AMINO ACID STABILITY IN THE EARLY OCEANS.  E. T. Parker¹, K. L.Britton², A. S. Burton³, D. P. Glavin⁴, J. P. Dworkin⁵, and J. L. Bada⁶, ¹School of Chemistry and Biochemistry, Georgia Institute of Technology (erictparker@gatech.edu), ²School of Mathematics, Science & Engineering, Southern Weste rn College, ³Astromaterials Research and Exploration Science Directorate, NASA Johnson Space Center, ⁴Solar System Exploration Division, NASA Goddard Space Flight Center, ⁵Geophysical Research Division Scripps Institution of Oceanography, University of California, San Diego.

Introduction: It is likely that a variety of amino acids existed in the early oceans of the Earth at the time of the origin and early evolution of life. “Primordial soup” [1]-, hydrothermal vent [2]-, and meteorite [3]-based processes could have contributed to such an inventory. Several “protein” amino acids were likely present, however, based on prebiotic synthesis experiments and carbonaceous meteorite studies, non-protein amino acids, which are rare on Earth today, were likely the most abundant. An important uncertainty is the length of time these amino acids could have persisted before their destruction by abiotic and biotic processes. Prior to life, amino acid concentrations in the oceans were likely regulated by circulation through hydrothermal vents. Today, the entire ocean circulates through vent systems every 10⁷ years. On the early Earth, this value was likely smaller due to higher heat flow and thus marine amino acid life-time would have been shorter. After life, amino acids in the oceans could have been assimilated by primitive organisms.

Original Experimental Setup: To better constrain primitive amino acid lifetimes, Scripps pier seawater and baked (500 °C) Scripps beach sand were placed in a glass container in spring, 1992. The container was spiked with protein and non-protein amino acids that could have been present on the early Earth. These included D/L-alanine (Ala), D/L-valine (Val), α-aminoisobutyric acid (α-AIB), and D/L-isovaline (Isoval). Amino acid abundances were measured several times over 706 days using chiral derivatization with reverse phase high performance liquid chromatography-fluorescence detection (HPLC-FD) [4]. The experiment was then spiked with glycine and monitored for a total of 2500 days. The abundances of D/L-Ala, glycine, and L-Val were found to rapidly decrease, apparently by microbes in the mixture. Yet minimal degradation of the non-protein amino acids, and surprisingly, D-Val, was observed, even after 2500 days.

Recent Sample Analysis: The La Jolla beach sand/seawater experiment was re-visited in summer, 2012 to determine if amino acids were still present. Sample analysis was performed by HPLC-FD and a triple quadrupole mass spectrometer (QqQ-MS), based on methods outlined elsewhere [5]. Findings indicated α-AIB, D/L-Isoval, and D-Val were present in detectable quantities. The disappearance of glycine indicates that microorganisms still live in the mixture and are capable of readily consuming protein amino acids. These findings demonstrate that the non-protein amino acids initially added to the samples over 20 years ago are minimally affected by biological degradation and thus have very long lifetimes under these conditions.

Conclusions: These findings suggest that non-protein amino acids derived from direct prebiotic synthesis, or meteorite in-fall were likely far more resistant to degradation by primitive marine microorganisms, than protein amino acids. Such robust non-protein molecules could have accumulated in the early oceans and reached a steady state-concentration dependent on the circulation of the oceans through hydrothermal systems, and input from various synthesis processes. We are presently trying to estimate this concentration.


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