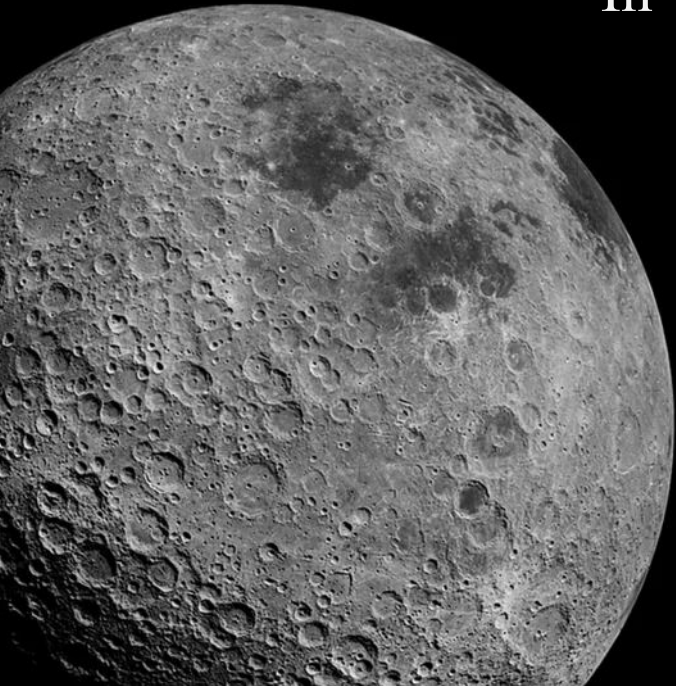


NASA SIBatt-3D

In-Space and On-Surface 3D Printing
Sodium-Ion Batteries from ISRU Materials

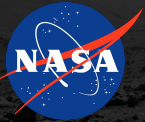
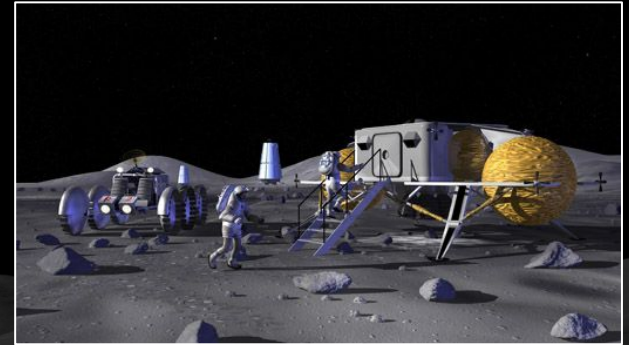


Presented by Cameroun Sherrard



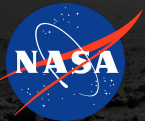
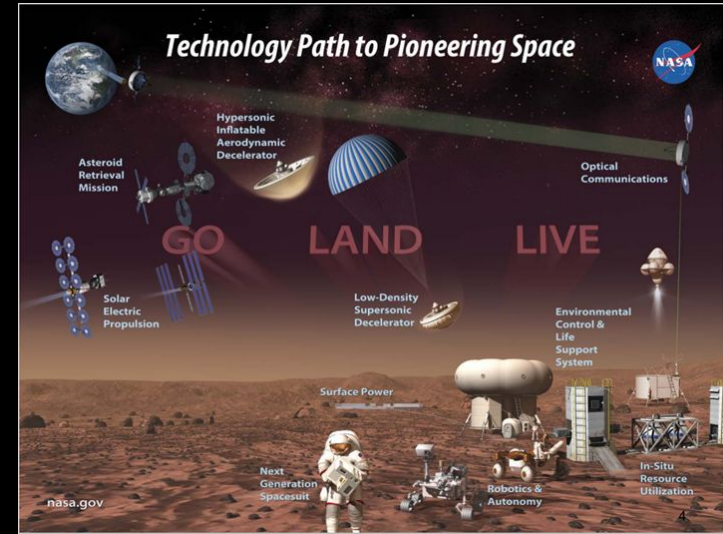
The Inspiration

- ★ Resupply/Replenish Power Storage Devices on Extended Space Mission
 - Build new batteries as needed to suit different requirements to reduce the variety brought from earth
 - Using versatile multi-use feedstocks to bring less material and save mass on extended missions
- ★ Provide a sustainable Power Sources for Surface Missions on the Moon and Mars
 - Use materials found in martian and lunar regolith to provide a sustainable power source and reduce required resupply payloads
 - Develop scalable techniques that can eventually be used power large scale systems such as habitats



NASA Mission Alignment

- ★ Space Technology Mission Directorate
 - Technology research and development for in-space and lunar/martian surface applications
- ★ In-Space Manufacturing
 - Development of manufacturing and recycling systems and processes - “Make it Don’t Take It”
- ★ On-Demand Manufacturing of Electronics
 - Additive manufacturing of metals, electronics, and sensors in microgravity



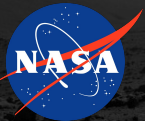
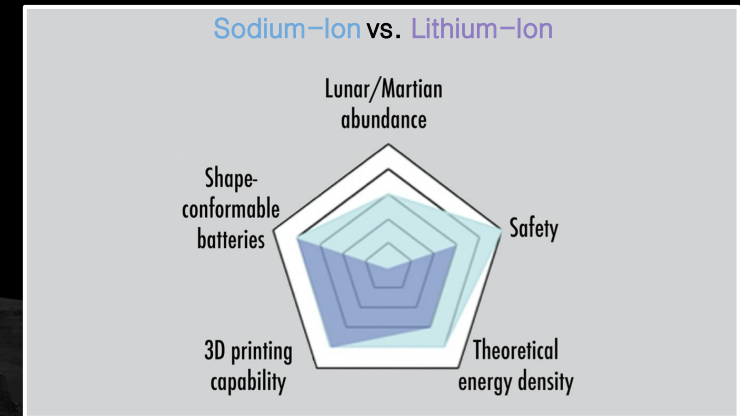
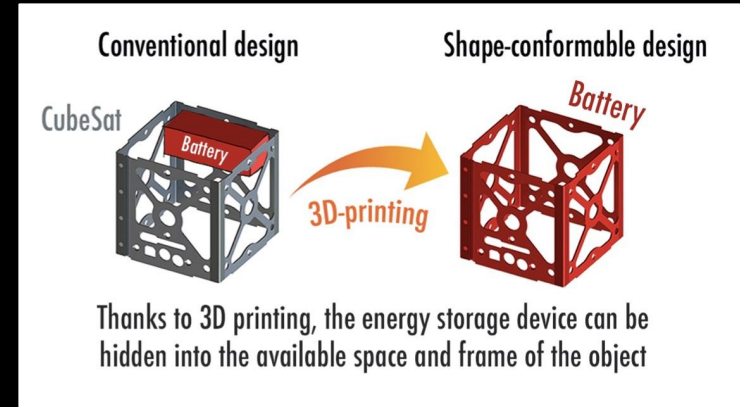
The SIBatt-3D Solution

★ 3D printing

- Versatile manufacturing technique
 - On-demand prints
 - Reduced waste
- Shape-conformable printing
 - Reduce dead volume
 - Potentially act as structures

★ Sodium-Ion Chemistry

- Relative abundance of active materials on the moon and mars
- Improved safety and sustainability over other chemistries



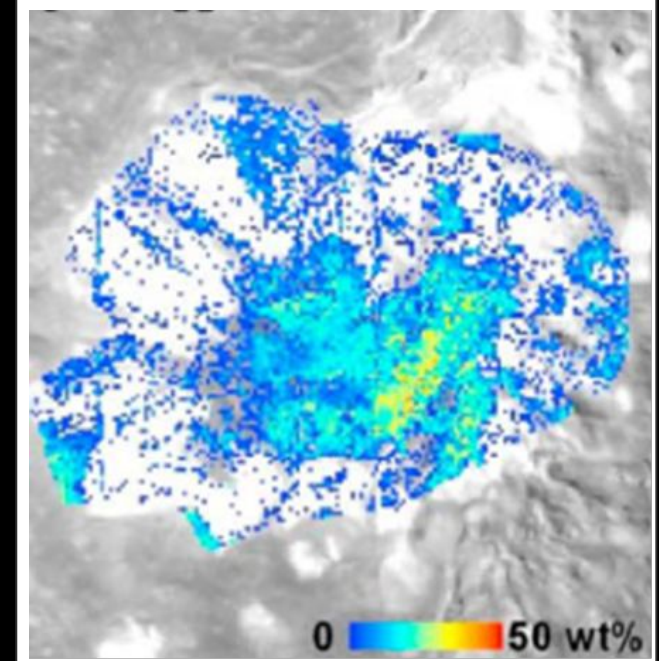
The SIBatt-3D Solution

★ ISRU

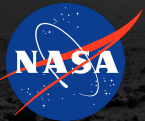
- Use materials found on the moon and mars for sustainable power on extended missions
- Reduce cost reducing shipments to replenish material

★ Supporting Systems

- Print supporting electrical and ceramic components using the same system for easy integration of batteries into other systems



Mineral mapping result for Jackson Crater for clinopyroxene taken from Lemelin [1]



Terrestrial Applications

★ Sustainability

- Better for the environment than Lithium and Cobalt mining
- Additive manufacturing processes are generally less wasteful

★ Safety

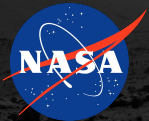
- Sodium-ion chemistries are less prone to overheating and catching fire
- Solid electrolytes

★ Shape-Conformability

- Allow for device miniaturization
- Potentially act as structures

★ On-Demand Manufacturing

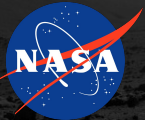
- Create batteries for different applications as needed



Material Identification

Possible ISRU Materials for SIBs						
Negative Electrode Materials						
$\text{Na}_3\text{V}_2(\text{PO}_4)_2\text{F}_2$	$\text{Na}_3\text{V}_2(\text{PO}_4)_3$	V_2O_5	NaFePO_4	$\text{NaFe}(\text{SO}_4)_2$	NaMnO_2	NaFeO_2
						$\text{Na}_{0.44}\text{MnO}_2$
Positive Electrode Materials						
FeTiO_3	Hard carbon	TiO_2	TiS_2	FeS_2	$\text{Na}_2\text{Ti}_3\text{O}_7$	
Separator for gel electrolyte or Solid electrolyte						
Filler for separator and/or solid polymer electrolyte		Salt for solid or gel polymer electrolyte		Polymer matrix for solid polymer electrolyte		Ceramic electrolyte
Nano SiO_2	TiO_2	NaPF_6	NaClO_4	Polyethylene oxide		$\text{Na}_3\text{Zr}_2\text{Si}_2\text{PO}_{12}$ (NASICON)
						Na_3PS_4

Regolith	Potential Regolith	Recycling Only	Cold Volatile
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Material Properties and Extraction

	Positive electrode candidates		Negative electrode candidates	
Material	NaFeO_2	$\text{Na}_{0.44}\text{MnO}_2$	Hard carbon	TiO_2
Voltage (vs. Na/Na^+)	3.1 V	2.5-3.5 V	0.2 V	1.0 V
Theoretical capacity (mAh/g)	241.8	121	300	335
Practical?	Yes	Yes	Yes	Yes

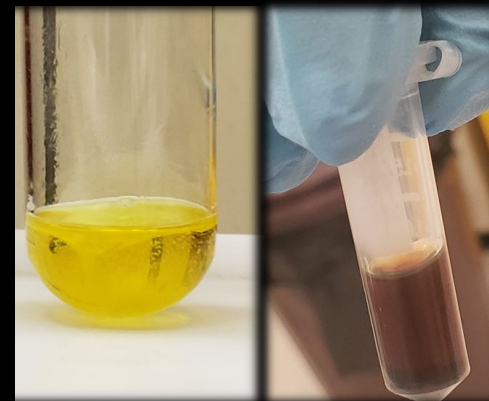


Image of ionic liquid trihexyltetradecylphosphonium pentachloride ($\text{P}_{66614}\text{Cl}_5$) before (left) and after (right) Ilmenite digestion



Print Technologies

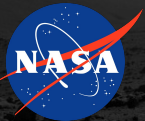
Material Extrusion - Direct Ink Write

A form of material extrusion printing that uses thick film inks as feedstocks

High TRL - Well understood technology that has been tested in microgravity

Integrated part of the Advanced Toolplate multi-material 3D printer built for NASA ISM

Difficult to achieve high resolution and high loading



Print Technologies

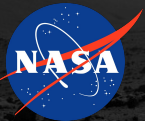
Vat Photopolymerization - Digital Light Processing

Printing using a photopolymer that hardens when exposed to light

High TRL - Well understood technology that has been tested in microgravity

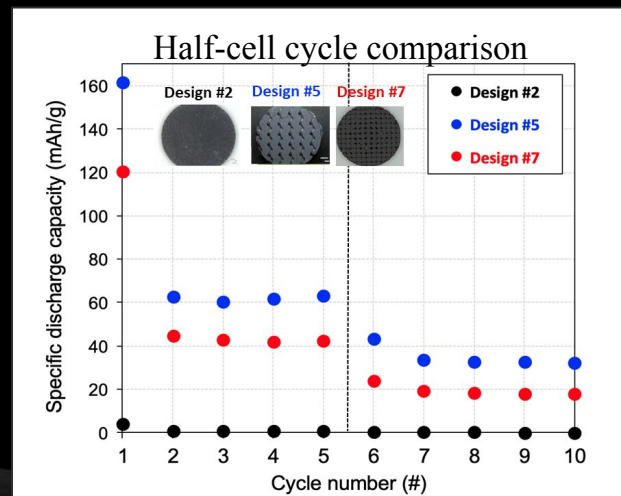
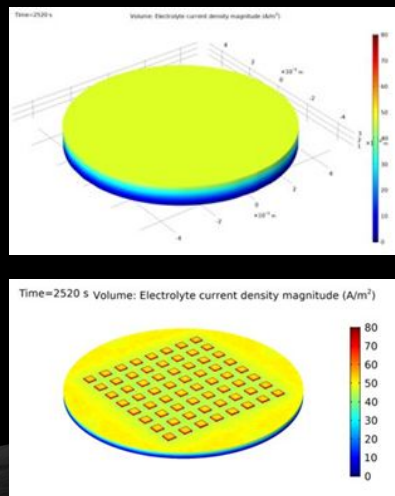
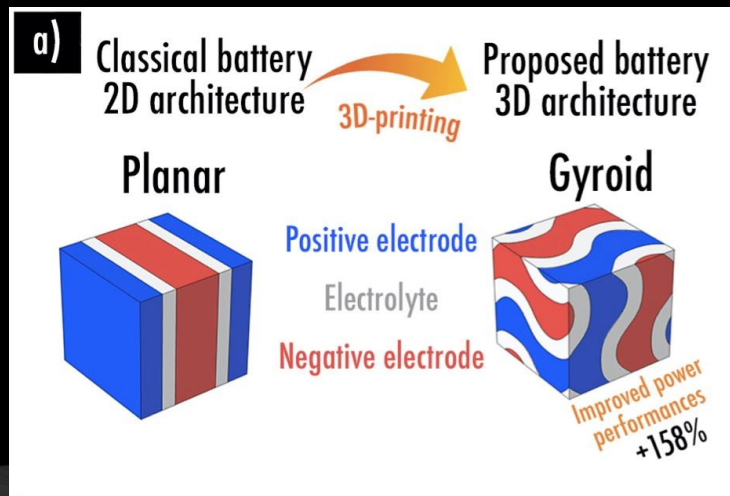
High Resolution - enhances complex geometries

Complex and energy intense post processing steps and issues with loaded materials mean the a special printer is required



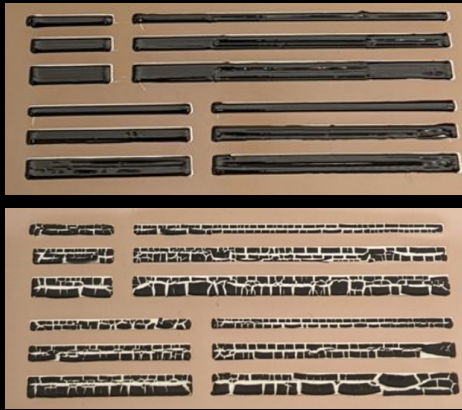
Electrode Design

Complex 3d electrode designs have been shown to improve performance and can help compensate for the losses generally associated with 3d printed electronics

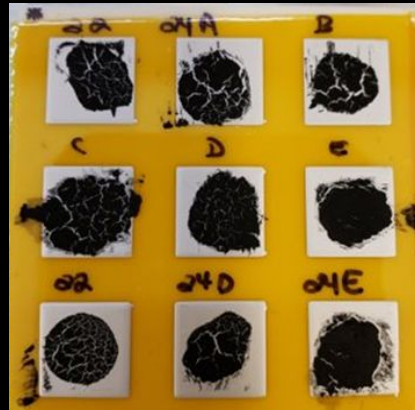


Direct Ink Write Prints

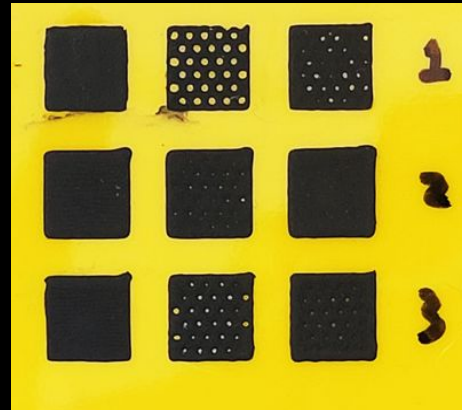
Custom inks were developed for each battery component. Inks are tested for an acceptable level of printability and adhesion while maintaining a high percentage of active material



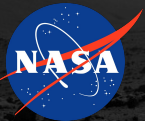
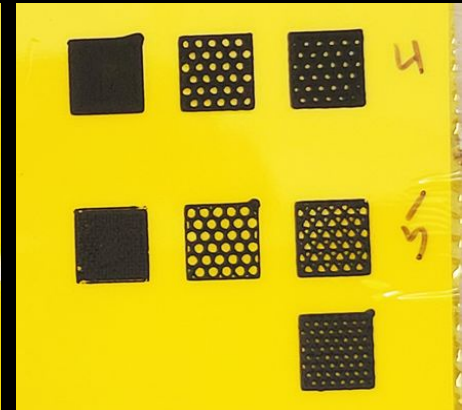
Hard Carbon ink tests on printed silver



Tests of different solvents in hard carbon inks to reduce cracking

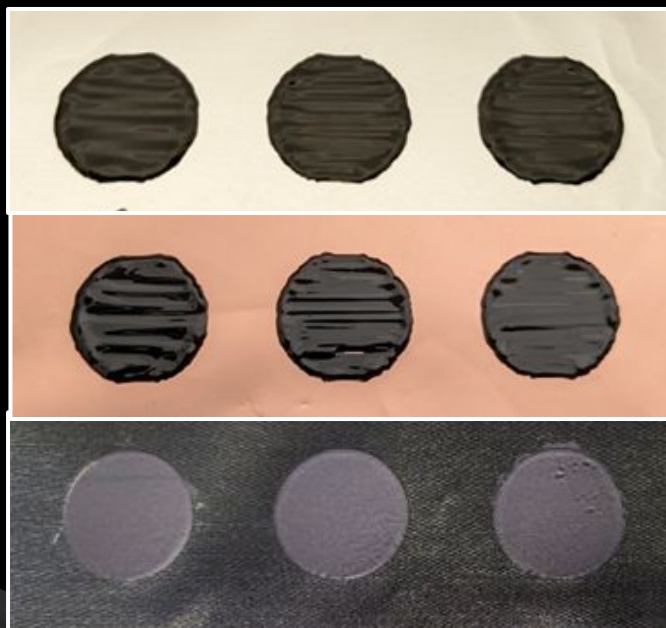


Testing Na_{0.44}MnO₂ ink formulations to in 1cm squares with different infill percentages to improve printability, formulation 5 was selected



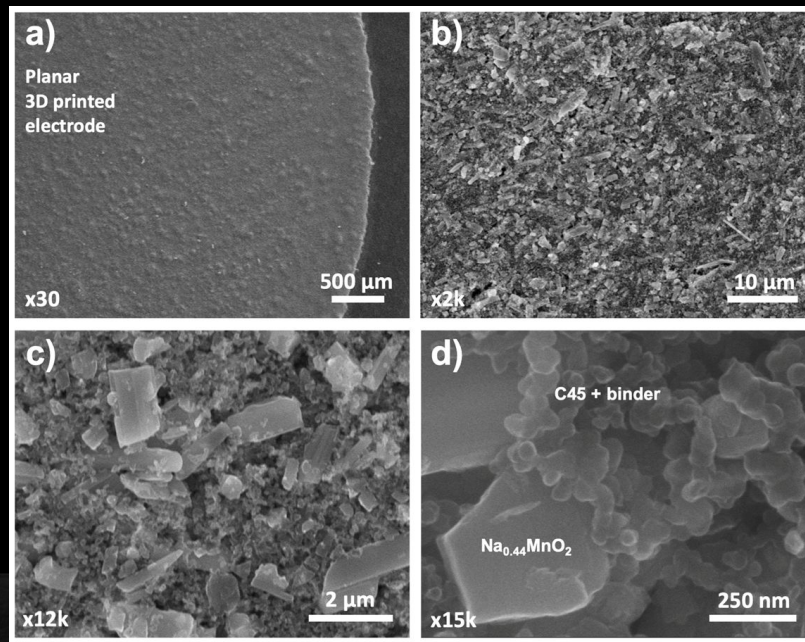
Direct Ink Write Prints

DIW Printed Electrodes for Coin Cell Tests



Printed $\text{Na}_{0.44}\text{MnO}_2$ electrodes on different foil substrates (Aluminium, Copper, Carbon Coated Al)

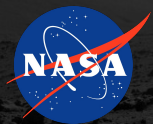
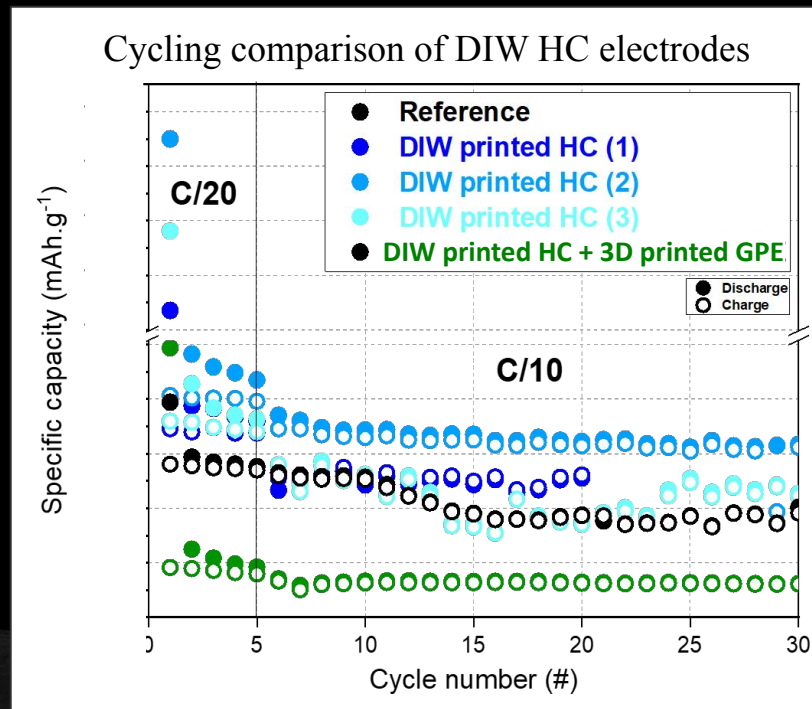
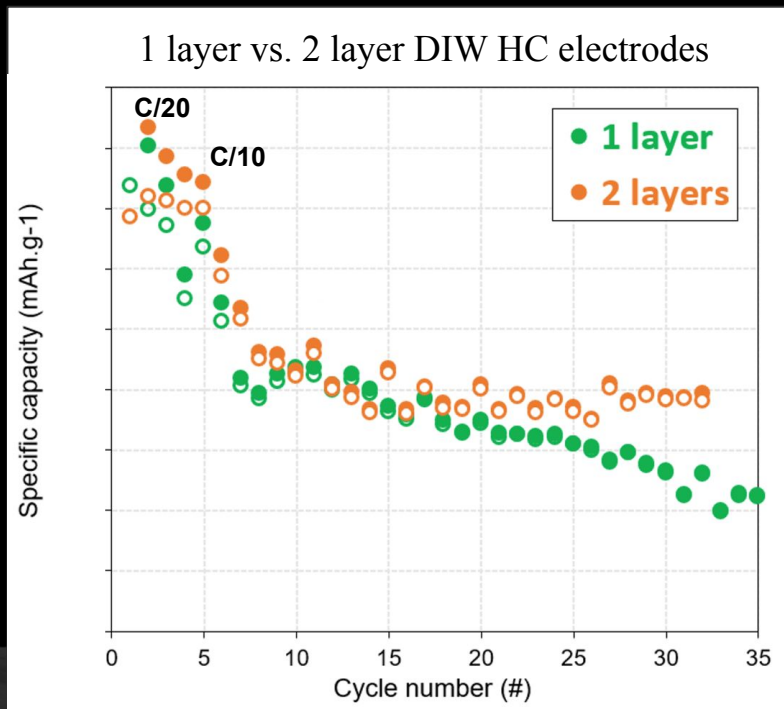
SEM observation of a 3D printed $\text{Na}_{0.44}\text{MnO}_2$ electrode



$\text{Na}_{0.44}\text{MnO}_2$ electrode printed at MSFC, very homogeneous, good dispersion of components considerable microporosity



Direct Ink Write Electrode Tests

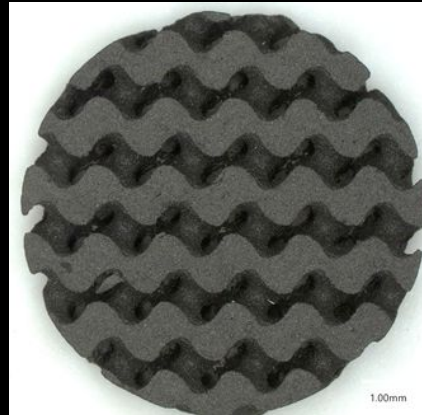


Vat Photopolymerization Prints

Custom resins were developed for each battery component. Different formulations of base resin, active material and conductive additives were tested along with a variety of print parameters and sintering and debinding processes



Solid TiO₂ Electrode (Design 1)



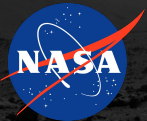
Gyroid TiO₂ Electrode (Design 5)



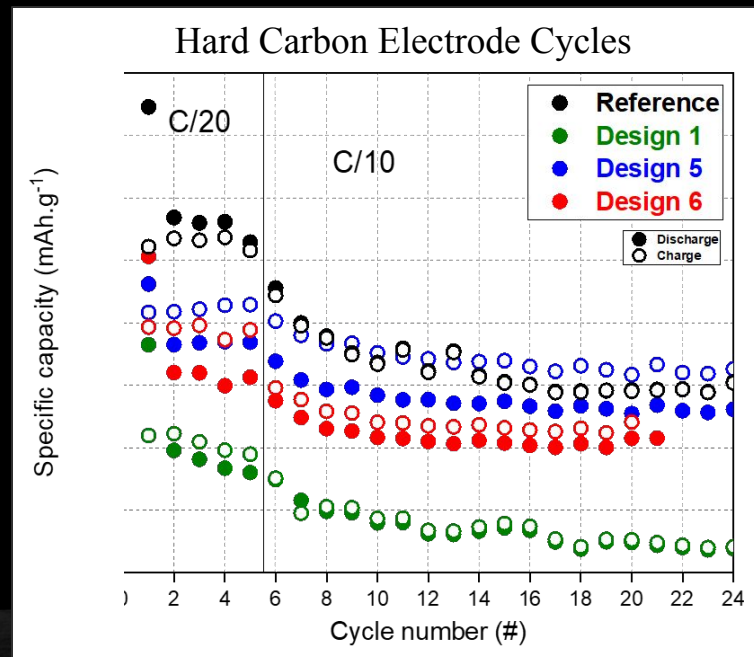
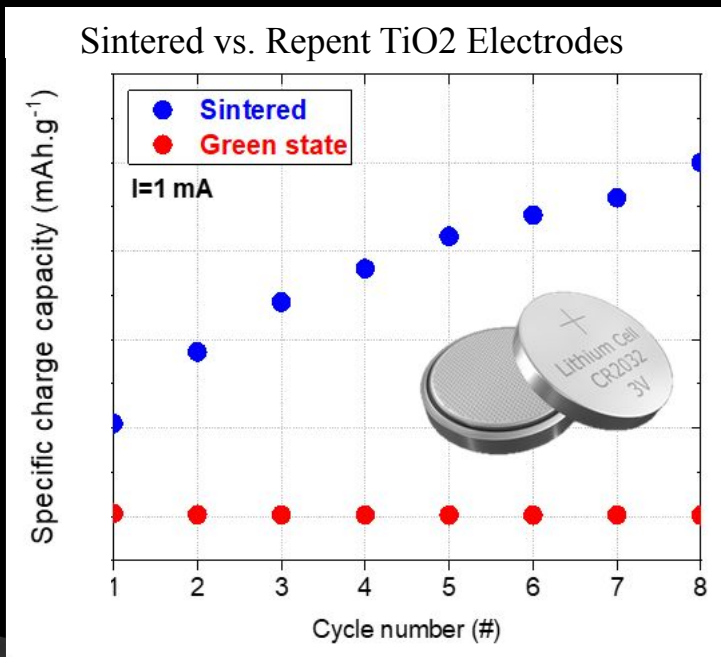
Grid TiO₂ Electrode (Design 6)



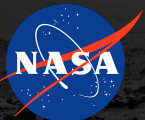
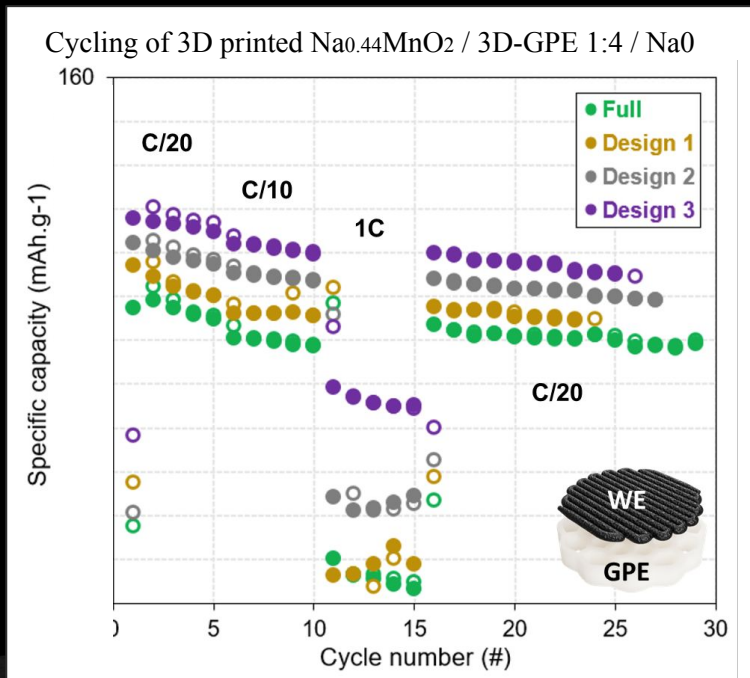
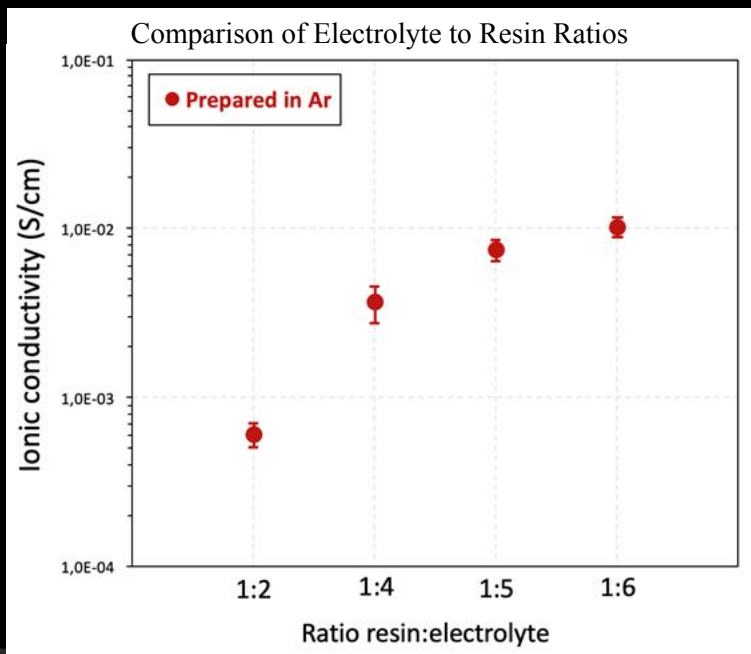
Grid TiO₂ Electrode (Design 7)



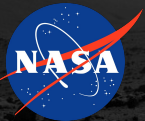
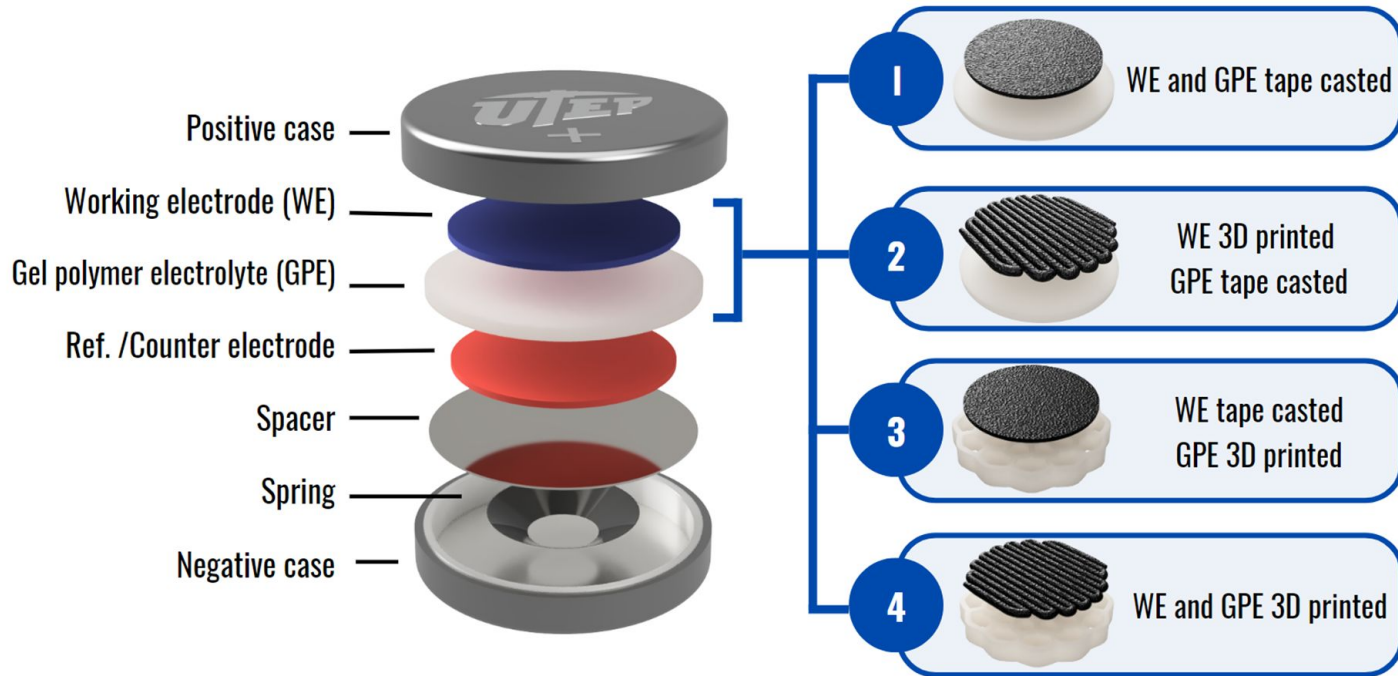
Vat Photopolymerization Electrode Tests



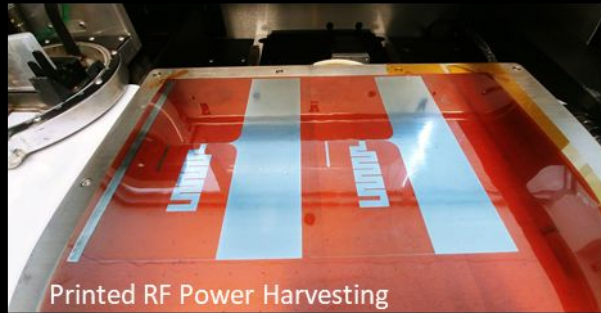
Printed Electrolytes



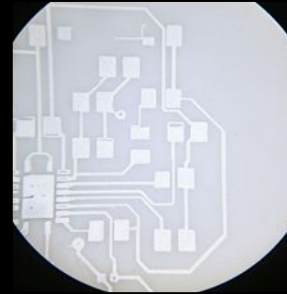
Complete Printed Cells



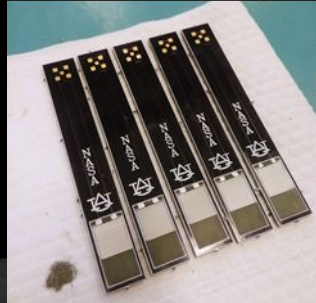
Printed Supporting Elements (ODME)



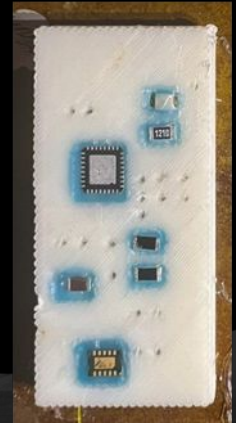
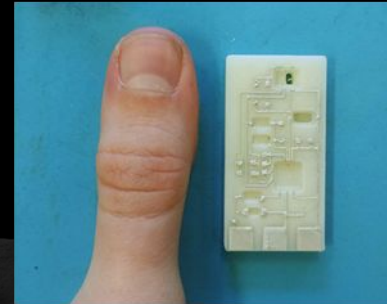
Printed RF Power Harvesting



Printed BLE
Arduino Board

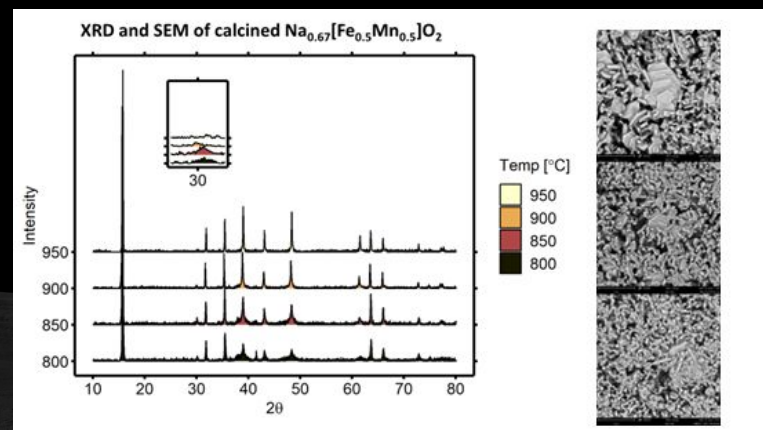
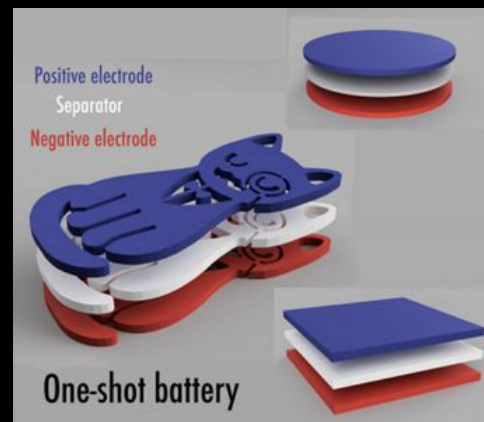


Printed multilayer CO2 Sensor



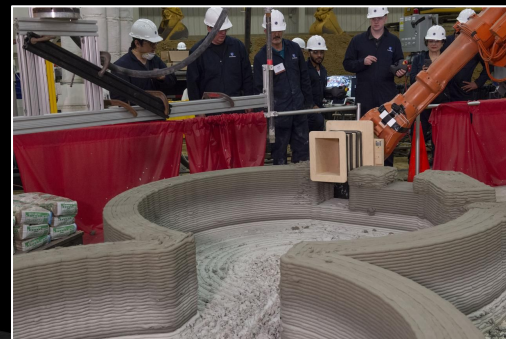
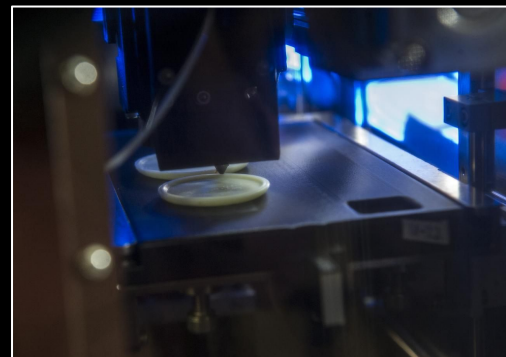
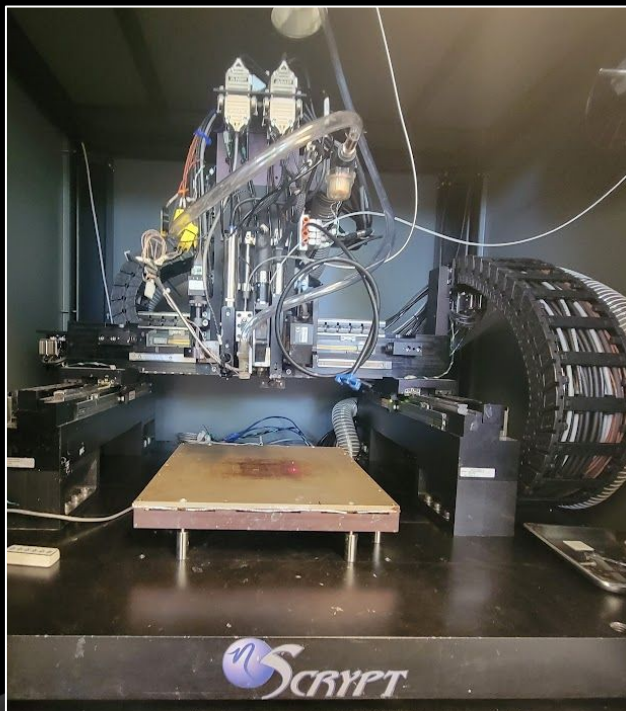
Next Steps

- ★ Test full battery cells
- ★ Demonstrate a shape conformable battery
- ★ Single shot complete battery prints
- ★ Solid electrolyte development
- ★ Improve ISRU and material recycling



Infusion Paths

- ★ Further Raise the TRL by printing in a relevant environment
- ★ Integrate with Advanced Toolplate multi-material printer
- ★ Embed a printed battery in a printed structure.



Publications

IEEE Access – Towards High Resolution 3D Printing of Shape-Conformable Batteries via Vat Photopolymerization: Review and Perspective

Perspective: What would battery manufacturing on the Moon and Mars look like?

Upcoming:

Creation of a carbon coated TiO_2 with dual functionality facilitates 3D printing and improved electrochemical performances of the electrode

Gel Polymer Electrolyte Article



SIBatt-3D Team



Cameroun Sherrard

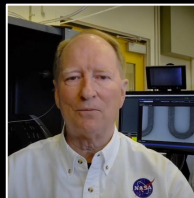
NASA Marshall Space Flight Center



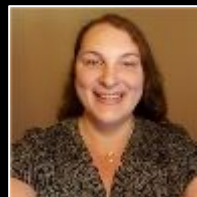
Christopher Henry



Dr. Jennifer Edmondson



Curtis Hill



Dr. Jennifer Jones



Eric MacDonald



Dr. Ana C. Martinez

The University
of Texas at El Paso



Dr. Alexis Maurel



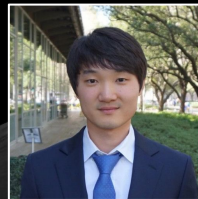
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Sina Bakhtar Chavari



Dr. Bharat Yelamanchi



Questions?



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References and Permissions

Slide 2: <https://www.nasa.gov/image-feature/the-soyuz-tma-15m-spacecraft-0>

<https://www.nasa.gov/exploration/multimedia/ifa18833.html>

Slide 3: NASA STMD

Slide 4: Graphics provided by the university of texas at el paso(UTEP)

Slide 5: Nasa Graphic

Slide 6. Image from NASA graphic from [Recycling Symbol - Download the Original Recycle Logo](#)

Slide 7: Table produced by SIBatt-3D project

Slide 8: Table produced by SIBatt-3D project - Nasa images

Slides 9: Images provided by NASA

Slides 10: Images provided by NASA

Slide 11: Graphics from UTEP and NASA ARC for SIBatt-3D

Slide 12: NASA Images taken for this presentation

Slide 13: NASA and UTEP Images taken for this presentation

Slide 13: NASA and UTEP Images taken for this presentation

Slide 14: Graphs provided by UTEP for SIBatt 3D project

Slide 15: Images provided by YSU

Slide 16: Graphs provided by UTEP

Slide 17: Graphs and images provided by UTEP

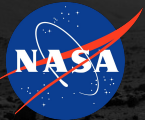
Slide 18: Graphic provided by UTEP

Slide 19: Nasa Images

Slide 20: Graphic provided by UTEP Chart created by NASA

Slide 21: Images provided by NASA

Slide 23: Team images used with approval



What is SIBatt-3D?

- ★ Developing additively manufactured battery cells and supporting systems
 - In-Space/On-Surface
 - Sodium Ion Chemistry
 - ISRU Feedstocks
 - Integrated printed supporting elements

