

# Thermochemical Degradation of $\text{HfSiO}_4$ by Molten CMAS

Jamesa L. Stokes<sup>1</sup>, Narottam P. Bansal<sup>1</sup>, Valerie L. Wiesner<sup>2</sup>

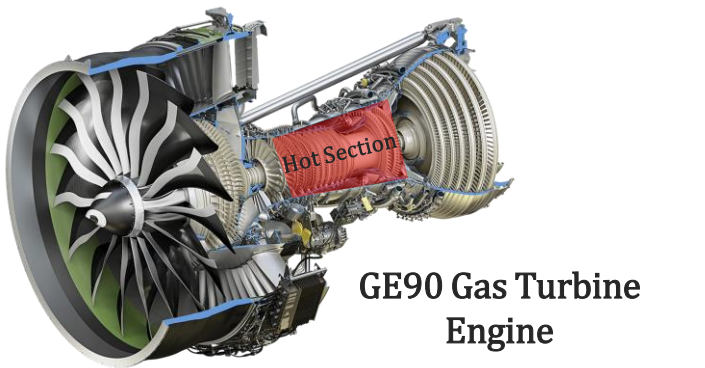
<sup>1</sup>*NASA Glenn Research Center*

<sup>2</sup>*NASA Langley Research Center*

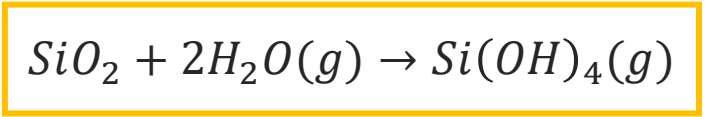
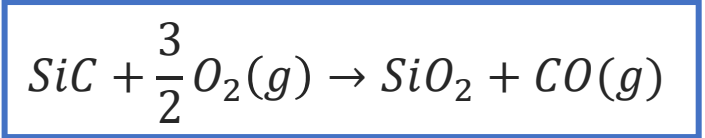
## Acknowledgments

NASA Transformational Tools and Technologies Program

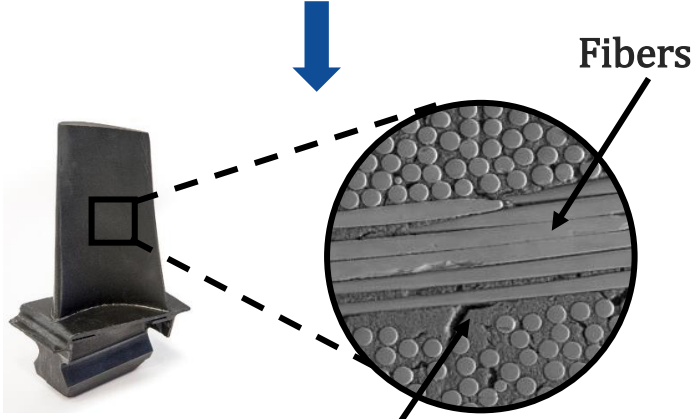
- CMC turbine engine components offer high temperature stability, but recess in high temperature water vapor environments



GE90 Gas Turbine Engine



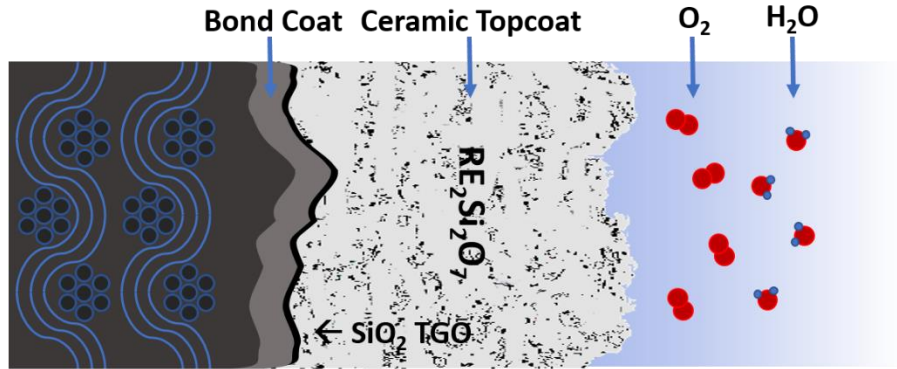
## Environmental Barrier Coatings (EBCs)



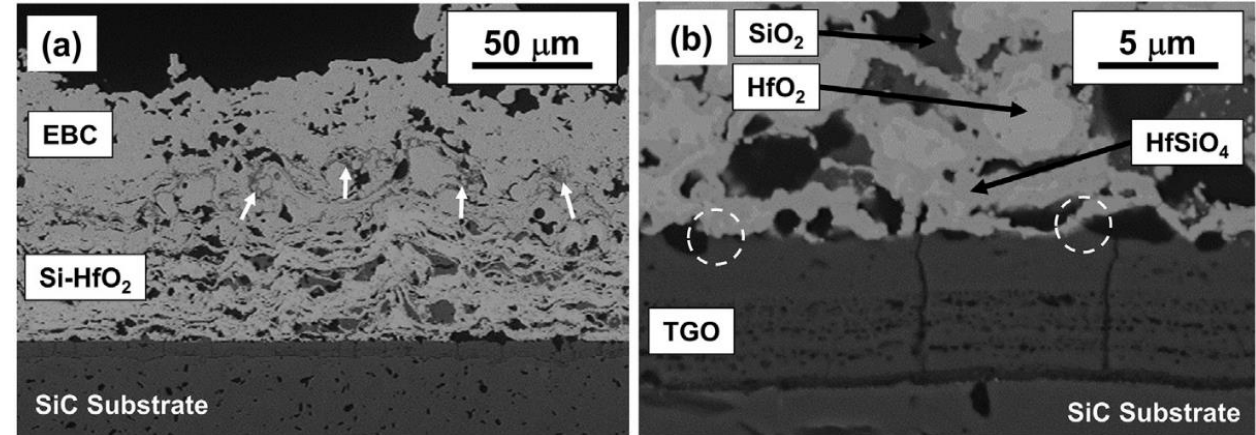
CMC Turbine Blade

Matrix

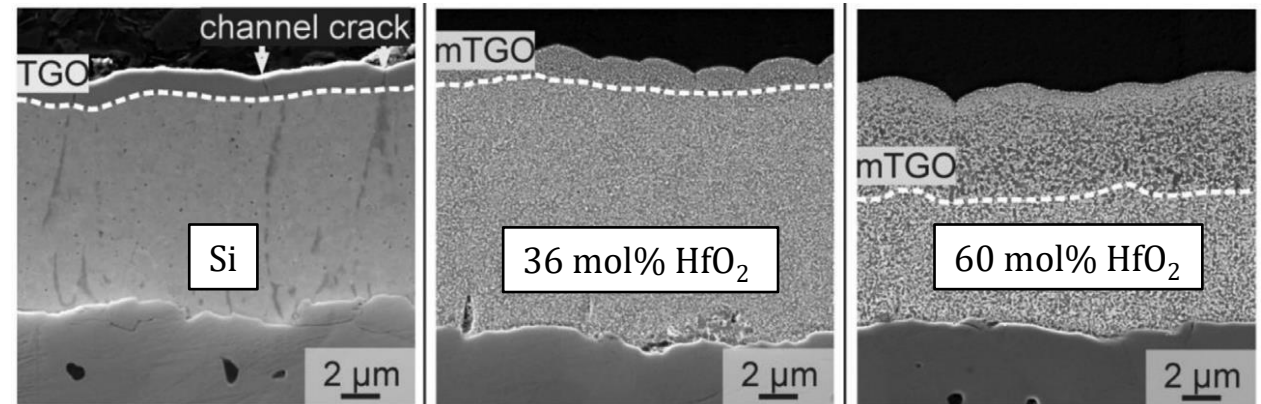
Fibers



- Si metal
  - Melting temperature  $\sim 1414^{\circ}\text{C}$
- $\text{HfO}_2$ -Si Composite Bondcoats
  - Higher temperature capability
  - Similar oxidation rate to Si at low  $\text{HfO}_2$  content
  - Higher oxidation rate with high  $\text{HfO}_2$  content



B.J. Harder, *Surface and Coatings Technology* **384** (2020)



R. Anton et al., *Acta Materialia* **183** (2020)

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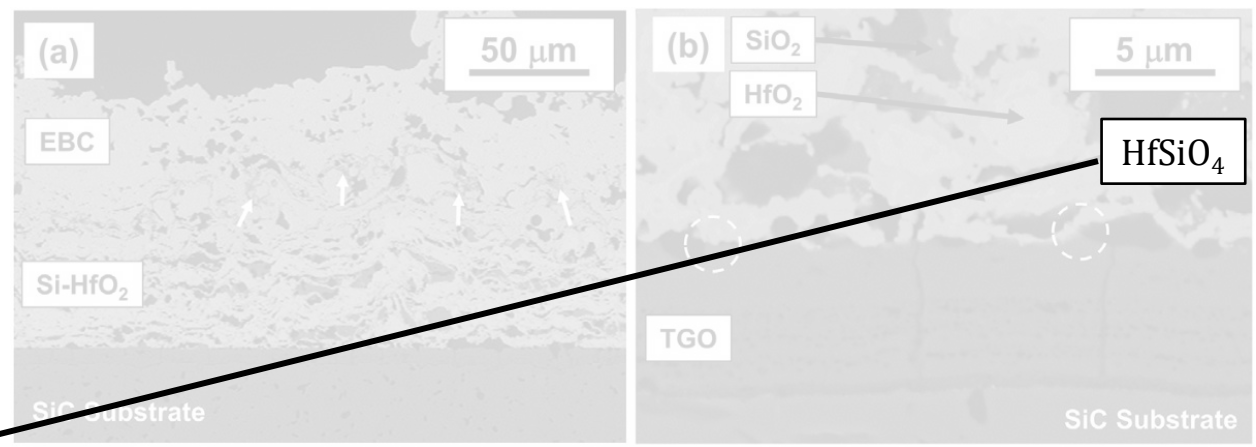
- Hafnon,  $\text{HfSiO}_4$ 
  - Reaction product of  $\text{SiO}_2$  TGO and  $\text{HfO}_2$
  - Similar CTE to Si-based CMCs

Ridley et al., *Journal of the American Ceramic Society* **104** (2021)

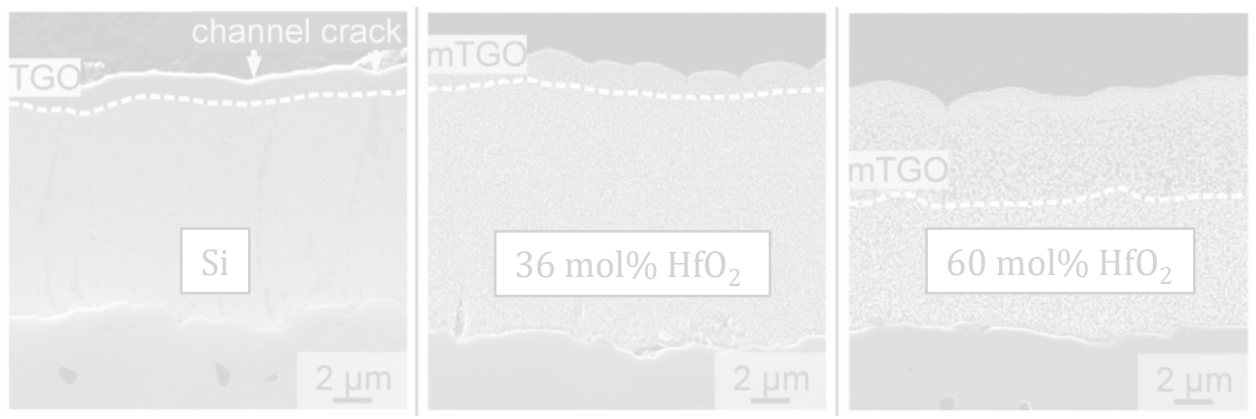
d) 1000 cycles

1	At%
Hf	71.2
Yb	28.8

Deijkers et al., *Acta Materialia* **217** (2021)



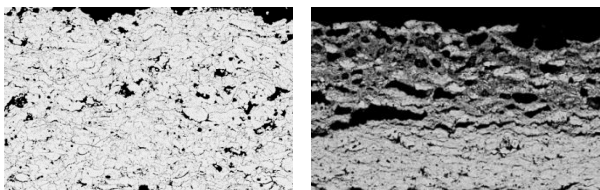
B.J. Harder, *Surface and Coatings Technology* **384** (2020)



R. Anton et al., *Acta Materialia* **183** (2020)

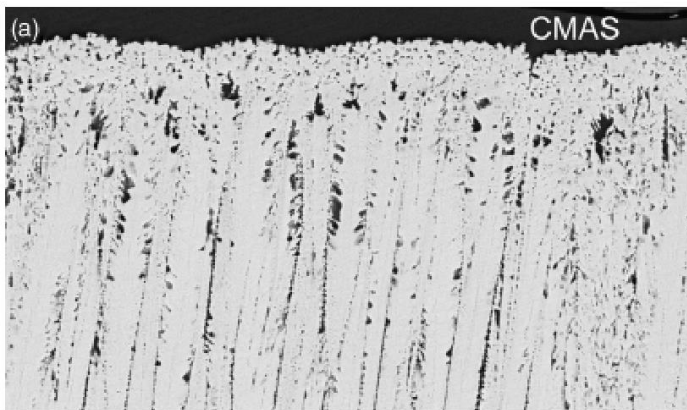
- HfO<sub>2</sub>

### APS YSZ



Aygun et al., *Acta Materialia* 55 (2007)

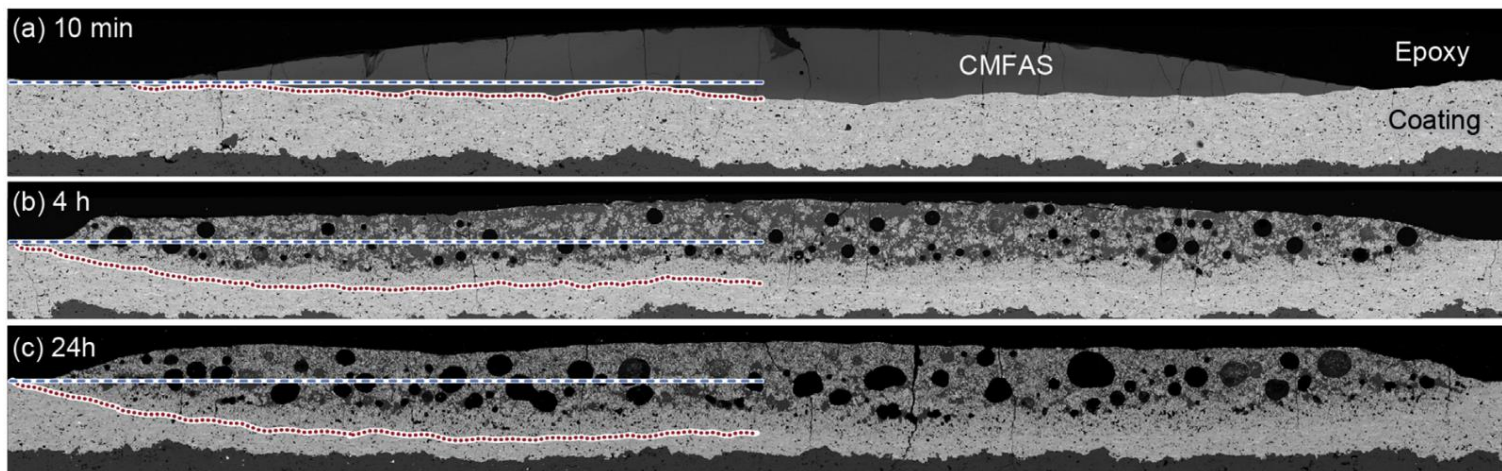
### EB-PVD YSZ



→ m-ZrO<sub>2</sub> + YO<sub>1.5</sub> leaching into glass

Krämer et al., *Journal of the American Ceramic Society* 89 (2006)

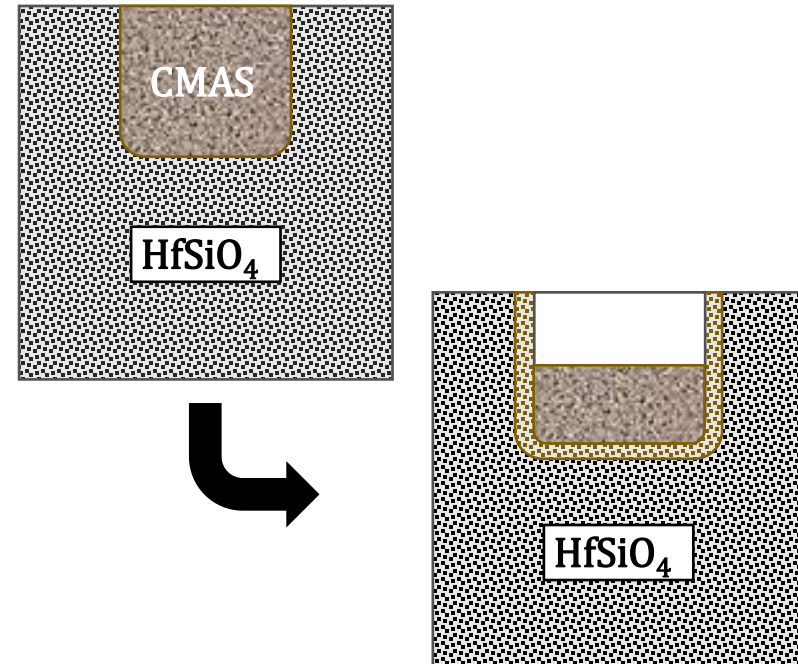
- SiO<sub>2</sub>



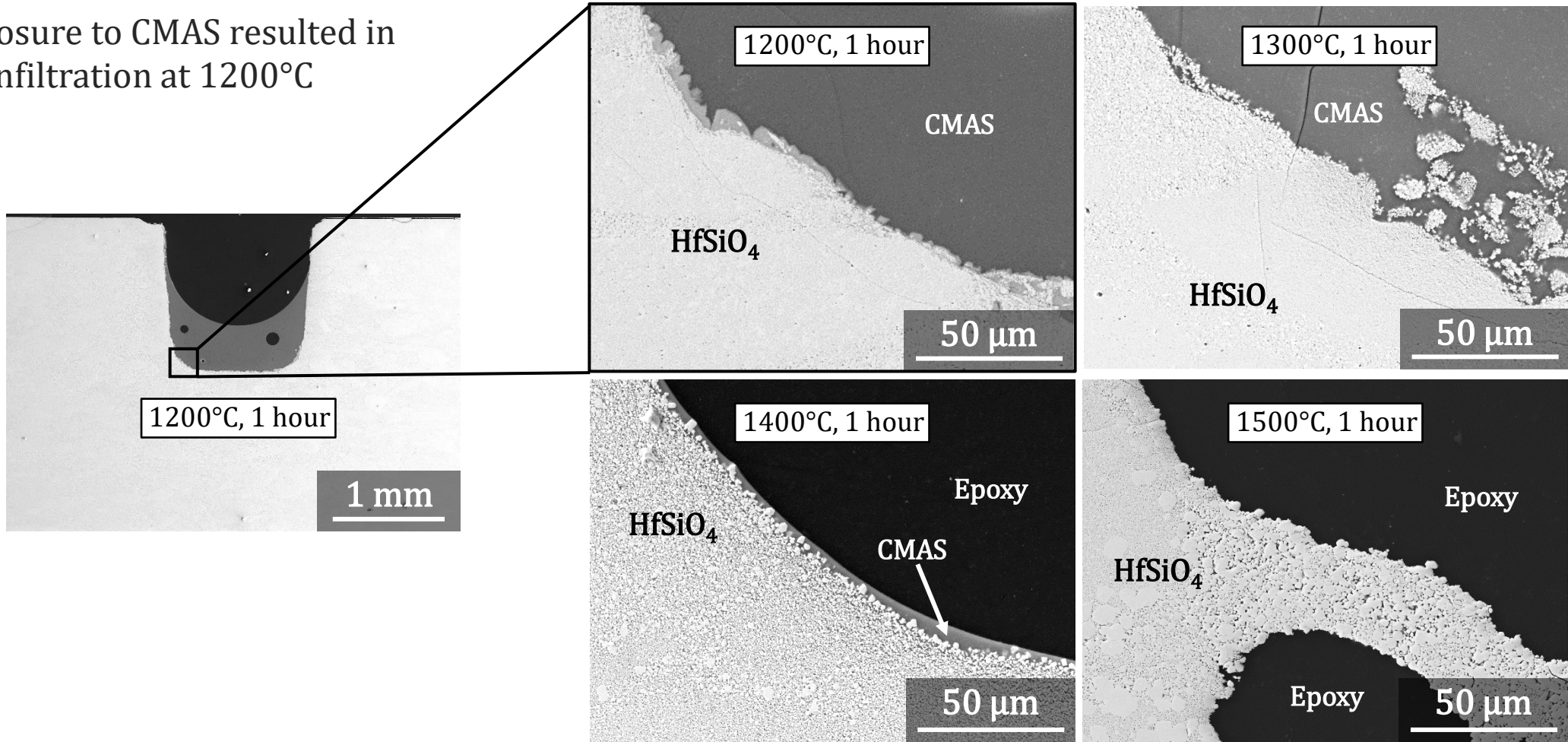
Poerschke et al., *Acta Materialia* 145 (2018)

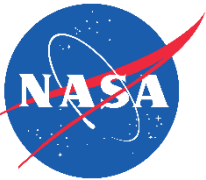
- HfO<sub>2</sub> – anisotropic CTE, higher CTCE than EBC materials
- Excess SiO<sub>2</sub> results in greater consumption of the coating materials needed to achieve melt saturation

- Hot pressed  $\text{HfSiO}_4$
- AFRL-02 CMAS,  $\sim 35 \text{ mg/cm}^2$  in drilled wells
  - 34 wt.% quartz ( $\text{SiO}_2$ ), 30 wt.% gypsum ( $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ ), 17 wt.% aplite ( $\text{SiO}_2 + \text{KAlSi}_3\text{O}_8$ ), 14 wt.% dolomite ( $\text{CaMg}(\text{CO}_3)_2$ ), 5 wt.% salt ( $\text{NaCl}$ )
  - $21.85\text{CaO}-6.27\text{MgO}-6.08\text{AlO}_{1.5}-61.25\text{SiO}_2-4.02\text{Na}_2\text{O}-0.49\text{K}_2\text{O}-0.04\text{FeO}_{1.5}$  (mol.%)
- Samples heat treated at  $1200^\circ\text{C}$ ,  $1300^\circ\text{C}$ ,  $1400^\circ\text{C}$ , and  $1500^\circ\text{C}$  for 1, 10 and 50 hours
- Reaction products and infiltration depth measured using SEM/EDS

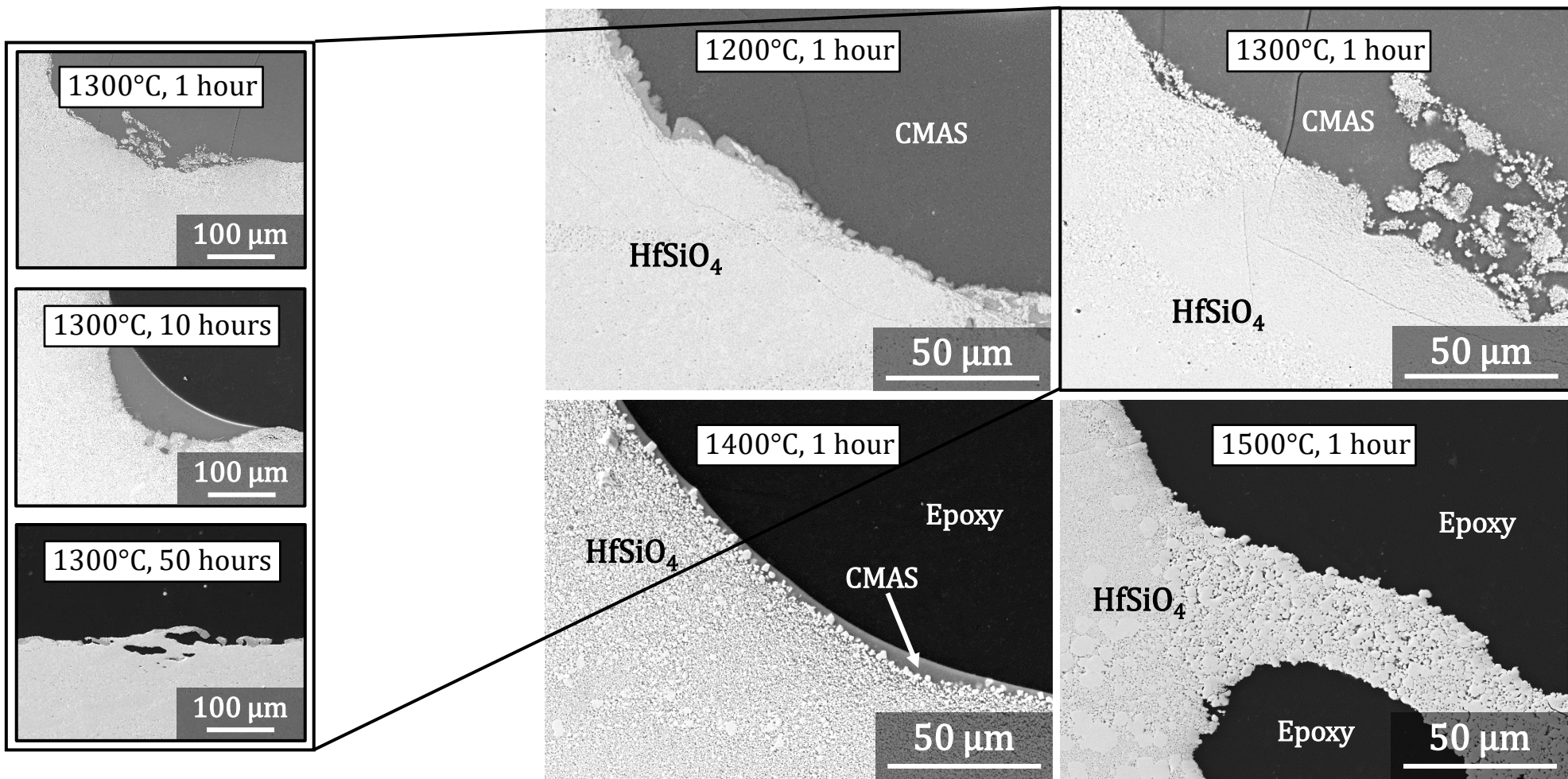


- Exposure to CMAS resulted in no infiltration at 1200°C

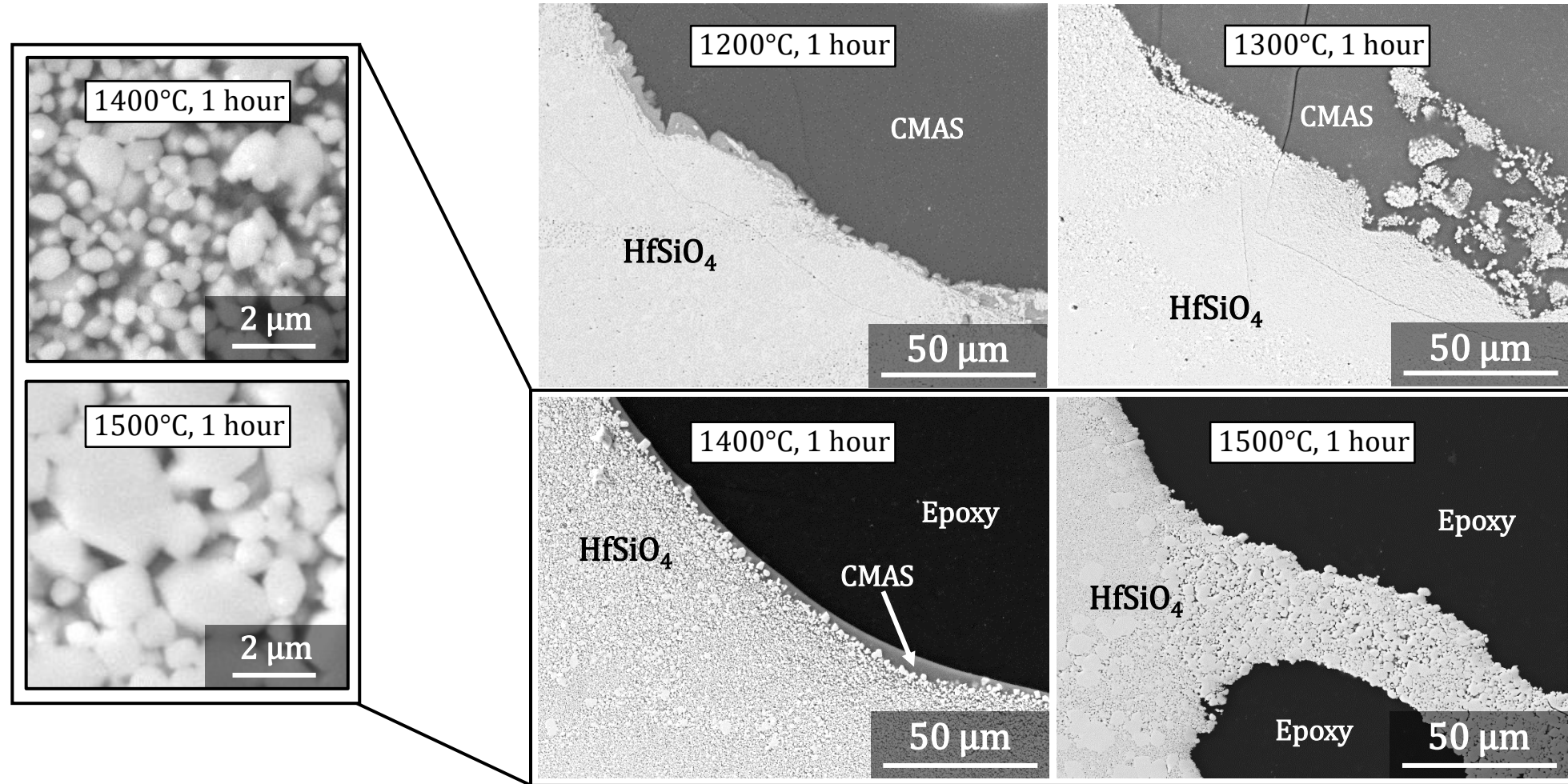




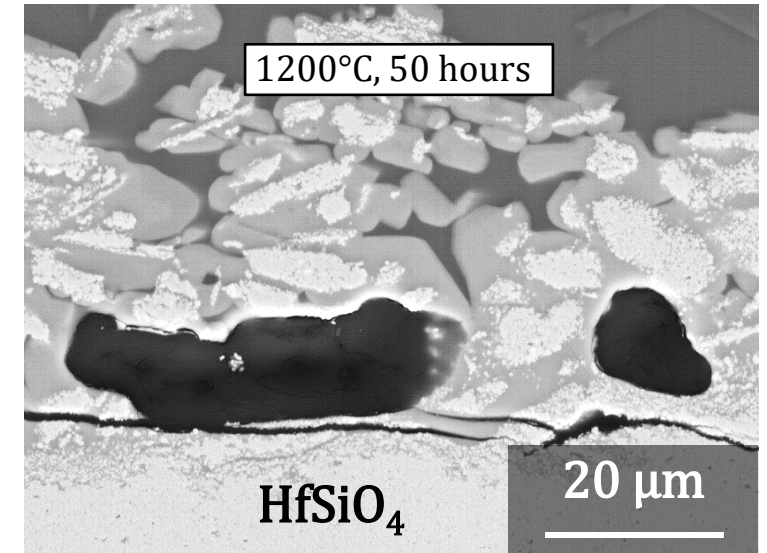
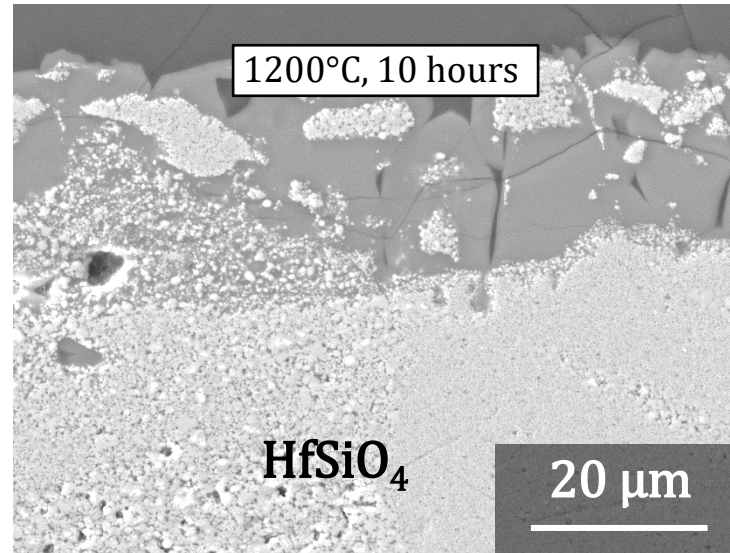
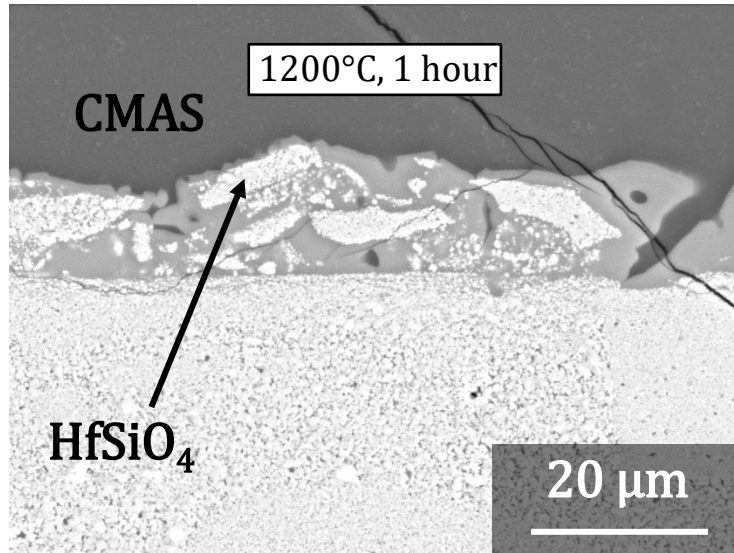
# Results – CMAS Infiltration



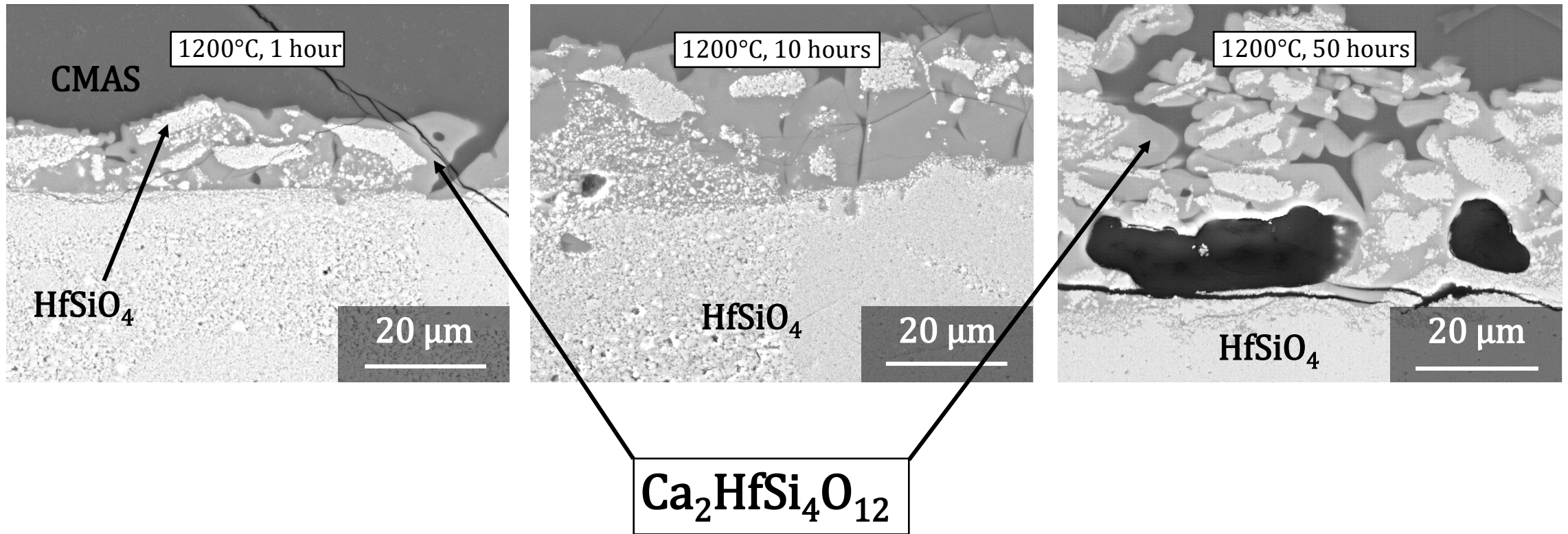




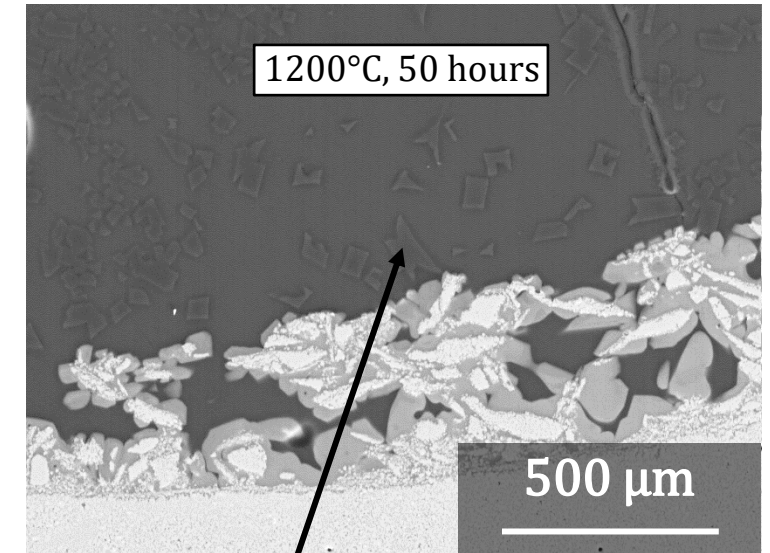
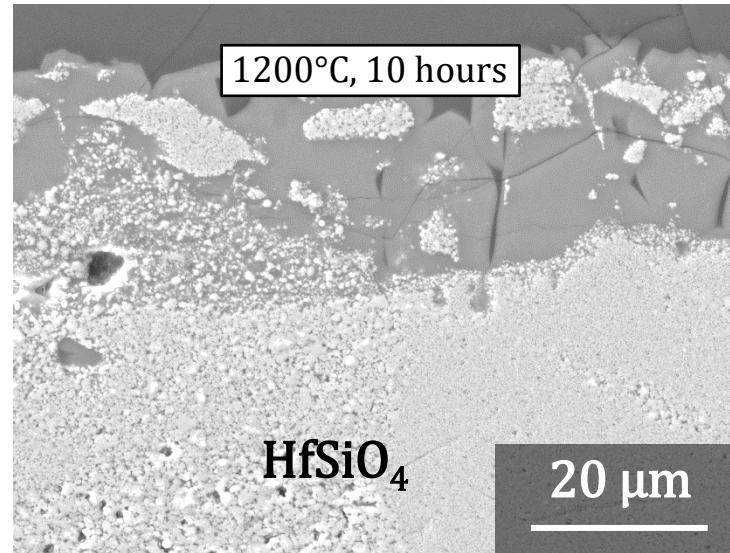
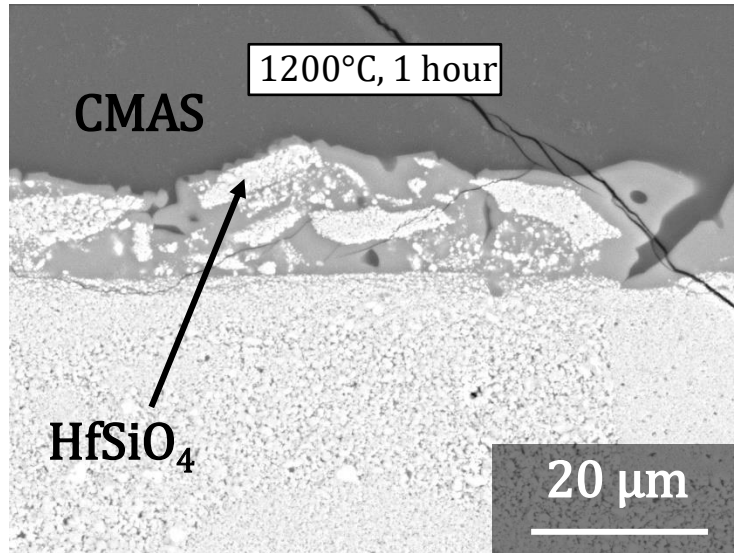
- Discontinuous interaction layer former at 1200°C increased in thickness with time
- Clusters of  $\text{HfSiO}_4$  remain in interaction region



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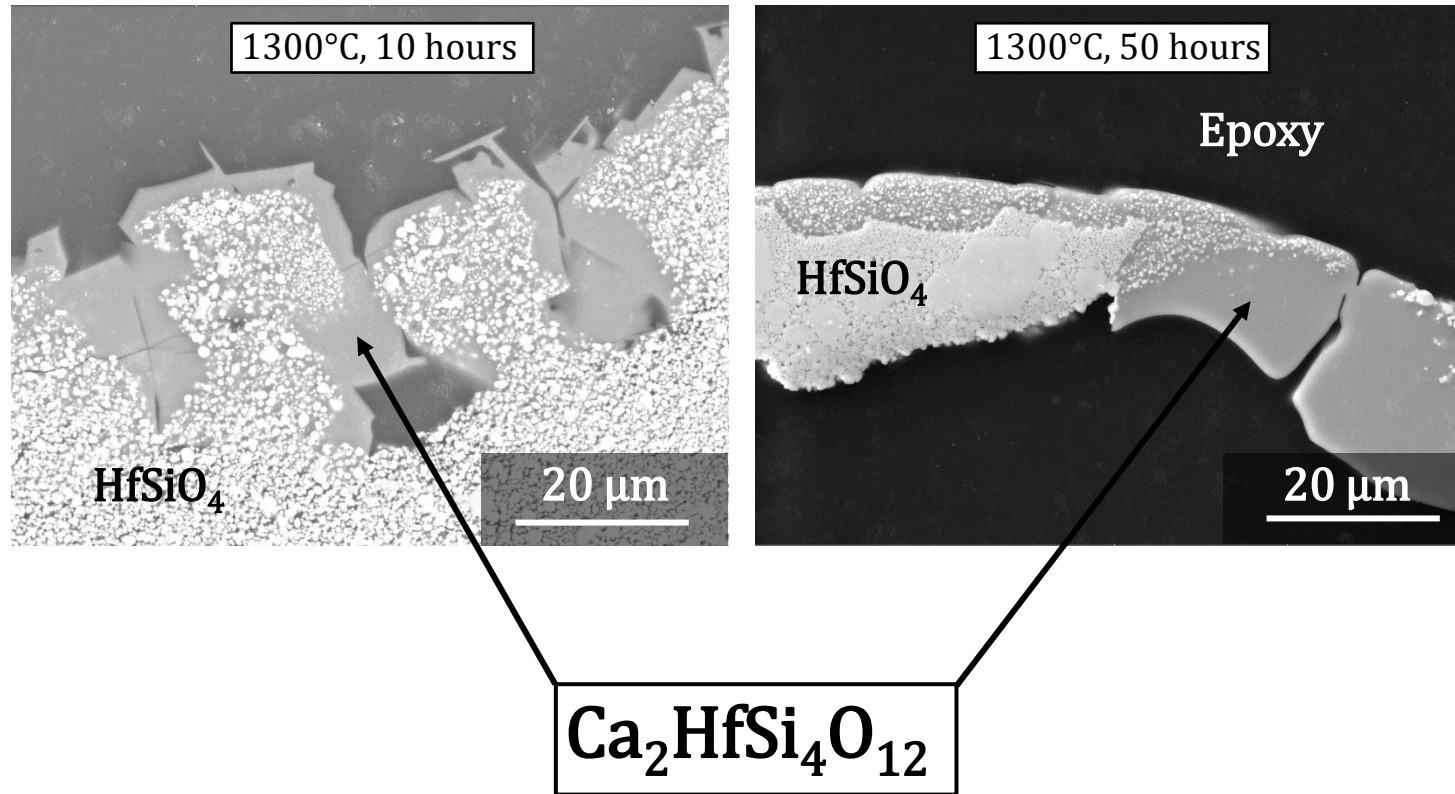


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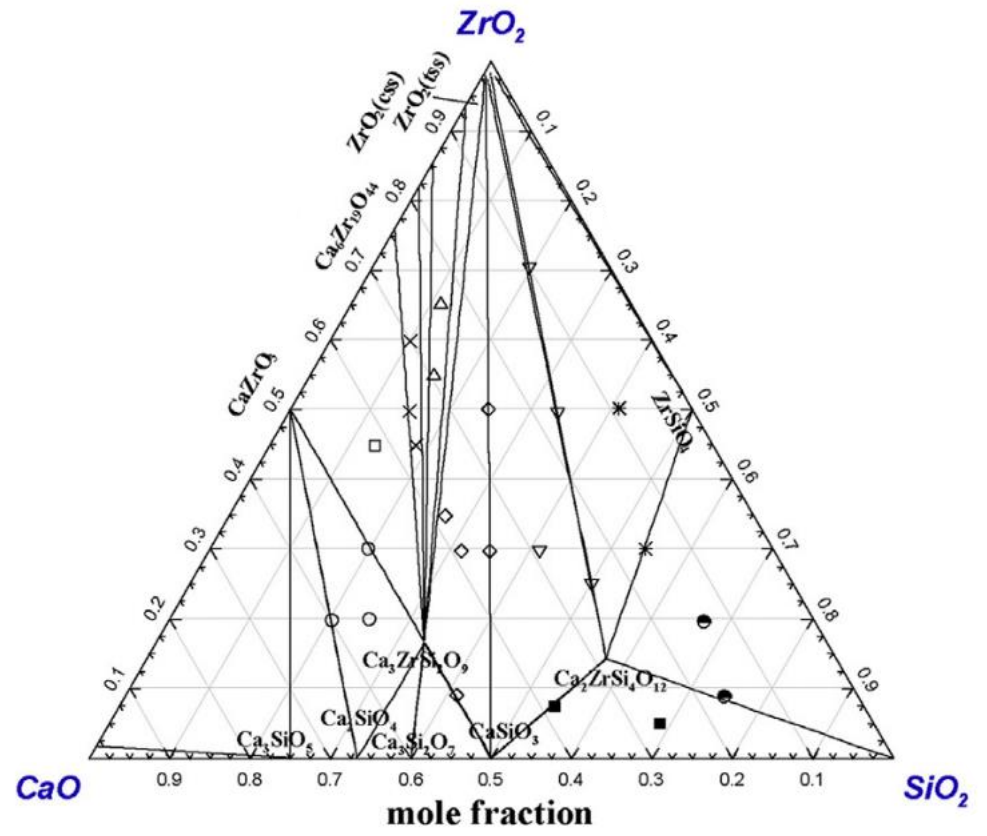


$\text{Ca}(\text{Mg},\text{Al})\text{Si}_2\text{O}_6$

- Cyclosilicate phase was observed at longer times at 1300°C but not after 1 hour

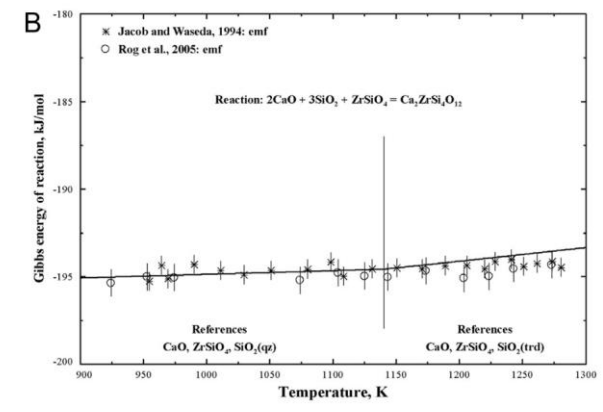


- Thermodynamic assessment of CaO-ZrO<sub>2</sub>-SiO<sub>2</sub> system at 1300°C

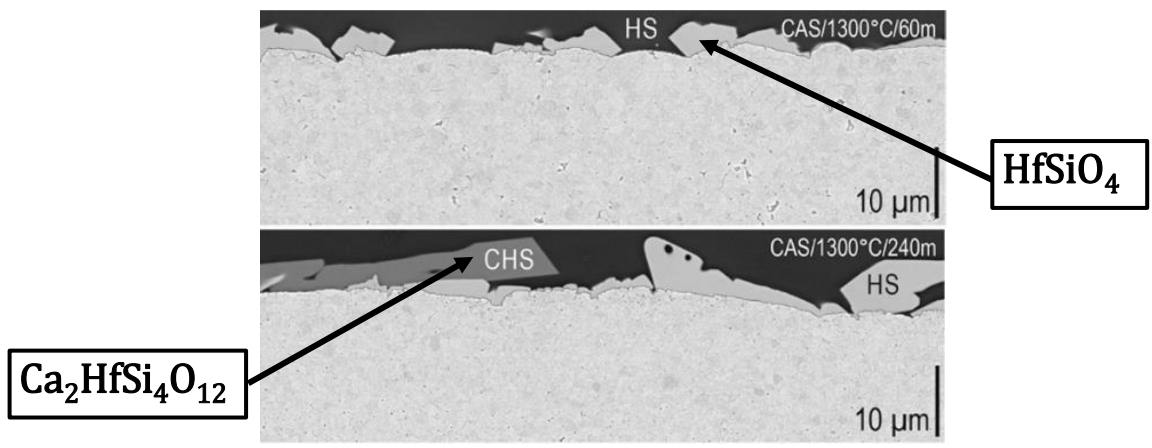


Kwon et al., *Journal of the European Ceramic Society* 37 (2017)

- Gibbs Free Energy Reaction

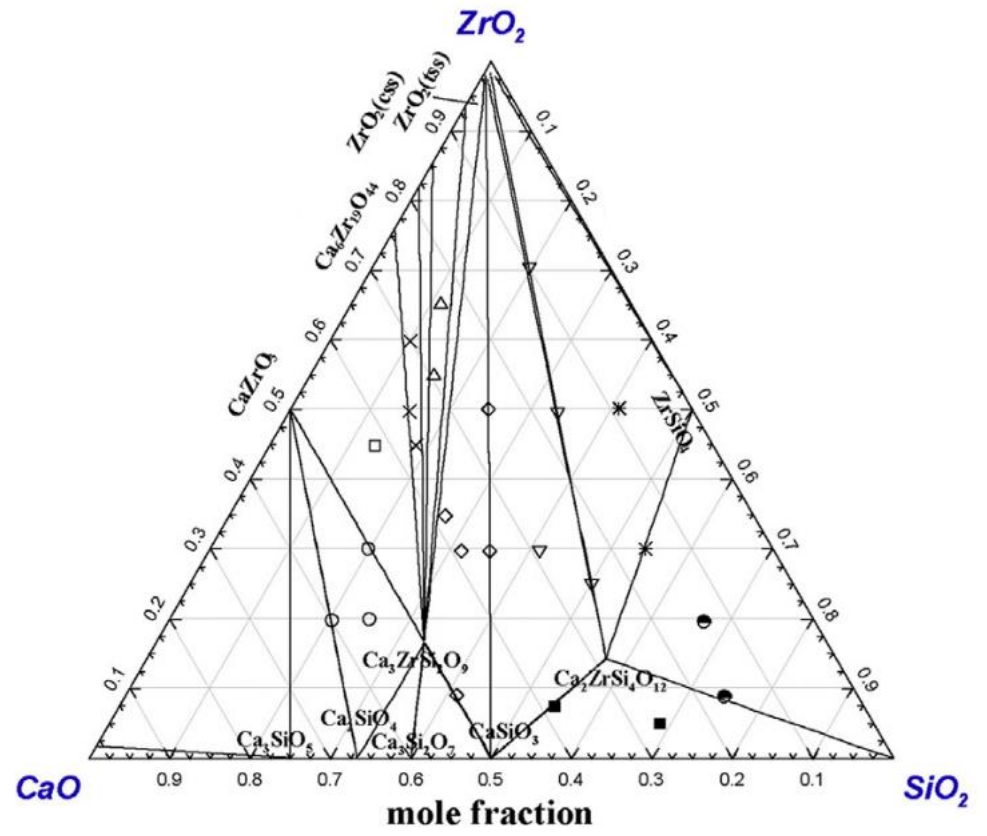


- CMAS Interaction with HfO<sub>2</sub>



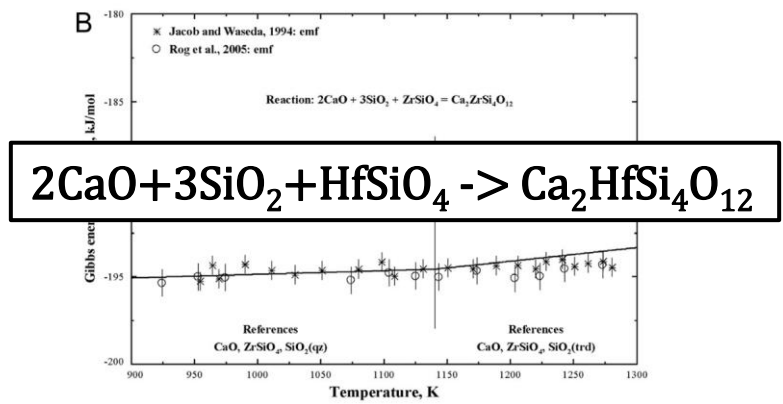
Holgate et al., *Journal of the European Ceramic Society* 41 (2021)

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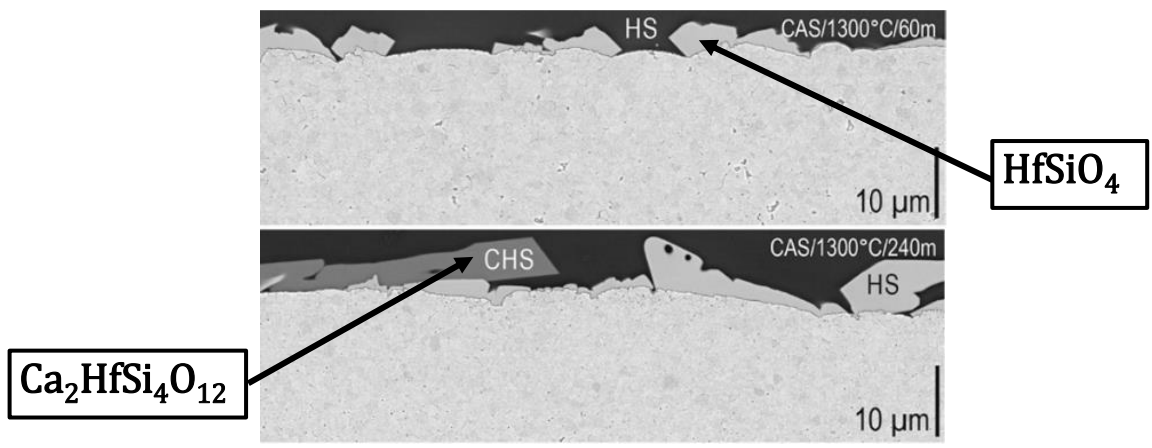


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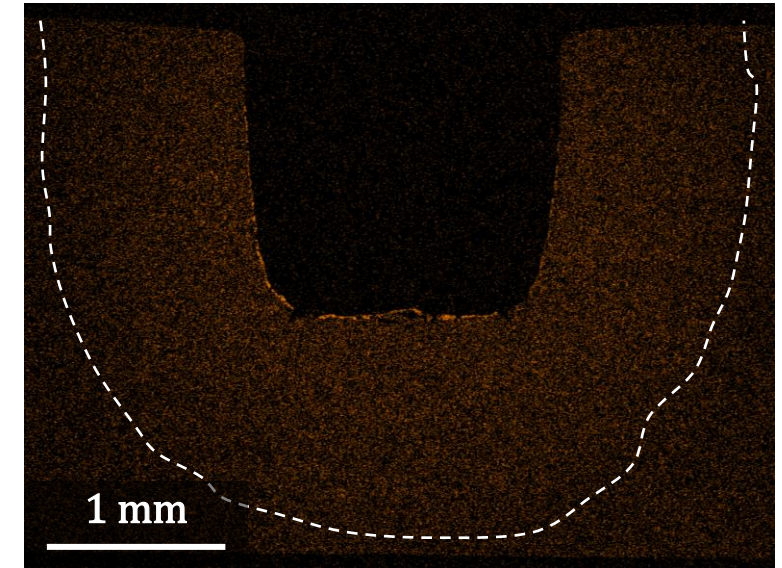
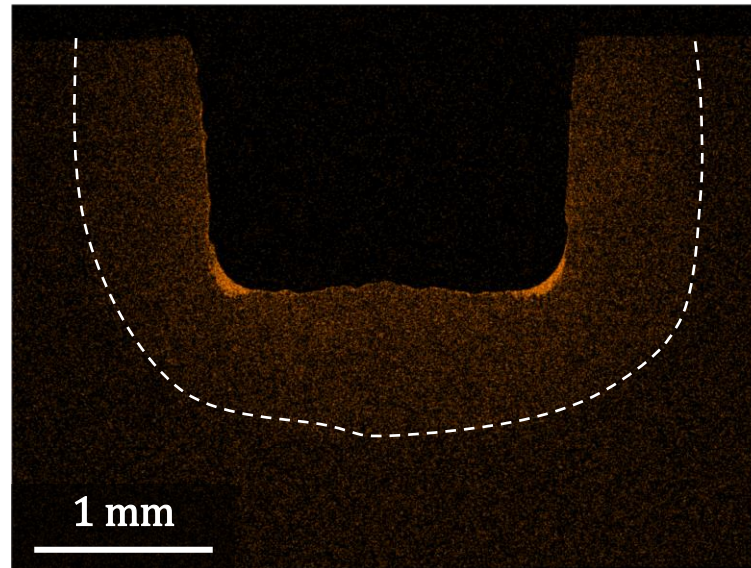
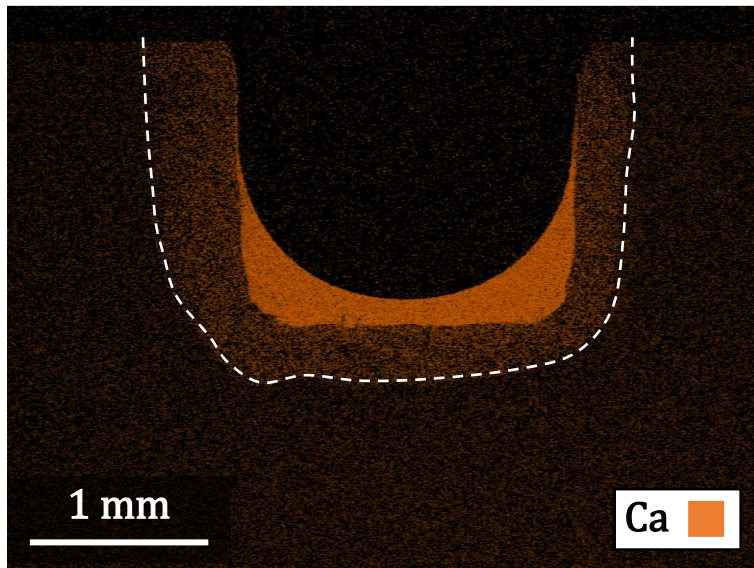


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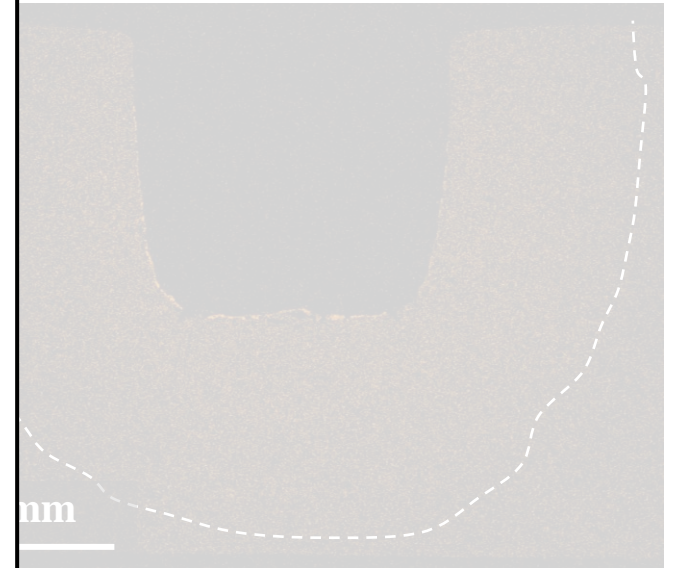
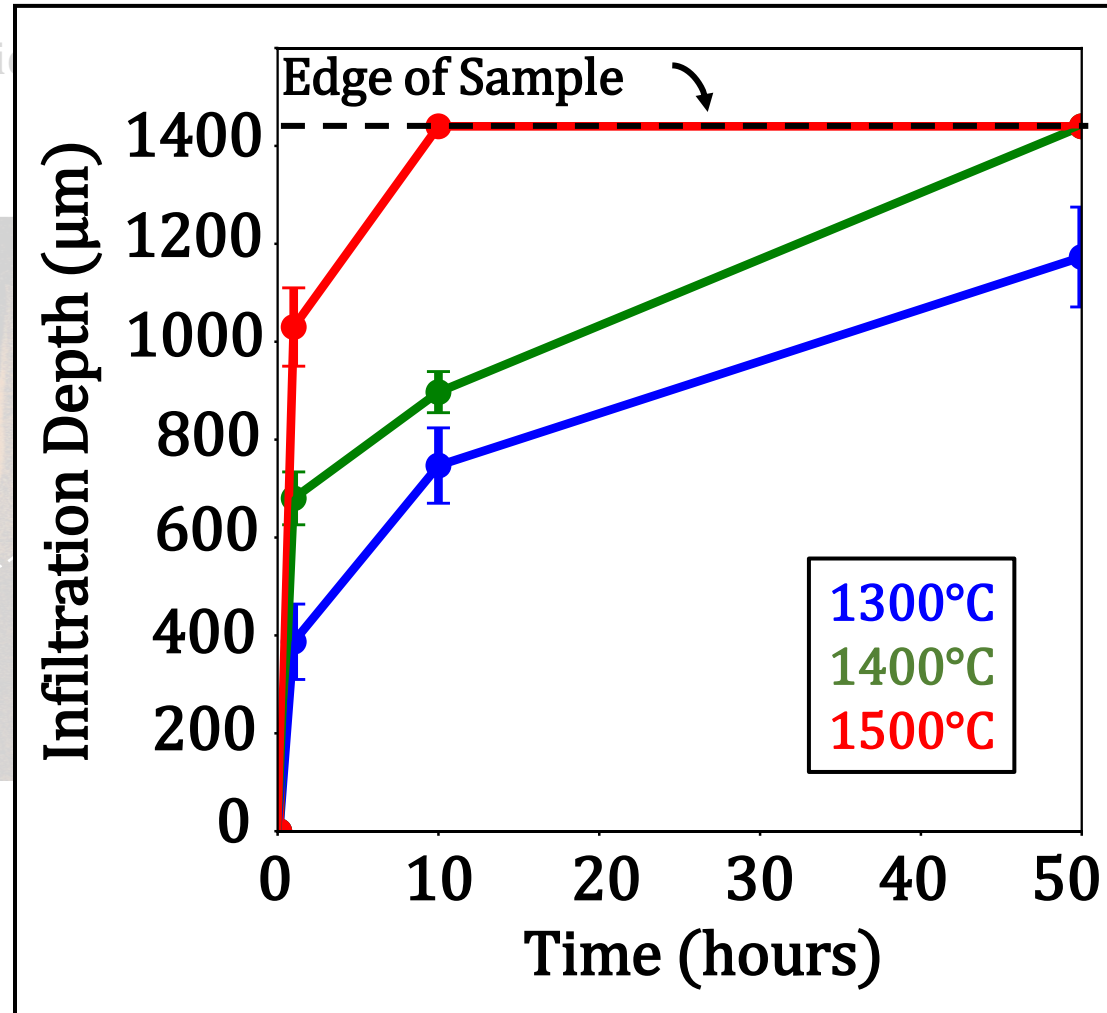
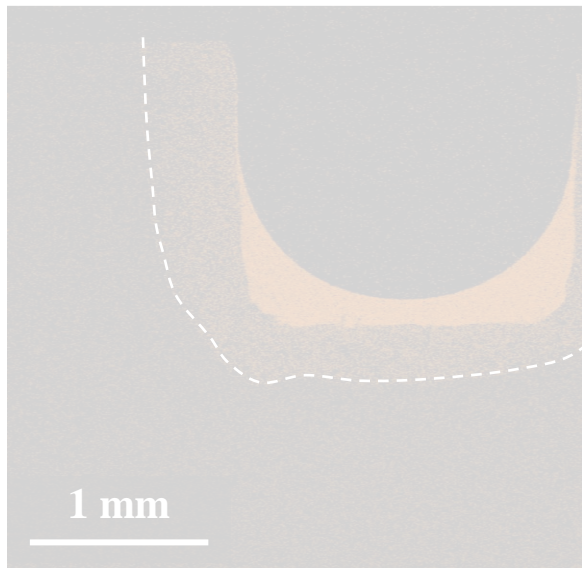
Holgate et al., *Journal of the European Ceramic Society* 41 (2021)

- EDS maps of CaO distribution at 1300°C

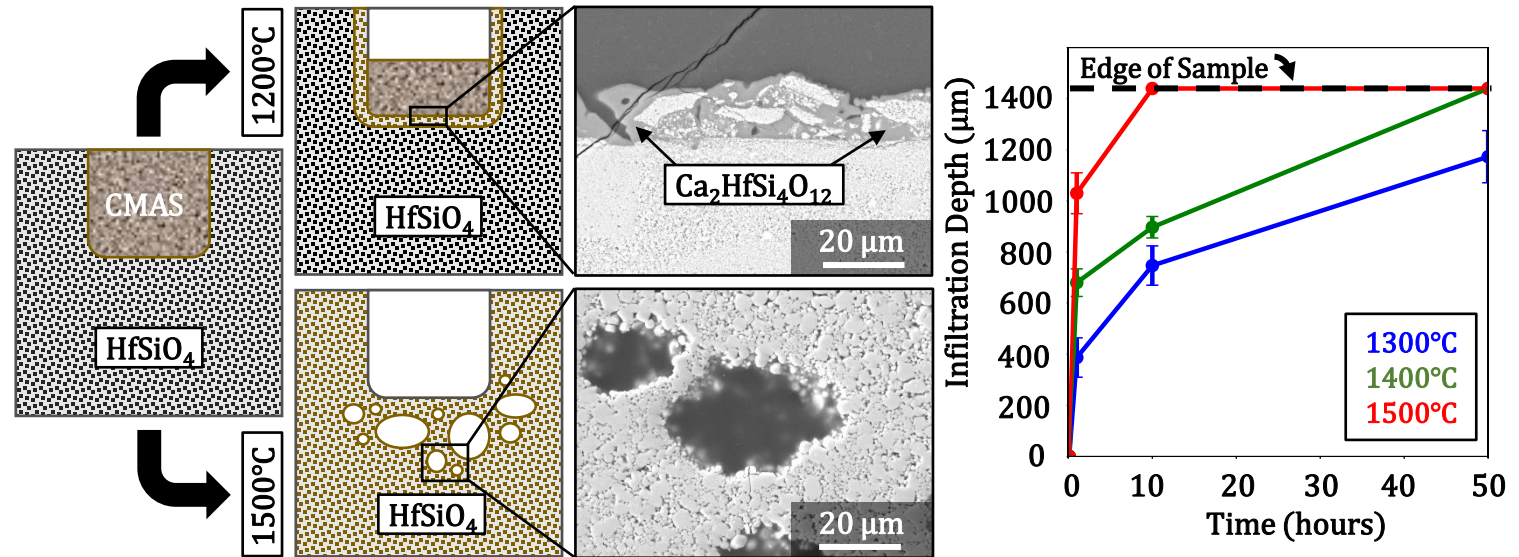




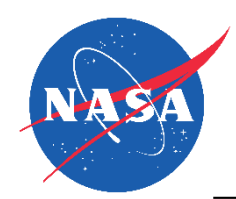
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- Hot pressed  $\text{HfSiO}_4$  was reacted with CMAS at 1200°C, 1300°C, 1400°C, and 1500°C.
- At temperatures above 1200C, CMAS rapidly infiltrates  $\text{HfSiO}_4$  through grain boundaries
- A slow growing cyclosilicate ( $\text{Ca}_2\text{HfSi}_4\text{O}_{12}$ ) phase crystallizes at 1200°C and 1300°C, but was not beneficial in halting CMAS ingress



Stokes, J.L., Bansal, N.P., Wiesner, V.L., “Thermochemical Degradation of  $\text{HfSiO}_4$  by Molten CMAS”, *Accepted, Ceramics International* (2022)



Thank You!