

MLtool++ package for machine learning and its applications to materials data

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Abstract

We are developing MLtool++ package of software programs for machine learning (ML). Given the MLtool Python code, we create a faster C++ code with the potential for parallelization. We have extracted materials data from the literature. One dataset contains melting temperatures of stoichiometric 1:1 metallic compounds XZ, composed by elements X={Al, Ti, V, Cr, Zr, Nb, Mo, Hf, Ta, W} and Z={Co, Ni, Cu, Rh, Pd, Ag, Ir, Pt, Au}, and another contains solid-solid symmetry-breaking phase transition temperatures. We studied dependencies of temperatures on composition, found several correlations, and parametrized them by analytical functions. MLtool++ package is generic and applicable to any tabulated numeric data.

MLtool++ software implementation

MLtool++ reads two text files, containing data and input parameters. The data is fitted by the selected analytical model; choices include polynomials, rational functions, gaussians, plane waves, or arbitrary user-defined functions. MLtool Python code cross-validates all models, returns a ranked list, and points at the best predictive models; MLtool++ package will be faster, parallel, with a similar functionality. MLtool++ package is object-oriented and multithreaded; there are plans for its further parallelization and optimization.

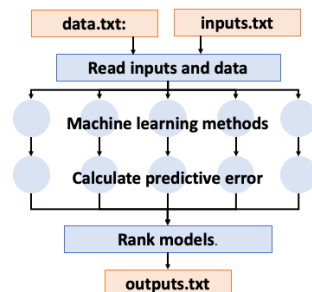


Figure 1: Flowchart of MLtool++ package.

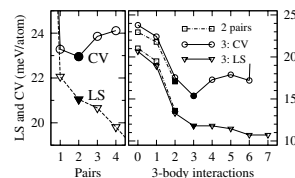


Figure 2: Cross-validation (CV) predictive and least-squares (LS) fit errors versus number of parameters, see Fig. 2 in [10].

Applications to materials data

Data sets were extracted from ASM phase diagrams database. Over 90 binary X-Z phase diagrams were considered with chemical elements X={Al, Ti, V, Cr, Zr, Nb, Mo, Hf, Ta, W} and Z={Co, Ni, Cu, Rh, Pd, Ag, Ir, Pt, Au}.

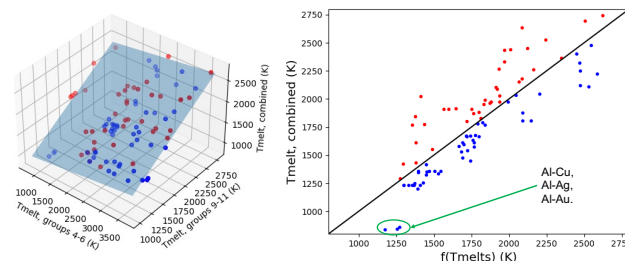


Figure 3: Melting temperatures T_{melt} [K] of 90 XZ stoichiometric 1:1 compounds versus T_{melt} of 10 X (group 4-6) and 9 Z (9-11) elemental metals, fitted by the linear function $T_{\text{melt}}(XZ) = \alpha \cdot \bar{T}_{\text{melt}}$, where $\alpha = [106 \text{ [K]} \ 0.0992 \ 0.791]$ and $\bar{T}_{\text{melt}} = [1 \ T_{\text{melt}}(X) \ T_{\text{melt}}(Z)]$.

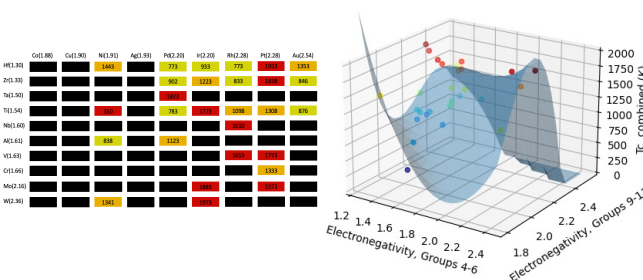


Figure 4: Solid-solid transition temperatures T_c [K] of 31 stoichiometric 1:1 compounds versus electronegativity of 10 X (group 4-6) and 9 Z (group 9-11) metals.

Considering solid-solid symmetry-breaking phase transition temperatures T_c [K] of stoichiometric 1:1 compounds XZ versus properties of elemental metals X and Z, we found a moderate correlation and approximated it by a third-degree polynomial:

$$T_c [\text{K}] = \alpha \cdot \bar{\epsilon n};$$
$$\alpha = [57.5 \ -23.4 \ -63.4 \ 2.77 \ 17.5 \ 23.2 \ -0.0398 \ -1.17 \ -3.13 \ -2.84] \cdot 10^4 \text{ K},$$
$$\bar{\epsilon n} = [1 \ \epsilon n_X \ \epsilon n_Z \ \epsilon n_X^2 \ \epsilon n_X \epsilon n_Z \ \epsilon n_Z^2 \ \epsilon n_X^3 \ \epsilon n_X^2 \epsilon n_Z \ \epsilon n_X \epsilon n_Z^2 \ \epsilon n_Z^3]$$

Conclusion

We started implementing components of MLtool++ software package, considered 90 ASM phase diagrams, performed data mining, and extracted relevant data sets. We applied our software to our data, found correlations and parametrized them by analytical equations.

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References

- [1] N.A. Zarkevich. **Mod. Phys. Lett. B** 35:2130003 (2021).
- [2] N.A. Zarkevich. **Complexity** 11:36 (2006).
- [3] N.A. Zarkevich, T.M. Smith, E.N. Baum, J.W. Lawson. **Crystals** 12:1049 (2022).
- [4] T.M. Smith, N.A. Zarkevich, A.J. Egan, et al., **Commun. Mater.** 2:106 (2021).
- [5] ASM binary phase diagrams (assessed 8 August 2022): <https://matdata.astminternational.org>
- [6] Bethany Wu, Shreyas Honrao, N.A. Zarkevich. "MLtool: A universal machine learning tool for fitting tabulated numerical data" (Poster). USRA virtual intern symposium (7 May 2021).
- [7] R.R. Curtin, M. Edel, M. Lozhnikov, Y. Mentekidis, S. Ghaisas, S. Zhang., **J. Open Source Software** 3:26 (2018).
- [8] D. Kingma, J. Ba. **International Conference on Learning Representations**, 2018.
- [9] G.A.F. Seber, C.J. Wild. "Nonlinear Regression". (New York: Wiley, 1989). ISBN: 0471617601.
- [10] N.A. Zarkevich, D.D. Johnson. **Phys. Rev. Lett.** 92 (25): 255702 (2004).

Provided a versatile fitting tool for any tabulated numeric data

References (Data Mining)

- Allibert C.H., Bernard C., Valignat N., and Dombre M., Co-Cr binary system: Experimental re-determination of the phase diagram and comparison with the diagram calculated from the thermodynamic data, **J. Less-Common Met.**, Vol. 59, 1978, p 211-228
- Balun J., and Inden G., Phase equilibria in the binary Rh-Ti system, **Intermetallics**, Vol. 14, 2006, p 260-271
- Baren, M.R. The Ag-Nb (Silver-Niobium) system. **Bulletin of Alloy Phase Diagrams** **10**, 640 (1989).
- Baren, M.R. The Ag-Ta (Silver-Tantalum) system. **Bulletin of Alloy Phase Diagrams** **9**, 244–245 (1988).
- Baren, M.R., The Ag-Hf (Silver-Hafnium) system. **Bulletin of Alloy Phase Diagrams** **10**, 110 (1989).
- Biggs T., Cornish L.A., Witcomb M.J., and Cortie M.B., Revised phase diagram for the Pt-Ti system from 30 to 60 at.% platinum, **J. Alloys Compd.**, Vol. 375, 2004, p 120-127
- C.W. Bale, E. Bélisle, P. Chartrand, S.A. Decterov, G. Eriksson, A.E. Gheribi, K. Hack, I.-H. Jung, Y.-B. Kang, J. Melançon, A.D. Pelton, S. Petersen, C. Robelin, J. Sangster, P. Spencer, M-A. Van Ende, FactSage thermochemical software and databases, 2010–2016, **Calphad**, Volume 54, 2016, Pages 35-53.
- Cahn, R.W. (1991), Binary Alloy Phase Diagrams—Second edition. T. B. Massalski, Editor-in-Chief; H. Okamoto, P. R. Subramanian, L. Kacprzak, Editors. **ASM International**, Materials Park, Ohio, USA. December 1990.
- Davydov A., and Kattner U.R., Thermodynamic Assessment of the Co-Mo System, **J. Phase Equilib.**, Vol. 20, 1999, p 5-16
- English J.J., Binary and ternary phase diagrams of columbium, molybdenum, tantalum, and tungsten, **Battelle Meml. Inst., Def. Met. Inf. Cent.**, DMIC Rep., Vol. 152, 1963, p 1-127
- Eremenko V.N., Kriklya L.S., Khoruzhaya V.G., and Shtepa T.D., Interaction of hafnium with ruthenium and iridium, **Sov. Powder Metall. Met. Ceram.**, Vol. 30, 1991, p 765-770
- Eremenko V.N., Shtepa T.D., Velikanova T.Y., Kriklya L.S., and Petyukh V.M., The Hf-Rh phase diagram, **Izvestiya Akademii Nauk SSSR; Russ. Metall.**, no.6, 1991, p 132-138
- Ghosh G., and Olson G.B., Thermodynamic Modeling of the Cr-Pd and Mo-Pd Systems, **J. Phase Equilib.**, Vol. 21, 2000, p 32-39
- Kaufman L., Coupled thermochemical and phase diagram data for tantalum based binary alloys, **CALPHAD: Comput. Coupling Phase Diagrams Thermochem.**, Vol. 15, 1991, p 243-259
- Khoruzhaya, V.G., Kornienko, K.E., Martsenyuk, P.S. et al. Phase equilibria in the system Al-Rh. **Powder Metall Met Ceram** 45, 251–258 (2006).
- Kleykamp H., Constitution and thermodynamics of the Mo-Ru, Mo-Pd, Ru-Pd and Mo-Ru-Pd systems, **J. Nucl. Mater.**, Vol. 167, 1989, p 49-63
- Kneller E.F., Khan Y., and Gorres U., The Alloy System Copper-Zirconium. Part I. Phase Diagram and Structural Relations, **Z. Metallkd.**, Vol. 77, 1986, p 43-48
- Leonov M.P., Bochvar N.R., and Ivanchenko V.G., Chromium-copper phase diagram, **Dokl. Phys. Chem.**, Vol. 290, 1986, p 886-887
- Markiv V.Y., Pet'kov V.V., Storozhenko A.I., Ivanchenko V.G., and Gorskii V.V., The equilibrium diagram of the Hf-Cu system, **Izv. Akad. Nauk SSSR** **2**, 209-213 (1974); **Russ. Metall.**, Vol. 2, 1974, p 123-125
- Meshkov L.L., Guzei L.S., and Sokolovskaya E.M., Thermodynamics of Nickel-Molybdenum Alloys, **Russ. J. Phys. Chem.**, Vol. 49, 1975, p 1128-1129
- Ocken H., and Van Vucht J.H.N., Phase equilibria and superconductivity in the molybdenum platinum system, **J. Less-Common Met.**, Vol. 15, 1968, p 193-199
- Okamoto H., Ag-Zr (Silver-Zirconium), **J. Phase Equilib.**, Vol. 18, 1997, p 312
- Okamoto H., Al-Au (Aluminum-Gold), **J. Phase Equilib.**, Vol. 12, 1991, p 114-115
- Okamoto H., Al-Ir (Aluminum-Iridium), **J. Phase Equilib.**, Vol. 21, 2000, p 409
- Okamoto H., Al-Ni (Aluminum-Nickel), **J. Phase Equilib.**, Vol. 14, 1993, p 257-259
- Okamoto H., Al-Pd (Aluminum-Palladium), **J. Phase Equilib.**, Vol. 24, 2003, p 196
- Okamoto H., Au-Hf (Gold-Hafnium), **J. Phase Equilib.**, Vol. 21, 2000, p 410
- Okamoto H., Au-Zr (Gold-Zirconium), **J. Phase Equilib.**, Vol. 23, 2002, p 111
- Okamoto H., Co-Nb (Cobalt-Niobium), **J. Phase Equilib.**, Vol. 21, 2000, p 495
- Okamoto H., Cr-Ni (Chromium-Nickel), **J. Phase Equilib.**, Vol. 18, 1997, p 221
- Okamoto H., Cu-Ti (Copper-Titanium), **J. Phase Equilib.**, Vol. 23, 2002, p 549-550
- Okamoto H., Ir-Ti (Iridium-Titanium), **J. Phase Equilib.**, Vol. 13, 1992, p 329-331
- Okamoto H., Nb-Ni (Niobium-Nickel), **J. Phase Equilib.**, Vol. 19, 1998, p 289
- Okamoto H., Ni-Ta (Nickel-Tantalum), **J. Phase Equilib.**, Vol. 21, 2000, p 497
- Okamoto H., Pd-Ta (Palladium-Tantalum), **J. Phase Equilib. Diffus.**, Vol. 28, 2007, p 233
- Okamoto H., Pd-Zr (Palladium-Zirconium), **J. Phase Equilib.**, Vol. 23, 2002, p 290
- Ran H., and Du Z., Thermodynamic assessment of the Ir-Zr system, **J. Alloys Compd.**, Vol. 413, 2006, p 101-105
- Stalick J.K., and Waterstrat R.M., The zirconium-platinum phase diagram, **J. Alloys Compd.**, Vol. 430, 2007, p 123-131
- Stalick J.K., Waterstrat R.M. The Hafnium-Platinum Phase Diagram. **J. Phase Equilib. Diffus.** 35, 15–23 (2014).
- Zhu J.H., and Liu C.T., Defect Structures in ZrCo₂ Laves Phase, **Acta Mater.**, Vol. 48, 2000, p 2339-2347.