Air Flow and Pressure Drop Measurements Across Porous Oxides

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Abstract

This report summarizes the results of air flow tests across eight porous, open cell ceramic oxide samples. During ceramic specimen processing, the porosity was formed using the sacrificial template technique, with two different sizes of polystyrene beads used for the template. The samples were initially supplied with thicknesses ranging from 0.14 to 0.20 in. (0.35 to 0.50 cm) and nonuniform backside morphology (some areas dense, some porous). Samples were therefore ground to a thickness of 0.12 to 0.14 in. (0.30 to 0.35 cm) using dry 120 grit SiC paper. Pressure drop versus airflow is reported. Comparisons of samples with thickness variations are made, as are pressure drop estimates. As the density of the ceramic material increases the maximum corrected flow decreases rapidly. Future sample sets should be supplied with samples of similar thickness and having uniform surface morphology. This would allow a more consistent determination of air flow versus processing parameters and the resulting porosity size and distribution.

Introduction

Porous, open cell ceramics are used in a variety of engineering, industrial and aerospace applications. These include hot gas and liquid filters, catalyst and sensor supports, solid oxide fuel cells and chemical reactors (refs. 1 and 2). A comprehensive review of the techniques used to create porous ceramics is found in reference 3. There are three major processing routes (1) replica template, (2) sacrificial template, and (3) direct foaming.

The replica technique uses either a synthetic template or natural cellular template that is infused with a ceramic suspension. Upon drying/firing, the finished ceramic displays the same structure as the template. Templates include naturally occurring sponge, coral, and wood as well as synthetic polymer and carbon foams. In the second technique, a sacrificial phase is homogeneously dispersed within a continuous matrix of ceramic powder particles or a ceramic slip. After processing, pores remain in place of the sacrificial phase. Materials used for the template include polymer beads, salts, and carbon. Natural organic materials for templates include cellulose, sucrose, and wax. Barea et al. (ref. 4) used sacrificial starch to process porous mullite. Finally, liquids such as oils and freeze-dried water or camphene have also been used for templates. Araki and Halloran (ref. 5) used frozen camphene as the template when making porous alumina. The third technique introduces gaseous bubbles into a ceramic suspension. The resulting foam is then stabilized with surfactants to keep the foam structure intact during the firing phase.
Surfactants can be non-ionic, anionic, or cationic. Setting of the foam can be via thermosetting, sol-gel, or gel-casting.

The purpose of the present study is to measure air flow properties of oxide ceramics manufactured using the sacrificial template technique (polymer beads). The desire was to create an open pore structure leading to an air-permeable material. The processing parameters used for the highest-flowing ceramics will be identified. The results will be used by the ceramic manufacturer to guide further processing work.

**Procedure**

Table I contains initial dimensions and weights of the ceramic specimens. The porosity was formed using the sacrificial template technique. Polystyrene beads were used, with the bead diameter for the PB-1 samples (0.6 to 1 mm) twice that for the PB-2.5 samples (0.36 to 0.5 mm). The rest of the processing specifics remains proprietary to the originator including volume percent of beads used. Photographs of each specimen are contained in appendix A.

The flow apparatus is based on one described in an unpublished report by Sims (ref. 6) The test equipment (figs. 1 and 2) consists of an aluminum “sample assembly”, a stainless steel tube (49.5-in. long, 2.350-in. ID, 0.065-in. wall) to insure uniform static pressure at the exit, a clamshell retainer ring, two dial-type 0-35 psig pressure gauges (Pennwalt, Wallace & Tiernan Division), two type-K thermocouples, and a either a 0.32-to-3.2 standard cubic feet per minute (SCFM) or 0.80-to-8.0 SCFM flow meter (Omega FL1501-A or FL4 611-V, respectively). The latter was needed for the three highest flowing specimens: PB2.5-1, PB1-2, and PB2.5-4.

Filtered shop air is routed through the flow meter to the sample assembly. The assembly (figs. 3, 4, and 5) contains a flow orifice, the ceramic sample, a rubber gasket and a downstream flow orifice. Three flow orifices are available to use: 0.07-in., 0.125-in. or 0.250-in. inside diameter. If the samples are somewhat restrictive to air flow, only the largest orifice provides useful data—as was the case in this study.

The inside diameter of the downstream flow orifice is 1.747 in., providing an exit area of 2.397 in.$^2$. The outer diameters of the flow orifices are sealed in the sample assembly with O-rings. Initial calibration of each flow orifice was conducted with no ceramic sample in place. The results are displayed in figure 6. Inlet pressure and temperature are measured upstream of the sample as shown in figure 3; the exit pressure and temperature is measured at the far end of the steel tube. The flow is so low at these measurement locations that total and static pressure are identical.

In preparation for the test each sample is “potted” into an aluminum ring using room-temperature vulcanizing rubber (RTV). Figure 7 is a representative cross section of such a ring. The thicknesses of the available aluminum rings range from 0.11 to 0.13 in. Because of this and the fact that the backside morphology of the as-received samples showed various levels of porosity—or lack thereof (see Results section and appendix A), the backside of each sample was ground (dry) using 120 grit SiC paper. The resulting thicknesses are listed in table II.

Once each sample was loaded into the flow apparatus, the upstream sample pressure was increased in 1 psi increments up to a maximum of 15 psi over ambient by adjusting the flow meter control knob. Airflow in SCFM was manually logged in a laboratory notebook. Upstream sample inlet temperature, downstream outlet temperature, and outlet pressure were logged at test start; these numbers did not vary throughout the flow test. The contribution of pressure loss by the orifice to the sample ΔP is negligible.

Because all samples exhibited good flow throughout the 15 Δpsi pressure range, a corrected ΔP was calculated by taking into account the ambient air pressure and measured sample inlet temperature. The procedure (ref. 6) is as follows, where
\[ \text{corrected } \Delta P = \sigma \times \text{measured } \Delta P \]
\[ \sigma = \frac{\rho_{\text{test}}}{\rho_{\text{stp}}}, \text{ and} \]
\[ \rho_{\text{stp}} = 0.0765 \text{ lb/ft}^3 \]
\[ \rho_{\text{test}} = (\text{Ambient atmospheric pressure + measured } \Delta P) \times 144 \]
\[ \text{(Inlet temperature + 459.67)} \times 53.352 \]

Atmospheric pressure and \( \Delta P \) are measured in psi and inlet temperature is measured in °F. SCFM versus corrected \( \Delta P \) was input into a spreadsheet and graphed (appendix B).

**Results and Discussion**

The top two photographs in each appendix A figure are of the as-received material, with the side having the identification number designated with the front side. Note that in most cases the as-received backside had an inconsistent morphology with some areas exhibiting porosity and other areas seemingly very dense. The backside surfaces in general were also not flat. For this reason, the backside of each sample was ground (dry) using 120 grit SiC paper. The resulting thicknesses are listed in table II.

Appendix B contains the air flow data for each material. None of the specimens flowed well with the 0.125-in. flow orifice installed. Air did flow through all samples using the 0.250-in. flow orifice up to 15 delta psi. Specimens PB1-2 and PB1-3 became cracked (the latter more than the former) at some point during the flow testing. These cracks were not catastrophic, but likely allowed somewhat more air to pass during the testing. Note the anomaly in flow for sample PB1-3 in figures 8 and 9. The crack appears to have occurred as the corrected \( \Delta P \) was raised from 42 to 54 psid (\( \Delta P \) from 4 to 5 psid).

Figure 8 compares the sample results individually presented in appendix B. Figure 9 uses the data but normalizes the \( \Delta P \) with the inlet pressure and divides by the sample thickness. This allows for comparisons of samples with thickness variations and provides some data for pressure drop estimates as the thickness is changed. Since the inlet pressure increases with \( \Delta P \) for a sample test, both SCFM and mass flow will also increase. The characterization of sample \( \Delta P \) with flow can better be determined if corrected mass flow were used.

\[ \text{Corrected mass flow} = \frac{\text{mass flow} \times \sqrt{\text{Inlet temperature °F} + 459.67}}{\text{Inlet pressure/Pstd}} \times \sqrt{\text{Tstd}} \]

where Pstd = 14.696 psia and Tstd = 518.67 R.

The corrected mass flow is also divided by the flow area of the 0.250-in. orifice (0.04909 in.²).

Figure 9 shows that the mass flow of the samples becomes constant or choked as the \( \Delta P \) increases.

Figure 10 presents the pressure drop data for constant levels of corrected airflow as a function of the sample density. Figure 11 takes estimates of the maximum or choked corrected airflow from figure 9 and are plotted against sample density. As the density of the ceramic material increases, the maximum corrected flow decreases rapidly.

The ranking of the specimens in order from those allowing the highest flow rate at 15 psi \( \Delta P \) to the lowest is listed in table III. Also listed are the initial and final sample thicknesses, as well as the change in thickness due to grinding. Note that the sample that exhibited the most cracking (PB1-3) was the second lowest flowing specimen. The pressure build-up causes such cracks to form.

**Conclusion**

The PB2.5 processing route results in more-porous specimens that flow greater amounts of air than those processed via PB1. It should be noted that any future sample set should be supplied with each specimen in the batch having the same thickness. A uniform backside morphology with a consistent porosity is also highly desired. Eliminating these variables would allow a more consistent determination...
of air flow versus processing parameters/porosity size/distribution. The nonuniformity of the sample density/surface, combined with the small flow orifice, likely caused variations in the reported data. More consistent results could be achieved with a larger orifice.

### TABLE I.—INITIAL (THICK) SAMPLE WEIGHT/DIMENSION/DENSITY

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<th>Wgt, g</th>
<th>Wgt, lb</th>
<th>Thickness, in.</th>
<th>Dia., in.</th>
<th>Vol., in.³</th>
<th>Density, lb/in.³</th>
<th>Thickness, cm</th>
<th>Dia., cm</th>
<th>Vol., cc</th>
<th>Density, g/cc</th>
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### TABLE II.—AS-TESTED (GROUND THICKNESS) SAMPLE WEIGHT/DIMENSION/DENSITY

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<th>Thickness, in.</th>
<th>Dia., in.</th>
<th>Vol., in.³</th>
<th>Density, lb/in.³</th>
<th>Thickness, cm</th>
<th>Dia., cm</th>
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### TABLE III.—RANKING OF HIGHEST FLOW AT Δ15 psi

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<td>in.</td>
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<td>cm</td>
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Figure 1.—Complete flow apparatus.

Figure 2.—Pressure gauges and flowmeter.
Figure 3.—Sample assembly in place.

Figure 4.—Flow orifices: 0.07, 0.125, 0.250 in. (L to R). All disc ODs = 2.495 in.

Figure 5.—Sample insertion order: flow orifice, potted sample (RTV faces upstream), gasket, downstream orifice.
Figure 6.—Air flow rate (SCFM) versus delta pressure (psid) for each flow orifice alone (no sample loaded).

Figure 7.—Test set-up cross-section.
Figure 8.—Sample comparisons, pressure drop versus airflow.

Figure 9.—Sample corrected pressure drop over inlet pressure per sample thickness versus corrected mass flow per orifice flow area.
Figure 10.—Cross-plot of figure 9 showing corrected pressure drop per inch at constant corrected mass flows per area versus sample density.

Figure 11.—Estimated choked corrected mass flow per unit area from figure 9 versus sample density.
Appendix A

Macrographs
(1) Front side (with ID);
(2) Backside as-received;
(3) Backside ground

PB1-1
PB1-2
PB1-3
PB1-4

PB2.5-1
PB2.5-2
PB2.5-3
PB2.5-4
Figure 12.—Sample PB1-1 (front and back as-received; back as-sanded).
Figure 13.—Sample PB1-2 (front and back as-received; back as-sanded).
Figure 14.—Sample PB1-3 (front and back as-received; back as-sanded).
Figure 15.—Sample PB1-4 (front and back as-received; back as-sanded).
Figure 16.—Sample PB2.5-1 (front and back as-received; back as-sanded).
Figure 17.—Sample PB2.5-2 (front and back as-received; back as-sanded).
Figure 18.—Sample PB2.5-3 (front and back as-received; back as-sanded).
Figure 19.—Sample PB2.5-4 (front and back as-received; back as-sanded).
Appendix B

SCFM versus corrected ΔP

PB1-1
PB1-2
PB1-3
PB1-4

PB2.5-1
PB2.5-2
PB2.5-3
PB2.5-4
### Sample PB1-1 Results

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### Sample PB1-2 Results

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Sample PB1-3 Results

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**Sample PB2.5-1 Results**

![Sample PB2.5-1 Results](image)

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![Sample PB2.5-2 Results](image)
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5b. **GRANT NUMBER**
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   John H. Glenn Research Center at Lewis Field
   Cleveland, Ohio 44135-3191

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13. **SUPPLEMENTARY NOTES**
    Rodger A. Werner (retired), NASA Glenn Research Center. Responsible person, Dennis S. Fox, organization code RXD0, 216-433-3295.

14. **ABSTRACT**
   This report summarizes the results of air flow tests across eight porous, open cell ceramic oxide samples. During ceramic specimen processing, the porosity was formed using the sacrificial template technique, with two different sizes of polystyrene beads used for the template. The samples were initially supplied with thicknesses ranging from 0.14 to 0.20 in. (0.35 to 0.50 cm) and nonuniform backside morphology (some areas dense, some porous). Samples were therefore ground to a thickness of 0.12 to 0.14 in. (0.30 to 0.35 cm) using dry 120 grit SiC paper. Pressure drop versus air flow is reported. Comparisons of samples with thickness variations are made, as are pressure drop estimates. As the density of the ceramic material increases the maximum corrected flow decreases rapidly. Future sample sets should be supplied with samples of similar thickness and having uniform surface morphology. This would allow a more consistent determination of air flow versus processing parameters and the resulting porosity size and distribution.

15. **SUBJECT TERMS**
   Ceramics; Porosity; Air flow

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   **b. ABSTRACT** U
   **c. THIS PAGE** U

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