A method for the reactivation of a tin oxide-containing catalyst of a CO2 laser is provided. First, the catalyst is pretreated by a standard procedure. When the catalyst experiences diminished activity during usage, the heated zone surrounding the catalyst is raised to a temperature which is the operating temperature of the laser and 400° C. for approximately one hour. The catalyst is exposed to the same laser gas mixture during this period. The temperature of the heated zone is then lowered to the operating temperature of the CO2 laser.

5 Claims, No Drawings
REACTIVATION OF A TIN OXIDE-CONTAINING CATALYST

ORIGIN OF THE INVENTION

The invention described herein was made jointly in the performance of work under NASA Contract Nos. L-79510B and L-83135B with Chemicon and NASA Contract NAS1-17099 with Old Dominion Research Foundation and employees of the United States Government. In accordance with 35 USC 202, the contractors have elected not to retain title.

This application is a continuation-in-part of co-pending application, Ser. No. 874,320, filed June 12, 1986, abandoned.

BACKGROUND OF THE INVENTION

This invention relates generally to the reactivation of a catalyst and in particular to the reactivation of a tin oxide-containing catalyst.

New approaches for extending the life of lasers used in a wide variety of applications are under investigation. One aspect of the rapid progress in this area is that new techniques for long lifetime space applications of high pulse energy, common and rare isotopes, closed-cycle CO2 lasers are being studied. The high-energy, pulsed CO2 lasers must be operated closed-cycle to conserve gas, especially if rare-isotope gases are used. Rare-isotope gases, such as C18O2, are used for improved transmission of the laser beam in the atmosphere.

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A temperature of 225° C. was experimentally found to provide an adequate temperature for reactivation.

The catalyst is exposed to the same circulating stream of gas that was continually drawn from and returned to the laser envelope. This laser gas mixture may contain a rare oxygen isotope in the CO₂ molecule. The oxide containing catalyst may contain the same rare oxygen isotope so that contained in the CO₂ molecule.

This constant heating and exposing the catalyst to the laser gas mixture is maintained for an hour. However, neither the time of heating and exposing nor the amount of heating is specific. Rather, they are functions of the integrity of the surface of the catalyst and the desired reactivation time, respectively. Note, however, that while the reactivation time may be decreased by increasing the temperature, this also increases the risk of damaging the integrity of the surface of the catalyst. Thus, the temperature and heating time will vary with the catalyst material.

After heating and exposing for an appropriate amount of time, the heated zone around the catalyst is allowed to return to the nominal operating temperature of the CO₂ laser. This temperature normally resides in the range of 23° C. to 100° C.

Catalyst activity can be measured as the percentage conversion of CO to CO₂. In the specific embodiment described above, the initial steady state conversion percentage was 70%. After four days this conversion percentage decreased to 67%. No decrease in activity is acceptable because the catalyst must maintain its activity for long periods of time. However, after being subjected to the reactivation process of the present invention, the conversion percentage rose to 77%. Such a reactivation not only returned the catalyst to its initial steady state, but resulted in a 10% improvement over the initial steady state value.

Accordingly, the present invention allows for a restoration to the initial steady state activity of the catalyst of a CO₂ laser. In addition, the present invention exceeds this steady state value, thus resulting in improved combination activity. The present invention accomplishes these desirable results simply and inexpensively.

We claim:

1. A method for reactivating a catalyst comprised of tin oxide and a noble metal, wherein said catalyst is used to combine a laser gas mixture of CO and O₂ to form CO₂ for use in a CO₂ laser having an operating temperature between about 23° C. and 110° C., said method comprising:

   exposing the catalyst to a stream of the laser gas mixture which is drawn from and returned to the laser;
   increasing the ambient temperature surrounding the catalyst between a temperature above the operating temperature of the laser and 400° C. for a time sufficient to reactivate the catalyst;
   reducing said ambient temperature to the operating temperature of the laser;
   whereby the combination activity of the catalyst is improved.

2. The method of claim 1 wherein the ambient temperature surrounding the catalyst is increased to 225° C.

3. The method of claim 2 wherein said temperature of 225° C. is maintained for one hour prior to reduction to the operating temperature of the laser.

4. The method of claim 1 wherein the laser gas mixture contains a rare oxygen isotope in the CO₂ molecule.

5. The method of claim 4 wherein the oxide of the catalyst contains the same rare oxygen isotope.

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