A Battery Health Monitoring Framework for Planetary Rovers

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Abstract—Batteries have seen an increased use in electric ground and air vehicles for commercial, military, and space applications as the primary energy source. An important aspect of using batteries in such contexts is battery health monitoring. Batteries must be carefully monitored such that the battery health can be determined, and end of discharge and end of usable life events may be accurately predicted. For planetary rovers, battery health estimation and prediction is critical to mission planning and decision-making. We develop a model-based approach utilizing computationally efficient and accurate electrochemistry models of batteries. An unscented Kalman filter yields state estimates, which are then used to predict the future behavior of the batteries and, specifically, end of discharge. The prediction algorithm accounts for possible future power demands on the rover batteries in order to provide meaningful results and an accurate representation of prediction uncertainty. The framework is demonstrated on a set of lithium-ion batteries powering a rover at NASA Ames Research Center using real experimental field test data.

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1. INTRODUCTION

Batteries have seen an increased use as the primary energy source in electric ground and air vehicles for commercial, military, and space applications. In such contexts, battery health monitoring (BHM) becomes a critical issue, since failure of the battery directly affects the availability of the asset. Batteries must be carefully monitored such that the battery health can be determined, and end of discharge and end of usable life events may be accurately predicted. In planetary rovers, a battery health management (BHM) system is especially important, due to its role in short- and long-term mission planning and decision-making [1, 2].

A BHM system consists of both battery state estimation and prediction components. While much research has been carried out on battery modeling and battery state estimation, the prediction component, i.e., prognostics, has only recently begun to receive attention [3–5]. Even more recently, battery prognostics has begun to be applied to electric vehicles. In [6, 7], end of discharge prediction was developed for an unmanned aerial vehicle platform. In [8], an approach for predicting the remaining driving range of a battery-powered car was developed. These approaches differ by the type of battery model used, and the underlying estimation and prediction algorithms. Empirical battery models are used in [6], and equivalent circuit models are used in [7–9]. Equivalent circuit models are popular because they are relatively simple and computationally efficient. For estimation, the particle filter algorithm is used in [2, 3, 6], whereas the unscented Kalman filter (UKF) is used in [7–9]. The UKF is generally preferred over the particle filter when the model assumptions required for the UKF can be met, due to the much higher computational requirements of the particle filter [10]. Prediction algorithms used are typically sample-based algorithms, such as Monte Carlo sampling. They differ due to how future inputs are characterized and how that uncertainty is incorporated into the predictions [11].

Our approach to a BHM system is different from previous approaches in several ways. First, we use a new electrochemistry-based battery model recently developed in [12]. Unlike similar models, it is composed only of ordinary differential equations, thus leading to a model that is as computationally efficient as equivalent circuit models. We use the UKF for state estimation, based on this new model. For prediction, we use an operator-centric approach that provides best-, average-, and worst-case usage predictions. The prognosis framework applied here is based on previous work presented in [9, 11].

We apply our BHM system to a planetary rover testbed developed at NASA Ames Research Center, in which we monitor a set of 24 lithium-ion batteries that are used to power the rover motors [13]. We provide experimental results validating the rover battery model, and demonstrate battery state estimation and end-of-discharge prediction using real field test data.

The paper is organized as follows. Section 2 describes the model-based prognostics framework. Section 3 summarizes the battery model. Sections 4 and 5 discuss the estimation and prediction approaches, respectively. Section 6 presents the experimental results. Section 7 concludes the paper.

2. MODEL-BASED PROGNOSTICS

In this section, we first formulate the prognostics problem, and present the uncertainty representation framework [11]. We then provide an architecture for model-based prognostics that will be applied for BHM.
For describing the probability distribution of a generic trajectory $A_k$, we introduce a set of *surrogate* random variables $\lambda_n = [\lambda^1_n, \lambda^n_2, \ldots]$ [11]. We describe a trajectory using $\lambda_n$ and instead define $p(\lambda_n)$, which in turn defines $p(A_k)$. These surrogate variables can be used to describe trajectories in myriad ways. For the parameter, input, and process noise trajectories we have the surrogate variables $\lambda_\theta$, $\lambda_u$, and $\lambda_v$. Additional discussion on the use of surrogate variables can be found in [11].

**Prognostics Architecture**

We adopt a model-based prognostics architecture [11, 15], in which there are two sequential problems, (i) the estimation problem, which requires determining a joint state-parameter estimate $p(x(k), \theta(k)|y(k_0:k_P))$ based on the history of observations up to time $k$, $y(k_0:k_P)$, and (ii) the prediction problem, which determines at $k_P$, using $p(x(k), \theta(k)|y(k_0:k_P))$, $p(\lambda_\theta)$, $p(\lambda_u)$, and $p(\lambda_v)$, a probability distribution $p(k_E(k_P)|y(k_0:k_P))$. The distribution for $k_E(k_P)$ can be trivially computed from $p(k_E(k_P)|y(k_0:k_P))$ by subtracting $k_P$ from $k_E(k_P)$.

The prognostics architecture is shown in Fig. 1 [11]. In discrete time $k$, the system is provided with inputs $u_k$ and provides measured outputs $y_k$. The estimation module uses this information, along with the system model, to compute an estimate $p(x(k), \theta(k)|y(k_0:k))$. The prediction module uses the joint state-parameter distribution and the system model, along with the distributions for the surrogate variables, $p(\lambda_\theta)$, $p(\lambda_u)$, and $p(\lambda_v)$, to compute the probability distribution $p(k_E(k_P)|y(k_0:k_P))$ at given prediction times $k_P$.

### 3. Battery Modeling

In order to apply model-based prognostics, we require a model. For this purpose, we employ an electrochemistry-based lithium ion battery model developed in [12]. In contrast to the empirical and equivalent circuit battery models used in previous approaches [3, 8, 9], the new model presented in [12] is based on the underlying electrochemical equations, but at a level of abstraction high enough that the model is still efficient with the improved fidelity. The model is represented as a set of ordinary differential equations and can be converted to a discrete-time representation and solved efficiently with a sample time of 1 s. We summarize here the model equations and refer the reader to [12] for additional details.

The battery model computes the voltage as a function of time given the current drawn from the battery. Several electrochemical processes contribute to the cell’s potential. The different potentials are summarized in Fig. 2 (adapted from [16]). The overall battery voltage $V(t)$ is the difference between the potential at the positive current collector, $\phi_p(0, t)$, and the negative current collector, $\phi_n(L, t)$, minus resistance losses at the current collectors (not shown in the

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**Figure 1. Prognostics architecture.**

![Diagram showing the workflow of a prognostics system with estimation and prediction modules.](image-url)
the positive electrode, so discharge, Li ions move out of the negative electrode and into ions in the lithium-intercalated host material [17]. During the number of electrons transferred in the reaction (positive), $i$ are given in Table 1 [12].

The potentials at the current collectors are described by several voltage terms. At the positive current collector is the equilibrium potential $V_{U,p}$. This voltage is then reduced by $V_{s,p}$ due to the solid-phase ohmic resistance, and $V_{e,p}$ the surface overpotential. The electrolyte ohmic resistance then causes another drop $V_{c}$. At the negative electrode, there is a drop $V_{n,n}$ due to the surface overpotential, and a drop $V_{e,n}$ due to the solid-phase resistance. The voltage drops again due to the equilibrium potential at the negative current collector $V_{U,n}$. We describe each of these voltages in the following subsections.

Equilibrium Potential

The equilibrium potential (also known as the open-circuit voltage) is captured using the Nernst equation:

$$V_{U,i} = U_0 + \frac{RT}{nF} \ln \left( \frac{1 - x_i}{x_i} \right) + V_{\text{INT},i}, \quad (5)$$

where $i$ refers to the electrode ($n$ for negative or $p$ for positive), $U_0$ is a reference potential, $R$ is the universal gas constant, $T$ is the electrode temperature (in K), $n$ is the number of electrons transferred in the reaction ($n = 1$ for Li-ion), $F$ is Faraday’s constant, $x$ is the mole fraction of lithium ions in the lithium-intercalated host material [17]. During discharge, Li ions move out of the negative electrode and into the positive electrode, so $x_n$ decreases while $x_p$ increases, and $V_{U,p} - V_{U,n}$ will decrease.

Here, $V_{\text{INT},i}$ is the activity correction term (0 in the ideal condition). We use the Redlich-Kister expansion:

$$V_{\text{INT},i} = \frac{1}{nF} \left( \sum_{k=0}^{N_i} A_{i,k} \left( (2x_i - 1)^{k+1} - \frac{2x_i k(1-x_i)}{2x_i - 1} \right) \right). \quad (6)$$

Here we use $N_p = 12$ and $N_n = 0$. The identified parameters are given in Table 1 [12].

We let $q_i$ represent the amount of Li ions in electrode $i$, as measured in Coulombs. The flow of Li ions is opposite to the flow of current, so $q_i$ changes in the same direction as $x_i$, and we have:

$$x_i = \frac{q_i}{q_i^{\text{max}}}, \quad (7)$$

where $q_i^{\text{max}} = q_p + q_n$ refers to the total amount of available Li ions. It follows that $x_p + x_n = 1$. When fully charged, $x_p = 0.4$ and $x_n = 0.6$. When fully discharged, $x_p = 1$ and $x_n = 0$.

Concentration Overpotential

When a battery is discharged, the reactions take place at the surface of the electrode and this results in a concentration gradient across the cell. The model accommodates this by splitting the total electrode volume into two individual control volumes (CVs), one for the bulk (with subscript $b$) and one for the surface (with subscript $s$).

For the volumes, the concentration of Li ions is computed as

$$c_{b,i} = \frac{q_{b,i}}{v_{b,i}}, \quad (8)$$
$$c_{s,i} = \frac{q_{s,i}}{v_{s,i}}, \quad (9)$$

where, for CV $v$ in electrode $i$, $c_{v,i}$ is the concentration and $v_{v,i}$ is the volume. We define $v_i = v_{b,i} + v_{s,i}$. Note now that the following relations hold:

$$q_p = q_{s,p} + q_{b,p}, \quad (10)$$
$$q_n = q_{s,n} + q_{b,n}, \quad (11)$$
$$q^{\text{max}} = q_{s,p} + q_{b,p} + q_{s,n} + q_{b,n}. \quad (12)$$

As the battery discharges, Li ions move from the surface layer at the negative electrode, through the bulk, and to the surface layer at the positive electrode, in order to match the flow of electrons. Li ions also move from the bulk CV to the surface due to the concentration gradient. The diffusion rate from the bulk to the surface is expressed as

$$q_{bs,i} = \frac{1}{D} (c_{b,i} - c_{s,i}), \quad (13)$$

where $D$ is the diffusion constant.

Table 1. Battery Model Parameters

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>$U_{0,p}$</td>
<td>4.03 V</td>
</tr>
<tr>
<td>$A_{p,0}$</td>
<td>-33642.23 J/mol</td>
</tr>
<tr>
<td>$A_{p,1}$</td>
<td>0.11 J/mol</td>
</tr>
<tr>
<td>$A_{p,2}$</td>
<td>23506.89 J/mol</td>
</tr>
<tr>
<td>$A_{p,3}$</td>
<td>-74679.26 J/mol</td>
</tr>
<tr>
<td>$A_{p,4}$</td>
<td>14359.34 J/mol</td>
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<tr>
<td>$A_{p,5}$</td>
<td>307849.79 J/mol</td>
</tr>
<tr>
<td>$A_{p,6}$</td>
<td>85053.13 J/mol</td>
</tr>
<tr>
<td>$A_{p,7}$</td>
<td>-1075148.06 J/mol</td>
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<td>$A_{p,8}$</td>
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<td>$A_{p,9}$</td>
<td>991586.68 J/mol</td>
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<td>$A_{p,10}$</td>
<td>283423.47 J/mol</td>
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<tr>
<td>$A_{p,11}$</td>
<td>-163020.34 J/mol</td>
</tr>
<tr>
<td>$A_{p,12}$</td>
<td>-470297.35 J/mol</td>
</tr>
<tr>
<td>$U_{0,n}$</td>
<td>0.01 V</td>
</tr>
<tr>
<td>$A_{n,0}$</td>
<td>86.19 J/mol</td>
</tr>
</tbody>
</table>

3 For Li$_x$CoO$_2$, $x$ must be at least 0.4; Li cannot be reversibly removed beyond that [17].
The $q$ variables are described as

$\dot{q}_{s,p} = i_{\text{app}} + \dot{q}_{b,s,p}$  \hspace{1cm} (14)

$\dot{q}_{b,p} = -\dot{q}_{b,s,p} + i_{\text{app}}$  \hspace{1cm} (15)

$\dot{q}_{b,n} = -\dot{q}_{b,s,n} + i_{\text{app}}$  \hspace{1cm} (16)

$\dot{q}_{s,n} = -i_{\text{app}} + \dot{q}_{b,s,n}$.  \hspace{1cm} (17)

where $i_{\text{app}}$ is the applied electric current. Initially, $c_{s,i} = c_{b,i}$ thus diffusion is zero.

The concentration overpotential is the difference in voltage between the two CVs due to the difference in concentration. Using the expression for equilibrium potential, we can compute the potential for the bulk volume and the potential for the surface layer; the difference between them is the concentration overpotential. So, we can explicitly account for the concentration overpotential simply by using as the expression for equilibrium potential, the equilibrium potential of the surface layer [12], i.e.,

$$V_{U,i} = U_0 + \frac{RT}{nF} \ln \left( \frac{1 - x_{s,i}}{x_{s,i}} \right) + V_{\text{INT},i},$$  \hspace{1cm} (18)

where $x_{s,i}$ is computed using

$$x_{s,i} = \frac{q_{s,i}}{q_{\text{max},s,i}},$$  \hspace{1cm} (19)

and

$$q_{\text{max},s,i} = q_{\text{max},v_{s,i},i}/i_{\text{i}}.$$  \hspace{1cm} (20)

**Ohmic Overpotential**

The voltage drops due to the solid-phase ohmic resistances, the electrolyte ohmic resistance, and the resistances at the current collectors are constant and lumped together into resistance $R_o$:

$$V_o = i_{\text{app}}R_o.$$  \hspace{1cm} (21)

**Surface Overpotential**

The overpotentials due to charge transfer resistance and SEI kinetics are described by the Butler-Volmer equation, which, for Li ion batteries, reduces to

$$V_{\eta,i} = \frac{RT}{F\alpha} \text{arcsinh} \left( \frac{J_i}{2J_{\text{d}0}} \right),$$  \hspace{1cm} (22)

where $J_i$ is the current density, and $J_{\text{d}0}$ is the exchange current density. The current densities are defined as

$$J_i = \frac{i}{S_i}$$  \hspace{1cm} (23)

$$J_{\text{d}0} = k_i(1 - x_{s,i})^\alpha(x_{s,i})^{1-\alpha},$$  \hspace{1cm} (24)

where $k_i$ is a lumped parameter of several constants including a rate coefficient, electrolyte concentration, and maximum ion concentration.

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**State of Charge**

The state of charge (SOC) of a battery is defined to be 1 when the battery is fully charged and 0 when the battery is fully discharged by convention. In this model, it is analogous to the mole fraction $x_{s,n}$, but scaled from 0 to 1. We distinguish here between nominal SOC and apparent SOC [12]. Nominal SOC is computed based on the combination of the bulk and surface layer CVs in the negative electrode, whereas apparent SOC is be computed based only on the surface layer. When a battery reaches the voltage cutoff, apparent SOC is 0, and nominal SOC is greater than 0 (how much greater depends on the difference between the diffusion rate and the current drawn). Once the concentration gradient settles out, the surface layer will be partially replenished and apparent SOC will rise while nominal SOC remains the same. Nominal ($n$) and apparent ($a$) SOC are defined using

$$SOC_n = \frac{q_n}{0.6q_{\text{max},n}}$$  \hspace{1cm} (25)

$$SOC_a = \frac{q_{s,a}}{0.6q_{\text{max},s,a}}.$$  \hspace{1cm} (26)

where $q_{\text{max},s,a} = q_{\text{max},v_{s,a},i}/i_{\text{i}}$.

**Battery Voltage**

Battery voltage can now be expressed as follows:

$$V = V_{U,p} - V_{U,n} - V_{\eta,p}' - V_{\eta,n}' - V_{o}'$$  \hspace{1cm} (27)

where

$$V_{\eta,p}' = \frac{(V_{n,p} - V_{n,p}')}{\tau_{\eta,p}}$$  \hspace{1cm} (28)

$$V_{\eta,n}' = \frac{(V_{n,n} - V_{n,n}')}{\tau_{\eta,n}}.$$  \hspace{1cm} (29)

and the $\tau$ parameters are empirical time constants (used since the voltages do not change instantaneously).

The model contains as states $x$, $q_{s,p}$, $q_{b,p}$, $q_{b,n}$, $q_{s,n}$, $V_o'$, $V_{\eta,p}'$, and $V_{\eta,n}'$. The single model output is $V$. Identified model parameters are given in Table 2 [12].

**Model Validation**

Measured and predicted voltage curves for a 0.044 A discharge (approximately equal to open-circuit voltage) are shown in Fig. 3, and for a 2 A discharge in Fig. 4. The data here was obtained from a laboratory setting using the rover batteries. Clearly, the model is very accurate. Measured and predicted voltage curves for battery data from a rover field test are shown in Fig. 5. The model is still quite accurate in this case. Note that the spike in predicted voltage occurring near 4000 s is due to a gap in the recorded data, causing an incorrect sample time to be given to the model.

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4. **ESTIMATION**

In order to accurately predict the future behavior of the battery, we must first estimate its state. For this purpose, we use the unscented Kalman filter (UKF) [18, 19]. Among nonlinear filters, the UKF generally has better accuracy than

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4Note that SOC of 1 corresponds to the point where $q_{s,a} = 0.6q_{\text{max},s,a}$, since the mole fraction at the positive electrode cannot go below 0.4, as described earlier.
Table 2. Battery Model Parameters

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>$q_{max}$</td>
<td>$1.32 \times 10^4$ C</td>
</tr>
<tr>
<td>$R$</td>
<td>8.314 J/mol/K</td>
</tr>
<tr>
<td>$T$</td>
<td>292 K</td>
</tr>
<tr>
<td>$F$</td>
<td>96487 C/mol</td>
</tr>
<tr>
<td>$n$</td>
<td>1</td>
</tr>
<tr>
<td>$D$</td>
<td>$7.0 \times 10^6$ mol s/C/m³</td>
</tr>
<tr>
<td>$\tau_o$</td>
<td>10 s</td>
</tr>
<tr>
<td>$\alpha$</td>
<td>0.5</td>
</tr>
<tr>
<td>$R_o$</td>
<td>0.085 Ω</td>
</tr>
</tbody>
</table>

| $S_p$  | $2 \times 10^{-4}$ m² |
| $k_p$  | $2 \times 10^4$ A/m²  |
| $v_{x,p}$ | $2 \times 10^{-6}$ m³ |
| $v_{b,p}$ | $2 \times 10^{-5}$ m³ |
| $\tau_{g,p}$ | 90 s               |

| $S_n$  | $2 \times 10^{-4}$ m² |
| $k_n$  | $2 \times 10^4$ A/m²  |
| $v_{x,n}$ | $2 \times 10^{-6}$ m³ |
| $v_{b,n}$ | $2 \times 10^{-5}$ m³ |
| $\tau_{g,n}$ | 90 s               |

Figure 3. Comparison of predicted and measured open-circuit voltage.

Figure 4. Comparison of predicted and measured voltage for a constant 2 A discharge.

random variable $y$ by some nonlinear function $y = g(x)$, and computes the mean $\bar{y}$ and covariance $P_{yy}$ using a set of deterministically selected weighted samples, called sigma points [18]. $\mathcal{X}^i$ denotes the $i$th sigma point from $x$ and $w^i$ denotes its weight. The sigma points are always chosen such that the mean and covariance match those of the original distribution, $x$ and $P_{xx}$. Each sigma point is passed through $g$ to obtain new sigma points $\mathcal{Y}^i$, i.e.,

$$\mathcal{Y}^i = g(\mathcal{X}^i) \quad (31)$$

with mean and covariance

$$\bar{y} = \sum_i w^i \mathcal{Y}^i \quad (32)$$

$$P_{yy} = \sum_i w^i (\mathcal{Y}^i - \bar{y})(\mathcal{Y}^i - \bar{y})^T \quad (33)$$

We use the symmetric unscented transform, which selects $2n_x + 1$ sigma points symmetrically about the mean [19]:

$$w^i = \begin{cases} 
\frac{\kappa}{(n_x + \kappa)}, & i = 0 \\
\frac{1}{2(n_x + \kappa)}, & i = 1, \ldots, 2n_x 
\end{cases} \quad (34)$$

$$\mathcal{X}^i = \begin{cases} 
x, & i = 0 \\
\bar{x} + \left(\sqrt{(n_x + \kappa)P_{xx}}\right)^i, & i = 1, \ldots, n_x \\
\bar{x} - \left(\sqrt{(n_x + \kappa)P_{xx}}\right)^i, & i = n_x + 1, \ldots, 2n_x 
\end{cases} \quad (35)$$

where $\left(\sqrt{(n_x + \kappa)P_{xx}}\right)^i$ refers to the $i$th column of the matrix square root of $(n_x + \kappa)P_{xx}$ (e.g., computed using the Cholesky decomposition). The number $\kappa$ is a free parameter that can be used to tune the higher order moments of the distribution, and if $x$ is assumed Gaussian, then selecting $\kappa = 3 - n_x$ is recommended [18].

The UKF assumes the general nonlinear form of the state and output equations, but is restricted to additive Gaussian noise. First, $n_x$ sigma points $\mathcal{X}_{k-1|k-1}$ are derived from the current mean $\bar{x}_{k-1|k-1}$ and covariance estimates $P_{k-1|k-1}$.
The prediction step is:
\[
\hat{x}_{k|k-1} = f(\hat{x}_{k-1|k-1}, u_{k-1}), \quad i = 1, \ldots, n_s
\]
(36)
\[
\hat{y}_{k|k-1} = h(\hat{x}_{k|k-1}), \quad i = 1, \ldots, n_s
\]
(37)
\[
\hat{x}_k = \sum_{i} w_i^f \hat{x}^i_{k|k-1}
\]
(38)
\[
\hat{y}_k = \sum_{i} w_i^f \hat{y}^i_{k|k-1}
\]
(39)
\[
P_{k|k-1} = Q + \sum_{i} w_i^f (\hat{x}^i_{k|k-1} - \hat{x}_{k|k-1})(\hat{x}^i_{k|k-1} - \hat{x}_{k|k-1})^T
\]
(40)
where \(Q\) is the process noise covariance matrix.

The update step is:
\[
P_{yy} = R + \sum_{i} w_i^f (\hat{y}^i_{k|k-1} - \hat{y}_{k|k-1})(\hat{y}^i_{k|k-1} - \hat{y}_{k|k-1})^T
\]
(41)
\[
P_{xy} = \sum_{i} w_i^f (\hat{x}^i_{k|k-1} - \hat{x}_{k|k-1})(\hat{y}^i_{k|k-1} - \hat{y}_{k|k-1})^T
\]
(42)
\[
K_k = P_{xy} P_{yy}^{-1}
\]
(43)
\[
\hat{x}_k = \hat{x}_{k|k-1} + K_k (y_k - \hat{y}_{k|k-1})
\]
(44)
\[
P_k = P_{k|k-1} - K_k P_{yy} K_k^T
\]
(45)
where \(R\) is the sensor noise covariance matrix.

For the battery, we have 7 states, resulting in 15 sigma points. An estimate of SOC can be computed from the estimates of the states. This can be computed using the unscented transform, in which the sigma points for the state are transformed into sigma points for SOC, using the equations for the state estimate from these transformed sigma points.

### 5. Prediction

Prediction is initiated at a given time \(k_P\) using the current joint state-parameter estimate, \(p(x(k_P), \theta(k_P)|y(k_0:k_P))\). The goal is to compute \(p(k_E(k_P)|y(k_0:k_P))\) using the state-parameter estimates and assumptions about uncertainty regarding the future parameter, input, and process noise values.

In this work, we assume all parameters are known, so \(\theta(k)\) is empty. Further, we assume that, because the model is so accurate, process noise is negligible. Of course, process noise is not zero, but in this case the uncertainty in the future inputs dominates significantly, so including process noise has virtually no effect on the prediction results. In the following, we describe the approach for the general case as originally developed in [11].

For one realization of each of the uncertain quantities at prediction time \(k_P\): the state \(x(k_P)\), the parameter trajectory \(\theta_{k_P}\), the input trajectory \(U_{k_P}\), and the process noise trajectory \(V_{k_P}\), the corresponding realization of \(k_E\) can be computed with the system model as shown in Algorithm 1 [11]. In Algorithm 1, the function \(\mathbb{P}\) simulates the system model until the threshold \(T_E\) evaluates to 1.

This algorithm requires computing first realizations of the state-parameter distribution, the parameter trajectory, the input trajectory, and the process noise trajectory. The distribution for the state comes from the UKF, and the distributions for the parameter, input, and process noise trajectories are defined indirectly by the set of surrogate variables. So, we are interested in computing the distribution for \(k_E\) from the distributions for \(p(x(k_P))\), \(p(\theta_{k_P})\), \(p(U_{k_P})\), and \(p(V_{k_P})\) (see Fig. 1).

In this paper, since only future input uncertainty is considered, we need to define only the surrogate variables for the future input trajectory. Here, future input trajectories may take many complex forms. Instead of representing these complexities directly, we consider instead an equivalent constant-loading distribution for the future inputs. That is, we assume that the future power usage for a battery will be constant with the value drawn from some distribution.

Prediction methods differ by how they sample from the given distributions, and how they call \(\mathbb{P}\). Different methods were investigated in [11], including Monte Carlo sampling, unscented transform sampling, and the inverse first-order-reliability method (FORM). In the case of the rover, the operator really only needs to know EOD predictions for best-, average-, and worst-case usage scenarios. For the state estimate, we use as samples the sigma points provided by the UKF. Each sample is simulated forward three times, once for each of these cases. From this we obtain best-, average-, and worst-case EOD predictions, each with some small variance (due to the state estimate variance).

If instead the operator desires a probability distribution, then an equivalent constant-loading distribution can be selected. We can determine the appropriate statistics for an equivalent constant-loading distribution by analyzing field experiments. Assuming a normal distribution, we can compute the average power for each field experiment, then compute the mean and variance of the average power. For a particular realization of the input, we can sample the average power from this distribution, and use that power for \(k \geq k_P\). In this paper we use a mean power draw of 3.3 W with a variance of 1.0.

In this case, we have one surrogate variable needed to describe the future input, which is used to define the constant power draw, as defined by the determined statistics. In order to sample efficiently from this distribution, we use the unscented transform method, as originally described in [9]. The unscented transform described in Section 4 is used to sample from the future input distribution, and in this case only 3 samples are needed since there is only a single
surrogate variable. This is much more efficient than the hundreds or thousands of samples required for Monte Carlo sampling. Prediction then proceeds in the normal way, with the \( P \) function being called for the sigma point combinations (each state sigma point is simulated for each future input sigma point). Using the unscented transform equations we can then determine the statistics of \( EOD \), namely, the mean and variance. (For additional statistical moments, extended versions of the unscented transform or Monte Carlo sampling are required. Alternatively, the inverse FORM method can be used to determine cumulative distribution function values at desired points.) Mathematical details of this approach are provided in [9, 11].

6. RESULTS

In this section, we present experimental results obtained from the planetary rover testbed at NASA Ames Research Center [13]. The rover motors are powered by 24 lithium-ion batteries, with two strings of 12 batteries in series, connected in parallel. So, each battery sees only about half of the total current required to operate the motors. Fully charged, each battery supplies 4.2 V. The batteries can be safely discharged to 2.5 V, but the drop from about 3.3 V down to 2.5 V happens relatively quickly, so to prevent overdischarge, the voltage cutoff used is \( V_{EOD} = 3.3 \) V.

The BHM system monitors the health state of each battery, and periodically provides \( EOD \) predictions, to let the rover operator know how much longer the rover can be used in the field. We demonstrate the BHM system as it performs on a field test of the rover. In this scenario, the rover was executing various maneuvers, e.g., moving straight, turning at various speeds, and stopping. The current drawn from the batteries during these maneuvers is shown in Fig. 6.

As described in Section 3, the battery model is very accurate. As a result, the battery state can be confidently estimated, as shown in Fig. 5. The SOC estimate for a single battery is shown in Fig. 7. We plot here the apparent SOC, which is computed with respect to the charge available at the electrode surface. Due to the development of concentration gradients, apparent SOC may increase when the diffusion rate from the bulk to the surface layer exceeds the discharge rate of the battery, as observed in the figure. On larger time scales, SOC is seen to decrease steadily, and rover operation ceases once SOC reaches around 5%.

The future power demands on the battery are unknown, especially in the driving scenario presented here, where an operator is deciding spontaneously where to drive the rover. If the future usage of the rover is known exactly, the BHM system should be able to obtain accurate predictions, since the model is accurate. Prediction results in this case are shown in Fig. 8, and the results are very accurate. There is some slight positive bias, which is due in part to sensor noise, because \( EOD \) is determined based on measured battery voltage, which, due to noise, will be earlier than actual \( EOD \) because the voltage threshold will be reached first due to noise. Overall, results are still very accurate, and have virtually no spread, since the estimation results had very little variance. Relative accuracy of the predictions averages to 98%, with a spread of only 0.16% relative standard deviation. Accuracy can only be improved with a better model, and, given the model we have, represents the best possible performance. This serves as a baseline for the scenarios where the future inputs are not known.

If the future battery usage is unknown, we provide the operator with prediction results for three separate cases, best-, average-, and worst-case power demands. Prediction results in this case are shown in Fig. 9. Clearly, predictions assuming average-case usage are not accurate, as the actual power usage is less than the average case up until 4000 s; after that the average power usage of the remainder of the scenario is greater than the average case. The average relative accuracy in this case is 83.3% based on the predictions corresponding to the average case. The predictions based on average usage still falls within the bounds set by the best- and worst-case usage. These results are with respect to continued usage, so
the operator knows how much driving time is remaining, i.e., the predictions do not include time spent while the rover is stopped.

Using the equivalent constant-loading distribution, we obtain the predictions given in Fig. 10. Here, because the actual mean power drawn in this scenario is larger than in the collection of field experiments, the median $\Delta k_E$ predictions are larger than the true values. Relative accuracy computed using these values is only 63%. Because of the large variety of scenarios in the field experiments, the spread is large, at 45% relative standard deviation. These numbers can be improved only if additional information is known about the future usage and that information is captured in the future input characterization, e.g., through the use of additional surrogate variables [11].

7. CONCLUSIONS

In this paper, we described a battery health monitoring framework for a rover. The prognostics framework was detailed, along with algorithms for state estimation and end-of-discharge prediction. The provided battery model was validated with both laboratory and field test data. Validation of the BHM system was demonstrated using data from an actual field test of the rover.

Future work will include end-of-life prediction for the batteries, which must track the degradation of the battery health over multiple usage cycles. Initial results suggest that the battery model described here is applicable in this case [12]. In addition, when there is some knowledge available as to the future operation of the rover, e.g., a set of waypoints, or a large set of past driving scenarios, future inputs can be characterized in a more complex way than presented here, which can potentially lead to more accurate results with much less spread [11]. However, this depends on what kind of information is desired by the operator, as in many cases, predictions for best-, average-, and worst-case usage are sufficient.

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REFERENCES


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