ABSTRACT

Four phenylethynyl amine compounds—3 and 4-aminophenoxy-4’-phenylethynylbenzophenone, and 3 and 4-aminophenylethynylbenzophenone—were readily prepared and were used to endcap imide oligomers. Phenylethynyl-terminated amide acid oligomers and phenylethynyl-terminated imide oligomers with various molecular weights and compositions were prepared and characterized. These oligomers were cured at 300°C to 400°C to provide crosslinked polyimides with excellent solvent resistance, high strength and modulus and good high temperature properties. Adhesive panels, composites, films and moldings from these phenylethynyl terminated imide oligomers gave excellent mechanical performance.
PHENYLETHYNYL AMINE

ORIGIN OF THE INVENTION

The invention described herein was jointly made by employees of the United States Government and during the performance of work under a NASA contract and is subject to the provisions of Section 305 of the National Aeronautics and Space Act of 1958, as amended, Public Law 85-568 (72 Stat. 435; 42 USC 2457).

This is a divisional of application Ser. No. 08/209,512 filed on Mar. 3, 1994, now U.S. Pat. No. 5,412,066.

BACKGROUND OF THE INVENTION

1. Field of the Invention

The present invention relates generally to polyimides, especially imide oligomers. It relates particularly to a series of phenylethynyl-terminated imide oligomers which can be thermally cured to resins that are useful as adhesives, composite matrices, moldings, films and coatings. The invention also relates particularly to a series of amino phenylethynyl compounds used to terminate the imide oligomers.

2. Description of Related Art

Polyimides are condensation polymers that can be prepared by various routes. The most popular route involves the reaction of an aromatic dianhydride with an aromatic diamine. The intermediate polyamic acid is either thermally or chemically cyclodehydrated to form the polyimide which has a repeat unit of the general type

\[
\begin{array}{c}
\text{O} \\
\text{A} \\
\text{N} \\
\text{Ar} \\
\text{O}
\end{array}
\]

where Ar is a tetravalent aromatic radical, which can be as simple as 1,2,4,5-tetrasubstituted benzene, or it may be a bis-4-(o-diphenylene) having the general structure

where X=nil, O, S, SO_{2}, C=O, etc. Ar may be any other appropriate tetravalent radical. Ar is a divalent aromatic radical which may include 1,3-phenylene, 1,4-phenylene, 4,4'-oxydiphenylene, 4,4'-biphenylene, 4,4'-thiodiphenylene, 4,4'-carbonilodiphenylene, 4,4'-methylendiphenylene or any other appropriate divalent radical.


SUMMARY OF THE INVENTION

The primary object of the present invention is to provide a series of phenylethynyl-terminated imide oligomers of various molecular weights that can be thermally cured to provide materials that are crosslinked, insoluble in common organic solvents and have excellent high temperature mechanical properties. Another object is to provide novel amino phenylethynyl compounds that can be used to terminate oligomers.

According to the present invention, the foregoing and additional objects were obtained by synthesizing phenylethynyl-terminated imide oligomers with different polymer backbones and different molecular weights. After curing 1 hour at 350°C, the resulting polymers were crosslinked and insoluble in common organic solvents. Molecular weights prepared include 3000, 6000 and 9000 g/mole. However, this should not be viewed as a limitation since essentially any molecular weight from ~500 to ~20000 g/mole is easily prepared by adjusting the monomer stoichiometry. A modified Carothers equation, \[ D_w = (1+r)/(1-r) \] where \( r \) = monomer ratio and \( D_w \) = degree of polymerization, provides a means to calculate the monomer ratio necessary to produce the desired molecular weight. The general synthetic procedure is shown in the drawing. Data for theoretical number average molecular weights (\( \bar{M}_n \)), inherent viscosities (\( \eta_{inh} \)) and glass transition temperatures of uncured and cured polymers.
are included in Table 1. Data for the thermal stability of the cured polymer in air and nitrogen and the thin film properties are included in Table 2. Data for the titanium to titanium adhesive properties are included in Table 3. The phenylethynyl-terminated imide oligomers of the present invention are eminently suitable as adhesives, composite matrices, moldings, films and coatings.

Two significant advantages of phenylethynyl-terminated imide oligomers compared to linear polyimides are better compression moldability (more flow, better consolidation under less pressure) and the ability to react forming a structure containing some crosslinks which typically increases the solvent resistance, glass transition temperature, and modulus. These properties are important for aerospace applications as films, moldings, adhesives and composites.

The advantage of phenylethynyl-terminated imide oligomers compared to polyimides containing pendant ethynyl or phenylethynyl groups is that the phenylethynyl-terminated imide oligomers have low to moderate molecular weights and therefore have lower melt viscosity for easier processing into moldings, composites and adhesive bonds.

The advantage of phenylethynyl-terminated imide oligomers compared to ethynyl-terminated polyimides is the higher reaction temperature of the phenylethynyl group compared to the ethynyl group. The phenylethynyl group begins to react at -350°C (beginning of the exotherm in the DSC trace when run at 20°C/min) compared to -200°C for the ethynyl group. This higher reaction temperature provides a larger window to melt process these oligomers whereas ethynyl-terminated imide oligomers begin to react and, therefore, are either not melt processable or have a very small window to melt process.

**BRIEF DESCRIPTION OF THE DRAWING**

For a more complete understanding of the present invention, including its primary objects and attending benefits reference should be made to the Description of the Preferred Embodiments, which is set forth in detail below. This description should be read together with the accompanying drawing, wherein the single drawing FIGURE is an equation showing the general synthetic scheme for the preparation of phenylethynyl terminated imide oligomers according to the present invention.

**DESCRIPTION OF THE PREFERRED EMBODIMENTS**

The approach according to the present invention was to provide phenylethynyl-terminated polyimides by employing any of four novel phenylethynyl amine compounds to endcap imide oligomers. By adjusting the monomer stoichiometry, the molecular weight of the resulting phenylethynyl-terminated polyimide could be controlled. The four phenylethynyl amine compounds are 3 and 4-aminophenoxy-4'-phenylethynylbenzophenone, and 3 and 4-amino-4'-phenylethynylbenzophenone, which have the following structures, respectively:

![Structures](image)

The phenylethynyl group is the functionality which reacts at elevated temperatures, while the amine group allows the compound to be incorporated into any system which reacts with an amine. Any compound having the following general structural formula can also be used as an endcapping agent:

![General Structural Formula](image)

where X is nil or a meta or para substituted phenoxy group and Y is an activating group consisting of C(CF₃)₂, CO, or SO₂.

The phenylethynyl group has two important advantages over the ethynyl group. These are stability towards conditions of aromatic nucleophilic displacement reactions and stability at moderate temperatures (i.e. 150°-250°C). For example, when 4-phenylethynyl-4'-fluorobenzophenone was treated with a slight excess of 3 or 4-aminophenol in the presence of an alkali metal base such as potassium carbonate in a polar aprotic solvent such as DMF or DMAc at 143°C, 3 or 4-aminophenoxy-4'-phenylethynylbenzophenone was afforded in nearly quantitative yield. However, several attempts to make the 3-aminophenyl-4'-ethynylbenzophenone from the 4-ethynyl-4'-fluorobenzophenone under similar reaction conditions afforded a mixture of products. In addition, imide model compounds of the phenylethynyl endcapper react at a moderate to fast rate in air at 300° to 350°C whereas analogous model compounds containing the ethynyl group react at a moderate to fast rate in air at 200° to 250°C. Phenylethynyl containing imide model compounds are reasonably stable at 200°C. (see Example 6) whereasthynyl containing imide model compounds react at 200°C.

When an imide oligomer terminated with these activated phenylethynyl endcappers has been cured in a steel mold for 1 hour at 300 psi and 350°C, the resulting glass transition temperatures (Tg) are 15° to 20°C higher than those of the cured imide oligomer terminated with nonactivated phenylethynyl endcappers. (See Example 7 below and the accompanying Table). Cured imide oligomers terminated with the activated phenylethynyl endcappers offered higher use temperatures than that of cured imide oligomers from nonactivated phenylethynyl endcappers.

**EXAMPLE 1**

**Synthesis of 4-bromo-3'-nitrobenzophenones**

In a 250 mL flask equipped with a nitrogen inlet, overhead stirring assembly, and reflux condenser were placed 3-nitrobenzoyl chloride (28 g, 0.15 mol), bromobenzene (130 g, 0.83 mol), and aluminum chloride (22 g, 0.16 mol). The mixture was heated for 3 h at 100°C, then cooled to room temperature and stirred for an additional 16 h. The mixture was poured into 1 L of acidic (HCl) ice water, and the
organics were extracted using methylene chloride, dried over magnesium sulfate, and the solvents were removed under reduced pressure to yield a yellow powder. Recrystallization from acetone afforded 42.1 g (90% yield) of white crystals: m.p. 150°-151° C. Recrystallization from acetone afforded 42.1 g (90% yield) of white crystals: m.p. 150°-151° C.

Synthesis of 4-bromo-4'-nitrobenzophenone

In a 250 mL flask equipped with a nitrogen inlet, overhead stirring assembly, and reflux condenser were placed 4-nitrobenzoyl chloride (22.6 g, 0.122 mol), bromobenzene (100 g, 0.63 mol), and aluminum chloride (17.3 g, 0.13 mol). The mixture was heated for 3 h at 100° C., then cooled to room temperature and stirred for an additional 16 h. The mixture was poured into 1 L of acidic (HCl) ice water, and the organics were extracted using methylene chloride, dried over magnesium sulfate, and the solvents were removed under reduced pressure to yield a yellow powder. Recrystallization from acetone afforded 29.2 g (85% yield) of yellow crystals: m.p. 123°-124° C. (Lit. 122° C.). Recrystallization from 1-butanol afforded 13.73 g (85% yield) of white crystals: m.p. 185°-186° C. IR (KBr) 3086 cm-1 (aromatic); 2209 cm-1 (C=O); 1654, 1612 cm-1 (aromatic C-H).

Anal Calcd for C21H15NO3Br: %C=77.05 %H=4.00 %N=4.28 Found: %C=75.94 %H=3.94 %N=3.81.

EXAMPLE 2

Synthesis of Phenylethynylbenzophenones

Synthesis of 4-fluoro-4'-phenylethynylbenzophenone

In a 250 mL flask equipped with a nitrogen inlet, overhead stirring assembly, and reflux condenser were placed 4-fluorobenzene (117 g, 1.2 mol) and 4-bromobenzoyl chloride (39.8 g, 0.181 mol). The mixture was cooled to 0° C. and aluminum chloride (27 g, 0.20 mol) was added. The mixture was stirred for 15 minutes at 23° C., followed by heating to reflux for 4 h. The reaction was then cooled to 23° C. and stirred for 16 h. The resulting slurry was poured into 2 L of acidic (HCl) water. The organics were extracted with methylene chloride, and dried over magnesium sulfate. The slurry was filtered and the methylene chloride was removed under reduced pressure to yield an off-white powder. Recrystallization from toluene afforded 43.3 g (86%) of 4-bromo-4'-fluorobenzophenone as white crystals: m.p. 193°-195° C. (Lit. 197°-198° C.).

Anal Calcd for C21H13N03F: %C=77.05 %H=4.00 %F=6.83 Found: %C=76.84 %H=3.94 %F=6.16.

Synthesis of 3-nitro-4'-phenylethynylbenzophenone

In a 250 mL flask equipped with a nitrogen inlet and reflux condenser were placed 4-bromo-3'-nitrobenzophenone (15.31 g, 0.050 mol), phenylacetylene (5.62 g, 0.055 mol), triphenylphosphine (0.100 g), bis(triphenylphosphine)chloridium (II) chloride (0.050 g), and 100 mL of triethylamine. The mixture was heated to 50°-60° C. Copper (I) iodide (0.050 g), and additional 50 mL of triethylamine were added. The reaction was heated to reflux for 2 h, then cooled to room temperature. The resulting precipitate was collected by filtration, washed with acidic (HCl) water, and dried at 100° C. Recrystallization from 1-butanol afforded 29.2 g (85% yield) of white crystals: m.p. 123°-124° C., (Lit. 122° C.). Recrystallization from acetone afforded 23 g (75%) of white crystals: m.p. 150°-151° C.

Anal Calcd for C21H15NO3Br: %C=51.01 %H=2.63 %Br=26.01 Found: %C=50.08 %H=2.82 %Br=25.81.

Synthesis of 4-nitro-4'-phenylethynylbenzophenone

In a 250 mL flask equipped with a nitrogen inlet and reflux condenser were placed 4-bromo-3'-nitrobenzophenone (19.3 g, 0.063 mol), phenylacetylene (6.5 g, 0.063 mol), triphenylphosphine (0.15 g). bis(triphenylphosphine)pallidium (II) chloride (0.08 g), copper (I) iodide (0.08 g), and additional 50 mL of triethylamine were added. The mixture was heated to 65° C. for 16 h, then cooled to room temperature. The resulting mixture was poured into acidic (HCl) water, the precipitate was collected by filtration and dried in vacuo at 110° C. Recrystallization from toluene afforded 9.5 g (47% yield) of yellow-white crystals: m.p. 193°-195° C. IR (KBr) 3086 cm-1 (aromatic); 2209 cm-1 (C=O); 1654, 1612 cm-1 (C=O); and 1602 cm-1 (aromatic C-H). Mass Spec. m/e (relative intensity): M+, 327.1 (53%); Fw=327.34 g/mol.

Anal Calcd for C21H13NO3Br: %C=77.05 %H=4.00 %Br=25.81 Found: %C=76.94 %H=3.94 %Br=3.81.
EXAMPLE 3

Synthesis of Phenylethynylamino Endcappers

Synthesis of 4-(3-aminophenoxy)-4'-phenylethynylbenzophenone (3-APEB)

In a 500 mL flask equipped with nitrogen inlet, overhead stirring assembly, Dean-Stark trap, and reflux condenser were placed 3-aminophenol (18.5 g, 0.17 mol), 4-phenylethynyl-4'-fluorobenzophenone (43.2 g, 0.14 mol), potassium carbonate (22 g, 0.16 mol), toluene (30 mL), and DMAc (200 mL). The mixture was heated to 142°C for 8 h, then cooled to room temperature. The reaction mixture was poured into water and the resulting light yellow powder was collected by filtration. Recrystallization from ethanol afforded the desired product in 85% yield: m.p. 151°-153°C. IR (KBr): 3414 and 3342 cm⁻¹ (NH₂); 3040, 1599, and 1499 cm⁻¹ (aromatic C-H); 2216 cm⁻¹ (C≡C), and 1646 cm⁻¹ (C=O): Mass Spec. m/e (relative intensity): M⁺, 389.4 (100); FW=389.44 g/mol.

Anal Calcd for C₃₁H₃₄NO₆: %C=83.28 %H=4.92 %N=4.71 Found: %C=84.60 %H=5.26 %N=4.52

4-(4-Aminophenoxy)-4'-phenylethynylbenzophenone (4-APEB)

In a 250 mL flask equipped with nitrogen inlet, overhead stirring assembly, Dean-Stark trap, and reflux condenser were placed 4-aminophenol (5.46 g, 0.05 mol), 4-phenylethynyl-4'-fluorobenzophenone (15.02 g, 0.05 mol), potassium carbonate (7.6 g, 0.055 mol), toluene (20 mL), and DMAc (45 mL). The mixture was heated to 150°C for 2 h, then cooled to room temperature. The reaction mixture was poured into water and the resulting light yellow powder was collected by filtration. Recrystallization from ethanol afforded 5.09 g (69% yield) of a pale yellow solid: m.p. 140°-142°C. IR (KBr): 3463 3356 and 3208 cm⁻¹ (NH₂); 2217 cm⁻¹ (C=C); and 1646 cm⁻¹ (C=O): Mass Spec. m/e (relative intensity): M⁺, 394.4 (100); FW=394.44 g/mol.

Anal Calcd for C₃₁H₃₄NO₆: %C=82.80 %H=4.86 %N=3.60 Found: %C=82.80 %H=4.86 %N=3.60

EXAMPLE 4

Synthesis of Imide Model Compounds

Synthesis of 4-phenylethynylbenzoyl-4'-phenoxy-3'-phenylphthalimide

In a 200 mL flask equipped with a reflux condenser were placed phthalic anhydride (1.33 g, 8.9 mmol), 3-APEB (3.50 g, 8.9 mmol), and acetic acid (100 mL). The solution was stirred for 1.5 h at 23°C and heated to reflux for 3 h. The solution was poured into 2 L of water, the precipitate was collected by filtration, washed in warm ethanol, and dried at 100°C. Recrystallization from ethanol afforded 5.41 g (61% yield) of a pale yellow solid: m.p. 195°-196.5°C. IR (KBr): 3242, 3340, and 3224 cm⁻¹ (NH₂); 2216 cm⁻¹ (C≡C), and 1645 cm⁻¹ (C=O): Mass Spec. m/e (relative intensity): M⁺, 297.2 (47); FW=297.36 g/mol.

Anal Calcd for C₂₀H₁₇NO₄: %C=84.82 %H=5.26 %N=4.52

Synthesis of 4-aminophenylbenzoyl-4'-phenoxy-3'-phenylphthalimide

In a 200 mL flask equipped with a reflux condenser were placed phthalic anhydride (1.33 g, 8.9 mmol), 3-APEB (3.50 g, 8.9 mmol), and acetic acid (100 mL). The solution was stirred for 1.5 h at 23°C and heated to reflux for 3 h. The solution was poured into 2 L of water, the precipitate was collected by filtration, washed in warm ethanol, and dried at 100°C. Recrystallization from ethanol afforded 5.41 g (61% yield) of a pale yellow solid: m.p. 195°-196.5°C. IR (KBr): 3242, 3340, and 3224 cm⁻¹ (NH₂); 2216 cm⁻¹ (C≡C), and 1645 cm⁻¹ (C=O): Mass Spec. m/e (relative intensity): M⁺, 297.2 (47); FW=297.36 g/mol.

Anal Calcd for C₂₀H₁₇NO₄: %C=84.82 %H=5.26 %N=4.52

5,599,993
was heated to reflux for 16 h. The solution was poured into 1 L of water, and the precipitate was collected by filtration, washed in boiling methanol, and dried at 100°C in vacuo to afford 2.5 g (95% yield) of light tan powder; visual m.p. 202°–205°C. IR (KBr); 3050–3020 and 1600 cm⁻¹ (aromatic C—H), 2217 cm⁻¹ (C==C), 1772 and 1714 cm⁻¹ (imide), and 1648 cm⁻¹ (C==O).

Anal Calcd for C₉₋₅H₁₁N₂O₄: %C=80.91 %H=4.07 %N=2.70 Found: %C=78.88 %H=4.06 %N=2.31

**Synthesis of 4-phenylethynylbenzoyl-4'-phenylphthalimide**

In a 100 mL flask equipped with a reflux condenser were placed 4-amino-4'-phenylethynylbenzophenone (2.97 g, 0.010 mol), phthalic anhydride (1.48 g, 0.010 mol), and 50 mL of glacial acetic acid. The mixture was heated to reflux for 16 h, then cooled to room temperature. The resulting solid was collected by filtration and dried at 100°C. Recrystallization from 1-butanol afforded 3.05 g (71.4% yield) of a white solid: m.p. 202°–205°C. IR (KBr) 3050–3020 and 1600 cm⁻¹ (aromatic C—H), 2217 cm⁻¹ (C==C), 1772, 1718 cm⁻¹ (imide C=O), and 1654 cm⁻¹ (C=O). Mass Spec. m/e (relative intensity): M+, 427.4 (6); FW=427.46 g/mol.

**EXAMPLE 5**

4-Phenylethynylbenzoyl-4'-phenoxy-3''-phenylphthalimide, 4-phenylethynylbenzoyl-4'-phenoxy-4''-phenylphthalimide, and 4-phenylethynylbenzoyl-4'-phenylphthalimide were individually placed in a DSC cell and heated at a rate of 10° C/min to 450°C. The model compounds displayed melting point maxima of 172°, 209° and 208° C. with a
melting enthalpies of 101, 109, and 107 J/g respectively. The exothermic onsets occurred at 354°, 347°, and 319° C. with maximum onsets at 415°, 402°, and 399° C. and curing enthalpies of 248, 287, and 408 J/g respectively. These compounds were also held at 350° C. for 1 h. The resulting DSC thermograms of these thermally treated compounds displayed an absence of both melting points and exothermic peaks.

EXAMPLE 6

4-Phenylethynylbenzoyl-4'-phenoxy-3'-phenylphthalimide, 4-phenylethynyl-benzoyl-4'-phenoxy-4'-phenylphthalimide, and 4-phenylethynylbenzoyl-4'-phenylphthalimide were individually placed in a DSC cell and held at 200° C. for 20 h. The melting enthalpies of these compounds were 90.7, 73.2, and 95.7 J/g respectively indicating that 89%, 68%, and 89% of the starting material remained intact.

EXAMPLE 7

Synthesis of a 9000 g/mol phenylethynyl-terminated imide oligomer

In a 250 mL flask equipped with a nitrogen inlet, overhead stirring assembly, and reflux condenser were placed 3,4'-ODA (3.7992 g, 0.019 mol), ODPA (6.2045 g, 0.200 mol), and the appropriate endcapper (0.8007 mol), m-cresol (54 g), and isoquinoline (0.2 g). The solution was stirred at 23° C. for 16 h, and heated to reflux for 3.5 h. The reaction was cooled to 23° C. and poured into stirring methanol. The white precipitate was collected by filtration, extracted with methanol for 48 h and dried in vacuo at 180° C. for 10 h to yield 9.2 g (98%) of white powder.

Thermal Properties of a 9000 g/mol Imide Oligomer Synthesized from ODPA and 3,4'-ODA Using Several Phenylethynyl Endcaps

<table>
<thead>
<tr>
<th>Endcap</th>
<th>( \eta_{inh} ) (dL/g)*</th>
<th>( T_g ) (°C) Uncured</th>
<th>( T_g ) (°C) Cured†</th>
</tr>
</thead>
<tbody>
<tr>
<td>H₂N[O-O][C]</td>
<td>0.35</td>
<td>218</td>
<td>243</td>
</tr>
<tr>
<td>H₂N[O-C][C]</td>
<td>0.36</td>
<td>220</td>
<td>228</td>
</tr>
<tr>
<td>H₂N[O]</td>
<td>0.41</td>
<td>226</td>
<td>242</td>
</tr>
<tr>
<td>H₂N[O][C]</td>
<td>0.37</td>
<td>213</td>
<td>223</td>
</tr>
</tbody>
</table>

*Measured in m-cresol at 25° C. using a conc. of 0.5 g/dL.
†Cured in a mold at 350° C. for 1 hour at 200 psi.

The general reaction scheme for the phenylethynyl-terminated imide oligomers is shown in the Drawing where \( n \) is an integer from 1-100 repeat units. The solvent can be N, N-dimethylacetamide (DMAc), N-methylpyrrolidinone (NMP), N, N-dimethylformamide (DMF) or m-cresol. Ar' can be as simple as 1,2,4,5-tetrasubstituted benzene, or it may be a bis-4-(o-diphenylene) having the general structure

\[
\begin{align*}
\text{Ar} & = \text{aryl, arildyl, diaryl, etc., and mixtures thereof.} \\
\text{Ar} & \text{ in the Drawing is selected from a group of radicals consisting of} \\
\end{align*}
\]

![Chemical structure](image-url)
wherein Y=nil, O, S, C=O, SO₃, CH₃, C(CH₃)₂, C(CF₃)₂, etc., and mixtures thereof.

Phenylethynyl-terminated imide oligomers with a wide range of molecular weights (≈500 to ≈20,000 g/mole) are readily prepared by offsetting the ratio of one monomer relative to the other by a calculated amount and adding an appropriate amount of either 3- or 4-aminophenoxy-4'-phenylethynylbenzophenone or 3- or 4-aminophenylethynylbenzophenone. Prior to curing, low molecular weight polymers have low melt viscosity and are readily processed into moldings, composites and adhesive bonds. When cured, different molecular weight polymers display different properties. For example, low molecular weight phenylethynyl-terminated imide oligomers provide a higher crosslink density than higher molecular weight phenylethynyl-terminated imide oligomers. Materials containing high crosslink density normally produce better solvent resistance, higher modulus, better high temperature properties and lower toughness than those containing low crosslink density. Therefore, the polymers can be designed with the proper combination of properties for the desired application by controlling the molecular weight. Specific examples follow.

for 16 h followed by heating at reflux (≈195°C.) for 4 h. After cooling, the reaction was poured into methanol, washed in boiling methanol and dried at 110°C. for 72 h to afford an off-white solid in >95% yield. The (ηinh) resulting phenylethynyl-terminated polyimide had an inherent viscosity of 0.20 dL/g in m-cresol at 25°C. Additional properties for this polymer are listed in Tables 1, 2 and 3.

**TABLE 1**

<table>
<thead>
<tr>
<th>Mn, g/mole</th>
<th>Endcapper</th>
<th>ηinh, dL/g</th>
<th>Initial*</th>
<th>Final*</th>
</tr>
</thead>
<tbody>
<tr>
<td>3000</td>
<td>3-APEB</td>
<td>0.12</td>
<td>173</td>
<td>233</td>
</tr>
<tr>
<td>3000</td>
<td>4-APEB</td>
<td>0.20</td>
<td>195</td>
<td>246</td>
</tr>
<tr>
<td>6000</td>
<td>3-APEB</td>
<td>0.22</td>
<td>208</td>
<td>248</td>
</tr>
<tr>
<td>6000</td>
<td>4-APEB</td>
<td>0.22</td>
<td>208</td>
<td>245</td>
</tr>
<tr>
<td>9000</td>
<td>3-APEB</td>
<td>0.36</td>
<td>220</td>
<td>249</td>
</tr>
<tr>
<td>9000</td>
<td>4-APEB</td>
<td>0.41</td>
<td>222</td>
<td>250</td>
</tr>
</tbody>
</table>

*Thermally imidized in m-cresol at -195°C. for 4 h with isoquinoline catalyst.

<table>
<thead>
<tr>
<th>Mn, g/mole</th>
<th>Endcapper</th>
<th>Temperature at 5% Weight Loss*, °C.</th>
<th>Tensile Strength, ksi</th>
<th>Tensile Modulus, ksi</th>
<th>Elongation, %</th>
</tr>
</thead>
<tbody>
<tr>
<td>3000</td>
<td>3-APEB</td>
<td>443</td>
<td>498</td>
<td>—</td>
<td>—</td>
</tr>
<tr>
<td>3000</td>
<td>4-APEB</td>
<td>493</td>
<td>497</td>
<td>—</td>
<td>—</td>
</tr>
<tr>
<td>6000</td>
<td>3-APEB</td>
<td>439</td>
<td>468</td>
<td>—</td>
<td>—</td>
</tr>
<tr>
<td>6000</td>
<td>4-APEB</td>
<td>486</td>
<td>493</td>
<td>—</td>
<td>—</td>
</tr>
<tr>
<td>9000</td>
<td>3-APEB</td>
<td>433</td>
<td>465</td>
<td>19.6</td>
<td>558</td>
</tr>
<tr>
<td>9000</td>
<td>4-APEB</td>
<td>473</td>
<td>502</td>
<td>19.3</td>
<td>455</td>
</tr>
</tbody>
</table>

*Heating rate of 2.5°C/min in flowing air or nitrogen.

**TABLE 3**

<table>
<thead>
<tr>
<th>Mn, g/mole</th>
<th>Molding Pressure, psi</th>
<th>Exposure Conditions</th>
<th>Test Temperature, °C.</th>
<th>Shear Strength, psi</th>
</tr>
</thead>
<tbody>
<tr>
<td>3000</td>
<td>15</td>
<td>none</td>
<td>23</td>
<td>4730</td>
</tr>
<tr>
<td>3000</td>
<td>25</td>
<td>none</td>
<td>23</td>
<td>5270</td>
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<td>50</td>
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<td>23</td>
<td>5270</td>
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</table>

**EXAMPLE 8**

The following example illustrates the synthesis of a phenylethynyl-terminated imide oligomer with theoretical number average molecular weight of 3000 g/mole. 3,4'-Oxydianiline (ODA) (4.268 mmole, 0.8546 g), 4,4'-oxydiphthalic dianhydride (ODPA) (5.000 mmole, 1.5511 g), 4-(4-aminophenoxy)-4'-phenylethynylbenzophenone (4-APEB) (1.494 mmole, 0.5819 g, 2% excess), m-cresol (14 g), and isoquinoline (0.2 g) were added to a 100 mL three neck flask equipped with a mechanical stirrer, condenser and nitrogen inlet. The reaction was stirred at 25°C.
After cooling, the reaction was poured into methanol, 4-(4-aminophenoxy)-4'-phenylethynylbenzophenone (3-APEB) (5.864 mmole, 2.2837 g), m-cresol (55 g), and isoquinoline (0.2 g) were added to a 100 mL three neck flask equipped with a mechanical stirrer, condenser and nitrogen inlet. The reaction was stirred at 25°C for 16 h followed by heating at reflux (~195°C) for 4 h. After cooling, the reaction was poured into methanol, washed in boiling methanol and dried at 110°C for 72 h to afford an off-white solid in >95% yield. The resulting phenylethynyl-terminated polyimide had η_i,ns of 0.12 dL/g in m-cresol at 25°C. Additional properties for this polymer are listed in Tables 1, 2 and 3.

EXAMPLE 12

The following example illustrates the synthesis of a phenylthényl-terminated imide oligomer with theoretical number average molecular weight of 6000 g/mole. 3,4'-Oxydianiline (ODA) (18.48 mmole, 3.7001 g), 4,4'-oxydiphthalic dianhydride (ODPA) (20.00 mmole, 6.2045 g), 4-(3-aminophenoxy)-4'-phenylethynylbenzophenone (3-APEB) (3.044 mmole, 1.1855 g), m-cresol (55 g), and isoquinoline (0.2 g) were added to a 100 mL three neck flask equipped with a mechanical stirrer, condenser and nitrogen inlet. The reaction was stirred at 25°C for 16 h followed by heating at reflux (~195°C) for 4 h. After cooling, the reaction was poured into methanol, washed in boiling methanol and dried at 110°C for 72 h to afford an off-white solid in >95% yield. The resulting phenylethynyl-terminated polyimide had η_i,ns of 0.22 dL/g in m-cresol at 25°C. Additional properties for this polymer are listed in Tables 1, 2 and 3.

EXAMPLE 13

The following example illustrates the synthesis of a phenylthényl-terminated imide oligomer with theoretical number average molecular weight of 9000 g/mole. 3,4'-Oxydianiline (ODA) (18.97 mmole, 3.7992 g), 4,4'-oxydiphthalic dianhydride (ODPA) (20.00 mmole, 6.2045 g), 4-(3-aminophenoxy)-4'-phenylethynylbenzophenone (3-APEB) (2.056 mmole, 0.8007 g), m-cresol (55 g), and isoquinoline (0.2 g) were added to a 100 mL three neck flask equipped with a mechanical stirrer, condenser and nitrogen inlet. The reaction was stirred at 25°C for 16 h followed by heating at reflux (~195°C) for 4 h. After cooling, the reaction was poured into methanol, washed in boiling methanol and dried at 110°C for 72 h to afford an off-white solid in >95% yield. The resulting phenylethynyl-terminated polyimide had η_i,ns of 0.36 dL/g in m-cresol at 25°C. A film cast from m-cresol solution of this oligomer and cured 1 h at 350°C was tough and finger nail creasible. The film was completely insoluble and did not swell when soaked for 48 h in DMAc and completely insoluble but swelled slightly (~5% increase in area) when soaked for 48 h in methylene chloride. Additional properties for this polymer are listed in Tables 1, 2 and 3. This material has been designated LARC™-PETI-1.

EXAMPLE 14

The following example illustrates the synthesis of a phenylthényl-terminated imide oligomer with theoretical number average molecular weight of 9000 g/mole. 3,4'-Oxydianiline (ODA) (50.00 mmole, 10.0123 g), 3,3',4,4'-benzophenone tetracarboxylic dianhydride (BTDA) (47.37

---

**TABLE 3-continued**

<table>
<thead>
<tr>
<th>Theoretical Mn, g/mole</th>
<th>Molding Pressure, psi</th>
<th>Exposure Conditions</th>
<th>Test Temperature, °C.</th>
<th>Shear Strength, psi</th>
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<td>48 h soak*</td>
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<td>3860</td>
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<td>3900</td>
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<tr>
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<td>50</td>
<td>3000 h/177°C</td>
<td>177</td>
<td>3890</td>
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</table>

*Chevron HyJet IV hydraulic fluid.
The following example illustrates the synthesis of a phenylethynyl-terminated imide oligomer with theoretical number average molecular weight of 9000 g/mole. 3,4'-Oxydianiline (ODA) (28.46 mmole, 5.6988 g), 4,4'-oxydiphthalic dianhydride (ODPA) (30.00 mmole, 9.3066 g), and isoquinoline (2 g) were added to a 250 mL three neck flask equipped with a mechanical stirrer, condenser and nitrogen inlet. The reaction was stirred at 25°C for 16 h to form the phenylethynyl-terminated polyamide acid followed by adding acetic acid (14 g) and heating at reflux (~130°C) for 4 h. A film was cast from a portion of this solution and cured 1 h at 350°C in air. The tensile strength, tensile modulus and elongation at break were 18,500 psi, 424,500 psi and 53%, respectively. At 25°C, these properties were 8,600 psi, 279,500 psi and 65%, respectively.

**EXAMPLE 19**

The following example illustrates the synthesis of a phenylethynyl-terminated imide oligomer with theoretical number average molecular weight of 9000 g/mole. This material has the same chemical composition as that is in Example 13 and is designated LARCTM-PET-1. 3,4'-Oxydianiline (ODA) (42.69 mmole, 8.5432 g), 4,4'-oxydiphthalic dianhydride (ODPA) (45.00 mmole, 13.9599 g), 3-amino-4'-phenylethynylbenzophenone (3.08 mmole, 0.9159 g), m-cresol (81 mL), and N-methylpyrrolidinone (NMP) (54 mL) were added to a 250 mL three neck flask equipped with a mechanical stirrer, condenser and nitrogen inlet. The reaction was stirred at 25°C for 24 h. A film was cast from a portion of this solution and cured 1 h at 350°C in air. The tensile strength, tensile modulus and elongation at break were 18,500 psi, 424,500 psi and 53%, respectively. At 177°C, these properties were 8,600 psi, 279,500 psi and 65%, respectively.
jet fuel, hydraulic fluid and toluene. The films on glass were

decantation doctoried onto clean, dry plate glass and dried to a

tack free form in a low humidity chamber. The films on glass

were dried in flowing air at 100°, 225° and 350° C. for 1 h
each. Thin film tensile properties were determined according
to ASTM D882. LARC™-PETI-1 films gave tensile strength, tensile modulus and elongation at break at 25° C. of
17,600 psi, 425,000 psi and 42% and at 177° C. of 7400 psi, 284,000 psi and 76%, respectively. The air cured film
had a Tg of 249° C. Solvent resistance was determined on
stressed (0.2±0.02 in) film strips immersed in various solvents for 2 weeks at ambient temperature. The specimens
were secured in a circular form by multiple wraps around a
standard paper clip. Films from LARC™-PETI-1 exhibited excellent resistance to ethylene glycol, methyl ethyl ketone,
jet fuel, hydraulic fluid and toluene.

EXAMPLE 20

Unoriented thin films were prepared in the following manner. NMP solutions (≈30% solids) of the phenylethynyl
terminated amide acid oligomers were centrifuged, the
decantate doctoried onto clean, dry plate glass and dried to a

tack free form in a low humidity chamber. The films on glass

were dried in flowing air at 100°, 225° and 350° C. for 1 h
each. Thin film tensile properties were determined according
to ASTM D882. LARC™-PETI-1 films gave tensile strength, tensile modulus and elongation at break at 25° C. of
17,600 psi, 425,000 psi and 42% and at 177° C. of 7400 psi, 284,000 psi and 76%, respectively. The air cured film
had a Tg of 249° C. Solvent resistance was determined on
stressed (0.2±0.02 in) film strips immersed in various solvents for 2 weeks at ambient temperature. The specimens
were secured in a circular form by multiple wraps around a
standard paper clip. Films from LARC™-PETI-1 exhibited excellent resistance to ethylene glycol, methyl ethyl ketone,
jet fuel, hydraulic fluid and toluene.

EXAMPLE 21

Molded specimens were prepared in the following man-
ner. Powdered imide oligomers (approximately 8.5 g) were

compression molded in a 1.25" stainless steel mold by heating to a temperature of 350° C. under 150-200 psi for
1 h. Miniature compact tension specimens (0.62 inx0.62 inx0.375 in thick) were machined from LARC™-PETI-1 moldings and subsequently tested to determine their fracture toughness (K,, critical stress intensity factor)
according to ASTM E399 using four specimens per test condition. Fracture energy (G,, critical strain energy release
rate) was calculated using the mathematical relationship,

\[ G_{ic} = (K_{ic})^2/E \]

where E is the tensile modulus of the material. LARC™-PETI-1 moldings gave G,, at 25° C. of 13.8 in

\[ \text{lb}-\text{in.} \] and a final Tg of 233° C. was measured after 1 h at 350° C.

LARC™-PETI-1 moldings gave G,, at 25° C. of 13.8 in

\[ \text{lb}-\text{in.} \] and a final Tg of 233° C. was measured after 1 h at 350° C.

EXAMPLE 22

LARC™-PETI-1 adhesive specimens were prepared in the following manner. Adhesive tape was prepared by coating
112 E-glass (A1100 finish) cloth with phenylethynyl
terminated imide oligomer (PETAA) solution in

NMP and drying at a final temperature of 250° C. to a volatile content of approximately 0.7%. Several coats were obtained
adequate tape thickness. Titanium (Ti-6Al-4V) with a Pasa-
directional prepregs (tapes) were prepared by coating
unsized IM-7 carbon/graphite fiber on a tape machine with
30%
NMP solutions of the PETAA of LARC™-PETI. The
volatile contents of the tapes varied from 7 to 12% and were
controlled by the residence time and temperature of an
in-line furnace in the tape machine. Resin content in the
tapes (minus volatiles) varied from 31 to 34% by weight. Small composites (3 in.x6 in.x8 to 16 plies) were fabricated in a stainless steel mold by heating to 350° C. under 200 psi and
holding at 350° C. under 200 psi for 1 h. Larger laminates (6 in.x6 in.x16 to 32 plies) were fabricated in an
autoclave by heating to 350° C. under 200 psi and holding for 1 h at 350° C. under 200 psi.

The composite panels were ultrasonically scanned (C-scanned) to show no voids, cut into specimens and tested for
flexural properties according to ASTM D790, short beam shear strength according to ASTM D2344 and compression
strength according to ASTM D3410 Procedure B.

EXAMPLE 23

Composites were prepared in the following manner. Unidirectional prepregs (tapes) were prepared by coating unsized IM-7 carbon/graphite fiber on a tape machine with 30% NMP solutions of the PETAA of LARC™-PETI. The
volatile contents of the tapes varied from 7 to 12% and were
controlled by the residence time and temperature of an
in-line furnace in the tape machine. Resin content in the
tapes (minus volatiles) varied from 31 to 34% by weight. Small composites (3 in.x6 in.x8 to 16 plies) were fabricated in a stainless steel mold by heating to 350° C. under 200 psi and
holding at 350° C. under 200 psi for 1 h. Larger laminates (6 in.x6 in.x16 to 32 plies) were fabricated in an
autoclave by heating to 350° C. under 200 psi and holding for 1 h at 350° C. under 200 psi.

The composite panels were ultrasonically scanned (C-scanned) to show no voids, cut into specimens and tested for
flexural properties according to ASTM D790, short beam shear strength according to ASTM D2344 and compression
strength according to ASTM D3410 Procedure B.
We claim:

1. A phenylethynyl amine having the general structure:

\[
\text{H}_2\text{N} - \text{X} \quad \text{Y} \quad \text{C} \quad \text{C} \quad \text{C} \\
\]

wherein X is a bond or X is a meta or para substituted phenoxy group, and Y is an activating selected from the group selected consisting of \( \text{C(CF}_3\text{)_2} \), \( \text{CO} \), and \( \text{SO}_2 \).

2. A phenylethynyl amine according to claim 1 having a specific structure selected from the group consisting of:

\[
\begin{align*}
\text{H}_2\text{N} &- \text{O} - \text{C} - \text{C} - \text{C} \\
\text{H}_2\text{N} &- \text{O} - \text{C} - \text{C} - \text{C} \\
\text{H}_2\text{N} &- \text{C} - \text{C} - \text{C} \\
\text{H}_2\text{N} &- \text{O} - \text{C} - \text{C} - \text{C} \\
\end{align*}
\]