

Solid-state Precipitation in Molecular Dynamics: kMC-MD Hybrid Simulations

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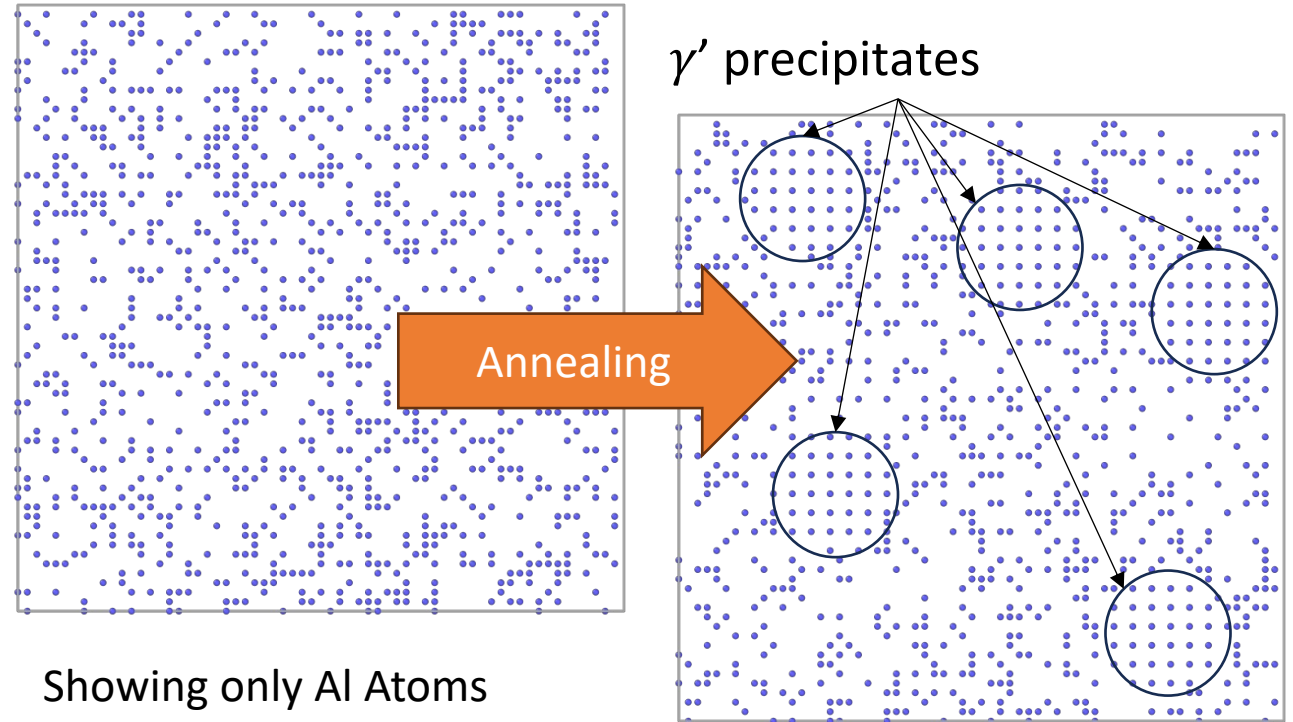
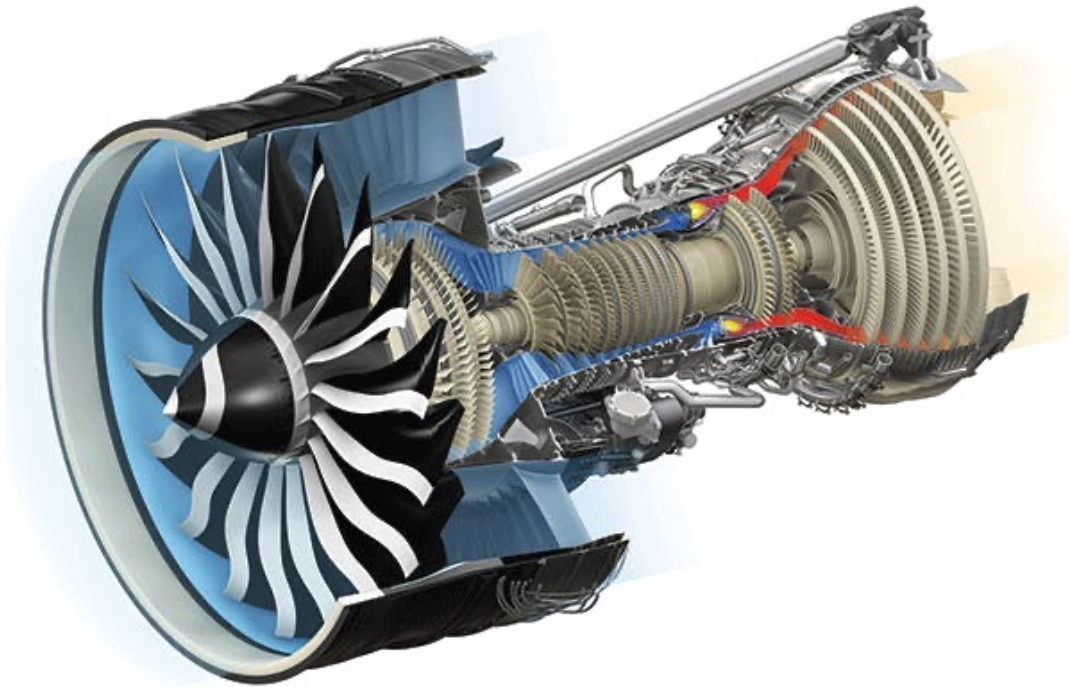
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Ni-based Superalloys

- γ' ($\text{Ni}_3\text{Al} - \text{L}_{12}$) precipitates are key in controlling creep properties in advanced superalloys
- Determining annealing conditions to produce precipitates of specific size and distribution through experimental trial-and-error
- Fundamental parameters of nucleation difficult to obtain through experiments



Solid-State Phase Precipitation

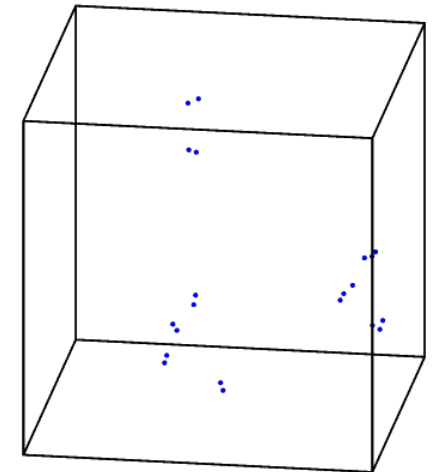
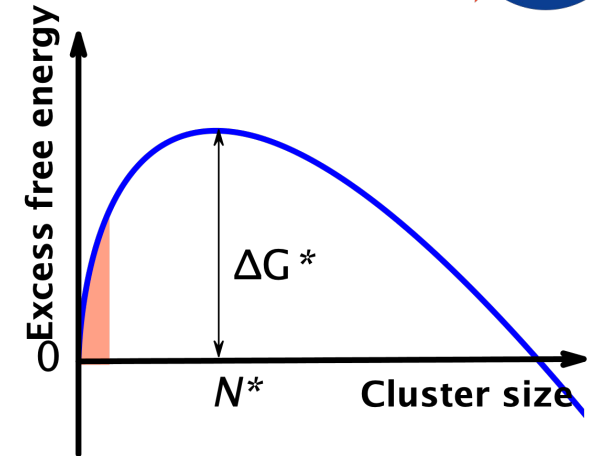
- Precipitation of new phases relies on solid-state diffusion to distribute chemical species evenly
- Majority of time spent in sampling small cluster sizes
- Solved for Liquid-Solid (Persistent Embryo Method)
- Precipitate grows once energy penalty for nucleation is overcome

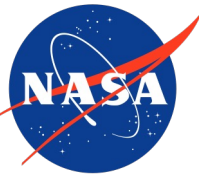
$$\Delta G = V\Delta\mu + A\sigma$$

- Calculation of interfacial energy through determining critical nucleus size

$$\frac{R^* \Delta G}{2} = \sigma$$

Solid-state Diffusion is too slow for observing precipitation in MD simulation





Solid State Diffusion – kMC in MD

Challenge:

- Inaccessible to MD simulation due to short timescales
- Atom-based diffusion models use lattice-based kMC technique
- How can we combine kMC diffusion with MD-based physics?

Solution:

- Use Voronoi-based neighbor selection to moderate solute movement between nearest-neighbors under Monte-Carlo based sampling, implemented in LAMMPS

Examples:

- Test by illustrating replication of ideal diffusion profiles
- Show precipitate growth in Ni-based superalloy
- Use kMC-MD simulation to calculate interfacial energy from critical nucleus

KMC-MD Solute Diffusion

Predicted ideal solute diffusion vs. KMC

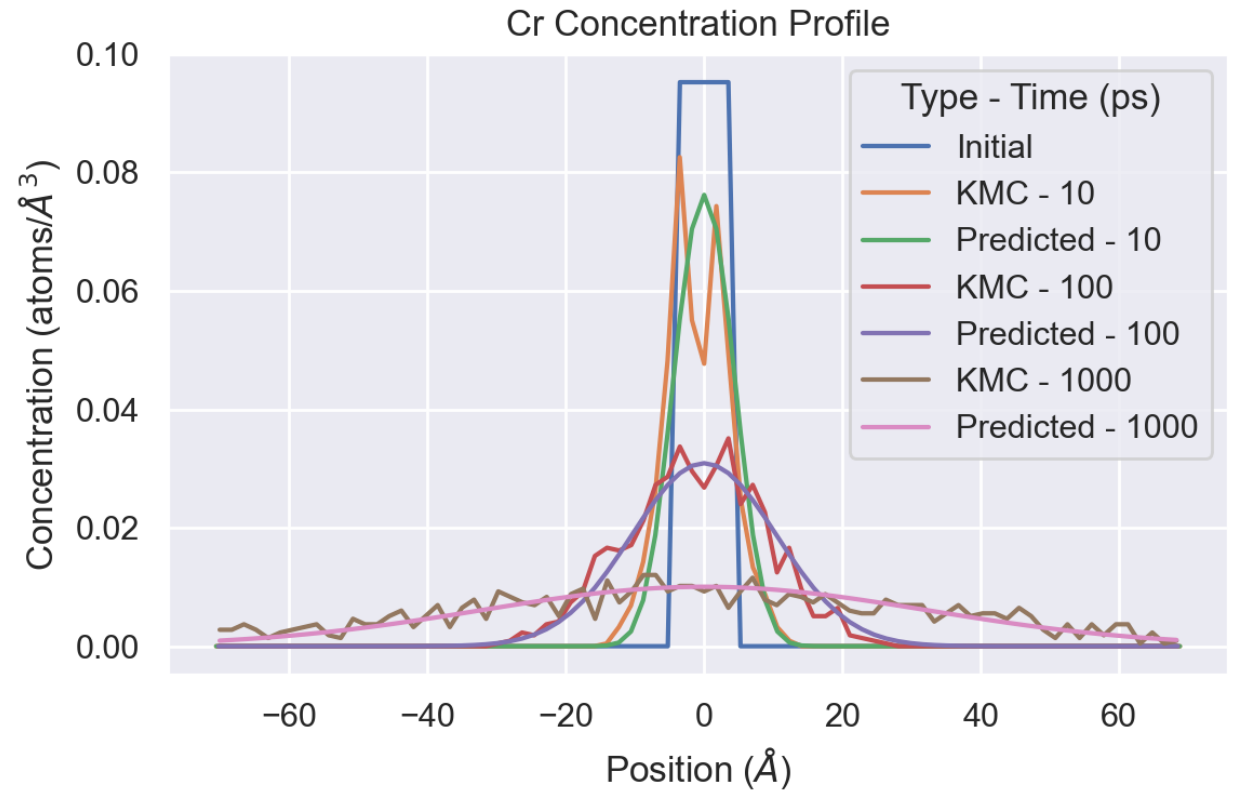
- Initialize fully-periodic cell of Ni with central concentration of Cr
- 3.5 x 3.5 x 14 nm
- 1000K MD/MC Temperature
- Swap rate of 1 atom/fs
- $D = 0.001 \text{ \AA}^2/\text{fs}$

$$c(x, t) = \frac{c_0}{2} \left(\operatorname{erf} \left(\frac{x + \frac{w}{2}}{\sqrt{4Dt}} \right) - \operatorname{erf} \left(\frac{x - \frac{w}{2}}{\sqrt{4Dt}} \right) \right)$$

w = Initial Pulse Width, D = Diffusivity

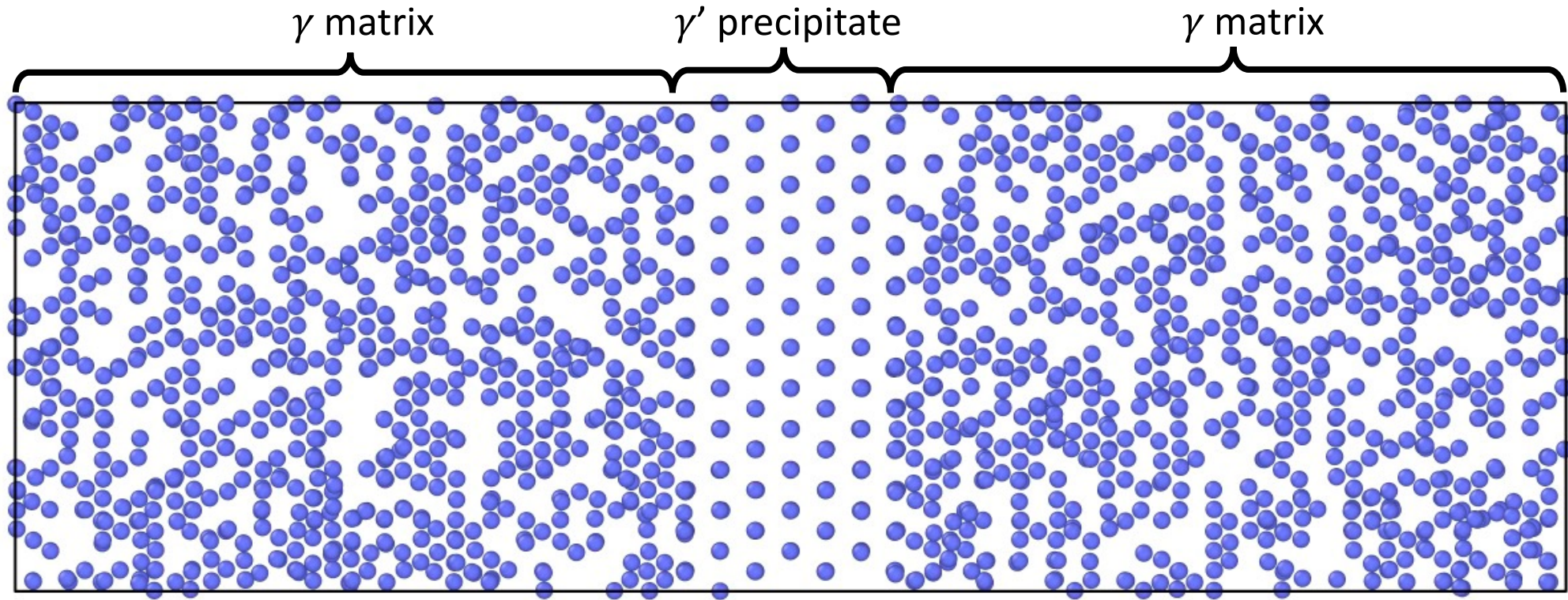
x = Position, t = Time

$$D = \frac{r^2}{12}$$



Good agreement between KMC-MD simulation and diffusion equation

KMC-MD Precipitate Growth



Swap 1% of Al atoms
every 50 MD steps

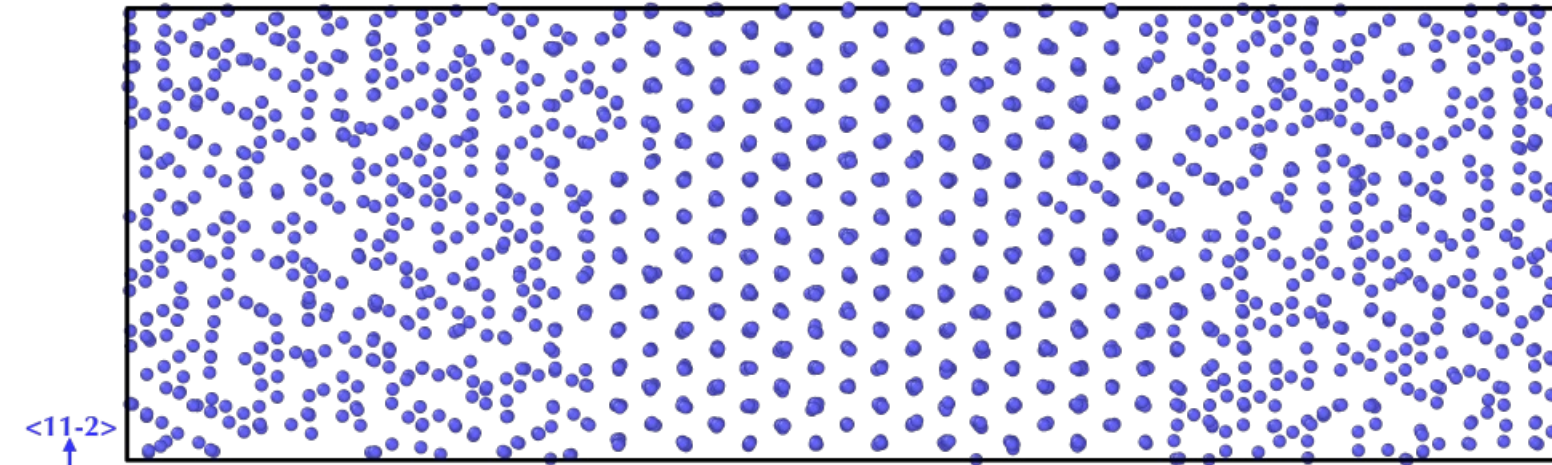
Showing only Al Atoms

2128 Al Atoms
1000K, 0 Bar NPT

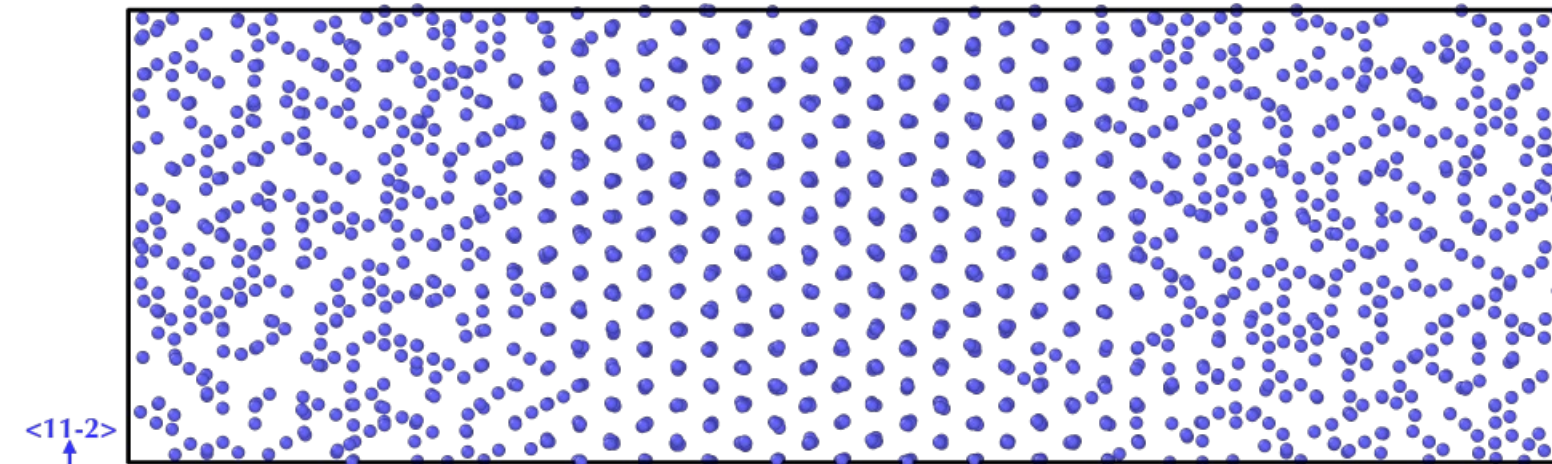
Final Structures – Al atoms

kMC-MD

Swap 1% of Al atoms every 50MD steps



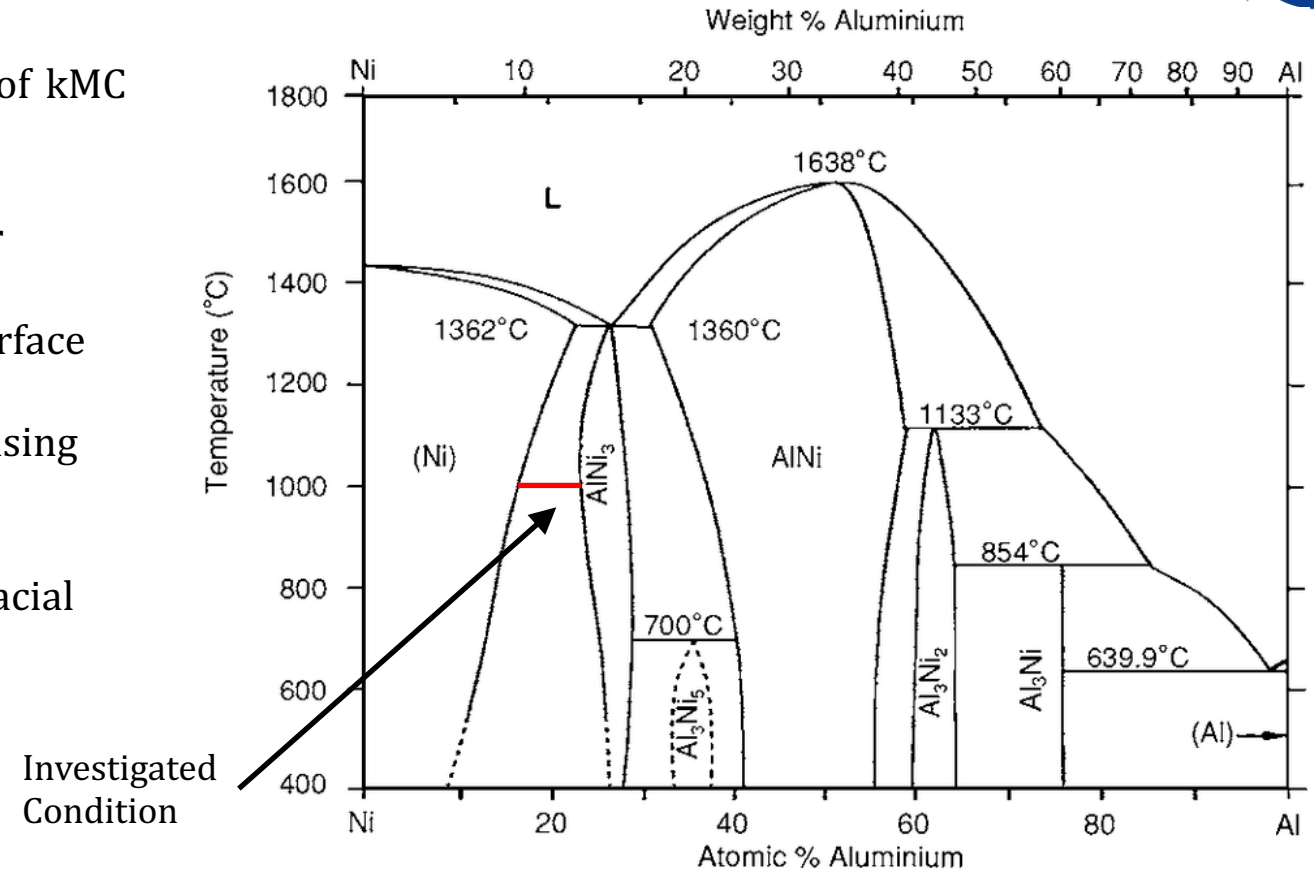
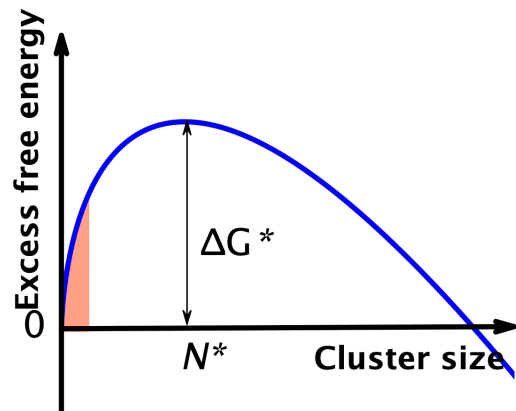
MC-MD



**2128 Al Atoms
1000K, 0 Bar NPT**

MD Simulation of Precipitation

- ❖ kMC-MD algorithm combines the time resolution of kMC and correct physics of MD simulation in LAMMPS.
- ❖ Demonstrated accelerated solute diffusion behavior
- ❖ Verified solid-phase growth behavior using flat interface
- Simulate phase precipitation from a solid solution using novel MD/kMC simulation technique.
- Determine critical nucleus size N^* , convert to interfacial energy



Equilibrium Condition @ 1000K: FCC Ni ~ 12.5% Al
 L_{12} Ni-Al ~ 24% Al

Simulation Setup

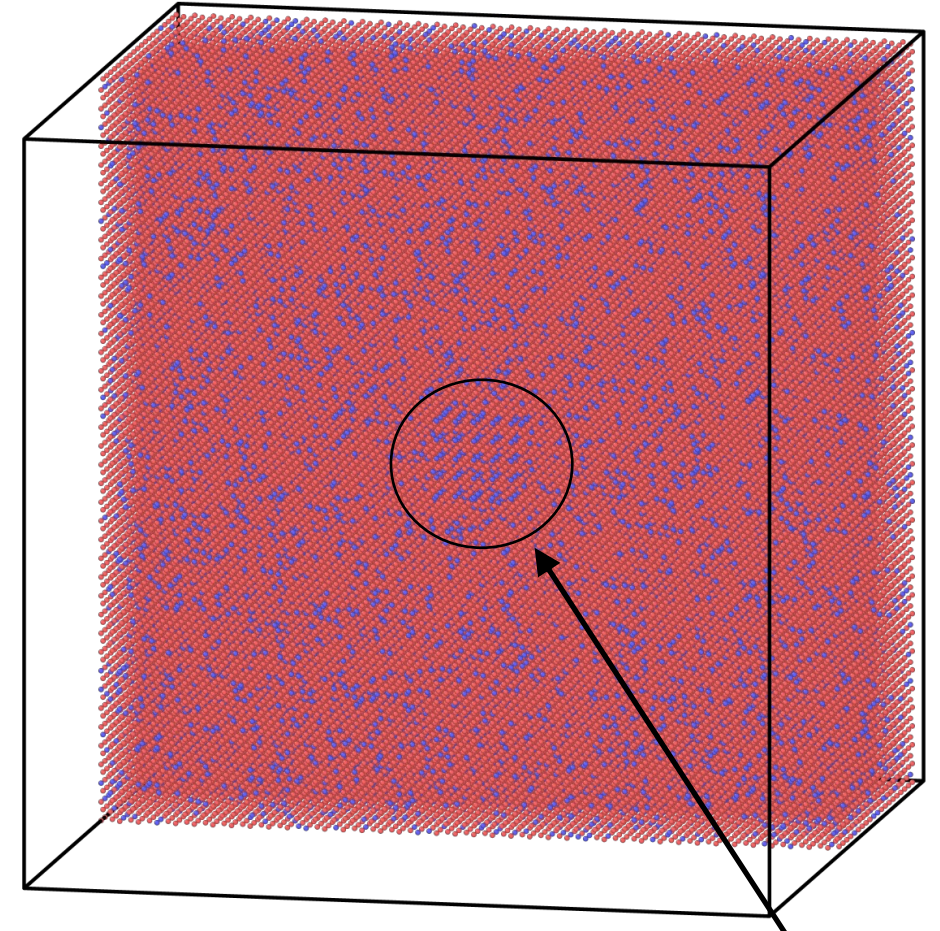
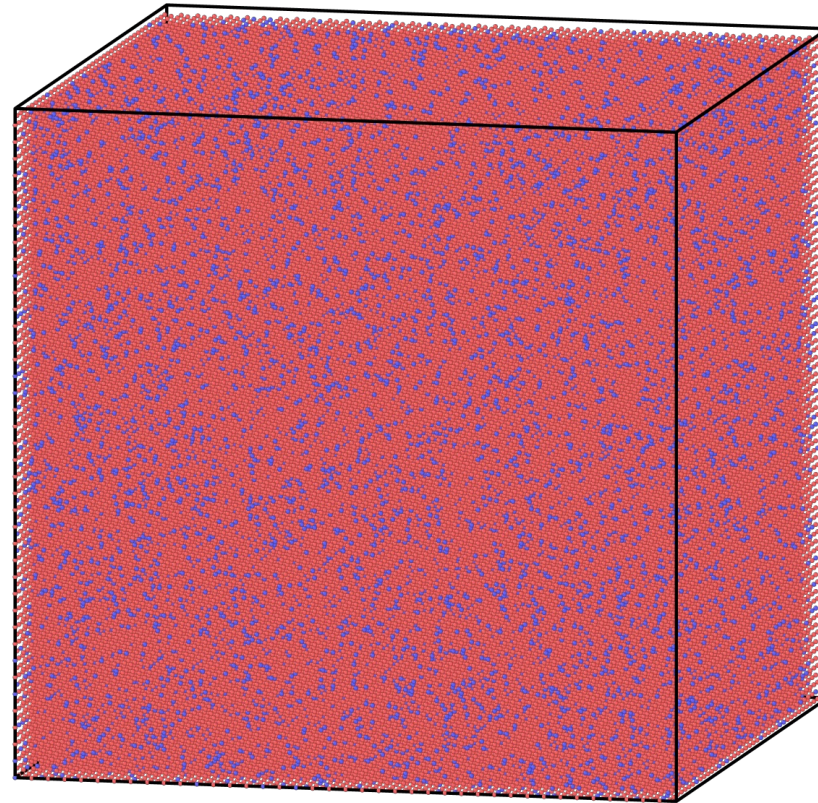
□ Need to identify critical nucleus size N^*

- Ni-Al system

- Initialize Cuboidal or Spherical L_{12} Precipitate Nucleus inside FCC Ni
- Randomly distribute Al in remaining Ni matrix
- Equilibrium matrix concentration $\sim 12.5\%$

- Evolve system with KMC/MD

- Swap Al atoms with neighboring Ni atoms to simulate diffusion process
- Accelerating diffusion process accelerates precipitate growth

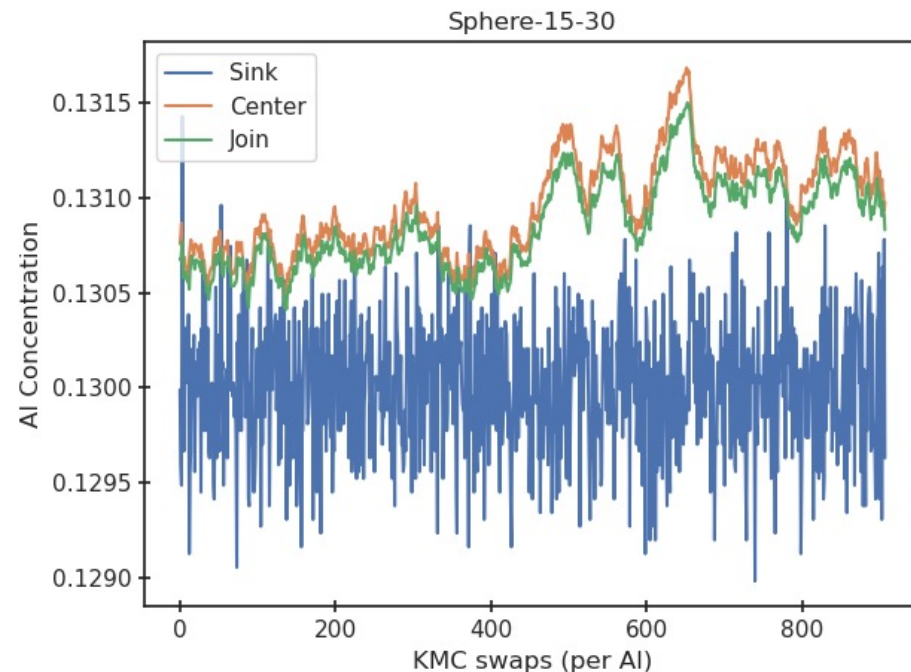


Ni atoms – Red
Al atoms – Blue

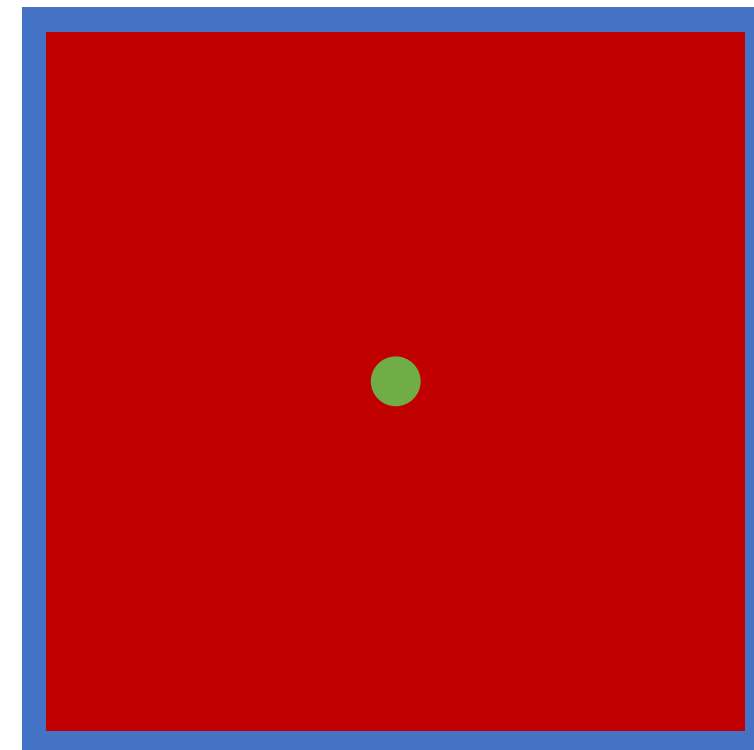
Initial Precipitate

Constant Driving-Force MD Simulation

- Developed new procedure in LAMMPS for fixing composition of regional composition to specific concentration, varying the global concentration over the simulation
 - Calculate concentration in sink region
 - Determine difference from target concentration
 - Change required number of atoms to new type to meet target



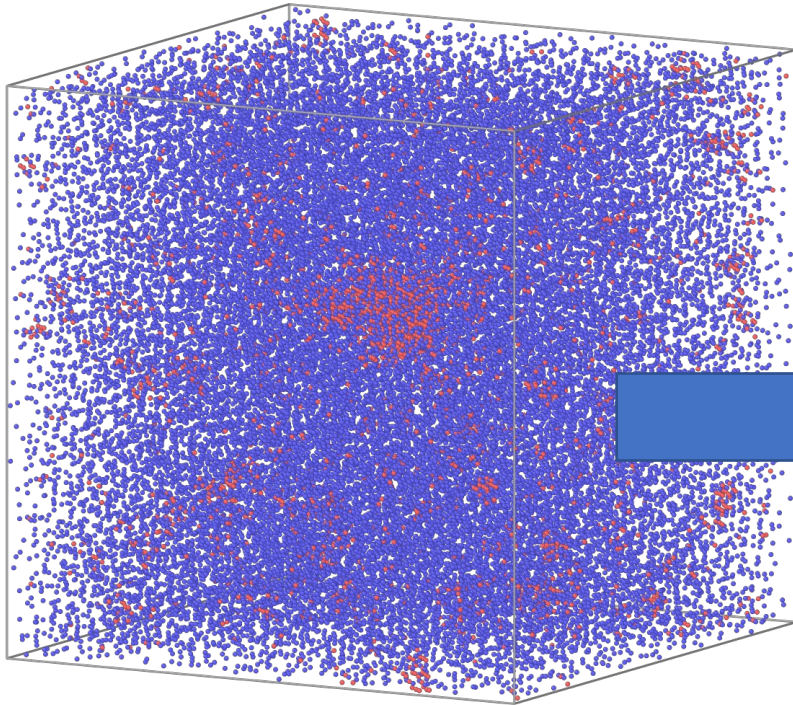
Slice through center of simulation



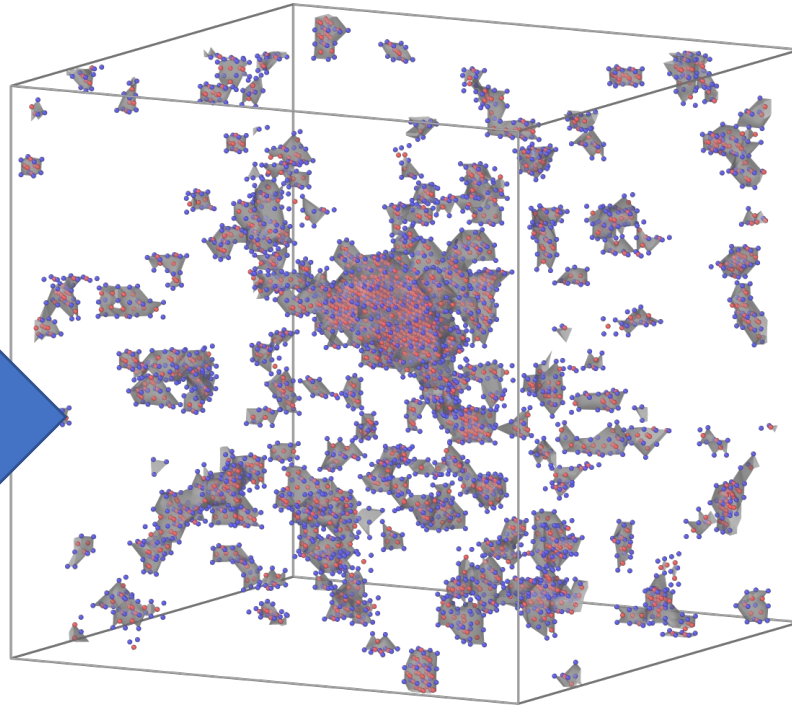
- Source/Sink
- Matrix
- Precipitate

Identifying Precipitate Phase

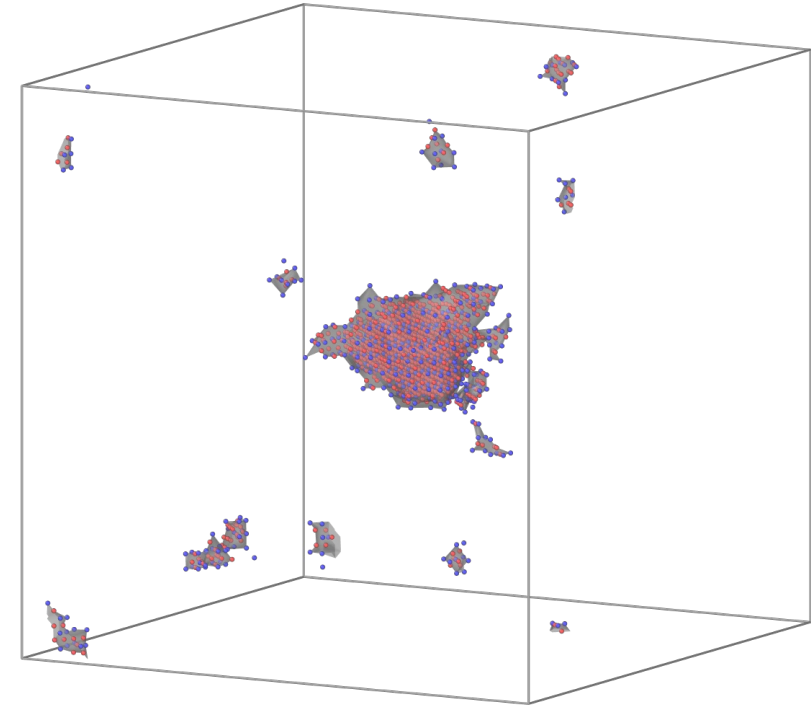
Polyhedral Template Matching (PTM) ^φ



PTM + Clustering



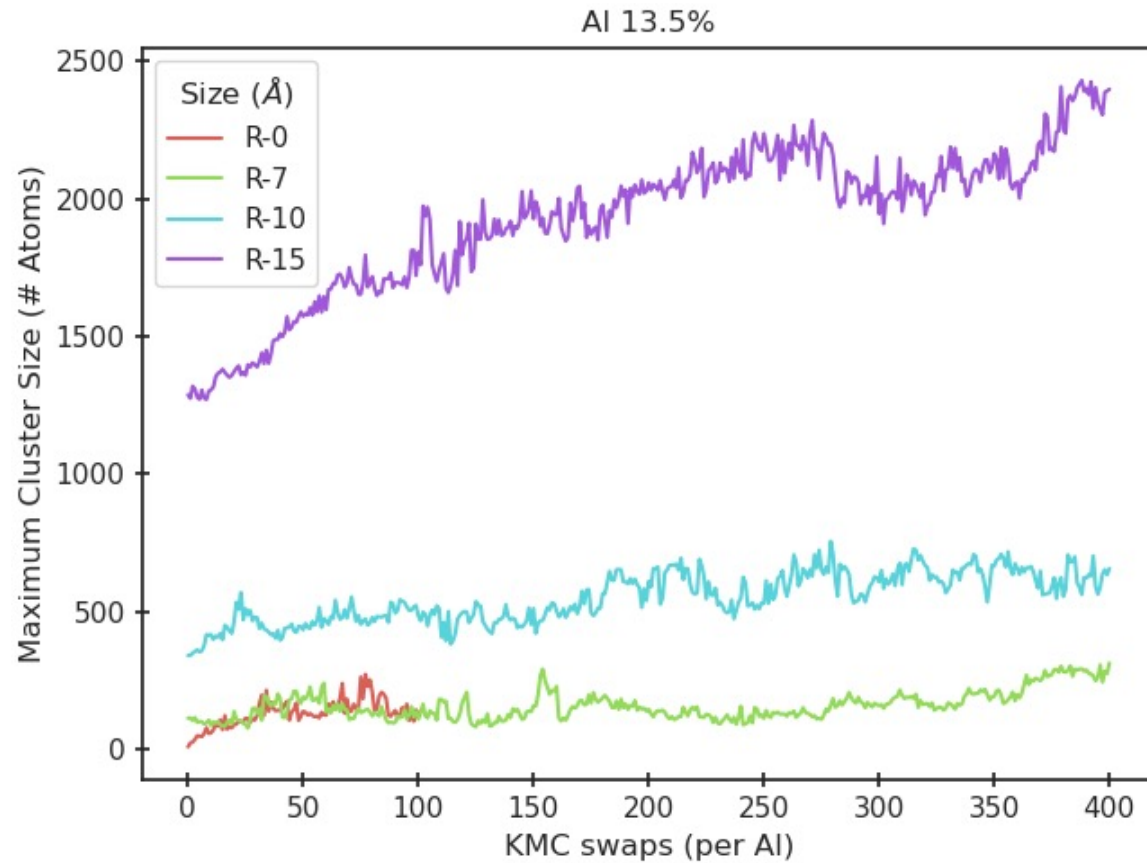
New Algorithm – Neighbor Shell Concentration



- All images are the same atomic structure
- PTM implemented in OVITO
- Existing tools cannot select for L_{12} ordered phase with precision

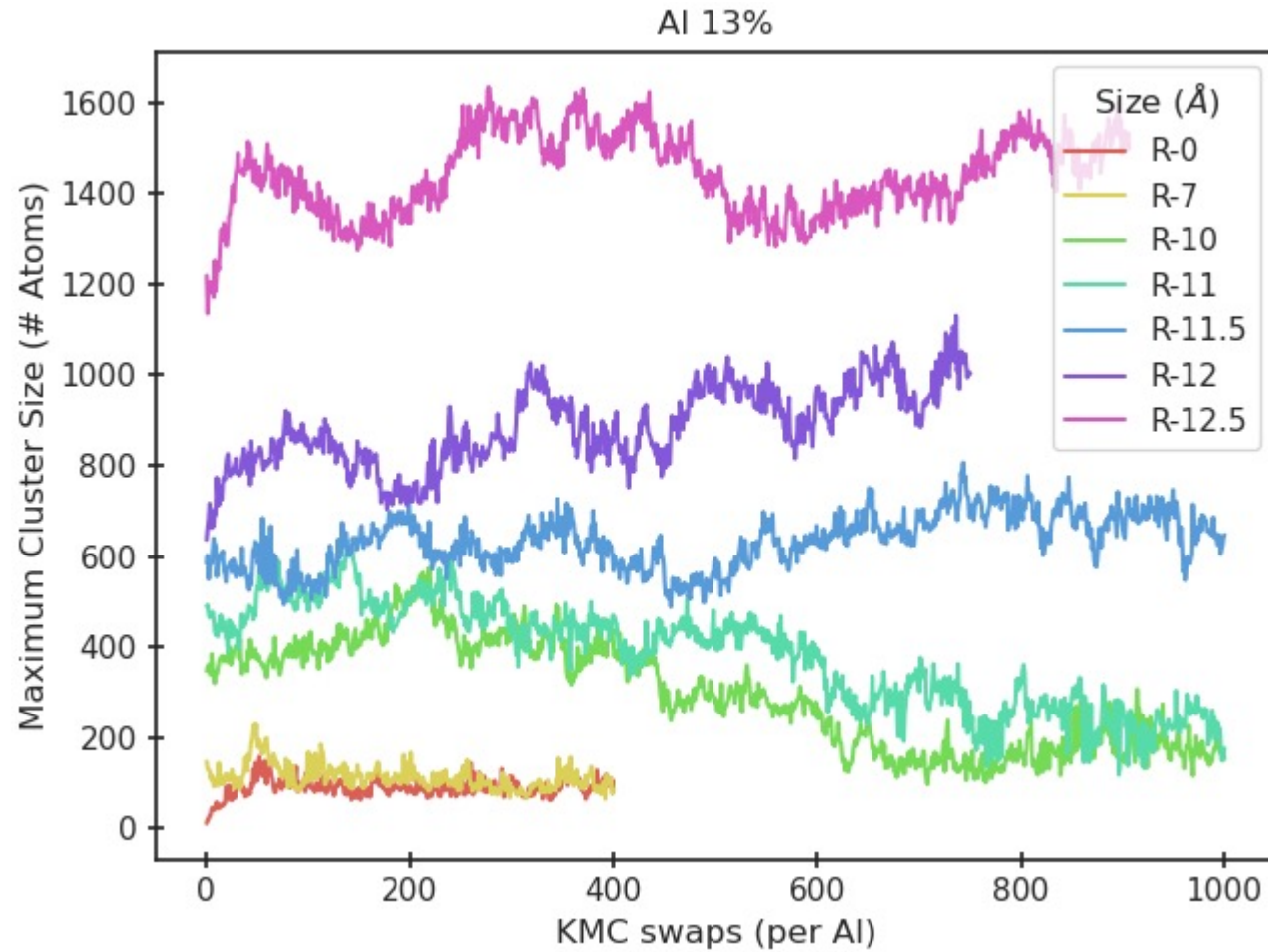
- Developed new algorithm for identifying L_{12} nuclei
- Based on local concentration of neighbor shells

13.5% Al Matrix Phase

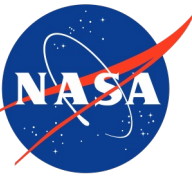


- 7 Å initial precipitate no different growth than starting without initial nuclei
- Driving force for precipitation too large

13% Al Matrix Phase



- Constant driving force from source/sink allows for clearer picture of precipitate size change.
- Critical nucleus size between 11 and 12 Å initial precipitate radius



Precipitate Comparison

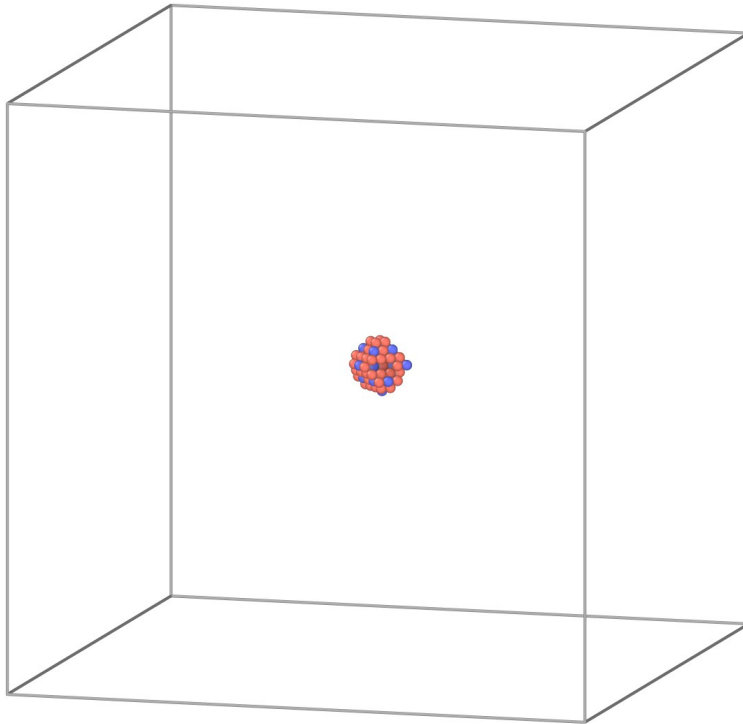
CALPHAD
kMC-MD Simulations

$$\frac{\Delta G^* R^*}{2} = \sigma$$

Interfacial Energy

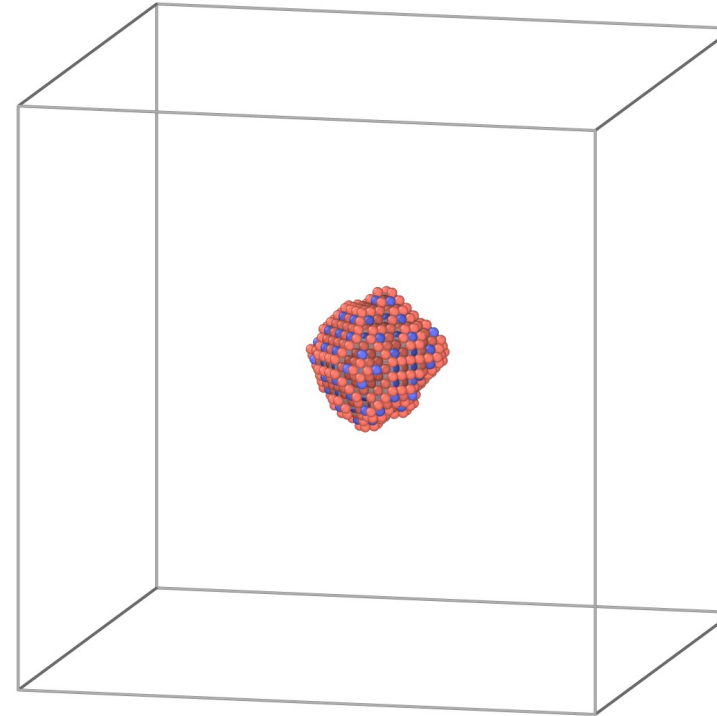
Only Viewing
Precipitate Atoms

7 Å

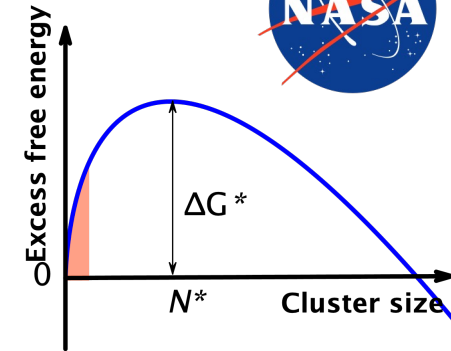


Smaller Precipitate Shrinks

15 Å



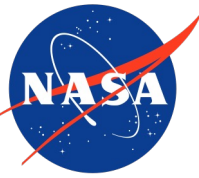
Larger Precipitate Grows



Interfacial energy
at 1000K

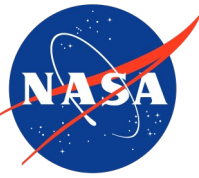
Experiment:
 $18.2 \pm 3.5 \text{ mJ/m}^2$

Simulation:
 17.4 mJ/m^2



Conclusions

- Hybrid kinetic Monte Carlo simulation approach enables simulation of substitutional solute diffusion through Molecular Dynamics
- New kMC-MD simulations allow for simulation of previously untapped atomistic behaviors such as solid-solid phase transformations
- Identified γ - γ' interfacial energy of 17.4 mJ/m² within 5% of experimental value (18.2 mJ/m²) at 1000K
- Improved over previous computational method for determining γ - γ' interfacial energy using capillary fluctuations at 12.9mJ/m² @1000K ^φ



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