CAPP
International Research Project on the Effects of Chemical Ageing of Polymers on Performance Properties

Interim Report on
Chemical and Thermal Analysis
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Chemical And Thermal Analysis

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# TABLE OF CONTENTS

## Synopsis

1.0 HIGH TEMPERATURE AGING OF COFLOM FOR MECHANISTIC ANALYSIS ................................................................. 1
   1.1 Test Apparatus ........................................................................................................ 1
   1.2 Sample Preparations ........................................................................................... 1
   1.3 Test Conditions .................................................................................................. 2

2.0 CHEMICAL ANALYSIS ASSESSMENT OF AGING EFFECTS .......... 2
   2.1 Gel Permeation Chromatography of Aged Coflon ......................................... 2
   2.2 X-Ray Diffraction - Crystallinity Changes ....................................................... 4
   2.3 FTIR and Ultraviolet Analysis ...................................................................... 5
   2.4 GC/Mass Spectrometry Analysis .................................................................. 5

3.0 THERMAL STABILITY AND COMPOSITIONAL ANALYSIS .......... 6
   3.1 Creep Analysis using TMA .......................................................................... 6
   3.2 DSC of Aged Materials .............................................................................. 7

4.0 APPENDICES
SYNOPSIS

Work during the past six months has included significant research in several areas aimed at further clarification of the aging and chemical failure mechanism of thermoplastics (PVDF or Tefzel) pipes. Among the areas investigated were the crystallinity changes associated with both the Coflon and Tefzel after various simulated environmental exposures using X-ray diffraction analysis. We have found that significant changes in polymer crystallinity levels occur as a function of the exposures. These crystallinity changes may have important consequences on the fracture, fatigue, tensile, and chemical resistance of the materials. We have also noted small changes in the molecular weight distribution. Again these changes may result in variations in the mechanical and chemical properties in the material. We conducted numerous analytical studies with methods including X-ray Diffraction, Gel Permeation Chromatography, Fourier Transform Infrared Spectroscopy, Ultra-Violet Scanning Analysis, GC/Mass Spectrometry, Differential Scanning Calorimetry and Thermomechanical Analysis. In the ultra-violet analysis we noted the presence of an absorption band indicative of triene formation. We investigated a number of aged samples of both Tefzel and Coflon that were forwarded from MERL. We also cast films at SWT and subjected these films to a refluxing methanol 1% ethylene diamine solution. An updated literature search was conducted using Dialog and DROLLS to identify any new papers that may have been published in the open literature since the start of this project. The updated literature search and abstracts are contained in the Appendix section of this report.

1.0 HIGH TEMPERATURE AGING OF COFLON FOR MECHANISTIC ANALYSIS

1.1 Test Apparatus

The apparatus for the experimentation involved a round bottom flask fitted with a reflux condenser for aging of the Coflon film. Into the reflux condenser the diamine containing methanol solution was charged, and the Coflon film was added.

1.2 Sample Preparations

The virgin Coflon was dissolved in hot DMAC and cast into thin films. The solvent was allowed to evaporate prior to the beginning of the aging experiments.
1.3 **Test Conditions**

The tests were conducted by immersing the Coflon films in the refluxing methanol ethylene diamine solution. The test films were removed periodically for later chemical analysis. Photographs that exhibit the visual changes of the Coflon upon aging follow on the next page.

2.0 **CHEMICAL ANALYSIS ASSESSMENT OF AGING EFFECTS**

2.1 **Gel Permeation Chromatography of Aged Coflon**

Gel Permeation Chromatography was conducted on a number of environmentally aged Coflon test pieces. The solvent was NMP and the flow rate was 2mL/minute. The injection volume was 100 uL, the detector temperature was set at 30°C and the columns were run at 30°C.

<table>
<thead>
<tr>
<th>Sample Identification</th>
<th>Visual Appearance</th>
<th>Molecular Weights Mn, Mw, Mz</th>
<th>Exposure</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control Coflon</td>
<td>White, Translucent</td>
<td>81,700-538,900-3,217,000</td>
<td>Laboratory Exposure and Conditions</td>
</tr>
<tr>
<td>T-75 Coflon</td>
<td>White</td>
<td>177,460-536,509-1,062,095</td>
<td>Fluid F, 140°C 3 day</td>
</tr>
<tr>
<td>T-77 Coflon</td>
<td>Brown</td>
<td>170,504-532,717-1,052,564</td>
<td>Fluid F, 140°C 8 days</td>
</tr>
<tr>
<td>T-74 Coflon</td>
<td>White</td>
<td>152,905-534,361-1,080,943</td>
<td>Fluid F, 120°C 2 weeks</td>
</tr>
<tr>
<td>T-87 Coflon</td>
<td>White</td>
<td>171,265-538,897-1,067,315</td>
<td>27 days 120-135°C air oven</td>
</tr>
<tr>
<td>T-88 Coflon</td>
<td>White</td>
<td>177,880-549,827-1,084,282</td>
<td>Fluid F, 120°C 5 ksi 1 weeks</td>
</tr>
<tr>
<td>T-89 Coflon</td>
<td>White</td>
<td>173,803-539,103-1,073,977</td>
<td>Fluid F, 120°C 5 ksi 3 weeks</td>
</tr>
<tr>
<td>T-76 Coflon</td>
<td>White</td>
<td>173,329-534,908-1,059,213</td>
<td>Fluid F, 140°C 5 day</td>
</tr>
<tr>
<td>T-53 Coflon</td>
<td>Brown</td>
<td>158,990-519,823-990,690</td>
<td>Fluid F, 120°C 5 ksi 4 weeks gas phase</td>
</tr>
</tbody>
</table>
After Aging in Methanol + 1% Ethylene Diamine at Reflux

PVDF Solution (NMP) Dried on Hot Plate for 3 hours (Temp. 150-160°C)

New PVDF Films
X-Ray Diffraction - Crystallinity Changes

X-ray Diffraction was conducted on both samples from the exposed Tefzel and Coflon thermoplastics. The exposure fluids for the environmentally aged samples are detailed below.

1. Fluid A - 100% Methanol
2. Fluid B - 97/3 CH4/CO2 with saturated water vapor
3. Fluid C - 97/3 CH4/CO2
4. Fluid D - 94/5/1 CH4/CO2/H2S
5. Fluid E - 94/5/1 CH4/CO2/H2S with saturated water vapor
6. Fluid F - As fluid E plus 1% ethylene diamine
7. Fluid G - As fluid A plus 1% ethylene diamine
8. Fluid H - As Fluid B plus 1% ethylene diamine
9. Fluid I - MERL Formulated aromatic and aliphatic hydrocarbon solution

The percentage crystallinity was calculated on the samples. Twelve polymer samples were analyzed, ten PVDF (Coflon) samples and two Tefzel samples. The degree of crystallinity ranges from as low as 9% to as high as 48%. The Coflon samples were more crystalline. Below are the results of the analysis including the fluids the samples were exposed to prior to testing. Color changes were slight with both the Tefzel and Coflon materials in this batch of samples.

<table>
<thead>
<tr>
<th>Sample Identification</th>
<th>Visual Appearance</th>
<th>Percentage Crystallinity</th>
<th>Exposure</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control Tefzel</td>
<td>White Translucent</td>
<td>26.0%</td>
<td>Laboratory Temperature and Conditions</td>
</tr>
<tr>
<td>T-105 Tefzel</td>
<td>White Translucent</td>
<td>9%</td>
<td>Fluid A, 6 days 140°C vapor pressure</td>
</tr>
<tr>
<td>Control</td>
<td>White Translucent</td>
<td>41.5%</td>
<td>Laboratory Exposure and Conditions</td>
</tr>
<tr>
<td>Coflon</td>
<td>Translucent</td>
<td>39%</td>
<td>Fluid F, 3 months 100°C</td>
</tr>
<tr>
<td>T-73 Coflon</td>
<td>Beige</td>
<td>35</td>
<td>Fluid F, 120°C, 2 weeks</td>
</tr>
<tr>
<td>T-74 Coflon</td>
<td>Beige</td>
<td>31% prev. 26%</td>
<td>Fluid F, 140°C 3 days</td>
</tr>
<tr>
<td>T-75 Coflon</td>
<td>Beige</td>
<td>34% prev. 29%</td>
<td>Fluid E, 140°C 8 days</td>
</tr>
<tr>
<td>T-77 Coflon</td>
<td>White Translucent</td>
<td>40%</td>
<td>After 27 days at 120-130°C in air oven</td>
</tr>
<tr>
<td>T-87 Coflon</td>
<td>Beige</td>
<td>29% prev. 39%</td>
<td>Fluid F, 140°C 5 ksi 2 weeks</td>
</tr>
<tr>
<td>T-90 Coflon</td>
<td>Beige</td>
<td>37%</td>
<td>Fluid F, 140°C 5 ksi 4 weeks</td>
</tr>
<tr>
<td>T-91 Coflon</td>
<td>Brown</td>
<td>39%</td>
<td>Fluid G, 65°C, reflux 2 weeks</td>
</tr>
<tr>
<td>T-100 Coflon</td>
<td>Beige</td>
<td>48%</td>
<td>Fluid I, 140°C, 5 ksi 10 weeks</td>
</tr>
</tbody>
</table>
We can see clearly that all of the exposures for both materials had an effect on the percentage crystallinity. The greatest decrease in crystallinity for both polymers was observed with the Tefzel Fluid G exposure in methanol with ethylene diamine. We also note that the percentage of crystallinity for Coflon increases or decreases depending on the exposure. At the end of this report in Appendix we have attached a number of plots pertaining to the X-ray diffraction analysis of the Coflon and Tefzel test pieces.

2.3 **FTIR and Ultraviolet Analysis**

The Coflon films described in Section 1.0 were analyzed using Fourier Transform Infrared Spectroscopy (FTIR) and Ultraviolet (UV) spectroscopy. Changes in the molecular structure of PVDF were therefore determined as a function of exposure time. The aging media for the films was methanol with 1% ethylene diamine (EDA). Exposure times were 0, 1, 2, 3, 4 and 17 hours.

Film thicknesses were 0.01 mm for all tests. FTIR analyses were performed on a Nicolet 20 SX spectrophotometer in the transmission mode. A Varian DMS 200 UV/Vis spectrophotometer was used for the ultraviolet measurements.

**FTIR Results**

Appendix 4.1 displays the complete FTIR spectra for each Coflon film tested. Figure 2-1 compares an enlarged section of the baseline and 17 hour aged specimens. A clear indication of C=C formation is observed at 1700 cm\(^{-1}\). The decrease in the intensity of the peak at 1400 cm\(^{-1}\) may be due to loss of plasticizer (carbonyl) or carbon-fluorine.

**Ultraviolet Results**

The UV absorption at 290 nm of the PVDF films aged in methanol/EDA were recorded at room temperature. Absorption at this frequency is known to be associated with C=C double bonds, particularly conjugated trienes. The figure displayed in Appendix 4.2 shows the results for the five aged specimens and the baseline sample. An increase in the absorbance was found after aging of the PVDF films. This growth in absorbance translates into a decrease in percent transmittance in the samples. The absorbance change was found to increase with time of exposure to the methanol/amine mixture.

2.4 **Gas Chromatography couples with Mass Spectrometry**

We conducted GC/Mass Spec on Sample T-89 extracted with methanol. The extract showed primarily the presence of the dibutyl ester of decanedioic acid. Secondarily evident was the free acid decanedioic acid. The decanedioic acid is a degradation product of the plasticizer. The chromatograms are included in Appendix 4.4.
Figure 2-1. FTIR Spectra of Baseline and Aged Coflon Films

3.0 THERMAL STABILITY AND COMPOSITIONAL ANALYSIS

3.1 Creep Analysis using TMA

The Coflon films described in Section 1.0 were subjected to creep testing in TRI's 943 Thermomechanical Analyzer (TMA) after aging in methanol/EDA. The films were cut into 0.15 inch (3.8 mm) widths. The specimens were mounted in a tensile fixture. A dead load was applied sufficient to impose a stress of 500 psi. The TMA furnace was then placed around the sample compartment. The sample temperature was ramped up 10°C per minute to 150°C. In this manner the creep behavior was observed while the specimen was undergoing an increase in temperature.

Appendix Figure 4.1 displays the results for the unaged Coflon and three of the aged specimens. The specimen which was aged for 17 hours was to brittle to
fixture and is not indicated in Appendix Figure 4.1 for this reason. A significant increase in the resistance to creep with temperature was noted in the aged specimens. This behavior may be a result of dehydrofluorination and possible subsequent crosslinking within the polymer.

3.2 DSC of Aged Materials

A separate set of PVDF films was prepared for the DSC tests. The films were prepared as described in Section 1.0 with the exception of having a slightly thicker cross section. These specimens were again exposed to methanol/EDA for 0.25, 1.0, and 15 hours respectively. After exposure, sections of the films were subjected to Differential Scanning Calorimetry (DSC) testing. Sections were cut from the dry aged films and placed into tared DSC sample pans. The sample temperature was raised at a constant rate of 10°C per minute to a maximum of 200°C. Melt onset onset, peak temperature, and heat of fusion were thus determined.

DSC Results

DSC plots are displayed in Appendix 4.2. Tabular data describing the melt aspects and heats of fusion is shown in Table 3.1.

<table>
<thead>
<tr>
<th>Coflon Film Sample</th>
<th>Melt Onset (°C)</th>
<th>Melt Peak (°C)</th>
<th>Heat of Fusion (J/g)</th>
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<tr>
<td>Baseline</td>
<td>161.45</td>
<td>173.74</td>
<td>58.49</td>
</tr>
<tr>
<td>Aged 0.25 hrs</td>
<td>162.11</td>
<td>177.12</td>
<td>56.89</td>
</tr>
<tr>
<td>Aged 1.0 hrs</td>
<td>160.76</td>
<td>174.55</td>
<td>68.21</td>
</tr>
<tr>
<td>Aged 15 hrs</td>
<td>153.87</td>
<td>169.39</td>
<td>40.17</td>
</tr>
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</table>

A slight increase in melt temperatures was noted with the 0.25 hour aged specimen. At the 1 hour and 15 hour exposure level the temperatures appear to be decreasing. Most notable are the values obtained from the specimen which was aged 15 hours. Here the melt onset and peak temperatures dropped significantly as did the heat of fusion. This results lends credibility to the hypothesis that initial aging of PVDF in this fluid causes some crosslinking and at longer exposure times lower molecular weight species begin to appear which may lower the melt points and heats of fusion.
4.0 Appendix

4.1 TMA
Creep Behavior of Coflon Films
Aged in Methanol + 1% EDA

Graph showing percent strain versus temperature for different aging times.
4.2 FTIR and Ultraviolet Absorption Plots
FTIR Spectra
Coflon Film Unaged Baseline

[Graph showing FTIR spectroscopy data with absorbance on the y-axis and wavenumber in cm⁻¹ on the x-axis.]
FTIR Spectra
Coflon Film Aged 1 hr MeOH + 1% EDA Reflux
ITIR Spectra
Coflon Film Aged 2 hrs MeOH + 1% EDA Reflux
FTIR Spectra
Coflon Film Aged 3 hrs MeOH + 1% EDA Reflux

![FTIR Spectra Graph](image-url)
FTIR Spectra
Coflon Film Aged 4 hrs MeOH + 1% EDA Reflux
FTIR Spectra
Coflon Film Aged 17 hrs MeOH + 1% EDA Reflux
Aged Coflon Films
Changes in UV Absorption at 290 nm
After Exposure to Methanol + 1% EDA

Exposure Time (hrs)

Percent Transmission at 290 nm

0 5 10 15 20
4.2 DSC Plots
Sample: COFLON BASELINE
Size: 18.6000 mg
Method: 10R300

File: COF107.01
Operator: CPT
Run Date: 5-Dec-96 15:40

Heat Flow (W/g)

Temperature (°C)

161.45°C
58.49 J/g

173.74°C
Sample: COFLON 1HR MEOH + 1% EDA
Size: 24.5000 mg
Method: 10R300

DSC
File: COF107.05
Operator: CPT
Run Date: 5-Dec-96 16:26

Heat Flow (W/g)

Temperature (°C)

160.76°C
68.21 J/g

174.55°C
4.3 X-Ray Diffraction Plots
T90 shows reduced crystallinity

and additional phases forming
4.4 GC/Mass Spectrometry Chromatograms T-89
### Information from Data File:
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- **Operator**: Don Clay
- **Acquired**: 4 Dec 96 5:09 pm using AcqMethod MEOHSCAN.M
- **Sample Name**: MeOH extract of PVDF 12/04/96
- **Misc Info**: Vial Number: 1

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- C:\HPCHEM\DATABASE\WILEY.L

### Unknown Spectrum:
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### Integration Params:
- current RTEINT parameters

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Abundance

m/z -->

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Abundance

m/z ->

m/z ->

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#35446: Decanedioic acid, dibutyl ester (*)

m/z -->

m/z -->
4.5 Literature Search
(fluoropolymer
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coflon
polyvinylidene fluoride
tefzel
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lacs=cofalon
lacs=polyvinylidene fluoride
lacs=tefzel
end
\flexible pip
degradation
lacs=flexible pip
lacs=degradation
end

-- TECHNICAL REPORTS STATISTICS PAGE 1 OF 6 NOV 22, 1996

--TOTAL-SEARCH FINDS--------------------- 5  ARMY--  0
-- FIRST LEVEL FINDS-------------------- 217  NAVY--  0
-- FIRST AND SECOND LEVEL FINDS--------  5  AF----  0
-- 1+2+3 LEVEL FINDS--------------------  0  OTHLR-  0
-- 1+2+3+4 LEVEL FINDS------------------  0
-- 1 - AD NUMBER: C046455
-- 2 - FIELDS AND GROUPS: 1/3, 11/1, 1/3.3
-- 3 - ENTRY CLASSIFICATION: UNCLASSIFIED
-- 5 - CORPORATE AUTHOR: MATERIALS RESEARCH LABS ASCOT VALLE (AUSTRALIA)
-- 6 - UNCLASSIFIED TITLE: CHEMICAL DISEALED OF A FLUOROSILICONE SEALANT.
-- 8 - TITLE CLASSIFICATION: UNCLASSIFIED
-- 9 - DESCRIPTIVE NOTE: TECHNICAL REP.
-- 10 - PERSONAL AUTHORS: HANNA, FLER K. J.; HUANG, ROBERT H.; MAZURK.
-- 11 - REPORT DATE: MAR, 1990
-- 12 - PAGINATION: 31P MEDIA COST: $ 6.00 PRICE CODE: AA
-- 14 - REPORT NUMBER: MRL-1K-90-6
-- 18 - MONITOR ACRONYM: DODA
-- 19 - MONITOR SKILLS: AR-06-296
-- 20 - REPORT CLASSIFICATION: CONFIDENTIAL
-- 22 - LIMITATIONS (ALPHA): DISTRIBUTION AUTHORIZED TO DOD ONLY; OTHER
-- REQUESTS SHALL BE REFERRED TO EMBASSY OF AUSTRALIA. ATTN: JOAN
-- E.LISS. HEAD. PUB. SEC.-DL/SCI., 1601 MASSACHUSETTS AVE., NW.
-- WASHINGTON, DC 20036.
-- COMPOUNDS, LEAKAGE(FLUID), FLUOROPOLYMERS, POLYMERS, DEGRADATION.
-- 24 - DESCRIPTOR CLASSIFICATION: UNCLASSIFIED
-- 25 - IDENTIFIERS: *WING FUEL TANKS, 04-2817 FLUOROSILICONE SEALANT.
-- RAAF F-111 AIRCRAFT, F-111 AIRCRAFT.
-- 26 - IDENTIFIER CLASSIFICATION: UNCLASSIFIED
-- 29 - INITIAL INVENTORY: 2
-- 32 - REGRADE CATEGORY: 
-- 33 - LIMITATION CODES: 14
-- 35 - SOURCE CODE: 409014
-- 36 - ITEM LOCATION: DTIC
-- 40 - GEOPOlITICAL CODE: AU
-- 41 - TYPE CODE: 8
-- 43 - IAC DOCUMENT TYPE: 
--- ******
-- 2 - OF 5
-- 1 - AD NUMBER: B209155
-- 2 - FIELDS AND GROUPS: 11/9, 11/10, 18/6
-- 3 - ENTRY CLASSIFICATION: UNCLASSIFIED
-- 5 - CORPORATE AUTHOR: BATTIELE MEMORIAL INST. COLUMBUS OH RADIATION
-- EFFECTS INFORMATION CENTER
-- 6 - UNCLASSIFIED TITLE: THE EFFECT OF NUCLEAR RADIATION ON
-- FLUOROPOLYMERS.
-- 8 - UNCLASSIFIED CLASSIFICATION: UNCLASSIFIED
-- 9 - DESCRIPTIVE NOTE: MEMORANDUM RLPL.
-- 10 - PERSONAL AUTHORS: BROADWAY, N. J.; PALINCHAK, S.
-- 11 - REPORT DATE: JUN 30, 1959
-- 12 - PAGINATION: 31P MEDIA COST: $ 6.00 PRICE CODE: AA
-- 14 - REPORT NUMBER: REIC-M-17
-- 15 - CONTRACT NUMBER: AF 33(616)-6564
-- 16 - PROJECT NUMBER: 2133
-- 18 - MONITOR ACRONYM: XC
-- 19 - MONITOR SKILLS: USAF
-- 20 - REPORT CLASSIFICATION: UNCLASSIFIED
BECAUSE OF THEIR CHEMICAL INERTNESS, HIGH-TEMPERATURE RESISTANCE, AND GOOD ELECTRICAL PROPERTIES, SINCE THEY ARE RESISTANT TO FUMING NITRIC ACID AND OTHER OXIDIZING AGENTS, SUCH MATERIALS AS TEFON AND KEL-F ARE ESSENTIAL MATERIALS FOR VALVES, HOSES, AND ELECTRICAL INSULATION. THEIR HEAT RESISTANCE HAS MADE THEM INVALUABLE FOR HIGH-TEMPERATURE APPLICATIONS. THIS RESISTANCE TO HEAT AND CHEMICALS IS DUE TO THE CARBON-FLUORINE BOND WHICH IS CHEMICALLY ONE OF THE MOST STABLE OF ORGANIC LINKAGES. BECAUSE OF THEIR IMPORTANT APPLICATIONS, THESE MATERIALS WERE INVESTIGATED FOR USE IN A RADIATION ENVIRONMENT. IT WAS FOUND THAT THE RADIATION RESISTANCE OF FLUOROPOLYMERS IS NOT AS GOOD AS THEIR CHEMICAL AND HEAT RESISTANCE. TEFON HAS VERY POOR RADIATION RESISTANCE, WHILE THE OTHER FLUOROPOLYMERS ARE ONLY SLIGHTLY BETTER. THIS MEMORANDUM SUMMARIZES THE AVAILABLE DATA ON THE RADIATION STABILITY OF TEFON AND KEL-F PLASTICS, AND KEL-F, POLYFLUOROBUTYL ACRYLATE (POLYFBA) HEXAFLUOROPENTAMETHYLENE ADIPATE (A POLYESTER), AND SILASTIC LS-53 (A FLUORINATED SILICONE) ELASTOMERS.
PROJECI NUMBER: 3005
TASK NUMBER: 50
MONITOR ACRONYM: WRDC

PROJECTIONS

REPORT LHOOIFICAT ZOn'4" UNCLASSIFIED

DESCRIPTIONS:

*BIODEGREATION, *RUBBER COATINGS, *ELASTOMERS,
*MICROORGANISMS, *PAINTS, *SEALING COMPOUNDS, AGAR, AIR FORCE,
AIRCRAFT, BIOLOGY, CHEMICALS, COMPOSITE MAILKIALS, CORROSION
INHIBITION, DAMAGE, DEGRADATION, DETERIORATION, EFFICIENCY, ENZYMES,
FLUORINE COMPOUNDS, FUNGICIDES, GERMICIDES, HAZARDS, MAINTENANCE,
METABOLISM, NUTRIENTS, POLYSULFIDES, REMOVAL, REPAIR, SILICONES,
SOUP, SURFACES, TEL, FLUOROTXYLYLENE RESINS, FLUOROPOLYMERS.

PROJECTED CLASSIFICATION: UNCLASSIFIED
IDENTIFIERS:
PL65502F, WWDC3055090, EXPORT CONTROL,
FLUOROSILICONE POLYSULFIDE, POLYSULFIDE ELASTOMERS.
IDENTIFIER CLASSIFICATION: UNCLASSIFIED
INITIAL INVENTORY: 1
LIMITATION CODEs: 3 57
SOURCE SERIES: F
SOURCE CODE: 397355

ITEM LOCATION: DTIC
GEOPOLITICAL CODE: 4902
TYPE CODE: 4
IAC ACCESSION NUMBER: PL-053019
IAC DOCUMENT TYPE:
PLASTEC-MICROFICHE
IAC SUBJECT TERMS:
P-(U)CORROSION RESISTANCE, MICROBIOLOGICAL
DETERIORATION, BIODEGRADATION, POLYSULFIDES, FLUOROSILICONE,
SEALANTS, FLUOROCARBONS, MONITORING, DEGRADATION, FOURIER TRANSFORM
IR, SPECTROSCOPY, PAINTS, ENZYME DEGRADATION, TEFLON, ELASTOMERS,
P1F, COATINGS, ZZ CONTROLLED USGO.,;

---

1 OF 5

AD NUMBER: A294734
FIELDS AND GROUPS: 7/3, 7/4, 7/6, 8/8
ENTRY CLASSIFICATION: UNCLASSIFIED
CORPORATE AUTHOR: COLD REGIONS RESEARCH AND ENGINEERING LAB
HANOVER NH
UNCLASSIFIED TITLE: SUSCEPTIBILITY OF ABS, FEP, TFE, FEP, PTFE,
AND PVC WELL CASINGS TO DEGRADATION BY CHEMICALS.
TITLE CLASSIFICATION: UNCLASSIFIED
DESCRIPTION NOTE: SPECIAL REPT.

PERSONAL AUTHORS: KANNLY, THOMAS A.; PARRLER, LOUISE V.
REPORT DATE: JAN, 1995
PAGINATION: 24P
MLUDIA COST: 1 6.00 PRICE CODE: AA
REPORT NUMBER: CRREL-SR-95-1
MONITOR ACRONYM: 3FIN-MEC-LT, XA
LESS COMMONLY USED MATERIALS FOR CASING GROUNDWATER MONITORING

WELLS: ACRYLONITRILE BUTADIENE STYRENE (ABS), FLUORINATED ETHYLENE

PROPYLENE (FEP), FIBERGLASS-REINFORCED EPOXY (FRE), AND FIBERGLASS-

REINFORCED PLASTIC (FRP), WITH TWO MORE COMMONLY USED CASING

MATERIALS: POLYVINYL CHLORIDE (PVC) AND POLYTETRAFLUOROETHYLENE

(PFTE). THE SIX MATERIALS WERE EXPOSED TO 28 NEAR ORGANIC COMPOUNDS

(INCLUDING ONE ACID) AND 10 EXTREMELY ACIDIC AND ALKALINE

CONDITIONS FOR UP TO 112 DAYS. THIS WAS DONE TO SIMULATE SOME OF

THE MOST AGGRESSIVE ENVIRONMENTS THAT MONITORING WELL CASINGS MAY

BE EXPOSED TO. THE CASINGS WERE OBSERVED FOR CHANGES IN WEIGHT AND

SIGNS OF PHYSICAL DEGRADATION (SWELLING, SOFTENING, DECREASE IN

STRENGTH, DETERIORATION, OR DISSOLUTION). AS EXPECTED, THE TWO

FLUORINATED POLYMERS (FEP AND PFTE) WERE THE MOST INERT MATERIALS

TESTED. THEY WERE NOT DEGRADED BY ANY OF THE TEST CHEMICALS,

ALTHOUGH SAMPLES EXPOSED TO A FEW ORGANIC CHEMICALS DID SHOW A

SLIGHT WEIGHT GAIN (APPROX. 1%). AMONG THE NONFLUORINATED PRODUCTS

TESTED, FRE WAS THE MOST INERT. ALL ORGANIC CHEMICALS CAUSED

PARTICLES TO FLAKE FROM THE FRE SURFACE, FOLLOWED BY SEPARATION OF

THE GLASS FIBERS, AND TWO ORGANIC CHEMICALS CAUSED WEIGHT GAINS

EXCEEDING 10%. ALSO, HIGHLY ACIDIC CONDITIONS (PH<1) DEGRATED THIS

MATERIAL, AND THIS MAY LIMIT THE USE OF THIS MATERIAL IN ACIDIC

ENVIRONMENTS. JG

ABSTRACT CLASSIFICATION: UNCLASSIFIED
EXPERIMENTAL RESIN C8/1SA AS A STRUCTURAL MATERIAL AND FOR USE IN BLENDS AND COMPOSITES.

TITLE CLASSIFICATION: UNCLASSIFIED

PERSONAL AUTHORS: TWARDOWSKI, T. E.; GEIL, P. H.

REPORT DATE: JUN 01, 1990

PUBLISHING AGENCY: U1UC-NCCN-99-0013

--

DESCRIPTION: TECHNICAL

PERSONAL AUTHORS:

TWAROWSKI, T. E.; GEIL, P. H.

--

PRICE: AA

--

ABSTRACT: EPOXY THERMOSETS ARE CURRENTLY THE MATERIAL OF CHOICE FOR HIGH PERFORMANCE COMPOSITE MATERIALS. THEY HAVE HIGH MODULUS, LOW WEIGHT, EXCELLENT ADHESION TO THE FILLER COMPONENT, AND HIGH DIMENSIONAL AND THERMAL STABILITY AS WELL AS PROCESSING CHARACTERISTICS MUCH MORE TRACTABLE THAN THERMOPLASTICS. UNFORTUNATELY, EPOXIES ARE OFTEN ATTACKED BY ENVIRONMENTAL ELEMENTS, ESPECIALLY WATER, RESULTING IN A DEGRADATION OF PROPERTIES. AS SUCH, MODIFICATION OF EPOXIES TO IMPROVE THEIR ENVIRONMENTAL RESISTANCE IS VALUABLE CONSIDERATION. EPOXY RESINS INCORPORATING LARGE FLUORINE CONTENTS HAVE MET THIS CHALLENGE, SHOWING LOW CONTACT ANGLE, LOW MEASURE UPTAKE AND IMPROVED FLAME RESISTANCE. WHAT REMAINS IS THE NEED TO INVESTIGATE THE SUITABILITY OF SUCH RESINS FOR USE IN STRUCTURAL MATERIALS.

--

ABSTRACT CLASSIFICATION: UNCLASSIFIED

--

SOURCE CODE: 422821

--

ITEM LOCATION: U1IC/NI1S

--

GEOGRAPHICAL CODE: 1719

--

TYPE CODE: 0

--

IAC ACCESSION NUMBER: PL-053137

--

IAC DOCUMENT TYPE: PLASTEC-MICROFICHE

--

IAC SUBJECT TERMS: (U)MODIFIER MOLECULE, INTERFACE STRENGTH, CHARACTERIZATION, FLUORINATED EPOXY, GLASS TRANSITION, DSC, MOISTURE ABSORPTION, BLENDS, COMPOSITES, EPON 828, SWELLING, EPOXY 828, AGING, STRESS STRAIN, ADHESION, THERMAL STABILITY, PROPERTY DEGRADATION, WATER IMMERSION, TENSILE PROPERTIES, ZZ UNLIMITED.

--

<<ENTER NEXT COMMAND>>
### Status: Path 1 of [Dialog]

### Status: Initializing Port COM2 using (Baud 19200 Hendshake XON/XOFF DataBits 7 Parity Even StopBits 1)

### Status: Initializing modem ...
ATEIQ0V1X4&Cl&D2
OK
### Status: Dialing primary number (448-4611)...

ATDT448-4611
PROTOCOL: LAP-M
COMPRESSION: NONE

CONNECT 19200
### Status: Connection established at 19200 baud
/ARQ

please type your terminal identifier
-3523:01-005-
please log in: DIALOG

DIALOG: call connected
### Status: Connected

DIALOG INFORMATION SERVICES
PLEASE LOGON:
******** HHHHHHHH SSSSSSSS?
### Status: Signing onto Dialog
********
ENTER PASSWORD:
******** HHHHHHHH SSSSSSSS? ********
Welcome to DIALOG
### Status: Connected

Dialog level 42.10.06D

Last logoff: 22nov96 09:36:40
Logon file405 22nov96 10:54:50
Banner display set OFF.
HILIGHT set on as '**'
KWIC is set to 5.
BLIP set on
NOTICE set ON to $25.00
You will be prompted to confirm each TYPE or PRINT request where format charges exceeds $25.00
COST = ONESEARCH.
Please enter SUBACCOUNT name/number:
?bulluck/n7402
Is BULLUCK/N7402 the SUBACCOUNT you want to use? (Y/N)
?y
Subaccount is set to BULLUCK/N7402
SYSTEM: HOME
Menu System II: D2 version 1.7.8 term=ASCII
Terminal set to DLINK

*** DIALOG HOMEBASE(SM) Main Menu ***

Information:
1. Announcements (new files, free connect time, price changes, etc.)
2. Database, Rates, & Command Descriptions
3. Help in Choosing Databases for Your Topic
4. Customer Services (telephone assistance, training, seminars, etc.)
5. Product Descriptions
Connections:
6. DIALOG Menus (SM)
7. DIALOG Business Connection (R) and DIALOG Headlines (SM)
8. KR SourceOne (SM) Document Delivery
9. Data Star (R)
10. Other Online Menu Services & Files (MoneyCenter(R), OAG, TNT, etc.)

(c) 1995 Knight-Ridder Information, Inc.

/H = Help                  /L = Logoff                  /NOMENU = Command Mode

Enter an option number to view information or to connect to an online service. Enter a BEGIN command plus a file number to search a database (e.g., B1 for ERIC).

File 411: DIALINDEX(R)

DIALINDEX(R)
(c) 1996 Knight-Ridder Info

*** DIALINDEX search results display in an abbreviated ***
*** format unless you enter the SET DETAIL ON command. ***

?sf chemlit, chemeng, plastics, material
You have 43 files in your file list.
(To see banners, use SHOW FILES command)

?s ((pvdf or coflon or polyvinylidene()fluoride or tefzel) and (flexible()pip? or degradation))/ti,de,id

Your SELECT statement is:

s ((pvdf or coflon or polyvinylidene()fluoride or tefzel) and (flexible()pip? or degradation))/ti,de,id

<table>
<thead>
<tr>
<th>Items</th>
<th>File</th>
</tr>
</thead>
<tbody>
<tr>
<td>4</td>
<td>2: INSPEC 1969-1996/Nov W3</td>
</tr>
<tr>
<td>10</td>
<td>8: EI Compendex(R) 1970-1996/Dec W4</td>
</tr>
</tbody>
</table>

>>>Term "ID" is not defined in file 144 and is ignored

14 144: Pascal 1973-1996/Oct

>>>Term "ID" is not defined in file 322 and is ignored

1 322: Polymer Online

>>>Term "DE" is not defined in file 340 and is ignored


>>>Term "ID" is not defined in file 340 and is ignored

2 340: CLAIMS(R)/US PATENTS ABS 1950-1996/SEP

>>>File 399 processing for PIP? stopped at PIPERAZINYLALKANOYLDIHYDROBENZOFURANS

84 399: CA SEARCH(R) 1967-1996/UD=12522


6 6: NTIS 64-1996/Dec W5

2 293: Eng Materials Abs(R) 1986-1996/Dec

>>>Term "ID" is not defined in file 108 and is ignored

1 108: Aerospace Database 1962-1996/Nov

13 files have one or more items; file list includes 43 files.
One or more terms were invalid in 26 files.
?save temp
Temp SearchSave "TD228" stored
?b hits
  22nov96 10:57:23 User036172 Session D803.2
  Sub account: BULLUCK/N7402
  $1.50  0.050 Hrs File411
  $1.50  Estimated cost File411
  $0.60 TyMNet
  $2.10  Estimated cost this search
  $2.11  Estimated total session cost  0.051 Hrs.

SYSTEM: OS - DIALOG OneSearch
*File 399: Use is subject to the terms of your user/customer agreement. 
For format prices, including formats 6 & 8, see HELP RATES 399.
You have 13 files in your file list.
(To see file names, coverage dates, and copyright notices, enter SHOW FILES.) 

  --- ----- --------
  Set Items Description
  ?exs td228
  Hilight option is not available in file(s) 6, 293, 399
  HILIGHT set on as '*' 
  KWIC is set to 5.
  KWIC option is not available in file(s) 6, 293, 399
  >>>Term "ID" is not defined in one or more files
  >>>Term "DE" is not defined in one or more files
  >>>File 399 processing for PIP? stopped at PIPERAZINYLALKANOYLDIHYDROBENZO 
  FURANS 
  Processed 10 of 13 files ...
  Completed processing all files 
  4969 PVDF/TI, DE, ID 
  3 COFLON/TI, DE, ID 
  12502 POLYVINYLIDENE/TI, DE, ID 
  194611 FLUORIDE/TI, DE, ID 
  9055 POLYVINYLIDENE/TI, DE, ID(W) FLUORIDE/TI, DE, ID 
  141 TEFZEL/TI, DE, ID 
  121920 FLEXIBLE/TI, DE, ID 
  365537 PIP?/TI, DE, ID 
  900 FLEXIBLE/TI, DE, ID(W) PIP?/TI, DE, ID 
  268842 DEGRADATION/TI, DE, ID 
  S1 272 ((PVDF OR COFLON OR POLYVINYLIDENE()FLUORIDE OR TEFZEL)) 
  AND (FLEXIBLE() PIP? OR DEGRADATION))/TI, DE, ID 

?rd s1
  >>>Duplicate detection is not supported for File 94.
  >>>Duplicate detection is not supported for File 322.
  >>>Duplicate detection is not supported for File 340.

  >>>Records from unsupported files will be retained in the RD set. 
  ...examined 50 records (50) 
  ...examined 50 records (100) 
  ...examined 50 records (150) 
  ...examined 50 records (200) 
  ...examined 50 records (250) 
  ...completed examining records 
  S2 253 RD S1 (unique items)

?s s2/eng
  >>>Term "ENG" is not defined in one or more files

S3 183 S2/ENG

?s s3 and py=1992:8888
  >>>One or more prefixes are unsupported
  >>> or undefined in one or more files. 
  Processed 10 of 13 files ...
  Completed processing all files 
  183 S3 
  13684327 PY=1992 : PY=8888
Title: Piezo- and pyroelectric properties of dehydrofluorinated *PVDF* films
Conference Title: Proceedings of the 8th International Symposium on Electrets (ISE 8)

Title: Advanced *flexible* *pipe* materials for aggressive hydrocarbon service

Title: Application of TGA/FTIR to the thermal *degradation* mechanism of tetrafluoroethylene-propylene copolymers

Title: XPS studies of radiation-induced structural changes in *polyvinylidene* *fluoride*

Title: X-ray induced *degradation* of poly(vinylidene fluoride) films.

Systematic Peptide Fragmentation of Polyvinylidene Difluoride(*PVDF*)-Immobilizes Proteins Prior to Microsequencing.

Thermal decomposition kinetics of a commercial fluoropolymer

KeV ion beam irradiation of *polyvinylidene* *fluoride* (*PVDF*)

Temperature influence on the gas desorption from *polyvinylidene* *fluoride* (*PVDF*) irradiated with helium beams

A FTIR study of *PVDF* irradiated by means of swift heavy ions

Utilization of *PVDF* sensors to determine impact damage in graphite/epoxy plates by acousto-ultrasonic technique
4/6/12 (Item 6 from file: 144)
11349383 PASCAL No.: 94-0171771
Monosaccharide and oligosaccharide analysis of glycoproteins
electrotransferred onto *PVDF* membranes

4/6/13 (Item 7 from file: 144)
10964625 PASCAL No.: 93-0474091
Retention of beer spoilage microorganisms by *polyvinylidene* *fluoride*
microporous membranes with various retention ratings

4/6/14 (Item 1 from file: 323)
00585817
TITLE: FAMILY FORTUNES

4/6/15 (Item 2 from file: 323)
00581044
TITLE: RADIATION EFFECTS ON FLUOROPOLYMERS: RADIATION-INDUCED STRUCTURAL
AND CRYSTALLINITY CHANGES OF *TEFZEL*'

4/6/16 (Item 3 from file: 323)
00577457
TITLE: COMPARISON OF HIGH PERFORMANCE, CLEAR FILMS USED FOR LONG TERM
PRODUCTION

4/6/17 (Item 4 from file: 323)
00577237
TITLE: POLYMER COMPOSITIONS INTENDED FOR THE MANUFACTURE OF CABLES AND
*FLEXIBLE* *PIPES* AND ARTICLES BASED ON THESE COMPOSITIONS

4/6/18 (Item 5 from file: 323)
00576607
TITLE: ACID-BASE INTERACTIONS AT POLYMER INTERFACES

4/6/19 (Item 6 from file: 323)
00576590
TITLE: CRYSTALLISATION BEHAVIOUR AND PHASE DIAGRAM OF EXTENDED-CHAIN
CRYSTALS OF POLY(VINYLIDENE FLUORIDE) UNDER HIGH PRESSURE

4/6/20 (Item 7 from file: 323)
00564145
TITLE: RADIATION-INDUCED ENHANCEMENT OF CRYSTALLINITY IN POLYMERS

4/6/21 (Item 8 from file: 323)
00559287
TITLE: ELECTROCHEMISTRY AS THE WAY TO TRANSFORM POLYMERS

4/6/22 (Item 9 from file: 323)
00559286
TITLE: ELECTROCHEMICALLY INDUCED FUNCTIONALISATION OF FLUOROCONTAINING
POLYOLEFINS

4/6/23 (Item 10 from file: 323)
00559052
TITLE: BLENDS OF GLYCIDYL METHACRYLATE(GMA)/METHYL METHACRYLATE(MMA)
COPOLYMERS WITH *POLYVINYLIDENE* *FLUORIDE*

4/6/24 (Item 11 from file: 323)  00554472
TITLE: POLYMER SUPPORTS IN SYNTHESIS

4/6/25 (Item 12 from file: 323)  00549288
TITLE: YIELD BEHAVIOUR OF *PVDF* AND THE DEFORMATION PROCESS AT HIGH TEMPERATURE

4/6/26 (Item 13 from file: 323)  00509576
TITLE: STUDY OF THE THERMAL *DEGRADATION* OF POLYCHLOROTRIFLUOROETHYLENE, *POLYVINYLIDENE* *FLUORIDE* AND COPOLYMERS OF CHLOROTRIFLUOROETHYLENE AND VINYLIDENE FLUORIDE

4/6/27 (Item 14 from file: 323)  00496424
TITLE: RADIATION EFFECTS ON *PVDF*

4/6/28 (Item 15 from file: 323)  00492182
TITLE: GRAFTING OF SILICON PHTHALOCYANINE DICHLORIDE ONTO *PVDF* FILM SURFACES BY ULTRASOUND

4/6/29 (Item 16 from file: 323)  00474223
TITLE: PLASTICS HEAT EXCHANGERS

4/6/30 (Item 17 from file: 323)  00472912
TITLE: SEEKING THE PERFECT BLEND

4/6/31 (Item 18 from file: 323)  00470711
TITLE: X-RAY PHOTOELECTRON SPECTROSCOPY(XPS) STUDIES OF RADIATION-INDUCED STRUCTURAL CHANGES IN *POLYVINYLIDENE* *FLUORIDE*

4/6/32 (Item 19 from file: 323)  00448524
TITLE: REACTIONS OF ATOMIC OXYGEN WITH POLYMER FILMS

4/6/33 (Item 1 from file: 340)  2620850 9516422
C/POLYMER COMPOSITIONS INTENDED FOR THE MANUFACTURE OF CABLES AND *FLEXIBLE* *PIPES* AND ARTICLES BASED ON THESE COMPOSITIONS; *POLYVINYLIDENE* *FLUORIDE* BLEND

4/6/34 (Item 2 from file: 340)  2439164 9402057
C/METHOD FOR EXTRUDING POLYOLEFINS CONTAINING VINYLIDENE CHLORIDE POLYMERS; COATING EQUIPMENT WITH *POLYVINYLIDENE* *FLUORIDE* TO PREVENT *DEGRADATION* OF THE CHLOROPOLYMER
KeV-MeV ion irradiation of polyvinylidene fluoride (PVDF) films

Photodegradation of fluorocarbon resin films. A new evaluation method using an ESR spectrometer

Testing the utility of EIS measurements to predict and monitor the behavior of organically coated aluminum during atmospheric exposure

Swift heavy ion modification of polymers

Radical in heavy ion-irradiated polyvinylidene fluoride

Photoacoustic detection of the decomposition kinetics of polymers: interpretation of acoustic signals

Crystallite damage studies on irradiated poly(vinylidene fluoride)

Derivatization of polyvinylidene difluoride membranes for the solid-phase sequence analysis of a phosphorylated sea urchin embryo histone H1 peptide

Surface analysis of poly(vinylidene difluoride) membranes

Low dose $\gamma$-irradiation of some fluoropolymers: effect of polymer chemical structure
Title: THERMAL-DECOMPOSITION KINETICS OF A COMMERCIAL FLUOROPOLYMER (Abstract Available)

4/6/46 (Item 1 from file: 6)
1787467 NTIS Accession Number: PB94-886397/XAB
Polyvinylidene Fluoride: Structure and Degradation. (Latest citations from the Rubber and Plastics Research Association Database)
(Published Search)
NTIS Prices: PC N01/MF N01

4/6/47 (Item 1 from file: 293)
126857
Thermal Degradation of Commercial Fluoropolymers in Air.

### Status: Signing Off...
logoff
22nov96 11:01:37 User036172 Session D803.3
Sub account: BULLUCK/N7402
$7.38 0.083 Hrs FileOS
$3.50 47 Types
$10.88 Estimated cost FileOS
OneSearch, 13 files, 0.083 Hrs FileOS
$1.00 TYMNET
$11.88 Estimated cost this search
$13.99 Estimated total session cost 0.134 Hrs.

### Status: Signed Off. (7 minutes)
### Status: Path 1 of [Dialog]

### Status: Initializing Port COM2 using (Baud 19200 Handshake XON/XOFF DataBits 7 Parity Even StopBits 1)

### Status: Initializing modem ...
ATEIQOVIX4&C16D2
OK
### Status: Dialing primary number (448-4611)...
ATDT448-4611
PROTOCOL: LAP-M

COMPRESSION: NONE

CONNECT 19200
### Status: Connection established at 19200 baud
/ARQ

please type your terminal identifier
-3523:01-004-
please log in: DIALOG

DIALOG: call connected
### Status: Connected

DIALOG INFORMATION SERVICES
PLEASE LOGON:
   ******** HHHHHHHH SSSSSSSS?
### Status: Signing onto Dialog
********
ENTER PASSWORD:
   ******** HHHHHHHH SSSSSSSS? ********
Welcome to DIALOG
### Status: Connected

Dialog level 42.10.06D

Last logoff: 22nov96 11:01:38
Logon file405 22nov96 15:00:32
Banner display set OFF.
HILIGHT set on as '*'
KWIC is set to 5.
BLIP set on
NOTICE set ON to $25.00
You will be prompted to confirm each TYPE or PRINT request where format charges exceeds $25.00
COST = ONESEARCH.
Please enter SUBACCOUNT name/number:
?bulluck/n7402
Is BULLUCK/N7402 the SUBACCOUNT you want to use? (Y/N)
?y
Subaccount is set to BULLUCK/N7402
SYSTEM:HOME
Menu System II: D2 version 1.7.8 term=ASCII
Terminal set to DLINK

*** DIALOG HOMEBASE(SM) Main Menu ***

Information:
1. Announcements (new files, free connect time, price changes, etc.)
2. Database, Rates, & Command Descriptions
3. Help in Choosing Databases for Your Topic
4. Customer Services (telephone assistance, training, seminars, etc.)
5. Product Descriptions
Enter an option number to view information or to connect to an online service. Enter a BEGIN command plus a file number to search a database (e.g., B1 for ERIC).

?b8

22nov96 15:00:36 User036172 Session D804.1
Sub account: BULLUCK/N7402
$0.00 0.001 Hrs FileHomeBase
$0.00 Estimated cost FileHomeBase
$0.01 TYMNET
$0.01 Estimated cost this search
$0.01 Estimated total session cost 0.001 Hrs.

File 8:Ei Compendex(R) 1970-1996/Dec W4
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Set Items Description
--- ----- -------------
?k 03681363;k 03641106
S0 1 03681363
S0 2 03641106
?t 0/5/all

0/5/1
DIALOG(R) File 8:Ei Compendex(R)
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03681363 E.I. No: EIP93081046180
Title: Advanced flexible pipe materials for aggressive hydrocarbon service
Author: Hill, R.T.; Measamer, J.C.
Corporate Source: Wellstream Corp, Panama City, FL, USA
Conference Location: Singapore, Singapore Conference Date: 19930606-19930611
E.I. Conference No.: 18719
Publication Year: 1993
Language: English
Document Type: CA; (Conference Article) Treatment: X; (Experimental)
Journal Announcement: 9309W4
Abstract: The increasing development of marginal offshore hydrocarbon deposits has resulted in production of increasingly corrosive fluids. This has increased the need for pipelines capable of operating at elevated temperatures in the presence of high concentrations of H//2S and CO//2 gases. Conventional pipelines requires the use of stainless steels or corrosion resistant alloys which drives up the cost of materials and fabrication/installation of the system. The use of flexible pipe systems is assuming as important role in these applications where stainless steel and
CRA carcass materials may be economically utilized. This paper discusses testing of flexible pipe carcass materials and a new PVDF homopolymer fluid barrier material. (Author abstract) 7 Refs.

Descriptors: *Offshore pipelines; Submarine pipelines; Corrosion resistance; Seawater effects; Plastic pipe
Identifiers: Flexible pipes; Plasticized homopolymer; Corrosive fluids; PVDF homopolymer. Carcass layer

Classification Codes:
619.1.2 (Pipe Materials)
619.1 (Pipe, Piping & Pipelines); 539.2 (Corrosion Protection); 511.2 (Oil Field Equipment)
619 (Pipes, Tanks & Accessories); 539 (Metals Corrosion & Protection);
511 (Oil Field Equipment & Production Operations)
61 (PLANT & POWER ENGINEERING); 53 (METALLURGICAL ENGINEERING); 51 (PETROLEUM ENGINEERING)

Title: Application of TGA/FTIR to the thermal degradation mechanism of tetrafluoroethylene-propylene copolymers
Author: Schild, H.G.
Publication Year: 1993
CODEN: JPACEC ISSN: 0887-624X
Language: English
Document Type: JA; (Journal Article) Treatment: A; (Applications); G; (General Review); X; (Experimental)
Journal Announcement: 9307W4

Abstract: Aflas, produced by Asahi Glass Co. Ltd., Tokyo, Japan and marketed in the United States by 3M, is a high-performance elastomeric tetrafluoroethylene (TFE)/propylene copolymer with superior high temperature stability, electrical, and chemical resistance properties. (Edited author abstract) 14 Refs.

Descriptors: *Copolymers; Polypropylenes; Polytetrafluoroethylenes; Pyrolysis; Thermoanalysis; Infrared spectroscopy
Identifiers: Aflas; Tefzel; Thermogravimetric analysis; Fourier transform infrared spectroscopy; Tetrafluoroethylene; Reaction mechanism

Classification Codes:
815.1.1 (Organic Polymers)
815.1 (Polymeric Materials); 802.2 (Chemical Reactions); 801.1 (Chemistry, General)
815 (Plastics & Polymeric Materials); 802 (Chemical Apparatus & Plants);
801 (Chemical Analysis & Physical Chemistry)
81 (CHEMICAL PROCESS INDUSTRIES); 80 (CHEMICAL ENGINEERING)
Thermal decomposition kinetics of a commercial fluoropolymer (Tefzel) in the temperature range 359-550 Degree C by thermogravimetry (TG) and 235-270 Degree C by differential scanning calorimetry (DSC) was investigated under nitrogen flux. The temperature range studied (225-550 Degree C) includes the range of recommended continuous use and processing temperature of these commercial resins. The activation energy (DELTA E) for decomposition was 53.33 k.cal mole SUP -1 in TG at 10 Degree C/min rate, whereas it was 9.49 k.cal mole SUP -1 in DSC. The effect of heating rate on TG and DSC thermograms was studied.

English Descriptors: Fluoroelastomer; Ethylene copolymer; Tetrafluoroethylene copolymer; Commercial form; Thermal degradation; Kinetics; Rate constant; Activation energy; Heating rate; Experimental study

French Descriptors: Caoutchouc fluor; Ethylene copolymere; Ethylene(tetrafluoro) copolymere; Forme commerciale; Degradation thermique; Cinetique; Constante vitesse; Energie activation; Vitesse chauffage; Etude experimentale; Tefzel
ABSTRACT: Tensile yield stress measurements on PVDF were carried out, covering as many as seven decades of strain rates at temps. from -50 to 150°C. The data were analysed on the basis of four Ree-Eyring processes acting in parallel, three of which were identified. At high temps. and low strain rates, a threshold yield stress was observed. A model was proposed, permitting description of the yield behaviour both at the threshold and in its adjacent strain rate dependent range. It consisted of a modification of the Ree-Eyring theory using an asymmetrical free energy barrier for the rate process. The data were consistent with the above model, which implied that yielding occurred only when the stress was higher than a threshold. In this case, a rejuvenation (or de-ageing) effect should be assumed in the deformation process which could consist of a melting of a chain segment located between two folds. 21 refs.

SUBJECT HEADING (RAPRA): TENSILE PROPERTIES, PVDF; VINYLIDENE FLUORIDE POLYMERS, tensile properties

GEOGRAPHIC LOCATION: BELGIUM; EUROPEAN COMMUNITY; EUROPEAN UNION; WESTERN EUROPE

DESCRIPTORS: DATA; DEFORMATION; FLUOROPOLYMER; FREE ENERGY; GRAPH; HIGH TEMPERATURE; LOW TEMPERATURE; MECHANICAL PROPERTIES; MELTING; MODEL; PLASTIC; POLYVINYLIDENE FLUORIDE; PROPERTIES; PVDF; STRAIN RATE; TABLES; TECHNICAL; TEMPERATURE; TENSILE PROPERTIES; THEORY; THERMAL DEGRADATION; THERMOPLASTIC; YIELD; YIELD STRESS

RAPRA CLASSIFICATION CODE: 42C386; 95111
CATEGORİE CODES: UG; KM

05/09/76

TITLE: STUDY OF THE THERMAL DEGRADATION OF POLYCHLOROTRIFLUOROETHYLENE, POLYVINYLIDENE FLUORIDE AND COPOLYMERS OF CHLOROTRIFLUOROETHYLENE AND VINYLIDENE FLUORIDE

AUTHOR(S): Zulfiqar S; Zulfiqar M; Rizvi M; Munir A; McNeill I C

CORPORATE SOURCE: Quaid-i-Azam, University; Glasgow, University


CODEN: PDSTDW JOURNAL ANNOUNCEMENT: 9406 RAPRA UPDATE: 9410

DOCUMENT TYPE: Journal

LANGUAGE: English

SUBFILE: (R) RAPRA

ABSTRACT: A systematic study of the thermal degradation of a series of homopolymers and copolymers of chlorotrifluoroethylene and vinylidene fluoride was carried out using TGA and thermal volatilisation analysis (TVA). Volatile products were separated by sub-ambient TVA and characterised by means of IR spectroscopy and mass spectrometry. On degradation, polychlorotrifluoroethylene gave the monomer as the major product and carbon dioxide, C2F2C12, C3F5Cl and C2F3C13 in traces. Polyvinylidene fluoride formed hydrogen fluoride in appreciable amount along with the monomer and C4H3F3. The copolymers showed a similar type of degradation pattern. The structural changes which took place during degradation were also studied and mechanisms of formation of the various products are discussed. 12 refs.

SUBJECT HEADING (RAPRA): DEGRADATION, thermal, PCTFE, PVDF; CHLOROTRIFLUOROETHYLENE POLYMERS, thermal degradation; VINYLIDENE FLUORIDE POLYMERS, thermal degradation

IDENTIFIERS (Non-Polymer Terms): CARBON DIOXIDE; CHLORINE COMPOUND; CHLOROPENTAFLUOROPROPENE; DICHLORODIFLUOROPROPENE; FLUORINE COMPOUND; HYDROFLUORIC ACID; HYDROGEN FLUORIDE; TRICHLOROTRIFLUOROPROPENE; TRIFLUOROBUTADIENE
Polymer compositions based on fluoropolymers intended especially for the manufacture of electrical cables and of flexible pipes, comprising, by weight, (A) from 25 to 75% of PVDF homopolymer; (B) from 25 to 75% of a thermoplastic copolymer of VF2 and of at least one other fluoromonomer, exhibiting a content of 5 to 25% of this other monomer. This other fluorocomonomer may be in particular CTFE (chlorotrifluoroethylene), HFP (hexafluoropropylene) or TrFE (trifluoroethylene). Articles produced from these compositions have good mechanical properties at low temperature.

Exemplary Claim:
1. A polymer composition based on polyvinylidene fluoride (PVDF) homopolymer and at least one fluorocopolymer, comprising: a mixture of fluoropolymers comprised of, by weight: (A) from about 25 to about 75% of at least one PVDF homopolymer; and (B) from about 25 to about 75% of at least one thermoplastic copolymer comprising vinylidene fluoride (VF2) and from about 5 to about 25% by weight of at least one other fluoromonomer selected from the group consisting of hexafluoropropylene and chlorotrifluoroethylene.
Photoacoustic detection of the decomposition kinetics of polymers: interpretation of acoustic signals

AUTHOR(S): Kukreja, L. M.; Hess, P.
LOCATION: Institute of Physical Chemistry, University of Heidelberg, Im Neuenheimer Feld 253, Heidelberg, Germany, 69120

SECTION:
CA237004 Plastics Manufacture and Processing
CA273XXX Optical, Electron, and Mass Spectroscopy, and Other Related Properties

IDENTIFIERS:
polyimide degrdn kinetics photoacoustic detection, PVDF piezoelec transducer polyimide ablation

DESCRIPTORS:
Photoacoustic effect...
in laser-induced ablation of polymides
Polyethers,polyimide-,properties... Polyimides,polyether-,properties...
laser-induced degrdn. of, PVDF for photoacoustic detection of
Ablation,laser-induced... Polymer degradation,laser-induced...
of polyimides, photoacoustic detection of
Kinetics of polymer degradation,photochem.... with laser source, of polyimides, photoacoustic detection of

CAS REGISTRY NUMBERS:
25036-53-7 25038-81-7 laser-induced degrdn. of, PVDF for photoacoustic detection of
24937-79-9 piezoelec. foil transducer, for photoacoustic detection of polyimide degrdn.

?s (crystallite and irradiated and poly(vinylidene()fluoride))/ti
783 CRYSTALLITE/TI
24726 IRRADIATED/TI
Crystallite damage studies on irradiated poly(vinylidene fluoride)

AUTHOR(S): Zhao Zhudi; Chu Jin; Chen Xinfang

130023

CODEN: RPCHDM ISSN: 0146-5724 LANGUAGE: English

SECTION: CA_36005 Physical Properties of Synthetic High Polymers

IDENTIFIERS: PVDF crystallite radiation damage, polyvinylidene fluoride crystallite irradn

DESCRIPTORS:
Crystallites...
in poly(vinylidene fluoride), damage of, by irradn.
Polymer degradation, radiochem....
of poly(vinylidene fluoride), crystallite damage in
Crystallinity.... Heat of fusion and Heat of freezing.... Polymer morphology, cryst., spherulitic...
of poly(vinylidene fluoride), radiation effect on

CAS REGISTRY NUMBERS:
24937-79-9 crystallite damage of irradiated

53770 SURFACE/TI (SEE ?IGNOTE)
306491 ANALYSIS/TI (SEE ?IGNOTE)
195990 SURFACE/TI
53770 MEMBRANES/TI

S4 20 (SURFACE AND ANALYSIS AND MEMBRANES)/TI

S4 and poly

909548 POLY

S5 6 S4 AND POLY

S5 and vinylidene

13851 VINYLIDENE

S5 AND VINYLIDENE

Surface analysis of poly(vinylidene difluoride) membranes


D-7400

CODEN: JMESDO ISSN: 0376-7388 LANGUAGE: English

SECTION: CA238003 Plastics Fabrication and Uses

IDENTIFIERS: PVDF membrane surface analysis spectroscopy
DESCRIPTORS:
Membranes, microporous...
  hydrophilic and hydrophobic PVDF, surface anal. of, by x-ray, IR and mass spectroscopy
Fluoropolymers...
  membranes, hydrophilic and hydrophobic, surface anal. of, by x-ray, IR and mass spectroscopy
Surface analysis...
  of hydrophilic and hydrophobic PVDF membranes
Polymer degradation, radiochem....
  of hydrophilic and hydrophobic PVDF membranes, by x-ray
Surface structure...
  of hydrophilic and hydrophobic PVDF membranes, surface anal. of, by x-ray, IR and mass spectroscopy
Mass spectra, secondary-ion...
  of hydrophilic and hydrophobic PVDF membranes, surface structure in relation to
CAS REGISTRY NUMBERS:
24937-79-9 membranes, hydrophilic and hydrophobic, surface anal. of, by x-ray and IR and mass spectroscopy
?
### Status: Signing Off...
logoff
22 nov 96 15:09:01 User 036172 Session D804.6
Sub account: BULLUCK/N7402
  $12.00 0.100 Hrs File 399
  $5.85 3 Types
$17.85 Estimated cost File 399
$1.20 TYMNET
$19.05 Estimated cost this search
$34.15 Estimated total session cost 0.168 Hrs.

### Status: Signed Off. (9 minutes)