NASA's Corrosion Technology Laboratory at the Kennedy Space Center: Anticipating, Managing, and Preventing Corrosion

2014 INTERNATIONAL WORKSHOP ON ENVIRONMENT AND ALTERNATIVE ENERGY
"Increasing Space Mission Resiliency through Sustainability"
October 21-24, 2014
Kennedy Space Center, FL, USA

Luz Marina Calle, Ph.D.
NASA's Corrosion Technology Laboratory
Kennedy Space Center, FL, 32899, USA
Outline

- Introduction
- Corrosion
  - Definition, impact, and cost
  - Corrosion grand challenges; coatings pollution
  - Natural and Launch environments at NASA’s Kennedy Space Center (KSC)
  - Corrosion at KSC
  - Qualifying coatings for NASA’s launch facilities
  - Computerized corrosion management at KSC
  - NASA’s Corrosion Technology Laboratory Website
- Corrosion evaluation
  - Atmospheric exposure testing
  - Accelerated tests
  - Electrochemical measurements
  - Surface analysis
- Corrosion Engineering Projects
  - Environmentally driven projects
- Technology Development
  - Smart coatings
  - Environmentally friendly corrosion protective compounds (CPCs)
  - New accelerated corrosion test method
Introduction

• NASA has been dealing with corrosion since the inception of the Space Program in 1962 because it launches from the most naturally corrosive environment in North America
• The beachside atmospheric exposure test site was established in 1966 to test materials coatings, and maintenance procedures near the launch pads
• In 1981, corrosion conditions at the launch pads became even more severe due to solid rocket booster (SRB) exhaust products from the Space Shuttle
• In 1985, electrochemical corrosion testing begins
• In 2000, The Corrosion Technology Laboratory is created to achieve KSC’s goal of increased participation in research and development
• In 2000 a computerized corrosion data management system is implemented
• In 2004, the corrosion technology laboratory starts developing smart coatings based on microencapsulation technology specifically designed for corrosion control applications (U.S. Patents No. 7,790,225, 2010 and 20130017612).
Introduction

• In 2011, NASA-STD-5008B revision updates the standard and adds a paragraph on environmental stewardship:
  
a. Environmental, health, and safety impacts of processes and materials shall be taken into account when employing protective coating methods and techniques.
  
b. Alternative, environmentally friendly materials that do not contain hexavalent chromium, lead, cadmium, or hazardous air pollutants (HAPs), such as methyl ethyl ketone, toluene, and xylene, shall be considered when determining the correct coating method/technique for each protective coating application.

• 2014, NASA’s Space Technology Roadmap includes corrosion control technologies as one of the areas needed to lower the cost and improve the sustainability and efficiency of its ground operations in support of future launch activities.

• This presentation provides a chronological overview of NASA’s Corrosion Technology Laboratory at the Kennedy Space Center role in anticipating, managing, and preventing corrosion throughout the history of NASA’s Space Program.
What is Corrosion?

- Corrosion is the deterioration of a material due to reaction with its environment (M.G. Fontana). It literally means to "gnaw away"
- Degradation implies deterioration of the properties of the material.

KSC Crawler/Transporter
Structural Steel Corrosion
Impact of Corrosion
At US $2.2 (1.6 €) trillion, the annual direct cost of corrosion worldwide is over 3% of the world's GDP.*

Direct costs do not include the environmental damage, waste of resources, loss of production, or personal injury.

Gross Domestic Product - GDP The monetary value of all the finished goods and services produced within a country’s borders in a specific time period, though GDP is usually calculated on an annual basis.

*World Corrosion Organization 2010

(1 Trillion = 10^{12} = 1 billion)
Cost of Corrosion Control at KSC

Cost of corrosion control at KSC Launch Pads estimated as $1.6M/year

1 Estimate based on corrosion control cost of launch pads (39A and 39B) and the 3 Mobile Launch Platforms (MLPs) in 2001
Corrosion Grand Challenges*

- Development of cost-effective, environment-friendly, corrosion-resistant materials and coatings.
- High-fidelity modeling for the prediction of corrosion degradation in actual service environments.
- Accelerated corrosion testing under controlled laboratory conditions. Such testing would quantitatively correlate with the long-term behavior observed in service environments.
- Accurate forecasting of remaining service time until major repair, replacement, or overhaul becomes necessary. i.e., corrosion prognosis.

*Research Opportunities in Corrosion Science and Engineering, Committee on Research Opportunities in Corrosion Science and Engineering; National Research Council (2010)99
Coatings Pollution

Surface Preparation or Cleaning
- Pretreatment, solvents, conversion coatings

Coating Application Process
- Type of process: spray, roller coating, electrocoat
- Type of coating: high VOC, toxic pigments (hexavalent chromium)
- Waste: leftover paint, cleaning solvents/thinners, Air emissions (VOCs & HAPs), Spray booth filters, Soiled rags, Expired shelf-life inventory

Coating removal
- Waste (may be toxic)

VOC: Volatile Organic Compound; HAP: Hazardous Air Pollutants
Where Are We?

Orlando

Kennedy Space Center

Miami
The Kennedy Space Center in Florida, USA, is a special place where we launch rockets from a wild life refuge in one of the most corrosive areas in the world.
KSC Natural Environment
KSC Launch Environment
Positioned within 1,000 ft (305 m) of the Atlantic Ocean, KSC’s launch facilities are exposed to salty air that blows from the ocean, high ambient air temperatures, and an extensive amount of UV Light. The high temperature of the engine exhaust is up to nearly 5,000 F (2,760 °C). Close to 70 tons of hydrochloric acid (HCl) are generated by the combustion products of a rocket’s solid propellant.
In 1981 the Space Shuttle introduced acidic deposition products.
Natural Salt Fog Chamber
## Corrosion Rates of Carbon Steel

### Corrosion rates of carbon steel calibrating specimens at various locations*

<table>
<thead>
<tr>
<th>Location</th>
<th>Type Of Environment</th>
<th>µm/yr</th>
<th>Corrosion ratea mils/yr</th>
</tr>
</thead>
<tbody>
<tr>
<td>Esquimalt, Vancouver Island, BC, Canada</td>
<td>Rural marine</td>
<td>13</td>
<td>0.5</td>
</tr>
<tr>
<td>Pittsburgh, PA</td>
<td>Industrial</td>
<td>30</td>
<td>1.2</td>
</tr>
<tr>
<td>Cleveland, OH</td>
<td>Industrial</td>
<td>38</td>
<td>1.5</td>
</tr>
<tr>
<td>Limon Bay, Panama, CZ</td>
<td>Tropical marine</td>
<td>61</td>
<td>2.4</td>
</tr>
<tr>
<td>East Chicago, IL</td>
<td>Industrial</td>
<td>84</td>
<td>3.3</td>
</tr>
<tr>
<td>Brazos River, TX</td>
<td>Industrial marine</td>
<td>94</td>
<td>3.7</td>
</tr>
<tr>
<td>Daytona Beach, FL</td>
<td>Marine</td>
<td>295</td>
<td>11.6</td>
</tr>
<tr>
<td>Pont Reyes, CA</td>
<td>Marine</td>
<td>500</td>
<td>19.7</td>
</tr>
<tr>
<td>Kure Beach, NC (80 ft. from ocean)</td>
<td>Marine</td>
<td>533</td>
<td>21.0</td>
</tr>
<tr>
<td>Galeta Point Beach, Panama CZ</td>
<td>Marine</td>
<td>686</td>
<td>27.0</td>
</tr>
<tr>
<td><strong>Kennedy Space Center, FL (beach)</strong></td>
<td><strong>Marine</strong></td>
<td><strong>1070</strong></td>
<td><strong>42.0</strong></td>
</tr>
</tbody>
</table>

*Two-year average  

A mil is one thousandth of an inch
Changes In Corrosion Rate with Distance From The Ocean

Comparison of Average Corrosion Rate (Weight Loss) of UNS G10080 and Atmospheric Salt Content at Various Distances from the Seacoast

- Weight Loss, UNS G10080
- Salt Collection Rate (Funnel Samples)

Distance from Seacoast (Feet)

Milligrams, NaCl/m²/hr

Weight Loss, grams
Corrosion Comprehension: Operating in a Corrosive Environment, DoD Video, March 2012. DoD established an Environmental Severity Index (ESI) derived from 10 years of observations of steel and aluminum alloy samples (or “coupons”) left exposed to the elements at 130 military installations around the world.

**Determined experimentally using an Alternating Seawater Spray System at NASA’s Beachside Corrosion Test Site at KSC.**

*Corrosion Comprehension: Operating in a Corrosive Environment, DoD Video, March 2012. DoD established an Environmental Severity Index (ESI) derived from 10 years of observations of steel and aluminum alloy samples (or “coupons”) left exposed to the elements at 130 military installations around the world.

**Determined experimentally using an Alternating Seawater Spray System at NASA’s Beachside Corrosion Test Site at KSC.**
Examples of Launch Pad Corrosion

Enclosed / Inaccessible Areas

Dissimilar Metals

KSC Launch tower structural steel corrosion

Under the LC 39B Flame Trench
Corrosion Failures

Tubing split caused by Pitting

Hidden corrosion
Corrosion evaluation studies began at KSC in 1966 during the Gemini/Apollo Programs.
The KSC Beachside Corrosion Test Site was established at that time to conduct controlled corrosion studies for protective coatings.
Qualifying Coatings for NASA Launch Facilities

• Over the years, the Corrosion Technology Laboratory has developed proven methodologies to evaluate and test materials and coatings for use in NASA’s unique corrosive environments

• Based upon this knowledge, experience and expertise, NASA-STD-5008B, “Protective Coating of Carbon Steel, Stainless Steel, and Aluminum on Launch Structures, Facilities, and Ground Support Equipment” was developed to test and evaluate protective coatings to control corrosion of these assets.
NASA-STD-5008B


- Governs maintenance at John F. Kennedy Space Center and other NASA Centers.
- Establishes practices for the protective coating of ground support equipment and related facilities.
- Zones of Exposure are established to define coating system Requirements for specific environments.

Launch complex 39 zones of exposure
In order for a coating system to be used at NASA, it must be listed on the NASA-STD-5008B Approved Products List. Coating systems on this list are qualified according to the requirements of NASA-STD-5008B by the Corrosion Technology Laboratory.

Typical protocol requires laboratory adhesion tests, color measurements, gloss measurements, and corrosion evaluations on the coatings exposed at the NASA KSC Beachside Corrosion Test Site.
Atmospheric Exposure

Real world exposure at a site that mimics actual performance requirements

NASA Technical Standard for Protective Coatings (NASA-STD-5008B) requires 18 months of good performance for preliminary approval and continued good performance for 5 years for final approval of a coating system.
Coatings Qualification

Coating samples at 18 months (top) and 60 months (bottom) of exposure. This coating was approved for use.

Coating samples at 18 months (top) and 60 months (bottom) of exposure. This coating was not approved for use.
Since 2000, a computerized corrosion management program has been used to keep track of corrosion in more than 3,600 critical components and 7-8 million ft\(^2\) of surface area.

- Launch complex components
  - Launch towers and structures
  - Sound suppression water systems
  - Cryogenic fuel tanks and associated piping
  - Access towers
  - High-pressure gas tanks
  - Camera towers
  - Lighting protection
  - Mobile launch platforms
- Metallic structures outside launch area
Information stored in the database includes the location of the structure, the type of structure, the surface area of the structure, the substrate material, and the current condition of the coating system. Photos visually document condition ratings.
This series of photos tracks how corrosion of a water pipe at KSC’s launch complex 39B has progressed over four years.
The Corrosion Technology Laboratory at the NASA Kennedy Space Center is a network of capabilities—people, equipment, and facilities that provide technical innovations and engineering services in all areas of corrosion for NASA and external customers.

**The Corrosion Technology Laboratory:**

- Provides consulting and testing services for NASA and external customers
- Conducts applied research
- Develops new corrosion detection and control technologies
- Investigates, evaluates, and determines materials performance and degradation in different environments in support of NASA, other government organizations, industry, and educational institutions
- Participates in educational outreach activities

Corrosion Evaluation at KSC (current)

- Field Exposure
- Accelerated testing
- Electrochemical testing
- Technology development
KSC Beachside Corrosion Test Site

Launch Complex 39A
Launch Complex 39B

KSC Beachside Corrosion Test Site

• Full Seawater Immersion Exposure
• Tidal Exposure
• Seawater Spray/Splash (Splash Zone) Exposure

Atmospheric exposure racks

On-site laboratory

Atlantic Ocean
NASA’s Corrosion Technology Laboratory

The Corrosion Technology Laboratory at NASA’s Kennedy Space Center provides technical innovations and engineering services in all areas of corrosion/materials degradation for NASA and external customers.

Capabilities

• Beachside Atmospheric Exposure
• Full Seawater Immersion Exposure
• Tidal Exposure
• Seawater Spray/Splash (Splash Zone) Exposure
• Corrosion Engineering Services
• Accelerated Corrosion Testing
• Concrete Testing
• Cathodic Protection
• Coating Development
• Electrochemistry
• Surface Analysis
• Coating Application and Evaluation
• Website: http://corrosion.ksc.nasa.gov/
Environmentally Driven Projects

- Non-Chrome Systems Testing
- Hexavalent Chrome Free Coatings for Electronics
- Alternative to Nitric Acid Passivation
- Environmentally Preferable Coatings for Structural Steel (Launch Structures)
- Low VOC Topcoats for Thermal Spray Coatings
- Environmentally Friendly Corrosion Protective Compounds (CPCs)
- Smart and Multifunctional Corrosion Protective Coating Development
Alternative to Nitric Acid Passivation

Expected Results
• Provide the data necessary to verify that citric acid can be used as an environmentally preferable alternative to nitric acid for passivation of stainless steel

Benefits of Citric Acid
• Citric acid does not remove nickel, chromium, and other heavy metals from alloy surfaces
• Reduced risk associated with worker health and safety
• Reduced hazardous waste generation resulting in reduced waste disposal costs
• Reduced Nitrogen Oxide (NOx) emissions that are a greenhouse gas, contribute to acid rain and smog, and increased nitrogen loading (oxygen depletion) in bodies of water
Smart Coatings for Corrosion Control

The use of "smart coatings" for corrosion sensing and control relies on the changes that occur when a material degrades as a result of its interaction with a corrosive environment.

Such transformations can be used for detecting and repairing corrosion damage.

The Corrosion Technology Laboratory is developing a coating that can detect and repair corrosion at an early stage.

This coating is being developed using pH sensitive microcontainers that deliver their contents when corrosion starts to:

- Detect and indicate the corrosion location
- Deliver environmentally friendly corrosion inhibitors
- Deliver healing agents to repair mechanical coating damage.

http://upload.wikimedia.org/wikipedia/commons/1/12/Cape_Dwarf_Chameleon.jpg
**Microencapsulation-based Smart Coatings**

Corrosion indication, detection, and healing of mechanical damage can be achieved using microencapsulation technology.

**What are microcontainers?**

Particles or liquid drops coated in polymers. These microcontainers can carry any material that needs protection or controlled release.

**Why microencapsulate a material?**

- Incorporate active materials while maintaining coating integrity
- Achieve controlled-release
- Make active materials easier/safer to handle.
- Incorporate multiple component systems.
- Prevent undesired leaching
- Versatility

![Diagram of Core, Shell, Microcapsules, Microparticles](image)
Types of Feedback-Active Microcontainers for Corrosion Detection and Control

- Containers with an active ingredient-rich core and stimuli-responsive shell (microcapsules)
- Containers with an active ingredient incorporated into a stimuli-responsive matrix (microparticles)
- Containers with a porous ceramic core impregnated by inhibitor and enveloped by a stimuli-responsive polyelectrolyte (PE) shell*

*D. Grigoriev, D. Akcakayiran, M. Schenderlein, and D. Shcukin, Corrosion, 70
pH-triggered Release Microcapsules

Microcapsule containing pH indicator (inhibitor, self healing agents)

The shell of the microcapsule breaks down under basic pH (corrosion) conditions

pH indicator changes color and is released from the microcapsule when corrosion starts
Smart Coating Response to Corrosion and Mechanical Damage

Corrosion indicators

Corrosion inhibitors

Self-healing agents

Microcapsules are incorporated into smart coating

Mechanical damage causes capsule to rupture

Corrosion (basic pH) causes capsule to rupture

Indication of hidden corrosion by color change
Environmentally Friendly Corrosion Protective Coatings
And Corrosion Preventative Compounds (CPCs)

- Progressively stricter environmental regulations are driving the coating industry to abolish many corrosion protective coatings and corrosion preventative compounds (CPCs) that are not environmentally friendly.

- The objective of this project is to identify, test, and develop qualification criteria for environmentally friendly corrosion protective coatings and corrosion preventative compounds (CPCs) for flight hardware and ground support equipment.
Corrosion Preventive Compounds (CPCs)

Example: Ascent Wind Profiler, World’s Largest Doppler Radar Site
Located at the north side of the NASA KSC Shuttle Landing Facility
Areas of Dissimilar Metal and Crevice Corrosion
Corrosion Testing

CPCs offer a viable solution that can be easily incorporated into the current work constraints (ease of application and reapplication schedule)

Initial salt fog chamber results

Initial atmospheric exposure results

Example cable clamp test article

Corrosion around the SS bolt attached to the aluminum clamp after 2 weeks

Bare

Coated with CPC
Technology Development

- Long-term prediction of corrosion performance from accelerated tests.
- Coating development (Smart coatings for corrosion detection and control).
- Detection of hidden corrosion.
- Self-healing coatings.

1010 steel (UNS 10100) panels after prolonged exposure

Correlation?

1 mile from launch pad to test racks

~100 feet from high tide line to test racks
Timescale Correlation between Marine Atmospheric Exposure and Accelerated Corrosion Testing

Alternating Seawater Spray System with exposure panels, and modification for panels used for surface analysis (left). Wet candles exposed to KSC beachside atmospheric conditions and used to measure chloride concentration per month (right).
Summary

KSC is located in one of the most naturally corrosive areas in North America and has more than 4 decades of experience dealing with corrosion.

Acidic exhaust from SRBs exacerbate natural corrosive conditions at the launch pads.

NASA has encountered challenges in corrosion protection since the inception of the Space Program. Some of these challenges have been environmentally driven.

NASA’s Corrosion Technology Laboratory has been actively engaged in anticipating, managing, and preventing corrosion since its foundation in 2000.

NASA is engaged in projects aimed at identifying more environmentally friendly corrosion protection coatings and technologies.

Current technology development efforts target the development of smart coatings for corrosion detection and control and the development of a new accelerated corrosion test method that correlates with long-term corrosion test methods.

Website: [http://corrosion.ksc.nasa.gov/](http://corrosion.ksc.nasa.gov/)