Integrated Simulations of the Sabatier and Carbon Vapor Deposition Reactor to Understand Its Impacts to Operations and Performance

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The carbon vapor deposition (CVD) reactor is a technology developed by Honeywell Aerospace to convert methane, at high temperatures, into hydrogen and solid carbon. This element is coupled with a Sabatier reactor to support a closed-loop environmental control and life support system with the aim of achieving nearly complete oxygen recovery (>95%). Initial open-loop, brassboard CVD reactor tests and simulations have shown the CVD's ability to achieve moderately high methane conversion and high hydrogen selectivity. However, in an integrated system, additional deficiencies are expected due to recycling of unreacted or extraneous species from the Sabatier reactor (e.g., carbon dioxide, hydrogen, water) and CVD reactor (e.g., hydrocarbons, methane, etc.). Sabatier and CVD reactor models were integrated and simulated to predict potential impacts to individual reactors' and the overall system's performance. The simulations showed that increasing the recycle of the CVD effluent hydrogen combined with decreasing the system inlet hydrogen flow rate (i.e., drawing a stoichiometric flow rate from an electrolyzer) can lead to an oxygen recovery of > 95%. However, system integration comes at a detriment to the individual reactors. The simulations show the initial conversion from the integrated system (Sabatier = 87% and CVD = 59%) to be lower than the standalone systems (Sabatier = 91% and CVD = 69%). Furthermore, transient simulations show substrate densification, leading to worsening methane conversion coupled with increasing acetylene production, which is commensurate with soot formation. Simulations predict a shortening of the maintenance interval (i.e., time until CVD methane conversion drops below 50%) in the integrated system, which would increase the consumable substrate mass. These analyses highlight the importance of long-duration, integrated tests to corroborate these findings as well as suggest potential modifications (e.g., intermediate gas separations) to improve performance.

Nomenclature

ACM	=	Aspen Custom Modeler	H_2	=	hydrogen
CDRA	=	carbon dioxide removal assembly	$H_2:CO_2$	=	hydrogen-to-carbon-dioxide ratio
CDRS	=	carbon dioxide removal system	H_2O	=	water
CH_4	=	methane	ISS	=	International Space Station
C_2H_2	=	acetylene	OGA	=	oxygen generation assembly
C_6H_6	=	benzene	PCI	=	Precision Combustion, Inc.
СМ	=	crewmember	SA	=	Sabatier assembly
CO_2	=	carbon dioxide	SDU	=	Sabatier development unit
CVD	=	carbon vapor deposition	X_{CH4}	=	methane conversion
ECLSS	=	environmental control and life support			
		system			

FDUU2	=	4-beu	carbon	aloxide	scrubbel
			<i><i>xaxxxxxxxxxxxxx</i></i>	<i>x i x x i x x i x x i x x i x x i x x i x x x i x x x x</i>	

GC = gas chromatography

GIEM = Gateway Integrated ECLSS Model

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I. Introduction

In consideration of the closed-loop environmental control and life support system (ECLSS) to facilitate long-duration human exploration, spacecraft life support technologies, such as a Sabatier¹⁻⁵ and carbon vapor deposition (CVD) reactor system⁶⁻⁹ have been developed and we have demonstrated the capability of each individual system to provide sufficient function to recycle and transform the prevalent metabolic gas product carbon dioxide (CO₂) into breathing oxygen (O₂).

Figure 1 shows a notional closed-loop ECLSS architecture. CO_2 is collected and concentrated by a CO_2 removal system, such as the carbon dioxide removal assembly $(CDRA)^{10,11}$ or 4-bed carbon dioxide scrubber (FBCO2).¹²⁻¹⁵ The accumulated CO_2 is then delivered in conjunction with hydrogen (H₂), which is produced by the oxygen generation assembly (OGA), to a CO_2 reduction system (i.e., Sabatier reactor) to convert the CO_2 into water (H₂O) and byproduct methane (CH₄). Various Sabatier-based CO₂ reduction technologies have been developed. The Sabatier assembly had been operated on the International Space Station from 2011 to 2017 until recent return and upgrades to support improved reliability and stability for exploration.¹⁻³ Another Sabatier system was developed by Precision Combustion, Inc. (PCI)^{4.5}. PCI's Sabatier development unit (SDU) utilizes their Microlith® catalytic technology to produce a compact Sabatier reactor system that is able to achieve high CO_2 conversion and CH₄ selectivity at high space velocities.



Figure 1. Notional closed-loop ECLSS architecture with CO2 reduction.

The Sabatier reaction (as shown in Equation 1) produces H_2O from CO_2 , which is inevitably delivered to the OGA to regenerate O_2 . However, its byproduct CH_4 was vented as part of the International Space Station Air String demonstration with the Sabatier assembly¹⁶. This loss of H_2 in the form of CH_4 limits the ECLSS's maximum O_2 recovery to 50%.

$$CO_2 + 4 H_2 \rightleftharpoons CH_4 + 2 H_2O$$
 Equation 1

Processing and recovering H₂ from CH₄, hence, becomes pertinent to fully closing the spacecraft O₂ balance when using a Sabatier-based ECLSS system. The CVD (or CH₄ pyrolysis) reactor developed by Honeywell is one such technology for closing the air loop to enable near 100% O₂ recovery.⁶⁻⁹ As shown in Figure 1, the CVD reactor system takes in CH₄ and produces H₂ and solid carbon (C(s)), where the former may be used to supplement the requisite H₂ needs of the Sabatier reactor. The CVD reactor uses the non-oxidative thermal decomposition of CH₄ at high temperatures (i.e., > 1,000 °C), also known as CH₄ pyrolysis, to produce H₂, C(s), and other carbonaceous species. The CVD reactor uses a carbon fiber substrate within its reactor volume to immobilize solid pyrolytic carbon rather than allowing the formation of soot entrained in the process gas flow. This simplifies the separations and cleanup of the process gases. Honeywell's brassboard CVD reactor has demonstrated the ability to convert 4 crewmembers (CMs) worth of CH₄ with reasonably high conversion (50 - 80%) and high H₂ selectivity (> 95%).

To assess these CO₂ reduction systems (i.e., Sabatier and CVD reactor) under more realistic operational scenarios, integrated testing is needed to elucidate the effects of integration, recycling of byproducts, incomplete conversion, and imperfect selectivity. By leveraging existing models of the individual Sabatier and CVD reactors, ¹⁷ a model of the integrated system may be developed to perform predictive performance assessments closer to their real operational state. The integrated model may then be used to provide a priori insights into the expected behavior to inform integrated test development. Integrated modeling and simulation is powerful in that the models are more flexible compared to real hardware to account for different test configurations; they allow for rapid analysis of different operating conditions; running the simulations requires fewer resources compared to hardware integration and testing; and the models provide detailed tracking of all process parameters (e.g., compositions and flow rates) that may otherwise be cumbersome to acquire in testing. The objective of this integrated modeling and simulation effort for the combined Sabatier and CVD reactor systems is to provide a priori predictions on the effects of integration (e.g., recycling of byproducts, incomplete conversion, etc.) on the individual reactor and overall system performances.

II. Integrated Model Formulation

A. Sabatier Reactor Model Description

The Sabatier reactor model used for this integrated modeling effort was based on PCI's SDU, which was delivered to NASA as part of Contract NNX11CC05CSDU.^{4,5} The SDU includes the Sabatier reactor portion, which was modeled as a plug flow reactor due to its high operating space velocities, and a H₂O condenser, which is modeled as a vapor liquid equilibrium at a fixed outlet setpoint temperature. To keep the model simple but insightful, the SDU model solely includes the major reactions of CO₂ methanation (Equation 1), CO methanation (Equation 2), and the reverse water gas shift reactions (Equation 3).

$$CO + 3 H_2 \rightleftharpoons CH_4 + H_2O$$
 Equation 2

$$CO_2 + 2 H_2 \rightleftharpoons CO + H_2O$$
 Equation 3

The kinetic parameters for these reactions were fitted using test data of the CO₂ conversion and CH₄ selectivity at different steady-state conditions (55 combinations of temperature, pressure, and hydrogen-to-carbon-dioxide ratio [H₂:CO₂]) as well as time-resolved gas chromatography (GC) data. Over the steady-state test points, the model was able to predict CO₂ conversion with an average error of \pm 2.8% and standard deviation of 1.9%. The time-resolved GC data was able to be simulated with an average error on the mole percent of CO₂, H₂, CH₄, and carbon monoxide of \pm 3.2 mol%, \pm 4.0 mol%, \pm 5.5%, and \pm 0.3%, respectively. The model is able to achieve good agreement with the test data.

B. Carbon Vapor Deposition Reactor Model Description

A CVD reactor model was developed to capture the multitude of physicochemical phenomena that constitute this highly dynamic system.¹⁷ The model considers the homogeneous gas-phase CH₄ pyrolysis reactions, the heterogeneous solid carbon deposition and substrate densification phenomenon, and the axial gas transport through the reactor volume and its multiple carbon fiber substrate layers. The 1-D CVD reactor model material balance was based on a general packed-bed, axial dispersion plug flow reactor model, and the original kinetic model included the consideration of 59 different gaseous and surface species and 243 total reactions. Based on the sensitivity analysis over the entire reaction network as described in Chen, *et al.* (2024), the most insensitive reactions were removed from the model to improve model run times while maintaining nearly equivalent model results. The reaction network for the CVD reactor model was reduced from 243 reactions to 136 reactions, which led to a marked reduction in computational run times and improved model stability.

C. Integrated SDU and CVD Reactor Model Description

Both the SDU and CVD reactor models were developed using Aspen Custom Modeler (ACM) due to its ability to allow for custom model development while leveraging its ordinary and partial differential equation solvers as well as Aspen's large chemical properties database. The ACM software also allows already built models to be readily connected to simulate integrated systems and processes. An example of such a model is the Gateway Integrated ECLSS Model (GIEM) that includes various subsystems of the Gateway air revitalization system such as the CO₂ and

humidity control, CO₂ removal system, condensing heat exchanger, and more.¹⁸ The SDU and CVD reactor models were integrated by connecting the SDU gas outlet to the CVD reactor model's inlet and recycling the CVD gas outlet to the SDU inlet to close the loop. The integrated models are shown in Figure 2.

Some key features of the integrated model are noted. Firstly, the integrated system assumes no intermediate H_2 separations between the CVD reactor and the SDU inlet, where all the CVD effluent may be directed to recycle to the SDU. If a viable H_2 separator for spacecrafts exists, the model may be updated to reflect that unit process. Additionally, the SDU was designed for a 3.24 crewmember (CM) processing rate, whereas the CVD reactor was designed to be able to process a 4 CM load. Therefore, the integrated model includes scale-up and scale-down process units that can be used to arbitrarily increase the total molar flow while maintaining the same composition. This illustrates one of the strengths of modeling, where it is easily adjustable compared to hardware testing, which would involve more complex gas supplement and purge designs, flow controller selection and calibration, and potentially in-line gas analyzers to determine instantaneous gas effluent compositions. The integrated SDU and CVD reactor gas effluent back to the SDU versus the portion that is purged. Real-world chemical process systems that employ a recycle stream will typically purge a portion of the reactor effluent out of the process to prevent buildup of undesirable products that may dilute the reactants and limit system performance. The integrated SDU and CVD reactor model will be used to assess different recycle and purge flow rates to determine their effect on performance (i.e., individual reactor conversion and system O_2 recovery).



Figure 2. Depiction of the integrated SDU and CVD reactor models in ACM.

III. Integrated Model Assumptions

To perform the integrated modeling and simulation of the combined SDU and CVD reactor system, the set of operating conditions that define the individual subsystem's nominal operations is established. **Table 1** lists the nominal operating conditions for the standalone SDU and CVD reactor systems. However, under integrated system operations, these values will likely deviate from the nominal values due to incomplete conversion and imperfect selectivity that will result in the recycling of unreacted reactants (e.g., CO_2 and CH_4) and byproducts (e.g., acetylene [C_2H_2]). Additionally, due to the dynamic nature of the CVD reactor, whose substrate fills with solid carbon over time, the real operations of the integrated system will also be dynamic with worsening CH_4 conversion in the CVD reactor over time and hence less H_2 being recycled back to the SDU. This integrated model has not implemented any feedback control mechanism to adjust the H_2 flow over the course of a run to maintain a specific $H_2:CO_2$ but rather recycles a specified fraction of the CVD reactor effluent, whose composition will inherently change over time. The simulation results may be informative as to whether a more sophisticated flow control scheme will be necessary.

Operating Parameters	Nominal Values			
Sabatier Reactor				
CO ₂ Feed Rate	1.3 SLPM (or 3.24 CM equivalent)			
$H_2:CO_2$	4, 4.25, 4.5			
Average Reactor Temperature	~350 °C			
CVD Reactor				
CH ₄ Feed Rate	1.6 SLPM (or 4 CM equivalent)			
Setpoint Temperature	1170 °C			

Table 1. SDU and CVD Reactor Nominal Operating Conditions

IV. Integrated SDU and CVD Reactor Model Results

The integrated SDU and CVD reactor model is useful for providing predictions on the overall system performance versus trying to justify system capabilities based solely on operation of the standalone systems. The integrated model may also help to provide input into integrated system test planning. The use of models to perform this type of integrated analysis leverages existing models of the two subsystems (i.e., the SDU and CVD reactor) and may be performed over a wide array of conditions with less resource overhead compared to hardware integration and testing.



Figure 3. Examples of the integrated Sabatier + CVD reactors configurations.

Figure 3 depicts some of the scenarios that were simulated using the integrated SDU and CVD reactor model. The open-loop Sabatier and CVD reactors (Figure 3a) shows how, without recycling any of the H₂ derived from CH₄, the system's O₂ recovery is limited to a maximum of 50% where the H₂O produced by the system carries half of the H₂ that was fed into the system. In an ideal system (Figure 3c), both the Sabatier and CVD reactors operate at 100% conversion and 100% product selectivity to entirely close the O₂ balance (i.e., 100% O₂ recovery). The ideal system is able to reduce CO₂ using H₂ into H₂O and C(s) according to their stoichiometric ratios (i.e., 1 CO₂ + 2 H₂O \rightarrow C (s) + 2 H₂O) without any gaseous byproduct formation and a solid carbon waste. In reality, both the Sabatier and CVD reactors operate at CVD reactors operate at < 100% selectivity, so a situation like that shown in Figure 3b is expected. Due to recycling of the CVD reactor effluent, the integrated system's O₂ recovery may be markedly improved over an ECLSS with only a Sabatier reactor or the open-loop system (Figure 3a). However, incomplete conversion and imperfect selectivity may inherently limit the O₂ recovery to < 100%.

The integrated SDU and CVD reactor model was used to assess the following cases: (1) steady-state open-loop integration with no recycle, (2) steady-state closed-loop integration, and (3) transient closed-loop integration. The steady-state open-loop integrated system is used to represent the baseline effect of a pure SDU effluent feed to the CVD reactor. The closed-loop integrations are used to provide insight into how recycling the CVD reactor effluent affects the overall system performance and necessary maintenance interval (i.e., time between substrate changeouts).

A. Open-loop Integration and Baseline Results

The integrated SDU and CVD reactor was simulated in an open-loop configuration to establish the baseline effect of a pure SDU effluent on the downstream CVD reactor. The system was assessed with a H_2 :CO₂ of 4, 4.25, and 4.5. According to the Sabatier reaction stoichiometry (Equation 1), a H_2 :CO₂ of 4 is necessary if 100% conversion were achieved. However, the reaction thermodynamics will limit the maximal equilibrium conversion of CO₂ and H_2 to CH₄ and H₂O. Therefore, it is common to operate the Sabatier reactor with an excess amount of H_2 to shift the equilibrium toward the products according to Le Chatelier's principle.

Table 2. Open-loop integration Simulation Results						
		H2:CO2				
	4	4.25	4.5			
Operating Parameter						
SDU Inlet Flow Rate	15.95 mol/hr	16.75 mol/hr	17.55 mol/hr			
SDU Inlet Composition	20.0% CO ₂ 80.0% H ₂	20.0% CO2 19.0% CO2 80.0% H2 81.0% H2				
Performance Parameter						
H ₂ O Production Rate	5.50 mol/hr	5.70 mol/hr	5.84 mol/hr			
SDU Conversion	90.5%	93.8%	96.3%			
CVD Reactor Conversion	69.5%	67.7%	65.4%			
SDU O ₂ Recovery	86.2%	89.3%	91.6%			
System O ₂ Recovery	43.1%	42.0%	40.7%			

 Table 2. Open-loop Integration Simulation Results

The results from the open-loop integrated simulation are shown in **Table 2**. The SDU results show the reactor to operate with > 90% conversion, which increases with H_2 :CO₂. Similarly, the SDU O₂ recovery increases from 86.2% to 91.6% with an increase in the H_2 :CO₂ from 4 to 4.5. However, feeding an SDU gas effluent into the CVD reactor does appear to lead to a detriment in performance. Due to H_2 inhibition and dilution effects, the CVD reactor conversion worsens with increasing H_2 :CO₂. These open-loop simulations already show how integration of the individual systems can lead to a difference in performance level of the integrated system compared to what may be expected based on solely standalone testing.

The system O_2 recovery is determined by the H_2O condensed from the SDU divided by the larger of either the O_2 or H_2 fed into the integrated system (Equation 4). The CVD reactor gas effluent in the open-loop configuration is assumed to be vented. Therefore, the system O_2 recovery tends to be dictated by the recovery of H_2 . Therefore, although the SDU O_2 recovery is high, the system O_2 recovery is significantly lower and even decreases from 43.1%

to 40.7% when the H_2 :CO₂ is increased from 4 to 4.5. The reason is that additional H_2 is being fed into the system for marginally more H_2O production. This is an interesting insight that appears to subvert the logic that improving the Sabatier reactor's conversion is beneficial for the entire spacecraft when in reality it may be disadvantageous.

System
$$O_2$$
 Recovery = $\frac{SDU H_2O Product}{max(O_2 \text{ or } H_2 \text{ feed})}$ Equation 4

These initial open-loop simulations define the baseline conversions and recoveries that the closed-loop system will be compared against.

B. Closed-loop Integration Results

The integrated SDU and CVD reactor was simulated in a closed-loop configuration with different recycle flow rates to determine how recycling the CVD effluent would incrementally affect the individual reactor and integrated system performances. Two types of simulations were performed: steady-state simulations with a fresh substrate (i.e., without carbon deposition) and transient simulations with a dynamic substrate (i.e., with carbon deposition).

1. Fresh Substrate Simulation Results

Steady-state simulations with a fresh substrate were performed with recycle flow rates of approximately 5-70 mol/hr and system inlet H₂:CO₂ of 2.03 - 3.00. This represents an integrated system with partial recycle of the CVD effluent to nearly complete loop closure. The closer the system is to full loop closure, the greater the recycle flow rates become and the lower the system inlet H₂:CO₂ can be to sustain the Sabatier reaction. Figure 4 shows plots of the mixture H₂:CO₂ (Figure 4a), Sabatier conversion (Figure 4b), and CVD conversion (Figure 4c) at different recycle flow rates and system inlet H₂:CO₂. The green box in the plots bounds the SDU inlet H₂:CO₂ between 4 and 4.5, which is its nominal operating range during standalone SDU testing. Figure 4b and Figure 4c also show the baseline Sabatier and CVD reactor conversions from the open-loop simulations at a H₂:CO₂ of 4, 4.25, and 4.5 as points of comparison. Figure 4 is used to describe the effect of the CVD effluent recycle on the individual reactor performance.

The mixture $H_2:CO_2$ represents the inlet into the SDU after the system feed is mixed with the recycled CVD effluent. As shown in Figure 4a, the mixture $H_2:CO_2$ increases with the recycle flow rate at a constant system inlet $H_2:CO_2$. It makes sense that, as you recycle more of the CVD effluent, which is predominantly H_2 , the $H_2:CO_2$ going into the SDU will also increase. Another way to read Figure 4a is that, as the recycle flow rate increases, the system is able to operate with a lower system inlet $H_2:CO_2$ (i.e., lower H_2 feed into the system) while maintaining the requisite stoichiometry for the Sabatier reaction (i.e., minimum of 4 H_2 to 1 CO₂). The goal is to achieve a system inlet $H_2:CO_2$ of 2:1, which mimics the generation of H_2 and O_2 from water electrolysis and would result in 100% O_2 recovery.

However, as shown in Figure 4b and Figure 4c, the greater recycle flow rate leads to marked effects on the individual SDU and CVD reactor. An increased recycle flow rate leads to greater Sabatier conversion at a constant system inlet H_2 :CO₂ due to the greater amount of excess H_2 that improves the reaction rate and shifts equilibrium toward the products (CH₄ and H₂O). The unreacted H_2 that flows through the SDU, however, leads to a significant detriment to the CVD reactor conversion due to its inhibitory effects on carbon deposition and dilution effects on the reactant residence time in the CVD reactor. As seen in both Figure 4b and Figure 4c, the sensitivity of the Sabatier and CVD conversion to the recycle flow rate is much greater at the higher recycle flow rates, which may exacerbate the need for very precise flow controllers for hardware testing if complete loop closure is desired. In general, if trying to maintain the H_2 :CO₂ into the SDU between 4 and 4.5, a greater recycle flow rate appears to trend both the SDU and CVD reactor toward worsening performance (i.e., conversion) as indicated by the negative slope of the bounding green box in the figures below.



Figure 4. Plots of the (a) mixture H₂:CO₂, (b) Sabatier conversion, and (c) CVD conversion at different recycle flow rates and system inlet H₂:CO₂.

Figure 5 plots the mixture H_2O production rate from the SDU (Figure 5a), SDU O_2 recovery (Figure 5b), and system O_2 recovery (Figure 5c), which is defined by Equation 4, at different recycle flow rates. As with the prior figures, the open-loop configuration simulations at a H_2 :CO₂ of 4, 4.25, and 4.5 are also plotted as points of comparison. The figures below are representative of the simulated integrated system performance.

Despite the worsening individual reactor performances demonstrated in Figure 4b and Figure 4c, the integrated system generates more H_2O , and is able to achieve greater O_2 recovery. The H_2O production increases because not only is H_2 being recovered from the SDU CH₄ byproduct via the CVD reactor effluent, but also unreacted CO₂ from the first pass of the SDU is being recycled and given more opportunities to reduce. The SDU O_2 recovery (Figure 5b) mirrors the H_2O production rate trends and shows a flattening around 97%. This is a significant improvement from the open-loop configuration, which achieved a SDU O_2 recovery of 91.6% at a H_2 :CO₂ of 4.5.

The most significant effect of an increased recycle flow rate and decreased system inlet $H_2:CO_2$ is that the integrated system balance operates near the stoichiometry of water electrolysis and hence achieves a much greater system O_2 recovery compared to the open-loop simulations. The open-loop simulations predict a system O_2 recovery above 95%. Another result to note is that the system O_2 recovery is far more sensitive to the system inlet $H_2:CO_2$, whose decrease is facilitated by a higher recycle flow rate of the CVD effluent, rather than strictly increasing the recycle flow rate at a fixed system inlet $H_2:CO_2$. This result makes sense where the system O_2 recovery is mainly dictated by the difference between the system inlet $H_2:CO_2$ versus the output from a water electrolysis system (i.e., $H_2:O_2 = 2$). These results suggest that the integrated system is capable of operating together to achieve a high level of O_2 loop closure that may sustain long exploration missions with limited H_2 resupply. However, the reduction in the individual reactor performances is interesting and may suggest system modifications (e.g., intermediate H_2 separations to limit byproduct gas recycle) to improve performance.



Figure 5. Plots of the (a) H₂O production rate, (b) SDU O₂ recovery, and (c) system O₂ recovery at different recycle flow rates and system inlet H₂:CO₂.

2. Transient Simulation Results

Transient simulations over a dynamic substrate (i.e., with carbon deposition) were performed with a recycle flow rate of 10.3 mol/hr and system inlet H₂:CO₂ of 2.03. These operating conditions represent one of the high-performing fresh substate simulation points that achieved approximately 95% system O₂ recovery. The transient simulation was run for 250 hr of simulation time. Figure 6 plots the CVD CH₄ conversion (Figure 6a), CVD effluent C₂H₂ mole fraction (Figure 6b), and CVD effluent composition (Figure 6c) versus the simulated time on stream. For their brassboard testing, Honeywell's stopping criteria that dictate when to end a run were (1) CH₄ conversion below 50%, (2) estimated density of any substrate layer exceeding 1.8 g/cm³, or (3) C₂H₂ concentration in the CVD effluent greater than 1 mol%.⁹ These criteria were established to provide sufficient H₂ to sustain the Sabatier reactor and to avoid risk of soot generation, which tends to be preceded by an increasing C₂H₂ concentration.

As shown in Figure 6a, the substrate starts at a lower CH₄ conversion in the integrated SDU and CVD reactor system (58.5%) compared to the standalone CVD reactor (68.5%) due to the excess H₂ and other byproducts that are fed along with CH₄ to the CVD reactor. The decreasing CH₄ conversion of the integrated system with time follows a similar trend as the standalone system but is offset by the difference in the initial conversion between the two configurations. The integrated system reaches the 50% CH₄ conversion stop criterion after approximately 110 hr compared to 275 hr for the standalone CVD reactor. Despite the uncertainty that may be present in the simulation (as will be discussed in Section IV.C), the results suggest an integrated system consisting of the Sabatier and CVD reactors will lead to a significant decrease in the maintenance interval, which is accompanied by an increase in the consumable substrate mass.

Furthermore, the integrated system shows higher C_2H_2 concentration in its CVD effluent compared to the standalone CVD reactor (Figure 6b), which appears to be commensurate with a rapidly increasing ethylene and C6+

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concentrations (Figure 6c). The high concentration of C6+ species in the CVD effluent is indicative of soot precursor formation and suggests that operations past the 110 hr time on stream may produce a significant amount of the undesirable soot product that may be entrained within the gas stream. Thus, although system integration aids in achieving high system O_2 recovery (> 95%), it may also significantly reduce the maintenance interval and increase the necessary consumables mass for the system.



Figure 6. Plots of the (a) CH₄ conversion and (b) CVD effluent acetylene mole fraction versus time for the standalone and integrated systems as well as the (c) integrated system's CVD effluent composition versus time.

Figure 7 shows the integrated system O_2 recovery versus time. Interestingly, despite the significant reduction in the CVD reactor's CH₄ conversion and reaching the reactor stop criteria after 110 hr, the system O_2 recovery sees a much smaller reduction from 95.4% to 93.2%. After 250 hr, the system recovery decreases much more significantly to 85.3%. Therefore, within its operational bounds up until 110 hr, the simulation suggests that the integrated system is able to maintain a high level of performance. Thus, the major detriment of SDU and CVD reactor integration is the significant reduction in the maintenance interval.

Next, Figure 8 provides insights into the some of the more specific tracked parameters within the simulation including the mixture H_2 :CO₂ (Figure 8a) and recycle (or CVD effluent) composition of the CH₄, CO₂, and H₂



Figure 7. Integrated system O₂ recovery over time.

(Figure 8b). Immediately upon simulation start, the mixture H_2 :CO₂ continuously decreases. The decrease in the mixture H_2 :CO₂ is a result of the CVD reactor substrate dynamics. As the solid carbon deposits and densifies within the substrate, the CVD reactor conversion decreases, and hence, its H_2 generation rate also decreases. The lower H_2

generation rate reduces the H_2 : CO_2 at the SDU inlet after mixing the system feed with the recycled CVD effluent. In the real system, if it is desired to maintain a constant mixture H_2 : CO_2 , then a supplemental H_2 will need to be fed that adjusts itself according to the recycle flow rate and composition. These simulation results indicate the potential need for a more sophisticated and precise flow control scheme.

The result of the lower mixture H_2 :CO₂ into the SDU is reflected in the recycle (or CVD effluent) composition, which shows a decrease in the CH₄ and H₂ concentrations and a significant increase in the unreacted CO₂ concentration over time. Therefore, the simulations show how dynamics in one of the reactor systems can lead to intensified effects in the other reactor system that may worsen over time if left unchecked. The results from this analysis provide valuable insight into how operations of an integrated CO₂ system may differ from the expectations from standalone testing.



Figure 8. Plots of the (a) mixture H₂:CO₂, (b) recycle flow rate, and (c) recycle stream composition versus time for the integrated system.

C. Analysis Sources of Uncertainty

This analysis of the integrated Sabatier and CVD reactor is not without its sources of uncertainty, including the discrepancy between the model and brassboard reactor testing with regard to H_2 inhibition, the lack of byproduct reactions in the Sabatier reactor model, and the loosely heuristic nature of the stop criterion for the CVD reactor. This section intends to discuss these uncertainties and their expected effect on the quantitative accuracy of the analysis results. However, it is believed that the qualitative trends predicted by the models should hold.

As shown in Chen, *et al.* (2024), the CVD reactor model, which is based on literature kinetic parameters, differs from the brassboard CVD reactor testing in that it does include an inhibitory effect of H_2 on conversion, which was not observed during testing. H_2 inhibition was expected based on literature observations of carbon vapor infiltration systems where H_2 is believed to block the active sites for deposition.¹⁹⁻²² However, the brassboard CVD reactor experiments showed no such H_2 inhibition effect. This difference between the CVD reactor model and brassboard CVD reactor may cause the CVD model to underpredict the CH₄ conversion as the H_2 concentration in the feed increases. Therefore, the model may predict reaching the stop criterion of a minimum CVD reactor conversion of 50% conversion sooner than reality. This would effectively cause underprediction of the maintenance interval and overprediction of the consumable substrate mass required.

Additionally, the Sabatier reactor model excludes the consideration of byproduct (e.g., C_2 hydrocarbons and larger) reactions since it only considers the CH₄ steam reforming and H₂O gas shift reactions. Some of the hydrocarbon byproducts of CH₄ pyrolysis may be consumed within the Sabatier reactor via steam reformation. Future updates to the integrated model will consider the C₂H₂ steam reforming reactions within the Sabatier reactor. However, the current model may underpredict H₂ production as a result of steam reformation of byproduct hydrocarbons (i.e., C₂H₂ and ethane), albeit in relatively small amounts (< 2 mol%), and overpredict accumulation of the major byproduct C₂H₂.

The last source of uncertainty for our conclusions is in the stop criterion (i.e., minimum 50% conversion for the CVD reactor). This criterion is based on a loose heuristic that too much CH_4 at the inlet to the Sabatier will prevent light off. However, there has been no experimental verification of this criterion. This 50% conversion criterion may be more thoroughly assessed for appropriateness and to understand the feasibility of operating at lower conversion. By being able to operate at below 50% conversion, the time interval between substrate changeout would increase. This additional consideration, however, does not affect the comparison between the standalone and integrated case

made herein since both use the same stop criterion. Overall, these sources of uncertainty may suggest using these modeling results to elucidate qualitative trends in system performance.

V. Conclusion

In this work, a model of the integrated Sabatier (i.e., SDU) and CVD reactor systems was developed to assess the effects of system integration on their combined performance and compared to their standalone reactor performances. The integrated SDU and CVD reactor model development leveraged existing models of the individual SDU and CVD reactors. The use of modeling and simulation to support technology development by providing a priori insights into these hardware integrations is a powerful tool in that it is more flexible than testing real hardware, allows for rapid assessment of different operating conditions and configurations, and requires fewer resources to implement.

Simulation and analysis of the integrated SDU and CVD reactor system with fresh substrate at steady state showed that recycling the CVD effluent improves the system O_2 recovery, with > 95% O_2 recovery being achievable while reducing the necessary H_2 : CO_2 into the system. However, the system integration leads to a detriment of the individual reactor performances with decreasing SDU and CVD reactor conversions as the recycle flow rate of CVD effluent increases. This reduction in individual reactor performance is, in part, due to the increased amount of byproducts and the dilution effects they impart on the reaction kinetics.

Transient simulation and analysis of the integrated SDU and CVD reactor system were used to show how the dynamically changing CVD reactor, whose substrate densifies over time due to the depositing carbon, leads to worsening system performance over time, with a reduction in the system O_2 recovery from 95.4% to 93.2% over 110 hr and a further reduction to 85.3% over 250 hr. Albeit, the system performance reduction is rather small, the larger effect on operations may come from a reduction in the maintenance interval of the integrated reactor system. The transient simulation showed the integrated system reached its below 50% CH₄ conversion stop criteria after 110 hr, which is a shorter maintenance interval compared to that from the standalone brassboard reactor testing. These simulation results suggest that, despite the sources of uncertainty, the integrated Sabatier and CVD reactor systems would lead to a shortening of the maintenance interval and hence an increase in the consumable substrate mass that would be needed on a long-duration exploration mission. The simulation results also indicate that the CVD reactor dynamics lead to decreasing H₂:CO₂ into the SDU for the closed-loop system, which may necessitate supplemental H₂ feed and a precise flow control system if it is desired to maintain a constant mixture H₂:CO₂ over the course of a run.

Note, however, that the integrated SDU and CVD model is not without its sources of uncertainty, which may skew the simulation results. In particular, some model aspects that may be updated to improve its fidelity include a more complete definition of the CO_2 and CO interconversions within the CVD reactor model, fine tuning the H_2 inhibition effects in the CVD reactor to achieve agreement between the model and brassboard reactor, expanding the Sabatier reactor's kinetic model to include some of the major byproducts (e.g., C_2H_2), and reconsideration or further assessment of the operational stop criterion for the system. Model advancement may also be supported by integrated Sabatier and CVD reactor testing where the test data can be used to improve the models in an iterative model development cycle. Despite these uncertainties, the integrated model of the SDU and CVD reactor system has provided valuable insight that may be beneficial for test planning and operations of the real-world counterparts.

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