Using oxygen and carbon stable isotopes, <sup>53</sup>Mn-<sup>53</sup>Cr isotope  
21 systematics, and petrology to constrain the history of carbonates and water in the  
22 CR and CM chondrite parent bodies. ProQuest Dissertations And Theses; Thesis  
23 (Ph.D.) -- The University of New Mexico, 2013; Publication Number: AAT  
24 3588093; ISBN: 9781303260063; Source: Dissertation Abstracts International,  
25 Volume: 74-11(E), Section: B.; 235 p.p.  
26  
27  
28 Urey, H. C., Craig H., 1953. The composition of the stone meteorites and the origin of  
29 the meteorites. Geochim. Cosmochim. Acta 4, 36-82.  
30  
31  
32 van Dishoeck, E. F., 2004. ISO spectroscopy of gas and dust: from molecular clouds to  
33 protoplanetary disks. Ann. Rev. Astron. Astrophys. 42, 119-167.  
34  
35  
36 van Schmus, W. R., Wood, J. A., 1967. A chemical-petrologic classification for the  
37 chondritic meteorites. Geochim. Cosmochim. Acta 31, 755-765.  
38  
39  
40 Velbel, M. A., Long, D. T., Gooding, J. L., 1991. Terrestrial weathering of Antarctic  
41 stone meteorites - Formation of Mg-carbonates on ordinary chondrites. Geochim.  
42 Cosmochim. Acta 55, 67-76.  
43  
44  
45 Wasson J. T., 1996. Chondrule formation: Energetics and length scales. *In* Chondrules  
46 and the Protoplanetary Disk. Cambridge Univ. Press (ed. R. H. Hewins, R. H.  
47 Jones and E. R. D. Scott), pp. 45-54.  
48  
49  
50 Wasson, J. T., Rubin, A. E., 2014. Absence of matrix-like chondrule rims in CR2 LAP  
51 02342. Meteorit. Planet. Sci. 49, 245-260.  
52  
53  
54 Wasson, J. T., Rubin, A. E., 2010. Metal in CR chondrites. Geochim. Cosmochim. Acta  
55 74, 2212-2230.  
56  
57  
58 Wasson, J. T., Rubin, A. E., 2009. Composition of matrix in the CR chondrite LAP  
59 02342. Geochim. Cosmochim. Acta 73, 1436-1460.  
60  
61  
62  
63  
64  
65

1  
2  
3  
4 Weisberg, M. K., Huber, H., 2007. The GRO 95577 CR1 chondrite and hydration of the  
5 CR parent body. *Meteorit. Planet. Sci.* 42, 1495-1503.  
6

7  
8 Weisberg, M. K., McCoy, T. J., Krot, A. N., 2006. Systematics and evaluation of  
9 meteorite classification. *In Meteorites and the Early Solar System II* (eds. D.S.  
10 Laurretta and H.Y. McSween, Jr.). University of Arizona Press. pp. 19-52.  
11

12  
13 Weisberg, M. K., Connolly, H. C., Jr., Ebel, D. S., 2004. Petrology and origin of  
14 amoeboid olivine aggregates in CR chondrites. *Meteorit. Planet. Sci.* 39, 1741-  
15 1753.  
16

17  
18 Weisberg, M. K., Prinz, M., Clayton, R. N., Mayeda, T. K., Grady, M. M., Pillinger, C.  
19 T., 1995. The CR chondrite clan. *Antarctic Meteorite Res.* 8, 11.  
20

21  
22 Weisberg, M. K., Prinz, M., Clayton, R. N., Mayeda, T. K., 1993. The CR (Renazzo-  
23 type) carbonaceous chondrite group and its implications. *Geochim. Cosmochim.*  
24 *Acta* 57, 1567-1586.  
25

26  
27 \*Weisberg, M. K., Prinz, M., Clayton, R. N., Mayeda, T. K., 1992. Formation of layered  
28 chondrules in CR2 chondrites: A petrologic and oxygen isotopic study. *Meteorit.*  
29 27, 306.  
30

31  
32 \*Weisberg, M. K., Prinz, M., 2001. The Grosvenor Mountains 95577 CR1 chondrite and  
33 hydration of the CR chondrites (abstract). *Meteorit. Planet. Sci.* 35, A168.  
34

35  
36 Wiik, H. B., 1956. The chemical composition of some stony meteorites. *Geochim.*  
37 *Cosmochim. Acta* 9, 279-289.  
38

39  
40 \*Wlotzka F. 1993. A weathering scale for the ordinary chondrites (abstract). *Meteorit.*  
41 28, 460.  
42

43  
44 \*Wlotzka, F., Jull, A. J. T., Donahue D. J. 1995. Carbon-14 terrestrial ages of meteorites  
45 from Acfer, Algeria. *In Meteorites from Cold and Hot Deserts.* LPI Tech. Rpt. 95-  
46 02, 72-73.  
47

48  
49 Wood, J. A., 1962. Metamorphism in chondrites. *Geochim. Cosmochim. Acta* 26, 739-  
50 742.  
51

52  
53 Wood, J. A., 1963a. Chondrites and chondrules. *Scientif. Amer.* 4, 64-82.  
54

55  
56 Wood, J. A., 1963b. On the origin of chondrules and chondrites. *Icarus* 2, 152-180.  
57

58  
59 Wood, J. A., 1967. Chondrites: Their metallic minerals, thermal histories, and parent  
60 planets. *Icarus* 6, 1-49.  
61  
62  
63  
64  
65

- 1  
2  
3  
4 Wood, J. A., McSween, H. Y. Jr., 1977. Chondrules as condensation products. *In*  
5 Comets, asteroids, meteorites: Interrelations, evolution and origins; *Proced.*  
6 *Thirty-ninth Internat. Colloquium*, 365-373.  
7  
8  
9 \*Zanda, B., Bourot-Denise, M., Hewins, R. H., Cohen, B. A., Delaney, J. S., Humayun,  
10 M., Campbell, A. J., 2002. Accretion textures, iron evaporation and re-  
11 condensation in Renazzo chondrules (abstract). *33<sup>rd</sup> Lunar Planet. Sci. Conf.*  
12 1852.  
13  
14  
15 Zanda, B., Yu, Y., Bourot-Denise, M., Hewins, R., 1997. The history of metal and  
16 sulfides in chondrites (abstract). *Workshop on Parent-Body and Nebular*  
17 *Modification of Chondritic Materials*, Lunar Planet. Institute, p.68.  
18  
19  
20 Zanda, B., Bourot-Denise, M., Perron, C., Hewins, R. H., 1994. Origin and metamorphic  
21 redistribution of silicon, chromium, and phosphorus in the metal of chondrites.  
22 *Science* 265, Issue 5180, 1846-1849.  
23  
24  
25 Zega, T. J., Buseck, P. R., 2003. Fine-grained-rim mineralogy of the Cold Bokkeveld CM  
26 chondrite. *Geochim. Cosmochim. Acta* 67, 1711–1721.  
27  
28  
29 Zhao, X., Floss, C., Lin, Y., Bose, M., 2013. Stardust investigation into the CR chondrite  
30 Grove Mountain 021710. *Astrophys. J.* 769, article id. 49, 16 pp.  
31  
32  
33 Zinner, E., 2014. Presolar Grains, *In Meteorites and Cosmochemical Processes*, Volume  
34 1 (Ed. Andrew Davis), *Treatise on Geochemistry (Second Edition)*. Elsevier,  
35 Oxford, pp. 181-213.  
36  
37  
38 Zolensky, M. E., Abreu, N. M., Velbel, M. A., Rubin, A., Chaumard, N., Michikami T.,  
39 2018. Physical, chemical, and petrological characteristics of chondritic materials  
40 and their relationships to small Solar System bodies *In Chapter 2 of Primitive*  
41 *Meteorites and Asteroids. Physical, Chemical and Spectroscopic Observations*  
42 *Paving the Way to Exploration* (N. Abreu, editor); Elsevier, Amsterdam;  
43 <https://doi.org/10.1016/B978-0-12-813325-5.00004-5> pp.  
44  
45  
46 Zolensky, M. E., Barrett, R., Browning, L., 1993. Mineralogy and composition of matrix  
47 and chondrule rims in carbonaceous chondrites. *Geochim. Cosmochim. Acta* 57,  
48 3123-3148.  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60  
61  
62  
63  
64  
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## FIGURE CAPTIONS

### Fig. 1.

Back-scattered electron images of petrographic thin sections of Antarctic CR chondrites showing the range of different asteroidal features recorded by this group of meteorites. (a) Weakly altered CR EET 92105. Chondrules and metal show no signs of aqueous alteration. (b) Moderately aqueously altered LAP 04720. At the mm- $\mu$ m scale, few appreciable mineralogical changes are noted. Chondrules and metal show no signs of aqueous alteration. (c) Extensive aqueously altered GRO 95577. Chondrules and metal have been extensively replaced by phyllosilicates and oxides. (d) Heated and recrystallized CR GRA 06100. CHO = chondrule; DI = dark inclusion; Fe-Ni = FeNi metal nodule; Mtx = matrix.

### Fig. 2.

Back-scattered electron images showing the characteristics of Fe-Ni metal in type I chondrules from different members of the EET 87111 pairing group that have been affected by terrestrial weathering to different degrees. Although weathering processes progressively replace Fe-Ni metal nodules, there is significant heterogeneity in alteration features within individual stones. Abundant Fe-Ni metal grains are preserved in stones from weathering grade C. Fe-Ni = Fe-Ni metal nodule.

### Fig. 3.

Three-oxygen isotope plot for bulk isotopic analyses from CR chondrites. Data from: Weisberg et al. (1993); Clayton and Mayeda (1999); Schrader et al. (2011, 2014); Harju et al. (2014). CCAM = Carbonaceous chondrite anhydrous minerals line. Y&R = Young and Russell line. TFL = Terrestrial fractionation line. Samples measured by multiple studies are denoted with an asterisk.

### Fig. 4.

Back-scattered electron micrographs of the most common types of chondrules in CR chondrites. (a) Layered MgO-rich, type I chondrule (CHO) in MIL 090657, where the boundaries between the core and outer layers are outlined by Fe-Ni metal nodules. (b) Close up from (a), showing details on the outer chondrule shell, igneous rim with smaller Fe-Ni metal grains compared with interior nodules, and fine-grained rim containing tochilinite. (c) FeO-rich, type II chondrule in EET 92105. (d) Close up from (c), showing

opaque assemblages (Fe-Ni metal and Fe-Ni sulfides) in a type II chondrule in EET 92105. CHO = chondrule; Fe-Ni = FeNi metal nodule.

**Fig. 5.**

Electron micrographs of CR mesostasis in type I and type II chondrules. (a) BSE image of region within a type I chondrule in EET 96259. A box shows the area where a FIB section was retrieved. Notice that the metal grains show signs of weathering. (b) Bright-field TEM image from a FIB section from mesostasis in a type I chondrule in EET 96259. The mesostasis is made up of crystalline andesine and Mn-bearing augite. The band that was partially replaced was identified as predominantly SiO<sub>2</sub>, with <5 wt.% combined total for other oxides (e.g., Al<sub>2</sub>O<sub>3</sub>, Cr<sub>2</sub>O<sub>3</sub>, MgO, FeO, and S). (c) Z-contrast, high-angle annular dark-field (HAADF) micrograph of the area of mesostasis boxed in (b). (d) X-ray elemental map overlay. Al= blue; Ni=white; Ca= orange; Na=green; Fe=yellow; K=magenta; Mg=red; Mn=pink. (e) BSE image of FeO-rich chondrule fragment with a rim of coarse-grained Fe and Fe-Ni sulfides from GRA 95229. Sulfides are also distributed interstitially to pyroxene bars. A box shows the area where a FIB section was retrieved. (f) Z-contrast, high-angle annular dark-field (HAADF) micrograph of a region in the FIB section of fayalitic olivine (Ol), pyrrhotite (Po), and quartz-pyroxene normative glass (Glass).

**Fig. 6.**

BSE images showing the different textural occurrences of fine-grained materials in CR chondrites.

(a) Micrograph of fine-grained matrix in MIL 090657, dominated by fine-grained silicates, with micron and sub-micron Fe-sulfides and Fe-oxides. (b) BSE image of a fine-grained rim around a type I chondrule in GRA 95229. Unlike interchondrule matrix, CR rims do not contain clastic materials (e.g., chondrule fragments). (c) BSE of a region in a dark inclusion in MET 00426, which is nearly featureless, but does contain some chondrule fragments. Dark inclusions in CRs are variable in textures. (d) BSE of a clast in MIL 090657, containing angular chondrule fragments set in fine-grained materials that are more FeO-rich (brighter in BSE) than the matrix in the host meteorite.

**Fig. 7.**

Three-oxygen isotope plot for matrix isotopic analyses from CR chondrites. Data from: Weisberg et al. (1993); Clayton and Mayeda (1999); Schrader et al. (2014). CCAM =

Carbonaceous chondrite anhydrous minerals line. Y&R = Young and Russell line. TFL = Terrestrial fractionation line. Samples measured by multiple studies are denoted with an asterisk.

**Fig. 8.**

TEM of images of the matrix of LAP 02342. (a) Z-contrast, high-angle annular dark-field (HAADF) micrograph of a matrix region from a FIB section. An assemblage of Fe-Mg amorphous silicates, Ca-bearing pyroxene (pyx), Mg-rich olivine, tochilinite, pentlandite, and organics is shown. (b) X-ray elemental map overlay. Ca= yellow; S=green; Fe=turquoise; Si=indigo; and C=red.

**Fig. 9.**

Electron micrographs of a matrix region located in mildly heated GRO 03116. (a) BSE image of the region of clastic matrix from which a FIB section was extracted (FIB-Mtx1). (b) Bright-field TEM image of a FIB section, showing an angular type I chondrule (CHO) fragment and fine-grained matrix materials. (c) Bright-field TEM image of Fe-Mg-rich phyllosilicates from the fine-grained matrix in (b). (d) High-resolution TEM image of Fe-Ni-metal (FeNi) embedded in Fe-Mg-rich amorphous silicates in matrix, inlaid diffraction pattern. (e) High-resolution TEM image of a rounded olivine (ol) crystal in matrix.

**Fig. 10.**

Presolar grain abundances in CR chondrites. The petrographic grades are 2 unless otherwise noted. Average abundances are shown for meteorites in which multiple surveys were conducted. All presolar grain surveys in CRs were conducted by raster ion imaging using a Cameca NanoSIMS ion probe, except for the study of NWA 530 (Nagashima et al., 2004), which used a SCAPS-equipped Cameca IMS 1270 of poorer spatial resolution. The reported abundance for this meteorite is a lower limit and cannot be directly compared to abundances determined through NanoSIMS analyses. (Data from Davidson et al., 2015, 2014a; Floss and Stadermann, 2005, 2009a,b; Koch and Floss, 2017; Leitner et al., 2011, 2012a,b, 2015, 2016a,b; Nagashima et al., 2004; Nguyen et al., 2010; Zhao et al., 2013).

**Fig. 11.**

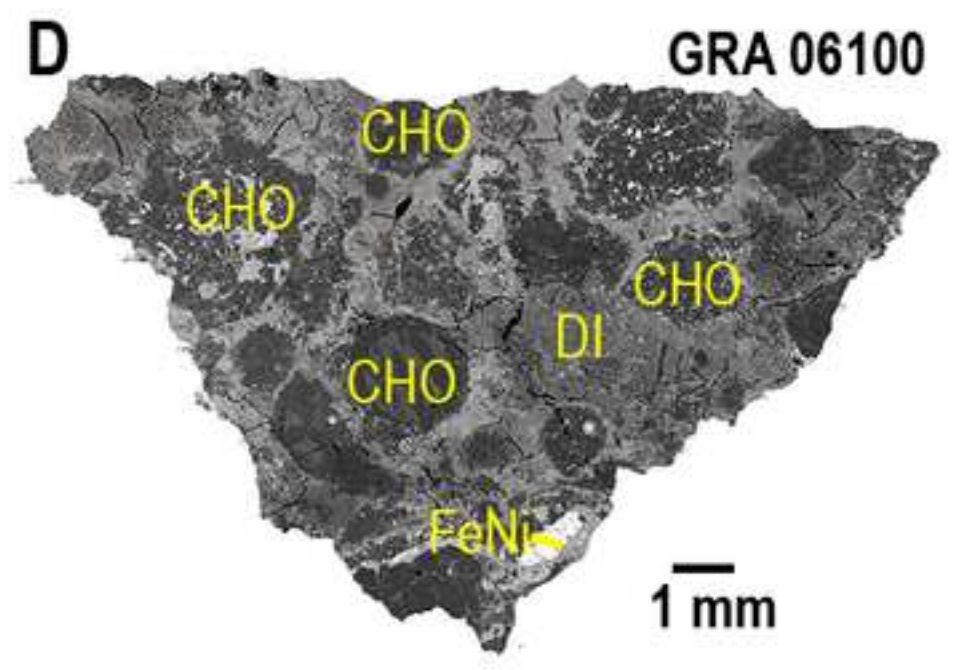
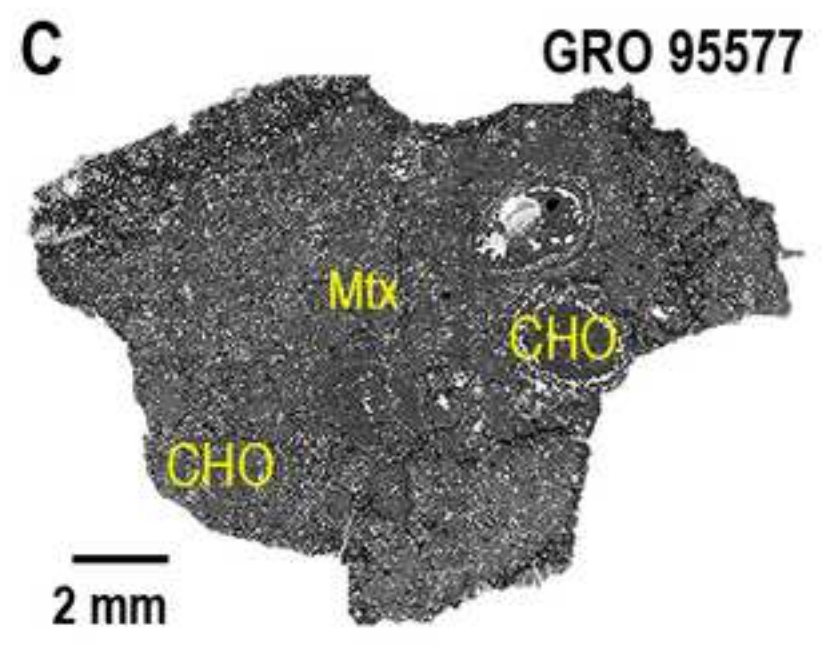
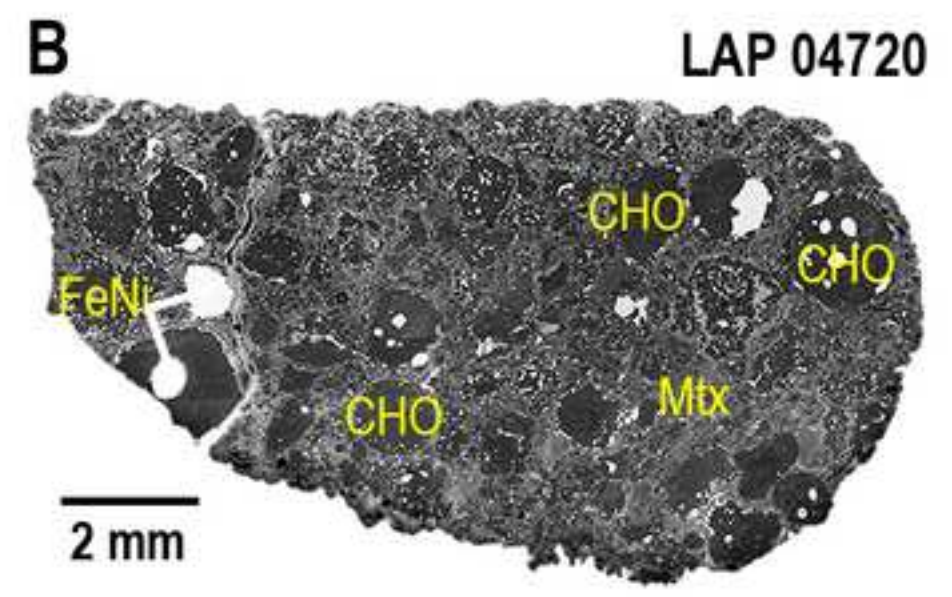
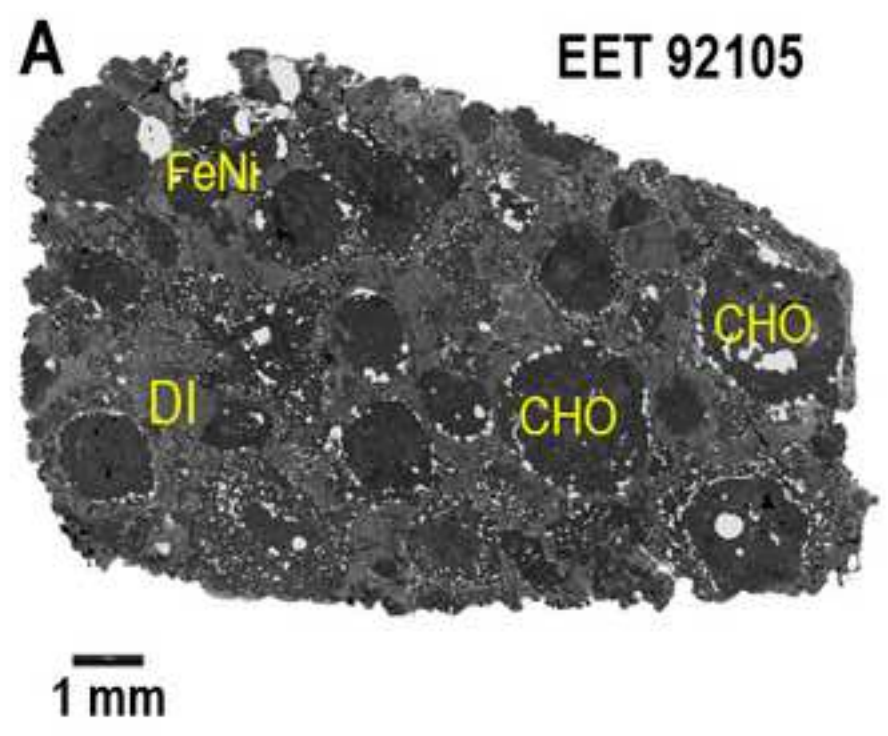
Summary of the effects of aqueous alteration recorded by the components of CR chondrites from different petrologic types.

**Fig. 12.**

Electron micrographs of an omphacite and graphite-bearing clast in pristine QUE 99177, modified from Abreu (2013). (a) BSE image of clast and contacts with host meteorite. (b) Map of the distribution of different components and minerals based on X-ray compositional maps collected using EPMA. Note that the areas labeled as chondrules have significantly larger grains sizes and no graphite. (c) BSE image of a chondrule-like area, showing triple-junctions between olivine (Ol) crystals, interstitial hornblende, and omphacite (Hbl-Omp) crystals and Fe-sulfides. (d) BSE image of a matrix-like area, containing olivine (Ol), pargasite (Prg), troilite (Tro), and large graphite laths.

**Fig. 13.**

Infrared spectra of CR chondrites from different petrologic types. (a) Reflectance spectra (0.3-2.5  $\mu\text{m}$ ) of powdered samples of a CR1, CR2, and CR3 carbonaceous chondrite (from Cloutis et al., 2012). (b) Reflectance spectrum (0.35-2.5  $\mu\text{m}$ ) of a powdered sample of the thermally metamorphosed GRA 06100 CR3.6 chondrite. (c) Infrared reflectance spectra (2-5  $\mu\text{m}$ ) of powdered samples of some CR chondrites. (d) Infrared reflectance spectra (5-15  $\mu\text{m}$ ) of powdered samples of some CR chondrites.





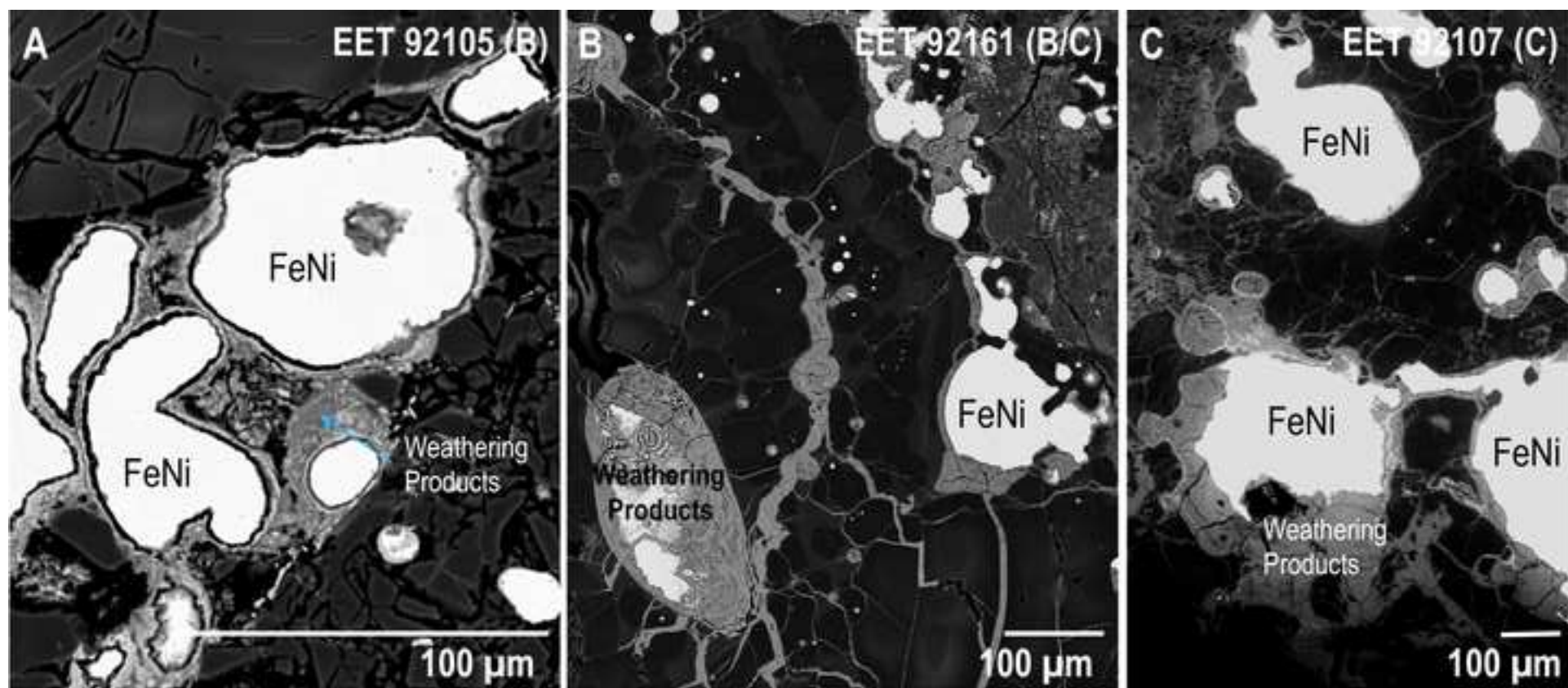
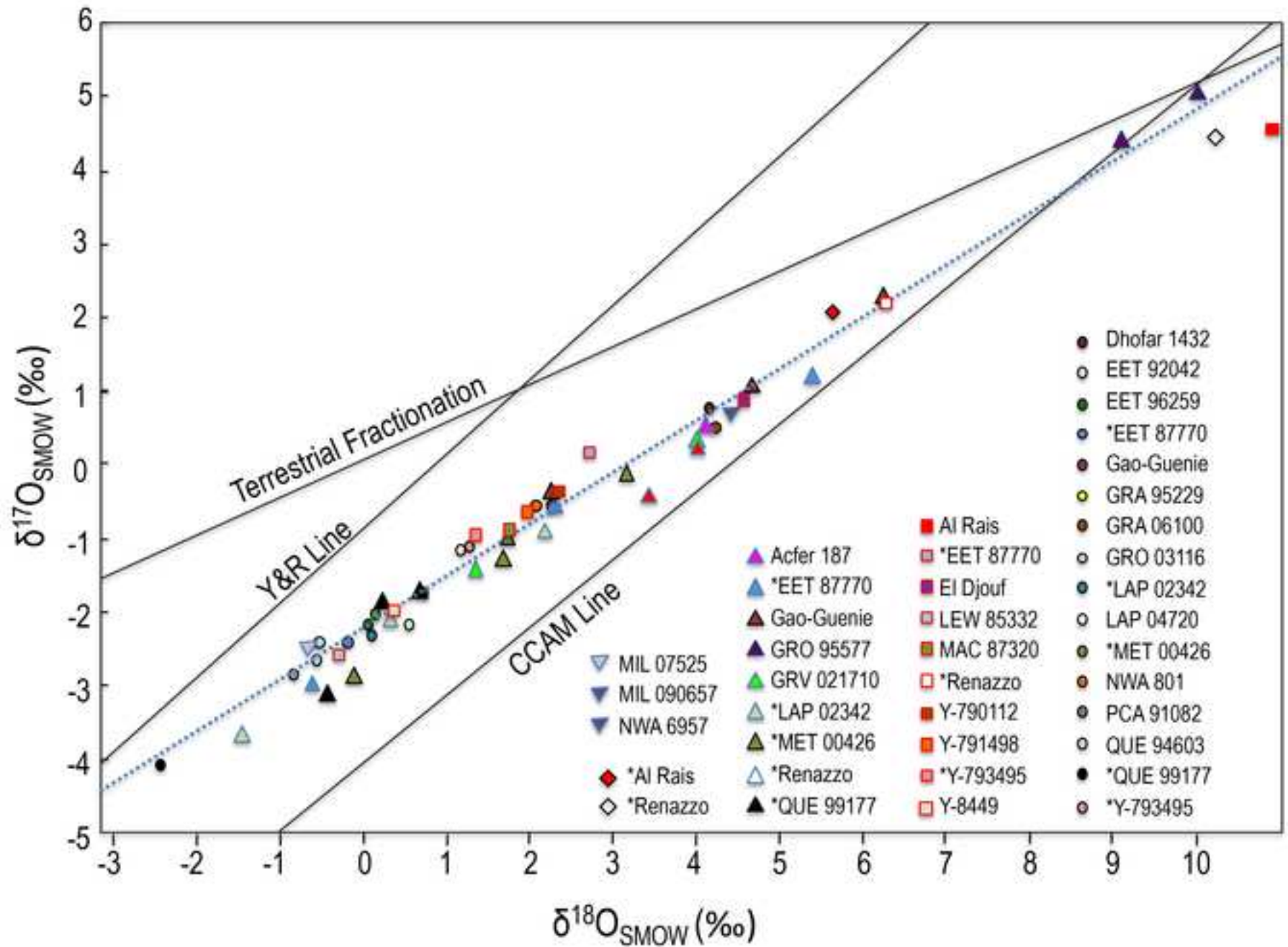
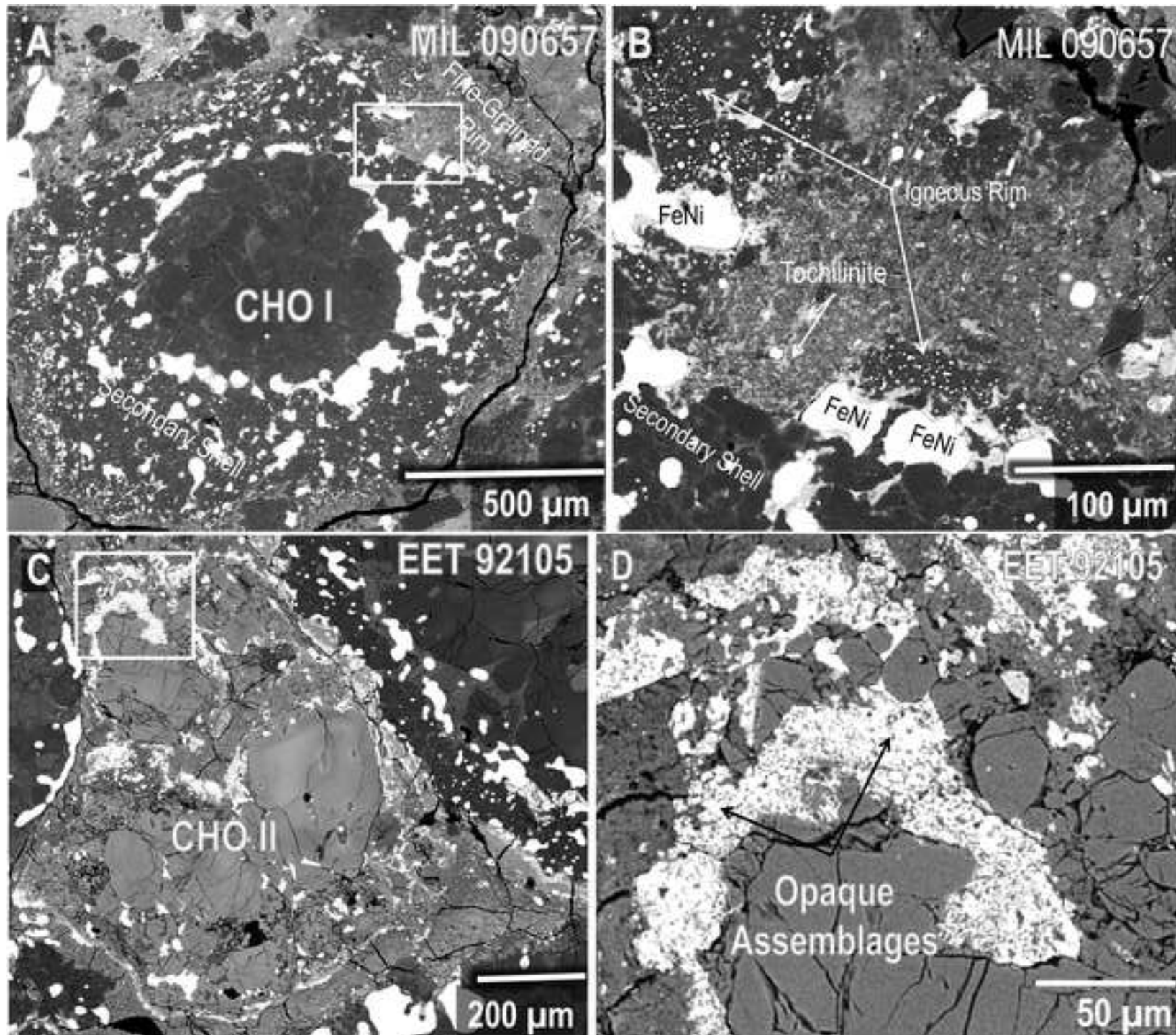
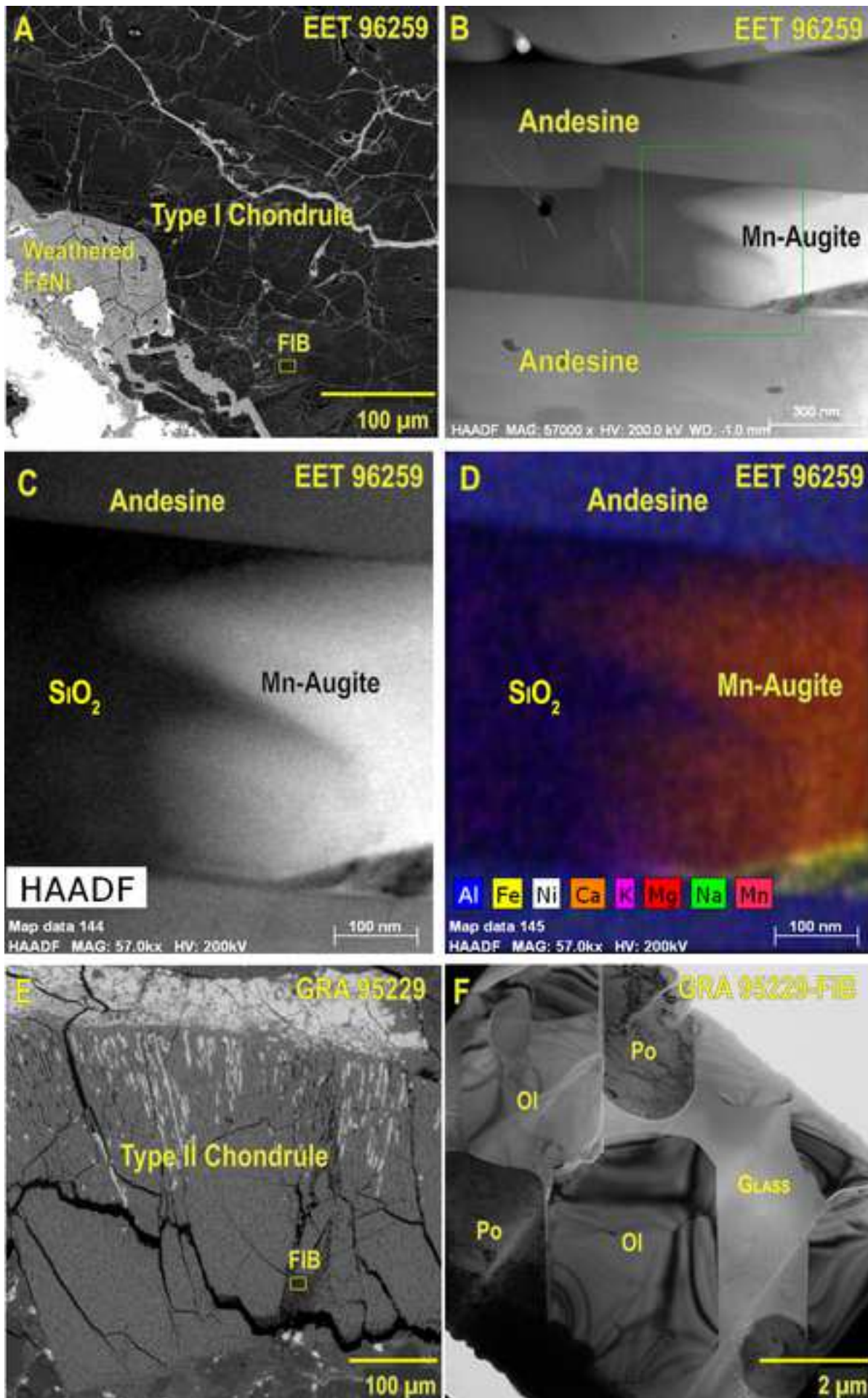


Figure 3







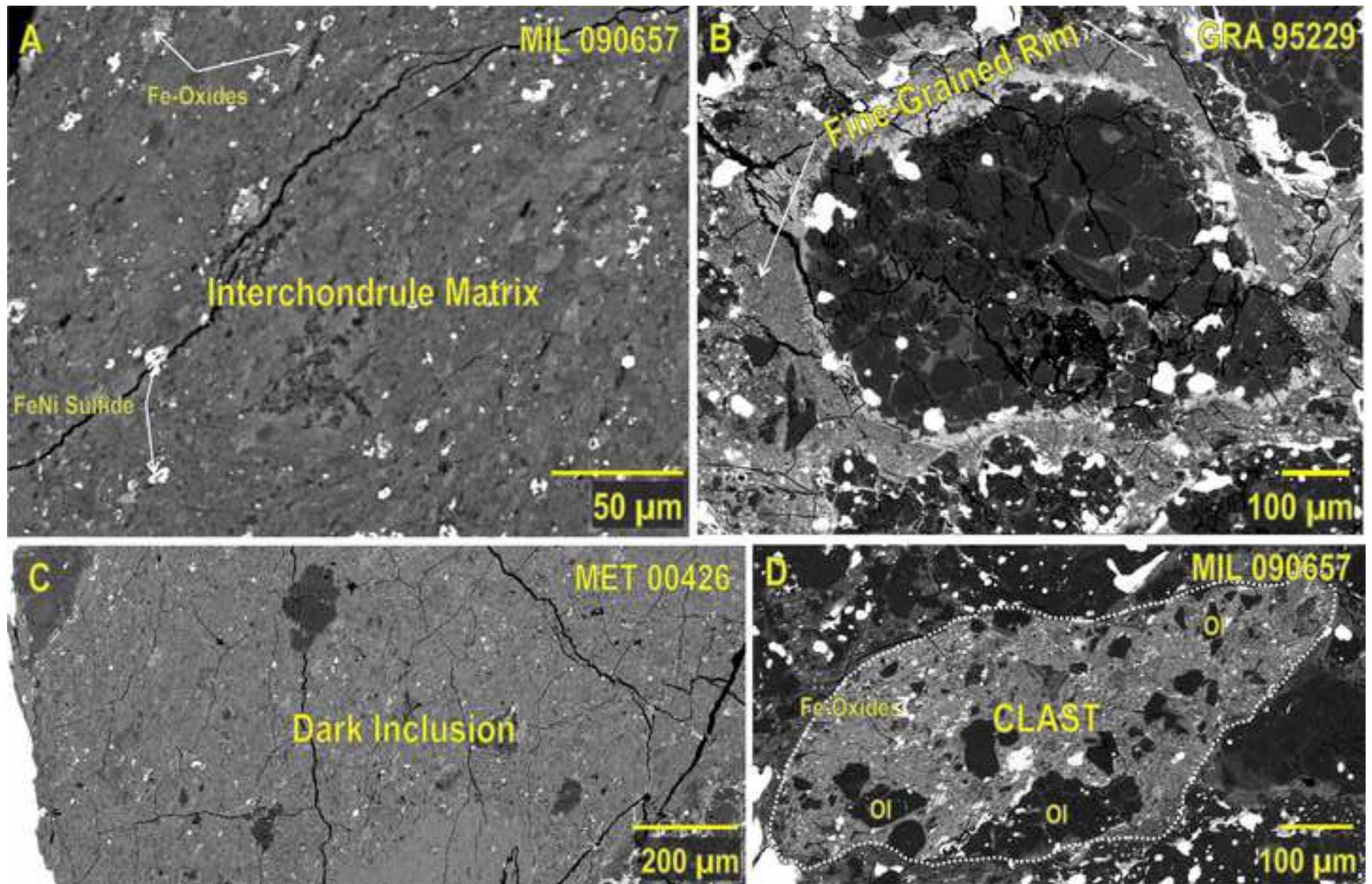
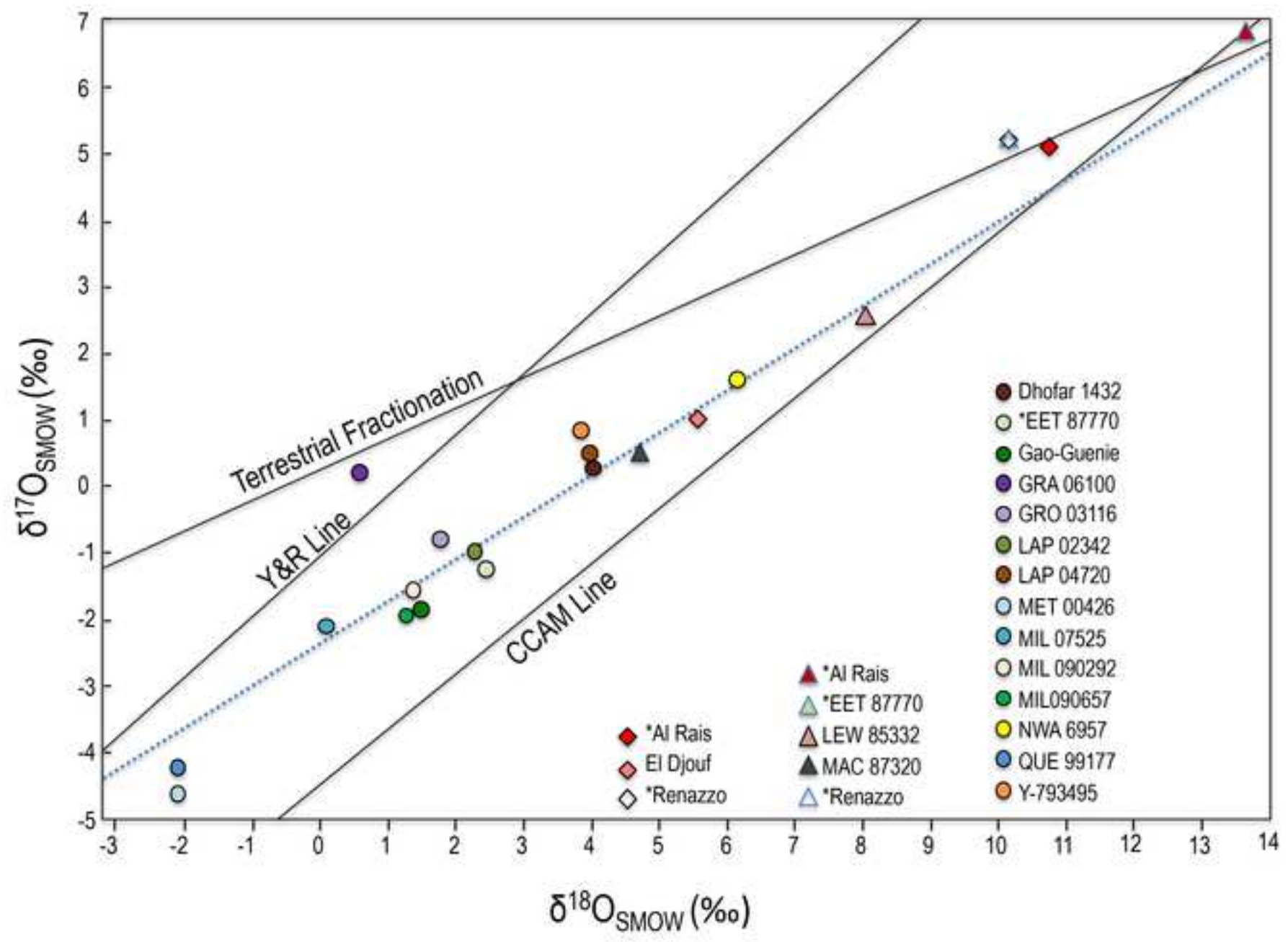
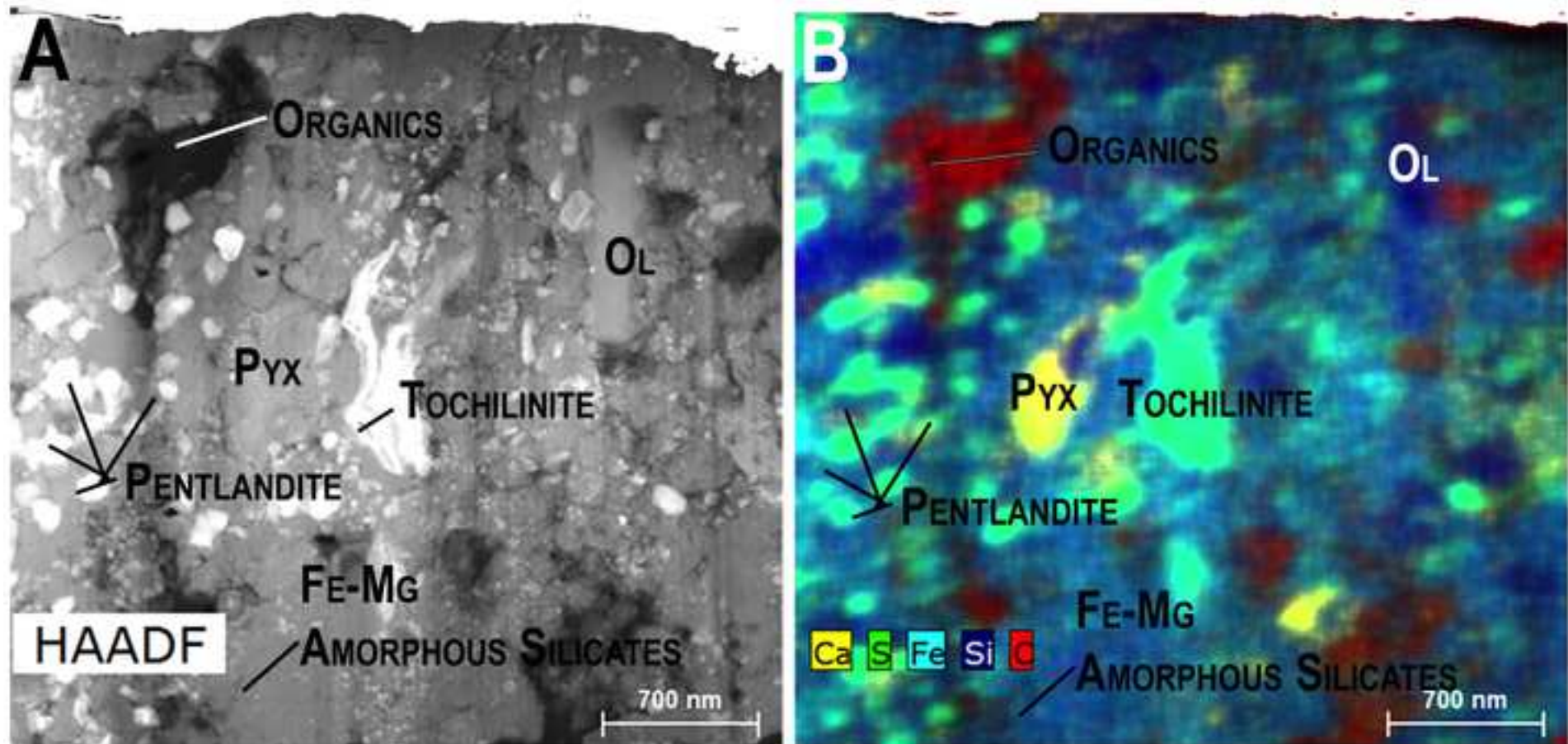
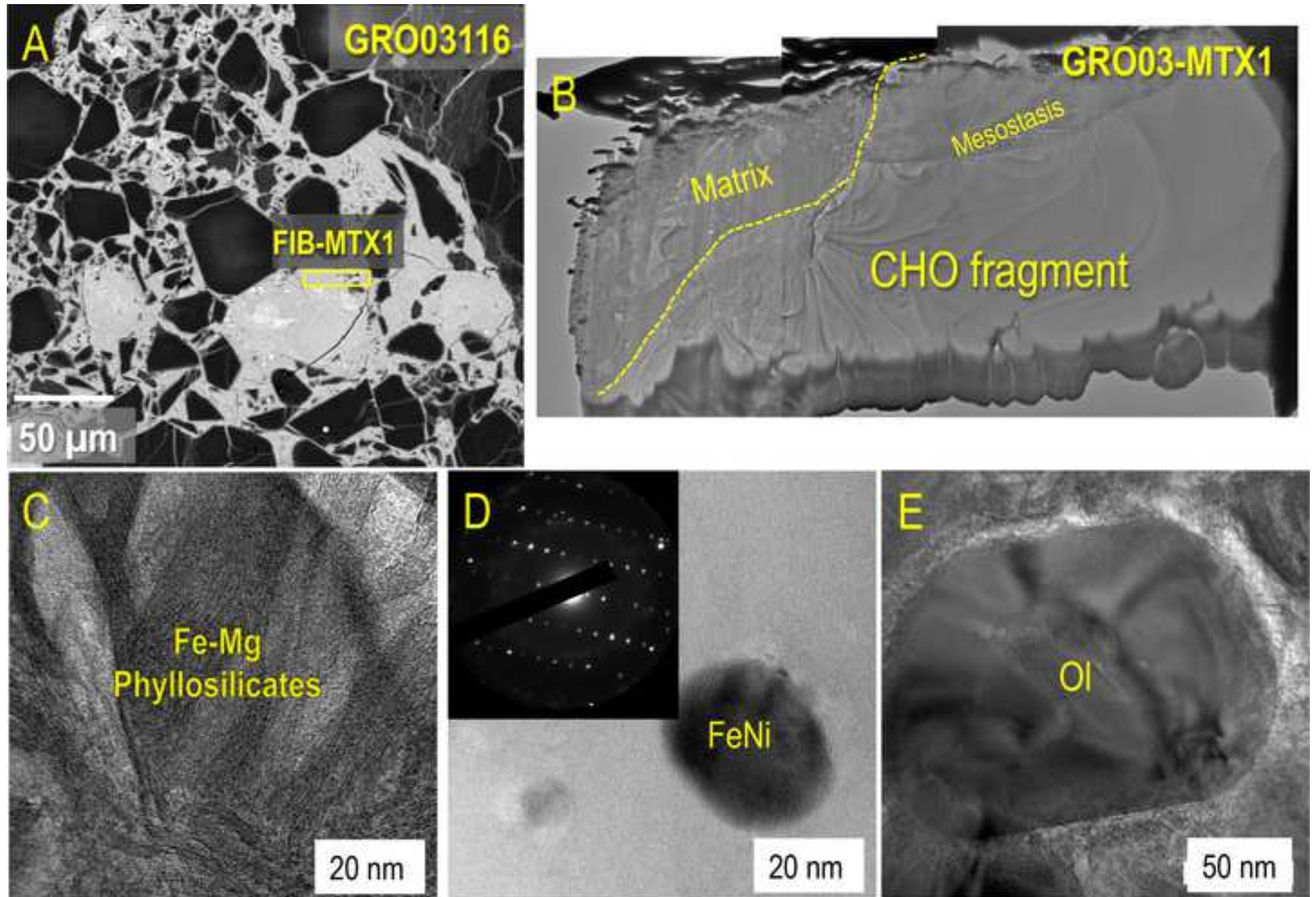


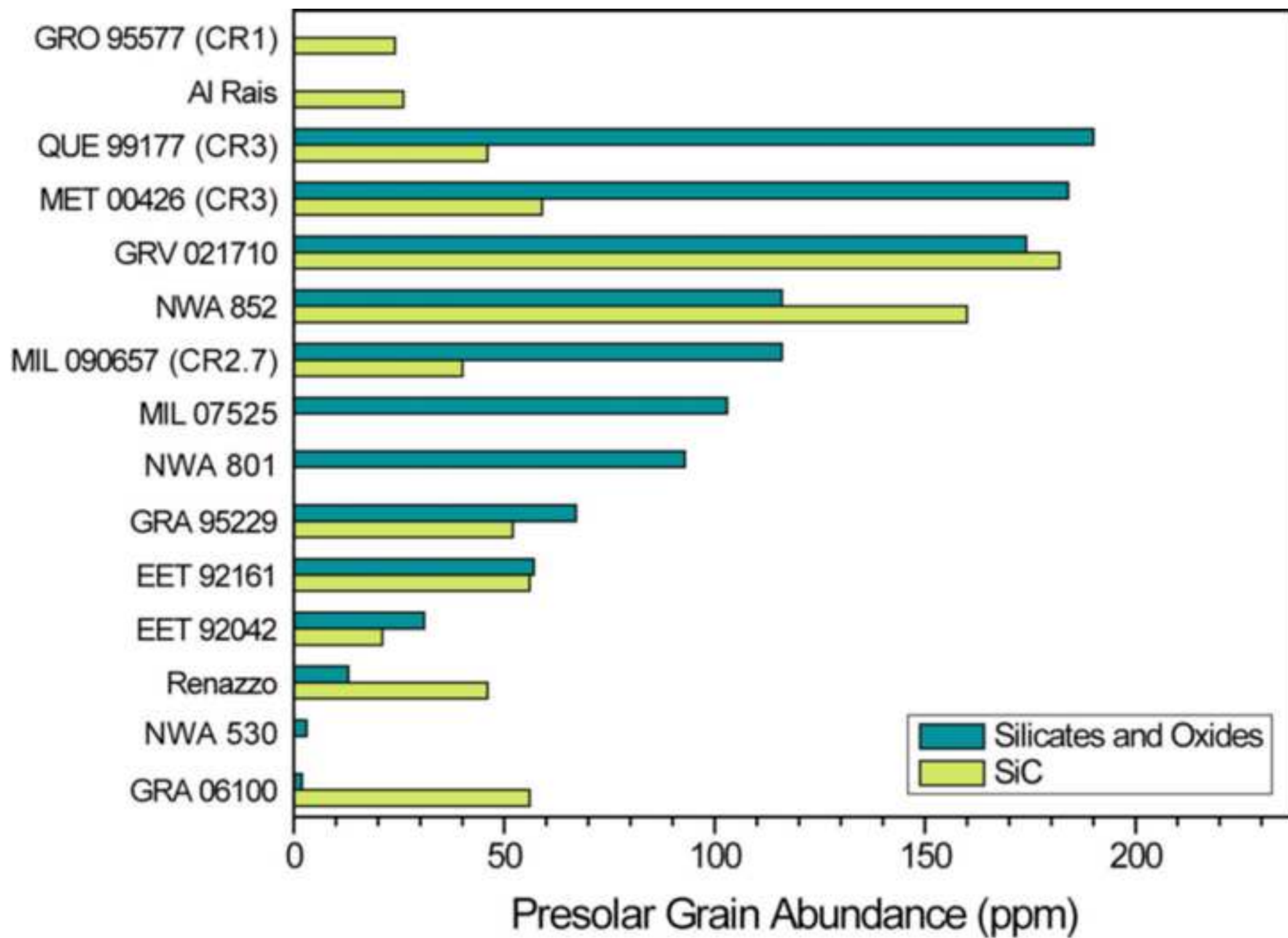
Figure 7











# AQUEOUS ALTERATION OF CR CARBONACEOUS CHONDRITES

CR1

CR2

CR3

Higher abundance of isotopically exotic organics and minerals formed by stellar and interstellar processes

## Aqueous Alteration

- Replacement of chondrule ol & en phenocrysts by phyllosilicates
- Complete replacement of Fe-Ni by layered oxides
- Extensive formation of carbonates

- Replacement of matrix mineralogy by oxides, phyllosilicates, sulfides, sulfates, carbonates
- Partial/complete replacement of mesostasis
- Partial replacement of Fe-Ni by layered oxides

- Widespread nanophase sulfides & amorphous silicates in matrix
- Nanocrystalline and/or glassy CHO mesostasis
- Intact Fe-Ni metal



Impact Events and associated deformation and hydrothermal activity

Materials range in size from mm to nm => Imaging and chemical characterization resolution gap between SEM/EMPA and TEM techniques

