



National Aeronautics and Space Administration

Ames Research Center • Entry Systems and Technology Division

Thermochemically-Closed Sonic-Flow Inversion for Enthalpy and Temperature in Multispecies Arc-Jet Flows

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CREATED AT NASA AMES RESEARCH CENTER, MOFFETT FIELD, CALIFORNIA 94035-1000.

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TP-03, ABLATION I AND AEROTHERMODYNAMICS III (PART 1).

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Motivation

Why should we care about bulk enthalpy?



Bulk enthalpy is used to characterize the total energy content upstream of a nozzle

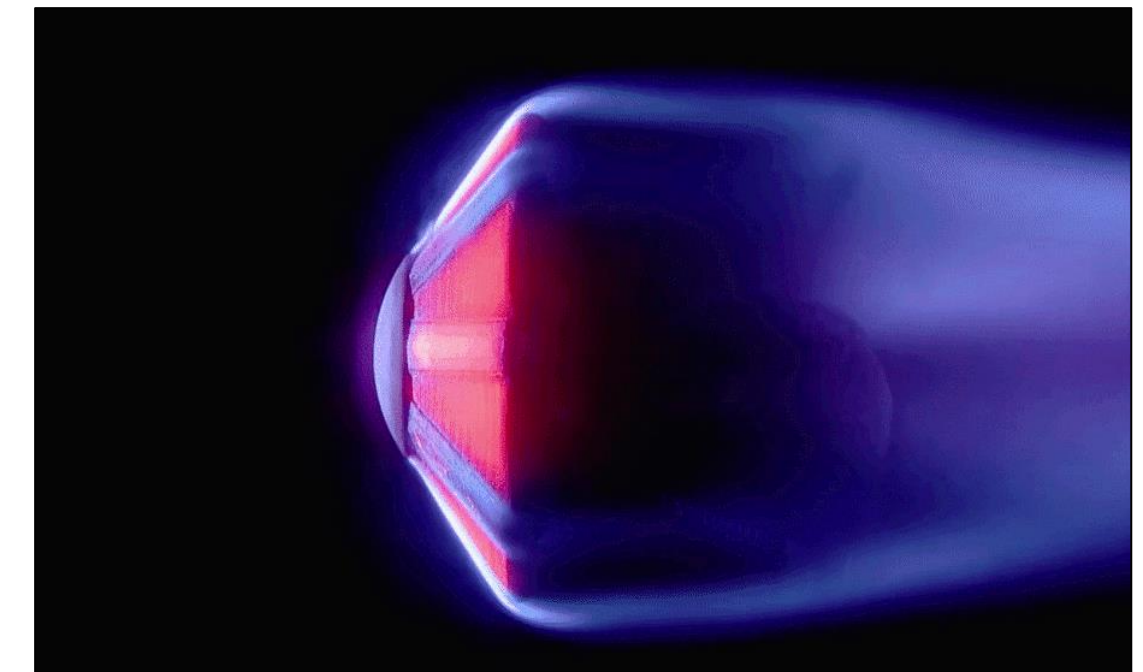
From an experimental point of view:

- Key for establishing ground-to-flight similarity as it links an arc-jet's total energy content to an equivalent kinetic-energy velocity scale:

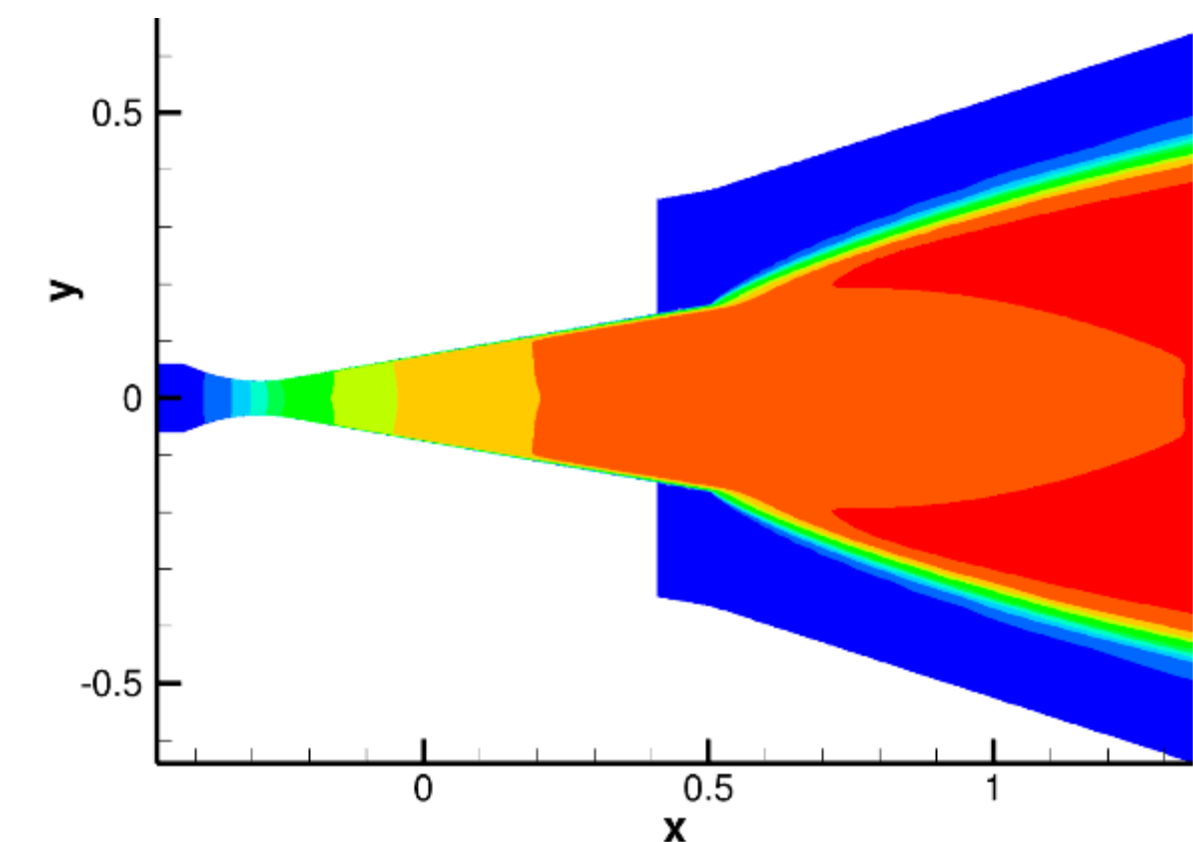
$$h_{\text{bulk}} \cong \frac{v_{\infty}^2}{2}$$

From a numerical point of view:

- Experimental estimates for h_{bulk} are used as primary in-flow boundary conditions for CFD simulations
- This input affects the accuracy of numerical predictions (e.g., $\dot{q}_0$, p_0), directly impacting any design/engineering decision made thereafter



ADEPT arc-jet test – 60 MW IHF
Credit: NASA/TSF



Palmer et al., AIAA Aviation 2025

Bulk enthalpy quantification techniques

Bulk enthalpy is not directly measured as a primitive thermodynamic state variable

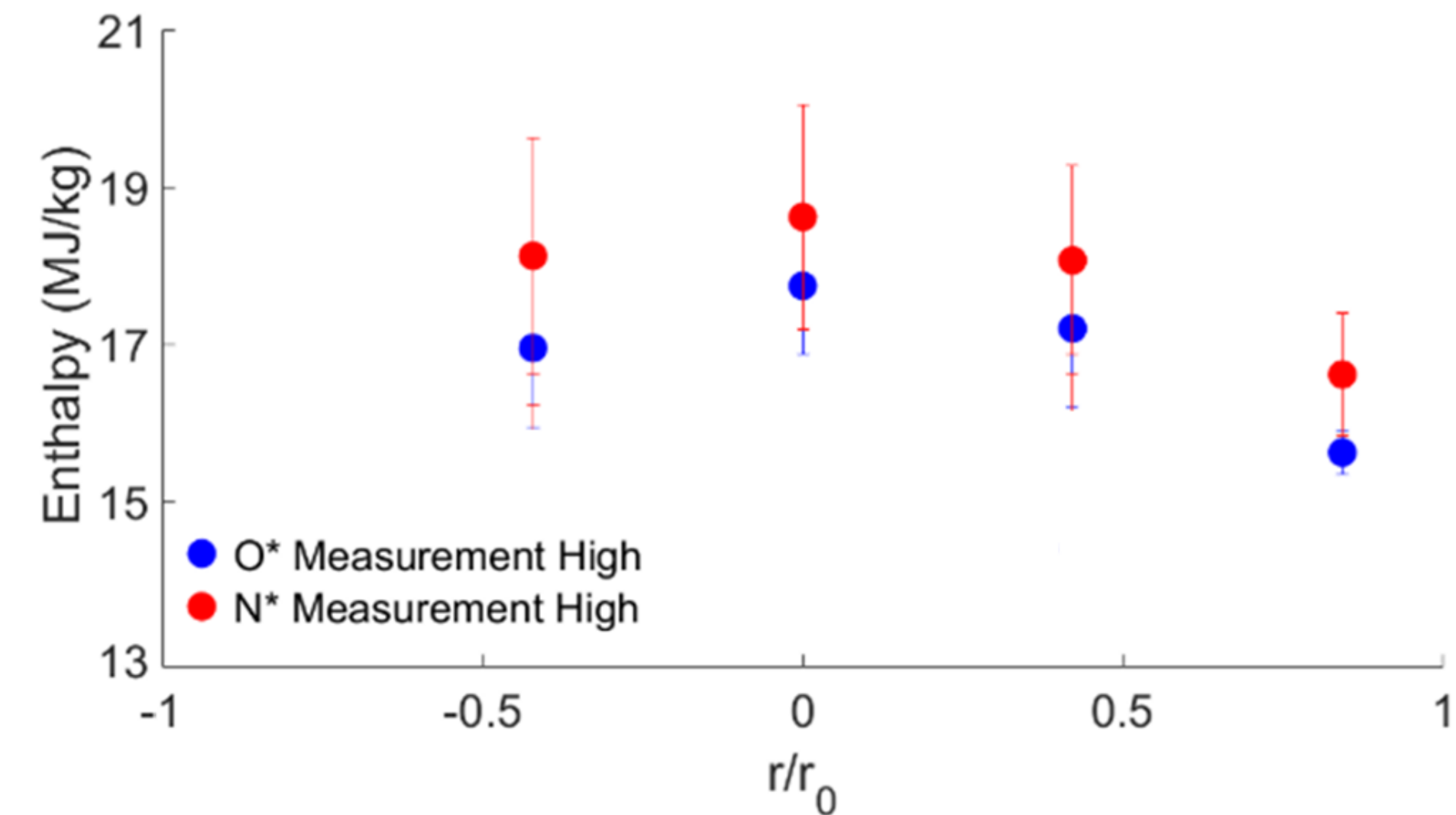
Techniques:

1. Optical & laser diagnostics
2. Energy balance
3. Sonic-flow

1. Optical & laser diagnostics



- High-frequency, non-intrusive local measurements
- Detailed insight into challenging environments (e.g., absorption, fluorescence, scattering)
- To obtain bulk properties from local measurements:
 1. Assumptions about the flow (e.g., axisymmetry)
 2. Meticulous uncertainty quantification and data analysis



TDLAS: Finch et al., AIAA SciTech 2023

Pros	Cons
1. Resolves radial structure 2. Provides local thermochemical information 3. High-frequency data acquisition possible 4. Non-intrusive	1. Requires optical access 2. Non-trivial optical setup and alignment 3. Optical setup may be species/gas dependent 4. Enthalpy can vary by diagnostic/tagged species

2. Energy balance

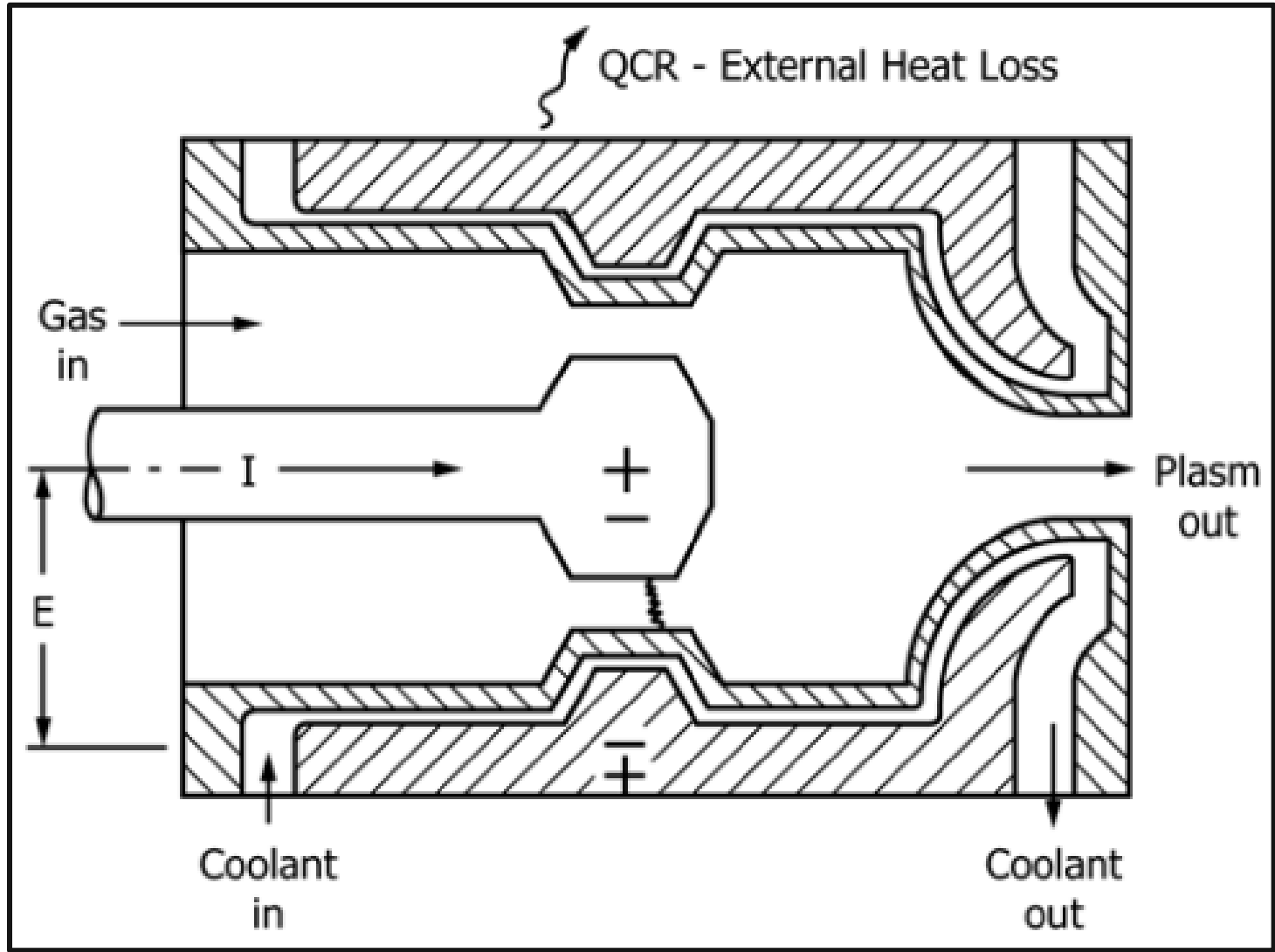
ASTM E341-08 standard



Also known as “enthalpy-by-energy balance” or “EB²” based on:

- 1. Energy IN: current and voltage supply
(minus)
- 2. Energy OUT: ΔT from the cooling of heater components (cathode, disks, anode, and nozzle)

$$h_{\text{bulk}} = \frac{P_{\text{arc}} - \Delta P_{\text{cool}}}{\dot{m}_{\text{test gas}}}$$



ASTM E341-08 standard 2020

Pros	Cons
1. Simple implementation in small arc-jet facilities 2. Gas-agnostic 3. Non-intrusive	1. Difficult implementation in larger arc-jet facilities 2. Large uncertainty if ΔT is low

3. Sonic-flow



Developed over many decades based on:

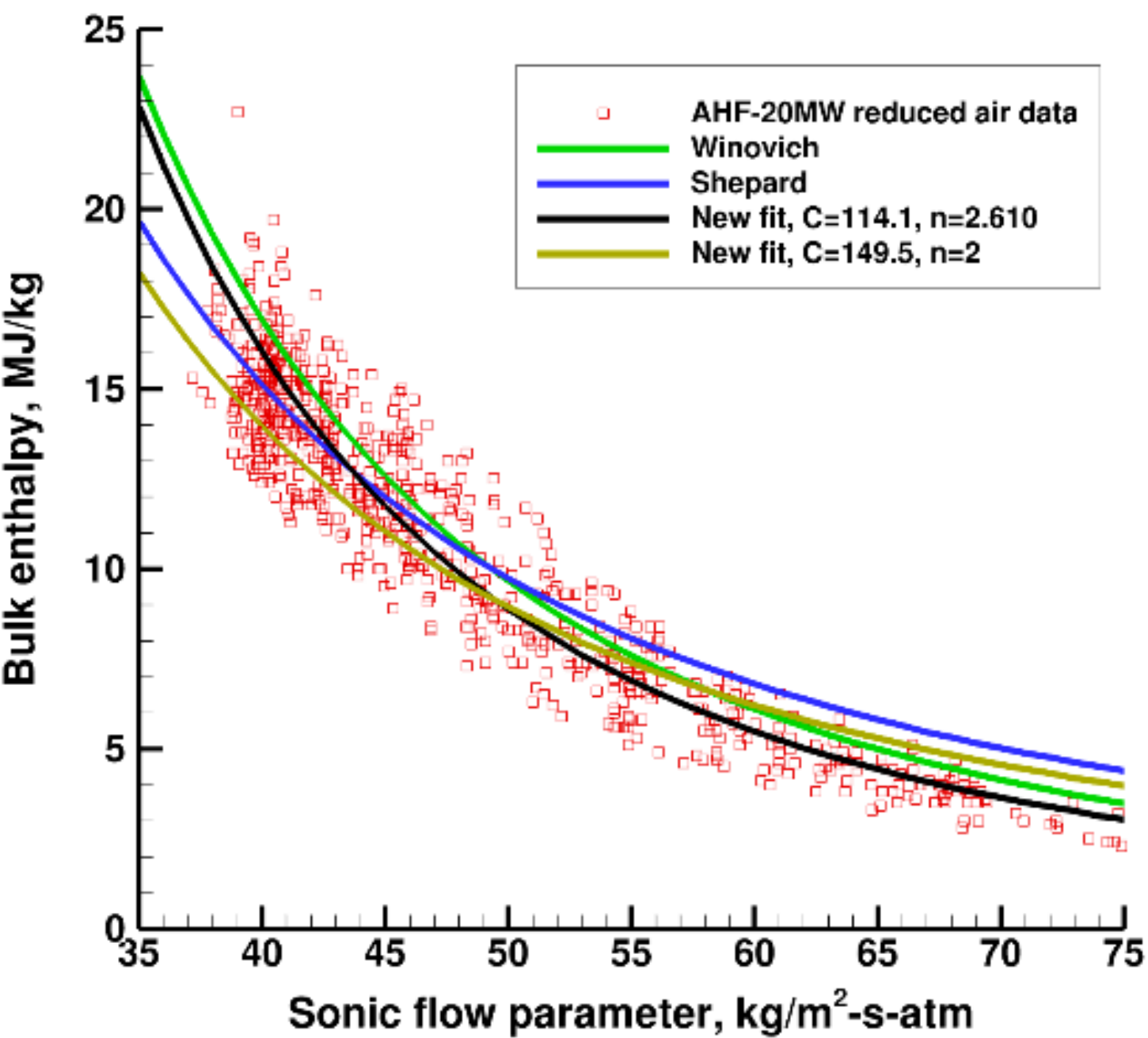
- Compressible flow theory (Winovich 1964)
- Numerical solutions (Shepard 1993)
- EB² data (Prabhu 2011; Palmer 2025)

Results can take three general forms:

$$h_{\text{bulk}} \equiv \left(\frac{C}{\sigma}\right)^2 \equiv \left(\frac{C}{\sigma}\right)^n \equiv \left(\frac{C + \sigma/\sigma_0}{\sigma}\right)^n$$

where the sonic-flow parameter σ is:

$$\sigma = \frac{\dot{m}_{\text{test gas}}}{p_0 A_t}$$



Palmer et al., AIAA Aviation 2025

Pros	Cons
1. Established in literature	1. Sensitive to scatter in EB ² data
2. Good initialization for numerical simulations	2. Gas-specific correlations required
3. Non-intrusive	3. Fit form / constraints affect results
4. Fast implementation once correlations are available	4. Limited extrapolation beyond calibration range

TSIF

Thermochemically-Closed Sonic-Flow Inversion Framework

TSIF: Framework Overview



TSIF is comprised of two steps:

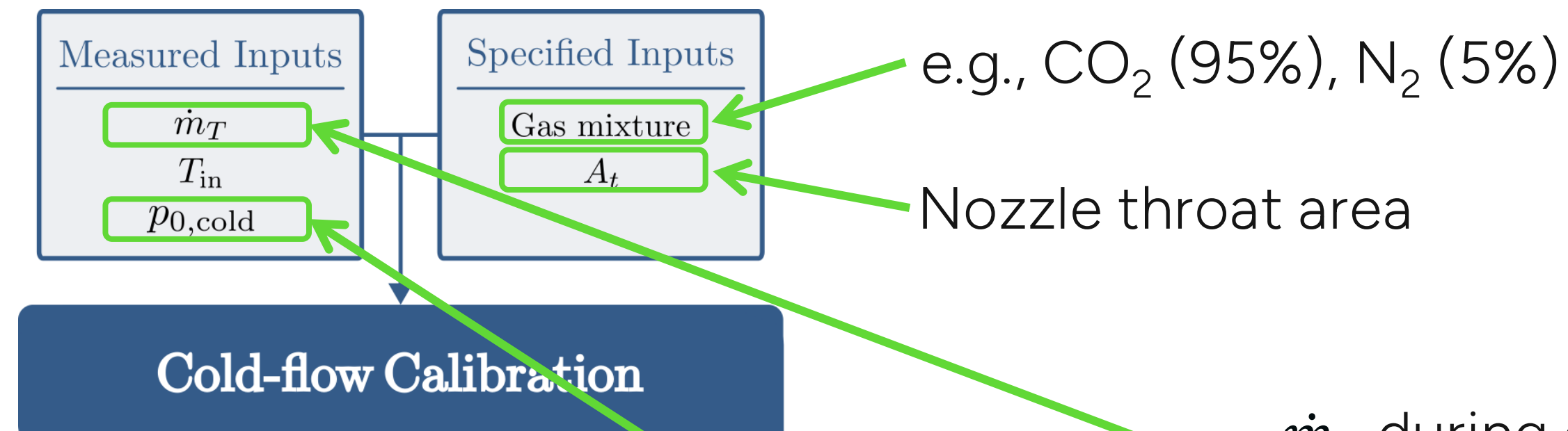
1. Cold-flow Calibration
2. Hot-flow Inversion

Cold-flow Calibration

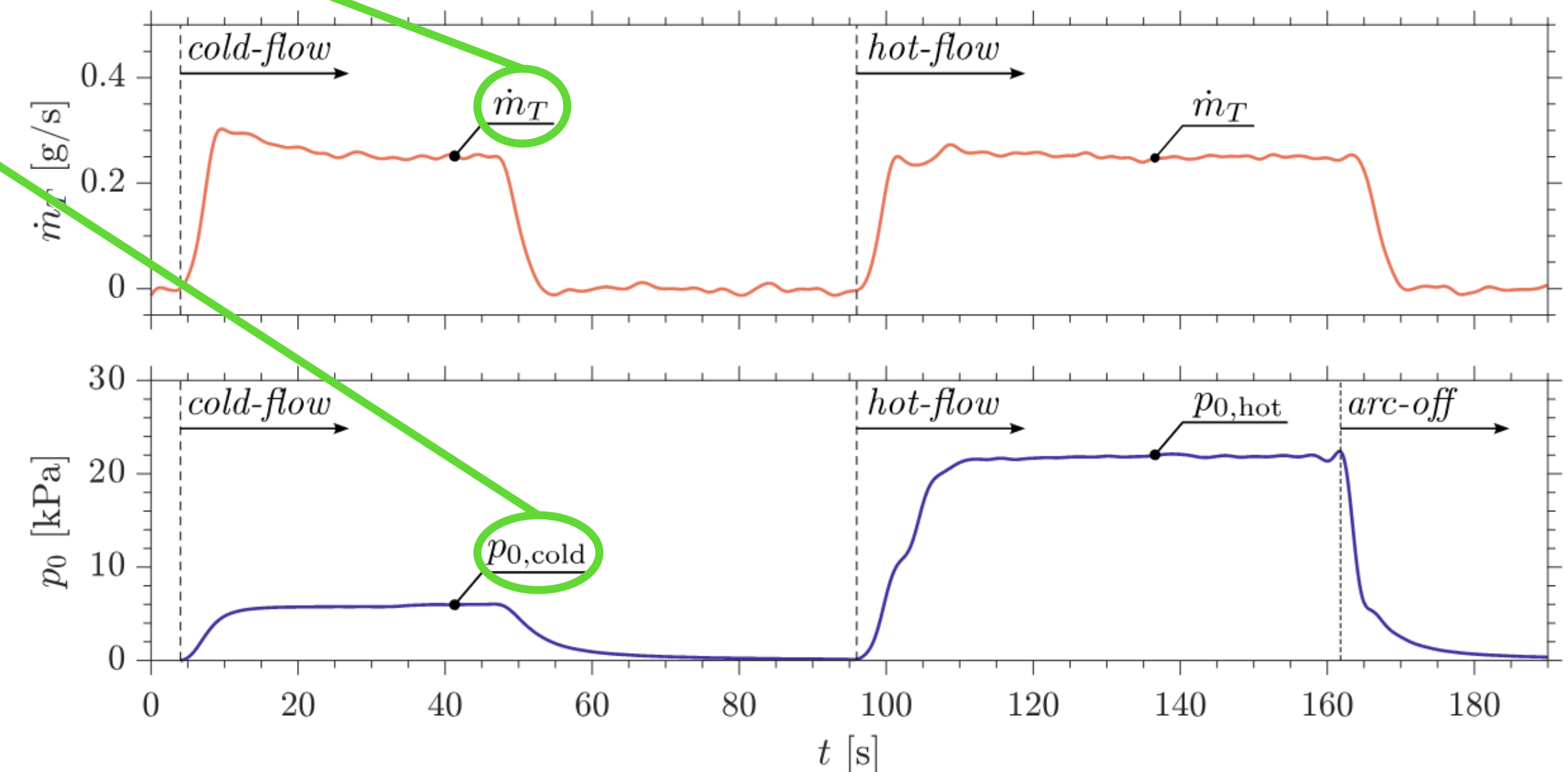
Hot-flow Inversion

Step 1: Cold-flow Calibration

Inputs

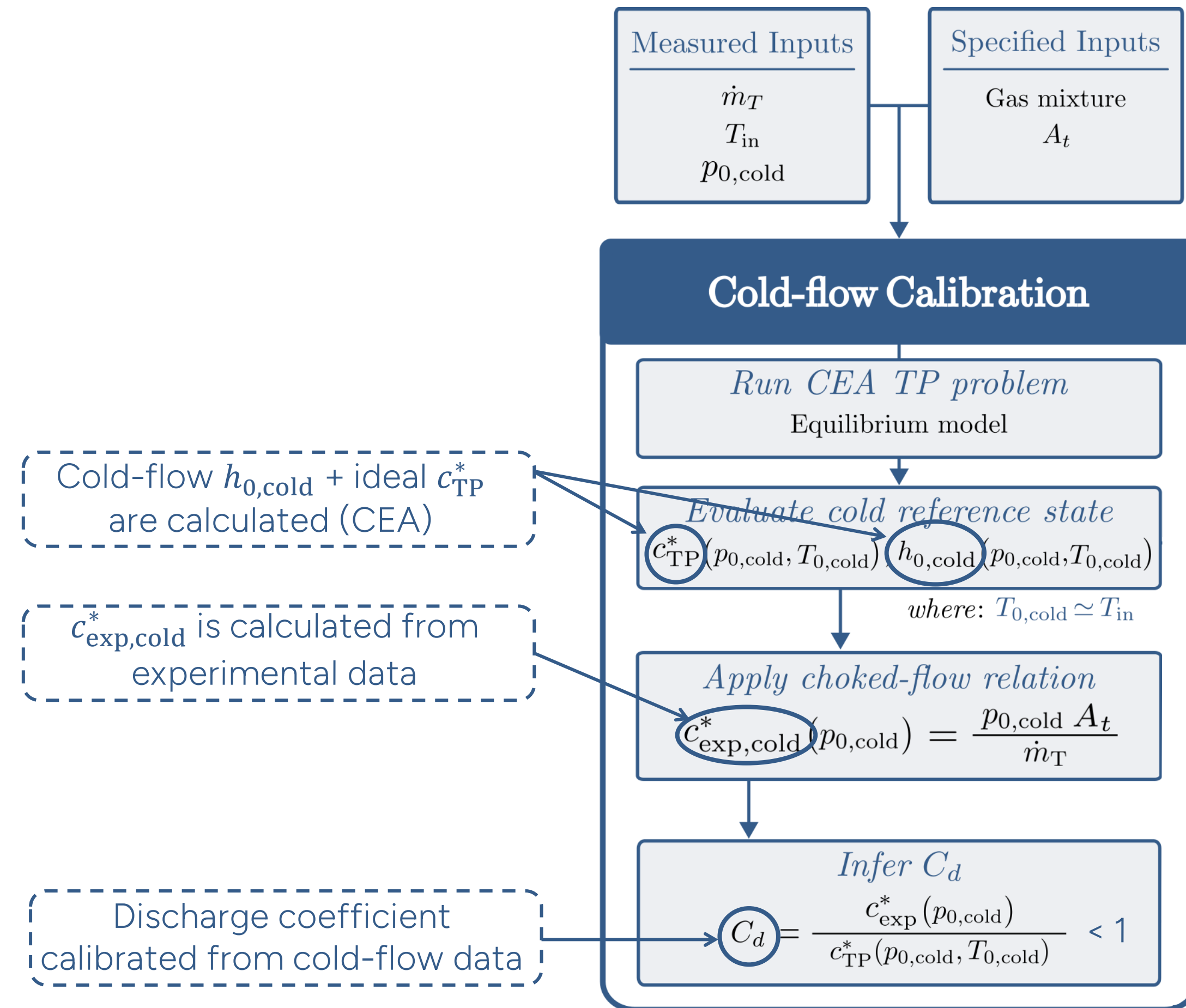


$\dot{m}_T$ during cold-flow and hot-flow must be the same
(cold-flow can be done before or after arc-on)



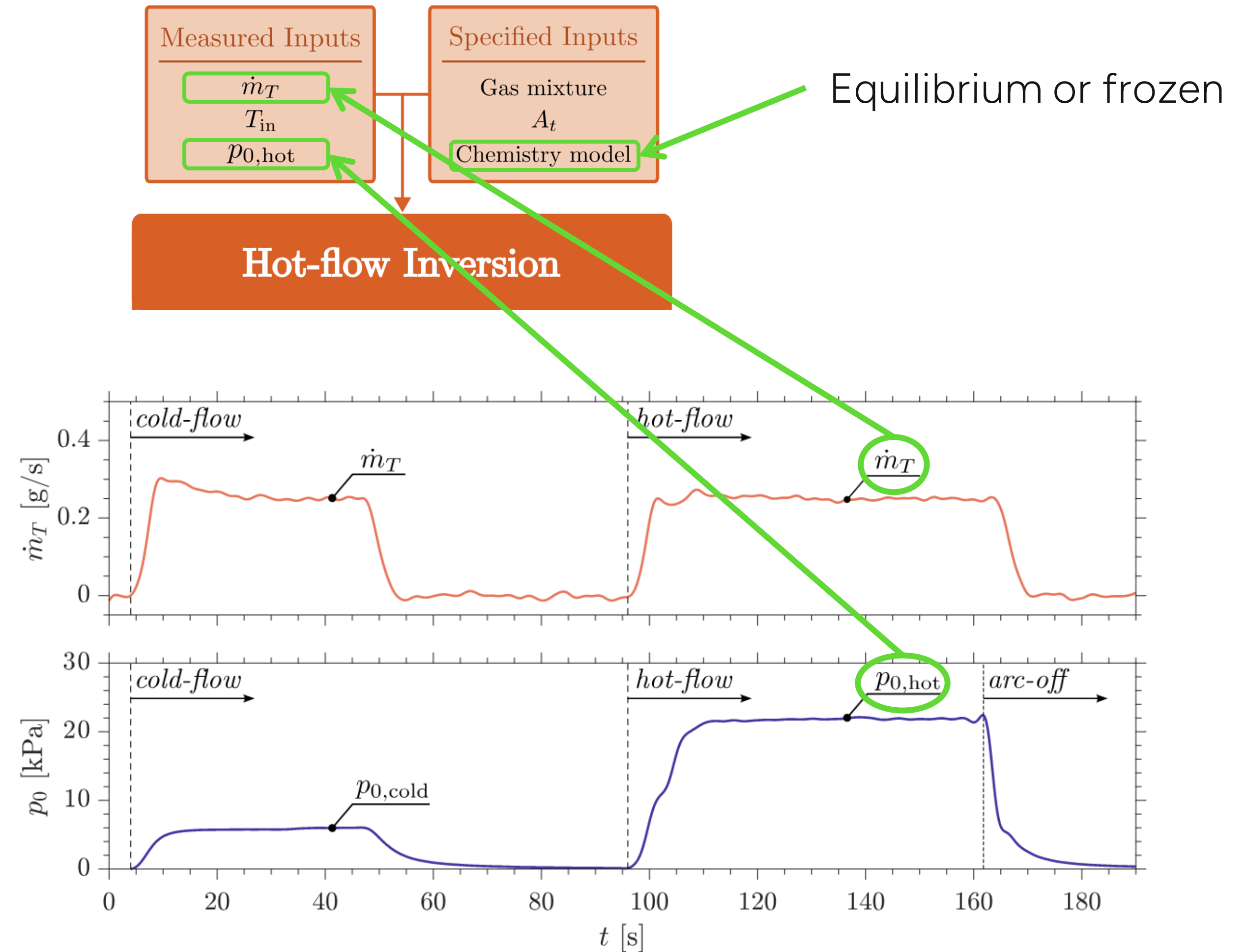
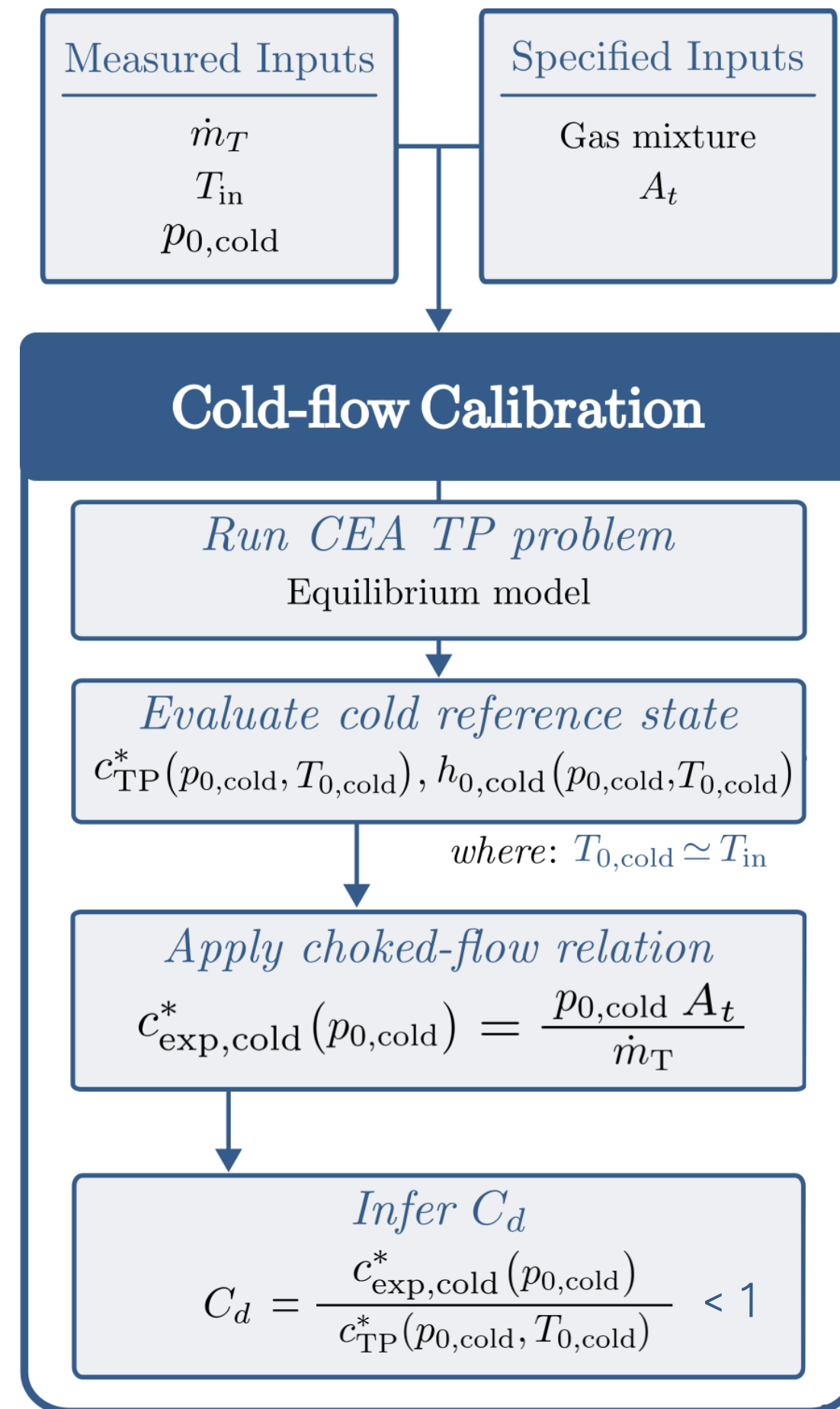
Step 1: Cold-flow Calibration

CEA equilibrium case using Temperature–Pressure (TP) problem



Step 2: Hot-flow Inversion

CEA using ROCKET problem



Step 2: Hot-flow Inversion

3D visualization of c^*-h_0 manifold (CO₂/N₂ mixture)



Hot-flow Inversion

Run CEA ROCKET problem

Equilibrium or Frozen model

Build ideal chemistry map

$c_{\text{ROCKET}}^*, h_s(p_{0,\text{hot}}, \{T_{0,k}\})$

INPUTS:

- measured/constant $p_{0,\text{hot}}$
- gridded T_0 ($1000 \leq T_0 \leq 12,000$ K)

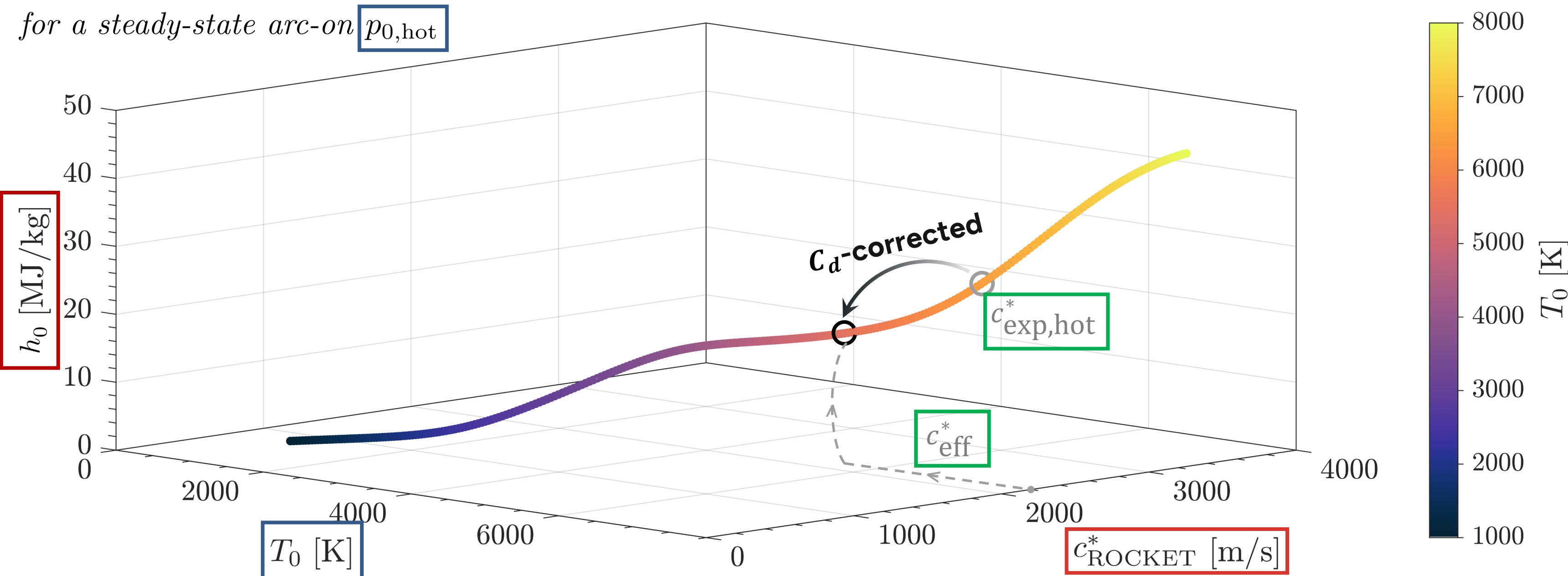
OUTPUTS:

- c_{ROCKET}^*
- h_0

FINAL STEPS:

- Compute $c_{\text{exp,hot}}^* = p_{0,\text{hot}} A_t / \dot{m}_T$
- Use C_d to correct $c_{\text{exp,hot}}^*$ to c_{eff}^*

for a steady-state arc-on $p_{0,\text{hot}}$



Step 2: Hot-flow Inversion

3D visualization of c^*-h_0 manifold (CO₂/N₂ mixture)



Hot-flow Inversion

Run CEA ROCKET problem

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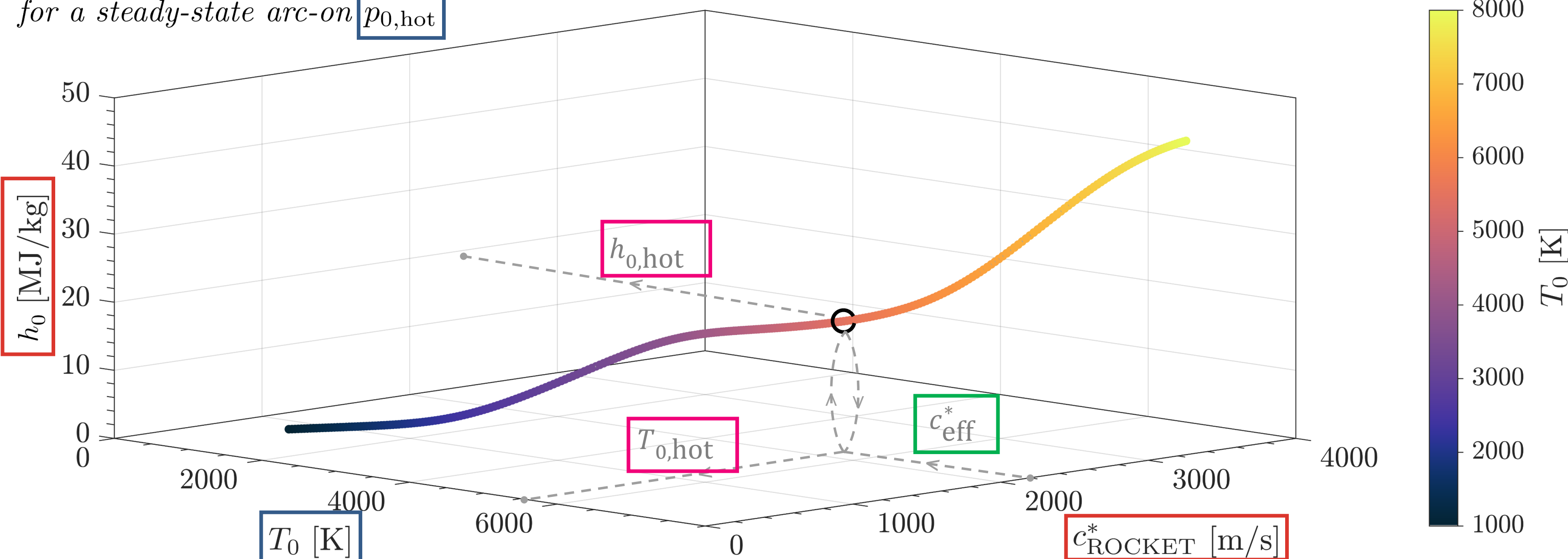
OUTPUTS:

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FINAL STEPS:

- Compute $c_{\text{exp,hot}}^* = p_{0,\text{hot}} A_t / \dot{m}_T$
- Use C_d to correct $c_{\text{exp,hot}}^*$ to c_{eff}^*
- Infer $h_{0,\text{hot}}$ and $T_{0,\text{hot}}$ by inverting at c_{eff}^* query point

for a steady-state arc-on $p_{0,\text{hot}}$



Uncertainty analysis

$$\left(\frac{\sigma_{c_{\text{eff}}^*}}{c_{\text{eff}}^*}\right)^2 \approx \left(\frac{\sigma_{C_d}}{C_d}\right)^2 + \left(\frac{\sigma_{p_{0,\text{hot}}}}{p_{0,\text{hot}}}\right)^2 + \left(\frac{\sigma_{A_t}}{A_t}\right)^2 + \left(\frac{\sigma_{\dot{m}_T}}{\dot{m}_T}\right)^2$$

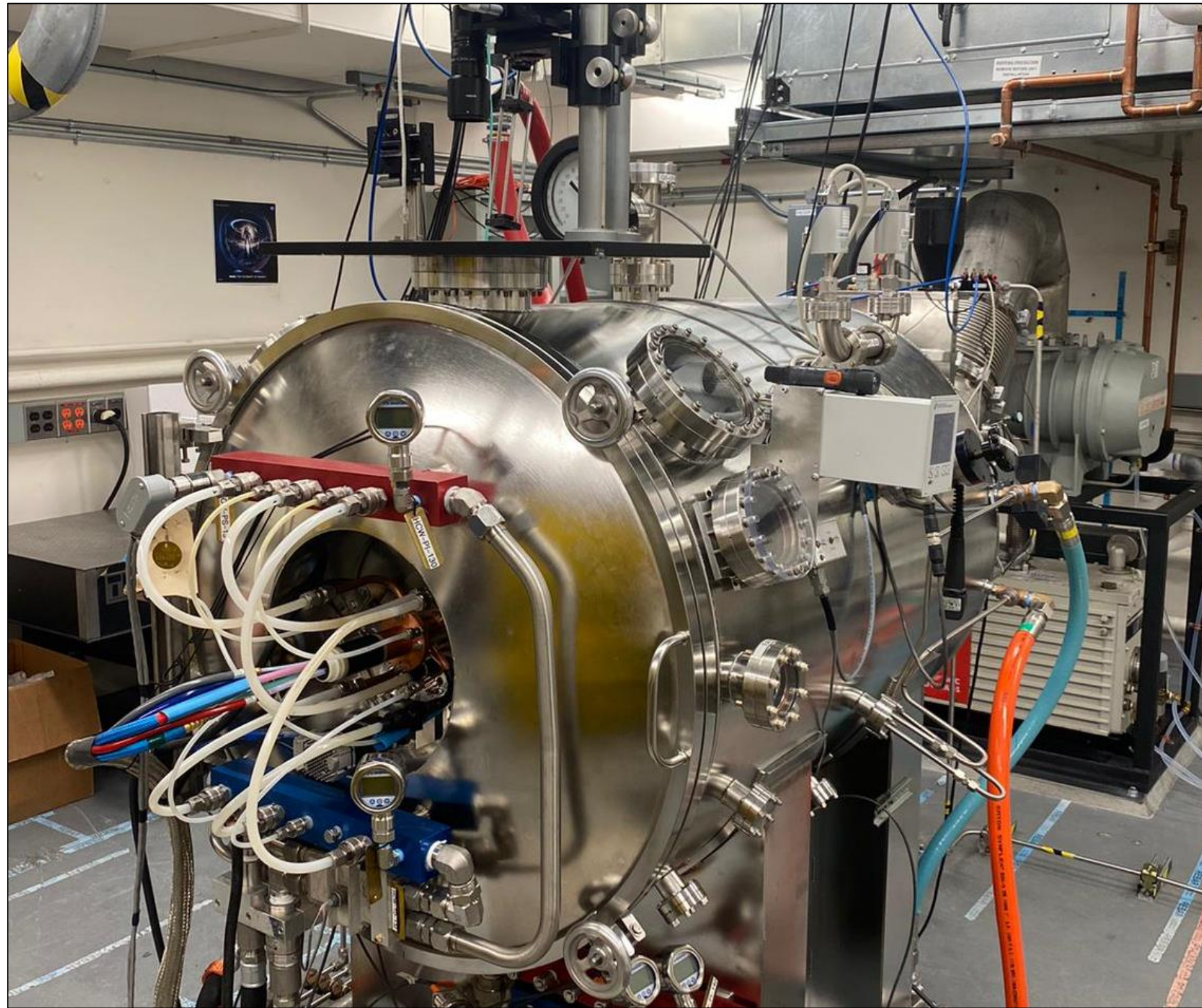
$$\sigma_{h_{0,\text{hot}}} \approx \left| \left(\frac{\partial h_0}{\partial c_{\text{ROCKET}}^*} \right)_{p_{0,\text{hot}}} \right| \sigma_{c_{\text{eff}}^*}$$

Bulk enthalpy

$$h_{\text{bulk}} = \Delta h_0 = h_{0,\text{hot}} - h_{0,\text{cold}}$$

Experimental details

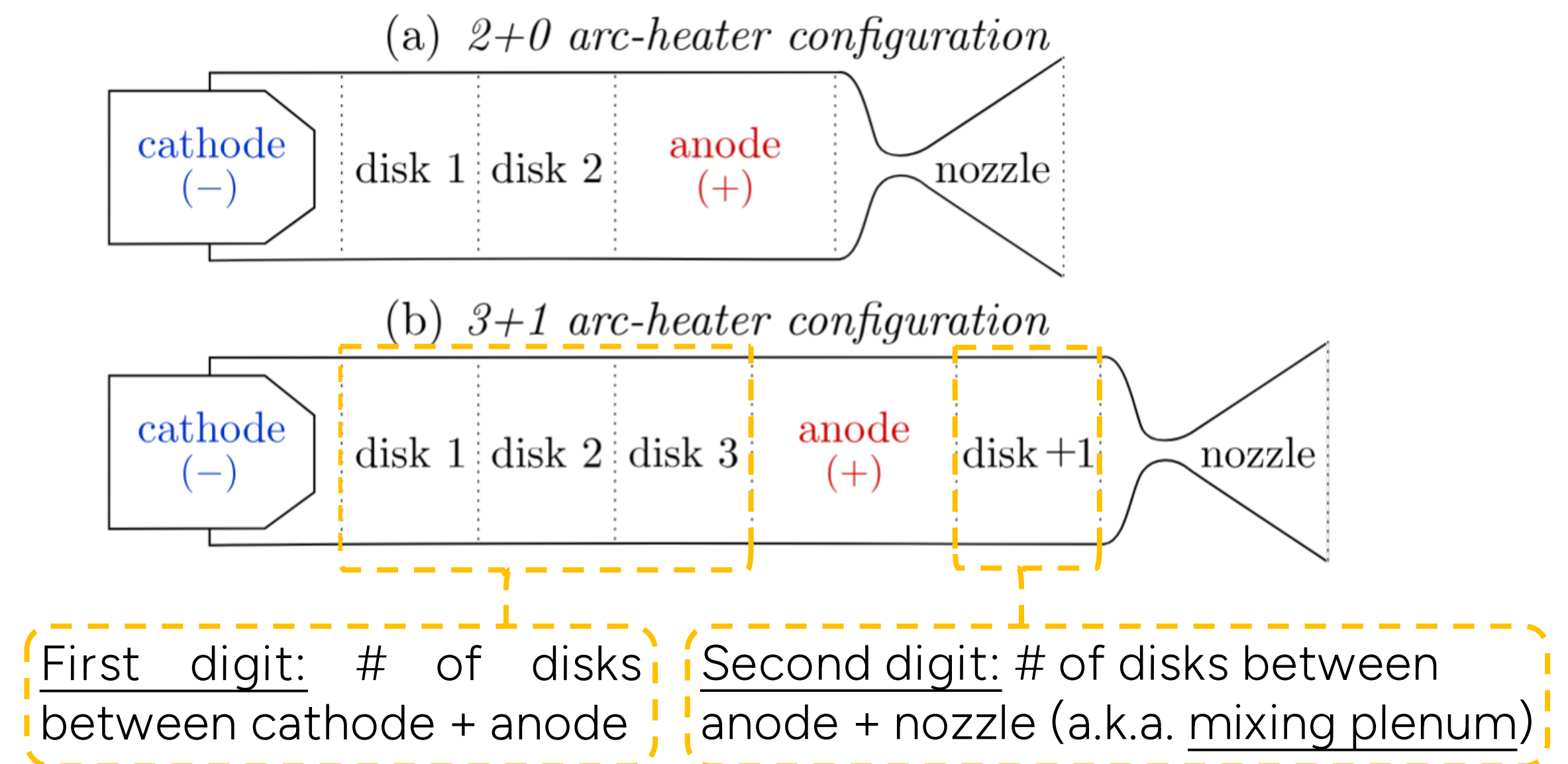
30 kW mARC II: segmented-constricted arc-jet



AIAA papers:

- MacDonald et. al, Aviation 2023, DOI: [10.2514/6.2023-3298](https://doi.org/10.2514/6.2023-3298)
- Rodrigues et. al, Aviation 2024, DOI: [10.2514/6.2024-3553](https://doi.org/10.2514/6.2024-3553)
- Rodrigues et. al, SciTech 2025, DOI: [10.2514/6.2025-2443](https://doi.org/10.2514/6.2025-2443)
- Palmer et al. Aviation 2025, DOI: [10.2514/6.2025-3480](https://doi.org/10.2514/6.2025-3480)
- Edwards et. al, Aviation 2026 (this conference)
- Rodrigues, Journal of Thermophysics and Heat Transfer, 2026 (accepted)

Two configurations: "2+0" and "3+1"



Test conditions:

- Test gases: CO_2/N_2 (95%/5%), Air (100%), N_2 (100%)
- Arc current: 40–200 A
- Total flow rate: 0.25–0.80 g/s

Results

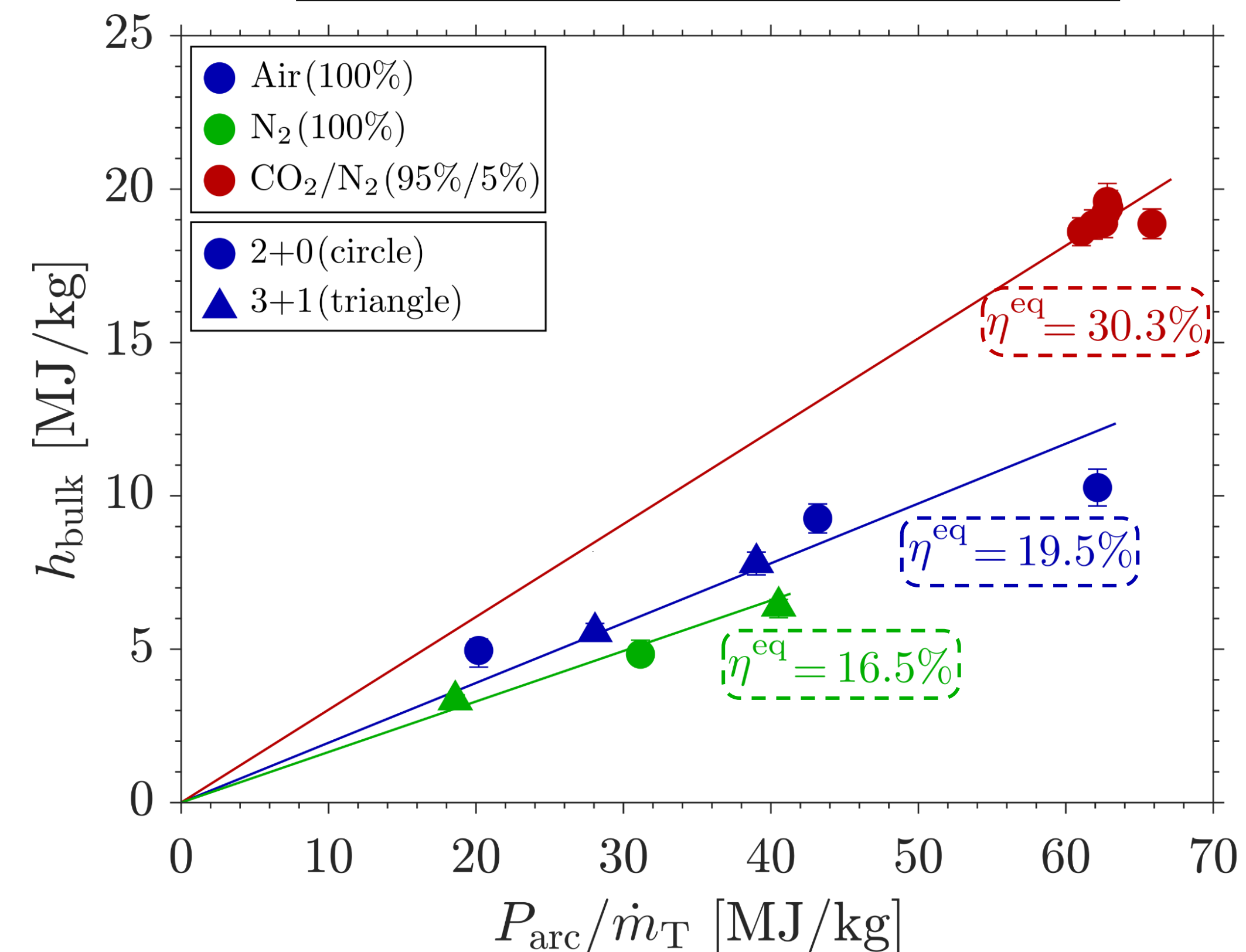
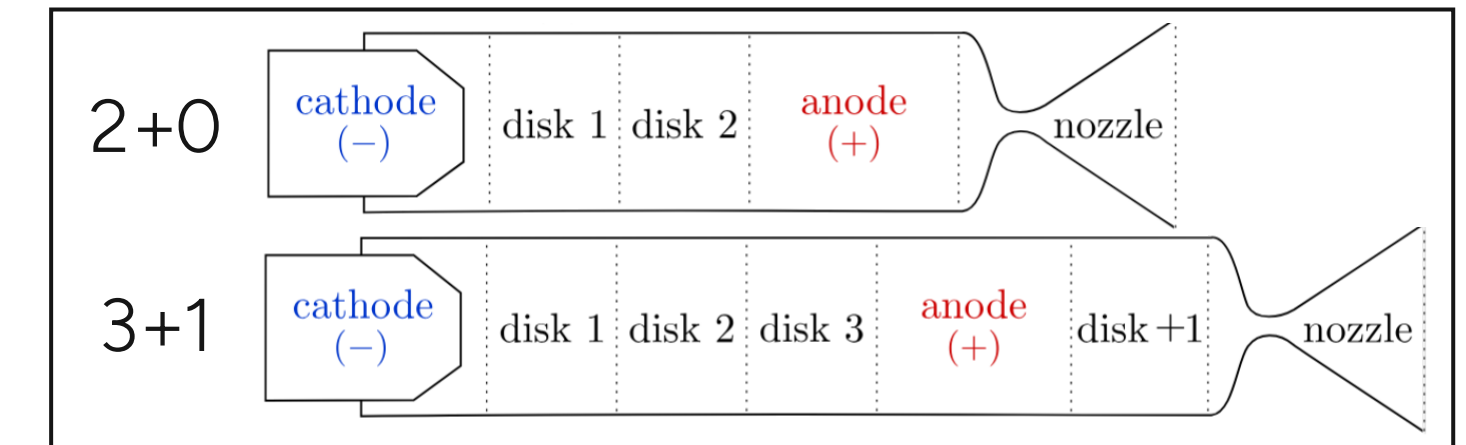
Inferred bulk enthalpy h_{bulk}



Bulk enthalpy increases ~linearly with specific energy input

- Gas dependence follows expected trends:
 $\text{CO}_2/\text{N}_2 > \text{air} > \text{N}_2$
- More data needed to assess linearity
- Assuming a linear scaling for each mixture enables inference of TSIF-based arc-jet efficiency:

$$h_{\text{bulk}} = \eta \left(\frac{P_{\text{arc}}}{\dot{m}_{\text{T}}} \right)$$

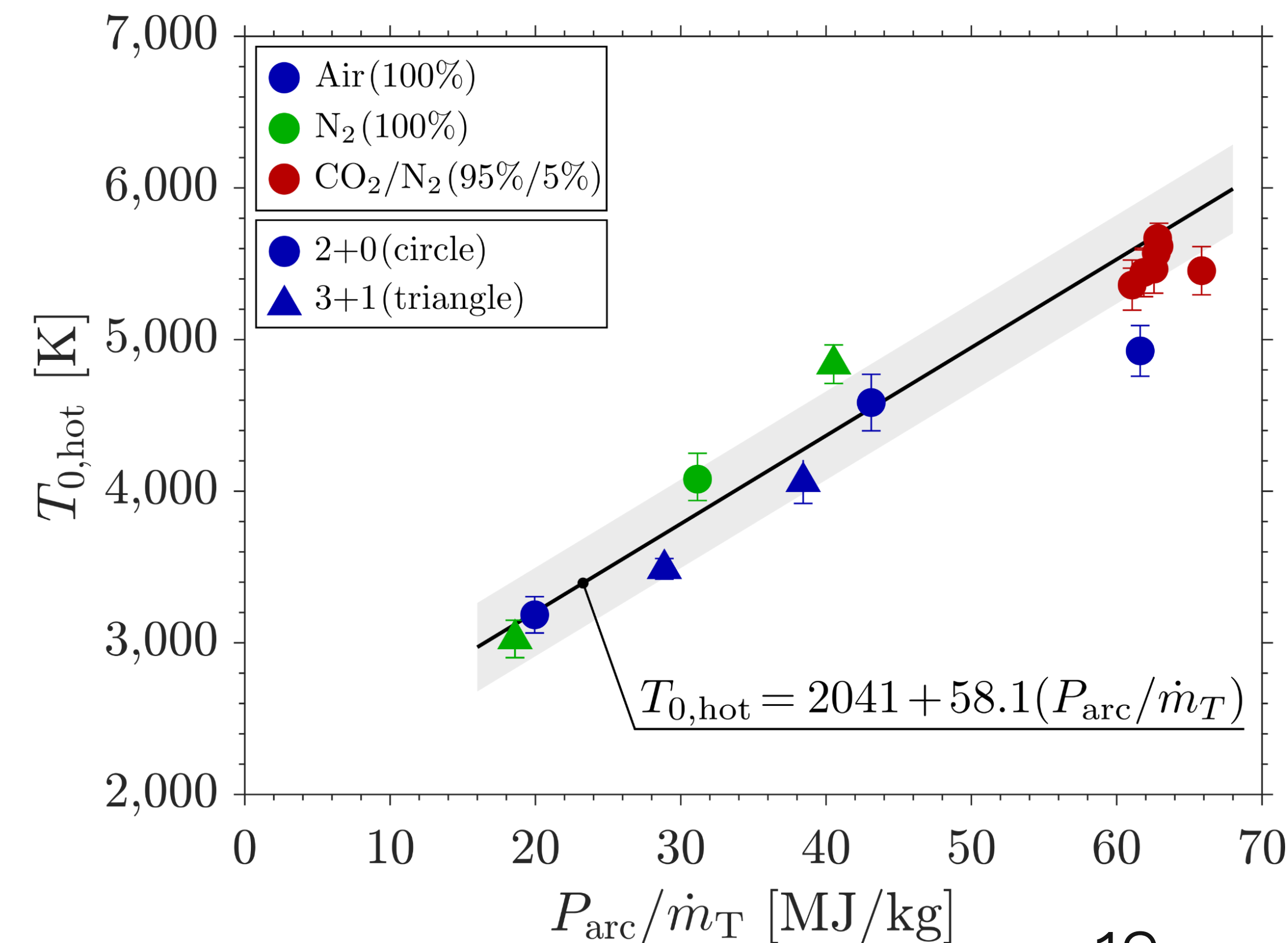
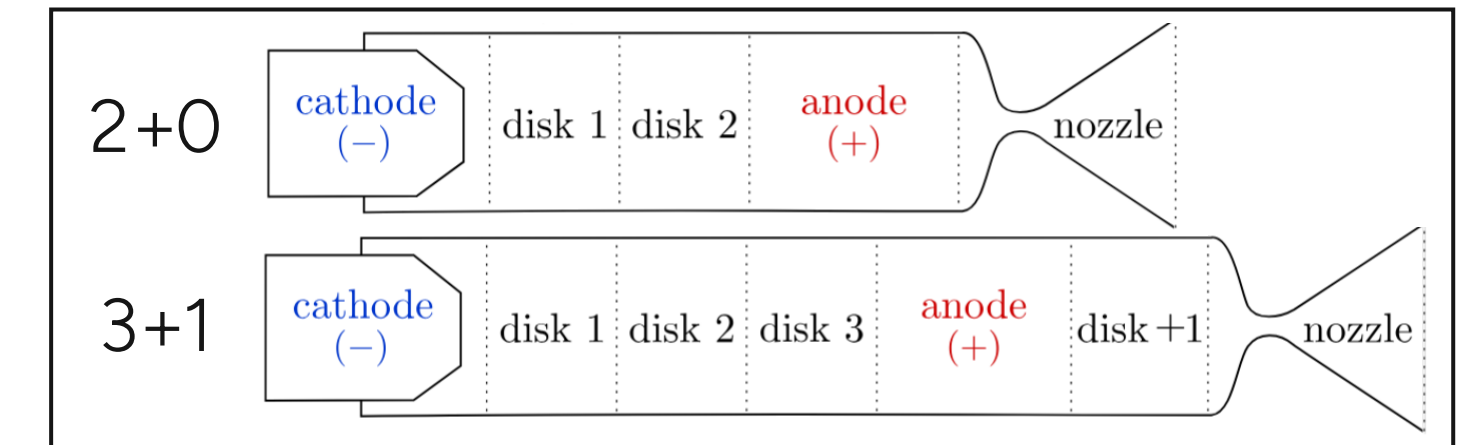


Inferred total temperature $T_{0,\text{hot}}$



Across the tested conditions, total temperature varies approximately linearly with specific energy input

- Data for air, N_2 , and CO_2/N_2 collapse reasonably well onto a common local trend
- The linear fit should be interpreted as a 'back-of-the-envelope' relation over the tested range, not a physical low-energy limit



Recovering the classic sonic-flow scaling for air

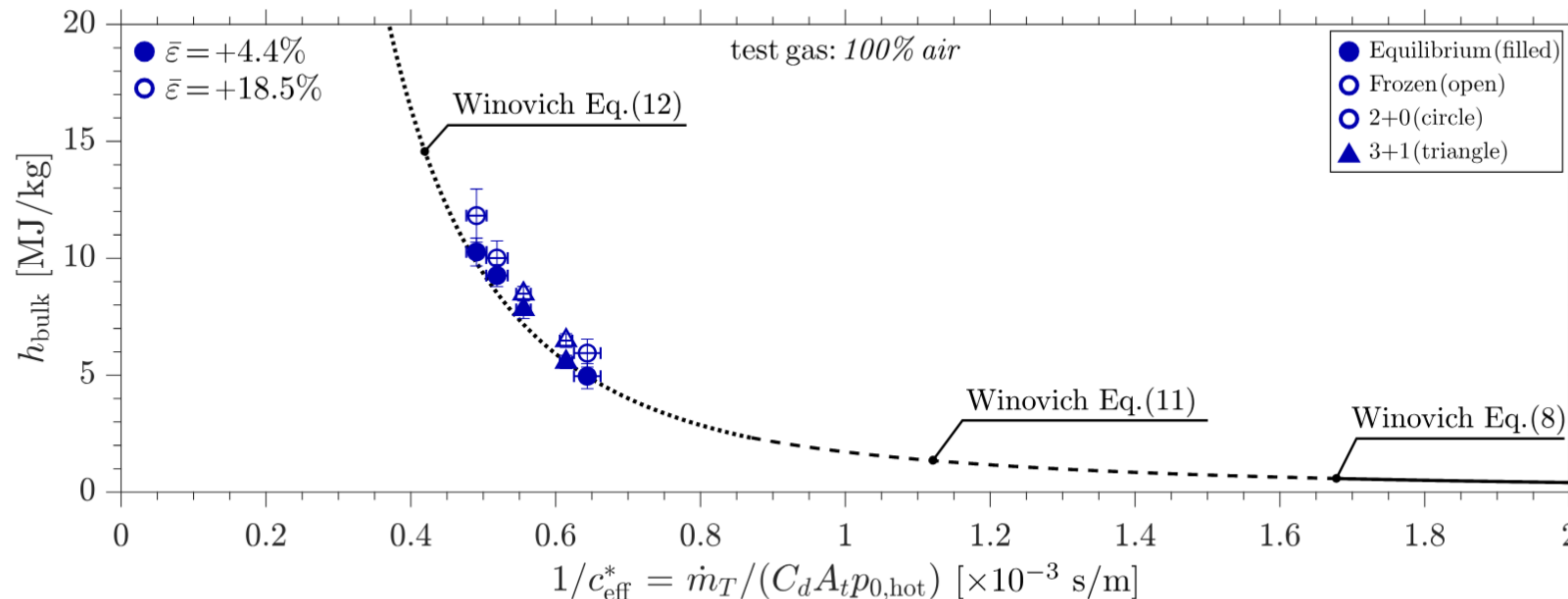


Frozen–air chemistry (○): produces a uniform upward shift (+18.5%):

- Suppressed recombination – higher bulk enthalpy is needed for the same choked flow constraint
- Highlights thermochemical effects rather than data scatter or framework error

Equilibrium–air chemistry (●): agrees with Winovich scaling (+4.4%):

- Winovich equilibrium behavior emerges naturally without empirical fitting
- Configuration (2+0 vs 3+1) does not affect the solution



Concluding Remarks

1. Framework:

- TSIF recasts Winovich's (1964) work by coupling first-principles compressible flow with an open-source thermochemical solver to infer h_{bulk} and T_0 in c^* space
- Mixture-flexible with selectable equilibrium/frozen chemistry models, clear uncertainty propagation, and explicitly quantified facility-specific discharge calibration

2. Physical consistency:

- Inferred thermochemical states follow expected trends
- Demonstrated sensitivity to multi-species gas composition and chemistry closure model
- Recovered Winovich's equilibrium-air scaling without any constraints

3. Practical impact:

- Only requires measured upstream quantities: total flow rates and total pressures!
- Can be applied to choked-flow facilities with readily available data at no added cost

1. Apply TSIF to other arc-jet facilities' datasets:
 - Please contact me to get bulk enthalpy estimates! (jocelino.rodriques@nasa.gov)
2. Compare TSIF with other enthalpy quantification methods:
 - ASTM E341 for upstream bulk enthalpy
 - ASTM E637 for downstream centerline enthalpy
3. Feed TSIF data into numerical simulations:
 - Analyze measured and predicted outputs – how do they compare?
4. Submit TSIF for balloting as new ASTM standard practice:
 - Chair of E21.08 Thermal Protection subcommittee suggested idea
 - Submitted work item — currently being drafted before going to ballot

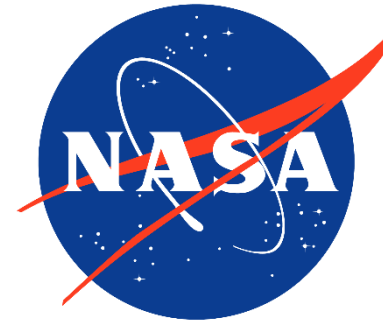
Customer testing



mARC II has officially begun customer testing this year (SpaceX, Blue Origin)

If you are interested in low-cost ground-testing at Ames, please contact our POC for more details:

National Aeronautics and Space Administration



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Thermophysics Facilities Branch (TSF)
Ames Research Center

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 650-448-4089



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- TS leadership: Peter Race, Scott Eddleman, Dr. David Hash

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- CEA software development team: Mark Leader et al. at NASA Glenn Research Center

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