



Building colloidal crystals under microgravity

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NASA Advanced Colloids Experiments (ACE) Program on ISS:



Results of experiments:

ACE –T11: Colloids of spherical particles, Nov 2020 - Jun 2021
ACE-TR Ellipsoids: Colloids of ellipsoidal particles, Sept 2021 - Feb 2021

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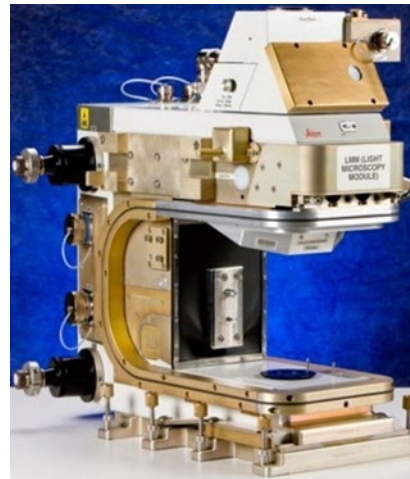


Why study colloids in microgravity?

- On Earth gravitational sedimentation confirms most colloidal systems to 2D with numerous artifacts due to floors and boundaries.
- Long-duration microgravity on the ISS literally adds another dimension to the problem making available 3D motion, 3D processing, 3D fundamental science, 3D protocols and 3D materials and devices

Light Microscopy Module (LMM)

Approach: Use high-resolution confocal (3D) microscopy in long-duration microgravity on the ISS to investigate the colloidal structure at the particle level.

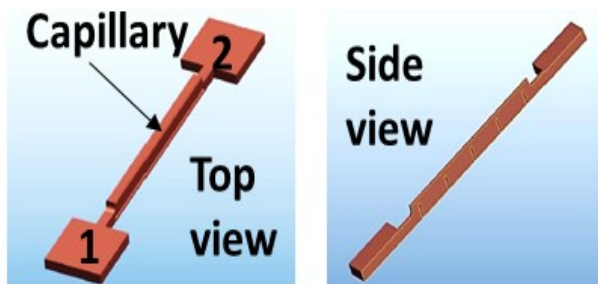


Sample Module

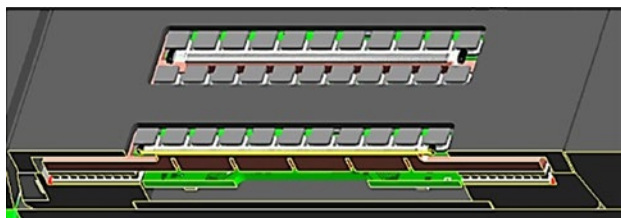
3D Microscope

Fluids Integrated Rack

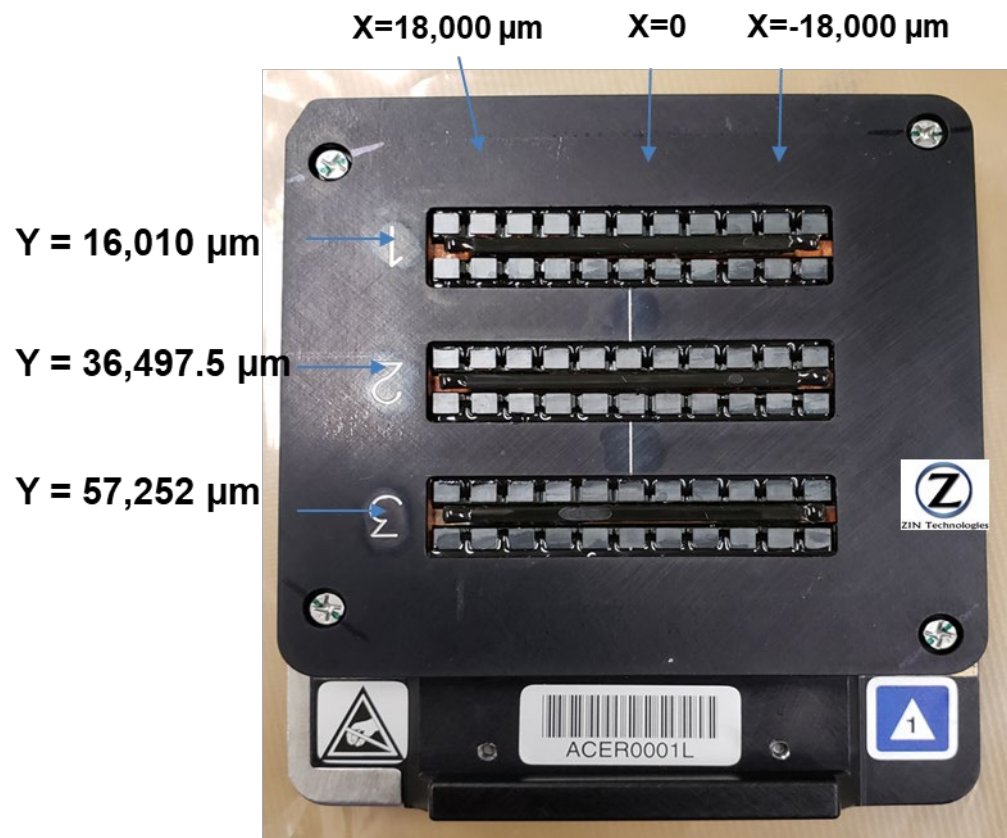
NASA GRC & ZIN Tech images



The capillary of diameter 0.140 mm is mounted between hot and cold thermistors 1 and 2.



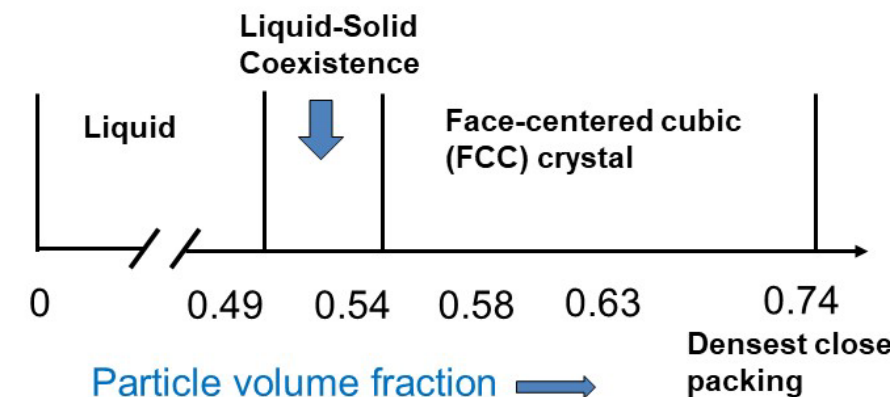
Capillary sample cells surrounded by inductors that are used for walking a turning stirring bar for mixing a sample.



➤ Hard spheres is the basic model of particles in the statistical mechanical theory of fluids and solids. Defined as impenetrable spheres that cannot overlap in space, they mimic the extremely strong repulsion that atoms and spherical molecules experience at very close distances. The hard-sphere colloid provides a generic model that explains the quasi-universal structure and dynamics of simple liquids.

➤ **The hard-sphere model is the cornerstone of our understanding of liquid crystallization!**

Most recent computer simulations predict the following **equilibrium phase diagram** for hard spheres



- In terrestrial experiments heterogeneous polycrystalline materials with crystallites $< 100 \mu m$ form in sediments as the density increases toward the bottom of the samples.
- Due to gravitational settling of particles, terrestrial experiments fail to confirm the crystalline phases predicted.

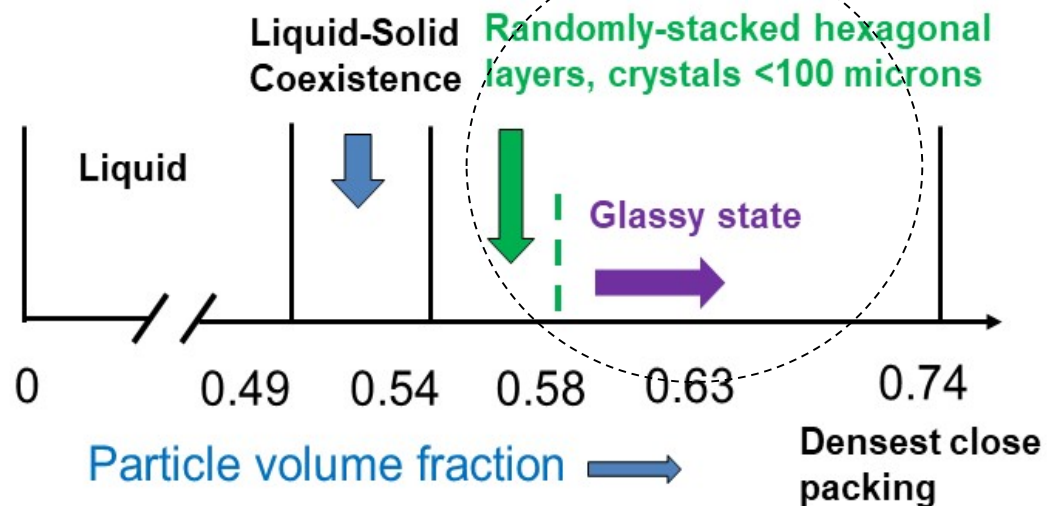
Experimental phase diagram of hard spheres

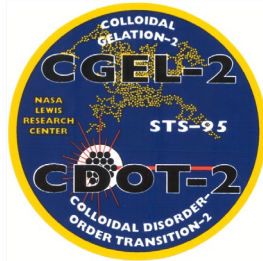


0.637 0.621 0.595 0.578 0.553 0.528 0.512 0.502 0.478

Glass Crystal Fully Coexistence Liquid
Heterogeneous crystallized

Crystallization for densities beyond 0.57 – 0.58 did not occur! Interpreted as the **glassy state**

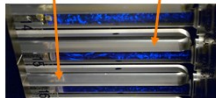
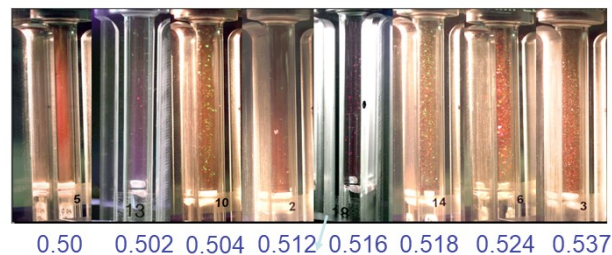




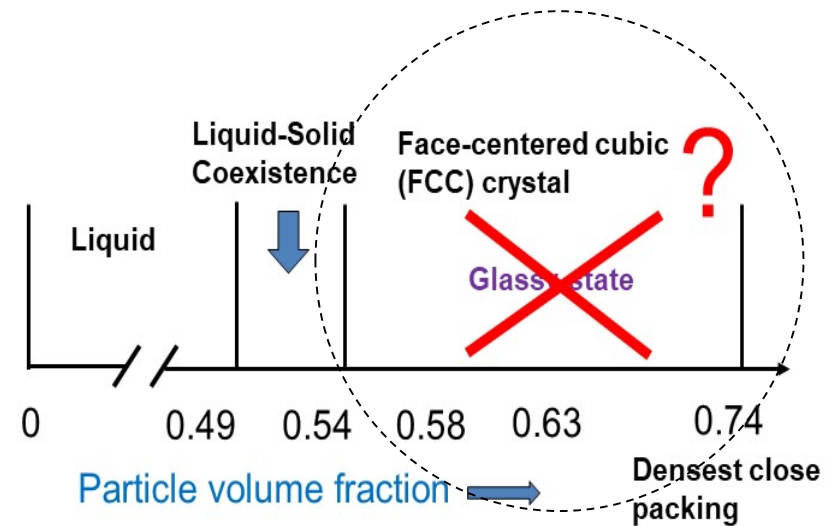
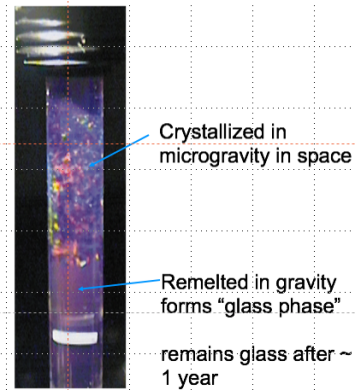
Colloidal Disorder-Order Transitions (CDOT) experiments:
 First: **STS-73, Nov 1995**. Second: **STS-95, Oct 1998**.

Theory revised maybe works???

CDOT 1998: Chaikin et al, Nature, 387, 883 (1997)



Volume fraction 60%

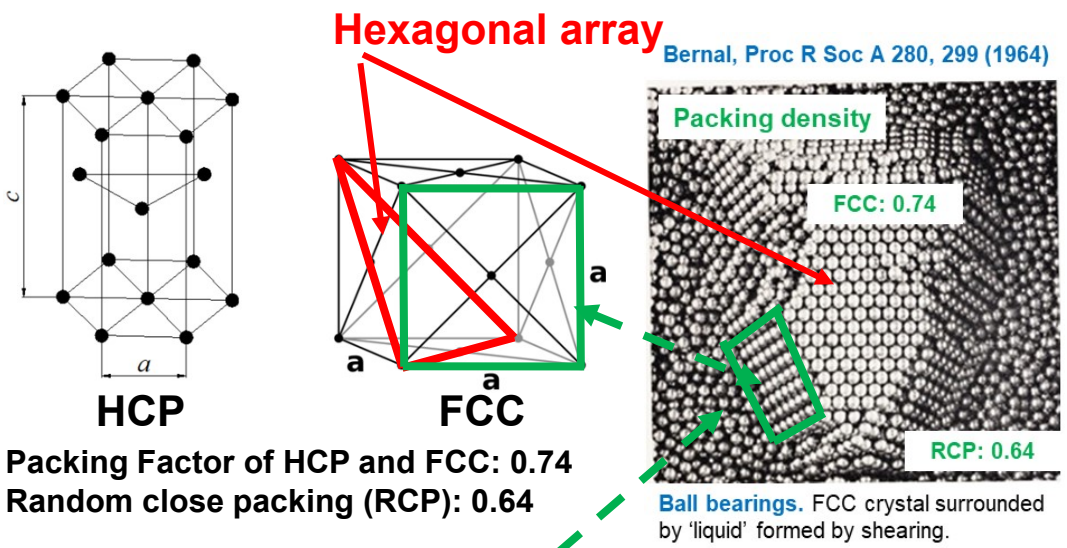


Equilibrium Crystal Structure

Most metals adopt hexagonal close packed (HCP) and face-centered cubic (FCC) structures

[https://en.wikipedia.org/wiki/Periodic_table_\(crystal_structure\)](https://en.wikipedia.org/wiki/Periodic_table_(crystal_structure))

1 H HEX	2 He HCP	3 Li BCC	4 Be HCP	5 B RHO	6 C HEX	7 N HEX	8 O SC	9 F SC	10 Ne FCC
11 Na BCC	12 Mg HCP	13 Al FCC	14 Si DC	15 P ORTH	16 S ORTH	17 Cl ORTH	18 Ar FCC	19 K BCC	20 Ca FCC
21 Sc HCP	22 Ti HCP	23 V BCC	24 Cr BCC	25 Mn BCC	26 Fe BCC	27 Co HCP	28 Ni FCC	29 Cu FCC	30 Zn HCP
31 Ga ORTH	32 Ge DC	33 As RHO	34 Se HEX	35 Br ORTH	36 Kr FCC	37 Rb BCC	38 Sr FCC	39 Y HCP	40 Zr HCP
41 Nb BCC	42 Mo BCC	43 Tc HCP	44 Ru FCC	45 Rh FCC	46 Pd FCC	47 Ag FCC	48 Cd HCP	49 In TETR	50 Sn TETR
51 Sb RHO	52 Te HEX	53 I ORTH	54 Xe FCC	55 Ba FCC	56 La FCC	57 Ce FCC	58 Pr FCC	59 Nd FCC	60 Pm FCC
61 Sm FCC	62 Eu FCC	63 Gd FCC	64 Tb FCC	65 Dy FCC	66 Ho FCC	67 Er FCC	68 Tm FCC	69 Yb FCC	70 Lu FCC
71 Hf BCC	72 Ta BCC	73 W BCC	74 Re HCP	75 Os HCP	76 Ir FCC	77 Pt FCC	78 Au FCC	79 Hg FCC	80 Tl FCC
81 Pb FCC	82 Bi FCC	83 Po FCC	84 At FCC	85 Rn FCC	86 Fr FCC	87 Ra FCC	88 Ac FCC	89 Th FCC	90 Pa FCC
91 U FCC	92 Np FCC	93 Pu FCC	94 Am FCC	95 Cm FCC	96 Bk FCC	97 Cf FCC	98 Es FCC	99 Fm FCC	100 Md FCC
101 No FCC	102 Lr FCC	103 La FCC	104 Ce FCC	105 Pr FCC	106 Nd FCC	107 Pm FCC	108 Sm FCC	109 Eu FCC	110 Gd FCC
111 Tb FCC	112 Dy FCC	113 Ho FCC	114 Er FCC	115 Tm FCC	116 Yb FCC	117 Lu FCC	118 Og FCC	119 Ts FCC	120 Og FCC



Packing Factor of HCP and FCC: 0.74
Random close packing (RCP): 0.64

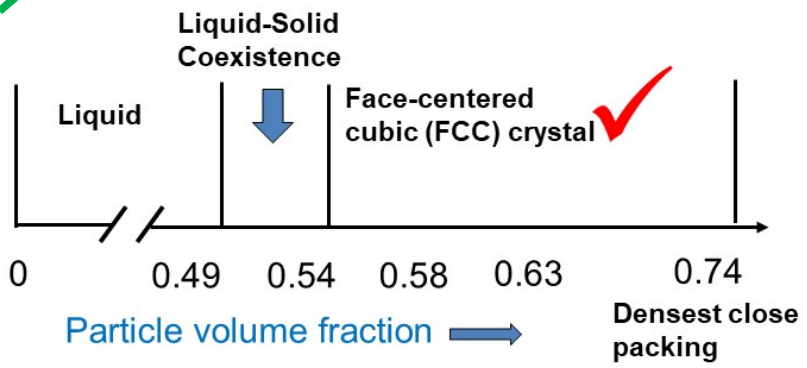
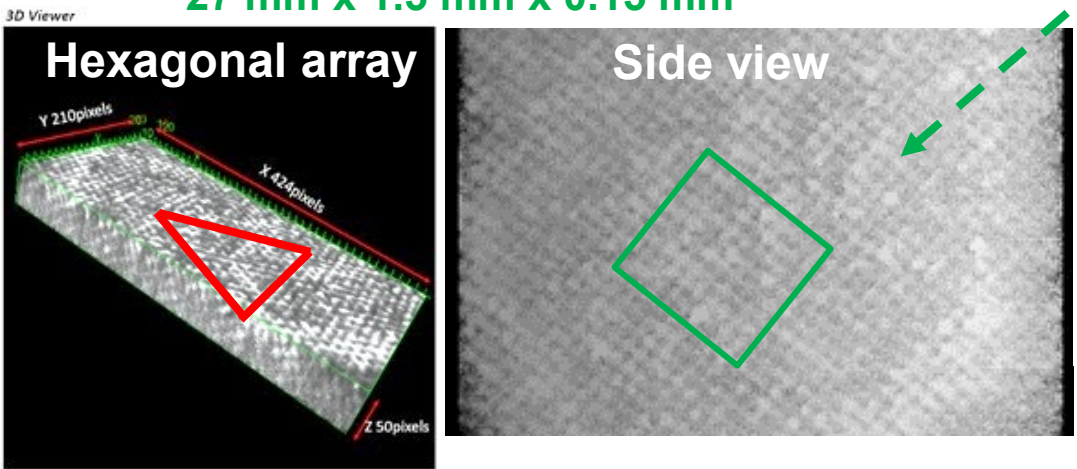
Theory does work!!!

First confirmation of theory

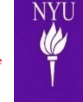
~100 Gb confocal images and videos

FCC crystal is of the size of the capillary

27 mm x 1.5 mm x 0.15 mm

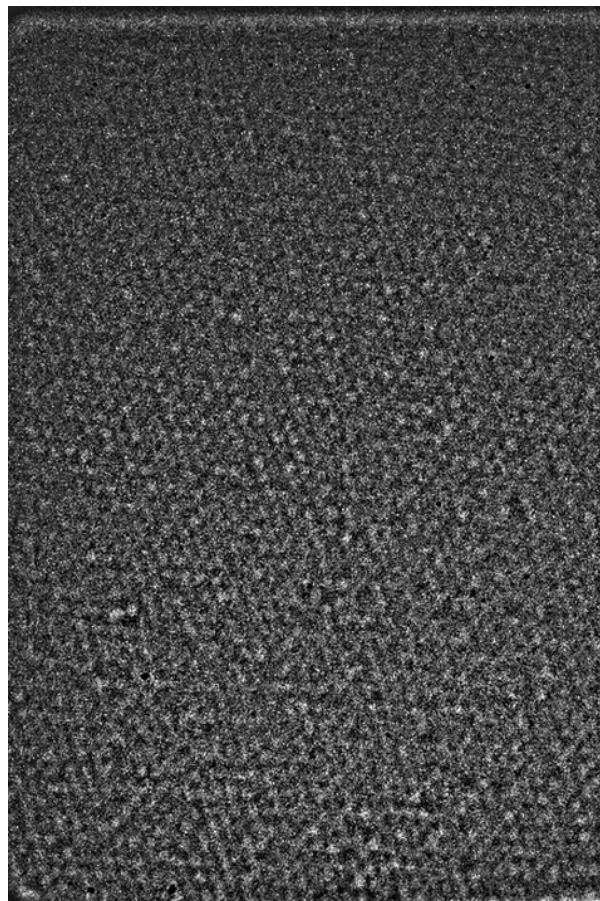


Monodisperse fluorescent polymethyl methacrylate (PMMA) particles $1.29 \pm 0.02 \mu\text{m}$

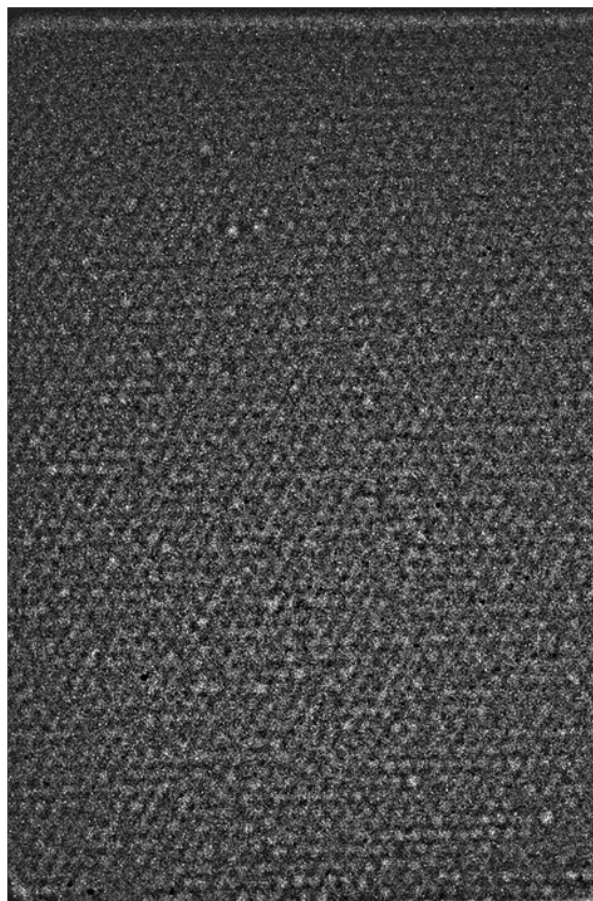


Crystallization of spherical particles on ISS

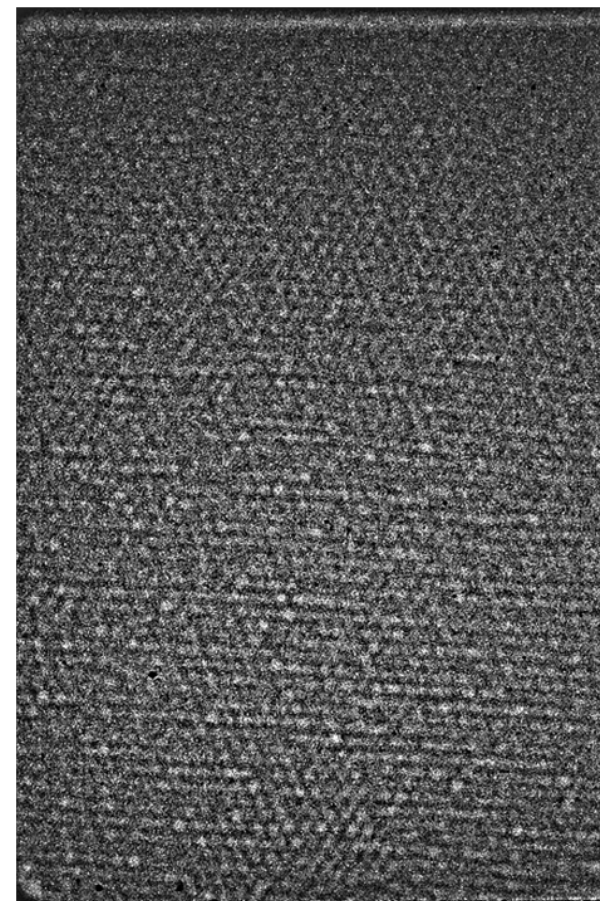
Confocal microscopy: Magnification 100 X, oil immersion objective



X= -10.885 mm



X= -2.836 mm



X= +16.921 mm

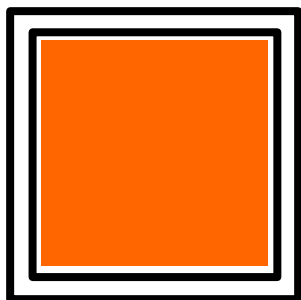
← 27 mm across the capillary →

Parallel crystalline lines in every portion of the images across the entire sample capillary.

The colloid of hard ellipsoids provides a generic model that explains the quasi-universal structure and dynamics of liquid crystals. Compared to molecules in liquid crystals which are too small to be seen with an optical microscope, the colloidal particles are huge in size and can be used to test predictions of computer simulations. .

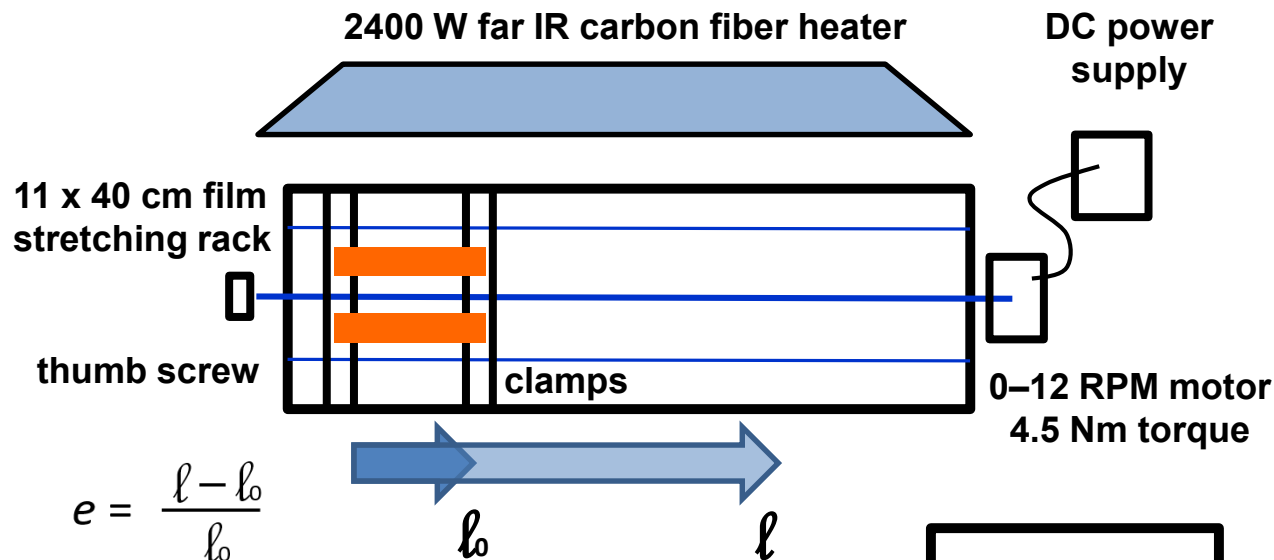
Synthesis of ellipsoids: Andrew D. Hollingsworth, NYU

- 1 to 2 micron dia. stabilized with poly(12-hydroxylstearic acid) PMMA spheres
- fluorescent labeling
- disperse PMMA spheres in PDMS prepolymer
- cast/cure PDMS film
- thermomechanical stretching / T soak / quench
- harvest ellipsoids
- photocrosslinking
- post-processing

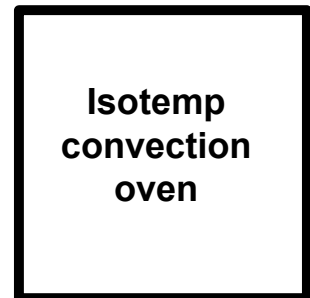


22 x 22 cm plexi mold (1.27 cm sheet)

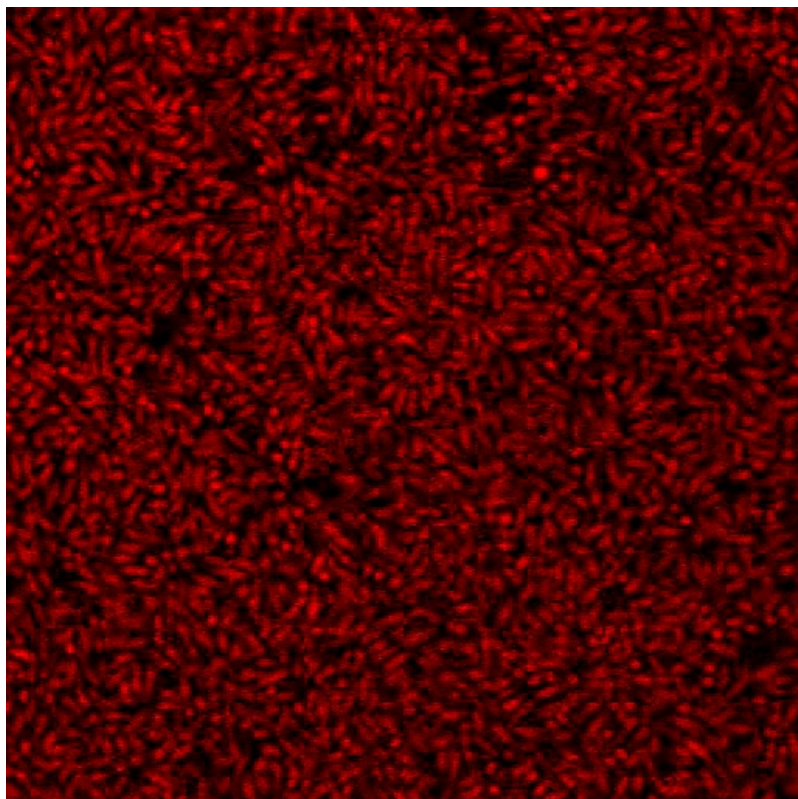
Thermomechanical stretching PDMS film



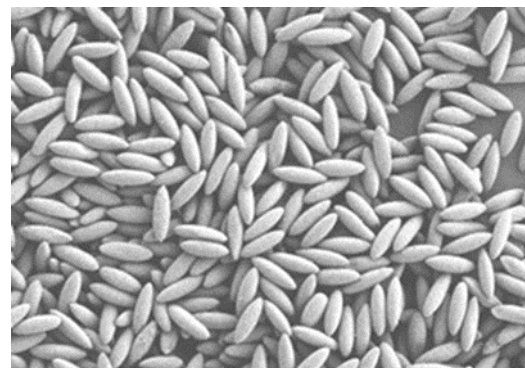
Film thickness: 0.15 cm
 Particle density: $10^{10} / \text{cm}^3$
 Pull velocity: 0.5 cm/min
 Film cure T: 60 h at RT; 3 h at 120°C
 Annealing T: 30 min at 170°C



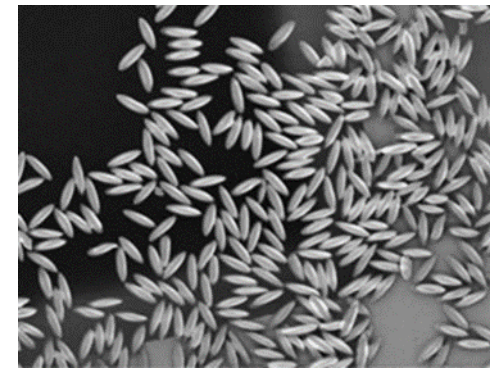
Isotemp
convection
oven



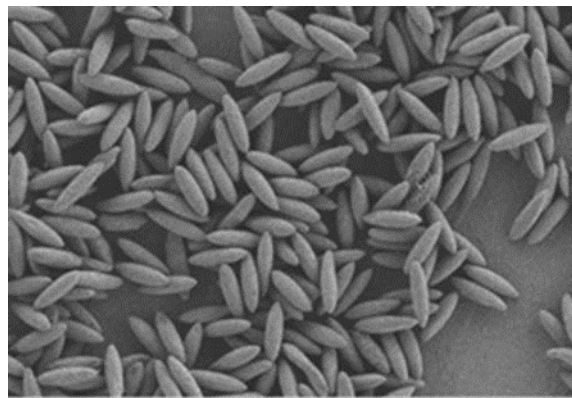
Leica SP5 confocal
63 X glycerol, 543 nm HeNe



CSF-02-139A pulls 5 6b 7 8 9
 $\alpha \approx 3.0$



CSF-02-139B pull 1
 $\alpha \approx 3.7$



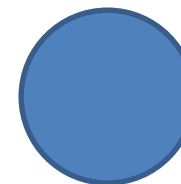
CSF-02-139C pull 10
 $\alpha \approx 3.8$



$$\alpha = 3.8$$

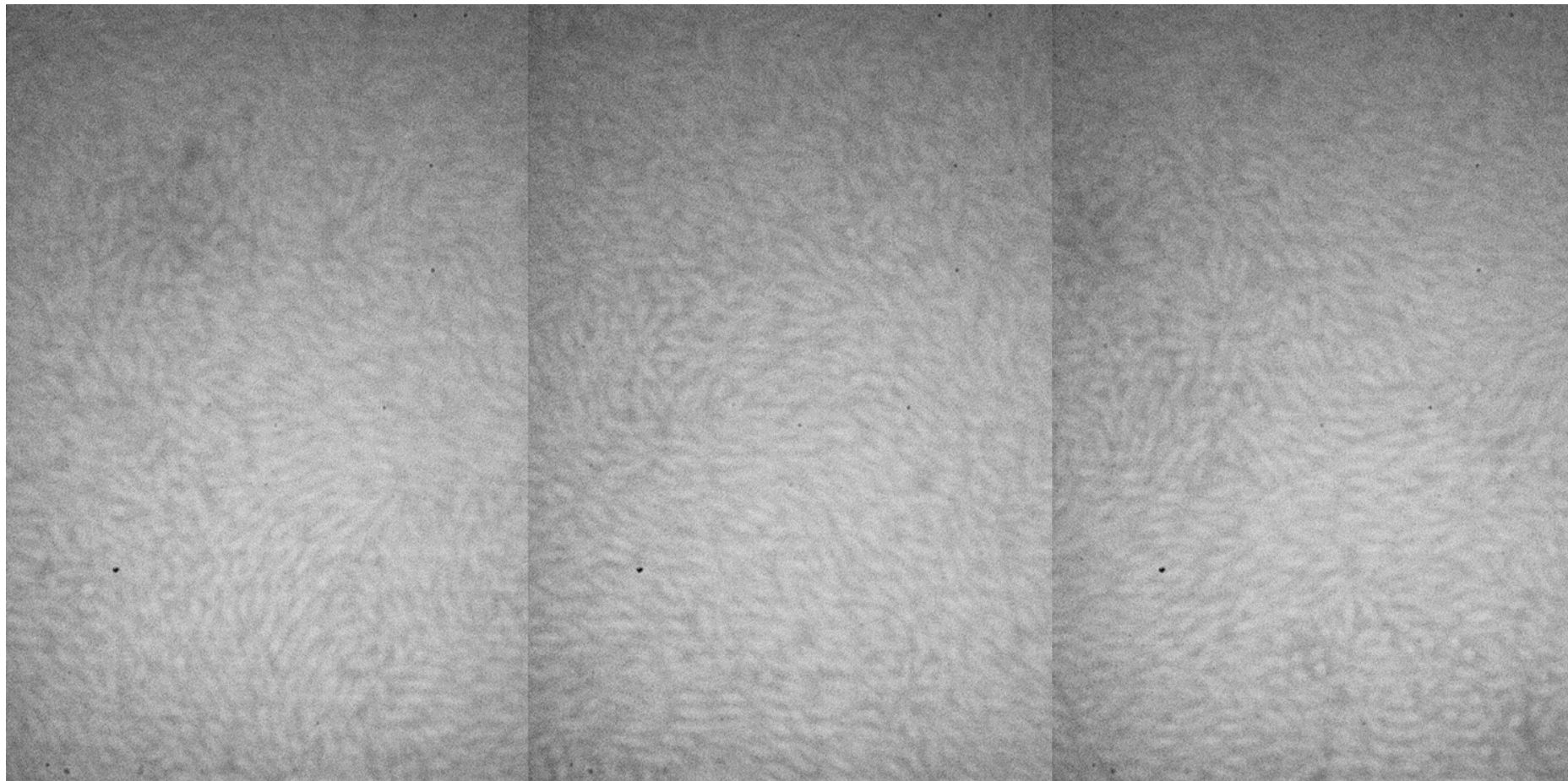


$$\alpha = 3.0$$



$$\alpha = 1.0$$

Large nematic domains in scanning along the capillary axis on ISS



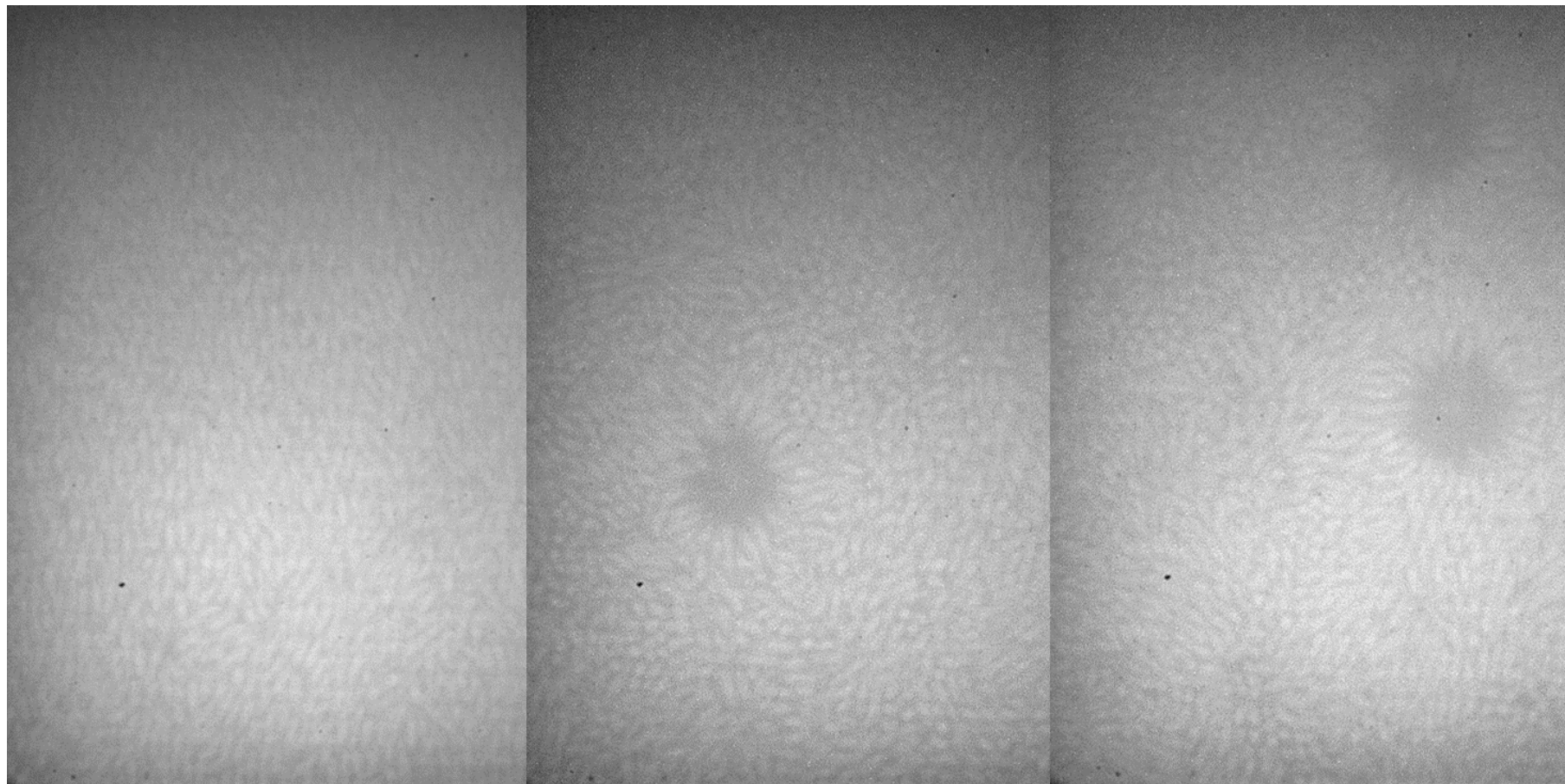
2020/10/05, 17:21:25.535, CnFcl, ACETR,
S3001, C3, X100, X-16674, **Z0+15**, P4,
E458961, 00945

2020/10/05, 17:21:29.535, CnFcl, ACETR,
S3001, C3, X100, X-16566, **Z0+15**, P5,
E458961, 00946

2020/10/05, 17:21:33.535, CnFcl, ACETR,
S3001, C3_X100, X-16457, **Z0+15**, P6,
E458961, 00947

Finding that microscopic colloidal ellipsoids mimic the molecular-sized structures, along with their defects, commonly called ‘hedgehogs’ in the liquid crystal community provides an important support for theoretical predictions.

Nematic domains and associated “hedgehog” defects on ISS



2020/09/24, 22:14:56.768, CnFcl, ACETR,
S3001, C3, X100, X-13964, P29, E443412,
00250

2020/09/24, 22:16:56.833, CnFcl, ACETR,
S3001, C3, X100, X-10712, P59, E443412,
00280

2020/09/24, 22:15:40.701, CnFcl, ACETR,
S3001, C3, X100, X-12771, P40, E443412,
00261

Up until now theories for the simplest, most common, most fundamental Crystallization Transition from Liquid to FCC Crystal had not been confirmed.

Now it is!

- Large FCC crystals observed on spherical colloids on ISS
- Large nematic domains observed in ellipsoidal colloids on ISS

Simulations of equilibrium* work only when gravitational effects are unimportant

Build for the Future†

Use simulation/computation for equilibrium* materials processing in μg

Develop non-equilibrium** processing and dynamic materials/devices in μg

Proceed to understand effects of 1g in materials processing and design

In μg manufacture better materials for ceramics, paints, photonics, pastes for additive manufacturing for industry on Earth and for space exploration

* **Equilibrium** – no net flows of matter or energy

** **Non-equilibrium** – forces, stresses, activity, energy conversion

† **See Grand Challenges in Soft Matter Science: Prospects for Microgravity Research - NASA/CP-20205010493**

Thank you!

Questions

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