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3 On the Effects of the Ocean on Atmospheric CFC-11 Lifetimes And 4 Emissions

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6 Peidong Wang^a, Jeffery R. Scott^a, Susan Solomon^a, John Marshall^a, Andrew R. Babbin^a, Megan
7 Lickley^a, David W. J. Thompson^b, Timothy DeVries^c, Qing Liang^d, Ronald G. Prinn^a

8 ^aEarth, Atmospheric, and Planetary Sciences, Massachusetts Institute of Technology, Cambridge,
9 MA 02139; ^bDepartment of Atmospheric Science, Colorado State University, Fort Collins, CO

10 80523; ^cDepartment of Geography, University of California, Santa Barbara, CA 93106;

11 ^dAtmospheric Chemistry and Dynamics Laboratory, NASA Goddard Space Flight Center,
12 Greenbelt, MD 20771

13 *Peidong Wang

14 Email: pdwang@mit.edu

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22 **This PDF file includes:**

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28 **Abstract**

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29 The ocean is a reservoir for CFC-11, a major ozone-depleting chemical. Anthropogenic
30 production of CFC-11 dramatically decreased in the 1990s under the Montreal Protocol, which
31 stipulated a global phase-out of production by 2010. However, studies raise questions about
32 current overall emission levels, and indicate unexpected increases of CFC-11 emissions of about
33 10 Gg yr⁻¹ after 2013 (based upon measured atmospheric concentrations and an assumed
34 atmospheric lifetime). These findings heighten the need to understand processes that could affect
35 the CFC-11 lifetime, including ocean fluxes. We evaluate for the first time how ocean uptake and
36 release through 2300 affects CFC-11 lifetimes, emission estimates, and the long-term return of
37 CFC-11 from the ocean reservoir. We show that ocean uptake yields a shorter total lifetime and
38 larger inferred emission of atmospheric CFC-11 from 1930 to 2075 compared to estimates using
39 only atmospheric processes. Ocean flux changes over time result in small but not completely
40 negligible effects on the calculated unexpected emissions change (decreasing it by 0.4 ± 0.3 Gg
41 yr⁻¹). Moreover, it is expected that the ocean will eventually become a source of CFC-11,
42 increasing its total lifetime thereafter. Ocean outgassing should produce detectable increases in
43 global atmospheric CFC-11 abundances by the mid-2100s, with emission of around 0.5 Gg yr⁻¹;
44 this should not be confused with illicit production at that time. An illustrative model projection
45 suggests that climate change is expected to make the ocean a weaker reservoir for CFC-11,
46 advancing the detectable change in global atmospheric mixing ratio by about 5 years.

47 **Significance Statement**

48 Manufactured CFC-11 is depleting the Antarctic ozone layer. CFC production has been strictly
49 controlled by the Montreal Protocol, but emission estimates are very sensitive to choices of
50 lifetimes, which are often assumed to be constant over time. We employ a hierarchy of models to
51 study the effect of the ocean on the time-dependent uptake and release of atmospheric CFC-11.
52 The ocean is a sink for CFC-11 and significantly affects its total lifetime and hence the emission
53 inferred from concentration data of past decades. This has not been explicitly included in
54 international ozone assessments. We show that as anthropogenic production ceases, ocean
55 fluxes become more important, suggesting a need for further studies with high-resolution global
56 models linking atmospheric chemistry and ocean processes.

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59 **Main Text**

60

61 **Introduction**

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63 Man-made chlorofluorocarbons (CFCs) are the primary cause of the Antarctic ozone hole
64 (1). The atmospheric lifetimes of these chemicals range from about 50 – 500 years. The Montreal
65 Protocol agreed to a complete phase out of CFC production and consumption worldwide by 2010.
66 Evidence for healing of the Antarctic ozone layer has indeed emerged (2, 3), indicating the overall
67 success of the Montreal Protocol. Atmospheric loss processes of CFC-11, the most abundant
68 ozone-destroying chlorofluorocarbon, are due to photolysis and reaction with excited oxygen
69 (O¹D) once the gas reaches the stratosphere. The atmospheric lifetime of CFC-11 is assumed to
70 be inversely related to the atmospheric abundance of the molecule, with due consideration of the
71 lag times between tropospheric and stratospheric burdens (4). Given its lifetime of about 50 – 60
72 years and continued emissions from storage banks such as chillers and building insulation foams
73 (5), the CFC-11 inventory in the atmosphere is decreasing slowly. However, the rate of decrease
74 in atmospheric concentrations has been slowing down since about 2012, suggesting higher
75 overall emission and an unexpected additional post-2013 emission increase of CFC-11 of about 7
76 – 13 Gg yr⁻¹ (10 – 20 % of the total global emission during that time; 6, 7). The latter is clearly
77 inconsistent with the zero global new production that has been agreed by the Montreal Protocol.

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78 CFC-11 is soluble in water, and therefore the ocean has absorbed some CFC-11 from
 79 the atmosphere. CFC-11 ocean uptake is greatest in high latitudes where cold sea surface
 80 temperatures (SST) enhance CFC-11 solubility (8), and mixing and transport from the surface
 81 into the deep ocean is enhanced. By 1994, the ocean had stored up to 1 % of the total
 82 anthropogenic emissions of CFC-11 (9), and by 2014 the ocean held roughly 110 Gg of CFC-11
 83 (10), or about 5 – 10 % of the CFC-11 inventory in the various anthropogenic storage banks.
 84 While some CFC-11 is removed in sulfidic anoxic waters (11), this effect is small for the current
 85 climate, and CFC-11 has long been employed as a useful passive tracer to study ocean
 86 circulation (e.g. 12, 13). Early studies using a global model incorporating CFC-11 air-sea fluxes
 87 suggested that the ocean's effects on atmospheric CFC-11 lifetimes and concentrations were
 88 negligible in the 1980s, when anthropogenic emissions were high (14). However, now that
 89 anthropogenic emissions have dramatically decreased and attention is focused on unexpected
 90 emissions of 10 Gg or even less, changes in ocean uptake of CFC-11 could be affecting the
 91 atmospheric CFC-11 inventory enough to influence emission estimates, and could introduce a
 92 time-dependent effect on its total lifetime. Further, as anthropogenic emissions continue to
 93 decrease in the future, the ocean must eventually become supersaturated with CFC-11, and turn
 94 into a source instead of a sink. No study has yet estimated when that should be expected to
 95 occur, and what its magnitude will be.

96 Here, we address the following questions: (i) How is the ocean affecting the atmospheric
 97 CFC-11 inventory, the lifetime of CFC-11 in the atmosphere and its time dependence, and how
 98 does this in turn influence emission estimates based on observed concentrations?; (ii) When will
 99 the ocean become a source of CFC-11 to the atmosphere, and how much will ocean outgassing
 100 affect the apparent emission and atmospheric mixing ratio in the future? (iii) How will climate
 101 change affect ocean CFC-11 uptake in the future?

102 For a conceptual understanding, we use a hierarchy of models starting with a simple 6-
 103 box model that simulates the CFC-11 inventory in the atmosphere, ocean mixed layer, and deep
 104 ocean layers (each layer has 2 boxes representing the two hemispheres, see the schematic in
 105 Figure 1a). CFC-11 in each box is assumed to be well mixed in this illustrative model. The
 106 atmospheric CFC-11 lifetime is kept constant at 55 years and estimated emissions are taken from
 107 published work (15). We assume constant inter-hemispheric exchange timescales for each layer,
 108 and constant cross-layer timescales for mixed layer to deep ocean exchange (see Table S1).
 109 Atmospheric CFC-11's vertical distribution does affect its lifetime and surface concentration.
 110 Here, we subsume stratosphere-troposphere exchange into our adopted atmospheric lifetime
 111 estimates assuming a well-mixed atmosphere, and focus on the ocean's effect on atmospheric
 112 CFC-11. We then replace the four ocean boxes with a more sophisticated albeit low-resolution
 113 representation of the ocean, the MIT general circulation model (MITgcm; 16, 17), which includes
 114 a physics-based CFC-11 air-sea flux and transport into the interior ocean, and treats CFC-11 as a
 115 conservative tracer in the ocean (depicted in Figure 1b). The MITgcm (for brevity, we refer to the
 116 combined coupled box model atmosphere-ocean model simply as the MITgcm) is run in two
 117 modes. First, we use the model forced with climatological average wind stress and buoyancy
 118 fluxes (Hist run) to assess the influence of parameters including SST, wind stress, etc. on air-sea
 119 CFC-11 fluxes. Second, we force the MITgcm using global monthly RCP8.5 output from the MPI-
 120 ESM-LR (Max-Planck-Institute Earth System Model low resolution version) fully coupled global
 121 climate model (RCP8.5 run; 18, 19). This model has been shown to provide a realistic response
 122 of the Southern Ocean (55 – 70 °S), the region that stores the most CFC-11, to the southern
 123 annular mode (20). In the RCP8.5 run, interannual variability within the MPI-ESM-LR output
 124 provides changes in the forcing of the ocean applied after 1930, but variability in the atmospheric
 125 circulation is not explicitly incorporated into the box model atmosphere. We compare these runs
 126 to a “no ocean” run in which the CFC-11 air-sea flux is turned off. Both the box model and
 127 MITgcm runs extend from 1930 (essentially the start of emission of this anthropogenic gas) to
 128 2300.

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 130 **Results**
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133 **Near-term CFC-11**

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135 To evaluate the performance of the box model and MITgcm, we compared the computed
136 CFC-11 atmospheric concentrations with observations (Figure 2a). While the box model is
137 essentially tuned, the MITgcm setup is a fairly standard, off-the-shelf coarse resolution global
138 ocean model without any specific tuning for this application. Both the box model and MITgcm
139 agree well with observations, capturing the increase in CFC-11 before 1990 due to the large
140 anthropogenic emissions, as well as the concentration decrease after 1990 given the decline in
141 the emissions and losses due to chemical reactions in the atmosphere and exchange with CFC-
142 11 depleted ocean waters. The gradient in CFC-11 between the northern hemisphere (NH) and
143 southern hemisphere (SH) is also well captured. Because most of the emissions (around 90%)
144 occur in the NH and the CFC-11 lifetime is long, the NH minus SH difference can be up to 17 ppt
145 when anthropogenic emissions are large (comparable to 20 ppt reported in (21)). As the
146 emissions decrease, the inter-hemispheric exchange brings the NH and SH CFC-11 abundances
147 closer to each other. Both the box model and MITgcm slightly underestimate the observed CFC-
148 11 mixing ratio after 1990, as emissions decline (2.0 ppt lower in box model; 2.7 ppt lower in
149 MITgcm averaged from 1990 to 2017). Reasons for the underestimation of the atmospheric CFC-
150 11 concentration could be either 1) the ocean uptake is overestimated; 2) the CFC-11
151 atmosphere-only lifetime at that time is larger than our adopted constant value; and/or 3) CFC-11
152 emissions are higher than those prescribed in this simulation.

153 Figures 2b and c provide a qualitative comparison of the MITgcm global ocean column-
154 integrated CFC-11 with observations for 1994 (9). The MITgcm captures the spatial distribution of
155 observed CFC-11 in the ocean rather well given a well-mixed atmospheric CFC-11 distribution,
156 indicating that local dynamics is the driving factor for ocean uptake. Intercomparisons between
157 the MITgcm and other models of similar resolution as well as observations (22) also indicate that
158 the MITgcm does a favorable job in simulating CFC-11 distribution. In the MITgcm, CFC-11 is
159 overestimated in the Weddell Sea, although the Weddell Sea only stores less than 4.5 % of the
160 global ocean CFC-11 inventory (not shown) and has only a minor impact on the global estimate
161 (also note (22) showed large model spread in the Southern Ocean inventory due to ventilation
162 differences among models). Most CFC-11 in the NH ocean is stored in the North Atlantic, with
163 subduction into the thermocline and the Atlantic Meridional Overturning Circulation (AMOC)
164 playing key roles (13). The SH is the major reservoir for CFC-11, which is transported in Antarctic
165 Intermediate Water and Subantarctic Mode Water (23) and accumulates in the interior ocean
166 between 40 °S to 60 °S. In 1994, the best estimate of the global ocean CFC-11 inventory from the
167 World Ocean Circulation Experiment (WOCE) is 75.6 Gg (with cumulative error of 16.5 Gg; 9),
168 while the MITgcm suggests about 82 Gg of CFC-11 in the same year, which is slightly larger but
169 well within the uncertainty range of the observational value.

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172 **Box model sensitivity tests**

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174 For a conceptual understanding of the primary factors affecting CFC-11 inventories in the
175 atmosphere and ocean, we perturbed key parameters (Table S1) in the box model by $\pm 15\%$
176 (Figure S1a). Before 1990, all the sensitivity tests produce similar CFC-11 inventories for both the
177 atmosphere and ocean, underscoring that high anthropogenic emission dominated the behavior
178 during that time. After 1990 as global emissions decrease, the importance of other drivers in
179 affecting CFC-11 inventories increases. For example, a 15 % change in the prescribed
180 atmospheric CFC-11 lifetime affects the atmospheric inventory by up to about 570 Gg in 2050s,
181 or about 15 % of the total atmospheric CFC-11 inventory at that time. A 15 % change in mixed
182 layer depth (MLD; a proxy in the box model for the rate of ventilation of intermediate waters)
183 affects the atmospheric inventory by up to 13 Gg in 1990s. The biggest impact of MLD on CFC-
184 11 inventory is expected to occur in the 20th century because larger CFC-11 emissions and
185 undersaturated ocean waters result in the highest ocean uptake then. Changing the piston
186 velocity only has a small effect on the CFC-11 atmospheric inventory, up to 0.1 Gg. Changes in
187 inter-hemispheric exchange constants adopted for the atmospheric and ocean reservoirs only
188 affect the NH to SH gradient, but do not affect the total inventory in each reservoir, and $\pm 15\%$
189 changes in this parameter only generate differences within computational error. However, in the
190 real world if the exchange timescales between different CFC-11 reservoirs (for example between
191 the atmosphere and ocean, or the atmospheric loss in the stratosphere versus the troposphere)
192 are significantly different in each hemisphere, the effects of inter-hemispheric exchange could
193 become more significant.

194 The MLD in the box model affects the CFC-11 concentration in the shallow ocean boxes.
195 A deeper MLD implies that the ocean has a larger capacity to store CFC-11. This is crucial to
196 determining whether the ocean is supersaturated or undersaturated with CFC-11 at the air-sea
197 interface. Our box model assumes a constant MLD in time. In the real world and in more complex
198 ocean models, ocean circulation changes can be expected to be dominant factors driving surface
199 ocean CFC-11 concentration, and changes in the meridional overturning circulation with climate
200 change are likely to be important. This highlights the importance of using an ocean model with
201 realistic ocean dynamics to understand CFC-11 evolution in the atmosphere and ocean, as done
202 here with the MITgcm (albeit with low spatial resolution in this configuration of the MITgcm).
203 Figure S1b shows CFC-11 inventories using the MITgcm. Ocean inventories in the box model
204 and MITgcm agree well before 1990 because emission is the driving factor for CFC-11 air-sea
205 fluxes, but they deviate significantly in the future, when ocean dynamics begins to drive changes
206 in surface ocean CFC-11 concentration and the air-sea flux. Our box model only has two ocean
207 layers, which equilibrate CFC-11 between the atmosphere and ocean more rapidly than the
208 MITgcm. Further, some CFC-11 can be transported very deep in the ocean. With 15 ocean layers
209 in the MITgcm, the ocean is able to sequester more CFC-11 in the interior and it takes more time
210 to release that CFC-11 back to the atmosphere, such that the ocean CFC-11 inventory peaks in
211 the year 2075 in the MITgcm, roughly 80 years after the peak in atmospheric CFC-11
212 concentrations.

213 Box model results should be considered illustrative rather than quantitative regarding the
214 future CFC-11 inventory. Nonetheless, although some parameters in the box model may have co-
215 dependencies in the real world, our sensitivity analysis qualitatively highlights the importance of
216 two key factors that affect the CFC-11 inventory as anthropogenic emissions drop: the
217 atmospheric lifetime and ocean dynamics. We next focus on the MITgcm results to further
218 explore these issues.

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221 **Effect of the ocean on atmospheric CFC-11 concentration**

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223 We first present results using the climatological ocean forcing adopted in the Hist
 224 scenario. Figure 3a shows the difference in CFC-11 atmospheric inventories and abundances
 225 between the MITgcm run and no ocean runs. A similar plot but for dichlorodifluoromethane, or
 226 CFC-12, is shown in Figure S2. CFC-11 and CFC-12 are treated in the same manner in the
 227 MITgcm but CFC-12 is less soluble; therefore results for CFC-12 generally follow the same
 228 pattern as CFC-11 but the magnitude is smaller. Under this forcing, the cumulative effect of the
 229 ocean reaches its maximum in 2009, at which point the atmospheric CFC-11 inventory is 76.6 Gg
 230 less with the presence of the ocean (equivalent to 3.5 ppt less mole fraction) compared to the no
 231 ocean run. As anthropogenic emissions further decrease, the CFC-11 gradient between the
 232 atmosphere and the ocean decreases, decreasing the flux going into the ocean. Atmospheric
 233 CFC-11 differences between the ocean and no ocean runs reach zero around 2135. After that,
 234 the atmosphere accumulates more CFC-11 due to release from the ocean, and this outgassing
 235 accumulates in the atmosphere. Based on the current typical detection precision of CFC-11
 236 measurements (24) for the AGAGE (Advanced Global Atmospheric Gases Experiment) network,
 237 the net increase of global CFC-11 released from the ocean is expected to become detectable by
 238 2145 or earlier based on this model. At that point, the atmosphere is expected to contain about
 239 0.5 ppt more average CFC-11 compared to a no ocean run (and the global average abundance of
 240 CFC-11 is about 50 ppt at that time). Future instrument improvements may allow earlier
 241 detection. By 2225, the atmosphere contains about 16 Gg more CFC-11 than in the no ocean run
 242 (about 0.8 ppt). The ocean keeps releasing CFC-11 back to the atmosphere until the end of our
 243 study period. By the end of the run in 2300, the effect of the ocean on atmospheric CFC-11
 244 remains significant.

245 The calculated CFC-11 hemispherically integrated air-sea flux is shown in Figure 3b.
 246 Most of the uptake in the NH happens in northward-flowing western boundary currents of the
 247 North Atlantic and North Pacific (Figure S3a-d), due to local cooling and the upwelling of deep
 248 undersaturated water favorable for CFC-11 uptake in the subpolar gyres. Even though 90 % of
 249 the emissions are in the NH, inter-hemispheric exchange in the atmosphere mixes the CFC-11
 250 concentration quickly. The Southern Ocean is a hotspot of CFC-11 uptake due to cold SSTs,
 251 upwelling of CFC-poor deep waters, and strong surface winds conducive to CFC-11 uptake.
 252 Upwelling brings undersaturated circumpolar deep water to the surface south of the Antarctic
 253 polar front in the Southern Ocean, inducing CFC-11 uptake. These surface waters are
 254 transported northward and are ultimately subducted into the interior ocean with intermediate and
 255 mode water formation in the sub-Antarctic, accumulating CFC-11 in the interior ocean and
 256 preventing it from readily escaping back to the atmosphere in the near future (Figure S3e-h). Due
 257 to these processes, more than twice as much CFC-11 is effectively stored in the SH ocean
 258 reservoir. The loss of CFC-11 into the global ocean via the air-sea flux is about 8.8 % of the loss
 259 in the atmosphere in the 1950s (Table 1). The flux going into the ocean reaches a maximum in
 260 the 1980s at 3.6 Gg yr^{-1} . As anthropogenic emissions increase and more CFC-11 accumulates in
 261 the atmosphere given its long lifetime, loss in the atmosphere reaches a maximum in the 1990s
 262 (at 103.1 Gg yr^{-1}). By the 2010s, the flux going into the ocean is about 1.2 Gg yr^{-1} , or about 1.3 %
 263 of the loss occurring in the atmosphere at that time. It is noteworthy that the change in percent
 264 loss due to the ocean as compared to the atmosphere decreases by 7.5 % from 1950 to 2020,
 265 suggesting a similar fractional increase of the overall CFC-11 lifetime due to the weakening of the
 266 ocean uptake.

267 The calculated global net flux is expected to reverse direction around 2075, with the NH
 268 displaying an earlier release of CFC-11 to the atmosphere in 2067, while the SH begins
 269 outgassing in 2077 in this model. The reason for the late release of the CFC-11 flux in the SH is
 270 due to more CFC-11 being transported into the deeper ocean, which then takes longer to get
 271 back to the surface (Figure S3). The maximum flux of CFC-11 out of the ocean occurs in the
 272 2120s, with up to 0.5 Gg yr^{-1} of flux coming back into the atmosphere globally. By the end of
 273 2300, the total flux from the ocean is still 0.2 Gg yr^{-1} . At this point, the loss of CFC-11 in the
 274 atmosphere is only 1.4 Gg yr^{-1} given the low atmospheric burden. The effect of the ocean source
 275 is counteracting the atmospheric loss by 14 % in the 2290s, suggesting that the CFC-11 lifetime
 276 should continue to increase far into the future.

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279 **Effect of the ocean on CFC-11 lifetime and emission estimates**

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281 The effects of the ocean on CFC-11 lifetimes and therefore on emissions inferred from
 282 concentration data are significant. Figure 4a presents lifetimes calculated by taking the model-
 283 calculated atmospheric abundances of CFC-11 and dividing by the loss rates in the atmosphere-
 284 only and in the atmosphere and ocean together (and similar results for CFC-12 are also given in
 285 Figure S4). As expected, when only the atmospheric loss is considered, the lifetime is a constant
 286 55-year as prescribed, but the results including the ocean loss are quite different, at around 50-
 287 year in 1950, increasing to about 54-year by 2000 and 60-year by 2250.

288 To evaluate the effect of the ocean on inferred emissions estimates, we adopt the
 289 concentrations from the MITgcm as if they were measured data and infer emissions considering
 290 different lifetime assumptions (see Methods section). We then compare these inferred emissions
 291 to the emissions used to drive the model. As expected, inferred emission using the dynamic
 292 lifetime that includes both the ocean and atmosphere loss (the red curve in Figure 4a) fully
 293 recovers the input emissions that drive the MITgcm. Because knowing the exact atmospheric loss
 294 rate is not possible in real world, assumed constant atmospheric lifetimes are typically used to
 295 estimate emissions. We thus tested using constant 52- 55- and 58-year lifetimes to explore the
 296 range of uncertainty in emission estimates. From the 1970s to 1990s, when the ocean uptake
 297 was large, inferred emissions using a constant atmospheric lifetime of 52 years provide a closer
 298 match to the prescribed emissions that were input to the MITgcm. From 2000 and beyond, when
 299 the ocean uptake is small, inferred emissions using a constant 55-year atmospheric lifetime
 300 provide a closer match to the prescribed emissions, showing how the large ocean uptake in
 301 earlier decades is equivalent to having a shorter atmosphere-only CFC-11 lifetime. Thus, the
 302 ocean CFC-11 uptake acts to decrease the overall atmosphere plus ocean lifetime over 1970 –
 303 by about 3 years.

304 For the key period from 2002 – 2012 to 2014 – 2016, the increase in the input emission
 305 for the MITgcm is 11.2 Gg yr^{-1} , while the increase in the inferred emission assuming a constant
 306 55-year lifetime is 11.6 Gg yr^{-1} . This highlights the time-dependent influence of the ocean on
 307 atmospheric loss rates of CFC-11. If the ocean's role is ignored, and a constant atmosphere-only
 308 lifetime is assumed, then inferring emissions from concentration changes for 2014 – 2016
 309 compared to 2002 – 2012 would overestimate the unexpected emission of CFC-11 by 0.4 ± 0.3
 310 Gg yr^{-1} (assuming a constant lifetime of 55 ± 3 years).

311 In addition, the atmospheric CFC-11 lifetime has also been shown to be time-dependent
 312 rather than constant, largely as a result of the lag time between surface release and stratospheric
 313 loss (4). Figure S5 overlays calculated atmosphere-only lifetimes from a suite of chemistry-
 314 climate models studied in the SPARC (Stratosphere-troposphere Processes And their Role in
 315 Climate) intercomparison. While atmospheric processes alone act to decrease the calculated total
 316 lifetime from 1930-2010, ocean processes have the opposite effect. The total lifetime would be
 317 best captured by models including both effects, which offset each other to some extent. Changes
 318 in atmospheric lifetimes likely explain why our model underestimates the CFC-11 mole fraction
 319 after 1990 in Figure 2, since we used a constant atmospheric CFC-11 lifetime throughout those
 320 model runs.

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323 **Effect of climate change on CFC-11 ocean uptake**

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325 The MITgcm simulation under the MPI model's RCP8.5 scenario makes the ocean a
 326 weaker reservoir for CFC-11, leading to less uptake in the earlier period and less outgassing in
 327 the later period, and climate change affects the timing at certain critical periods. The global ocean
 328 starts to release CFC-11 in 2075 under Hist forcing, but the outgassing begins 10 years earlier
 329 under the RCP8.5 scenario (Figure 3b). Without climate change, the ocean's effect on the
 330 atmospheric concentration of CFC-11 becomes detectable after 2145, compared to 2140 under
 331 RCP8.5, suggesting that climate change accelerates the shift towards outgassing CFC-11.

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332 We tested the drivers of these changes using the MITgcm simulations with only certain
 333 ocean forcing fields changing under RCP8.5, in order to identify which factors dominate CFC-11
 334 ocean uptake under a changing climate. Zooming in on the period between 2050 and 2090 in
 335 Figure 3b, the flux of CFC-11 in the NH under the full RCP8.5 forcing most closely follows that
 336 obtained under SST + Qnet (surface heat flux, calculated as latent heat + sensible heat +
 337 shortwave + longwave) only forcing. The additional warming of the surface ocean under this
 338 forcing leads to more stratified conditions, and reduces the solubility of CFC-11 in seawater,
 339 which results in earlier outgassing in the NH. In SST + Qnet only and full RCP8.5 runs, the AMOC
 340 decreased similarly (Figure S6c), suggesting that changes in ocean circulation are also playing a
 341 major role in weakening uptake, as found in other model studies (13). In the SH, the SST does
 342 not increase as much as in the NH, due to the upwelling of deep cold water (25, 26); as such, we
 343 find that the SH air-sea CFC-11 flux is mainly affected by changes in salinity as forced by
 344 changes in evaporation-precipitation (E-P) and surface restoring of SSS. In particular, increases
 345 in net precipitation in the Southern Ocean (Figure S7f) decrease mixed layer depths, leading to
 346 weaker ventilation of the intermediate and deep ocean. Thus, more CFC-11 is stored in shallower
 347 ocean depths in response to these changes, leading to an earlier outgassing of CFC-11 to the
 348 atmosphere in the SH. Note that changes in SSS + E-P forcing do not cause an appreciable
 349 weakening of the AMOC (Figure S6c) in this model, unlike changes in SST + Qnet. We
 350 emphasize that other models could have different responses to these forcings, and this analysis
 351 is intended to be illustrative rather than quantitative.

352 When the ocean is acting as a sink for CFC-11, the atmosphere has up to 5.0 Gg more
 353 CFC-11 under full RCP8.5 forcing than under Hist forcing (Figure S8). In contrast, when the
 354 ocean turns into a source of CFC-11, the atmosphere has up to 2.4 Gg less CFC-11 with the full
 355 RCP8.5 scenario, which is due to weaker outgassing from the ocean. The combined effect of
 356 changes in SST, SSS, and buoyancy fluxes exceeds those in the full RCP8.5 forcing run,
 357 because the effects of wind stress on ocean circulation, and of sea ice fraction on air-sea CFC-11
 358 exchange partially counteract the ocean CFC-11 uptake due to thermal and saline changes in this
 359 model. Under RCP8.5, there is a poleward intensification of the SH westerly winds (Figure S7),
 360 which modifies the ventilation rate and transport of CFC-11 into the ocean (27). In a simulation
 361 that isolates the effects of changes in wind stress on the ocean dynamics, there is enhanced
 362 ocean uptake, especially over 50 – 60 °S during the early ocean sink period (Figure S3), and the
 363 atmosphere has up to 0.8 Gg less CFC-11 compared to the Hist forcing run. When the ocean
 364 turns into a source of CFC-11, changes in wind stress forcing can lead to 1.4 Gg more CFC-11 in
 365 the atmosphere, due to enhanced equatorward Ekman transport and stronger upwelling in the
 366 Southern Ocean (28).

367 Discussion

371 Previous work has not explicitly analyzed the effects of the ocean on atmospheric CFC-
 372 11 and has generally assumed that the effects of ocean uptake and outgassing can be accounted
 373 for by adjusting the uncertainty in atmospheric lifetimes. The results shown here reveal that
 374 ocean uptake and outgassing have a much more pronounced effect on our understanding of the
 375 lifetime of atmospheric CFC-11 than previously anticipated. The results have small but significant
 376 implications for past CFC-11 emission estimates, and key conceptual implications for the future.

377 Here we summarize our findings on the three primary questions posed in the introduction:
 378 First, our model suggests that the ocean's CFC-11 uptake ability varies significantly in time,
 379 translating to time-dependence in the total CFC-11 lifetime if the ocean's effect is subsumed into
 380 the atmospheric lifetime estimate. This result does not significantly affect calculated ozone
 381 depletion or radiative forcing, which often employ prescribed concentrations based on
 382 observations. The significance of our work is that knowledge of lifetimes is required to estimate
 383 emissions from concentrations and, in turn, to examine emissions sources and consistency with
 384 the Montreal Protocol. The calculated 7.5 % increase in lifetime from the 1950s to the 2010s due
 385 to weakening ocean uptake affects estimates of CFC-11 emissions by up to 4 Gg yr⁻¹, and it also
 386 affects their time dependence, compared to calculations neglecting this effect. We estimate that

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387 the ocean's influence reduces inferred unexpected emission of CFC-11 after 2013 (6, 7) by about
388 $0.4 \pm 0.3 \text{ Gg yr}^{-1}$ (assuming a constant lifetime of 55 ± 3 years) compared to calculations that
389 neglect the ocean effect. This is because the ocean's weakening sink leads to an increased
390 accumulation of CFC-11 in the atmosphere, which biases estimates of new emissions if the
391 ocean's effect is unaccounted for.

392 Second, a global net flux coming out of the ocean is projected to begin around 2075, and
393 the release of CFC-11 from this bank implies an accumulating influence on atmospheric CFC-11
394 abundances that should become detectable in the global average after about 2145, with
395 outgassing up to 0.5 Gg yr^{-1} . Detectable signals could be greatly enhanced and occur sooner if
396 observation sites are located close to ocean upwelling regions, where stronger CFC-11
397 outgassing can be expected (Figure S3). The ocean ultimately leads to up to a 0.8 ppt increase in
398 the global average atmospheric abundance by 2225. Such observations will signal the return of
399 CFC-11 from the ocean, rather than new production outside the Montreal Protocol at that time.

400 Finally, an illustrative model projection suggests that climate change will likely make the
401 ocean turn into a source of CFC-11 about 10 years earlier, and will make the effect on
402 atmospheric mixing ratio detectable 5 years earlier according to the scenario presented here.
403 Different models or scenarios could yield differences in detail regarding these findings but are
404 unlikely to alter the general result.

405 In closing, we note that our results illustrate the importance of the ocean in the new era of
406 the Montreal Protocol in which global anthropogenic productions of ozone depleting substances
407 (ODSs) has dramatically decreased, which means that small sources, sinks, or differences in
408 estimates of lifetimes have now become extremely important because they affect emissions
409 estimates. Atmospheric CFC-11 is not the only ODS taken up to some extent by the ocean. Other
410 gases including for example CFC-12, CCl_4 (carbon tetrachloride) and CH_3CCl_3 (methyl
411 chloroform) are also subject to significant ocean uptake and sequestration, even though it has
412 been demonstrated that CCl_4 and CH_3CCl_3 are not entirely conserved within the ocean (29, 30,
413 31, 32). Indeed CFC-11 is also not entirely conserved in sufficiently anoxic water characterized by
414 sulfide accumulation (11). Whether this effect could become more significant in future climates
415 depends on where and how deep the ocean sequesters CFC-11, and if sizable regions of anoxic
416 conditions develop in future oceans. Together with changes in ocean temperatures and
417 circulation patterns, these effects could be important in the future for detection of global and
418 regional sources of ODSs. This work highlights the need for the atmospheric chemistry and
419 oceanography communities to further examine these questions involving other ODSs. High-
420 resolution global atmosphere-ocean models, and continued observational programs for global
421 monitoring of ODSs in both the atmosphere and ocean, will be key tools for predicting and
422 detecting these changes in the future.

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423 **Materials and Methods**

424

425 **Emission data:**426 Global CFC-11 emissions up to 2016 (15) are split into NH and SH (33). We adopted a simple
427 constant declining rate of CFC-11 emission from 1995 to 2016 to predict emissions beyond 2016.
428 Emissions in both hemispheres reached zero at around 2100 and were kept zero to the end
429 2300. A similar treatment is applied for CFC-12 emissions.

430

431 **Box model equations:**

432

$$\frac{dm_{nh}^a}{dt} = E_{nh} + F_{nh} - (T_{n2s}^a + L_{nh}^a) m_{nh}^a + T_{s2n}^a * m_{sh}^a \quad (2)$$

$$\frac{dm_{sh}^a}{dt} = E_{sh} + F_{sh} - (T_{s2n}^a + L_{sh}^a) m_{sh}^a + T_{n2s}^a * m_{nh}^a \quad (3)$$

$$\frac{dm_{nh}^t}{dt} = -F_{nh} - T_{n2s}^t * m_{nh}^t + T_{s2n}^t * m_{sh}^t \quad (4)$$

$$\frac{dm_{sh}^t}{dt} = -F_{sh} - T_{s2n}^t * m_{sh}^t + T_{n2s}^t * m_{nh}^t \quad (5)$$

$$\frac{dm_{nh}^d}{dt} = - (T_{nh}^{d2t} + T_{n2s}^d) * m_{nh}^d + T_{nh}^{t2d} * m_{nh}^t + T_{s2n}^d * m_{sh}^d \quad (6)$$

$$\frac{dm_{sh}^d}{dt} = - (T_{sh}^{d2t} + T_{s2n}^d) * m_{sh}^d + T_{sh}^{t2d} * m_{sh}^t + T_{n2s}^d * m_{nh}^d \quad (7)$$

$$F_{nh} = -k ([CFC]_{nh}^a - [CFC]_{nh}^t) \quad (8)$$

$$F_{sh} = -k ([CFC]_{sh}^a - [CFC]_{sh}^t) \quad (9)$$

433 where m indicates mass of CFC-11, L is the CFC-11 atmospheric loss rate, T is the exchange
434 timescale between each box, E is the CFC-11 emission, F is the flux of CFC-11 between the
435 atmosphere and ocean, k is the piston velocity at 10 cm hr⁻¹ (34), $[CFC]$ indicates the
436 concentration in the atmosphere and mixed layer boxes. Superscripts indicate layers and
437 subscripts indicate the hemisphere. Table S1 shows a description of each term as well as the
438 numerical values associated with each parameter.

439

440 **MITgcm model:**

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441 Emission, atmospheric loss rate, and the atmosphere inter-hemispheric exchange timescale for
 442 CFC-11 in the MITgcm runs are the same as used in the box model setup. The CFC-11 air-sea
 443 flux equation is the same as shown in equations 8 – 9 (35), except that each term is calculated
 444 locally and dynamically given changes in the ocean forcing. Transport of CFC-11 within the ocean
 445 is done by the MITgcm. The MITgcm ocean model used here has $2.8^\circ \times 2.8^\circ$ horizontal resolution
 446 and 15 vertical layers down to 5000 m. This is a fairly coarse resolution, but appears sufficient to
 447 provide a first order estimate of CFC-11 uptake in the past based on Figure 2, and hence is used
 448 to estimate the long-term future. A mesoscale eddy parameterization (36) is used with an eddy
 449 diffusivity set to $1000 \text{ m}^2 \text{ s}^{-1}$. Convective adjustment is applied to statically unstable water
 450 columns, and background vertical diffusivity is set to $5 \times 10^{-5} \text{ m}^2 \text{ s}^{-1}$. Improved resolution and
 451 adjustments in ocean model parameters have been shown to impact details of CFCs storage
 452 within the ocean (13, 37), but we would not expect such changes to alter our qualitative results.
 453 Figure S6 shows the resulting residual mean meridional overturning circulation in the Atlantic
 454 Ocean, which appears reasonable.

455

456 **MITgcm forcing fields:**

457 The MITgcm is forced by monthly net E-P-R and Qnet climatologies (38). In addition, SST and
 458 SSS in the upper 50 m ocean layer are restored to a monthly climatology based on survey data
 459 taken from 1950 to 1990 (39, 40), with restoring timescales of 60 and 90 days respectively. Wind
 460 stress and wind speed climatologies over this period are from the European Centre for Medium-
 461 Range Weather Forecasts (ECMWF; 41, 42). Wind speed is decoupled from the wind stress in
 462 this setup and is only used to calculate the piston velocity for CFC-11 air-sea flux (i.e., it has no
 463 impact on ocean dynamics, in contrast with applied wind stress). The monthly sea ice
 464 concentration climatology (43, 44), like the surface wind speed, is only used in the calculation of
 465 CFC uptake; note our MITgcm setup does not include a prognostic sea ice model. The ocean
 466 was subject to 5900 years of spinup with the above climatology to produce an equilibrium state,
 467 followed by runs with CFC-11 from 1930 to 2300.

468

469 **MITgcm RCP8.5 setup:**

470 In this simulation the ocean dynamical model is spun up to equilibrium in the same way as the
 471 Hist run. Modified forcing fields using RCP8.5 output from MPI-ESM-LR (18, 19) during the
 472 transient simulation from 1930-2300 were constructed as follows. First we coarse-grained the
 473 output from MPI-ESM-LR using a nearest neighbor algorithm to MITgcm's $2.8^\circ \times 2.8^\circ$ resolution,
 474 and constructed a base period from 1850 to 1930 (using MPI-ESM-LR's historical simulation).
 475 Monthly anomalies from 1930 to 2300 compared to the base period were then added to the
 476 MITgcm Hist forcing discussed above (anomalies were imposed beginning in the early 20th
 477 century to avoid any abrupt changes in forcing fields that might cause unphysical changes in
 478 ocean circulation). Figure S7 shows the spatial patterns of the RCP8.5 anomalies for all the
 479 forcing variables nudged in MITgcm. Unlike in the Hist simulation, in the RCP8.5 simulation the
 480 ocean circulation and air-sea gas exchange piston velocity both evolve over time. AMOC strength
 481 in the MITgcm RCP8.5 run decreases by 60 % from 1930 – 2000 to 2200 – 2300 (Figure S6),
 482 comparable to a 56 % decrease obtained in the MPI-ESM-LR.

483

484 **Top-down inferred emission based on the MITgcm output:**

485 We inferred the emission and the atmospheric lifetime from the MITgcm output of the CFC-11
 486 (and CFC-12) inventory by:

487

$$E_{inf} = \frac{dm}{dt} + \frac{m}{\tau} \quad (10)$$

43

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$$\tau_{atm} = \frac{m}{L_{nh}^a + L_{sh}^a} \quad (11)$$

$$\tau_{atm+ocn} = \frac{m}{L_{nh}^a + L_{sh}^a + F_{nh} + F_{sh}} \quad (12)$$

488 where E_{inf} is the inferred emission, m is the mass of CFC-11 (and CFC-12) as calculated from
 489 the MITgcm, $\frac{dm}{dt}$ is the tendency of the CFC-11 (and CFC-12) inventories. τ_{atm} is the inferred
 490 CFC-11 (and CFC-12) lifetime only considering the atmospheric loss, and $\tau_{atm+ocn}$ is the total
 491 lifetime considering both the atmospheric loss and the bi-directional flux from the ocean. τ in
 492 equation 10 is replaced with inferred lifetimes from equations 11 – 12 as well as assumed
 493 constant 52- and 58-year lifetime for the inferred emission estimates in Figure 4b for CFC-11
 494 (assumed constant 90- and 100-year lifetime in Figure S4b for CFC-12).

495

496 **Data availability:**

497 The MITgcm output under different forcing runs and the code used here are available publicly on
 498 Zenodo (DOI: 10.5281/zenodo.4435502).

499

500

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502

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510 **References**

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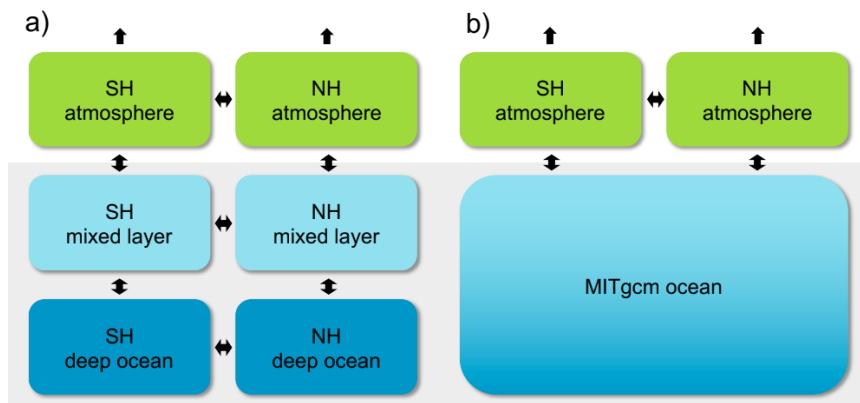
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617 **Figures and Tables**

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619

620 **Figure 1.** Schematic diagrams showing a) the box model; and b) the MITgcm setup. The box
 621 model has three layers that represent the atmosphere, ocean mixed layer and deep ocean. Each
 622 layer has two boxes that indicate the NH and the SH. The MITgcm setup replaces the four ocean
 623 boxes with the MITgcm ocean but keeps the atmospheric boxes unchanged. One-way arrows
 624 indicate CFC-11 atmospheric loss; two-way arrows indicate CFC-11 transport into/out of the box.

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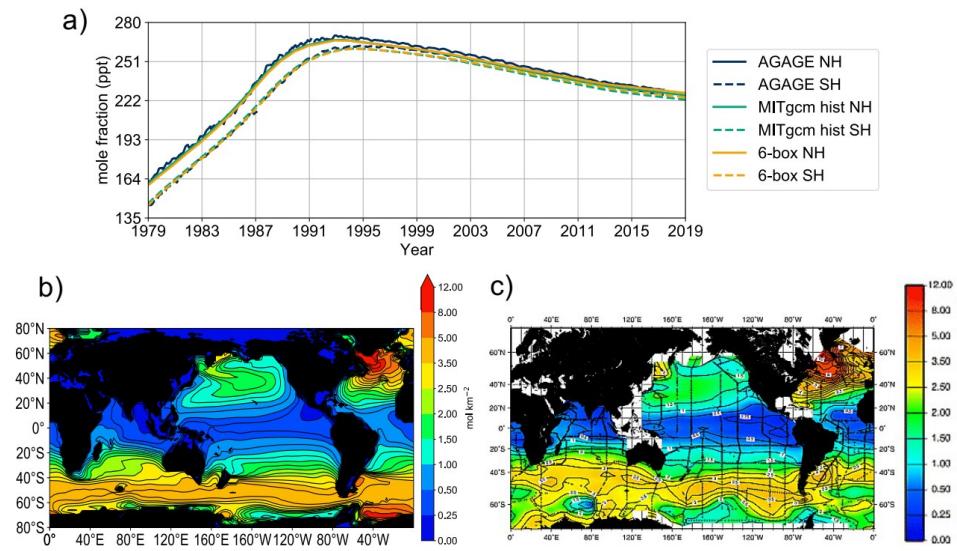
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 629 **Figure 2.** a) Model validation of CFC-11 atmospheric abundances compared to CFC-11
 630 atmospheric surface data (14); b) the MITgcm ocean column integrated CFC-11 under Hist run;
 631 and c) observations of ocean column integrated CFC-11 (9). Panels b and c both show the CFC-
 632 11 inventories in the year 1994, and are in the same units of mol km⁻².
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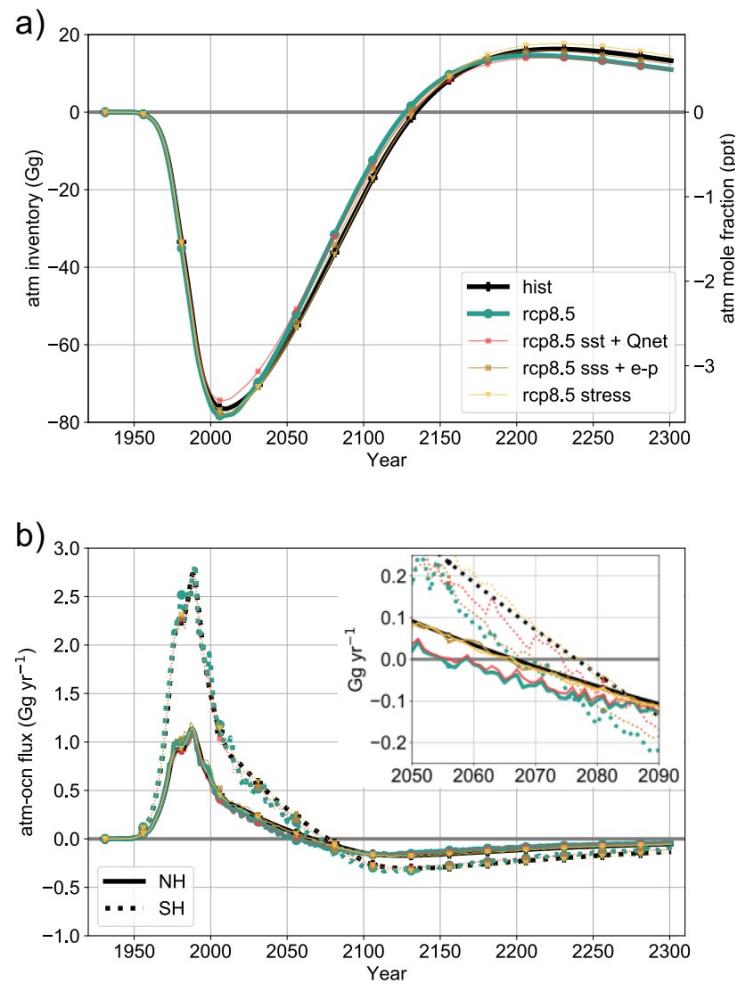
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635 **Figure 3.** a) Atmospheric CFC-11 inventory (left axis) and abundance (right axis) for different
 636 ocean forcing tests (different colors and markers) in the MITgcm minus the atmosphere under no
 637 ocean run; results are shown both for the Hist and RCP8.5 runs. The differences between sea ice
 638 only and the wind speed only forcing results are nondifferentiable from the result using historical
 639 forcing, especially after 2100, and are omitted from this figure; b) MITgcm hemispherically-
 640 integrated CFC-11 air-sea flux under different ocean forcings. Positive values indicate fluxes
 641 going from the atmosphere to the ocean. Solid lines are the NH integrated flux and dashed lines
 642 are the SH integrated flux. Bottom panel is zoomed in between 2050 and 2090, when the flux
 643 changes sign.

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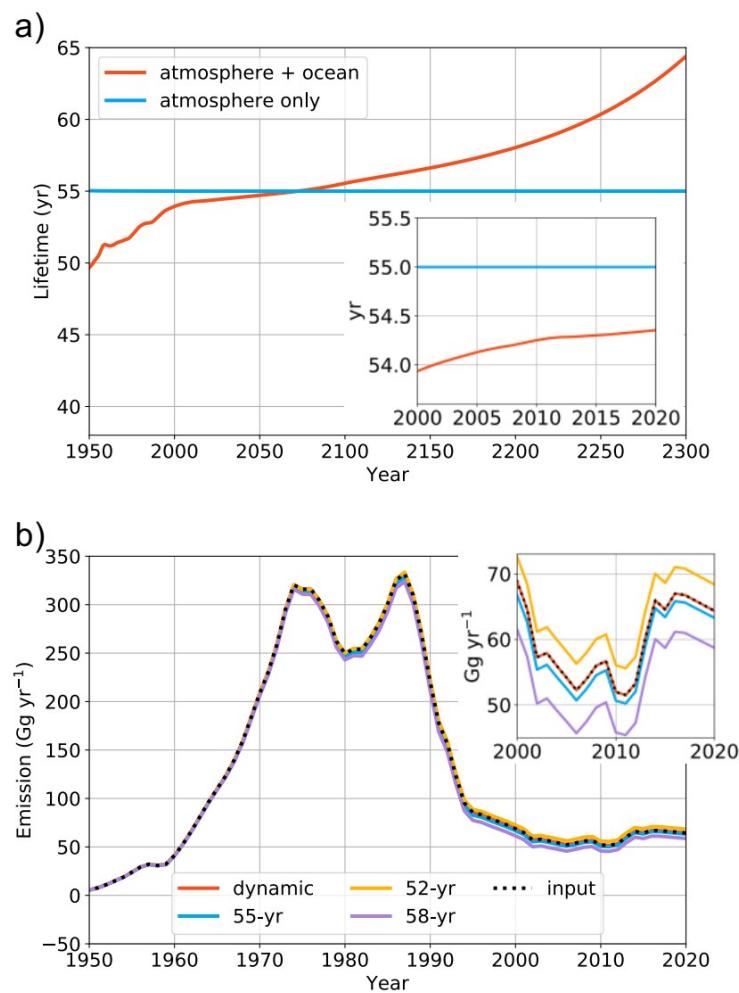


Figure 4. a) Lifetime of atmospheric CFC-11 in the model calculated by including only atmospheric losses (blue line) and including both atmosphere and ocean (red line); b) Comparison between emissions prescribed in the MITgcm runs to those inferred by treating the model calculated concentrations as data and ignoring the effect of the ocean on the lifetime. Black dashed line shows the emission input to the run. We use a 1-box model to do a top-down estimate of the emission (see Methods) given the CFC-11 concentrations that are output from the MITgcm using: a dynamic CFC-11 atmospheric-ocean lifetime (same as the red line in panel a); constant 55-year atmosphere-only lifetime (same as the blue line in panel a); and constant 52- and 58-year lifetimes to test the sensitivity of the inferred emissions. Inferred emission is shown from 1950 to 2020; emission after 2020 approaches zero linearly.

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74

659 **Table 1.** Calculated loss of CFC-11 in the atmosphere assuming a constant 55-year lifetime, loss
 660 of CFC-11 to the ocean as air-sea flux (positive values indicate a flux of CFC-11 from the
 661 atmosphere to the ocean), and relative loss in the ocean compared to that in the atmosphere
 662 from the MITgcm simulations. Values for several decades around the period of maximum loss in
 663 the atmosphere (1990s), the period of maximum flux of CFC-11 from the ocean to the
 664 atmosphere (2120s) and the outgassing late in the 23rd century are shown in the table. Error bars
 665 indicate ± 1 standard deviation associated with the decadal average. Loss rates in the 1930s and
 666 1940s are very small compared to the error bars, and are not shown.
 667

Time	Atmosphere loss (Gg yr ⁻¹)	Ocean loss (Gg yr ⁻¹)	Ocean/atmosphere loss (%)
1951 – 1960	2.3 ± 1.4	0.2 ± 0.1	8.8 ± 1.1
1961 – 1970	14.2 ± 6.5	1.0 ± 0.4	7.1 ± 0.3
1971 – 1980	50.4 ± 12.7	2.8 ± 0.4	5.7 ± 0.7
1981 – 1990	88.3 ± 10.5	3.6 ± 0.2	4.1 ± 0.3
1991 – 2000	103.1 ± 1.2	2.6 ± 0.4	2.5 ± 0.4
2001 – 2010	96.6 ± 2.3	1.6 ± 0.2	1.6 ± 0.2
2011 – 2020	90.0 ± 1.4	1.2 ± 0.1	1.3 ± 0.0
...
2101 – 2110	35.0 ± 1.8	-0.4 ± 0.0	-1.2 ± 0.1
2111 – 2120	29.3 ± 1.5	-0.5 ± 0.0	-1.6 ± 0.1
2121 – 2130	24.5 ± 1.3	-0.5 ± 0.0	-2.0 ± 0.1
2131 – 2140	20.5 ± 1.0	-0.5 ± 0.0	-2.3 ± 0.1
2141 – 2150	17.2 ± 0.9	-0.5 ± 0.0	-2.7 ± 0.1
...
2281 – 2290	1.6 ± 0.1	-0.2 ± 0.0	-12.7 ± 0.4
2291 – 2300	1.4 ± 0.1	-0.2 ± 0.0	-14.0 ± 0.4

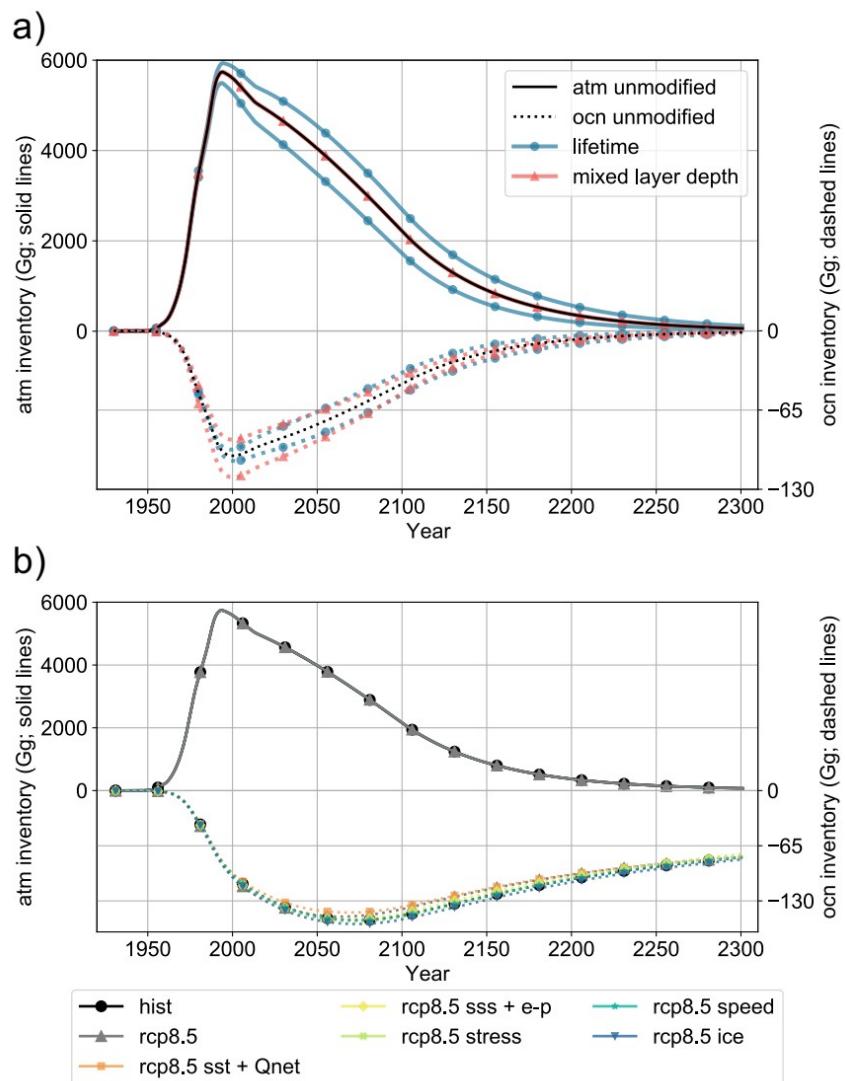
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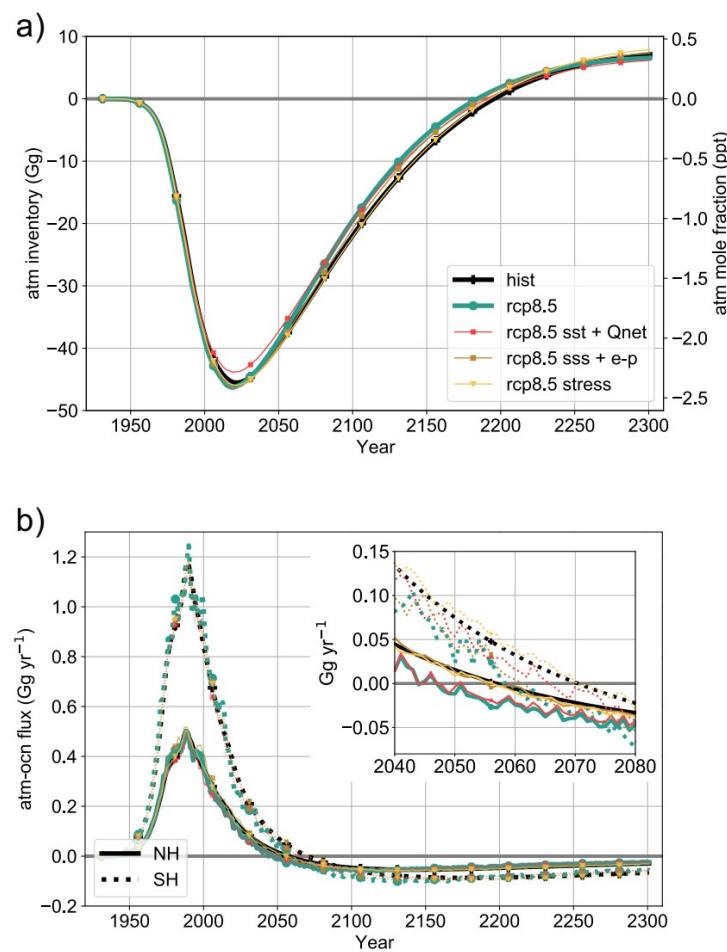
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672 **Figure S1. a)** Box model CFC-11 inventories for the atmosphere (left axis and solid lines) and the
673 ocean (right axis and dashed lines). Black lines indicate CFC-11 inventories with unmodified
674 parameters as in Table S1, and colored lines display $\pm 15\%$ changes of these associated
675 parameters (only the results from most sensitive parameters are shown in the figure). b) MITgcm
676 CFC-11 inventories under additional ocean forcing runs. In both panels, the ocean inventory is
677 shown inversed in sign to facilitate comparison.
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Figure S2. Similar to Figure 3, but for CFC-12.

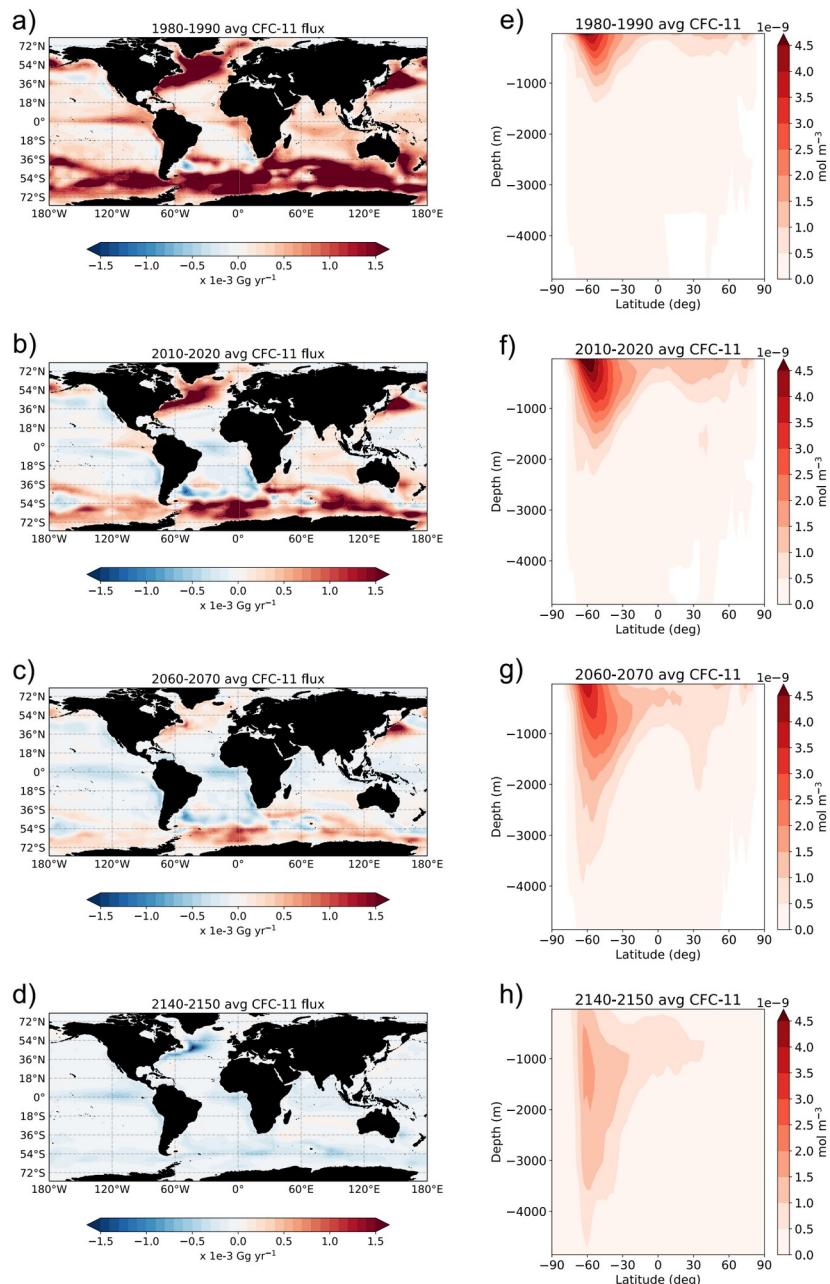
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Figure S3. a – d) Decadal averages of CFC-11 air-sea flux (area integrated) for different time periods in the MITgcm Hist run (positive indicate fluxes going into the ocean); e – h) Zonal mean

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684 CFC-11 concentration in the ocean for the same averaging periods shown on the left.

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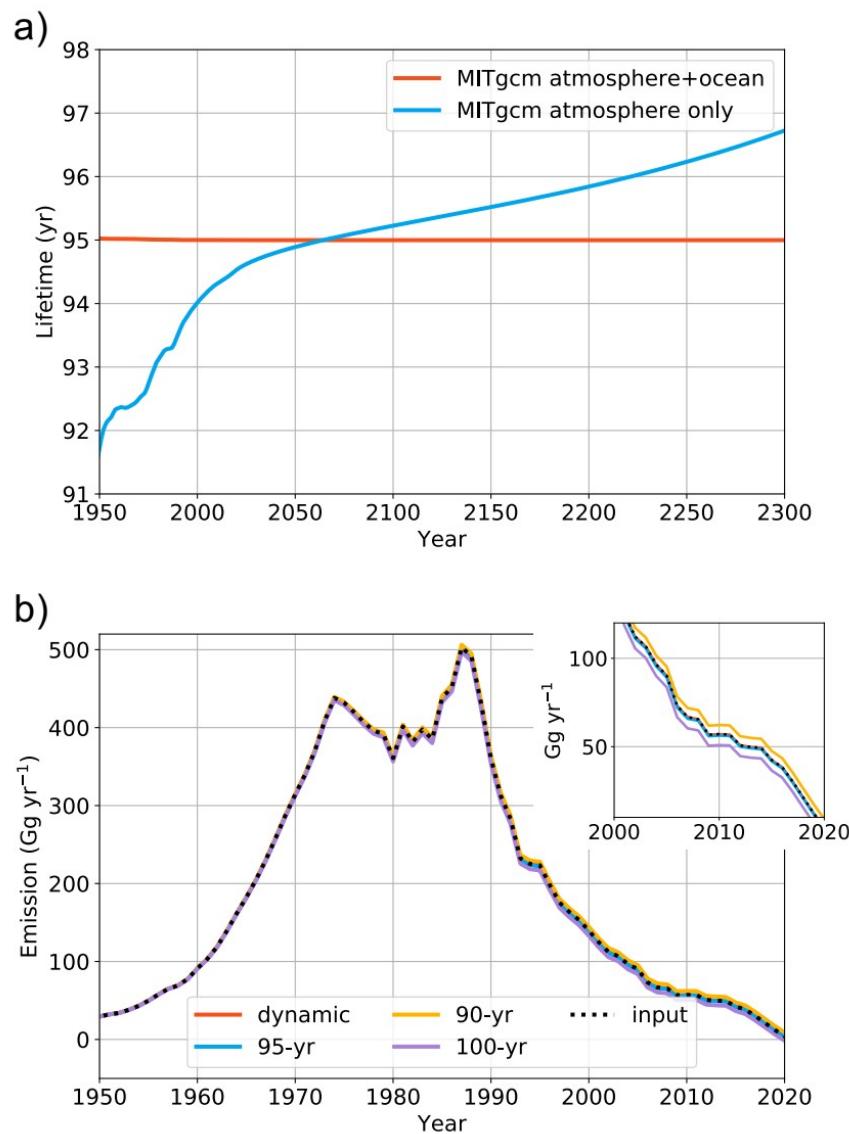
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687 **Figure S4.** Similar to Figure 4, but for CFC-12.

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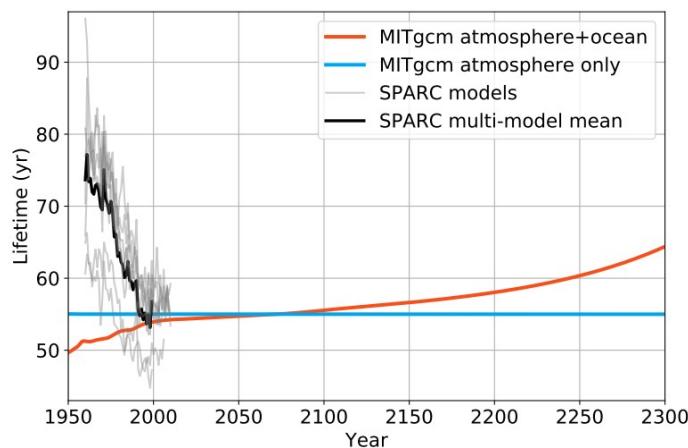
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Figure S5. Same as Figure 3, except that atmosphere-only CFC-11 lifetime calculated from SPARC chemistry-climate models are overlaid for comparison. Thin gray lines are calculated time-dependent atmosphere-only lifetimes from the SPARC models from 1960 to 2010, while the heavy black line is the SPARC multi-model mean.

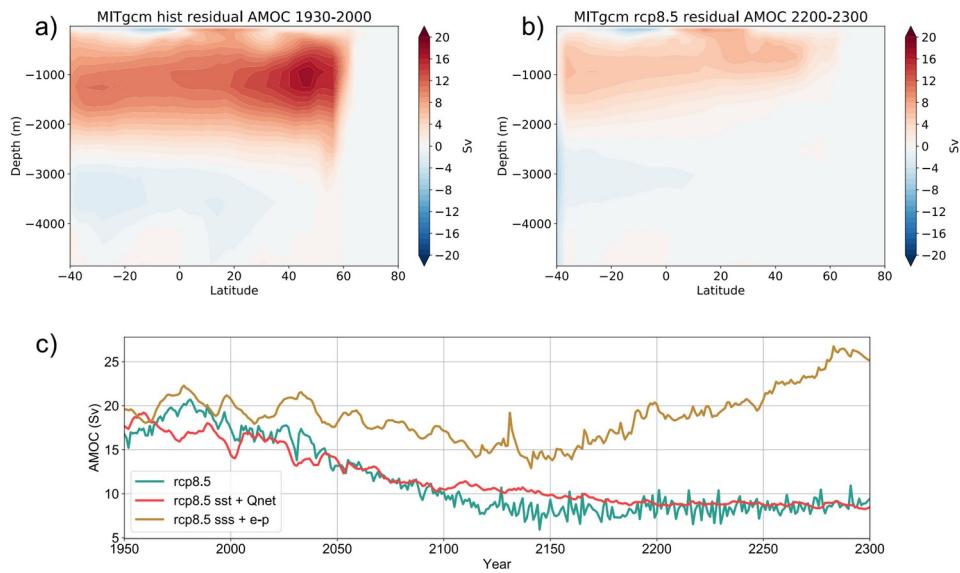
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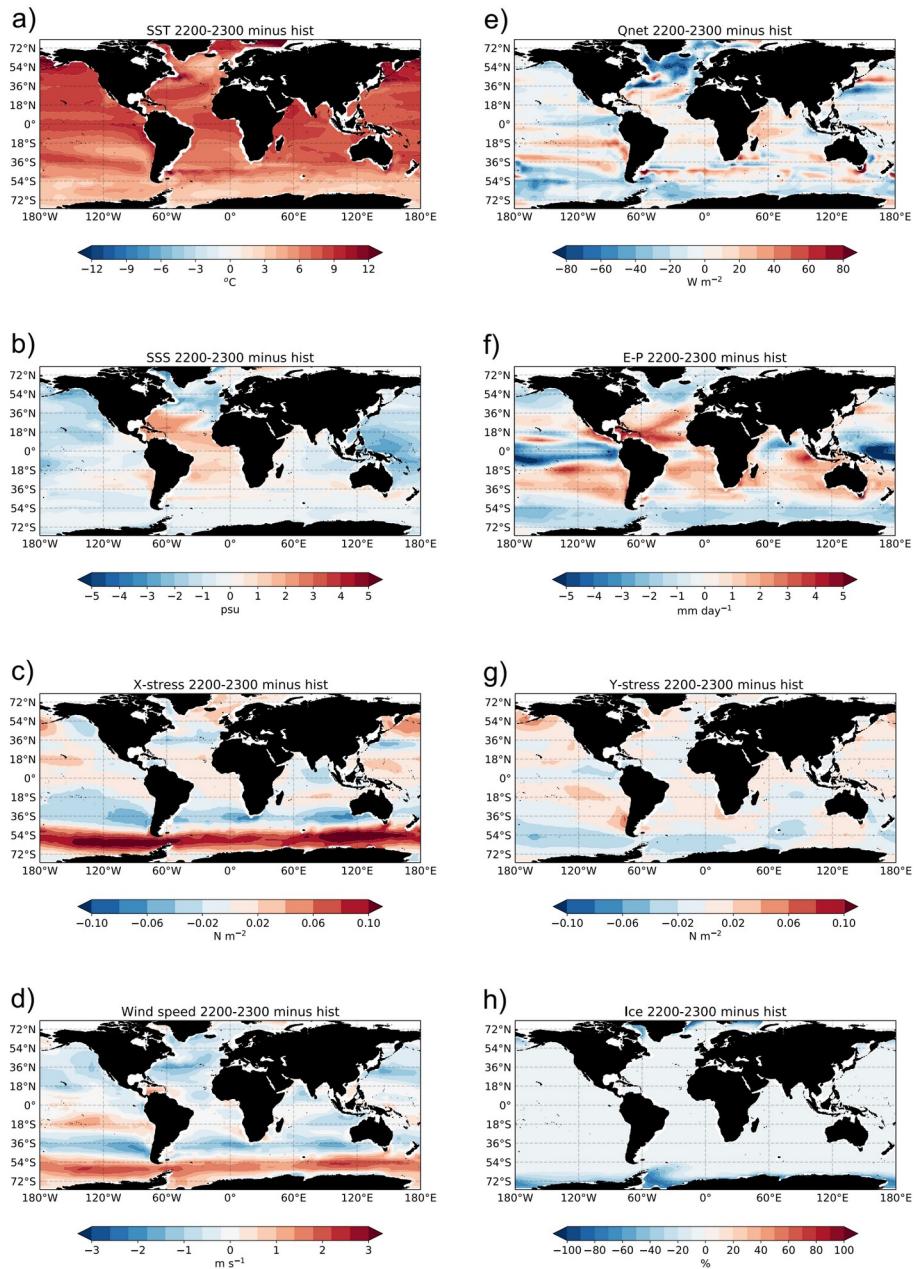
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696 **Figure S6.** a) Zonal mean residual AMOC under Hist run averaged in 1930 – 2000; b) Zonal
697 mean residual AMOC under RCP8.5 run averaged in 2200 – 2300; c) Time series of AMOC
698 strength (maximum between 20 – 50 °N) under full RCP8.5, SST + Qnet only and SSS + E-P
699 only. Under full RCP8.5, AMOC in the MITgcm has decreased by 60 % from 1930 – 2000 to 2200
700 – 2300. This percent decrease is comparable to MPI-ESM-LR, which shows a 56 % decrease
701 during the same period, except that the AMOC climatology in MPI-ESM-LR started at a higher
702 value than the MITgcm at the beginning of 1930.
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705 **Figure S7.** Anomaly maps of all the forcing fields from the MPI-ESM-LR RCP8.5 scenario. The
 706 maps indicate averages of the last 100 years (2200 – 2300) minus the base period (1850 –
 707 1930).

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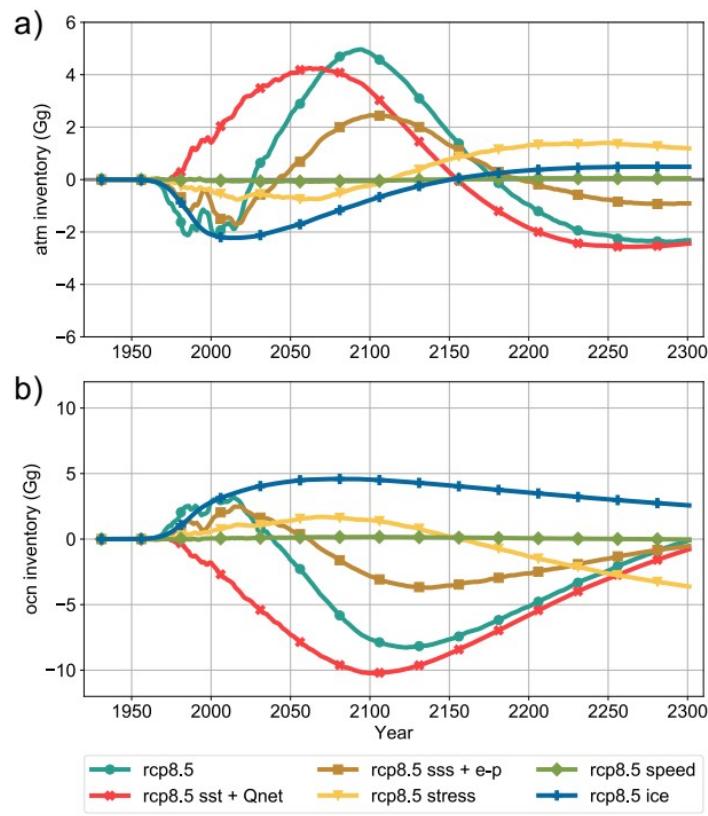
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711 **Figure S8.** CFC-11 inventory difference between each RCP8.5 forcing run and Hist forcing run
 712 for a) atmosphere inventories; and b) ocean inventories.

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715 **Table S1.** A list of parameters and variables used in the box model.

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Name	Value Used	Description
m_{nh}^a, m_{sh}^a	Calculated in the model	CFC-11 mass in the atmospheric boxes, subscript indicates the NH and SH.
m_{nh}^t, m_{sh}^t	Calculated in the model	CFC-11 mass in the mixed layer boxes, subscript indicates the NH and SH.
m_{nh}^d, m_{sh}^d	Calculated in the model	CFC-11 mass in the deep ocean boxes, subscript indicates the NH and SH.
F_{nh}, F_{sh}	Calculated in the model	CFC-11 air-sea flux.
E_{nh}, E_{sh}	Discussed in the text	CFC-11 emission in the NH and SH.
L	$1/55 \text{ yr}^{-1}$	CFC-11 atmospheric loss rate.
T_{n2s}^a, T_{s2n}^a	$1/1.3 \text{ yr}^{-1}$	Atmospheric exchange rate between the NH and SH.
T_{n2s}^t, T_{s2n}^t	$1/50 \text{ yr}^{-1}$	Mixed layer exchange rate between the NH and SH.
T_{n2s}^d, T_{s2n}^d	$1/100 \text{ yr}^{-1}$	Deep ocean exchange rate between the NH and SH.
$T_{nh}^{t2d}, T_{sh}^{t2d}$	$1/3 \text{ yr}^{-1}$	Mixed layer to deep ocean exchange rate in the NH and SH.
$T_{nh}^{d2t}, T_{sh}^{d2t}$	$1/5 \text{ yr}^{-1}$	Deep ocean to mixed layer exchange rate in the NH and SH.
k	10 cm hr^{-1}	Piston velocity.
H^a	13 km	Atmosphere height.
A^a	$5.1 \times 10^{14} \text{ m}^2$	Atmospheric total surface area.
H^t	150 m	Mixed layer depth.
A^t	$3.1 \times 10^{14} \text{ m}^2$	Ocean total surface area (assume 15% sea ice)

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