

1 VolKilau: **Volcano** rapid response balloon campaign during the 2018 **Kilauea** eruption

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11 A Rapid balloon deployment during the 2018 Kilauea eruption provides a unique set of *in situ*
12 data to understand volcanic plume, chemical, and microphysical properties

13 **Abstract**

14 After nearly 35 years of stable activity, the Kilauea volcanic system in Hawai'i, went
15 through sudden changes in May 2018 with the emergence of 20 volcanic fissures along the
16 Lower Eastern Rift Zone (LERZ) destroying 700 homes in Leilani Estates and forcing more than
17 2000 people to evacuate. Volcanic emissions lasted for several months between June and
18 September 2018 leading to low visibility and poor air quality in Hawai'i and across the Western
19 Pacific. The NASA-funded VolKilau mission was rapidly mounted and conducted between 11-
20 18 June 2018 to: i) Profile volcanic emissions with SO₂ and aerosol measurements, ii) Validate
21 satellite observations and iii) Be prepared for the next large volcanic eruption. Through a series

1 of balloon-borne measurements with tethered and free released launches, we measured SO₂
2 concentration, aerosol concentration and optical properties 60 to 80 km downwind from the
3 volcanic fissures using gas sensors, optical particle counters, backscatter sondes, and an aerosol
4 impactor. While most of measurements made during the Kilauea eruption were ground-based,
5 the VolKilau mission represented a unique opportunity to characterize plume properties,
6 constrain emission profiles, study early chemistry involving the conversion of SO₂ into sulfuric
7 acid and understand the influence of water clouds in the removal of SO₂. Our team continues to
8 be better prepared for deploying in the event of a major volcanic eruption.

9 1. Introduction.

10 Volcanic eruptions have shaped life on the Earth through the combination of geologic
11 and atmospheric impacts. Major eruption can cool the climate system for several years (Robock
12 et al., 2000) from the veil of aerosols produced in the stratosphere (McCormick et al., 1995) that
13 reduces the amount of solar radiation reaching the surface. Tropospheric effusive eruption can
14 strongly deteriorate air quality with the release of sulfur-bearing gas and aerosols (Hansell et al.,
15 2004; 2006) and potentially affect the 450 million people worldwide living in proximity of
16 volcanoes (Small et al., 2001). The Laki eruption in 1783-84 led to an increase of mortality in
17 Iceland by >20% and devastating impacts on air quality and crop production throughout Europe
18 (Grattan et al., 2003; Witham and Oppenheimer, 2004). Cardiorespiratory and other acute health
19 effects have been observed on the people of Hawai'i from the recent eruptions of Kilauea (Longo
20 et al., 2013). Improving forecast system of dangerous life-threatening sulfur gases and
21 Particulate Matter (PM) is key to limit local population exposure. The complex physico-chemical
22 processes involved in the evolution of volcanic plumes from the few seconds after their
23 emissions start are not well understood. Observations of sulfate aerosols near volcanic sources

1 (Mather et al., 2003; Kroll et al., 2015) suggest rapid SO₂ oxidation and/or the production of
2 sulfate itself in the magma (Mather et al., 2003). Roberts et al. (2019) recently showed through
3 kinetic-based simulations that high temperature chemistry quickly forms oxidants such as OH,
4 HO₂ and H₂O₂ that can lead to the production of sulfate within a few second after the emission of
5 SO₂ starts. The aqueous pathways for sulfate formation have been studied with photochemical
6 model including isotope markers showing that TMI-O₂ could indeed dominate the formation of
7 sulfate (Galeazzo et al., 2018) relative to other pathways through H₂O₂ and O₃, which can be
8 rapidly titrated at low pH. In addition, the production of sulfate through new particle formation
9 has been shown to be more frequent under the influence of volcanic emissions compared to
10 background level conditions (Rose et al., 2019, Boulon et al., 2011, Sahyoun et al., 2019).
11 Finally, halogen chemistry leading to ozone depletion has been observed in volcanic plumes
12 (Surl et al., 2015) that could affect oxidation processes. Therefore, measurements in fresh and
13 aged volcanic plumes are extremely important to understand the different pathways of SO₂
14 oxidation from OH in gas-phase and/or H₂O₂, O₃ and O₂ in liquid phase (Galeazzo et al., 2018).

15 Balloon-based measurements in fresh volcanic plumes are very limited but can provide
16 essential information on the vertical extension of SO₂ and aerosol concentration when the plumes
17 are lofted above the ground due to increased buoyancy. Pieri et al. (2013) reported one of the
18 first blimp and tethered balloon coincident observations of SO₂ and aerosol concentration at the
19 Turrialba volcano in Costa Rica to validate satellite observations from the Advanced Space
20 borne Thermal Emission and Reflection (ASTER). Balloon-borne measurements during the
21 Holuhraun eruption in Iceland (Vignelles et al., 2016) provided one of the first aerosol size
22 distribution of volcanic aerosols in very fresh lofted (age ~15 min) plume.

1 NASA Langley Research Center (LaRC) scientists teamed up with researchers from
2 Universities of Colorado and Costa Rica to rapidly deploy gas-aerosol-instrumented balloons
3 during the **Volcano** rapid response balloon campaign during the **Kilauea** eruption (VolKilauea).
4 Through a series of 8 balloon flights in Hawai‘i between June 11th and June 18th 2018, we
5 measured the atmospheric composition in the outflow plumes from the volcanic fissures and the
6 Halema‘uma‘u main crater.

7 The purpose of this paper is to describe the preparation, the deployment and preliminary
8 results obtained during VolKilauea. With SO₂ and aerosol measurements, we characterized the
9 Kilauea volcanic plumes with high vertical resolution balloon data. We show how our
10 measurements can be used to validate satellite observations and model dispersions and help shed
11 light on microphysical plume evolution.

12 2. The 2018 Kilauea eruption

13 2. a. Sudden geological changes of the Kilauea volcanic system

14 The state of Hawaii forms a chain of islands created over the last 60 million years through the
15 connection of a hotspot bringing magma from the inner mantle. Hawai‘i is the youngest of the
16 archipelagos and Kilauea the most recent volcano which started to erupt 800 000 years ago. After
17 nearly 35 years of stable activity, in May 2018, the Kilauea volcano erupted from 20 new
18 fissures and sent lava flowing over streets and neighborhoods (Neal et al., 2019). Wiping out 700
19 homes and moving more than 2000 people, no US eruption was as disruptive since Mount St
20 Helens blew ash up to the stratosphere in 1980. The 2018 Kilauea eruption marked profound
21 changes of the volcanic system with the collapse of the Pu‘u ‘Ō‘ō crater, the drain of the lava
22 lake of the Halema‘uma‘u crater, also called Kilauea summit, and the emergence of volcanic

1 fissures along the Lower Eastern Rift Zone (LERZ) (Figure 1). Following a 6.9 magnitude
2 earthquake, volcanic fissures opened in the Leilani Estates area on May, 3rd and more than 20
3 fissures were reported the following weeks. The United States Geological Survey monitored the
4 extension of the lava fields through thermal Infrared camera mounted on helicopters (Neal et al.,
5 2019). In addition, NASA flew the Gulfstream III (G-III) to assess the volume of lava spilled out
6 by the volcanic fissures with a Synthetic Aperture Radar from the Jet Propulsion Laboratory.
7 Aside from lava destruction, volcanic fissures continuously emitted 30-60 kt of sulfur dioxide
8 (SO₂) in the atmosphere, which turned into sulfuric acid droplets forming the so-called volcanic
9 fog (vog) especially pronounced on the Southwestern side of the Island and over the Western
10 Pacific (Fig.3).

11 2. b. SO₂ emissions during the Kilauea eruption

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13 The first of the 20 volcanic fissures from the 2018 outbreak was observed on May 3rd in
14 the Leilani Estates area with levels of SO₂ threatening air quality safety with violation of the
15 National Ambient Air Quality Standard for SO₂ and concentrations greater than 75 ppb. SO₂
16 emitted from the Kilauea summit and the LERZ have been measured daily by the Ozone
17 Mapping Profiling Suite (OMPS). OMPS nadir mapper is a hyperspectral ultraviolet
18 spectrometer (Flynn et al., 2014; Seftor et al., 2014) on board the NASA-NOAA National Polar
19 Partnership (NPP) satellite measuring SO₂ total vertical column density in Dobson Units (1 DU
20 = 2.69×10^{16} SO₂ molecules per cm²) using Principal Component Analysis (PCA) algorithm
21 (Zhang et al., 2017; Li et al., 2017). An assumed SO₂ profile shape, represented by its center of
22 Mass Altitude (CMA) near 2.5 km was used to derive the lower tropospheric (TRL) SO₂ map
23 shown in Fig.2 (top) (Ialongo et al., 2015). Figure 2 (bottom) shows a time series of total daily

1 mass of SO₂ (in kilotons: kt) between June and September 2018 (daily OMPS SO₂ maps and
2 tonnages are available on-line at https://so2.gsfc.nasa.gov/pix/daily/0718/hawaii_0718p.html).

3 The Kilauea eruption lasted for 3 months between May and August 2018 (Fig.2) with a
4 maximum of 60 kt of SO₂ emitted on June 15th as inferred from OMPS. We note that this
5 estimate could be affected and underestimated several factors such as retrieval algorithm,
6 instrument resolution and the presence of clouds. The SO₂ plumes were transported by trade
7 winds and affected air quality across Hawai‘i and Western Pacific (Fig.2/bottom).

8 During the 2018 eruption at Kilauea’s LERZ, USGS reported SO₂ emissions over 30kt per day, a
9 value that could be affected by instrument saturation and wind speed assumptions
10 ([https://www.usgs.gov/faqs/how-much-sulfur-dioxide-so2-gas-does-k-lauea-emit?qt-](https://www.usgs.gov/faqs/how-much-sulfur-dioxide-so2-gas-does-k-lauea-emit?qt-news_science_products=0#qt-news_science_products)
11 [news_science_products=0#qt-news_science_products](https://www.usgs.gov/faqs/how-much-sulfur-dioxide-so2-gas-does-k-lauea-emit?qt-news_science_products=0#qt-news_science_products)). Figure 3 shows steam and gas emitted by the
12 volcanic cone produced by Fissure 8. The subsequent transformation of SO₂ into sulfate aerosols,
13 the vog is observed on a photo taken 20 km north from Kona on June, 10th. The large extension
14 of the vog is also evident on the true color image from the VIIRS/NPP on June, 7th characterized
15 by a grey-blueish veil observed hundreds of miles downwind from Hawai‘i (Fig.3). While the
16 impacts of the Kilauea eruption on aviation was minimal, the vog presented a serious threat to air
17 quality across the Island and largely affected the tourist Industry (Neal et al., 2019).

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1 3. Campaign preparation and planning

2 Volcanic eruptions are by nature challenging to study. Being quasi-unpredictable and
3 sudden, scientists often rely on satellite observations to assess their atmospheric impacts.

4 Extreme conditions associated with volcanic eruptions make field measurements extremely
5 dangerous and thus rare. Finally, the composition of volcanic emissions, their dependency with
6 geological processes and the different stages of plume physico-chemical processing when
7 entering the atmosphere result in complex atmospheric impacts. Heat buoyancy created near
8 volcanic sources generate aloft plume that cannot always be captured by ground-based
9 measurements enhancing the value of airborne deployments.

10 Our team is involved with the NASA Disaster program ([https://disasters.nasa.gov/kilauea-hawaii-
11 eruption-2018](https://disasters.nasa.gov/kilauea-hawaii-eruption-2018)) and the Volcano Response (VolRes) international initiative to deploy during and
12 after volcanic eruptions. Within 15 days, VolKilau was on the ground of Hawai'i, nearly 60 km
13 downwind from fissures and 20 km from the Halema'uma'u crater to study SO₂ and aerosol
14 emissions from the Kilauea eruption. We deployed a variety of balloon measurement payloads
15 just downwind from the eruptive vents to profile SO₂ and aerosol concentrations in the plume to
16 improve model simulations and validate satellite observations SO₂ and aerosols. We used
17 forecast maps of SO₂ provided by the University of Hawaii (Businger et al. 2015) to choose a
18 deployment location and sample the outflows of Kilauea emission zones. The NOAA Hybrid
19 Single Particle Lagrangian Integrated Trajectory (HYSPLIT) model (Draxler and Hess (1997);
20 Stein et al. 2015) was initialized between 50 and 700m above the local terrain to account for
21 buoyancy created at the emission sources. The volcanic plumes originated from three different
22 locations (Fissure 8 in the LERZ, Pu'u 'Ō'ō crater and Halema'uma'u crater). Fissure 8 became
23 the largest source of SO₂ with flux between 30-60 kt/day a few weeks after the beginning of the

1 fissure outbreak, forming a volcanic cone (photo, Fig.3). The Pu‘u ‘Ō‘ō crater, which rapidly
2 collapsed after the new fissures started erupting became a minor source. However, the Kilauea
3 summit occasionally produced ash clouds up to 10 000 ft and was an additional relatively small
4 source of SO₂ compared to Fissure 8.

5 After exploring several options remotely, it became obvious that choosing a location near the
6 Leilani Estates area could be dangerous given the potential high levels of SO₂ measured on the
7 ground at the proximity of the fissures but also complicated since the area was restricted to only
8 emergency responders and local people. The best option would become to deploy downwind
9 from the Fissure 8 where trade winds were pushing volcanic emissions toward us at acceptable
10 concentrations for outdoor activities. The Kapapala ranch (19.263, -155.445), located along road
11 11 between the Kapapala forest reserve and the Hawai‘i Volcano national Park (See Fig.1)
12 became a viable option given past deployment experience of the team.

13 4. The VolKilau campaign

14 a. Concept

15 The VolKilau campaign was designed to study the atmospheric outflow from the Kilauea
16 volcanic emissions with gas and aerosol measurements. The objectives of this mission were to i)
17 profile volcanic emissions, ii) validate satellite observations and iii) be prepared for the next
18 major volcanic eruption. The first and second objectives relied on obtaining vertically resolved
19 data of SO₂ and aerosols through the atmosphere. VolKilau was based on three types of
20 ballooning activities (Fig.4) : 1) Free released balloon to sample volcanic outflows from the
21 fissures but also high altitude ash cloud injected occasionally from the Kilauea summit, 2) Low-
22 level cut down balloon flight to optimize the measurements in the plume and terminate the flight
23 early for recovery and 3) Tethered balloon activities to sample the volcanic plume below 2 km.

1 b. Payloads

2 The payloads selected for the campaign measure SO₂ and aerosol optical, physical and chemical
3 properties. It included:

4 **LASP Light Weight OPC (ROPC):** A lightweight and low-cost balloon borne optical particle
5 counter, based on a commercially available optical head (Met One 9722) integrated with a flow
6 control system and radio sonde telemetry system. This is the smallest of the balloon borne OPCs
7 developed at LASP and is well suited to measurements in the troposphere with relatively high
8 aerosol loadings. The size range in radius is between 0.15 – 5 µm, with number concentration
9 ranging from 1 – 3x10⁵ #/l. The ROPC reaches a coincident counting threshold near 300 000
10 particles per liter. Above this limit, there is a statistically significant probability that more than
11 one particle may pass through the laser beam at one time, leading to an undercounting of
12 particles. While the instrument will still respond to particle concentrations above the coincidence
13 threshold, the measurement becomes qualitative in nature due to the increasing probability of
14 coincident particles. Mather et al. (2003) reported sulfate size distributions in the plume of the
15 Masaya volcano with a peak near 0.15 µm radius. Thus, our OPC measurements do not fully
16 capture the entire size distribution of sulfate and therefore likely provide lower limit estimates of
17 total aerosol number concentration.

18 **LOPC :** Another lightweight OPC developed at NASA Langley and adapted for weather balloon
19 flight applications, also using the MetOne 9722 particle counter with a small external pump from
20 *Particle Plus, Inc.* The data logging is based on Raspberry PI and is compatible with an Imet
21 meteorological radiosonde. The OPC includes 8 channels with sizes between 0.15 – 10 µm and
22 the measurement range is similar to the ROPC.

1 **COBALD** is a lightweight (~500 g) balloon-borne backscatter sonde developed at the Institute
2 of Atmospheric and Climate Science, Swiss Federal Institute of Technology (ETH), Zurich.
3 COBALD consists of two high power (250 mW) Light Emitting Diodes (LEDs) which emit light
4 at 455 nm (blue) and 940 nm (near infrared) wavelengths. A silicon photo-diode placed between
5 the two LEDs collects the light backscattered from air-borne particles (molecules, aerosols and
6 clouds). When connected with an iMet radiosonde, the instrument transmits backscatter counts at
7 both of these wavelengths, pressure, temperature, relative humidity, wind speed and wind
8 direction at 1 Hz to the receiving ground station. The concurrent profiles of pressure, temperature
9 and backscattered counts are used to calculate Scattering Ratio (SR), the ratio of total
10 backscattering coefficient ($\beta = \beta_p + \beta_m$) to the molecular backscattering coefficient (β_m) at a
11 given wavelength. The ratio of $(SR_{940}-1)$ at 940 nm to $(SR_{455}-1)$ at 455 nm, defined as the color
12 index (CI), gives qualitative information about particle size (Cirisan et al., 2014; Vernier et al.,
13 2015).

14 **MiniGAS** system is an integrated suite of instruments for UAV applications including
15 meteorological (pressure, temperature) and up to 5 chemical sensors. The system uses a
16 microcomputer, a data logger and telemetry system from which gas concentrations can be
17 displayed in real time on a laptop (Diaz et al., 2015). The miniGAS PRO Version (Fig. 5) is
18 designed for volcanic gas measurements on small UAVs, multicopter drones and tethered
19 balloons/blimps for < 1.5km agl altitude flights. The miniGAS-PRO was deployed during the
20 2018 VolKilauea campaign for measuring SO₂ concentration from the Kilauea volcanic plume. The
21 instrument collected SO₂ (0-20ppm range) and CO₂ (0-2000ppm range) on two tethered flights
22 coincidentally with OPC measurements. SO₂ and CO₂ measurements are based on electrochemical
23 and Non Dispersive InfraRed (NDIR) techniques, respectively. A 0.2 μm PTFE teflon filter is

1 included at the entrance of the sensor to limit the intrusion of droplets inside the miniGAS and
2 affect the measurements. However, we cannot completely rule out potential measurement
3 artefacts at high relative humidity that may affect *in cloud* measurements.

4 **Ozonesonde** is a payload using the Electro Chemical Cell (ECC) measurement technique to
5 retrieve ozone concentration.

6 Ozone reacts with potassium iodide following: $2KI + O_3 + H_2O \rightarrow 2KOH + I_2 + O_2$ (1)

7 anode reaction: $3I^- \rightarrow I_3^- + 2e^-$ (2)

8 cathode reaction: $I_2 + 2e^- \rightarrow 2I^-$ (3) (exposed to ambient air and the presence of O_3)

9 However, SO_2 interferes with those measurements and reacts with water following the reaction:

10 $SO_2 + 2H_2O \rightarrow 2e^- + SO_4^{2-} + 4H^+$ (4)

11 Each molecule of SO_2 produces 2 electrons, replenishing the cathode (3) and reducing the current
12 between the anode and the cathode. The interference of SO_2 has been used to infer SO_2 information
13 (Morisson et al., 2010) using one sonde with and without an SO_2 filter. With the DualSonde
14 technique, SO_2 is inferred by subtracting from the unfiltered signal ($O_3 + SO_2$), the filtered signal
15 (O_3). During the campaign, we attempted to use the Dualsonde technique but quickly realized that
16 the levels of SO_2 were too high and led to the saturation of the signal. Thus, we used the
17 ozonesonde only has an indicator to find the bottom (during the ascent) and top (during the descent)
18 of the volcanic plume.

19 **Impactor.** Volcanic aerosol composition is investigated using an aerosol impactor, which was
20 initially prepared for tethered balloon applications but only run on the ground because of weight
21 limitations from Federal Aviation Authorities regulations and technical constrains. The MPS-3
22 Microanalysis Particle Sampler, a 3-stage impactor, was connected to a small air pump

1 delivering 1.4 lpm of flow and was run for 30 min to sample 42 l of air. Inertial impaction type
2 of sample collector is associated with particle size cut-points of 1.4 μm (Stage-1,S-1), 0.2 μm
3 (Stage-2, S-2) and down to 0.05 μm (Stage-3, S-3). After collection, the filters were placed in
4 2ml sterile polypropylene vials with deionized water (resistivity: 18.2 Ωcm^{-1}) and extracted for 3
5 intervals of 10 minutes each, by ultrasonication. The subsequent solutions were analyzed by ion
6 chromatography (DIONEX CS 5000) to detect water-soluble cations (Na^+ , NH_4^+ , Mg^{2+} , K^+ ,
7 Ca^{2+}) and anions (Cl^- , NO_3^- , SO_4^{2-} , NO_2^-) using a cation column (DIONEX IonPacTM CS16) and
8 anion column (DIONEX IonPacTM AS23), coupled to their respective suppressors CERS/AERS
9 500, before the conductivity detector (Thermoscientific P/N 061830) (Patel and Rastogi, 2018).

10 c. Balloon Flights

11 The VolKilau campaign consisted of 8 balloon flights between June 11th and June 18th (table 1)
12 One flight took place from the Whittington Beach (19.086, -155.550) and the rest from the
13 Kapapala Ranch (19.263, -155,445). We launched 4 free-released, 4 tethered and one low-level
14 cut flights during this time frame. The trajectories of the free released balloon flights are shown
15 in Fig.6.

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17 d. HYSPLIT trajectory model description

18 A custom version of the HYSPLIT model, referred to as the ‘vog model’, was run in real-time at
19 the University of Hawai’i at Mānoa to produce forecasts of SO_2 and sulfate aerosol (Businger et
20 al. 2015). The meteorological model input is a custom version of the Weather Research and
21 Forecasting – Advanced Research Weather (WRF-ARW) model with 900 m horizontal
22 resolution over the main Hawaiian Islands (Cherubini et al. 2008). The vog forecasts rely on

1 emission rates of SO₂ from the U.S. Geological Survey and Hawaiian Volcanoes Observatory.
2 Prior to the Lower East Rift Zone (LERZ) eruptive event, the vog model incorporated real-time
3 SO₂ emission estimates as weekly averages that were updated on a daily basis. These estimates
4 were made using a UV spectrometer array (Elias et al. 2018) and acted as the primary source
5 term for volcanic emissions in the vog model. The LERZ event interrupted operation of the UV
6 spectrometer array. Less frequent estimates made with mobile instrumentation were used instead
7 with DOAS UV spectrometer. The frequency of updates had a direct and notable impact on vog
8 model forecasts of concentrations (L. Holland et al., 2019, accepted).

9 5. Campaign preliminary results.

10 The VolKilauea campaign dataset is analyzed together with satellite measurements of SO₂ and
11 aerosols, and simulations using the HYSPLIT trajectory model.

12 a. Free release flight into the Kilauea plume

13 After a tethered balloon test flight on June, 11th from the Kapapala Ranch, the first flight of the
14 campaign took place on June, 13th near 3am UTC with a combined flight with an Optical Particle
15 Counter (OPC) and an ECC for SO₂ information. Since SO₂ and wind forecasts showed that the
16 volcanic plume was further south than the Kapapala Ranch, we decided to move our equipment
17 along the shore of Hawai'i near the Whittington Beach (19.086, -155.549). Figure 7 shows
18 aerosol concentration profiles for $r > 0.15 \mu\text{m}$, $0.25 \mu\text{m}$, $0.35 \mu\text{m}$, $0.5 \mu\text{m}$, $1 \mu\text{m}$, $1.5 \mu\text{m}$, 2.5
19 and $5 \mu\text{m}$ associated with Relative Humidity (RH) and temperature profiles. Corresponding true
20 color image from VIIRS as well as SO₂ lower tropospheric column map from OMPS instrument
21 3-4 h before the balloon flight are also shown. The SO₂ cloud is seen over Hawai'i and extending
22 further South-SouthWest across the Pacific Ocean. The true colored image shows a white cloud

1 (delineated by yellow contours) originating from Fissure 8. The ozone profile is shown to
2 decrease rapidly from 0.02 ppmv to negative values near 1.3 km high due to the presence of SO₂
3 , consistent with the base of the volcanic plume. To reduce SO₂ in the ECC cell, it takes time, so
4 the signal remain below 0 for a good portion of the flight, even though it was already out of the
5 Plume as indicated by the OPC data. Near 1.3 km, the aerosol concentration for $r > 0.15 \mu\text{m}$
6 increases by a factor 10 within a few hundred meters, also indicating the base of the volcanic
7 plume. Volcanic aerosols are enhanced for all sizes when $\text{RH} > 100\%$ likely indicating the
8 presence of sulfate aerosol in saturated conditions. The volcanic cloud is observed between 1.5
9 and 3.5 km height with a maximum of concentration for $r > 0.15 \mu\text{m}$ near 2.5 km with values
10 reaching $3 \cdot 10^5 \text{ \#}/\text{l}$ where the RH is below 100 %. Above saturation, aerosol concentration of $r > 5$
11 μm peaks near $1 \cdot 10^4 \text{ \#}/\text{l}$ (near 2.2 km), nearly 4 orders of magnitudes larger than at 2.5 km. The
12 wind direction (40-60°) trace back the 1.5-3.5 km layer to volcanic emissions from Fissure 8, 80
13 km away and wind speed (10-11m/s) indicates a plume age of ~2.2 h. Relative variations of
14 aerosol concentration in function of RH reveals potential interaction between volcanic aerosols
15 and water clouds. Saturated levels seem to favor the growth and dissolution of sulfate aerosols in
16 liquid water clouds with large sulfuric acid droplets while sub-saturated conditions generate
17 smaller particles likely through new particle formation (Rose et al., 2019, Boulon et al., 2011)
18 from photooxidation and the growth of sulfate aerosols from SO₂.

19 b. High Resolution OPC data

20 We conducted a low-level balloon flight (<5 km) using a radio-controlled cut down system to
21 sample the plume and terminate the flight to optimize the chance of recovery of the payloads.
22 This balloon flight was launched on June, 14th near 1am from the Kapapala Ranch downwind
23 from volcanic emissions of fissure 8 (Fig.8) and nearly 2h after an overpass of the MODerate

1 resolution Imaging radio Spectrometer (MODIS) on the Aqua satellite. The flight was terminated
2 at 5 km height but despite the accurate location of the landing; we could not recover the
3 payloads. The ascent and descent profiles (Fig.8) show high concentration of aerosols up to 3 km
4 like the previous day with the presence of larger aerosols within the cloud. The volcanic plume
5 is capped by a trade wind inversion at 2.9 km with dry air from the free troposphere above this
6 level. Wind speed within the layer between 4-9 m/s from 0-70°N is consistent with its origin
7 from fissure 8, 2-4 h prior to our measurements. While the optical particle counter was at the
8 coincidence counting limit throughout the plume ($3 \times 10^5 \text{ l}^{-1}$) from 1.2-2.9 km, the measurements
9 qualitatively show significant structure within in the plume with a series of ~100m thick layers
10 from 1.9 – 2.5km on ascent with significantly varying aerosol size distributions. After the flight
11 was remotely terminated, the instrument descended through the volcanic plume, re-entering the
12 plume approximately 2km north of the ascent profile. This descent profile shows a similar
13 overall vertical extent to the plume, but the structure within the plume is significantly different.
14 Throughout the descent the RH in the plume was greater than 100% probably causing the
15 majority of the aerosol to grow to radii above $1 \mu\text{m}$ reducing the vertical structure within the
16 plume. The descent took place in a cloudy area as observed by MODIS (Fig.8) confirming the
17 presence of large water droplets as observed by the OPC. During the descent and below 2.5 km,
18 RH is blocked at 104% that may suggest instrument hysteresis issues in cloud interstitial
19 conditions. Both the internal structure observed during ascent profile and the differences between
20 the relatively closely spaced ascent and descent profiles reinforce the importance of considering
21 inhomogeneity within the plume as well as cloud/aerosol interactions.

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1 c. Satellite comparison

2 One of the objectives of the VolKilau campaign was to compare and validate satellite
3 observation using our balloon data. A nearly coincident balloon flight was planned at 6am UTC
4 on June, 15th with a nearby overpass of the CALIPSO satellite (within 200 km and 6h)
5 downwind from fissure 8 and our balloon measurements. The measurements included an ECC, a
6 backscatter sonde (COBALD) and an Optical Particle Counter (OPC) shown in Figure 9. The
7 ECC saturated with high concentration of SO₂ (black dotted line near 1.2 km, Fig9b), thus the
8 unfiltered ECC data went to Zero coincidently with an increase of aerosol concentration. A
9 second aerosol layer is observed near 4-4.5 km on both aerosol instruments consistent with the
10 volcanic plume. Both layers are correlated with increasing RH between 40-95 % a consistent
11 with expected aerosol growth in humid environment as observed near the Etna volcano (Roberts
12 et al., 2018). Those moist layers very likely originate from fissure 8 with wind speed near 3-7
13 m/s coming from the 0-60°N direction, yielding a plume age near 2-5h. The CALIPSO overpass
14 200 km downwind and 6 h also indicate a double layer, one in the boundary layer capping at 3
15 km and in the free troposphere near 4 km. SO₂ and AOD maps from OMPS/NPP and
16 MODIS/Terra/Aqua also indicates the dispersion of the volcanic cloud westward downwind
17 from fissure 8, across the Western Pacific. CALIPSO intersected the volcanic plume associated
18 with high SO₂ and AOD from MODIS and OMPS and coincident with high backscatter values
19 and low depolarization (not shown) consistent with the presence of liquid sulfate aerosols for
20 both layers. Wind speed and direction near 4 km height observed on the balloon flight (6 m/s,
21 50-60 degree) are consistent with the movement of air masses moving South-Southeast and
22 crossing the CALIPSO overpass after 6h. Those measurements indicate that volcanic materials
23 from fissure 8 had a complex vertical and horizontal distribution depending on weather

1 conditions and atmospheric instability. Reports from USGS indicated convective cloud forming
2 above fissure 8 that could have transported volcanic materials up to the tropopause level.
3 However, no trace of volcanic aerosols above 5 km were observed by satellites or during the
4 VolKilauea campaign. Nevertheless, those observations indicate that parts of Kilauea volcanic
5 plume were transported in the free troposphere above the Boundary Layer.

6 d. Volcanic plume chemical and microphysical changes.

7 Volcanic plume chemistry is investigated by combining temperature, RH, aerosols, SO₂ and
8 CO₂/SO₂ using a tethered balloon flight on June, 15th 2018 from the Kapapala Ranch (VK04).
9 The time series of those parameters between 23:16 and 23:45 UTC is shown in Figure 10
10 together with geolocation of the balloon flight on Google maps colored with SO₂ measurements.
11 Levels of SO₂ near the surface were approaching 0.3 ppmv and increased up to 0.5 ppmv at
12 23:31 just before entering a water cloud near 1 km asl as indicated by RH near 100% and a
13 sudden increase of particle concentrations for sizes between 0.5 to 10 μm. The tethered balloon
14 was brought down after encountering the clouds since FAA regulations required a continuous
15 sight of view with the balloon during those activities. Outside the water cloud, the RH dropped to
16 70% and SO₂ levels regain their levels near 0.5 ppmv while the concentration of aerosols
17 dropped. CO₂ concentration is nearly constant within the plume so that the increase of CO₂/SO₂
18 ratio observed in Fig.10 is driven by the sharp drop of SO₂ and likely suggest the absence of SO₂
19 within clouds. However, we cannot exclude that cloud droplets affected the miniGAS
20 measurements despite the presence of a PTFE filter at the entrance to prevent the intrusion of
21 sulfuric acid droplets. Coincident measurements of SO₂ and aerosols near the vent of volcano
22 likely reveal the interaction between water vapor, SO₂ and aerosols (Roberts et al., 2018). *In-*
23 *cloud* SO₂ decreased rapidly probably because of different aqueous pathways leading to the

1 formation of sulfate either from the oxidation of H_2O_2 (Carn et al., 2011), O_3 and/or the TMI-O₂
2 pathway (Galeazzo et al., 2018) even if we cannot rule out potential measurement artifacts of
3 SO_2 in high RH conditions. The vog was observed downwind from Fissure 8 across the Pacific
4 Ocean.

5 e. Volcanic aerosol composition

6 Volcanic composition is investigated using an aerosol impactor. We show the ionic composition
7 of two samples, HS1(Hawai'i Sample 1) and HS2(Hawai'i Sample 2) collected on June 12th and
8 14th 2018 at 5.30 am and 7.44am for 30 min with a flow of 1.4 lpm for a total of 42 l of air
9 sampled. Stage 1 and 2 for the two set of samples hold similar concentration of major ions in
10 aerosols between 3950 and 5900 ng/m³. Here, Na^+ and Cl^- ions dominate the total ionic mass
11 with a smaller contribution of SO_4^{2-} in stage 1 ($r > 1.4 \mu\text{m}$) and stage 2 ($1.4 \mu\text{m} < r < 0.05 \mu\text{m}$). The
12 major difference observed between the two samples is on stage 3 ($r \leq 0.05 \mu\text{m}$) where the
13 concentration of SO_4^{2-} and NH_4^+ aerosols are 6.2 and 7.9 times, respectively, higher in HS2
14 compared to HS1 (Fig. 11). Those measurements are compared with HYSPLIT simulations at
15 the time of the collections. HYSPLIT also shows much larger concentration of SO_2 (by a factor
16 of 2 to 10) during HS2 compared to HS1 consistent with favorable meteorological conditions
17 bringing volcanic material toward the measurement on June 14th compare to June 12th when trade
18 winds pushed volcanic aerosols offshore. Further, Cl^-/Na^+ ratios on stages 1 and 2 of samples
19 HS1 and HS2 were similar to their sea-salt ratio (1.8); however, on stage 3 it was about 3.3 i.e.,
20 ~2 times higher than expected from sea-salts. This observation suggests the presence of excess
21 Cl^- on the stage 3 in both the samples. Relatively high Cl^- on stage 3 indicates likely a significant
22 contribution of Cl^- from volcanic emissions resulting from the hydrochloric acid emissions from
23 lava entering the ocean (Resing et al., 1999). Further, non-sea-salt (nss) fraction of SO_4^{2-} ,

1 estimated using $\text{SO}_4^{2-}/\text{Na}^+$ ratio in sea-salt as a reference, was low on stages 1 and 2 but close to
2 100% on stage 3. We note also the presence of NH_4^+ on stage 3 of HS1 and HS2 bearing
3 somewhat similar relative concentration with non-sea-salt (nss)- SO_4^{2-} , indicating the presence of
4 SO_4^{2-} aerosol in the form of $(\text{NH}_4)_2\text{SO}_4$. While the presence of SO_4^{2-} is consistent with the rapid
5 conversion of SO_2 into sulfate aerosols in the troposphere, its apparent association with NH_4^+ is
6 interesting. The neutralization of acidic volcanic sulfate by ammonia has already been observed
7 after the 2011 Grimsvötn volcano 3000 km from the sources. This process is size dependent with
8 the neutralization efficiency enhanced for small particles (Šakalys et al., 2016). Our results also
9 suggest that the neutralization of sulfate near the ground was dominant for particles smaller than
10 50 nm diameter. However, the level of neutralization was much larger than those found by Kroll
11 et al. (2015) which reported a ratio $\text{SO}_4^{2-}/\text{NH}_4^+$ between 10-20 at Pahala less than 10 km away
12 from our measurement site. The proximity of livestock could eventually explained the enhance
13 levels of neutralization. However, a significant fraction of SO_4^{2-} was present as free acid as
14 indicated by observed mass ratio of sulfate/ammonium (3.6) in comparison to expected mass
15 ratio (2.7) if all sulfate would have existed as $(\text{NH}_4)_2\text{SO}_4$.

16 6. Conclusion

17 The 2018 Kilauea eruption was the most significant volcanic event in the US since Mt St Helen
18 erupted almost 4 decades ago. Our team deployed quickly in Hawai‘i to i) obtain vertically
19 resolved profiles of SO_2 , aerosol and meteorological parameters, ii) validate satellite
20 observations and iii) be better prepared for the next large volcanic eruption. We deployed within
21 15 days in Hawai‘i between June 11th and June 18th 2018 with 8 flights through tethered-, low-
22 level cut and free released balloon activities from a Ranch downwind from active volcanic
23 fissures and the Kilauea main crater. We flew multiple times in the volcanic plume from Fissure

1 8 emissions and found aerosol number concentration up to 3×10^5 #/l. Coincident measurements
2 between SO₂, aerosol concentration and RH suggest rapid conversion of SO₂ into aerosols in
3 supersaturated conditions leading to the formation of the vog. Satellite observations and balloon
4 data demonstrates the complexity of volcanic emissions with multiple volcanic layer observed up
5 to 4.5-5 km asl in the free troposphere. Finally, chemical analysis of ground-based samples
6 shows the recurrent presence of Sea Salt (NaCl) together with volcanic sulfate. The mass
7 concentration of sulfate can vary by a factor of 6 within a few days depending on wind
8 conditions as revealed by HYSPLIT simulations. Excess of Cl⁻ relative to the Cl⁻/Na⁺ ratio of
9 sea salt reveals the influence of hydrochloric acid emissions from lava entering the ocean
10 (Resing et al., 1999). The presence of ammonium in the samples suggest that acidic volcanic
11 sulfate are rapidly neutralized with the ammonia released by livestock, which were at the
12 proximity of the measurement location. The VolKilauea mission represented a unique opportunity
13 to combined SO₂ and aerosol measurements to understand aerosol microphysics in the Kilauea
14 volcanic plume. We tested instruments in the tropospheric volcanic plume environment so that
15 they can be used during future large volcanic eruptions.

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2 Disasters programs by K. Jucks, J. Kaye and D. Green at NASA HQ; We thank the owners of the
3 Kapapala Ranch for using their property during the field mission. We thank Tara for using her
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5 Kapapala Ranch owner. We thank Jason from EPA for helping us get access to the restricted area
6 in Leilani Estates.

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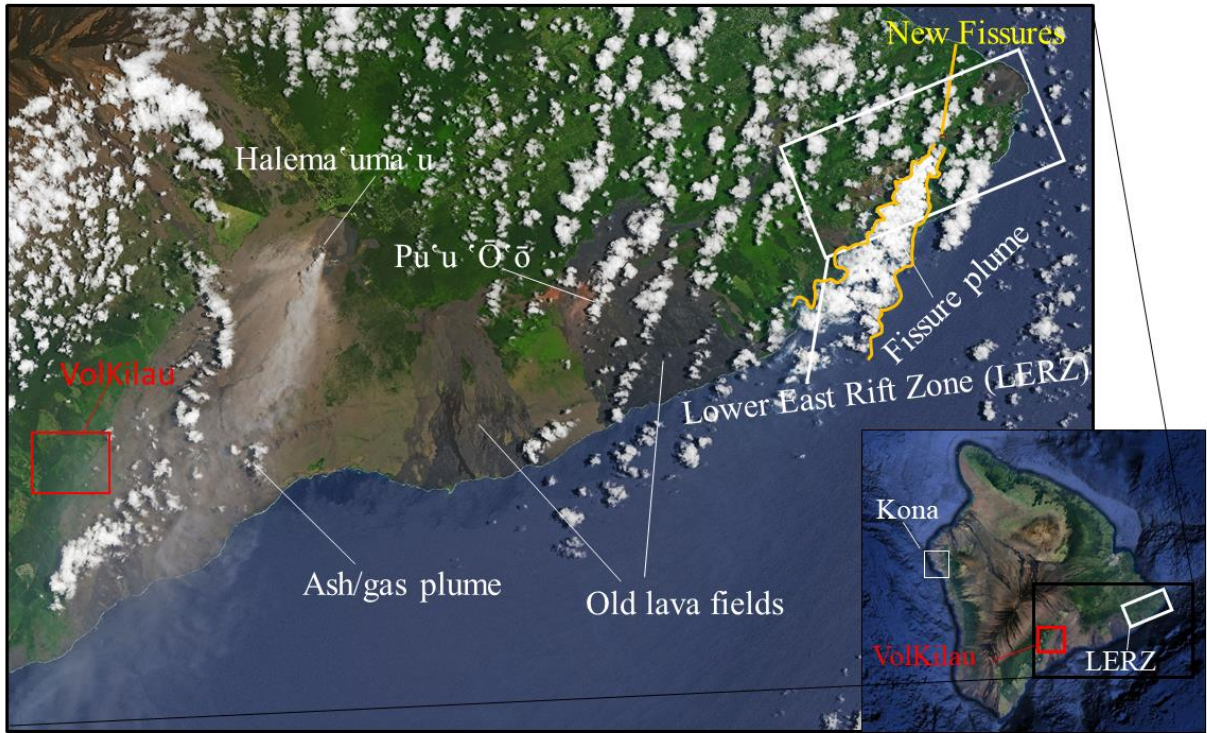
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1 8. Figures



2 9.

3 *Figure 1. True Color image from the NASA/USGS Landsat 8 Operational Land Imager on May,*
4 *14th 2018 in south Hawai'i. The Pu'u 'Ō'ō crater, Halema'uma'u summit, Lower East Rift Zone*
5 *(LERZ) and the location of the VolKilau campaign are shown.*

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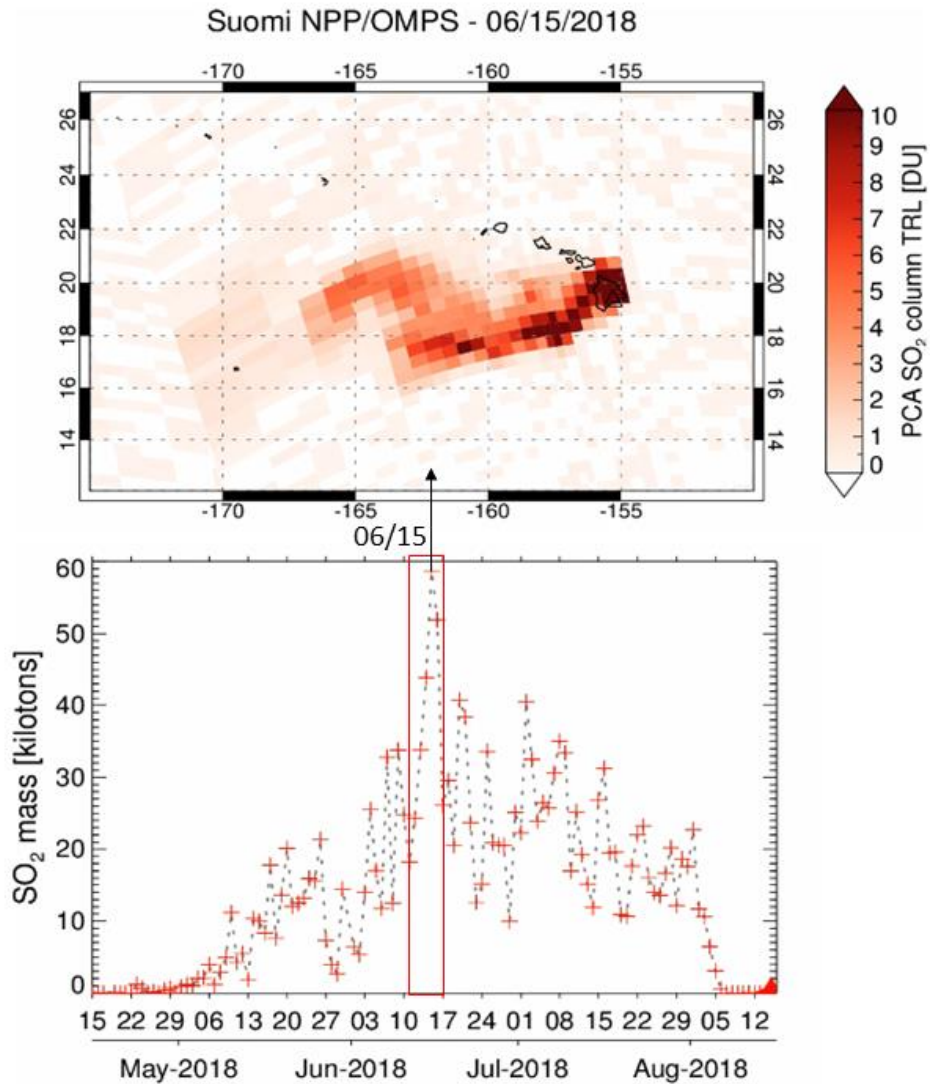
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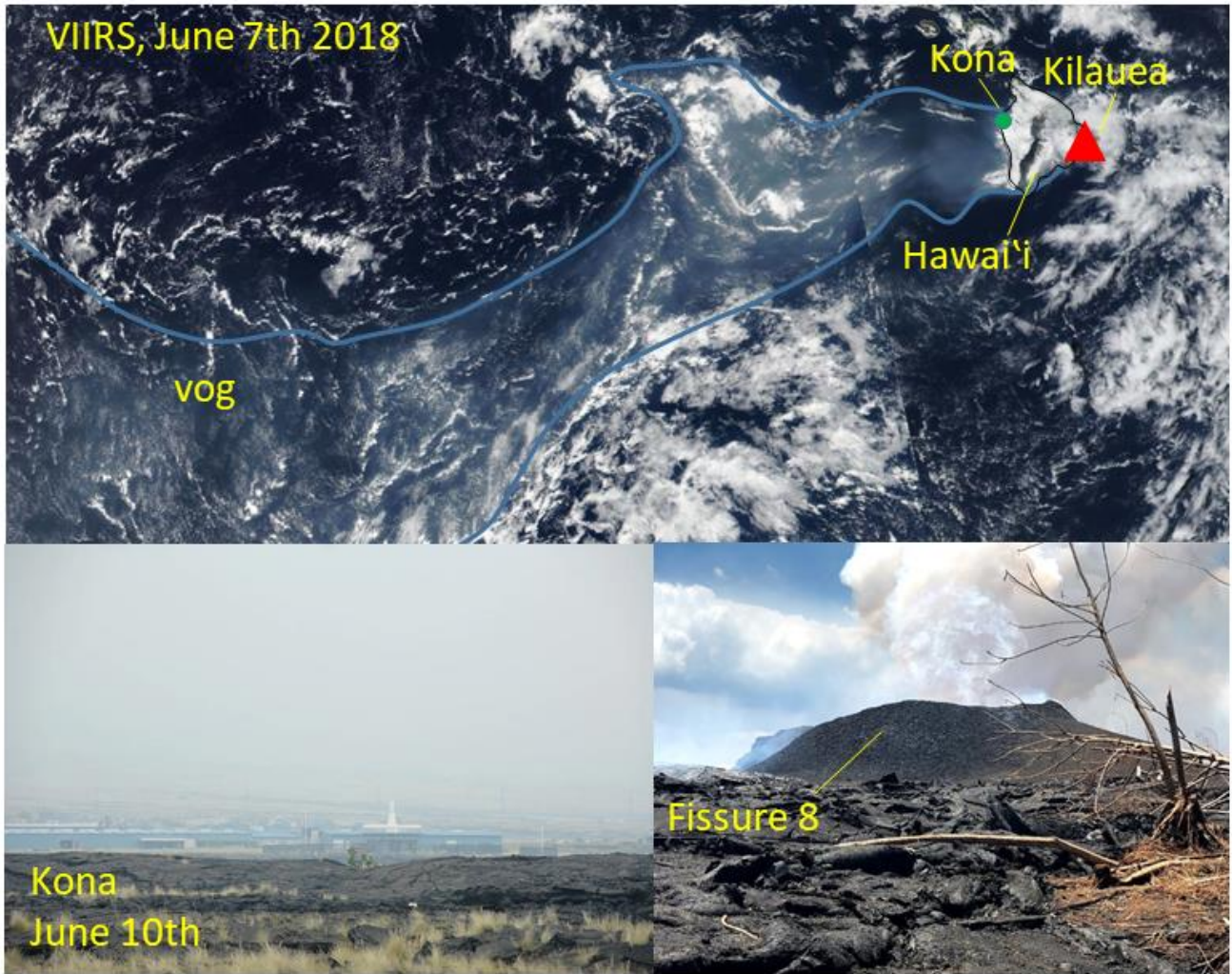
2 *Figure 2. (Bottom) Time series of SO₂ mass (in kilotons) from OMPS/NPP nadir mapper during*
 3 *the 2018 Kilauea eruption and (top) map of lower tropospheric (TRL) SO₂ on June 15th 2018*
 4 *using the Principal Component Analysis (PCA) algorithm.*

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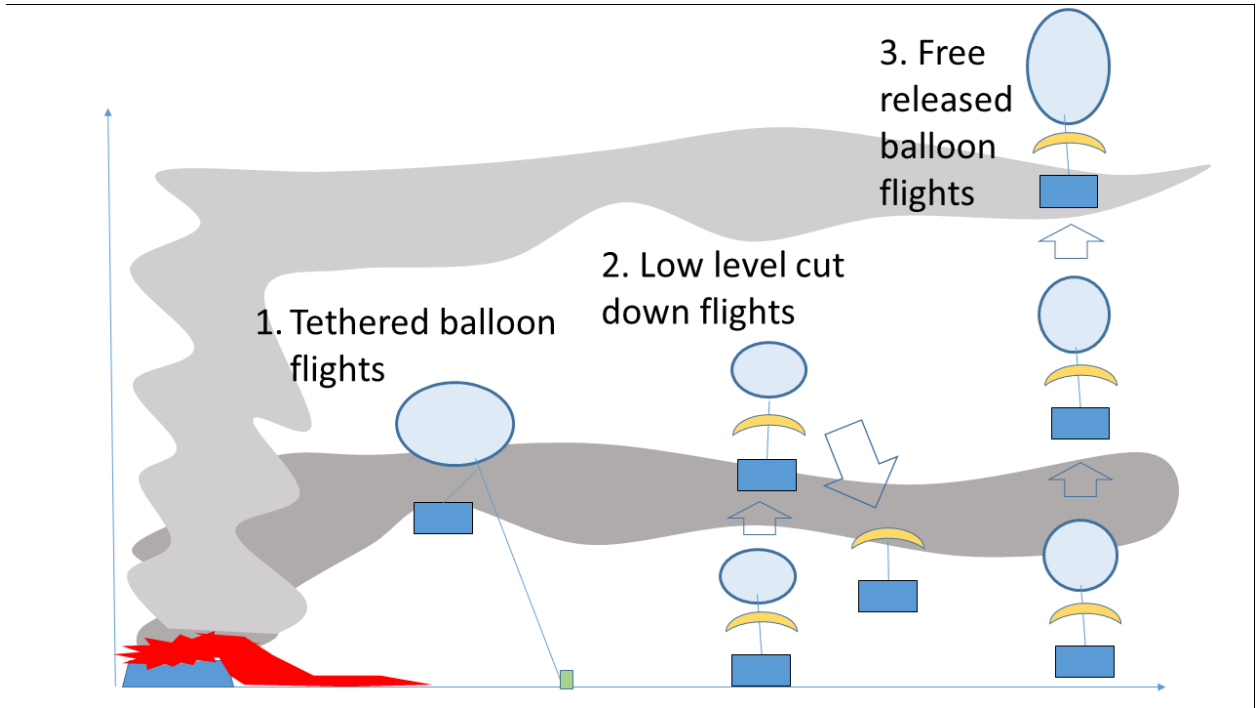
3 *Figure 3. (Top) True color image shows the vog across the Western Pacific observed by*
4 *VIIRS/NPP on June, 7th 2019. (Bottom) photos of the vog in Kona on June, 10th and volcanic*
5 *emissions from fissure 8 on June, 18th (photo credit. J-P. Vernier).*

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3 *Figure 4. Balloon activities during the VolKilauea campaign to sample volcanic plumes emitted at*
4 *low altitudes from the fissures in the LERZ and at higher levels from explosive events at the*
5 *Kilauea summit.*

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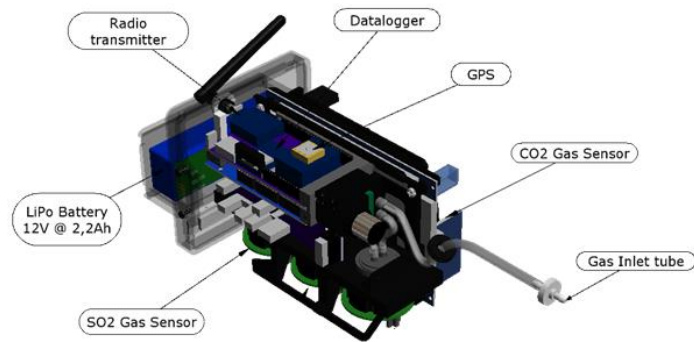
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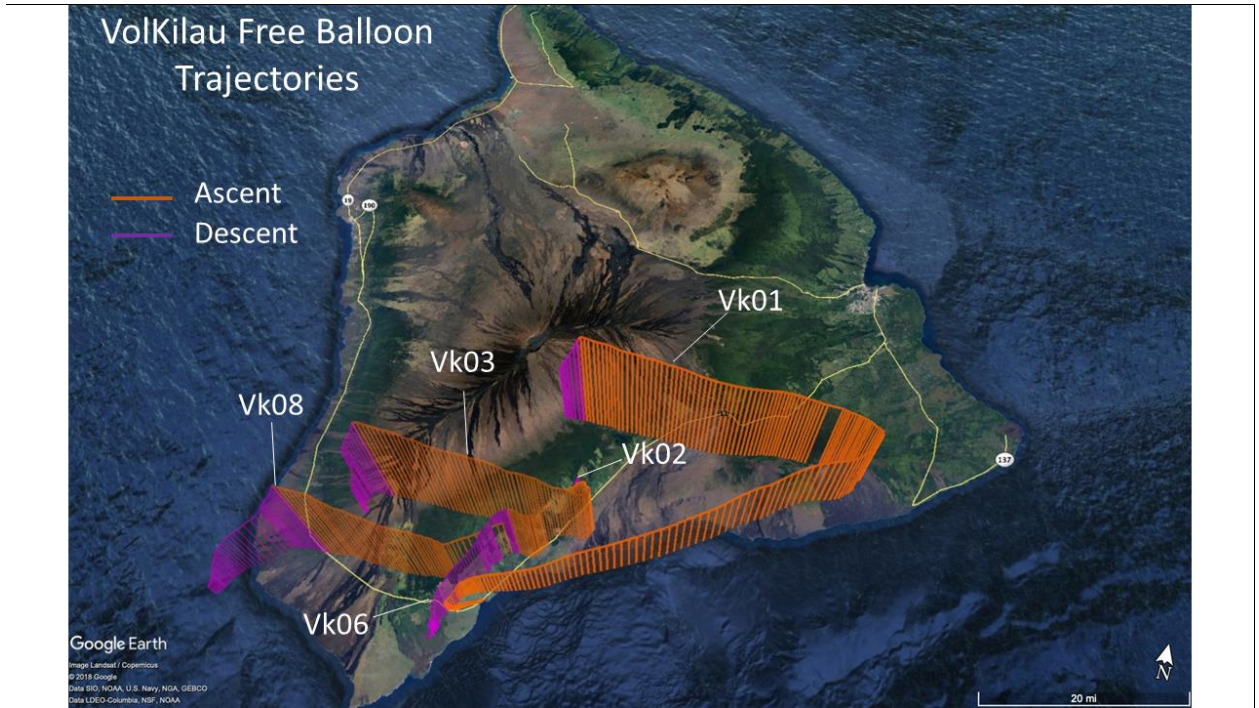
3 *Figure 5. Schematic and picture of the miniGAS PRO, a 1.5kg miniature multiple gas payload*
4 *which incorporates waterproof case, GPS, SO₂ and CO₂, temperature and relative humidity*
5 *sensors with onboard datalogger and telemetry, designed for UAV/ drone or tethered*
6 *balloon/blimp measurements.*

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3 *Figure 6. Trajectories of the 5 free released balloon flights during the VolKilau campaign.*

4 *Balloon ascent is shown in orange and descent in magenta.*

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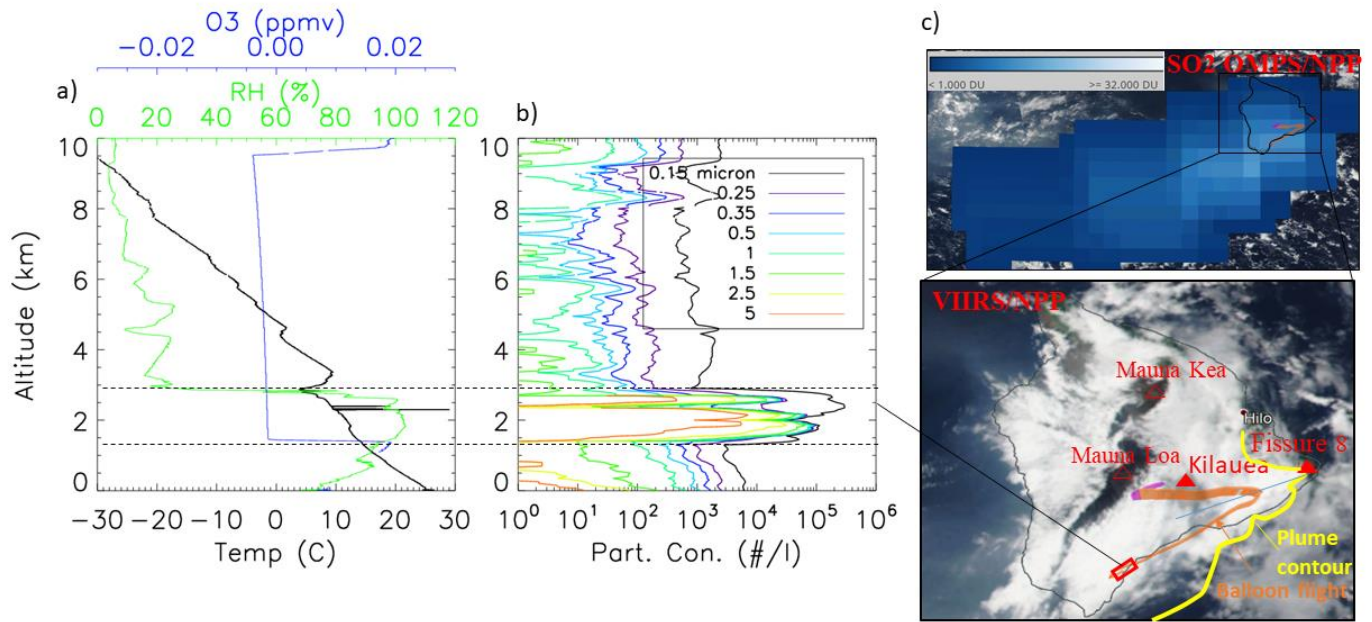
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3 *Figure 7. Data from a free-released balloon flight launched on June, 13th 2018 near 3am (UTC)*

4 *from the Whittington beach (19.086, -155.550). Profiles of (a) temperature, Relative Humidity*

5 *(RH), ozone mixing ratio, and (b) aerosol concentration profiles for $r(\text{radius}) > 0.15 \mu\text{m}$,*

6 *$r > 0.25 \mu\text{m}$, $r > 0.35 \mu\text{m}$, $r > 0.5 \mu\text{m}$, $r > 1 \mu\text{m}$, $r > 1.5 \mu\text{m}$, $r > 2.5 \mu\text{m}$ and $r > 5 \mu\text{m}$. (c) Corresponding*

7 *SO₂ lower tropospheric column map from OMPS/NPP (top) and true color image from*

8 *VIIRS/NPP near 23UTC on June, 12th superimposed with balloon flight trajectory (bottom).*

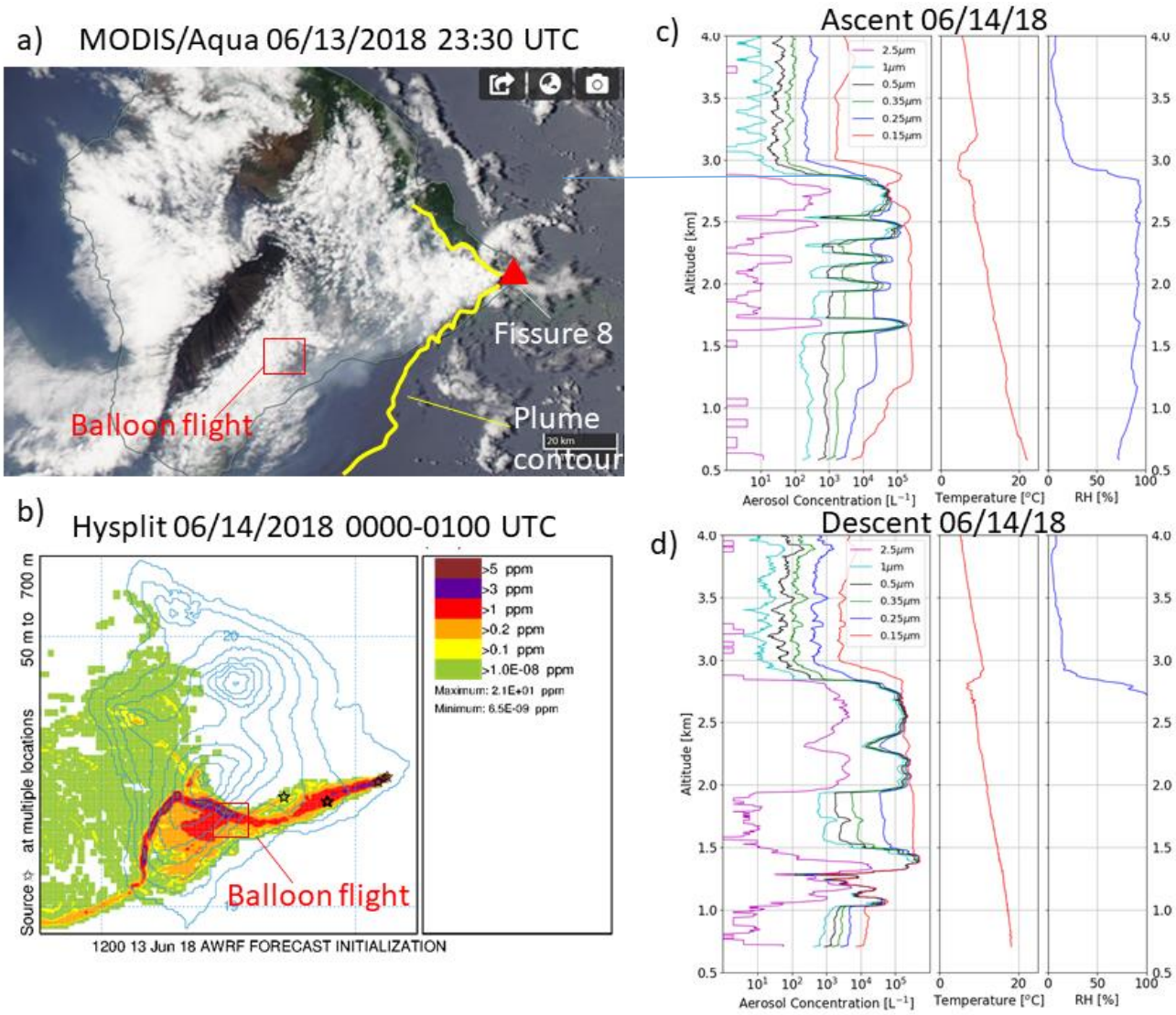
9 *Volcanic plume emitted from fissure 8 in the LERZ is drawn with a yellow contour.*

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2 *Figure 8. (a) True color image from the MODerate resolution Imaging radioSpectrometer*

3 *(MODIS) on the Aqua satellite on June, 13th near 23:30 UTC (b) and SO₂ mixing ratio derived*

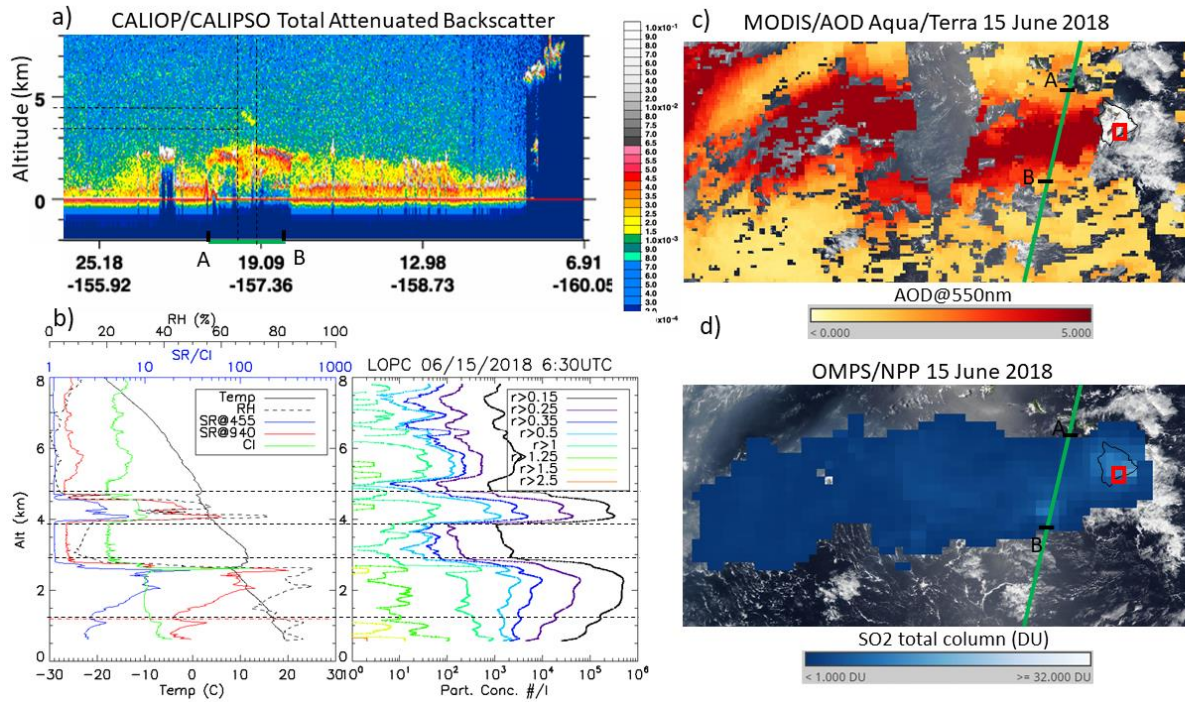
4 *from HYSPLIT between 00:00 and 01:00 UTC on June, 14th. (c&d) Temperature, RH, aerosol*

5 *concentration profiles for sizes ranging from 0.15 μm to 5 μm during the ascent and descent of*

6 *the balloon flight (VK02) launched from the Kapapala Ranch (19.263, -155.445) near 1am UTC*

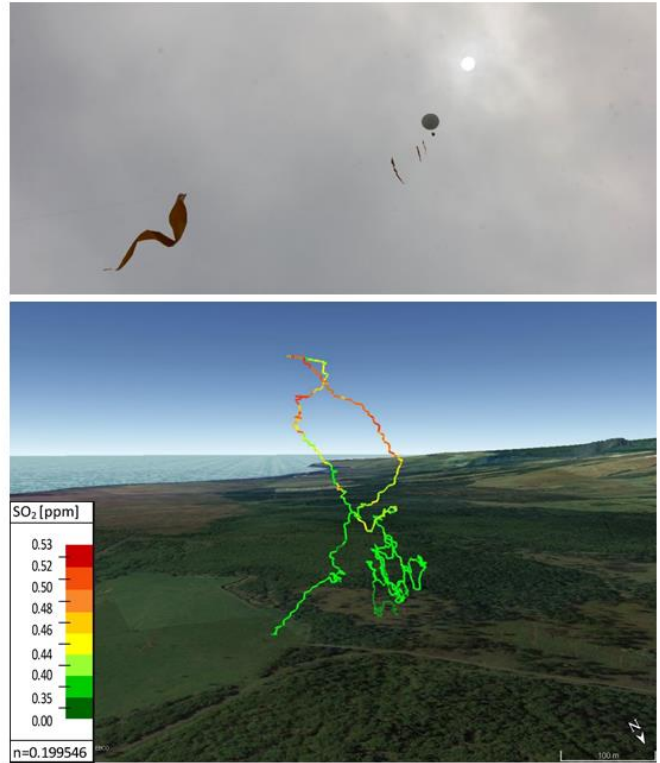
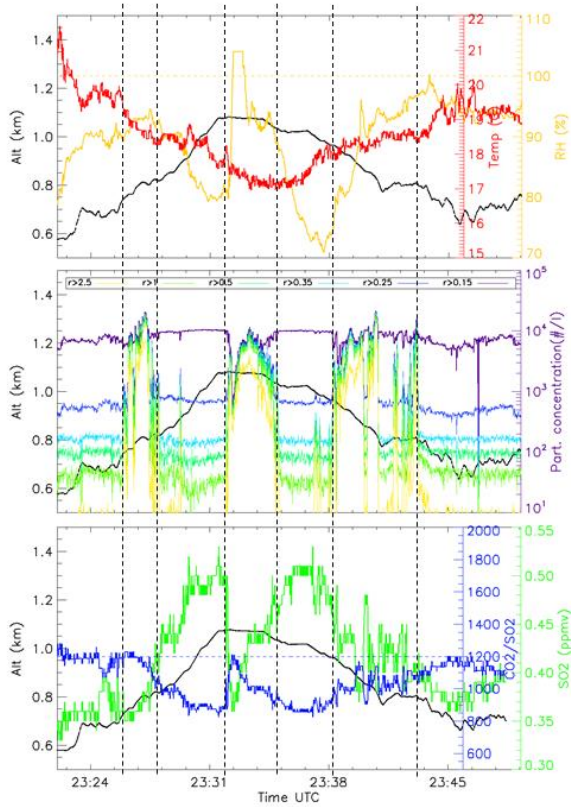
7 *on June, 14th.*

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 2 *Figure 9. (a) Total attenuated backscatter from CALIOP/CALIPSO on June, 15th at 12am UTC.*
 3 *(b) Temperature (Temp), Relative Humidity (RH), aerosol concentration profiles from the*
 4 *Langlely Optical Particle Counter (LOPC) and Scattering Ratios (SR@455nm,SR@940), Color*
 5 *Index (CI=SR@940/SR@455) from the COBALD backscatter sonde on June 15th launched near*
 6 *6am UTC. (c&d) MODIS Aerosol Optical Depth from Aqua/Terra and SO₂ lower tropospheric*
 7 *column from OMPS/NPP with the CALIPSO orbit track position shown in green (a). OMPS and*
 8 *MODIS have been obtained from <https://worldview.earthdata.nasa.gov/>.*

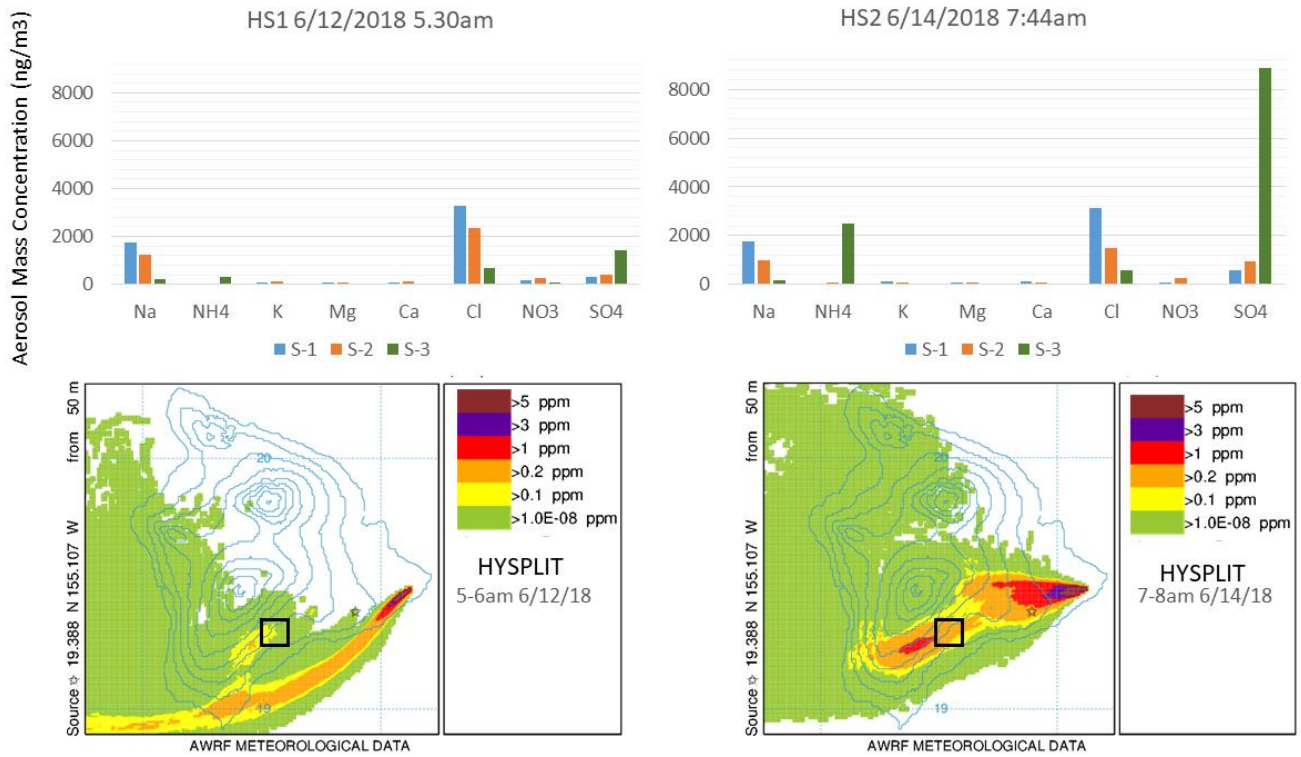
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 2 *Figure 10. (Right) Upper looking view of the tethered system and balloon position colored with*
 3 *SO₂ mixing ratio on Google map. (Left) Time series (from top to bottom) of temperature, RH,*
 4 *GPS altitude, aerosol concentration, SO₂ mixing ratio and CO₂/SO₂ ratio on June, 15th between*
 5 *23:20 and 23:50 UTC. Aerosol and SO₂ concentrations were measured by the ROPC and*
 6 *miniGAS, respectively.*

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3 *Figure 11. (Top) Ionic concentrations from aerosol samples collected on June 12th and 14th*
4 *(Hawai‘i Sample 1&2; HS1 and HS2) at 5.30 am and 7:44am, respectively. (Bottom) SO₂ mixing*
5 *ratio between 0-100m from HYSPLIT near the time of the sampling.*

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Date (UTC)	Location	Time (UTC)	Flight Nb	Flight type	Max Alt	Payloads	Weight	balloon
11-Jun	Kapapala	4am	test	Tethered	100m	ROPC	0.7 kg	300g
13-Jun	Whittingto Beach	3am	VK01	Free Released	28 km	LOPC,ECC,Imet	4.6kg	1200g
14-Jun	Kapapala	1am	VK02	Free Released/Low cut	5km	ROPC,miniGAS,Filter	2 kg	1200g
15-Jun	Kapapala	6am	VK03	Free released	25 km	LOPC,COBALD,ECC	3.5 kg	1200g
15-Jun	Kapapala	11pm	VK04	Tethered	1.2 km	ROPC,miniGAS	2 kg	300g
16-Jun	Kapapala	10pm	VK05	Tethered	1.1 km	LOPC,Impactor, Imet	2.5 kg	1200 g
16-Jun	Kapapala	11pm	VK06	free released	25 km	Imet	0.2g	1200 g
17-Jun	Kapapala	10pm	VK07	Tethered	1.6 km	ROPC,MiniGAS, Impactor	1.2 kg	2000 g
18-Jun	Kapapala	1am	VK08	free released	28 km	LOPC	2.6 kg	2000 g

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2 *Table 1. List of balloon flights during the VolKilau campaign together with date, time, location,*
3 *flight type, payload, weight and balloon size information. The following acronyms correspond to*
4 *ROPC: LASP Light Weight Optical Particle Counter, LOPC: Langley Optical Particle Counter,*
5 *ECC: Ozonesonde, Imet: meteorological radiosonde, Impactor (G): Ground samples with the*
6 *aerosol impactor, COBALD: backscatter sonde.*

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Con. ng/m ³		Na ⁺	NH ₄ ⁺	K ⁺	Mg ²⁺	Ca ²⁺	Cl ⁻	NO ₂ ⁻	NO ₃ ⁻	SO ₄ ²⁻
HS1 (06/12/2018)	S1	1747	0	0	0	0	3300	0	159	327
5.30am UTC	S2	1249	0	145	0	144	2364	0	245	418
	S3	204	307	0	0	0	674	0	0	1429
HS2 (06/14/2018)	S1	1759	0	122	0	122	3151	0	0	599
7:44am UTC	S2	991	0	0	0	0	1474	0	263	928
	S3	182	2466	0	0	0	577	0	0	8862

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3 *Table 2. Ionic concentration for the two ground aerosol samples taken during the VolKilau*
4 *campaign on 06/12/2018 and 06/14/2018 at 5.30 am and 7.44am respectively.*