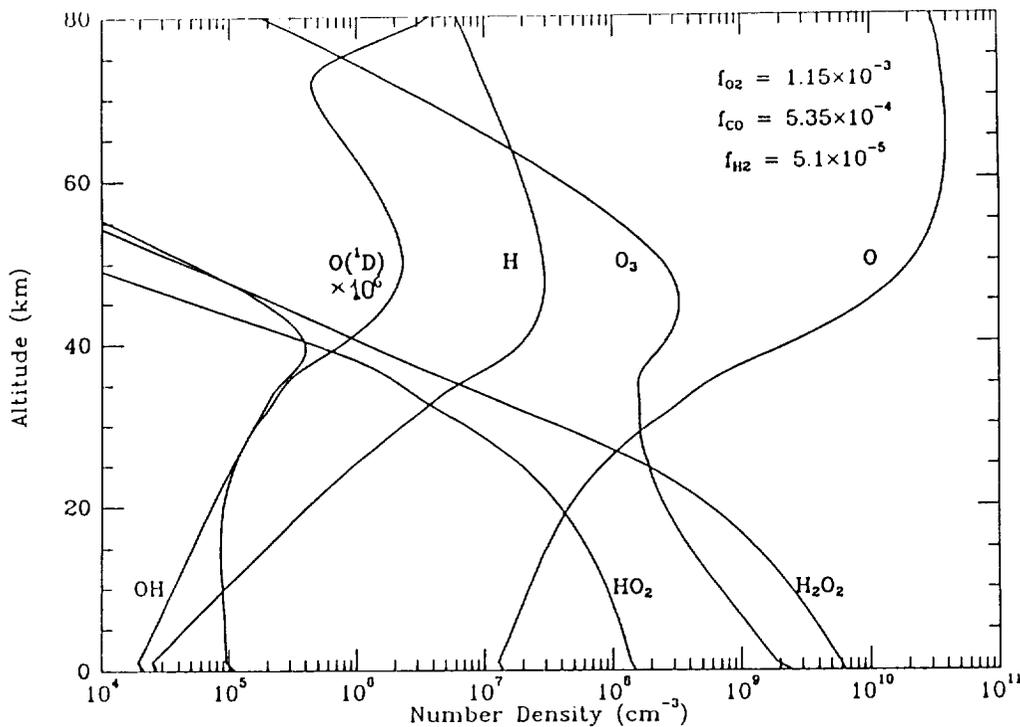


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UNIQUENESS OF A SOLUTION OF A STEADY-STATE PHOTOCHEMICAL PROBLEM: APPLICATIONS TO MARS; V. A. Krasnopolsky, NASA/Goddard SFC, Greenbelt, MD 20771

Based on conservation of chemical element in chemical reactions, a rule is proved that the number of boundary conditions given by densities and/or non-zero velocities should not be less than the number of chemical elements in the system, and the components given by densities and velocities should include all elements in the system. Applications of this rule to Mars are considered. It is proved that a problem of CO2-H2O chemistry in the lower and middle atmosphere of Mars, say, in the range of 0-80 km does not have an unique solution, if only CO2 and H2O densities are given at the lower boundary, while all other boundary conditions are fluxes. Two models of this type are discussed. These models fit the same boundary conditions, are balanced with a relative uncertainty of 10^-4 for H2 (and much better for other species), and predict the O2, CO, and H2 mixing ratios which differ by orders of magnitude. One more species density, e. g. that of O2, should be specified at the boundary to obtain the unique solution. The situation is better if the upper boundary is extended to the exobase where thermal escape velocities of H and H2 can be specified. However in this case either oxygen nonthermal escape rate (and hence the total hydrogen escape rate) or the O2 (or other species) density at the surface should be given as the boundary condition. Two models of Mars' photochemistry, with and without nitrogen chemistry, are considered. The oxygen nonthermal escape rate of 1.2x10^8 cm^-2s^-1 is given at 240 km and is balanced with the total hydrogen escape rate within uncertainty of 1% for both models. Both models fit the measured O2 and CO mixing ratios, the O3 line absorption at 9.6 micrometers and the O2 1.27 micrometers dayglow within the uncertainties of the measured values, though the model without nitrogen chemistry fits better. The importance of nitrogen chemistry in the lower and middle atmosphere of Mars depends on a fine balance between productions of NO and N in the upper atmosphere which is not known within the required accuracy.



Model of Mars' photochemistry without nitrogen chemistry. The effective absorbing O<sub>3</sub> abundance is equal to 0.44 μm, and the O<sub>2</sub> 1.27 μm dayglow intensity is equal to 3.2 MR (the measured values are 0.54±0.15 μm and 3±1 MR, respectively).

