TO:            KSI/Scientific & Technical Information Division  
               Attn: Miss Winnie M. Morgan  
FROM:          GP/Office of Assistant General Counsel for Patent Matters  
SUBJECT:        Announcement of NASA-Owned U.S. Patents in STAR

In accordance with the procedures agreed upon by Code GP and Code KSI, the attached NASA-owned U.S. Patent is being forwarded for abstracting and announcement in NASA STAR.

The following information is provided:

U.S. Patent No.                  : 3,801,617
Government or Corporate Employee : Hughes Aircraft Co.
Supplementary Corporate Source (if applicable) : Los Angeles, CA
NASA Patent Case No.             : MFS-22,411-1

NOTE - If this patent covers an invention made by a corporate employee of a NASA Contractor, the following is applicable:

YES [ ]               NO [ ]

Pursuant to Section 305(a) of the National Aeronautics and Space Act, the name of the Administrator of NASA appears on the first page of the patent; however, the name of the actual inventor (author) appears at the heading of column No. 1 of the Specification, following the words "with respect to an invention of ..."

Bonnie L. Woerner  
Enclosure
The invention is directed to thiophenyl ether disiloxanes and trisiloxanes embodying the formula

\[ \text{[3-(phenylthio)phenyl]-1,1,3,3-tetramethyldisiloxane} \]

where X is an atom of hydrogen or a halogen such as chlorine or bromine; an alkyl radical such as methyl, ethyl, propyl or the like; an aryl radical such as phenyl, biphenylyl, naphthyl, tolyl or xlyl; an aralkyl radical such as benzyl or phenylethyl or an alkaryl radical; n and n₂ are integers from 1 to 3, inclusive, and R₁ and R₂ are methyl or phenyl radicals. Compounds embodying the invention exhibit favorable properties for use as lubricants and functional fluids under severe conditions of temperature, radiation and vacuum.

BACKGROUND OF THE INVENTION

The invention described herein was made in the performance of work under a NASA contract and is subject to the provisions of Section 305 of the National Aeronautics and Space Act of 1958, Public Law 85-568 (72 Stat. 435, 42 U.S.C. 2457).

ABSTRACT OF THE DISCLOSURE

Thiophenyl ether disiloxanes and trisiloxanes useful as lubricant fluids have been found to result in a marked improvement in lubricity, while the siloxane units provide substantially reduced pour points.

It is therefore an object of this invention to provide improved lubricant and functional fluids for service over a wide range of temperatures.

Another object is to provide lubricant fluids that are resistant to nuclear and solar radiation.

Yet another object is to provide aromatic-group-containing siloxane compounds having low pour points and good lubricity.

ORIGIN OF THE INVENTION

The present invention is directed to thiophenyl ether disiloxanes and trisiloxanes useful as lubricant fluids or other functional fluids for service under extreme conditions. Substitution of sulfur for the phenox group oxygen of ether siloxane compounds has been found in this invention to result in a marked improvement in lubricity, while the siloxane units provide substantially decreased pour points.

It is therefore an object of this invention to provide improved lubricant and functional fluids for service over a wide range of temperatures.

Still another object is to provide lubricant fluids having a low vapor pressure.

Yet another object is to provide aromatic-group-containing siloxane compounds having low pour points and good lubricity.

Other objects and advantages of the invention will be apparent from the following detailed description.

DESCRIPTION OF THE PREFERRED EMBODIMENTS

Thiophenyl ether disiloxanes and trisiloxanes embodying the invention are represented by the formula

\[ \text{[3-(phenylthio)phenyl]-1,1,3,3,5,5-hexamethyldisiloxane} \]

Novel siloxane compounds of this invention can be prepared by procedures similar to those previously used for preparing ether siloxanes, with a sulfur-containing reactant being substituted for the corresponding oxygen-containing ether compound. Disiloxanes are preferably prepared by reacting the Grignard reagent of a bromine-substituted phenylbenzene with a dichlorosilane and hydrolyzing the resulting arylthiophenylchlorosilane with water. Trisiloxanes can be prepared by reacting the above-mentioned Grignard reagent with a trisiloxane compound obtained by partial hydrolysis of the appropriate dichlorosilane such as dimethyl-dichlorosilane.

Compounds embodying the invention exhibit favorable overall properties for use as lubricant fluids in space or other severe environments. In addition to their good lubricity, resistance to radiation and liquidity over a wide temperature range, these compounds have a low vapor pressure, resistance to oxidation and thermal degradation at temperatures over 500°F and good miscibility with lubricant additives. Other properties of the compounds, in particular, high bulk moduli and high surface tension, indicate their suitability for use as working fluids for use under severe conditions, for example, as hydraulic fluids in spacecraft hydraulic servomechanisms.

The invention is further illustrated by the following specific examples.
siloxane was prepared by the following procedure. To a solution of 108 grams of sodium methoxide in 750 ml. of absolute ethanol was added 220 grams of benzenethiol. After refluxing for four hours, the solvent was distilled under vacuum, leaving 261.5 grams of sodium thiophenoxybromine. 5. Water. The brown oil, which separated, was washed several times with water, decolorized with charcoal, dried over potassium carbonate and distilled. After reacted, m-dibromobenzene was collected. 121.5 grams of product 1-bromo-3-(phenylthio)benzene distilled at 157-160° C. at 3 torr was recovered. A solution of 115 grams of the 1-bromo-3-(phenylthio)benzene in 100 ml. of tetrahydrofuran was added with stirring during three hours to 17 grams of magnesium. Formation of the Grignard reagent was initiated with a small amount of 1-iodobutane. After stirring for 3 hours longer, the Grignard reagent was decanted and added slowly during 2 hours to 122 ml. of dimethyldichlorosilane. After stirring overnight, the tetrahydrofuran and excess dimethyldichlorosilane were distilled. The residue was washed with water and with methanol, then was dissolved in carbon tetrachloride. This solution was passed through a column of neutral alumina. The solvent was evaporated under vacuum, and the residue was filtered through a fiber glass filter. 42.1 grams of 1,3-bis[3-(4-chlorophenylthio)phenyl]-1,1,3,3-tetramethyldisiloxane was recovered.

EXAMPLE 2

1,3-bis[3-(4-chlorophenylthio)phenyl]-1,1,3,3-tetramethyldisiloxane was prepared by the following procedure. To a solution of 108 grams of sodium methoxide in 750 ml. of absolute ethanol was added 289.3 grams of 4-chlorobenzene. After refluxing with stirring for 6 hours, the solvent was distilled from the product sodium 4-chlorothiophenoxide. A solution of 333 grams of the product in 700 ml. of bis[2-(methoxymethyl)ethy] ether was added slowly during four hours to a refluxing solution of 944 grams of m-dibromobenzene in 800 ml. of the same solvent. The mixture was refluxed with stirring for 41 hours. It was then cooled, diluted with 1000 ml. of benzene, washed twice with a 4 percent aqueous sodium hydroxide solution, twice with a 4 percent aqueous sodium chloride solution and three times with water. Benzene and excess m-dibromobenzene were distilled, and 350 grams of the product 1-bromo-3-(4-chlorophenylthio)benzene distilled at 170-175° C. at 1 torr was recovered. A solution of 150 grams of the latter product in 500 ml. of anhydrous tetrahydrofuran was added slowly to 10.6 grams of magnesium. Formation of Grignard reagent was completed in 5 hours. The reagent was then added dropwise over 1.5 hours to a solution of 57.7 grams of the trisiloxane fraction in 500 ml. of anhydrous tetrahydrofuran. The mixture was refluxed 2 hours, cooled and treated with water. The solvent in the organic phase was removed by heating in vacuum. The residue was dissolved in carbon tetrachloride and passed through a column of neutral alumina. The solvent was evaporated under vacuum, leaving 108.5 grams of 1,5-bis[3-(phenylthio)phenyl]-1,1,3,3,5,5-hexamethyldisiloxane.

Samples prepared in the above examples were subjected to elemental analyses to confirm their purity, and their structures were established by infrared and nuclear magnetic resonance spectroscopy.

Physical properties of the compound of Example 3 were determined to be as follows: pour point, —53° F.; viscosity, kinematic, 22,740 centistokes at —45° F., 8,226 centistokes at —35° F., 599 centistokes at 0° F., 20.49 centistokes at 100° F., 4.99 centistokes at 210° F., and 1.6 centistokes at 400° F. (extrapolated); vapor pressure 1 x 10^-6 torr at 100° F., 1 x 10^-3 torr at 350° F.; thermal and oxidation stability, to 500° F.; bulk moduli (isothermal secant), 250,000 p.s.i. at 100° F. and 132,000 p.s.i. at 350° F.; surface tension at 75° F., 35 dynes per centimeter; and LFW-1 lubricant rating, 2. Samples irradiated with a cobalt-60 source to a dosage level of 10^10 ergs per gram showed only slight changes after irradiation. The compounds of Examples 1 and 2 displayed similar properties, except that the chlorinated product (Example 2) had a substantially higher pour point and higher viscosity, but its lubricity was better than for the material of Example 3.

It is to be understood that the above examples are merely illustrative of the invention and that various changes and modifications in the procedures described could be employed without departing from the invention.

What is claimed is:

1. Thiophenyl ether siloxanes having the formula
where X is a hydrogen atom, a halogen atom or an alkyl, aryl, alkaryl or aralkyl radical, \( n_1 \) and \( n_3 \) are integers from 1 to 3, inclusive, \( n_2 \) is an integer from 1 to 2, inclusive, and \( R_1 \) and \( R_2 \) are methyl or phenyl radicals.

2. Siloxane compounds of claim 1 wherein X is a hydrogen atom, a halogen atom or a hydrocarbon radical selected from the group consisting of methyl, ethyl, propyl, phenyl, biphenyl, naphthyl, tolyl, xylyl, benzyl or phenylethyl radicals.

3. The compound of claim 2 wherein X is a hydrogen atom, \( n_1 \) and \( n_3 \) are 1, \( n_2 \) is 2 and \( R_1 \) and \( R_2 \) are methyl radicals.

4. The compound of claim 2 wherein X is a hydrogen atom, \( n_1 \) and \( n_3 \) are 1, \( n_2 \) is 1 and \( R_1 \) and \( R_2 \) are methyl radicals.

5. The compound of claim 2 wherein X is chlorine, \( n_1 \) and \( n_3 \) are 1, \( n_2 \) is 1 and \( R_1 \) and \( R_2 \) are methyl radicals.