COSMOGENIC RADIONUCLIDES
IN STONE METEORITES

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This document presents the techniques and compilation of results of cosmogenic $^{26}$Al measurements at Goddard Space Flight Center on 91 samples of 76 stone meteorites. Short-lived radionuclides, including $^{22}$Na, $^{46}$Sc, $^{54}$Mn, and $^{60}$Co, were measured in 13 of these meteorites. About one-third of these data has not previously been published. The results are discussed briefly in terms of (1) depletion of $^{26}$Al and natural potassium due to weathering, (2) possible exposure of several chondrites to an unusually high cosmic-ray flux, (3) comparison of $^{26}$Al, $^{22}$Na, $^{46}$Sc, and $^{54}$Mn in chondrites with the spallation $^{22}$Ne/$^{21}$Ne ratio as a shielding indicator, and (4) comparison of $^{26}$Al - $^{22}$Ne/$^{21}$Ne data for achondrite classes with the chondrite trend.
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INTRODUCTION

The purpose of this document is to present a compilation of the results of a decade of measurements of cosmic-ray-induced radioactivity in stone meteorites. Part of the data has appeared in the published literature (References 1 through 8) and is included for completeness, but is not extensively discussed.* Results of incomplete projects are included here in the hope that investigators actively engaged in meteorite research will find the information of value.

The short- and long-lived radionuclide contents of a meteorite specimen depend on the chemical composition of the sample, on the depth from the preatmospheric surface of the meteorite, on the preatmospheric size of the meteorite, its terrestrial and cosmic-ray exposure ages, and the cosmic-ray flux to which the meteorite was exposed in space. The knowledge of nuclear production systematics in meteorites has grown in recent years in the areas of target element production rates (References 2 and 9) and depth effects (References 6 and 8). The interpretation of radionuclide measurements is beginning to fulfill its potential as a tool in understanding the cosmic-ray environment and the origins of meteorites.

MEASUREMENT SYSTEM

Whole rock and powdered meteorite specimens were measured for gamma-ray emitting radionuclides in a nondestructive, dual parameter, gamma coincidence counting system. The essential components of the detection system are:

- A clean steel shield, 137-cm outside length, 81-cm outside width and height, 15-cm wall thickness, with a 5-cm thickness of lead bricks on top;
- A plastic scintillator anticoincidence detector, 51 cm long, 41-cm outside diameter, and 18-cm inside diameter, with six RCA 8054 photomultiplier tubes at 60° spacings on one end;
- A 5-cm thickness of borated paraffin on top of the anticoincidence detector (inside the shield); and
- Two low background 10- by 10-cm (4- by 4-in.) NaI(Tl) detectors, mounted in lucite tubes for geometric reproducibility, and each including
  - 7.6-cm unactivated NaI light pipe,
  - OFHC copper shell,
  - RCA 4521 phototube,
  - <1 ppm K contamination, and
  - 8 percent resolution for $^{137}$Cs gamma-ray (0.661 MeV).

Anticoincidence gating is achieved by summing outputs from opposed phototubes and routing the resulting three signals through separate amplifiers and baseline restorers into single channel analyzers. The analyzer windows are set so that upper level discriminator outputs correspond to cosmic-ray events ($E > 3$ MeV, up to hundreds of MeV), while the windows correspond primarily to natural and to Compton-scattered events ($0.05 < E < 3$ MeV). These two sets of outputs are routed respectively to two coincidence modules which are set to deliver gating outputs when any two inputs are in time coincidence (~1 μs). The outputs are applied to the anticoincidence inputs to the pulse height analyzer.

Sample pulses are amplified, passed through digital gain/zero stabilizers, and entered into a 4096-channel dual parameter pulse height analyzer. The analyzer stores pulses as coincident or noncoincident events, with a resolving time of ~1 μs. The customary counting matrix is 64 by 64 channels, encompassing 0.1 to 3 MeV from each detector. The 64-channel conversion provides only adequate effective resolution for noncoincident events, but is excellent for coincident events; the latter are of primary importance to these studies.

SAMPLE AND BACKGROUND COUNTING

Whole or powdered meteorite samples were placed in lucite sample holders (selected because of typically high radiochemical purity) for nondestructive counting. For whole samples,
Lucite rings (approximately 1- or 6-mm thickness, and with 5-, 7.6-, 10.2-, or 12.7-cm inside diameter) were used with ~1-mm thick end plates. For powdered samples, lucite containers (2.5-, 5.1-, 7.6-, or 10.2-cm inside diameters and 0.6, 1.2, 1.9, or 2.5 cm deep) were used. The assembled sample holder was bolted (stainless steel threaded rods) to the end of one of the sodium iodide detector holders, ensuring that sample position could be duplicated with calibration standards.

**BACKGROUNDS**

Background measurements were made using 150-μm dunite or nickel powder (depending on sample density). Initially the background samples were counted in the same geometries as the meteorite specimens. For powdered specimens, a lucite container was filled with dunite powder to the sample's level. For whole specimens, a mock-up was made (and later used for calibration sources). The meteorite specimen was wrapped in aluminum foil. A second layer was carefully formed around the sample and coated with a surface-hardening epoxy resin. A seam was left so that the shell could later be opened and the sample removed. The empty shell was then sealed and further coated, leaving an opening at the top (in its counting position) through which nickel powder was introduced to the desired weight. Under these conditions the dunite and nickel powdered background samples had bulk and electronic densities closely approximating those of powdered and whole stone meteorite samples, respectively.

After many exact mock-up backgrounds were run, it became clear that background levels varied systematically with the separation distance of the two sample detectors and were apparently not influenced by sample shape or even, for the range of samples studied, sample mass. Accumulated background counting rates (B) for the energy regions of interest were correlated with detector spacing (S) by unweighted least squares regression analysis, yielding equations of the form:

\[ B = a_0 + a_1 S + a_2 S^2 \]

In no case was there statistical justification for a third order fit; in some cases the second order term was unjustified statistically. Standard deviations were calculated from the differences between observed and calculated background counting rates. These calculated uncertainties for the background equations were similar in magnitude to measured uncertainties in typical 7000- to 10000-min background runs. Backgrounds calculated by this means were considered more reliable than the measured backgrounds for any individual sample. The procedure of measuring a background for each sample was discontinued. Thereafter, dunite or nickel backgrounds were measured about once a month to cover the range of expected detector spacings, to improve the confidence level of the equations, and to ensure the continued validity of the equations over time.
Table 1 lists the radioactive species of interest in this work, the appropriate gamma-ray energies and coincidences, the calculated background equations (in counts per minute for S in centimeters), the $R^2$ goodness of fit of the equations to the input data, and the calculated standard deviations to be used with the equations. The summation symbol (Σ) indicates that data along two analyzer axes are summed; for example,

$$\sum X_{0.51} Y_{1.27} + X_{1.27} Y_{0.51 \text{ MeV}}$$

Energy ranges used in counting rate calculations varied from ±0.05 MeV at low energies to ±0.10 MeV for higher energies. In cases where $R^2 \approx 0.7$, the counting rates were so low and varied so little with S that statistical fluctuations in counting rate measurements were major factors in the $R^2$ calculations.

Table 1
Background Equations

<table>
<thead>
<tr>
<th>Radioactivity</th>
<th>Energy, MeV</th>
<th>Background, cpm</th>
<th>±σ</th>
<th>$R^2$</th>
</tr>
</thead>
<tbody>
<tr>
<td>$^{22}\text{Na, }^{26}\text{Al}$</td>
<td>$0.51 \times 0.51$</td>
<td>$0.0719 - 0.129S + 0.00101S^2$</td>
<td>0.0025</td>
<td>0.96</td>
</tr>
<tr>
<td>$^{22}\text{Na}$</td>
<td>$\Sigma 0.51 \times 1.27$</td>
<td>$0.0489 - 0.0101S + 0.00086S^2$</td>
<td>0.0036</td>
<td>0.86</td>
</tr>
<tr>
<td>$^{22}\text{Na, }^{26}\text{Al}$</td>
<td>$\Sigma 0.51 \times 1.80$</td>
<td>$0.0839 - 0.0184S + 0.00160S^2$</td>
<td>0.0050</td>
<td>0.91</td>
</tr>
<tr>
<td>$^{26}\text{Al}$</td>
<td>$\Sigma 0.51 \times 2.32$</td>
<td>$0.0179 - 0.0031S + 0.00025S^2$</td>
<td>0.0016</td>
<td>0.78</td>
</tr>
<tr>
<td>$\text{U}$</td>
<td>$\Sigma 0.61 \times 0.77$</td>
<td>$0.0431 - 0.0094S + 0.00087S^2$</td>
<td>0.0031</td>
<td>0.85</td>
</tr>
<tr>
<td>$\text{U}$</td>
<td>$\Sigma 0.61 \times 1.12 \text{ to } 1.24$</td>
<td>$0.0452 - 0.0082S + 0.00058S^2$</td>
<td>0.0032</td>
<td>0.86</td>
</tr>
<tr>
<td>$\text{U}$</td>
<td>$\Sigma 0.61 \times 1.38 \text{ to } 1.51$</td>
<td>$0.0361 - 0.0081S + 0.00067S^2$</td>
<td>0.0025</td>
<td>0.87</td>
</tr>
<tr>
<td>$\text{Th}$</td>
<td>$\Sigma 0.28 \times 2.61$</td>
<td>$0.0254 - 0.0021S$</td>
<td>0.0021</td>
<td>0.72</td>
</tr>
<tr>
<td>$\text{Th}$</td>
<td>$\Sigma 0.58 \times 2.61$</td>
<td>$0.0281 - 0.0049S + 0.00031S^2$</td>
<td>0.0017</td>
<td>0.87</td>
</tr>
<tr>
<td>$\text{Th}$</td>
<td>$\Sigma 0.86 \times 2.61$</td>
<td>$0.0113 - 0.0021S + 0.00014S^2$</td>
<td>0.0014</td>
<td>0.66</td>
</tr>
<tr>
<td>$^{60}\text{Co}$</td>
<td>$\Sigma 1.17 \times 1.33$</td>
<td>$0.0111 - 0.0021S + 0.00019S^2$</td>
<td>0.0010</td>
<td>0.69</td>
</tr>
<tr>
<td>$^{40}\text{K}$</td>
<td>$\Sigma 1.46$</td>
<td>$2.158 + 0.0612S + 0.0136S^2$</td>
<td>0.053</td>
<td>0.95</td>
</tr>
</tbody>
</table>

Figure 1 shows the data and calculated correlation lines for the $0.51- \times 0.51$- and $0.51- \times 2.32$-MeV background counting rates versus detector spacing. The trend toward decreasing background with increasing detector spacing is common to all but one case listed in table 1. The singular exception is the $^{40}\text{K}$ background, shown in figure 2 for each detector and for the sum. It is clear that the increasing total background with spacing is due primarily to the background variation of one sample detector. The position in the shield of that detector (Y) varies with sample thickness, while the X detector is kept in a fixed position. The Y detector moves nearer to the end of the anticoincidence mantle as sample thickness, and
thus detector spacing, increases. Six conventional photomultiplier tubes, containing small but measurable amounts of K, U, and Th, are affixed to that end of the anticoincidence detector. Thus, as the Y detector is moved closer to these phototubes, the noncoincident background due to these naturally radioactive species increases.

CALIBRATION

Standardized sources of $^{22}$Na, $^{26}$Al, $^{40}$K, $^{46}$Sc, $^{54}$Mn, $^{60}$Co, U, and Th were used, as appropriate, to calibrate counting yields for individual meteorite sample geometries. All sources were obtained commercially ($^{40}$K as reagent-grade KCl) with listed calibration uncertainties of $\leq$5 percent. All calibrations were compared with certified standards used by Battelle Pacific Northwest Laboratories, resulting in revising the specific activities of the $^{26}$Al, $^{46}$Sc, and $^{54}$Mn sources. The $^{26}$Al source was corrected by quantitatively comparing coincident and single counting rates with a National Bureau of Standards $^{22}$Na source ($\pm$ 2 percent) in the same geometry. The revised $^{26}$Al calibration was confirmed with the Battelle $^{26}$Al standard.
Standards were mixed with 150-μm dunite or iron powders in order to simulate the densities of powdered or whole rock stone meteorite samples, respectively. The appropriate powder mixtures were placed directly in lucite holders, or into epoxy-coated mockups (discussed above), to duplicate sample geometries. In a few cases where samples were thin (<0.5 cm) slices of larger specimens, epoxy shells were not made. Instead, a thick layer of aluminum foil was shaped around the sample, filled with the proper mass of the standard of interest, and manually smoothed flat to fit the sample’s holder.

Counting duplicate small samples demonstrated source uniformity at least down to the 10-g level. Repeated filling, mounting, and counting mockups or shaped foils yielded agreement between runs within counting statistics. All cosmogenic radionuclide measurements were calibrated by direct comparison with standards in identical geometries. This was generally true of $^{40}$K, and less so for U and Th. In some cases in which the K, U, or Th contents were not required to high accuracy, counting yields were estimated from other samples of similar size/shape; uncertainties in calibration estimates were included in the calculations of element concentration.
Table 2 lists measured counting yields for a range of representative sample geometries. The samples are identified by mass and detector spacing. For coincident events, yields were obtained from the difference between standard and background counting rates, divided by the source specific activity (or mass in the cases of U and Th). For K, the source counting rate for the 1.46-MeV full energy peak was first corrected for the counting rate in adjacent nonpeak channels. A net background counting rate, obtained similarly (see table 1), was subtracted, and the difference divided by the mass of potassium in the source. For actual samples, this technique minimized potential interference due to scattering of higher energy gamma rays. Thus, for the specific channels used in this work:

\[ \epsilon_K = \frac{R_{(ch\ 30-33)} - R_{(28 + 29 + 34 + 35)}}{M_K} \]

No calibrated source was available for \(^{48}\)V (0.98 × 1.31 MeV). When this nuclide was detected in the coincidence spectrum of a recently-fallen chondrite, the counting yield was interpolated from those of \(^{60}\)Co (1.17 × 1.33 MeV) and \(^{46}\)Sc (0.89 × 1.12).

Table 2
Representative Counting Yields

<table>
<thead>
<tr>
<th>Nuclide</th>
<th>Energy (MeV)</th>
<th>Sample Mass and Detector Spacing</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>203 g</td>
</tr>
<tr>
<td></td>
<td></td>
<td>4.9 cm</td>
</tr>
<tr>
<td>(^{22})Na (cpm/dpm)</td>
<td>(0.51 \times 0.51)</td>
<td>0.0111</td>
</tr>
<tr>
<td></td>
<td>(\Sigma 0.51 \times 1.27)</td>
<td>0.00247</td>
</tr>
<tr>
<td></td>
<td>(\Sigma 0.51 \times 1.78)</td>
<td>0.00231</td>
</tr>
<tr>
<td>(^{26})Al (cpm/dpm)</td>
<td>(0.51 \times 0.51)</td>
<td>0.0121</td>
</tr>
<tr>
<td></td>
<td>(\Sigma 0.51 \times 1.81)</td>
<td>0.00228</td>
</tr>
<tr>
<td></td>
<td>(\Sigma 0.51 \times 2.32)</td>
<td>0.00201</td>
</tr>
<tr>
<td>(^{60})Co (cpm/dpm)</td>
<td>(\Sigma 1.17 \times 1.33)</td>
<td>0.00179</td>
</tr>
<tr>
<td>(^{40})K (cpm/gK)</td>
<td>(\Sigma 1.46)</td>
<td>12.7</td>
</tr>
<tr>
<td>U (cpm/mgU)</td>
<td>(\Sigma 0.61 \times 0.77)</td>
<td>0.377</td>
</tr>
<tr>
<td></td>
<td>(\Sigma 0.61 \times 1.12) to 1.24</td>
<td>0.781</td>
</tr>
<tr>
<td></td>
<td>(\Sigma 0.61 \times 1.38) to 1.51</td>
<td>0.191</td>
</tr>
<tr>
<td>Th (cpm/mg Th)</td>
<td>(\Sigma 0.28 \times 2.61)</td>
<td>0.133</td>
</tr>
<tr>
<td></td>
<td>(\Sigma 0.58 \times 2.61)</td>
<td>0.163</td>
</tr>
<tr>
<td></td>
<td>(\Sigma 0.86 \times 2.61)</td>
<td>0.027</td>
</tr>
</tbody>
</table>
COUNTING RATE AND SPECIFIC ACTIVITY CALCULATIONS

Sample counting times varied from 4000 to about 10,000 minutes, depending primarily on mass and/or expected activity level. For coincident events where the sample counting rate is $R_s$ and the total counts in the selected energy region $G$,

$$R_s = \frac{G}{t}, \text{ and } \sigma_s = \pm \frac{(G)^{1/2}}{T}$$

The net sample counting rate, $R_N$, is the difference between $R_s$ and the background rate, $R_B$; $\sigma_N = \pm (\sigma_s^2 + \sigma_B^2)^{1/2}$. This net rate can be corrected, if necessary, for contributions from other nuclides. The corrected net counting rate is divided by the radionuclide counting yield $\varepsilon_N$ and by the sample mass in kilograms to find the specific activity in disintegrations per minute per kilogram of sample (dpm/kg).

For $^{26}$Al (in the absence of $^{22}$Na), U, or Th, specific activities were calculated for three coincidence energy regions. The weighted average of the three calculated activity values was taken as the sample specific activity. Thus:

$$A = \frac{\sum (\frac{A_1}{\sigma_1^2} + \frac{A_2}{\sigma_2^2} + \frac{A_3}{\sigma_3^2})}{\sum (\frac{1}{\sigma_1^2} + \frac{1}{\sigma_2^2} + \frac{1}{\sigma_3^2})}$$

and

$$\sigma_A = \left[ \sum \frac{1}{\sum (\frac{1}{\sigma_1^2} + \frac{1}{\sigma_2^2} + \frac{1}{\sigma_3^2})} \right]^{1/2}$$

The agreement in the pattern of counting rates for the three energy regions between sample and calibration source was checked by:

$$\chi^2_R = \frac{\left( \frac{A - A_1}{\sigma_1} \right)^2 + \left( \frac{A - A_2}{\sigma_2} \right)^2 + \left( \frac{A - A_3}{\sigma_3} \right)^2}{2}$$

A value $\leq 1$ indicates satisfactory agreement. If $\chi^2_R > 1$, the calculated $\sigma_A$ is multiplied by $\left( \chi^2_R \right)^{1/2}$ to provide a more realistic estimate of the activity uncertainty.
Where $^{22}$Na and $^{26}$Al were both present, constituting a mutual interference, the net counting rate data from the four coincidence energy regions (table 1) were treated as an overdetermined set of simultaneous linear equations. The four equations, of the form $Y \pm \sigma_Y = a_1 x_1 + a_2 x_2$, were reduced by least squares regression, weighted by $(\sigma_Y)^2$. In this specific case, $Y$ is the net counting rate, $a_1$ is the counting yield for one component (e.g., $^{22}$Na) in a given energy region, $x_1$ is the concentration of that component, and $a_2$ and $x_2$ refer likewise to the second component (e.g., $^{26}$Al).

Potassium was determined from the net 1.46-MeV peak extending above the combined non-coincident gamma-ray spectrum. Thus the counting rate, $R_S$, of the sample was obtained from the counts, $C$, in and on either side of the full energy peak:

$$R_S = \frac{C_{(ch 30-33)} - C_{(28 + 29 + 34 + 35)}}{T}$$

and

$$a_s = \frac{[C_{(30-33)} + C_{(28 + 29 + 34 + 35)}]^{\frac{1}{2}}}{T}$$

The net K counting rate was found, as in the coincidence counting case, by $R_S - R_B$.

RESULTS

Measured specific activities, in dpm/kg at time of fall, for recently fallen chondrite samples studied in this laboratory are given in table 3. All U and Th values were calculated assuming that measured $^{208}$Tl and $^{214}$Bi were in radioactive equilibrium with their respective parents, $^{232}$Th and $^{238}$U. The Murchison sample was obtained from the University of Melbourne, Australia; the Havero sample consisted of numerous small fragments obtained from the University of Turku, Finland. Half-lives of the cosmogenic radionuclides are given in each column. Data that have already been published are so cited in the reference column.

Measured specific activities of finds and old observed falls, the former identified by year only, are given in table 4. The Keyes samples are portions of cores obtained from D. Bogard of Johnson Space Center. The Pribram samples were obtained from Z. Ceplecha of the Ondrejov Observatory, Czechoslovakia, through R. McCrosky of the Smithsonian Astrophysical Observatory. The Wellington samples were provided by E. King, University of Houston. The Zerga amphoterite sample was provided by R. Fudali, Smithsonian Institute.

Table 5 contains radioactivities in achondrites measured in this laboratory. The Moama sample was provided by J. Lovering, University of Melbourne. The Norton County samples were obtained from K. Keil, University of New Mexico.
<table>
<thead>
<tr>
<th>Meteorite</th>
<th>Class</th>
<th>Recovered Mass - kg</th>
<th>Date of Fall</th>
<th>Sample ID</th>
<th>Sample Size - g</th>
<th>$^{22}\text{Na}$ (dpm/kg)</th>
<th>$^{26}\text{Al}$ (dpm/kg)</th>
<th>$^{46}\text{Sc}$ (dpm/kg)</th>
<th>$^{48}\text{V}$ (dpm/kg)</th>
<th>$^{54}\text{Mn}$ (dpm/kg)</th>
<th>$^{60}\text{Co}$ (dpm/kg)</th>
<th>K ppm</th>
<th>Ref</th>
</tr>
</thead>
<tbody>
<tr>
<td>Barwell</td>
<td>L6</td>
<td>47</td>
<td>12/24/65</td>
<td>NMNH*</td>
<td>1265</td>
<td>840 ±40</td>
<td>510 ±40</td>
<td>90 ±30</td>
<td>850 ±50</td>
<td>60 ±10</td>
<td>1150 ±50</td>
<td>1</td>
<td></td>
</tr>
<tr>
<td>St Séverin</td>
<td>LL6</td>
<td>277</td>
<td>6/27/66</td>
<td>DIH1</td>
<td>1293</td>
<td>1020 ±30</td>
<td>550 ±50</td>
<td>50 ±10</td>
<td>160 ±50</td>
<td>640 ±60</td>
<td>190 ±70</td>
<td>790 ±20</td>
<td></td>
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<tr>
<td>Tahmatli</td>
<td>L6</td>
<td>2</td>
<td>10/5/67</td>
<td>NMNH</td>
<td>192</td>
<td>850 ±30</td>
<td>590 ±30</td>
<td>4.4 ±2</td>
<td>190 ±120</td>
<td>590 ±60</td>
<td>28 ±0.5</td>
<td>740 ±30</td>
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<tr>
<td>Allende</td>
<td>C3</td>
<td>&gt;2000</td>
<td>2/8/69</td>
<td>NMNH 3530</td>
<td>604</td>
<td>810 ±20</td>
<td>480 ±20</td>
<td>7.2 ±1.3</td>
<td>70 ±40</td>
<td>1080 ±40</td>
<td>770 ±20</td>
<td>270 ±20</td>
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<td></td>
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<td></td>
<td></td>
<td>3531</td>
<td>598</td>
<td>840 ±20</td>
<td>520 ±20</td>
<td>4.4 ±1.2</td>
<td>110 ±3.0</td>
<td>1040 ±40</td>
<td>900 ±20</td>
<td>240 ±20</td>
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<td>3529a</td>
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<td>625 ±2.2</td>
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<tr>
<td>Marchuson</td>
<td>C2</td>
<td>9/28/69</td>
<td>#22</td>
<td>152</td>
<td>900 ±2.0</td>
<td>430 ±2.0</td>
<td>7.8 ±3.5</td>
<td>101 ±2.0</td>
<td>79 ±3.0</td>
<td>390 ±20</td>
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*NMNH = National Museum of Natural History
Table 4
Radioactivities in Chondrites

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*aNMNH = National Museum of Natural History  
ASU = Arizona State University  
AML = American Meteorite Laboratory  
JSC = Johnson Space Center, NASA  
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*ME = Field Museum of Natural History, Chicago
AML = American Meteorite Laboratory
ASU = Arizona State University
AMNH = American Museum of Natural History
NMNH = National Museum of Natural History

†Herzog, G.F. and P.J. Cressy, "$^{26}$Al Losses from Weathered Chondrites," Meteoritics, 1976, in press.
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<td>Clovis #2, a</td>
<td>H6</td>
<td>34</td>
<td>1960</td>
<td>AML H8 19</td>
<td>80</td>
<td>75.4 ±2.6</td>
<td>641 ±40</td>
<td>31 ±14</td>
<td>72 ±49</td>
<td></td>
</tr>
<tr>
<td>Clovis #2, b</td>
<td></td>
<td></td>
<td></td>
<td>AML H8 66</td>
<td>102</td>
<td>70.2 ±2.0</td>
<td>735 ±44</td>
<td>64 ±12</td>
<td>&lt;80</td>
<td></td>
</tr>
<tr>
<td>Menow</td>
<td>H4</td>
<td>11</td>
<td>10/7/1862</td>
<td>ME 1388</td>
<td>24</td>
<td>53.7 ±5.8</td>
<td>794 ±127</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Bath</td>
<td>H4</td>
<td>21</td>
<td>8/29/1892</td>
<td>ME 1782</td>
<td>93</td>
<td>66.0 ±2.4</td>
<td>775 ±34</td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

*NMNH = National Museum of Natural History  
ME = Field Museum of Natural History, Chicago  
AML = American Meteorite Laboratory  
CAL = Calcutta, India  
Table 4 (Continued)

<table>
<thead>
<tr>
<th>Meteorite</th>
<th>Class</th>
<th>Recovered Mass · kg</th>
<th>Date of Fall/Find</th>
<th>Sample ID*</th>
<th>Sample Size · g</th>
<th>$^{26}$Al dpm/kg</th>
<th>K ppm</th>
<th>U ppb</th>
<th>Th ppb</th>
<th>Ref</th>
</tr>
</thead>
<tbody>
<tr>
<td>Bledsoe</td>
<td>H</td>
<td>31</td>
<td>1970</td>
<td>AML H121.36</td>
<td>109</td>
<td>44.8 ± 2.0</td>
<td>386 ± 36</td>
<td>32 ± 14</td>
<td>76 ± 44</td>
<td>†</td>
</tr>
<tr>
<td>Tieschutz</td>
<td>H3/C3</td>
<td>28</td>
<td>7/15/1878</td>
<td>ME 1726</td>
<td>21</td>
<td>78.0 ± 7.0</td>
<td>160 ± 160</td>
<td></td>
<td></td>
<td>6</td>
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<tr>
<td>Felt</td>
<td></td>
<td></td>
<td></td>
<td>AML H114.9</td>
<td>93</td>
<td>65.5 ± 3.8</td>
<td>610 ± 35</td>
<td>39 ± 13</td>
<td>&lt;78</td>
<td></td>
</tr>
<tr>
<td>Leoville</td>
<td>C3</td>
<td>8</td>
<td>1961</td>
<td>AML H40.72</td>
<td>62</td>
<td>60.1 ± 2.7</td>
<td>151 ± 45</td>
<td>52 ± 16</td>
<td>&lt;209</td>
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<tr>
<td>Seminole #2</td>
<td>1</td>
<td>1965</td>
<td></td>
<td>AML H43.3</td>
<td>88</td>
<td>48.9 ± 3.1</td>
<td>589 ± 64</td>
<td>217 ± 25</td>
<td>&lt;160</td>
<td>†</td>
</tr>
<tr>
<td>Seminole #3</td>
<td>1.7</td>
<td>1967</td>
<td></td>
<td>AML H73.4</td>
<td>15</td>
<td>64.6 ± 6.0</td>
<td>663 ± 171</td>
<td>175 ± 60</td>
<td></td>
<td>§</td>
</tr>
<tr>
<td>Zerga</td>
<td>LL6</td>
<td>0.1</td>
<td>1973</td>
<td>NMNH 5740</td>
<td>74</td>
<td>51.5 ± 3.4</td>
<td>1018 ± 60</td>
<td>186 ± 28</td>
<td>&lt;170</td>
<td></td>
</tr>
</tbody>
</table>

*AML = American Meteorite Laboratory  
ME = Field Museum of Natural History, Chicago  
NMNH = National Museum of Natural History  
ORDINARY CHONDRITES

Chondrite chemical analyses typically show K = 700 to 900 ppm and U on the order of 20 ppb. A number of the weathered meteorite finds in table 4 have significantly lower K content (see figure 3) and/or higher U (or its daughters). These alterations can be attributed to the weathering process itself, sometimes leaching K from the meteorite and sometimes introducing U (and Th) or their decay products into the sample. Although these weathering effects do not always correspond to diminished $^{26}$Al activities in table 4, the converse is true; significantly reduced $^{26}$Al levels (not attributable to cosmic-ray shielding or short exposure age) are invariably associated with low K and/or high U contents. Specifically:

<table>
<thead>
<tr>
<th>Meteorite</th>
<th>$^{26}$Al (dpm/kg)</th>
<th>K (ppm)</th>
<th>U (ppb)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Etter</td>
<td>42.2</td>
<td>538</td>
<td>24</td>
</tr>
<tr>
<td>Dalgety Downs</td>
<td>40.3 to 49.9</td>
<td></td>
<td>207 to 630</td>
</tr>
<tr>
<td>Bledsoe</td>
<td>44.8</td>
<td>386</td>
<td>32</td>
</tr>
<tr>
<td>Seminole #2</td>
<td>48.9</td>
<td>589</td>
<td>217</td>
</tr>
</tbody>
</table>

Expected $^{26}$Al activities under average cosmic-ray shielding conditions can be calculated from chemical compositions using target element production rates (Reference 2). This average production rate does not take into account the effects of sample depth and meteorite pre-atmospheric size (References 6 and 8). The spallation $^{22}$Ne/$^{21}$Ne ratio has been shown to be an indicator of irradiation hardness (Reference 10), and thus (References 6, 8, 11, and 12) related to shielding (size and depth).* In figure 4, the ratio of measured $^{26}$Al to that calculated from chemical composition ($^{26}$Al)/($^{26}$Al$_Q$) is plotted against $^{22}$Ne/$^{21}$Ne for meteorites in which both measurements were made on the same sample. The upper line represents the trend reported by Herzog and Cressy (Reference 6) for data obtained from a number of chondrites. The lower line was calculated from the data (•) obtained on core samples of the Keyes chondrite (References 8 and 11). The first relationship reflects a variety of unknown sizes and depths, while the latter is due only to depth variations in a single fairly large stone meteorite.

The difference in slope between the two lines, and the tendency of the data to scatter around them, suggests that the $^{22}$Ne/$^{21}$Ne ratio is not entirely adequate as an indicator of both size and depth effects on $^{26}$Al production (Reference 6). The Herzog-Cressy line is most likely an approximation to a family of curves, similar to the Keyes line (although not necessarily linear) but with slopes dependent on meteorite pre-atmospheric size.

The Dalgety Downs data (○) fall conspicuously below the trend of the other data in the diagram, probably because of terrestrial weathering.† The St. Séverin (SS) data point

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†Herzog, G.F. and P.J. Cressy, "$^{26}$Al Losses from Weathered Chondrites," Meteoritics, 1976, in press.
Table 5
Radioactivities in Achondrites

<table>
<thead>
<tr>
<th>Meteorite</th>
<th>Class</th>
<th>Recovered Mass - kg</th>
<th>Date of Fall/Find</th>
<th>Sample ID*</th>
<th>Sample Size - g</th>
<th>$^{26}$Al dpm/kg</th>
<th>K ppm</th>
<th>U ppb</th>
<th>Th ppb</th>
<th>Ref</th>
</tr>
</thead>
<tbody>
<tr>
<td>Kenna</td>
<td>U</td>
<td>10.9</td>
<td>1972</td>
<td>AML H159.3</td>
<td>42</td>
<td>49.3 ±3.7</td>
<td>92 ±68</td>
<td>&lt;84</td>
<td>&lt;174</td>
<td>†</td>
</tr>
<tr>
<td>Moama</td>
<td>Eu</td>
<td>3.4</td>
<td>1940</td>
<td>Melbourne</td>
<td>61</td>
<td>86.5 ±3.5</td>
<td>47 ±57</td>
<td>&lt;55</td>
<td>&lt;191</td>
<td></td>
</tr>
<tr>
<td>Nuevo Laredo</td>
<td>Eu</td>
<td>0.5</td>
<td>1950</td>
<td>NMNH 1783</td>
<td>132</td>
<td>81.5 ±2.1</td>
<td>355 ±38</td>
<td>109 ±22</td>
<td>385 ±57</td>
<td></td>
</tr>
<tr>
<td>Cachari</td>
<td>Eu</td>
<td>23.5</td>
<td>1921</td>
<td>NMNH 5309</td>
<td>203</td>
<td>69.8 ±2.2</td>
<td>402 ±26</td>
<td>196 ±22</td>
<td>376 ±59</td>
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</tr>
<tr>
<td>Yurtuk</td>
<td>Ho</td>
<td>1</td>
<td>4/2/36</td>
<td>NMNH 1422</td>
<td>31</td>
<td>71.1 ±4.7</td>
<td>&lt;340</td>
<td>134 ±38</td>
<td>185 ±124</td>
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</tr>
<tr>
<td>Bununu</td>
<td>Ho</td>
<td>0.4</td>
<td>fell 1942</td>
<td>NMNH 1571</td>
<td>77</td>
<td>99.1 ±3.4</td>
<td>214 ±47</td>
<td>55 ±20</td>
<td>189 ±75</td>
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</tr>
<tr>
<td>Bida</td>
<td>Ho</td>
<td>5.5</td>
<td>1912</td>
<td>NMNH 1577</td>
<td>72</td>
<td>97.4 ±3.1</td>
<td>72 ±49</td>
<td>&lt;53</td>
<td>81 ±75</td>
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<tr>
<td>Shallowater</td>
<td>Au</td>
<td>4.7</td>
<td>1936</td>
<td>ASU 318 9</td>
<td>135</td>
<td>62.8 ±2.0</td>
<td>230 ±22</td>
<td>50 ±9</td>
<td>86 ±30</td>
<td>§</td>
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<tr>
<td>Bustee</td>
<td>Au</td>
<td>2</td>
<td>2/12/1852</td>
<td>NMNH 5622</td>
<td>82</td>
<td>75.9 ±3.3</td>
<td>205 ±47</td>
<td>&lt;30</td>
<td>&lt;160</td>
<td>§</td>
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<tr>
<td>Khor Temiki</td>
<td>Au</td>
<td>3</td>
<td>4/8/32</td>
<td>NMNH 1551</td>
<td>83</td>
<td>89.9 ±3.5</td>
<td>432 ±77</td>
<td>&lt;40</td>
<td></td>
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<tr>
<td>Pesyanoe</td>
<td>Au</td>
<td>3</td>
<td>10/2/33</td>
<td>NMNH 1425</td>
<td>66</td>
<td>98.4 ±3.4</td>
<td>584 ±57</td>
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<td>Norton County</td>
<td>Au</td>
<td>&gt;1100</td>
<td>2/18/48</td>
<td>N23-5 (NM)</td>
<td>214</td>
<td>97.6 ±3.3</td>
<td>210 ±30</td>
<td>&lt;60</td>
<td></td>
<td>§</td>
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<tr>
<td></td>
<td></td>
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<td></td>
<td>N102</td>
<td>456</td>
<td>79.0 ±2.7</td>
<td>&lt;30</td>
<td>&lt;15</td>
<td>&lt;46</td>
<td>§</td>
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<tr>
<td>Garland</td>
<td>D</td>
<td>0</td>
<td>~1955</td>
<td>NMNH 2140</td>
<td>94</td>
<td>84.1 ±3.8</td>
<td>&lt;150</td>
<td>&lt;58</td>
<td>&lt;180</td>
<td>‡</td>
</tr>
<tr>
<td>Tatahoune</td>
<td>D</td>
<td>12</td>
<td>7/27/31</td>
<td>ME 2651</td>
<td>83</td>
<td>79.4 ±3.3</td>
<td>&lt;60</td>
<td>&lt;19</td>
<td>&lt;58</td>
<td>‡</td>
</tr>
</tbody>
</table>

*AMU = American Meteorite Laboratory  
NMNH = National Museum of Natural History  
ASU = Arizona State University  
ME = Field Museum of Natural History, Chicago

Figure 3. The distribution of measured potassium contents (tables 3 and 4) for H, L, and LL chondrite finds and observed falls.

(Reference 13) was obtained from a very near surface sample; other data from this meteorite fall much nearer the Herzog-Cressy line. Ucera (U) appears to have experienced a two-stage irradiation (Reference 14). The $^{22}$Ne/$^{21}$Ne ratio primarily reflects shielding conditions during the first, relatively long irradiation, while the $^{16}$Al activity was produced during the last few million years, after Ucera had been reduced to a smaller body by an intervening collision.

Five meteorites lie considerably above the field of data. Malakal experienced a uniquely high cosmic-ray exposure (Reference 7). The same explanation may apply to Shelburne (S), Clovis #2 (C), Kabo (K), and Tieschitz (T) as well (Reference 6). Appley Bridge, Cavour, and Morland were not plotted because noble gas data indicate short cosmic-ray exposure ages and thus $^{26}$Al undersaturation.* However, their $^{26}$Al activities are inconsistently large for the estimated ages and may be best explained by an increased cosmic-ray flux also.

**FRESHLY-FALLEN METEORITES**

Although the relationship of $^{26}$Al production to the neon isotope ratio is not as tight as one would like, the empirical relationship calculated by Herzog and Cressy (Reference 6) does provide an improved basis for estimating $^{21}$Al production rates in individual meteorite samples.

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Figure 4. The ratio of observed to calculated $^{26}\text{Al}$ activity plotted against spallation $^{22}\text{Ne}/^{21}\text{Ne}$ ratio in H, L, and LL chondrites. Diamonds (*) are core data from the Keyes chondrite (References 8 and 11). Open circles (○) are data for Dalgety Downs. Symbols are identified in the text. The equation for the upper line (Reference 6) is: $^{26}\text{Al}/^{26}\text{Al}_0 = 3.2 - 2.0\left(^{22}\text{Ne}/^{21}\text{Ne}\right)$. The lower line is taken from the Keyes data: $^{26}\text{Al}/^{26}\text{Al}_0 = 5.55 - 4.24\left(^{22}\text{Ne}/^{21}\text{Ne}\right)$. The data are taken from tables 3 and 4 and from the literature (References 6, 9 through 18, and others).

Unfortunately, not nearly as much data exist on short-lived radionuclides as on long-lived $^{26}\text{Al}$, due to the rarity of meteorite falls. The data in table 3 probably represent more than one-fourth of all such data measured to date in meteorites shortly after their fall. With such a paucity of data, it is difficult to search for trends. Data for meteorites in which $^{22}\text{Na}$, $^{46}\text{Sc}$, $^{54}\text{Mn}$, and noble gases were measured in the same sample are plotted in figure 5. No clear correlations are apparent. Larger specific activities of these nuclides have been reported in some meteorites, but are not plotted because no corresponding neon isotope data exist. An additional complicating factor exists because of the relatively short half-lives of these nuclides. Their time-of-fall specific activities are potentially sensitive to spatial variations in cosmic-ray flux density and may be significantly affected by exposure to solar flare activity.

Figure 5. $^{46}\text{Sc}$ and $^{54}\text{Mn}$, normalized to target element composition, and $^{22}\text{Na}$, plotted against $^{22}\text{Ne}/^{21}\text{Ne}$ ratio. The open circle (o) represents the Haverö ureilite; its $^{22}\text{Ne}/^{21}\text{Ne}$ ratio has been normalized to chondrite chemistry using the Mg/Si weight ratio.*

Figure 6 is a plot of $^{22}\text{Na}$ against $^{26}\text{Al}$. A general positive correlation is apparent; note the Malakal point with its conspicuously high $^{26}\text{Al}$ activity. In figure 7, $^{46}\text{Sc}$ is also positively correlated with $^{54}\text{Mn}$, although standard deviations of $^{46}\text{Sc}$ activities are typically about 20 percent. The correlation is somewhat surprising in this case, because $^{46}\text{Sc}$ is a relatively high-energy product of cosmic-ray interactions with iron, while $^{54}\text{Mn}$ is definitely a low-energy product.

**$^{26}\text{Al}$ IN ACHONDITRES**

Achondrite isotope data can not be compared directly with chondrite data because of the considerable differences in chemical composition. The ratio of $^{26}\text{Al}$ against $^{26}\text{Al}_0$, the production rate calculated from target element chemistry (Reference 2), will in principle

---

account for this effect; however, for extremely different chemistries, as in eucrites and howardites, the corrections are less certain. Cressy and Bogard* empirically determined an equation to adjust $^{22}$Ne/$^{21}$Ne ratios for chemistry differences:

$$\left( F_{22/21}\right)_{\text{chem}} = 1.122 - 0.224 \frac{[\text{Mg}]}{[\text{Si}]} + 0.09 \left( \frac{[\text{Mg}]}{[\text{Si}]} \right)^2$$

Dividing measured neon isotope ratios by \((F_{22/21})_{\text{chem}}\) yields the chondrite chemistry equivalent of the observed ratios. Thus, plots of \(^{26}\text{Al}/^{26}\text{Al}_0\) against the adjusted \(^{22}\text{Ne}/^{21}\text{Ne}\) ratio can be compared with the corresponding plot for ordinary chondrites (figure 4).

Figure 8 is such a plot for howardites, eucrites, and the Nakhla achondrite (included here because it is a single-meteorite class whose chemical composition most closely resembles the calcium-rich achondrites). Several features are apparent from this plot.

- Eucrite and howardite data seem to be randomly intermingled, although their Mg, Ca, and Al contents are significantly different. The \(^{26}\text{Al}\) production rates \((^{26}\text{Al}_0)\) calculated (Reference 2) from average eucrite and howardite chemical compositions are 136 and 112 dpm/kg, respectively. The difference is due primarily to the production rate from Al \((P_{^{26}\text{Al}})\). An error in this rate would produce distinctly different values of \(^{26}\text{Al}/^{26}\text{Al}_0\) for the two classes; the pattern of eucrite and howardite data in figure 8 supports \(P_{^{26}\text{Al}} = 113\) dpm/kg Al (Reference 2).

- Yet, for the most part, the Eu and Ho \(^{26}\text{Al}\) data fall appreciably beneath not only the Herzog-Cressy chondrite trend, but the Keyes depth correlation as well. On the other hand, Nakhla, with a lower, almost chondritic, Al content, lies much nearer the chondrite data. An overestimated \(P_{^{26}\text{Al}}\) could yield the kind of discrepancy observed between Nakhla and the aluminum-rich achondrites. As noted above however, this would also have resulted in the segregation of eucrites and howardites from each other.

Another possible explanation was suggested several years ago (Reference 2) on the basis of less information. It is conceivable that eucrites and howardites as a group may have been exposed to a lower cosmic-ray flux than that experienced by most chondrites. This would require an unknown but exotic orbit for essentially two entire classes (perhaps related) of meteorites, despite whatever collisions they may have undergone in their individual histories.

- None of the samples in figure 8 have adjusted \(^{22}\text{Ne}/^{21}\text{Ne}\) ratios greater than 1.14; some are as low as 1.03. The raw data for these samples lie between 1.10 and 1.22, slightly above a typical chondrite range. This pattern could be due at least in part to the Mg/Si normalization. The Bruderheim data used for the normalizing equation* did not include samples having such low Mg/Si ratios. Data from Elenovka olivine (Reference 19), Mg/Si <0.2, was used to fix the curve at the low-Mg end. It is possible the equations could have “overadjusted” the \(^{22}\text{Ne}/^{21}\text{Ne}\) ratios for these meteorites \((\text{Mg/Si} = 0.2\) to 0.4), but not by more than about 0.02 units.

---

Another possibility is that eucrites and howardites are typically much larger than most chondrites. It has been shown by a number of investigators that $^{22}\text{Ne}/^{21}\text{Ne}$ ratios decrease with increasing size and/or depth, due to the increasing proportion of low-energy secondary cosmic rays produced within the meteorite. In principle, sufficiently large size/depth should result in decreased $^{26}\text{Al}$ production also. This has not been unambiguously observed experimentally, due presumably to the rarity of such large chondrites. All in all, this possibility does not seem too likely.

A third possibility is exposure to an unusually soft cosmic-ray spectrum; that is, a high proportion of low-energy radiation. This alternative would again require unique orbits for these meteorites. The most likely region would be relatively near the sun, where the large flux of low-energy solar protons and enhanced solar modulation of galactic cosmic rays could yield the observed low values of $^{22}\text{Ne}/^{21}\text{Ne}$ and $^{26}\text{Al}/^{26}\text{Al}_0$. The frequently-observed $^3\text{He}$ loss for these meteorites could then be explained by solar heating. It must be noted that this speculation benefits from our poor knowledge of other regions of the solar system.
The aubrite data in figure 9 roughly parallel the chondrite trend, but with higher $^{26}$Al ratios.* Although the discrepancies are not great, they form a consistent trend. If real, this trend suggests an intriguing possibility. The Malakal chondrite was shown to have experienced an uniquely high cosmic-ray flux (Reference 7). If aubrites were exposed to a similar but less extreme flux, their $^{26}$Al activities would be consistent with the data in figure 9. There is some circumstantial evidence for this "unique orbit" alternative.

Figure 9. Variation of $^{26}$Al/$^{26}$Al in aubrites with the $^{22}$Ne/$^{21}$Ne ratio. The diamonds (*) are Norton County data (Reference 20).* As before, solid points represent same sample measurements. No chemistry adjustment to $^{22}$Ne/$^{21}$Ne was required.

Aubrite cosmic-ray exposure ages are typically greater than $30 \times 10^6$ years; the mean for 9 aubrites is 36 with a range of 8 (Aubres) to 76 (Norton County). Bishopville, Bustee, Cumberland Falls, and Khor Temiki have apparent ages between 36 and 40 million years, and may represent products of a single collision. The mean collisional lifetime for aubrites is appreciably longer than that of any other stone meteorite class. Either aubrites are much

"tougher" than other stones (demonstrably not the case), or they traverse orbits in which the probability of encountering other objects is much less than for chondrites. This hypothesis is worthy of further study.

The existing diogenite data are shown in figure 10 and have been discussed in some detail by Herzog and Cressy.* The data generally lie near the chondrite trend, although Garland may have an anomalously high $^{26}$Al activity. The noble gas data clearly imply* that eight of the nine known diogenites are products of just two collisions that occurred 14 and 23 million years ago, respectively.

The ureilite data (figure 11) are discussed in more detail elsewhere.† Recovered masses are typically small, and $^{22}$Ne/$^{21}$Ne ratios rather high, suggesting that the preatmospheric masses of ureilites are also generally small. With the exception of Dyalpur, the data cluster very well along the chondrite trend. The pattern is convincing. The earlier suggestion of a unique orbit and a low cosmic-ray flux to explain low ureilite $^{26}$Al activities (References 5 and 21) is clearly unnecessary.†

![Figure 10. $^{26}$Al/$^{26}$Al₀ in diogenites against adjusted $^{22}$Ne/$^{21}$Ne. Solid points are same sample measurements. The $^{22}$Ne/$^{21}$Ne adjustment lowered measured values about 0.02 units.](image)

Figure 11. $^{26}\text{Al}/^{26}\text{Al}_0$ in ureilites and Chassigny (c) against adjusted $^{22}\text{Ne}/^{21}\text{Ne}$. Chassigny is a single meteorite class whose chemical composition most closely resembles those of ureilites. Kenna (o) is the only ureilite for which the same sample criterion is not known to apply. Chemistry normalization raised $^{22}\text{Ne}/^{21}\text{Ne}$ about 0.02 units.

Goddard Space Flight Center
National Aeronautics and Space Administration
Greenbelt, Maryland April 6, 1976
REFERENCES


"The aeronautical and space activities of the United States shall be conducted so as to contribute to the expansion of human knowledge of phenomena in the atmosphere and space. The Administration shall provide for the widest practicable and appropriate dissemination of information concerning its activities and the results thereof."

—National Aeronautics and Space Act of 1958

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